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Trifluoroacetimidoyl halides: a potent synthetic origin

The preparation of fluorine-containing compounds has attracted considerable attention due to the important applications of their related chemicals. Among the candidates, trifluoroacetimidoyl halides are considered as potent trifluoromethyl synthons to construct a wide variety of trifluoromethyl-containing compounds and trifluoromethyl-substituted N-heterocycles, which have found extensive applications in the fields of organic synthesis, pharmaceuticals, agrochemicals, and materials science. In the review, recent advances in the synthetic applications of trifluoroacetimidoyl halides are summarized. We specially focused on two different reaction modes upon trifluoroacetimidoyl halides, namely, coupling and annulation reactions to illustrate their synthetic applications and potentials in the construction of valuable trifluoromethyl-containing molecules. Their preparations were covered as well.

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Introduction

The properties of organic compounds with C-F bonds including electronegativity, bioavailability, metabolic stability and lipophilicity have gained significant interest. 1 It is known that approximately 20-25% of the current approved drugs and 30-40% of agrochemicals contain at least one fluorine atom.² Therefore, the incorporation of fluorine atoms or fluorinated groups into building blocks to assemble fluorine-containing compounds has emerged as a major research focus.³ Among these, considerable efforts have been devoted to construct trifluoromethylated compounds,4 which have raised widespread concerns in the fields of organic synthesis, pharmaceuticals, agrochemicals, and materials science.⁵

The most popular pathway to access trifluoromethyl-containing molecules involves transition-metal-catalyzed or metalfree trifluoromethylation reactions with versatile trifluoromethyl reagents such as Togni's reagent, 4b,6 Umemoto's reagent, Ruppert-Prakash reagent and Langlois reagent. 3f,9 Besides, another mainstream approach is the reaction of diverse trifluoromethyl-containing synthons with suitable



Zhengkai Chen

Zhengkai Chen was born in 1988 in Henan, China. He obtained his B.S. degree from Zhengzhou University in 2010. He had received a Ph.D. degree at Zhejiang University under the supervision of Prof. Yuhong Zhang in 2015. Then he joined the Department of Chemistry, Zhejiang Sci-Tech University. His current research interests focus on transition metal-catalyzed carbonylative transformations and new methods for the synthesis of functionalized nitrogencontaining heterocycles.



Sipei Hu

Sipei Hu was born in 1995 in Zhejiang, China. He obtained his B.S. degree from Heilongjiang University of Science and Technology in 2017. He is currently studying towards a M.S. degree at Zhejiang Sci-Tech University under the supervision of Dr Zhengkai Chen and Prof. Xiao-Feng Wu. His current research interests focus on transition metal-catalyzed or metalfree synthesis of nitrogen-containing heterocycles.

^aDepartment of Chemistry, Zhejiang Sci-Tech University, Hangzhou 310018, China. E-mail: xiao-feng.wu@catalysis.de

^bLeibniz-Institut für Katalyse e.V. an der Universität Rostock, Albert-Einstein-Straße 29a, 18059 Rostock, Germany

coupling substrates. It is worth mentioning that 2,2,2-trifluorodiazoethane (CF₃CHN₂) is recognized as the most frequently used trifluoromethyl-containing synthon. 10 Trifluoroacetimidoyl halides, which seemingly have not received enough attention, are also usually applied as useful building blocks for the assembly of trifluoromethyl-substituted scaffolds in synthetic organofluorine chemistry. The present review highlights the recent advancements enabling the synthesis of diverse trifluoromethyl-containing molecules by using trifluoroacetimidoyl halides and their related compounds.

Preparation of trifluoroacetimidoyl halides

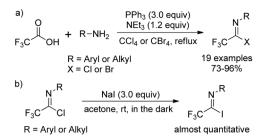
The traditional methods for the synthesis of trifluoroacetimidoyl halides are based on two pathways. One is the reaction of N-substituted trifluoroacetamides with phosphorus pentachloride (PCl₅)¹¹ and the other one is the thermal-induced or copper-catalyzed addition of perfluoroalkyl iodide to isocyanides. 12 However, the above-mentioned methods suffer from poor yields, harsh reaction conditions and not readily available reagents. Until 1993, Uneyama and co-workers developed an efficient and streamlined one-pot approach for the synthesis of trifluoroacetimidoyl halides in high yields (Scheme 1, eqn (a)). 13 Trifluoroacetimidoyl chlorides/bromides can be afforded by heating a mixture of trifluoroacetic acid and a primary amine in CCl₄/CBr₄ in the presence of PPh₃ and NEt₃. Trifluoroacetimidoyl iodides can be easily prepared by the substitution of the chloride group of trifluoroacetimidoyl chlorides by iodide ions, in which almost quantitative yields are achieved (Scheme 1, eqn (b)). In 2005, Saidi and co-workers



Xiao-Feng Wu

Xiao-Feng Wu was born in China. He studied chemistry in Zhejiang Sci-Tech University (China), where he got his bachelor's degree in science (2007). In the same year, he went to Rennes 1 University (France) and earned his master's degree in 2009. Then he joined Matthias Beller's group in Leibniz-Institute for Catalysis (Germany), where he completes his PhD defense in January 2012. Subsequently, he started his independent research

at LIKAT and ZSTU, where he was promoted to professor in 2013. In March 2017, Xiao-Feng defended his Habilitation successfully from Rennes 1 University (France). Xiao-Feng has authored >290 publications in international journals, meanwhile he is also the editor or author of 10 books. In 2019, Xiao-feng was invited to publish his Author Profile in Angew. Chem., Int. Ed. (Angew. Chem., Int. Ed., 2019, 58, 8624).



Scheme 1 The preparation of trifluoroacetimidoyl halides.

further applied the reaction to prepare trifluoroacetimidoyl chlorides containing additional functional groups. 14 By using this method, a wide range of structurally diverse trifluoroacetimidovl halides could be readily obtained in moderate to excellent yields. In addition, other perfluoroalkaneimidoyl chlorides were also smoothly prepared by the same method using the corresponding perfluoroalkanoic acids as starting materials.

Coupling reaction of trifluoroacetimidoyl halides

The trifluoroacetimidoyl halides can be applied as useful synthetic blocks to participate in diverse coupling reactions with suitable coupling partners for the construction of a range of trifluoromethyl-containing compounds.

3.1. C-C bond formation

As early as in 1991, Uneyama and co-workers developed a palladium-catalyzed coupling reaction of trifluoroacetimidoyl iodides with olefins and 1-alkynes to vield trifluoromethylated α,β-unsaturated imines (Scheme 2a and b). 15 It was found that trifluoroacetimidoyl iodides exhibit higher reactivity and

a)
$$Pd_2(dba)_3 \bullet CHCI_3 (5 mol\%)$$
 $Pd_2(dba)_3 \bullet CHCI_3 (5 mol\%)$ $Pd_3(dba)_3 \bullet CHCI_3 (6 m$

Scheme 2 Pd-Catalyzed coupling of trifluoroacetimidoyl iodides with olefins, 1-alkynes and arylboronic acids.

undergo palladation process smoothly to form imidoylpalladium intermediates, while the corresponding chlorides remain intact under the same conditions. The transformation tolerates diverse 1-alkenes and 1-alkynes, whether activated or non-activated ones. The resultant trifluoromethylated α,β-unsaturated imines could be easily transformed into nitrogen-containing heterocycles bearing a CF₃ group.

