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Application of Pauson–Khand reaction in the total synthesis of terpenes

Majid M. Heravi * and Leila Mohammadi

The Pauson–Khand reaction (PKR) is a formal [2 + 2 + 1] cycloaddition involving an alkyne, an alkene and carbon monoxide mediated by a hexacarbonyldicobaltalkyne complex to yield cyclopentenones in a single step. This versatile reaction has become a method of choice for the synthesis of cyclopentenone and its derivatives since its discovery in the early seventies. The aim of this review is to point out the applications of PKR in the total synthesis of terpenes.

Department of Chemistry, School of Physics and Chemistry, Alzahra University, Vanak, Tehran, Iran. E-mail: mmh1331@yahoo.com; mmheravi@alzahra.ac.ir; Tel: +98 2188041344; +98 9121329147



Majid M. Heravi was born in 1952 in Mashhad, Iran. He received his B. Sc. Degree from the National University of Iran in 1975 and his M. Sc. and Ph. D. degrees from Salford University, England, in 1977 and 1980. He completed his doctoral thesis under the supervision of the late Jim Clark. He started his career as a research fellow in Dar-oupakhsh (a pharmaceutical company) in 1981 Tehran, Iran

and joined as an assistant professor at Ferdowsi University of Mashhad, Iran, in 1983 and was promoted to associate professor in 1993 and full professor in 1997. In 1999 he moved to Alzahra University of Tehran, Iran as professor of chemistry where he is still working. He has previously been a visiting professor at UC Riverside, California, USA and Hamburg University, Hamburg, Germany. His research interests focus on heterocyclic chemistry, catalysis, organic methodology and green synthetic organic chemistry. He has so far published 880 ISI cited papers including 12 chapters in *Advances in heterocyclic chemistry* book series published by Elsevier and one chapter in *Alkaloids* (book series published by Elsevier). He is the main author of two published book by Elsevier, namely *Recent Applications of Name Reactions in Multicomponent Reactions*, *Recent Applications of Selected Name Reactions in the Total synthesis of Alkaloids*, *Heteropolyacids as Highly Efficient and Green Catalysts Applied in Organic Transformations* (in press) and *Polyoxometalate-Based Hybrids and their Applications* (Approved and under preparation).



Leila Mohammadi was born in Tehran, Iran. She received her B.Sc. in 2006. Then, she received her M.Sc. degree under the supervision of Dr Sakineh Asghari on the synthesis of heterocyclic compounds at the University of Mazandaran (Babolsar). She is currently working on the Ph.D. thesis in organic chemistry under the supervision of Prof. Dr Majid M. Heravi at Alzahra University,

Tehran, Iran in 2021.

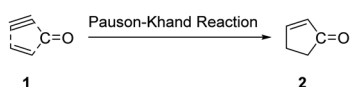


1. Introduction

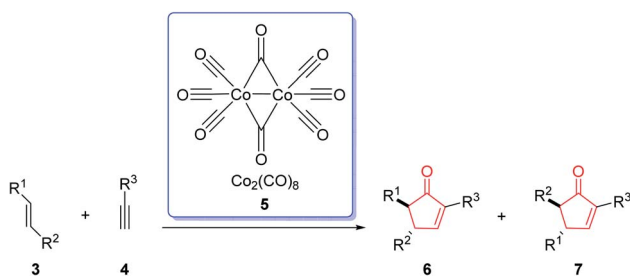
The Pauson–Khand reaction (PKR or PK-type reaction) is formally a chemical reaction in which a triple bond, a double bond and carbon monoxide are subjected to a [2 + 2 + 1] cycloaddition reaction to form an α,β -cyclopentenone.¹ This entails the formation of three new bonds and one or two rings in the intermolecular or intramolecular fashion, respectively (Scheme 1).¹

The first example of this reaction was reported in 1973.² In this reaction, norbornene reacted with the phenyl acetylene–hexacarbonyldicobalt complex to afford the corresponding cyclopentenone in 45% yield by using a stoichiometric amount of dicobalt octacarbonyl $[\text{Co}_2(\text{CO})_8]$ (Scheme 2).³ This reaction was discovered by Ihsan Khand (1935–1980), who was working as a postdoctoral fellow with Peter Pauson (1925–2013),⁴ at the University of Strathclyde in Glasgow.

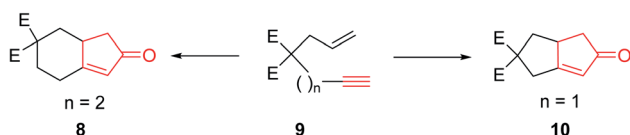
The original PKR had several drawbacks such as only being proceeded in the presence of stoichiometric amount of dicobalt octacarbonyl $[\text{Co}_2(\text{CO})_8]$ as the only cluster, performed thermally under relatively harsh reaction conditions which resulted in desired transformations but with low efficiency. It also showed limited substrate scope covering narrow range of substrates. The use of strained olefins was necessary to obtain acceptable yields. In addition, the reactions typically afforded a mixture of regioisomers if unsymmetrical alkynes and alkenes were used. In several cases, the PKRs showed poor conversions and especially selectivities (chemo-, regio- and stereo-selectivities). Therefore, an important breakthrough was



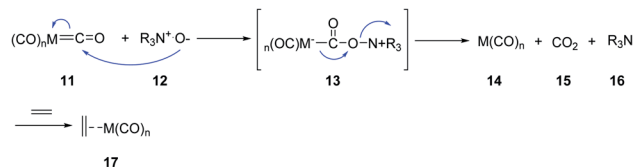
Scheme 1 Formation of three new bonds in PKR resulting in the formation of cyclopentanones.



Scheme 2 The first example of Pauson–Khand reaction without regioselectivity.



Scheme 3 Regioselective intramolecular PKR.



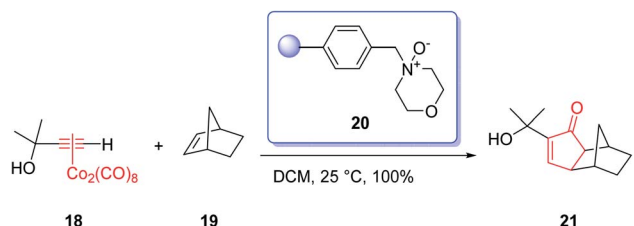
Scheme 4 Formation of a vacant site in the cobalt cluster.

required which was obtained by Schore and co-workers,^{5,6} who reported that carbon-tethered enyne precursors can be subjected to an intramolecular Pauson–Khand reaction (IPKR) in good yields with high regioselectivity. In other words, it was not necessary to use strained olefins as starting materials. In 1983, Schore and coworkers achieved and demonstrated the intramolecular PKR (IPKR).⁶ A series of 5,5-disubstituted enynes **9** were subjected to PKR conditions to obtain 5,6-used bicycles **8** or **10** under PKR (thus, IPKR) conditions its good efficiency and conversions⁷ (Scheme 3).⁵

The PKR has received much attention of chemical community, especially synthetic organic chemists, as a method of choice for scientific studies because of the increase in diversity of the available starting materials. Therefore, it has rapidly improved and developed during these since its discovery.^{8–15}

As an example an essential contribution to improvement of the promotion of PKR Smit and Caple in 1986.¹⁶ They immobilized the reagents onto various solid supports. In these cases, PKRs were performed at lower temperatures and completed in shorter reaction times¹⁶ [although other metals were found to catalyze the PKR, use of $\text{Co}_2(\text{CO})_8$ in stoichiometric amount showed several merits. The $\text{Co}_2(\text{CO})_8$ complex is inexpensive and commercially available, it tolerates a wide spectrum of functional groups, exhibits activity toward both terminal and internal alkynes].

Jeong *et al.*¹⁷ Schreiber and coworkers independently circumvented other problems such as requirement of high temperatures and CO pressures, as well as long reaction times.¹⁸ They found that trimethylamine *N*-oxide and *N*-methylmorpholine *N*-oxide can accelerate the PKR dramatically. A wide range of cyclopentenones were obtained in good yields even at room temperature in the presence of *N*-oxides. It is presumed that *N*-oxides affect the reaction *via* oxidative liberation of the CO ligands of the metal complex facing up a coordination site. Thus, the following oxidative alkene addition becomes, the rate-determining step, accelerating the PKRs (Scheme 4). Other reagents were also found to assist PKRs to proceed smoothly



Scheme 5 A PKR performed on solid state.



and rapidly; there include silica gel,^{13,16} molecular sieves,^{19,20} alkyl methyl sulfides,²¹ and primary amines.²²

The most commonly used amine *N*-oxides are trimethylamine *N*-oxide (TMANO) and *N*-methylmorpholine *N*-oxide (NMO). The latter has recently been immobilized onto a solid support which facilitates the work-up of the reaction and results in excellent yields of the products of PKRs (Scheme 5).²³

Due to the innovation of IPKR, bicyclic frameworks which are prevalent in naturally occurring compounds can be constructed in one step *via* IPKR. Therefore, IPKR has found several applications in the total synthesis of natural products frequently used as the key step.^{6,24–26} Synthetic organic chemistry has developed momentarily owing to the exponential growth in transition-metal-catalyzed reactions. The effect of transition-metal-catalyzed reactions on organic synthesis could be evaluated by the wide range of molecules that have been developed from simple compounds to complex natural products. In 2002, J. Yoshida and coworkers intelligently expanded the metal-catalyzed organic transformations to PKR.²⁷ They used the Ru carbonyl complex [Ru₃(CO)₁₂] **24** in catalytic amount (0.5 mol%) in the reaction of olefins containing an easily removable pyridisilyl group, readily obtained from alkynes under one atmosphere of CO. This metal catalyzed-PKR promotes the reaction efficiently since the pyridyl group directs it to proceed *via* the coordination of the nitrogen with the metal that provided complete regioselectivity with unsymmetrical olefins (Scheme 6).²⁷

These achievements encouraged several research groups to investigate the scope and limitations of using other metals as catalyst in PKR. In this regard, various alkenes, alkynes, and carbon monoxide in the presence of different transition metals

under PKR conditions were converted into the corresponding cyclopentenone derivatives.^{28,29} For this purpose, various carbonyl complexes of iron (iron pentacarbonyl),^{29–32} tungsten (tungsten pentacarbonyl),³² chromium, molybdenum (molybdenum hexacarbonyl),³³ heterobimetallic cobalt/tungsten complexes,³⁴ and cobalt complex created *in situ* from alkyne cobalt complex and triethylsilane,³⁵ titanocene complexes,^{34,36–38} Co₂(CO)₈ with high-intensity visible light system,³⁹ highly purified Co₂(CO)₈,⁴⁰ and other ruthenium complexes^{35,38,41,42} were successfully used under PKR conditions.^{43–46} Noticeably, the above-mentioned metal-catalyzed IPKR reactions had to be performed under medium or high pressure of carbon monoxide. In addition to ruthenium, rhodium complexes efficiently catalyze PKR which has attracted considerable attention. In 2002, Jeong and co-workers,⁴⁷ successfully used several rhodium complexes in various reactions under PKR conditions. Noticeably, some need activation with AgOTf prior to be used. After the appearance of the first IPKR in 1984,^{16,48} a plethora of papers were published regarding the successful catalytic IPKR.^{2,49–51}

In the nineties, several other developments in the PKR were achieved, including the introduction of an asymmetric variant (APKR). In general, bicyclic cyclopentenones as PKR products are valuable synthetic targets. The PKR is a powerful tool for the construction of such structural units. On the other hand, bicyclic cyclopentenones are prevalent scaffolds in natural products. Therefore, the asymmetric variant of PKR (APKR) is a unique and useful technique for the stereoselective construction of bicyclic cyclopentenones as a key step in the multistep total synthesis of natural products. Shibata and coworkers, reported the first example of a catalytic APKR in 2000.²⁹



Scheme 6 The catalytic intermolecular Pauson–Khand reaction directed by a pyridisilyl group.

Their strategy was based on a chiral *ansa*-metallocene complex of titanium. Subsequently, other catalytic APKR were supplemented by a plethora of promising reports, comprising chiral catalysts derived from binaphthyl phosphines and iridium, cobalt^{33,52} or rhodium precatalysts.³²

The efficiency of enantioselectivity of APKR mainly depends on the selection of correct substrates to induce asymmetry in the PKR. For this purpose, four approaches are available (a) using chiral substrate; (b) employing appropriate chiral auxiliary (c) using chiral metal complex and (d) the chiral promoter. The best results have been obtained using chiral substrates or chiral auxiliaries. Although, the use of chiral metal complex to give the best results, but it is still in the beginning stages and will be developed in a way to give excellent results expectedly in the future.

During investigation on the Rh-catalyzed PKR, two chiral catalysts were reported for APKR under an atmospheric pressure of CO. The chiral catalysts were (*S*)-BINAP/Co₂(CO)₈,^{33,52} (*S*)-tolBINAP/[Ir(cod)Cl]₂ (ref. 29) and (*S*)-BINAP/[RhCl(CO)₂]₂.^{2,31,32}

1.1. Mechanism of Pauson–Khand reaction

Although the Pauson–Khand reaction was discovered in 1973, a plausible mechanism for it was proposed by Magnus *et al.* in 1985 as illustrated in (Scheme 7).⁵³ Nowadays, this mechanism is believed to be undisputable since it has recently been confirmed by detailed negative ion electrospray collision testing.⁵⁴ Accordingly, the PKR is believed to commence with the generation of the alkyne–Co₂(CO)₈ complex **29**, loss of a carbonyl ligand to vacate a coordination site, olefin coordination **30**, followed by insertion, occurring at the end of the alkyne, which is less hindered to produce *in situ* the metallacycle **31**. The latter reacts promptly with inserted CO ligand to produce complex **32** followed by reductive elimination of **33** proceeds to furnish the desired target, cyclopentenone **34**. It is noteworthy that all the bond-forming steps took place on just

one cobalt atom. The other cobalt atom which is present in the complex is supposed to function as an anchor which has extra electronic effects on the bond-forming metal atom *via* the present metal–metal bond⁵⁵ (Scheme 7).⁵³

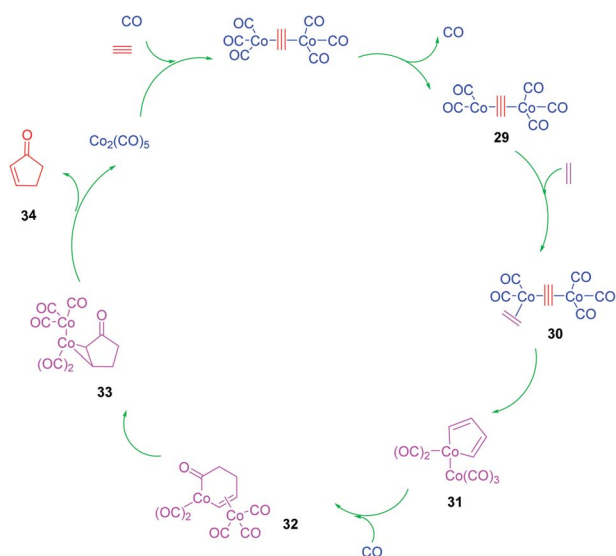
Although the prominence and importance of all kinds of PKR have been extensively covered by the previously published reviews,^{56,57} its applications in total synthesis of natural products have largely been overlooked and limited to a subsection that mentions some total syntheses using PKR. These natural products are vincristine (Oncovin), Navelbine (vinorelbine), etoposide (VP-16), teniposide (VM-26), Taxol (paclitaxel) and most recently in 1996, Taxotere (docetaxel), topotecan (Hycamtin), sexestobergsterol, spatane, daphne, iridomirmecin, dendrobin, kalmanol and β-cuparenone were mentioned.^{58,59}

2. Applications of Pauson–Khand reaction in the total synthesis of terpenes

2.1. Monoterpenes

In 1968, Naya and co-workers initially isolated ether **50** which is a monoterpene from the Japanese hop “Shinshu-Wase”.⁶⁰ This naturally occurring compound is also present in Spalter hops.⁶¹ Fascinatingly, it is thought to contribute to both the taste and aroma in low concentrations in a number of beers.^{62–66} In 2016, Park research group produced Japanese hop extracts employing a standardized method.⁶⁷ This method facilitates the production of improved diagnostic and immunotherapeutic reagents.⁶⁷

The first reported total synthesis of Japanese hop ether was reported by Imagawa *et al.* in 1979 which was claimed unreliable^{65,68} and since that time few alternative pathways have also been claimed.⁶⁹ In 2005, Kerr *et al.* achieved and reported a concise (fourteen steps) total synthesis of monoterpene Japanese hop ether in 29% overall yield. They used an intramolecular Pauson–Khand reaction under mild *N*-oxide promoted conditions and with complete retention of alkene configuration (for both *cis*- and *trans*-alkenes) in the product cyclopentenone, as the crucial key step.⁶⁵ They started with propargyl alcohol **35** and transformed it to the corresponding THP-protected derivative **36** in almost quantitative yield. The latter upon deprotonation, using *n*-BuLi, with subsequent reaction with solid *para*-formaldehyde under ultrasound irradiation furnished the mono-protected ynediol **37** in excellent yield. The *Z*-olefin **38** was obtained in virtually quantitative yield by hydrogenation of ynediol **37** over Lindlar's catalyst and the obtained allylic alcohol was transformed to the bromide **39** in high yield upon being treated with CBr₄ and PPh₃. The bromide **39** was then reacted with the alkoxide anion of dimethylpropargyl alcohol **40** to afford allyl propargyl ether in high yield. In a key step (PKR), compound **41** underwent complexation with dicobalt of a carbonyl to afford **42** in excellent yield. Eventually, the *Z*-enyne complex **42** converted into the product **43** in high yield by use of TMANO·2H₂O in acetone under air. Having compound **43** available in hand, it was deoxygenated affording **48** after several steps. The latter was then directly converted into



Scheme 7 A plausible mechanistic proposal for the Pauson–Khand reaction.





Scheme 8 Total synthesis of Japanese hop ether 50.

organoselenium species **49** upon reaction with *o*-nitrophenylselenenyl cyanate and tri-*n*-butylphosphine in excellent yield. The latter was simply treated with H_2O_2 permitting the *in situ* generation of the selenoxide, which was subjected to elimination to furnish the desired natural product Japanese hop ether **50** in 94% yield (Scheme 8).⁶⁵

Iwabuchi and co-workers in 1997,⁷⁰ initially isolated (+)-mintlactone **54** and (–)-isomintlactone **55** as *endo* α,β -unsaturated monoterpene- γ -lactones from the oil of the wood of *Bursera graveolens* (Palo Santo).⁷⁰ In 1968, Muraki and co-workers isolated their enantiomers, *ent*-(–)-**54** and *ent*-(+)-**55**, from *Mentha cardica*.⁷⁰ Also, The last two *p*-menthanolides

were both isolated from *Mentha arvensis*⁷¹ and are present as minor constituents of the commercial essential oil.⁷² Interestingly, the total synthesis of the above-mentioned bicyclic monoterpene attracted much attention of synthetic organic chemists.^{70,72–78} Recently, Bates *et al.*⁷⁹ achieved and reported a brief ten-step synthesis of (–)-mintlactone starting from the THP ether of propargyl alcohol *via* a highly asymmetric tin(II) chloride-catalyzed intramolecular propargylic Barbier reaction with subsequent allenol cyclocarbonylation. Furthermore, Shishido *et al.*⁷⁸ accomplished and reported a concise total synthesis of (–)-mintlactone in ten steps commencing from citronellal. In 2009, Zhai and co-workers⁷⁰ design a pathway for

total synthesis of (+)-mintlactone starting from (–)-citronellol **51**, using molybdenum-mediated intramolecular hetero-Pauson–Khand reaction as a key step. This pathway started with (–)-citronellol **51** which upon treatment with nitrous acid following Abidi's protocol gave alkynol **52** directly in three steps in 26% overall yield.⁸⁰ The latter was oxidized upon treatment with PCC in CH₂Cl₂ at ambient temperature to give ynal **53** as suitable precursor for PKR. The ynal **53** was then treated with freshly prepared organomolybdenum Mo(CO)₃(DMF)₃,⁸¹ as catalyst, in THF at ambient temperature to afford the desirable natural product (+)-mintlactone (**54**, 39%) along with its inseparable diastereomer, apparently (–)-isomintlactone (**55**, 3%), in an optimized combined yield of 42%. Interestingly, in this total synthesis, during the PKR, one stereogenic center, two rings, and three covalent bonds (1 C–O and 2 C–C) are generated which proceeded in high diastereoselectivity (C-3, dr = 12 : 1). Due to more stability of a chair conformation (TS-1) in its transition state which theoretically has lower energy than a twist boat one (TS-2), thus (+)-mintlactone **54** must have emerged as a major product. In conclusion, the total synthesis of (+)-mintlactone was accomplished *via* a three-step assemblage which can be exemplified as a new concept in the art of synthetic organic chemistry as “step economy”⁸² and “strategic efficiency”.⁸³ Important aspects of this total synthesis involved HNO₂-induced formal isopropylidene “demethanation” and the Mo(CO)₃(DMF)₃-promoted intramolecular PKR (Scheme 9).⁷⁰

The generic name iridoid monoterpenes is derived from the names iridomyrmecin, iridolactone, and iridodial which initially isolated from special species of *Iridomyrmex*, ants, secreting them as defensive existing either in the glycosidic or in the non-glycosidic form,⁸⁴ these naturally occurring compounds were found the dynamic and active components of folk medicinal plants being used conventionally as medicine for long time as antiviral, antibacterial, and anti-inflammatory.^{85,86} In addition, iridoids are also commercially important since can be used as sex pheromones against some agriculturally

important species such as aphids.⁸⁷ Structurally, iridoids contain a confined cyclopenta[*c*]pyran framework as shown for some important members of family. Nevertheless, controlling their stereochemical complexities and the diverse oxygenation patterns, which frequently are confined in *cis*-fused bicycle, are challenging for synthetic organic chemist and render them striking targets for total synthesis. Thus, several fantastic synthetic pathways for their total synthesis have been reported.⁸⁸ Some of them are accomplished and reported by Suh and co-workers^{89,90} and recently one reported by Chakraborty and co-workers.⁹¹ In spite of the appearance of these reports among the others, a literature survey disclosed few reports on the total synthesis of diverse oxy-functionality pattern observed on the cyclopentane ring of the iridoid framework.^{89,92,93} In 2019, Khan and co-workers reported an efficient and economic strategy to have access to several iridoids (**65**, **68**, **68'**, **70**, **71**, **72**, **75**, **75'**, **79**)⁹⁴ using an intramolecular Pauson–Khand reaction (IPKR) as the crucial step to access ten iridoids in a stereo-selective fashion. In their strategy, bicyclic ether **61** was found to be an important intermediate, since it contains the iridoid scaffold as well as bearing a carbonyl functionality at the C6 position and tosylate at the C4 position. The manipulation of the C6 carbonyl moiety in **61** makes the construction of cyclopentane ring of the desired natural products (**65**, **68**, **68'**, **70**, **71**, **72**, **75**, **75'**, **79**) possible whereas the tosylate present at C4 can be exploited for the construction of tetrahydropyran ring, which can be extended to the δ -lactone ring or easily discarded in the framework of scholarein A **80**. In addition, total synthesis of some other natural products can be simplified by having easy access to an iridoid. For example, the total synthesis of 7-*epi*-boschnialactone **68'**.

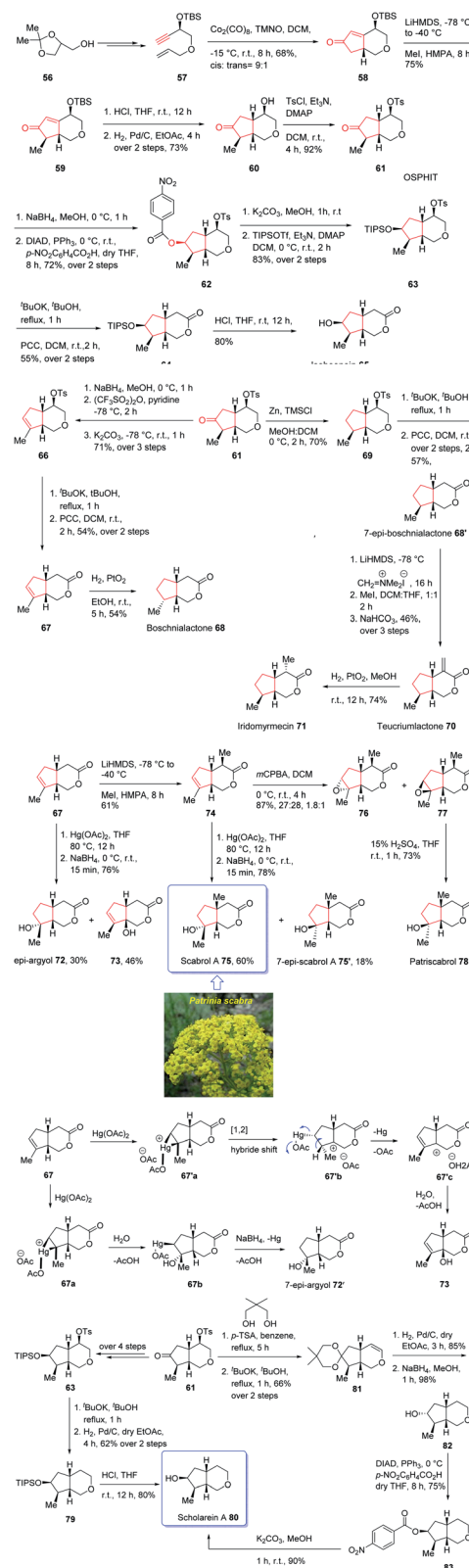
This strategy is started with the easily accessible glycerol acetone **56**,^{88,95} which in two steps was converted into the enyne **57** as an appropriate precursor of PKR. The latter in crucial step was subjected to diastereoselective conventional IPKR⁸⁸ in the presence of Co₂(CO)₈, TMNO to afford compound



Scheme 9 Total synthesis of (+)-mintlactone **54** and (–)-isomintlactone **55**.

58 which was then transformed into the important intermediate **61** in three steps. This important intermediate **61**, was initially treated with Zn/TMSCl to undergo Clemmensen reduction⁹⁶ to give compound **69** which after two steps involving manipulation of its C4-tosylate moiety, **69** was converted into the δ -lactone affording the C7-epimer of boschnialactone **68'** in good yield, with identical spectroscopic data to that previously reported.⁹⁷ On the other hand, intermediate **61** was converted into TIPS protected bicyclic lactone **64**. TIPS deprotection in compound **64** in the presence of 10% HCl in THF gave the desired natural product iridolactone isoboonein **65** in satisfactory yield. The spectral data of this synthetic compound **65** were found being identical to those recorded for natural isoboonein.⁹³ After successful synthesis of natural product **65**, the synthesis of iridoids **71**, **68** and **70** from the key intermediate **61** was contemplated. On the already prepared epimeric natural product **68'**, an exocyclic methylene group was introduced at the C4 position to obtain the other desired natural product **70** and another natural product **71** was subsequently provided through the stereoselective exocyclic double bond reduction from the *exo* face. The spectral data of the synthetic compounds **71** and **70** were compared with those already recorded and reported for the natural products,^{94,98} and found being identical. Finally, noriridoid scholarein **80** was synthesized from intermediate **61**. The latter was first converted into intermediate **63** as a TIPS ether in several steps. Then, with **63** available in hand, its C4-tosylate group in the tetrahydropyran ring was eliminated and the resultant dihydropyran was oxidized to obtain the δ -lactone, the desired natural product isoboonein **65** (Scheme 10).⁹⁴

Hamigeran A–D are members of a family of metabolites which are isolated from the extract of sponge *Hamigera tarangaensis* by Cambie and co-workers in 2000.⁹⁹ They actually are a small class of brominated terpenes. Among members of this family, hamigeran-B **97** has exhibited remarkable biological potency as it showed 100% virus inhibitory property toward both herpes and polio viruses with negligible cytotoxicity.¹⁰⁰ Its tricyclic structural backbone comprising an aromatic ring, has attracted the attention of organic synthetic chemists. Therefore, several research groups have focused on its total synthesis. The first asymmetric synthesis was achieved by Nicolaou and co-workers relied on an asymmetric Diels–Alder reaction as key step,¹⁰¹ Clive *et al.* employed radical cyclization for the construction of the five-membered ring conducting both a racemic and asymmetric synthesis.¹⁰² Thereafter, Trost and co-workers applied an asymmetric allylic alkylation as the source of asymmetry in a novel synthetic strategy for total synthesis of hamigeran-B.¹⁰³ Wright *et al.* accomplished and reported the synthesis of hamigeran scaffold by employing an effective electro-oxidative coupling reaction.¹⁰⁴ Very recently, Lovely and co-workers completed the framework of the tricyclic structure of hamigeran using Pauson–Khand reaction. In this strategy, cyclization only occurred when the olefin-containing group was tethered to the aromatic backbone to diminish its conformational movement. To this purpose, they selected silylene protecting group. Then, effective formation of the aryl enyne from a salicylic acid derivative was achieved through *ortho*-lithiation and Sonogashira cross-coupling reaction.



Scheme 10 Total synthesis of noriridoid scholarein A 80.

A concise strategy for the total synthesis of hamigeran B **97** commenced from the easily accessible *tert*-butyl amide **84** (ref. 105) which was provided by standard acylation of salicylic acid.





Scheme 11 Total synthesis of hamigeran B 97.

