



Contact engineering for 2D Janus MoSSe/metal junctions†

Cite this: *Nanoscale Horiz.*, 2024, 9, 264

Received 10th October 2023,
Accepted 16th November 2023

DOI: 10.1039/d3nh00450c

rsc.li/nanoscale-horizons

Yu Shu,^a Ting Li,^a Naihua Miao,^b Jian Gou,^c Xiaochun Huang,^{*d} Zhou Cui,^a Rui Xiong,^a Cuilian Wen,^{id a} Jian Zhou,^{id b} Baisheng Sa^{id *a} and Zhimei Sun^{id *b}

The flourish of two-dimensional (2D) materials provides a versatile platform for building high-performance electronic devices in the atomic thickness regime. However, the presence of the high Schottky barrier at the interface between the metal electrode and the 2D semiconductors, which dominates the injection and transport efficiency of carriers, always limits their practical applications. Herein, we show that the Schottky barrier can be controllably lifted in the heterostructure consisting of Janus MoSSe and 2D vdW metals by different means. Based on density functional theory calculations and machine learning modelings, we studied the electrical contact between semiconducting monolayer MoSSe and various metallic 2D materials, where a crossover from Schottky to Ohmic/quasi-Ohmic contact is realized. We demonstrated that the band alignment at the interface of the investigated metal-semiconductor junctions (MSJs) deviates from the ideal Schottky–Mott limit because of the Fermi-level pinning effects induced by the interface dipoles. Besides, the effect of the thickness and applied biaxial strain of MoSSe on the electronic structure of the junctions are explored and found to be powerful tuning knobs for electrical contact engineering. It is highlighted that using the sure-independence-screening-and-sparsifying-operator machine learning method, a general descriptor $W_M^3/\exp(D_{int})$ was developed, which enables the prediction of the Schottky barrier height for different MoSSe-based MSJ. These results provide valuable theoretical guidance for realizing ideal Ohmic contacts in electronic devices based on the Janus MoSSe semiconductors.

New concepts

As transistor dimensions continue to shrink, achieving efficient electrical contacts in semiconductor devices becomes increasingly important. Two-dimensional (2D) materials have atomic thickness and no dangling bonds on the surface, making them the core materials for next-generation electronic devices. Furthermore, the interface formed by the 2D metal and 2D semiconductor can effectively alleviate the Fermi-level pinning (FLP) effects. Therefore, it is necessary to conduct research on the characteristics of metal-semiconductor interfaces and precisely regulate the Schottky barrier to develop ideal metal-semiconductor junctions (MSJs). Herein, we achieved Ohmic contact and controllable Schottky barrier by selecting 2D metals with different work functions together with monolayer MoSSe, an important 2D Janus transition metal dichalcogenide. It is proved that the existence of interface dipole is the main reason for the FLP effects. Furthermore, we modulate the interface properties of 2D MSJs by varying the thickness and applied biaxial strain. Finally, we developed a universal descriptor capable of predicting the Schottky barrier heights of different 2D MSJs using the machine learning method. Our study provides a new strategy for the understanding and realization of ideal electrical contacts in 2D materials-based MSJs.

1. Introduction

Transition metal dichalcogenide (TMD) monolayers with extraordinary electronic and optoelectronic properties have great potential for next-generation silicon-beyond devices.^{1–4} All TMD monolayers comprise three stacked atomic-layers with the stacking configurations of X–M–X, where M is a transition metal and X is a chalcogen.⁴ Janus TMDs is a special class of TMDs in which the uppermost and lowermost layers of the sandwich structure consist of different chalcogen elements with stacking configurations of X–M–Y, where X and Y indicate different dichalcogenide elements.⁵ The broken mirror symmetry in Janus TMDs generates an intrinsic dipole moment along the vertical direction, leading to novel physical properties, such as second-harmonic generation response, enhanced in-plane piezoelectric polarization, and great catalytic performance.^{6,7} Monolayer MoSSe is a typical Janus TMD material that has been experimentally synthesized since 2017.⁸

^a Multiscale Computational Materials Facility, College of Materials Science and Engineering, Fuzhou University, Fuzhou 350108, P. R. China. E-mail: bssa@fzu.edu.cn

^b School of Materials Science and Engineering, Beihang University, Beijing 100191, P. R. China. E-mail: zmsun@buaa.edu.cn

^c School of Physics, Zhejiang University, Hangzhou 310058, P. R. China

^d Department of Physics, University of Hamburg, D-20355 Hamburg, Germany. E-mail: huangxiaochun0321@gmail.com

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d3nh00450c>

Theoretical studies have shown that monolayer MoS₂ has a tunable direct-bandgap and a high carrier mobility, possessing a potential application in atomically-thin advanced electronic devices.^{5,9}

It is known that when the size of the devices is reduced to the atomic scale, the interfacial electronic structure between the two-dimensional (2D) semiconductor and the electrode materials becomes critical in realizing high-performance applications.^{10,11} In particular, the energy barrier of the metal-semiconductor junctions (MSJs), namely, the Schottky barrier, limits the carrier injection and the contact-resistance characteristics of the devices.^{12–14} In general, the Schottky barrier is induced by two main reasons: (1) the energy difference in the work functions between the semiconductor and the electrode materials and (2) the strong Fermi level pinning (FLP) effect at the interface of the junction.^{15–17} To overcome this issue, proper metals need to be screened out for electrode materials. In contrast to traditional 3D bulk metals, metallic 2D materials enable the fabrication of van der Waals (vdW)-type heterojunction devices, where the FLP effect can be significantly suppressed.^{18–21} Thereinto, 2D metallic TMDs and 2D transition metal carbides/nitrides (MXenes) show excellent mechanical stability and electrical conductivity, making them ideal electrodes for various electronic devices and electrochemical systems.^{4,22,23} With the emergence of new synthesis methods for MXenes, their surface functional groups now encompass a rich chemical diversity, evolving from the initial F, O, and OH to include Cl, Br, and I.^{24,25} These diverse functional groups offer tunable work functions and variable 2D lattices, making MXenes potential candidates for 2D electrodes.

In this work, using first-principles calculations, we studied the interfacial contacting properties of the devices built from monolayer MoS₂ and various 2D metals MXenes (Nb₂C and Nb₃C₂ including functional group of F, O, OH) and MX₂ (M = V, Nb, Ta; X = S, Se). The calculated band structures of these devices show that transition from Schottky contact to Ohmic contact can be achieved by selecting 2D metals with different work functions. We demonstrated that the presence of interface dipoles results in the FLP effect, leading to a deviation of the band alignment from the prediction of the ideal Schottky–Mott limit. Besides, we unraveled that the band edge position of MoS₂ is sensitive to both the layer number and the biaxial strain applied on the film; hence, adjustable Schottky barrier height (SBH) and Ohmic contact can be achieved. Furthermore, we constructed a general formula to describe the SBH of 2D metal–MoS₂ contact based on the sure-independence-screening-and-sparsifying-operator (SISSO) machine learning method.

2. Computational details

Density functional theory (DFT) calculations were performed with the Vienna *ab initio* simulation package (VASP),²⁶ utilizing the Projector-Augmented Wave (PAW)²⁷ method and the ALKEMIE²⁸ platform. The exchange–correlation function was described using

the Perdew–Burk–Ernzerhof (PBE)²⁹ pseudopotential and the generalized gradient approximation (GGA).³⁰ To optimize the electronic properties, *k*-point meshes of 12 × 12 × 1 and 16 × 16 × 1 in the first Brillouin zone (BZ) were used for structure and electronic properties calculations, respectively. The relaxation of atom positions was continued until the maximum residual force per atom was less than 0.01 eV Å⁻¹ and the energy difference was smaller than 10⁻⁶ eV. The plane-wave kinetic energy cut-off was 500 eV. A vacuum thickness of 20 Å was applied to prevent interactions along the *z*-direction between periodic images, and van der Waals correction was applied using the Grimme's DFT-D3 method.^{31,32} The dipole correction method was utilized to eliminate errors in electrostatic potential, total energy, and atomic force resulting from asymmetric arrangement under periodic boundary conditions. The SISSO³³ machine learning method was employed to predict the SBH of metal–semiconductor contacts.

3. Results and discussion

Fig. 1a depicts the geometrical structure of the MoS₂ monolayer, which has a hexagonal structure similar to 2H-MoS₂ monolayers with the space group *P*3̄*m*1.⁵ The MoS₂ monolayer is formed by Mo layer sandwiched between S and Se layers, and the optimized lattice constant *a* is 3.25 Å. As shown in Fig. 1b, the monolayer MoS₂ is a direct bandgap semiconductor (the bandgap is 1.49 eV). The calculated effective electrostatic potential in Fig. 1c demonstrated that there is a net electrostatic potential difference of Δ*V* = 0.74 eV between two sides of the Janus MoS₂ monolayer, which leads to an intrinsic dipole pointing from Se to S (the calculated dipole moment is *D*_{int} = 0.23). These calculated results are in good agreement with previous works.^{34,35}

Next, we investigated the contact characteristics of MSJs formed between MoS₂ monolayer and various 2D metals, including metallic MXenes Nb₂C, Nb₃C₂, Nb₃CT₂ (T = F, O, OH), Nb₃C₂T₂ (T = F, O, OH), and metallic TMDs MX₂ (M = V, Nb, Ta; X = S, Se). Fig. 1d shows the geometrical structures of the studied 2D metals. The band structures of all these 2D metals have many electronic states crossing the Fermi level, which are represented in Fig. S1 (ESI[†]), indicating their excellent conductivity and potential as metal electrodes. Fig. 1e shows the band edge alignments of MoS₂ monolayer and the work functions of the studied 2D metals. To ensure a comprehensive study of metal–MoS₂ contacts, we selected 2D metals with a wide range of work function values from 2.10 to 6.08 eV. Herein, the minimum (maximum) work functions of 2D metals are lower (larger) than the *E*_C (*E*_V) of MoS₂ monolayer. Meanwhile, the work functions of the 2D metals can be continuously varied within the bandgap of MoS₂ monolayer and thus can effectively study the interaction law of the metal–MoS₂ contacts. The schematic diagram of MoS₂-based FET model is presented in Fig. 1f, where the 2D metal serves as the source and drain electrodes, and the MoS₂ monolayer serves as the channel material of the device.