A palladium-catalyzed Suzuki cross-coupling reaction of bromodifluoroacetimidoyl halides with arylboronic acids to synthesize bromodifluoromethyl ketimines was realized by Wu and co-workers (Scheme 2c).16 The bromodifluoroacetimidoyl halides were prepared through the reaction of bromodifluoroacetic acid, triphenyl phosphine, arylamine and triethylamine in tetrachloromethane. In the reaction, the reactivity of bromodifluoroacetimidoyl chloride was found to be moderate even at prolonged reaction time.

In 1992, Uneyama and co-workers reported an efficient route to the formation of α -imino perfluoroalkanoates via palladium-catalyzed carboalkoxylation of fluorinated imidoyl iodides with various alcohols (Scheme 3).17 The obtained α-imino perfluoroalkanoates can be further transformed into α-amino perfluoroalkanoic acids. The mechanism of the carboalkoxylation reaction was considered to be initiated by the oxidative addition of palladium(0) to carbon-halide bonds, followed by the insertion of carbon monoxide into the palladium-carbon bond of 3-A to give an acylpalladium(II) intermediate 3-B. Then, a nucleophilic attack of the alcoholic hydroxyl group to 3-B can afford ester products, and the active palladium catalyst will be regenerated under the assistance of a base to accomplish the catalytic cycle. However, the reaction proceeded sluggishly with bulky alcohols, such as tert-butyl

In 2000, a modified protocol upon the reaction was proposed by the same group (Scheme 4).18 The efficiency of the reaction was improved by the addition of aprotic polar solvents (DMF or DMI) as an additive. It was speculated that the nucleophilicity of tert-butyl alcohol could be enhanced in the presence of DMF or DMI (1,3-dimethyl-2-imidazolidinone). However, the polar solvent might act as a weak ligand to reduce the aggregation of the intermediate palladium species and to accelerate the coordination of carbon monoxide to pal-

Scheme 3 Palladium-catalyzed alkoxycarbonylation of fluorinated

Scheme 4 Palladium-catalyzed tert-butoxycarbonylation of trifluoroacetimidoyl iodides and the synthesis of fluorinated α -amino acids.

ladium. More importantly, a series of fluorinated α -amino acids were afforded by the reaction of fluorinated iminoester with a range of nucleophiles and the subsequent removal of an N-protecting group (PMP) and an O-protecting group (t-Bu).

The highly enantioenriched β -fluorinated α -amino esters could be obtained by the asymmetric hydrogenation of α-fluorinated iminoester. 19 The transformation proceeded smoothly in the presence of a Pd(OCOCF₃)₂-BINAP complex in fluorinated alcohol solvents under hydrogen pressure with high efficiency and enantioselectivity.

Trifluoroacetimidoyl lithium, a synthetic equivalent of trifluoroacetyl carbanion, could be prepared from the iodinelithium exchange of N-aryl trifluoroacetimidoyl iodides with *n*-BuLi in ether at −78 °C (Scheme 5).²⁰ Nucleophilic reactions of the obtained imidoyl lithiums 5-A with various electrophiles took place to deliver trifluoromethyl-containing molecules 5-C. The 5-A species were very unstable and prone to isomerize to N-lithium carbene species 5-B, which underwent dimerization to afford diamine 5-D in the absence of other reactive electrophiles. Substituents on the N-aryl ring of trifluoroacetimidoyl iodides, solvent and reaction temperature greatly influenced the efficiency of nucleophilic reactions.

Uneyama and co-workers further investigated the generation and chemical properties of the trifluoroacetimidoyl lithiums (Scheme 6).21 The common imidoyl lithiums are stable only below -60 °C due to the highly ionic nature of a carbon-lithium bond. As for the benzoylation reaction with benzoyl chloride, the solvent of the reaction was complex and the lower temperature was beneficial for the reaction because the yield sharply dropped with the increase in temperature. Substituents on the aryl ring exerted a profound effect on the

Scheme 5 Generation and nucleophilic reactions of trifluoroacetimi-

Review

$$F_{3}C$$

$$Ph$$

$$F_{3}C$$

$$Ph$$

$$PhCOCI$$

$$F_{3}C$$

$$Ph$$

$$F_{3}C$$

$$F_{3$$

Scheme 6 Nucleophilic reactions of trifluoroacetimidoyl lithium with diverse electrophiles.

stability of the imidoyl lithium, where a large steric factor could promote the benzovlation reaction. Therefore, nucleophilic reactions of imidoyl lithium species bearing a 2,6-dimethylphenyl group with various electrophiles were tested to determine the scope and limitation of the transformation. Most of electrophiles could be compatible with the present reaction to give the corresponding products in moderate to good yields.

Uneyama's group also disclosed an approach to access trifluoroacetimidoyl zinc species, another trifluoroacetimidoyl metal, which were readily generated by the oxidative addition of imidoyl halides to activated zinc powder at room temperature (Scheme 7).22 In contrast to the imidoyl lithium mentioned above, the imidoyl zinc was more stable and could be handled at room temperature. The reaction was accelerated in the presence of ultrasound irradiated mixtures of zinc powders and aluminum powders in HMPA.

A metal-free trifluoroacetimidoyl carbanion was prepared by Uneyama and co-workers through the reaction of trifluoroacetimidoyl chlorides with trimethylsilyl cuprate and the subsequent fluoride ion-catalyzed generation of carbanions (Scheme 8).23 In the reaction, silyl cuprate was formed by the metal exchange of (trimethylsilyl)lithium with copper(1) and the replacement of lithium with tetraalkylammonium cation could stabilize the carbanion 8-B and promote the C-C bond

Scheme 7 The reaction of N-aryl trifluoroacetimidoyl zinc halide with benzaldehyde.

$$F_3$$
C CI $+$ Me $_3$ SiLi $\xrightarrow{\text{CuCN}}$ $\xrightarrow{\text{THF-HMPA}}$ F_3 C $\xrightarrow{\text{SiMe}_3}$ 0 °C to -55 °C $\xrightarrow{\text{8-A}}$ $\xrightarrow{\text{8-B}}$ $\xrightarrow{\text{R-CHO}}$ $\xrightarrow{\text{NY-Ar}}$ $\xrightarrow{\text{NY-Ar}}$ $\xrightarrow{\text{R-CHO}}$ $\xrightarrow{\text{NY-Ar}}$ $\xrightarrow{\text{NY-Ar}}$ $\xrightarrow{\text{R-CHO}}$ $\xrightarrow{\text{NY-Ar}}$ $\xrightarrow{\text{NY-Ar}}$ $\xrightarrow{\text{R-B}}$ $\xrightarrow{\text{R-B}}$

Scheme 8 Preparation of (trifluoroacetimidoyl)trimethylsilanes and the reaction with electrophiles.

formation at the imine carbon. With the employment of tetrabutylammonium as a counter cation, the stability of the imidoyl carbanions was enhanced and the reaction could be performed at room temperature with relatively high efficiency.

