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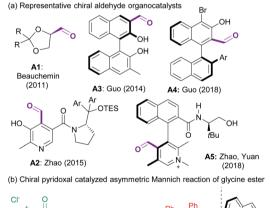
# Chiral Lewis acid-bonded picolinaldehyde enables enantiodivergent carbonyl catalysis in the Mannich/condensation reaction of glycine ester†

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A new strategy of asymmetric carbonyl catalysis *via* a chiral Lewis acid-bonded aldehyde has been developed for the direct Mannich/condensation cascade reaction of glycine ester with aromatic aldimines. The co-catalytic system of 2-picolinaldehyde and chiral Yb<sup>III</sup>-*N,N'*-dioxides was identified to be efficient under mild conditions, providing a series of trisubstituted imidazolidines in moderate to good yields with high diastereo- and enantioselectivities. Enantiodivergent synthesis was achieved *via* changing the sub-structures of the chiral ligands. The reaction could be carried out in a three-component version involving glycine ester, aldehydes, and anilines with equally good results. Based on control experiments, the X-ray crystal structure study and theoretical calculations, a possible dual-activation mechanism and stereo-control modes were provided to elucidate carbonyl catalysis and enantiodivergence.

### Introduction

Asymmetric carbonyl catalysis, inspired by enzyme catalysis, 2 represents a type of useful organo-covalent-activation method via formation of various species, such as hemiacetal,3 or imine,4 as well as dioxirane from chiral ketone for epoxidation.5 Substantial efforts in developing aldehyde-based chiral organocatalysts have been made after pioneering reports by Kuzuhara,6 Breslow and co-workers.7 Excellent contributions have been documented in recent years by Beauchemin,8 Guo,9 Zhao and Yuan10 et al. (Scheme 1a), and the reactions involve Copetype hydroamination,8 alkylation,9d transamination,10a and addition reactions of glycine esters. 9a,b,10b For instance, chiral Nquaternized pyridoxal analogue A5 containing a chiral axis and a carbon stereogenic center allowed a biomimetic asymmetric Mannich reaction of tert-butyl glycine ester and N-diphenylphosphinyl imines (Scheme 1b). 10b Cooperative imine/ hydrogen-bond bifunctional activation guaranteed high activity and stereoselectivity. On top of that, the dual-catalysis by Lewis acid-bonded picolinaldehyde has been explored by Aron's group in the synthesis of chiral amino acids. <sup>11</sup> They found that the metal ion chelated quinonoid intermediate facilitated racemization and Tsuji–Trost allylation, and



(c) Chiral Lewis acid-chelated picolinaldehyde for Mannich/condensation reaction

Scheme 1 Carbonyl catalysts and selected examples of  $\alpha$ -functionalization of glycine esters.

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<sup>†</sup> Electronic supplementary information (ESI) available:  $^1\text{H}$ ,  $^{13}\text{C}\{^1\text{H}\}$  and  $^{19}\text{F}\{^1\text{H}\}$  NMR spectra, HPLC spectra, and CD spectra (PDF). X-ray crystallographic data for 3n, L-RaEt<sub>2</sub>/Yb(OTf)<sub>3</sub> and L-TQtBu/Yb(OTf)<sub>3</sub> (CIF). CCDC 1983265, 2011003 and 2011004. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/d08c07052a

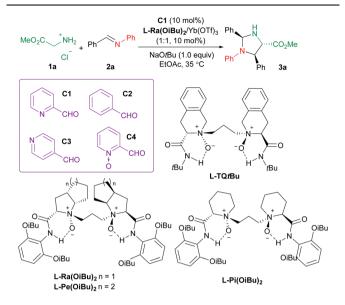
enantiomerically enriched amino acids could be obtained in concert with Alacase<sup>11a</sup> or a chiral palladium catalyst.<sup>11b</sup>

