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## Optical temperature sensing of rare-earth ion doped phosphors

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Accurate and reliable temperature measurement of many special inaccessible objects is a challenging task. Optical temperature sensing is a promising method to achieve it. The current status of optical thermometry of rare-earth ion doped phosphors is reviewed in detail. Based on the mechanisms of optical temperature sensing of different phosphors, temperature dependent luminescence spectra, the fluorescence intensity ratio technique in the data fitting process, and errors of the energy difference between thermally coupled levels, we describe the recent developments in the use of optical thermometry materials. The most important results obtained in each case are summarized, and the main challenges that we need to overcome are discussed. Research in the field of phosphor sensors has shown that they have significant advantages compared to conventional sensors in terms of their properties like greater sensitivity, freedom from electromagnetic interference, long path monitoring, and independence of compatibility with electronic devices.

Received 22nd August 2015  
Accepted 28th September 2015

DOI: 10.1039/c5ra16986k

www.rsc.org/advances

### 1. Introduction

Temperature is a key parameter to induce some changes of cellular events, physical and chemical properties of functional nanomaterials, functional degradation of microcircuits, and so on. Contact temperature measurement, just like

a thermometer, is an ideal method to study some phenomena that change with temperature. However, especially at the sub-micron scale, for temperature fluctuations of microcircuits and intracellular liquids, the conventional thermometer is not able to make measurements.<sup>1–3</sup> Brites and Jaque reported some examples of luminescent and non-luminescent thermometers working at the nanometric scale.<sup>4,5</sup> Even so, it is difficult to measure directly the temperatures of inaccessible objects, such as high-voltage power stations, coal mines, and volcanic and corrosive circumstances.<sup>6–9</sup> Therefore, the ability to glean the temperature of the above objects could have valuable repercussions leading to novel insight about their properties induced by a temperature change.

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Optical temperature sensing, in contrast to other thermometers, is a promising method to achieve contactless measurement and large-scale imaging. Optical temperature sensing is based on monitoring the emission intensity change of luminescent materials induced by temperature when they interact with physical systems. Phosphors with an intense emission intensity are used as temperature detectors to convert the measured spectrum to temperature. It is easy to calibrate and calculate the temperature change range only through the main parameters of luminescence including intensity, effective bandwidth, spectrum shape, spectral shift and lifetime.

Recently, CdSe and CdTe semiconducting quantum dots were used as optical temperature nanoprobe in the low temperature range, since the peak position of the emission wavelength changes as a function of temperature.<sup>10,11</sup> Organic compounds, such as Rhodamine B, the Ru-phen complex, DPTB dissolved in MOE, and so on, were used for dye-based intensity luminescence nanothermometry, due to their spectral properties depending on many factors, such as the solvent, concentration, pH and temperature.<sup>5,11,12</sup> These quantum dot nanocrystals and organic compounds are easily oxidized at high temperature, and are only

available in the low temperature range. To overcome this, rare-earth ion doped oxide crystals, glass, core-shell heterojunction nanoparticles, and transparent glass ceramics containing fluoride nanocrystals were fabricated and used as optical temperature sensors. The present review aims to describe the latest progress of optical thermometry based on the luminescence of rare-earth ion doped phosphors. It contains fundamental principles, different rare-earth ion doped phosphors for optical thermometry, a conclusion and the main challenges.

## 2. Fundamental principles of optical temperature sensing

Phosphor thermometry is a non-contact technique that uses luminescence signals to measure temperature remotely. As shown in Fig. 1, phosphors are composed of a host and some luminescent ions, and will emit visible, infrared, or ultraviolet radiation upon excitation from an external light source. The intensity, wavelength, and lifetime of the emission bands are used to determine the temperature of a surface. Thus, the choice of luminescent ions is important to measure the temperature change around phosphors. Among the metal ions



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used as luminescence centers, the trivalent rare-earth ions are especially attractive, due to abundant energy levels located at a wide wavelength range from ultraviolet to infrared.<sup>13</sup> Possessing real intermediate energy levels, trivalent rare-earth ions can give out desired emissions *via* various energies, due to abundant energy levels of 4f configurations. Inheriting their native intra-configurational transitions, trivalent rare-earth ion activated luminescent materials have received consistent attention due to large Stokes/anti-Stokes shifts, long luminescence lifetimes, and sharp band emissions.<sup>14</sup> Additionally, some trivalent rare-earth ions own a couple of adjacent levels with a very small energy gap ( $\Delta E$ ) of about 100–2000  $\text{cm}^{-1}$ , such as  $\text{Er}^{3+}$ :  $^2\text{H}_{11/2}$  and  $^4\text{S}_{3/2}$ ;  $\text{Tm}^{3+}$ :  $^3\text{F}_{2,3}$  and  $^3\text{H}_4$ ;  $\text{Ho}^{3+}$ :  $^5\text{S}_2$  and  $^5\text{F}_4$ ;  $\text{Nd}^{3+}$ :  $^4\text{F}_{5/2}$  and  $^4\text{F}_{3/2}$ ;  $\text{Dy}^{3+}$ :  $^4\text{I}_{15/2}$  and  $^4\text{F}_{9/2}$ ;  $\text{Eu}^{3+}$ :  $^5\text{D}_1$  and  $^5\text{D}_0$ , and so on. As shown in Fig. 1, in the photoluminescence process, these adjacent energy levels, the upper level and lower level, can be thermally populated and depopulated through changing the environmental temperature around the phosphors. The two adjacent energy levels were called thermally coupled energy levels (TCL). The luminescence intensity ratio between  $I_U$  and  $I_L$  will change regularly with the temperature increase. A function relation between the luminescence intensity ratio and temperature can be determined through fitting some data points at different temperatures.

The luminescence intensity of an emission band can be expressed as:

$$I_{ij} = hvA_{ij}N_i \quad (1)$$

where  $hv$  is the transition energy per photon from an  $i$  state to a  $j$  state,  $A_{ij}$  is the spontaneous radiative emission probability, and  $N_i$  is the state population of the  $i$  state.<sup>15</sup>

The term  $v$  is expressed as:

$$v = \frac{c}{\lambda} \quad (2)$$

where  $\lambda$  is the mean wavelength of the transition,  $h$  is Planck's constant, and  $c$  is the speed of light.

In conclusion,  $I_{ij}$  can be expressed as:

$$I_{ij} = \frac{hc}{\lambda} A_{ij} N_i \quad (3)$$

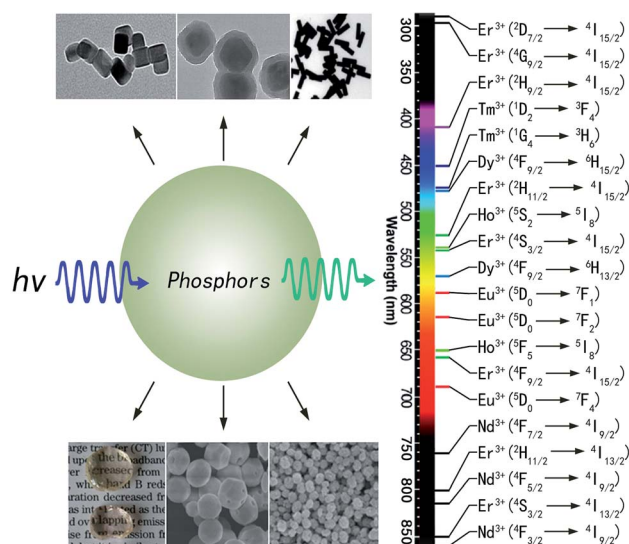


Fig. 2 Different types and luminescence transitions of rare-earth ion doped phosphors used for optical thermometry. Data are summarized according to ref. 21–80.

The ratio of the luminescence from each thermally coupled level of active ions is modified as:

$$R = \frac{I_U}{I_L} \quad (4)$$

where  $I_U$  and  $I_L$  are the fluorescence intensities generated by the radiative transitions from the upper and lower thermally coupled levels to the ground level.

Thus, the term  $R$  can be given as the following:

$$R = \frac{\lambda_L A_{Uj} N_U}{\lambda_U A_{Lj} N_L} \quad (5)$$

where  $\lambda_L$  is the mean wavelength of the  $L \rightarrow j$  transition,  $\lambda_U$  is the mean wavelength of the  $U \rightarrow j$  transition,  $A_{Uj}$  is the spontaneous radiative emission probability of the  $U \rightarrow j$  transition, and  $A_{Lj}$  is the spontaneous radiative emission probability of the  $L \rightarrow j$  transition.  $N_U$  and  $N_L$  are the state populations of the upper and lower levels.

The terms  $A_{Uj}$ ,  $A_{Lj}$ ,  $N_U$ , and  $N_L$  are dependent on the temperature. The population process of the  $N_U$  and  $N_L$  levels

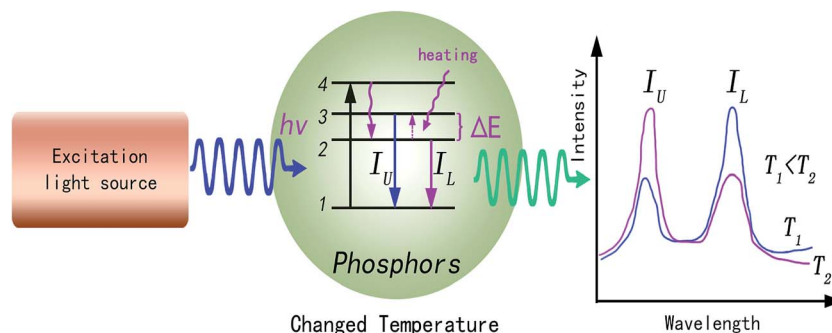


Fig. 1 Schematic illustration of the basic mechanism of the optical thermometry process in rare-earth ion doped phosphors.

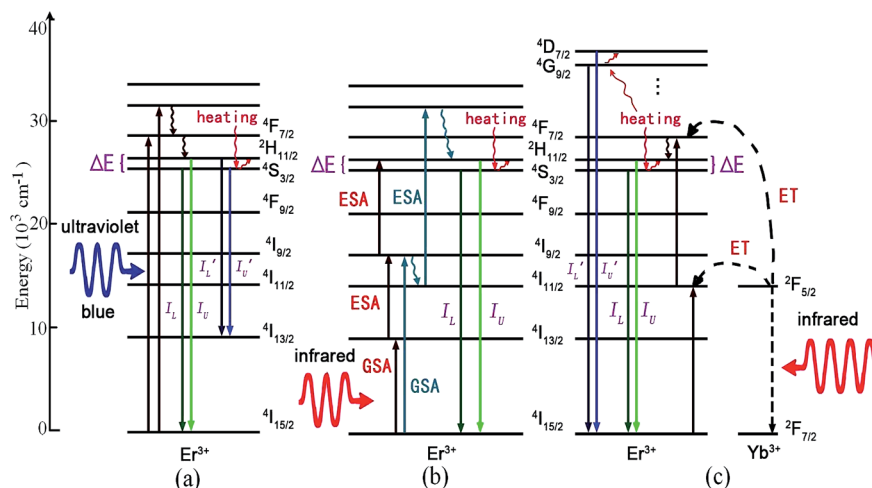


Fig. 3 The mechanism of optical thermal sensing through (a)  $\text{Er}^{3+}$  down-conversion under ultraviolet (blue) excitation, (b)  $\text{Er}^{3+}$  up-conversion under infrared excitation, and (c)  $\text{Yb}^{3+}$ – $\text{Er}^{3+}$  energy transfer under infrared excitation.

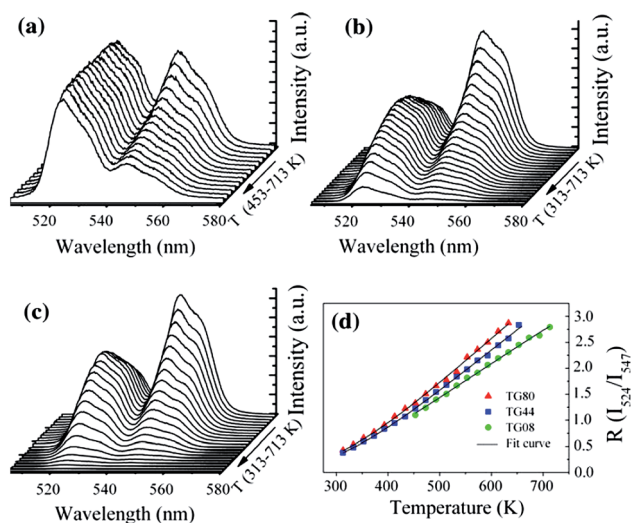


Fig. 4 Temperature dependent  $\text{Er}^{3+}$  green emission spectra of 0.5 mol%  $\text{Er}^{3+}$  doped (a)  $\text{Te}_{80}\text{Zn}_{9.5}\text{Na}_{10}$  (TG08) glass, (b)  $\text{Te}_{40}\text{Ge}_{40}\text{Zn}_{9.5}\text{Na}_{10}$  (TG44) glass, (c)  $\text{Ge}_{80}\text{Zn}_{9.5}\text{Na}_{10}$  (TG80) glass, and (d) temperature dependent  $R$  values of the two green emissions of  $\text{Er}^{3+}$ . Reprinted from ref. 21 with permission of Springer.

obeys the Boltzmann distributing law.<sup>16</sup> Thus,  $R$  can be expressed as:

$$R = A \exp \left[ \frac{-\Delta E_f}{KT} \right] + B \quad (6)$$

where  $A$  is a fitting constant that depends on the experimental system and intrinsic spectroscopic parameters;  $\Delta E_f$  is the fitting energy difference between thermally coupled levels;  $K$  is the Boltzmann constant ( $298 \text{ K}$ ,  $KT \approx 210 \text{ cm}^{-1}$ );  $T$  is the absolute temperature; and  $B$  is an offset.<sup>17,18</sup> The term  $B$  includes the overlapping of fluorescence peaks originating from the two individual thermally coupled energy levels, and stray light

originating from other energy levels or from the excitation.  $B$  is expressed as:

$$B = \frac{m}{n} \quad (7)$$

where  $n$  defines the fraction of the total fluorescence intensity, and  $m$  defines the fraction of the total intensity.

The sensitivity is a key parameter to value the possibility of practical applications. The sensitivity of optical thermometry is the rate of change of  $R$  in response to the variation of temperature.<sup>17,19,20</sup> To allow comparison between the sensitivities obtained from different thermally coupled levels, the relative sensitivity  $S_R$  and the absolute sensitivity  $S_A$  are defined as:

$$S_R = \frac{dR}{dT} = R \frac{\Delta E_f}{KT^2} \quad (8)$$

and

$$S_A = \frac{1}{R} \frac{dR}{dT} = \frac{\Delta E_f}{KT^2} \quad (9)$$

where  $\Delta E_f$  is the fitting energy difference between thermally coupled levels.

