



Cite this: *Soft Matter*, 2017,
13, 6363

Nanoconfinement greatly speeds up the nucleation and the annealing in single-DNA collapse[†]

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Received 25th June 2017,
Accepted 29th August 2017

DOI: 10.1039/c7sm01249g

rsc.li/soft-matter-journal

Manipulating and measuring single-molecule dynamics and reactions in nanofluidics is a rapidly growing field with broad applications in developing new biotechnologies, understanding nanoconfinement effects *in vivo*, and exploring new phenomena in confinement. In this work, we investigate the kinetics of DNA collapse in nanoslits using single T4-DNA (165.6 kbp) and λ -DNA (48.5 kbp), with particular focus on the measurement of the nucleation and annealing times. Fixing the ethanol concentration at 35% and varying the slit height from 2000 to 31 nm, the nucleation time dramatically decreases from more than 1 hour to a few minutes or less. The increased collapsed rate results from the larger free energy experienced by coiled DNA in confinement relative to compacted DNA. Our results also shed light on other conformational transitions in confinement, such as protein folding.

1. Introduction

Nanofluidics provides a platform to explore new phenomena in nanoconfinement,^{1–4} develop biotechnologies to manipulate and measure biomolecules,^{5–8} as well as understand nanoconfinement effects by membranes and cytoskeleton *in vivo*.⁹ The effects of nanoconfinement are usually caused by spatial constraint,¹⁰ high surface–volume ratio,¹ and altered hydrodynamics as well as transport properties.¹ Nanoconfinement may greatly affect conformation,¹⁰ dynamics¹⁰ and reactions^{11,12} of molecules, or provide conveniences in observing conformations and reactions¹³ through molecular stretching.

DNA molecules in nanofluidic devices have been extensively investigated to develop genome mapping technology, explore new scaling regimes of confined polymers,^{14–17} and discover novel ways of manipulating single DNA molecules.^{7,18–20} Quantitative data for the equilibrium conformation and dynamics of DNA in nanofluidics with well-defined geometries has led to great successes in critical examinations of classic theories about confined polymers and the improvement of those theories.^{10,14,15,21} In this work, we attempt to extend such quantitative analysis

from DNA equilibrium properties to non-equilibrium properties; more specifically, toward describing DNA collapse in nanofluidics. We note that other non-equilibrium DNA behaviors in nanoconfinement have also been investigated previously.^{22–26} The DNA collapse we investigate here is a kinetic process, the effect of nanoconfinement on DNA collapse can provide more general insights about how confinement affects reaction rates. In addition, DNA collapse can be used to reversibly control gene expression.²⁷ *In vivo*, DNA collapse is necessary for dense packaging of the genome.

DNA collapse in confinement has been explored in several experiments. In these experiments, the DNA size rapidly decreases after adding condensing agents (e.g. crowders,³ proteins,^{28,29} and cationic surfactant¹³). While these studies used nanofluidic confinement to visualize the collapse process, they did not study the nucleation and annealing processes, which will be described shortly, and the effect of nanoconfinement on these processes.

In this work, we subject DNA to poor solvent conditions, using a buffer based on a 35% ethanol solution, which is marginally sufficient to initiate DNA collapse in bulk and allows us to study the process in detail. Accordingly, the nucleation can take more than 1 hour³⁰ and the change in nucleation time in nanoconfinement can be readily measured.

2. Materials and methods

2.1 Overview of our experimental setup

Fig. 1 shows the overview of our experimental system to study DNA collapse. We drive a DNA molecule into a nanoslit, *i.e.* a shallow

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[†] Electronic supplementary information (ESI) available: More information about experiments, data analysis and images and videos of DNA collapse. See DOI: 10.1039/c7sm01249g

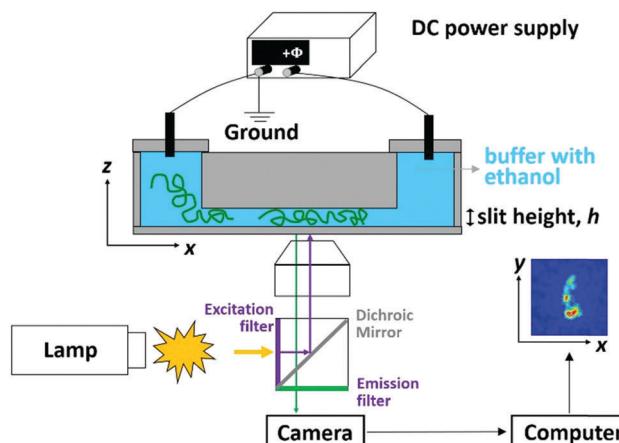


Fig. 1 Overview of our experimental system to study DNA collapse in slits. DNA molecules are driven into a nanoslit (two parallel walls with a gap in z -direction) by an electric field. The electric field is turned off once a DNA molecule enters a nanoslit. The DNA conformation projected on the x - y plane (R_{\parallel}) is recorded by fluorescent microscopy with a camera.

region confined by two parallel walls, by an electric field, and then turn off the electric field immediately after the DNA enters the slit. The DNA molecules are initially stretched by the electric field gradient at the entrance to the slit, which will be discussed in more detail below. A DNA molecule eventually collapses from an extended to a compact conformation due to the concentration of ethanol in the buffer. To obtain quantitative information about the collapse kinetics, we use fluorescent microscopy to record the DNA conformation in the nanoslits. The fluorescence intensity profile is used to calculate the in-plane radius of gyration R_{\parallel} . In addition, we also calculate the apparent aspect ratio R_M/R_m of every conformation, where R_M and R_m are the major and minor principal axes of DNA conformations, respectively. Here, we define $R_M \equiv 2\sqrt{\lambda_1}$ and $R_m \equiv 2\sqrt{\lambda_2}$, where λ_1 and λ_2 are the large and small eigenvalues of the radius of gyration tensor,³¹ respectively. To obtain the relaxation time of a DNA molecule, we calculate the self-correlation time τ_{relax} for the direction of the major principal axis of the gyration tensor.³¹ We use an exponential function to fit the self-correlation $C(t)$ starting from $C(t) \sim 0.3$. We calculate the relaxation time only for coiled conformations, not for globular conformations. See the ESI† for more details about the calculation of R_{\parallel} and τ_{relax} via image analysis.

2.2 Nanoslit fabrication and DNA sample preparation

The nanoslits are fabricated from fused-silica wafers (Mark Optics) using two layers of contact photolithography and reactive ion etching (CHF₃/O₂). The detail of nanoslit fabrication was described previously.³² The slit height h is 31, 89, 250, 549, or 2000 nm. We also perform DNA collapse experiments in bulk using a channel with the height of 100 μ m.