These studies, together with the ingenious imine-based transient directing group strategy in transition metal-catalyzed C-H activation,12 intrigue us to develop chiral Lewis acidbonded aldehyde as a new route for asymmetric carbonyl catalysis in α-functionalization of glycine esters, 13 which might compensate for the design and synthesis of chiral aldehyde organocatalysts de novo. Our research group focused on chiral Lewis acid catalysts of N,N'-dioxides which are capable of a number of asymmetric transformations due to their easy preparation and structural modification.14,15 It is anticipated that the stereo-environment created by the chiral Lewis acid could deliver to the bonded-aldehyde for asymmetric carbonyl catalysis, for instance, the asymmetric nucleophilic addition of the azomethine ylides, generated from picolinaldehyde and glycine ester (Scheme 1c). The study by Wang's group reported the utilization of metallated azomethine ylides in [3 + 2] cycloadditions with a variety of electron deficient alkenes.16 The improved reactivity and efficient enantiocontrol from metalbonded amino esters shed light on our new chiral iminebonded activation method for the Mannich-type reaction of glycine ester. Herein, we wish to disclose a new co-catalytic system of 2-picolinaldehyde and a chiral YbIII complex of N,N'-dioxides, 17 which was optimized to be efficient to catalyze the diastereo- and enantioselective Mannich/condensation cascade reaction of glycine ester with aromatic aldimines under mild conditions (Scheme 1c). Various enantioenriched imidazolidines were provided in good yields with high stereoselectivities, even in the three-component version with glycine ester, aldehydes and alinines. Interestingly, changing the substructure of Feng N,N'-dioxides enabled enantiodivergent synthesis. 18 Two different working modes were provided to give a rationale for carbonyl catalysis and enantiodivergence on the basis of the X-ray crystal structures of chiral Yb(III) complexes and DFT calculations.

#### Results and discussion

Initially, methyl glycine ester hydrochloride (1a) and N-phenyl phenylmethanimine (2a)19 were chosen as the model substrates to optimize the reaction conditions. In the preliminary screening, the chiral complex of Yb(OTf)<sub>3</sub> with Feng N,N'dioxide L-Ra(OiBu)2 was employed as the Lewis acid catalyst. NaOtBu (1.0 equivalent) was used as the base to generate methyl glycine ester from 1a in situ, and 2-picolinaldehyde C1 (10 mol%) was loaded as the carbonyl co-catalyst. The reaction took place smoothly in EtOAc at 35 °C, affording imidazolidine 3a as the final product in 40% yield, 93:7 dr, and 95% ee after 24 hours (Table 1, entry 1), which is generated from the Mannich reaction of aldimine 2a with glycine ester, following condensation with benzaldehyde (see Fig. 1 for details). The comparison experiments in the absence of 2-picolinal dehyde or in the presence of other aldehydes confirmed that both pyridine and aldehyde groups were indispensable. No reaction occurred without the addition of 2-picolinaldehyde (entry 2) or in the presence of benzaldehyde C2 (entry 3), or isonicotinaldehyde C3

Table 1 Optimization of reaction conditions



Entry	Variation	Yield <sup>b</sup> (%)	dr <sup>c</sup>	ee <sup>c</sup> (%)
1	None	40	93:7	95(+)
2	Without C1	N.R.	_	_
3	C2	N.R.	_	_
4	C3	N.R.	_	_
5	C4	45	81:19	9(+)
6	L-Pi(OiBu) <sub>2</sub>	43	90:10	91(+)
7	L-Pe(OiBu) <sub>2</sub>	34	92:8	97(+)
8	L-TQtBu	34	86:14	87(-)
9	C1 (20 mol%)	44	93:7	95(+)
$10^d$	C1 (20 mol%), EtOH	46	94:6	97(+)
$11^{d,e}$	C1 (20 mol%), EtOH	62	95:5	97(+)
$12^{d,e}$	L-TQtBu, C1 (20 mol%)	50	88:12	95(-)