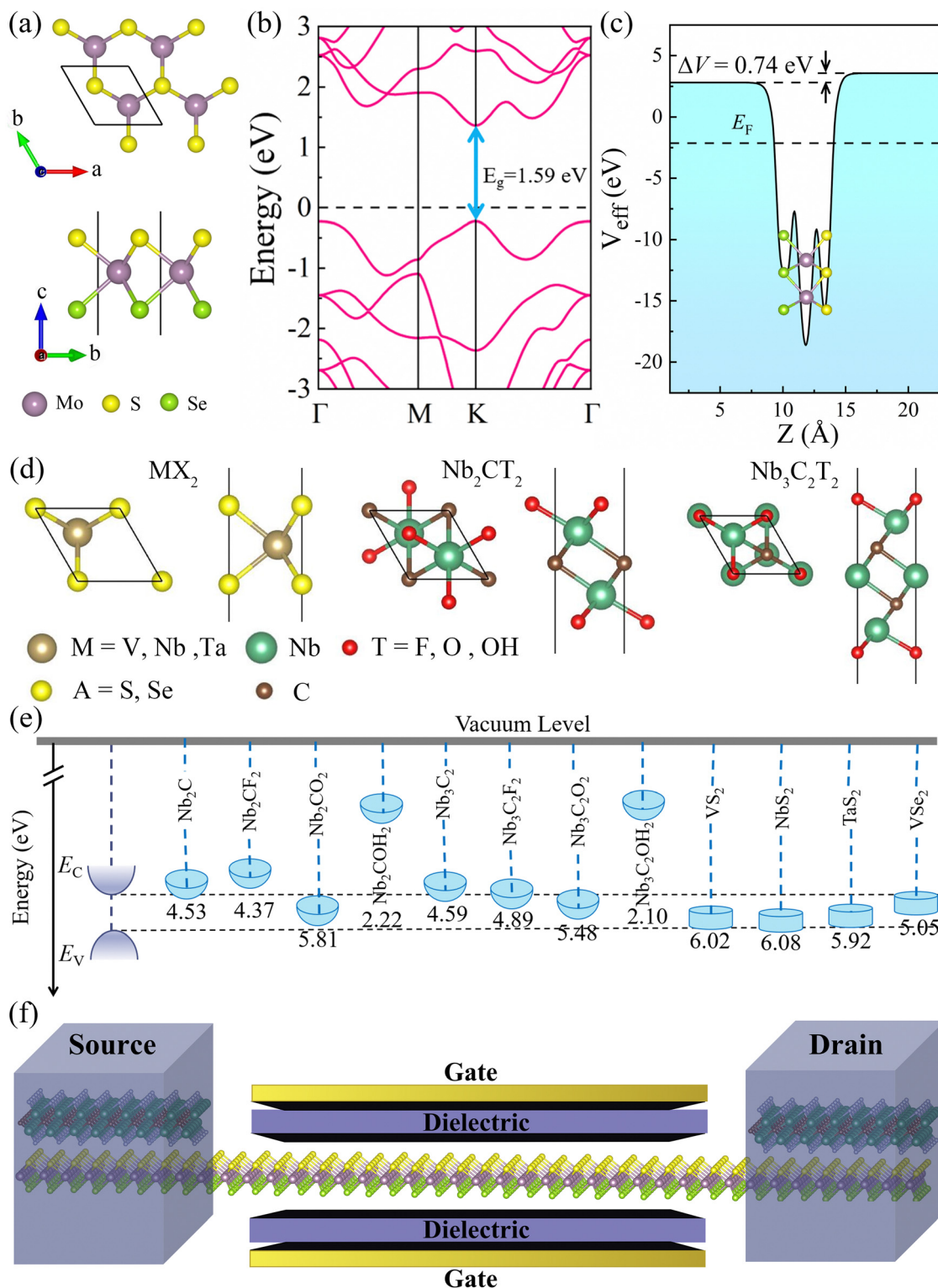


Fig. 1 (a) Top and side views of the optimized structure, (b) band structure, and (c) effective electrostatic potential of the MoSSe monolayer. (d) Top and side views of 2D metals. (e) The band alignments between MoSSe and work functions of 2D metals, E_C and E_V represent the energy levels of CBM and VBM for MoSSe monolayer, respectively. (f) Schematic diagram of 2D metal–MoSSe FETs.

Among these 2D metals, the optimized lattice constants lie in the range from 3.11 to 3.35 \AA , leading to less than 5% lattice mismatch (as listed in Table 1) relative to the MoSSe monolayer

(3.25 \AA). In the following, 2D metal–MoSSe contacts formed by placing 2D metals on the top of the Se and S sides of the MoSSe monolayer, where 2D metals contact with the Janus MoSSe

Table 1 Lattice mismatch (ε), metal work function (W_M), interlayer distance (d), binding energy (E_b) and Schottky barrier height (Φ_{SBH}) (n and p represent n-type and p-type contact, respectively) of 2D metal–MoSSe contacts (Se and S correspond to the Se and S sides in contact with the 2D metal)

2D metal	ε (%)	W_M (eV)	d (Å)		E_b (meV Å ⁻²)		Φ_{SBH} (eV)	
			Se	S	Se	S	Se	S
Nb ₂ C	4.17	4.53	2.62	1.68	-155.17	-201.38	—	—
Nb ₂ CF ₂	1.56	4.37	2.82	2.59	-29.07	-28.41	0.77 (n)	0.18 (n)
Nb ₂ CO ₂	3.83	5.81	2.90	2.73	-35.29	-32.82	0.03 (p)	0.41 (p)
Nb ₂ COH ₂	1.56	2.22	3.09	3.09	-56.21	-69.37	-0.10 (n)	-0.17 (n)
Nb ₃ C ₂	2.77	4.59	2.61	1.68	-161.56	-212.77	—	—
Nb ₃ C ₂ F ₂	4.50	4.89	2.85	2.64	-30.27	-29.64	0.50 (p)	0.33 (n)
Nb ₃ C ₂ O ₂	3.18	5.48	2.82	2.83	-35.79	-37.29	0.05 (p)	0.58 (p)
Nb ₃ C ₂ (OH) ₂	3.80	2.10	1.99	2.08	-57.55	-71.35	-0.33 (n)	-0.28 (n)
VS ₂	2.20	6.02	3.12	2.93	-30.22	-29.37	0.06 (p)	0.47 (p)
NbS ₂	3.08	6.08	3.01	2.86	-32.39	-30.13	0.08 (p)	0.46 (p)
TaS ₂	2.77	5.92	3.02	2.95	-31.07	-29.16	0.11 (p)	0.57 (p)
VSe ₂	3.38	5.46	3.11	3.01	-30.85	-30.10	0.50 (p)	0.43 (n)

monolayers on its Se and S sides, are named metal–SeMoS contacts and metal–SMoSe contacts, respectively. In addition, 2D metal–MoSSe contact represents both metal–SeMoS and metal–SMoSe cases. For each type of 2D MSJs, there are six different possible heterostructure stacking configurations, as shown in Fig. S2 (ESI†). We summarized the relative total energies of 2D metal–MoSSe contacts in Tables S1–S3 (ESI†). Furthermore, the 2D MSJs with stacking configuration with the lowest total energy are selected for the subsequent calculations. We defined the metal–semiconductor interlayer distance as the average distance between the Se or S atoms on top of MoSSe and the atoms at the bottom of the 2D metal.³⁶ In most 2D metal–MoSSe contacts, the equilibrium interlayer distance ranges from 2.59 and 3.12 Å, which are typical vdW interactions.³⁷ However, the Nb₂C and Nb₃C₂ systems without and with –OH surface functional groups have smaller interlayer distances, indicating relatively strong interactions. Among them, the Nb₂C and Nb₃C₂ systems with –OH surface functional groups have strong interactions due to hydrogen bonding. The bare Nb₂C and Nb₃C₂ systems have a large number of dangling bonds on the surface, resulting in strong orbital coupling with the MoSSe layer. To further understand the binding strength of the MSJ interface, the binding energy (E_b) was calculated as³⁸

$$E_b = \frac{E_{\text{metal–MoSSe}} - E_{\text{metal}} - E_{\text{MoSSe}}}{A} \quad (1)$$

where $E_{\text{metal–MoSSe}}$, E_{metal} , and E_{MoSSe} represent the total energy of metal–MoSSe contact, isolated 2D metal and MoSSe monolayers, respectively. A is the interface area of the MSJ. Table 1 summarizes the calculated binding energies. The results indicate that all 2D metal–MoSSe contacts exhibit negative binding energies, implying structural stability and experimental feasibility. Notably, except for the Nb₂C and Nb₃C₂ systems without and with –OH surface functional groups, the binding energy of 2D metal–MoSSe contact is about -30 meV Å^{-2} , belonging to typical vdW interactions, which are consistent with the conclusions from the interlayer distance results.

The contact properties of MSJs are mainly dominated by the SBH, which is defined as the energy difference between the Fermi level and the conduction band minimum (CBM) or

valence band maximum (VBM) of the semiconductor in the MSJ.^{39,40}

$$\Phi_n = E_C - E_F \quad (2)$$

$$\Phi_p = E_F - E_V \quad (3)$$

where Φ_n and Φ_p represent the electron and hole SBHs, respectively. E_F is the Fermi level in the MSJs, and E_C and E_V are the energy level of the CBM and VBM of MoSSe monolayer, respectively. The projected band structures of 2D metal–MoSSe contacts are illustrated in Fig. S3 (ESI†). The different types of Schottky barriers obtained from the projected band structures of 2D metal–MoSSe contacts are listed in Table 1. As shown in Fig. S3 (ESI†), after contacting Nb₂C and Nb₃C₂ with MoSSe, the band structures of MoSSe are obviously metalized due to the presence of dangling bonds on the surface of Nb₂C and Nb₃C₂, resulting in strong covalent-like orbital coupling between the MoSSe monolayer and 2D metals (Nb₂C and Nb₃C₂). The formation of chemical bonds at the interface due to such strong orbital coupling leads to the metallization of MoSSe. On the contrary, for the MSJs with van der Waals bonding and hydrogen bonding interactions, the band structure of MoSSe monolayer is well preserved. Based on the obtained SBHs, the MSJs can be classified into three different types given in Fig. 2: Schottky contact, Ohmic contact, and quasi-Ohmic contact. If the work function of 2D metal is closer to the work function of MoSSe, the formed MSJ tends to form Schottky contact. For example, as shown in Fig. 2a, the projected band structure of the VSe₂–MoSSe contacts shows that the Fermi level of VSe₂ is located between the CBM and VBM of MoSSe. Due to the larger work function of S-side in the MoSSe monolayer, the Fermi level moves up close to the CBM of the MoSSe monolayer to form n-type Schottky contact in VSe₂–SMoSe contact. Furthermore, due to high work functions metals (for instance, Nb₂CO₂, Nb₃C₂O₂, VS₂, NbS₂, TaS₂, and VSe₂) contacting the Se side, the Fermi level shifts less, resulting in the formation of p-type Schottky contact, as shown in Fig. S3 (ESI†). For MSJs with Ohmic contacts ($\Phi_{\text{SBH}} \leq 0$) as shown in Fig. 2b, it should select metals with small work function (such as Nb₂C(OH)₂ and

