Chemical Science



REVIEW

View Article Online
View Journal | View Issue



Cite this: Chem. Sci., 2025, 16, 2154

Recent advances and perspectives in synthetic applications of silylboronates as silyl radical precursors†

Zhihua Cai, D‡ab Qing-Qing Bu,‡ab Xi-Yu Wang,c Shengchao Yang,*ab Jian Zhou Dc and Jin-Sheng Yu D*c

Silylboronates, as powerful and versatile reagents, have been widely used in synthetic chemistry over the past few decades, due to their ability to incorporate silicon and boron atoms into organic molecules. With the rapid development of radical chemistry, the use of silylboronates as silyl radical precursors has recently become a research focus in organic synthesis. Significant achievements have been made in the synthetic applications of silylboronates as silyl radical sources for various C–Si and C–X bond forming transformations. This review summarizes these recent advances, discusses their advantages and limitations, and illustrates the synthetic chances still open for further research and applications in this emerging area.

Received 7th October 2024 Accepted 23rd December 2024

DOI: 10.1039/d4sc06777k

rsc.li/chemical-science

1 Introduction

Organosilicon compounds, owing to their unique chemical and physical properties, have found widespread applications in the fields of synthetic chemistry, pharmaceuticals, agrochemicals,

"School of Chemistry and Chemical Engineering, State Key Laboratory Incubation Base for Green Processing of Chemical Engineering, Shihezi University, Shihezi, Xinjiang 832003, P. R. China. E-mail: shengchao.yang@shzu.edu.cn

^bXinjiang Key Laboratory of Organosilicon Functional Molecules and Materials, Turpan, Xinjiang 838200, P. R. China

State Key Laboratory of Petroleum Molecular & Process Engineering, Shanghai Engineering Research Center of Molecular Therapeutics and New Drug Development, School of Chemistry and Molecular Engineering, East China Normal University, Shanghai 200062, P. R. China. E-mail: jsyu@chem.ecnu.edu.cn

- \dagger Dedicated to Professor Yong Tang on the occasion of his 60th birthday.
- \ddagger These authors contributed equally to this work.

and materials science. 1-3 Considerable efforts have hence been focused on synthesizing value-added organosilanes with structural diversity, and plenty of efficient silylation protocols have been accordingly developed by using various organosilicon reagents.4 Among them, the use of silylboronates,5 powerful and versatile reagents to incorporate silicon and/or boron atoms into organic molecules, has increased exponentially over the past few decades,6 since the pioneering discovery that the Si-B bond could oxidatively add to low-valent transition metals (e.g. platinum, palladium, nickel, etc.) by the groups of Ito7a,b and Tanaka.7c Several major strategies for Si-B bond activation, including oxidative addition, transmetalation, Lewis base activation, and carbenoid insertion, have been subsequently established, which enabled a large number of elegant transformations involving silylboronates, 6a,d such as 1,n-additions of C-C multiple bonds,8 1,2-additions of C=X bonds,9 1,4-



Zhihua Cai

Zhihua Cai received his BS and MS degrees in 2009 and 2012 from Shihezi University under the guidance of Prof. Lin He. He received his PhD degree from the University of Chinese Academy of Sciences in 2019 under the guidance of Prof. Gang Li. Then he joined Shihezi University, where he is now an associate professor. His research interests include organosilicon and benzyne chemistry.



Qing-Qing Bu

Qing-Qing Bu received his PhD degree from Goettingen University in 2018 under the guidance of Prof. Lutz Ackermann. Then she joined Shihezi University as a Young Scholar. Her research interests include the hydrosilylation reaction, the synthesis of silane coupling agents and functional silicone oil, and the application of calcium carbide.

additions of α,β -unsaturated carbonyl related and compounds,10 allylic and propargylic substitution,11 functionalization of strained-ring compounds or carbenoids and related compounds,12 cycloadditions of multiple-bond systems,13 C-H bond silylations, 14a,b cross-coupling of $C(sp^2)$ -X or $C(sp^3)$ -X bonds, 14c,d and others.15

In parallel with these impressive advances,6 the synthetic applications of silylboronates as silyl radical precursors are also gaining increasing attention and become a hot research topic in recent years, along with the renaissance of radical chemistry,16 because silyl radical species¹⁷ are valuable intermediates for accessing diverse organosilanes via regio- and chemoselective silylation in organic synthesis.

As a consequence, a variety of silyl radical generation methods from silvlboronates through photochemical, electrochemical or base activation strategies and their synthetic applications in various silvl radical involved transformations have been elegantly established for constructing C-Si or C-X bonds in the past few years. As illustrated in Fig. 1, these protocols can be classified into the following four categories,

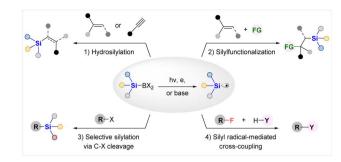


Fig. 1 Synthetic applications of silylboronates as silyl radical precursors.

according to the kind of reaction developed: (1) hydrosilylation of alkenes or alkynes; (2) silyl radical involved silylfunctionalization; (3) selective radical silylation via C-X bond cleavage; (4) silyl radical mediated cross-coupling reactions.

Despite great advances made in this emerging area, there lacks a timely review article to summarize these latest advances



Xi-Yu Wang

Xi-Yu Wang was born in Shaanxi, China. She received her BSc degree from Xi'an Medical University in 2020 and obtained her Master's degree from the Guizhou University of Traditional Chinese Medicine in 2023. Currently, she is pursuing her Doctoral degree under the guidance of Prof. Jin-Sheng Yu. Her thesis work focuses on the catalytic asymmetric construction of Si-stereogenic chiral organosilicon compounds.



Jian Zhou

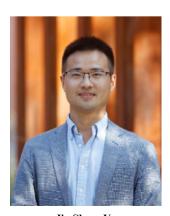
Jian Zhou obtained his PhD degree in 2004 from the Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, under the guidance of Prof. Yong Tang. After spending one year working as a postdoctoral fellow with Professor Shū Kobayashi at the University of Tokyo and three years with Professor Benjamin List at the Max-Planck-Institut für Kohlenforschung, he joined the Shanghai Key Laboratory of

Green Chemistry and Chemical Processes at ECNU as a Professor at the end of 2008. His research interests include the development of new chiral catalysts and asymmetric reactions for the construction of tetrasubstituted carbon stereocenters.



Shengchao Yang

Shengchao Yang received his BS degree from Central South University and Monash University (2 + 2 joint-supervision) in 2013. He received his PhD degree from Monash University in 2018. Then he joined Shihezi University as an associate professor in 2019 and was promoted to full professor in 2023. His research interests include organosilicon chemistry and silicon-based functional materials.



Jin-Sheng Yu

Jin-Sheng Yu received his PhD degree from East China Normal University (ECNU) in 2016 under the guidance of Prof. Jian Zhou. After two years as a JSPS postdoctoral fellow with Prof. Masakatsu Shibasaki at the Institute of Microbial Chemistry, he joined ECNU as a Zijiang Young Scholar and was later promoted to full professor. His research interests include organosilicon and organofluorine chemistry, asymmetric

catalysis, and biochemical pesticides.

in exploring the synthetic potential of silylboronates as silyl radical precursors in organic synthesis; in sharp contrast, several elegant reviews have been published by Oestreich and coworkers to introduce the applications of silylboronates as silyl anion equivalents. In light of this, together with the flourishing of Si–B chemistry, we feel that it is necessary to present a review article that focuses on elucidating the latest technological innovations and the reactions for synthetic applications of silylboronates as silyl radical precursors, outlining the remaining synthetic opportunities, thereby facilitating the researchers with some references and inspiration to develop more efficient approaches for silyl radical generation from Si–B agents and diverse synthetic applications of the thus obtained silyl radicals. In this review, we will introduce the advances according to the classification shown in Fig. 1.

2 Hydrosilylation of alkenes or alkynes with silylboronates

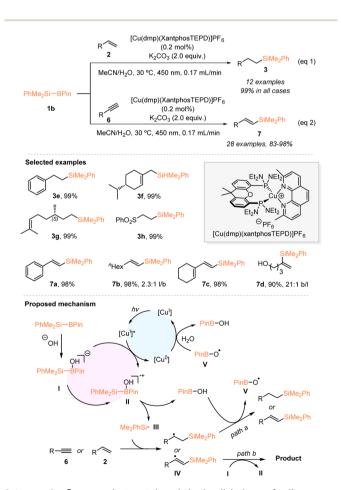
Hydrosilylation of alkenes or alkynes is an important and powerful way of preparing organosilicon compounds in industry and the laboratory.¹⁸ With the rapid development of radical chemistry,¹⁶ silyl radical-mediated hydrosilylation has been extensively explored in the past decade and has been one of the hot research areas in organic synthesis.^{17a,b} In this context, the use of silylboronates as silyl radical precursors for developing hydrosilylation has recently received increasing attention, since the seminal work by Ito, who demonstrated the generation of silyl radicals from a silylaminoboronate *via* homolysis under UV irradiation.¹⁹ Accordingly, a series of hydrosilylation reactions of alkenes or alkynes with silylboronates have been developed by using photochemical or electrochemical methods, allowing the access of various silylated alkanes or alkenes.