The latter upon treatment with *n*-BuLi leading to the formation of aryl lithium which was trapped with iodine to give **85**. Upon cleavage of the methyl ether in **85** with excess BBr_3 , amide **86** was obtained upon treatment with Meerwein's salt and then aqueous base. The methyl ester was provided in 89% yield *via* three sequential steps. Compound **87** was then converted into **88** in two steps. The latter was then subjected to Sonogashira cross-coupling with 2-methyl-2-butynol which upon treatment with **90**, hydrolysis of the amide employing the two-step reaction *via* the imitate gave **91**. Then, compound **91** underwent desilylation to afford **92**. Oxidation state adjustment of **92** *via* reduction using DIBAL-H gave benzyl alcohol and oxidation with MnO_2 afforded the corresponding aldehyde **93** as key intermediate. Reaction of the latter with methallylmagnesium chloride proceeded cleanly to give the diol **94**, which upon treatment with $(t-Bu)_2Si(OTf)_2$ afforded the silylene derivative **95**

in high yield. Pleasantly, the enyne **95** was converted into the $Co_2(CO)_8$ complex, with subsequent thermal activation at $70^\circ C$ in toluene under PKR resulted in the construction of desired tetracyclic adduct **96** in 70% yield as a sole diastereomer.¹⁰⁶ The desired stereochemical biases was induced by intramolecular PKR resulting in the placement of the peripheral *exo* substituents.¹⁰⁷ Finally, the latter was converted to the desired natural product hamigeran B **97** in two steps (Scheme 11).¹⁰⁴

2.2. Terpenoids

Terpenoids represent a highly diverse group of natural products with wide applications. Terpenoids, also known as isoprenoids, are the most numerous and structurally diverse natural products found in many plants. Several studies, *in vitro*, preclinical, and clinical have confirmed that this class of compounds



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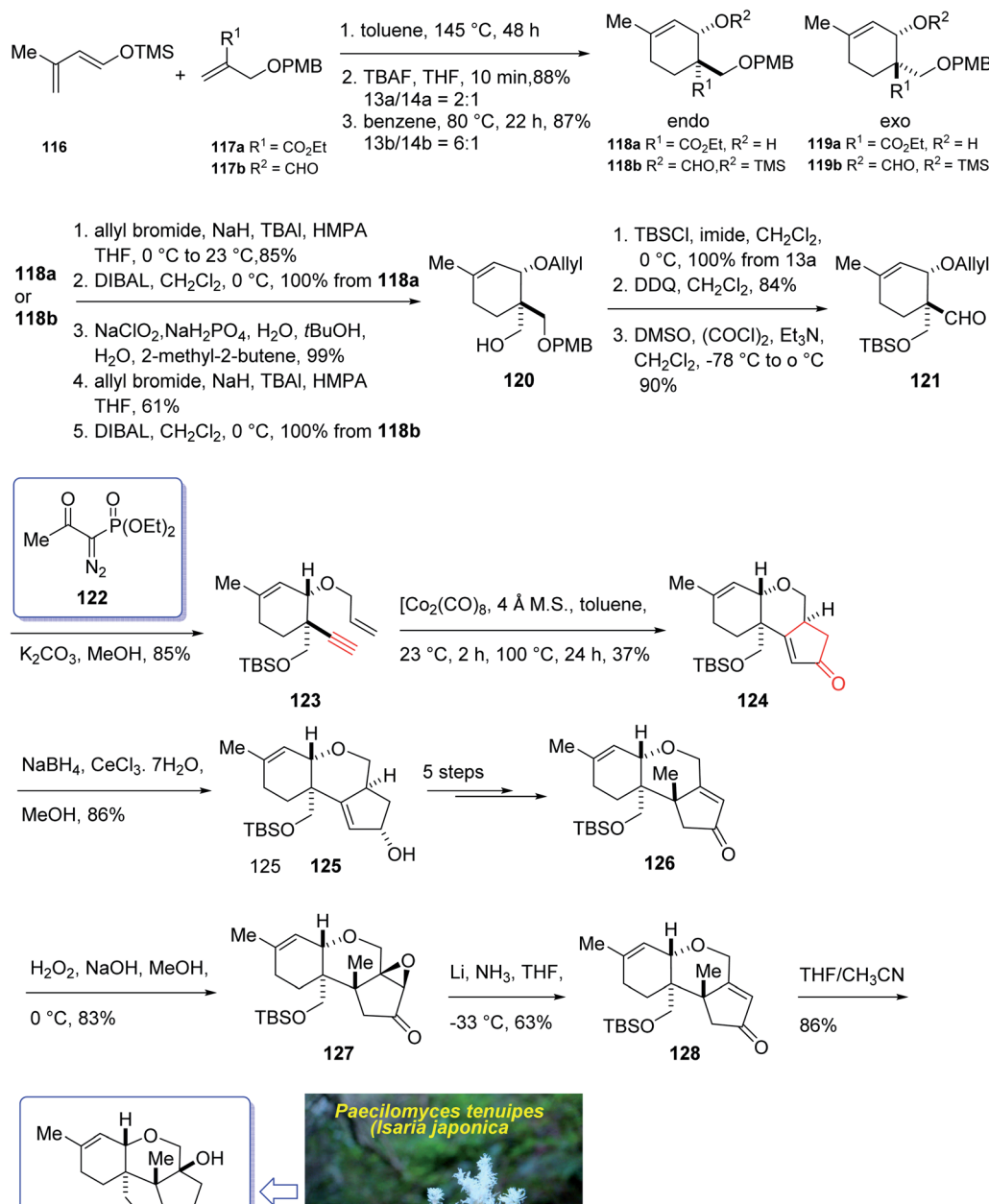
stereoselective introduction of ad¹-synthon into cyclopentenone **100** *via* intermolecular PKR.¹²³

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cyclopentanone **103** was then subjected to the retro-Diels–Alder conditions reported by Grieco¹²⁵ using AlMeCl_2 as a Lewis acid and maleic anhydride as a cyclopentadiene scavenger to afford cyclopentenone **104** in 86% yield. At this stage, the latter was reduced using DIBAL-H at low temperature to give allyl alcohol **105** in satisfactory yield. Upon allylic substitution, compound **105** was readily derivatized to the corresponding acetate **107** and carbonate **106**. Gratifyingly, reaction of carbonate **106** using sodium hydride as a base afforded the key nucleoside in 84% yield with a 4 : 1 dr of N9/N7 regioisomers. The desired regioisomer at N9 **109** was obtained in 67% yield after chromatographic purification, and ee > 99% determined by chiral HPLC. Finally, the key intermediate **110** was converted into

desired natural products *i.e.* enantiomerically pure (–)-carbovir **111** and (–)-abacavir **115** using compound **110** and **114**, respectively (Scheme 12).¹¹⁵

In 2004, Oshima *et al.*¹²⁶ isolated several paecilomyces tenuipes terpenoids from extract of cultured fruiting bodies of *Paecilomyces tenuipes* (*Isaria japonica*). It was actually a common entomopathogenic fungus employed as traditional remedy and healthy foods in China.¹²⁶ Among them, compound **129** (paecilomycine A), at 10 nm, is able to promote neurite outgrowth in PC 12 cells. It was also known that paecilomycine A **129** is considerably more potent than scabronine G in increasing NGF levels. In 2007, the appearance of a fascinating skeleton for the total synthesis of paecilomycine A **129** (ref. 127) was stimulated



Scheme 13 Total synthesis of paecilomycine A **129**.



Review

by Danishefsky and co-workers for the isolation and structure elucidation. In 1961, Martin and Hill reported an initial total synthesis of racemic **129** *via* Diels–Alder reaction to many early hindrances in their efforts¹²⁸ as well as using Pauson–Khand

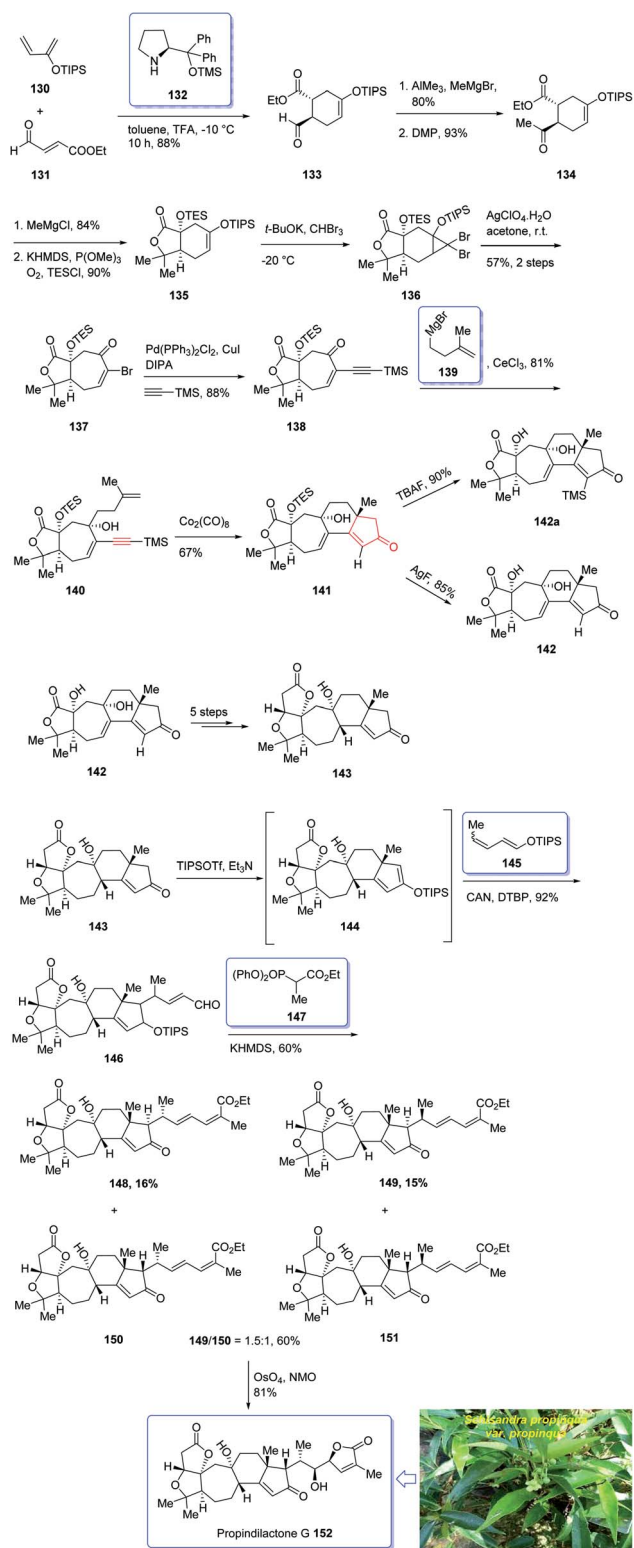
reaction. Initially, the Diels–Alder reaction of **116** (ref. 129) and **117a**,¹³⁰ under severely thermal conditions (at approximately 145 °C in toluene) followed by deprotection, afforded a 2 : 1 mixture of **118a** and **119a** with 88% yield. As illustrated, O-alkylation performed to give the corresponding allyl ether. Reduction of the ester group produced compound **120** which followed by protection as a silyl ether and deprotection of the PMB group followed by oxidation of the obtained alcohol to aldehyde afforded compound **121**. The compound **123** was afforded *via* Bestmann–Ohira reagent (dimethyl 1-diazo-2-oxopropyl phosphonate) with compound **121**. An important compound available in hand **123**, was submitted to intramolecular Pauson–Khand reaction at 100 °C (ref. 19 and 131) to afford the sole stereoisomer **124** in 37% yield. The compound **126** was transformed to desired natural product paecilomycine A **129**, involving several group functional transformation as illustrated in (Scheme 13).¹²⁷

In 2008, Sun *et al.*¹³² for the first time isolated propindilactone G **152** (ref. 132) which is a member of a novel family of nor-triterpenoids¹³³ from different species of Schisandraceae family (*Schisandra propinqua* var. *propinqua*) from Southeast Asia.¹³² It has been used as folk herb in China for liver protection and immune-regulation for a long time.¹³³ Structurally, propindilactone G **152** contains an exceptional 5/5/7/6/5 pentacyclic scaffold having seven chiral centers which three of them are quaternary stereogenic centers.¹³⁴ Primary biological screening of propindilactone G **152** specified that these kinds of nortriterpenoids show auspicious anti-HIV potency.¹³⁵ Their, interesting chemical structures in combination with their insufficiency in nature, which restricts their further biological screening, have prompted great interest among¹³⁶ synthetic organic chemists to design a pathway for the total synthesis of propindilactone G1.

In 2015, a brief total synthesis of (+)-propindilactone G **152** using Pauson–Khand reaction as a key step was accomplished and reported by Yang *et al.*¹³⁷

This strategy started with an asymmetric Diels–Alder reaction of diene **130** and dienophile **131** in the presence of a chiral ligand (Jorgensen–Hayashi catalyst) **132** which afforded (–)-ester **133** in high chemical yield and in excellent ee (98%). After several steps, the latter was transformed into enyne **140** as a sole isomer. Then, the latter was subjected to PKR conditions which upon treatment with Co₂(CO)₈ in the presence of Celite¹³⁸ in refluxing toluene afforded cyclopentenone subunit **141** bearing an all-carbon quaternary chiral centers. After several steps, the latter was converted to a mixture of **148**, **149** and **150/151**. Pleasantly, compound **150** was converted into the desired natural product propindilactone G **152** upon treatment with OsO₄ as oxidant and in the presence of NMO¹³⁹ as a co-oxidant. These reactions desired the total synthesis of (+)-propindilactone G **152** in only twenty steps (Scheme 14).¹⁶⁶

In 2005, Sun and his research group initially isolated lancifodilactone G **181**, as one of the most important members of the schinortriterpenoids family, from the extract of *Schisandra lancifolia*¹⁴⁰ which had been used as anti-hepatitis, antitumor, and anti-HIV agents as traditional medication.¹³³ Due to these



Scheme 14 Total synthesis of (+)-propindilactone G **152**.



Scheme 15 Total synthesis of lancifodilactone G 181.

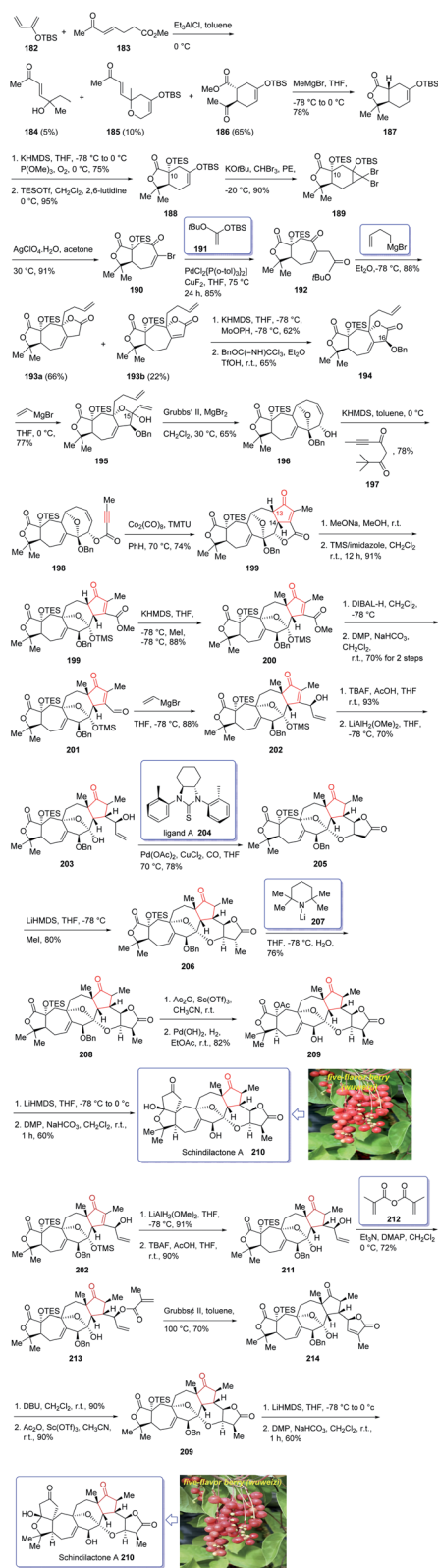
biological activities and interesting chemical structure of **181**, many attempts were made to their total synthesis,¹⁴¹ with the goal of fast-tracking of the assessment of its pharmacological

activity. Asymmetric total synthesis of structurally fascinating and highly oxygenated lancifodilactone acetate **G 7** was accomplished in twenty-eight steps from commercially available 2-(triisopropylsiloxy)-1,3-butadiene **153** reported by Yang *et al.* in 2017.¹⁴² The total synthesis started with asymmetric intermolecular Diels–Alder reaction¹⁴³ of diene **153** with dienophile **154** catalyzed by oxazaborolidine **155** (ref. 144) to provide ketoester **156** in satisfactory chemical yield and 87% ee. After several steps, the latter was transformed into enyne **168** as an appropriate PKR precursor. The enyne **168** was then subjected a PKR upon treatment with the complex of tetramethyl thiourea (TMTU) and $\text{Co}_2(\text{CO})_8$ under already secured optimal conditions to afford enone **169** in 73% yield as a single isomer. Next, the enone **169** was transformed after several steps into ketone **178**. The latter can be converted to lancifodilactone **G** acetate **179** in two steps including a Pd/C-catalyzed hydrogenation. On the other hand, ketone **178** was converted into **180** which after several steps was converted to the desired natural product lancifodilactone **G 181** (Scheme 15).¹⁴²

(±)-Schindilactone **A 210** is a member of family of Schisan-draceae which is valuable from both economic and medicinal points of view.¹⁴⁵ More than 20 species of Schisandraceae were found in China which have extensively been used as traditional medicines¹⁴⁶ over 2000 years in 2008 Han-Dong Sun research group isolated over 70 nortriterpenoids from Schisan-draceae.^{147,148} Among them, schindilactone **A 210** (ref. 149 and 150) was found being conspicuous member with eminent biological potencies including inhibition of tumor growing and hepatitis, and also as anti-HIV-1.¹⁴⁷ In spite of its prominence, only small amounts of schindilactone **A 210** can be obtained from natural sources even for its biological potencies screening, thus, its total synthesis has attracted much attention of synthetic organic chemists.

In 2012, Yang *et al.*¹⁴⁵ accomplished and reported the total synthesis of schindilactone **A 210**. Their strategy involved (a) an Ag-mediated ring-expansion reaction to obtain vinyl bromide **180** from dibromocyclopropane **189**; (b) a Pd-catalyzed cross-coupling of vinyl bromide **190** with a copper enolate to provide ketoester **192**; (c) a RCM reaction to obtain oxabicyclonononol **196** from diene **195**; (d) construction of cyclopentenone segment in substrate **199** via catalyzed Pauson–Khand reaction. This total synthesis started with hetero-Diels–Alder reaction between diene **182** and dienophile **183** in toluene at 0 °C in the presence of Et_2AlCl ¹⁵¹ to furnish a mixture containing compound **184** (about 5% obtained from the alkylation of Et_2AlCl to the keto group in dienophile), compounds **185** and **186** (about 10%). Then, upon the treatment of ester **186** with MeMgBr , lactone **187** was obtained in good yield. The latter was converted to enyne **198**, as an appropriate PKR precursor, after several steps involving various functional group transformations. In a key step, enyne **198** was subjected to PKR under the secured optimal PKR conditions ($\text{Co}_2(\text{CO})_8$ /TMTU in dry benzene under a CO atmosphere (balloon) at 70 °C for 4 hours) affording compound **199** in satisfactory yield. Then, the latter converted into alcohol **209** after several steps. Next, alcohol **209** was subjected to a Dieckmann-type condensation, upon treatment with LiHMDs in THF at −78 °C, followed by





Scheme 16 Total synthesis of (±)-schindilactone A 210.

oxidation using DMP to afford the final desired natural product schindilactone A 210 in 60% overall yield over two steps (Scheme 16).¹⁴⁵

2.3. Diterpenes and diterpenoids

The active component isolated from the extracts of the branch of a *Daphnopsis americana* tree in Costa Rica was so-called guanacastepene A. It is actually a diterpene having a unique carbon skeleton structure.¹⁵² It exhibited very high potency towards faecal (VREF) pathogens, methicillin-resistant *S. aureus* (MRSA) and vancomycin-resistant E.¹⁵² Guanacastepene A 225 has a unique carbon skeleton and a highly functionalized upper half, thus it has attracted much attention of synthetic organic chemists worldwide, as an interesting compound for total synthesis. Due to these waves of interest, the total synthesis of 225 has been attempted and successfully achieved by several research groups.¹⁵³ Danishefsky *et al.*^{154,155} accomplished and reported the first total synthesis of guanacastepene A 225, but before long thereafter Snider *et al.* achieved and reported total synthesis of 225.¹⁵⁶ Interestingly, both of these protocols started with formation of the five-membered ring with subsequent annulations of the seven- and six-membered rings. A conceptually new pathway to the highly functionalized tricyclic core of guanacastepene A 225 was designed through the use of rhodium(i)-catalyzed intramolecular allenic Pauson–Khand reaction for the construction of seven-membered rings.¹⁵⁷ This strategy commenced with Smith's enone 215 (ref. 158) which in enolate form was alkylated with an appropriate alkyne iodide in the presence of LDA to afford compound 218 in moderate yield.¹⁵⁹ Reaction of the enone 218 with lithium acetylide ethylenediamine complex *via* sequential hydrolysis of the crude product with HCl gave enone 219 in 65% yield over two steps. The hydroxyl group of the latter was protected using TBSCl to afford the corresponding silyl ether 220 in 95% yield. The carbonyl group of the latter, upon reduction following the Luche protocol provided the corresponding secondary alcohol 221 as a mixture of two diastereomers (3 : 1 ratio). Protection of the allylic alcohol gave *tert*-butyldimethylsilyl ether 222 in 88% yield over the two steps. Deprotonation of the terminal alkyne using *n*-BuLi at -78°C , followed by the addition of para-formaldehyde afforded the corresponding propargylic alcohol 223 in satisfactory yield. The alcohol 223 upon treatment with Et_3N and MsCl produced a mesylate, which without purification was added directly to $(\text{Me}_2\text{PhSi})_2\text{Cu}(\text{CN})\text{Li}_2$ at -85°C to provide the 3,3-disubstituted allene 224 in 90% yield over two steps. Having pure allenyne 224 available in hand, it was treated *via* Pauson–Khand reaction conditions (10% mol $[\text{Rh}(\text{CO})_2\text{Cl}]_2$ in toluene at 80°C) to give the 4-alkylidene cyclopentenone 225 as a single product in 65% yield which in fact is the tricyclic ring core system of natural product, guanacastepene A 225 (Scheme 17).¹⁶⁰

Ingenol 233 is a highly oxygenated tetracyclic diterpene which in racemic form imitates the task of diacylglycerol, being the endogenous activator of protein kinase C.¹⁶¹ It is actually isolated from Euphorbia family and its structure was unambiguously elucidated and reported in 1968 by Hecker *et al.*¹⁶² This natural product 233 shows weak activity against protein kinase C.¹⁶³ Ingenol 233 which is the parent compound of several dozen of natural product ingenanes, has the same carbon scaffold but diverse peripheral functionalities.¹⁶⁴ Besides their fascinating “inside-outside” bridged BC ring system,¹⁶⁵ the ingenanes show diverse biological activities.¹⁶⁶ In



1980s, Winkler *et al.* inspired by this significant biological function and complex architecture attempted the total synthesis of **233**.¹⁶⁷ In 1997, the same group used the intramolecular dioxenone photocycloaddition to create its exceptional stereochemical feature.¹⁶¹ In 2002, Winkler *et al.*¹⁶⁷ accomplished and reported the first total synthesis of racemic ingenol. The total synthesis of **233** was completed in forty-two steps in overall yield of 0.042% from commercially available.

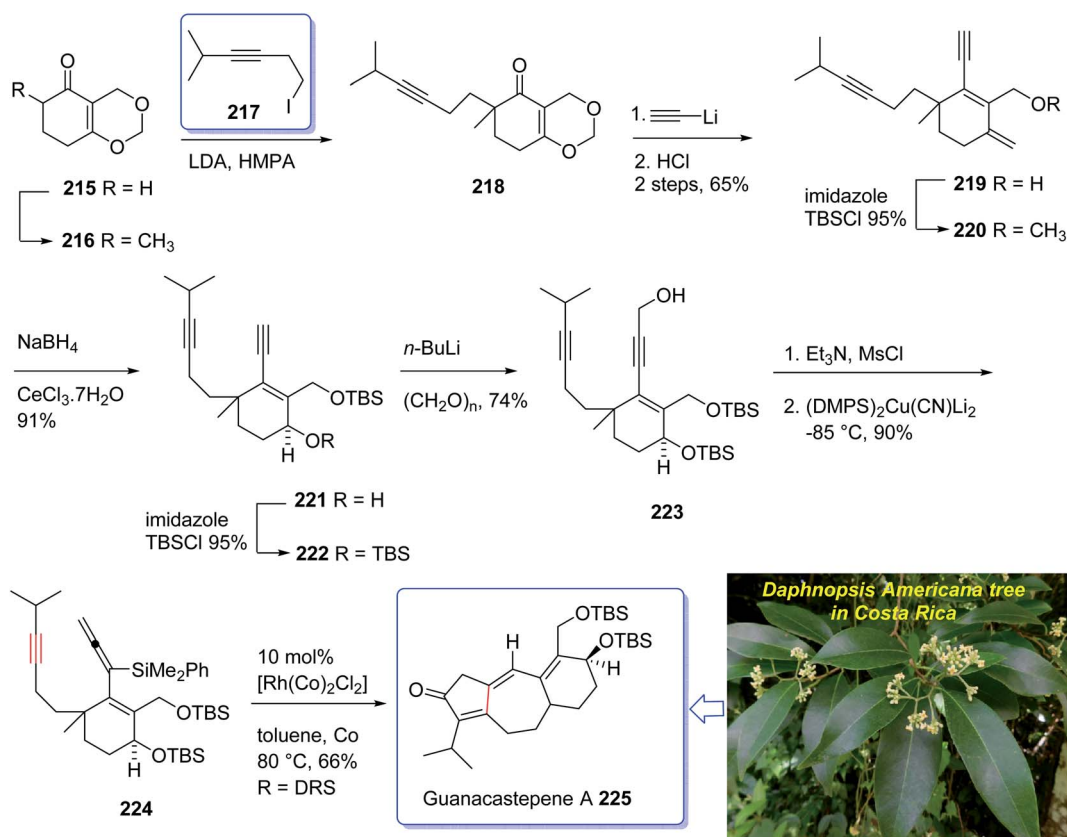
In 2005, Winkler *et al.*¹⁶⁸ achieved and reported the total synthesis of racemic ingenol **233** beginning from unsaturated aldehyde 2-methyleneoct-7-enal **226** which was synthesized in a one-pot fashion involving Swern oxidation of 7-octen-1-ol followed by reaction of the intermediate aldehyde with Eschenmoser's salt.¹⁶⁹ Compound **226** then reacted with the conjugate base of *tert*-butyl acetate to afford compound **227**, which upon oxidation by MnO₂ gave ketoester **228**. Treatment of **228** under dioxenone-forming conditions (TFAA, TFA, Ac₂O, Me₂CO) resulted in the formation of the dioxenone photo-substrate **229** in satisfactory yield. After five steps, the desired methylene photoadduct **230** was obtained. Alkylation of the conjugate base of **230** (LDA, THF, DMPU, -78 °C) with 3-trimethylsilylpropargyl bromide followed by desilylation with TBAF (THF, 100%) the Pauson-Khand substrate **231** was provided. It is worthwhile to mention that the Pauson-Khand reaction of **231** in the presence of the Me₃N-*N*-oxide dehydrate was substantially more effective than the reaction employing anhydrous Me₃N-*N*-oxide. After several steps involving various

functional group transformations, compound **232** was converted to the desired natural product ingenol **233**. In summary, the target **233** was obtained from **226** in overall yield of 0.042% (Scheme 18).¹⁶⁸

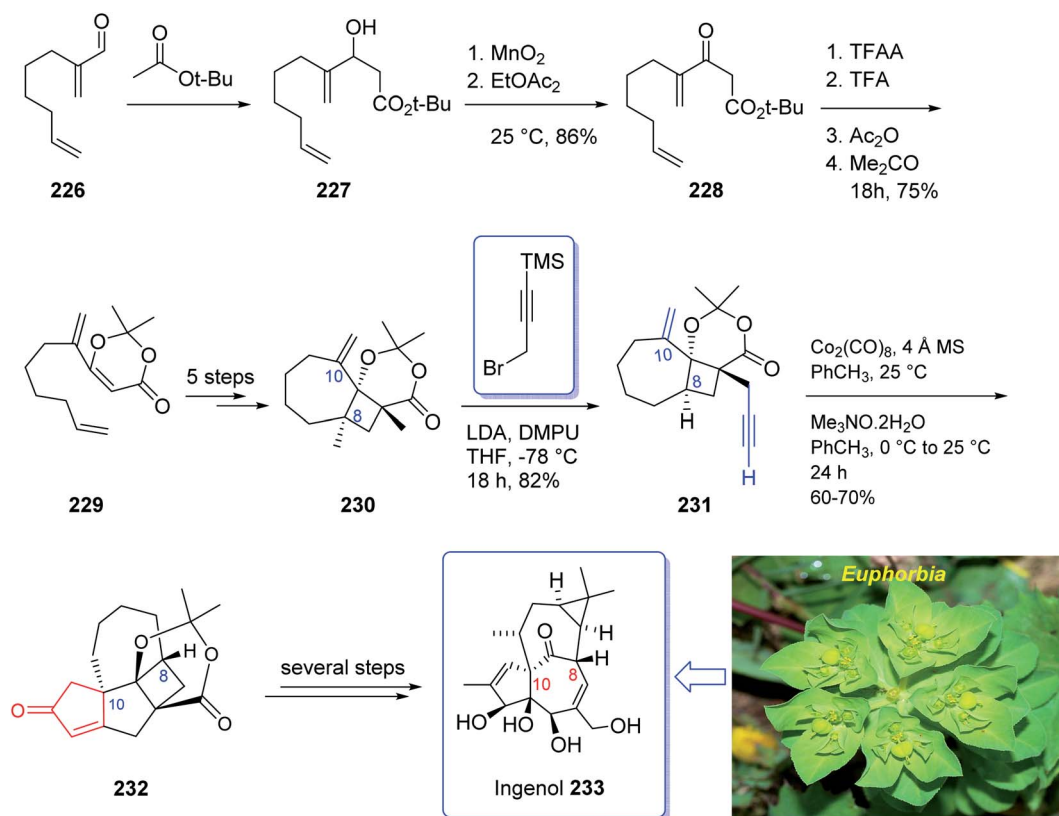
The aquariolides are actually classified as cyclic diterpenes. They were initially isolated from *Erythropodium caribaeorum*, but in 2002 aquariolide A **244** was identified from cultured specimens of this gorgonian¹⁷⁰ by the Andersen research group. Andersen and co-workers in 2003 achieved and reported the isolation of aquariolides A, B, and C from animals growing in the wild.¹⁷¹ In fact, the distinctive feature of these naturally occurring compounds is the "aquarian" backbone, which contains two five-membered rings fused to a nine-membered ring *E. caribaeorum* was found being a source of briarane diterpenes,¹⁷² and the aquariolides which are supposed to generate biosynthetically from a briarane precursor through a di- π -methane rearrangement followed by vinyl-cyclopropane rearrangement.¹⁷¹ A brief biological screening disclosed that aquariolides B and C showed moderate *in vitro* cytotoxicity against human breast cancer MCF-7 cells.²⁰²

In 2006, the total synthesis of **244**, as a prelude was accomplished and reported by Burnell and co-workers involving a diastereoselective Pauson-Khand reaction and subsequent ring expansion.¹⁷³

In this approach, the total synthesis of aquariolide A **244** is commenced from benzyl ester **234**, which was converted into ynone **235** in 85% yield *via* Yamaguchi procedure.¹⁷⁴ Then,



Scheme 17 Total synthesis of guanacastepene A **225**.



Scheme 18 Total synthesis of ingenol 233.