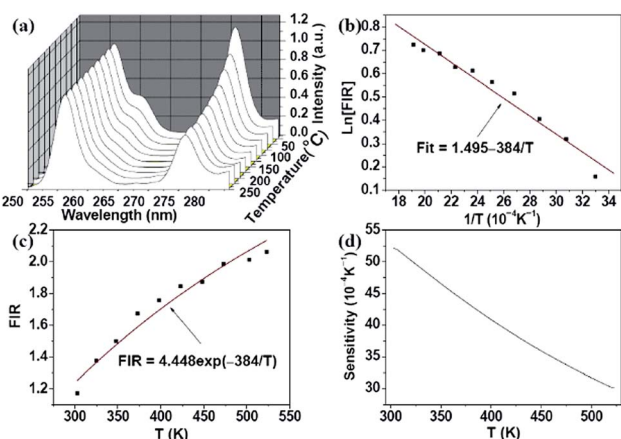
The terms  $S_R$  and  $S_A$  are dependent on  $\Delta E_f$ . If  $\Delta E_f$  agrees well with the experimental energy difference  $\Delta E_m$ , the values of  $S_R$  and  $S_A$  are accurate. The error  $\delta$  between  $\Delta E_f$  and  $\Delta E_m$  is expressed as:

$$\delta = \frac{|\Delta E_f - \Delta E_m|}{\Delta E_m} \quad (10)$$

where  $\Delta E_m$  is calculated from the two analyzed peaks in the spectrum. The error  $\delta$  is a key parameter to determine whether the value of  $\Delta E_f$  agrees well with the experimental value of  $\Delta E_m$ . A large value of  $\delta$  means that the energy transfer between thermally coupled levels and other levels is not neglected, and the population of thermally coupled levels at high temperature is induced by the routes of the Boltzmann distribution and energy transfer.<sup>4,14,17</sup>

**Table 1** The maximum sensitivity values of  $\text{Er}^{3+}$  doped phosphors by the fluorescence intensity ratio technique are presented, and the involved transitions from the TCL, excitation wavelength, emission wavelength, and temperature range are included

| Rare-earth ions                                  | Host                                                                         | $\lambda_{\text{ex}}$ (nm) | $\lambda_{\text{em}}$ (nm) | Transitions                                                         | Temperature range (K) | $S_R$ (maximum)                                              | $S_A$         | Ref. |
|--------------------------------------------------|------------------------------------------------------------------------------|----------------------------|----------------------------|---------------------------------------------------------------------|-----------------------|--------------------------------------------------------------|---------------|------|
| $\text{Er}^{3+}$                                 | Tellurite glass                                                              | 379                        | 524, 547                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 313–713               | $0.0085 \text{ K}^{-1}$ (596 K)                              | $1108.3/T^2$  | 21   |
| $\text{Er}^{3+}$                                 | In–Zn–Sr–Ba glass                                                            | 406                        | 523, 545                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 125–425               | $0.0028 \text{ K}^{-1}$ (425 K)                              | $1167.6/T^2$  | 22   |
| $\text{Er}^{3+}$                                 | $\alpha\text{-NaYF}_4$ glass ceramics                                        | 488                        | 525, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 300–720               | $0.0024 \text{ K}^{-1}$ (540 K)                              | $1051.51/T^2$ | 23   |
| $\text{Er}^{3+}$                                 | PKAZLF glass                                                                 | 488                        | 524, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 298–773               | $0.0079 \text{ K}^{-1}$ (630 K)                              | $1238.8/T^2$  | 24   |
| $\text{Er}^{3+}$                                 | Te–Pb–Al glass                                                               | 488                        | 530, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 100–573               | $0.0079 \text{ K}^{-1}$ (541 K)                              | $1149.43/T^2$ | 25   |
| $\text{Er}^{3+}$                                 | Sr–Ba–Nb–B glass                                                             | 532                        | 800, 850                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{13/2}$ | 300–700               | $0.0017 \text{ K}^{-1}$ (600 K)                              | $1255.14/T^2$ | 26   |
| $\text{Er}^{3+}$                                 | Te–Pb–Al glass                                                               | 800                        | 530, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 300–550               | $0.0054 \text{ K}^{-1}$ (540 K)                              | $1085.57/T^2$ | 27   |
| $\text{Er}^{3+}$                                 | ZBLALiP glass                                                                | 805                        | 522, 546                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 150–850               | $0.0023 \text{ K}^{-1}$ (495 K)                              | $981/T^2$     | 28   |
| $\text{Er}^{3+}$                                 | Ba(Zr,Ca)TiO <sub>3</sub> ceramics                                           | 980                        | 525, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 200–443               | $0.0044 \text{ K}^{-1}$ (443 K)                              | $1135.5/T^2$  | 29   |
| $\text{Er}^{3+}$                                 | Si–B–Ba–Na glass                                                             | 978                        | 534, 549                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 296–673               | $0.0023 \text{ K}^{-1}$ (296 K)                              | $335/T^2$     | 30   |
| $\text{Er}^{3+}$                                 | BaTiO <sub>3</sub> nanocrystals                                              | 980                        | 526, 547                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 322–466               | $1200 \text{ K}^{-1}$                                        | $940/T^2$     | 31   |
| $\text{Er}^{3+}$                                 | PLZT ceramics                                                                | 980                        | 534, 565                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 310–883               | $0.004 \text{ }^\circ\text{C}^{-1}$ (610 $^\circ\text{C}$ )  | $1096.92/T^2$ | 32   |
| $\text{Er}^{3+}, \text{Mo}^{6+}$                 | YbAG                                                                         | 976                        | 522, 546                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 295–973               | $0.0048 \text{ K}^{-1}$ (467 K)                              | $900.8/T^2$   | 33   |
| $\text{Er}^{3+}$                                 | ZnO                                                                          | 978                        | 536, 553                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 353–973               | $0.0062 \text{ }^\circ\text{C}^{-1}$ (170 $^\circ\text{C}$ ) | $880.11/T^2$  | 34   |
| $\text{Er}^{3+}$                                 | Chalcogenide glass                                                           | 1540                       | 530, 555                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 293–493               | $0.0102 \text{ }^\circ\text{C}^{-1}$                         | $1135.23/T^2$ | 35   |
| $\text{Er}^{3+}$                                 | $\text{Na}_{0.82}\text{Ca}_{0.08}\text{Er}_{0.16}\text{Y}_{0.853}\text{F}_4$ | 1540                       | 523, 542                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 5–300                 | $0.0022 \text{ K}^{-1}$ (338 K)                              | $958.28/T^2$  | 36   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\alpha\text{-NaYF}_4$                                                       | 980                        | 525, 545                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 298–318               | $0.0030 \text{ K}^{-1}$ (515 K)                              | $1028/T^2$    | 37   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\beta\text{-NaLuF}_4$                                                       | 980                        | 256, 276                   | $^4\text{D}_{7/2}, ^4\text{G}_{9/2} \rightarrow ^4\text{I}_{15/2}$  | 303–523               | $0.0052 \text{ K}^{-1}$ (303 K)                              | $384/T^2$     | 38   |
| $\text{Er}^{3+}, \text{Yb}^{3+}, \text{Eu}^{3+}$ | $\text{Y}_2\text{O}_3$                                                       | 980                        | 523, 551                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 303–593               | $0.0103 \text{ K}^{-1}$ (593 K)                              | $1062.50/T^2$ | 39   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{Y}_2\text{O}_3$                                                       | 978                        | 539, 564                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 150–300               | $0.0528 \text{ K}^{-1}$ (150 K)                              | $1173.39/T^2$ | 40   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{Y}_2\text{O}_3$                                                       | 980                        | 528, 556                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 93–613                | $0.0044 \text{ K}^{-1}$ (427 K)                              | $886.08/T^2$  | 41   |
| $\text{Er}^{3+}, \text{Yb}^{3+}, \text{Eu}^{3+}$ | $\text{Y}_2\text{O}_3$                                                       | 980                        | 523, 551                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 301–403               | $0.0008 \text{ K}^{-1}$ (327 K)                              | $682.6/T^2$   | 42   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{Al}_2\text{O}_3$                                                      | 978                        | 523, 545                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 295–973               | $0.0051 \text{ K}^{-1}$ (495 K)                              | $964.1/T^2$   | 43   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{BaMoO}_4$                                                             | 980                        | 531, 552                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 303–523               | $0.0206 \text{ K}^{-1}$ (463 K)                              | $873.38/T^2$  | 44   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{Y}_2\text{SiO}_5$ pulsed                                              | 975                        | 530, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 300–600               | $0.0070 \text{ K}^{-1}$ (600 K)                              | $1226/T^2$    | 45   |
| $\text{Er}^{3+}, \text{Yb}^{3+}, \text{Nd}^{3+}$ | $\text{Y}_2\text{SiO}_5$                                                     | 808                        | 528, 556                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 298–753               | $0.00095 \text{ K}^{-1}$ (439 K)                             | $710.87/T^2$  | 46   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{SrWO}_4$                                                              | 980                        | 525, 547                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 300–518               | $0.01498 \text{ K}^{-1}$ (403 K)                             | $866.17/T^2$  | 47   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{YVO}_4$                                                               | 980                        | 524, 554                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 300–485               | $0.01169 \text{ K}^{-1}$ (380 K)                             | $774.1/T^2$   | 48   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{CaWO}_4$                                                              | 980                        | 384, 408                   | $^4\text{G}_{11/2}, ^2\text{H}_{9/2} \rightarrow ^4\text{I}_{15/2}$ | 303–873               | $0.0073 \text{ K}^{-1}$ (873 K)                              | $2109.31/T^2$ | 51   |
| $\text{Er}^{3+}, \text{Mo}^{6+}$                 | $\text{Yb}_2\text{Ti}_2\text{O}_7$                                           | 976                        | 532, 546                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 290–610               | $0.0074 \text{ K}^{-1}$ (340 K)                              | $679.2/T^2$   | 52   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{LiNbO}_3$                                                             | 980                        | 530, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 285–453               | $0.0075 \text{ K}^{-1}$ (310 K)                              | $1250/T^2$    | 54   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3$ ceramics                        | 980                        | 525, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 173–553               | $0.0035 \text{ K}^{-1}$ (493 K)                              | $1017.12/T^2$ | 49   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{NaBiTiO}_3$ ceramics                                                  | 980                        | 525, 550                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 163–613               | $0.0031 \text{ K}^{-1}$ (400 K)                              | $827.26/T^2$  | 50   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{Bi}_7\text{Ti}_4\text{NbO}_{21}$                                      | 980                        | 547, 670                   | $^4\text{S}_{3/2}, ^4\text{F}_{9/2} \rightarrow ^4\text{I}_{15/2}$  | 153–553               | $0.0044 \text{ K}^{-1}$                                      |               | 53   |
| $\text{Er}^{3+}, \text{Yb}^{3+}$                 | $\text{TeO}_2\text{-WO}_3$ glass                                             | 980                        | 527, 551                   | $^2\text{H}_{11/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ | 300–690               | $0.0029 \text{ K}^{-1}$ (690 K)                              | $976.75/T^2$  | 55   |



**Fig. 5** Temperature dependent (a) spectrum of  $\text{Er}^{3+}$  ions from  $^4\text{D}_{7/2}$  and  $^4\text{G}_{9/2}$  levels; (b) log plot of the FIR; (c) FIR relative to the temperature; and (d) sensor sensitivity under 980 nm excitation. Reprinted from ref. 38 with permission of Optical Society of America.

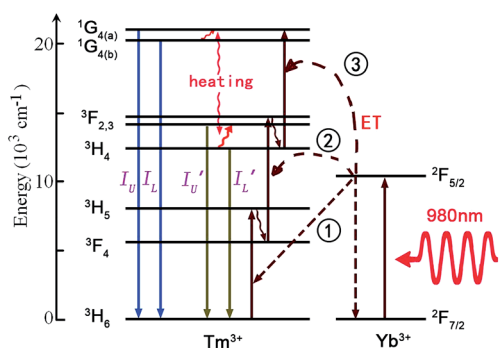
### 3. Rare-earth ion doped phosphors for optical thermometry

Phosphors are composed of inorganic oxides and ceramic materials which means that phosphors are resistant to oxidation in high temperature environments and are non-reactive with harsh chemicals. To obtain excellent optical temperature sensors, trivalent lanthanide ion doped phosphors were synthesized widely.<sup>21–80</sup> At present, trivalent lanthanide ions, such as  $\text{Er}^{3+}$ ,  $\text{Tm}^{3+}$ ,  $\text{Ho}^{3+}$ ,  $\text{Nd}^{3+}$ ,  $\text{Dy}^{3+}$ , and  $\text{Eu}^{3+}$ , have been used as luminescence centers (or activators) in the process of optical thermometry, as shown in Fig. 2. Host materials used for optical thermometry involve nanorods,<sup>60</sup> spherical and tetragonal nanoparticles,<sup>59</sup> core-shell particles,<sup>61</sup> hollow nanoparticles, glass ceramics containing fluoride nanocrystals,<sup>23,68,71</sup> fibers,<sup>17</sup> and oxide bulks.<sup>34,39–54,58</sup> Phosphors are excited with a light source, and the emitted luminescence can be in the ultraviolet, visible, or even in the infrared region.

For optical temperature sensors, the TCL should satisfy some conditions that depend strongly on the host matrix into