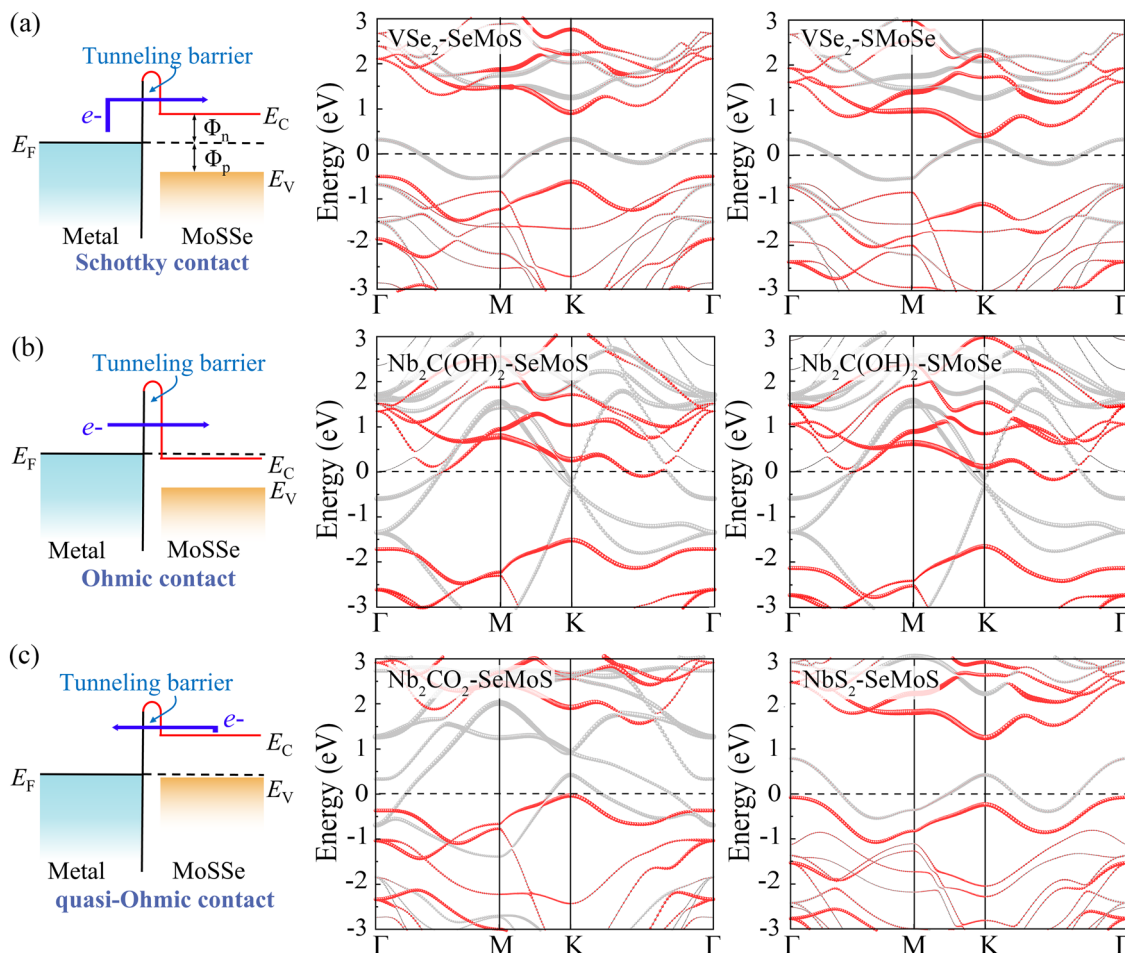


Fig. 2 Schematic diagram and the projected band structures of the corresponding metal–MoS₂ with (a) Schottky contact, (b) Ohmic contact, (c) quasi-Ohmic contact. The red and gray spheres represent the contributions from MoS₂ and metal layers, respectively.

Nb₃C₂(OH)₂) to ensure the spontaneous transfer of electrons from the metal side to the semiconductor side. For instance, the projected band structure of the 2D Nb₂C(OH)₂–MoS₂ contact shows that the CBM of the Nb₂C(OH)₂ layer crosses the Fermi level to form an Ohmic contact (Fig. 2b), which indicates that electrons can be freely injected from Nb₂C(OH)₂ into MoS₂ layers. For quasi-Ohmic contact, as shown in Fig. 2c, it refers to MSJs with properties between Schottky contact and Ohmic contact, in which SBH is very close to 0. For the high work function metals contacting with Se side, ultra-low Schottky barriers are obtained to form quasi-Ohmic contact. For example, the projected band structure of Nb₂CO₂–SeMoS and NbS₂–MoS₂ (Fig. 2c) shows that the VBM of the MoS₂ layer is close to the Fermi level, leading to quasi-Ohmic contact. Furthermore, we found that 2D metal–SeMoS contacts tend to form p-type contact, while 2D metal–SMoSe contacts are more likely to form n-type contact, as summarized in Table 1.

To further analyze the local chemical bonding environments, Fig. 3a depicts the 2D plots of electron localization function (ELF) for typical 2D metal–MoS₂ contacts. The ELF is calculated on a 3D grid space using a single determinantal wavefunction derived from the calculated Kohn–Sham orbitals.^{41,42}

The ELF value is a quantitative description of the electron density localization, ranging from 0 to 1, where 1 indicates perfect localization and 0.5 represents gas-like pair probability. The strength of a bond can be characterized by the bond point in the ELF, which corresponds to the saddle point with two negative eigenvalues and one positive eigenvalue.⁴³ Herein, the interfacial bond points of 2D metal–MoS₂ contacts can be simply reflected by the minimum value at the center of the ELF line profile curves, as shown in Fig. 3b and Fig. S4 (ESI†).⁴⁴ Due to the similar characteristics displayed by the contacts of 2D metals with Se and S sides, we analyzed Nb₂C–SeMoS and MX₂–SeMoS as examples to evaluate the bond strength, as shown in Fig. 3a and b. The Nb₂C–SeMoS contact exhibits the largest ELF bond point value of 0.18 for the Nb–Se bond, indicating strong orbital overlap and covalent-like interfacial interactions. In contrast, the values of the ELF bond points of X–Se (X = F, O, S, Se) in the other Nb₂CT₂–SeMoS contacts are lower, indicating weak atomic orbital overlap and interfacial interactions. Among the Nb₂CT₂–SeMoS contacts (T = F, O, OH), the H–Se bond point in the Nb₂C(OH)₂–SeMoS contact exhibits relatively larger value, which also corresponds to their relatively larger binding energy and smaller interlayer distance. The other 2D metal–MoS₂ contacts exhibited similar results, as shown in Fig. S4 (ESI†).

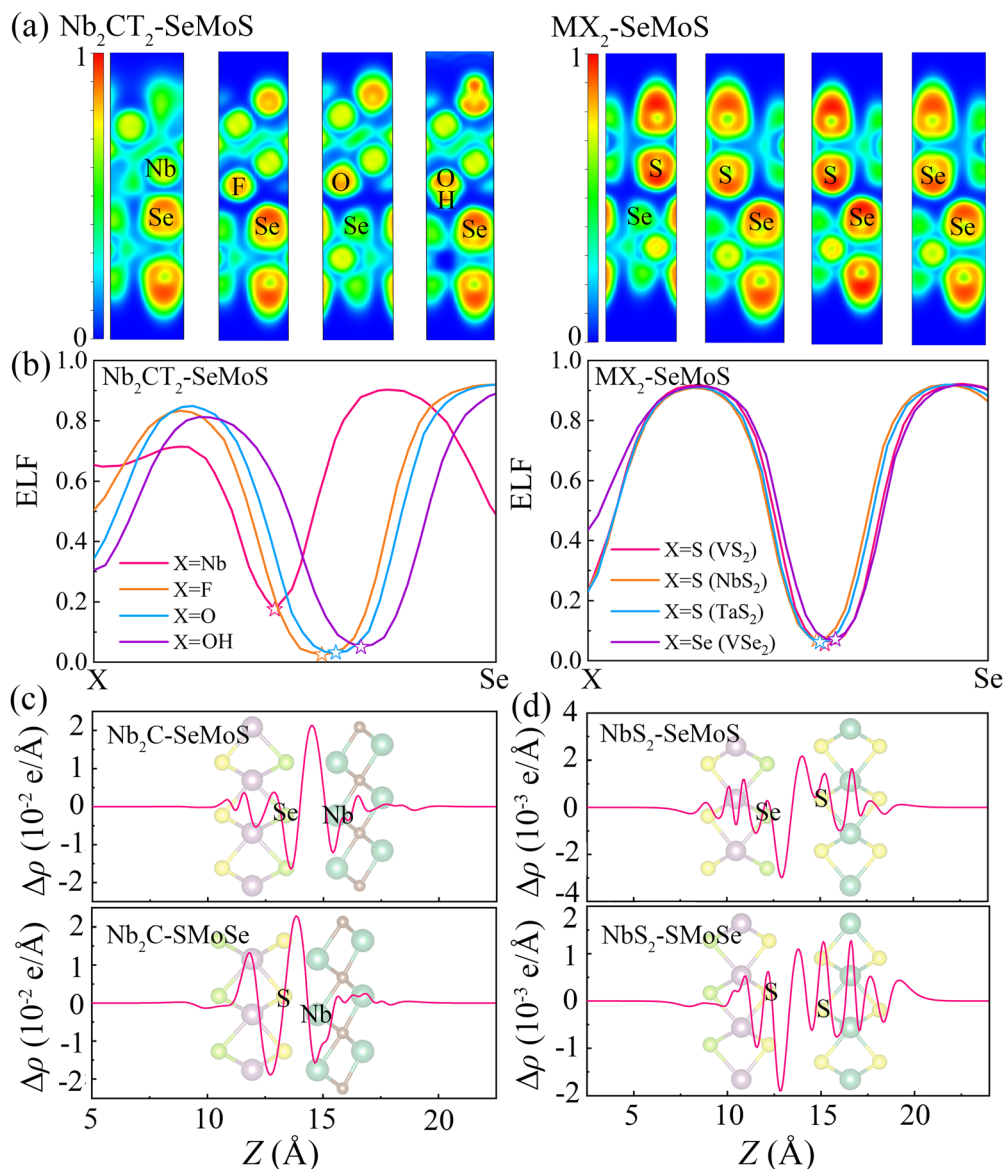


Fig. 3 The ELF 2D line profiles and bond points of (a) X–Se bonds in 2D metal–SeMoS contacts and (b) X–S bonds in 2D metal–SMoSe contacts. The pentagrams represent the positions of the bond points. The plane averaged charge density differences for (c) Nb₂C–SeMoS and (d) NbS₂–SMoSe contacts.

