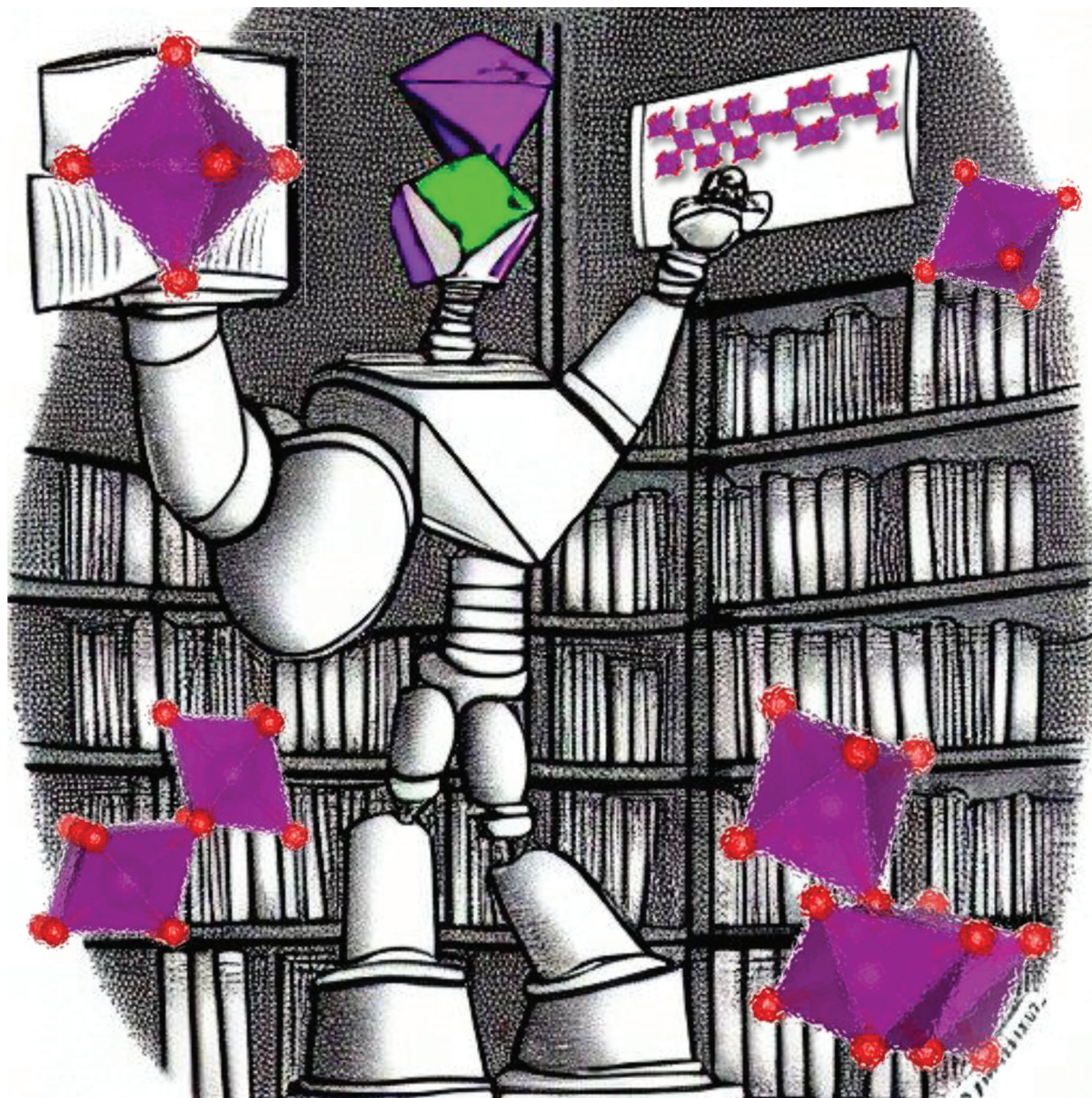


Dalton Transactions

An international journal of inorganic chemistry

rsc.li/dalton



ISSN 1477-9226

PAPER

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using structure-mining: the case of γ - MnO_2

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View Article Online
View Journal | View IssueCite this: *Dalton Trans.*, 2022, **51**, 17150Characterisation of intergrowth in metal oxide materials using structure-mining: the case of γ -MnO₂[†]

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Manganese dioxide compounds are widely used in electrochemical applications e.g. as electrode materials or photocatalysts. One of the most used polymorphs is γ -MnO₂ which is a disordered intergrowth of pyrolusite β -MnO₂ and ramsdellite R-MnO₂. The presence of intergrowth defects alters the material properties, however, they are difficult to characterise using standard X-ray diffraction due to anisotropic broadening of Bragg reflections. We here propose a characterisation method for intergrown structures by modelling of X-ray diffraction patterns and pair distribution functions (PDF) using γ -MnO₂ as an example. Firstly, we present a fast peak-fitting analysis approach, where features in experimental diffraction patterns and PDFs are matched to simulated patterns from intergrowth structures, allowing quick characterisation of defect densities. Secondly, we present a structure-mining-based analysis using simulated γ -MnO₂ superstructures which are compared to our experimental data to extract trends on defect densities with synthesis conditions. We applied the methodology to a series of γ -MnO₂ samples synthesised by a hydrothermal route. Our results show that with synthesis time, the intergrowth structure reorders from a R-like to a β -like structure, with the β -MnO₂ fraction ranging from ca. 27 to 82% in the samples investigated here. Further analysis of the structure-mining results using machine learning can enable extraction of more nanostructural information such as the distribution and size of intergrown domains in the structure. Using this analysis, we observe segregation of R- and β -MnO₂ domains in the manganese oxide nanoparticles. While R-MnO₂ domains keep a constant size of ca. 1–2 nm, the β -MnO₂ domains grow with synthesis time.

Received 5th July 2022,
Accepted 29th August 2022

DOI: 10.1039/d2dt02153f

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Introduction

Due to the increasing need for sustainable technologies, a high interest has grown for functional materials based on earth abundant elements. Manganese dioxide MnO₂ is an interesting compound in that perspective. Manganese oxide crystallises in several polymorphs including layered structures, tunnel structures, as well as spinel, rutile, and hollandite structures. These show a range of different electrochemical properties, allowing for targeted materials design for e.g., energy technologies.^{1–6} In this study, we focus on the γ -MnO₂ polymorph in the MnO₂ family. γ -MnO₂ is of great interest as it is frequently used in commercial batteries as an electrode material and is investigated for secondary battery systems.^{7–10} However, the structural complexity of γ -MnO₂ has made it

difficult to fully map its relations between synthesis, structure and properties. The structure can be represented as an intergrowth of β -MnO₂ ($P4_2/mnm$) and R-MnO₂ ($Pnam$) slabs (Fig. 1). β -MnO₂ crystallises in the rutile structure and is built from single-chains of edge-sharing [MnO₆] octahedra, which are connected through corner-sharing octahedra along the a -

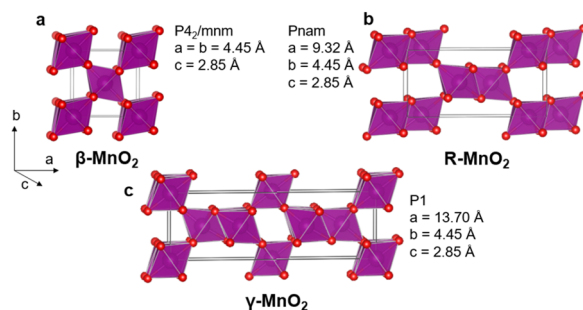


Fig. 1 β -MnO₂ (a) and R-MnO₂ (b) can randomly be stacked along their common a -axis to form an intergrowth γ -MnO₂ structure (c).

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[†] Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d2dt02153f>

and *b*-axis. R-MnO₂ is closely related but contains double-chains of edge-sharing [MnO₆] octahedra along the *a*-axis. The mixing, or intergrowth, of these two parent structures within the same O²⁻ hcp sublattice results in the γ -MnO₂ structure as illustrated in Fig. 1.^{11–14} The intergrowth structure can be considered a sub-ordered sequence of faults along a common crystallographic direction, here the *a*-axis, within the O²⁻ hcp sublattice.^{12,14} The structure is also referred to as containing “De Wolff defects”, named after P. M. De Wolff who was one of the first to work on the characterisation of this phase.¹⁵ A consequence of intergrowth is a huge variability in the structure of different γ -MnO₂ samples, which heavily influences the material properties.^{3,16–18}

However, the defects make the characterisation of γ -MnO₂ challenging. The structural disorder gives rise to anisotropic broadening of Bragg reflections in powder diffraction patterns from γ -MnO₂, and analysis beyond *e.g.*, standard Rietveld refinement is required. In the past decades, several approaches using X-ray diffraction (XRD) have been developed to characterise the γ -MnO₂ structure. A first approach was proposed by De Wolff,¹⁵ who related the broadening of specific Bragg peaks in R-MnO₂-like γ -MnO₂ to the amount of β -MnO₂ domains. Later, Chabre and Pannetier¹² interpreted De Wolff defects using a statistical model of the intergrowth structure. By defining a De Wolff defect distribution to describe the intergrowth rate, they modelled a library of γ -MnO₂ XRD patterns. From this library, they were able to find patterns in the line position and broadening and determined empirical rules to estimate the intergrowth rate of γ -MnO₂ for real samples from their diffraction patterns. Another approach has been used by Simon *et al.*, who performed multi-phase Rietveld refinements on patterns from γ -MnO₂ samples using the R-MnO₂ and β -MnO₂ structures, as well as a highly disordered ϵ -MnO₂ model, where Mn⁴⁺ ions partially occupy all octahedral sites in the O²⁻ hcp sublattice.¹⁹ This provided good agreement between the model and experimental data and allowed quantification of the different phases. The intergrowth in γ -MnO₂ have also been characterised using Pair Distribution Function (PDF) analysis. The PDF is obtained by Fourier transforming the normalised and corrected total scattering data, and can provide information on the local and medium range structure in materials. Using PDF, Galliez *et al.*²⁰ characterised γ -MnO₂ samples by fitting their PDFs with the β -MnO₂ and R-MnO₂ parent structures. This approach gave information about the fractions of parent structures in γ -MnO₂ samples, but was not able to account for the De Wolff defect distribution in the structure.

