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## EOM-CC guide to Fock-space travel: The C<sub>2</sub> edition

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Despite their small size, C<sub>2</sub> species pose big challenges to electronic structure, owing to their extensive electronic degeneracies and multi-configurational wave functions, which lead to a dense manifold of electronic states. We present detailed electronic structure calculations of C<sub>2</sub>, C<sub>2</sub><sup>-</sup>, and C<sub>2</sub><sup>2-</sup>, emphasizing spectroscopically relevant properties. We employ double ionization potential (DIP) and ionization potential (IP) variants of the equation-of-motion coupled-cluster method with single and double substitutions (EOM-CCSD) and a dianionic reference state. We show that EOM-CCSD is capable of describing multiple interacting states in C<sub>2</sub> and C<sub>2</sub><sup>-</sup> in an accurate, robust, and effective way. We also characterize the electronic structure of C<sub>2</sub><sup>2-</sup>, which is metastable with respect to electron detachment.

### 1 Introduction

Ironically, the smallest form of neat carbon, the C<sub>2</sub> molecule, features the most complex electronic structure. The complexity stems from the inability of the eight valence electrons of the two carbons to form a quadruple bond (remarkably, the bonding in C<sub>2</sub> is still hotly debated<sup>1–8</sup>). Because the optimal electron pairing cannot be attained, multiple electronic configurations have similar likelihood, leading to a dense manifold of low-lying electronic states. This results in rich spectroscopy: C<sub>2</sub> features multiple low-lying electronic transitions, which have been extensively studied experimentally<sup>9–14</sup>. Nevertheless, C<sub>2</sub> continues to generate interest. For example, recently, new band systems have been identified<sup>15–17</sup>.

Besides its obvious fundamental importance, C<sub>2</sub> (and its anionic forms, C<sub>2</sub><sup>-</sup> and C<sub>2</sub><sup>2-</sup>), play a role in combustion<sup>18</sup>, plasma<sup>19–21</sup>, and astrochemistry<sup>19,22</sup>. For example, C<sub>2</sub> and C<sub>2</sub><sup>-</sup> have been observed in comet tails, protoplanetary nebulae, the atmospheres of stars, and in the diffuse interstellar medium<sup>22–27</sup>. C<sub>2</sub> is responsible for the color of blue flames<sup>18</sup>. It is also a prominent product of electrical discharges containing hydrocarbons<sup>20</sup>.

From the theoretical point of view, C<sub>2</sub> is arguably the most difficult molecule among homonuclear diatomics from the first two rows of the periodic table. Electronic near-degeneracies lead to multiconfigurational wave-functions. Small energy separations between different electronic states also call for high accuracy. Because of its complex electronic structure, C<sub>2</sub> has often been described as the poster child of multi-reference methodology. The

availability of high-quality spectroscopic data, complex electronic structure, and its small size make C<sub>2</sub> a popular benchmark system for quantum chemistry studies<sup>28–32</sup>. Among recent theoretical studies of the low-lying states of C<sub>2</sub>, the most comprehensive are the tour-de-force MR-CISD (multi-reference configuration interaction with single and double excitations) calculations by Schmidt and coworkers<sup>33</sup> and by Szalay and co-workers<sup>34</sup>. In both studies, the effect of basis set and higher-order corrections have been carefully investigated. To correct MR-CISD energies for violation of size-extensivity, Davidson's quadruple correction was used. Szalay and co-workers have also reported results obtained with an alternative strategy, the so-called MR-average quadratic coupled-cluster (AQCC) method. In both studies, the theoretical values of the reported equilibrium distances (*r<sub>e</sub>*) and term energies (*T<sub>ee</sub>*) agreed well with the experimental data.

The anionic forms of C<sub>2</sub>, C<sub>2</sub><sup>-</sup> and C<sub>2</sub><sup>2-</sup>, have received less attention. C<sub>2</sub><sup>-</sup> is produced in plasma discharge from acetylene<sup>35,36</sup>. Electronically excited C<sub>2</sub><sup>-</sup> has been observed in a carbon-rich plasma via fluorescence<sup>21</sup>. Recently, C<sub>2</sub><sup>-</sup> has been proposed as a candidate for laser cooling of anions<sup>37</sup>; this makes these species interesting in the context of quantum information storage. Ervin and Lineberger<sup>38</sup> have measured the photoelectron spectrum of C<sub>2</sub><sup>-</sup> using 3.53 eV photons; they reported the adiabatic electron affinity (AEA) of C<sub>2</sub> to be 3.269±0.006 eV. A similar value (3.273±0.008 eV) has been derived by Neumark and coworkers<sup>39</sup>, who reported vibrationally resolved photodetachment spectra using 4.66 eV radiation. Feller has reported an AEA of 3.267 eV calculated using a composite method based on coupled-cluster (CC) methods<sup>40</sup>.

Because of its highly unsaturated character, C<sub>2</sub> has relatively large electron attachment energy, so that even the two lowest excited states of C<sub>2</sub><sup>-</sup> are bound electronically. In contrast, C<sub>2</sub><sup>2-</sup> is metastable with respect to electron detachment. Its existence,

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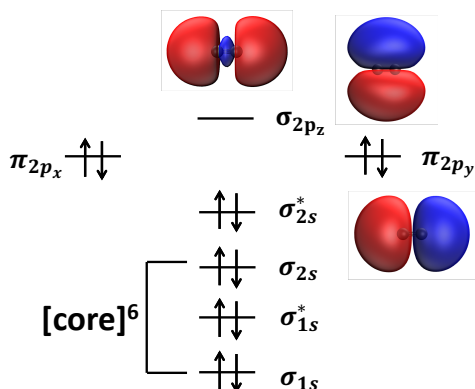
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which has been postulated on the basis of features observed<sup>41,42</sup> in electron scattering from  $C_2^-$ , has been confirmed by calculations<sup>43,44</sup>.

In this contribution, we present detailed electronic structure calculations of  $C_2$ ,  $C_2^-$ , and  $C_2^{2-}$ , with an emphasis on spectroscopically relevant properties. We employ an alternative methodology based on CC and equation-of-motion CC (EOM-CC) theory<sup>45–49</sup>. We show that electronic states of  $C_2$  and  $C_2^-$  are well described by the double ionization potential (DIP)<sup>50</sup> and ionization potential (IP)<sup>51,52</sup> variants of EOM-CCSD (EOM-CC with single and double substitutions) using a dianionic reference state. Formulated in a strictly single-reference fashion, the EOM-CC family of methods provides an accurate, robust, and effective alternative to cumbersome multi-reference calculations<sup>45–49</sup>. To describe metastable species, such as  $C_2^{2-}$ , we employ the complex-variable extension of CCSD and EOM-CCSD via the complex absorbing potential (CAP) approach<sup>53–55</sup>.

## 2 Molecular orbital framework and essential features of electronic structure of $C_2$ species



**Fig. 1** Molecular orbital diagram. The three lowest orbitals that remain doubly occupied in the low-energy manifold of electronic states of  $C_2$  and  $C_2^-$  are denoted as 'core'. The electronic states of  $C_2$  are derived by distributing six additional electrons over four upper orbitals,  $\sigma_{2s}^*$ ,  $\pi_{2px}/\pi_{2py}$ , and  $\sigma_{2pz}$ . Shown is the leading electronic configuration of the ground state,  $X^1\Sigma_g$ . Low-lying states of  $C_2^-$  are derived by distributing five electrons over the four upper orbitals. In  $C_2^{2-}$ , all four upper orbitals are doubly occupied. Shown are Dyson orbitals (isovalue = 0.05) computed with EOM-IP-CCSD/aug-cc-pVTZ from the dianionic reference.

Fig. 1 shows the molecular orbital diagram and describes orbital occupation patterns in  $C_2$ ,  $C_2^-$ , and  $C_2^{2-}$ . Due to orbital near-degeneracies, various electronic configurations of six electrons over the upper four orbitals have similar energies, leading to closely lying electronic states and multi-configurational wave-functions. In  $C_2^-$ , there are four important configurations in which the unpaired electron resides on one of the upper orbitals. In  $C_2^{2-}$ , which is isoelectronic with  $N_2$ , all four upper orbitals are doubly occupied, resulting in the electronic configuration  $[core]^6(\sigma_{2s}^*)^2(\pi_{2px})^2(\pi_{2py})^2(\sigma_{2pz})^2$ . Consequently, the ground state of  $C_2^{2-}$  is a well-behaved closed-shell state dominated by a

single Slater determinant; thus, it can be well described by single-reference methods such as, for example, CCSD. From this reference state, EOM-IP and EOM-DIP operators can generate all important electronic configurations needed for describing the electronic states of  $C_2^-$  and  $C_2$ , respectively, as illustrated in Fig. 2.