In our experiments, λ -DNA (48.502 kbp, New England Biolabs) or T4 DNA (165.6 kbp, Nippon gene) suspended at a concentration of 0.69 μ g mL⁻¹ is stained with YOYO-1 (Invitrogen) intercalating dye at a base pair to dye ratio of 4:1 and allowed to sit overnight. YOYO-1 staining significant increases DNA length,³³ yielding

λ -DNA with a length \sim 22 μ m and T4-DNA with a length of \sim 75 μ m. Immediately before experiments, DNA samples are diluted to produce workable concentrations of DNA. The ethanol concentration is always 35% for slit experiments, but varies for the bulk experiments as shown in Fig. 4. The experimental buffer consisted of 1.5 \times TBE (134 mM Tris base, 134 mM boric acid, and 3 mM EDTA), 4 vol% β -mercaptoethanol (BME, Cabiochem) and ethanol. The ionic strength is approximately 57 mM.³¹

2.3 Data acquisition and analysis

In order to avoid photobleaching over the experiment's duration, we turn on the fluorescent lamp and record 15–30 seconds of video every few minutes until a molecule has fully collapsed. Note that the slit confinement restricts DNA motion in the z -direction, and hence DNA molecules stay in focus during the imaging period. Sample experimental videos are included in the ESI.† For the bulk/free space experiments, DNA molecules move out of focus in a few seconds as indicated by the decrease of the total fluorescence intensity of an individual DNA. Accordingly, for the bulk experiments, we measure the hydrodynamic size of DNA based on the in-plane (X - Y) diffusivity as described below.

Fig. 2 shows examples of the time evolution of R_{\parallel} for a T4-DNA molecule at different time intervals. We analyze about 20–30 different molecules for each channel height h at various times T_{start} . We find that the variance of R_{\parallel} among these molecules is small (Fig. 3). We further average R_{\parallel} for these molecules with the same h and T_{start} . Eventually, we obtain the time evolution of R_{\parallel} for each slit height as shown in Fig. 5. Similar data analysis is also applied to obtain the apparent aspect ratio R_M/R_m .

Direct calculation of the DNA size from the intensity profile is limited in resolution and accuracy due to the point-spread function of the microscope.³¹ This problem is more profound when DNA assumes a highly compact conformation. As a result,

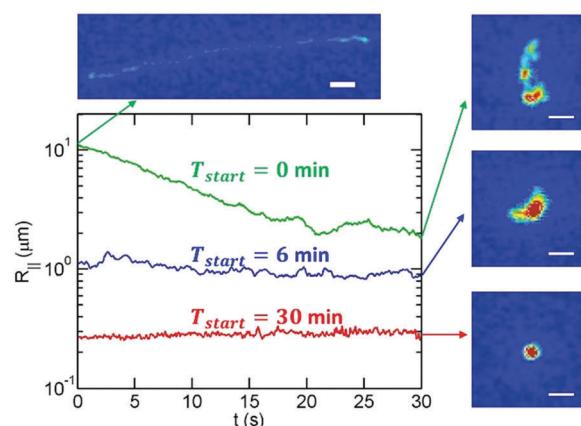


Fig. 2 The time evolution of R_{\parallel} for T4-DNA in a nanoslit with $h = 89$ nm and 35% ethanol. The top image shows a stretched DNA molecule induced by the electric field that drives the DNA into a nanoslit. In order to avoid photobleaching over the long-time experiments, we turn on the fluorescent lamp and take 15–30 s video every few minutes until a molecule has fully collapsed. The green, blue, and red curves correspond to videos taken at the times $T_{\text{start}} = 0, 6, 30$ min, respectively. The scale bars are 2 μ m.

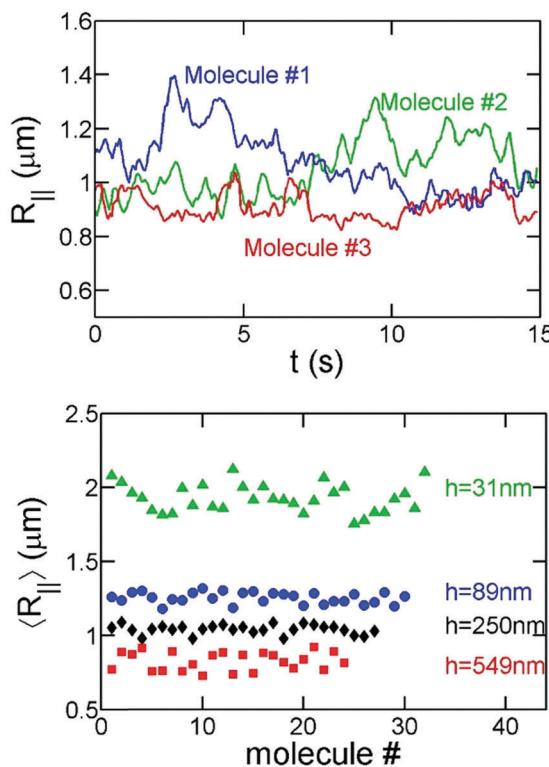


Fig. 3 Comparison of R_{\parallel} for different T4-DNA molecules. (top) Three curves correspond to videos taken for three different DNA molecules with $h = 250$ nm and 35% ethanol at $T_{\text{start}} = 2$ min. (bottom) Each data point correspond to R_{\parallel} for a T4-DNA molecule averaged over 15–30 s. Different symbols correspond to different slit heights. All data points are measured at the same $T_{\text{start}} = 2$ min.

we also measure the DNA diffusion coefficient D , and then calculate the DNA hydrodynamic radius through $R_h = k_B T / (6\pi\eta D)$. The viscosity η for different ethanol concentrations are determined by measuring the mean squared displacement of polystyrene beads of known radius. We use this method for our measurement of DNA size in bulk, *i.e.* free space.

It is worth mentioning that a given DNA molecule is stretched by the electric field when driven into a nanoslit as shown by the top-left image in Fig. 2. The green curve in Fig. 2 indicates that the effect of initial stretching in T4-DNA vanishes after about 20 seconds for the slit height of 89 nm. Our previous experiments³⁴ systematically measured the duration τ_{str} for a stretched DNA molecule to reach its equilibrium conformation in a nanoslit and found the empirical formula $\tau_{\text{str}} \sim L_{\text{DNA}}^{2.2} h^{-0.5}$. We estimate that the effect of initial stretching by the electric field will vanish in less than 1 minute for all slit heights used in the current study.