To further understand the electronic properties of MSJs, the plane-averaged charge density difference ($\Delta\rho$) is calculated as

$$\Delta\rho = \rho_{\text{metal-MoSSe}} - \rho_{\text{metal}} - \rho_{\text{MoSSe}} \quad (4)$$

where $\rho_{\text{metal-MoSSe}}$, ρ_{metal} , and ρ_{MoSSe} are the charge densities of metal–MoSSe contact, isolated metal, and MoSSe monolayers, respectively. Due to the stronger covalent-like interaction of Nb₂C–MoSSe interfaces, Fig. 3c observes a significant charge assembling in the interface areas. This is consistent with our proposal that the strong covalent orbital coupling system leads to the metallization of the band. On the other hand, an interface dipole and a built-in electric field are formed due to the significant charge transfer at the vdW MSJ interfaces, which is evident from the apparent charge accumulation and charge depletion shown in Fig. 3d and Fig. S5 (ESI[†]).^{45,46} Fig. 3d illustrates the relatively weaker charge transfer in NbS₂–MoSSe

contacts, indicating that relatively small interface dipoles are formed. Similar results are observed in the Nb₂CT₂, Nb₃C₂T₂ (T = F and O), and MX₂ (M = V, Ta; X = S, Se) systems. The Nb₂C(OH)₂–MoSSe and Nb₃C₂(OH)₂–MoSSe interfaces exhibit relatively strong charge transfer, resulting in larger interface dipoles.

The desired 2D MSJs can be achieved by tuning the SBH through the use of metals with different work functions. It is essential to comprehend the FLP effect to analyze the interface properties and the role of interface dipoles. The FLP factor S is calculated as follows.^{19,47}

$$S = \frac{d\Phi_n}{dW_M} \quad (5)$$

The ideal Schottky–Mott limit corresponds to $S = 1$, indicating that there is no FLP effect. However, for $S = 0$, a strong FLP

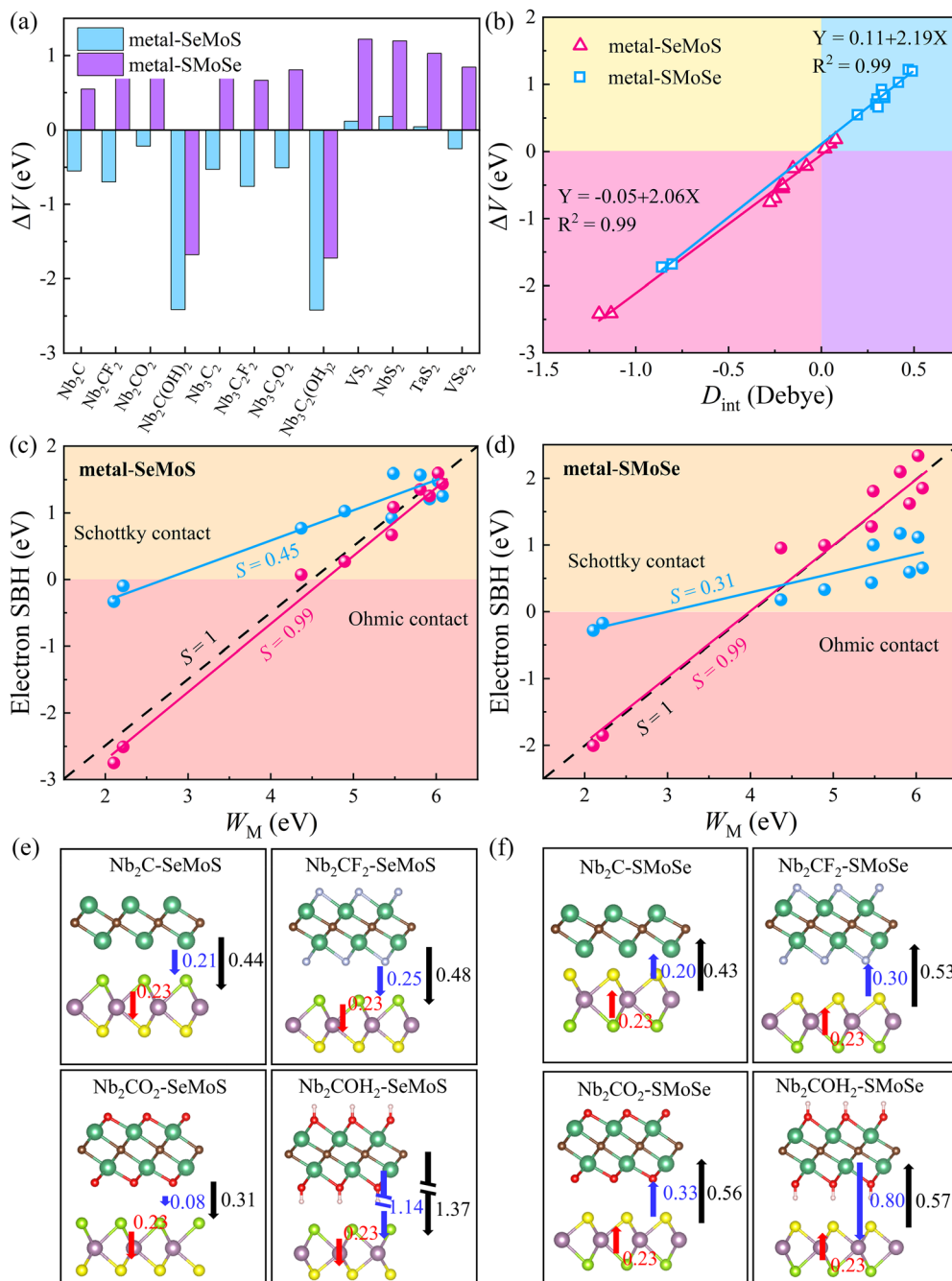


Fig. 4 (a) The electronic potential difference (ΔV) in 2D metal–MoS₂ contacts. (b) The change in ΔV with interface dipole (D_{int}) for 2D metal–MoS₂ contacts. The variation in the electron SBH as a function of metal work functions (W_M) for (c) metal–SeMoS (d) metal–SMoSe contacts. The azure and pink lines represent fitting results without and with ΔV correction, respectively. The black dotted line represents the Schottky–Mott limit $S = 1$. The Fermi level pinning factor S is defined as the value of the fitted slope. The dipole moment distributions in selected (e) 2D metal–SeMoS and (f) 2D metal–SMoSe contacts. The red, blue, and black arrows represent the intrinsic dipole, interface dipole, and total dipole, respectively.

effect occurs. The Schottky–Mott limit is difficult to reach in 2D MSJs due to the presence of interface dipoles,^{48,49} resulting in a certain deviation of SBH from the ideal value.⁵⁰ As shown in Fig. 4a, the increase in the work function difference between the 2D metal and MoS₂ monolayer leads to an increase in ΔV . Herein, at a larger value of ΔV , a larger interface dipole in the metal–MoS₂ interfaces is presented. However, due to the different work functions of the Se and S sides in the MoS₂

monolayer, we have obtained different ΔV values for metal–SMoSe and metal–SeMoS contacts. Notably, the low work function metals ($\text{Nb}_2\text{C(OH)}_2$ and $\text{Nb}_3\text{C}_2\text{(OH)}_2$) led to a significant work function difference and a large ΔV . Furthermore, the interface dipole (D_{int}) shows a good linear scaling relationship with ΔV , as shown in Fig. 4b, which is consistent with the Helmholtz equation:⁵¹ $\Delta V = D_{\text{int}} \cos \varphi / \epsilon \epsilon_0$. Therefore, we can correct the SBH by ΔV . The larger ΔV indicates that the FLP

effect is stronger, resulting in more deviation from the Schottky–Mott limit. Therefore, the modification of SBH by Φ_n minus ΔV can be achieved. Fig. 4c and d show that the FLP factors S of the metal–SeMoS and metal–SMoSe contacts are 0.45 and 0.31, respectively. Thus, the FLP ability of the Se side of MoSSe monolayer is weaker than that of the S side due to the larger ΔV in the S side observed in Fig. 4b. The corrected FLP factor S of metal–SeMoS and metal–SMoSe contacts can reach 0.99, almost reaching the Schottky–Mott limit. This phenomenon further verifies that the deviation from the Schottky–Mott limit of the FLP effect in 2D MSJs is mainly caused by the interface dipole, where the deviation strength is related to the magnitude of the interfacial dipoles. Therefore, it prospectively hinders the FLP effect by weakening the interlayer coupling between the 2D metal and MoSSe monolayer. As a result, the strength of the interfacial dipoles can be reduced, which leads to an ideal band alignment and a reduced SBH. Furthermore, it is noted that the strength of D_{int} is related to the electronegativity (which represents the ability of an element to attract electrons) of the terminal atoms on the contact interface.⁴⁸ Since the electronegativity of the S element (2.58) is greater than that of the Se element (2.55), a larger D_{int} is exhibited when the S side of MoSSe monolayer is in contact with the 2D metals, resulting in a stronger FLP effect, further deviating from the ideal Schottky–Mott limit.

On the other hand, intrinsic dipoles are also presented in the MoSSe monolayer. Therefore, we further studied the role of the total dipole moment $D_{\text{tot}} = D_0 + D_{\text{int}}$, where D_0 is the intrinsic dipole. As shown in Fig. 4e, f and Fig. S6 (ESI[†]), when

the 2D metal contacts MoSSe monolayer with the Se side, the interface dipoles and intrinsic dipoles of VS₂, NbS₂, and TaS₂ are in opposite directions, resulting in a smaller D_{tot} . For the other metal–MoSSe contacts, the larger D_{tot} was generated due to the same orientation of the interface dipoles and intrinsic dipoles. In contrast, when the 2D metal contacts MoSSe monolayer with the S side, the interface dipoles and intrinsic dipoles of the low work function metals (Nb₂C(OH)₂ and Nb₃C₂(OH)₂) are in opposite directions, resulting in a minor D_{tot} . In comparison, the other metals in contact with the Se sides lead to larger D_{tot} due to the same orientation of interface dipoles and intrinsic dipoles. The superposition effect of interface and intrinsic dipoles increases the ΔV , resulting in a stronger FLP effect deviating from the Schottky–Mott limit. On the other hand, ΔV is also linearly related to the total dipole moment D_{tot} for 2D metal–MoSSe contacts, as shown in Fig. S7 (ESI[†]).