Mathematically, the EOM-CCSD target states are described by the following ansatz<sup>46–48</sup>:

$$\Psi = (\hat{R}_1 + \hat{R}_2)e^{\hat{T}_1 + \hat{T}_2}\Phi_0, \quad (1)$$

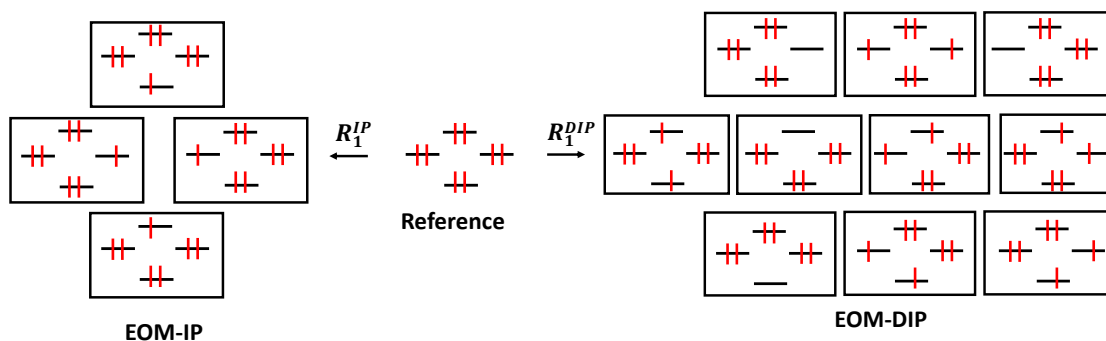
where  $e^{\hat{T}_1 + \hat{T}_2}\Phi_0$  is the reference CCSD wave function (the amplitudes of the excitation operator  $\hat{T}$  are determined by the CCSD equations for the reference state) and operator  $\hat{R}$  is a general excitation operator. In EOM-IP-CCSD,  $\hat{R}$  comprises all  $1h$  (one hole) and  $2h1p$  (two hole one particle) operators<sup>51,52</sup>, whereas in EOM-DIP-CCSD it comprises all  $2h$  and  $3h1p$  operators. In EOM-EE-CCSD (EOM-CCSD for excitation energies<sup>56</sup>) and EOM-SF-CCSD (spin-flip EOM-CCSD<sup>57,58</sup>),  $\hat{R}$  is particle-conserving and includes  $1h1p$  and  $2h2p$  operators (in the SF variant,  $\hat{R}$  changes the number of  $\alpha$  and  $\beta$  electrons). In the EA (electron attachment) variant<sup>59</sup>, the operator  $R$  is of the  $1p$  and  $1h2p$  type. The amplitudes of  $\hat{R}$  are found by diagonalization of the similarity-transformed Hamiltonian,  $\bar{H}$ :

$$\bar{H} = e^{-T}He^T, \quad (2)$$

$$\bar{H}R^k = E_kR^k. \quad (3)$$

Linear parameterization ensures that different configurations can mix and interact. There are no assumptions about their relative importance—the relative weights of different configurations are defined by the EOM eigen-problem and can span the entire range of situations, from those dominated by a single electronic configuration to those of equal contributions from multiple determinants. The EOM-CC ansatz is capable of reproducing exact degeneracies (such as between the two components of  $\Pi$  states in linear molecules or Jahn-Teller degeneracies), which are violated by state-specific MR treatments. Since all important configurations appear at the same excitation level, they are treated in a balanced way. As a multi-state method, EOM-CC produces the entire manifold of electronic states without requiring user input regarding state character. These features make EOM-CC very attractive for treating multiple electronic states and extensive degeneracies<sup>49</sup>. Recent applications illustrating the power of EOM-CC include calculations of electronic states of copper oxide anions<sup>60</sup>, Cvetanović diradicals<sup>61</sup>, and molecules with several unpaired electrons<sup>62,63</sup>.

The success of EOM-CC in treating a particular electronic structure depends on whether a proper well-behaved reference can be found from which the manifold of target states can be reached by an appropriately chosen  $\hat{R}_1$ . As illustrated in Fig. 2, the electronic structure of  $C_2$  is best described by EOM-DIP using the dianionic reference state. The DIP method is capable of describing electronic degeneracies beyond two-electrons-in-two-orbitals or three-electrons-in-three-orbitals patterns<sup>50,60,61,64–68</sup>; however, its applications are limited by complications due to the use of the dianionic reference.



**Fig. 2** EOM-IP (left) and EOM-DIP (right) manifolds generated from the dianionic reference (center). Only configurations generated by  $\hat{R}_1$  from the top four orbitals from Fig. 1 are shown. EOM-IP enables access to the ground and electronically excited states of  $C_2^-$ , whereas EOM-DIP describes the ground and excited states of  $C_2$ .

Isolated dianions of small molecules are usually unstable with respect to electron detachment and exist only as transient species (resonances).<sup>69</sup> In dianions, resonances emerge due to the competition between the two factors: (i) long-range repulsion between the anionic core and an extra electron and (ii) short-range stabilizing valence interactions. Together, these lead to a repulsive Coulomb barrier. The extra electron is trapped behind this barrier but can escape the system by tunneling. This is similar to metastable radical monoanions, where the extra electron is trapped behind an angular-momentum barrier, which affords resonance character. In a computational treatment using a sufficiently large basis, the wave function of a resonance becomes more and more diffuse, approximating a continuum state corresponding to an electron-detached system and a free electron<sup>70–72</sup>.

Resonances can be described by a non-Hermitian extension of quantum mechanics<sup>73</sup> by using, for example, complex absorbing potential (CAP)<sup>74,75</sup>. If one is interested in the dianionic state itself, then the CAP-based extension of CC theory can be used<sup>55</sup>. However, in practical calculations using EOM-DIP-CC, the dianionic state just serves as a reference for generating target configurations. Thus, less sophisticated approaches can be used to mitigate complications arising from its metastable character. The easiest and most commonly used approach is to use a relatively small basis set, such that the reference state is artificially stabilized<sup>50,60,61,64–66,76</sup>. Kuś and Krylov have investigated an alternative strategy: stabilization of the resonance using an artificial Coulomb potential with a subsequent de-perturbative correction<sup>71,72</sup>. Here we show that in the case of  $C_2$ , using the aug-cc-pVTZ basis provides a robust description of the dianionic reference, which delivers accurate results for the target states. To further validate these calculations, we carried out CAP-EOM-IP-CCSD calculations in which the dianionic reference is stabilized by the CAP and compare the potential energy curves of  $C_2^{2-}$  and  $C_2^-$  obtained by these two calculations.

In the CAP approach<sup>74,75</sup>, the Hamiltonian is augmented by a purely imaginary confining potential  $i\eta W$  (the parameter  $\eta$  controls the strength of the potential). This transformation converts the resonances into  $L^2$ -integrable wave functions with complex energies

$$E = E_{res} - \frac{i\Gamma}{2}, \quad (4)$$

where the real and imaginary parts correspond to the resonance position ( $E_{res}$ ) and width ( $\Gamma$ ). In a complete basis set, the exact resonance position and width can be recovered in the limit of  $\eta \rightarrow 0$ . In finite bases, the resonance can only be stabilized at finite values of  $\eta$ . The perturbation due to the finite-strength CAP can be removed by applying first-order de-perturbative correction<sup>53,54</sup> and identifying the special points of  $\eta$ -trajectories at which the dependence of the computed energy on  $\eta$  is minimal. When combined with the EOM-CCSD ansatz, this approach has been shown to yield accurate and internally consistent results for both bound and metastable states<sup>55</sup>. For example, these calculations yield smooth potential energy curves<sup>77–79</sup> and in many cases correctly identify the points where resonances become bound. We note, however, that in some polyatomic molecules spurious widths of about 0.04 eV for bound states persist<sup>79</sup>. In our previous calculations<sup>53,55,78–82</sup>, we used CAP-EOM-CCSD to describe metastable EOM states from stable (bound) CCSD references. Here we present the first example of a calculation where the CCSD reference is metastable, but the target EOM-CCSD states are bound.