The electric field that is used to drive DNA into a nanoslit may introduce concentration polarization at the microchannel–nanochannel–microchannel interfaces wherein the concentrations of ions in the buffer do not distribute homogeneously.³⁵ Prior work has shown that concentration polarization primarily depends on an inverse Dukhin number $Fhzc/\sigma$, where F is Faraday's constant, h is the slit height, z is the valence of the ion, c is the concentration of the ion, and σ is the surface charge

density of slit walls.³⁵ Substituting $z = 1$, $c \approx 57$ mM, and $\sigma \approx 60$ mC m⁻²,³⁶ we obtain the inverse Dukhin numbers of approximately 183, 50, 23, 8.2 and 2.8 for $h = 2000$, 549, 250, 89 and 31 nm, respectively. These inverse Dukhin numbers are greater than unity, which suggests that concentration polarization is not significant in our experiments.

3. Results and discussions

3.1 DNA collapse in free space

We first present our experimental results of DNA collapse in free space. The free-space experiments are used to estimate the critical ethanol concentration for DNA collapse, and later we pick an ethanol concentration in the vicinity of the critical concentration so that the duration of DNA collapse is sufficiently long and we can measure the effect of confinement on DNA collapse more conveniently.

Fig. 4a shows the time evolution of the hydrodynamic radius, R_h , of T4-DNA for a range of ethanol concentration c_{ethanol} from 0% to 60%. For $c_{\text{ethanol}} \lesssim 30\%$, T4-DNA molecules always assume coiled conformations in 24 hours and show significant conformational fluctuations (Fig. 4b). Accordingly, we consider that T4-DNA or λ -DNA collapse do not take place for $c_{\text{ethanol}} \lesssim 30\%$. Note that the typical relaxation time of DNA conformations is on the order of seconds and our maximal observation time 24 hours is many orders of magnitude longer than the relaxation time. For $30\% \lesssim c_{\text{ethanol}} \lesssim 45\%$, T4-DNA molecules collapse towards globular conformations (Fig. 4b). For globular conformations, we can still observe conformational fluctuations, such as a short DNA fragment occasionally stretching out from a globular conformation (see experimental videos in ESI†). For $c_{\text{ethanol}} \gtrsim 45\%$, T4-DNA molecules eventually reach more compact conformations with little conformational fluctuations (Fig. 4b). These compact conformations are expected

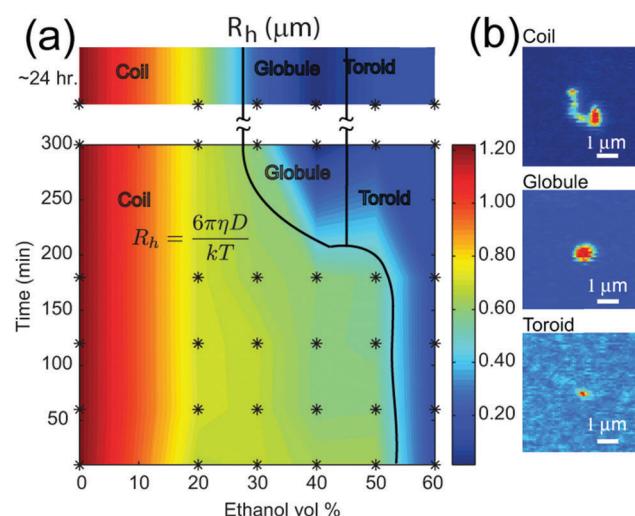


Fig. 4 (a) Time evolution of hydrodynamic size R_h (based on a diffusivity measurement) of T4 DNA in solution with different ethanol concentrations. (b) Experimental images of T4-DNA molecules in three states: a coil, a globule and a toroid.

to be toroidal based on previous experimental^{37,38} and simulation^{39,40} studies of condensed DNA. The duration until DNA collapse depends on the ethanol concentration. For $30\% \lesssim c_{\text{ethanol}} \lesssim 50\%$, DNA collapse takes hours.

Single-molecule DNA collapse under a wide range of ethanol concentration has also been investigated using magnetic tweezers and Atomic Force Microscopy (AFM) by Wang *et al.*³⁷ They found that λ -DNA collapse occurs within a few hours incubation time for $c_{\text{ethanol}} \gtrsim 30\%$, and the condensed λ -DNA structures become more compact with the increasing ethanol concentration. At $c_{\text{ethanol}} = 50\%$, they found that DNA reaches condensed states between two and three hours. All of these observations are in agreement with our results.

Based on these results for DNA in bulk, we pick the ethanol concentration of 35% for nanoslit experiments. With 35% ethanol, T4-DNA is fully collapsed after about 220 minutes and eventually reaches globular conformations.

3.2 DNA collapse in nanoslits

Now, we move to the experimental results of DNA collapse in nanoslits using the fixed ethanol concentration of 35%. To monitor the conformation change during DNA collapse, we calculate both the average in-plane radius of gyration $\langle R_{\parallel} \rangle$ and the apparent aspect ratio $\langle R_M/R_m \rangle$ of single DNA conformations based on the DNA intensity profiles, as shown in Fig. 5. Recall that R_M and R_m are the major and minor principal axes of DNA conformation.⁴¹ The observations of $\langle R_{\parallel} \rangle$ (two top plots in Fig. 5) and $\langle R_M/R_m \rangle$ (two bottom plots in Fig. 5) are consistent with each other. During DNA collapse, the DNA size becomes smaller and the apparent aspect ratio decreases towards unity, which indicates nearly isotropic conformations.

With the decrease of the slit height from $h = 2000$ nm to $h = 31$ nm, the collapse time decreases from more than one hour

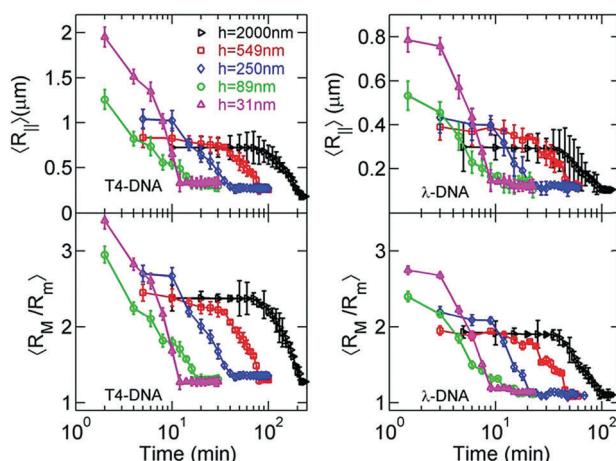


Fig. 5 In-plane radius of gyration $\langle R_{\parallel} \rangle$ (top) and apparent aspect ratio of conformation $\langle R_M/R_m \rangle$ (bottom) as a function of the incubation time in 35% ethanol for T4 DNA (left) and λ -DNA (right) confined to different slit channels. We calculate $\langle R_{\parallel} \rangle$ and $\langle R_M/R_m \rangle$ based on the DNA intensity profile. The error bars indicate the standard deviations among the $\langle R_{\parallel} \rangle_{\text{interval}}$ of 20–30 DNA molecules under a certain condition, where $\langle R_{\parallel} \rangle_{\text{interval}}$ is the average size of a given DNA over an interval of 15–30 seconds.

