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## Correction: The role of ion solvation in lithium mediated nitrogen reduction

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Correction for 'The role of ion solvation in lithium mediated nitrogen reduction' by O. Westhead *et al.*, *J. Mater. Chem. A*, 2023, <https://doi.org/10.1039/D2TA07686A>.

The authors regret an error in their calculation of the yield rate in Fig. 1b. Due to an error with the unit conversion the peak yield rate at 0.6 M LiClO<sub>4</sub> was incorrectly given as  $60 \pm 3 \text{ nmol cm}^{-2} \text{ s}^{-1}$  ( $n = 3$ ). The corrected yield rate is  $0.53 \pm 0.04 \text{ nmol cm}^{-2} \text{ s}^{-1}$  ( $n = 3$ ) and the corrected version of Fig. 1b is provided herein.

An independent expert reviewed the data provided by the authors and concluded that it does not change the discussion or conclusions presented in the article.

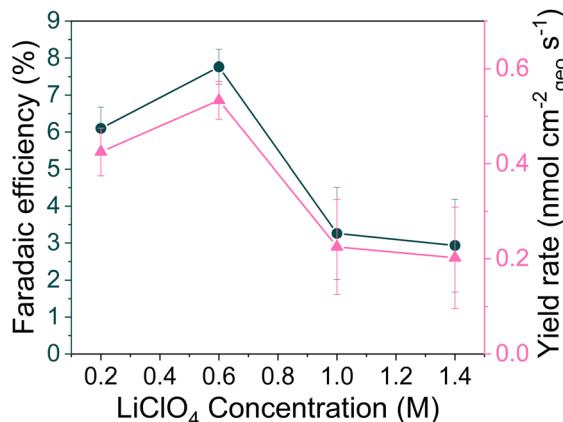


Fig. 1 (b) The change in faradaic efficiency and yield rate with LiClO<sub>4</sub> concentration ( $n = 3$  separate experiments, error bar is standard error in the mean) for a chronopotentiometry experiment at an applied constant current of  $-2 \text{ mA cm}^{-2}$  until  $-10\text{C}$  is passed.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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