In the presence of different metal species, trifluoroacetimidoyl halides are readily transformed into trifluoroimidoyl metals, which could be applied as versatile carbanion equivalents to react with most of the electrophiles. The stabilities of trifluoroimidoyl metals are quite different on the basis of the covalency between carbon and the metal, which is attributed to the electronegativity difference between carbon and the metals.24

Yuan and co-workers reported a simple and efficient onepot reaction for the synthesis of α,β -unsaturated trifluoromethyl ketones (Scheme 9).25 In this transformation, diethoxyphosphorylmethyllithium was in situ generated from diethyl methylphosphonate and butyllithium, which reacted with N-phenyltrifluoroacetimidoyl chloride at -70 °C and quenched to afford a mixture of imino phosphonate and enamine. The treatment of the mixture with butyllithium gave a lithium salt, followed by the reaction with aldehydes though Horner-Emmons olefination and subsequent acid hydrolysis to provide the α,β -unsaturated trifluoromethyl ketone products in good yields. The protocol represents a convenient and economic route to access α,β-unsaturated trifluoromethyl ketones in contrast to Ishihara's method, which requires a four-step

$$R^{1} \xrightarrow{P} (OEt)_{2} \xrightarrow{LDA (2.0 \text{ equiv})} \xrightarrow{F_{3}C} \xrightarrow{Ph} \xrightarrow{Ph} P(OEt)_{2}$$

$$R^{2} \xrightarrow{CHO} (EtO)_{2} \xrightarrow{N-Ph} \xrightarrow{N-Ph} \xrightarrow{N-Ph} \xrightarrow{Ph} P(OEt)_{2}$$

$$R^{2} \xrightarrow{F_{3}C} \xrightarrow{R^{1}} \xrightarrow{N-Ph} \xrightarrow{R^{2}} \xrightarrow{Ph} P(OEt)_{2}$$

$$R^{2} \xrightarrow{R^{1}} \xrightarrow{N-Ph} \xrightarrow{N-Ph} \xrightarrow{N-Ph} \xrightarrow{N-Ph} \xrightarrow{N-Ph} \xrightarrow{N-Ph} \xrightarrow{N-Ph} \xrightarrow{N-Ph} P(OEt)_{2}$$

$$R^{2} \xrightarrow{R^{1}} \xrightarrow{N-Ph} \xrightarrow{N$$

Scheme 9 Synthesis of α , β -unsaturated trifluoromethyl ketones.

N-substituted Scheme 10 Synthesis of trifluoroacetimidoyl ketones.

preparation of the imino phosphonate from pentafluoropropionic acid.26

The synthesis of diverse N-substituted trifluoroacetimidoyl aryl ketones by 1,3-dimethylimidazolium iodide-catalyzed aroylation of trifluoroacetimidoyl chlorides with aromatic aldehydes in the presence of sodium hydride was disclosed by Yuan and co-workers as well (Scheme 10).27 In the reaction, the aroyl anions attacked the C=N bond of the imidoyl chloride to yield trifluoroacetimidoyl ketones with the displacement of the chlorine atom. Uneyama and co-workers also reported a pathway to produce N-aryltrifluoroacetimidoyl aryl/alkyl ketone in the early stages, 21 where N-aryltrifluoroacetimidoyl lithium was generated by an iodine-lithium exchange reaction between N-aryltrifluoroacetimidoyl iodide and n-BuLi, and then trapped by the acyl chloride. The methodology suffered from low reaction temperatures (-100 °C to -78 °C) and limited substrate scope.

The plausible reaction mechanism was depicted in Scheme 11. The interaction of NaH on the C-2 hydrogen of 1,3dimethylimidazolium iodide formed the ylide (B-), which attacked the aldehyde to give an O-anion species 11-A. The latter could isomerize to carbanion 11-B, followed by the displacement of the chlorine of imidoyl chloride to afford the intermediate 11-C, which was treated with NaH to give a new O-anion intermediate 11-D. The release of the ylide (B⁻) from O-anion could deliver the final aroyl product.

$$Ar - CHO \qquad Ar - C - B \qquad Ar -$$

Scheme 11 Plausible reaction mechanism.

Scheme 12 Preparation of β -polyfluoroalkyl enaminones and their application in the synthesis of fluorinated heterocycles.

The nucleophilic substitution of N-aryl polyfluoroalkyl imidoyl iodides with the carbanion of an α-methyl ketone β-polyfluoroalkyl enaminones, which β-dicarbonyl equivalents to synthesize fluorinated heterocyclic compounds (Scheme 12).²⁸ Treatment of β-polyfluoroalkyl enaminones with an excess of hydrazine monohydrate could furnish 3-polyfluoroalkyl pyrazoles. 6-Polyfluoroalkyl pyrimidines were obtained in satisfactory yields by the reaction of β-polyfluoroalkyl enaminones with amidines.

As early as 1989, Uneyama and co-workers have reported the coupling reaction of trifluoroacetimidoyl chlorides with various carbon nucleophiles, including Grignard reagents and active methylene compounds.29 As for Grignard reagents, the alkylation of the trifluoroacetimidoyl chlorides by the replacement of the chlorides was faster than the addition of Grignard reagents to the product imines. With respect to active methylene compounds, the obtained products were enamines rather than imines because of the relative stability of the enamine derivatives.

The reaction of fluorinated imidoyl chlorides with lithium derivatives of enantio-pure methyl sulfoxides to produce chiral β-imino sulfoxides was described by Fustero and co-workers (Scheme 13).30 The reaction proceeded well with moderate to high efficiency under mild conditions and could be completed within 20 minutes. The configurational stability of tautomers and geometric isomers of β-imino sulfoxides 13-B were also investigated. Notably, some β-imino sulfoxides were afforded via the aza-Wittig reaction of γ-fluoro-β-keto sulfoxides with N-aryl imino triphenylphosphoranes.

Scheme 13 Synthesis of fluorinated chiral β -imino sulfoxides

Scheme 14 Synthesis fluorinated vinvlogous amidines and β-enamino ketones.

Scheme 15 Synthesis of fluorinated β-enamino ketones and 1,3-dicarbonyl derivatives

Fustero and co-workers reported an efficient method for the assembly of fluorinated vinylogous amidines starting from imidoyl chlorides and enolizable ketimines (Scheme 14).³¹ In the presence of two equivalents of LDA, the fluorinated 1,3diimine products 14-A were formed, which could be exclusively isolated as vinylogous amidine tautomers 14-B. The reaction scope was broad, as demonstrated by the compatibility of structurally various ketimines and even chiral substituents at one or both nitrogen atoms. Regioselective hydrolysis mediated by 6 N H₂SO₄ occurred to release the corresponding fluorinated β-enamino ketones 14-D, where only the imine group bearing the fluoroalkyl substituent (R_F) could be regioselectively hydrolyzed (Scheme 14).

The fluorinated β-enamino ketones could also be generated by the reaction of imidoyl chlorides and ketones with the modification of the reaction temperature. The isolated fluorinated β-enamino ketones exist as a mixture of enamino-imino tautomers 15-A and 15-B, which could be readily hydrolyzed to yield the corresponding 1,3-dicarbonyl derivatives in the enol form. However, the β-enamino ketones 14-D remain unchanged under the same hydrolysis conditions, probably due to the strong intramolecular hydrogen bond (R_F) O···H-N (Scheme 15).