to about 10 minutes. To elaborate the process of DNA collapse, we re-plot the time evolution of $\langle R_{\parallel} \rangle$ for T4-DNA in Fig. 6 but with a linear time-scale for $h = 2000$ nm, $h = 250$ nm, and $h = 31$ nm. For $h = 2000$ nm, the collapse process exhibits three stages: a nucleation state where $\langle R_{\parallel} \rangle$ remains approximately constant, an annealing stage where $\langle R_{\parallel} \rangle$ rapidly decreases, and a final stage where the DNA is collapsed. These stages are similar to those observed in previous experiments of single-DNA collapse.³⁰ The nucleation in our experiments is reasonable considering that the DNA–DNA attraction induced by 35% ethanol is marginally larger than the critical attraction and the condensed state is stable only when its size exceeds a certain value, *i.e.* the nucleation size. We define the nucleation time τ_{nuc} for the first stage and the annealing time τ_{anneal} for the second

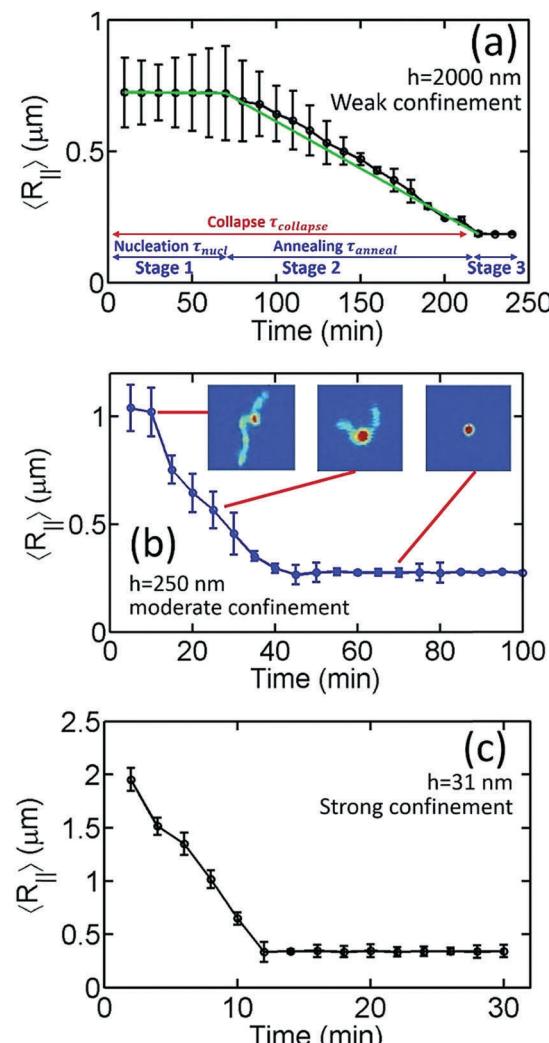


Fig. 6 Plot of $\langle R_{\parallel} \rangle$ for T4-DNA in (same as in Fig. 5), but with a linear scale for the incubation time for (a) $h = 2000$ nm, (b) $h = 250$ nm, and (c) $h = 31$ nm. For (a), the collapse process exhibits three stages. The intersection of the two green lines is used to determine the nucleation time. Three inset images in (b) show one T4-DNA molecule at $t = 10$, 25 and 70 min. The error bars indicate the standard deviations among the $\langle R_{\parallel} \rangle_{\text{interval}}$ of 20–30 DNA molecules under a certain condition, where $\langle R_{\parallel} \rangle_{\text{interval}}$ is the average size of a given DNA over an interval of 15–30 seconds.

stage as shown in Fig. 6a. We also define the collapse time $\tau_{\text{collapse}} = \tau_{\text{nuc}} + \tau_{\text{anneal}}$ for the entire process.

Now we proceed to more quantitative analysis of the time scales τ_{nuc} , τ_{anneal} , and τ_{collapse} for DNA collapse. We determine τ_{collapse} as follows. We calculate the average value and standard deviation of R_{\parallel} over the time interval that the DNA is clearly in the collapsed stage 3; e.g. $t \geq 230$ minutes for T4-DNA with $h = 2000$ nm, and $t \geq 14$ minutes for T4-DNA with $h = 31$ nm. We refer the average value as $\langle R_3 \rangle$ and the standard deviation as σ_3 . Then, τ_{collapse} is defined as the earliest time that R_{\parallel} is less than $\langle R_3 \rangle + \sigma_3$. The collapse times are listed in Table 1 and plotted in Fig. 7. We determine τ_{nuc} as follows. We approximate that the evolution of R_{\parallel} in stage 1 and stage 2 can be fit by two straight lines, respectively, and the intersection of these two straight lines correspond to τ_{nuc} . We adjust the lines to minimize the error in the fit (details provided in the ESI†). This method yields $\tau_{\text{nuc}} = 70$ min for $h = 2000$ nm, $\tau_{\text{nuc}} = 35$ min for $h = 549$ nm, and $\tau_{\text{nuc}} = 10$ min for $h = 250$ nm in the case of T4-DNA. We stress that this is an approximate method to estimate the time at which the DNA transitions from stage 1 to stage 2 that we define as τ_{nuc} . For T4-DNA with $h = 89$ nm and $h = 31$ nm, there is no clear distinction between stages 1 and 2, which can be caused by three possible reasons: (i) no nucleation exists for these two slit heights (see more discussion in Section 3.3); (ii) the nucleation is less than a few minutes, which cannot be resolved by our data sampling rate; (iii) the nucleation is less than a few minutes, and the plateau in the evolution of $\langle R_{\parallel} \rangle$ is convoluted with the initial DNA stretching due to the electric field (see Fig. 2 for an example). The third possible reason is unlikely, because the effect of the initial DNA stretching vanishes in less than one minute as discussed in the methods section. It is worth pointing out that the data in Fig. 6 show a clear trend that the nucleation time is shorter for a smaller slit height from $h = 2000$ nm to 89 nm, and the transition from stage 2 to 3 is clearly seen. The speeds of DNA collapse for $h = 89$ nm and $h = 31$ nm are close and so it is unclear which one is faster. Note that it is also difficult to precisely identify the spatial location of the nucleation core directly from image analysis, because the high-intensity spots seen in nucleation cores are hard to distinguish from similar spots caused by thermal fluctuations in the projected DNA conformation (see an example of the coil in Fig. 4b and more images in the ESI†). Only in a few experiments are we able to observe the nucleation process as shown by the imaging of Fig. 6b. In the previous AFM experiments by Wang *et al.*,³⁷ DNA collapse sometimes starts from non-local contacts of DNA segments, which are separated