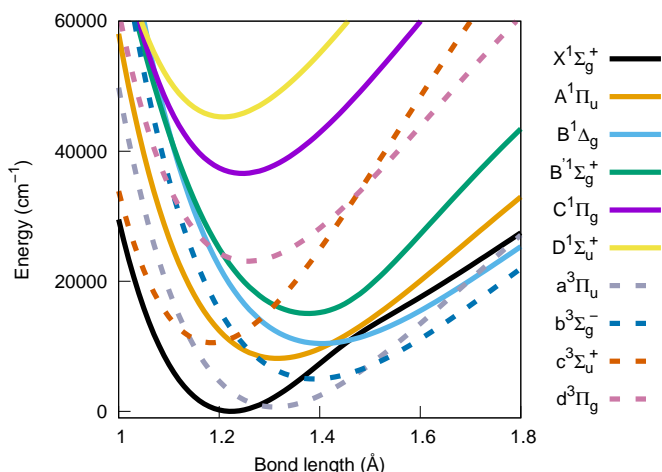
### 3 Computational details

As explained above, we describe the electronic states of  $C_2^-$  and  $C_2$  by EOM-IP-CCSD and EOM-DIP-CCSD, respectively, using the dianionic reference (see Fig. 2). In real-valued EOM-CCSD calculations, we used the aug-cc-pVTZ basis. In the CAP-augmented CCSD and EOM-IP-CCSD calculations, we used the aug-cc-pVTZ+3s3p and aug-cc-pCVTZ+6s6p6d basis sets (the exponents of the additional diffuse sets were generated using the same protocol as in our previous studies<sup>54,81</sup>). Two core orbitals,  $\sigma_{1s}$  and  $\sigma_{1s}^*$ , were frozen in correlated calculations except when employing the aug-cc-pCVTZ basis. In the calculations using aug-cc-pVTZ+3s3p, the CAP onset was set according to the expectation value of  $R^2$  of the triplet UHF wave function of  $C_2$  (at  $r_{cc}=1.28$  Å, the onsets were:  $x_0 = y_0 = 1.6$  Å,  $z_0 = 2.6$  Å). In the calculations with aug-cc-pCVTZ+6s6p6d, the CAP onset was set according to the expectation value of  $R^2$  of the dianion computed using CCSD/aug-cc-pCVTZ (at  $r_{cc}=1.2761$  Å, this gave  $x_0 = y_0 = 2.4$  Å,  $z_0 = 3.6$  Å). First-order correction<sup>53</sup> was applied to the computed total energy and then optimal values of  $\eta$  were

determined from these corrected trajectories using our standard protocol<sup>53,54</sup>. All electronic structure calculations were carried out using the Q-Chem package<sup>83,84</sup>. The calculations of partial widths were carried out using ezDyson<sup>85</sup>.

## 4 Results and discussion

### 4.1 C<sub>2</sub>



**Fig. 3** Potential energy curves of low-lying singlet and triplet states of C<sub>2</sub>.

Fig. 3 shows the potential energy curves of low-lying singlet and triplet states of C<sub>2</sub> computed using EOM-DIP-CCSD/aug-cc-pVTZ. The respective electronic configurations, equilibrium distances, and term values are summarized in Table 1. Table 1 also presents MR-CISD+Q/cc-pVTZ results from Ref. 34 and the experimental values. As one can see, C<sub>2</sub> features 10 electronic states within  $\sim 24,000$  cm<sup>-1</sup> (about 3 eV).

The results illustrate that EOM-DIP-CCSD is capable of tackling the complexity of C<sub>2</sub> rather well. It describes the entire manifold of the low-lying states with an accuracy comparable to that of much more cumbersome and labor-intensive multi-reference calculations. When compared to the experimental values, the root-mean-square (RMS) errors in the equilibrium bond lengths and term energies computed with EOM-DIP-CCSD/aug-cc-pVTZ are 0.0165 Å and 1661 cm<sup>-1</sup>. The errors in bond lengths are only marginally bigger than those of MR-CISD+Q/cc-pVTZ values (0.0114 Å). Remarkably, the errors in energy are consistently smaller than a conservative estimate of the EOM-CCSD error bars, which is roughly 0.3 eV (2420 cm<sup>-1</sup>). The relative state ordering is also correctly described. MR-CISD+Q/cc-pVTZ yields, on average, smaller errors in term energies (RMS of 469 cm<sup>-1</sup>), however, for three out of nine states, the EOM-DIP-CCSD/aug-cc-pVTZ values are closer to the experiment.

For a fair comparison, it is important to stress that the EOM-DIP-CCSD ansatz is very compact and includes only  $2h$  and  $3h2p$  configurations, whereas in MR-CISD+Q and AQCC, the size-extensivity corrections entail contributions of up to quadruply excited configurations. As with other EOM-CCSD methods, perturbative or explicit inclusion of connected triple excitations is expected to significantly reduce the errors. We note that higher

excitations can also describe orbital relaxation thus mitigating the effect of the unstable dianionic reference.

To put the results presented in Table 1 in a perspective, it is instructive to compare the performance of various flavors of multi-reference methods and to discuss the effects of basis set increase and higher-order corrections. Szalay *et al.* carried out<sup>34</sup> extensive comparisons between MR-CISD, MR-CISD+Q, and MR-AQCC for thirteen states of C<sub>2</sub>. The effects of higher-order corrections have also been investigated by Jiang and Wilson<sup>31</sup> in the framework of the correlation-consistent composite approach (MR-ccCA) based on the complete active space self-consistent field (CASSCF) theory with second-order perturbative corrections (CASPT2).

The size-extensivity correction is significant—the errors of MR-CISD decrease when either Davidson’s correction or MR-AQCC is employed. Without size-extensivity corrections, the RMS in the equilibrium bond lengths and term energies computed with MR-CISD/cc-pVTZ are 0.0117 Å and 623 cm<sup>-1</sup>. The effect of the basis set on the term energies is less systematic<sup>34</sup>. The RMS error in bond lengths within MR-AQCC/cc-pVTZ is 0.0115 Å (to be compared to 0.0114 Å of MR-CISD+Q). The errors in term energies were also comparable to MR-CISD+Q/cc-pVTZ. We note that in the MR-AQCC(TQ) calculations, the largest errors in term energies were observed for <sup>1</sup>Δ<sub>u</sub> and <sup>e</sup>³Π<sub>g</sub> (999 cm<sup>-1</sup> and 722 cm<sup>-1</sup>). Both MR-AQCC and MR-CISD+Q calculations were sensitive to the orbital choice and showed improved performance when using state-averaged CASSCF orbitals. Extrapolation to the complete basis set based on the cc-pVTZ and cc-pVQZ calculations results in a systematic decrease of equilibrium bond lengths by 0.01 Å.

Several studies have also investigated the magnitude of higher-order corrections, with an aim to achieve spectroscopic accuracy<sup>31,86</sup>. Schmidt and co-workers showed that the inclusion of core-valence correlation combined with scalar relativistic corrections in the framework of MR-CISD+Q brings the spectroscopic constants within 1% from the experimental values<sup>33</sup>. Jiang and Wilson have reported similar trends<sup>31</sup>.

In addition to the states shown in Table 1, we also computed two electronic states, <sup>1</sup>Δ<sub>u</sub> and <sup>e</sup>³Π<sub>g</sub>, which have been recently identified experimentally<sup>15–17</sup>. The electronic configurations of these states are: [core]<sup>6</sup>(σ<sub>2s</sub><sup>\*</sup>)<sup>2</sup>(π<sub>2p<sub>x</sub></sub>)<sup>2</sup>(π<sub>2p<sub>y</sub></sub>)<sup>1</sup>(π<sub>2p<sub>x</sub></sub><sup>\*</sup>)<sup>1</sup> and [core]<sup>6</sup>(σ<sub>2s</sub><sup>\*</sup>)<sup>2</sup>(π<sub>2p<sub>x</sub></sub>)<sup>1</sup>(π<sub>2p<sub>y</sub></sub>)<sup>1</sup>(σ<sub>2p<sub>z</sub></sub>)<sup>1</sup>(π<sub>2p<sub>x</sub></sub><sup>\*</sup>)<sup>1</sup>. Thus, they cannot be generated by the  $2h$  operator from the dianionic reference, so that the norm of the  $3h1p$  EOM amplitudes becomes large ( $\approx 1$ ). Consequently, the computed term energies are too high. In order to describe these states with the same accuracy as the states dominated by  $2h$  configurations, the EOM-DIP ansatz needs to be extended up to  $4h2p$  operators.

We note that several lowest state of C<sub>2</sub> can also be described by EOM-SF-CCSD using a high-spin triplet reference, e.g., [core]<sup>6</sup>(σ<sub>2s</sub><sup>\*</sup>)<sup>2</sup>(π<sub>2p<sub>x</sub></sub>)<sup>2</sup>(π<sub>2p<sub>y</sub></sub>)<sup>1</sup>(σ<sub>2p<sub>z</sub></sub>)<sup>1</sup>. Using ROHF-EOM-SF-CCSD/aug-cc-pVTZ, vertical excitation energy from <sup>1</sup>Σ<sub>g</sub><sup>+</sup> to a<sup>3</sup>Π<sub>u</sub> at  $r_{cc}=1.2425$  of C<sub>2</sub> is 319 cm<sup>-1</sup>, to be compared with 1924 cm<sup>-1</sup> computed by EOM-DIP-CCSD/aug-cc-pVTZ. To quantify the bonding pattern in C<sub>2</sub>, we also computed Head-Gordon’s index<sup>87</sup>, which characterizes the number of effectively unpaired electrons. For the EOM-SF-CCSD wave function of the ground state of C<sub>2</sub> at



**Table 1** Equilibrium bond lengths ( $r_e$ , Å) and term energies ( $T_{ee}$ , cm<sup>-1</sup>) of the low-lying states of C<sub>2</sub>.