In a broader context, several tools have been developed for characterisation of planar faults in various crystal structures. In DIFFaX,²¹ faulted crystal structures are modelled by defining the unit layers that the system consists of along with a stacking fault probability for each layer, which allows calculation of scattering intensities. The scattered intensity of the faulted structure is calculated analytically using the structure factor of individual unit layers and an inter-layer interference term that account for the stacking probability of the different layers. This method was used by Chabre and Pannetier in their

analysis of the effect of De Wolff defects in γ -MnO₂ in XRD patterns.¹² DIFFaX+, developed by Leoni *et al.*²² allows to directly refine stacking probabilities in order to quantitatively estimate defect densities. A similar methodology has been implemented in the FullProf suite software package through the FAULTS program, developed by Casas-Cabanas *et al.*²³ A tool for simulating XRD patterns from structures with stacking faults has also been implemented in TOPAS by Coelho *et al.*^{24,25} This approach is based on the generation of supercells, from which individual scattering signals can be calculated. Simulated XRD patterns are then obtained by averaging over these signals. Other approaches in the literature are also based on modelling supercells for calculation of scattering patterns.^{26,27} Planar defects in nanocrystalline materials have furthermore been studied with PDF analysis, where the unit layers can be accurately described by the local range of the PDF and medium range by modelling faulted supercells.^{28–33}

In this paper, we develop a new method for intergrowth characterization. We combine analysis of γ -MnO₂ XRD patterns and PDFs, which allows to investigate which of the two methods is better suited for characterization of intergrown systems such as γ -MnO₂. First, a large database of intergrowth superstructures with varying De Wolff defect distributions is constructed, using the statistical model developed by Chabre and Pannetier,¹² and XRD patterns and PDFs from these structures are simulated. We then use the modelled patterns to establish trends in XRD/PDF peak intensity and shape with intergrowth parameters, which allows development of a peak-fitting method for estimation of defect densities. Secondly, we use the structures in the database for ‘structure-mining’ through Rietveld analysis against experimental data from hydrothermally synthesized manganese oxide nanoparticles. This analysis allows identification of the best fitting structure for a given dataset, so that trends on defect densities in a series of samples can be extracted. Thirdly, the structure-mining results are given to a machine learning algorithm which makes it possible to extract information on the size of domains in the intergrowth samples. This gives new insight into the formation and growth of γ -MnO₂ intergrowths.

Methods

Sample synthesis

All γ -MnO₂ samples have been synthesised using a soft chemistry route. The synthesis method is adapted from the protocol used by Wang *et al.*³⁴ MnSO₄ and (NH₄)₂S₂O₈ are dissolved in deionised (DI) water. The synthesis was conducted hydrothermally at 150 °C for 2 h, 3.5 h and 5 h with a starting concentration of Mn²⁺ ions of 0.7 M and a Mn²⁺ : S₂O₈²⁻ ratio of 5 : 3 using a Teflon-lined steel autoclave. The autoclaves were then left to cool down at room temperature after heating. The as-synthesised black solid was filtered and washed 3 times each with DI water and ethanol before being left to dry at room temperature. The samples were labelled sample 1, 2 and 3, corresponding to heating time of 2, 3.5 and 5 h, respectively.



XRD and X-ray total scattering measurements

The samples were put in Kapton tubes and XRD as well as X-ray total scattering patterns were measured at the I15-1 beamline at Diamond Light Source, United Kingdom. Scattering patterns from an empty Kapton tube and from a Si standard sample were measured as well. The Kapton tube data were used for background subtraction and the Si standard was used to account for instrumental broadening of the Q -space peaks. The total scattering data were acquired on a Perkin Elmer XRD 1611 CP3 2D detector with a pixel size of $100 \times 100 \mu\text{m}^2$. The XRD data were collected on a Perkin Elmer XRD 4343 CT 2D detector with a pixel size of $150 \times 150 \mu\text{m}^2$. The sample to detector distances were 283 mm (total scattering) and 670 mm (XRD) using a wavelength of $0.1618 \text{ \AA}$.

The 2D images were integrated using Dawn.³⁵ PDF data were obtained by Fourier transformation of the normalised and corrected total scattering patterns using xPDFSuite³⁶ with a Q_{min} value of $1.0 \text{ \AA}^{-1}$ and a Q_{max} value of $23 \text{ \AA}^{-1}$ after subtracting the scattering contribution of the Kapton tube.

Transmission electron microscopy (TEM)

Transmission electron microscopy images have been recorded with a Tecnai T20 G2 S-TEM with HT 200 keV and a LaB₆ filament. The samples were prepared by dropcasting a suspension of MnO₂ nanoparticles in ethanol on lacey carbon coated copper TEM grids. The ImageJ³⁷ software was used for image processing and analysis.

Results and discussion

Three γ -MnO₂ samples with different amounts of De Wolff defects were synthesised. To achieve different defect densities, the samples were hydrothermally treated for 2 h (sample 1), 3.5 h (sample 2) and 5 h at 150°C (sample 3), respectively. Fig. 2a and b shows experimental XRD patterns and PDFs obtained for sample 1, 2 and 3 along with calculated patterns of β -MnO₂ and R-MnO₂. The preparation of the samples differs only by synthesis time, and we can observe that the longer the synthesis (sample 3), the structure of the samples becomes closer related to β -MnO₂. The samples prepared with shorter syntheses times, sample 1 and 2, show reflections from both phases, indicating a structure containing features from both β -MnO₂ and R-MnO₂. For all three samples, some reflections in the XRD patterns are broader than others, such as the 110 peak of R- and β -MnO₂ at 1.5 and $2.0 \text{ \AA}^{-1}$ in comparison to the 011 peak of β - and R-MnO₂ around 2.5 and $2.6 \text{ \AA}^{-1}$, highlighted by the dotted line in Fig. 2a. This anisotropic broadening is coupled with a small shift of the peak position of the 110 peaks of the parent structures from one sample to the other. According to DIFFaX simulations from Chabre and Pannetier,¹² this peak shift is related to changes in the De Wolff defect distribution, confirming the presence of intergrown structures in our γ -MnO₂ samples. Micro-twinning along the 201 and 601 planes of the R-MnO₂ parent structure can also happen in γ -MnO₂ samples, but only affects the peak

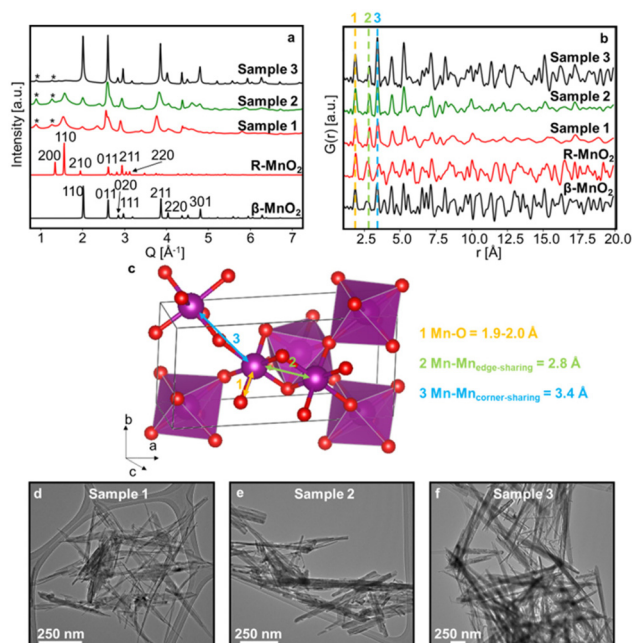


Fig. 2 Experimental XRD patterns (a) and PDFs (b) of samples 1, 2 and 3 showing diverse intergrowth rates as relative intensity of reflections of XRD and peaks of PDF varies. The presence of a secondary phase α -MnO₂ is indicated by (*) in the XRD patterns. Theoretical XRD and PDF patterns of R- (red) and β -MnO₂ (black) parent structures are shown as well. Yellow, green and blue lines in (b) correspond to the first Mn–O and Mn–Mn distances found in these structures as emphasised in (c) in the R-MnO₂ structure. TEM images of samples 1 (d), 2 (e) and 3 (f).

position of these particular reflections, according to DIFFaX calculations of Chabre and Pannetier.¹² We do not observe any such significant effect here. In addition to γ -MnO₂, we observe impurities of α -MnO₂, indicated by stars in Fig. 2a. No evidence of a significant fraction of remaining Mn²⁺ ions from the precursors could be observed. Mn²⁺ containing compounds would likely give rise to a Mn–O peak around $2.2 \text{ \AA}$ in the PDF,³⁸ which is not observed in our experimental data. Furthermore, synthesis of manganese oxides often leads to the formation of manganese dioxide hydrate MnO(OH)₂ that give rise to a peak on the XRD data around $0.5 \text{ \AA}^{-1}$.³⁹ Such a peak is not observed in our experimental data.