Table 1 The time scales for the collapse of T4-DNA and λ -DNA. The data for λ -DNA are shown in square brackets. A nucleation time of 0 min is assigned if there is no clear transition from stage 1 to stage 2

	τ_{collapse} (min)	τ_{nuc} (min)	τ_{anneal} (min)
$h = 2000$ nm	220 [95]	70 [35]	150 [60]
$h = 549$ nm	80 [45]	35 [24]	45 [21]
$h = 250$ nm	45 [21]	10 [9]	35 [12]
$h = 89$ nm	16 [12]	0 [0]	16 [12]
$h = 31$ nm	12 [9]	0 [0]	12 [9]

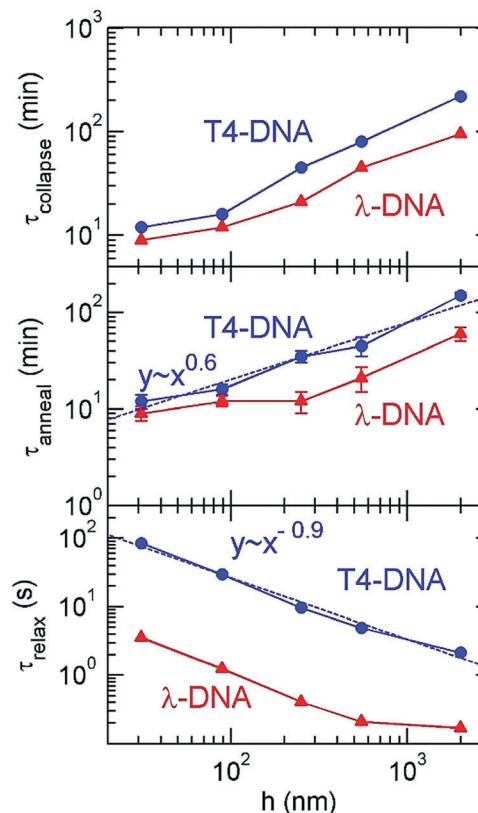


Fig. 7 The collapse time, annealing time and rotational relaxation time as a function of the slit height for T4 DNA and λ -DNA.

by a big loop that assumes a loose conformation. If the nucleation occurs due to non-local contacts, it is even more difficult to identify the nucleation core location from the image analysis.

The five slit heights in our experiments vary from $h = 2000$ nm to $h = 31$ nm. To evaluate the strength of confinement by these slits, we compare the slit height with the DNA persistence length in 35% ethanol and the DNA size in bulk. The DNA persistence length decreases from $L_p \approx 50$ nm to $L_p \approx 22$ nm after adding 35% ethanol, which was determined by previous experiments by Wang *et al.*³⁷ The smallest slit height $h = 31$ nm is close to $L_p \approx 22$ nm. To calculate the size of DNA coils in bulk, we first analyze the effective DNA diameter w in 35% ethanol. The value of w should be close to zero, because 35% ethanol concentration is close to the critical value for DNA collapse, and the solvent quality is close to the θ -condition. As such, we approximate YOYO-1 stained DNA in 35% ethanol as a worm-like chain with $L_p \approx 22$ nm and $w \approx 0$ nm. We obtain λ -DNA size $\langle R_{\parallel, \text{bulk}}^{\lambda\text{-DNA}} \rangle \approx 0.33$ μm and T4-DNA size $\langle R_{\parallel, \text{bulk}}^{\text{T4-DNA}} \rangle \approx 0.61$ μm after using $L^{\lambda\text{-DNA}} \approx 22$ μm , $L^{\text{T4-DNA}} \approx 75$ μm and applying the theoretical formula $\langle R_{\parallel, \text{bulk}} \rangle = \sqrt{LL_p/3} \times \sqrt{2/3}$, where $\sqrt{2/3}$ is used for the in-plane rather than three dimensional radius of gyration. The largest slit height $h = 2000$ nm is about six times of $\langle R_{\parallel, \text{bulk}}^{\lambda\text{-DNA}} \rangle \approx 0.33$ μm , and hence λ -DNA experiences negligible confinement, and the initial λ -DNA size for $h = 2000$ nm is $\langle R_{\parallel} \rangle \approx 0.30$ μm , which is close to $\langle R_{\parallel, \text{bulk}}^{\lambda\text{-DNA}} \rangle$. For T4-DNA, The largest slit

height $h = 2000$ nm is about three times of $\langle R_{\parallel, \text{bulk}}^{\text{T4-DNA}} \rangle \approx 0.61 \mu\text{m}$, and T4-DNA experiences weak confinement. The initial T4-DNA size for $h = 2000$ nm is $\langle R_{\parallel} \rangle \approx 0.72 \mu\text{m}$, which is larger than $\langle R_{\parallel, \text{bulk}}^{\text{T4-DNA}} \rangle$ due to DNA compression by slit walls. Note that our previous experiment³¹ obtained the λ -DNA size $\langle R_{\parallel, \text{bulk}}^{\lambda\text{-DNA}} \rangle \approx 0.52 \mu\text{m}$ in the absence of ethanol, which is larger than $0.33 \mu\text{m}$ in the current study.

Our experiments employ two types of DNA molecules: T4-DNA with $L \approx 75 \mu\text{m}$ and λ -DNA with $L \approx 22 \mu\text{m}$. Despite of the difference in the contour length by a factor of 3.41, both DNA molecules exhibit the same trend in confinement. Quantitatively, the difference in the contour length leads to the differences in confinement free energy and the relaxation time τ_{relax} , and then the difference in time scales of DNA collapse. Based on the blob-model scaling $\tau_{\text{relax}} \sim L^{-5/2}$ in confinement,³¹ the relaxation of T4-DNA is expected to be slower than λ -DNA by a factor of 21, which is close to $\tau_{\text{relax}}^{\text{T4-DNA}}/\tau_{\text{relax}}^{\lambda\text{-DNA}} \approx 24$ observed in our experimental measurement of rotational relaxation time over the range from $h = 31 \text{ nm}$ to 549 nm . The relaxation time of T4-DNA appears to follow a scaling $\tau_{\text{relax}} \sim h^{-0.9}$ from $h = 31$ to 2000 nm , while the relaxation time of λ -DNA significantly deviates from this scaling at $h = 2000 \text{ nm}$, because λ -DNA experiences negligible confinement at $h = 2000 \text{ nm}$ due to the insufficiency in the contour length. Although T4-DNA and λ -DNA differ in the relaxation time by a factor of about 24, the difference in the collapse time is only by a factor of about 1.8, and the difference in the annealing times is by a factor of about 2.1 (see ESI†). As previously investigated by Yoshinaga,⁴² the annealing speed may be determined by the motion of the collapsed domain or the coil domain depending on the stage of collapse process. If it is the former, the annealing time is insensitive to the chain length, while if it is the latter, the annealing time is related to the relaxation time of the entire length. The mixture of both motions in the collapse process may result in a weak dependence of the annealing time or collapse time on the chain length.