Table 2 Host dependent  $\Delta E_f$ ,  $\Delta E_m$ , and  $\delta$  in  $\text{Er}^{3+}$  doped and  $\text{Yb}^{3+}$ – $\text{Er}^{3+}$  co-doped phosphors

| Samples                                                                                   | $\Delta E_f$ ( $\text{cm}^{-1}$ ) | $\Delta E_m$ ( $\text{cm}^{-1}$ ) | $\delta$ (%) | Ref. |
|-------------------------------------------------------------------------------------------|-----------------------------------|-----------------------------------|--------------|------|
| $\text{Er}^{3+}$ doped tellurite glass                                                    | 781                               | 802                               | 2.62         | 21   |
| $\text{Er}^{3+}$ doped In–Zn–Sr–Ba glass                                                  | 861                               | 771.8                             | 11.55        | 22   |
| $\text{Er}^{3+}$ doped $\alpha$ - $\text{NaYF}_4$ glass ceramics                          | 741                               | 865.8                             | 14.41        | 23   |
| $\text{Er}^{3+}$ doped PKAZLF glass                                                       | 873                               | 902.2                             | 3.24         | 24   |
| $\text{Er}^{3+}$ doped Te–Pb–Al glass                                                     | 810                               | 686                               | 18.1         | 25   |
| $\text{Er}^{3+}$ doped Sr–Ba–Nb–B glass                                                   | 872.3                             | 748                               | 16.62        | 26   |
| $\text{Er}^{3+}$ doped Te–Pb–Al glass                                                     | 765                               | 810                               | 5.56         | 27   |
| $\text{Er}^{3+}$ doped ZBLALiP glass                                                      | 681                               | 700                               | 2.72         | 28   |
| $\text{Er}^{3+}$ doped Ba(Zr,Ca)TiO <sub>3</sub> ceramics                                 | 789                               | 850                               | 7.18         | 29   |
| $\text{Er}^{3+}$ doped Si–B–Ba–Na glass                                                   | 236                               | 511.7                             | 53.89        | 30   |
| $\text{Er}^{3+}$ doped BaTiO <sub>3</sub> nanocrystals                                    | 662.4                             | 729.9                             | 9.25         | 31   |
| $\text{Er}^{3+}$ doped PLZT ceramics                                                      | 773                               | 1027.5                            | 24.77        | 32   |
| $\text{Er}^{3+}$ , $\text{Mo}^{6+}$ co-doped YbAG                                         | 634.8                             | 842.1                             | 24.62        | 33   |
| $\text{Er}^{3+}$ doped ZnO                                                                | 611                               | 573.5                             | 6.54         | 34   |
| $\text{Er}^{3+}$ doped chalcogenide glass                                                 | 800                               | 849.9                             | 5.87         | 35   |
| $\text{Na}_{0.82}\text{Ca}_{0.08}\text{Er}_{0.16}\text{Y}_{0.853}\text{F}_4$              | 675.3                             | 670.27                            | 0.75         | 36   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\alpha$ - $\text{NaYF}_4$                   | 724.4                             | 766.1                             | 5.44         | 37   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\beta$ - $\text{NaLuF}_4$                   | 270.6                             | 2830.6                            | 90.44        | 38   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ , $\text{Eu}^{3+}$ tri-doped $\text{Y}_2\text{O}_3$   | 738.54                            | 800                               | 7.68         | 39   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{Y}_2\text{O}_3$                       | 826.9                             | 810.6                             | 2.01         | 40   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{Y}_2\text{O}_3$                       | 615.7                             | 800                               | 23.0         | 41   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ , $\text{Eu}^{3+}$ tri-doped $\text{Y}_2\text{O}_3$   | 474.9                             | 971                               | 51.09        | 42   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{Al}_2\text{O}_3$                      | 679.4                             | 771.8                             | 11.97        | 43   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{BaMoO}_4$                             | 607                               | 716                               | 15.22        | 44   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{Y}_2\text{SiO}_5$                     | 781                               | 686.1                             | 13.83        | 45   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ , $\text{Nd}^{3+}$ tri-doped $\text{Y}_2\text{SiO}_5$ | 498                               | 601.7                             | 17.23        | 46   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{SrWO}_4$                              | 602                               | 766.1                             | 21.42        | 47   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{YVO}_4$                               | 538                               | 693                               | 22.36        | 48   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3$ | 706.8                             | 800                               | 11.65        | 49   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{NaBiTiO}_3$                           | 574.8                             | 800                               | 28.15        | 50   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{CaWO}_4$                              | 1455                              | 1530                              | 4.90         | 51   |
| $\text{Er}^{3+}$ , $\text{Mo}^{6+}$ co-doped $\text{Yb}_2\text{Ti}_2\text{O}_7$           | 478.6                             | 482                               | 0.71         | 52   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{Bi}_7\text{Ti}_4\text{NbO}_{21}$      | 775                               | 820.7                             | 5.57         | 53   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{LiNbO}_3$                             | 860                               | 686.2                             | 25.33        | 54   |
| $\text{Er}^{3+}$ , $\text{Yb}^{3+}$ co-doped $\text{TeO}_2$ – $\text{WO}_3$ glass         | 678.94                            | 826.5                             | 17.85        | 55   |

Fig. 6 The mechanism of optical thermal sensing through  $\text{Yb}^{3+}$ – $\text{Tm}^{3+}$  energy transfer under 980 nm infrared excitation.

which the rare-earth ions are doped. The factors affecting the TCL of the rare-earth ions are as follows:

1. The separation between the TCL should be more than  $200\text{ cm}^{-1}$  to avoid strong overlapping of the two fluorescence wavelengths, and be less than  $2000\text{ cm}^{-1}$  to avoid too small a population in the upper level for the temperature range of interest.

2. In order to achieve a sufficient fluorescence intensity from the upper level transition, the radiative transitions from the upper level should dominate its non-radiative transitions.

TCL proven by experiment are listed as follows:  $\text{Er}^{3+}$ :  $^2\text{H}_{11/2}/^4\text{S}_{3/2}$ , and  $^4\text{D}_{7/2}/^4\text{G}_{9/2}$ ;  $\text{Tm}^{3+}$ :  $^3\text{F}_{2,3}/^3\text{H}_4$ , and  $^1\text{G}_{4(a)}/^1\text{G}_{4(b)}$ ;  $\text{Ho}^{3+}$ :  $^5\text{S}_2/^5\text{F}_4$ ,  $^5\text{F}_{2,3}/^3\text{K}_8$ , and  $^5\text{G}_6/^5\text{F}_1$ ;  $\text{Nd}^{3+}$ :  $^4\text{F}_{5/2}/^4\text{F}_{3/2}$ ,  $^4\text{F}_{7/2}/^4\text{F}_{3/2}$ , and  $^4\text{F}_{7/2}/^4\text{F}_{5/2}$ ;  $\text{Dy}^{3+}$ :  $^4\text{I}_{5/2}$  and  $^4\text{F}_{9/2}$ ; and  $\text{Eu}^{3+}$ :  $^5\text{D}_1/^5\text{D}_0$ . The TCL are populated by the up-conversion and down-conversion processes of the above trivalent rare-earth ions. Optical thermometry has been achieved by analyzing the temperature dependent luminescence properties originating from the transitions from the TCL to the other levels.

### 3.1 Optical thermometry based on $\text{Er}^{3+}$ doped phosphors

Trivalent erbium  $\text{Er}^{3+}$  has a  $4f^{11}$  electronic configuration and dense energy levels located at a wide wavelength range from ultraviolet to infrared, which is suitable to absorb and emit ultraviolet, visible and infrared luminescence. The  $\text{Er}^{3+}$  ion has two couples of adjacent TCL,  $^2\text{H}_{11/2}$  and  $^4\text{S}_{3/2}$ , and  $^4\text{D}_{7/2}$  and  $^4\text{G}_{9/2}$ , whose relative emission intensities are strongly temperature dependent. Among the different rare-earth ions capable of single-center phosphors,  $\text{Er}^{3+}$  is probably the most used one,

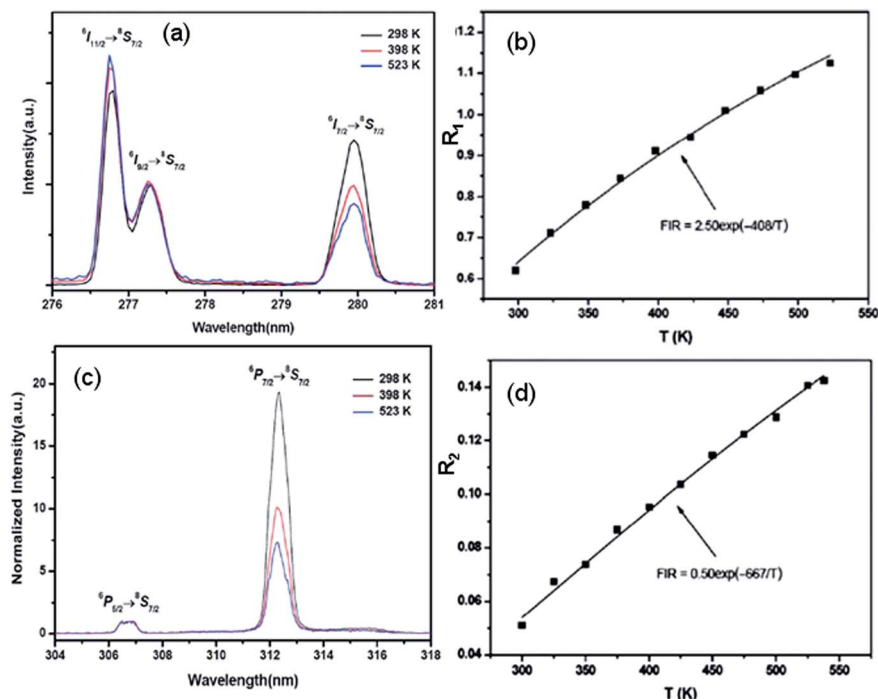


Fig. 7 Temperature dependent (a) ultraviolet spectra in the range of 276–281 nm, and (b)  $R_1$  of the emissions from the transitions of  $^6I_{9/2}/^8S_{7/2}$  and  $^6I_{7/2}/^8S_{7/2}$  of NaLuF<sub>4</sub>:Yb<sup>3+</sup>, Tm<sup>3+</sup>, Gd<sup>3+</sup> microcrystals. Temperature dependent (c) ultraviolet spectra in the range of 304–318 nm, and (d)  $R_2$  of the emissions from the transitions of  $^6P_{5/2}/^8S_{7/2}$  and  $^6P_{7/2}/^8S_{7/2}$  of NaLuF<sub>4</sub>:Yb<sup>3+</sup>, Tm<sup>3+</sup>, Gd<sup>3+</sup> microcrystals. Reproduced from ref. 60 with permission of the Royal Society of Chemistry.

due to its very intense two green emission bands originating from the  $^2H_{11/2}$  and  $^4S_{3/2}$  TCL. As shown in Fig. 3, the mechanism of optical thermometry based on Er<sup>3+</sup> doped phosphors can be summarized as follows:

(1) Down-conversion of Er<sup>3+</sup> ions under ultraviolet (or blue) excitation:<sup>21–26</sup> under ultraviolet (or blue) excitation, Er<sup>3+</sup> ions are excited directly to the higher excited states by ground state absorption (GSA), as shown in Fig. 3(a). Some ions in the excited states relax to the next lower energy levels  $^2H_{11/2}$  and  $^4S_{3/2}$  through the process of non-radiative relaxation, making the  $^2H_{11/2}$  and  $^4S_{3/2}$  levels populated. Er<sup>3+</sup> ions in the  $^2H_{11/2}$  and  $^4S_{3/2}$  levels radiatively relax to the  $^4I_{15/2}$  level, giving two green emissions with the intensity of  $I_U$  and  $I_L$ . Er<sup>3+</sup> ions in the  $^2H_{11/2}$  and  $^4S_{3/2}$  levels radiatively relax to the  $^4I_{13/2}$  level, giving two green emission bands with the intensity of  $I'_U$  and  $I'_L$ . The

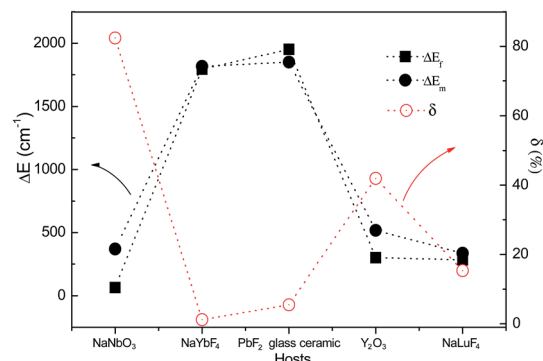


Fig. 8 Host dependent  $\Delta E_f$  and  $\Delta E_m$  between the TCL of Tm<sup>3+</sup> ions, and the corresponding  $\delta$  values.

**Table 3** The maximum sensitivity values of Tm<sup>3+</sup> doped phosphors by the fluorescence intensity ratio technique are presented, and the involved transitions from the TCL, excitation wavelength, emission wavelength, and temperature range are included

| Rare-earth ions                                        | Host                                 | $\lambda_{ex}$ (nm) | $\lambda_{em}$ (nm) | Transitions                                        | Temperature range (K) | $S_R$ (maximum)                 | $S_A$                  | Ref. |
|--------------------------------------------------------|--------------------------------------|---------------------|---------------------|----------------------------------------------------|-----------------------|---------------------------------|------------------------|------|
| Tm <sup>3+</sup> , Yb <sup>3+</sup>                    | NaNbO <sub>3</sub>                   | 976                 | 480, 486            | $^1G_4, ^3F_{2,3}, ^3H_4 \rightarrow ^3H_6$        | 293–353               | 0.0008 °C <sup>−1</sup> (25 °C) | 93.53/T <sup>2</sup>   | 56   |
| Tm <sup>3+</sup> , Yb <sup>3+</sup>                    | PbF <sub>2</sub> glass ceramics      | 980                 | 700, 800            | $^3F_{2,3}, ^3H_4 \rightarrow ^3H_6$               | 293–703               | 0.0006 K <sup>−1</sup> (1360 K) | 2829.5/T <sup>2</sup>  | 57   |
| Tm <sup>3+</sup> , Yb <sup>3+</sup>                    | Y <sub>2</sub> O <sub>3</sub>        | 976                 | 476, 488            | $^1G_{4(a)}, ^1G_{4(b)} \rightarrow ^3H_6$         | 303–753               | 0.0035 K <sup>−1</sup> (303 K)  | 452.51/T <sup>2</sup>  | 58   |
| Tm <sup>3+</sup> , Yb <sup>3+</sup>                    | Y <sub>2</sub> O <sub>3</sub>        | 978                 | 454, 815            | $^1D_2 \rightarrow ^3F_4, ^3H_4 \rightarrow ^3H_6$ | 10–300                | 0.078 K <sup>−1</sup> (270 K)   | 566.91/T <sup>2</sup>  | 59   |
| Tm <sup>3+</sup> , Yb <sup>3+</sup> , Gd <sup>3+</sup> | NaLuF <sub>4</sub>                   | 980                 | 307, 312.4          | $^6P_{5/2}, ^6P_{7/2} \rightarrow ^8S_{7/2}$       | 298–523               | 0.0004 K <sup>−1</sup> (333 K)  | 667/T <sup>2</sup>     | 60   |
|                                                        |                                      | 980                 | 277.3, 279.9        | $^6I_{9/2}, ^6I_{7/2} \rightarrow ^8S_{7/2}$       | 298–523               | 0.0029 K <sup>−1</sup> (298 K)  | 408/T <sup>−2</sup>    | 60   |
| Tm <sup>3+</sup>                                       | NaYbF <sub>4</sub> @SiO <sub>2</sub> | 980                 | 697, 798            | $^3F_{2,3}, ^3H_4 \rightarrow ^3H_6$               | 100–700               | 0.00054 K <sup>−1</sup> (100 K) | 2677.39/T <sup>2</sup> | 61   |

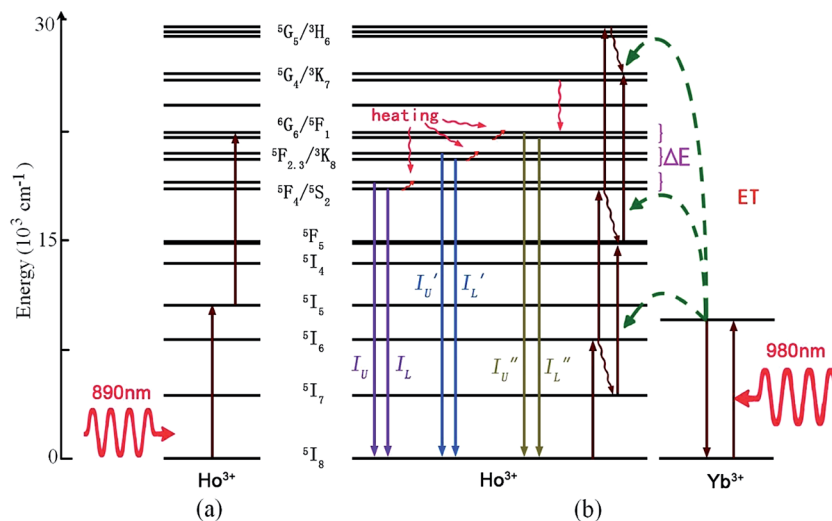


Fig. 9 The mechanism of optical thermal sensing through (a) up-conversion of the  $\text{Ho}^{3+}$  ion under 890 nm infrared excitation, and (b)  $\text{Yb}^{3+}$  –  $\text{Ho}^{3+}$  energy transfer under 980 nm infrared excitation.

luminescence intensity ratios,  $I_U/I_L$  and  $I_U'/I_L'$ , change with the temperature of the phosphors. These temperature dependent fluorescence intensity ratios are used as the precise evaluation scale of optical temperature sensing.