Early in 2000, the Ito group pioneered the generation of silyl radicals from organosilylboronate by using a photochemical method. Under UV irradiation of a high pressure Hg lamp, a dimethylphenylsilyl radical was effectively generated from $PhMe_2Si-B(N^iPr_2)_2$ (1a) via homolytic photolysis of the Si-B bond, which was confirmed by trapping experiments with

Scheme 1 Photochemically induced silyl radical hydrosilylation of alkenes with $PhMe_2Si-B(N^iPr_2)_2$ (1a).

TEMPO (Scheme 1).¹⁹ The thus generated organosilyl radical enabled the regioselective hydrosilylation of monosubstituted alkenes 2 and delivered the desired organosilanes 3 with moderate yields. Interestingly, under UV irradiation, PhMe₂Si–B(NⁱPr₂)₂ (1a) could promote the radical polymerization of vinyl acetate, methyl methacrylate and acrylate, affording the corresponding polymers containing a dimethylphenylsilyl at the polymer termini. Moreover, the radical pathway was confirmed by the experimental fact that no polymerization occurred in the presence of TEMPO or without irradiation.

In 2021, by using organosilylboronate as the radical precursor, Poisson and coworkers developed a copper-photocatalyzed hydrosilylation of alkenes under continuous flow (eqn (1), Scheme 2).²⁰ Under blue LED irradiation, the use of 0.2 mol% of their developed [Cu(dmp)(XantphosTEPD)]PF₆ effectively enabled the hydrosilylation of various monosubstituted alkenes 2 with the Suginome reagent PhMe₂Si-Bpin (1b), in the presence of 2 equiv. K₂CO₃, which afforded the desired silylated alkanes 3 with 99% yields in all cases. Furthermore, the authors established the hydrosilylation of alkynes 6 with PhMe₂Si-Bpin (1b) catalyzed by a Cu photocatalyst under blue LED irradiation and continuous flow conditions, allowing access to a variety of silylated alkenes 7 with excellent yields and good to excellent linear selectivities



Scheme 2 Copper-photocatalyzed hydrosilylation of alkenes or alkynes with $PhMe_2Si-Bpin$.

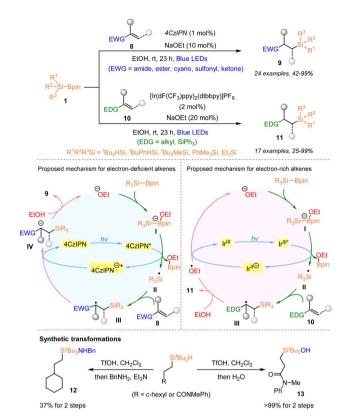
species or chain propagation.

(eqn (2), Scheme 2). Based on the mechanistic studies, a possible reaction mechanism is proposed in Scheme 2. A borate species I was first generated from PhMe₂Si-Bpin under the action of the Lewis base hydroxide, which underwent oxidation with an excited Cu photocatalyst [Cu^I]* to give the radical cation II. Subsequently, the Si-B bond cleavage of intermediate II delivered the dimethylphenylsilyl radical III, which then reacted with alkenes or alkynes. The thus formed β -silylated carbon-centered radical species IV delivered the desired products after undergoing H-abstraction from HO-Bpin

One year later, Ohmiya, Sumida, and coworkers described a visible-light-driven organosilyl radical generation method from silylboronates (Scheme 3).²¹ By using 1 mol% 4CzlPN, they realized the hydrosilylation of alkenes 2 with organosilylboronates 1 under blue LED irradiation, in the presence of 10 mol% Lewis base DMAP, affording the desired silylated products 3 in 30–93% yields. Notably, the corresponding deuterated organosilane 3n was achieved with 48% yield and 95% deuterium incorporation, when methanol-d₄ was used instead of MeOH.

In 2023, Tanaka and Nagashima reported a photocatalytic protocol for in situ generation of both secondary and tertiary organosilyl radicals from silylboronates via one-electron oxidation of ate complexes of silylboronates and alkoxide co-catalysts (Scheme 4).22 By employing such an "anionic SET strategy", highly efficient hydrosilylation and deuterosilylation of both electron-deficient alkenes 8 and electron-rich alkenes 10 with tertiary or secondary silylboronates 1 were achieved. It was found that a series of electron-deficient alkenes 8 reacted well with silylboronates by using a combination of 1 mol% 4CzlPN and 10 mol% NaOEt, under blue LED irradiation and afforded the corresponding organosilanes 9 with 42-99% yields, whilst the use of [Ir(dF(CF₃)ppy)₂(dtbbpy)]PF₆ (2 mol%) and NaOEt (20 mol%) proved to be more efficient for the hydrosilylation of electron-rich alkenes 10, furnishing the desired products 11 with 25-99% yields. Notably, such a catalytic system was suitable for the generation of secondary silyl radicals, although the use of secondary silyl radicals has been problematic in silylation

Scheme 3 Alkene hydrosilylation with silylboronate catalyzed by 4CzIPN.



Scheme 4 Organophotocatalytic hydrosilylation of electron-rich and -deficient alkenes with silylboronates.

due to their instability. As shown in Scheme 4, the plausible reaction mechanism was proposed according to the experimental results and DFT calculations. In addition, the elaboration of the product *via* the functionalization of the Si–H bond further highlighted the synthetic usefulness of this method.

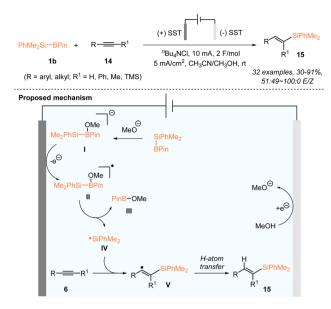
Almost simultaneously, Zhang, Wang, and coworkers reported a visible light-induced organophotocatalytic hydrosilylation of electron-deficient alkenes 8 with organosilylboronates 1 (Scheme 5).23 In the presence of 3 mol% 4CzlPN, a variety of hydrosilylation products 9 were obtained in 30-85% yields, under the irradiation of 24 W blue LEDs (eqn (1), Scheme 5). The necessity of both a photocatalyst and visible light was demonstrated by control experiments, since no target was detected in the absence of a photocatalyst or in darkness. The mild reaction conditions, broad substrate scope, and late-stage functionalization of three plant terpenoids further highlighted its practicability. A possible reaction mechanism was proposed, in which the in situ generated redox-active complex I from silvlboronate and methanol via $O \rightarrow$ B coordination underwent a SET process with the photocatalyst in a photoexcited state (PC*) to deliver the silyl radical II and reduced PC' species, accompanied by the formation of MeO-Bpin (III). Subsequently, the formed silvl radical II reacted with alkenes to afford the desired products after photoreduction and protonation, along with the regeneration of the photocatalyst 4CzlPN. Moreover, this strategy was successfully extended to the hydrosilylation of electron-rich alkenes 10, by slightly tuning the reaction conditions (eqn (2), Scheme 5). Notably, the authors

Scheme 5 Photochemically induced homolytic cleavage of Si-B bonds.

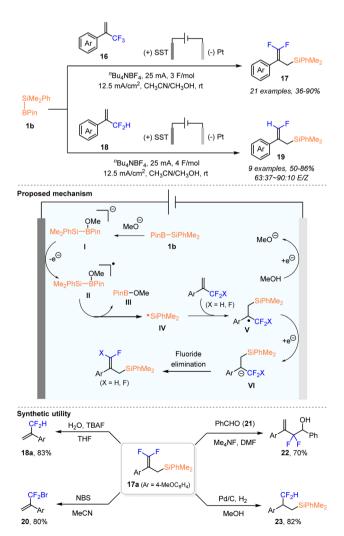
explored the hydrosilylation of alkynes with PhMe₂Si–Bpin, under the catalysis of 3 mol% 4CzlPN and blue LED irradiation, allowing the access of monosilylated alkenes *via* hydrosilylation or bissilylated alkanes *via* bis-silylation depending on the amount of PhMe₂Si–Bpin employed.

In addition to photocatalysis, the generation of silyl radicals from silylboronates proves to be feasible under electrochemical conditions. The electrochemical hydrosilylation of both alkenes and alkynes with PhMe₂Si–Bpin (**1b**) has been accordingly established. In 2022, the Poisson group reported the electrochemical generation of silyl radicals from silylboronates for the first time and realized the electrochemical hydrosilylation of alkynes **14** with PhMe₂Si–Bpin (Scheme 6).²⁴

Various terminal or internal alkynes 14 worked smoothly, in an undivided cell using stainless steel electrodes at both the anode and the cathode, at a constant current of 5 mA and a total charge of 2 F mol⁻¹ in a 0.1 M solution of ⁿBu₄NCl in a mixed solvent of MeCN/MeOH, delivering a series of monosilylated alkenes 15 in 30–91% yields with a 51 : $49 \sim 100 : 0 E/Z$ ratio. The mechanistic study suggested the involvement of a dimethylphenylsilyl radical in this alkyne hydrosilylation. A plausible reaction mechanism is proposed in Scheme 6. The methoxide (MeO) anion, generated in situ from the reduction of MeOH at the cathode, first reacted with PhMe2Si-Bpin to give a borate species I, which underwent anodic oxidation to give the radical species II. The silyl radical IV was subsequently generated from II, accompanied by the production of MeO-Bpin. The addition of the silyl radical to alkyne substrates 14 produced the reactive vinyl radical V, which delivered the targets 15 after undergoing H abstraction from MeOH.



Scheme 6 Electrochemical hydrosilylation of alkynes.