Three years later, the same group developed a two-step route to construct racemic and/or chiral nonracemic γ-fluorinated-β-amino acid derivatives (16-D/E) through reduction of fluorinated β-enamino esters 16-C, which could be prepared from the coupling of fluorinated imidoyl chlorides

Scheme 16 Stereoselective synthesis of fluorinated β -amino acids.

and lithium enolates of alkyl esters (Scheme 16).32 In order to achieve good yield and chemo- and stereoselectivity of the fluorinated β-amino acids, optimization of the reduction conditions was performed. The optimal condition was the combination of an excess of anhydrous ZnI2 and NaBH4 in dry dichloromethane at room temperature. Mechanistic investigation implied that a metal-chelated six-membered model was involved in the reaction and the choice of (-)-8-phenylmenthol as a chiral auxiliary could enable the best diastereoselectivity to reduce the chiral β-enamino esters.

Fustero and co-workers explored a simple route to the synthesis of chiral nonracemic fluorinated β-amino sulfones by the employment of imidoyl halides as fluorinated building blocks (Scheme 17).³³ The reaction of fluorinated chiral imidoyl chloride with sulfonyl carbanion could afford chiral fluorinated β-imino sulfones as a tautomeric imino/enamino mixture in high yields. Then, the stereoselective reduction occurred to give the corresponding β-amino sulfones with NaBH₄ as the reducing agent. The isolation of the predominant diastereoisomer through chromatographic separation and the subsequent hydrogenolysis over Pd/C produced enantiomerically pure β-amino sulfones.

$$\begin{array}{c} \text{Ar} & \text{Me} \\ \text{N} & \text{H} \\ \text{N} & \text{H} \\ \text{R}_{F} & \text{Cl} \\ \end{array} + \begin{array}{c} \text{O} & \text{O} \\ \text{Me} & \text{SAr} \\ \end{array} \begin{array}{c} \text{LDA} \ (2.0 \ \text{equiv}) \\ \text{THF}, -78 \ ^{\circ}\text{C} \\ \text{68-93\%} \\ \end{array} \begin{array}{c} \text{R}_{F} & \text{SAr} \\ \text{(tautomers)} \\ \end{array} \\ \begin{array}{c} \text{Me} \\ \text{Me} \\ \text{NH} \\ \text{SO}_{2}\text{Ar} \\ \text{NH} \\ \text{SO}_{2}\text{Ar} \\ \end{array} \begin{array}{c} \text{Me} \\ \text{NH} \\ \text{SO}_{2}\text{Ar} \\ \text{MeOH} \\ \text{V80-97\%} \\ \end{array} \\ \begin{array}{c} \text{NH}_{2} \\ \text{SO}_{2}\text{Ar} \\ \text{NH}_{2} \\ \text{SO}_{2}\text{Ar} \\ \end{array} \\ \begin{array}{c} \text{NH}_{2} \\ \text{SO}_{2}\text{Ar} \\ \end{array}$$

Scheme 17 Asymmetric synthesis of fluorinated β -amino sulfones.

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$$\begin{array}{c} R \searrow NH \\ R_F & \times SO_2Ar \end{array} \begin{array}{c} \text{i) n-BuLi (2.0 equiv)} \\ -78 \, ^\circ\!\!\!\!\!^\circ\!\!\!\!^\circ\!\!\!\!^\circ C, 2 \, h, THF \\ \text{ii) CICH}_2I (1.0 equiv) \\ \text{iii) MeLi (1.0 equiv)} \\ \text{iv) NH}_4CI \ \text{sat.} \end{array} \begin{array}{c} R \searrow NH \\ F_F & \times SO_2Ar \\ \end{array} \\ = 4\text{-OMePh} \\ R_F = CF_3, CF_2CI, CF_2Ph \\ R_F & \times SO_2Ar \\ Li \\ \end{array}$$

Scheme 18 Asymmetric synthesis of allylic amines.

The developed strategy was also applied to synthesize enantiomerically pure α-fluoroalkyl allylic amines, which were regarded as versatile building blocks (Scheme 18). A Julia-type methylenation-desulfonylation reaction of β-amino sulfones was involved in the reaction. In the presence of *n*-butyllithium and the consecutive addition of ClCH2I and methyllithium, the organolithium intermediate 18-A was formed, which underwent β-elimination process to furnish fluorinated allylic amines in good yields. It is worth mentioning that the author described a new synthesis of fluorinated allylamines through the reaction of 2-(trimethylsilyl)ethyl sulfones and sulfoxides with imines and imino esters, which utilized TBAF to enable the fragmentation of 2-(trimethylsilyl)ethyl sulfones to release the desired allylic amines.³⁴

Uneyama and co-workers presented a palladium(0)-catalyzed reductive dimerization of the fluorinated imidoyl iodides in the presence of carbon monoxide to yield fluorinated α-diimines, which were considered as a new class of catalyst ligands (Scheme 19).35 In the transformation, carbon monoxide acted as a reducing agent for regeneration of the catalytically active Pd(0) species. The α-diimines could be flexibly designed and synthesized to meet the catalytic activity through tuning of the electronic and steric properties of the N-substituent of imine moieties. Notably, a chiral diimine ligand could also be generated by the use of this method.

The fluorinated α -diimines can also be produced from samarium iodide-mediated reductive coupling of imidoyl iodide and the subsequent oxidation by manganese dioxide (Scheme 20).36 X-ray diffraction analysis showed the imine groups in the E,E-geometry, with a substantial nonplanarity of

Scheme 19 Carbon monoxide-promoted reductive homocoupling of fluorinated imidoyl iodides.

Scheme 20 Sm-Mediated reductive coupling of trifluoroacetimidoyl iodide.

$$\begin{array}{c} \text{R}_{F} \times \text{R} + = \text{R} \times \frac{\text{Cul (10 mol\%)}}{\text{CH}_{3}\text{PO}_{4} (1.2 \text{ equiv})} \\ \text{R}_{F} = \text{R} \times \text{N}, \text{Ph, CO}_{2}\text{Me, TMS} \\ \text{R}_{F} = \text{CF}_{3}, \text{CF}_{2}\text{Br} \\ \text{(X = Cl, I)} \end{array}$$

Scheme 21 Cu(ı)-Catalyzed coupling reactions of fluorinated imidoyl halides with terminal alkynes.

the N=C-C=N moiety. Therefore, the resulting α -diimines failed to form chelate complexes of palladium and platinum.

In 2011, Wu and co-workers demonstrated a Cu(1)-catalyzed coupling reaction of fluorinated imidoyl halides with terminal alkynes to construct fluorinated alkynyl imines (Scheme 21).³⁷ Compared with a similar work of synthesizing fluorinated alkynyl imines via a Pd/Cu-catalyzed coupling reaction of bromodifluoroacetimidoyl iodide with terminal alkynes previously reported by the same group,³⁸ the reaction could be realized by a Cu(1) catalyst only. The transformation proceeds smoothly with respect to bromodifluoromethylated imidoyl halides and trifluoromethylated imidoyl halides, whereas the latter required stoichiometric amounts of Cu(1). The plausible pathway is initiated by the insertion of the generated copper alkynide 21-A to C-X bond of fluorinated imidoyl halides. The intermediate 21-B is herein formed, followed by the reductive elimination to deliver fluorinated alkynyl imines.