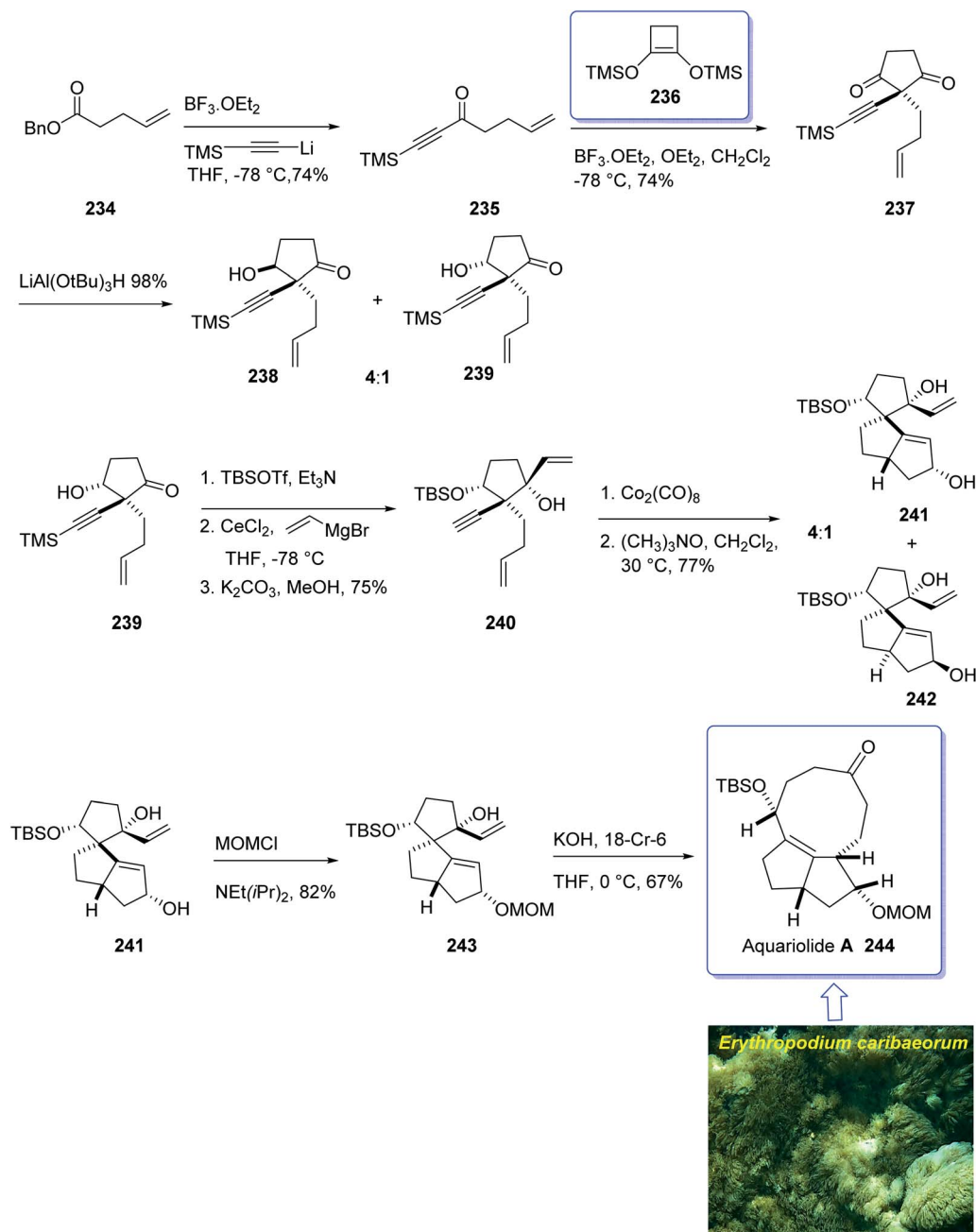
ketone **235** was subjected into geminal acylation upon reaction with **236** in the presence of $\text{BF}_3 \cdot \text{OEt}_2$ to provide diketone **237**. Protection of the latter hence provided the extra advantage of further directing addition of a vinyl Grignard reagent *via* blocking the alkenyl face of the cyclopentanone. Thus, the mono reduction of diketone **237** using lithium tri(*tert*-butoxy)-aluminumhydride afforded diastereomeric alcohols (**238** and **239**) in a ratio of 4 : 1. Keto-alcohol **239**, the other product from the reduction of **237**, was reacted with triethylsilane in TFA, and then protected by TBSCl and reacted with vinylmagnesium bromide in the presence of anhydrous CeCl_3 ,¹⁷⁵ followed by basic methanolysis of the TMS group to give compound **7** as the sole product. It was obtained upon treatment of the latter with $\text{Co}_2(\text{CO})_8$ and trimethylamine *N*-oxide, under the Pauson–Khand reaction conditions (in dichloromethane at 30°C) as a 4 : 1 ratio of inseparable diastereomers **241** and **242** (in a ratio of about 4 : 1 determined by NMR).

Protection of the major alcohol **241** as a MOM ether afforded **243**. As expected, subjecting **243** to appropriately basic conditions smoothly provided ketone **244**. The latter *via* viable routes to the ring system of the aquariolide diterpenes have been accomplished in two steps *via* generation of **241** and **243** in which the latter was transformed to desired natural product aquariolide A **244** in 67% overall yield (Scheme 19).¹⁷³

Cyanthiwiggins¹⁷⁶ **257** contain a cyclohepta[*e*]indene core and structurally belongs to the diverse cyathane class of diterpene.¹⁷⁷ They were isolated from the extract of the marine

sponges *Epipolasis reiswigi* and *Mermekioderma styx*. Biologically, cyanthiwiggins were found being cytotoxic towards A549 lung cancer cells and primary tumor cells.¹⁷⁸ Cyanthiwiggins have 5-6-7 tricarboxylic core with carbons in different oxidation states. An efficient total synthesis of cyanthiwiggins **257** was achieved and reported by Phillips *et al.* in 2005.¹⁷⁹ Then, the Stoltz research group developed and reported a brief strategy involving double asymmetric catalyzed alkylation and an RCM reaction as the key steps in the total synthesis of cyanthiwiggins B, F, and G.¹⁸⁰ In 2013, Gao *et al.* accomplished a brief synthetic strategy for the total synthesis of cyanthiwiggins A, C, G, and H¹⁸¹ comprising a common intermolecular [4 + 2] cycloaddition and RCM reaction as vital steps.

In 2018, Yang *et al.* developed a strategy for the total synthesis of 5-*epi*-cyanthiwigin I **257** using Pauson–Khand reaction (PKR) as a vital step. The total synthesis of **257** began with **245** which was reacted with allyl bromide **246** in the presence of LDA to afford diene **247** in high yield.¹⁸² The latter was further treated with Grubbs II catalyst **248** (ref. 183) to provide ketoester **249**, which upon alkylation with 4-iodo-2-methylbut-1-ene **250** in the presence of KO^tBu in *tert*-BuOH afforded ketone **251** in high yield. The key intermediate **254** was obtained by converting the ketone group of the latter upon treatment with LDA to the corresponding enolate, which was subsequently reacted with Comins' reagent **252** to provide the respective vinyl triflate **253** in excellent yield. The latter was reacted with 3-methylbut-1-yne under Sonogashira coupling



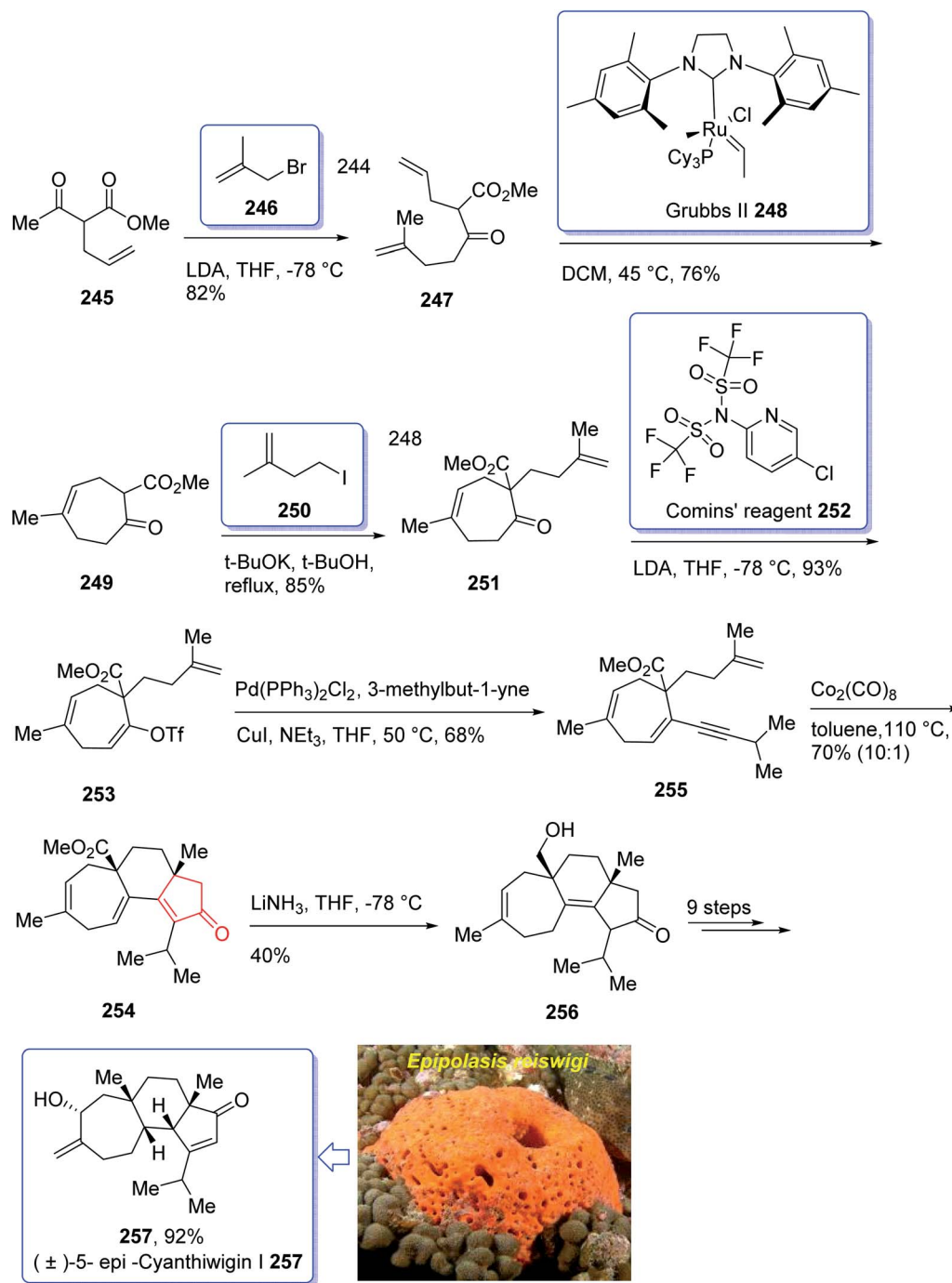
Scheme 19 Total synthesis of aquariolide A 244.

reaction conditions with enyne **254** in good yield. Next, enyne **254** was treated with Co₂(CO)₈ (1.2 equiv.) in refluxing toluene under PKR conditions for the construction of the indene core **255** of cyanthiwigins in good yield. Upon treatment of **255** with Li/NH₃ in THF at -78 °C, compound **256** was obtained in 40% yield which after nine steps was transformed to the desired C5-*epi*-cyanthiwigin I **257** in 92% yield. The developed chemistry enables the total synthesis of 5-*epi*-cyanthiwigin I **257** in seventeen steps and can be used for the total synthesis of other cyanthiwigins (Scheme 20).¹⁷⁷

Naturally occurring compound, ryanodine^{184,185} and its hydrolysis product ryanodol **270**,^{185,186} are among the most

highly oxidized diterpenoids reported so far. Ryanodol **270** was isolated from the extracted tropical shrub *Ryania speciosa* Vahl reported by Pepper and Carruth in 1945,¹⁸⁷ and then by other research groups.¹⁸⁸ It showed insecticidal properties¹⁸⁴ and is an important family of ion channels that regulate intracellular Ca²⁺ release and play a key role in signal transduction.¹⁸⁹ In 2016, Chuang *et al.*¹⁹⁰ achieved and reported a brief total synthesis of (+)-ryanodol in fifteen steps starting from the commercially available terpene (*S*)-pulegone **258**. In this strategy, the utilization of a Pauson-Khand reaction rapidly construct the carbon skeleton along with a SeO₂-catalyzed oxidation to assemble three oxygen atoms *via* a single step. In this approach, reaction





Scheme 20 Total synthesis of (±)-5-epi-cyanthiwigin 257.

of (*S*)-pulegone **258** with KHMDS at $-78\text{ }^\circ\text{C}$ followed by drop-wise addition of **259** gave α,α' -diol **260** which was isolated as a single diastereomer in 42% yield. Treatment of diol **260** with excess benzyl chloromethyl ether resulted in the protection of both alcohols as benzyloxymethyl ethers to afford **261**. At this point, the D-ring was created by an effective four-step sequence. Initially, propynylmagnesium bromide was added to **261** at $0\text{ }^\circ\text{C}$ which proceeded in 5 : 1 dr to give the equatorially disposed alkyne in 81% yield. The latter upon ozonolysis was cleaved to give methyl ketone **262**. Then, the ketone was effectively

transformed to α,β -unsaturated lactone **263** via 1,2-addition of ethoxyethynylmagnesium bromide with subsequent sequential Ag-catalyzed cyclization and elimination reactions.¹⁹¹ Having lactone **263** available in hand, 1,4-addition of magnesium divinyl cuprate gave the respective enyne **264** as a sole diastereomer in satisfactory yield. In a crucial step, compound **264** was treated with 1 mol% $[\text{RhCl}(\text{CO})_2]_2$ (ref. 30) under an atmosphere of carbon monoxide (PKR) gave the desired intermediate **265** as a sole diastereomer and in high chemical yield. This intermediate was transformed after four steps to afford



Scheme 21 Total synthesis of (+)-ryanodol 270.

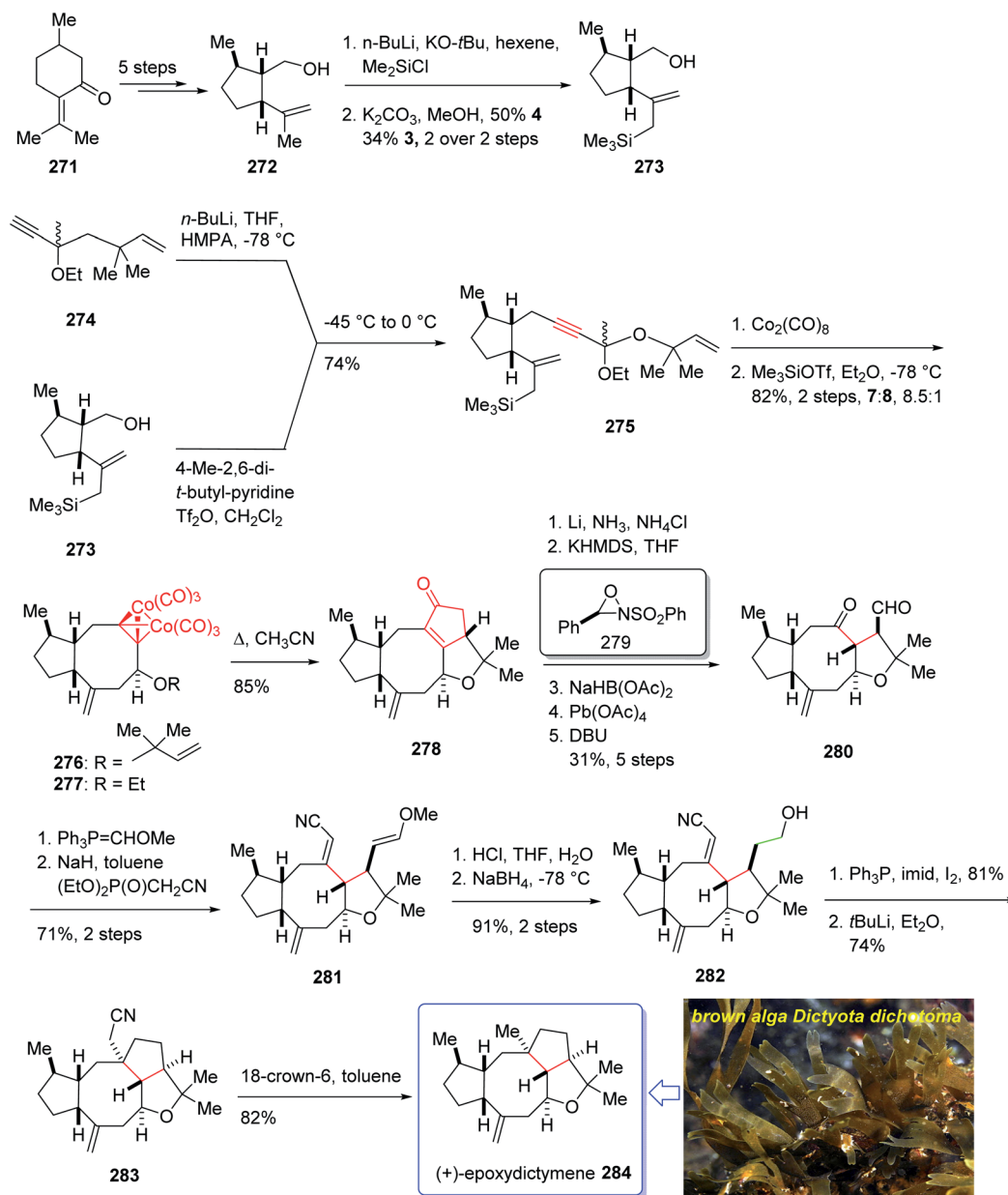
(+)-anhydroryanodol **269** in 61% yield. The latter was then treated with trifluoroperacetic acid to give epianhydroryanodol epoxide which is subjected to reductive cyclization to give the desired natural product (+)-ryanodol **270** in 0.42% overall yield over fifteen steps starting from commercially available (*S*)-pulegone **258** (Scheme 21).¹⁹⁰

Epoxydictymene **284** is a diterpene naturally occurring compound which is present in the brown alga *Dictyota dichotoma* isolated.¹⁹² In the early 1980s, Matsumoto *et al.*¹⁹³ isolated cyclononane and hydroazulene diterpenoids and in 1983 Ishida

*et al.*¹⁹⁴ elucidated its structure as epoxydictymene **284**. It contains a 5-8-5-5 (ref. 195) tetracyclic scaffold containing a strained *trans*-3-oxabicyclo [3.3.0] octane (*trans*-5-5).¹⁹⁶

The total synthesis of **284** was achieved and reported by Schreiber and co-workers in 1994 (ref. 197) by employing Pauson-Khand reaction as a key step. The total synthesis started from commercially available (+)-pulegone **271** which after several steps was converted into compound **272**. The latter was transformed in two steps involving protection of hydroxyl group by TMSCl in the presence of *n*-BuLi to afford compound **273**.





Scheme 22 Total synthesis of epoxydictymene 284.

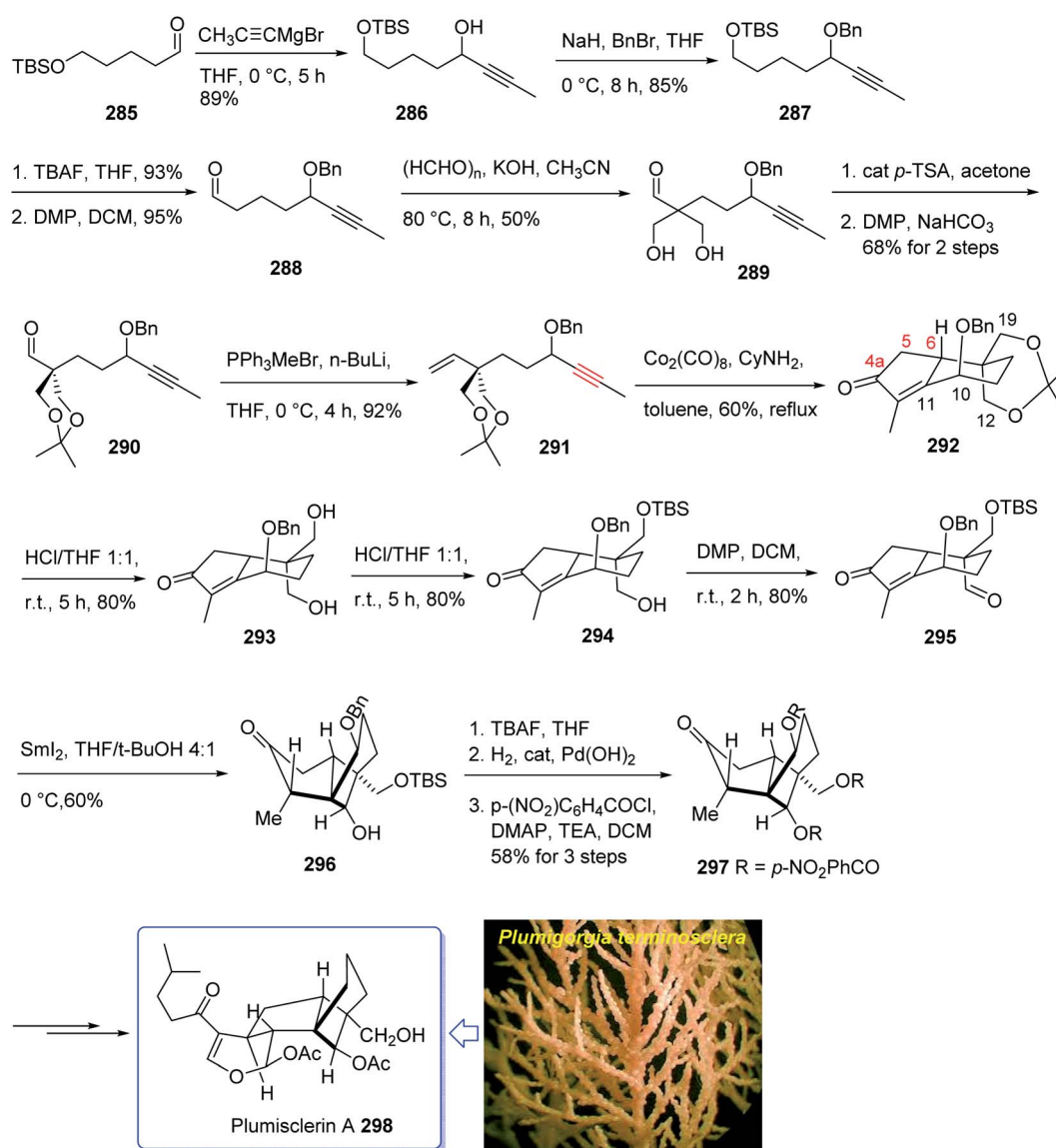
Reaction of lithium anion of 274 and alcohol 273 through displacement of a triflate ester under carefully controlled conditions, furnished alkyne 275 as suitable PKR precursor in high yield. In a crucial stage, the latter was treated with $\text{Co}_2(\text{CO})_8$ under PKR conditions to afford the desired organometallic cluster, which was then subjected to catalyzed Lewis acid cyclization¹⁹⁷ (Me_3SiOTf , Et_2O , -78°C , 15 min) to furnish ethers 276 and 277 in 82% overall yield as an 8.5 : 1 mixture of diastereomerically pure compounds. After several steps, ether 276 was converted into nitrile 283 containing the complete framework and configurations of the target natural product. Reductive decyanation¹⁹⁸ of nitrile 283 gave the desired natural product (+)-epoxydictymene 284 (Scheme 22).¹⁹²

Plumisclerin A 298, an exceptional marine diterpenoid, was initially isolated by Reyes research group in 2010 from the samples of *Plumigorgia terminosclera* collected at Mayotte Island.¹⁹⁹ Structurally, plumisclerin A 298 has a complicated and compact ring system bearing a fully-substituted cyclobutane (C ring), a bridged cyclohexane (D ring), a poly-substituted cyclopentane (B ring), as well as a fused dihydropyran ring (A ring). Its distinctive rigid tricyclo [4,3,1,0^{1,5}] decane framework probably is the first reported of such a plumisclerane framework in the field of naturally occurring compounds. Dihydropyran ring (A ring) in plumisclerin A 298, is *trans*-fused to the cyclopentane ring (B ring), while, most terpenoids have a *cis*-fused dihydropyran ring.²⁰⁰ It also bears seven chiral centers involving two all-carbon quaternary chiral

centers compactly spread in the molecule. Plumisclerin A **298** exhibits modest cytotoxicities towards numerous common tumor cells such as lung, colon cancer and breast cancers.¹⁹⁹ The interesting structure and its valuable biological activity, make plumisclerin A **298** as an important target for synthetic organic chemists. An effective strategy for the synthesis of the tricyclo [4,3,1,0^{1,5}] decane core (B/C/D rings) of plumisclerin A **295** was designed and successfully accomplished by Yao and co-workers in 2015.²⁰¹ In this strategy, the Pauson–Khand reaction and a SmI₂-catalyzed radical 1,4-conjugate addition play vital roles in the construction of fully functionalized 5,6-fused rings and the very strained cyclobutanol moiety with exact relative stereochemistries, respectively.

This attempt started with alcohol **286** which in turn is provided from the addition of propynyl magnesium to the known aldehyde **285**.²⁰² Alcohol **286** is converted after several steps to enyne **291** as a suitable precursor of PKR *via* various

functional group transformations as well as protecting-deprotecting processes. Having enyne **291** available in hands, it was investigated in a PKR^{5,7} explaining various combinations to find an optimal reaction conditions. Eventually, enyne **291** was converted to the desired compound **292** in modest yield when PKR was catalyzed by Co₂(CO)₈ in the presence of cyclohexylamine (CyNH₂). Worthy to mention that, the C10-OBn of enyne **291** in PKR, played a vital role to control the newly generated chiral center at C6 position, when the requisite cyclopentenone ring was constructed. Then, compound **292** was converted to compound **295** in three steps in which the latter upon treatment with SmI₂ in THF and *t*-BuOH (4 : 1, v/v) as mixed solvents at 0 °C afforded the anticipated bridged-compound **296** in good yield. After three steps, the latter was unambiguously transformed into the respective tri-*p*-nitro-benzoate **297** (confirmed by single crystal X-ray analysis). The latter was then transformed to the desired plumisclerin A **298** after several steps. In



Scheme 23 Total synthesis of plumisclerin A **298**.



Marrulibacetal **315**, a diterpenoid, was initially isolated from the aerial parts of *Marrubium globosum* ssp. *libanoticum* by Borrelli *et al.* in 2009.²⁰³ The same group reported the structural elucidation of marrulibacetal **315** in the same year.^{203,204}

Marrubium globosum ssp. *libanoticum* have been used for a long time as medicinal plant which are used as hypoglycemic, febrifuge, antispasmodic, and anti-inflammatory drugs in Northern Lebanon.²⁰³

In 2016, Nakamura *et al.*²⁰⁵ achieved and reported a stereoselective total synthesis of (+)-marrubiin commencing from a chiral scaffold *via* the CyNH₂-catalyzed Pauson–Khand reaction followed by oxidative cleavage of the resultant



Scheme 24 Total synthesis of marrulibacetal **315**.

cyclopentenone ring. This strategy started from enyne **299** (ref. 206) which was subjected to PKR using $\text{Co}_2(\text{CO})_8$ in CH_2Cl_2 at ambient temperature with subsequent addition of cyclohexylamine, followed by dilution with dichloromethane, and refluxing the mixture to afford tricyclic enone **300**. The latter was transformed to a 1 : 1 diastereomeric mixture of *cis*-diols **312** and **313** after several steps, including various functional group transformations. Ultimately, internal transacetalization of *cis*-diols **312** and **313** in the presence of TsOH in benzene gave the desired natural product, (–)-marrulibacetal **315**, along with its isomer (Scheme 24).^{205,207}

Crinipellin A **331** is classified as diterpenoid.²⁰⁸ It was isolated and reported in 1979 by Steglich *et al.* from the fungus *Crinipellis stipitaria* (Agaricales).^{208,209} It has an interesting chemical structure bearing α -methylene ketone moiety and an exceptional tetraquinane core, containing eight chiral centers, in which three of them are adjoining all-carbon quaternary carbons. From biological points of view, **331** and **332** were found to exhibit antibiotic potencies.^{208,209} The total synthesis of racemic **332** was accomplished in twenty-two steps *via* Barbier annulation by Piers and co-workers and reported in 1993.²¹⁰ In 2014, Lee and co-workers achieved and reported the first asymmetric total synthesis of **331** through a tandem [3 + 2] cycloaddition reaction for the construction of its tetraquinane scaffold containing three successive quaternary chiral centers.^{136,211,212} In 2018, Yang and co-workers²¹³ accomplished and reported the asymmetric total syntheses of (–)-crinipellin A **331** and (–)-crinipellin B **332** in eighteen steps from the commercialized phenol **316**, respectively. These total syntheses featured a developed thiourea/Pd-catalyzed intramolecular Pauson–Khand reaction for the asymmetric construction of the tetraquinane scaffold present in crinipellins. As a matter of fact, the vital step was PKR which provided compound **321** containing two *cis*-configured vicinal chiral centers.²¹⁴ The required enyne ester **320**, as a required precursor of PKR, which was provided *via* the Trost methodology²¹⁵ from already known compound (R)-4-isopropyl-3-methylcyclohex-2-en-1-one **317**. The latter in turn was synthesized in four steps from market purchasable 4-isopropyl-3-methylphenol **316**.²¹⁶ Having compound **318** in hand, it was subjected to an asymmetric Weitz–Scheffer-type epoxidation,²¹⁷ to give ketone **319** which was condensed with *p*-NO₂ArSO₂NHNH₂ followed by treatment with NaHCO₃ to undergo Eschenmoser fragmentation²¹⁸ to give the acetylene ketone **319** in 68% yield. The latter first underwent a Wittig reaction, and the resultant enyne, upon sequential direct treatment with BuLi and ethyl chloroformate afforded enyne **320** as a suitable precursor for PKR. This one-pot transformation is essential to ensure a high yield because the intermediate enyne is volatile. After considerable experimentation, it was found by a crucial step, enyne **320** was treated with a stoichiometric quantity of $\text{Co}_2(\text{CO})_8$ at ambient temperature for a while and then the resultant enyne/Co complex was gradually heated to 76 °C in the presence of 4-methylmorpholine *N*-oxide (NMO) for relatively long time (36 h), to obtain **322** in modest yield but excellent ee (98% ee) after crystallizations. This obtained ketoester **322** was reacted with an organocopper reagent (provided from treatment of allylmagnesium chloride

and $\text{CuBr} \cdot \text{Me}_2\text{S}^{219}$ at –78 °C) proceeded *via* highly diastereoselective conjugate addition reaction, to give the anticipated enolate which was reacted with Me_2SO_4 in the presence of Cs_2CO_3 and tetrabutylammonium fluoride (TBAF)²²⁰ to afford methyl vinyl ether **322** as a sole isomer in satisfactory yield. On