Scheme 7 Electrochemical synthesis of gem-difluoro- and γ -fluoroallyl silanes.

Review Chemical Science

In the same year, Poisson and coworkers developed a facile electrochemical protocol for the synthesis of *gem*-difluoro and γ -fluoroallyl silanes 17 and 19 through a silyl radical mediated transformation between α -trifluoromethylstyrenes 16 or α -difluoromethylstyrenes 18 and PhMe₂Si-Bpin (Scheme 7).²⁵

It was found that the in situ electro-generated silyl radical reacted smoothly with a series of α -trifluoromethylstyrenes 16 to afford the desired gem-difluoroallyl silanes 17 with 36-90% yields, in an undivided cell using a stainless steel electrode at the anode and a Pt electrode at the cathode, in a constant current of 25 mA and a total charge of 3 F mol⁻¹. By slightly varying the reaction conditions, α-difluoromethylstyrenes 18 could be effectively converted into the corresponding γ-fluoroallyl silanes 19 with 50–86% yields and a 63:37 \sim 90:10 E/Zratio. A similar mechanism is proposed in Scheme 7. Additionally, the excellent functional group tolerance, diversity and versatility of the thus obtained fluorinated building blocks further highlighted the synthetic value of this method, as exemplified by the preparation of α-difluoroalkyl substituted styrenes 18a and 20, difluoromethylated homoallylic alcohol 22, and β -difluoromethyl silane 23.

Later in 2023, also by using an electrochemical method, Tang, Pan, and coworkers realized a highly efficient radical hydrosilylation of electron-withdrawing alkenes 8 with PhMe₂-Si-Bpin 1b (Scheme 8).²⁶ In an undivided cell using stainless steel electrodes at both the anode and the cathode, a large panel of electron-withdrawing alkenes 8 successfully reacted with reactive dimethylphenylsilyl radical IV, which were electrogenerated from PhMe₂Si-Bpin, enabling the formation of various hydrosilylated adducts 24 with 60–87% yields. This direct electrochemical hydrosilylation did not require

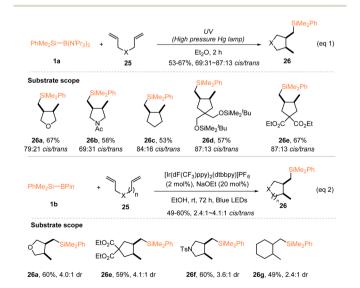
| Columbia | Characteristics |

Scheme 8 Electrochemical hydrosilylation of alkenes with $PhMe_2Si-Bpin$.

exogenous oxidants and catalysts, making it a green, efficient and sustainable approach for the synthesis of organosilanes. In addition, this method was applied for the late-stage hydrosilylation of natural product acrylate derivatives such as piperitol, Boc-L-prolinol, and cholesterol, allowing the access of the corresponding silylated natural product derivatives. The mechanistic study revealed that a silyl radical was involved in this electrochemical alkene hydrosilylation. As depicted in Scheme 8, a reaction mechanism was proposed, which was similar to that of Poisson's work shown in Schemes 6 and 7.

3 Silyl radical involved silylfunctionalization with silylboronates

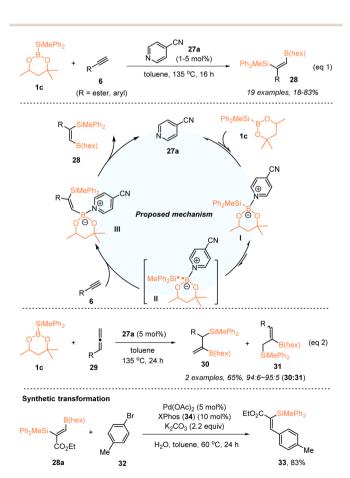
Along with the development of silyl radical hydrosilylations with silylboronates, a large number of radical silylfunctionalizations have been established using silvlboronates as the radical sources, for the diverse synthesis of functionalized organosilanes. Early in 2000, Matsumoto and Ito investigated a photochemical intramolecular cyclization of 1,6-dienes 25 with PhMe₂Si-B(NⁱPr₂)₂ (1a), during their study of the silyl radical-induced alkene hydrosilylation. In the presence of 1.2 equiv. PhMe₂Si-B(NⁱPr₂)₂ (1a), irradiation of various 1,6-dienes 25 with a high pressure Hg lamp afforded 5-exo-cyclized products **26** exclusively in 53–67% yields with $69:31 \sim 87:13$ cis/ trans selectivities (eqn (1), Scheme 9).19 It should be mentioned that different functional groups, such as ester, silyl ether, ether, and amide, were tolerated in this radical-induced silylative cyclization. 23 years later, by using a combination of 2 mol% [Ir(dF(CF₃)ppy)₂(dtbbpy)]PF₆ and 20 mol% NaOEt, Tanaka and Nagashima reported a similar radical silylative cyclization of oxygen-, tosylamide-, and malonate-linked 1,6-dienes or methylene-linked 1,7-diene 25 with PhMe2Si-Bpin 1b, allowing the preparation of five- or six-membered cycles featuring



Scheme 9 Photochemical silylative cyclization of dienes with silylboronates.

a PhMe $_2$ Si group with 49–60% yields and 2.4:1 \sim 4.1:1 *cis/trans* selectivities, under the irradiation of blue LEDs (eqn (2), Scheme 9). 22

In 2019, Suginome, Ohmura, and coworkers developed a regioselective syn-1,2-silaboration of terminal alkynes 6 with hexylene glycol-derived silylboronic ester PhMe₂Si-B(hex) (1c) by using a pyridine-based organocatalyst (Scheme 10).27 Under the catalysis of 1-5 mol% 4-cyanopyridine 27a, a series of aromatic alkynes and alkyl propiolates 6 underwent regio- and stereoselective addition with PhMe₂Si-B(hex), to afford (Z)-3boryl-2-silylacrylates 28 with 18-83% yields. Based on the initial experimental studies, a plausible reaction mechanism is proposed in Scheme 10. PhMe2Si-B(hex) was first activated using 4-cyanopyridine 27a via coordination with the B atom, leading to a homolytic cleavage of the Si-B bond to deliver a radical pair II consisting of a silyl radical and boron radical stabilized by 4-cyanopyridine. Subsequently, the radical pair was added to the C-C triple bond of alkynes 6 in a syn fashion, and the formed species III would deliver the targets 28, accompanied by the regeneration of 27a. The observed regioselectivity might stem from the fact that the more nucleophilic pyridine-boryl radical preferentially attacked the more electrondeficient alkyne terminus. Interestingly, terminal alkyl allenes were also suitable substrates in this 4-cyanopyridine catalysed with PhMe₂Si-B(hex) 1c, furnishing

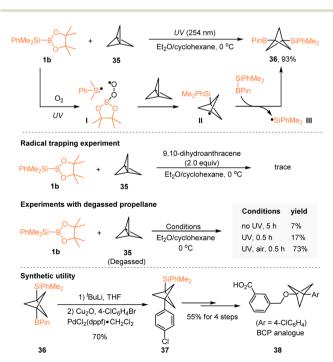


Scheme 10 Regioselective *syn-*1,2-silaboration of alkynes catalyzed by 4-cyanopyridine.

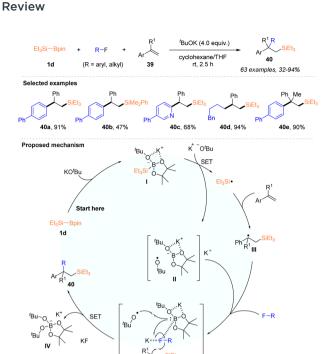
borylallylsilanes **30** as the major products. Additionally, the thus obtained silaboration adducts were useful synthetic intermediates in the synthesis of stereo-defined functionalized organosilanes via the elaboration of the boryl group, as exemplified by the preparation of (Z)-3-aryl-2-silylacrylate **33** through Suzuki–Miyaura coupling.

In 2020, Uchiyama, Kanazawa, and coworkers reported a photochemical silaboration of [1.1.1] propellane 35 with PhMe₂Si-Bpin 1b, which enabled the direct incorporation of B and Si functionalities onto the bicyclo[1.1.1]pentane (BCP) scaffold (Scheme 11).28 Under UV irradiation, the generated dimethylphenylsilyl radical I from PhMe2Si-Bpin 1b and O2 underwent the addition of [1.1.1] propellane 35 to give the BCP radical species II, which then reacted with another molecular PhMe₂Si-Bpin to deliver the product 36, accompanied by the formation of silyl radical III. Control experiments, together with radical trapping experiments, demonstrated the involvement of a radical chain mechanism in this silaboration, probably initiated by oxygen. Remarkably, the synthetic utility of this methodology was highlighted by its diverse elaborations toward various BCP scaffolds featuring a silyl group. For example, a BCP analogue 38 of the bioactive compound, as a potential bioisostere of the biaryloxy scaffold,29 was effectively prepared from 36 via six step reactions.

In 2021, the Shibata group disclosed a catalyst-free and tBuOK enabled regioselective carbosilylation of alkenes using silylboronate and organic fluorides, allowing the introduction of silyl and carbon groups across the C–C double bond of alkenes (Scheme 12). Under the action of 4 equiv. tBuOK , a variety of silylated alkanes 40 with β -tertiary or quaternary carbon centers were synthesized in 32–94% yields from styrenes or α -substituted styrenes 39, in the presence of Et_3Si -Bpin 1d



Scheme 11 Silaboration of [1.1.1]propellane with PhMe₂Si-Bpin.