<sup>a</sup> Unless otherwise noted; all reactions were carried out with **1a** (0.10 mmol), **2a** (2.0 equiv.), NaOtBu (1.0 equiv.), carbonyl catalyst (10 mol%) and ligand/Yb(OTf)<sub>3</sub> (1:1, 10 mol%) in EtOAc (0.17 M) at 35 °C for 24 hours. N.R. = no reaction. <sup>b</sup> Isolated yield of **3a** based on **1a**. <sup>c</sup> Determined by HPLC on a chiral stationary phase. <sup>d</sup> EtOH (8.0 equiv.). <sup>e</sup> **2a** (3.0 equiv.) for 48 hours.

(entry 4). Although the addition of 2-formylpyridine N-oxide C4 benefitted the generation of imidazolidine 3a, the enantioselectivity dropped dramatically (entry 5, 9% ee vs. 95% ee). We rationalized that the stronger chelated 2-picolinaldehyde and the resultant azomethine ylide were involved in initiating the Mannich reaction. We also identified other critical parameters for this asymmetric transformation (see ESI Tables S1-S8† for details). Subsequent investigation of Feng N,N'-dioxides with different chiral backbones indicated that the (S)-2-pipecolic acid-derived L-Pi(OiBu)2 exhibited comparable activity with slightly lower stereoselectivity, while L-perindopril-derived L-Pe(OiBu)<sub>2</sub> showed lower activity with similar stereoselectivity (entries 6 and 7). Notably, the introduction of (S)-1,2,3,4tetrahydroisoguinoline-3-carboxylic acid-derived N,N'-dioxide ligand L-TQtBu, led to the enantiodivergent synthesis of the corresponding product ent-3a in 34% yield, and 86: 14 dr with 87% ee (entry 8). With the increase of the amount of co-catalyst

Fig. 1 Proposed catalytic cycle.

C1 to 20 mol%, the product was isolated in a slightly higher yield with comparable stereoselectivity (entry 9, 44% yield, 93:7 dr, 95% ee). Besides, the addition of EtOH as a protonic additive (entry 10), and increasing the amount of aldimine 2a (entry 11) resulted in the formation of 3a with 62% yield, 95:5 dr and 97% ee. Under such conditions, the product *ent*-3a could be afforded in 50% yield, 88:12 dr and 95% ee by the use of ligand L-TQtBu instead (entry 12). Due to the occurence of several side reactions, the attempt to improve the yield of the product was unsuccessful (see the ESI† and control experiments below for details).

With the optimized reaction conditions established (Table 1, entry 11), the generality of the aldimines was explored (Table 2, conditions A). Firstly, a series of aromatic aldehyde-derived imines were investigated (3b-3p). Regardless of the electronic nature or steric hindrance of the substituents on the phenyl ring (Ar<sup>1</sup>), the imidazolidine products 3b-3k were obtained in 31-66% yields with excellent stereoselectivities (82:18-95:5 dr, 73-97% ee). Aldimine derived from 3,4-dimethylbenzaldehyde was tolerated (31, 39% yield, 93: 7 dr and 90% ee). The reaction of aldimines derived from 2-naphthaldehyde or heteroaryl aldehydes proceeded well, affording 3m-3p with comparative results (50-56% yields, 90:10-94:6 dr, 94-97% ee). The following screening of the N-aryl substituent (Ar<sup>2</sup>) of aldimines showed that para-substituted aniline based aldimine had a limited effect on the reaction (3q-3t, 35-46% yields, 90:10-94:6 dr, 95-97% ee). The 4-methoxyaniline derived one resulted in a high enantioselectivity with a decreased yield (3r, 35% yield, 92:8 dr, 95% ee). It should be noted that for several products, the use of the ligand L-Pi(OiBu)2 or L-Pe(OiBu)2 instead was necessary to get higher yields without the change of stereo-preference. The absolute configuration of the product 3n was determined to be (2S,4S,5S)-isomer by X-ray single crystal analysis.20 The stereochemistry of other products was assigned by comparing their CD spectra with those of the product 3n.