Fig. 5a demonstrates that a p-type Schottky contact is formed more favorably when Nb₃C₂F₂ with lower work function contacts the Se side. The work function of the Se side is greater than that of the Nb₃C₂F₂ side after contact, leading to a negative ΔV in the Nb₃C₂F₂–SeMoS contact. On the other hand, Fig. 5b shows that n-type Schottky contact formation is also favored when Nb₃C₂F₂ contacts the S side; the work function of the S side is lower than that of the Nb₃C₂F₂ side, resulting in a positive ΔV . Furthermore, due to the extremely low work function of Nb₂C(OH)₂ and (Nb₃C₂(OH)₂), ΔV is consistently negative regardless of the contact with the Se or S side. Conversely, when high work function metals contact the Se or S sides, p-type contact formation is more favorable, and ΔV is always

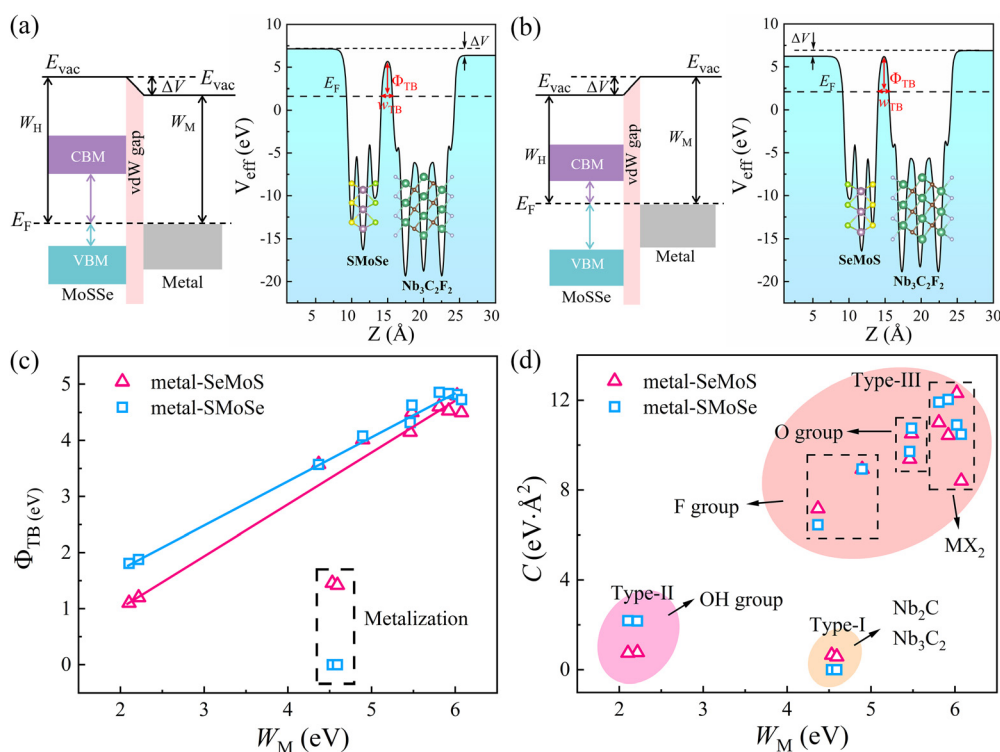


Fig. 5 The energy level schematic diagram and effective electrostatic potential of (a) Nb₃C₂F₂–SeMoS and (b) Nb₃C₂F₂–SMoSe contacts. The variation of (c) tunnel barrier height Φ_{TB} and (d) comprehensive factor C as a function of metal work functions for 2D metal–MoSSe contacts.

positive. In short, the metal-SeMoS and metal-SMoSe contacts typically generate oppositely-oriented interface dipoles. This phenomenon is also known as the pushback effect or pillow effect, which is caused by the redistribution of electrons by Pauli exchange repulsion.⁵² The effective electrostatic potential illustrated in Fig. 5a, b and Fig. S8, S9 (ESI[†]) determines the electrostatic potential difference ΔV and tunneling barrier (TB). Herein, a large TB formed at the MSJs can impede the carrier injection efficiency. We proposed the tunnel barrier height Φ_{TB} and width w_{TB} as two parameters to characterize TB, as listed in Table S4 (ESI[†]).

The efficiency of carrier injection can be evaluated by the tunneling probability P_{TB} ; the tunneling probability equal or close to 100% can promote the carrier injection efficiency. The tunneling probability P_{TB} is defined as⁵⁰

$$P_{\text{TB}} = \exp\left(-\frac{2w_{\text{TB}}}{\hbar}\sqrt{2m\Phi_{\text{TB}}}\right) \quad (6)$$

where m and \hbar are the mass of the free electron and reduced Planck's constant, respectively. Φ_{TB} and w_{TB} represent the tunnel barrier height and width, respectively. For clarity, a comprehensive factor $C = w_{\text{TB}}^2\Phi_{\text{TB}}$ can be introduced to reach a fast estimate of P_{TB} . A smaller value of C indicates a larger P_{TB} . The specific values of C from Φ_{TB} and w_{TB} are listed in Table S4 (ESI[†]) as well. Interestingly, we observed a linear relationship between the tunneling barrier height Φ_{TB} and W_{M} (except for the metalized Nb_2C -MoSse and Nb_3C_2 -MoSse contacts), where the tunneling barrier height Φ_{TB} gradually increases with the metal work functions, as depicted in Fig. 5c. As summarized in Table S4 (ESI[†]), the metalized Nb_2C -MoSse and Nb_3C_2 -MoSse contacts have the smallest C values, followed by the Nb_2COH_2 -MoSse and $\text{Nb}_3\text{C}_2\text{OH}_2$ -MoSse contacts, indicating a higher tunneling probability. In particular, the C values of the Nb_2C -MoSse and Nb_3C_2 -MoSse interfaces are 0, indicating high tunneling probabilities of 100%. On the other hand, the C values of the Nb_2CT_2 -MoSse and $\text{Nb}_3\text{C}_2\text{T}_2$ -MoSse contacts with F and O functional groups are relatively larger. Additionally, MX_2 -MoSse contacts present much greater C values. Additionally, Fig. 5d represents a map by connecting the comprehensive factor C and metal work function for different 2D MSJs. It is noted that the comprehensive factor C follows a similar rule to the tunnel barrier height Φ_{TB} . Interestingly, according to Fig. 5d, we can divide the studied 2D MSJs into three different types: Type-I includes Nb_2C -MoSse and Nb_3C_2 -MoSse contacts with small C values and large W_{M} , Type-II includes Nb_2COH_2 -MoSse and $\text{Nb}_3\text{C}_2\text{OH}_2$ -MoSse contacts with small C values and small W_{M} , and Type-III includes other MSJs with large C values and large W_{M} .

To further investigate the role of interface dipoles in 2D MSJs, we calculated the electronic properties of 2D metal-MoSse contacts with different MoSse layer numbers and biaxial strains. As shown in Fig. 6a and b, the results revealed that the band gap of the semiconductor in the $\text{Nb}_3\text{C}_2\text{F}_2$ -SeMoS contact gradually decreases with increasing MoSse layer numbers. Additionally, the valence band moves closer to the Fermi level, resulting in a quasi-Ohmic contact. In contrast, for the

$\text{Nb}_3\text{C}_2\text{F}_2$ -SMoSe contact, the conduction band moves down and crosses the Fermi level, forming an Ohmic contact with increasing MoSse layer numbers. Furthermore, it can be observed from Fig. 6c and d that as the number of MoSse layers increases, the Φ_{n} and Φ_{p} values of the $\text{Nb}_3\text{C}_2\text{F}_2$ -MoSse contacts also gradually decrease. Simultaneously, the interface dipoles gradually increase and the band gap of the semiconductor layer decreases, indicating that interlayer coupling between $\text{Nb}_3\text{C}_2\text{F}_2$ and MoSse is gradually enhanced with increasing MoSse layer numbers. Tables S5 and S6 (ESI[†]) list the detailed results for electronic SBH Φ_{n} , hole SBH Φ_{p} , interface dipole D_{int} , total dipole D_{tot} , potential step ΔV , work function difference ΔW , and metal work functions W_{M} for 2D MSJs with different MoSse layer numbers.

Based on the Φ_{n} and Φ_{p} values obtained from the energy band diagrams, we plot heatmaps of Φ_{n} and Φ_{p} for 2D MSJs with different numbers of MoSse layers in Fig. 6e and f, respectively. Negative Φ_{n} and Φ_{p} values are set to zero since the Ohmic contact is achieved when the SBHs become negative. It is noted that the Φ_{n} and Φ_{p} values of 2D metal-MoSse both show a decreasing trend as the number of layers increases. Furthermore, the Φ_{n} and Φ_{p} values of 2D MSJs are strongly influenced by two important parameters: firstly, the 2D metal-MoSse contact with Se and S sides leads to different SBHs. Secondly, the use of metals with different work functions leads to different SBHs. Interestingly, as the number of MoSse layers increases, the Φ_{n} values of $\text{Nb}_2\text{C}(\text{OH})_2$ and $\text{Nb}_3\text{C}_2(\text{OH})_2$ in contact with MoSse are always 0, indicating the formation of Ohmic contacts. Additionally, the 2D metal-SeMoS contacts are more inclined to form p-type contacts. As the number of MoSse layer number increases, significantly low Φ_{p} values are obtained to form p-type quasi-Ohmic contacts for Nb_2CO_2 -SeMoS, Nb_3CO_2 -SeMoS, VS_2 -SeMoS, and VSe_2 -SeMoS cases. Conversely, n-type Ohmic contacts are preferred for 2D metal-SMoSe contacts. Therefore, the variation in the number of semiconductor layers in MSJs can regulate the values of SBH. To further understand the influence of the strength of the interface dipole on the FLP, we studied the variation in the electronic SBH and the metal work function of 2D metal-MoSse contacts. By fitting the slope, we obtained the value of the FLP factor S . As shown in Fig. S10 (ESI[†]), the increase in the number of semiconductor layers leads to a decrease in the FLP factor, indicating an enhanced FLP effect. The decrease in the FLP factor also corresponds to the enhancement of the interface dipole, as shown in Fig. S10c and d (ESI[†]).