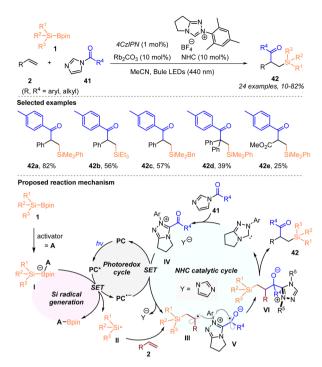


Scheme 12 ^tBuOK enabled regioselective carbosilylation of alkenes.

and aryl or alkyl fluorides. Moreover, an intramolecular carbosilylation of fluoroarenes featuring an alkenyl side chain with Et₃Si-Bpin 1d was also established. The mild conditions at room temperature, broad substrate scope and excellent chemoand site-selectivity demonstrated the practicability of this protocol. It should be mentioned that electron-deficient acrylates and acrylamides, as well as internal styrenes, were not suitable under this reaction system. As depicted in Scheme 12, the authors proposed a possible mechanism involving the silyl radical pathway, according to the initial mechanistic study. Et₃Si-Bpin 1d was first activated by ^tBuOK, and the formed intermediate I underwent a SET process with a t-butoxy anion (*BuO⁻) to deliver a triethylsilyl radical (*SiEt₃) via the cleavage of the Si-B bond, accompanied by the generation of boron species II. The addition of 'SiEt₃ to styrene gave rise to a carboncentered radical adduct III, which underwent radical reactions with fluorides, 'OBu^t, and borate anions, as shown in the transition state TS-I, where the C-F bond of fluorides was activated by both K⁺ and boron atoms. Finally, the desired carbosilylated adducts were obtained after forming a C-C bond, with the concomitant generation of a stable borate complex $[Bpin(O^tBu)_2]K$ **IV** and KF.

In 2022, by combining visible-light-induced silyl radical generation from silylboronates with radical NHC catalysis, Ohmiya, Sumida, and coworkers accomplished the acylsilylation of alkenes 2 with silylboronates 1 and acylimidazoles 41 (Scheme 13).²¹

In the presence of 4CzIPN (1 mol%), the NHC catalyst and Rb_2CO_3 (10 mol%, each), the radical relay three-component coupling process involving a silyl radical allows the introduction of acyl and organosilyl moieties into alkenes, affording a variety of β -acyl substituted silanes 42 with 10–82% yields,



Scheme 13 Radical relay three-component coupling of alkenes with silylboronates and acylimidazoles.

under blue LED irradiation. A possible reaction mechanism is illustrated in Scheme 13, in which the silyl radical II was first formed *via* SET oxidation of the activated silylboronates I with the excited photocatalyst (PC*). The radical addition of silyl radical II to alkenes 2 led to the generation of carbon-centered radical species III. The generated acyl azolium complex IV underwent a SET reduction with the reduced photocatalyst (PC'-) in the NHC catalytic cycle, to deliver the persistent ketyl radical V. Subsequently, radical-radical coupling between III and V occurred to give the intermediate VI, which produced the targets 42 with the regeneration of the NHC catalyst. Notably, as mentioned by the authors, such acylsilylation was difficult to achieve with HAT-promoted silyl radical generation methods.

Also in 2022, Li, Wang and co-workers described a metal-free 1,2-silylpyridylation of alkenes 43 with PhMe₂Si–Bpin (1b) and 4-cyanopyridines 27 *via* pyridine-mediated B–B and B–Si bond activation (Scheme 14).^{31a} In the presence of DABCO and B₂pin₂ (50 mol%, each), a wide range of alkenes 43 and substituted 4-cyanopyridines 27 were tolerated and provided various C4-silylalkylated pyridines 44 in 30–76% yields with excellent functional group compatibility.

More importantly, this method could be easily applied to the late-stage modification of complex bioactive molecules. A proposed mechanism with a silyl radical addition and sequential radical–radical coupling process is demonstrated in Scheme 14, in which the addition of the silyl radical to the alkene was the rate-limiting step. The author proposed that the 4-cyanopyridine might play dual roles in this 1,2-silylpyridylation, including the homolytic cleavage of the Si–B bond to give the silyl radical and pyridine-boryl radical, and the homolytic

Scheme 14 1,2-Silylpyridylation of aryl alkenes with PhMe₂Si-Bpin.

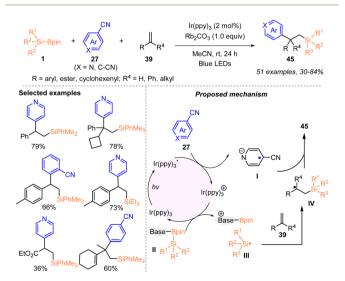
cleavage of the B–B bond of B_2pin_2 to produce a persistent pyridine-boryl radical for the subsequent radical–radical coupling process. Additionally, the presence of DABCO might inhibit the competitive Si–C bonding events of the silyl radical and 4-cyanopyridinyl-boryl radical.

Later on, the same group reported a visible-light catalyzed arylsilylation of alkenes **39** with silylboronates **1** and (hetero) aryl nitriles **27**, which enabled the diverse construction of valuable silicon-containing **1,1**-diaryl derivatives (Scheme **15**). Under blue LED irradiation, the arylsilylation of mono- or disubstituted alkenes **39** proceeded smoothly to deliver various silylated **1,1**-diaryl compounds **45** with 30-84% yields, in the presence of 2 mol% Ir(ppy)₃ and 1 equiv. Rb₂CO₃. Based on control experiments, the authors proposed a plausible mechanism, in which aryl nitriles underwent single-electron reduction

with excited $Ir(ppy)_3^*$ to generate aryl nitrile radical anion **I** and oxidizing $Ir(ppy)_3^+$. The activated silylborane **II** was oxidized by $Ir(ppy)_3^+$ to afford the silyl radical **III** after undergoing Si–B bond cleavage, along with the regeneration of $Ir(ppy)_3$. Subsequently, the addition of silyl radical **III** to alkenes led to the formation of carbon-centered radical **IV**, which coupled with aryl nitrile radical anion **I** to give the desired products **45**.

In 2024, Wang, Jia and coworkers disclosed a novel strategy to generate silyl radicals from silylboronates 1 via nucleohomolytic substitution of boron with aminyl radicals and applied such a strategy to realize an efficient silyl-oximation of alkenes 2 utilizing silylboronates 1 and N-nitrosamine 46 (Scheme 16).32 Under the irradiation of 425 nm LEDs, a variety of α-oximinoesters 47 bearing a silyl group were obtained in 31-87% vields. The radical-trapping experiment with TEMPO and the radical clock experiment with vinyl cyclopropane supported the involvement of a silvl radical intermediate in this visible-lightinduced catalyst-free silyl-oximation of alkenes. Accordingly, a plausible mechanism is proposed in Scheme 16, whereby photo-induced homolytic cleavage of the N-NO bond in Nnitrosamine produces an amino radical I and a persistent NO' radical. Subsequently, a nucleohomolytic substitution of silylboronates with amino radical I released a silyl radical III, which then reacted with an alkene to afford carbon-centered radical IV that was captured by the persistent NO' radical to deliver the target after tautomerization.

Very recently, Lian and coworkers developed a triphasic 1,2-hydroxysilylation of alkenes **39** with PhMe₂Si–Bpin **1b** under an oxygen atmosphere by mechanically piezoelectric catalysis through a single-electron-transfer (SET) pathway (Scheme 17).³³ A series of aromatic and aliphatic alkenes **39** worked well with PhMe₂Si–Bpin **1b** under an oxygen atmosphere to deliver the corresponding 1,2-hydroxysilylation products **48** with 32–94% yields. Moreover, this method could be used for the late-stage modification of drug-derived alkenes. Notably, this strategy pioneers the research in mechanic force-induced solid-liquid-



Scheme 15 Visible light-catalyzed arylsilylation of alkenes.

Scheme 16 Silyl-oximation of alkenes with silylboronate and *N*-nitrosamine.

Scheme 17 Triphasic hydroxysilylation of alkenes by mechanically piezoelectric catalysis

gas triphase reactions under ambient conditions and indicates that silylboronate can be transformed into a silyl radical by highly polarized Li₂TiO₃ particles and oxygen under ball-milling conditions for the first time. The excellent substrate scope with good functionality tolerance, simple operation, multiple recycles of the catalyst, and solvent-free conditions further highlighted the practicability of the current protocol. A silyl radical process involving reaction pathway was proposed according to the mechanistic studies, as illustrated in Scheme 17. The highly polarized Li₂TiO₃ particles were first generated by agitating Li₂TiO₃ particles via ball-milling, which reduced oxygen to give superoxide radicals ('O2"). Subsequently, the intermediate I, which was produced from superoxide radicals ('O₂⁻) via coordinating with the boron center of silylboronate 1b, underwent SET oxidation to afford the silyl radical species II, accompanied by the reduction of the oxidized Li₂TiO₃ particles. The trapping of the silyl radical II with alkenes delivered alkyl radical intermediate III, which then interacted with triplet oxygen to give the peroxyl radical species **IV** via the formation of the C-O bond. Finally, the target 48 was generated from intermediate IV after undergoing a HAT process and reduction with silylboronate.