Recently, based on down-conversion luminescence emissions centered at 524 nm ( $^2\text{H}_{11/2} \rightarrow ^4\text{I}_{15/2}$ ) and 547 nm ( $^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ ) of  $\text{Er}^{3+}$  ions, as shown in Fig. 4, optical temperature sensing in the range from 313 K to 713 K was studied by Sui *et al.* through analyzing the temperature dependent term  $R$  in  $\text{Er}^{3+}$  doped Te–Ge–Zn–Na glass under 379 nm excitation.<sup>21</sup> A maximum sensitivity of  $0.0085 \text{ K}^{-1}$  at 596 K was obtained in the

$\text{Te}_{40}\text{Ge}_{40}\text{Zn}_{9.5}\text{Na}_{10}$  glass. Under 406 nm excitation, the intensity ratio of green emissions at 523 nm and 545 nm was studied by González *et al.* in  $\text{Er}^{3+}$  doped fluorindate glass with a maximum sensitivity of  $0.0028 \text{ K}^{-1}$  for 425 K.<sup>22</sup> Under 488 nm excitation, the thermalized levels  $^4\text{S}_{3/2}$  and  $^2\text{H}_{11/2}$  were studied in  $\text{Er}^{3+}$  doped  $\text{NaYF}_4$  nanocrystalline glass ceramic with a maximum sensitivity of  $66 \times 10^{-4} \text{ K}^{-1}$  for 570 K.<sup>23</sup> Similarly, the maximum sensitivity of  $79 \times 10^{-4} \text{ K}^{-1}$  for 630 K was obtained in  $\text{Er}^{3+}$  doped zinc fluorophosphate glass, and  $79 \times 10^{-4} \text{ K}^{-1}$  for 541 K was obtained in  $\text{Er}^{3+}$  doped fluorotellurite glass.<sup>24,25</sup> Under 532 nm excitation, González *et al.*

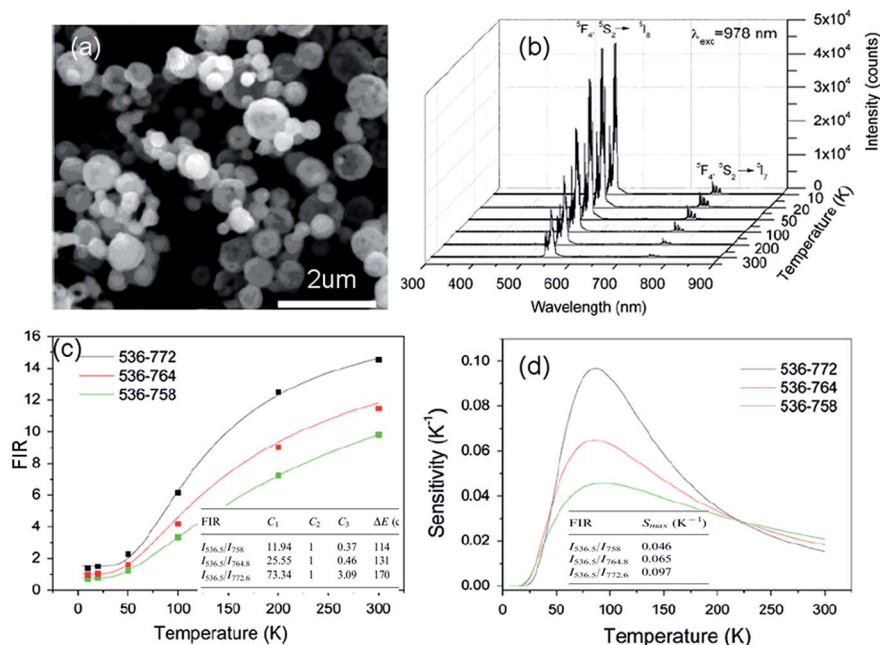
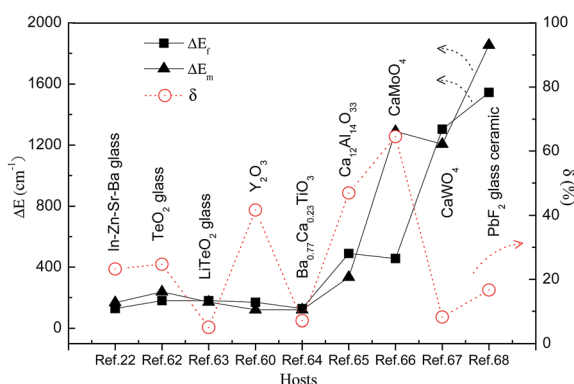


Fig. 10 (a) SEM image, (b) up-conversion spectra under 978 nm excitation, (c) FIR of emission at 536 nm relative to 758 nm, 764 nm and 772 nm emissions, and (d) the temperature dependence of sensitivity of  $\text{Y}_2\text{O}_3:\text{Yb}^{3+}, \text{Ho}^{3+}$  particles. Reproduced from ref. 59 with permission of Elsevier.

**Table 4** The maximum sensitivity values of  $\text{Ho}^{3+}$  doped phosphors by the fluorescence intensity ratio technique are presented, and the involved transitions from the TCL, excitation wavelength, emission wavelength, and temperature range are included

| Rare-earth ions                  | Host                                           | $\lambda_{\text{ex}}$<br>(nm) | $\lambda_{\text{em}}$ (nm) | Transitions                                                                         | Temperature<br>range (K) | $S_R$ (maximum)                   | $S_A$        | Ref. |
|----------------------------------|------------------------------------------------|-------------------------------|----------------------------|-------------------------------------------------------------------------------------|--------------------------|-----------------------------------|--------------|------|
| $\text{Ho}^{3+}$                 | In-Zn-Sr-Ba glass                              | 473                           | 545, 750                   | $^5\text{F}_4/^5\text{S}_2 \rightarrow ^5\text{I}_8, ^5\text{I}_7$                  | 20–300                   | $0.0036 \text{ K}^{-1}$ (59 K)    | $181.64/T^2$ | 22   |
| $\text{Ho}^{3+}$                 | $\text{TeO}_2$ glass                           | 890                           | 538, 545                   | $^5\text{F}_4, ^5\text{S}_2 \rightarrow ^5\text{I}_8$                               | 265–440                  | $0.0098 \text{ K}^{-1}$ (130 K)   | $255/T^2$    | 62   |
| $\text{Ho}^{3+}$                 | $\text{LiTeO}_2$ glass                         | 890                           | 538, 543                   | $^5\text{F}_4, ^5\text{S}_2 \rightarrow ^5\text{I}_8$                               | 265–383                  | $0.0063 \text{ K}^{-1}$ (265 K)   | $255/T^2$    | 63   |
| $\text{Ho}^{3+}, \text{Yb}^{3+}$ | $\text{Y}_2\text{O}_3$                         | 978                           | 536.5, 772.6               | $^5\text{F}_4/^5\text{S}_2 \rightarrow ^5\text{I}_8, ^5\text{I}_7$                  | 10–300                   | $0.0097 \text{ K}^{-1}$ (85 K)    | $241.2/T^2$  | 60   |
| $\text{Ho}^{3+}, \text{Yb}^{3+}$ | $\text{Ba}_{0.77}\text{Ca}_{0.23}\text{TiO}_3$ | 980                           | 546, 754                   | $^5\text{F}_4/^5\text{S}_2 \rightarrow ^5\text{I}_8, ^5\text{I}_7$                  | 93–300                   | $0.0053 \text{ K}^{-1}$ (93 K)    | $182.3/T^2$  | 64   |
| $\text{Ho}^{3+}, \text{Yb}^{3+}$ | $\text{Ca}_{12}\text{Al}_{14}\text{O}_{33}$    | 980                           | 467, 492, 542, 552         | $^5\text{G}_4/^5\text{G}_5, ^5\text{F}_4/^5\text{S}_2 \rightarrow ^5\text{I}_8$     | 298–500                  |                                   |              | 65   |
| $\text{Ho}^{3+}, \text{Yb}^{3+}$ | $\text{CaMoO}_4$                               | 980                           | 460, 489                   | $^5\text{F}_3, ^3\text{K}_8 \rightarrow ^5\text{I}_8$                               | 303–543                  | $0.0066 \text{ K}^{-1}$ (353 K)   | $648.5/T^2$  | 66   |
| $\text{Ho}^{3+}, \text{Yb}^{3+}$ | $\text{CaWO}_4$                                | 980                           | 460, 487                   | $^5\text{G}_6/^5\text{F}_1, ^5\text{F}_{2,3}/^3\text{K}_8 \rightarrow ^5\text{I}_8$ | 303–923                  | $0.0050 \text{ K}^{-1}$ (923 K)   | $1890/T^2$   | 67   |
| $\text{Ho}^{3+}, \text{Yb}^{3+}$ | $\text{PbF}_2$ glass ceramic                   | 980                           | 445, 485                   | $^5\text{G}_6/^5\text{F}_1, ^5\text{F}_{2,3}/^3\text{K}_8 \rightarrow ^5\text{I}_8$ | 303–643                  | $0.00102 \text{ K}^{-1}$ (1119 K) | $2191.9/T^2$ | 68   |

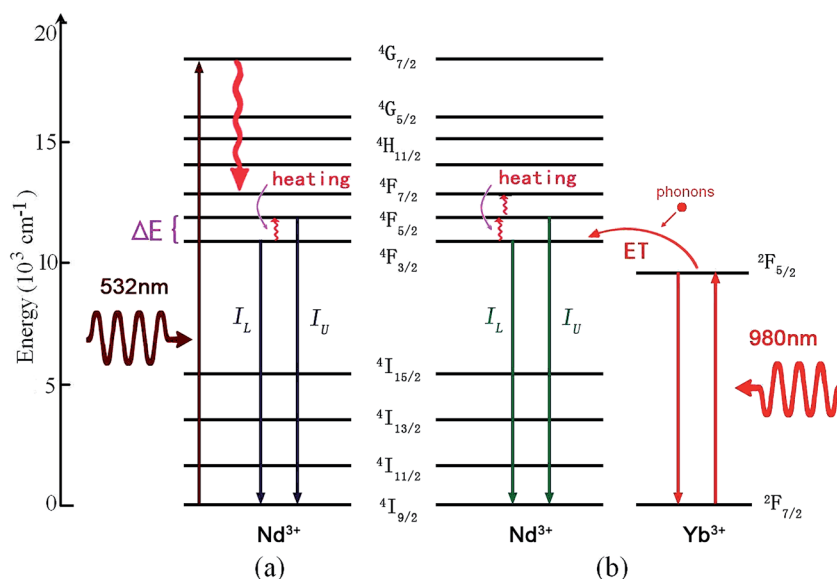


**Fig. 11** Host dependent  $\Delta E_T$  and  $\Delta E_m$  between the TCL of  $\text{Ho}^{3+}$  ions, and the corresponding  $\delta$  values.

demonstrated novel optical temperature thermometry based on infrared emissions centered at 800 nm ( $^2\text{H}_{11/2} \rightarrow ^4\text{I}_{13/2}$ ) and 850 nm ( $^4\text{S}_{3/2} \rightarrow ^4\text{I}_{13/2}$ ) in  $\text{Er}^{3+}$  doped Sr-Ba-Nb-B glass ceramic.<sup>26</sup> A

maximum sensitivity of  $0.0017 \text{ K}^{-1}$  for 600 K was obtained. The hosts, excitation wavelength ( $\lambda_{\text{ex}}$ ), emission wavelength ( $\lambda_{\text{em}}$ ), the involved transitions,  $S_R$  and  $S_A$  values, and the temperature range for phosphors doped with  $\text{Er}^{3+}$  ions are summarized in Table 1.

(2) Up-conversion of  $\text{Er}^{3+}$  ions under infrared excitation: under infrared excitation, the excited states of  $\text{Er}^{3+}$  ions are populated by the GSA and successive excited state absorption (ESA),<sup>27–36</sup> as shown in Fig. 3(b). It needs three 1540 nm infrared photons and two 800 nm (or 980 nm) infrared photons to populate the thermally coupled  $^2\text{H}_{11/2}$  and  $^4\text{S}_{3/2}$  energy levels. The energy transfer between  $\text{Er}^{3+}$  ions is thought of as the main up-conversion mechanism, only when the  $\text{Er}^{3+}$  concentration of the optically active center is high enough.<sup>13</sup> The ions in the high energy excited states relax to the next lower energy levels  $^2\text{H}_{11/2}$  and  $^4\text{S}_{3/2}$  through non-radiative relaxation, giving two green emissions with the intensities of  $I_U$  and  $I_L$ . The temperature dependent fluorescence intensity ratio,  $I_U/I_L$ , is used as the precise evaluation scale of optical temperature sensing.



**Fig. 12** The mechanism of optical thermal sensing through (1) down-conversion of the  $\text{Nd}^{3+}$  ion under 532 nm green excitation, and (2)  $\text{Yb}^{3+}$ – $\text{Nd}^{3+}$  energy transfer under 980 nm infrared excitation.