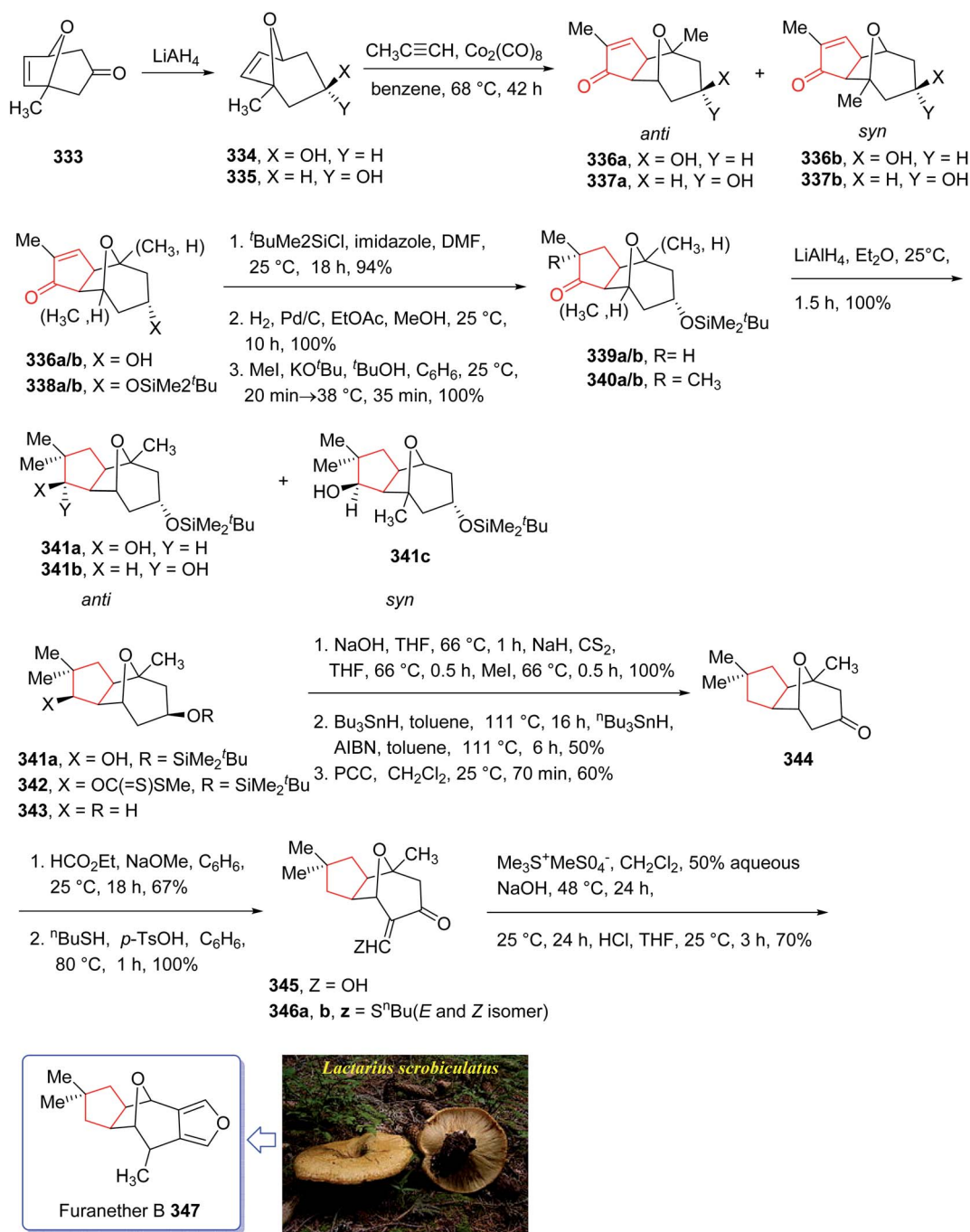


Scheme 25 Total synthesis of crinipellin A **331**, crinipellin B **332**.



the other hand, compound **323** was converted to another key intermediate **325**, in two steps involving regio- and stereo-selective assemblage of quaternary chiral center. This vital intermediate ester **325** was reduced with DIBAL-H, with subsequent oxidation of the resultant using the Dess–Martin reagent followed by reaction with the Bestmann reagent²²¹ to afford enyne **326** as a suitable PKR precursor in good yield. Enyne **326** was then subjected to Pd-catalyzed PKR for the successful construction of tetraquinane **328**. It is worthwhile mentioning that use of TU-1 ligand **327** (ref. 222) improved the

diastereoselectivity of this PKR affectedly. Treatment of **328** with dimethyldioxirane (DMDO) in a Na₂HPO₄ solution²²³ and H₂O₂/NaHCO₃ in sequence afforded the epoxides **329a** and **329b** after several steps in 7% and 38% yields, respectively. Then, compound **329a** was subjected to modified Eschenmoser methylenation upon treatment with *N*-methylanilinium trifluoroacetate and paraformaldehyde in THF at 70 °C to afford the desired natural product crinipellin A **331** in 86% yield.²⁰⁹ Having **329b** in hands, it was converted to the thermodynamically stable compound **330** via isomerization²²⁴ of its α -hydroxy



Scheme 26 Total synthesis of furanether B **347**.



ketone motif. This isomerization of **329b** to **330** was successfully accomplished by treatment of **329b** with various acidic and basic reagents. Compound **330** was then treated with *N*-methylanilinium trifluoroacetic acid (TFA) and paraformaldehyde in THF at 70 °C to afford the other natural product, crinipellin B **332** in 74% yield (Scheme 25).²¹³

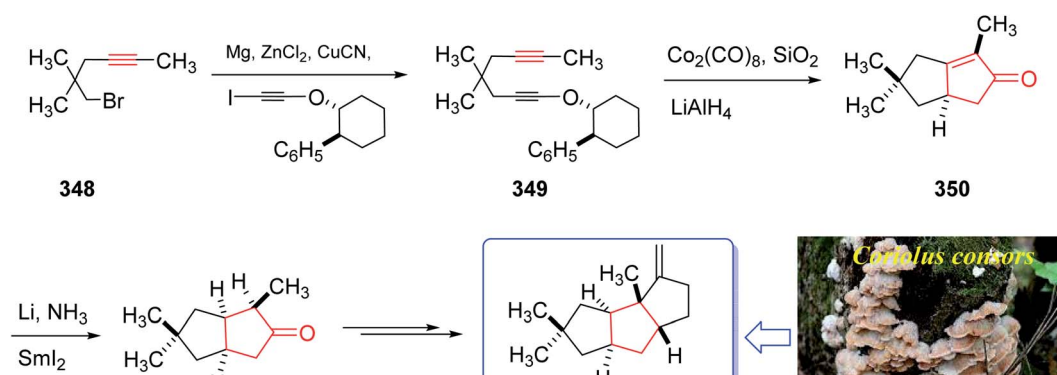
2.4. Sesquiterpenes

Furanether B **347**, a member of the lactarane class of sesquiterpenes, was initially isolated by Vita-Finzi and co-workers in 1980.²²⁵ The total synthesis of furanether B **347** was achieved and reported by Schore in co-workers in 1989 (ref. 226) using PKR as a key step. This strategy started with ketone 1-methyl-8-oxabicyclo [3.2.1] oct-6-en-3-one **333**, provided *via* a reported method by Noyori *et al.* from 2-methylfuran and tetra-bromoacetone.²²⁷ Ketone **333**, upon reduction by lithium aluminum hydride as reducing agent gave *exo* : *endo* isomeric alcohols **334** and **335** in 2 : 1 ratio and in 94% yield. Stereoisomeric alcohols **334** and **335** were reacted with propyne under PKR conditions (Co₂(CO)₈, benzene, heat) to afford a mixture of four isomeric tricyclic ketones (**336a**, **336b** and **337a**, **337b**). Pauson-Khand cycloaddition by reduction of **333** with lithium aluminum hydride. Pauson-Khand cycloaddition of propyne and the mixture of stereoisomeric alcohols **334** and **335** gave 75% yield of a mixture of four isomeric tricyclic ketones (**336a**, **336b** and **337a**, **337b**). Among them, **336a** was converted to ketone **344** after several steps. The latter was then formylated regiospecifically to afford compound **333** in good chemical yield.²²⁸ In solution, **345** contains 75–80% intramolecular hydrogen-bonded (*Z*)- β -hydroxyketone, the remainder being the *E*-isomer and traces of ketoaldehyde. Compound **345** was subjected to Ireland's procedure for the synthesis of the thiomethylene derivative to give **346a** and **346b** in almost quantitative yield.²²⁹ This mixture with treated with trimethylsulfonium methylsulfate in a two-phase system to give an epoxide that upon rearrangement on standing at ambient temperature for 24 h (ref. 230) followed by aromatization in the presence of HCl in THF gave the desired natural product, furanether B **347** in moderate yield.²³¹ Spectral data obtained for synthetic **334** were in agreement with those obtained from the isolated natural product (Scheme 26).²²⁶

The mold metabolite, hirsutene **352**, is parent member of an important class of linear triquinane sesquiterpene. It was initially isolated from the hydrocarbon extracts of fermented mycelium of *Coriolus consors* by Nozoe and co-workers in 1976.²³² Hirsutene **352** did not show any significant biological potency whereas its derivative, diketocoriolin B, exhibited a prominent cytotoxic,²³³ antibiotic and antitumor potencies.²³⁴ Therefore, its total synthesis attracted much attention of synthetic organic chemists.²³⁴ Hirsutene **352** was fully characterized by analysis of combined data, obtained from various spectroscopic techniques, commonly used for the structural elucidation of organic compounds.²³² In 1990, Pericas and co-workers²³⁵ achieved and reported an efficient total synthesis of hirsutene **352** using intramolecular Pauson-Khand reaction.²³² Homochiral diyne **352** was easily synthesized from coupling of **348** *via* a Cu-mediate coupling reaction²³⁶ involving the zinc reagent. In this strategy, enyne **349** as an appropriate PKR precursor was converted initially to the respective *E*-enol ether by treatment of the latter with LiAlH₄ in THF and the subjection of the resultant to PKR conditions (Co₂(CO)₈, SiO₂, purification, 85%) as key bicyclization under mild reaction conditions^{5,237} to afforded enone **350** diastereoselectively. The latter upon either Birch reduction or less effectively, catalytic hydrogenation²³⁸ gave **351** in racemic form. This bicyclic ketone **351**, was transformed to the desired natural product hirsutene **352** in several steps following the previously reported procedure (Scheme 27).²³⁵

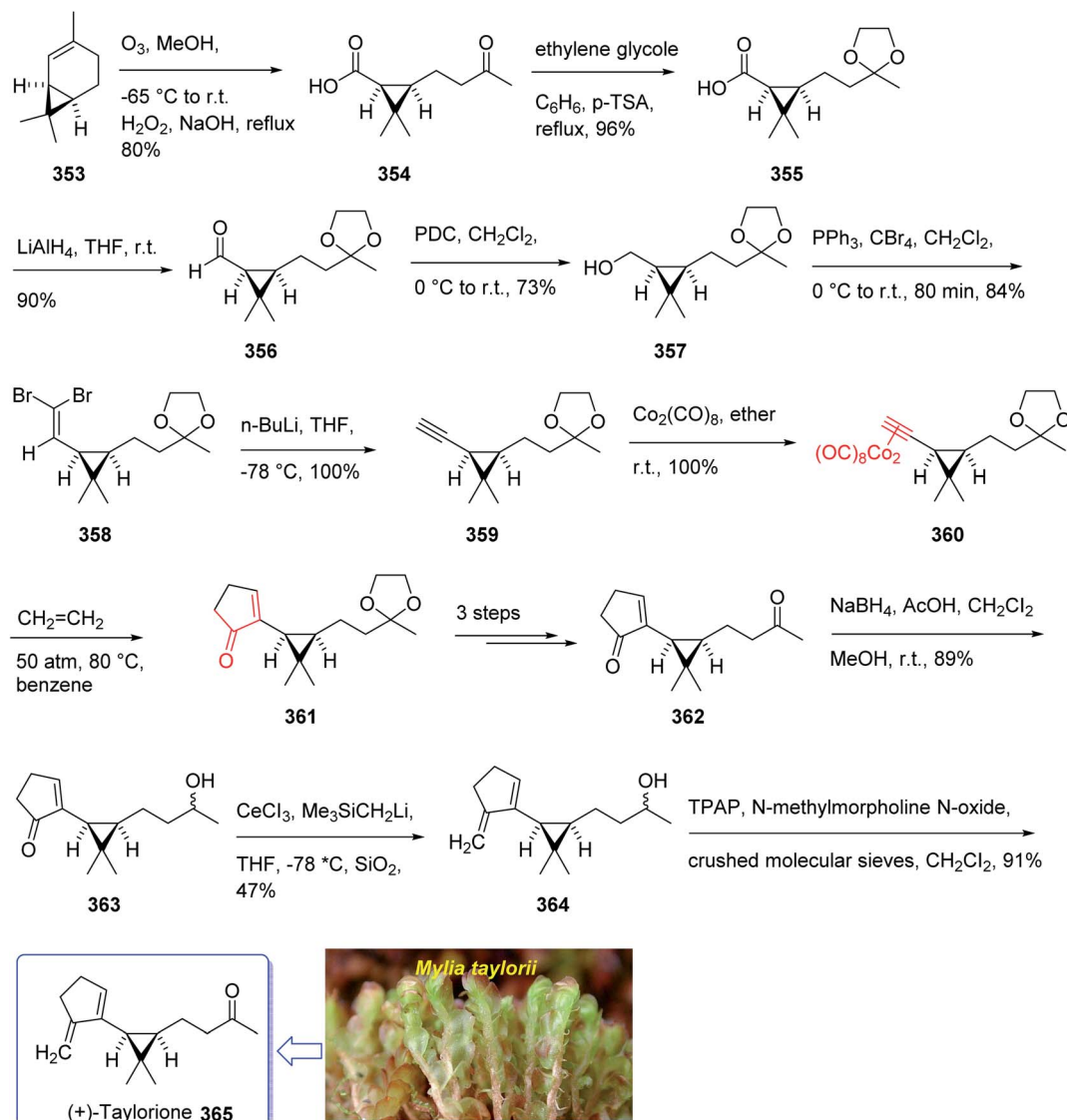
(+)-Taylorione **365** is the pure enantiomer of the principal sesquiterpene isolated from extract of the common leafy liverwort *Mylia taylorii*²³⁹ found on the northern hemisphere. The structure of (+)-taylorione **365** was elucidated and its absolute configuration was determined by combination of its spectral data analysis and degradation investigation.²⁴⁰

The total synthesis (+)-taylorione **365** was accomplished and reported by Kerr *et al.* in 1996 (ref. 241) using the Pauson-Khand annulation reaction as the key step. It began from already prepared (+)-2-carene **353** (ref. 241 and 242) which upon ozonolysis and subsequent oxidation with basified hydrogen peroxide, gave the keto acid **354** in 80% yield. The latter was converted to the ketal **351** upon the reaction with ethylene glycol in the presence of *p*-TSA in refluxing benzene. Since compound **355** found to be unstable to storage, it was instantly reduced to the corresponding alcohol **356** by treatment with LiAlH₄ in THF



Scheme 27 Total synthesis of hirsutene **352**.



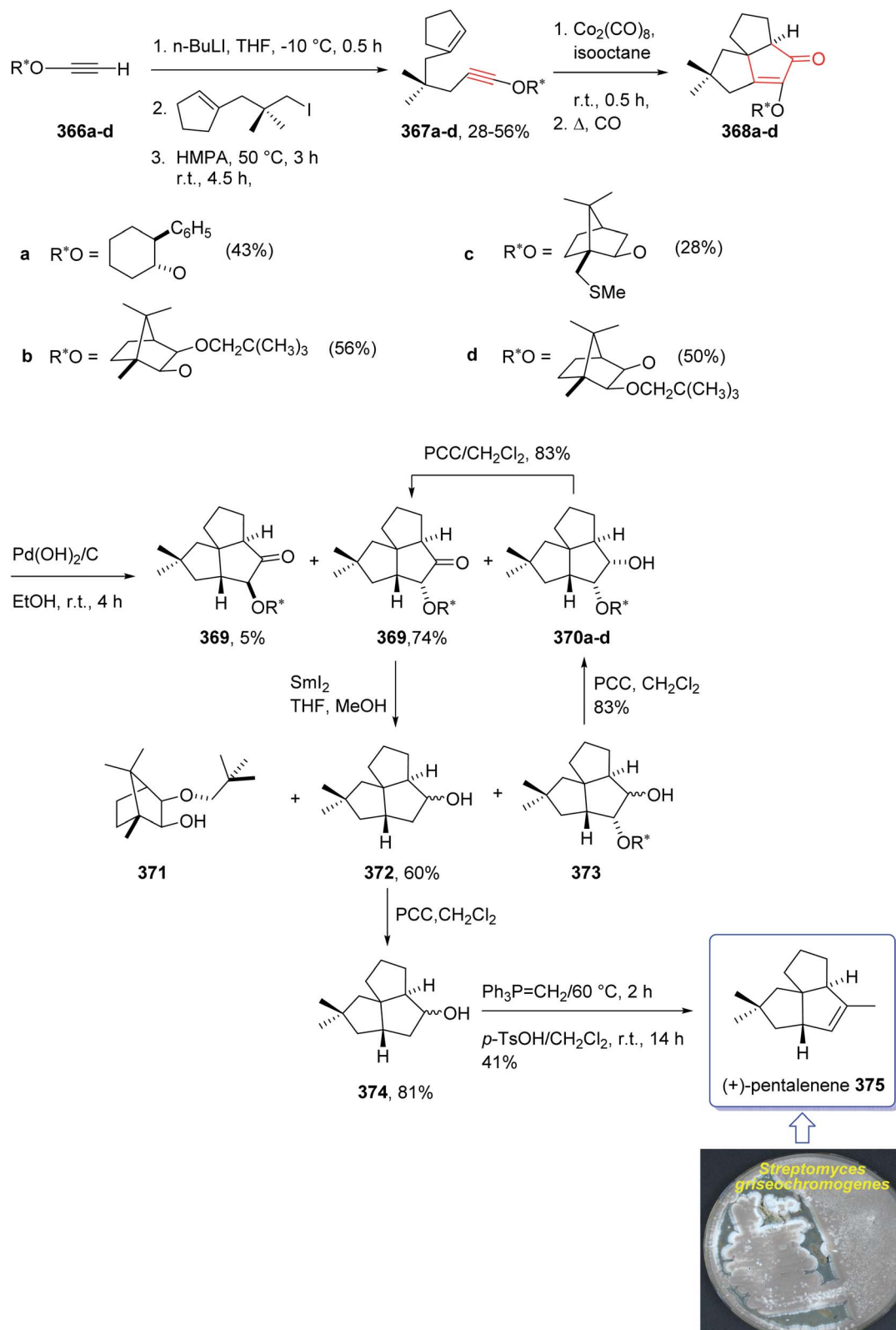


Scheme 28 Total synthesis of (+)-taylorione 365.

followed by oxidation with PDC to the corresponding aldehyde 357 in 73% yield. Having aldehyde 357 in hand, the synthesis of the essential alkyne complex 360 was contemplated. The latter was treated with CBr_4 in CH_2Cl_2 to obtain the desired dibromo olefin 358 in 84% yield in pure form after column chromatography. The dihalo alkene 358 was transformed to the pure terminal alkyne 359 rapidly upon treatment with $n\text{-BuLi}$ followed by aqueous work up. Then, alkyne 359 was reacted with octacarbonyldicobalt at ambient temperature to afford the dicobalt complex 360 in almost quantitative yield. The alkyne complex 360 was then underwent PK annulation reaction (C_2H_4 , 50 atmospheres, 80°C , benzene) to afford the desired cyclopentenone 361 in 38%. Pleasantly, upon treatment of cyclopentenone 361 with PPh_3 and CCl_4 in CH_2Cl_2 at 0°C to ambient temperature, the required diketone 362 was provided in an excellent yield. Having 362 available in hands, the remaining steps in the synthesis proceeded with no complication. The latter was subjected to selective carbonyl reduction under Ward

conditions²⁴³ (NaBH_4 in $\text{CH}_2\text{Cl}_2/\text{AcOH}/\text{MeOH}$) to provide the hydroxy ketones 363. After conversion of the latter to the diene 364 in moderate yield, to obtain optimum oxidation, it was reacted with Griffith–Ley tetrapropylammonium perruthenate (TPAP) agent to obtain the desired natural product 365 in 91% yield. In conclusion, the total synthesis of enantiopure (+)-taylorione 365 was accomplished starting from readily accessible chiral pool reagent (+)-2-carene 353, in a brief manner (in twelve steps) in a good overall yield of 12% (Scheme 28).²⁴¹

(+)-Pentalenene 375 (1*R*,3*aS*,5*aS*,8*aR*)-1,2,3,3*a*,5*a*,6,7,8-octahydro-1,4,7,7-tetramethyl cyclopenta[*c*]pentalene, was initially isolated by Seto and co-workers in 1980 (ref. 244) from the extract of *Streptomyces griseochromogenes*. Its structure elucidation revealed it has angularly fused triquinanes.^{245,246} This tricyclic sesquiterpene 375 is involved in the biosynthesis of neopentalenolactone antibiotic.



Scheme 29 Total synthesis of (+)-pentalenene 375.

Its total synthesis has been benchmark of various strategies involving the regio- and stereo-selective assembly of cyclopentanoid systems.^{246,276}

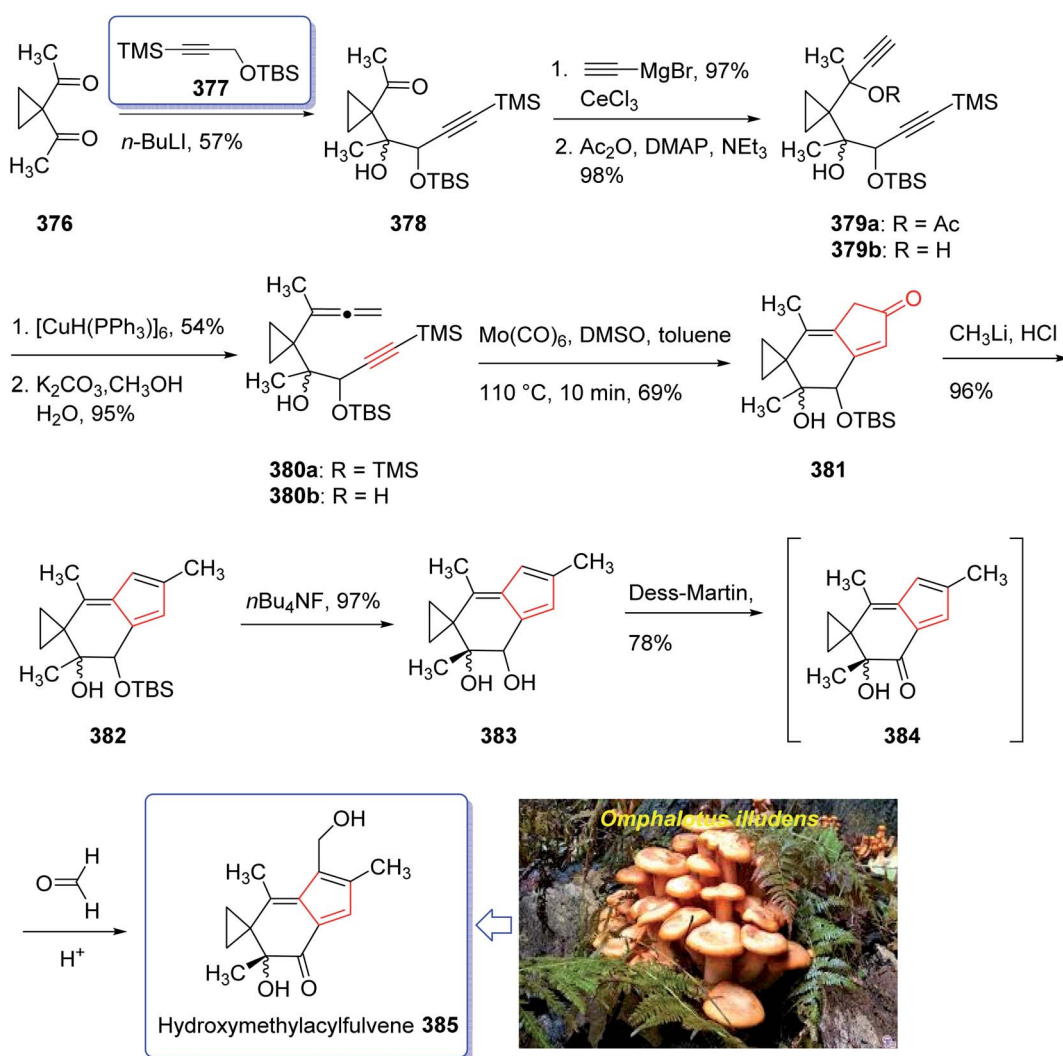
In 1997, Moyano and co-workers achieved and reported an effective total synthesis of 375 using Pauson–Khand reaction as key step.²⁴⁷ Accordingly, the total synthesis started from lithium

acetylide derived from (\pm)-(trans-2-phenylcyclohexyloxy)ethyne **366a**, created in dry tetrahydrofuran as solvent to furnish in enyne **367a** in moderate yield. The latter was subjected to a diastereoselective intramolecular Pauson–Khand reaction, following the procedure, already reported by Schore *et al.* who have synthesized the triquinane system of racemic **375** (ref. 245) in this way, the enyne **367** underwent a diastereoselective cyclization to provide the key intermediate **368**. Noticeably, it became evident that for such diastereoselective Pauson–Khand reaction, the best choice of chiral auxiliary is 3-(neopentyloxy) isoborneol. After several steps involving various functional group transformations, compound **368** converted to ketone **374** as illustrated in (Scheme 29). The configuration of **374** was determined from absolute configuration of **368c**, **369**, **370**, and **372** thus, had been established, unambiguously.

In the final step of this total synthesis, ketone **374** was transformed into the desired natural product, (+)-pentalene **375**, via Wittig olefination and acid-promoted isomerization of the exocyclic double bond (Scheme 29).²⁴⁸

Sesquiterpene, illudin S, exhibited high antitumor activity.²⁴⁹ Later, illudin analogues were synthesized showing highly improved effectiveness in comparison with the parent compounds.²⁵⁰ One of such analogues is hydroxymethylacetylfulvene **381** (HMAF, also called MGI 114). Since it was found to be active against breast, lung, and colon tumors, it has attracted much attention of synthetic organic chemists while showing intensely abridged toxicity. In addition, HMAF²⁵¹ as also found to be potent towards the MDR phenotype.²⁵²

The hydroxymethylacetylfulvene **385** can be provided semi-synthetically from the naturally occurring sesquiterpene illudin S. Illudin S is generated in cultures of *Omphalotus illudens* (Jack o'-lantern mushroom). Upon treatment of illudin S with formaldehyde in 1 N H₂SO₄ solution, HMAF can be obtained through a reverse Prins reaction to give the intermediate acylfulvene **384** which next can be subjected to an ene reaction with formaldehyde.²⁵³ The first total synthesis of HMAF was achieved and reported by McMorris *et al.* in 1997 (ref. 254) comprising a Padwa kind carbonyl ylide 1,3-dipolar cycloaddition²⁵⁵ to achieve the illudin scaffold.



Scheme 30 Total synthesis of hydroxymethylacetylfulvene (HMAF) **385**.

A brief synthetic approach involving eleven steps of HMAF was accomplished and reported by Brummond and co-workers in 1999,²⁵⁶ including an intramolecular [2 + 2 + 1] Pauson–Khand reaction. In this strategy, the easily accessible 1,1-

diacetylcyclopropane **376**,²⁵⁶ reacted with the lithio derivative of the *tert*-butyl-dimethylsilyl ether of 3-trimethylsilylpropyn-1-ol **377** to give the expected ketone **378** as a 1.3 : 1 mixture of diastereomers in good yield. After several steps, the latter was

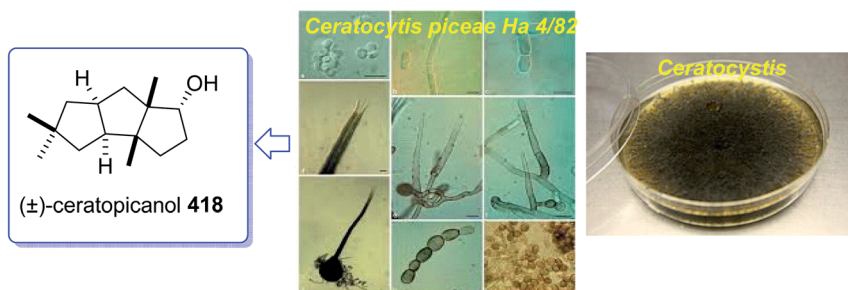
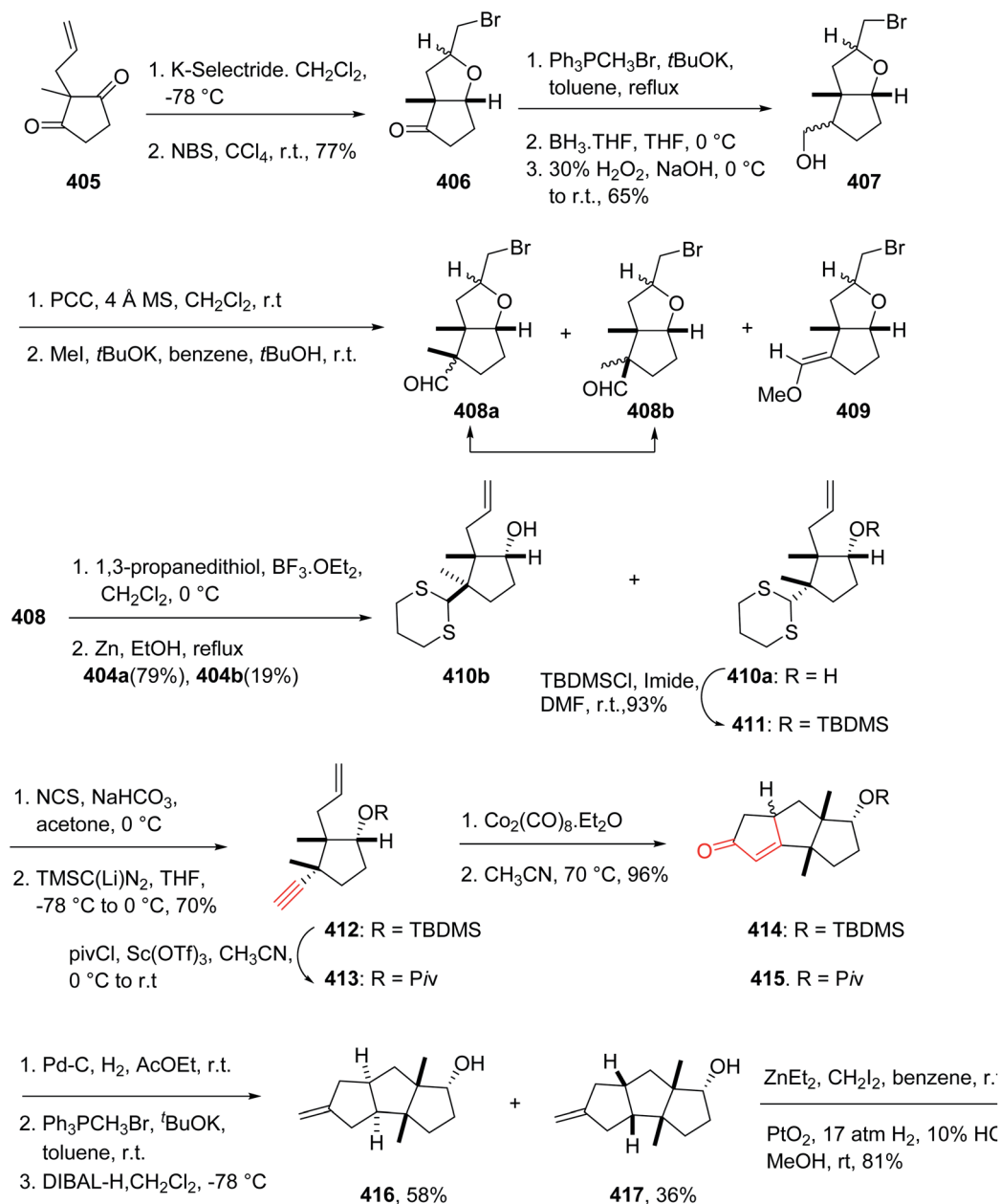


Scheme 31 Total synthesis of asteriscanolide **404**.



transformed into alkynyl allene **380b** which was subjected to fast cycloaddition under the conventional PKR conditions $[\text{Mo}(\text{CO})_6]$, DMSO, toluene, 110°C ^{257,258} to afford the 4-

alkylidene cyclopentenone **381** as single product in good yield. The latter was transformed into the secondary alcohol **383** after several steps which was oxidized using Dess–Martin oxidative



Scheme 32 Total synthesis of (+)-ceratopicanol **418**.



reagent to the corresponding ketone, acylfulvene **384**, in good yield. The latter was then reacted with formaldehyde in the presence of H_2SO_4 in acetone/water to give HMAF **385** in good yield (Scheme 30).²⁵⁶

Recently, asteriscanolide **404** which is a cyclooctane sesquiterpene lactone has attracted much attention of organic synthetic chemists. It was initially isolated in 1985 by the Feliciano research group from the hexane extract of *Asteriscus aquaticus*.²⁵⁹ Several successful attempts have been reported. Wender and co-workers,²⁶⁰ the Paquette group²⁶¹ and very recently the Snapper research group²⁶² have reported the total synthesis of asteriscanolide **404**.