To explore the strain engineering of electronic structure and the SBHs for 2D MSJs, we further investigated the 2D metal-MoSse contacts under biaxial strains from -6% to 6%. By taking the TaS_2 -MoSse contacts as examples, we found that the bandgap of the MoSse layer changes gradually with increasing tensile or compressive strains, as shown in Fig. 7a and b. Herein, both the valence and conduction bands tend to gradually move closer to the Fermi level with increasing tensile strain for both TaS_2 -SeMoS and TaS_2 -SMoSe contacts. It is interesting to note that the valence band moves up through the Fermi level to form an Ohmic contact with 4% tensile strain for the TaS_2 -

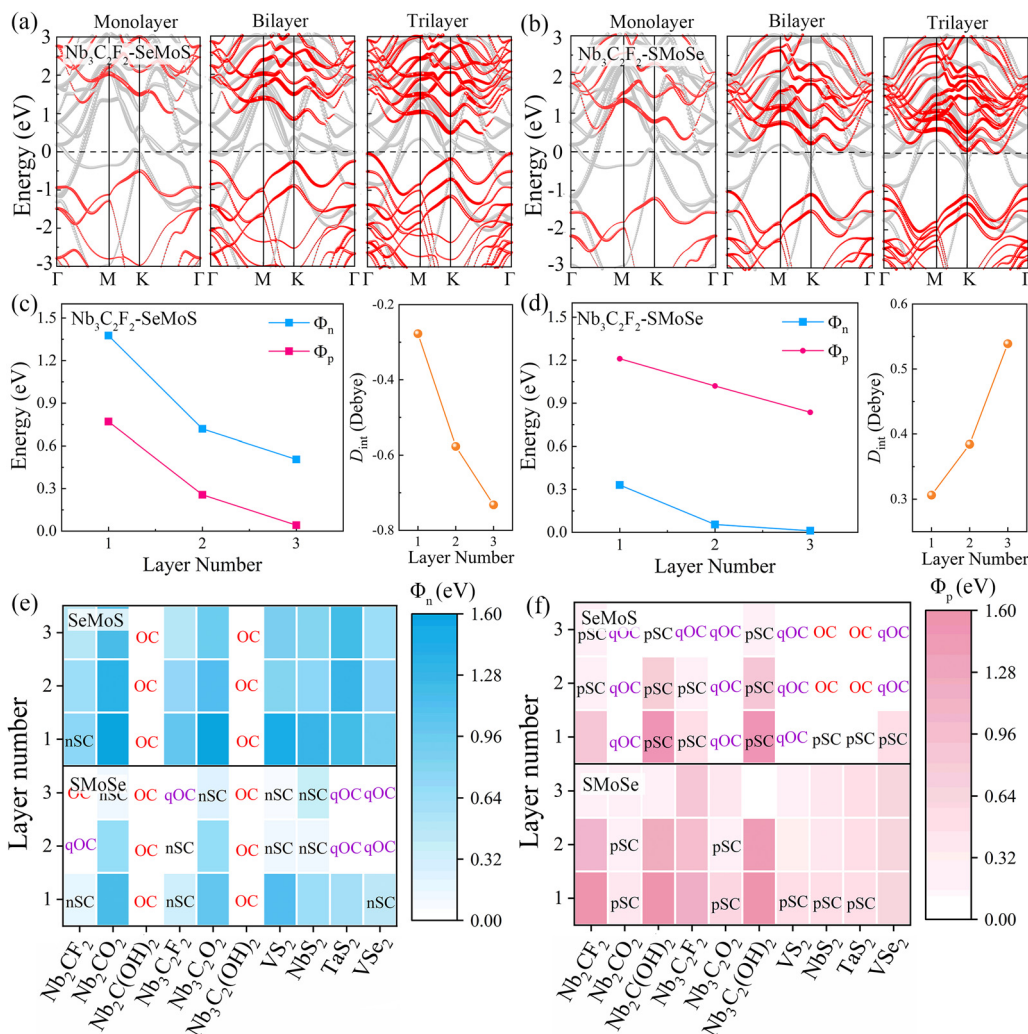


Fig. 6 The projected band structures of (a) $\text{Nb}_3\text{C}_2\text{F}_2\text{-SeMoS}$ and (b) $\text{Nb}_3\text{C}_2\text{F}_2\text{-SMoSe}$ contacts with monolayer, bilayer, and trilayer MoSSe. Schottky barriers and interface dipole of (c) $\text{Nb}_3\text{C}_2\text{F}_2\text{-SeMoS}$ and (d) $\text{Nb}_3\text{C}_2\text{F}_2\text{-SMoSe}$ contacts as a function of MoSSe layer number. Schottky barrier heights (e) Φ_n and (f) Φ_p of 2D metal–MoSSe contacts with different MoSSe layer numbers. OC means Ohmic contact, qOC means quasi-Ohmic contact (Φ_n or $\Phi_p < 0.05$ eV), nSC and pSC represent n-type and p-type Schottky contacts, respectively.

SeMoS contact. It is noted that the calculated Φ_n and Φ_p values are significantly affected by the variation of strain values, as shown in Fig. 7c and d. The $\text{TaS}_2\text{-SeMoS}$ contact transforms into p-type Ohmic contact at 6% tensile strain, while the $\text{TaS}_2\text{-SMoSe}$ contact transforms from p-type to n-type Schottky contact at compression strains greater than -2% . At the same time, regardless of whether TaS_2 contacts the MoSSe monolayer with the Se or S sides, the strain free interface dipole is the smallest, indicating weak interlayer coupling between TaS_2 and MoSSe at this point. The results show the possibility of Schottky to Ohmic contact transition under strain. Tables S7 and S8 (ESI[†]) list the detailed results for electronic SBH Φ_n , hole SBH Φ_p , interface dipole D_{int} , total dipole D_{tot} , potential step ΔV , work function difference ΔW , and metal work functions W_M for 2D MSJs under different strains.

Fig. 7e and f present the Φ_n and Φ_p heatmaps of 2D metal–MoSSe contacts under different biaxial strains, respectively. Interestingly, the contact types of $\text{Nb}_2\text{C}(\text{OH})_2$, $\text{Nb}_3\text{C}_2(\text{OH})_2$,

Nb_2CO_2 , and $\text{Nb}_3\text{C}_2\text{O}_2$ contacts remain unchanged within the studied strain range, where the n-type Ohmic contact ($-\text{OH}$ terminated) and p-type quasi-Ohmic contact ($-\text{O}$ terminated) characteristics are always maintained. As the biaxial strain changes from -6% to 6% , the Φ_p value of the high work function metals (VS_2 , NbS_2 , TaS_2) contact gradually decreases and approaches 0, forming a p-type quasi-ohmic contact. The opposite trend is observed for 2D metal-SMoSe contacts, which favors the formation of n-type Ohmic contact. Therefore, we proposed that the SBH can also be tuned by applying biaxial strains. The FLP strength under different biaxial strains is fitted from the relationship between SBH and metal work function in Fig. S11a and b (ESI[†]). For the metal-SeMoS contacts, the fitted FLP factor S obtained is hardly affected by the compressive strain. Applied -6% compression strain could only slightly increase the FLP factor S to 0.46 (from 0.45 without strain). With the increase in tensile strain, the FLP factor S gradually decreases. The FLP factor S of 0.26 at 6% indicates that the

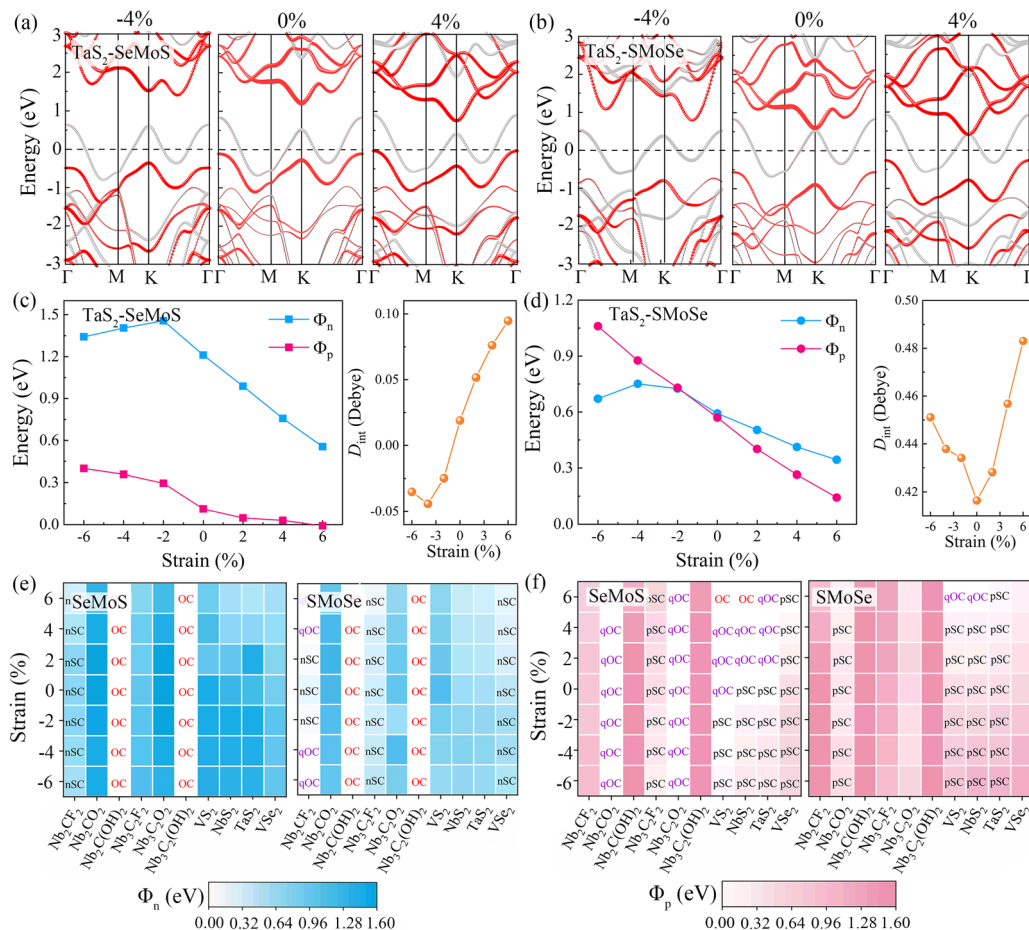


Fig. 7 The projected band structures of (a) TaS₂-SeMoS and (b) TaS₂-SMoSe contacts with different in-plane biaxial strains. Schottky barriers and interface dipole of (c) TaS₂-SeMoS and (d) TaS₂-SMoSe as a function of biaxial strains. Schottky barrier heights (e) Φ_n and (f) Φ_p of 2D metal-SeMoS and metal-SMoSe contacts with different biaxial strains. OC means Ohmic contact, qOC means quasi-Ohmic contact (Φ_n or $\Phi_p < 0.05$ eV), nSC and pSC represent n-type and p-type Schottky contacts, respectively.

applied tensile strains lead to a stronger FLP effect. On the other hand, the 2D metal-SMoSe contact exhibits a similar phenomenon, the only difference is that the value of the FLP factor S is slightly decreased to 0.29 under -6% compressive strain (from 0.31 without strain). This phenomenon is primarily attributed to the variation of the interface dipole values under strains, as shown in Fig. S11c and d (ESI[†]). As the tensile strain increases, the interface dipoles are continuously enhanced, leading to a larger ΔV and deviation from the ideal Schottky-Mott limit restrictions. This deviation results in a stronger FLP effect, which is indicated by an increase in the FLP factor S . Conversely, under different compressive strains, the interface dipole values for most 2D metal-MoSSe contacts are relatively close, resulting in a limited change in the FLP factor S .