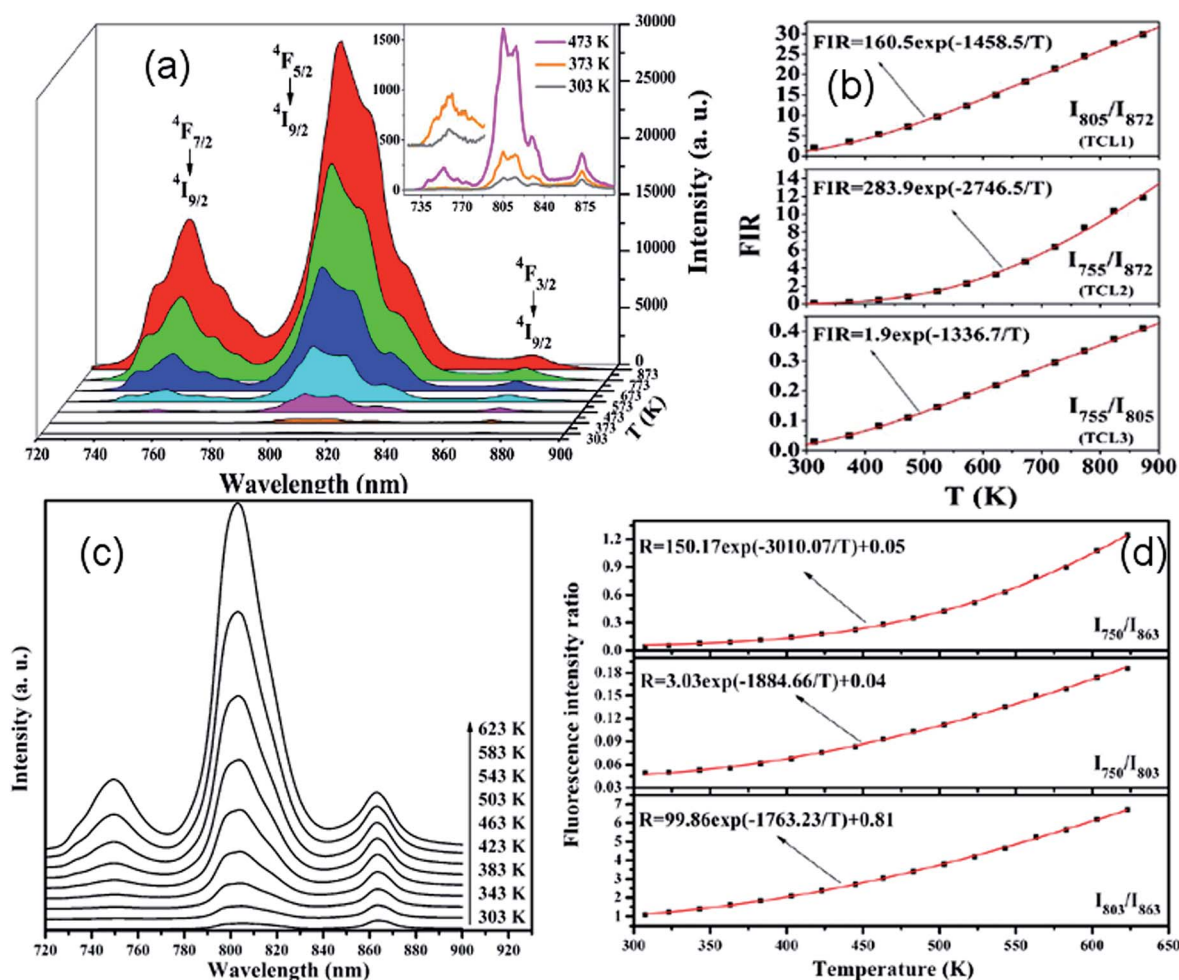


Fig. 13 Temperature dependent (a) near infrared emission spectra and (b) fluorescence intensity ratio (FIR) in  $\text{Nd}^{3+}/\text{Yb}^{3+}$  co-doped  $\text{CaWO}_4$  phosphors excited by a 980 nm laser. Temperature dependent (c) near infrared emission spectra and (d) FIR in  $\text{Nd}^{3+}/\text{Yb}^{3+}$  co-doped oxyfluoride glass ceramic containing  $\text{PbF}_2$  nanocrystals excited by a 980 nm laser. Reproduced from ref. 70 and 71 with permission of Optical Society of America and Elsevier.

Based on the up-conversion luminescence emissions centered at 524 nm ( $^2\text{H}_{11/2} \rightarrow ^4\text{I}_{15/2}$ ) and 547 nm ( $^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ ) of  $\text{Er}^{3+}$  ions, the optical temperature thermometry was studied by Luis *et al.* in a fluorotellurite glass under 800 nm excitation.<sup>27</sup> A better behaviour as a temperature sensor has been obtained for the less  $\text{Er}^{3+}$  concentrated glass with a maximum sensitivity of  $54 \times 10^{-4} \text{ K}^{-1}$  at 540 K. Based on the thermalization effects

between the upper levels responsible for green fluorescence at 522 nm and 546 nm under 805 nm laser excitation, new optical temperature sensing using the micrometer sized ZBLALiP spherical cavity was explored by Cai *et al.*<sup>28</sup> Using a diode laser emitting at about 980 nm as the excitation source, the optical temperature sensing properties based on green emissions of  $\text{Er}^{3+}$  ions were studied in BZT-BCT ferroelectric ceramics,

Table 5 The maximum sensitivity values of  $\text{Nd}^{3+}$  doped phosphors by the fluorescence intensity ratio technique are presented, and the involved transitions from the TCL, excitation wavelength, emission wavelength, and temperature range are included

| Rare-earth ions                  | Host                           | $\lambda_{\text{ex}}$ (nm) | $\lambda_{\text{em}}$ (nm) | Transitions                                                                                         | Temperature range (K) | $S_R$ (maximum)                | $S_A$         | Ref. |
|----------------------------------|--------------------------------|----------------------------|----------------------------|-----------------------------------------------------------------------------------------------------|-----------------------|--------------------------------|---------------|------|
| $\text{Nd}^{3+}$                 | P-K-Ba-Al glass                | 532                        | 810, 880                   | $^4\text{F}_{5/2}, ^4\text{F}_{3/2} \rightarrow ^4\text{I}_{9/2}$                                   | 300–850               | 0.0153 $\text{K}^{-1}$ (300 K) | $1306.47/T^2$ | 69   |
| $\text{Nd}^{3+}, \text{Yb}^{3+}$ | $\text{CaWO}_4$                | 980                        | 755, 872                   | $^4\text{F}_{7/2}, ^4\text{F}_{3/2} \rightarrow ^4\text{I}_{9/2}$                                   | 303–873               |                                | $2746.5/T^2$  | 70   |
|                                  |                                |                            | 805, 872                   | $^4\text{F}_{5/2}, ^4\text{F}_{3/2} \rightarrow ^4\text{I}_{9/2}$                                   |                       |                                | $1458.5/T^2$  |      |
|                                  |                                |                            | 755, 805                   | $^4\text{F}_{7/2}, ^4\text{F}_{5/2} \rightarrow ^4\text{I}_{9/2}$                                   |                       |                                | $1336.7/T^2$  |      |
|                                  |                                |                            | 750, 863                   | $^4\text{F}_{7/2}/^4\text{S}_{3/2}, ^4\text{F}_{3/2} \rightarrow ^4\text{I}_{9/2}$                  |                       |                                | $3010.07/T^2$ |      |
| $\text{Nd}^{3+}, \text{Yb}^{3+}$ | PbF <sub>2</sub> glass Ceramic | 980                        | 750, 803                   | $^4\text{F}_{7/2}/^4\text{S}_{3/2}, ^2\text{H}_{9/2}/^4\text{F}_{5/2} \rightarrow ^4\text{I}_{9/2}$ | 303–623               |                                | $1884.66/T^2$ | 71   |
|                                  |                                |                            | 803, 863                   | $\text{F}_{5/2}/^2\text{H}_{9/2}, ^4\text{F}_{3/2} \rightarrow ^4\text{I}_{9/2}$                    |                       |                                | $1763.23/T^2$ |      |
|                                  |                                |                            |                            |                                                                                                     |                       |                                |               |      |

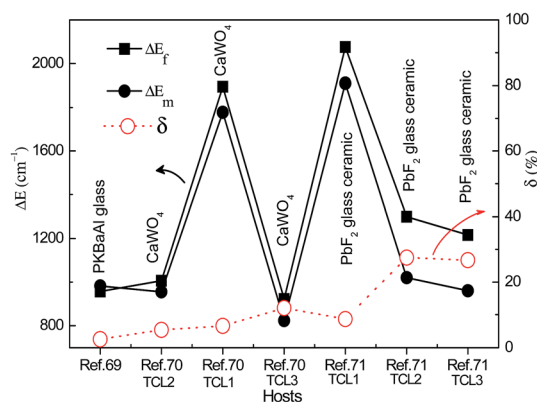


Fig. 14 Host dependent  $\Delta E_f$  and  $\Delta E_m$  between the TCL of  $\text{Nd}^{3+}$  ions, and the corresponding  $\delta$  values.

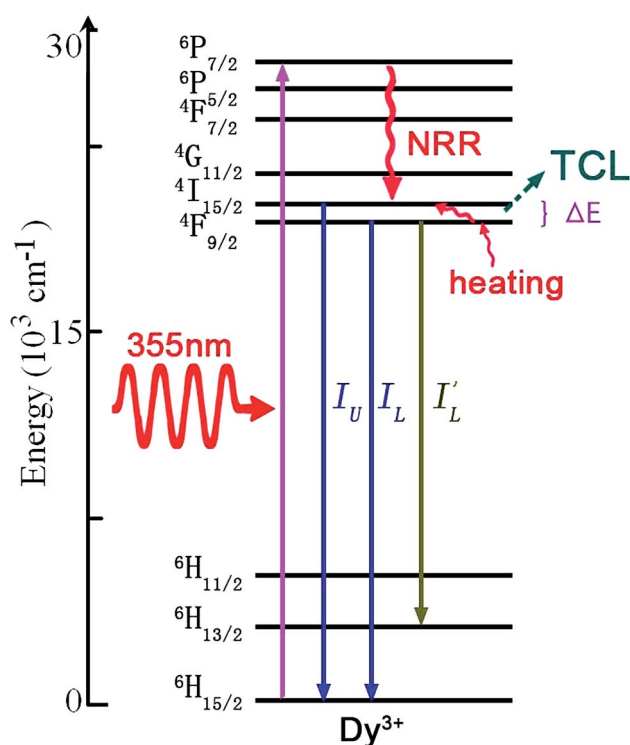


Fig. 15 The mechanism of optical thermal sensing through down-conversion of the  $\text{Dy}^{3+}$  ion under 355 nm ultraviolet excitation.

silicate glass,  $\text{BaTiO}_3$  nanocrystals, PLZT transparent ceramics,  $\text{Yb}_3\text{Al}_5\text{O}_{12}$  nanocrystals, and  $\text{ZnO}$  nanocrystals, respectively.<sup>29–34</sup> The maximum sensitivity of  $48 \times 10^{-4} \text{ K}^{-1}$  for 467 K was obtained in  $\text{Er}^{3+}$  doped  $\text{Yb}_3\text{Al}_5\text{O}_{12}$  nanocrystals. Under 1540 nm infrared excitation, based on green luminescence emissions of  $\text{Er}^{3+}$  ions, the optical temperature sensing properties were studied in  $\text{Ga}_2\text{S}_3\text{-La}_2\text{O}_3$  chalcogenide glass, and the  $\text{Na}_{0.82}\text{-Ca}_{0.08}\text{Er}_{0.16}\text{Y}_{0.853}\text{F}_4$  phosphor.<sup>35,36</sup> The corresponding results are summarized in Table 1.

(3)  $\text{Yb}^{3+}\text{-Er}^{3+}$  energy transfer under infrared excitation: at high temperature, fluorescence quenching of emission bands of  $\text{Er}^{3+}$  ions induced by the temperature was obvious at 423 K.<sup>21–35</sup>

It is necessary to enhance the emission intensity of phosphors in the process research of optical temperature sensing.

Among the rare-earth ions, the  $\text{Yb}^{3+}$  ion consists of only two levels, and has only one electronic excited state,  $^2\text{F}_{5/2}$ , that is located in the near infrared region at about 980 nm.<sup>13</sup> Moreover, high power InGaAs diode lasers are available to directly pump the  $\text{Yb}^{3+}$  absorption band around 980 nm. Fluorescent materials with  $\text{Yb}^{3+}$  as a sensitizer can convert short infrared into visible/ultraviolet light *via* energy transfer between lanthanide ions. As reported, the doping content of  $\text{Yb}^{3+}$  was usually kept at 18% or higher, because the large energy gap between the excited state  $^2\text{F}_{5/2}$  and ground state  $^2\text{F}_{7/2}$  blocks multiphoton cross-relaxation.<sup>14</sup> Thus, in order to enlarge the photo-absorption cross section of 980 nm infrared light, the  $\text{Yb}^{3+}$  ion is chosen to be a sensitizer for the  $\text{Er}^{3+}$  ion. Under 980 nm excitation, the two successive energy transfers from  $\text{Yb}^{3+}$  to  $\text{Er}^{3+}$  can induce the population of the thermally coupled  $^2\text{H}_{11/2}$  and  $^4\text{S}_{3/2}$  levels of  $\text{Er}^{3+}$ ,<sup>37,39–50</sup> and five successive energy transfers from  $\text{Yb}^{3+}$  to  $\text{Er}^{3+}$  can induce the population of the thermally coupled  $^4\text{D}_{7/2}$  and  $^4\text{G}_{9/2}$  levels of  $\text{Er}^{3+}$ ,<sup>38</sup> as shown in Fig. 3(c). The temperature dependent fluorescence intensity ratios,  $I_U/I_L$  and  $I'_U/I'_L$ , are used to evaluate the optical temperature sensing properties.

Recently, in the low temperature range, based on green luminescence emissions of  $\text{Er}^{3+}\text{-Yb}^{3+}$  co-doped  $\text{NaYF}_4$  nanoparticles, Vetrone *et al.* demonstrated the optical temperature sensing properties in the internal temperature of the living HeLa cervical cancer cell from 25 °C to 45 °C.<sup>37</sup> Notably, the temperature dependence of the five-photon 256 nm ( $^4\text{D}_{7/2} \rightarrow ^4\text{I}_{15/2}$ ) and 276 nm ( $^4\text{G}_{9/2} \rightarrow ^4\text{I}_{15/2}$ ) ultraviolet up-conversion luminescence in  $\text{Yb}^{3+}\text{-Er}^{3+}$  co-doped  $\beta\text{-NaLuF}_4$  nanocrystals was studied firstly by Zheng *et al.* from 303 K to 523 K with a maximum sensitivity of  $0.0052 \text{ K}^{-1}$  at 303 K,<sup>38</sup> as shown in Fig. 5.