State	Configuration	EOM-DIP-CCSD <sup>a</sup>		MR-CISD+Q <sup>b</sup>		Expt. <sup>c</sup>	
		$r_e$	$T_{ee}$	$r_e$	$T_{ee}$	$r_e$	$T_{ee}$
$X^1\Sigma_g^+$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>2</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>2</sup>	1.224	-	1.2536	-	1.2425	-
$A^1\Pi_u$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>2</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>1</sup> ( $\sigma_{2p_z}$ ) <sup>1</sup>	1.316	8127	1.3294	8000	1.3184	8391
$B^1\Delta_g$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>2</sup> ( $\pi_{2p_x}$ ) <sup>1</sup> ( $\pi_{2p_y}$ ) <sup>1</sup> ( $\sigma_{2p_z}$ ) <sup>2</sup>	1.404	10408	1.3972	11684	1.3855	12082
$B'^1\Sigma_g^+$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>2</sup> ( $\pi_{2p_x}$ ) <sup>1</sup> ( $\pi_{2p_y}$ ) <sup>1</sup> ( $\sigma_{2p_z}$ ) <sup>2</sup>	1.377	15012	1.3897	15134	1.3774	15409
$C^1\Pi_g$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>1</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>1</sup> ( $\sigma_{2p_z}$ ) <sup>2</sup>	1.246	36489	1.2682	34788	1.2552	34261
$D^1\Sigma_u^+$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>1</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>2</sup> ( $\sigma_{2p_z}$ ) <sup>1</sup>	1.208	45166	1.2521	43810	1.2380	43239
$a^3\Pi_u$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>2</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>1</sup> ( $\sigma_{2p_z}$ ) <sup>1</sup>	1.310	694	1.3228	256	1.3119	716
$b^3\Sigma_g^-$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>2</sup> ( $\pi_{2p_x}$ ) <sup>1</sup> ( $\pi_{2p_y}$ ) <sup>1</sup> ( $\sigma_{2p_z}$ ) <sup>2</sup>	1.390	4971	1.3786	5794	1.3692	6434
$c^3\Sigma_u^+$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>1</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>2</sup> ( $\sigma_{2p_z}$ ) <sup>1</sup>	1.185	10531	1.2170	9618	1.2090	9124
$d^3\Pi_g$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>1</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>1</sup> ( $\sigma_{2p_z}$ ) <sup>2</sup>	1.258	23025	1.2777	20382	1.2661	20022

<sup>a</sup> aug-cc-pVTZ basis set.<sup>b</sup> MR-CISD with Davidson correction using the cc-pVTZ basis set from Ref. 34.<sup>c</sup> From Refs. 9–14.

equilibrium,  $n_{u,nl}=0.29$ . This value indicates that C<sub>2</sub> has substantial diradical character, comparable<sup>63</sup> to that of singlet methylene (0.25) or *meta*-benzynes (0.26). In other words, there is no support for a quadruple bond, which would be manifested by  $n_{u,nl} \approx 0$ .

## 4.2 C<sub>2</sub><sup>-</sup>

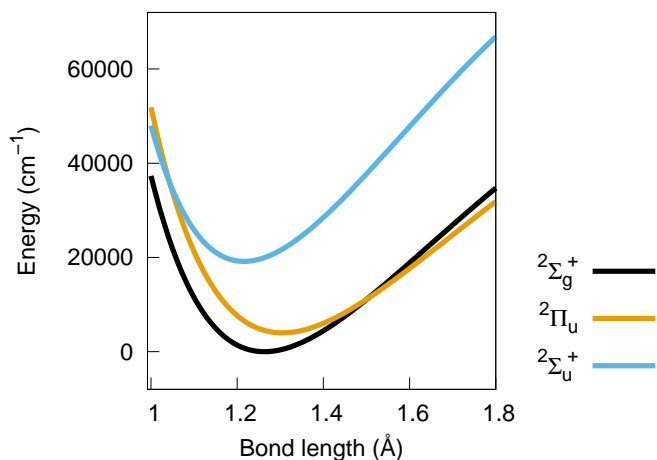
**Fig. 4** Potential energy curves of the three lowest states of C<sub>2</sub><sup>-</sup>.

Fig. 4 shows the potential energy curves of the three bound states of C<sub>2</sub><sup>-</sup> computed using EOM-IP-CCSD/aug-cc-pVTZ. The respective electronic configurations, equilibrium distances, and term values are given in Table 2. The Dyson orbitals<sup>89</sup> representing the unpaired electrons in C<sub>2</sub><sup>-</sup> are shown in Fig. 1.

As one can see, the computed equilibrium distances and term energies are in excellent agreement with the experimental data. The computed oscillator strengths show that transitions to both excited states are optically allowed. The computed  $T_{ee}$  of the  $^2\Sigma_u^+ \rightarrow ^2\Sigma_g^+$  transition is 2.37 eV. Vertically, at the equilibrium geometry of the  $^2\Sigma_u^+$  state, the energy gap between two states is 2.29 eV, which is exactly equal to the fluorescence signal observed in

Ref. 21. Thus, our results confirm that fluorescence observed in Ref. 21 can be attributed to the  $^2\Sigma_u^+ \rightarrow ^2\Sigma_g^+$  transition of C<sub>2</sub><sup>-</sup>.

We also computed AEA of C<sub>2</sub>. Using EOM-DIP-CCSD/aug-cc-pVTZ total energy of  $X^1\Sigma_g^+$  and EOM-IP-CCSD/aug-cc-pVTZ total energy of the  $X^2\Sigma_g^+$  state at the respective  $r_e$ , the computed value of AEA is 4.57 eV (without zero-point energy), which is more than 1 eV larger than the experimental value<sup>38,39</sup> of 3.27 eV and high-level *ab initio* estimates<sup>40</sup>. This suggests that the current correlation level is insufficient to describe relative position of the two manifolds. The two relevant states,  $X^1\Sigma_g^+$  and  $^2\Sigma_g^+$ , can also be computed using an alternative EOM-CC scheme, via SF and EA using the high-spin triplet reference, [core]<sup>6</sup>( $\sigma_{2s}^*$ )<sup>2</sup>( $\pi_{2p_x}$ )<sup>2</sup>( $\pi_{2p_y}$ )<sup>1</sup>( $\sigma_{2p_z}$ )<sup>1</sup>. These calculations yield AEA of 3.44 eV when using UHF triplet reference and 3.42 eV when using the ROHF reference. The analysis of the total energies shows that the EOM-EA energy of the anion is very close to the corresponding EOM-IP energy whereas the EOM-SF energy of the neutral state is significantly lower than the EOM-DIP energy. We attribute this to orbital relaxation effects—while the dianionic orbitals are reasonably good for the anion, they are too diffuse for the neutral and the EOM-DIP ansatz with only  $2h$  and  $3h1p$  operators is not sufficiently flexible to account for that.

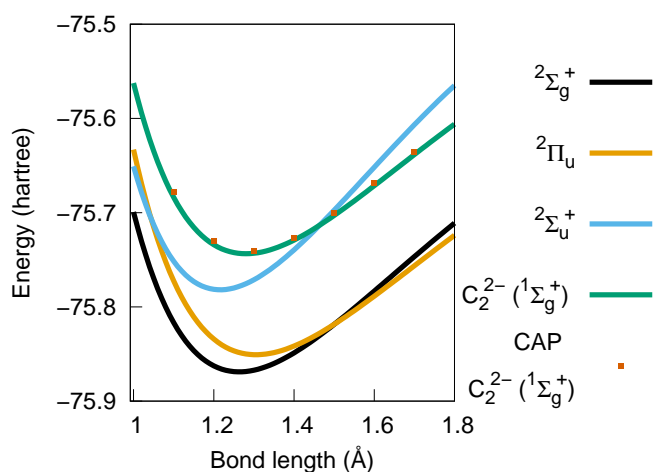
## 4.3 C<sub>2</sub><sup>2-</sup>

Fig. 5, which shows potential energy curves of C<sub>2</sub><sup>2-</sup> and C<sub>2</sub><sup>-</sup>, clearly illustrates the metastable nature of C<sub>2</sub><sup>2-</sup>. Adiabatically, C<sub>2</sub><sup>2-</sup> is 3.41 eV (at the EOM-IP-CCSD/aug-cc-pVTZ level) above the ground state of C<sub>2</sub><sup>-</sup> and can decay into any of the 3 states of the anion. The squared norms of the respective Dyson orbitals<sup>89</sup> computed using the EOM-IP-CCSD/aug-cc-pVTZ wave functions at the equilibrium bondlength of C<sub>2</sub><sup>2-</sup> (1.28 Å) are 0.86, 0.80, and 0.86 for the  $^2\Sigma_g^+$ ,  $^2\Pi_u$ , and  $^2\Sigma_u^+$  states, respectively. These values indicate that each of these channels corresponds to a one-electron detachment process. In the case of the autoionization, the shape of the Dyson orbital represents the state of the outgoing electron. Thus, the lowest channel ( $^2\Sigma_g$ ) corresponds to a *d*-wave whereas the two other channels correspond to *p*-waves. This qualitative

**Table 2** Equilibrium bond lengths ( $r_e$ , Å) and term energies ( $T_{ee}$ , cm<sup>-1</sup>) of bound electronic states of C<sub>2</sub><sup>-</sup>. EOM-IP-CCSD vertical excitation energies ( $E_{ex}$ , cm<sup>-1</sup>) and oscillator strengths ( $f_l$ ) are also shown.