In 2023, Tanaka and coworkers developed a chemo-, regioand stereo-selective dearomatic triple elementalization (carbosila-boration) of quinolines through the addition of organolithium and sequential visible light-induced silaboration, allowing the efficient synthesis of various carbo-sila-borated

Scheme 18 Dearomative triple elementalization (carbo-sila-boration) of quinolines

tetrahydroquinolines 50 that are expected to be versatile synthetic platforms (Scheme 18).34 The tandem sequence started with the addition of organolithium to quinolines 49. The generated lithium anilides then reacted with silylboronates to afford the anilide-silylborane ate complexes II, which underwent Si-B bond cleavage to form silyl radicals under blue LED irradiation, thus enabling the occurrence of carbo-sila-boration to deliver the desired products 50. Control experiments and DFT calculation studies revealed the involvement of a Si-B bond activation process in this sequence, in which visible light excitation of anilide-silylborane ate complexes II produced a silyl radical rather than a silyl anion. Notably, the synthetic utility of this method was highlighted by the potential of carbo-silaborated tetrahydroquinolines as synthetic platforms, as exemplified by the chemo- and stereospecific conversions of C-B/C-Si bonds to C-C, C-O, and C-H/D bonds. In addition, the asymmetric carbo-sila-boration of quinoline was explored by combining an asymmetric alkylation of quinoline with n-BuLi enabled by chiral ligand L1, delivering chiral carbo-sila-borated tetrahydroquinoline (2S,3R,4S)-50a.

Selective silylation *via* C-X bond cleavage

Aside from the radical silylfunctionalization of unsaturated hydrocarbons, e.g. alkenes or alkynes, the use of silylboronate as a silyl radical precursor has recently been applied for selective silylation via C-X bond cleavage. In 2023, Zhang, Wang, and employed visible-light-induced a photocatalytic strategy to realize a radical Minisci-type C-H silylation of N-heteroarenes with silylboronates 1 (Scheme 19).23

Scheme 19 Organophotocatalytic silylation \emph{via} C-H or C-I bond cleavage.

The use of 3 mol% 4CzlPN enabled the facile synthesis of sily-lated N-heteroarenes 51 with 48–72% yields, in the presence of TFA (1.0 equiv.), under 24 W blue LED irradiation. Moreover, the authors found that the silyl radicals generated *via* such an organophotocatalytic strategy could react with ethynylbenziodoxolones (EBXs) 52 through C–I bond cleavage, thus realizing the alkynylation of silylboronates 1 with EBXs 52 for the preparation of alkynylsilanes 53 with 56–82% yields.

A year later, Wang and Jia *et al.* also reported the preparation of synthetically useful alkynylsilanes by using silyl radicals generated from silylboronates *via* nucleohomolytic substitution of boron with aminyl radicals (Scheme 20).³² Under 425 nm LED irradiation, the alkynylation of silylboronates 1 with ethynyl phenyl sulfones (EPSs) 54 proceeded smoothly to afford a series of silylated alkynes 53 with 50–92% yields, in the presence of *N*-nitrosamine 46. The process for silyl radical generation was the same as the pathway of Scheme 16. The transformation of alkynylsilane 53a into 3-phenylindole further highlighted the utility of this protocol.

In 2023, Feng, Xue, and Liu accomplished a ^tBuONa enabled site-selective silylation of alkyl/aryl thioethers 57 with Et₃Si-Bpin (1d) *via* C(alkyl)–S bond cleavage (Scheme 21).³⁵

It was found that a series of primary or secondary alkyl aryl sulfides 57 were tolerated and afforded various alkylsilanes 3 with 40–98% yields, in the presence of 4.0 equiv. ^tBuONa. Et₃Si-Bpin was activated by ^tBuONa to form complex **I**, which further interacted with another ^tBuONa molecule to produce the active silyl radical *via* a single-electron transfer process. Subsequently,

Scheme 20 Visible light-induced alkynylation of silylboronates.

Scheme 21 Site-selective silylation of thioethers with ${\rm Et_3Si-Bpin}\ via$ C-S bond cleavage.

the resulting silyl radical prefers to work with the C(alkyl)–S bond, facilitating the C(alkyl)–S bond cleavage to yield alkylsilanes. Furthermore, the gram scale synthesis and late-stage silylation of bioactive molecules highlighted the usefulness of this protocol. Radical inhibition and radical clock experiments confirmed the involvement of silyl radical species in this reaction system.

Also in 2023, Cui, Cao, Shi, and coworkers reported a dearomatization silylation of benzofurans or furopyridines *via* silyl radical addition and subsequent C–O bond scission (Scheme 22).³⁶ Under the action of 2.0 equiv. ^tBuOK, a series of benzofurans or furopyridines without substituents at the C2 or C3 position worked well with Et₃Si–Bpin **1d** to deliver the ringopening vinylsilanes **61** with 43–92% yields. Benzofurans, featuring an aryl at the C2 position or a methyl at the C3 position, were tolerated as well and afforded the corresponding vinylsilanes with moderated to good yields.

It should be mentioned that benzofurans with a bulky *tert*-butyl group at the C2-position or an aryl group at the C3-position underwent^{1,5} Brook rearrangement following silyl radical addition and subsequent C–O bond scission, producing vinylphenoxyl silanes **63** with 80–93% yields. Based on initial mechanistic studies, the authors proposed a reaction mechanism, as shown in Scheme 22. The triethylsilyl radical EtSi was first generated *via* the reaction of Et₃Si–Bpin with a ^tBuO radical that was formed from ^tBuOK and O₂. The addition of EtSi to benzofurans or furopyridines proceeded to give the carboncentered radical species **I**, whilst the direct interaction of Et₃Si–Bpin with ^tBuOK led to the formation of complex **II**. Subsequently, radical species **I** underwent a SET process with

Me (R = H, aryl)

R 'BuOK (2.0 equiv.)

DME, rt

OH Me
(R = H, aryl)

Ar'

28 examples, 43-92%

Et₃Si—Bpin

1d

Ar'

Bu

OSIEt₃

(eq 1)

62

(2.0 equiv.)

63

(eq 2)

Scheme 22 Dearomatization radical silylation of benzofurans or furopyridines.

THF, rt

Trimethoxyl

611

2) DEAD, Ph₃P

3) PFA, AcOH, TFA

63% for 3 steps

65

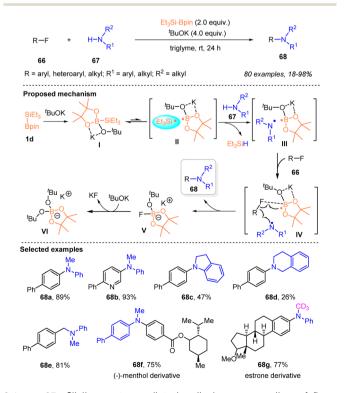
Doxenin analogi

complex II to afford intermediate III-A or III-B. Then III-A underwent β -elimination to give the product **61** after protonation, while III-B underwent sequential β -elimination and [1,5]-Brook rearrangement to furnish vinylphenoxyl silanes **63**. Moreover, the diverse product elaboration demonstrated the practicability of this protocol, as evidenced by the preparation of trimethoxyl oxyresveratrol **64** and doxepin analogue **65**.

5 Silyl radical mediated crosscoupling with silylboronates

With the application of silylboronates as silyl radical precursors for constructing organosilanes *via* C–Si bond forming reactions, silylboronates have also been identified as an effective medium for enabling C–F bond cleavage of organic fluorides to enable cross coupling reactions. In 2023, the Shibata group pioneered this study and established a transition metal-free silylboronate-mediated cross-coupling of organic fluorides **66** with amines **67**

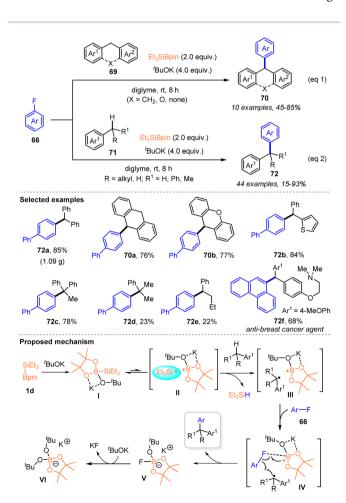
inert C-F bond activation under mild conditions (Scheme 23).37 A series of (hetero)aryl or alkyl fluorides reacted smoothly with cyclic or acyclic N-alkylanilines or secondary dialkylamines 67, in the presence of 2 equiv. Et₃Si-Bpin and 4 equiv. ^tBuOK, to afford various tertiary amines **68** with 18–98% yields. The coordination of Et₃Si-Bpin and ^tBuOK enabled the cross-coupling to proceed at room temperature, thus avoiding the high potential barriers associated with thermally induced S_N2 or S_N1 amination. A single-electron-transfer/radicalmediated defluorinative amination involving a frustrated radical pair chemistry mechanism is illustrated in Scheme 23, according to the mechanistic studies. The intermediate I was first generated from Et₃Si-Bpin with KO^tBu, which then produced a frustrated radical pair II that comprises Et₃Si' and a boron-radical species via the homolytic cleavage of the Si-B bond. The formed Et₃Si' underwent hydrogen abstraction from amines 67 to provide a frustrated radical pair III consisting of an amino radical and the boron radical species, along with the generation of HSiEt3. The frustrated radical pair III subsequently interacted with organic fluorides 66 via transition state IV, where the C-F bond was activated by the interaction between the F atom and B center, and the amino radical selectively attacked the carbon center of the C-F bond, to deliver the desired cross-coupling product, accompanied by the formation of stable $[Bpin(O^tBu)_2]K$ (VI). The significant feature of this transformation included the selective activation of the C-F bond of organofluorides by silvlboronate without affecting potentially cleavable C-O, C-Cl, C-H or C-N bonds and the CF₃ group, mild reaction conditions, broad substrate scope, and late-stage modification of fluorine-containing drugs.