In the high temperature range, special phosphors are needed with high thermal stability and high solubility between hosts and lanthanide ions. Among the oxide materials,  $\text{Y}_2\text{O}_3$  is a good candidate host, due to its wide bandgap, high melting point, high solubility between  $\text{Y}^{3+}$  and  $\text{Er}^{3+}$ , and good transparency from ultraviolet to infrared. Recently, in  $\text{Er}^{3+}\text{-Yb}^{3+}$  co-doped  $\text{Y}_2\text{O}_3$  nanoparticles, optical temperature sensing properties were studied in the temperature range from 93 K to 613 K through analyzing the temperature dependent fluorescence intensity ratio of the two green emissions.<sup>39–42</sup> A maximum sensitivity of  $528 \times 10^{-4} \text{ K}^{-1}$  for 150 K was obtained in the  $\text{Y}_{1.97}\text{Yb}_{0.02}\text{Er}_{0.01}\text{O}_3$  nanophosphors.<sup>40</sup> The fluorescence intensity ratio of the green up-conversion emissions at 523 nm and 545 nm in  $\text{Er}^{3+}\text{-Yb}^{3+}$  co-doped  $\text{Al}_2\text{O}_3$  was studied by Dong *et al.* as a function of temperature with a maximum sensitivity of  $0.0051 \text{ K}^{-1}$  at 495 K.<sup>43</sup> Additionally, oxide salts, such as  $\text{BaMoO}_4$ ,  $\text{Y}_2\text{SiO}_5$ ,  $\text{MWO}_4$  ( $\text{M} = \text{Ca}, \text{Sr}$ ),  $\text{YVO}_4$ ,  $\text{BaTiO}_3$ ,  $\text{Yb}_2\text{Ti}_2\text{O}_7$ ,  $\text{Bi}_7\text{Ti}_4\text{-NbO}_{21}$ , and  $\text{LiNbO}_3$ , were doped with  $\text{Er}^{3+}$  and  $\text{Yb}^{3+}$  ions, and were studied as optical temperature sensors by the temperature dependent fluorescence intensity ratio of the two green emissions.<sup>43–54</sup> From Table 1, one can find that the maximum  $S_R$ ,  $0.0528 \text{ K}^{-1}$ , is obtained at 150 K in  $\text{Er}^{3+}\text{-Yb}^{3+}$  co-doped  $\text{Y}_2\text{O}_3$  phosphors under 978 nm excitation, and the maximum  $S_A$ ,

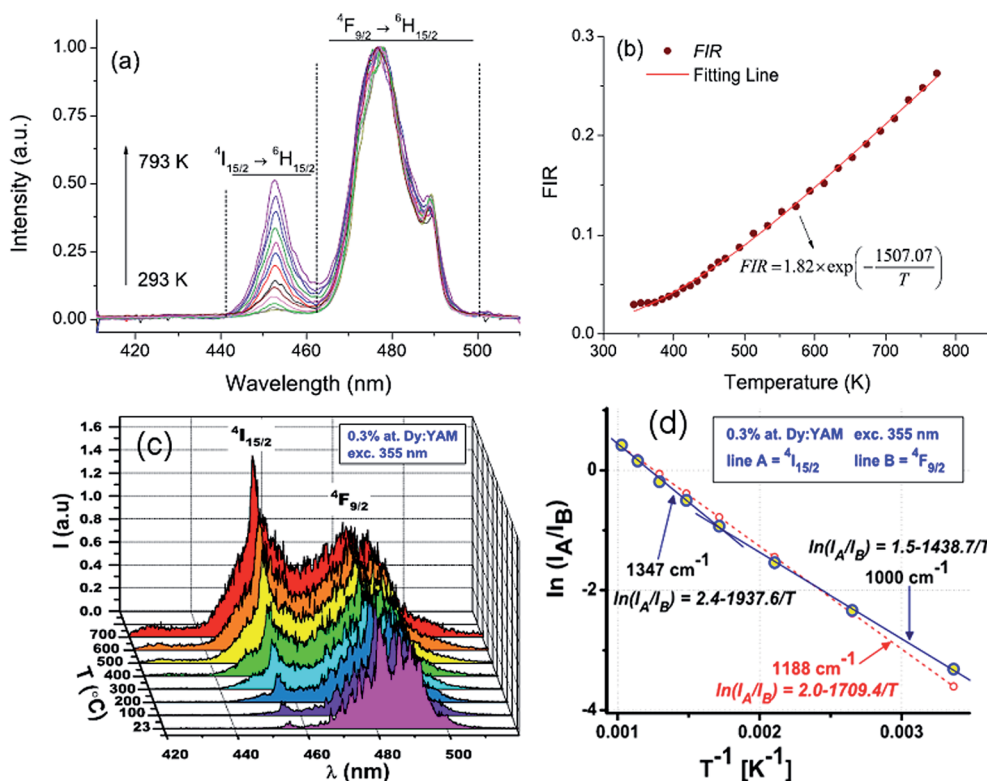


Fig. 16 Temperature dependent (a) emission spectra and (b) fluorescence intensity ratio (FIR) between the 455 nm and 478 nm emissions for the BaYF<sub>5</sub>:2% Dy<sup>3+</sup> sample. Temperature dependent (c) emission spectra and (d) Arrhenius plot of the intensity ratio for Dy<sup>3+</sup> doped Y<sub>4</sub>Al<sub>2</sub>O<sub>9</sub> crystals. Reproduced from ref. 72 and 73 with permission of Optical Society of America and Elsevier.

2109.31/ $T^2$ , is obtained in Er<sup>3+</sup>–Yb<sup>3+</sup> co-doped CaWO<sub>3</sub> phosphors under 978 nm excitation.

The term  $\delta$  is a key parameter used to judge the calculation accuracy of the fluorescence intensity ratio technique in the data fitting process.  $\Delta E_f$  and  $\Delta E_m$  between the TCL of Er<sup>3+</sup> ions and the corresponding  $\delta$  values are calculated and summarized in Table 2 through analyzing ref. 21–55. In Table 2, we can see that most of the Er<sup>3+</sup> doped phosphors have a  $\delta$  value of more than 5%, and the maximum  $\delta$  value is about 53.89% for the Er<sup>3+</sup> doped Si–B–Ba–Na glass. The  $\delta$  values less than 5% are obtained in Er<sup>3+</sup> doped tellurite glass, Na<sub>0.82</sub>Ca<sub>0.08</sub>Er<sub>0.16</sub>Y<sub>0.853</sub>F<sub>4</sub> bulks, PKAZLF glass, and ZBLALiP glass, respectively. The small  $\delta$  values for tellurite, PKAZLF, and ZBLALiP glass materials can be explained as follows: the successive excited state absorption overcomes the energy transfer between the thermally coupled levels and other levels, due to the abnormal local ligand fields around the Er<sup>3+</sup> sites in the tellurite, PKAZLF, and ZBLALiP

glass hosts.<sup>14,17</sup> The population of the TCL of Er<sup>3+</sup> ions at high temperature obeys the Boltzmann distribution. As shown in Table 2, most of the Er<sup>3+</sup>–Yb<sup>3+</sup> co-doped phosphors have a value of  $\delta$  of more than 10%, and the maximum  $\delta$  value is about 90% for the Er<sup>3+</sup>–Yb<sup>3+</sup> co-doped  $\beta$ -NaLuF<sub>4</sub> nanocrystals. Values of  $\delta$  of less than 5% are observed in Er<sup>3+</sup>–Yb<sup>3+</sup> co-doped Y<sub>2</sub>O<sub>3</sub>, and Er<sup>3+</sup>–Mo<sup>6+</sup> co-doped Yb<sub>2</sub>Ti<sub>2</sub>O<sub>7</sub>. The large  $\delta$  value may be ascribed to the actual deviation of eqn (6) induced by energy transfer from the TCL to the other excited states.<sup>14</sup>

### 3.2 Optical thermometry based on Yb<sup>3+</sup>–Tm<sup>3+</sup> co-doped phosphors

The trivalent Tm<sup>3+</sup> ion has a 4f<sup>12</sup> electronic configuration, and is reported as one of the most efficient blue luminescence center ions. It has two couples of adjacent thermally coupled levels, such as <sup>3</sup>F<sub>2,3</sub> and <sup>3</sup>H<sub>4</sub>, and <sup>1</sup>G<sub>4(a)</sub> and <sup>1</sup>G<sub>4(b)</sub>, whose relative emission intensity is strongly temperature dependent. As

Table 6 The maximum sensitivity values of Dy<sup>3+</sup> doped phosphors by the fluorescence intensity ratio technique are presented, and the involved transitions from the TCL, excitation wavelength, emission wavelength, and temperature range are included

| Rare-earth ions  | Host                                          | $\lambda_{ex}$ (nm) | $\lambda_{em}$ (nm) | Transitions                                                                                     | Temperature range (K) | $S_R$ (maximum)                | $S_A$          | Ref. |
|------------------|-----------------------------------------------|---------------------|---------------------|-------------------------------------------------------------------------------------------------|-----------------------|--------------------------------|----------------|------|
| Dy <sup>3+</sup> | BaYF <sub>5</sub>                             | 355                 | 455, 478            | <sup>4</sup> I <sub>15/2</sub> , <sup>4</sup> F <sub>9/2</sub> → <sup>6</sup> H <sub>15/2</sub> | 293–773               | 0.02 K <sup>-1</sup> (700 K)   | 1507.07/ $T^2$ | 72   |
| Dy <sup>3+</sup> | Y <sub>4</sub> Al <sub>2</sub> O <sub>9</sub> | 355                 | 455, 481            | <sup>4</sup> I <sub>15/2</sub> , <sup>4</sup> F <sub>9/2</sub> → <sup>6</sup> H <sub>15/2</sub> | 296–573               |                                | 1438.7/ $T^2$  | 73   |
|                  |                                               |                     |                     |                                                                                                 | 573–973               | 0.003 °C <sup>-1</sup> (973 K) | 1937.6/ $T^2$  |      |

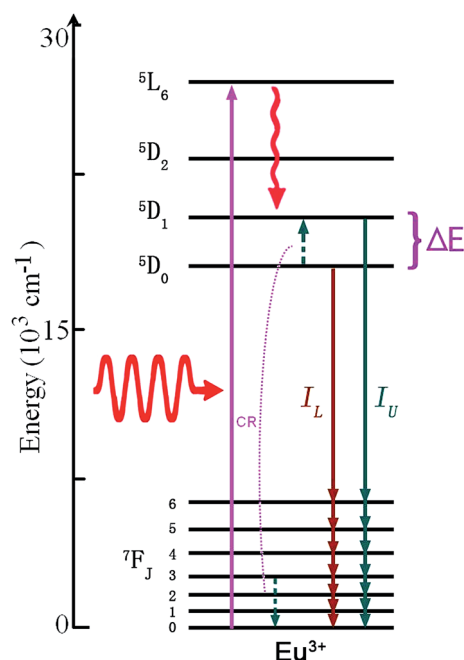


Fig. 17 The mechanism of optical thermal sensing through down-conversion of the  $\text{Eu}^{3+}$  ion under infrared excitation.

shown in Fig. 6, the energy transfer (ET) from  $\text{Yb}^{3+}$  to  $\text{Tm}^{3+}$  is the main population mechanism of optical thermometry based on up-conversion luminescence of the  $\text{Tm}^{3+}$  ion.<sup>60</sup> Under 980 nm infrared excitation, the thermally coupled levels  $^1\text{G}_{4(a)}$  and  $^1\text{G}_{4(b)}$  are populated by three successive ETs from  $\text{Yb}^{3+}$  to  $\text{Tm}^{3+}$ . The thermally coupled levels  $^3\text{F}_{2,3}$  and  $^3\text{H}_4$  are populated by two successive ETs from  $\text{Yb}^{3+}$  to  $\text{Tm}^{3+}$ . The ions in the high energy excited states, such as  $^1\text{G}_{4(a)}$  and  $^1\text{G}_{4(b)}$ ,  $^3\text{F}_{2,3}$  and  $^3\text{H}_4$ , relax to the next lower energy levels through non-radiative relaxation, giving two blue emissions with the intensities  $I_U$  and  $I_L$ , and two infrared emissions with the intensities  $I'_U$  and  $I'_L$ . The temperature dependent fluorescence intensity ratios,  $I_U/I_L$  and  $I'_U/I'_L$ , are used to evaluate the optical temperature sensing properties.

The spectral properties of the  $\text{Tm}^{3+}$  luminescence band at 480 nm from the  $^1\text{G}_4 \rightarrow ^3\text{H}_6$  transition in  $\text{Tm}^{3+}$  and  $\text{Yb}^{3+}$  co-doped  $\text{NaNbO}_3$  nanocrystals were analyzed in the biophysical temperature range from 297 K to 353 K with a 980 nm excitation source.<sup>56</sup> Novel TCL,  $^1\text{G}_{4(a)}$  and  $^1\text{G}_{4(b)}$ , were observed firstly by Pereira *et al.* Experiments proved that the population redistribution among the thermally coupled stark levels could be successfully used for ratiometric thermal sensing. By analyzing the  $R$  value between the 700 nm and 800 nm up-conversion emissions, the optical temperature sensing properties of the  $\text{Tm}^{3+}/\text{Yb}^{3+}$  co-doped oxyfluoride  $\text{PbF}_2$  glass ceramic were studied by Xu *et al.* in the temperature range of 293–703 K.<sup>57</sup> The optical

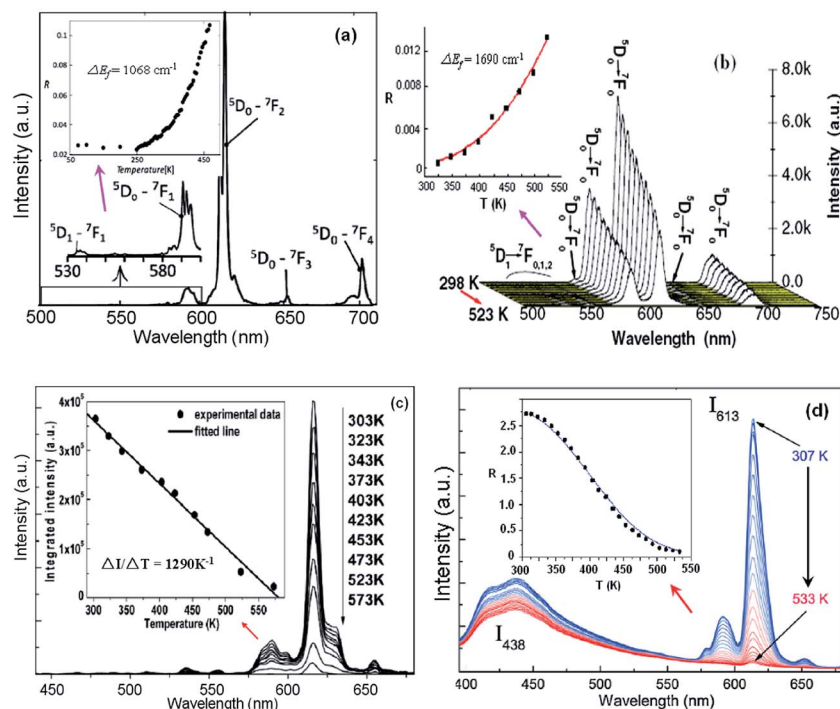


Fig. 18 (a) Emission spectrum of  $\text{CaGd}_{1.8}\text{Eu}_{0.2}(\text{WO}_4)_4$  upon excitation at 395 nm. The inset is the integrated intensity ratio of the  $^5\text{D}_1$  to  $^5\text{D}_0$  emissions. (b) Temperature dependent emission spectra of  $\text{NaEuF}_4$  spheres under 394 nm excitation. The inset is the emission intensity ratio of the  $^5\text{D}_1$  to  $^5\text{D}_0$  emissions with temperature. (c) Temperature dependent emission spectra of  $\text{Sr}_2\text{CeO}_4:\text{Eu}^{3+}$  under 365 nm excitation. The inset shows the integrated emission total intensity of 591 nm, 616 nm, and 654 nm at different temperatures. (d) Temperature dependent emission spectra of  $\text{TiO}_2:\text{Eu}^{3+}$  nanopowder. The inset is the emission intensity ratio of the 438 nm and 613 nm emissions with temperature. Reproduced from ref. 74–77 with permission of Optical Society of America and Elsevier.