In 2001, Krafft *et al.* reported an intermolecular Pauson-Khand cycloaddition and a ring-closing metathesis as vital steps.²⁶³ This strategy incorporates the cyclooctane chiral center prior to ring construction. Remarkably, the ring-closing metathesis creates a new eight-membered ring with an “in-out” intrabridgehead relationship which come across the principles as mentioned above.

The total synthesis started from the protected 3-butyln-1-ol **386** as the corresponding *tert*-butyldimethylsilyl (TBS) ether which was converted into the corresponding alkynoate **387** *via* treatment with *n*-BuLi in THF at -78°C to create the lithio alkyne and then added to ethyl chloroformate in THF at -78°C . Treatment of **387** with dicobalt octacarbonyl in petroleum ether gave the desired precursor for PKR, hexacarbonyl-dicobalt complexed alkyne **388**. The latter was then reacted with

propene in methylene chloride followed by incremental addition of *N*-methyl-morpholine-*N*-oxide monohydrate under PKR conditions. This afforded the highly functionalized cyclopentenone **389** which comprises different functional groups suitably placed for further employing of the side chains. After several steps, the latter was converted into trisubstituted cyclooctene **403**. The latter was a key intermediate for the successful total synthesis of desired natural product, asteriscanolide A **404**, after several steps (Scheme 31).²⁶³

Optically active (+)-ceratopicanol **418**, as a novel triquinane sesquiterpene was initially isolated from the extract of fungus *Ceratocystis piceae* Ha 4/82 (ref. 264) by Hanssen and co-workers in 1998.²⁶⁴ Its structure was elucidated and its relative stereochemistry was determined as $(1R^*, 2S^*, 6S^*, 8S^*, 9R^*)$ -1,4,4,8-tetramethyltricyclo[6.3.0.0^{2,6}]-undecan-9-ol.²⁶⁴ The absolute configuration of (+)-ceratopicanol **418** was determined unambiguously when the total synthesis of its unnatural (–)-stereoisomer was completed and its X-ray analysis was compared to the (+)-natural product **418**.²⁶⁵ Ceratopicanol **418** has a fascinating and exceptional structural characteristic containing five stereogenic centers among which are two adjoining bridgehead quaternary carbon centers. Due to these exceptional features, ceratopicanol **419** was selected as a target to be synthesized by several research groups.^{265,266} In 2002, Mukai *et al.*²⁶⁷ studied the stereoselective reduction of the 1,3-dicarbonyl **405** resulting in the formation of a compound containing hydroxyl and allyl moieties with *cis*-relationship. They found out when compound



Scheme 33 Total synthesis of α -cedrene **428a** and β -cedrene **428b**.



405 is treated with *K*-selectride, the highest stereoselectivity was obtained. To remove the undesired *trans*-hydroxy compound, a mixture obtained from reduction step was subsequently treated with *N*-bromosuccinimide (NBS) in CCl_4 to provide a mixture containing two stereoisomers of the 2-oxabicyclo [3.3.0]octa-6-one derivative **406**, because of the presence of C_3 stereogenic center, in satisfactory overall yield. After the next steps including different functional group transformations such as treatment of **408** with 1,3-propanedithiol in the presence of $\text{BF}_3 \cdot \text{OEt}_2$ which provided separable **410a** (75%) and **410b** (19%), conversion of secondary hydroxyl group of **410a** to **411** in 93% yield, followed by treatment with lithio-trimethylsilyldiazomethane, the alkyne derivative **412** was obtained as key intermediate. In accordance with the classical Pauson–Khand reaction procedure, compound **412** was treated with $[\text{Co}_2(\text{CO})_8]$ in Et_2O to give the corresponding alkyne–cobalt complex. This complex upon heating at 70°C in CH_3CN ²⁶⁸ gave compound **414** in high yield. Then, to a pivaloyl group was introduced on the secondary hydroxyl group of **412** to produce **413**. After treatment with $\text{Co}_2(\text{CO})_8$, the latter was transformed to the corresponding cobalt-complex **415**. Elimination of the pivaloyl group resulted in the isolation of the desired compound **416** together with **415**. Finally, compound **416** in pure form, treated with diethyl zinc and diiodomethane in benzene under the Simmons–Smith reaction to obtain the cyclopropane derivative which upon hydrogenation in the presence of PtO_2 under pressure gave the desired natural product (\pm)-ceratopicanol **418** in 81% yield (Scheme 32).²⁶⁷

The tricyclic sesquiterpenes α -cedrene **428a** and β -cedrene **428b** were initially isolated by Barrero *et al.*,²⁶⁹ in 1996 from *Juniperus cedrus* and *Juniperus thurifera*.^{269–271} With these natural products, α -cedrene **428a** and β -cedrene **428b**, a range of accurately relative oxygenated terpenoid analogues were also isolated from the same source. Due to their fascinating [5.3.1.0^{1,5}] tricyclic structure, the cedrene family and their relative naturally occurring compounds have attracted much attention of synthetic organic chemists over the years since initial characterization of α -cedrene **428a** and β -cedrene **428b** in 1953 by Stork research group.²⁷²

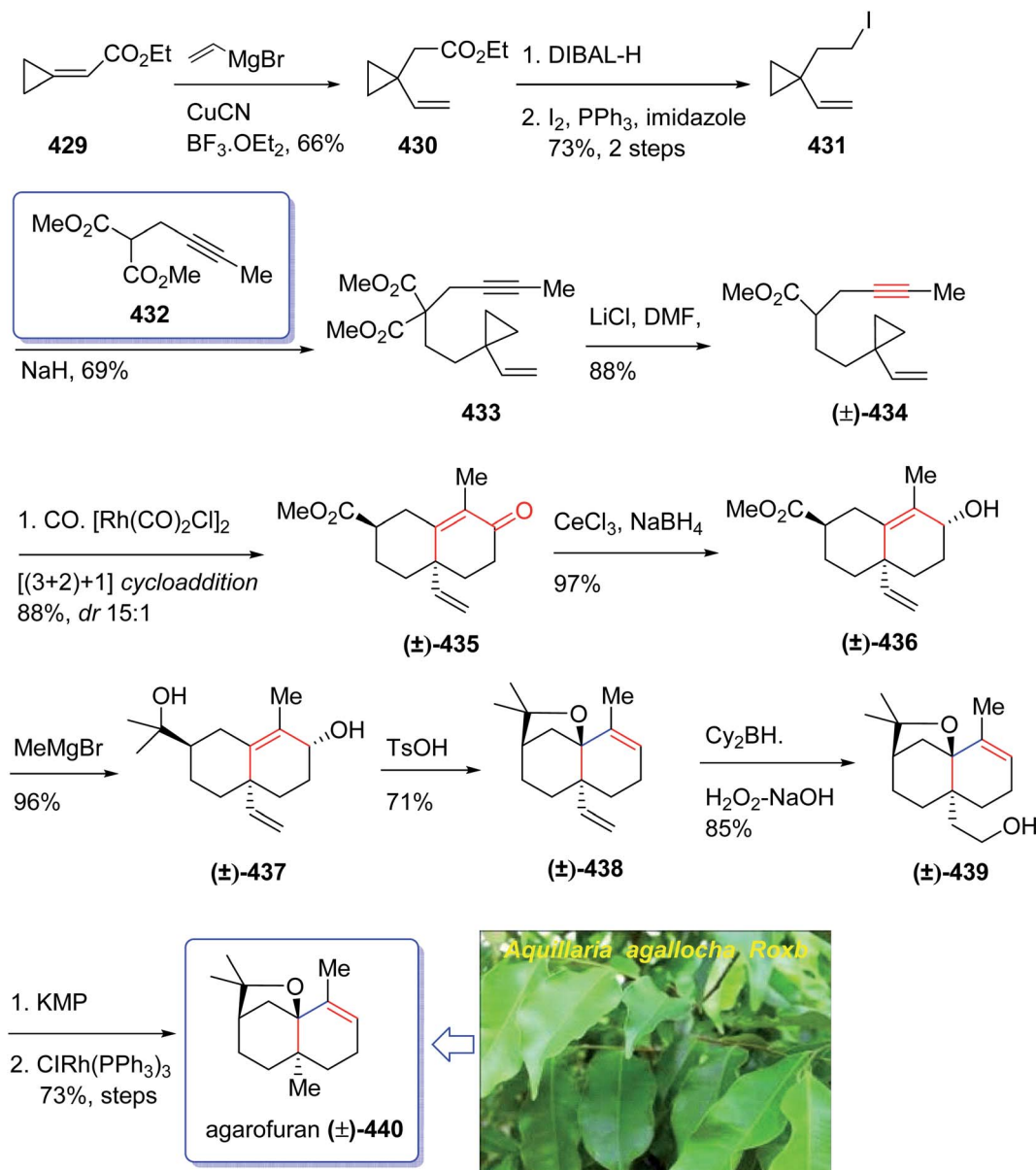
In 2006, Kerr and co-workers²⁷³ developed a pathway involving an intramolecular Pauson–Khand reaction in which the cedrene carbon scaffold was effectively installed from a simple monocyclic precursor directly.

A small number of further synthetic manipulations provided a concise formal total synthesis of α - and β -cedrene. The cyclisation precursor was readily prepared with a stereoselective ketone alkenylation selectively providing the olefin required for efficient access to the natural target. Using a simple monocyclic precursor led to the direct and highly effective formation of the interesting tricyclic [5.3.1.0^{1,5}] carbon scaffold of α -cedrene **428a** (and β -cedrene **428b**). It is noteworthy that in the crucial key Pauson–Khand annulation step, the olefinic precursors reacted with retention of configuration to afford the expected cyclopentenone epimer for synthesis of the desired natural products in excess. For further magnifying the effectiveness of this total synthetic design, the remaining and undesired cyclopentenone was also transformed into the vital isomer by

a simple base-prompted epimerization course. In sequence, the desired cyclopentenone intermediate was further expanded to cedrone **427**, therefore, establishing a formal total synthesis of α - and β -cedrene. Ultimately, it is worthwhile to notice that compound **425a** underneath a reaction sequence comparable to those conducted on **425b** resulting in the synthesis of *epi*-cedrone.

As a matter of fact, this synthetic pathway is commenced with the introduction of α,β -unsaturation into the market purchasable cyclohexanedione monoethylene acetal **419**. After three steps, compound **419** was converted into **420** *via* Saegusa oxidation. The latter was then subjected to standard Wittig reaction at 0°C employing the ethyltriphenylphosphonium bromide salt and *n*-BuLi to give olefins **425a/425b** in 92% yield as a 2 : 1 mixture of geometric isomers. The compound **421a/421b** was transformed into aldehydes **422ab** in 97% yield by oxidation after two steps.²⁷⁴ For the transformation of aldehydes **422ab** into alkynes,^{221,275} the Ohira–Bestmann technique (with the reagent dimethyl acetyldiazomethyl-phosphonate) was applied. In this case, this mild strategy gave alkynes **423a/423b** in 81% yield. In sequence, these were habitually complexed with octacarbonyldicobalt to provide the stable cyclisation precursors **424a/424b** in a virtually quantitative yield. At this vital step and with the essential complexes in hand, the Pauson–Khand annulation for the installation of the required tricyclic carbon α -cedrene scaffold was examined. Delightfully, intramolecular Pauson–Khand cyclisation of **424a/424b** proceeded smoothly to afford the enones **425a/425b** in high yield as a mixture of stereoisomers in the ratio of 2 : 1. This indicated that relative stereochemistry present in the initial olefins **421a/421b** had been carried *via* the cyclisation of precursors **424a/424b**. Using an efficient and selective approach to **425b**, an essential deoxygenated product **426** was provided after four steps. Upon treatment of **426** with $\text{Ph}_3\text{P/CBr}_4$, desired cedrone **428** was obtained in 99% yield.²⁷⁶ Then, the latter in two steps was converted to desired natural product α - and β -cedrene, **428a** and **428b** (Scheme 33).²⁷³

(\pm)- α -Agarofuran **440** is a racemic furanoid sesquiterpene natural product. In 1962, Bhattacharyya *et al.*,²⁷⁷ isolated α - and β -agarofuran from extract of agarwood C (*Aquilaria agallocha Roxb.*).²⁷⁸ Its structure was elucidated by chemical degradation and spectroscopic data analysis, chemical examination of fungus infected agarwood C (*Aquilaria agallocha Roxb.*).^{279,280} An approach to the total synthesis of α -agarofuran including Pauson–Khand reaction as a key step was presented by Yu *et al.* in 2010.²⁸¹ This strategy started from cyclopropylidene ester **429**,²⁸² which in two steps using conventional organic reactions, was converted to vinyl cyclopropane iodide **431**. Next, the latter was coupled with propargyl diester **446** to afford the *gem*-diester-tethered 1-yne-VCP **433**. Then, the latter was subjected to Krapcho decarboxylation²⁸³ to provide monoester-substituted 1-yne-VCP **434**. This compound can act as a suitable precursor for homo-Pauson–Khand cycloaddition reaction under already secured optimal reaction conditions to give bicyclic cyclohexenone **435** in 86% isolated yield with a good diastereoselectivity. Compound **438** having the tetrahydrofuran framework of agaroguran,²⁸⁴ was synthesized from **435** after



Scheme 34 Total synthesis of (±)-α-agarofuran 440.

several steps. Ultimately, the vinyl unit was converted respectively to its corresponding alcohol and then angular methyl group by a sequential hydroboration–oxidation–decarbonylation to afford the desired natural product as a racemate (±)-α-agarofuran 440 (Scheme 34).^{280,281}

Kitanaka²⁸⁵ and co-workers in 2005 for the first time isolated a new guaiane-type sesquiterpene, (+)-indicanone 454, from the extract root of *Wikstroemia indica* (Thymelaeaceae), collected from the southeast China. Notably, in this isolation, two already known and reported biflavonoids (*i.e.* sikokianin B and sikokianin C) were also obtained. *Wikstroemia indica* has been used to treat pneumonia, rheumatism, and bronchitis as folk medicine in China for long time. In 2012, the total synthesis of (+)-indicanone 454 was reported by Mukai and co-workers²⁸⁶

through the Rh(I)-catalyzed Pauson–Khand reaction of the allenyne derivative which was derived from (+)-limonene.

In this strategy, the total synthesis of (+)-indicanone 441 commenced with vinyl phosphate 442,²⁸⁷ prepared from commercially available (+)-limonene after four steps. Compound 442 after several steps involving different common functional group transformations as well as protection–deprotection of some functional groups was converted to the allenyl alcohol of 451 which upon protection by a silyl group gave compound 452 as an appropriate precursor for PKR. The latter was subjected to PKR, upon treatment with 5 mol% [RhCl(CO) dppp]₂ in toluene under reflux conditions and carbon monoxide (1 atm) supplied compound 453 in moderate yield. The latter was then desilylated using aqueous HCl gave the desired natural product (+)-indicanone 454. This strategy to the first

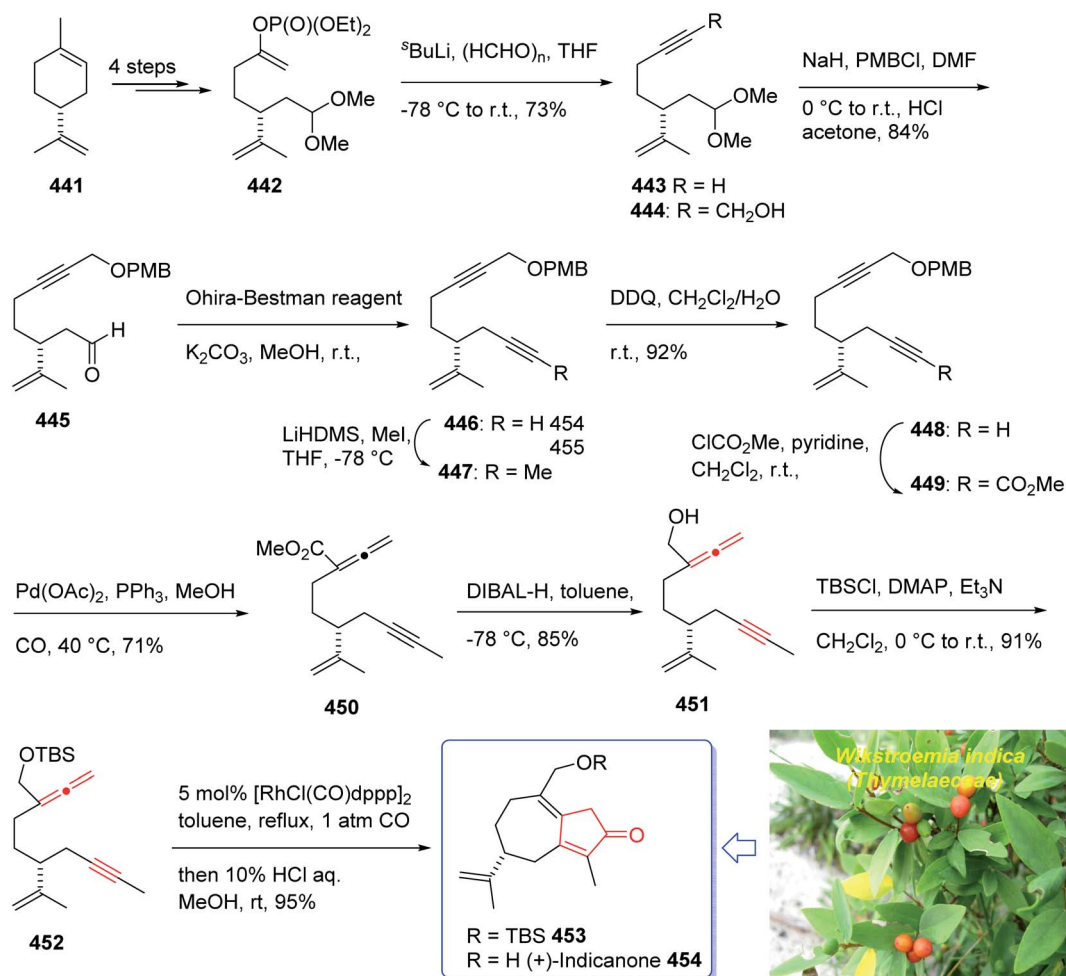


total synthesis of (+)-indicanone **454** was completed in ten steps starting from easily accessible known phosphate **442** in 29% overall yield. In addition, this total synthesis confirmed the complete structure and absolute stereochemistry of (+)-indicanone **454**, unambiguously (Scheme 35).²⁸⁸

In 2000, Fukuyama and his research group isolated merrilactone A **469**, a complex cage-shaped pentacyclic sesquiterpene, from extract of *Illicium merrillianum*.²⁸⁹ Its structure was fully characterized showing it has an oxetane moiety, two γ -lactone functionalities, and a highly substituted cyclopentane ring at its core. In addition, merrilactone A **469** contains seven adjoining stereogenic centers, involving five quaternary ones. From the biological point of view, merrilactone A **469** is a non-peptidal neurotrophic factor which stimulated neurite development in the values of fetal rat cortical neurons.²⁸⁹ Total synthesis of merrilactone A **469** has received much attention of organic synthetic chemists due to its exceptional and interesting structure as well as the reported activity against neurodegenerative diseases.²⁹⁰ Among them, Inoue and Hirama,^{291–293} Mehta,²⁹⁴ Frontier,²⁹⁵ Greaney²⁹⁶ and almost twenty years ago Danishefsky^{290,297} research groups have reported the total syntheses of merrilactone A **469**. Other synthetic approaches

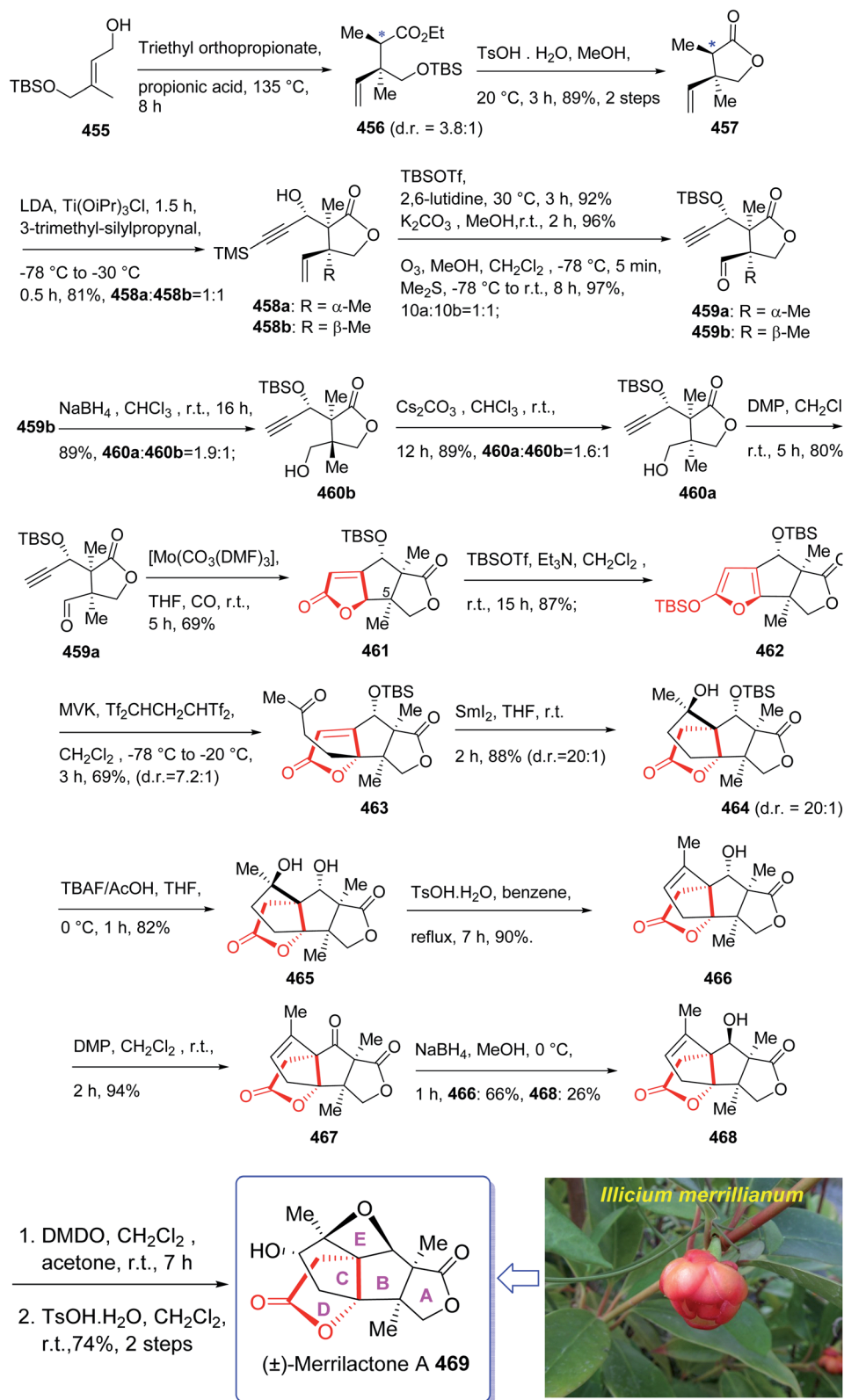
have also been renowned for the total synthesis of related natural products to merrilactone A **469**.^{298–301} Significantly, in 2012 Zhai and co-workers³⁰² achieved and reported a new and proficient strategy to the total synthesis of (\pm)-**469** involving the Pauson–Khand reaction (PKR)⁵⁸ and hetero-Pauson–Khand reaction (h-PKR) in 2012.^{303,304} An effective total synthesis of (\pm)-merrilactone A was developed by recognition of an efficient installation of (+)-mintlactone through an intramolecular ynal h-PKR.⁷⁰ It was commenced from the already reported alcohol **455**,³⁰⁵ which upon treatment with triethyl orthopropionate and propionic acid provided the Johnson–Claisen rearrangement^{306,307} product **456** (dr = 3.8 : 1) that upon desilylation and lactonization produced compound **457** (dr = 2.9 : 1), 89% over two steps from **455** mediated by TsOH·H₂O. Treatment of compound **457** with LDA, Ti(OiPr)₃Cl, and then 3-trimethylsilylpropynal resulted in 1 : 1 mixture of inseparable aldols **458a** and **458b** in 81% combined yield alongside with two other inseparable isomers in 9% overall yield.³⁰⁸

Alcohols **458a** and **458b** were transformed smoothly into the two separable ynals **459a** and **459b** (1 : 1). Upon sequential hydroxyl protection, alkyne desilylation,²⁵⁶ and selective ozonolysis.³⁰⁹ Conversion of **459b** into **459a** was achieved *via*



Scheme 35 Total synthesis of (+)-indicanone **454**.





Scheme 36 Total synthesis of merrilactone A 469.

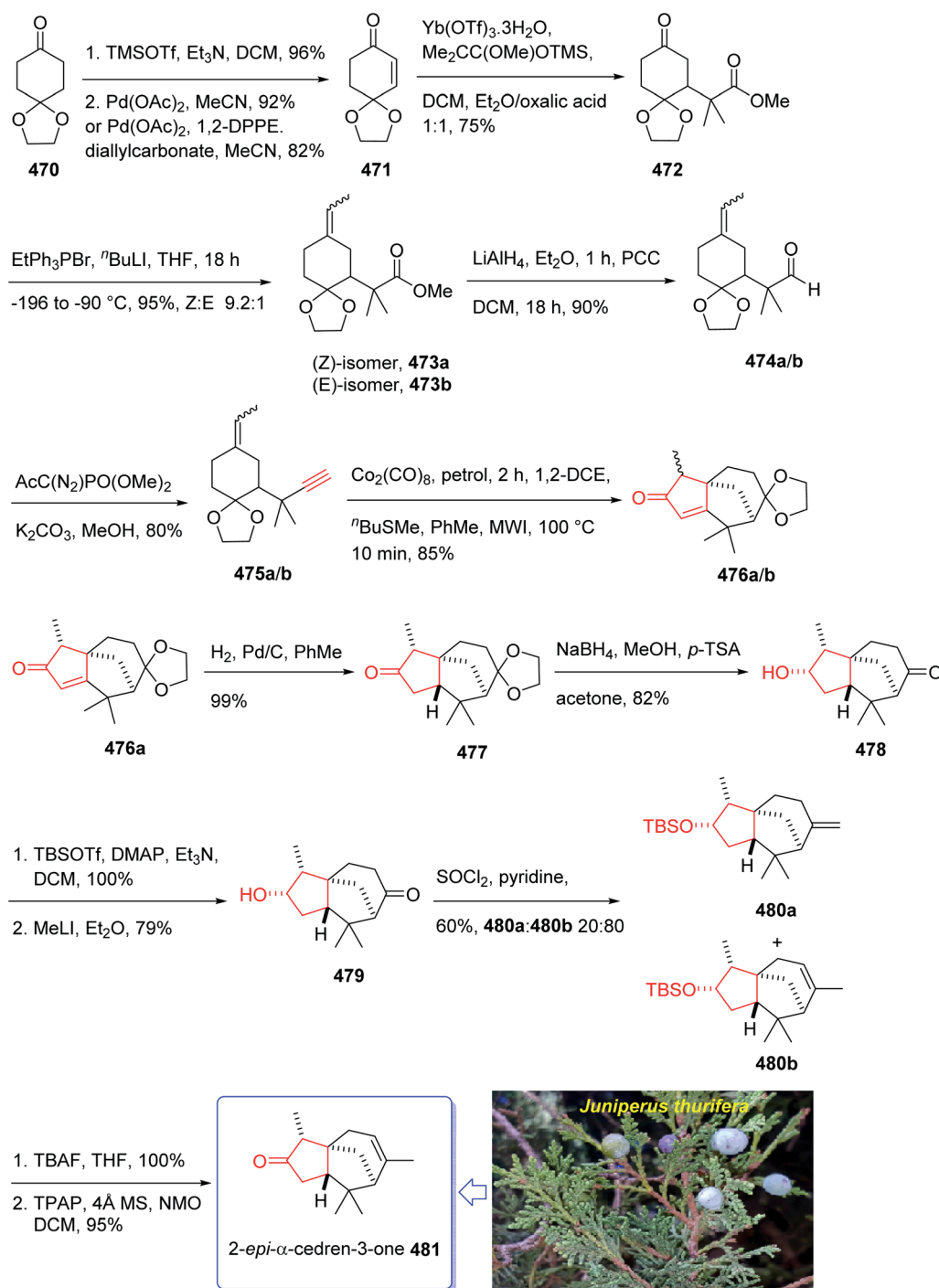
reduction of **459b** using NaBH_4 and intramolecular transesterification. Pleasantly, **460a** was provided from **460b** by initial transformation into a mixture of **460a** and **460b** (1.6 : 1)

in the presence of Cs_2CO_3 via intramolecular transesterification. Having ynal **461a** available in hand, the vital h-PKR was performed to obtain the B and D rings of the target



468. The latter was treated with $\text{Mo}(\text{CO})_3(\text{DMF})_3$ (ref. 70 and 304) in THF at room temperature and under argon atmosphere provided tricycle **461** in 69% yield. Worthy to mention that **459a** was not used up totally by replacing argon with carbon monoxide (CO). Pleasingly, when **459a** was exposed to $[\text{Mo}(\text{CO})_3(\text{DMF})_3]$ in THF at room temperature firstly under argon atmosphere for a while and then under CO atmosphere (balloon), compound **461** was obtained (69%). Then, α,β -unsaturated lactone **461** was transformed into silyloxyfuran