State	Configuration	EOM-IP-CCSD/aug-cc-pVTZ				Expt. <sup>a</sup>	
		$r_e$	$T_{ee}$	$E_{ex}$	$f_l$	$r_e$	$T_{ee}$
$2\Sigma_g^+$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>2</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>2</sup> ( $\sigma_{2p_z}$ ) <sup>1</sup>	1.260				1.268	
$2\Pi_u$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>2</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>1</sup> ( $\sigma_{2p_z}$ ) <sup>2</sup>	1.310	3989	4575	0.004	1.308	3986
$2\Sigma_u^+$	[core] <sup>6</sup> ( $\sigma_{2s}^*$ ) <sup>1</sup> ( $\pi_{2p_x}$ ) <sup>2</sup> ( $\pi_{2p_y}$ ) <sup>2</sup> ( $\sigma_{2p_z}$ ) <sup>2</sup>	1.219	19113	19801	0.085	1.223	18391

<sup>a</sup> From Ref. 88.



**Fig. 5** Potential energy curves of C<sub>2</sub><sup>-</sup> and C<sub>2</sub><sup>-</sup>. Total electronic energies are shown. Solid lines show CCSD/aug-cc-pVTZ and EOM-IP-CCSD energies. Orange squares show the results from CAP-CCSD/aug-cc-pVTZ+3s3p (first-order corrected energy).

analysis is supported by the calculations of partial waves using *ezDyson*.

**Table 3** First-order corrected energies of C<sub>2</sub><sup>-</sup> at optimal values of the  $\eta$  parameter and the corresponding trajectory velocities (in a.u.) computed with CAP-CCSD/aug-cc-pVTZ+3s3p.

$r_{CC}/\text{Å}$	$E^{Re}$	$\Gamma$	$\eta_{opt}$	$\eta _{\frac{dE}{d\eta}}$
1.1	-75.67858	0.02790	0.0176	$8.324 \times 10^{-5}$
1.2	-75.73037	0.02618	0.0164	$2.061 \times 10^{-5}$
1.28	-75.74059	0.02506	0.0156	$1.250 \times 10^{-4}$
1.3	-75.74022	0.02458	0.0148	$2.355 \times 10^{-4}$
1.4	-75.72646	0.02346	0.0140	$2.016 \times 10^{-4}$
1.5	-75.70046	0.02302	0.0128	$1.104 \times 10^{-4}$
1.6	-75.66865	0.02224	0.0120	$1.827 \times 10^{-4}$
1.7	-75.63523	0.02172	0.0116	$1.427 \times 10^{-4}$

To characterize lifetimes of the dianion and to quantify the effect of its resonance character on the computed quantities of C<sub>2</sub><sup>-</sup>, we carried out CAP-CCSD and CAP-EOM-IP-CCSD calculations. The results are summarized in Tables 3 and 4 and shown in Figs. 5 and 6.

As one can see from Fig. 5, the total energies of C<sub>2</sub><sup>-</sup> obtained from the CAP-augmented calculations are nearly identical to the real-valued results. Moreover, the impact on the computed term

energies of C<sub>2</sub><sup>-</sup> is also small: at  $r_{CC}=1.28$  Å, the differences in excitation energies of C<sub>2</sub><sup>-</sup> between the two calculations are  $\sim 0.03$  eV. The adiabatic energy gap between C<sub>2</sub><sup>-</sup> and C<sub>2</sub><sup>-</sup> is 3.16 eV computed with CAP-CCSD/aug-cc-pCVTZ+6s6p6d, only slightly smaller than the value obtained in real-valued calculations (3.41 eV).

Previous calculations using the charge-stabilization method<sup>43</sup> estimated that the closed-shell  $1\Sigma_g^+$  resonance of C<sub>2</sub><sup>-</sup> lies below 4 eV, roughly around 3.4 eV, above the ground state of C<sub>2</sub><sup>-</sup>. Later, CAP-augmented MR-CISD calculations<sup>44</sup> yielded  $E_{res}=3.52$  eV and  $r_e=1.285$  Å. Thus, our results confirm the findings of these earlier studies<sup>43,44</sup>.

The resonance position and width are rather sensitive to the basis set employed, as Table 3 illustrates. For example, at the equilibrium bond length ( $r_{CC}=1.28$  Å), the aug-cc-pVTZ+3s3p basis yields adiabatic  $E_{res}=3.7$  eV and  $\Gamma=0.68$  eV, whereas the aug-cc-pCVTZ+6s6p6d basis produces  $E_{res}=3.16$  eV and  $\Gamma=0.25$  eV. A distinct stabilization point of the  $\eta$ -trajectory is only obtained using the larger basis set (see Fig. 6); in the small basis only first-order corrected trajectory shows the stabilization point. Our best value for the resonance width (0.25 eV) is in very good agreement with the CAP-MR-CISD value (0.30 eV)<sup>44</sup> and also agrees qualitatively with the estimate from charge-stabilization calculations (0.26-0.55 eV)<sup>43</sup>. Compared to singlet resonances with open-shell character, for example, those of CN<sup>-</sup> that have  $\Gamma=0.48$ -0.56 eV<sup>81</sup>, C<sub>2</sub><sup>-</sup> is a narrow resonance. On the other hand, C<sub>2</sub><sup>-</sup> resonance is rather broad compared to other small dianions<sup>69</sup>, such as CO<sub>3</sub><sup>2-</sup> or SO<sub>4</sub><sup>2-</sup>.

We also estimated partial widths corresponding to the three decay channels. Within the Feshbach formalism, partial widths of autodetachment can be approximated by the following matrix element<sup>82</sup>:

$$\Gamma_c = \left( 2\pi \langle \xi_{\omega_c} | \hat{F} | \phi_c^d \rangle \right)^2, \quad (5)$$

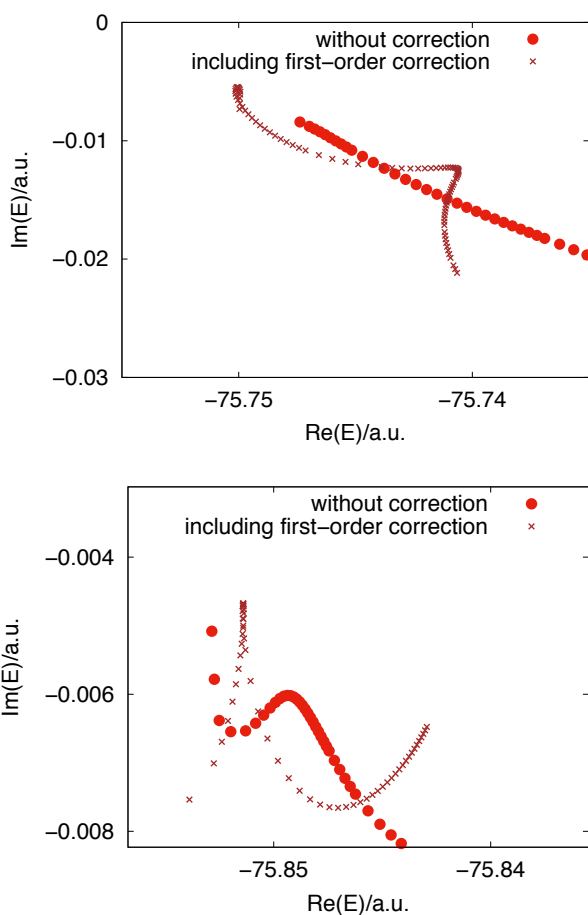
where  $\Gamma_c$  is the partial width corresponding to detachment channel  $c$ ,  $\omega_c$  and  $\phi_c^d$  are the respective detachment energy and Dyson orbital,  $\xi_{\omega_c}$  is the wave function of the free electron, and  $\hat{F}$  is the Fock operator. Given the localized nature of  $\hat{F}$ , this matrix element is bound by the value of the overlap between the Dyson orbital and the free-electron wave function. Thus, branching ratios  $x_p$  corresponding to different detachment channels can be estimated as follows:

$$x_p = \frac{\langle \xi_{\omega_p} | \phi_p^d \rangle^2}{\sum_c \langle \xi_{\omega_c} | \phi_c^d \rangle^2}, \quad (6)$$

giving rise to  $\Gamma_p = x_p \Gamma$ . Note that the contributions from the

**Table 4** First-order corrected energies of  $C_2^-$  at optimal values of the  $\eta$  parameter computed with CAP-CCSD/aug-cc-pCVTZ+6s6p6d and CAP-HF/aug-cc-pCVTZ+6s6p6d.