Next, we proceeded to prepare the antipodes of the product 3 by the employment of the chiral  $\mathbf{L}$ - $\mathbf{TQ}t\mathbf{Bu}/\mathbf{Yb}(\mathbf{OTf})_3$  complex as the Lewis acid catalyst. Representative aldimines were examined and the results are depicted in Table 2, condition B.

Table 2 Substrate Scope of Aldimines<sup>a</sup>

<sup>a</sup> Reactions were carried out with 1a (0.10 mmol), 2 (3.0 equiv.), NaOtBu (1.0 equiv.), C1 (20 mol%), EtOH (8.0 equiv.) and L/Yb(OTf)<sub>3</sub> (1:1, 10 mol%) in EtOAc (0.17 M) at 35 °C for 48 hours. The isolated yield of 3 based on 1a. The dr value was detected by  $^1$ H NMR analysis, and the ee value was determined by HPLC on a chiral stationary phase.  $^b$  L-Pi(OiBu)<sub>2</sub> was used as the ligand.  $^c$  L-Pe(OiBu)<sub>2</sub> was used as the ligand.

Generally, these reactions occurred smoothly to deliver the corresponding products *ent-3* with high enantio- and diaster-eoselectivity in slightly diminished yields (34–50% yields, 86: 14–95: 5 dr, 90–95% ee).

Table 3 Three-component reaction<sup>a</sup>

Entry	$\mathrm{Ar}^{1}/\mathrm{Ar}^{2}$	Yield (%)	dr	ee (%)
1	Ph/Ph (3a)	55	93:7	96
2	$4-\text{MeC}_{6}\text{H}_{4}/\text{Ph}$ (3b)	40	88:12	88
$3^b$	4-ClC <sub>6</sub> H <sub>4</sub> /Ph (3g)	53	94:6	91
4	$Ph/4-MeC_6H_4$ (3q)	39	94:6	93
$5^b$	$Ph/4-ClC_6H_4$ (3s)	44	94:6	97
6 <sup>c</sup>	$Ph/3-MeC_6H_4$ (3v)	41	92:8	87

<sup>&</sup>lt;sup>a</sup> Reactions were carried out with **1a** (0.10 mmol), **4** (3.0 equiv.), **5** (3.0 equiv.), NaOtBu (0.10 mmol), C1 (20 mol%), 4 Å MS (20.0 mg), and Yb(OTf)<sub>3</sub>/L-Ra(OiBu)<sub>2</sub> (1:1, 10 mol%) in EtOAc (0.17 M) at 35 °C for 48 hours. The isolated yield of **3** based on **1a**. The dr value was detected by <sup>1</sup>H NMR analysis and the ee value was determined by HPLC on a chiral stationary phase. <sup>b</sup> With L-Pi(OiBu)<sub>2</sub>. <sup>c</sup> With L-Pe(OiBu)<sub>2</sub>.

In order to improve the practicability of the reaction, we investigated the three-component synthesis of imidazolidines, from glycine ester, aldehyde and aniline. As shown in Table 3, the three-component reaction was accomplished under slightly modified conditions (see ESI Table S9† for details), and an array of aldehydes 4 and anilines 5 were surveyed. In comparison with the two-component reaction system, similar results were given (39–55% yields, 88:12–94:6 dr, 88–97% ee). By using this protocol, the product 3v was yielded in moderate yield with good diastereo- and enantioselectivity (Table 3, entry 6), which is not possible from the two-component reaction of *N-m*-tolyl substituted aldimine due to the separation and purification problem of aldimine. Unfortunately, aliphatic amines and aldehydes were not suitable in the current system.