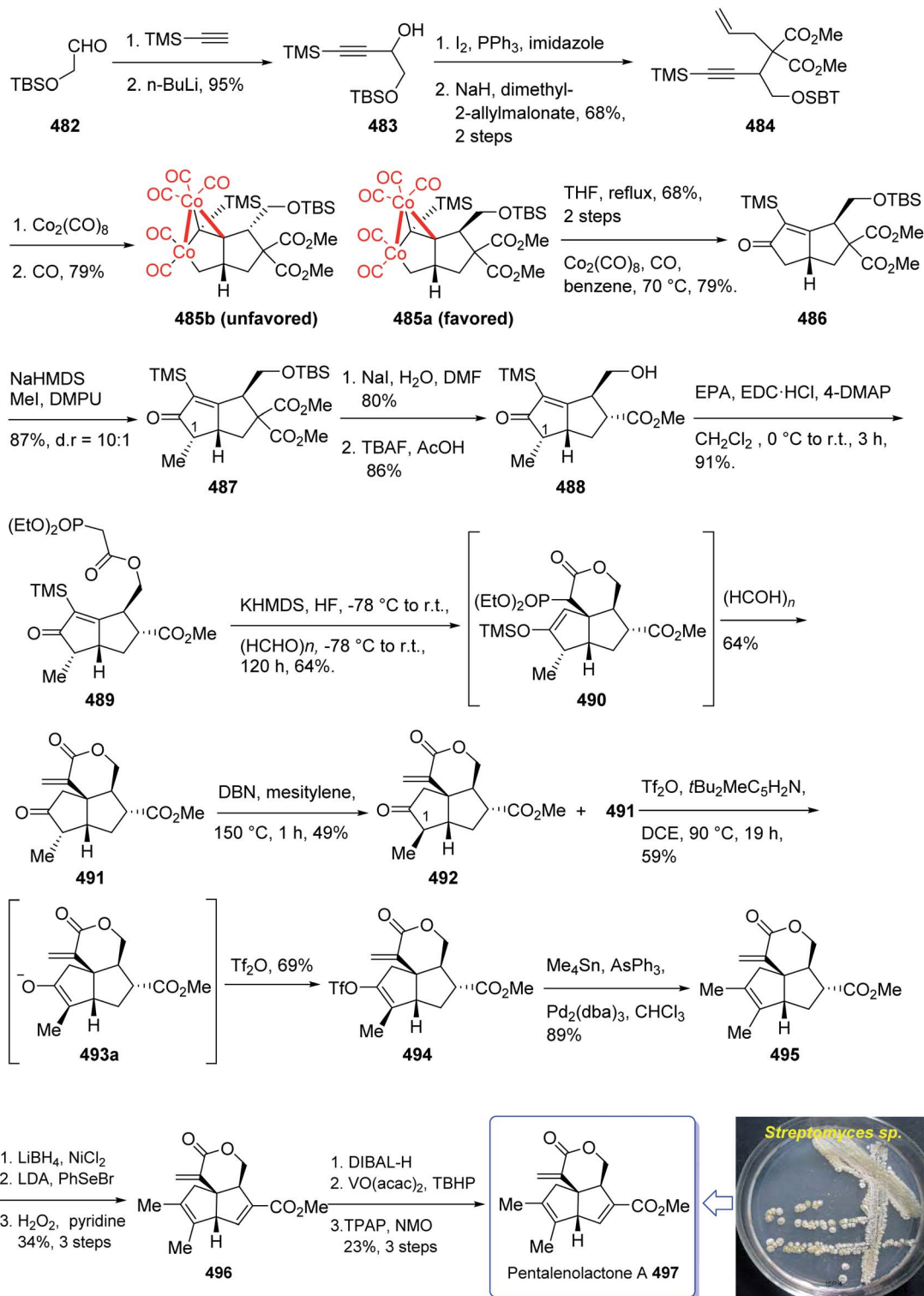
462,³¹⁰ upon treatment with MVK mediated by $\text{Ti}_2\text{CHCH}_2\text{CHTF}_2$ (ref. 311) under the Taguchi's strategy^{311,312} to give ketone **3** (61%) alongside with *epi*-**3** (8%) *via* a vinylogous Mukaiyama/Michael addition reaction. Expectedly, the C ring could be constructed *via* a reductive carbonyl-alkene coupling reaction.^{313,314} Compound **463** was expectedly cyclized to furnish the desired tetracycle **464** (88%) as basically a simple diastereo-isomer (*dr* = 20 : 1) upon treatment with SmI_2 in THF. Reaction of the latter with $\text{TsOH} \cdot \text{H}_2\text{O}$ in refluxing benzene with



Scheme 37 Total synthesis of 2-*epi*- α -cedren-3-one **481**.

concurrent dehydration and desilylation resulted in the formation of the trisubstituted alkene **465** in 91% yield. On contrary, treatment of **464** with TBAF and AcOH instead of desilylation,²⁹⁴ gave diol **465** in 82% yield. Compound **465** upon dehydration in the presence of TsOH·H₂O in refluxing benzene gave compound **466**.²⁹⁵ Precisely, compound **466** by oxidation

with DMP gave ketone **467** (94%) which after reduction with NaBH₄ furnished an easily separable mixture of **468** (26%) and **466** (66%). This procedure was repeated many times to collect adequate quantities of alcohol **468**. Ultimately, the latter was converted into the desired natural product, merrilactone, following a recognized procedure involving a stereoselective



Scheme 38 Total synthesis of pentalenolactone A 497.

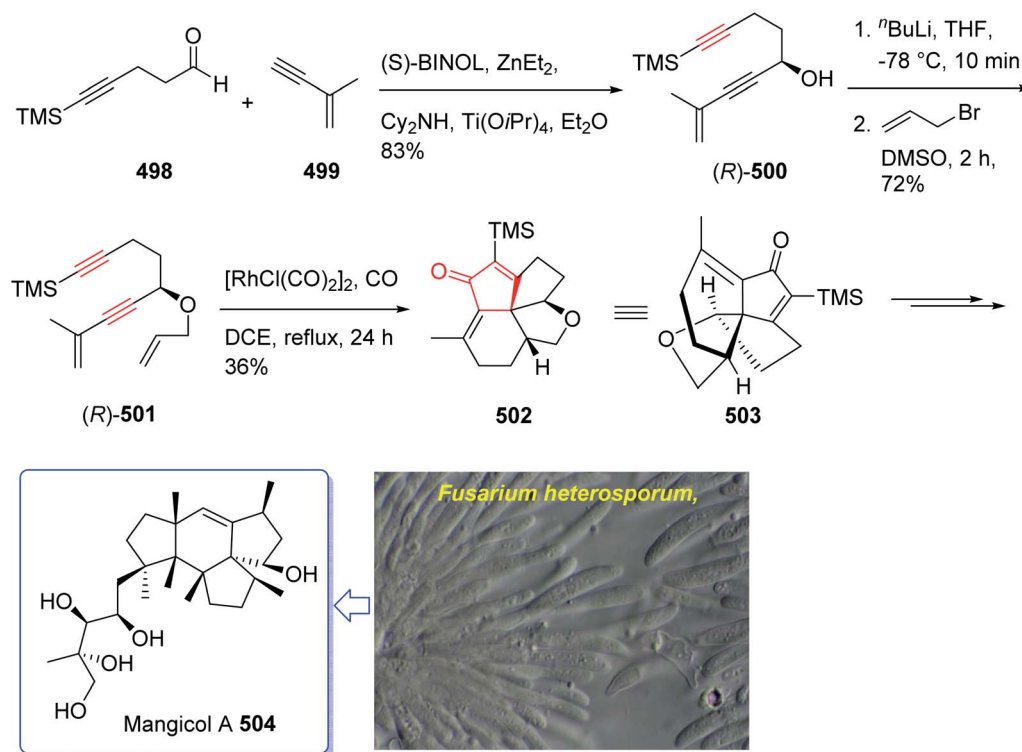


epoxidation and epoxide ring opening/oxetane generation (by homo-Payne rearrangement).^{291,315} The spectroscopic data of synthesized (\pm)-merrilactone A **469** was identical to those which already have reported for the natural product in the chemical literature.^{289,290,294,295} An efficient total synthesis of (\pm)-merrilactone A **469** has been reported in fifteen reaction steps for the shortest sequence from **7** which is a known compound (Scheme 36).^{302,305}

A sesquiterpene, 2-*epi*- α -cedren-3-one **481** is a natural compound, isolated from the essential oil of *Juniperus thurifera* in 2000 by Barrero and co-workers.²⁶⁹ Several efforts have been made to achieve its total synthesis due to its interesting molecular structure. A synthetic strategy was reported by Kerr and co-workers in 2018.³¹⁶ involving a highly (*Z*)-selective Wittig olefination reaction. This reaction was performed at very low temperature which was essential to obtain of the same configuration within the desired natural product **481** and catalyzed intramolecular Pauson–Khand cyclisation reaction being conducted under MWI for the construction of the required tricyclic core of compound **481**. This synthetic pathway started with market purchasable cyclohexanedione monoethylene acetal **470** which was converted to α,β -unsaturated ketone **471** *via* Pd-catalyzed Saegusa oxidation reaction.³¹⁷ The latter was then converted into ketoester **472** *via* a conjugate addition of the silyl ketene acetal of methyl isobutyrate, mediated by ytterbium(III) triflate trihydrate.³¹⁸ Conversion of the latter to obtain the required alkene moiety is needed for (*Z*)-selective olefination reaction to attain the relative stereochemistry, more unambiguously being aligned with the methyl group *syn* to the

methylene bridge in 2-*epi*- α -cedren-3-one, **481**. Wittig olefination reaction was performed at ambient temperature to give olefins **473a** and **473b** in excellent yield. At this point, compounds **473a** and **473b** were found to be practically inseparable. Therefore, aldehydes **474a/b** were provided *via* a two-step sequential reduction/oxidation. This mixture **474a/b** was treated with the Ohira–Bestmann reagent^{221,319} (dimethyl acetyldiazomethylphosphonate) to obtain the alkyne **475a/b** as suitable PKR precursors. Having enynes **475a/b** available in hand, the effectiveness of the PKR, for the construction of tricyclic skeleton as core should have been optimized. To the purpose, addition of *n*-butyl methyl sulfide, as a promoter of PKR initially recognized by Sugihara and Yamaguchi,³²⁰ along with employing sub-stoichiometric quantities of $\text{Co}_2(\text{CO})_8$ were successfully examined to obtain cyclopentenone **476a/b**. It is worthwhile mentioning, as proved and reported earlier,²⁷³ the ratio of inseparable stereoisomers obtained *via* the Wittig olefination relates always directly to the ratio of cyclopentenone **476a/b** provided in all PKRs reported previously. The mixture of **476a/b** was converted into **480a** and **480b** after several steps. This mixture **480a/480b** was separated and **480b** was subjected to sequential deprotection and oxidation afforded the desired natural product **481** in 95% overall yield over two final steps. In conclusion, the total synthesis of 2-*epi*- α -cedren-3-one has been accomplished in seventeen steps using PKR as a vital step (Scheme 37).³¹⁶

Pentalenolactone A **497** is a conspicuous member of naturally occurring antibiotics, generated by prokaryotic organisms.^{321–324} which was initially isolated and characterized as an



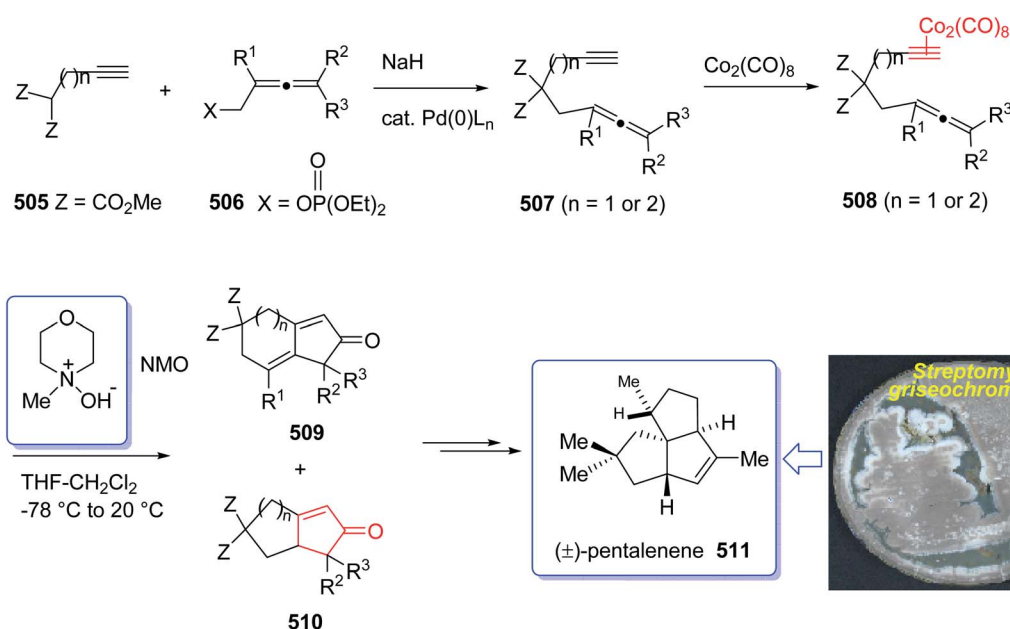
Scheme 39 Total synthesis of mangicol A **504**.

acidic lipophilic antibiotic. The sesquiterpene antibiotic pentalenolactone **497**, isolated from a variety of *Streptomyces* species, is a rare example of a cyclic terpenoid produced by a prokaryotic organism. Following the original isolation in 1957 by Koe and co-workers, McBride and co-workers,^{324,325} the structure and absolute configuration were eventually assigned in 1970 by combination of spectroscopic and X-ray crystallographic methods.^{326,327} The structure of pentalenolactone A **497** was determined by NMR analysis of its methyl ester derivative. The analysis showed that the structure consists of a highly compact carbon framework with a rare, angularly fused tricyclic pentanoid lactone having various oxidation states in each of the rings. It has a tricyclic core showing cytotoxic activity.^{328,329} Such compounds also show a broad spectrum of potencies towards bacteria, fungi, viruses, and tumors. Its stimulating properties and substantial chemical activity prompted Danishefsky *et al.* to develop a fascinating total synthesis of **497** in 1978.^{330–332}

In 2012, Li and co-workers³³² reported a protocol^{333,334} for the asymmetric total synthesis of methyl ester of pentalenolactone A **497** based on an intramolecular Pauson–Khand reaction. It began with aldehyde **493** which upon treatment with lithium ethynylate at $-78\text{ }^{\circ}\text{C}$ afforded propargyl alcohol **483** in 95% yield. The newly formed hydroxyl group in **483** was transformed into the respective iodo compound upon treatment with $\text{I}_2/\text{imidazole}$ mediated by Ph_3P and the resulting iodo species was next reacted with sodium dimethyl 2-allylmalonate to furnish the enyne **484** in 68% yield in two steps as an ideal precursor for the PKR. The latter was then converted to cyclopentenone **486** *via* PKR. The exclusive construction of **486** was most probably to minimize the steric interaction between the TMS and CH_2OTBS (TBS = *tert*-butyldimethylsilyl) moieties in complex **485a**^{53,247} relative to complex **485b**, which consequently resulted in the formation of the desired annulated product **486** as the sole

product. Having the latter available in hand, product **496** in an overall yield of 34% was converted to the desired natural product methyl ester of pentalenolactone A **497** after several steps following the already reported procedure by Danishefsky and co-workers in 1978 (Scheme 38).^{332,335}

In 2000, Fenical research group isolated Mangicol A **504** (ref. 336) from a marine fungal *Fusarium heterosporum*. They reported it as a novel type of sesterterpene polyols with an exceptional spirotricyclic scaffold bearing a quaternary chiral carbon core.³³⁶ Mangicol A **504**, among the other members of this class of naturally occurring compounds, has been found to be active anti-inflammatory agents, thus, has important therapeutic effect. Many unfinished efforts have been made for the total synthesis of this kind of compounds.^{337–340} For instance, Uemura and co-workers in 2004 reported the total synthesis of core structure of mangicols A **504** in twenty-nine steps employing a Diels–Alder reaction of a macrocycle.³³⁷ In 2012, Yu and co-workers^{14,15,56,58,341} designed an effective pathway to obtain the spirotricyclic core analogue of mangicol A. This route involved asymmetric Pauson–Khand reaction and intramolecular Diels–Alder reaction as key steps. The total synthesis began from the BINOL-based catalytic asymmetric addition of enynal **498** to enyne **499** leading to formation of the optically active propargylic alcohol (*R*)-**500**. Treatment of the latter with *n*-BuLi followed by reaction with allyl bromide in DMSO afforded the corresponding diene–diyne substrate (*R*)-**501** as an appropriate precursor for PKR. Thus, the latter was subjected to the rhodium(I)-based catalyst and then to the PKR conditions ($[\text{RhCl}(\text{CO})_2]_2$, CO, reflux, 24 h). The two cycloadditions of alkyne units took place to furnish compound **502**. The latter was converted after several steps to the desired natural product, the mangicol A **504** (Scheme 39).³⁴²



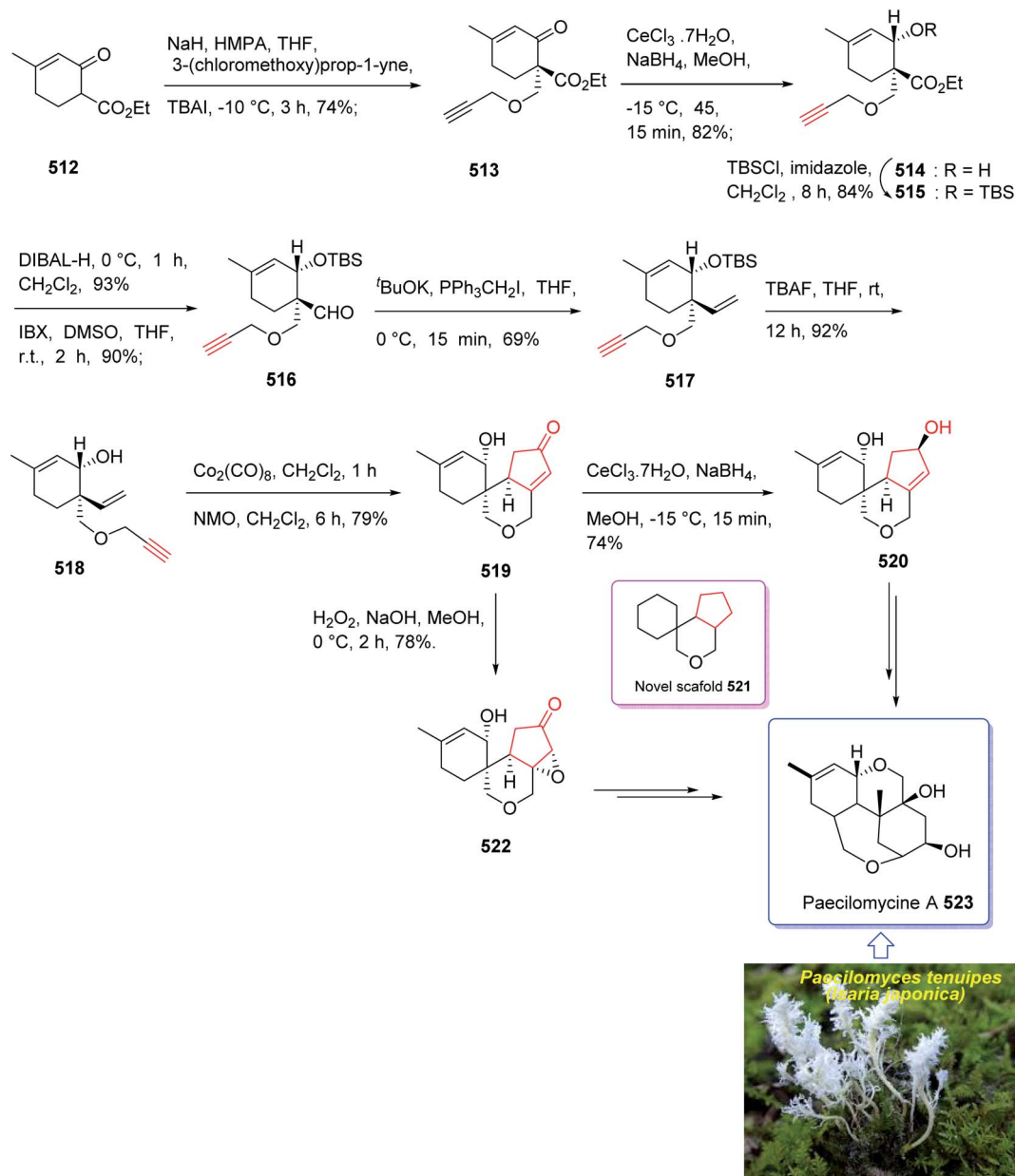
Scheme 40 Total synthesis of (\pm)-pentalenene **511**.



2.5. Sesquiterpenoids

Piers *et al.*³⁴³ in 1984 reported total synthesis of (±)-pentalenene **511** using Pauson–Khand reaction. The natural product (±)-pentalenene **511**, belongs to the pentalenolactone family of sesquiterpenoid antibiotics. It was initially isolated in 1980 by Seto and Yonehara from *Streptomyces griseochromogenes*.²⁴⁴ It has an interesting structure containing the parent hydrocarbon. In fact, (±)-pentalenene **511** has been found³⁴⁴ to be a biosynthetic precursor of pentalenolactone.^{327,345,346} Initially, several α,ω -allenynes **507** [1,2-dien-7-yne **507a–d** ($n = 1$) and 1,2-dien-8-yne **507e,f** ($n = 2$)] which were requisite for this total synthesis were provided *via* the Pd-catalyzed reaction of enyne **505** with the phosphates **506** in the presence of NaH in which

the phosphate moiety was substituted. Unambiguously, cyclization of the bicyclic enone *via* the tertiary amine oxide-promoted PKR of enyne **507** was anticipated to proceed in the expected desired stereochemical sense. Consequently, intramolecular cobalt-catalyzed PKR of α,ω -allenynes **507** as an appropriate PKR precursor resulted in the formation of bicyclic dienones **509** and α -alkyldienecyclopentenones **511**, regioselectively, *via* formation and conversion of intermediate hexacarbonyldicobalt complex **508a** depending on the substitution pattern of the allenic moiety. It seemed likely that transformation of α -alkyldienecyclopentenones **509** into (±)-**511** after several steps is straightforward leading to the formation of the desired natural product (±)-pentalenene **511** (Scheme 40).^{343,347}

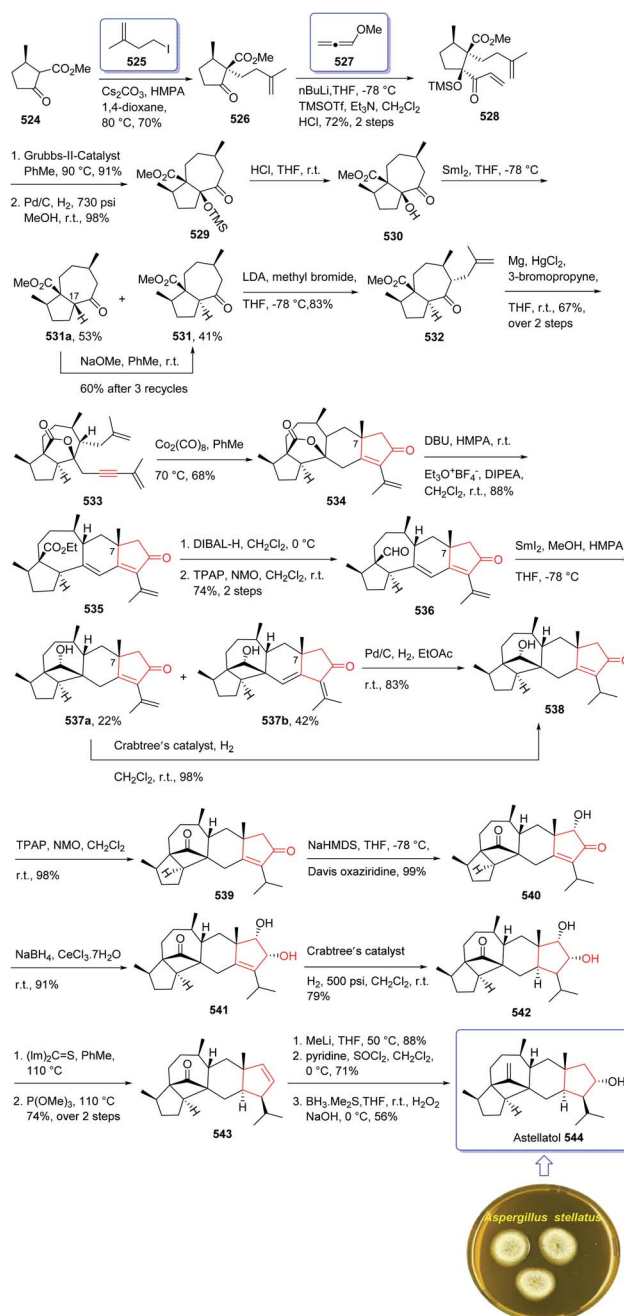


Scheme 41 Total synthesis of paecilomycine A 523.

Paecilomycine A **523**, is a tricothecane derived naturally occurring sesquiterpenoid which was initially isolated in 2004 by Oshima and co-workers, among other *Paecilomyces tenuipes* and terpenoids from extract of cultured fruiting bodies of *Paecilomyces tenuipes* (*Isaria japonica*).^{126,348} Paecilomycine A **523** showed an increased and inspiring neurite outgrowth in PC-12 cells and impressive neurotrophic activity.¹²⁶ The total synthesis of paecilomycine A **523** was reported by Mehta *et al.* in 2012 using intramolecular Pauson–Khand reaction.³⁴⁹ This strategy commenced with ethyl 4-methyl-2-oxocyclohex-3-enecarboxylate **512** which was submitted to propargyloxy-methylation using (propargyloxy) methyl chloride³⁵⁰ to give **513**. The latter was subjected to stereoselective Luche reduction³⁵¹ in compound **513** to afford α -hydroxy compound **514**. The latter then after several steps including protection–deprotection and various functional group transformations such as Wittig olefination, was converted to **518** as an appropriate PKR precursor. In a key step, the latter underwent stereoselective intramolecular PKR in the presence of $\text{Co}_2(\text{CO})_8$ and NMO in CH_2Cl_2 to provide the spiro-fused tricyclic hydroxy-enone **519**.

As a matter of fact, two derivatives were synthesized from **519**: (a) a tricyclic diol **520** obtained from stereoselective Luche reduction and (b) the respective epoxide **521** by nucleophilic epoxidation which was used for biological screening. Remarkably, all of the novel synthesized compounds **519**, **520**, **522** symbolizing the 2-oxa-spiro[5.5]undecane segment, which were already recognized to be neuroprotective in standard MTT and trypan blue for cell viability screening.^{352–354} Interestingly, both compounds, **520** and **522** were converted to the desired natural product paecilomycine A **523**. The expedition for the synthesis of novel structure exhibiting neurotrophic activity, encouraged by paecilomycine A **523**, has resulted in the design and synthesis of a novel framework containing 2-oxa-spiro [5.5] undecane core (Scheme 41).³⁴⁹

Among terpenes as a wide-spread family of naturally occurring small compounds, sesterterpenoids probably are the most enthralling molecules with complicated architectures showing various biological potencies. Several members showed anticarcinogenic, antimicrobial, anti-inflammatory, and cytotoxic potencies.^{355,356} One of these sesterterpenoids found to have a compound so-called astellatol **544**.^{357,358} Initially, in 1989 Sadler and co-workers isolated astellatol **544** from extract of *Aspergillus stellatus* and elucidated its structure.³⁵⁷ According to their structural determination, the left part of astellatol **544** structure possesses a crowded ring system with several stereogenic centers involving quaternary centers which makes its synthesis difficult. In addition, another major synthetic problem is the presence of highly substituted, notorious *trans*-hydrindane moiety on the right side.^{359,360} Astellatol **544** possesses an executional bicycle [4.1.1] octane moiety, ten stereogenic centers, a cyclobutane involving two quaternary centers, an *exo*-methylene group, and a sterically hindered isopropyl *trans*-hydrindane moiety. Due to the structural complexity of astellatol **544**, its total synthesis has been very challenging and stimulating which took about 30 years to be successfully achieved.



Scheme 42 Total synthesis of astellatol **544**.