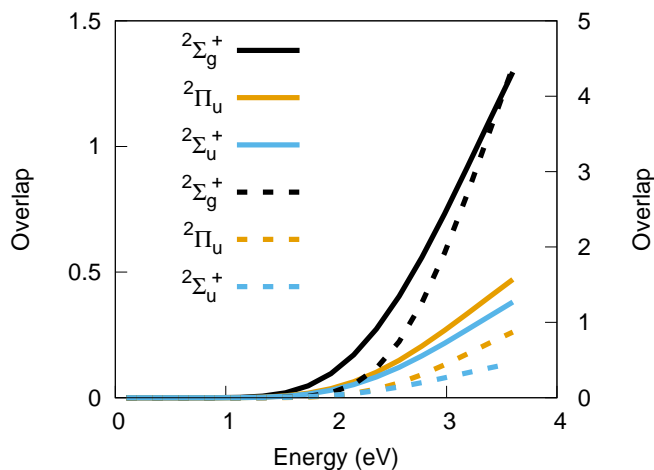
$r_{CC}/\text{\AA}$	CAP-CCSD			CAP-HF		
	$E^{Re}$	$\Gamma$	$\eta_{opt}$	$E^{Re}$	$\Gamma$	$\eta_{opt}$
1.0372	-75.730907	0.014406	0.0030	-75.298280	0.004573	0.0028
1.0901	-75.786473	0.012892	0.0030	-75.353147	0.003828	0.0026
1.1430	-75.821827	0.011584	0.0030	-75.387436	0.003267	0.0024
1.1959	-75.841880	0.010590	0.0028	-75.406232	0.002968	0.0020
1.2489	-75.850425	0.009711	0.0028	-75.413353	0.002924	0.0018
1.2761	-75.851410	0.009350	0.0028	-75.413483	0.003025	0.0018
1.3018	-75.850542	0.009089	0.0026	-75.411820	0.003178	0.0018
1.3547	-75.844523	0.008589	0.0026	-75.403959	0.003601	0.0018
1.4076	-75.834038	0.008303	0.0024	-75.391521	0.004147	0.0018
1.4605	-75.820500	0.007975	0.0024	-75.375802	0.004719	0.0018
1.5134	-75.804992	0.007837	0.0024	-75.357905	0.005367	0.0018
1.5664	-75.788195	0.007646	0.0024	-75.338755	0.005815	0.0020
1.6193	-75.770789	0.007618	0.0024	-75.318637	0.006414	0.0020
1.6722	-75.753150	0.007604	0.0022	-75.298126	0.007005	0.0020
1.7251	-75.735635	0.007578	0.0022	-75.277573	0.007564	0.0020
1.7780	-75.718455	0.007528	0.0022	-75.257580	0.007895	0.0020
1.8310	-75.701660	0.007486	0.0020	-75.237545	0.008336	0.0020



**Fig. 6** Uncorrected and first-order corrected CAP-CCSD using aug-cc-pCVTZ+3s3p (top) and aug-cc-pCVTZ+6s6p6d (bottom)  $\eta$ -trajectories for  $C_2^-$  at equilibrium bondlengths.

degenerate channels (such as  $\Pi_u$ ) should be multiplied by the respective degeneracy number (2 for  $\Pi$ -states). The overlap  $\langle \xi_{\omega_p} | \phi_p^d \rangle^2$  is proportional to the norm of  $\phi_c^d$  and depends strongly on the energy of the detached electron and the shape of the Dyson

orbital. Fig. 7 shows the energy dependence of the computed values of the squared overlap between the normalized Dyson orbitals and the free-electron wave function approximated by the Coulomb wave. As one can see, the overlap values are zero at low detachment energies and increase at higher energies. The trends for the  $\Sigma_u$  and  $\Pi_u$  channels are very similar, which is not surprising given the similar shape of the respective Dyson orbitals. Fig. 7 immediately suggest that the autodetachment process will be dominated by the channels producing the two lowest states of the anion,  $\Sigma_g$  and  $\Pi_u$ .



**Fig. 7** Squared overlap between Dyson orbitals and a Coulomb wave with charge=-1. Solid lines correspond to Dyson orbitals from EOM-IP-CCSD/aug-cc-pVTZ (scale on the left). Dashed lines correspond to Dyson orbitals (real part) from CAP-EOM-IP-CCSD/aug-cc-pVTZ+6s6p6d (scale on the right).

Table 5 lists the computed values using  $E_{res}=3.41$  eV (from EOM-IP-CCSD/aug-cc-pVTZ). As one can see, the contribution of the  $\Sigma_u$  is negligible and the  $\Sigma_g$  channel is dominant. When using lower energy value (3.16 eV, from CAP-EOM-IP-CCSD/aug-cc-pCVTZ+6s6p6d), the contribution from the  $\Sigma_u$  channels



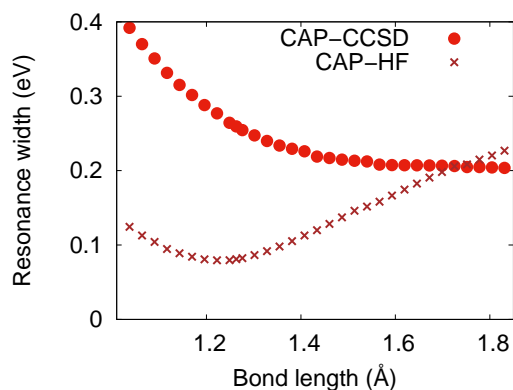
**Table 5** Calculation of partial widths using Coulomb wave and Dyson orbitals from real-valued and complex-valued EOM-IP-CCSD calculations.

Channel/DE <sup>a</sup>	$\ \phi_d\ ^2$	EOM-IP-CCSD/aug-cc-pVTZ			CAP-EOM-IP-CCSD/aug-cc-pVTZ+6s6p6d			
		Overlap <sup>b</sup>	$x_p$	$\Gamma_p$	$\ \phi_d\ ^2$	Overlap <sup>b</sup>	$x_p$	$\Gamma_p$
$^2\Sigma_g^+/3.41$	0.86	1.12	0.69	0.17	0.73	3.61	0.83	0.21
$^2\Pi_u/2.91$	0.86	0.25	0.31	0.08	0.71	0.39	0.17	0.04
$^2\Sigma_u^+/1.04$	0.80	$2.13 \times 10^{-4}$	$1.4 \times 10^{-4}$	$\sim 0$	0.62	$1.6 \times 10^{-4}$	$\sim 0$	$\sim 0$

<sup>a</sup> Adiabatic EOM-IP-CCSD/aug-cc-pVTZ energies (eV).

<sup>b</sup> Overlap (squared) is computed between normalized Dyson orbitals and the Coulomb wave with charge=-1 and kinetic energy corresponding to adiabatic detachment energy.

drops even further while the ratio between the  $\Sigma_g$  and  $\Pi_u$  channels remains unchanged. Using Dyson orbitals from the CAP-EOM-IP-CCSD/aug-cc-pVTZ+6s6p6d calculations leads to the increase of the relative weight of the  $\Sigma_g$  channel. These simple estimates are in qualitative agreement with partial widths computed using CAP-MR-CISD wave function and an approach based on the CAP projection<sup>44</sup>; their reported values correspond to  $x_p$  of 0.31, 0.66, and 0.02 for the  $\Sigma_g$ ,  $\Pi_u$ , and  $\Sigma_u$  channels. One important difference is that our calculations predict that the dominant decay channel is  $\Sigma_g$ , producing the ground-state of  $C_2^-$ . We note that using plane wave to describe the state of the free electron yields an entirely different picture: the overlaps are rather large around the threshold and change much slower, resulting in comparable branching ratios for all three channels.



**Fig. 8** First-order corrected resonance width of  $C_2^-$  as a function of bond length computed with CAP-CCSD and CAP-HF and the aug-cc-pVTZ+6s6p6d basis set.

Finally, we investigate the dependence of the resonance width on the bond length. As illustrated by Figure 8, the CAP-CCSD resonance width shrinks with an increasing bond length near the equilibrium distance while it is nearly constant beyond 1.6 Å. This is consistent with the potential energy curves of  $C_2^-$  and the  $^2\Sigma_g^+$  and  $^2\Pi_u$  states of  $C_2^-$  becoming nearly parallel at elongated bond distances (see Figure 5). However, the behavior is different from that of valence shape resonances in diatomic molecules (for example,  $H_2^-$  or  $N_2^-$ ) that become bound when the bond is stretched somewhat.<sup>55</sup> It is more reminiscent of dipole-stabilized resonances whose width is also rather insensitive to the changes of the bond lengths.<sup>80</sup> Figure 8 also shows that the resonance

width behaves differently at the CAP-CCSD and CAP-HF levels. Within the HF approximation,  $\Gamma$  has a minimum around the equilibrium structure (0.08 eV) and grows when the bond is stretched. This behavior is similar to the results reported in Ref. 44 where CAP-CIS and CAP-MR-CISD also yielded  $\Gamma$  increasing with bond length between 1.2 and 1.4 Å. A detailed investigation of these differences is beyond the scope of the present work, but we note that the resonance width of  $C_2^-$  has to vanish eventually, when the bond is stretched far enough, because the  $^4S$  ground state of  $C^-$  obtained in the dissociation limit is stable towards electron detachment.