The PDFs of the three samples (Fig. 2b) show similarities in peak positions, but differ in relative intensities. They also clearly reflect different sizes of the crystallites, or at least their coherent scattering domains, which increases from sample 1 to 3, as less damping of the PDF signal is seen at high r values. TEM images of the synthesised samples (Fig. 2d–f and Fig. S1†) do indeed show that samples 1, 2 and 3 show needle-shaped particles whose diameter increase with synthesis time, and we estimate them to be *ca.* 7 nm, 15 nm, and 54 nm, respectively. According to previous work, the long dimension of the needles can be assigned to the [001] direction of the β - and R-MnO₂ structures,^{7,40–43} such that the intergrowth direction is along the diameter of the needles. This anisotropic shape can also explain partially the anisotropic broadening of



the peaks on XRD data. However, the effect of De Wolff defects is prominent as it gives rise to even smaller intergrown domains within the sample.

Creation of γ -MnO₂ supercells with various De Wolff defect distributions

To characterise the defect chemistry of the three samples, we first investigate how different De Wolff defect distributions generally affect the XRD patterns and PDFs. The first step for the development of the methodology was therefore to generate a database of γ -MnO₂ supercells containing various De Wolff defect distributions using the approach sketched in Fig. 3. To do so, rules based on the study from Chabre and Pannetier¹² were applied. First, a pair of asymmetric units A and B (Fig. 3a) were identified and described in a primitive cell. Both consist of a pair of [MnO₆] octahedra sharing one corner. However, for A, one of the Mn atoms is placed at the origin of the cell and the other one at (1/2, 1/2, 1/2), whereas for B the Mn atoms are placed at (0, 1/2, 1/2) and (1/2, 1/2, 0). Stacking of 2 similar units A-A or B-B with a stacking vector (a_{β} , 0, 0), (a_{β} being the a cell parameter in the structure) yields a regular β -MnO₂ structure. In contrast, stacking of two different units A-B or B-A yields a regular R-MnO₂ structure. Here, the stacking vector is ($a_{\text{R}}/2$, $\pm 0.026b_{\text{R}}$, 0), with a_{R} and b_{R} being the a and b cell parameters of R-MnO₂. The slight shift (± 0.026) along the b -axis needs to be added to adjust for corner-sharing [MnO₆] octahedra in the R-MnO₂ structure. To keep a straight structure, the shift along the b -axis alternates from one sequence to the other. As such, the unit B will be shifted by $+0.026b_{\text{R}}$ along the b -axis with respect to unit A in a stacking sequence A-B, whereas an opposite shift of $-0.026b_{\text{R}}$ along the b -axis is applied in a B-A sequence.

The statistical approach of Chabre and Pannetier¹² involves the definition of De Wolff defect distribution probabilities P_{b} and P_{r} . P_{b} corresponds to the probability of having a regular

β -MnO₂ stacking, *i.e.* an A-A or a B-B sequence and P_{r} corresponds to the probability of having a regular R-MnO₂ stacking, *i.e.* an A-B or a B-A sequence. A combination of these two probabilities gives the respective phase fractions of β -MnO₂ domains (% β -MnO₂) and R-MnO₂ domains (%R-MnO₂) by eqn (1) and (2):

$$\% \beta - \text{MnO}_2 = \frac{1 - P_{\text{r}}}{2 - P_{\text{r}} - P_{\text{b}}} \times 100\% \quad (1)$$

$$\% \text{R} - \text{MnO}_2 = 1 - \% \beta - \text{MnO}_2 = \frac{1 - P_{\text{b}}}{2 - P_{\text{r}} - P_{\text{b}}} \times 100\% \quad (2)$$

We generate the supercells with varying P_{r} and P_{b} using a simple method based on iterative stacking of asymmetrical units A and B. The process is based on setting the value of P_{b} , P_{r} and of a random number x (between 0 and 1), which is generated in each stacking step as shown in Fig. 2b. The following rules were used: if the i^{th} and $(i-1)^{\text{th}}$ units are the same, *i.e.* A and A or B and B, the type of the $(i+1)^{\text{th}}$ unit to stack depends on P_{b} and x . If $x < P_{\text{b}}$, the stacking remains regular (A-A-A or B-B-B) representing β -MnO₂. If $x > P_{\text{b}}$, the opposite unit is stacked (resulting in A-B-A or B-A-B), thus generating stacking faults in the β -MnO₂ structure. Conversely, if the i -th and $(i-1)^{\text{th}}$ units are different, *i.e.* A-B or B-A, the nature of the $(i+1)^{\text{th}}$ unit depends on P_{r} . If $x < P_{\text{r}}$, the same unit present in the $(i-1)^{\text{th}}$ step is added, resulting in A-B-A or B-A-B structures, corresponding to a regular R-MnO₂ sequence. If $x > P_{\text{r}}$, the same unit is stacked (resulting in A-B-B or B-A-A) creating a stacking fault in the R-MnO₂ structure.

We generated such superstructures in the whole (P_{b} , P_{r}) parameter space, varying both probabilities between 0.01 and 0.99 with an increment of 0.01, yielding 9801 different supercells. The starting point was sequence A-A. Each of them consists of 250 stacked sequences. The a cell parameter of the supercell is calculated by summing up all individual a -components of the stacking vectors for the whole supercell, ranging from 900 Å to 950 Å.

Pattern simulation and peak-fitting analysis

Using the database of 9801 supercells generated above, we first use a simple, peak-fitting analysis to estimate the defect distribution in our experimental samples from the appearance of the scattering patterns and PDFs. XRD and PDF patterns were calculated from each of the supercells using TOPAS Academic.⁴⁴ In both cases, the isotropic atomic displacement parameters (ADP) of manganese and oxygen B_{iso} were set to 0.2 Å². A Gaussian profile with a peak width based on a Caglioti polynomial, $U \tan(\theta)^2 + V \tan(\theta) + W$ ⁴⁵ with $U = 0.42$, $V = -0.36$ and $W = 0.70$ was used.

PDFs from the superstructures were calculated in an r -range up to 20 Å with a Q_{damp} of 0.04 Å⁻¹ and a Q_{max} of 23 Å⁻¹ to model instrumental parameters close to those of a real experiment.

Fig. 4a and d show contour plots of the simulated XRD patterns and PDFs for supercells for which eqn (1) satisfies the condition $\% \beta\text{-MnO}_2 = P_{\text{b}}$ (and conversely eqn (2) satisfies the

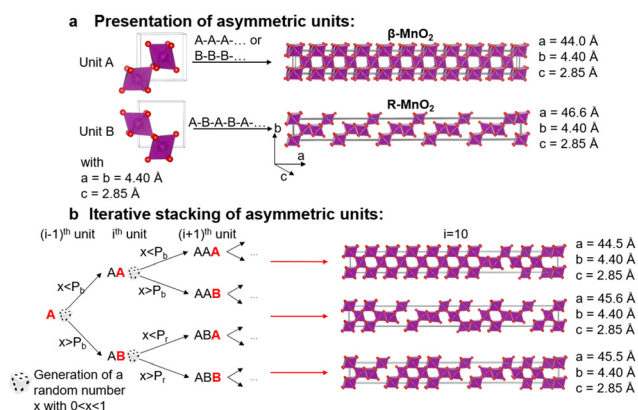


Fig. 3 Creation of supercells. The asymmetric units A and B (a) stacked in a regular fashion yield either the β - or R-MnO₂ structure. When stacked iteratively based on De Wolff defect distribution probabilities P_{b} and P_{r} (b), superstructures with different amounts of De Wolff defects are created. (b) gives an example of a stacking sequence where the $(i-1)^{\text{th}}$ unit A is assumed to be part of a β -MnO₂ block, and therefore the stacking of the i^{th} unit relies on P_{b} .



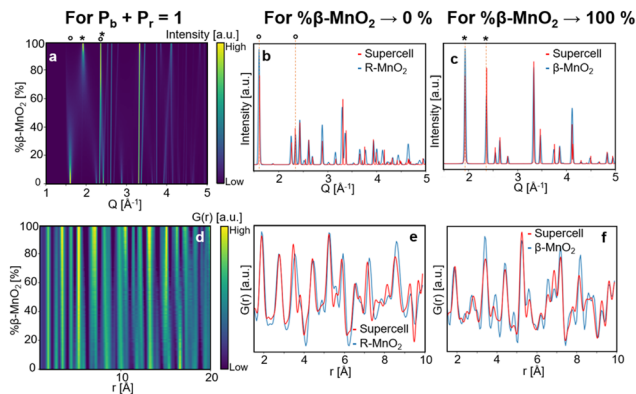


Fig. 4 Modelled XRD patterns (a) and PDFs (d) from supercells for which eqn (1) satisfies the condition $\% \beta\text{-MnO}_2 = P_b$, 110 ($1.5 \text{ \AA}^{-1}$ and $2.0 \text{ \AA}^{-1}$) and 011 (ca. $2.6 \text{ \AA}^{-1}$) reflections of R-MnO₂ and β -MnO₂ are shown by (*) and (*), respectively. Modelled XRD patterns (b) and PDFs (e) for $\% \beta\text{-MnO}_2 \rightarrow 0\%$ (red curve) compared to parent R-MnO₂ (blue curve). Modelled XRD patterns (c) and PDFs (f) for $\% \beta\text{-MnO}_2 \rightarrow 100\%$ (red curve) compared to parent β -MnO₂ (blue curve).

condition $\% \text{R-MnO}_2 = P_r$) in the range $0\% \leq \% \beta\text{-MnO}_2 \leq 100\%$. The modelled XRD patterns show good agreement with those previously reported.^{12,46} When $\% \beta\text{-MnO}_2$ is close to 0%, the PDF and XRD pattern show features and Bragg peaks of pure R-MnO₂ (Fig. 4b). Similarly, when $\% \beta\text{-MnO}_2$ tends towards 100%, XRD peaks sharpen, *i.e.* the structure orders back again. At $\% \beta\text{-MnO}_2 = 100\%$, the XRD patterns and PDFs show similar features to the ones calculated from β -MnO₂ as shown in Fig. 4c and f, respectively.