The total synthesis of astellatol **544** was successfully attempted by E. J. Corey and co-workers in 1985,³⁶⁰ and later by Paquette,^{361,362} Hudlicky,^{363,364} and Wender.³⁶⁵ In 2018, Xu *et al.* also accomplished **536** (ref. 366) and reported a brief and asymmetric synthesis of astellatol **544**. That was including a SmI_2 -catalyzed reductive radical 1,4-addition.³⁶⁷ This synthetic pathway started from the alkylation of the already reported chiral synthon **524** (ref. 368) with the homoallylic iodide **525** in the presence of Cs_2CO_3 and HPMA in dioxane under reflux to provide ketone **526**. The latter was attacked by the lithiated derivative of methoxypropadiene **527**,³⁶⁹ which upon protection and hydrolysis gave the enone **528**. The latter after several steps





Scheme 43 Total synthesis of A-ring component of nitiol 557.

was transformed to multigram quantities of 533, as an appropriate precursor for PKR. Pleasantly, the substrate 533 underwent PKR using Co₂(CO)₈, in 2015 by Yang *et al.*³⁷⁰ to afford the

expected cyclized product 534. After several steps, compound 534 was converted to alkene 543. Then, the latter was converted to the desired natural product astellatol 544 in two steps. In the

first step, alkene **543** was reacted with MeLi at 50 °C to give the tertiary alcohol, which successfully subjected to elimination in the presence of SOCl₂/pyridine conditions which leading to the *exo*-methylene functionality on the cyclobutane framework. In the second step, a hydroboration-oxidation occurred to give astellatol **544** in 56% yield. As a result, the total synthesis of the rare sesterterpenoid, astellatol **544**, was achieved in twenty-five steps and 0.63% overall yield starting from chiral synthon **524** (Scheme 42).^{366,371}

A concise and asymmetric total synthesis of A-ring component of nitiol **559** including diastereoselective Pauson–Khand cycloaddition was reported by Dake in 2001.³⁷² It was actually an attempt towards the total synthesis of the natural product, nitiol **558**.^{105,360,373} In 1995, Cordell and co-workers found out that the plant so-called *Gentianella* (G.) is a regular remedial plant that cultivates in the Andes area. Its extract contains several compounds with diverse structures showing various biological activities, extensively employed as folk medicine for the treatment of hepatitis, and obesity.^{374,375} Nitiol **558** was initially extracted from this whole plant and was separated several times and purified *via* different techniques by Kawahara research group in 1999 resulting in the isolation of, novel sesterterpenoid.³⁷⁶ Interestingly, its structural elucidation was made by the same group in the same year.³⁷⁶ Primary biological screening showed that compound **558** behaves as an active enhancer of interleukin-2 in human T cell lines.³⁷⁶ The total synthesis of nitiol **558** was accomplished and reported by Dake and co-workers in 2001.³⁷² The important steps in this strategy are an asymmetric Pauson–Khand cycloaddition and a norrish type 1 fragmentation reaction.^{105,360,373} The total synthesis started from diethyl methylmalonate **545** which upon alkylation with the appropriate tosylate in the presence of sodium hydride and catalytic amount of sodium iodide in DMF gave 2,2-disubstituted malonates **548** and **549**.^{377,378} The required enyne **551** and **550** were prepared *via* standard procedures involving, LAH reduction, TBSCl protection, oxidation, Corey–Fuchs or Nozaki reactions. Compound **551** was then subjected to Sugihara's conditions^{15,379,380} to provide desired bicyclooctenes **552a** and **552b** in satisfactory yields (57–74%). The prepared cyclopentenone **552a** then underwent conjugate reduction by utilization of lithium tri(*sec*-butyl) borohydride followed by quenching with methyl iodide to afford **553** in high isolated yield in pure form. Delightfully, this process afforded trisubstituted cyclopentane **554** in 50% yield. In this step, the ester functional group of **554** was transformed into an aldehyde **555** under typical conditions, in two steps including reduction using diisobutylaluminum hydride and Moffatt–Swern oxidation in 86% overall yield. The latter upon treatment with Kogen's Horner–Wadsworth–Emmons-type reagent **A**³⁸¹ gave the tri-substituted vinyl bromide **556** in excellent yield (>97%) and high stereoselectivity (*E* : *Z*, 14 : 1). Finally, the latter after several steps was converted to the desired natural products **557** in 40% overall yield (Scheme 43).³⁷²

3. Conclusion

In conclusion, in this review, we tried to draw the attention of readers, especially synthetic organic chemists to one of the most

useful name reactions in organic chemistry so-called Pauson–Khand reaction (PKR). We presented a brief summary of the catalytic PKR. As we mentioned there are just a few organic transformations that complement so much molecular complexity in one step as the PKR. The products of PKRs are cyclopentenones, which can be readily converted into various functionalized cyclopentanes, which exist as structural elements in the scaffolds of several natural products. Then, we discussed the merits and drawbacks, observed for PKR, during the years. As mentioned, the intramolecular PKR (IPKR) was developed. In addition, various transition metal catalysts were introduced. We showed that by using modified (chiral) catalysts and reaction conditions, the products of PKR can be synthesized, asymmetrically. Thus the asymmetric variant of this important name reaction, nowadays is quite possible, leading to optically active products. Asymmetric PKRs can be achieved by using different approaches, including use of chiral ligands and chiral metal complexes. Comprehension of the reaction mechanism will allow accurate prediction of the stereochemical outcome of the reaction. Thus, we focused on the applications of PKR in the total synthesis of natural products trying to encourage synthetic chemists to rely on this reaction when designing their synthetic pathways leading to total synthesis of an appropriate natural products. In addition, the content of this review has been arranged based on the family and types of plants, which the certain natural product has been isolated from and their biological activities were also mentioned. That makes this review in addition to organic synthetic chemists, useful and readable to natural products chemists, pharmacists and botanists.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 P. L. Pauson and I. U. Khand, *Ann. N. Y. Acad. Sci.*, 1974, **295**, 2–14.
- 2 I. U. Khand, G. R. Knox, P. L. Pauson, W. E. Watts and M. I. Foreman, *J. Chem. Soc., Perkin Trans. 1*, 1973, 977–981.
- 3 W. J. Kerr, M. McLaughlin and P. L. Pauson, *J. Organomet. Chem.*, 2001, **630**, 104.
- 4 H. Werner, *Angew. Chem., Int. Ed.*, 2014, **53**, 3309.
- 5 N. Schore and M. J. Croudace, *J. Org. Chem.*, 1981, **46**, 5436–5438.
- 6 C. Exon and P. J. Magnus, *J. Am. Chem. Soc.*, 1983, **105**, 2477–2478.
- 7 D. Das and T. K. Chakraborty, *Tetrahedron Lett.*, 2016, **57**, 3665–3677.



- 8 N. E. Schore, *Chem. Rev.*, 1988, **88**, 1081–1119.
- 9 (a) N. E. Schore, *Org. React.*, 1991, **40**, 1; (b) S. T. Ingate and J. Marco-Contelles, *Org. Prep. Proced. Int.*, 1998, **30**, 121.
- 10 C. Paradisi, B. M. Trost, L. Fleming and M. F. Semmelhack, in *Comprehensive Organic Synthesis*, Oxford, New York, Seoul, Tokyo, 1991, vol. 4.
- 11 N. Schore, *Transition Metal Alkyne Complexes: Pauson-Khand Reaction*, ed. L. S. Hege, Comprehensive Organometallic Chemistry-II, Pergamon, Oxford, 1995.
- 12 O. Geis and H. G. Schmalz, *Angew. Chem., Int. Ed.*, 1998, **37**, 911–914.
- 13 D. Strübing and M. Beller, The Pauson-Khand Reaction, in *Catalytic Carbonylation Reactions*, Springer, 2006.
- 14 S. E. Gibson and A. J. Stevenazzi, *Angew. Chem., Int. Ed.*, 2003, **42**, 1800–1810.
- 15 K. M. Brummond and J. L. Kent, *Tetrahedron*, 2000, **56**, 3263–3283.
- 16 W. Smit, A. Gybin, A. Shashkov, Y. Strychkov, L. Kyz'mina, G. Mikaelian, R. Caple and E. J. Swanson, *Tetrahedron Lett.*, 1986, **27**, 1241–1244.
- 17 N. Jeong, Y. K. Chung, B. Y. Lee, S. H. Lee and S. E. Yoo, *Synlett*, 1991, **1991**, 204–206.
- 18 S. Shambayani, W. E. Crowe and S. L. Schreiber, *Tetrahedron Lett.*, 1990, **31**, 5289–5292.
- 19 L. Pérez-Serrano, L. Casarrubios, G. Domínguez and J. J. Pérez-Castells, *Org. Lett.*, 1999, **1**, 1187–1188.
- 20 L. Pérez-Serrano, J. Blanco-Urgoiti, L. Casarrubios, G. Domínguez and J. J. Pérez-Castells, *J. Org. Chem.*, 2000, **65**, 3513–3519.
- 21 T. Sugihara, M. Yamada, M. Yamaguchi and M. J. Nishizawa, *Synlett*, 1999, **6**, 771–773.
- 22 T. Sugihara, M. Yamada, H. Ban, M. Yamaguchi and C. J. Kaneko, *Angew. Chem., Int. Ed.*, 1997, **36**, 2801–2804.
- 23 D. S. Brown, E. Campbell, W. J. Kerr, D. M. Lindsay, A. J. Morrison, K. G. Pike and S. P. Watson, *Synlett*, 2000, **2000**, 1573–1576.
- 24 D. H. Hua, M. J. Coulter and I. J. Badejo, *Tetrahedron Lett.*, 1987, **28**, 5465–5468.
- 25 A. Veretenov, W. Smit, L. Vorontsova, M. Kurella, R. Caple and A. J. Gybin, *Tetrahedron Lett.*, 1991, **32**, 2109–2112.
- 26 Z. J. Yang, *Acc. Chem. Res.*, 2021, **54**, 556–568.
- 27 I. K. Itami, K. Mitsudo and J. Yoshida, *Angew. Chem., Int. Ed.*, 2002, **114**, 3631–3634.
- 28 N. Schore, in *Comprehensive Organic Synthesis*, ed. B. M. Trost and I. Fleming, Pergamon, Oxford, 1991.
- 29 T. Shibata and K. J. Takagi, *J. Am. Chem. Soc.*, 2000, **122**, 9852–9853.
- 30 Y. Koga, T. Kobayashi and K. J. Narasaka, *Chem. Lett.*, 1998, **27**, 249–250.
- 31 N. Jeong, S. Lee and B. K. Sung, *Organometallics*, 1998, **17**, 3642–3644.
- 32 N. Jeong, B. K. Sung and Y. K. Choi, *J. Am. Chem. Soc.*, 2000, **122**, 6771–6772.
- 33 K. Hiroi, T. Watanabe, R. Kawagishi and I. J. Abe, *Tetrahedron Lett.*, 2000, **41**, 891–895.
- 34 F. A. Hicks, N. M. Kablaoui and S. L. Buchwald, *J. Am. Chem. Soc.*, 1996, **118**, 9450–9451.
- 35 N. Y. Lee and Y. K. Chung, *Tetrahedron Lett.*, 1996, **37**, 3145–3148.
- 36 F. A. Hicks, N. M. Kablaoui and S. L. Buchwald, *J. Am. Chem. Soc.*, 1999, **121**, 5881–5898.
- 37 F. A. Hicks, N. M. Kablaoui and S. L. Buchwald, *J. Am. Chem. Soc.*, 1996, **118**, 11688–11689.
- 38 F. A. Hicks and S. L. Buchwald, *J. Am. Chem. Soc.*, 1999, **121**, 7026–7033.
- 39 N. Jeong, S. H. Hwang, Y. W. Lee and J. S. Lim, *J. Am. Chem. Soc.*, 1997, **119**, 10549–10550.
- 40 B. Y. Lee, Y. K. Chung, N. Jeong, Y. Lee and S. H. Hwang, *J. Am. Chem. Soc.*, 1994, **116**, 8793–8794.
- 41 B. L. Pagenkopf and T. J. Livinghouse, *J. Am. Chem. Soc.*, 1996, **118**, 2285–2286.
- 42 T. Morimoto, N. Chatani, Y. Fukumoto and S. J. Murai, *J. Org. Chem.*, 1997, **62**, 3762–3765.
- 43 A. J. Pearson and R. A. Dubbert, *Organometallics*, 1994, **13**, 1656–1661.
- 44 T. R. Hoye and J. A. Suriano, *Organometallics*, 1992, **11**, 2044–2050.
- 45 T. R. Hoye and J. A. Suriano, *J. Am. Chem. Soc.*, 1993, **115**, 1154–1156.
- 46 N. Jeong, S. J. Lee, B. Y. Lee and Y. K. Chung, *Tetrahedron Lett.*, 1993, **34**, 4027–4030.
- 47 N. Jeong, B. K. Sung, J. S. Kim, S. B. Park, S. D. Seo, J. Y. Shin, K. Y. In and Y. K. Choi, *Pure Appl. Chem.*, 2002, **74**, 85–91.
- 48 D. C. Billington and D. J. Willison, *Tetrahedron Lett.*, 1984, **25**, 4041–4044.
- 49 D. C. Billington, *Tetrahedron Lett.*, 1983, **24**, 2905–2908.
- 50 P. Magnus, L. M. Principe and M. J. Slater, *J. Org. Chem.*, 1987, **52**, 1483–1486.
- 51 D. C. Billington, W. J. Kerr, P. L. Pauson and C. F. Farnocchi, *J. Organomet. Chem.*, 1988, **356**, 213–219.
- 52 K. Hiroi, T. Watanabe, R. Kawagishi and I. Abe, *Tetrahedron: Asymmetry*, 2000, **11**, 797–808.
- 53 P. Magnus and L. M. Principe, *Tetrahedron Lett.*, 1985, **26**, 4851–4854.
- 54 Y. Gimbert, D. Lesage, A. Milet, F. Fournier, A. E. Greene and J.-C. Tabet, *Org. Lett.*, 2003, **5**, 4073–4075.
- 55 M. Yamanaka and E. Nakamura, *J. Am. Chem. Soc.*, 2001, **123**, 1703–1708.
- 56 H. W. Lee and F. Y. Kwong, *Eur. J. Org. Chem.*, 2010, **2010**, 789–811.
- 57 J. H. Park, K.-M. Chang and Y. K. Chung, *Coord. Chem. Rev.*, 2009, **253**, 2461–2480.
- 58 J. Blanco-Urgoiti, L. Añorbe, L. Pérez-Serrano, G. Domínguez and J. Pérez-Castells, *Chem. Soc. Rev.*, 2004, **33**, 32–42.
- 59 K. Nicolaou, D. Vourloumis, N. Winssinger and P. S. Baran, *Angew. Chem., Int. Ed.*, 2000, **39**, 44–122.
- 60 Y. Naya and M. J. Kotake, *Tetrahedron Lett.*, 1968, **9**, 1645–1649.
- 61 R. Tressl, L. Fries, F. Fendesack, H. J. Koeppel and F. Chemistry, *J. Agric. Food Chem.*, 1978, **26**, 1426–1430.
- 62 V. E. Peacock and M. L. Deinzer, *J. Am. Soc. Brew. Chem.*, 1981, **39**, 136–141.



- 63 R. Tressl, L. Friese, F. Fendesack and H. J. Koeppler, *J. Agric. Food Chem.*, 1978, **26**, 1422–1426.
- 64 T. L. Peppard, S. A. Ramus, C. A. Witt and K. J. Siebert, *J. Am. Soc. Brew. Chem.*, 1989, **47**, 18–26.
- 65 J. J. Caldwell, I. O. Cameron, S. D. Christie, A. M. Hay, C. Johnstone, W. J. Kerr and A. J. Murray, *Synthesis*, 2005, **19**, 3293–3296.
- 66 K. C. Lam and M. L. Deinzer, *J. Am. Soc. Brew. Chem.*, 1986, **44**, 69–72.
- 67 K. Y. Jeong, M. Son, S.-Y. Choi, K. H. Park, H. J. Park, C.-S. Hong, J.-H. Lee and J.-W. J. Park, *Yonsei Med. J.*, 2016, **57**, 399.
- 68 T. Imagawa, N. Murai, T. Akiyama and M. J. Kawanisi, *Tetrahedron Lett.*, 1979, **20**, 1691–1692.
- 69 C. R. Johnson, R. C. Elliott and N. A. Meanwell, *Tetrahedron Lett.*, 1982, **23**, 5005–5008.
- 70 (a) H. Iwabuchi, *Abstracts of Papers: 41st Symposium on the Chemistry of Terpenes, Essential Oils, and Aromatics*, Morioka, Japan, 1997; (b) S. Muraki, *Abstracts of Papers: 41st Symposium on the Chemistry of Terpenes, Essential Oils, and Aromatics*, Shizuoka, Japan, 1968, p. 13.
- 71 I. Sakata and T. J. Hashizume, *Agric. Biol. Chem.*, 1973, **37**, 2441–2442.
- 72 K. Takahashi, T. Someya, S. Muraki and T. J. Yoshida, *Agric. Biol. Chem.*, 1980, **44**, 1535–1543.
- 73 G. T. Crisp and A. G. Meyer, *Tetrahedron*, 1995, **51**, 5831–5846.
- 74 C. S. Foote, M. Wuesthoff, S. Wexler, I. Burstain, R. Denny and G. Schenck, *Tetrahedron*, 1967, **23**, 2583–2599.
- 75 J. A. Hirsch and R. H. Eastman, *J. Org. Chem.*, 1967, **32**, 2915–2916.
- 76 V. Tadwalkar and A. J. Rao, *Chem. Informationsdienst*, 1972, **3**(21), DOI: 10.1002/chin.197221273.
- 77 A. Akhila, R. Srivastava and K. Rani, *Phytochemistry*, 1991, **30**, 485–489.
- 78 K. Shishido, O. Irie and M. J. Shibuya, *Tetrahedron Lett.*, 1992, **33**, 4589–4592.
- 79 R. W. Bates and S. J. Sridhar, *J. Org. Chem.*, 2008, **73**, 8104–8105.
- 80 S. J. Abidi, *Tetrahedron Lett.*, 1986, **27**, 267–270.
- 81 M. Pasquali, P. Leoni, P. Sabatino and D. Braga, *Synthesis, Structure and Reactivity of fac-Mo(CO)₃(DMF)₃-Effect of Lewis-Acids on the Substitution of DMF Ligands*, arpi.unipi.it, 1992.
- 82 P. A. Wender and B. L. Miller, *Organic Synthesis: Theory and Applications*, ed. T. Hudlicky, JAP Press, Greenwich, CT, 1993, vol. 2, p. 27.
- 83 F. J. Qiu, *Can. J. Chem.*, 2008, **86**, 903–906.
- 84 L. J. El-Naggar and J. L. Beal, *J. Nat. Prod.*, 1980, **43**, 649–707.
- 85 B. Dinda, S. Debnath and Y. J. Harigaya, *Chem. Pharm. Bull.*, 2007, **55**, 159–222.
- 86 R. Tundis, M. R. Loizzo, F. Menichini, G. A. Statti and F. J. Menichini, *Mini-Rev. Med. Chem.*, 2008, **8**, 399–420.
- 87 S. McElvain, P. M. Walters and R. D. Bright, *J. Am. Chem. Soc.*, 1942, **64**, 1828–1831.
- 88 M. S. Kaurav, V. Kumar, D. Kumar and T. J. Khan, *ChemistrySelect*, 2018, **3**, 5566–5570.
- 89 S. Lee, S.-M. Paek, H. Yun, N.-J. Kim and Y.-G. Suh, *Org. Lett.*, 2011, **13**, 3344–3347.
- 90 L. Candish and D. W. Lupton, *Org. Lett.*, 2010, **12**, 4836–4839.
- 91 H. P. A. Khan, D. Das and T. K. Chakraborty, *J. Org. Chem.*, 2018, **83**, 6086–6092.
- 92 A. D'Alfonso, M. Pasi, A. Porta, G. Zanoni and G. J. Vidari, *Org. Lett.*, 2010, **12**, 596–599.
- 93 J. Sim, L. Yoon, H. Yun, H. An and Y.-G. Suh, *Org. Biomol. Chem.*, 2016, **14**, 1244–1251.
- 94 A. Salam, S. Ray, M. A. Zaid, D. Kumar and T. J. Khan, *Org. Biomol. Chem.*, 2019, **17**, 6831–6842.
- 95 R. L. Kitching, E. Scheermeyer and R. Jones, *Biology of Australian Butterflies*, Csiro Publishing, 1999.
- 96 S. Xu, T. Toyama, J. Nakamura and H. J. Arimoto, *Tetrahedron Lett.*, 2010, **51**, 4534–4537.
- 97 D. E. Chavez and E. N. Jacobsen, *Org. Lett.*, 2003, **5**, 2563–2565.
- 98 U. Pagnoni, A. Pinetti, R. Trave and L. J. Garanti, *Aust. J. Chem.*, 1976, **29**, 1375–1381.
- 99 K. D. Wellington, R. C. Cambie, P. S. Rutledge and P. S. Bergquist, *J. Nat. Prod.*, 2000, **63**, 79–85.
- 100 E. Corey and T. Engler, *Tetrahedron Lett.*, 1984, **25**, 149–152.
- 101 K. C. Nicolaou, S. A. Snyder, T. Montagnon and G. Vassilikogiannakis, *Angew. Chem., Int. Ed.*, 2002, **41**, 1668–1698.
- 102 D. L. Clive and E. J. Stoffman, *Org. Biomol. Chem.*, 2008, **6**, 1831–1842.
- 103 B. M. Trost, *J. Org. Chem.*, 2004, **69**, 5813–5837.
- 104 J. B. Sperry and D. L. Wright, *Tetrahedron Lett.*, 2005, **46**, 411–414.
- 105 B. B. Snider, D. J. Rodini and J. Van Straten, *J. Am. Chem. Soc.*, 1980, **102**, 5872–5880.
- 106 R. M. Adlington and A. G. Barrett, *Acc. Chem. Res.*, 1983, **16**, 55–59.
- 107 J. P. Konopelski and M. A. Boehler, *J. Am. Chem. Soc.*, 1989, **111**, 4515–4517.
- 108 D. C. Baker and C. Chu, *Nucleosides and Nucleotides as Antitumor and Antiviral Agents*, Springer, 2013.
- 109 S. S. Nyandoro and C. Sciences, *Int. J. Biol. Chem. Sci.*, 2017, **11**, 3075–3085.
- 110 S. S. Nyandoro, *J. Pharmacogn. Phytochem.*, 2014, **3**, 147–157.
- 111 T. Fukumoto and R. Ikarashi, Process for producing and method of crystallizing 2-azabicyclo(2.2.1) hept-5-en-3-one, *US Pat.*, US6153765A, 2000.
- 112 B. D. Anderson and C.-Y. Chiang, *J. Pharm. Sci.*, 1990, **79**, 787–790.
- 113 W. H. Miller, S. Daluge, E. P. Garvey, S. Hopkins, J. E. Reardon, F. L. Boyd and R. L. Miller, *J. Biol. Chem.*, 1992, **267**, 21220–21224.
- 114 C. Simons, *Nucleoside Mimetics: Their Chemistry and Biological Properties*, CRC Press, 2000.
- 115 A. Vazquez-Romero, J. Rodríguez, A. Lledo, X. Verdager and A. J. Riera, *Org. Lett.*, 2008, **10**, 4509–4512.
- 116 *Biological Methylation and Drug Design — Experimental and clinical roles of S-adenosylmethione*, ed. R. T. Borchardt, C.



- R. Creveling and P. M. Ueland, Humana Press, Clifton, NJ, 1986.
- 117 F. Casu, M. A. Chiacchio, R. Romeo and G. Gumina, *Curr. Org. Chem.*, 2007, **11**, 999–1016.
- 118 B. G. Roy, P. K. Jana, B. Achari and S. B. Mandal, *Tetrahedron Lett.*, 2007, **48**, 1563–1566.
- 119 B. M. Trost, *Chem. Pharm. Bull.*, 2002, **50**, 1–14.
- 120 A. Lanver and H. G. Schmalz, *Nucleoside Analogs*, 2005, 1444–1458.
- 121 S. E. Gibson and N. J. Mainolfi, *Angew. Chem., Int. Ed.*, 2005, **44**, 3022–3037.
- 122 J. Solà, M. Revés, A. Riera and X. J. Verdager, *Angew. Chem., Int. Ed.*, 2007, **46**, 5020–5023.
- 123 A. Becheanu, T. Bell, S. Laschat, A. Baro, W. Frey, N. Steinke and P. J. Fischer, *Z. Naturforsch., B: J. Chem. Sci.*, 2006, **61**, 589–596.
- 124 Z. Benko, B. Fraser-Reid, P. S. Mariano and A. J. Beckwith, *J. Org. Chem.*, 1988, **53**, 2066–2072.
- 125 P. A. Grieco and N. J. Abood, *J. Org. Chem.*, 1989, **54**, 6008–6010.
- 126 H. Kikuchi, Y. Miyagawa, Y. Sahashi, S. Inatomi, A. Haganuma, N. Nakahata and Y. J. Oshima, *Tetrahedron Lett.*, 2004, **45**, 6225–6228.
- 127 S. J. Min and S. J. Danishefsky, *Angew. Chem., Int. Ed.*, 2007, **119**, 2249–2252.
- 128 J. G. Martin and R. K. Hill, *Chem. Rev.*, 1961, **61**, 537–562.
- 129 E. W. Colvin and I. G. Thom, *Tetrahedron*, 1986, **42**, 3137–3146.
- 130 H.-S. Byun, K. C. Reddy and R. J. Bittman, *Tetrahedron Lett.*, 1994, **35**, 1371–1374.
- 131 M. Ishizaki, K. Iwahara, K. Kyomura and O. J. Hoshino, *Synlett*, 1999, **1999**, 587–589.
- 132 C. Lei, S.-X. Huang, J.-J. Chen, L.-B. Yang, W.-L. Xiao, Y. Chang, Y. Lu, H. Huang, J.-X. Pu and H.-D. Sun, *J. Nat. Prod.*, 2008, **71**, 1228–1232.
- 133 Y.-M. Shi, W.-L. Xiao, J.-X. Pu and H.-D. Sun, *Nat. Prod. Rep.*, 2015, **32**, 367–410.
- 134 C. J. Douglas and L. E. Overman, *Proc. Natl. Acad. Sci. U. S. A.*, 2004, **101**, 5363–5367.
- 135 R.-T. Li, Q.-B. Han, Y.-T. Zheng, R.-R. Wang, L.-M. Yang, Y. Lu, S.-Q. Sang, Q.-T. Zheng, Q.-S. Zhao and H.-D. Sun, *Chem. Commun.*, 2005, **23**, 2936–2938.
- 136 Y. Tang, L. Deng, Y. Zhang, G. Dong, J. Chen and Z. J. Yang, *Org. Lett.*, 2005, **7**, 593–595.
- 137 L. You, X.-T. Liang, L.-M. Xu, L.-M. Y.-F. Wang, J.-J. Zhang, Q. Su, Y.-H. Li, B. Zhang, S.-L. Yang and J.-H. Chen, *J. Am. Chem. Soc.*, 2015, **137**, 10120–10123.
- 138 A. Farwick, J. U. Engelhart, O. Tverskoy, C. Welter, Q. A. Umlauf, F. Rominger, W. J. Kerr and G. J. Helmchen, *Adv. Synth. Catal.*, 2011, **353**, 349–370.
- 139 V. Van Rhee, R. Kelly and D. J. Cha, *Tetrahedron Lett.*, 1985, **1973**, 8.
- 140 W. L. Xiao, H.-J. Zhu, Y.-H. Shen, R.-T. Li, S.-H. Li, H.-D. Sun, Y.-T. Zheng, R.-R. Wang, Y. Lu, C. J. Wang and G. Lancifodilactone, *Org. Lett.*, 2005, **7**, 2145–2148.
- 141 X. Li, P. H. Cheong and R. G. Carter, *Angew. Chem., Int. Ed.*, 2017, **56**, 1704–1718.
- 142 D.-D. Liu, T.-W. Sun, K.-Y. Wang, Y. Lu, S.-L. Zhang, Y.-H. Li, Y.-L. Jiang, J.-H. Chen and Z. Yang, D-D Liu, T-W Sun, K-Y Wang, Y Lu, S-L Zhang, Y-H Li, Y-L Jiang, J-H Chen, and Z Yang, *J. Am. Chem. Soc.* 2017, **139**, 5732–5735, *J. Am. Chem. Soc.*, 2017, **139**, 5732–5735.
- 143 K. Nicolaou and S. Snyder, *Angew. Chem., Int. Ed.*, 2002, **41**, 1002.
- 144 E. J. Corey, *Angew. Chem., Int. Ed.*, 2009, **48**, 2100–2117.
- 145 W.-W. Ren, Z.-X. Chen, Q. Xiao, Y. Li, T.-W. Sun, Z.-Y. Zhang, Q.-D. Ye, F.-K. Meng and L. You, *Chem.-Asian J.*, 2012, **7**, 2341–2350.
- 146 T. W. Sun, W. W. Ren, Q. Xiao, Y. F. Tang, Y. D. Zhang, Y. Li, F. K. Meng, Y. F. Liu, M. Z. Zhao and L. M. J. Xu, *Chem.-Asian J.*, 2012, **7**, 2321–2333.
- 147 W.-L. Xiao, R.-T. Li, S.-X. Huang, J.-X. Pu and H.-D. Sun, *Nat. Prod. Rep.*, 2008, **25**, 871–891.
- 148 W.-L. Xiao, S.-Y. Yang, S.-Y. Yang, G.-Y. Yang, R.-R. Wang, H.-B. Zhang, W. Zhao, J.-X. Pu, Y. Lu and Y.-T. Zheng, *J. Nat. Prod.*, 2010, **73**, 221–225.
- 149 S.-X. Huang, R.-T. Li, J.-P. Liu, Y. Lu, Y. Chang, C. Lei, W.-L. Xiao, L.-B. Yang, Q.-T. Zheng and H.-D. Sun, *Org. Lett.*, 2007, **9**, 2079–2082.
- 150 R.-T. Li, Q.-S. Zhao, S.-H. Li, Q.-B. Han, H.-D. Sun, Y. Lu, L.-L. Zhang and Q.-T. Zheng, *Org. Lett.*, 2003, **5**, 1023–1026.
- 151 T. Mole and E. A. Jeffery, *Organoaluminium compounds*, Applied Chemistry, Elsevier Publishing Company, 1972.
- 152 S. F. Brady, M. P. Singh, J. E. Janso and J. Clardy, *J. Am. Chem. Soc.*, 2000, **122**, 2116–2117.
- 153 G. B. Dudley and S. J. Danishefsky, *Org. Lett.*, 2001, **3**, 2399–2402.
- 154 D. S. Tan, G. B. Dudley and S. J. Danishefsky, *Angew. Chem., Int. Ed.*, 2002, **41**, 2185–2188.
- 155 S. Lin, G. B. Dudley, D. S. Tan and S. J. Danishefsky, *Angew. Chem., Int. Ed.*, 2002, **41**, 2188–2191.
- 156 B. Shi, N. A. Hawryluk and B. B. Snider, *J. Org. Chem.*, 2003, **68**, 1030–1042.
- 157 K. M. Brummond, H. Chen, K. D. Fisher, A. D. Kerekes, B. Rickards, C. Sill and S. J. Geib, *Org. Lett.*, 2002, **4**, 1931–1934.
- 158 A. B. Smith III, B. D. Dorsey, M. Ohba Jr, A. T. Lupo and M. S. Malamas, *J. Org. Chem.*, 1988, **53**, 4314–4325.
- 159 K. M. Foote, C. J. Hayes, M. P. John and G. Pattenden, *Org. Biomol. Chem.*, 2003, **1**, 3917–3948.
- 160 K. M. Brummond and D. J. Gao, *Org. Lett.*, 2003, **5**, 3491–3494.
- 161 S. Kim and J. D. Winkler, *Chem. Soc. Rev.*, 1997, **26**, 387–399.
- 162 K. Zechmeister, F. Brandl, W. Hoppe, E. Hecker, H. Opferkuch and W. J. Adolf, *Tetrahedron*, 1970, **11**, 4075–4078.
- 163 C. M. Hasler, G. Acs and P. M. Blumberg, *Cancer Res.*, 1992, **52**, 202–208.
- 164 E. J. Hecker, *Cancer Res.*, 1968, **28**, 2338–2348.
- 165 R. W. Alder and S. P. East, *Chem. Rev.*, 1996, **96**, 2097–2112.
- 166 M. Blanco-Molina, G. C. Tron, A. Macho, C. Lucena, M. A. Calzado, E. Muñoz and G. Appendino, *Chem. Biol.*, 2001, **8**, 767–778.