The description of the decay channels reveals a shortcoming of the CAP-CCSD approach based on a metastable reference. The CAP-EOM-IP-CCSD energies of the three bound states of  $C_2^-$  feature sizable positive imaginary parts of more than 0.3 eV (at the equilibrium bond length and optimal  $\eta$  values for the dianionic resonance). This is despite that real parts of absolute CAP-EOM-IP-CCSD energies agree with the CAP-free values within  $\sim 0.1$  eV. Also, it is in stark contrast to the performance of CAP-EOM-CCSD based on bound reference states<sup>53,54</sup>, where the imaginary energies of bound states typically stay below 0.03 eV. We note that application of the de-perturbative correction<sup>53,54</sup> does not rectify this problem. This is not surprising as the original analysis of  $E(\eta)$  in terms of perturbation theory<sup>75</sup> was designed for resonances but not bound states. Furthermore, the imaginary energies of the three bound states of  $C_2^-$  differ by more than a factor of two so that a single, not state-specific, correction is not realistic. However, since a positive imaginary energy is unphysical and since no stabilization of the  $\eta$ -trajectory is observed for the CAP-EOM-IP-CCSD states, the problem is easily discernible. Importantly, despite this shortcoming, CAP-EOM-CCSD calculations using an unstable reference clearly distinguish bound and metastable states.

Experimentally<sup>41,42</sup>, the  $C_2^-$  resonance manifest itself as a broad feature around 10 eV in electron scattering detachment spectra from  $C_2^-$ , however, the interpretation of these spectra in terms of the position of the resonance is not straightforward, as explained by Sommerfeld and co-workers<sup>43</sup>. We hope that our results will stimulate further experimental efforts to characterize electronic structure of  $C_2^-$ .

## 5 Conclusion

We reported electronic structure calculations of  $C_2$ ,  $C_2^-$ , and  $C_2^{2-}$  using the CC/EOM-CC family of methods. Our results illustrate that EOM-CCSD provides an attractive alternative to MR ap-

proaches. The low-lying states of  $C_2$  and  $C_2^-$  are well described by EOM-DIP-CCSD and EOM-IP-CCSD using the dianionic closed-shell reference ( $C_2^{2-}$ ), despite its metastable nature.

EOM-DIP-CCSD offers a much simpler computational approach based on a single-reference formalism. In the EOM-DIP calculations, no active-space selection is required, and the results of the calculations do not depend on the number of states computed, in contrast to state-averaged MR schemes. One does not need to guess the electronic configurations of the states to be computed—once the user specifies how many states in each irrep are desired, the algorithm computes them.

The electronic structure of  $C_2^{2-}$  was characterized by CAP-augmented CCSD. The calculations placed the closed-shell  $C_2^{2-}$  resonance 3.16 eV adiabatically above the ground state of  $C_2^-$ . The computed resonance width is 0.25 eV, corresponding to a lifetime of 2.6 fs. The  $C_2^{2-}$  resonance can, in principle, decay into three open channels, producing the ground ( $X^2\Sigma_g^-$ ) or an excited ( $^2\Pi_u$  or  $^2\Sigma_u$ ) state of  $C_2^-$ . Our calculations of the partial widths suggest that the dominant decay channel (70-80%) corresponds to the ground state of the anion while the  $^2\Sigma_u$  channel is essentially blocked. The analysis of the respective Dyson orbitals reveals that the main effect controlling the branching ratios in this system is the energy of the outgoing electrons. Importantly, the CAP-augmented calculations yield detachment energies that are very close to the real-valued EOM-CCSD calculations with the aug-cc-pVTZ basis set, thus confirming the validity of the results obtained with EOM-DIP-CCSD and EOM-IP-CCSD using the dianion reference.

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## Notes and references

- 1 S. S. H. S. Rzepa and R. Hoffmann, *Angew. Chem. Int. Ed.*, 2013, **52**, 3020–3033.
- 2 L. T. Xu and T. H. Dunning, Jr., *J. Chem. Theory Comput.*, 2014, **10**, 195–201.
- 3 S. Shaik, D. Danovich, B. Braida and P. C. Hiberty, *Chem. Eur. J.*, 2016, **22**, 18977–18980.
- 4 G. Frenking and M. Hermann, *Chem. Eur. J.*, 2016, **22**, 18975–18976.
- 5 S. Shaik, D. Danovich, B. Braida and P. C. Hiberty, *Chem. Eur. J.*, 2016, **22**, 18977–18980.
- 6 A. L. Tchougreeff and R. Dronskowski, *Mol. Phys.*, 2016, **114**, 1423–1444.
- 7 D. W. O. de Sousa and M. A. C. Nascimento, *J. Chem. Theory Comput.*, 2016, **12**, 2234–2241.
- 8 M. Piris, X. Lopez and J. M. Ugalde, *Chem. Eur. J.*, 2016, **22**, 4109–4115.
- 9 K. Huber and G. Herzberg, *Constants of Diatomic Molecules (data prepared by J.W. Gallagher and R.D. Johnson, III)*, NIST Chemistry WebBook, NIST Standard Reference Database Number 69, Eds. P.J. Linstrom and W.G. Mallard, July 2001, National Institute of Standards and Technology, Gaithersburg MD, 20899 (<http://webbook.nist.gov>).
- 10 M. Douay, R. Nietmann and P. F. Bernath, *J. Molec. Spect.*, 1988, **131**, 261–271.
- 11 P. M. Goodwin and T. A. Cool, *J. Chem. Phys.*, 1988, **88**, 4548–4549.
- 12 P. S. Davis, J. G. P. M. C. Abrams and M. L. P. Rao, *JOSA B*, 1988, **5**, 2280–2285.
- 13 V. E. Bondybey, *J. Chem. Phys.*, 1976, **65**, 2296–2304.
- 14 J. Chauville, J. P. Maillard and A. W. Mantz, *J. Molec. Spect.*, 1977, **68**, 399–411.
- 15 M. Nakajima, J. A. Joester, N. I. Page, N. J. Reilly, G. B. Bacskay, T. W. Schmidt and S. H. Kable, *J. Chem. Phys.*, 2009, **131**, 044301.
- 16 O. Krechkivska, G. B. Bacskay, T. P. Troy, K. Nauta, T. D. Kreuzer, S. H. Kable and T. W. Schmidt, *J. Phys. Chem. A*, 2015, **119**, 12102–12108.
- 17 O. Krechkivska, B. A. Welsh, G. B. Bacskay, K. Nauta, S. H. Kable and T. W. Schmidt, *J. Chem. Phys.*, 2017, **146**, 134306.
- 18 W. H. A. Wollaston, *Philos. Trans. R. Soc. London*, 1802, **92**, 365–380.
- 19 A. N. Goyette, J. E. Lawler, L. W. Anderson, D. M. Gruen, T. G. McCauley, D. Zhou and A. R. Krauss, *J. Phys. D: Appl. Phys.*, 1998, **31**, 1975.
- 20 N. J. Reilly, T. W. Schmidt and S. H. Kable, *J. Phys. Chem. A*, 2006, **110**, 12355.
- 21 E. J. D. Mahoney, B. S. Truscott, M. N. R. Ashfold and Y. A. Mankelevich, *J. Phys. Chem. A*, 2017, **121**, 2760–2772.
- 22 E. Picazzio, A. A. de Almeida, S. M. Andrievskii, K. I. Churyumov and I. V. Luk'yanyk, *Adv. Space Res.*, 2007, **39**, 462–467.
- 23 W. Huggins, *Astron. Nachr.*, 1868, **71**, 382–384.
- 24 D. Crampton, A. P. Cowley and R. M. Humphreys, *Astrophys. J.*, 1975, **198**, L135–L137.
- 25 L. M. Hobbs, *Astrophys. J.*, 1979, **232**, L175–L177.
- 26 J. H. Goebel, J. D. Bregman, D. M. Cooper, D. Goorvitch, S. R. Langhoff and F. C. Witteborn, *Astrophys. J.*, 1983, **270**, 190–199.
- 27 V. G. Klochkova, R. Szczerba and V. E. Panchuk, *Astronomy Lett.*, 2000, **26**, 88–103.
- 28 J. D. Watts and R. J. Bartlett, *J. Chem. Phys.*, 1992, **96**, 6073–6084.
- 29 X. Li and J. Paldus, *Chem. Phys. Lett.*, 2006, **431**, 179–184.
- 30 D. Shi, X. Zhang, J. Sun and Z. Zhu, *Mol. Phys.*, 2011, **109**, 1453–1465.
- 31 W. Jiang and A. K. Wilson, *J. Chem. Phys.*, 2011, **134**, 034101.
- 32 O. Christiansen, H. Koch, P. Jorgensen and J. Olsen, *Chem. Phys. Lett.*, 1996, **256**, 185–194.
- 33 D. L. Kokkin, G. B. Bacskay and T. W. Schmidt, *J. Chem. Phys.*, 2007, **126**, 084302.
- 34 T. Muller, M. Dallos, H. Lischka, Z. Dubrovay and P. G. Szalay, *Theor. Chem. Acc.*, 2001, **105**, 227–243.