Despite extensive research into tuning the Schottky barrier, achieving optimal operational efficiency and device performance remains challenging. Therefore, we explored the factors that affect the SBH using modern machine learning models, specifically through the sure-independence-screening-sparsifying-operator (SISSO) method with first-principles calculation results as inputs.³³ The SISSO method is a data analysis technique based

on compressed sensing, which enables the identification of an optimal low-dimensional feature descriptor from a vast number of features.⁵³ For material data, SISSO can construct a formula that describes a specific property of the material by incorporating several relevant characteristics or descriptors.⁵⁴ The trained model can predict the desired property of new materials using their values of the descriptors. Tables S5–S8 (ESI[†]) list the 180 first-principles calculation results for the SISSO machine-learning study. Herein, we used the material property columns of interest, including Φ_n , Φ_p , D_{int} , D_{tot} , ΔV , ΔW , and W_M as descriptors to train the SISSO model. Using selected primary feature classes (PFs), SISSO iteratively generates feature combinations by applying various mathematical operators, such as $+$, $-$, \times , $/$, \exp , \exp^{-1} , 2 , 3 , and $\sqrt{\quad}$. Finally, the features generated by SISSO are ranked according to the root mean square error (RMSE) to identify the optimal set of descriptors.⁵⁵ This approach enables us to gain a deeper understanding of the complex factors that contribute to the SBH, which is critical for developing high-performance MSJs.

Since only the produced descriptors with high precision and low complexity are suitable for regulating SBH, Fig. 8a and b display the Pareto charts for six proposed formulas from SISSO

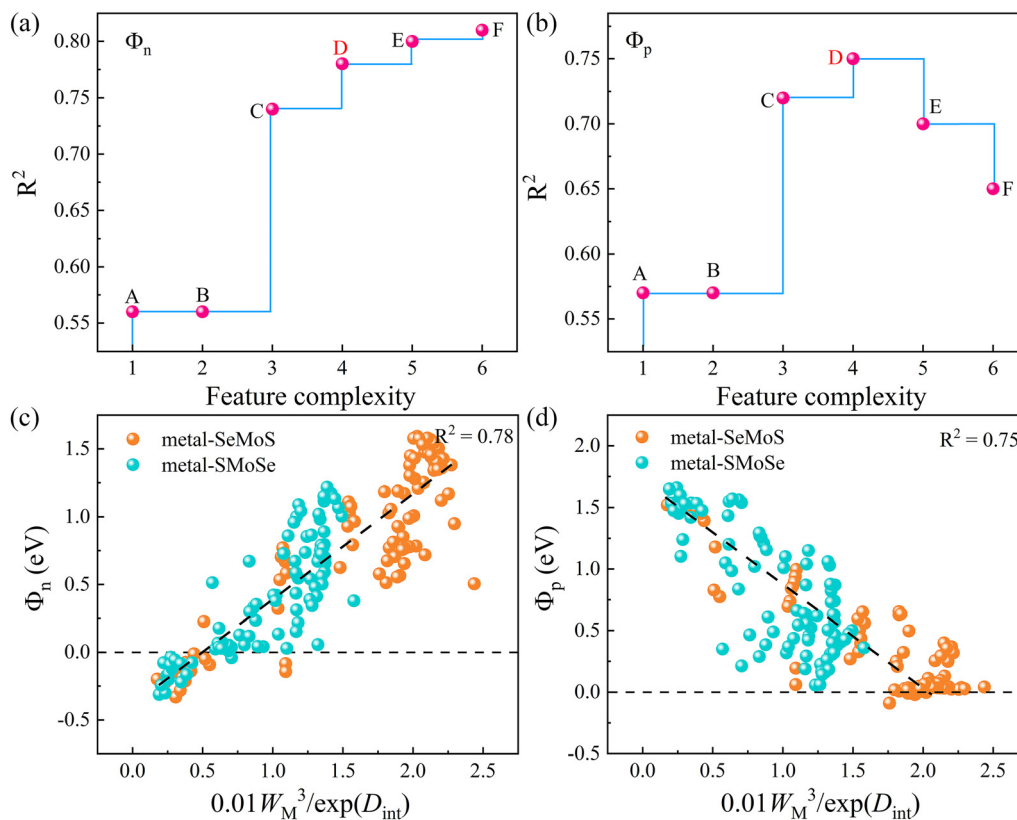


Fig. 8 Pareto chart of R^2 for (a) Φ_n and (b) Φ_p vs. complexity of the proposed mathematical formulas. The correlations of the calculated (c) Φ_n and (d) Φ_p values of metal–MoS₂ contacts vs. predicted $0.01W_M^3/\exp(D_{\text{int}})$ formula from SISSO. The data here are all derived from the DFT calculation results in this work.

Table 2 The mathematical formulas for the Pareto Chart in Fig. 8

Point	Formulas	Complexity	R^2	
			Φ_n	Φ_p
A	$0.29W_M$	1	0.56	0.57
B	$0.29W_M$	2	0.56	0.57
C	$0.05W_M \times (W_M - \Delta V)$	3	0.74	0.72
D	$0.01W_M^3/\exp(D_{\text{int}})$	4	0.78	0.75
E	$0.76(\exp(\text{cbrrt}(W_M)) - \exp(D_{\text{int}}))$	5	0.80	0.70
F	$0.32(\text{abs}((\log(W_M))^3 - (\exp(D_{\text{int}}))^2))$	6	0.81	0.65

that satisfy the criteria of both simplicity and accuracy. The detailed mathematical form and fitting error of the six formulas are shown in Table 2. It is noted that the increase in complexity cannot guarantee a decrease in the fitting error. Among them, the complexity of 4 is the best compromise between complexity and accuracy for both Φ_n and Φ_p . To clearly show the correlation, Fig. 8c and d depict the relationships of SBHs Φ_n and Φ_p with $0.01W_M^3/\exp(D_{\text{int}})$. The correlations between other mathematical formulas and Φ_n/Φ_p are also displayed in Fig. S12 (ESI[†]) for reference. It is observed that both Φ_n and Φ_p exhibit good linear relationships with $0.01W_M^3/\exp(D_{\text{int}})$. Thus, we can regulate the SBH by controlling the metal work function W_M and metal-semiconductor interface dipole D_{int} . Specifically, a large W_M and small D_{int} can result in an ideal n-type Ohmic contact

and *vice versa*, and small W_M and large D_{int} can produce a p-type Ohmic contact. Furthermore, adjusting W_M and D_{int} can also achieve quasi-Ohmic contacts with small SBH values. Therefore, by exploring the nonlinear combinations of existing features, the descriptors of W_M and D_{int} are proposed based on the SISSO to fast predict the target properties SBH in MSJs. As shown in Fig. S13 (ESI[†]), we further examined the SISSO method using supervised training data sampling (sequentially removing the data points with the smallest RMSE) to check the RMSE for different numbers of materials. The results show that the RMSE decreases sharply when the number of data points N_d ranges from 1 to 20. It is worth noting that when N_d falls within the range of 20 to 180, the trend of RMSE decline slows down. These findings suggest that the SISSO model can effectively identify the underlying physical laws even from small datasets in our study.

4. Conclusion

In summary, we have systematically investigated the geometric structure, electronic properties, and contacting barrier of MSJs formed by the contact of a series of 2D metals with Janus MoS₂Se using first-principles calculations. The projected band results indicate that the transition from Schottky contact to Ohmic

contact and tunable Schottky barrier can be achieved by selecting different 2D metals. For Nb₂C–MoSSe and Nb₃C₂–MoSSe contacts, strong interactions at the interface lead to the metalization of the bands. In contrast, weak interactions at the interface were observed for Nb₂CT, Nb₃C₂T₂ (T = F, O, OH), and MX₂ when in contact with the MoSSe monolayer. The FLP effect occurs due to the presence of interface dipoles at the MSJs interface, which deviates the band alignment from the prediction of the ideal Schottky–Mott limit. Furthermore, the values ΔV of 2D metal–MoSSe contacts are linearly related to the interface dipoles. Hence, weakening the interlayer coupling can effectively hinder the FLP effect. Additionally, the tunneling barriers of the 2D metal–MoSSe contacts depend on the interlayer coupling strength, and the Φ_{TB} values also show a good linear relationship with the work functions of 2D metals. Notably, the band edge position of MoSSe is sensitive to both layer numbers and biaxial strains. The results show that increasing the layer number and applying biaxial strain will alter the SBH and lead to the formation of Ohmic contacts. At the same time, the change in the layer number and biaxial strain will also generate different values of interfacial dipoles, affecting the FLP strength. Interestingly, we used the SISSO machine learning algorithm to construct a formula to describe the SBH of the 2D MSJs and tune the SBH. Our research offers valuable insights into tuning the Schottky barriers in MSJs and may facilitate the development of highly efficient Ohmic contacts in future electronic devices.

Author contributions

Shu Yu: investigation, formal analysis, writing – original draft. Ting Li: data curation, validation. Naihua Miao: visualization, formal analysis. Jian Gou: visualization, formal analysis. Xiaochun Huang: visualization, writing – review & editing. Rui Xiong: data curation, validation. Zhou Cui: data curation, validation. Cuilian Wen: formal analysis, writing – review & editing. Jian Zhou: formal analysis, writing – review & editing. Baisheng Sa: project administration, supervision, writing – review & editing. Zhimei Sun: supervision, formal analysis, writing – review & editing.

Conflicts of interest

The authors declare that they have no conflict of interest.

Acknowledgements

This work was supported by the National Key Research and Development Program of China (No. 2022YFB3807200), the National Natural Science Foundation of China (No. 21973012, 21905050, 52222101), the Natural Science Foundation of Fujian Province (No. 2021J06011, 2020J01351, 2020J01474, 2021H6011 and 2021J01590), and the “Qishan Scholar” Scientific Research Project of Fuzhou University.