Scheme 23 Silylboronate-mediated radical cross-coupling of fluorides and amines.

Shortly after, by employing the strategy of silyl radical-mediated cross-coupling of fluorides, the same group realized an efficient cross-coupling of aryl fluorides **66** with arylalkanes **69** or **71** enabled by a combination of Et₃Si-Bpin **1d** with ¹BuOK, allowing the diverse access of various triaryl- or diarylalkanes **70** or **72**, which are promising scaffolds for pharmaceuticals and functional materials, with moderate to excellent yields under very mild reaction conditions (Scheme 24).³⁸ A salient feature of this protocol was that the activation of the C-F and C-H bonds occurred at room temperature. The practicability was further highlighted by the fact that no transition metal and specialized ligands with high temperature, which are usually required in common cross-coupling reactions, were used in this method. A similar silyl radical-involved reaction mechanism is proposed in Scheme **24**, based on the ESR analysis and experimental studies.

Subsequently, they further utilized a silylboronate-mediated cross-coupling strategy to enable the cross coupling of alkyl fluorides and aryl alkanes for the formation of C(sp³)–C(sp³) bonds (Scheme 25).³9 Under the action of Et₃Si–Bpin with 'BuOK, a variety of alkyl fluorides **66** could react efficiently with mono- or diaryl alkanes **71** at 60 °C, producing diaryl or aryl alkanes **73** bearing a tertiary or quaternary carbon center in 22–98% yields. Moreover, alkyl chlorides, bromides, and iodides were also suitable substrates for this transformation. Although

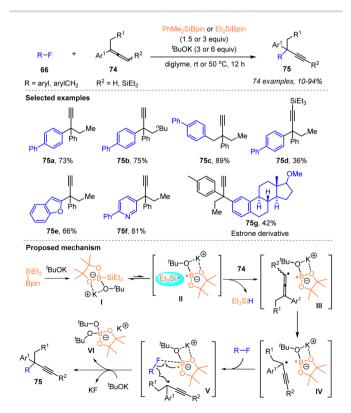


 $\begin{array}{ll} \textbf{Scheme 24} & \textbf{Silylboronate-mediated cross-coupling of benzylic C-H} \\ \textbf{with arylfluorides}. \end{array}$

Scheme 25 Silylboronate-mediated cross-coupling of alkylfluorides and aryl alkanes.

a radical-involving pathway was supported by radical trapping and ESR experiments, radical ring-opening experiments and DFT calculations suggested that an ionic process might occur in this cross-coupling. Notably, mechanistic studies using DFT calculations were conducted on silylboronate-mediated cross-coupling for the first time and supported the crucial role of diglyme in enhancing the reaction efficiency *via* encapsulation of potassium cations (Scheme 25).

More recently, Shibata and coworkers developed an efficient silylboronate-mediated cross-coupling between organic fluorides **66** and allenes **74** *via* a catalyst-free silyl radical relay strategy, which provided a facile protocol for the construction of α -alkynyl substituted all-carbon quaternary centers (Scheme 26).⁴⁰



Scheme 26 Radical cross-coupling of organic fluorides and allenes mediated by silylboronates.

Review Chemical Science

Under the action of ^tBuOK and Et₃Si-Bpin **1d** or PhMe₂Si-Bpin **1b**, a variety of aryl or benzyl fluorides **66** worked well with disubstituted allenes **74** to afford a library of coupling products **75** featuring an α-ethynyl-containing all-carbon quaternary center with up to 94% yields. The key to success was *in situ* radical rearrangement of the formed allenyl radicals to give bulky tertiary propargyl radicals. A salient feature of this cross-coupling was the ability of the *in situ* generated silyl radical to abstract a proton directly from the C(sp²)–H bond of allenes to form an allenyl radical, which was subsequently isomerized to a propargylic radical, as depicted in the proposed mechanism of Scheme **26**. Furthermore, mild reaction conditions without a transition metal, wide substrate scope, the late-stage functionalization of bioactive molecules and the modification of a liquid crystalline material demonstrated the practical utility of the current protocol.

6 Conclusion and outlook

The past few years have witnessed significant advancements in synthetic applications of silylboronates as silyl radical precursors, with the rapid development of radical chemistry. As summarized in this review article, a variety of C–Si or C–X bond forming transformations have been established for the diverse construction of organosilanes or other valuable molecules (*e.g.* tertiary amines, triaryl- or diarylalkanes and α -alkynyl substituted quaternary centers), including hydrosilylation of alkenes or alkynes; silyl radical involved silylfunctionalization; selective radical silylation via C–X bond cleavage; silyl radical mediated cross-coupling.

Despite these impressive achievements, there is still ample room for further exploration in synthetic applications of silylboronates as silyl radical precursors toward valuable chemicals. First, more successful and diverse transformations are needed to demonstrate the synthetic practicability and potential of silylboronates as silvl radical precursors, since most of the currently reexamples focus hydrosilylation ported on the silylfunctionalization. Second, the generation of silyl radical species from silylboronates mainly relies on the use of photochemical, electrochemical or stoichiometric base activation. Therefore, the development of new and efficient methods for the generation of silyl radicals from silylboronates will be highly desirable. Third, no asymmetric reactions using silylboronates as silyl radical sources have been reported, which provides another important direction to explore radical asymmetric transformations involving silylboronates for accessing chiral organosilanes, in particular, silicon-stereogenic chiral silanes. 41 With the continuous emergence of new silylboronate reagents, catalytic systems, and synthetic strategies, it can be anticipated that more and more attractive and facile silyl radical transformations involving silylboronates will be explored, which will greatly broaden the synthetic applications of silylboronates in the area of radical chemistry.

Data availability

No primary research results, software or code have been included and no new data were generated or analysed as part of this review.

Author contributions

All authors co-wrote the manuscript.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was financially supported by the National Natural Science Foundation of China (No. 22471073 and 22171087), the International Cooperation Project (No. DL2023047001L), the Major R&D projects (No. 2023B01019 and 2023AA004), the Key Laboratory Project of Shihezi City (No. 2023PT01), the Innovation Team Project of Shihezi City (No. 2023TD03), the Technology Innovation Project of Shanghai Municipal Agricultural Committee [No. HNK(T2023302)], the Innovation Program of Shanghai Municipal Education Commission (No. 2023ZKZD37), and the Ministry of Education (PCSIRT) and the Fundamental Research Funds for the Central Universities. We also acknowledge Prof. Kai Hong at East China Normal University for valuable discussion.

References

- 1 (a) S. E. Denmark and R. F. Sweis, Acc. Chem. Res., 2002, 35, 835–846; (b) T. Hiyama and M. Oestreich, Organosilicon Chemistry: Novel Approaches and Reactions, Wiley-VCH, Weinheim, Germany, 1st edn, 2019; (c) T. Komiyama, Y. Minami and T. Hiyama, ACS Catal., 2017, 7, 631–651.
- (a) A. K. Franz and S. O. Wilson, *J. Med. Chem.*, 2013, 56, 388–405; (b) R. Ramesh and D. S. Reddy, *J. Med. Chem.*, 2018, 61, 3779–3798; (c) E. Remond, C. Martin, J. Martinez and F. Cavelier, *Chem. Rev.*, 2016, 116, 11654–11684.
- 3 (a) L. Gai, J. Mack, H. Lu, T. Nyokong, Z. Li, N. Kobayashi and Z. Shen, *Coord. Chem. Rev.*, 2015, 285, 24–51; (b) E. A. Marro and R. S. Klausen, *Chem. Mater.*, 2019, 31, 2202–2211.
- 4 For a book: (a) Reagents for silicon-mediated organic synthesis (Handbook of Reagents for Organic Synthesis), ed. P. L. Fuchs, Wiley, 2013, For reviews:; (b) C. Chatgilialoglu, C. Ferreri, Y. Landais and V. Timokhin, Chem. Rev., 2018, 118, 6516-6572; (c) H. Ottosson and P. G. Steel, Chem.-Eur. J., 2006, 12, 1576–1585; (d) W. P. Weber, Silicon Reagents for Organic Synthesis, Springer, Berlin 1983; (e) L. A. Paquette, Science, 1982, 217, 793–800; (f) Z.-T. Ye, Z.-W. Wu, X.-X. Zhang, J. Zhou and J.-S. Yu, Chem. Soc. Rev., 2024, 53, 8546-8562; (g) X.-S. Hu, J.-S. Yu and J. Zhou, Chem. Commun., 2019, 55, 13638, For selected examples: ; (h) M. Liu, K. Dong, B. Xu, Z.-M. Zhang, Z. Wei and J. Zhang, Org. Chem. Front., 2024, **11**, 3821–3826; (*i*) X.-X. Zhang, Y. Gao, Y.-X. Zhang, J. Zhou and J.-S. Yu, Angew. Chem., Int. Ed., 2023, 62, e202217724; (j) W.-B. Wu, X. Yu, J.-S. Yu, X. Wang, W.-G. Wang and J. Zhou, CCS Chem., 2022, 4, 2140; (k) X.-S. Hu, J.-X. He, S.-Z. Dong, Q.-H. Zhao, J.-S. Yu and J. Zhou, Nat. Commun., 2020, 11, 5500; (l) B.-S. Mu, Y. Gao, F.-M. Yang, W.-B. Wu, Y. Zhang, X. Wang, J.-S. Yu and J. Zhou, Angew. Chem., Int.