- 35 B. D. Rehffuss, D. J. Liu, B. M. Dinelli, M. F. Jagod, W. C. Ho, M. W. Crofton and T. Oka, *J. Chem. Phys.*, 1988, **89**, 129–137.
- 36 M. Tulej, G. Knopp, T. Gerber and P. P. Radi, *J. Raman Spect.*, 2010, **41**, 853–858.
- 37 S. Gerber, J. Fescl, M. Doser and D. Comparat, *New J. Phys.*, 2018, **20**, 023024.
- 38 K. M. Ervin and W. C. Lineberger, *J. Phys. Chem.*, 1991, **95**, 1167–1177.
- 39 D. W. Arnold, S. E. Bradforth, T. N. Kitsopoulos and D. M. Neumark, *J. Chem. Phys.*, 1991, **95**, 8753–8764.
- 40 D. Feller, *J. Chem. Phys.*, 2016, **144**, 014105.
- 41 L. H. Andersen, P. Hvelplund, D. Kella, P. Mokler, H. B. Pedersen, H. T. Schmidt and L. Vejby-Christensen, *J. Phys. B*, 1996, L643.
- 42 H. B. Pedersen, N. Djurić, M. J. Jensen, D. Kella, C. P. Saffvan, L. Vejby-Christensen and L. H. Andersen, *Phys. Rev. Lett.*, 1998, **81**, 5302–5305.
- 43 T. Sommerfeld, U. V. Riss, H.-D. Meyer and L. S. Cederbaum, *Phys. Rev. Lett.*, 1997, **79**, 1237–1240.
- 44 T. Sommerfeld, F. Tarantelli, H.-D. Meyer and L. S. Cederbaum, *J. Chem. Phys.*, 2000, **112**, 6635–6642.
- 45 J. F. Stanton and J. Gauss, *Adv. Chem. Phys.*, 2003, **125**, 101–146.
- 46 A. I. Krylov, *Annu. Rev. Phys. Chem.*, 2008, **59**, 433–462.
- 47 K. Sneskov and O. Christiansen, *WIREs Comput. Mol. Sci.*, 2012, **2**, 566–584.
- 48 R. J. Bartlett, *WIREs Comput. Mol. Sci.*, 2012, **2**, 126–138.
- 49 A. I. Krylov, *Reviews in Comp. Chem.*, J. Wiley & Sons, 2017, vol. 30, ch. 4, pp. 151–224.
- 50 K. W. Sattelmeyer, H. F. Schaefer and J. F. Stanton, *Chem. Phys. Lett.*, 2003, **378**, 42–46.
- 51 J. F. Stanton and J. Gauss, *J. Chem. Phys.*, 1994, **101**, 8938–8944.
- 52 P. A. Pieniazek, S. E. Bradforth and A. I. Krylov, *J. Chem. Phys.*, 2008, **129**, 074104.
- 53 T.-C. Jagau, D. Zuev, K. B. Bravaya, E. Epifanovsky and A. I. Krylov, *J. Phys. Chem. Lett.*, 2014, **5**, 310–315.
- 54 D. Zuev, T.-C. Jagau, K. B. Bravaya, E. Epifanovsky, Y. Shao, E. Sundstrom, M. Head-Gordon and A. I. Krylov, *J. Chem. Phys.*, 2014, **141**, 024102.
- 55 T.-C. Jagau, K. B. Bravaya and A. I. Krylov, *Annu. Rev. Phys. Chem.*, 2017, **68**, 525–553.
- 56 J. F. Stanton and R. J. Bartlett, *J. Chem. Phys.*, 1993, **98**, 7029–7039.
- 57 A. I. Krylov, *Chem. Phys. Lett.*, 2001, **338**, 375–384.
- 58 A. I. Krylov, *Acc. Chem. Res.*, 2006, **39**, 83–91.
- 59 M. Nooijen and R. J. Bartlett, *J. Chem. Phys.*, 1995, **102**, 3629–3647.
- 60 N. Orms and A. I. Krylov, *J. Phys. Chem. A*, 2018, **122**, 3653–3664.
- 61 P. Pokhilko, R. Shannon, D. Glowacki, H. Wang and A. I. Krylov, *J. Phys. Chem. A*, 2018.
- 62 N. Orms and A. I. Krylov, *Phys. Chem. Chem. Phys.*, 2018, **20**, 13095–13662.
- 63 N. Orms, D. R. Rehn, A. Dreuw and A. I. Krylov, *J. Chem. Theory Comput.*, 2017, **14**, 638–648.
- 64 M. Nooijen and R. J. Bartlett, *J. Chem. Phys.*, 1997, **107**, 6812–6830.
- 65 M. Wladyslawski and M. Nooijen, ACS Symposium Series, 2002, pp. 65–92.
- 66 M. Musiał, A. Perera and R. J. Bartlett, *J. Chem. Phys.*, 2011, **134**, 114108.
- 67 R. J. Bartlett, *Mol. Phys.*, 2010, **108**, 2905–2920.
- 68 C. A. Taatjes, D. L. Osborn, T. M. Selby, G. Meloni, A. J. Trevitt, E. Epifanovsky, A. I. Krylov, B. Sirjean, E. Dames and H. Wang, *J. Phys. Chem. A*, 2010, **114**, 3355–3370.
- 69 A. Dreuw and L. S. Cederbaum, *Chem. Rev.*, 2002, **102**, 181–200.
- 70 D. Zuev, K. B. Bravaya, T. D. Crawford, R. Lindh and A. I. Krylov, *J. Chem. Phys.*, 2011, **134**, 034310.
- 71 T. Kuś and A. I. Krylov, *J. Chem. Phys.*, 2011, **135**, 084109.
- 72 T. Kuś and A. I. Krylov, *J. Chem. Phys.*, 2012, **136**, 244109.
- 73 N. Moiseyev, *Non-Hermitian quantum mechanics*, Cambridge University Press, 2011.
- 74 G. Jolicard and E. J. Austin, *Chem. Phys. Lett.*, 1985, **121**, 106–110.
- 75 U. V. Riss and H.-D. Meyer, *J. Phys. B*, 1993, **26**, 4503–4536.
- 76 R. Cvetanović, *J. Phys. Chem. Ref. Data*, 1987, **16**, 261–326.
- 77 T.-C. Jagau and A. I. Krylov, *J. Phys. Chem. Lett.*, 2014, **5**, 3078–3085.
- 78 Z. Benda, K. Rickmeyer and T.-C. Jagau, *J. Chem. Theory Comput.*, 2018, **14**, 3468–3478.
- 79 Z. Benda and T.-C. Jagau, *J. Chem. Theory Comput.*, 2018, **14**, 4216–4223.
- 80 T.-C. Jagau, D. B. Dao, N. S. Holtgrewe, A. I. Krylov and R. Mabbs, *J. Phys. Chem. Lett.*, 2015, **6**, 2786–2793.
- 81 W. Skomorowski, S. Gulania and A. I. Krylov, *Phys. Chem. Chem. Phys.*, 2018, **20**, 4805–4817.
- 82 W. Skomorowski and A. I. Krylov, *J. Phys. Chem. Lett.*, 2018, **9**, 4101–4108.
- 83 Shao, Y.; Gan, Z.; Epifanovsky, E.; Gilbert, A.T.B.; Wormit, M.; Kussmann, J.; Lange, A.W.; Behn, A.; Deng, J.; Feng, X., et al., *Mol. Phys.*, 2015, **113**, 184–215.
- 84 A. I. Krylov and P. M. W. Gill, *WIREs Comput. Mol. Sci.*, 2013, **3**, 317–326.
- 85 S. Gozem and A. I. Krylov, *ezDyson User's Manual*, 2015, ezDyson, <http://iopencshell.usc.edu/downloads/>.
- 86 K. A. Peterson, *J. Chem. Phys.*, 1995, **102**, 262–277.
- 87 M. Head-Gordon, *Chem. Phys. Lett.*, 2003, **372**, 508–511.
- 88 R. D. Mead, U. Hefter, P. A. Schulz and W. C. Lineberger, *J. Chem. Phys.*, 1985, **82**, 1723–1731.
- 89 C. M. Oana and A. I. Krylov, *J. Chem. Phys.*, 2007, **127**, 234106–14.