It is worth noting that the fluorinated alkynyl imines were applied to synthesize fluorinated propargylamines via selectively asymmetric reduction. In the transformation, phosphoric acid- and ruthenium-catalyzed chemoselective biomimetic hydrogenation of the carbon-nitrogen double bond of fluorinated alkynyl ketimines with the vicinal carbon-carbon triple bond intact was achieved by Zhou and co-workers in 2016.³⁹

The non-hydrogen-bridged 6-arylamino-3-imidoylfulvenes 22-B/C were rapidly prepared from the regioselective reaction of sodium cyclopentadienide with electrophilic trifluoroacetimidoyl chlorides (Scheme 22).40 The reaction was kinetically controlled and could be completed in two hours at room tempReview

Scheme 22 Synthesis of 6-arylamino-3-imidoylpentafulvenes.

erature. The monosubstitution of cyclopentadienide by an imidoyl chloride occurred to give imidoylcyclopentadiene, which was further deprotonated by another sodium cyclopentadienide to deliver a sodium imidovlcvclopentadienide 22-A. Electrophilic attack of trifluoroacetimidoyl chlorides at the less hindered 3-position of the 22-A furnished 1,3-bis-(imidoyl) cyclopentadienes 22-B/C with no intramolecular N-H-N hydrogen bond. The 1,3-bis(imidoyl)-pentamethylruthenocene metalloligands 22-D were generated from the deprotonation of 22-B/C by n-butyllithium and the subsequent reaction with Cp*Ru(CH₃CN)₃PF₆.

3.2. C-N bond formation

Uneyama and co-workers reported the preparation and thermolysis of N-[2,2,2-trifluoro-1-(tritylazo)ethylidene]anilines and the related azo compounds (Scheme 23).41 The reaction of imidoyl chlorides with hydrazine monohydrate could give 23-A almost quantitatively, which reacted with Ph3CCl and underwent oxidative dehydrogenation in the presence of Pb(OAc)₄ to deliver tritylazo-compounds 23-C. Imidoyl radicals 23-D could be generated via thermolysis of tritylazo-compounds 23-C. The kinetics of thermal decompositions and electronic effects of substituents on the N-aryl group were investigated. The N-aryl-2,2,2-trifluoroacetimidoyl radicals, generated from the photochemical homolysis of the imidoyl iodides or the thermal

$$F_{3}C \xrightarrow{\text{NH}_{2}\text{NH}_{2} \bullet \text{H}_{2}\text{O}} F_{3}C \xrightarrow{\text{NH}_{2}\text{NH}_{2} \bullet \text{H}_{2}\text{O}} F_{3}C \xrightarrow{\text{NH}_{3}\text{NH}_{2}} F_{3}C \xrightarrow{\text{NH$$

Scheme 23 Preparation and thermolysis of N-[2,2,2-trifluoro-1-(tritylazo)ethylidene]anilines.

Scheme 24 Synthesis of N-aryltrifluoroacetimidoyl phthalimide and succinimide.

homolysis of imidoyl azo-compounds, were smoothly utilized to synthesize 2-trifluoromethylated indole and quinoline derivatives by the same author.42

Saidi and co-workers demonstrated a base promoted synthesis of N-aryltrifluoroacetimidoyl phthalimide and succinimide from the C-N bond coupling reaction of trifluoroacetimidoyl chlorides with phthalimide and succinimide (Scheme 24).43 In the reaction, the chlorine was readily replaced with nitrogen nucleophiles in the presence of NaH under mild conditions.

Wu and co-workers described a nucleophilic substitution reaction of imidoyl chlorides with different amines to synthesize α -fluoro-substituted amidines in good to excellent yields (Scheme 25).44 With regard to the active amine, an addition-elimination route was involved in the reaction. As for unreactive amines, the coupling promoted by palladium is necessary. The obtained fluorinated amidine could be used as a component to perform Hantzsch three-component synthesis in the presence of ZnCl₂, and 1,4-dihydropyridine product was delivered in reasonable yields.

Darehkordi and co-workers applied N-aryltrifluoroacetimidoyl chloride as N-linkers to quinolones to construct diverse trifluoroacetimidoyl piperazinylquinolone derivatives (Scheme 26).⁴⁵ The trifluoromethyl group was successfully incorporated via the coupling reaction of trifluoroacetimidoyl chlorides with norfloxacin, ciprofloxacin and enoxacin under mild conditions. The selected trifluoroacetimidoyl ciprofloxacin and norfloxacin were found to display promising antibacterial activities.

$$\begin{array}{c} N \stackrel{Ar}{\longrightarrow} R_F = NH_2 \\ R = Aryl, Alkyl \\ R_F = CF_3, CF_2Br \\ \hline \\ N \stackrel{Ar}{\longrightarrow} \\ R_F = CF_3, CF_2Br \\ \hline \\ N \stackrel{Ar}{\longrightarrow} \\ CI \longrightarrow CHO \\ \end{array}$$

Scheme 25 Synthesis of α -fluoro-substituted amidines from imidoyl chlorides

Scheme 26 Synthesis of trifluoroacetimidoyl piperazinylquinolone derivatives

Scheme 27 Synthesis of different indole trifluoromethyl derivatives.

The reaction of indole molecules with different trifluoroacetimidoyl chlorides in the presence of NaH as a base to construct new trifluoromethylated indole derivatives was developed by the same group (Scheme 27).46 In the coupling reaction of common indole with trifluoroacetimidoyl chlorides, the nitrogen of pyrrole acted as a nucleophile to displace chlorine to provide the corresponding product 27-A. With regard to less reactive indole-3-carbaldehyde, the carbon atom at position 2 served as the nucleophile to yield the product 27-B. When indole-3-carboxylic acid was used, trifluoromethylated 4,9-dihydropyrano[3,4-b]indol-3(1H)-one 27-C was produced. The cyclization process was achieved by the intramolecular nucleophilic addition of oxygen of carboxylic acid into the carbon-nitrogen double bond of fluorinated imines.

After three years, the same author reported a similar method for the synthesis of trifluoromethylated benzimidazole, imidazole and benzothiazole derivatives via a base-promoted coupling reaction of trifluoroacetimidoyl chlorides with the corresponding N-heterocycles (Scheme 28).47 As for the benzothiazole derivatives with unequal N and S bidentate nucleophilic sites, only an N-substituted isomer was obtained in good yields in the presence of NEt₃.

3.3. C-O(S) bond formation

Benzyl N-phenyl-2,2,2-trifluoroacetimidate, which served as a new O-benzylation reagent, was reported by Yamada and coworkers (Scheme 29).48 The O-benzylation reagent was readily prepared by the NaH-promoted coupling reaction of benzyl alcohol with trifluoroacetimidoyl chlorides. The present tri-

Scheme 28 Synthesis of a new α -trifluoromethyl-substituted formamidine framework.

$$F_3C$$
 CI $+$ BnOH $\frac{\text{NaH (1.0 equiv)}}{\text{CH}_2\text{Cl}_2, 0 °C to rt}$ F_3C OBn $+$ R-OBn $+$ R= Alkyl $+$ R-OH $+$ R-OH

Scheme 29 Synthesis and application of a stable reagent for O-benzylation.

fluoroacetimidate was very stable and soluble in various solvents, which could react with sterically hindered alcohols and base-sensitive hydroxyl esters apart from common alcohols to deliver benzyl ethers in the presence of catalytic TMSOTf. The O-benzylation reagent features good stability and can be stored under air for two months without the loss of the reactivity.