**Table 7** The maximum sensitivity values of  $\text{Eu}^{3+}$  doped phosphors by the fluorescence intensity ratio technique are presented, and the involved transitions from the TCL, excitation wavelength, emission wavelength, and temperature range are included

| Rare-earth ions  | Host                                     | $\lambda_{\text{ex}}$ (nm) | $\lambda_{\text{em}}$ (nm) | Transitions                                           | Temperature range (K) | $S_{\text{R}}$ (maximum)        | $S_{\text{A}}$ | Ref. |
|------------------|------------------------------------------|----------------------------|----------------------------|-------------------------------------------------------|-----------------------|---------------------------------|----------------|------|
| $\text{Eu}^{3+}$ | $\text{CaEu}_2(\text{WO}_4)_4$ scheelite | 395                        | 535, 590                   | $^5\text{D}_1, ^5\text{D}_0 \rightarrow ^7\text{F}_1$ | 300–500               | $0.014 \text{ K}^{-1}$ (300 K)  | $1515.5/T^2$   | 77   |
| $\text{Eu}^{3+}$ | $\text{NaEuF}_4$ phosphor                | 394                        | 500–560, 560–720           | $^5\text{D}_1, ^5\text{D}_0 \rightarrow ^7\text{F}_1$ | 298–523               | $0.0043 \text{ K}^{-1}$         | $2398.2/T^2$   | 78   |
| $\text{Eu}^{3+}$ | $\text{Sr}_2\text{CeO}_4$                | 365                        | 591, 616, 654              | $^5\text{D}_0 \rightarrow ^7\text{F}_{1/2/3}$         | 303–573               | $1290 \text{ K}^{-1}$           |                | 79   |
| $\text{Eu}^{3+}$ | $\text{TiO}_2$ nanoparticles             | 360                        | 438, 613                   | Host trap, $^5\text{D}_0 \rightarrow ^7\text{F}_2$    | 307–533               | $0.0243 \text{ K}^{-1}$ (533 K) |                | 80   |

temperature sensing properties of  $\text{Tm}^{3+}$  were studied in  $\text{Tm}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{Y}_2\text{O}_3$  bulk and  $\text{Y}_2\text{O}_3$  sub-micronic spherical particles.<sup>58,59</sup> The  $S_{\text{A}}$  and  $S_{\text{B}}$  values obtained from the  $\text{Tm}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{Y}_2\text{O}_3$  sub-micronic spherical particles are larger than those from the  $\text{Tm}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{Y}_2\text{O}_3$  bulk.

Recently, trivalent rare-earth ion doped fluoride nanocrystals were reported as the most efficient up-conversion materials.<sup>13,14</sup> Based on ultraviolet up-conversion emissions from the  $^6\text{P}_{5/2}/^8\text{S}_{7/2}$  and  $^6\text{P}_{7/2}/^8\text{S}_{7/2}$  TCL, Zheng *et al.* demonstrated firstly the optical temperature sensing properties in  $\text{Tm}^{3+}$ – $\text{Yb}^{3+}$ – $\text{Gd}^{3+}$  tri-doped  $\text{NaLuF}_4$  microcrystals in the range of 298–523 K,<sup>60</sup> as shown in Fig. 7. The maximum sensor sensitivity of about  $0.0029 \text{ K}^{-1}$  was found at 298 K. The maximum sensor sensitivities were found to be about  $0.0004 \text{ K}^{-1}$  (333 K) and  $0.0029 \text{ K}^{-1}$  (298 K) by analyzing the ultraviolet emissions from the  $^6\text{P}_{5/2}/^8\text{S}_{7/2}$  and  $^6\text{P}_{7/2}/^8\text{S}_{7/2}$  levels, respectively. To our knowledge, fluoride nanocrystals are easily oxidized at high temperature. To overcome it,  $\text{NaYbF}_4:\text{Tm}^{3+}@\text{SiO}_2$  core-shell materials were synthesized, and their optical temperature sensing properties based on the  $^3\text{F}_2/^3\text{H}_4$  TCL of the  $\text{Tm}^{3+}$  ion were studied by Wang *et al.* in a wide temperature range from 100 K to 700 K.<sup>61</sup> A better behavior as a low temperature sensor has been obtained with a minimum sensitivity of  $5.4 \times 10^{-4} \text{ K}^{-1}$  at 430 K.

In Table 3 we summarize, for the sake of comparison, the different performance parameters of sensitivities and the involved transitions of  $\text{Tm}^{3+}$  doped phosphors. The maximum value of  $S_{\text{R}}$ ,  $0.078 \text{ K}^{-1}$ , is obtained in  $\text{Tm}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{Y}_2\text{O}_3$  sub-micronic spherical particles. The maximum value of  $S_{\text{A}}$ ,  $2829.5/T^2$ , is obtained in  $\text{Tm}^{3+}$ /Yb<sup>3+</sup> co-doped oxyfluoride  $\text{PbF}_2$  glass ceramic. To study the feasibility evaluation in application,  $\Delta E_{\text{f}}$ ,  $\Delta E_{\text{m}}$ , and  $\delta$  are studied in  $\text{Tm}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{NaNbO}_3$  nanocrystals, oxyfluoride  $\text{PbF}_2$  glass ceramic,  $\text{Y}_2\text{O}_3$ ,  $\text{NaLuF}_4$  microcrystals, and  $\text{NaYbF}_4$  under 980 nm excitation, as shown in Fig. 8. A large  $\delta$  value of more than 80% is observed in  $\text{Tm}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{NaNbO}_3$  nanocrystals. The large  $\delta$  value may be ascribed to the actual deviation of eqn (6) induced by energy transfer from the TCL to the other excited states.<sup>14</sup> It is accurate to use  $\text{NaYbF}_4:\text{Tm}^{3+}@\text{SiO}_2$  core-shell materials and  $\text{Tm}^{3+}$ /Yb<sup>3+</sup> co-doped oxyfluoride Si–Pb glass ceramic to evaluate the scale of optical temperature sensing.

### 3.3 Optical thermometry based on $\text{Ho}^{3+}$ doped phosphors

The trivalent  $\text{Ho}^{3+}$  ion has a  $4f^{10}$  electronic configuration, and is reported as one of the most efficient green luminescence center

ions, due to the intermediate  $^5\text{S}_2$  level with a long fluorescence lifetime.<sup>13,14</sup> It has three couples of adjacent TCL, such as  $^5\text{F}_4/^5\text{S}_2$ ,  $^5\text{F}_{2,3}/^3\text{K}_8$ , and  $^5\text{G}_6/^5\text{F}_1$ . As shown in Fig. 9, the two-step up-conversion of  $\text{Ho}^{3+}$  and successive energy transfer from  $\text{Yb}^{3+}$  to  $\text{Ho}^{3+}$  are the main mechanisms of optical thermometry based on  $\text{Ho}^{3+}$  doped phosphors.<sup>60–68</sup> In the case of a low doping concentration (<1%), the ESA process refers to a sequential absorption of two pump infrared photons at 890 nm by a single  $\text{Ho}^{3+}$  ion. Upon 980 nm laser irradiation,  $\text{Yb}^{3+}$  absorbs infrared photons with the generation of  $^2\text{F}_{7/2} \rightarrow ^2\text{F}_{5/2}$  upward transitions. Subsequently, it donates the energy to the adjacent  $\text{Ho}^{3+}$  through a phonon assisted energy transfer, with  $\text{Yb}^{3+}$  dropping back to its  $^2\text{F}_{7/2}$  ground state. This promotes the  $\text{Ho}^{3+}$  ions to their excited states, such as  $^5\text{I}_6$ ,  $^5\text{S}_2$ , and  $^5\text{G}_5$ . The thermally coupled  $^5\text{F}_4/^5\text{S}_2$ ,  $^5\text{F}_{2,3}/^3\text{K}_8$ , and  $^5\text{G}_6/^5\text{F}_1$  energy levels are populated by the process of non-radiative relaxation. The fluorescence intensity ratios,  $I_{\text{U}}/I_{\text{L}}$ ,  $I'_{\text{U}}/I'_{\text{L}}$ , and  $I''_{\text{U}}/I''_{\text{L}}$ , are dependent strongly on temperature, and are used to study the optical temperature sensing properties.

It was reported that there were two formulas used to fit  $\Delta E_{\text{f}}$  of the  $\text{Ho}^{3+}$  ions, due to the two different transitions, such as from the TCL to the other excited state, and from the TCL to the ground state. One formula is eqn (6), which is suitable to fit  $\Delta E_{\text{f}}$  corresponding to transitions from  $^5\text{F}_4/^5\text{S}_2$ ,  $^5\text{F}_{2,3}/^3\text{K}_8$ , and  $^5\text{G}_6/^5\text{F}_1$  to the  $^5\text{I}_8$  ground state. Using eqn (6) as a fitting formula, the temperature dependent fluorescence intensity ratios were studied in  $\text{Ho}^{3+}$  doped  $\text{TeO}_2$  glass and  $\text{LiTeO}_2$  glass under 890 nm excitation,<sup>62,63</sup> and  $\text{Ho}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{CaMoO}_4$ ,  $\text{CaWO}_4$ , and  $\text{PbF}_2$  glass ceramic under 980 nm excitation.<sup>66–68</sup> A maximum  $S_{\text{R}}$  of  $0.098 \text{ K}^{-1}$  for 130 K was obtained in  $\text{Ho}^{3+}$  doped  $\text{TeO}_2$  glass, and a maximum  $S_{\text{A}}$  of  $2191.9/T^2$  was obtained in  $\text{Ho}^{3+}$ – $\text{Yb}^{3+}$  co-doped glass ceramic.

The other formula is as follows:

$$R_1 = \frac{1 + C_1 \exp(-\Delta E_{\text{f}}/KT)}{C_2 + C_3 \exp(-\Delta E_{\text{f}}/KT)} \quad (11)$$

where  $C_1$ ,  $C_2$  and  $C_3$  are constants that depend on spontaneous emission rates, degeneracy of the energy levels and emission energies.<sup>60</sup> Eqn (11) is used to fit  $\Delta E_{\text{f}}$  originating from the transitions of  $^5\text{F}_4 \rightarrow ^5\text{I}_8$  and  $^5\text{S}_2 \rightarrow ^5\text{I}_7$ . Using eqn (11) as a fitting formula, the temperature dependent fluorescence intensity ratios were studied in  $\text{Ho}^{3+}$  doped In–Zn–Sr–Ba glass under 473 nm excitation,<sup>22</sup>  $\text{Ho}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{Y}_2\text{O}_3$  under 978 nm excitation,<sup>60</sup> and  $\text{Ba}_{0.77}\text{Ca}_{0.23}\text{TiO}_3$  and  $\text{Ca}_{12}\text{Al}_{14}\text{O}_{33}$  under 980 nm excitation.<sup>64,65</sup> The  $\text{Ho}^{3+}$ – $\text{Yb}^{3+}$  co-doped  $\text{Y}_2\text{O}_3$  powder was

obtained *via* the spray pyrolysis method; its SEM image in Fig. 10(a) shows that the resulting particles were spherical, sub-micronic in size and un-agglomerated. The up-conversion emission spectra in Fig. 10(b) show that the green (536 nm) and infrared (758, 764, and 772 nm) emissions were dependent strongly on temperature. Fig. 10(c) shows the experimental fluorescence intensity ratio (FIR) of emission at 536 nm relative to the emissions at 758 nm, 764 nm and 772 nm, and these were fitted with eqn (11). At temperatures of 85, 84 and 90 K, the sensitivity of  $\text{Ho}^{3+}$  in Fig. 10(d) exhibits maximum values of 0.097, 0.065 and 0.046  $\text{K}^{-1}$  for emissions at 536/772, 536/764 and 536/758 nm, respectively. The values of fitted  $\Delta E$  in the inset of Fig. 10(c) and sensitivity are dependent strongly on the FIR.

In Table 4 we summarize, for the sake of comparison, the different performance parameters of  $\text{Ho}^{3+}$  doped and  $\text{Ho}^{3+}\text{-Yb}^{3+}$  co-doped phosphors. The maximum value of  $S_R$ , 0.0098  $\text{K}^{-1}$ , is obtained in  $\text{Ho}^{3+}\text{-Yb}^{3+}$  co-doped  $\text{TeO}_2$  glass. The maximum value of  $S_A$ , 2191.9/ $T^2$ , is obtained in the  $\text{Tm}^{3+}/\text{Yb}^{3+}$  co-doped oxyfluoride  $\text{PbF}_2$  glass ceramic.

To value the fitting accuracy,  $\Delta E_f$  and  $\Delta E_m$  between the TCL of the  $\text{Ho}^{3+}$  ions and the corresponding  $\delta$  values are calculated in Fig. 11 through analyzing ref. 22, 60 and 62–68. The thermally coupled  $^5\text{F}_4/^5\text{S}_2$  energy levels were studied as a temperature function in ref. 22, 60 and 62–65. The minimum  $\delta$  value of 5.09% is obtained in  $\text{Ho}^{3+}$  doped  $\text{LiTeO}_2$  glass. The thermally coupled  $^5\text{F}_{2,3}/^3\text{K}_8$  energy levels were studied as a temperature function in ref. 66. A large  $\delta$  value of 64.6% was obtained in  $\text{Ho}^{3+}\text{-Yb}^{3+}$  co-doped  $\text{CaMoO}_4$ . The thermally coupled  $^5\text{G}_6/^5\text{F}_1$  energy levels were studied in ref. 67 and 68. A small  $\delta$  value of 8.2% was obtained in  $\text{Ho}^{3+}\text{-Yb}^{3+}$  co-doped  $\text{CaWO}_4$ . The large  $\delta$  value may be ascribed to the lack of correction on eqn (11). If eqn (11) is used directly to fit  $\Delta E_f$ , energy transfers from the TCL to the other excited states will be neglected. In fact, when the  $\text{Ho}^{3+}$  concentration is high enough, the cross-relaxation process,  $^5\text{S}_2 + ^5\text{I}_8 \rightarrow ^5\text{I}_4 + ^5\text{I}_7$ , occurs frequently among  $\text{Ho}^{3+}$  ions.<sup>13</sup>

### 3.4 Optical thermometry based on $\text{Nd}^{3+}$ doped phosphors

Among the rare-earth ions,  $\text{Nd}^{3+}$  has a  $4f^3$  electronic configuration, and is reported as one of the most efficient activated ions to obtain laser emissions. It is significant work to study the optical thermal sensing of  $\text{Nd}^{3+}$  doped phosphors, since the temperature of the laser crystals can be monitored through analyzing the temperature dependent spectrum of  $\text{Nd}^{3+}$ .  $\text{Nd}^{3+}$  has three couples of adjacent thermally coupled levels,  $^4\text{F}_{5/2}/^4\text{F}_{3/2}$  (TCL1),  $^4\text{F}_{7/2}/^4\text{F}_{3/2}$  (TCL2), and  $^4\text{F}_{7/2}/^4\text{F}_{5/2}$  (TCL3). As shown in Fig. 12, the mechanism of optical thermometry of  $\text{Nd}^{3+}$  doped phosphors is up-conversion and energy transfer from  $\text{Yb}^{3+}$  to  $\text{Nd}^{3+}$  assisted with phonons.<sup>69–71</sup> It needs one 532 nm green photon to populate the thermally coupled  $^4\text{F}_{5/2}$  and  $^4\text{F}_{3/2}$  energy levels through non-radiative relaxation. Upon 980 nm laser irradiation,  $\text{Yb}^{3+}$  absorbs infrared photons with the generation of  $^2\text{F}_{7/2} \rightarrow ^2\text{F}_{5/2}$  upward transitions. Subsequently, it donates the energy to the adjacent  $\text{Nd}^{3+}$  through a phonon assisted energy transfer, with  $\text{Yb}^{3+}$  dropping back to its  $^2\text{F}_{7/2}$  ground state. This promotes the

$\text{Nd}^{3+}$  ions to their excited state,  $^4\text{F}_{3/2}$ . The  $^4\text{F}_{5/2}$  energy level is populated by thermal excitation. The fluorescence intensity ratio,  $I_U/I_L$ , is dependent strongly on temperature, and is used to study the optical temperature sensing properties.