To evaluate the synthetic potential of the catalytic system, a scale-up preparation of imidazolidine 3h was carried out (Scheme 2a). Upon treatment of 3.0 mmol of 1a with 9.0 mmol of 2h under the optimized reaction condition A, the desired product 3h was obtained in 62% yield (0.97 g), with 92:8 dr and 93% ee after 60 hours. Furthermore (Scheme 2b), the product 3a was successfully converted into the chiral vicinal diamine product 6a via TsOH-mediated hydrolysis in 51% yield, 93:7 dr and 98% ee after two steps. The change of the ester group had an obvious effect on the transformation, and ethyl glycine ester **1b** led to a reduced yield (**6b**, 30% yield), and benzyl glycine ester 1c provided a lower diastereoselectivity (6c, 78:22 dr, 91%/93% ee). Similarly, ent-6b diamine was obtained in comparable yield when L-TQtBu was used (29%, 82: 18 dr, 91% ee). Noteworthily, the syn-diamine derivatives 6 were dominant in the current case (Scheme 2b), which is different from the chiral pyridoxal analogue A5 based catalytic system. 10b Reduction of the ester to a primary alcohol led to the imidazolidine derivative 7a in good yield with maintained stereoselectivity. In addition, chiral 2-imdazoline 8a was produced by oxidation with DDQ (Scheme 2c).

Scheme 2 Gram-scale synthesis and transformations

To get insight into the mechanism of the reaction, a series of control experiments were conducted (Scheme 3). Treatment of chiral diamine 6a with aldimine 2a in the presence of the L-Ra(OiBu)<sub>2</sub>/Yb(OTf)<sub>3</sub> complex delivered the imidazolidine 3a in 51% yield with maintained diastereoselectivity and enantioselectivity (eqn (1)). In contrast, a yield of 90% was obtained if benzaldehyde was used for the formation of imidazolidine 3a (eqn (2)). These results clearly indicated that the final imidazolidine product 3a was generated via a stepwise Mannich/ condensation cascade reaction with aldehyde released from the decomposition of aldimine 2a. Therefore, this reaction was different from the previous direct [3 + 2] cycloaddition of azomethine ylides with aldimines.21 When the Schiff base 9 of 2picolinaldehyde C1 and methyl glycine ester was used instead of 2-picolinaldehyde, the product 3a was isolated in 42% yield with 92:8 dr and 96% ee (eqn (3)). It confirmed that 2-

Scheme 3 Control experiments.

picolinaldehyde probably serves as the carbonyl catalyst to generate the Schiff base intermediate for the subsequent addition reaction. According to HRMS analysis of the reaction mixture, Schiff base 10 and picolinaldehyde-derived imine 11 might exist in the reaction system (eqn (4) and (5)). However, these species were unreactive under the current reaction conditions (see ESI page S24–S27 for detail†). These side reactions along with the instability of products were partly responsible for the moderate yields of the titled process.

The influence of the concentration of each component on the reaction rate was detected from operando IR profiles (see ESI page S19–S22 for details†). The kinetic study showed that the initial rate of the reaction was first-order depending on the chiral **L-Ra(OiBu)**<sub>2</sub>/Yb(OTf)<sub>3</sub> complex, glycine ester, 2-picolinaldehyde **C1** and aldimine **2**, indicating that these species are involved in the rate-determining process. Furthermore, the HRMS spectra of the mixture of **L-Ra(OiBu)**<sub>2</sub>, Yb(OTf)<sub>3</sub>, **C1**, **1a**, and NaOtBu (1:1:1:1) in MeOH exhibited the ion peak at m/z 1470.4521, referring to [**L-Ra(OiBu)**<sub>2</sub> + Yb<sup>3+</sup> + 2OTf<sup>-</sup> + 9]<sup>+</sup>

species (calculated m/z 1470.4516). It confirmed that chiral Lewis acid bonded 2-picolinal dehyde condenses with methyl glycine ester to generate the metal-chelated Schiff base I (Fig. 1).