3.3 Discussion about how confinement speeds up DNA collapse

Now we proceed to the theoretical explanation for the effect of confinement on DNA collapse. We highlight that as the slit becomes shallower, the relaxation dramatically slows down, but conversely the collapse kinetics greatly speeds up (Fig. 7). These two trends indicate that the faster collapse in confinement is driven by the effect of confinement on the free energy landscape and not by hydrodynamic effects. Fig. 8 illustrates the effect of confinement on the free energy landscape with respect to the compactness of DNA conformations. DNA collapse corresponds to a transition from a coiled conformation to a globular conformation. The confinement free energy experienced by the coiled conformation $F_{\text{coil}}^{\text{slit}}$ is larger than the one by the globular conformation $F_{\text{globule}}^{\text{slit}}$, because the coiled conformation is larger and more likely to be restricted by slit walls. The difference $\Delta F_{\text{slit}} = F_{\text{coil}}^{\text{slit}} - F_{\text{globule}}^{\text{slit}}$ reshapes the free energy landscape and thermodynamically promotes the coil-globule transition in confinement. In our previous simulation study,⁴³

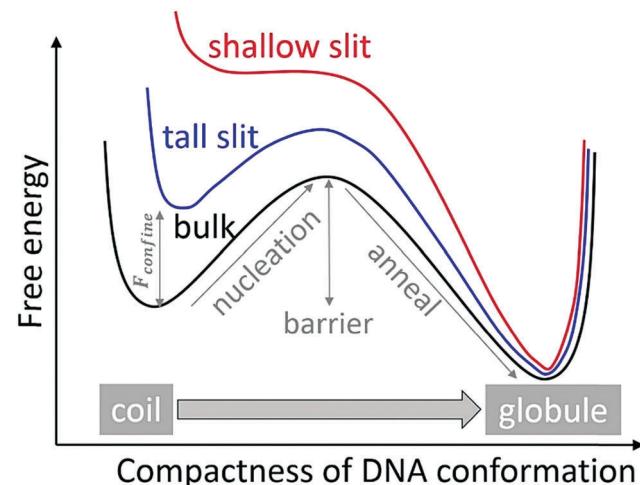


Fig. 8 Illustration of the effect of confinement on the free energy landscape with respect to the compactness of DNA conformation. DNA collapse corresponds to the transition from the left to the right with the increase of the compactness of DNA conformations. Because confinement significantly increases the free energy of the coiled conformation, the energy barrier separating the coil and globule shape is reduced in the shallow slit.

we have calculated the free energy landscape for the coil-globule transition in slit confinement using computer simulations and obtained similar curves as shown in Fig. 8. More generally, confinement promotes any reaction or conformation change that compacts the polymer conformation, such as protein folding^{44,45} and knot formation in confinement.⁴⁶

Due to the existence of plateaus in the time evolution of $\langle R_{\parallel} \rangle$, we expect that there is a free energy barrier along the path of DNA collapse. The free energy barrier is akin to that found in typical crystallization of small molecules, namely that the small nuclei are unstable due to a relatively large surface energy. Here, the surface energy refers to an energy increase because the surface monomers have a smaller number of attractive neighbors than the monomers in the core. The dependence of confinement free energy on the compactness of DNA conformation also reduces the free energy barrier for DNA collapse. The reduction of free energy barrier by confinement can explain our observation that the nucleation process becomes shorter or disappears in shallower slits.

Our experimental data can be understood quantitatively to a certain extent. We focus on the data of T4-DNA rather than λ -DNA, because T4-DNA is longer and it suffers less finite-length effects when applying scaling relationships. We estimate the confinement free energy per unit length f_{confine} by the scaling relationship¹⁵

$$\frac{f_{\text{confine}}}{k_B T} \approx \left(\frac{\pi^2}{3}\right) \left(\frac{h}{L_p}\right)^{-2} \frac{1}{L_p}, \quad (1)$$

where k_B is the Boltzmann constant, and T is the temperature. Substituting $L_p \approx 22 \text{ nm}$ and $h = 31, 89, 250, 549$ and 2000 nm , we have the confinement free energy per unit length as 75, 9.1, 1.2, 0.24, and $0.018 k_B T \mu\text{m}^{-1}$. To proceed with the quantitative

estimation using limited data points, we adopt a crude approximation for the nucleation time

$$\tau_{\text{nucl}} \sim \tau_{\text{relax}} \exp(F_{\text{barrier}}^{\text{slit}}/k_{\text{B}}T), \quad (2)$$

$$F_{\text{barrier}}^{\text{slit}} = F_{\text{barrier}}^{\text{bulk}} + f_{\text{confine}} L_{\text{nucl}} \quad (3)$$

where $F_{\text{barrier}}^{\text{bulk}}$ is the barrier in bulk, and L_{nucl} is the DNA contour length inside the nucleation core and is assumed to be unchanged with confinement. From $h = 2000$ nm to $h = 549$ nm, the relaxation time increases from 2.15 to 4.97 seconds by a factor of 2.31, and the nucleation time decreases from 70 to 35 min by a factor of 2. Accordingly, we have $(f_{\text{confine}}^{h=549\text{nm}} - f_{\text{confine}}^{h=2000\text{nm}}) L_{\text{nucl}} \approx \ln(2 \times 2.31) k_{\text{B}}T$. Eventually, we obtain $L_{\text{nucl}} \approx 6.9$ μm . Due to the confinement free energy experienced by the DNA fragment with a length of 6.9 μm , the free energy barrier of nucleation is lowered by $f_{\text{confine}}^{h=2000\text{nm}} L_{\text{nucl}} \approx 0.1 k_{\text{B}}T$ for $h = 2000$ nm, and by $f_{\text{confine}}^{h=549\text{nm}} L_{\text{nucl}} \approx 1.65 k_{\text{B}}T$ for $h = 549$ nm. The nucleation size of 6.9 μm is reasonable considering that the nucleation is a fraction of λ -DNA length of 22 μm . For example, the first image of Fig. 6b contains a nucleation core with DNA contour length of about 8.6 μm based on the analysis of the intensity profile.