- 167 J. D. Winkler, M. B. Rouse, M. F. Greaney, S. J. Harrison and Y. T. Jeon, *J. Am. Chem. Soc.*, 2002, **124**, 9726–9728.
- 168 J. D. Winkler, E. C. Lee and L. I. Nevels, *Org. Lett.*, 2005, **7**, 1489–1491.
- 169 J. D. Winkler, S. Kim, S. Harrison, N. E. Lewin and P. M. Blumberg, *J. Am. Chem. Soc.*, 1999, **121**, 296–300.
- 170 O. Taglialatela-Scafati, U. Deo-Jangra, M. Campbell, M. Roberge and R. J. Andersen, *Org. Lett.*, 2002, **4**, 4085–4088.
- 171 O. Taglialatela-Scafati, K. S. Craig, D. Rebérioux, M. Roberge and R. J. Andersen, *Eur. J. Org. Chem.*, 2003, **2003**, 3515–3523.
- 172 S. A. Look, W. Fenical, D. Van Engen and J. Clardy, *J. Am. Chem. Soc.*, 1984, **106**, 5026–5027.
- 173 P. D. Thornton and D. J. Burnell, *Org. Lett.*, 2006, **8**, 3195–3198.
- 174 M. Yamaguchi, K. Shibato, S. Fujiwara and I. Hirao, *Synthesis*, 1986, **5**, 421–422.
- 175 T. Imamoto, N. Takiyama, K. Nakamura, T. Hatajima and Y. J. Kamiya, *J. Am. Chem. Soc.*, 1989, **111**, 4392–4398.
- 176 D. Green, I. Goldberg, Z. Stein, M. Ilan and Y. Kashman, *Nat. Prod. Lett.*, 1992, **1**, 193–199.
- 177 Y. Chang, L. Shi, J. Huang, L. Shi, Z. Zhang, H.-D. Hao, J. Gong and Z. J. Yang, *Org. Lett.*, 2018, **20**, 2876–2879.
- 178 J. Peng, K. Walsh, V. Weedman, J. D. Bergthold, J. Lynch, K. L. Lieu, I. A. Braude, M. Kelly and M. T. Hamann, *Tetrahedron*, 2002, **58**, 7809–7819.
- 179 M. W. Pfeiffer and A. J. Phillips, *J. Am. Chem. Soc.*, 2005, **127**, 5334–5335.
- 180 J. A. Enquist Jr and B. M. Stoltz, *Nature*, 2008, **453**, 1228–1231.
- 181 C. Wang, D. Wang and S. J. Gao, *Org. Lett.*, 2013, **5**, 4402–4405.
- 182 T. Xu, C.-c. Li and Z. J. Yang, *Org. Lett.*, 2011, **13**, 2630–2633.
- 183 G. C. Fu and R. H. Grubbs, *J. Am. Chem. Soc.*, 1992, **114**, 5426–5427.
- 184 E. F. Rogers, F. R. Koniuszy Jr, J. Shavel and K. Folkers, *J. Am. Chem. Soc.*, 1948, **70**, 3086–3088.
- 185 K. Wiesner, Z. Valenta and J. A. Findlay, *Tetrahedron Lett.*, 1967, **8**, 221–223.
- 186 S. N. Srivastava and M. Przybylska, *Can. J. Chem.*, 1968, **46**, 795–797.
- 187 B. P. Pepper and L. J. Carruth, *J. Econ. Entomol.*, 1945, **38**, 59–66.
- 188 R. P. Mason and J. L. Holtzman, *Biochem. Biophys. Res. Commun.*, 1975, **67**, 1267–1274.
- 189 J. T. Lanner, D. K. Georgiou, A. D. Joshi and S. L. Hamilton, *Cold Spring Harbor Perspect. Biol.*, 2010, **2**, a003996.
- 190 K. V. Chuang and C. Xu, *Science*, 2016, **353**, 912–915.
- 191 M. Egi, Y. Ota, Y. Nishimura, K. Shimizu, K. Azechi and S. Akai, *Org. Lett.*, 2013, **15**, 4150–4153.
- 192 T. F. Jamison, S. Shambayati, W. E. Crowe and S. L. Schreiber, *J. Am. Chem. Soc.*, 1994, **116**, 5505–5506.
- 193 N. Enoki, R. Ishida and T. Matsumoto, *Chem. Lett.*, 1982, **11**, 1749–1752.
- 194 N. Enoki, A. Furusaki, K. Suehiro, R. Ishida and T. Matsumoto, *Tetrahedron Lett.*, 1983, **24**, 4341–4342.
- 195 N. A. Petasis and M. A. Patane, *Tetrahedron*, 1992, **48**, 5757–5821.
- 196 S.-J. Chang, D. McNally, S. Shary-Tehrany, S. M. Hickey and R. H. Boyd, *J. Am. Chem. Soc.*, 1970, **92**, 3109–3118.
- 197 S. L. Schreiber, T. Sammakia and W. E. Crowe, *J. Am. Chem. Soc.*, 1986, **108**, 3128–3130.
- 198 T. Ohsawa, T. Kobayashi, Y. Mizuguchi, T. Saitoh and T. J. Oishi, *Tetrahedron Lett.*, 1985, **26**, 6103–6106.
- 199 M. J. Martin, R. Fernandez, A. Francesch, P. Amade, S. S. de Matos-Pita, F. Reyes and C. Cuevas, *Org. Lett.*, 2010, **12**, 912–914.
- 200 B. Dinda, S. Debnath and Y. Harigaya, *Chem. Pharm. Bull.*, 2007, **55**, 159–222.
- 201 J.-P. Chen, W. He, Z.-Y. Yang and Z.-J. Yao, *Org. Lett.*, 2015, **17**, 3379–3381.
- 202 K. Hiroi, Y. Hiratsuka, K. Watanabe, I. Abe, F. Kato and M. J. Hiroi, *Synlett*, 2001, **2001**, 0263–0265.
- 203 D. Rigano, G. Aviello, M. Bruno, C. Formisano, S. Rosselli, R. Capasso, F. Senatore, A. Izzo and F. Borrelli, *J. Nat. Prod.*, 2009, **72**, 1477–1481.
- 204 H. M. Ge, Z. G. Yu, J. Zhang, J. H. Wu and R. X. Tan, *J. Nat. Prod.*, 2009, **72**, 753–755.
- 205 H. Yamakoshi, Y. Sawayama, Y. Akahori, M. Kato and S. Nakamura, *Org. Lett.*, 2016, **18**, 3430–3433.
- 206 Y. Akahori, H. Yamakoshi, Y. Sawayama, S. Hashimoto and S. Nakamura, *J. Org. Chem.*, 2014, **79**, 720–735.
- 207 E. Nieto-Mendoza, J. A. Guevara-Salazar, M. T. Ramírez-Apan, B. A. Frontana-Urbe, J. A. Cogordan and J. Cárdenas, *J. Org. Chem.*, 2005, **70**, 4538–4541.
- 208 J. Kupka, T. Anke, F. Oberwinkler, G. Schramm and W. Steglich, *J. Antibiot.*, 1979, **32**, 130–135.
- 209 T. Anke, J. Heim, F. Knoch, U. Mocek, B. Steffan and W. Steglich, *Angew. Chem., Int. Ed.*, 1985, **24**, 709–711.
- 210 E. Piers and J. J. Renaud, *J. Org. Chem.*, 1993, **58**, 11–13.
- 211 T. Kang, S. B. Song, W.-Y. Kim, B. G. Kim and H.-Y. Lee, *J. Am. Chem. Soc.*, 2014, **136**, 10274–10276.
- 212 Y. Tang, L. Deng, Y. Zhang, G. Dong, J. Chen and Z. J. Yang, *Org. Lett.*, 2005, **7**, 1657–1659.
- 213 Z. Huang, J. Huang, Y. Qu, W. Zhang, J. Gong and Z. Yang, *Angew. Chem., Int. Ed.*, 2018, **130**, 8880–8884.
- 214 N. Jeong, D. Kim and J. Choi, *Chem. Commun.*, 2004, **9**, 1134–1135.
- 215 B. M. Trost and V. K. Chang, *Synthesis*, 1993, **1993**, 824–832.
- 216 J. H. Lee and L. J. Deng, *J. Am. Chem. Soc.*, 2012, **134**, 18209–18212.
- 217 E. Weitz and A. J. Scheffer, *Ber. Dtsch. Chem. Ges.*, 1921, **54**, 2344–2353.
- 218 M. M. McLachlan, P. D. O'Connor, K. A. Fairweather, A. C. Willis and L. Mander, *Aust. J. Chem.*, 2010, **63**, 742–760.
- 219 J. S. Clark and C. Xu, *Angew. Chem., Int. Ed.*, 2016, **55**, 4332–4335.
- 220 M. Shigeta, T. Hakamata, Y. Watanabe, K. Kitamura, Y. Ando, K. Suzuki and T. Matsumoto, *Synlett*, 2010, **2010**, 2654–2658.
- 221 S. Müller, B. Liepold, G. J. Roth and H. J. Bestmann, *Synlett*, 1996, **1996**, 521–522.



- 222 C. B. Tripathi and S. J. Mukherjee, *J. Org. Chem.*, 2012, **77**, 1592–1598.
- 223 W. Adam, Y. Y. Chan, D. Cremer, J. Gauss, D. Scheutzow and M. Schindler, *J. Org. Chem.*, 1987, **52**, 2800–2803.
- 224 J. D. White, N. S. Cutshall, T.-S. Kim and H. J. Shin, *J. Am. Chem. Soc.*, 1995, **117**, 9780–9781.
- 225 R. Battaglia, M. De Bernardi, G. F. Mellerio, G. Vidari and P. Vita-Finzi, *J. Nat. Prod.*, 1980, **43**, 319–328.
- 226 M. E. Price and N. E. Schore, *J. Org. Chem.*, 1989, **54**, 5662–5667.
- 227 T. Sato, M. Watanabe and R. Noyori, *Tetrahedron Lett.*, 1979, **20**, 2897–2900.
- 228 W. S. Johnson and H. J. Posvic, *J. Am. Chem. Soc.*, 1947, **69**, 1361–1366.
- 229 R. E. Ireland and J. A. Marshall, *J. Org. Chem.*, 1962, **27**, 1615–1619.
- 230 P. Mosset and R. J. Grée, *Synth. Commun.*, 1985, **15**, 749–757.
- 231 J. L. Jessen, G. Schröder and W. J. Tochtermann, *Chem. Ber.*, 1985, **118**, 3287–3298.
- 232 M. Toyota, Y. Nishikawa, K. Motoki, N. Yoshida and K. Fukumoto, *Tetrahedron*, 1993, **49**, 11189–11204.
- 233 K. Ramig, M. A. Kuzemko, K. McNamara and T. Cohen, *J. Org. Chem.*, 1992, **57**, 1968–1969.
- 234 P. A. Wender and J. J. Howbert, *Tetrahedron Lett.*, 1982, **23**, 3983–3986.
- 235 J. Castro, H. Sorensen, A. Riera, C. Morin, A. Moyano, M. A. Pericas and A. Greene, *J. Am. Chem. Soc.*, 1990, **112**, 9388–9389.
- 236 M. C. Yeh and P. Knochel, *Tetrahedron Lett.*, 1989, **30**, 4799–4802.
- 237 M. C. Croudace and N. E. Schore, *J. Org. Chem.*, 1981, **46**, 5357–5363.
- 238 G. A. Molander and G. J. Hahn, *J. Org. Chem.*, 1986, **51**, 1135–1138.
- 239 A. Matsuo, S. Sato, M. Nakayama and S. J. Hayashi, *Tetrahedron Lett.*, 1974, **15**, 3681–3684.
- 240 A. Matsuo, S. Sato, M. Nakayama and S. Hayashi, *J. Chem. Soc., Perkin Trans. 1*, 1979, 2652–2656.
- 241 J. G. Donkervoort, A. R. Gordon, C. Johnstone, W. J. Kerr and U. J. Lange, *Tetrahedron*, 1996, **52**, 7391–7420.
- 242 H. Alper and J.-T. Edward, *Can. J. Chem.*, 1970, **48**, 1543–1549.
- 243 D. E. Ward, C. K. Rhee and W. M. Zoghaib, *Tetrahedron Lett.*, 1988, **29**, 517–520.
- 244 H. Seto and H. Yonehara, *J. Antibiot.*, 1980, **33**, 92–93.
- 245 E. G. Rowley and N. E. Schore, *J. Org. Chem.*, 1992, **57**, 6853–6861.
- 246 M. Hatanaka, F. Ueno and I. J. Ueda, *Tetrahedron Lett.*, 1996, **37**, 89–90.
- 247 P. Magnus, C. Exon and P. J. Albaugh-Robertson, *Tetrahedron*, 1985, **41**, 5861–5869.
- 248 J. Tormo, A. Moyano, M. A. Pericàs and A. J. Riera, *J. Org. Chem.*, 1997, **62**, 4851–4856.
- 249 T. McMorris, M. Kelner, W. Wang, M. Diaz, L. Estes and R. J. Taetle, *Experientia*, 1996, **52**, 75–80.
- 250 T. C. McMorris, J. Yu, Y. Hu, L. A. Estes and M. J. Kelner, *J. Org. Chem.*, 1997, **62**, 3015–3018.
- 251 J. R. MacDonald, C. C. Muscoplat, D. L. Dexter, G. L. Mangold, S.-F. Chen, M. J. Kelner, T. C. McMorris and D. D. Von Hoff, *Cancer Res.*, 1997, **57**, 279–283.
- 252 M. Kelner, T. McMorris, L. Estes, K. Samson, R. Bagnell and R. J. Taetle, *Eur. J. Cancer*, 1998, **34**, 908–913.
- 253 T. C. McMorris, M. J. Kelner, W. Wang, J. Yu, L. A. Estes and R. J. Taetle, *J. Nat. Prod.*, 1996, **59**, 896–899.
- 254 T. C. McMorris, Y. Hu, J. Yu and M. J. Kelner, *Chem. Commun.*, 1997, **3**, 315–316.
- 255 A. Padwa, E. A. Curtis and V. P. Sandanayaka, *J. Org. Chem.*, 1996, **61**, 73–81.
- 256 K. M. Brummond and J. J. Lu, *J. Am. Chem. Soc.*, 1999, **121**, 5087–5088.
- 257 K. M. Brummond, H. Wan and J. L. Kent, *J. Org. Chem.*, 1998, **63**, 6535–6545.
- 258 J. L. Kent, *J. Org. Chem.*, 1998, **63**, 6535.
- 259 A. San Feliciano, A. Barrero, M. Medarde, J. M. del Corral, A. Aramburu, A. Perales and J. Fayos, *Tetrahedron Lett.*, 1985, **26**, 2369–2372.
- 260 P. A. Wender, N. C. Ihle and C. R. Correia, *J. Am. Chem. Soc.*, 1988, **110**, 5904–5906.
- 261 L. A. Paquette, J. Tae, M. P. Arrington and A. H. Sadoun, *J. Am. Chem. Soc.*, 2000, **122**, 2742–2748.
- 262 J. Limanto and M. L. Snapper, *J. Am. Chem. Soc.*, 2000, **122**, 8071–8072.
- 263 M. Krafft, Y. Cheung and K. J. Abboud, *J. Org. Chem.*, 2001, **66**, 7443–7448.
- 264 H.-P. Hanssen and W.-R. Abraham, *Tetrahedron*, 1988, **44**, 2175–2180.
- 265 G. Mehta, N. Krishnamurthy and S. R. Karra, *J. Am. Chem. Soc.*, 1991, **113**, 5765–5775.
- 266 D. L. Clive and S. R. Magnuson, *Tetrahedron Lett.*, 1995, **36**, 15–18.
- 267 C. Mukai, M. Kobayashi, I. J. Kim and M. J. Hanaoka, *Tetrahedron*, 2002, **58**, 5225–5230.
- 268 Y. K. Chung, B. Y. Lee, N. Jeong, M. Hudecek and P. L. Pauson, *Organometallics*, 1993, **12**, 220–223.
- 269 A. F. Barrero, J. Q. del Moral and A. J. Lara, *Tetrahedron*, 2000, **56**, 3717–3723.
- 270 G. Stork and R. J. Breslow, *J. Am. Chem. Soc.*, 1953, **75**, 3291.
- 271 P. J. Walter, *Liebigs Ann. Chem.*, 1841, **39**, 247–251.
- 272 G. Stork and F. H. Clarke Jr, *J. Am. Chem. Soc.*, 1955, **77**, 1072–1073.
- 273 J. J. Crawford, W. J. Kerr, M. McLaughlin, A. J. Morrison, P. L. Pauson and G. J. Thurston, *Tetrahedron*, 2006, **62**, 11360–11370.
- 274 D. Dess and J. J. Martin, *J. Org. Chem.*, 1983, **48**, 4155–4156.
- 275 S. J. Ohira, *Synth. Commun.*, 1989, **19**, 561–564.
- 276 C. Johnstone, W. J. Kerr and J. S. Scott, *Chem. Commun.*, 1996, **3**, 341–342.
- 277 B. Prema and P. J. Bhattacharyya, *Appl. Environ. Microbiol.*, 1962, **10**, 524–528.
- 278 M. Maheshwari, K. Varma and S. J. Bhattacharyya, *Tetrahedron*, 1963, **19**, 1519–1525.



- 279 M. Maheshwari, T. Jain, R. Bates and S. J. Bhattacharyya, *Tetrahedron*, 1963, **19**, 1079–1090.
- 280 H. Barrett and G. J. Buechi, *J. Am. Chem. Soc.*, 1967, **89**, 5665–5667.
- 281 L. Jiao, M. Lin, L. G. Zhuo and Z.-X. J. Yu, *Org. Lett.*, 2010, **12**, 2528–2531.
- 282 J. R. Henderson, M. Parvez and B. A. Keay, *Org. Lett.*, 2007, **9**, 5167–5170.
- 283 A. P. Krapcho, G. A. Glynn and B. J. Grenon, *Tetrahedron Lett.*, 1967, **8**, 215–217.
- 284 J. W. Huffman and R. C. Desai, *J. Org. Chem.*, 1982, **47**, 3254–3258.
- 285 L. Y. Wang, T. Uehara and S. J. Kitanaka, *Chem. Pharm. Bull.*, 2005, **53**, 137–139.
- 286 Y. Hayashi, K. Ogawa, F. Inagaki and C. Mukai, *Org. Biomol. Chem.*, 2012, **10**, 4747.
- 287 R. Baudouy and P. J. Prince, *Tetrahedron*, 1989, **45**, 2067–2074.
- 288 Y. Hayashi, K. Ogawa, F. Inagaki and C. J. Mukai, *Org. Biomol. Chem.*, 2012, **10**, 4747–4751.
- 289 J.-m. Huang, R. Yokoyama, C.-s. Yang and Y. J. Fukuyama, *Tetrahedron Lett.*, 2000, **41**, 6111–6114.
- 290 V. B. Birman and S. J. Danishefsky, *J. Am. Chem. Soc.*, 2002, **124**, 2080–2081.
- 291 M. Inoue, T. Sato and M. J. Hiraama, *J. Am. Chem. Soc.*, 2003, **125**, 10772–10773.
- 292 M. Inoue, T. Sato and M. J. Hiraama, *Angew. Chem., Int. Ed.*, 2006, **118**, 4961–4966.
- 293 M. Inoue, N. Lee, S. Kasuya, T. Sato, M. Hiraama, M. Moriyama and Y. J. Fukuyama, *J. Org. Chem.*, 2007, **72**, 3065–3075.
- 294 G. Mehta and S. R. Singh, *Angew. Chem., Int. Ed.*, 2006, **45**, 953–955.
- 295 W. He, J. Huang, X. Sun and A. J. Frontier, *J. Am. Chem. Soc.*, 2008, **130**, 300–308.
- 296 L. Shi, K. Meyer and M. F. Greaney, *Angew. Chem., Int. Ed.*, 2010, **122**, 9436–9439.
- 297 Z. Meng and S. J. Danishefsky, *Angew. Chem., Int. Ed.*, 2005, **44**, 1511–1513.
- 298 K. Harada, H. Kato and Y. J. Fukuyama, *Tetrahedron Lett.*, 2005, **46**, 7407–7410.
- 299 K. Harada, H. Ito, H. Hioki and Y. J. Fukuyama, *Tetrahedron Lett.*, 2007, **48**, 6105–6108.
- 300 J. Iriando-Alberdi, J. E. Perea-Buceta and M. F. Greaney, *Org. Lett.*, 2005, **7**, 3969–3971.
- 301 G. Mehta and S. R. Singh, *Tetrahedron Lett.*, 2005, **46**, 2079–2082.
- 302 J. Chen, P. Gao, F. Yu, Y. Yang, S. Zhu and H. J. Zhai, *Angew. Chem., Int. Ed.*, 2012, **51**, 5897–5899.
- 303 C. Mukai, T. Yoshida, M. Sorimachi and A. J. Odani, *Org. Lett.*, 2006, **8**, 83–86.
- 304 J. Adrio and J. C. Carretero, *J. Am. Chem. Soc.*, 2007, **129**, 778–779.
- 305 A. Reichenberg, M. Hintz, Y. Kletschek, T. Kuhl, C. Haug, R. Engel, J. Moll, D. N. Ostrovsky, H. Jomaa and M. J. Eberl, *Bioorg. Med. Chem. Lett.*, 2003, **13**, 1257–1260.
- 306 K. Tadano, J. Ishihara, H. Yamada and S. J. Ogawa, *J. Org. Chem.*, 1989, **54**, 1223–1227.
- 307 L. Shi, X. Lei, J. Zhang and G. J. Lin, *Helv. Chim. Acta*, 2010, **93**, 555–564.
- 308 M. Nerz-Stormes and E. R. Thornton, *J. Org. Chem.*, 1991, **56**, 2489–2498.
- 309 A. B. Smith and G. R. Ott, *Angew. Chem., Int. Ed.*, 1998, **120**, 3935–3948.
- 310 S. K. Bagal, R. M. Adlington, R. A. Brown and J. E. Baldwin, *Tetrahedron Lett.*, 2005, **46**, 4633–4637.
- 311 A. Takahashi, H. Yanai, M. Zhang, T. Sonoda, M. Mishima and T. J. Taguchi, *J. Org. Chem.*, 2010, **75**, 1259–1265.
- 312 A. Takahashi, H. Yanai and T. J. Taguchi, *Chem. Commun.*, 2008, **20**, 2385–2387.
- 313 G. Matsuo, K. Kawamura, N. Hori, H. Matsukura and T. J. Nakata, *J. Am. Chem. Soc.*, 2004, **126**, 14374–14376.
- 314 W. Zi, S. Yu and D. J. Ma, *Angew. Chem., Int. Ed.*, 2010, **49**, 5887–5890.
- 315 J.-M. Huang, C.-S. Yang, R. Zhao, H. Takahashi and Y. J. Fukuyama, *Chem. Pharm. Bull.*, 2004, **52**, 104–107.
- 316 W. J. Kerr, M. McLaughlin, L. C. Paterson and C. M. Pearson, *Tetrahedron*, 2018, **74**, 5062–5068.
- 317 Y. Ito, T. Hirao and T. J. Saegusa, *J. Org. Chem.*, 1978, **43**, 1011–1013.
- 318 S. Kobayashi, I. Hachiya, T. Takahori, M. Araki and H. J. Ishitani, *Tetrahedron Lett.*, 1992, **33**, 6815–6818.
- 319 W. J. Kerr, M. McLaughlin, A. J. Morrison and P. L. Pauson, *Org. Lett.*, 2001, **3**, 2945–2948.
- 320 T. Sugihara, M. Yamada, M. Yamaguchi and M. J. Nishizawa, *Synlett*, 1999, **6**, 771–773.
- 321 S. A. Lopez, M. Pourati, H.-J. Gais and K. J. Houk, *J. Org. Chem.*, 2014, **79**, 8304–8312.
- 322 S. Takeuchi, *Tetrahedron Lett.*, 1969, **10**, 2737–2740.
- 323 D. Martin, G. Slomp, S. Mizsak, D. Duchamp and C. J. Chidester, *Tetrahedron Lett.*, 1970, **11**, 4901–4904.
- 324 D. E. Cane, J. K. Sohng and P. G. Williard, *J. Org. Chem.*, 1992, **57**, 844–851.
- 325 D. L. Pruess and J. P. Scannell, *Adv. Appl. Microbiol.*, 1974, **17**, 19–62.
- 326 J. H. Rigby, M. S. Laxmisha, A. R. Hudson, C. H. Heap and M. J. Heeg, *J. Org. Chem.*, 2004, **69**, 6751–6760.
- 327 D. J. Duchamp and C. G. Chidester, *Acta Crystallogr., Sect. B: Struct. Crystallogr. Cryst. Chem.*, 1972, **28**, 173–180.
- 328 W. G. Dauben and J. M. Gerdes, *Tetrahedron Lett.*, 1983, **24**, 3841–3844.
- 329 L. A. Paquette and M. A. Poupart, *J. Org. Chem.*, 1993, **58**, 4245–4253.
- 330 S. Danishefsky, M. Hiraama, K. Gombatz, T. Harayama, E. Berman and P. Schuda, *J. Am. Chem. Soc.*, 1978, **100**, 6536–6538.
- 331 A. R. Bassindale, S. J. Glynn and P. G. Taylor, *Activating and Directive Effects of Silicon*, John Wiley & Sons, Ltd., 2009.
- 332 Q. Liu, G. Yue, N. Wu, G. Lin, Y. Li, J. Quan, C. Li, G. Wang and Z. J. Yang, *Angew. Chem., Int. Ed.*, 2012, **124**, 12238–12242.



- 333 H. Brederbeck, G. Simchen, S. Rebsdats, W. Kantelehner, P. Horn, R. Wahl, H. Hoffmann and P. J. Grieshaber, *Chem. Ber.*, 1968, **101**, 41–50.
- 334 Y. Ban, N. Taga and T. J. Oishi, *Tetrahedron Lett.*, 1974, **15**, 187–190.
- 335 S. Danishefsky, M. Hiram, K. Gombatz, T. Harayama, E. Berman and P. J. Schuda, *J. Am. Chem. Soc.*, 1978, **100**, 6536–6538.
- 336 M. K. Renner, P. R. Jensen and W. J. Fenical, *J. Org. Chem.*, 2000, **65**, 4843–4852.
- 337 K. Araki, K. Saito, H. Arimoto and D. J. Uemura, *Angew. Chem., Int. Ed.*, 2004, **116**, 83–86.
- 338 S. Pichlmair, M. de Lera Ruiz, K. Basu and L. A. Paquette, *Tetrahedron*, 2006, **62**, 5178–5194.
- 339 S. Pichlmair, M. de Lera Ruiz, I. Vilotijevic and L. A. Paquette, *Tetrahedron*, 2006, **62**, 5791–5802.
- 340 J. L. Wood, B. G. Pujanauski and R. J. Sarpong, *Org. Lett.*, 2009, **11**, 3128–3131.
- 341 A. J. Fletcher and S. D. Christie, *J. Chem. Soc., Perkin Trans. 1*, 2000, **11**, 1657–1668.
- 342 W. Chen, J.-H. Tay, J. Ying, M. Sabat, X.-Q. Yu and L. J. Pu, *Chem. Commun.*, 2013, **49**, 170–172.
- 343 E. Piers and V. J. Karunaratne, *J. Chem. Soc., Chem. Commun.*, 1984, **15**, 959–960.
- 344 D. E. Cane and A. M. Tillman, *J. Am. Chem. Soc.*, 1983, **105**, 122–124.
- 345 D. G. Martin, G. Slomp, S. Mizsak, D. J. Duchamp and C. G. Chidester, *Tetrahedron Lett.*, 1970, **11**, 4901–4904.
- 346 D. J. Duchamp and C. G. Chidester, *Acta Crystallogr., Sect. B: Struct. Crystallogr. Cryst. Chem.*, 1972, **28**, 1092–1099.
- 347 M. Ahmar, C. Locatelli, D. Colombier and B. J. Cazes, *Tetrahedron Lett.*, 1997, **38**, 5281–5284.
- 348 G. Mehta, R. Samineni and P. J. Srihari, *Tetrahedron Lett.*, 2011, **52**, 1663–1666.
- 349 G. Mehta, R. Samineni, P. Srihari, R. G. Reddy and S. J. Chakravarty, *Org. Biomol. Chem.*, 2012, **10**, 6830–6833.
- 350 A. Hasan and P. C. Srivastava, *J. Med. Chem.*, 1992, **35**, 1435–1439.
- 351 J. L. Luche, *J. Am. Chem. Soc.*, 1978, **100**, 2226–2227.
- 352 M. Yamazaki and K. J. Chiba, *J. Health Sci.*, 2005, **51**, 687–692.
- 353 V. P. Kumar, R. G. Reddy, D. D. Vo, S. Chakravarty, S. Chandrasekhar and R. J. Grée, *Bioorg. Med. Chem. Lett.*, 2012, **22**, 1439–1444.
- 354 Y. P. Wang, Z. F. Wang, Y. C. Zhang and T. Qing, *Cell Res.*, 2004, **14**, 467–472.
- 355 Y. Liu, L. Wang, J. H. Jung and S. J. Zhang, *Nat. Prod. Rep.*, 2007, **24**, 1401–1429.
- 356 L. Wang, B. Yang, X.-P. Lin, X.-F. Zhou and Y. J. Liu, *Nat. Prod. Rep.*, 2013, **30**, 455–473.
- 357 I. H. Sadler and T. J. Simpson, *Chem. Commun.*, 1989, **21**, 1602–1604.
- 358 I. H. Sadler and T. J. Simpson, *Magn. Reson. Chem.*, 1992, **30**, S18–S23.
- 359 D. T. Hog, P. Mayer and D. J. Trauner, *J. Org. Chem.*, 2012, **77**, 5838–5843.
- 360 E. Corey, M. C. Desai and T. A. Engler, *J. Am. Chem. Soc.*, 1985, **107**, 4339–4341.
- 361 L. A. Paquette, J. Wright, G. J. Drtina and R. A. Roberts, *J. Org. Chem.*, 1987, **52**, 2960–2962.
- 362 J. Wright, G. J. Drtina, R. A. Roberts and L. A. Paquette, *J. Am. Chem. Soc.*, 1988, **110**, 5806–5817.
- 363 T. Hudlicky, L. Radesca-Kwart, L.-Q. Li and T. Bryant, *Tetrahedron Lett.*, 1988, **29**, 3283–3286.
- 364 T. Hudlicky and A. Fleming, *J. Am. Chem. Soc.*, 1989, **111**, 6691–6707.
- 365 P. A. Wender and S. K. Singh, *Tetrahedron Lett.*, 1990, **31**, 2517–2520.
- 366 N. Zhao, S. Yin, S. Xie, H. Yan, P. Ren, G. Chen, F. Chen and J. J. Xu, *Angew. Chem., Int. Ed.*, 2018, **57**, 3386–3390.
- 367 N. Zhao, S. Xie, G. Chen and J. J. Xu, *Chem.-Eur. J.*, 2016, **22**, 12634–12636.
- 368 J. N. Marx and L. R. Norman, *J. Org. Chem.*, 1975, **40**, 1602–1606.
- 369 J. Wencel-Delord, M. Mauduit and C. Crévisy, *Encyclopedia of Reagents for Organic Synthesis*, John Wiley & Sons, Ltd., 2001.
- 370 J. Zhang, X. Wang, S. Li, D. Li, S. Liu, Y. Lan, J. Gong and Z. Yang, *Chem.-Eur. J.*, 2015, **21**, 12596–12600.
- 371 N. Zhao, S. Xie, P. Tian, R. Tong, C. Ning and J. Xu, *Org. Chem. Front.*, 2019, **6**, 2014–2022.
- 372 M. S. Wilson and G. R. Dake, *Org. Lett.*, 2001, **3**, 2041–2044.
- 373 S. K. Attah-Poku, F. Chau, V. Yadav and A. G. Fallis, *J. Org. Chem.*, 1985, **50**, 3418–3419.
- 374 G. A. Cordell, *Phytochemistry*, 1995, **40**, 1585–1612.
- 375 P. J. Houghton, *J. Pharm. Pharmacol.*, 1999, **51**, 505–512.
- 376 N. Kawahara, M. Nozawa, A. Kurata, T. Hakamatsuka, S. Sekita and M. J. Satake, *Chem. Pharm. Bull.*, 1999, **47**, 1344–1345.
- 377 B. M. Trost, *J. Am. Chem. Soc.*, 1996, **118**, 6625–6633.
- 378 F. Björkling, J. Boutelje, S. Gatenbeck, K. Hult, T. Norin and P. J. Szmulik, *Tetrahedron*, 1985, **41**, 1347–1352.
- 379 S. M. Weinreb, B. Trost, I. J. Fleming and I. Fleming, *Comprehensive Organic Synthesis*, 1991.
- 380 R. H. Grubbs, S. H. Pine, B. M. Trost and I. J. Fleming, *Comprehensive Organic Synthesis*, Pergamon Press, New York, 1991.
- 381 K. Tago and H. J. Kogen, *Org. Lett.*, 2000, **2**, 1975–1978.