Barroca-Aubry and co-workers also developed O-benzylation reagent PMB-NPTFA, which was prepared from p-methoxybenzyl alcohol and trifluoroacetimidoyl chlorides in the presence of NaH (Scheme 30). 49 The PMB-NPTFA was presented in crystalline form, stable and soluble in a variety of solvents. The corresponding PMB ethers could be rapidly afforded in high yields by Bi(OTf)3 catalyzed reactions of primary or secondary alcohols with PMB-NPTFA in the presence of conventional 4 Å MS. Acid- and base-sensitive protecting groups were tolerated under the mild reaction conditions.

Scheme 30 Synthesis of PMB-NPTFA and application in mild acid catalyzed etherification.

$$F_3$$
C CI F_3 C OR F_3

Scheme 31 Synthesis of N-aryl trifluoroacetimidate-based benzyl and allyl protecting group reagents.

Pohl and co-workers discovered O-allyl and O-benzyl N-(pnitrophenyl)trifluoroimidates for the addition of allyl and benzyl protecting groups (Scheme 31).⁵⁰ The powerful reagents 31-A/B were achieved by the exploration of the role of para-substituents on the balance between the stability and reactivity of the N-phenyl trifluoroacetimidates. With catalytic amount of TMSOTf as the promoter, the reaction of reagents 31-A/B with different alcohols to give the benzylation or allylation products in high yields compared with the less stable allyl and benzyl trichloroacetimidate reagents.

It should be noted that trifluoroacetimidoyl chlorides were also successfully applied as mediators to promote the C-O bond formation in the synthesis of tetralactosaminyl O-glycothreonine^{51a} and Lewis A (Le^a) tandem repeat^{51b} under similar conditions.

Kuniyasu, Kambe and co-workers demonstrated a Pd-catalyzed iminothiolation of alkynes by using trifluoromethylated iminosulfides to construct 4-sulfur-functionalized 1-azadienes (Scheme 32).⁵² The iminosulfides could be readily prepared by the base-mediated coupling of trifluoroacetimidoyl chlorides

$$F_{3}C \xrightarrow{\text{CI}} + R^{1}\text{-SH} \xrightarrow{\text{NEt}_{3}} (2.0 \text{ equiv}) \xrightarrow{\text{benzene, reflux, 3 h}} F_{3}C \xrightarrow{\text{SR}^{1}} F_{3}C \xrightarrow{\text{SR}^{1$$

Scheme 32 Synthesis of trifluoromethylated iminosulfides and the application in iminothiolation of alkynes.

with thiols. In the iminothiolation reaction, the existence of the CF3 group was very essential for the successful transformation and the milder conditions enabled the addition of terminal alkynes compared with the internal alkynes. The reaction was initiated by the oxidative addition of iminosulfides to Pd (0) species to form Pd(II) complex 32-A, which underwent regio- and stereoselective insertion of alkynes into the Pd-S bond of 32-A to afford the intermediate 32-B. The final reductive elimination of 32-B resulted in the 1-azadiene derivatives with the release of Pd(0) complex.

3.4. C-P bond formation

The Yuan group developed an Arbuzov-type strategy for the preparation of 1-(N-aryl/alkyl amino)-2,2,2-trifluoroethylphosphonates, which was realized by the reaction of trialkyl phosphite with trifluoroacetimidoyl chlorides followed by subsequent reduction in the presence of sodium cyanoborohydride (Scheme 33).⁵³ Noteworthy was that low reactivity was observed by using sodium borohydride as a reducing agent, and N-substituents in the trifluoroacetimidoyl chloride exerted an obvious influence on the reaction. The experiment data implied that the transformation was initiated by the nucleophilic attack of trivalent phosphorus substrates on trifluoroacetimidoyl chlorides.

Onys'ko and co-workers reported a reaction of trifluoroacetimidoyl chlorides with trialkyl phosphites to give rise to N-aryltrifluoroacetimidoylphosphonates, in which existed as dynamic equilibrium mixture of Z,E isomers (Scheme 34).54 The obtained imidoylphosphonates easily reacted with O- or S-centered nucleophiles to yield the products of addition to the C=N bond. The trifluoroacetimidovlphosphonates could act as dipolarophiles to undergo cycloaddition reactions with

Scheme 33 Synthesis of 1-aminophosphonic acid derivatives bearing a trifluoromethyl moiety.

Scheme 34 Synthesis of N-aryltrifluoroacetimidoylphosphonates.

nitrile oxides, delivering the corresponding phosphorylated oxadiazolines, which included the fragment of aminophosphonic acid.

3.5. C-Si bond formation

In 2003, Uneyama, Amii and co-workers developed a Mg(0)/ Me₃SiCl system to promote double silylation of trifluoroacetimidoyl chlorides, providing bis-silylated difluoroenamines, which was regarded as a novel fluorinated dianion equivalent (Scheme 35). Initially, the Mg-mediated reductive cleavage of C–Cl bonds of the imidoyl chlorides occurred to give rise to imidoylmagnesium species 35-A, which reacted with chlorotrimethylsilane to afford imidoylsilanes 35-B. Subsequently, the two-electron reduction of the imidoylsilanes 35-B furnished the β -fluorinated organomagnesium species 35-C, followed by the β -elimination to produce bis-silylated difluoroenamines.

The resultant bis-silylated difluoroenamines could be applied as useful bifunctional synthetic blocks. The Lewis acid-catalyzed aldol-type reaction of bis(silyl)enamines with various aldehydes or aldimines in the presence of BF3·OEt2 in CH2Cl2 at 0 °C provided β -hydroxy- α , α -difluoro imidoylsilanes or β -amino- α , α -difluoro imidoylsilanes in good yields (Scheme 36). Notably, the bis(silyl)enamines could also react with two kinds of electrophiles to yield a variety of difluorinated imines.

A Mg-promoted selective activation of imidoyl carbon-halogen bonds to synthesize fluorinated imidoylsilanes was developed by the same group (Scheme 37).⁵⁶ When excess amounts of magnesium metal and TMSCl were used, the bis (silyl)enamine 35-D was selectively formed. In order to avoid

Scheme 35 Mg-Promoted double silylation of trifluoroacetimidoyl chlorides.

Me₃Si N Ar
$$R^{1}$$
 SiMe₃ R^{2} R^{1} SiMe₃ R^{2} R^{1} SiMe₃ R^{2} R^{1} SiMe₃ R^{2} R^{1} Aryl, Alkyl R^{2} = H, Me, 4-OMePh R^{1} H R^{2} R^{1} R^{2} R^{3} R^{4} R^{2} R^{4} R^{5} SiMe₃ R^{2} R^{4} R^{5} R^{5} R^{6} R^{5} R^{5} R^{6} R^{5} R^{5}

Scheme 36 The aldol-type reactions of bis-silylated difluoroenamine with aldehydes and ketones or aldimines.