When the β -MnO₂ fraction deviates from the two end members ($0\% < \% \beta\text{-MnO}_2 < 100\%$), reflections from both parent structures can be observed. Some reflections strongly broadened, indicating a loss of long-range order in the γ -MnO₂ structure. These are for instance the 110 reflections of both β - and R-MnO₂ reflecting disorder along the *a*-axis. On the contrary, 011 reflections are not affected by the intergrowth.

The simulated data show that an estimate of the β -MnO₂ fraction can be obtained from the intensity ratio between the 011 reflections of R- and β -MnO₂. For the simulated XRD patterns we extract the intensities and FWHMs by fitting Pseudo-Voigt functions to these peaks. An example of such a fit is shown in Fig. S2a.† With increasing β -MnO₂ fraction, a decrease of the intensity ratio ($I_{011R}/I_{011\beta}$) can be observed. By fitting an empirical quadratic function (eqn (3)) to the calculated data points, the intensities of the 011 reflection of the β - and R-phases can be linked to their structure factors $F(hkl)$ and to $\% \beta\text{-MnO}_2$ *via* eqn (3):

$$\frac{I_{011R}}{I_{011\beta}} = \frac{|F_{011R}|^2}{|F_{011\beta}|^2} \frac{1 - \% \beta - \text{MnO}_2}{\% \beta - \text{MnO}_2} \quad (3)$$

This procedure can be compared to a simple phase quantification in a binary mixture system, where the intensity ratio of two peaks is proportional to the ratio of the phase fractions, weighted by the ratio of their squared structure factors. The

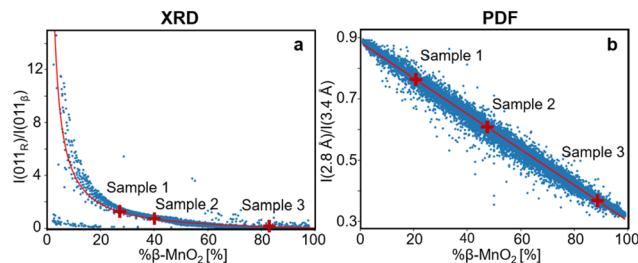


Fig. 5 (a) Intensity ratio between 011_R and 011_β lines from the modelled XRD patterns plotted against the β -MnO₂ fraction ($\% \beta\text{-MnO}_2$, blue dots) and fitted using eqn (3) (red solid line). (b) Intensity ratio between PDF peaks at 2.8 Å and 3.4 Å from the modelled PDFs plotted against $\% \beta\text{-MnO}_2$ (blue dots) and fitted using a linear function (red solid line).

fitted ($I_{011R}/I_{011\beta}$) ratio is plotted as function of $\% \beta\text{-MnO}_2$ in Fig. 5a. We note that some data points ($\% \beta\text{-MnO}_2 < 20\%$, Fig. 5a) do not follow the same trend. This can be attributed to the fact that the 011 reflections of β - and R-MnO₂ have very similar *d*-spacings. In the case where the structure of a γ -MnO₂ sample is closely related to those of one of the parents, its reflections will dominate the XRD pattern. In that situation it becomes difficult to fit two peaks as the XRD would show a strong reflection next to a weak one and some lack of convergence may thus occur.

While this intensity ratio can inform on $\% \beta\text{-MnO}_2$ from XRD patterns, a second relation is needed to fully quantify the stacking disorder within γ -MnO₂ intergrowth. Once $\% \beta\text{-MnO}_2$ is determined, we still need to find P_b or P_r as they are all related to each other through eqn (1). Determining these three values would allow not only to get an estimate of the fraction of both parent structures, but also how their domains are distributed within the intergrowth. For this purpose, the ratio between the FWHM of 011_R and 110_R is calculated for all simulated XRD patterns of the γ -MnO₂ superstructures. While the first reflection is not impacted by the intergrowth (its structure factor does not have any contribution from the position of atoms along the *a*-axis) the second one is. Therefore, a given fraction of R-MnO₂ in the intergrowth will impact the relative broadening of these two reflections which will depend on the domain size of R-MnO₂. The calculated ratio of XRD patterns of all supercells having the same β -MnO₂ fraction are then plotted against P_r and are shown in Fig. S3.† When values for P_r and $\% \beta\text{-MnO}_2$ have been established, P_b can be calculated using eqn (1).

We can now calculate the β -MnO₂ fraction, P_b and P_r for our three experimental XRD datasets of sample 1, 2 and 3. The Bragg peak intensity ratio was determined for each of the three samples by fitting Pseudo-Voigt functions to the 110_R, 110_β, 011_β and 011_R reflections of the experimental data as shown in Fig. S4a, d and g.† The corresponding β -MnO₂ fraction can be estimated from the ratio of intensities as presented in eqn (3). Once this value is determined, P_r is estimated using the fitted FWHM of 011_R and 110_R and, P_b is calculated using eqn (1). The resulting $\% \beta\text{-MnO}_2$, P_b and P_r for the three samples are listed in Table 1. The β -MnO₂ fraction gradually increases from



Table 1 Calculated % β -MnO₂, P_b and P_r values of samples 1, 2, 3 based on XRD and PDF peak-fitting analysis

Sample	% β -MnO ₂ XRD	P_b XRD	P_r XRD	% β -MnO ₂ PDF	P_b PDF	P_r PDF
Sample 1	27	0.53	0.83	21	0.4	0.84
Sample 2	40	0.72	0.81	48	0.69	0.72
Sample 3	82	0.97	0.9	87	0.98	0.87

sample 1 to 3. This can be related to the chemistry of the system. The chosen synthesis pathway leads initially to a metastable R-MnO₂ phase that reorders into β -MnO₂ as the synthesis progresses. Conversely, the structure becomes more and more ordered, as P_b and P_r both increase with synthesis time. Besides having more and more β -MnO₂ blocks in the structure, the β - and R-MnO₂ blocks tend to segregate with time.

Having established a simple method for intergrowth characterisation from XRD data, we can use a similar approach for PDF analysis. We first identify peaks arising from specific motifs in the β -MnO₂ and R-MnO₂ structure. The PDF peak at 2.8 Å can be attributed to the Mn–Mn distance between two [MnO₆] edge-sharing octahedra, while the peak at 3.4 Å arises from the Mn–Mn distance between two [MnO₆] corner-sharing octahedra as shown in Fig. 2c. R-MnO₂ contains more edge-sharing octahedra in its structure (4) than β -MnO₂ (2). Conversely, a manganese atom has less 2nd neighbours at 3.4 Å in R-MnO₂ (where it has 4) than β -MnO₂ (with 8 neighbours). The intensity ratio between the height of the two peaks can thus be used as a measure for % β -MnO₂. We calculated the ratio for each of the simulated PDFs by fitting Gaussian functions to the two peaks, an example of the fitting procedure is shown in Fig. S2b.† The results are plotted as a function of % β -MnO₂ in Fig. 5b. In contrast to the XRD analysis, a linear trend is observed for the PDF analysis. This can be explained by the fact that the amount of edge- and corner-sharing octahedra changes linearly when switching from blocks of β -MnO₂ to blocks of R-MnO₂. The obtained data can be fitted using a linear regression (red line, Fig. 5b). The % β -MnO₂ fractions of sample 1, 2 and 3 were calculated by this approach and are reported in Table 1.

As in the XRD analysis, a second trend is used to quantify P_b and P_r once % β -MnO₂ is determined. The intensity ratio between the peaks at 5.3 Å and 5.7 Å is calculated for γ -MnO₂ superstructures and plotted as a function of P_b for superstructures with a given % β -MnO₂ in Fig. S5.† Examples of fits are shown in Fig. S2c.† The two peaks used here correspond to second and third nearest neighbour in a (2 × 1) channel and while the former is present in both β - and R-MnO₂ structures, the latter can only be found in R-MnO₂. However, because in an intergrown structure a β -MnO₂ block can be next to a R-MnO₂ block, these two peaks may arise as well. Therefore, the ratio of these two peaks contain information about how many β -MnO₂ blocks are connected to R-MnO₂ blocks, *i.e.* the stacking prob-

abilities P_r and P_b . The calculated ratio can be fitted using a linear function. Again, we can use these correlations to determine the defect density in our three samples. For each experimental PDF, the % β -MnO₂ is first calculated and used to extract the corresponding P_b value shown in Table 1. Knowing both % β -MnO₂ and P_b , P_r can be determined using eqn (1). Using the described procedure, we can now estimate % β -MnO₂, P_b and P_r for the experimental samples 1, 2 and 3 and the fitting of the PDF peaks is shown in Fig. S4b, c, e, f, h, and i.† The results are summarised in Table 1. We are thus able to get a full picture of the intergrowth in γ -MnO₂ using PDF analysis.

As seen in Table 1, the values extracted from the XRD and PDF analysis are of the same order of magnitude. However, some variations can be observed, especially regarding the estimation of P_b and P_r . In general, the peak analysis from PDF gives higher values than those obtained from XRD. This could imply that some other effects influence the intensity of PDF peaks in the local range. Indeed, other types of defects, such as micro-twinning and proton-induced defects may be seen in R-MnO₂-like intergrowth samples (as in sample 1).^{12,13,47,48} Those may alter the short-range order in the crystal, and thus the intensity of the first peaks of the PDF. The proton-induced defects consist of substitution of Mn⁴⁺ by Mn³⁺ along with the formation of hydroxyl groups on the neighbouring O²⁻ site, or by compensation of a manganese vacancy by four hydroxyl groups. Such defects may have a large impact on the properties of the material,⁴⁹ however, characterization of these is outside the scope of this paper, and will be addressed in a future study. Finally, some stacking faults in the MnO₂ O²⁻ hcp sublattice has been predicted⁵⁰ but no experimental evidence of such stacking events has been reported.