Based on the aforementioned results, a plausible catalytic cycle is rationalized for the current reaction (Fig. 1). Initially, chiral ytterbium-bonded picolinaldehyde reacts with glycine ester to yield chiral cation bonded Schiff base species I. In view of the large and variable coordination number of the ytterbium ion, the tridentate coordination of Schiff base 9 and another anion is anticipated to enhance the stability and reactivity of the corresponding enolate species, as well as the quinonoid tautomer. Then, H-bond activated aldimine species undergoes the Mannich reaction *via* the enolate intermediate II to afford the intermediate IV. Then, hydrolysis of the intermediate IV produces the chiral diamine 6, regenerating the active catalytic species Ligand/Yb<sup>III</sup> bonded-picolinaldehyde. Eventually, the final product imidazolidine 3 was yielded by the condensation of vicinal diamine and aldehyde.

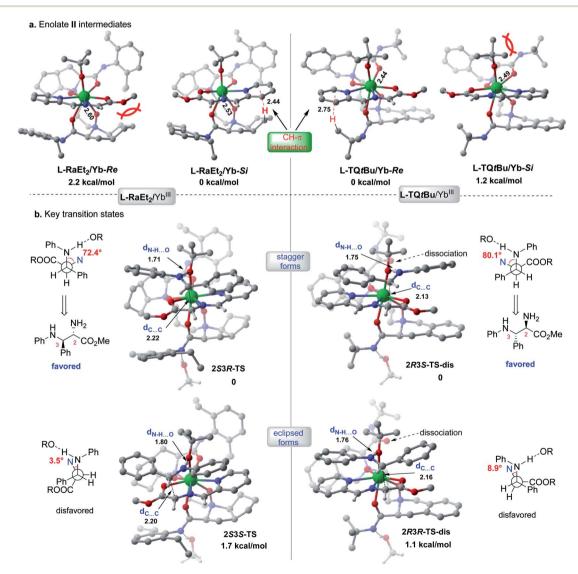


Fig. 2 Calculated stereo-models of transition states and intermediates for L-RaEt<sub>2</sub>/Yb<sup>III</sup> or L-TQtBu/Yb<sup>III</sup> complexes.

The X-ray crystal structures of the chiral Lewis acid complexes L-RaEt<sub>2</sub>/Yb(OTf)<sub>3</sub> and L-TQtBu/Yb(OTf)<sub>3</sub> (ref. 22) provide interesting spatial information (see ESI Fig. S2 and S3† for full pictures), which might account for the enantiodivergent outcomes. To unveil the different stereochemical control of L-RaEt<sub>2</sub> (ref. 23) and L-TQtBu in the formation of chiral intermediate IV, preliminary density functional theory (DFT) calculations were performed at the SMD-B3LYP-D3/MWB59/6-311+G(d,p) level of theory. Due to different thermodynamic stabilities, OtBu is more favorable to coordinate with YbIII for about -10.2 kcal mol<sup>-1</sup> rather than the triflate anion. Moreover, based on the binding orientation of the tridentate Schiff base 9 in different Yb<sup>III</sup>-chiral N,N'-dioxides complexes, OtBu always locates at the Si-face or Re-face site of the Schiff base for L-RaEt<sub>2</sub>/ Yb<sup>III</sup> or L-TQtBu/Yb<sup>III</sup>, respectively, which would block their opposite sites by substrate/ligand interactions and avoid the steric hindrance of the bulky ligand (bicycle or tert-butyl amide subunits). As shown in L-RaEt<sub>2</sub>/Yb<sup>III</sup> of Fig. 2a, the steric effect of the bicycle ring on the ligand would lead to relatively weak coordination 2.60 Å between ligands O<sup>N</sup> and Yb<sup>III</sup> in L-RaEt<sub>2</sub>/Yb-**Re**, and thus its thermodynamic free energy is 2.2 kcal mol<sup>-1</sup> higher than that of L-RaEt<sub>2</sub>/Yb-Si. On the other hand, the tertbutyl amide in L-TQtBu/YbIII displays major steric hindrance with OtBu that results in unfavorable coordination in L-TQtBu/Yb-Si.