Scheme 37 An improved synthesis of fluorinated imidoylsilanes.

further reduction involving the C-F bond cleavage, a reduced amount of magnesium metal and a low temperature should be implemented. Moreover, the employment of a catalytic amount of ethyl bromide to activate the surface of metallic magnesium was beneficial for the reaction. Except for trifluoromethyl imidoylsilanes, difluoromethyl, perfluoroalkyl and pentafluoroethyl imidoylsilanes could be prepared in moderate to good yields, which were applied as stable and useful fluorinated acyl anion equivalents.

3.6. Formation of trifluoroacetimidoyl $Pd(\pi)$ or $Ti(\pi)$ or $Pt(\pi)$ complexes

The intramolecular thiopalladation of thioanisole-substituted propargyl imines to construct perfluoroalkyl-substituted benzothiophene-based palladacycles under mild reaction conditions was described by Likhar and co-workers (Scheme 38).⁵⁷ The corresponding perfluoroalkyl propargyl imines could be prepared from Sonogashira coupling of perfluoroalkyl imidoyl iodides and alkynes. The structures of dimeric palladacycles were fully confirmed by 1 H/ 13 C NMR and IR. The protocol enables the formation of heterocyclic-based palladacycles in the absence of a heterocyclic ligand system.

The thiopalladacycles obtained from palladium catalysts and *ortho*-thioanisole-substituted propargyl imines were applied to undergo Heck-type coupling of alkenes for the synthesis of 3-alkenylbenzo[*b*]thiophene (Scheme 39).⁵⁸ The existence of imino moiety in the alkenylated benzothiophene **39-B** enabled the instability of the product, so further reduction by using a combination of NaBH₃CN in glacial acetic acid and methanol could give the secondary amine products **39-C**. Noteworthy was that the formation and involvement of the thiopalladacycle **39-A** was identified as the key step of the Heck-type coupling. The following alkene insertion and Pd–H elimination afforded the 3-alkenylbenzo[*b*]thiophene product.

Diverse fluorinated imidoyl palladium complexes were synthesized and their structures were identified by Uneyama,

$$R_{F} = CF_{3}, C_{4}F_{9}$$

$$PdCl_{2}(PPh_{3})_{2} (2 mol\%)$$

$$R_{F} = CF_{3}, C_{4}F_{9}$$

$$PdCl_{2}(PhCN)_{2} (1.0 equiv)$$

$$THF, 0 °C to rt$$

$$3 examples, 79-85\%$$

Scheme 38 Synthesis of benzothiophene-based palladacycles.

Review

Scheme 39 Synthesis of 3-alkenylbenzo[b]thiophenes.

$$F_{3}C \xrightarrow{N} X + Pd(PPh_{3})_{4} \xrightarrow{benzene} F_{3}C \xrightarrow{Pd} Pd \\ R = 4-OMePh, n-C_{6}H_{13} \\ + Pd_{2}(dba)_{3} \bullet CHCl_{3} \xrightarrow{92\%} F_{3}C \xrightarrow{N} PhSnBu_{3} \\ + Pd_{2}(dba)_{3} \bullet CHCl_{3} \xrightarrow{92\%} F_{3}C \xrightarrow{N} PMP$$

Scheme 40 Synthesis of trifluoroacetimidoyl palladium(II) complexes.

Amii and co-workers (Scheme 40).⁵⁹ Trifluoroacetimidoyl palladium(II) phosphine complexes 40-A were prepared from the oxidative addition reaction of trifluoroacetimidoyl iodide with Pd(PPh₃)₄ in benzene at room temperature. The obtained palladium(II) complexes 40-A could react with organotin compounds to afford trifluoromethylated imines in moderate yields. The phosphine-free trifluoroacetimidoyl palladium(II) complexes 40-B were generated from the reaction of imidoyl iodide with Pd₂(dba)₃·CHCl₃, whereas inferior reactive imidoyl chloride did not participate in the reaction. The present palladium(II) complex 40-B could be applied as a catalyst for carboalkoxylation.

The amino(trifluoromethyl)-carbene-palladium complex 41-**B** was formed by the treatment of imidoyl phosphine complex 41-A with trimethyloxonium tetrafluoroborate (Meerwein's reagent) (Scheme 41). The structure of complex 41-B was verified by the control experiments, which treated 41-B with an O2 atmosphere or an excess amount of elemental sulfur (S8) under heating conditions. The production of N-methyl-N-hexyl amide or N-methyl-N-hexyl thioamide indicated that an α-(dialkylamino)-α-trifluoromethylcarbene moiety was involved in the palladium complex 41-B. In addition, the reaction of imidoyl palladium(II) complex 41-A with methyl thioglycolate could provide the N,S-acetal of fluoral 41-C via insertion of carbene ligands into sulfur-hydrogen σ bonds, which underwent intramolecular amide formation to release thiazolinone 41-D.

$$F_{3}C Pd Ph_{3} PPh_{3} PPh$$

Scheme 41 Application of trifluoroacetimidoyl palladium(II) complexes.

F₃C

SeMe
$$\frac{PdCl_2(PhCN)_2}{THF, 0 \, ^{\circ}C, 2 \, h}$$

R¹
 R^2
 R^1
 $R^2 = Aryl$
 R^1
 $R^2 = Aryl$
 R^3
 R^4
 R^2
 R^4
 R^5
 R^6
 R^7
 R^7

Scheme 42 Synthesis of cyclopalladated benzoselenophene.

The cyclopalladated benzoselenophenes were readily constructed from the intramolecular cyclization of selenoanisolesubstituted propargyl imines in the presence of stoichiometric amounts of dichorobis(-triphenylphosphine)palladium(II) in dry THF (Scheme 42).60 The reactivity of cyclopalladated benzoselenophene complex was further explored via the insertion of different alkynes into the Pd-C bond of dimeric cyclopalladated benzoselenophenes to build benzoseleno[2,3-c]-pyridinones. The results demonstrated that the cyclopalladated heterocycles could be utilized to assemble various fused heterocycles through insertion reactions.

A library of titanium complexes bearing two regioisomeric trifluoromethyl-containing enaminoketonato ligands were synand characterized by Li and co-workers (Scheme 43).61 The treatment of trifluoroacetimidoyl chloride

Scheme 43 Synthesis of the titanium complex bearing trifluoromethylcontaining enaminoketonato ligands.

$$R_F$$
 XCH_3
 $PtCl_2$
 $Cl-Pt-CH_3$
 NAr
 $X = S$, Se
 $R_F = CF_3$, C_2F_5 , C_4F_9
 ArN
 ArN

Scheme 44 Synthesis of the titanium complex bearing trifluoromethylcontaining enaminoketonato ligands.

with heptanone in the presence of LDA could furnish the corresponding β-enaminoketone 43-A, which was transformed into lithium salt of β-enaminoketone 43-B to react with TiCl₄ to deliver the titanium complex 43-C. Other β -enaminoketones were obtained by the Claisen condensation between methyl ketones and ethyl trifluoroacetate and the subsequent condensation of the β -diketone with aniline under acidic conditions. The present titanium complexes were active towards ethylene polymerization with modified methylaluminoxane as a cocatalyst and the substituents attached on the ligands exerted a significant influence on the catalytic activities and the behavior of ethylene polymerization.