Recently, based on down-conversion luminescence emissions centered at 810 nm and 880 nm of  $\text{Nd}^{3+}$  ions, optical temperature sensing in the temperature range from 300 K to 850 K was studied by Rodríguez *et al.* in  $\text{Nd}^{3+}$  doped phosphate glass under 532 nm excitation.<sup>69</sup> Using a 980 nm diode laser as an excitation source, the temperature sensing properties based on the infrared luminescence emissions at 755 nm, 805 nm, and 872 nm of  $\text{Nd}^{3+}/\text{Yb}^{3+}$  co-doped  $\text{CaWO}_4$  powders has been discussed by Xu *et al.* in the temperature range from 303 K to 873 K,<sup>70</sup> as shown in Fig. 13(a). Three TCL, such as TCL1, TCL2, and TCL3, were proven by using the fluorescence intensity ratio technique, as shown in Fig. 13(b). Similarly, the temperature sensing properties based on near infrared emissions from TCL1, TCL2, and TCL3 of  $\text{Nd}^{3+}/\text{Yb}^{3+}$  co-doped oxyfluoride glass ceramic containing  $\text{PbF}_2$  nanocrystals were studied by Xu *et al.* as a function of temperature in the range of 303–623 K,<sup>71</sup> as shown in Fig. 13(c) and (d). In Table 5 we summarize, for the sake of comparison, the different performance parameters of  $\text{Nd}^{3+}$  doped and  $\text{Nd}^{3+}\text{-Yb}^{3+}$  co-doped phosphors.

The  $\Delta E_f$  and  $\Delta E_m$  values between the TCL of  $\text{Nd}^{3+}$  ions and the corresponding  $\delta$  values are calculated in Fig. 14 through analyzing ref. 69–71. The temperature dependent emissions at 810 nm and 880 nm from TCL1 were studied in  $\text{Nd}^{3+}$  doped phosphate glass (ref. 69), and a small  $\delta$  value of 2.55% was obtained. The temperature dependent emissions at 755 nm, 805 nm, and 872 nm from TCL1, TCL2, and TCL3 were studied in the  $\text{Nd}^{3+}\text{-Yb}^{3+}$  co-doped  $\text{CaWO}_4$  powders (ref. 70). The  $\delta$  values corresponding to TCL1, TCL2 and TCL3 are calculated to be 6.58%, 5.34%, and 12.03%. The temperature dependent emissions at 750 nm, 803 nm, and 863 nm from TCL1, TCL2, and TCL3 were studied in the  $\text{Nd}^{3+}\text{-Yb}^{3+}$  co-doped oxyfluoride glass ceramic (ref. 71). The corresponding  $\delta$  values are calculated to be 8.69%, 27.45%, and 26.67%. If TCL1 and TCL3 are used to study the optical temperature sensing properties, we can get small and large  $\delta$  values, respectively. It may be ascribed to the actual deviation of eqn (6) induced by the cross-relaxation process between the  $\text{Nd}^{3+}$  ions,  $^4\text{F}_{7/2} + ^4\text{I}_{9/2} \rightarrow ^4\text{F}_{3/2} + ^4\text{I}_{11/2}$ .<sup>13</sup>

### 3.5 Optical thermometry based on $\text{Dy}^{3+}$ doped phosphors

The  $\text{Dy}^{3+}$  ion with a  $4f^9$  electronic configuration has a couple of adjacent TCL,  $^4\text{I}_{5/2}$  and  $^4\text{F}_{9/2}$ . As shown in Fig. 15, the mechanism of optical thermometry based on luminescence of  $\text{Dy}^{3+}$  ions is down-conversion induced by ultraviolet excitation.<sup>72,73</sup> Under ultraviolet excitation,  $\text{Dy}^{3+}$  ions are excited directly to the higher excited states by the GSA transition. The ions in the excited states relax to the next lower energy level,  $^4\text{I}_{5/2}$  and  $^4\text{F}_{9/2}$ , through non-radiative relaxation. Three emissions with the intensities  $I_U$ ,  $I_L$ , and  $I'_L$  occur by the radiative transitions from  $^4\text{I}_{5/2}/^4\text{F}_{9/2}$  to  $^6\text{H}_{13/2}$  and  $^6\text{H}_{15/2}$ . The luminescence intensity ratios,  $I_U/I_L$  and  $I_U/I'_L$ , change with temperature, and are the precise evaluation scale of optical temperature sensing.

Optical temperature sensing of the  $\text{Dy}^{3+}$  doped  $\text{BaYF}_5$  nanoparticles was studied by Cao *et al.* in the temperature range from 293 K to 773 K.<sup>72</sup> The temperature dependent fluorescence spectra of  $\text{Dy}^{3+}$  ions under excitation of a 355 nm laser in Fig. 16(a) show that the intensity of the 455 nm emission increased with a temperature increase, and the intensity of the 478 nm emission was independent of temperature. The temperature dependent luminescence intensity ratio of the 455 nm and 478 nm emissions was fitted in Fig. 16(b), in which the line fitted very well with the experimental data at a temperature higher than 380 K. By analyzing the temperature dependent 455 nm and 481 nm blue emissions, Boruc *et al.* demonstrated optical temperature sensing of  $\text{Dy}^{3+}$  doped  $\text{Y}_4\text{Al}_2\text{O}_9$  crystals.<sup>73</sup> Temperature dependent emission spectra in Fig. 16(c) show that the intensity of the 455 nm emission increased with a temperature increase, and the intensity of the 481 nm emission decreased with a temperature increase. Temperature dependent luminescence intensity ratios of two blue emissions in Fig. 16(d) show different sensitivity values in two temperature ranges, such as  $296\text{ K} < T < 573\text{ K}$  and  $573\text{ K} < T < 973\text{ K}$ . The performance parameters of  $\text{Dy}^{3+}$  doped  $\text{BaYF}_5$  and  $\text{Y}_4\text{Al}_2\text{O}_9$  are summarized in Table 6. To value the fitting accuracy,  $\Delta E_f$  and  $\Delta E_m$  between the TCL of  $\text{Dy}^{3+}$  ions and the corresponding  $\delta$  values are calculated through analyzing ref. 72 and 73. The  $\delta$  value for the  $\text{Dy}^{3+}$  doped  $\text{BaYF}_5$  nanoparticles is 1.09%. For the  $\text{Dy}^{3+}$  doped  $\text{Y}_4\text{Al}_2\text{O}_9$  crystals, the  $\delta$  value is 18.8% at the temperature range  $296\text{ K} < T < 573\text{ K}$ , and it is 11.8% at the temperature range  $573\text{ K} < T < 973\text{ K}$ . It is accurate to use the  $\text{Dy}^{3+}$  doped  $\text{BaYF}_5$  nanoparticles to evaluate the scale of optical temperature sensing.

### 3.6 Optical thermometry based on $\text{Eu}^{3+}$ doped phosphors

Among the rare-earth ions,  $\text{Eu}^{3+}$  has a  $4f^6$  electronic configuration, and is reported as one of the most efficient activated ions to realize red emission in light emitting diodes (LEDs).<sup>74–76</sup> It is very significant to study optical thermal sensing of  $\text{Eu}^{3+}$  doped phosphors, since the temperature of LEDs can be monitored through analyzing the temperature dependent spectrum of  $\text{Eu}^{3+}$ .  $\text{Eu}^{3+}$  has a couple of adjacent thermally coupled levels,  $^5\text{D}_1$  and  $^5\text{D}_0$ . As shown in Fig. 17, the mechanism of optical thermometry based on luminescence of the  $\text{Eu}^{3+}$  ion is down-conversion induced by ultraviolet excitation.<sup>77–80</sup> Under ultraviolet excitation,  $\text{Eu}^{3+}$  ions are excited directly to the higher excited states by the GSA transition, and then relax to the next lower energy levels,  $^5\text{D}_1$  and  $^5\text{D}_0$ , through non-radiative relaxation. Two emissions with the intensity of  $I_U$  and  $I_L$  occur by the radiative transitions from  $^5\text{D}_1/^5\text{D}_0$  to  $^7\text{F}_J$  ( $J = 0, 1, 2, 3, 4, 5, 6$ ). At high temperature, for high  $\text{Eu}^{3+}$  concentration doped phosphors, the increase in the  $^5\text{D}_1$  emissions may be attributed to a thermally assisted cross-relaxation (CR) process with the involvement of two  $\text{Eu}^{3+}$  ions initially in the excited  $^5\text{D}_0$  and the  $^7\text{F}_3$  state.<sup>77</sup>

Meert *et al.* reported that the luminescence intensity ratios of the  $^5\text{D}_1$  to  $^5\text{D}_0$  emissions of the  $\text{Eu}^{3+}$  doped  $\text{CaEu}_2(\text{WO}_4)_4$  scheelites changed in an exponential form with a temperature increase,<sup>77</sup> as shown in Fig. 18(a). The temperature dependent

luminescence of spherical  $\text{NaEuF}_4$  phosphors with different particle sizes was studied by Tian *et al.*<sup>78</sup> Fig. 18(b) shows the temperature dependent emission spectra of the  $\text{NaEuF}_4$  spheres of 300 nm under 394 nm excitation.  $\Delta E_f$  was calculated by fitting the temperature dependent emission intensity ratios of the  $^5\text{D}_1/^5\text{D}_0$  emissions. The sensitivity values of the spherical  $\text{NaEuF}_4$  phosphors decreased with the size increase from 100 nm to 700 nm, and a sensitivity up to  $0.43\% \text{ K}^{-1}$  was achieved when the particle size was 100 nm. The temperature dependent luminescence spectrum of  $\text{Sr}_2\text{CeO}_4:\text{Eu}^{3+}$  was studied by Shi *et al.*, as shown in Fig. 18(c). The integrated luminescence total intensity of the  $^5\text{D}_0$  emissions at 591 nm, 616 nm, and 654 nm showed a line change relation in the temperature range from 373 K to 573 K.<sup>79</sup> Luminescence temperature sensing was studied by Nikolić *et al.* in  $\text{Eu}^{3+}$  doped  $\text{TiO}_2$  nanoparticles over a temperature range of 307–533 K.<sup>80</sup> The 438 nm blue emission associated with the trap emission of the  $\text{TiO}_2$  host and the emission peak of the  $\text{Eu}^{3+}$  ions at 613 nm are observed in Fig. 18(d) under continuous excitation at a wavelength of 360 nm. The temperature dependent luminescence intensity ratio of the 438 nm and 613 nm emissions was suitable to be used for temperature sensing. In Table 7 we summarize, for the sake of comparison, the different performance parameters of  $\text{Eu}^{3+}$  doped phosphors.

To value the fitting accuracy,  $\Delta E_f$  and  $\Delta E_m$  between the TCL of the  $\text{Eu}^{3+}$  ions and the corresponding  $\delta$  values are calculated through analyzing ref. 77 and 78. The  $\delta$  value for the  $\text{Eu}^{3+}$  doped  $\text{CaEu}_2(\text{WO}_4)_4$  scheelites is 38.7%, and the  $\delta$  value for the spherical  $\text{NaEuF}_4$  phosphors is 2.5%. As for the  $\text{Eu}^{3+}$  doped  $\text{TiO}_2$  and  $\text{Sr}_2\text{CeO}_4$ , the values of  $\delta$  are difficult to give, due to the incomplete data in ref. 76 and 77.

## 4. Conclusion and challenges

In summary, we have presented a detailed review of the  $\text{Er}^{3+}$ ,  $\text{Tm}^{3+}$ ,  $\text{Ho}^{3+}$ ,  $\text{Nd}^{3+}$ ,  $\text{Dy}^{3+}$ , and  $\text{Eu}^{3+}$  doped phosphors to date for the achievement of optical temperature sensing from the analysis of temperature dependent spectra. It has been shown that many phosphors with small  $\delta$  values can be used as basic light emitting materials for optical thermometry. In the case of  $\text{Er}^{3+}$ ,  $\text{Tm}^{3+}$ ,  $\text{Ho}^{3+}$ ,  $\text{Nd}^{3+}$ , and  $\text{Dy}^{3+}$  doped phosphors ( $\Delta E_m \leq 900 \text{ cm}^{-1}$ ) thermal equilibrium can be assumed because the population of the thermally coupled levels follows a Boltzmann distribution. Beyond this limit changes in the emission intensities are due to the effect of “coupling/decoupling” at higher or lower temperatures. As in the case of the  $\text{Eu}^{3+}$  doped phosphors, the  $^5\text{D}_1$  and  $^5\text{D}_0$  levels ( $\Delta E_f \sim 1725 \text{ cm}^{-1}$ ) are thermally decoupled at lower temperatures, with thermalization at higher temperatures. With the significant fundamental scientific and technological knowledge already accumulated by past research, optical thermometry represents an encouraging prospect in the development of new temperature sensors. However, from the point of view of application, many challenges remain in the areas of materials science and physical mechanisms.

One related challenge is the issue of trying to find novel phosphor materials that can be used as temperature sensors at temperatures greater than  $500^\circ\text{C}$  (773 K). Luminescence of rare-

earth ion doped phosphors mentioned in this review is very weak at temperatures greater than 773 K, due to intense thermal quenching. It is difficult to measure temperature change in the range of more than 500 °C through the conventional optical temperature sensing technology. New phosphors with low thermal quenching rates and high luminescence efficiency should be synthesized cheaply. Instead of rare-earth ions, new luminescence centers with high luminescence intensity at high temperature (>773 K) may be excellent candidate activated ions in novel fluorescence sensors.

Another main challenge facing the field of optical thermometry is the lack of understanding of the physical mechanisms determining the transitions between the electronic levels of both the host lattice and the rare-earth ions. A theoretical model describing thermal population between the TCL in eqn (6) is suitable to be used in the data fitting process only in the cases without energy transfer. In fact, energy transfer between the host and the rare-earth ions becomes active at high temperature, and energy transfer among the rare-earth ions becomes active at a high doping concentration. Thus, the error values of  $\Delta E_f$  for some phosphors mentioned in this review are large at more than 10%. In order to improve the calculation accuracy, a theoretical model describing the thermal population between the TCL in eqn (6) should be corrected for different phosphors. Additionally, the thermal affection of the excitation source should be considered in eqn (6).

Finally, we also believe that phosphor thermometry will be widely used in the future, not only for temperature detection in special environments, but also for some medical therapeutic equipment. In this sense, the development of phosphors with high thermal sensitivity and low thermal quenching of fluorescence will yield significant advances in the design of new optical temperature sensors. Colloidal phosphors with a random flightchain with biological tissues will be the target of innovation for disease diagnoses through the technology of optical temperature sensing.

## Acknowledgements

This work was supported by National Natural Science Foundation of China (11404171, 11374162), Natural Science Youth Foundation of Jiangsu Province (BK20130865), the Six Categories of Summit Talents of Jiangsu Province of China (2014-XCL-021), and the Natural Science Foundation of the Jiangsu Higher Education Institutions of China (14KJB430020).

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