Structure refinement and structure-mining

The peak-fitting-approach presented above gives a good first estimate of the defect density in the materials. However, to further extend our analysis and to identify the best model to describe the data, we use 'structure-mining'. Here, the structures generated above are fitted to the experimental data, and the trends in the goodness of fits of the different models are investigated. In order to optimise calculation time, P_b and P_r were limited to (0.0 < P_b < 1.0, 0.5 < P_r < 1.0) for sample 1, (0.5 < P_b < 1.0, 0.5 < P_r < 1.0) for sample 2 and (0.5 < P_b < 1.0, 0.0 < P_r < 1.0) for sample 3 according to the results of the peak-fitting analysis.

All the structure refinements were performed using TOPAS.⁴⁴ For the XRD analysis, the supercell cell parameters a , b and c and the isotropic atomic displacement parameters (ADP) for both manganese and oxygen were refined along with background and peak shape parameters, as described further in Table S1 in the ESI.† To minimise the number of refinement parameters, all atomic coordinates in the supercell are expressed as a function of the ones in the asymmetric unit. To achieve that, a translational operation in accordance with the stacking sequence in the supercell is applied to the atoms in each layer. In addition, the hollandite α -MnO₂ structure was added as a secondary phase. A similar approach was used for the PDF analysis.



Fig. 6 shows the R_{wp} values obtained for the fitting of mined structures for samples 1, 2 and 3. The R_{wp} values are represented in a colour map, where the P_r and P_b values used for the superstructure generation are plotted on the x- and y-axes. The results are also represented in one dimension as a function of β -MnO₂ fraction of the refined superstructures (Fig. 6b–q). The R_{wp} value corresponding to a simple two-phase refinement using the two parent structures (β -MnO₂ and R-MnO₂) on the same experimental dataset is shown as a red vertical line (Fig. 6b–q).

When performing refinements of the mined superstructures on the three samples, we see that the lowest R_{wp} values are systematically clustered in a region of the (P_b , P_r) space (dark blue regions). This is the case for both the XRD and PDF analysis, which means we can use this region as a β -MnO₂ fraction estimate: for sample 1, we estimate the β -MnO₂ fraction to be in the range 20–40%, sample 2 in the range 40–60% and for sample 3 in the range 80–100% (see Fig. 6b–q respectively). The fact that a whole range of super-

cells yielded similar fits, *i.e.* similar R_{wp} values, implies that in the case of such intergrown system, a single model is not sufficient to describe the material structure, as also seen from the poor fit quality for especially sample 1 which is further discussed below. This can be related to the random nature of the stacking of R- and β -MnO₂ blocks in the supercell. Indeed, some key structural motifs in different quantities may be present in all the superstructures that yielded a good fit in the blue area on Fig. 6a–g and j–p. The idea of structural motifs will be further investigated in the next section.

When comparing the results of the fitting of mined structures on sample 1, 2 and 3 for XRD and PDF (Fig. 6b–q respectively) as a function of % β -MnO₂, we observe that the dip in the R_{wp} around the best fits is deeper for the XRD superstructure refinements than for PDF. This indicates that XRD is more sensitive to disorder in the intergrowth sequence than PDF. XRD probes the average structure of a material while PDF probes the local structure. Defects such as intergrowth affect the long-range structure, which is better characterised in XRD.

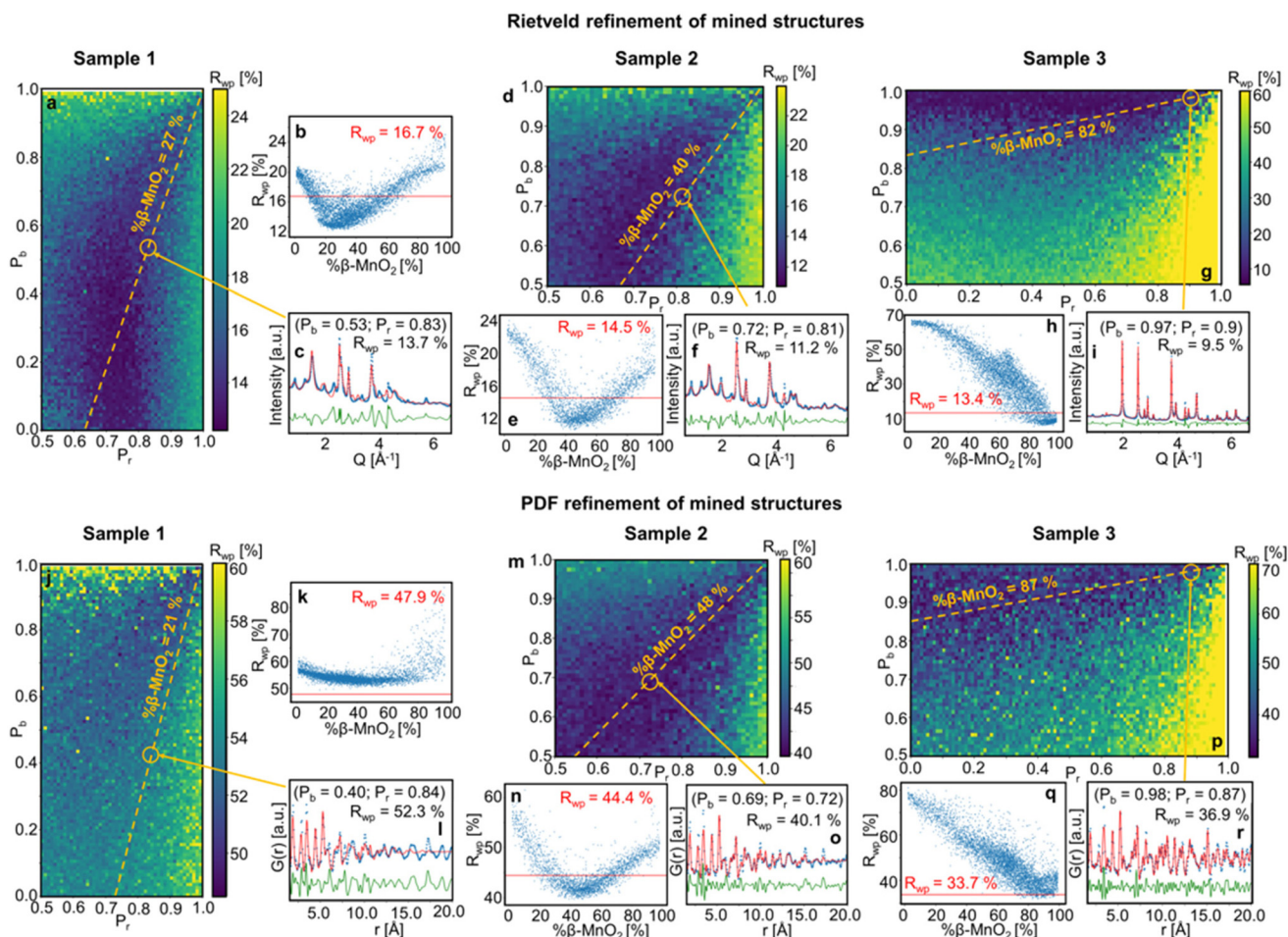


Fig. 6 Results from fitting of mined structures of sample 1, 2 and 3. Contour plots show R_{wp} values from Rietveld (a, d and g) and PDF refinements (j, m and p) as a function of (P_b , P_r) for each superstructure for samples 1, 2 and 3, respectively. The same R_{wp} values as a function of the β -MnO₂ fraction are shown in (b, e and h) and (k, n and q), respectively. The red solid line represents the R_{wp} value of the refinement of a two-phase mixture of parent structure β - and R-MnO₂. The Rietveld and PDF refinement of the superstructure with (P_b , P_r) values determined from the peak-fitting analysis are shown in (c, f and h) and (l, o and r), respectively.



The results of the peak-fitting analysis were also compared to the ones from the structure-mining approach. The fit of the superstructures corresponding to the (P_b , P_r) values obtained from the peak-fitting analysis of samples 1, 2 and 3 are shown in Fig. 6c–i (XRD) and l–r (PDF). The P_r and P_b values identified from the peak-fitting analysis fall into the region where superstructures fitted better to the experimental data than a two-phase mixture of the parent structures. However, some discrepancies are still observed, especially for sample 1 and 2. This is more clearly seen in Fig. S6,† showing a zoom of the low Q region of the Rietveld refinements along with the low r region or the PDF refinements presented in Fig. 6. For the Rietveld refinement of the XRD data, the intensity of some reflections is not fully described, including the 110_R and 110_β reflections. Furthermore, the 201_R reflection at $2.4 \text{ \AA}^{-1}$ is not matched in position or in intensity. In r -space, the intensity of the first peaks, at $1.90 \text{ \AA}$, $2.88 \text{ \AA}$ and $3.40 \text{ \AA}$ are not well described. The discrepancies in the fits can arise from different effects. Firstly, the limited size of the supercells (250 asymmetric units) may not provide a sufficient statistical average of the De Wolff defects distribution in the sample, which affects the description of the peak intensities in the XRD patterns. A second effect, as observed from TEM (Fig. 2 and Fig. S1†), is the particle shape, which affects the Bragg peak width and r -dependent PDF peak intensity. Finally, as discussed above, the $\gamma\text{-MnO}_2$ structures can contain several types of defects, *i.e.* De Wolff or intergrowth defects, microtwinning and H-related defects. All of them impact the XRD patterns and PDFs. The aim of the present work is to develop a simple and fast method, building on structure mining, which can reveal trends in De Wolff defect densities. Therefore, our structure refinements may miss some of the information hidden in the scattering signal of the samples as we focus here on only one aspect of the problem.