References

- H. Wang, H. Yuan, S. S. Hong, Y. Li and Y. Cui, *Chem. Soc. Rev.*, 2015, **44**, 2664–2680.
- Q. H. Wang, K. Kalantar-Zadeh, A. Kis, J. N. Coleman and M. S. Strano, *Nat. Nanotechnol.*, 2012, **7**, 699–712.
- K. F. Mak and J. Shan, *Nat. Photonics*, 2016, **10**, 216–226.
- R. Yang, Y. Fan, Y. Zhang, L. Mei, R. Zhu, J. Qin, J. Hu, Z. Chen, Y. Hau Ng, D. Voiry, S. Li, Q. Lu, Q. Wang, J. C. Yu and Z. Zeng, *Angew. Chem., Int. Ed.*, 2023, **62**, e202218016.
- R. P. Li, Y. C. Cheng and W. Huang, *Small*, 2018, **14**, 1802091.
- W. J. Yin, B. Wen, Q. X. Ge, D. F. Zou, Y. Xu, M. W. Liu, X. L. Wei, M. Y. Chen and X. L. Fan, *Prog. Natl. Sci.: Mater. Int.*, 2019, **29**, 335–340.
- C. X. Xia, W. Q. Xiong, J. Du, T. X. Wang, Y. T. Peng and J. B. Li, *Phys. Rev. B*, 2018, **98**, 165424.
- A.-Y. Lu, H. Zhu, J. Xiao, C.-P. Chuu, Y. Han, M.-H. Chiu, C.-C. Cheng, C.-W. Yang, K.-H. Wei and Y. Yang, *Nat. Nanotechnol.*, 2017, **12**, 744–749.
- Y. Ding, G. F. Yang, Y. Gu, Y. Z. Yu, X. M. Zhang, X. Tang, N. Y. Lu, Y. K. Wang, Z. C. Dai, H. Q. Zhao and Y. H. Li, *J. Phys. Chem. C*, 2020, **124**, 21197–21206.
- H. L. Chen, W. N. Zhang, M. L. Li, G. He and X. F. Guo, *Chem. Rev.*, 2020, **120**, 2879–2949.
- B. Jiang, Z. Y. Yang, X. Q. Liu, Y. Liu and L. Liao, *Nano Today*, 2019, **25**, 122–134.
- R. T. Tung, *Appl. Phys. Rev.*, 2014, **1**, 011304.
- J. H. Kang, W. Liu, D. Sarkar, D. Jena and K. Banerjee, *Phys. Rev. X*, 2014, **4**, 031005.
- J. L. Tao Shen, X. Liu, P. Cheng, J.-C. Ren, S. Li and W. Liu, *Adv. Funct. Mater.*, 2022, **32**, 2207018.
- X. C. Liu, M. S. Choi, E. Hwang, W. J. Yoo and J. Sun, *Adv. Mater.*, 2022, **34**, 2108425.
- C. Gong, L. Colombo, R. M. Wallace and K. Cho, *Nano Lett.*, 2014, **14**, 1714–1720.
- G. S. Kim, S. H. Kim, J. Park, K. H. Han, J. Kim and H. Y. Yu, *ACS Nano*, 2018, **12**, 6292–6300.
- P. C. Shen, C. Su, Y. X. Lin, A. S. Chou, C. C. Cheng, J. H. Park, M. H. Chiu, A. Y. Lu, H. L. Tang, M. M. Tavakoli, G. Pitner, X. Ji, Z. Y. Cai, N. N. Mao, J. T. Wang, V. C. Tung, J. Li, J. Bokor, A. Zettl, C. I. Wu, T. Palacios, L. J. Li and J. Kong, *Nature*, 2021, **593**, 211–217.
- Y. Liu, J. Guo, E. B. Zhu, L. Liao, S. J. Lee, M. N. Ding, I. Shakir, V. Gambin, Y. Huang and X. F. Duan, *Nature*, 2018, **557**, 696–700.
- Y. Y. Liu, P. Stradins and S. H. Wei, *Sci. Adv.*, 2016, **2**, e160006.
- R.-S. Chen, G. Ding, Z. Feng, S.-R. Zhang, W.-A. Mo, S.-T. Han and Y. Zhou, *Adv. Funct. Mater.*, 2022, **32**, 2204288.
- V. Kamysbayev, A. S. Filatov, H. C. Hu, X. Rui, F. Lagunas, D. Wang, R. F. Klie and D. V. Talapin, *Science*, 2020, **369**, 979–983.
- A. Iqbal, F. Shahzad, K. Hantanasirisakul, M. K. Kim, J. Kwon, J. Hong, H. Kim, D. Kim, Y. Gogotsi and C. M. Koo, *Science*, 2020, **369**, 446–450.

- 24 A. VahidMohammadi, J. Rosen and Y. Gogotsi, *Science*, 2021, **372**, eabf1581.
- 25 M. Li, X. Li, G. Qin, K. Luo, J. Lu, Y. Li, G. Liang, Z. Huang, J. Zhou, L. Hultman, P. Eklund, P. O. Å. Persson, S. Du, Z. Chai, C. Zhi and Q. Huang, *ACS Nano*, 2021, **15**, 1077–1085.
- 26 J. Hafner, *J. Comput. Chem.*, 2008, **29**, 2044–2078.
- 27 J. P. Perdew and Y. Wang, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1992, **45**, 13244–13249.
- 28 G. J. Wang, L. Y. Peng, K. Q. Li, L. G. Zhu, J. Zhou, N. H. Miao and Z. M. Sun, *Comput. Mater. Sci.*, 2021, **186**, 110064.
- 29 K. Parlinski, Z. Q. Li and Y. Kawazoe, *Phys. Rev. Lett.*, 1997, **78**, 4063–4066.
- 30 J. P. Perdew, K. Burke and Y. Wang, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1996, **54**, 16533.
- 31 S. Grimme, J. Antony, S. Ehrlich and H. Krieg, *J. Chem. Phys.*, 2010, **132**, 154104.
- 32 S. Grimme, S. Ehrlich and L. Goerigk, *J. Comput. Chem.*, 2011, **32**, 1456–1465.
- 33 R. Ouyang, S. Curtarolo, E. Ahmetcik, M. Scheffler and L. M. Ghiringhelli, *Phys. Rev. Mater.*, 2018, **2**, 083802.
- 34 S. D. Tao, B. Xu, J. Shi, S. Y. Zhong, X. L. Leio, G. Liu and M. S. Wu, *J. Phys. Chem. C*, 2019, **123**, 9059–9065.
- 35 F. P. Li, W. Wei, P. Zhao, B. B. Huang and Y. Dai, *J. Phys. Chem. Lett.*, 2017, **8**, 5959–5965.
- 36 Y. Shu, Y. Liu, Z. Cui, R. Xiong, Y. Zhang, C. Xu, J. Zheng, C. Wen, B. Wu and B. Sa, *Adv. Electron. Mater.*, 2023, **9**, 2201056.
- 37 J. H. Chen, X. J. He, B. S. Sa, J. Zhou, C. Xu, C. L. Wen and Z. M. Sun, *Nanoscale*, 2019, **11**, 6431–6444.
- 38 X. H. Yang, B. S. Sa, P. Lin, C. Xu, Q. Zhu, H. B. Zhan and Z. M. Sun, *J. Phys. Chem. C*, 2020, **124**, 23699–23706.
- 39 J. Wang, X. Y. Zhou, M. J. Yang, D. Cao, X. S. Chen and H. B. Shu, *J. Mater. Chem. C*, 2020, **8**, 7350–7357.
- 40 Q. Peng, C. Si, J. Zhou and Z. M. Sun, *Appl. Surf. Sci.*, 2019, **480**, 199–204.
- 41 B. Silvi and A. Savin, *Nature*, 1994, **371**, 683–686.
- 42 A. D. Becke and K. E. Edgecombe, *J. Chem. Phys.*, 1990, **92**, 5397–5403.
- 43 B. Sa, J. Zhou, Z. Sun, J. Tominaga and R. Ahuja, *Phys. Rev. Lett.*, 2012, **109**, 096802.
- 44 B. Sa, N. Miao, Z. Sun and B. Wu, *RSC Adv.*, 2015, **5**, 1690–1696.
- 45 R. Xiong, R. Hu, Y. G. Zhang, X. H. Yang, P. Lin, C. L. Wen, B. Sa and Z. M. Sun, *Phys. Chem. Chem. Phys.*, 2021, **23**, 20163–20173.
- 46 X. Zhang, L. Feng, S. Zhong, Y. Ye, H. Pan, P. Liu, X. Zheng, H. Li, M. Qu and X. Wang, *Sci. China: Mater.*, 2023, **66**, 811–818.
- 47 Q. Q. Wang, L. M. Cao, S. J. Liang, W. K. Wu, G. Z. Wang, C. H. Lee, W. L. Ong, H. Y. Yang, L. K. Ang, S. Y. A. Yang and Y. S. Ang, *npj 2D Mater. Appl.*, 2021, **5**, 71.
- 48 J. Li, W. Q. Liu, W. H. Zhou, J. L. Yang, H. Z. Qu, Y. Hu and S. L. Zhang, *Phys. Rev. Appl.*, 2022, **17**, 054009.
- 49 J. Li, W. H. Zhou, L. L. Xu, J. L. Yang, H. Z. Qu, T. T. Guo, B. Xu, S. L. Zhang and H. B. Zeng, *Mater. Today Phys.*, 2022, **26**, 100749.
- 50 T. Shen, J. C. Ren, X. Y. Liu, S. Li and W. Liu, *J. Am. Chem. Soc.*, 2019, **141**, 3110–3115.
- 51 M. Bruening, R. Cohen, J. F. Guillemoles, T. Moav, J. Libman, A. Shanzer and D. Cahen, *J. Am. Chem. Soc.*, 1997, **119**, 5720–5728.
- 52 M. Farmanbar and G. Brocks, *Phys. Rev. B*, 2016, **93**, 085304.
- 53 G. Cao, R. Ouyang, L. M. Ghiringhelli, M. Scheffler, H. Liu, C. Carbogno and Z. Zhang, *Phys. Rev. Mater.*, 2020, **4**, 034204.
- 54 B. Weng, Z. Song, R. Zhu, Q. Yan, Q. Sun, C. G. Grice, Y. Yan and W.-J. Yin, *Nat. Commun.*, 2020, **11**, 3513.
- 55 S. H. V. Oh, W. Hwang, K. Kim, J. H. Lee and A. Soon, *Adv. Sci.*, 2022, **9**, 2104569.