Ed., 2022, **61**, e202208861; (*m*) C.-W. Lei, X.-Y. Wang, B.-S. Mu, J.-S. Yu, Y. Zhou and J. Zhou, *Org. Lett.*, 2022, **24**, 8364–8369; (*n*) W.-B. Wu, B.-S. Mu, J.-S. Yu and J. Zhou, *Chem. Sci.*, 2022, **13**, 3519–3525.

Chemical Science

- 5 For typical synthetic methods, see: (a) T. Takeuchi, R. Shishido, K. Kubota and H. Ito, *Chem. Sci.*, 2021, 12, 11799–11804; (b) R. Shishido, M. Uesugi, R. Takahashi, T. Mita, T. Ishiyama, K. Kubota and H. Ito, *J. Am. Chem. Soc.*, 2020, 142, 14125–14133; (c) T. A. Boebel and J. F. Hartwig, *Organometallics*, 2008, 27, 6013–6019; (d) M. Suginome, T. Matsuda and Y. Ito, *Organometallics*, 2000, 19, 4647–4649, Also see: ; (e) T. Ohmura and M. Suginome, *Bull. Chem. Soc. Jpn.*, 2009, 82, 29–49.
- 6 (a) M. Oestreich, E. Hartmann and M. Mewald, Chem. Rev., 2013, 113, 402–441; (b) J. R. Wilkinson, C. E. Nuyen, T. S. Carpenter, S. R. Harruff and R. Van Hoveln, ACS Catal., 2019, 9, 8961–8979; (c) W. Xue and M. Oestreich, ACS Cent. Sci., 2020, 6, 1070–1081; (d) J.-J. Feng, W. Mao, L. Zhang and M. Oestreich, Chem. Soc. Rev., 2021, 50, 2010–2073.
- 7 (a) M. Suginome, H. Nakamura and Y. Ito, Chem. Commun.,
 1996, 2777–2778; (b) M. Suginome, T. Matsuda,
 H. Nakamura and Y. Ito, Tetrahedron, 1999, 55, 8787–8800;
 (c) S. Onozawa, Y. Hatanaka and M. Tanaka, Chem. Commun., 1997, 1229–1230.
- 8 For selected recent examples, see: (a) M. Suginome, H. Nakamura and Y. Ito, Angew. Chem., Int. Ed. Engl., 1997, **36**, 2516–2518; (b) T. Ohmura, K. Oshima, H. Taniguchi and M. Suginome, J. Am. Chem. Soc., 2010, 132, 12194-12196; (c) T. Fujihara, Y. Tani, K. Semba, J. Terao and Y. Tsuji, Angew. Chem., Int. Ed., 2012, 51, 11487-11490; (d) Y. Tani, T. Fujihara, J. Terao and Y. Tsuji, J. Am. Chem. Soc., 2014, 136, 17706–17709; (e) Z.-T. He, X.-Q. Tang, L.-B. Xie, M. Cheng, P. Tian and G.-Q. Lin, Angew. Chem., Int. Ed., 2015, 54, 14815-14818; (f) T. Iwamoto, T. Nishikori, N. Nakagawa, H. Takaya and M. Nakamura, Angew. Chem., Int. Ed., 2017, 56, 13298-13301; (g) H. Sakaguchi, M. Ohashi and S. Ogoshi, Angew. Chem., Int. Ed., 2018, 57, 328-332; (h) Z. Liu, J. Chen, H.-X. Lu, X. Li, Y. Gao, J. R. Coombs, M. J. Goldfogel and K. M. Engle, Angew. Chem., Int. Ed., 2019, 58, 17068-17073; (i) Y. Gu, Y. Duan, Y. Shen and R. Martin, Angew. Chem., Int. Ed., 2020, **59**, 2061–2065; (j) T. R. Pradhan, M. Paudel, T. Feoktistova, P. H.-Y. Cheong and J. K. Park, Angew. Chem., Int. Ed., 2022, e202116154; (k) G. Wang, M. Wei, T. Liu, W. Jin, Y. Zhang, B. Wang, Y. Xia and C. Liu, Adv. Synth. Catal., 2022, 364, 909-913; (l) Q. Chen, Z. Li and Y. Nishihara, Org. Lett., 2022, 24, 385–389; (m) Z.-Y. Chen, M.-W. Yang, Z.-L. Wang and Y.-H. Xu, Org. Lett., 2023, 25, 5242-5247; (n) H. Moniwa, M. Yamanaka and R. Shintani, J. Am. Chem. Soc., 2023, 145, 23470-23477; (o) Y. Ozawa, H. Koriyama, Y. Shiratori and H. Ito, ACS Org. Inorg. Au, 2023, 3, 104–108; (p) Q. Li, Z.-L. Wang and Y.-H. Xu, Chin. Chem. Lett., 2023, 34, 108150; (q) C. Ding, Y. Ren, Y. Yu and G. Yin, Nat. Commun., 2023, 14, 7670.
- 9 For selected recent examples, see: (*a*) V. Cirriez, C. Rasson, T. Hermant, J. Petrignet, J. Díaz Alvarez, K. Robeyns and

- O. Riant, Angew. Chem., Int. Ed., 2013, 52, 1785–1788; (b) A. Hensel, K. Nagura, L. B. Delvos and M. Oestreich, Angew. Chem., Int. Ed., 2014, 53, 4964–4967; (c) Y. Huo, P. Shen, W. Duan, Z. Chen, C. Song and Y. Ma, Chin. Chem. Lett., 2018, 29, 1359–1362; (d) M. Takeda, A. Mitsui, K. Nagao and H. Ohmiya, J. Am. Chem. Soc., 2019, 141, 3664–3669; (e) K. Yabushita, A. Yuasa, K. Nagao and H. Ohmiya, J. Am. Chem. Soc., 2019, 141, 113–117; (f) Z.-Y. Zhao, M. Cui, E. Irran and M. Oestreich, Angew. Chem., Int. Ed., 2023, e202215032.
- 10 For selected recent examples, see: (a) M. O'Brien and A. H. Hoveyda, J. Am. Chem. Soc., 2011, 133, 7712-7715; (b) R. T. H. Linstadt, C. A. Peterson, D. J. Lippincott, C. I. Jette and B. H. Lipshutz, Angew. Chem., Int. Ed., 2014, 53, 4159-4163; (c) J. A. Calderone and W. L. Santos, Angew. Chem., Int. Ed., 2014, 53, 4154-4158; (d) H. Wu, J. M. Garcia, Haeffner, S. Radomkit, A. R. Zhugralin and A. H. Hoveyda, J. Am. Chem. Soc., 2015, 137, 10585-10602; (e) T. Kitanosono, L. Zhu, C. Liu, P. Xu and S. Kobayashi, J. Am. Chem. Soc., 2015, 137, 15422-15425; (f) M. Wang, Z.-L. Liu, X. Zhang, P.-P. Tian, Y.-H. Xu and T.-P. Loh, J. Am. Chem. Soc., 2015, 137, 14830-14833; (g) C. Rasson, A. Stouse, A. Boreux, V. Cirriez and O. Riant, Chem.-Eur. J., 2018, 24, 9234-9237; (h) W. Mao, W. Xue, E. Irran and M. Oestreich, Angew. Chem., Int. Ed., 2019, 58, 10723-10726; (i) Q. Li, Z.-L. Wang, H.-X. Lu and Y.-H. Xu, Org. Lett., 2022, 24, 2832-2836; (j) Y. Xiao, Z.-Y. Zhao, S. Kemper, E. Irran and M. Oestreich, Angew. Chem., Int. Ed., 2024, e202407056, For reviews, see: ; (k) X. Tang, L. Xie, Y. Chen, P. Tian and G. Lin, Chin. J. Org. Chem., 2016, 36, 2011–2023; (l) B. Han, W. Li, S. Chen, Z. Zhang, X. Zhao, Y. Zhang and L. Zhu, Chin. J. Org. Chem., 2023, 43, 555-572, and ref. 6.
- 11 For selected examples, see: (a) D. J. Vyas and M. Oestreich, Angew. Chem., Int. Ed., 2010, 49, 8513–8515; (b)
 L. B. Delvos, D. J. Vyas and M. Oestreich, Angew. Chem., Int. Ed., 2013, 52, 4650–4653; (c) R. Shintani, R. Fujie, M. Takeda and K. Nozaki, Angew. Chem., Int. Ed., 2014, 53, 6546–6549; (d) Z.-L. Liu, C. Yang, Q.-Y. Xue, M. Zhao, C.-C. Shan, Y.-H. Xu and T.-P. Loh, Angew. Chem., Int. Ed., 2019, 58, 16538–16542.
- 12 For selected examples, see: (a) M. Suginome, T. Matsuda and Y. Ito, J. Am. Chem. Soc., 2000, 122, 11015–11016; (b) H. Lee, J. T. Han and J. Yun, ACS Catal., 2016, 6, 6487–6490; (c) T. Yamamoto, R. Murakami, S. Komatsu and M. Suginome, J. Am. Chem. Soc., 2018, 140, 3867–3870; (d) L. Zhang and M. Oestreich, Chem.–Eur. J., 2019, 25, 14304–14307; (e) M. Cui and M. Oestreich, Org. Lett., 2020, 22, 3684–3687; (f) M. Kondo, J. Kanazawa, T. Ichikawa, T. Shimokawa, Y. Nagashima, K. Miyamoto and M. Uchiyama, Angew. Chem., Int. Ed., 2020, 59, 1970–1974; (g) K. Sekine, D. Akaishi, K. Konagaya and S. Ito, Chem.–Eur. J., 2022, e202200657; (h) B. Yang, K. Cao, G. Zhao, J. Yang and J. Zhang, J. Am. Chem. Soc., 2022, 144, 15468–15474.
- 13 (*a*) T. Ohmura, I. Sasaki and M. Suginome, *Org. Lett.*, 2019, 21, 1649–1653; (*b*) I. Sasaki, T. Ohmura and M. Suginome,

Org. Lett., 2020, **22**, 2961–2966; (*c*) Y. Zhang, W. Xu, T. Gao, M. Guo, C.-H. Yang, H. Xie, X. Kong, Z. Yang and J. Chang, *Org. Lett.*, 2022, **24**, 7021–7025.