The R_{wp} values for the corresponding superstructures and for the two-phase refinement of the parent structures are compared and summarised in Table 2. The superstructure refinements on experimental data generally yield a better fit than the refinement of the simple mix of the parent structures, with R_{wp} values being *ca.* 3% systematically lower. An exception arises for the PDF refinements of sample 1 and 3. For sample 1, the R_{wp} values obtained for analysis in both Q - and r -space are higher than those of the two other samples. Broader reflec-

tions in Q -space as well as a rapid damping of the PDF in direct space imply that sample 1 contains smaller crystallites, as also seen from our TEM analysis.

Nanostructure analysis

In order to extract more information from the structure-mining process, we use our recently published approach for structure analysis, namely the Machine Learning-based Motif Extractor (ML-MotEx).⁵¹ The idea of ML-MotEx is to give the results from a large set of structure refinements to an explainable ML algorithm, which learns to predict the goodness-of-fit value for a structural refinement to the experimental data. Most importantly, the use of an explainable ML algorithm makes it possible to obtain information on the effect that each feature of the model has on the fit quality. ML-MotEx uses a Gradient Boosting Decision Tree (GBDT)⁵² to predict the R_{wp} value from structural information. SHAP (SHapley Additive exPlanation)^{53,54} values are used to explain each feature's effect on the R_{wp} value by taking into consideration the full dataset given as input. While the magnitude of the SHAP value quantifies the importance of a feature, its sign signifies whether the presence of the feature in the model affects the R_{wp} value positively or negatively, as described further below. The method is also described in detail in our recent publication.⁵¹

We here want to use the ML-MotEx method to extract information on the nanostructure of the samples, *i.e.* determine the size of the $R\text{-MnO}_2$ and $\beta\text{-MnO}_2$ domains present in the intergrowth structures. Results from each of the 2500 superstructure refinements, performed on both XRD and PDF data, were used as input parameters for ML-MotEx. The structural features investigated using explainable ML were the $\beta\text{-MnO}_2$ fraction, P_b and P_r , and the intergrowth sequence of each supercell. This input allows to investigate what structural features are important for the fit quality.⁴² The intergrowth sequence of a superstructure is generated from the asymmetric unit sequence used to build it (presented in Fig. 3). By taking the asymmetric unit at the i^{th} and $(i - 1)^{\text{th}}$ position we determine whether a “block” constituting the intergrowth sequence corresponds to β - or $R\text{-MnO}_2$.

Fig. 7a shows the results of using ML-MotEx on the PDF measured on sample 2. ML-MotEx on the XRD and PDF obtained for samples 1 and 3 are shown in Fig. S7 in the ESI.† For each fit, a SHAP value is obtained for all of the features. We first consider the $\beta\text{-MnO}_2$ fraction, which we also investigated from the XRD and PDF analysis above. In the figure, the feature value (*i.e.* here $\beta\text{-MnO}_2$) is shown on the colour axis, where red represents a low $\beta\text{-MnO}_2$ fraction, while blue represents a high $\beta\text{-MnO}_2$ fraction. The SHAP values obtained for this feature show that when $\beta\text{-MnO}_2$ is either high (red/ $\sim 100\%$) or low (blue/ $\sim 0\%$), the SHAP values are positive. This means that the R_{wp} values are increased in these fits. Conversely, medium $\beta\text{-MnO}_2$ values (purple, *ca.* 50%) show negative SHAP values and thus lower the R_{wp} value. This agrees well with the results from the analysis in Fig. 6e, where we determined $\beta\text{-MnO}_2$ to be 48% for sample 2. The SHAP

Table 2 R_{wp} values of Rietveld and PDF refinements of the superstructures corresponding to results from peak-fitting analysis and from refinements of parent structures β - and $R\text{-MnO}_2$ on samples 1, 2 and 3. Fits can be found in Fig. 6c, f and Fig. S6(a–c)†

Sample	Refinement	R_{wp} , super-structures (%)	R_{wp} , $\beta\text{-MnO}_2$ + $R\text{-MnO}_2$ (%)
Sample 1	Rietveld	13.7	16.7
	PDF	52.3	47.9
Sample 2	Rietveld	11.2	14.5
	PDF	40.0	44.4
Sample 3	Rietveld	9.5	13.4
	PDF	36.9	33.6



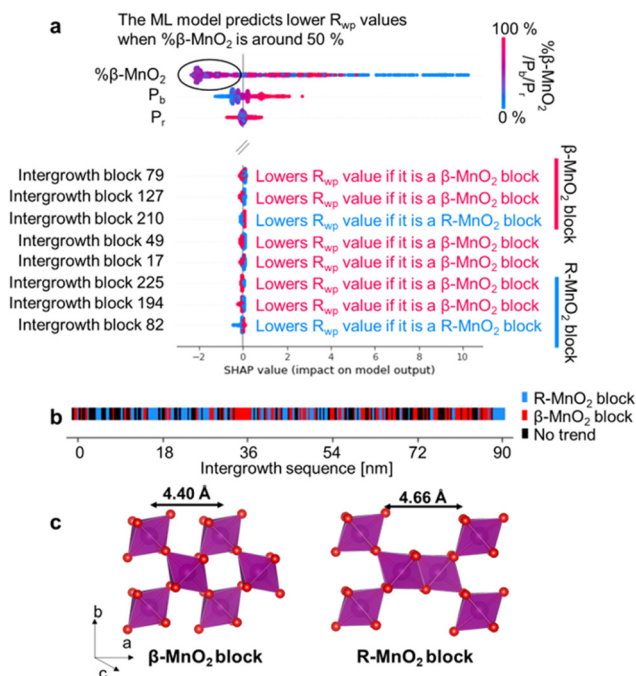


Fig. 7 Summary of the ML-MotEx analysis of the PDF data measured at sample 2. Violin plot of the SHAP values obtained in the analysis of the PDF data measured on sample 2, showing preferable % β-MnO₂, P_b and P_r values and if intergrowth blocks in the starting model are favorable rather as a β- or R-MnO₂ block for the fit quality. (a) Each block in the supercell is colored with respect to the results from the ML-MotEx analysis. If ML-MotEx prefers the block to be R-MnO₂, it is colored blue, if ML-MotEx does not differentiate between the blocks, it is colored black and if ML-MotEx prefers β-MnO₂ it is colored red. (b) Representation of a β-MnO₂ and of a R-MnO₂ block (c).

analysis also suggests that the R_{wp} value of the refinements can be pushed down by having a low P_b value. This is new information as refinements of the mined structures shown in Fig. 6m did not indicate a region of better fits for particularly low P_b values. Finally, the SHAP analysis shows that the P_r value does not seem to have a strong effect on the fit quality.

To get deeper understanding of the influence of De Wolff defect distribution on the fit, we next investigate the effect of the intergrowth sequence itself on the refinements through the SHAP analysis. We correlate here the nature of individual intergrowth block within the intergrowth sequence to the R_{wp} values using the SHAP analysis. As each block in the sequence is directly related to the structure refined to the experimental data, it is possible to establish a correlation between each intergrowth block and the R_{wp} value when investigating the whole range of supercells refined to one single experimental dataset. The influence of the intergrowth sequence on the fit quality is illustrated in Fig. 7b, where “intergrowth block X” parameters refer to a specific block in the 250-block long intergrowth sequence. Here, a positive feature value (red colour in Fig. 7a) refers to a β-MnO₂ block. Conversely, a negative feature value (blue colour in Fig. 7a) refers to a block of R-MnO₂. As an example, the SHAP value for intergrowth block 79 in Fig. 7a is red in the negative region which means that the R_{wp} value of the refinement is

decreased if the unit is a β-MnO₂ block. Conversely, the SHAP value is blue in the positive region so a R-MnO₂ block will increase the R_{wp} of the refinement. The SHAP analysis performed on XRD results of sample 2, along with PDF and XRD results of samples 1 and 3 are shown in Fig. S7.†

Based on the interpretation of SHAP values given above, we search for any continuous R- or β-MnO₂ domains. First, we sort the intergrowth sequence blocks by whether it is more beneficial for the refinement to have them as a β- or a R-MnO₂ block, *i.e.* when the SHAP value of a given block should rather be negative or positive to push down the R_{wp} value of the refinement. Then, we screen the complete intergrowth sequence to search for sequences of consecutive “preferentially β-MnO₂ block” or “preferentially R-MnO₂ block” and count the number of similar blocks in these sub-sequences. The result of this analysis is shown in Fig. 7b where blocks of R- and β-MnO₂ are shown in blue and red, respectively. One can notice that these blocks are not randomly distributed in the intergrowth sequence but rather segregated into some small domains of a few consecutive R- or β-MnO₂ blocks. We extract the size of these continuous domains of R-MnO₂ and β-MnO₂ by multiplying this number to the unit cell parameter a of R- and β-MnO₂. The number of domains of a given size is counted and normalised by the total number of domains. The resulting domain size distributions are plotted as a function of size to build a Domain Size Distribution (DSD) function. This process is carried out for the results from structural refinements performed on both XRD and PDF data for sample 1, 2 and 3 and shown in Fig. 8. The results show small domain