Further discussion will depend on the most favorable YbIII/ chiral N,N'-dioxide complexes with the Schiff base substrate and OtBu ligand (Fig. 2b). To control the stereoselectivity, an efficient inner-sphere hydrogen bond (around 1.7 Å) between the coordinated OtBu and protonated aldimine would drive the nucleophilic attacking to form stable transition states in the following Mannich reaction, compared to the outer-sphere hydrogen bond interaction that affords 6.1 kcal mol<sup>-1</sup> free energy higher in the transition state 2S3R-TS\_1 (see ESI Table S10†). For both the L-RaEt<sub>2</sub>/Yb<sup>III</sup> and L-TQtBu/Yb<sup>III</sup> complexes, the stagger forms of transition states were favored via Si-Si (2S3R-TS) and Re-Re (2R3S-TS-dis) facial selectivity, resulting in (2S,3R)-6a and (2R,3S)-6a respectively, which are in agreement with experimental observations (Fig. 2b). In contrast, the eclipsed forms of transition states via Si-Re and Re-Si facial attacking were calculated to be about 1-2 kcal mol<sup>-1</sup> energy higher than those of the most stable ones, indicating the important substrate interactions between the Schiff base and protonated aldimine under each chiral space, which account for the diastereoselectivity. Furthermore, the transient ligand dissociation of the L-TQtBu/YbIII complex at one carbonyl group of amide was inspected in the most stable transition states 2R3S-TS-dis and 2R3R-TS-dis via Re-Re and Re-Si facial selectivities, respectively, and the relative free energies are slightly favorable to the full coordinated transition states (see ESI Fig. S5-S7†), suggesting that the large steric hindrance of transition states in this ligand would reduce the coordination number also supported by the crystal structure of L-TQtBu/ Yb(OTf)3. However, the calculated ligand dissociation can be recovered when the protonated aldimine is released from the Yb<sup>III</sup> center displaying dynamic coordination.

# Conclusions

In summary, a new asymmetric carbonyl-catalysis strategy was developed for the Mannich/condensation of glycine ester with aldimine under mild conditions. The chiral N,N'-dioxide/ Yb(OTf)<sub>3</sub> complex bonded aldehyde enabled carbonyl activation of glycine ester for α-addition transformation. This protocol provided facile and feasible access to a variety of synthetically useful chiral imidazolidine derivatives in moderate yields (up to 66% yield), and excellent diastereo- and enantioselectivities (up to 95:5 dr, 97% ee). The reaction could be performed in either two- or three-component versions. Interestingly, enantiodivergent synthesis was accessible by modulating the substructure of the Feng N,N'-dioxide ligand in connection with a multi-coordinated vtterbium metal ion. Theoretical calculations suggested that the steric hindrance and CH- $\pi$  interaction between the substrate and ligand were responsible for this reversal. And the inner-sphere hydrogen bond and the stagger model of reaction substrates enable stabilization of key transition states leading to high stereoselectivity. Further application of this co-catalytic system in other reactions is under investigation.

# Conflicts of interest

There are no conflicts to declare.

# Acknowledgements

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- 23 Switching the ligand **L-Ra(OiBu)**<sub>2</sub> to **L-RaEt**<sub>2</sub>, comparable results (42% yield, 88:12 dr, 92%/90% ee νs. 40% yield, 93:7 dr, 95%/91% ee) were afforded after 24 h (see ESI Table S5, line 25†). Currently, we only obtained the X-ray crystal structure of the **L-RaEt**<sub>2</sub>/Yb(OTf)<sub>3</sub> complex. Therefore, the **L-RaEt**<sub>2</sub>/Yb(OTf)<sub>3</sub> complex was used for the following computational studies.