- 14 (a) Y. Gu, Y. Shen, C. Zarate and R. Martin, J. Am. Chem. Soc.,
 2019, 141, 127–132; (b) J. Zheng, H. Zhang, S. Kong, Y. Ma,
 Q. Du, B. Yi, G. Zhang and R. Guo, ACS Catal., 2024, 14,
 1725–1732; (c) S. Kamio, M. Nakamoto, T. Yamagishi,
 M. Oestreich and H. Yoshida, Chem. Commun., 2024, 60,
 6379–6382; (d) K. Asai, K. Hirano and M. Miura, Eur. J. Org
 Chem., 2022, e202101535.
- 15 (a) T. Takeuchi, A. Roy and H. Ito, J. Am. Chem. Soc., 2023,
 145, 16249–16260; (b) I. Sasaki, A. Maebashi, J. Li,
 T. Ohmura and M. Suginome, Eur. J. Org Chem., 2022,
 e202101573; (c) L. Zhou, J. Qiu, C. Wang, F. Zhang,
 K. Yang and Q. Song, Org. Lett., 2022, 24, 3249–3253.
- 16 (a) T. Kubo and M. Abe, Chem. Rev., 2024, 124, 4541–4542; (b)
 S. Crespi and M. Fagnoni, Chem. Rev., 2020, 120, 979–9833;
 (c) J. P. Barham and B. König, Angew. Chem., Int. Ed., 2020, 59, 11732–11747; (d) M. Yan, J. C. Lo, J. T. Edwards and P. S. Baran, J. Am. Chem. Soc., 2016, 138, 12692–12714; (e)
 A. Studer and D. P. Curran, Angew. Chem., Int. Ed., 2015, 55, 58–102; (f) C. Chatgilialoglu and A. Studer, Encyclopedia of Radicals in Chemistry, Biology and Materials, Wiley-VCH, Weinheim, 2012.
- 17 (a) X. Zhang, J. Fang, C. Cai and G. Lu, Chin. Chem. Lett.,
 2021, 32, 1280-1292; (b) L.-Q. Ren, N. Li, J. Ke and C. He,
 Org. Chem. Front., 2022, 9, 6400-6415; (c) J.-S. Li and J. Wu,
 ChemPhotoChem, 2018, 2, 839-846; (d) X. Shang and
 Z.-Q. Liu, Org. Biomol. Chem., 2016, 14, 7829-7831; (e)
 C. Chatgilialoglu, Chem. Rev., 1995, 95, 1229-1251.
- 18 (a) J. Sun and L. Deng, ACS Catal., 2016, 6, 290–300; (b)
 B. Marciniec, H. Maciejewski, C. Pietraszuk and P. Pawluć, In Hydrosilylation: A Comprehensive Review on Recent Advances, ed. B. Marciniec, Springer, Berlin, 2009, ch. 1; (c)
 S. Díez-González and S. P. Nolan, Acc. Chem. Res., 2008, 41, 349–358; (d) A. K. Roy, Adv. Organomet. Chem., 2007, 55, 1–59; (e) F. Buch, H. Brettar and S. Harder, Angew. Chem., Int. Ed., 2006, 45, 2741–2745; (f) L. D. Almeida, H.-L. Wang, K. Junge, X.-J. Cui and M. Beller, Angew. Chem., Int. Ed., 2021, 60, 550–565.
- A. Matsumoto and Y. Ito, *J. Org. Chem.*, 2000, **65**, 5707–5711.
 M. Zhong, X. Pannecoucke, P. Jubault and T. Poisson, *Chem.-Eur. J.*, 2021, **27**, 11818–11822.
- 21 N. Takemura, Y. Sumida and H. Ohmiya, *ACS Catal.*, 2022, **12**, 7804–7810.
- 22 R. Arai, Y. Nagashima, T. Koshikawa and K. Tanaka, *J. Org. Chem.*, 2023, **88**, 10371–10380.
- 23 Y. Wan, Y.-M. Zhao, J.-J. Zhu, Q.-Y. Yuan, W. Wang and Y.-Q. Zhang, *Green Chem.*, 2023, 25, 256–263.

- 24 T. Biremond, P. Jubault and T. Poisson, *ACS Org. Inorg. Au*, 2022, 2, 148–152.
- 25 M. Aelterman, T. Biremond, P. Jubault and T. Poisson, *Chem.-Eur. J.*, 2022, **28**, e202202194.
- 26 H.-Y. Zhou, L.-Q. Fei, J.-L. Zhang, Y.-M. Pan and H.-T. Tang, Adv. Synth. Catal., 2023, 365, 1591–1595.
- 27 Y. Morimasa, K. Kabasawa, T. Ohmura and M. Suginome, Asian J. Org. Chem., 2019, 8, 1092–1096.
- 28 M. Kondo, J. Kanazawa, T. Ichikawa, T. Shimokawa, Y. Nagashima, K. Miyamoto and M. Uchiyama, *Angew. Chem., Int. Ed.*, 2020, **59**, 1970–1974.
- 29 F. Lovering, J. Bikker and C. Humblet, *J. Med. Chem.*, 2009, 52, 6752–6756.
- 30 (a) J. Zhou, B. Jiang, Y. Fujihira, Z. Zhao, T. Imai and N. Shibata, *Nat. Commun.*, 2021, 12, 3749, For a related defluorosilylation of fluoroarenes or fluoroalkanes with silylboronates, see: ; (b) B. Cui, S. Jia, E. Tokunaga and N. Shibata, *Nat. Commun.*, 2018, 9, 4393.
- 31 (a) L. Gao, X. Liu, G. Li, S. Chen, J. Cao, G. Wang and S. Li, *Org. Lett.*, 2022, **24**, 5698–5703; (b) J. Cao, L. Gao, G. Wang and S. Li, *Green Chem.*, 2024, **26**, 4785–4791.
- 32 H. Lan, X. Huo, Y. Jia and D. Wang, *Org. Lett.*, 2024, **26**, 1011–1016.
- 33 X. Wang, X. Zhang, X. He, G. Guo, Q. Huang, F. You, Q. Wang, R. Qu, F. Zhou and Z. Lian, *Angew. Chem., Int. Ed.*, 2024, 63, e202410334.
- 34 S. Ishigaki, Y. Nagashima, D. Yukimori, J. Tanaka, T. Matsumoto, K. Miyamoto, M. Uchiyama and K. Tanaka, *Nat. Commun.*, 2023, 14, 652.
- 35 (a) S. Chen, X. Guo, H. Hou, S. Geng, Z. Liu, Y. He, X. Xue and Z. Feng, *Angew. Chem., Int. Ed.*, 2023, 62, e202303470, For their related work using silylboronates, see: ; (b) H. Zhang, E. Wang, S. Geng, Z. Liu, Y. He, Q. Peng and Z. Feng, *Angew. Chem., Int. Ed.*, 2021, 60, 10211–10218; (c) S. Wang, M. Sun, H. Zhang, J. Zhang, Y. He and Z. Feng, *CCS Chem.*, 2020, 2, 2164–2173.
- 36 B. Cui, Y. Tian, Y. Gao, S. Hua, Y. Shi and C. Cao, *J. Org. Chem.*, 2023, **88**, 11173–11185.
- 37 J. Zhou, Z. Zhao and N. Shibata, *Nat. Commun.*, 2023, **14**, 1847.
- 38 J. Zhou, Z. Zhao, B. Jiang, K. Yamamoto, Y. Sumiib and N. Shibata, *Chem. Sci.*, 2023, **14**, 4248–4256.
- 39 J. Zhou, Z. Zhao, T. Kiyono, A. Matsuno, J. Escorihuela and N. Shibata, *Chem. Sci.*, 2024, 15, 17418–17424.
- 40 J. Zhou, Z. Zhao, S. Mori, K. Yamamoto and N. Shibata, *Chem. Sci.*, 2024, **15**, 5113-5122.
- 41 X. Wang, C. Feng, J. Jiang, S. Maeda, K. Kubota and H. Ito, *Nat. Commun.*, 2023, **14**, 5561.