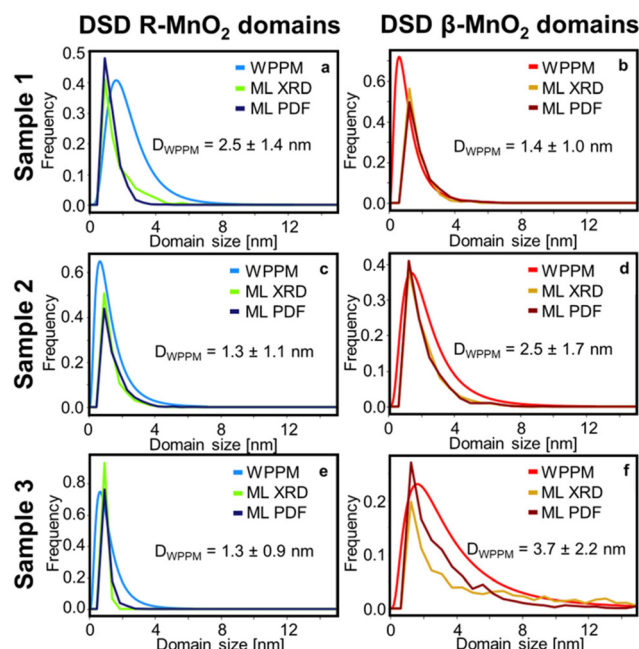


Fig. 8 Domain Size Distributions (DSD) determined by Whole Powder Pattern Modelling (WPPM), ML-MotEx analysis of XRD (ML XRD) and PDF (ML PDF) refinement results of R- and β-MnO₂ domains in samples 1 (a and b), 2 (c and d) and 3 (e and f).



sizes of 1–2 nm depending on the sample and the parent structure (either R- or β -MnO₂), as discussed below.

As a reference for the approach presented above, we also analysed the line profile of individual reflections for both R- and β -MnO₂ in the experimental data using Whole Powder Pattern Modelling (WPPM). We carry out this analysis independently of the structure mining approach. This method, detailed in ESI Notes 1,† was used in order to determine the domain size distribution in the intergrowth direction, *i.e.* the [100] direction for both β -MnO₂ and R-MnO₂ domains. The software PM2K developed by Leoni *et al.*⁵⁵ was used to estimate the DSD of β -MnO₂ and R-MnO₂ domains based on the 110_R and 110 _{β} reflections of the XRD patterns of the three samples. The coefficients of the Caglioti-based instrumental profile were determined by analyzing the XRD pattern from the Si standard beforehand and a Chebyshev polynomial function was used to model the background. A lognormal function was used to model the DSD of β -MnO₂ and R-MnO₂ domains and the average domain size and standard deviation in the corresponding directions were determined. The fitting of the peaks is shown in Fig. S8.† From the lognormal distribution center μ and variance σ^2 , the average crystallite size (D_{average}) and standard deviation of the crystallite size distribution (s.d.) can be derived as shown in eqn (4) and (5) below:

$$D_{\text{average}} = e^{(\mu + \frac{\sigma^2}{2})} \quad (4)$$

$$\text{s.d.} = (e^{2\mu + \sigma^2}(e^{\sigma^2} - 1)) \quad (5)$$

Finally, as we aim to determine the DSD of R- and β -MnO₂ domains along the [100] direction, we need to make some assumptions since we use the 110 reflection which has a contribution in the [100] and [010] directions. We assume that the size of the domains is large along the [010] direction (no intergrowth in this direction) and we correct the size by using the relation $(a^2 + b^2)^{1/2}/a$, *i.e.*, $2^{1/2}/2$ for β -MnO₂ domains, and *ca.* 1.11 for R-MnO₂ domains. By doing that, we can consider the same dimension for the results from DSD and ML-MotEx as the latter is already looking at number of intergrowth blocks along the [100] direction. The DSDs extracted from the WPPM analyses are plotted in Fig. 8 and compared to the ones obtained by ML-MotEx.

The DSD analysis show some differences between sample 1, 2 and 3. The R-MnO₂ DSDs, which is centred around 1.0 nm, become narrower with increasing synthesis time. For instance, the average crystallite size estimated from WPPM is 2.5 ± 1.4 nm for sample 1 and 1.3 ± 0.9 nm for sample 3. This corresponds to only one to two unit cells of regular R-MnO₂ domains within the intergrowth structure. Meanwhile, the β -MnO₂ DSD shows a narrow distribution of small domains of 1.4 ± 1.0 nm in sample 1, which broadens and shifts towards larger domain size in sample 2 (2.5 ± 1.7 nm) and 3 (3.7 ± 2.2 nm). Similar trends in size are seen from the ML-MotEx results. From this observation, it seems that in γ -MnO₂, the intergrowth structure is stabilised by small R-MnO₂ domains, around which β -MnO₂ domains may grow. This is in line with theoretical studies from Sun *et al.* where they showed that the R-MnO₂ structure is the most

thermodynamically stable MnO₂ polymorph at small crystallite sizes up to 5 nm.⁵⁶ Beyond that size, β -MnO₂ becomes the most thermodynamically stable MnO₂ polymorph. The calculated DSD of the supercells determined by the peak-fitting method are shown in Fig. S9† and show similar trends as those that we observe using WPPM and ML. While the R-MnO₂ DSD remains the same from sample 1 to 3, the β -MnO₂ domains in the supercell become larger, although shifting towards larger values than expected for sample 3, with domain sizes ranging up to 12 nm. Because we investigate here only one single supercell, there can only exist a limited number of large domains, in opposition to the ML results where these large domains are mitigated by having some intergrowth blocks that are estimated as neither “preferentially β -MnO₂ block” or “preferentially R-MnO₂ block”.

The similarity between the WPPM and our ML-MotEx-based method illustrates the crystal intergrowth observed can be compared to a stack of ultra-small nano-sized domains of each parent crystal structure, giving rise to the anisotropic line broadening in XRD patterns. Although the model did not account for all the effects impacting the XRD pattern of an intergrown crystal (peak asymmetry, microstrain), this approach gives a good description of the nanostructure.

For the 3 samples that we characterised, the size of the β - and R-MnO₂ domains are *ca.* 1–2 nm, which corresponds also to the range in which the PDF data are refined. This observation can explain why only a few superstructures yielded a better refinement than that done with the two separate parent structures. Because PDF probes short-range order in a material, intergrowth systems can be treated as a binary mixture of the parent structures. While PDF analysis can thus be used to estimate the β -MnO₂ and R-MnO₂ fractions, XRD is more sensitive with respect to the way the two crystal structures are intergrown, *i.e.* the effect of P_b and P_r . However, in a case of an intergrowth structure that shows a high amount of disorder and where Bragg peaks can no longer be observed, PDF can still provide the missing information.

Conclusion

We have developed a method to characterise crystal intergrowth in γ -MnO₂ using a combination of XRD and PDF analysis which is potentially applicable to many other systems. Our samples were hydrothermally synthesised γ -MnO₂, where different synthesis times led to different defect densities. Our aim was to characterize the intergrowth density, and analyse the size of the domains present in the structure.

We first simulated XRD and PDF patterns from a large database of γ -MnO₂ structure models generated with different intergrowth densities. By considering trends in the simulated patterns, we developed a simple peak-fitting approach to estimate the defect density. This peak-fitting approach allowed us to obtain an estimate for the three parameters characterising the intergrowth, namely the β -MnO₂ fraction, P_b and P_r .

To further characterise the materials, we used a structure-mining approach. Again using our database of γ -MnO₂ struc-



ture models, we refined large subsets of these structures against our experimental data. Using the fit qualities obtained, we picked the best fitting structure for each sample and its corresponding intergrowth parameters. As the synthesis time of γ -MnO₂ is increased from 2 to 5 hours, the intergrowth structure reorders from a R-like to a β -like structure, with segregation of R- and β -MnO₂ domains. Our structure-mining approach also shows that structures with a fairly broad distribution of P_b and P_r lead to similar fit qualities. This suggests that it is in general difficult to assign one specific model to describe the structure of such disordered samples.

To address this issue, and to analyse the nanostructure in the samples, we characterised the size of the intergrowth domains in the three samples. Here, we applied an explainable machine learning algorithm, ML-MotEx, to extract information from the 2500 fits done for the structure-mining approach. ML-MotEx extracts the size of domains of both parent structures that help to yield a better fit of the structure, thus comparing crystal intergrowth structures to a mere stack of ultra-small nano-sized domains of two parent structures. The results showed that while R-MnO₂ domains keep a constant size, β -MnO₂ domains tend to grow as synthesis is carried out for longer time. The study furthermore showed that XRD is more suitable than PDF to characterise crystal intergrowths. The intergrowth affects especially the long-range order, which is better investigated with XRD.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work is part of a project that has received funding from the European Research Council (ERC) under the European Union's Horizon 2020 Research and Innovation Programme (grant agreement no. 804066). We are grateful to the Villum Foundation for financial support through a Villum Young Investigator grant (VKR00015416). Funding from the Danish Ministry of Higher Education and Science through the SMART Lighthouse is gratefully acknowledged. We acknowledge Diamond Light Source for time on Beamline I15-1 under Proposal CY21791-4. We thank Phillip Chapter for collecting data during remote beamtime. TEM images were collected at the DTU Nanolab, the National Centre for Nano Fabrication and Characterization at the Technical University of Denmark.