We also adopt a crude approximation to quantitatively analyze the annealing time. We assume the annealing speed is proportional to the slope of free energy landscape as labeled in Fig. 8:

$$\tau_{\text{anneal}} \sim \tau_{\text{relax}}/\varepsilon_{\text{slit}}, \quad (4)$$

$$\varepsilon_{\text{slit}} = \varepsilon_{\text{bulk}} + (\pi^2/3)h^{-2}L_p \quad (5)$$

where $\varepsilon_{\text{slit}}$ is the decrease of the free energy during DNA collapse after crossing the free energy barrier. This formula is similar with the one proposed by Yoshinaga,⁴² $\tau_{\text{anneal}} \sim \eta/\varepsilon$, while we replace η by τ_{relax} . This model views DNA collapse as a ball moving downhill with the resistance produced by the solvent. We assume that $\varepsilon_{\text{bulk}}$ is much smaller than the second term in eqn (5) and is dropped, because we use an ethanol concentration close to the critical value for DNA collapse. Hence, we further simplify the annealing time to

$$\tau_{\text{anneal}} \sim \tau_{\text{relax}}/h^{-2}. \quad (6)$$

Using the de Gennes scaling $\tau_{\text{relax}} \sim h^{-7/6}$, we obtain $\tau_{\text{anneal}} \sim h^{5/6}$. Considering that many assumptions are made, the predicted exponent of 5/6 agrees fairly well with the exponent 0.6 ± 0.2 obtained by the fit to our experiment results of τ_{anneal} . In particular, we ignore $\varepsilon_{\text{bulk}}$ in eqn (5). The mixture of $\varepsilon_{\text{bulk}} \sim h^0$ and $(\pi^2/3)h^{-2}L_p$ would result in an apparent scaling exponent larger than -2 , and then an exponent in τ_{anneal} less than 5/6.

Previous simulations by Das and Chakraborty have also quantitatively investigated the effect of slit confinement on polymer collapse, however, only for short flexible chains.⁴⁷ After including hydrodynamic interactions, the slit confinement can speed up polymer collapse by a factor from 2.5 to 4 for the chain length from 80 to 200. If we extrapolate the confinement effect to longer chains, it may explain our experimental observation that the slit confinement speeds up DNA collapse by 18 times.

Recall that T4-DNA used in our experiments has a contour length ~ 75 μm , which is about 3400 times of DNA the persistence length ~ 22 nm in 35% ethanol.

3.4 Discussion about the surface charge of the slit walls

In addition to the steric repulsion, the slit walls also affect DNA behavior through the surface charge.³⁶ The silica nanoslits used here have negative surface charges. The electrostatic repulsion between slit walls and DNA makes the accessible height by DNA less than the actual slit height, *i.e.* $h_{\text{actual}} = h - 2\delta$, where δ is the effective thickness of the surface layer inaccessible by DNA. The calculation of δ is rather complicated, but δ depends on the Debye length and is on the order of nanometres in our studies.^{2,48} Accordingly, we expect the difference between h_{actual} and h is small in our studies and does not affect our overall conclusions.

The counterions (more specifically, H^+ in the current study) of surface charge also change the pH value of the buffer in nanoslits. Previous experiments by Bottenus *et al.*⁴⁹ measured the change in the pH value in nanochannels over a range of buffer conditions. In a channel with a cross-section of 100 nm \times 400 nm, the pH value shift is about 0.2 at an ionic strength of 57 mM.⁴⁹ This prior work suggests that the changes in pH value for $h = 2000$, 549, 250 and 89 nm are smaller than or around 0.2. Such a small change in pH value should not change DNA properties, such as charge density.

4. Conclusions

In summary, single molecule experiments show that nanoconfinement greatly speeds up the nucleation and the annealing of single-DNA collapse, despite the fact that DNA dynamics is slowed down by confinement. If we view DNA collapse as a reaction from a metastable coiled state to a globule state, then nanoconfinement can be considered as a catalyst to accelerate this reaction by reshaping the free energy landscape and promoting smaller conformations. What is different from a traditional catalyst is that the relative free energy between the coil (reactant) and the globule (product) is also altered by confinement. However, if DNA is driven into confinement before reaction and driven out of confinement after reaction, then confinement essentially acts as a catalyst by changing the free energy barrier. Such catalysis by confinement may be used by cells to speed up protein folding. Experiments^{44,50} and simulations⁴⁵ suggest that the nanocage-confinement by chaperonin may be a key factor, among others,⁵¹ facilitating protein folding. Confinement is also suggested to affect other conformation transitions, *e.g.* DNA denaturation,⁵² polymer looping⁵³ and protein dimerization.⁵⁴ The direct visualization of long DNA in our experiment allows for quantitative measurements and sheds light on conformation transitions in confinement.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work is supported by the Singapore-MIT Alliance for Research and Technology (SMART) and the National Science Foundation (NSF) grant CBET-1602406.