References

- 1 L. I. Hill, A. Verbaere and D. Guyomard, *J. Power Sources*, 2003, **119**–121, 226–231.
- 2 S. Devaraj and N. Munichandraiah, *J. Phys. Chem. C*, 2008, **112**, 4406–4417.
- 3 K. Galliez, P. Deniard, D. Lambertin, S. Jobic and F. Bart, *J. Nucl. Mater.*, 2013, **438**, 261–267.
- 4 D. M. Robinson, Y. B. Go, M. Mui, G. Gardner, Z. Zhang, D. Mastrogiiovanni, E. Garfunkel, J. Li, M. Greenblatt and G. C. Dismukes, *J. Am. Chem. Soc.*, 2013, **135**, 3494–3501.
- 5 Y. Meng, W. Song, H. Huang, Z. Ren, S. Y. Chen and S. L. Suib, *J. Am. Chem. Soc.*, 2014, **136**, 11452–11464.
- 6 T. R. Juran, J. Young and M. Smeu, *J. Phys. Chem. C*, 2018, **122**, 8788–8795.
- 7 F. Y. Cheng, J. Chen, X. L. Gou and P. W. Shen, *Adv. Mater.*, 2005, **17**, 2753–2756.
- 8 M. Chamoun, W. R. Brant, C.-W. Tai, G. Karlsson and D. Noréus, *Energy Storage Mater.*, 2018, **15**, 351–360.
- 9 I. Vasiliev, B. A. Magar, J. Duay, T. N. Lambert and B. Chalamala, *J. Electrochem. Soc.*, 2018, **165**, A3517–A3524.
- 10 M. M. Thackeray, M. H. Rossouw, A. de Kock, A. P. de la Harpe, R. J. Gummow, K. Pearce and D. C. Liles, *J. Power Sources*, 1993, **43**, 289–300.
- 11 C. Godart, M. Latroche, C. Fretigny and C. Levy-Clement, *Phys. Status Solidi A*, 1992, **132**, 253–268.
- 12 Y. Chabre and J. Pannetier, *Prog. Solid State Chem.*, 1995, **23**, 1–130.
- 13 A. Le Gal La Salle, S. Sarciaux, A. Verbaere, Y. Piffard and D. Guyomard, *J. Electrochem. Soc.*, 2000, **147**, 945.
- 14 L. I. Hill and A. Verbaere, *J. Solid State Chem.*, 2004, **177**, 4706–4723.
- 15 P. M. De Wolff, *Acta Crystallogr.*, 1959, **12**, 341–345.
- 16 C. Poinsignon, M. Amarilla and F. Tedjar, *J. Mater. Chem.*, 1993, **3**, 1227–1229.
- 17 S. Jouanneau, S. Sarciaux, A. Le Gal La Salle and D. Guyomard, *Solid State Ionics*, 2001, **140**, 223–232.
- 18 K. Galliez, P. Deniard, C. Payen, D. Lambertin, F. Bart, H.-J. Koo, M.-H. Whangbo and S. Jobic, *Inorg. Chem.*, 2015, **54**, 1194–1196.
- 19 D. Simon, R. Morton and J. Gislason, *J. Adv. X-ray Anal.*, 2004, **47**, 267–280.
- 20 K. Galliez, P. Deniard, P.-E. Petit, D. Lambertin, F. Bart and S. Jobic, *J. Appl. Crystallogr.*, 2014, **47**, 552–560.
- 21 M. M. J. Treacy, J. M. Newsam and M. W. Deem, *Proc. R. Soc. London, Ser. A*, 1991, **433**, 499–520.
- 22 M. Leoni, A. F. Gualtieri and N. Roveri, *J. Appl. Crystallogr.*, 2004, **37**, 166–173.
- 23 M. Casas-Cabanas, M. Reynaud, J. Rikarte, P. Horbach and J. Rodríguez-Carvajal, *J. Appl. Crystallogr.*, 2016, **49**, 2259–2269.
- 24 A. A. Coelho, J. S. O. Evans and J. W. Lewis, *J. Appl. Crystallogr.*, 2016, **49**, 1740–1749.
- 25 C. M. Ainsworth, J. W. Lewis, C.-H. Wang, A. A. Coelho, H. E. Johnston, H. E. A. Brand and J. S. O. Evans, *Chem. Mater.*, 2016, **28**, 3184–3195.
- 26 W. A. Ślawiński, D. S. Wragg, D. Akporiaye and H. Fjellvåg, *Microporous Mesoporous Mater.*, 2014, **195**, 311–318.
- 27 S. Bette, T. Takayama, V. Duppel, A. Poulain, H. Takagi and R. E. Dinnebier, *Dalton Trans.*, 2019, **48**, 9250–9259.
- 28 A. S. Masadeh, E. S. Bozin, C. L. Farrow, G. Paglia, P. Juhas, S. J. L. Billinge, A. Karkamkar and M. G. Kanatzidis, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **76**, 115413.



- 29 X. Yang, A. S. Masadeh, J. R. McBride, E. S. Božin, S. J. Rosenthal and S. J. L. Billinge, *Phys. Chem. Chem. Phys.*, 2013, **15**, 8480–8486.
- 30 A. Kudielka, S. Bette, R. E. Dinnebier, M. Abeykoon, C. Pietzonka and B. Harbrecht, *J. Mater. Chem. C*, 2017, **5**, 2899–2909.
- 31 S. Bette, T. Takayama, K. Kitagawa, R. Takano, H. Takagi and R. E. Dinnebier, *Dalton Trans.*, 2017, **46**, 15216–15227.
- 32 A. M. Pütz, M. W. Terban, S. Bette, F. Haase, R. E. Dinnebier and B. V. Lotsch, *Chem. Sci.*, 2020, **11**, 12647–12654.
- 33 S. Mangelsen, B. R. Srinivasan, U. Schürmann, L. Kienle, C. Näther and W. Bensch, *Dalton Trans.*, 2019, **48**, 1184–1201.
- 34 X. Wang and Y. Li, *J. Am. Chem. Soc.*, 2002, **124**, 2880–2881.
- 35 J. Filik, A. Ashton, P. Chang, P. Chater, S. Day, M. Drakopoulos, M. Gerring, M. Hart, O. Magdysyuk and S. Michalik, *J. Appl. Crystallogr.*, 2017, **50**, 959–966.
- 36 X. Yang, P. Juhas, C. L. Farrow and S. J. L. Billinge, 2014, eprint, DOI: [10.48550/arXiv.1402.3163](https://doi.org/10.48550/arXiv.1402.3163).
- 37 M. D. Abràmoff, P. J. Magalhães and S. J. Ram, *J. Biophotonics Int.*, 2004, **11**, 36–42.
- 38 K. Pussi, J. Gallo, K. Ohara, E. Carbo-Argibay, Y. V. Kolen'ko, B. Barbiellini, A. Bansil and S. Kamali, *Condens. Matter*, 2020, **5**, 19.
- 39 K. S. Abou-El-Sherbini, M. Askar and R. Schöllhorn, *Solid State Ionics*, 2002, **150**, 407–415.
- 40 F. Cheng, J. Zhao, W. Song, C. Li, H. Ma, J. Chen and P. Shen, *Inorg. Chem.*, 2006, **45**, 2038–2044.
- 41 C. Wang, Y. Zeng, X. Xiao, S. Wu, G. Zhong, K. Xu, Z. Wei, W. Su and X. Lu, *J. Energy Chem.*, 2020, **43**, 182–187.
- 42 X. Zhang, P. Yu, D. Zhang, H. Zhang, X. Sun and Y. Ma, *Mater. Lett.*, 2013, **92**, 401–404.
- 43 D. Su, H.-J. Ahn and G. Wang, *J. Mater. Chem. A*, 2013, **1**, 4845–4850.
- 44 A. Coelho, *J. Appl. Crystallogr.*, 2018, **51**, 210–218.
- 45 G. Caglioti, A. Paoletti and F. P. Ricci, *Nucl. Instrum. Methods*, 1958, **3**, 223–228.
- 46 O. Schilling and J. Dahn, *J. Appl. Crystallogr.*, 1998, **31**, 396–406.
- 47 J.-R. Hill, C. M. Freeman and M. H. Rossouw, *J. Solid State Chem.*, 2004, **177**, 165–175.
- 48 D. Balachandran, D. Morgan, G. Ceder and A. van de Walle, *J. Solid State Chem.*, 2003, **173**, 462–475.
- 49 P. Ruetschi, *J. Electrochem. Soc.*, 1984, **131**, 2737.
- 50 T. X. Sayle, C. R. A. Catlow, R. R. Maphanga, P. E. Ngoepe and D. C. Sayle, *J. Cryst. Growth*, 2006, **294**, 118–129.
- 51 A. S. Anker, M. Juelsholt, S. L. Skjærvø, I. Kantor, D. R. Sørensen, S. J. L. Billinge and R. Selvan, *ChemRxiv*, 2022, DOI: [10.26434/chemrxiv-2022-v26431vfx](https://doi.org/10.26434/chemrxiv-2022-v26431vfx).
- 52 T. Chen and C. Guestrin, Presented in part at the Proceedings of the 22nd ACM SIGKDD International Conference on Knowledge Discovery and Data Mining, San Francisco, California, USA, 2016.
- 53 S. M. Lundberg, G. Erion, H. Chen, A. DeGrave, J. M. Prutkin, B. Nair, R. Katz, J. Himmelfarb, N. Bansal and S.-I. Lee, *Nat. Mach. Intell.*, 2020, **2**, 56–67.
- 54 S. M. Lundberg and S.-I. Lee, Proceedings of the 31st International Conference on Neural Information Processing Systems, 2017, pp. 4765–4774.
- 55 M. Leoni, T. Confente and P. Scardi, *Z. Kristallogr.*, 2006, 249–254.
- 56 W. Sun, D. A. Kitchaev, D. Kramer and G. Ceder, *Nat. Commun.*, 2019, **10**, 573.