References

- 1 R. B. Schoch, J. Han and P. Renaud, *Rev. Mod. Phys.*, 2008, **80**, 839.
- 2 W. Reisner, J. N. Pedersen and R. H. Austin, *Rep. Prog. Phys.*, 2012, **75**, 106601.
- 3 C. Zhang, P. G. Shao, J. A. van Kan and J. R. van der Maarel, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 16651.
- 4 D. J. Mai, C. Brockman and C. M. Schroeder, *Soft Matter*, 2012, **8**, 10560.
- 5 E. T. Lam, A. Hastie, C. Lin, D. Ehrlich, S. K. Das, M. D. Austin, P. Deshpande, H. Cao, N. Nagarajan, M. Xiao and P. Y. Kwok, *Nat. Biotechnol.*, 2012, **30**, 771.
- 6 R. Marie, J. N. Pedersen, D. L. Bauer, K. H. Rasmussen, M. Yusuf, E. Volpi, H. Flyvbjerg, A. Kristensen and K. U. Mir, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, **110**, 4893–4898.
- 7 W. Reisner, N. B. Larsen, H. Flyvbjerg, J. O. Tegenfeldt and A. Kristensen, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 79–84.
- 8 W. Reisner, N. B. Larsen, A. Silahtaroglu, A. Kristensen, N. Tommerup, J. O. Tegenfeldt and H. Flyvbjerg, *Proc. Natl. Acad. Sci. U. S. A.*, 2010, **107**, 13294–13299.
- 9 H.-X. Zhou, G. Rivas and A. P. Minton, *Annu. Rev. Biophys.*, 2008, **37**, 375.
- 10 W. Reisner, K. J. Morton, R. Riehn, Y. M. Wang, Z. Yu, M. Rosen, J. C. Sturm, S. Y. Chou, E. Frey and R. H. Austin, *Phys. Rev. Lett.*, 2005, **94**, 196101.
- 11 A. Küchler, M. Yoshimoto, S. Luginbühl, F. Mavelli and P. Walde, *Nat. Nanotechnol.*, 2016, **11**, 409.
- 12 T. E. Sintra, S. P. Ventura and J. A. Coutinho, *J. Mol. Catal. B: Enzym.*, 2014, **107**, 140.
- 13 G. Henkin, D. Berard, F. Stabile, M. Shayegan, J. S. Leith and S. R. Leslie, *Anal. Chem.*, 2016, **88**, 11100.
- 14 Y. Wang, D. R. Tree and K. D. Dorfman, *Macromolecules*, 2011, **44**, 6594.
- 15 L. Dai, J. R. C. Van der Maarel and P. S. Doyle, *Macromolecules*, 2014, **47**, 2445.
- 16 L. Dai, C. B. Renner and P. S. Doyle, *Adv. Colloid Interface Sci.*, 2016, **232**, 80.
- 17 L. Dai, D. R. Tree, J. R. C. van der Maarel, K. D. Dorfman and P. S. Doyle, *Phys. Rev. Lett.*, 2013, **110**, 168105.
- 18 D. J. Berard, F. Michaud, S. Mahshid, M. J. Ahamed, C. M. McFaul, J. S. Leith, P. Bérubé, R. Sladek, W. Reisner and S. R. Leslie, *Proc. Natl. Acad. Sci. U. S. A.*, 2014, **111**, 13295.
- 19 A. R. Klotz, M. Mamaev, L. Duong, H. W. de Haan and W. W. Reisner, *Macromolecules*, 2015, **48**, 4742.
- 20 J.-W. Yeh, A. Taloni, Y.-L. Chen and C.-F. Chou, *Nano Lett.*, 2012, **12**, 1597.
- 21 C. Zhang, F. Zhang, J. A. van Kan and J. R. C. van der Maarel, *J. Chem. Phys.*, 2008, **128**, 225109.
- 22 J. Tang, D. W. Trahan and P. S. Doyle, *Macromolecules*, 2010, **43**, 3081–3089.
- 23 A. C. Klepinger, M. K. Greenier and S. L. Levy, *Macromolecules*, 2015, **48**, 9007–9014.
- 24 A. Khorshid, S. Amin, Z. Zhang, T. Sakaue and W. W. Reisner, *Macromolecules*, 2016, **49**, 1933–1940.
- 25 M. J. Ahamed, S. Mahshid, D. J. Berard, F. O. Michaud, R. Sladek, W. W. Reisner and S. R. Leslie, *Macromolecules*, 2016, **49**, 2853–2859.
- 26 D. Kim, C. Bowman, J. T. Del Bonis-O'Donnell, A. Matzavinos and D. Stein, *Phys. Rev. Lett.*, 2017, **118**, 048002.
- 27 A. Estevez-Torres and D. Baigl, *Soft Matter*, 2011, **7**, 6746.
- 28 C. Zhang, D. Guttula, F. Liu, P. P. Malar, S. Y. Ng, L. Dai, P. S. Doyle, J. A. van Kan and J. R. van der Maarel, *Soft Matter*, 2013, **9**, 9593.
- 29 C. Zhang, K. Jiang, F. Liu, P. S. Doyle, J. A. van Kan and J. R. van der Maarel, *Lab Chip*, 2013, **13**, 2821.
- 30 K. Yoshikawa and Y. Matsuzawa, *J. Am. Chem. Soc.*, 1996, **118**, 929.
- 31 J. Tang, S. L. Levy, D. W. Trahan, J. J. Jones, H. G. Craighead and P. S. Doyle, *Macromolecules*, 2010, **43**, 7368.
- 32 E. A. Strychalski, S. L. Levy and H. G. Craighead, *Macromolecules*, 2008, **41**, 7716.
- 33 K. Günther, M. Mertig and R. Seidel, *Nucleic Acids Res.*, 2010, **38**, 6526.
- 34 A. Balducci, C.-C. Hsieh and P. Doyle, *Phys. Rev. Lett.*, 2007, **99**, 238102.
- 35 T. A. Zangle, A. Mani and J. G. Santiago, *Chem. Soc. Rev.*, 2010, **39**, 1014.
- 36 D. Stein, M. Kruithof and C. Dekker, *Phys. Rev. Lett.*, 2004, **93**, 035901.
- 37 Y. Wang, S. Ran, B. Man and G. Yang, *Soft Matter*, 2011, **7**, 4425.
- 38 V. A. Bloomfield, *Curr. Opin. Struct. Biol.*, 1996, **6**, 334.
- 39 A. Montesi, M. Pasquali and F. MacKintosh, *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.*, 2004, **69**, 021916.
- 40 H. Noguchi, S. Saito, S. Kidoaki and K. Yoshikawa, *Chem. Phys. Lett.*, 1996, **261**, 527.
- 41 C. C. Hsieh, A. Balducci and P. S. Doyle, *Nano Lett.*, 2008, **8**, 1683.
- 42 N. Yoshinaga, *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.*, 2008, **77**, 061805.
- 43 L. Dai, C. B. Renner, J. Yan and P. S. Doyle, *Sci. Rep.*, 2015, **5**, 18438.
- 44 A. Brinker, G. Pfeifer, M. J. Kerner, D. J. Naylor, F. U. Hartl and M. Hayer-Hartl, *Cell*, 2001, **107**, 223.
- 45 J. Mittal and R. B. Best, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 20233.
- 46 L. Dai, C. B. Renner and P. S. Doyle, *Macromolecules*, 2015, **48**, 2812–2818.
- 47 S. Das and S. Chakraborty, *J. Chem. Phys.*, 2010, **133**, 174904.
- 48 G. K. Cheong, X. Li and K. D. Dorfman, *Phys. Rev. E*, 2017, **95**, 022501.

49 D. Bottenus, Y.-J. Oh, S. M. Han and C. F. Ivory, *Lab Chip*, 2009, **9**, 219–231.

50 Y.-C. Tang, H.-C. Chang, A. Roeben, D. Wischnewski, N. Wischnewski, M. J. Kerner, F. U. Hartl and M. Hayer-Hartl, *Cell*, 2006, **125**, 903.

51 A. I. Jewett and J.-E. Shea, *Cell. Mol. Life Sci.*, 2010, **67**, 255.

52 H. Li, Z. Wang, N. Li, X. He and H. Liang, *J. Chem. Phys.*, 2014, **141**, 044911.

53 J. Shin, A. G. Cherstvy and R. Metzler, *ACS Macro Lett.*, 2015, **4**, 202.

54 W. Wang, W.-X. Xu, Y. Levy, E. Trizac and P. Wolynes, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 5517.