A monomeric platinum(iv) complex, recognized as highly stable octahedral bis-ligated platinacycle, was formed by the cycloplatination of ortho thio(seleno)anisole-substituted perfluoroalkyl propargyl imines with PtCl₂ (Scheme 44).⁶² The perfluoroalkyl propargyl imines were prepared by palladiumcatalyzed Sonogashira coupling of perfluoroalkyl imidoyl iodides and alkynes. The obtained platinum(IV) complexes were fully characterized by ¹H/¹³C NMR, IR and CHN analysis. The emission spectra of platinacycles were also tested. The reactivity of Pt(IV)-complexes was further investigated via the reaction with diphenyl acetylene or excess triphenyl phosphine, which produced benzo[4,5]-thieno-[2,3-c]-pyridine skeletons or monoligated Pt(II) complexes in quantitative yields.

3.7. Reduction of fluorinated imidoyl chlorides

Uneyama and co-workers reported a simple pathway for the preparation of polyfluoroalkylated aldimines through the reduction of polyfluoroalkyl imidoyl chlorides by the use of LTBA (lithium tri-tert-butoxyaluminum hydride) as a reducing agent (Scheme 45).63 Other different reducing agents, such as NaBH₄, LiAlH₄ or DIBAL-H, showed poor reactivity. A variety of polyfluoroalkylated imidoyl chlorides were smoothly compatible with the transformation to furnish the corresponding aldimine products, irrespective of the length of fluorinated alkyl chains.

Scheme 45 Synthesis of polyfluoroalkylated aldimines.

Annulation reaction of trifluoroacetimidoyl halides

The trifluoroacetimidoyl halides have been known as versatile synthons to react with different substrates for the assembly of structurally diverse trifluoromethyl-substituted N-heterocycles.

4.1. Synthesis of fluoroalkyl 5-membered N-heterocycles

The formal [1 + 4] cyclization reaction of N-aryl trifluoroacetimidoyl iodides with phenyl-hydrazones to access 5-trifluoromethyl pyrazoles was achieved by Huang and co-workers (Scheme 46).64 The reaction proceeded well with excess sodium hydride in anhydrous DMF at room temperature and the trifluoroacetimidoyl iodide was regarded as a CF3 carbon building block. With respect to perfluoroalkyl imidoyl iodides, the corresponding fluoroalkyl pyrazoles could not be provided under the standard conditions.

Yuan and co-workers developed a convenient method for the synthesis of 5-trifluoromethyl tetrazoles from the trifluoroacetimidoyl chlorides and NaN3 without any catalyst or additive (Scheme 47).65 The azide intermediate 47-A was formed by the nucleophilic replacement of trifluoroacetimidoyl chlorides with azide anion, and the subsequent intramolecular cyclization yielded the tetrazole products. The success of the reaction requires two factors: one is the cis configuration of the lone pair of the imino nitrogen and azido group and another is the sufficiently rich electron density on imino nitrogen.

An efficient and green strategy for the preparation of bioactive 5-trifluoromethyl tetrazoles via microwave-assisted 1,3dipolar cycloaddition reaction of trifluoroacetimidoyl chlorides and NaN3 was demonstrated by Darehkordi and co-workers (Scheme 48).66 Compared with the common reaction at room temperature, the microwave-assisted protocol provided a remarkable improvement in yields and the tetrazole products were easily purified by simply washing with *n*-hexane.

Wu and co-workers presented a gold(1)-catalyzed synthesis of 2-fluoroalkyl imidazoles from a 5-exo-dig cyclization of fluorinated propargyl amidines, which were obtained by the coupling of fluorinated imidoyl chlorides with propargyl amines (Scheme 49).67 The transformation proceeded

Scheme 46 Synthesis of 5-trifluoromethyl pyrazoles.

$$\begin{array}{c|c}
N \stackrel{R}{\longrightarrow} & NaN_3 \text{ (1.0 equiv)} & R \\
\hline
F_3C \stackrel{N}{\longrightarrow} & CH_3CN \text{ or DMF, rt} & F_3C \stackrel{N}{\longrightarrow} & F_3C \stackrel{N}{\longrightarrow} & N \\
R = Aryl, Alkyl & 12 examples, 50-94% & 47-A
\end{array}$$

Scheme 47 Synthesis of 5-trifluoromethyl tetrazoles.

Scheme 48 Synthesis of 5-trifluoromethyl tetrazoles.

Scheme 49 Synthesis of 2-fluoroalkyl imidazoles.

smoothly with regard to a series of fluorinated imidoyl chlorides. The proposed mechanism is outlined in Scheme 49. The nucleophile attack of amidinonyl nitrogen to the C–C triple bonds activated by cationic Au(I) led to intermediate 49-A, which isomerized to the more stable imidazole 49-B by an H-1,3-shift. The photolysis of 49-B could give rise to the imidazole products 49-C. Noteworthy was that when 3-phenyl propargyl amine was employed as a starting reagent, a 6-endo-dig cyclization occurred to provide 1,4-dihydropyrimidine product in 73% yield.

When the fluorinated propargyl amidine was treated with electrophile NIS in the presence of Ph₃PAuCl/AgBF₄, imidazole-5-carbaldehyde was produced (Scheme 50). The trans-

Scheme 50 Synthesis of 2-fluoroalkyl imidazole-5-carbaldehyde.

formation tolerated various functional groups and only terminal alkynes were suitable for the reaction. With 3-phenyl propargyl amine as a substrate, 5-iodo-1,4-dihydropyrimidine was delivered in moderated yields. Mechanistic study revealed that the carbonyl oxygen was derived from O₂. Activated propargyl amidine reacted with NIS affords 5-iodomethyl imidazole **50-B**, which underwent the dissociation of the C–I bond to give the radical intermediate **50-C** and an iodine radical. Subsequently, the peroxy-intermediate **50-D** was formed in the presence of O₂. The final aldehyde product was provided by the removal of a hydroxyl radical from **50-D**.

A three-component reaction of trifluoromethylimidoyl chlorides with isocyanides and dimethyl or diethyl acetylene dicarboxylate without any catalyst for the construction of trifluoromethylated pyrroles was reported by Darehkordi and co-workers (Scheme 51).⁶⁸ The reaction was initiated by the nucleophilic addition to DMAD (dimethyl acetylene dicarboxylate) to generate zwitterion 51-A, which underwent nucleophilic addition to the imidoyl carbon to give the intermediate 51-B. Intramolecular cyclization of 51-B delivered 51-C, followed by the elimination of chlorine and protonation to yield the pyrrole product.

Darehkordi and co-workers disclosed a TiO₂-nanoparticle-catalyzed synthesis of trifluoromethyl-4,5-dihydro-1,2,4-oxadi-azoles and trifluoromethyl-1,2,4-oxadiazoles from the reaction of trifluoroacetimidoyl chlorides and amidoximes (Scheme 52).⁶⁹ Changing the reaction condition regarding

$$R^{1}-NC + R^{2}OOC - COOR^{2} + R^{1}OCR^{2} + R^{2}OOC - COOR^{2} + R^{1}OCR^{2} + R^{1}OCR^$$

Scheme 51 Synthesis of trifluoromethylated pyrroles.

Scheme 52 Synthesis of trifluoromethyl-4,5-dihydro-1,2,4-oxadiazoles and trifluoromethyl-1,2,4-oxadiazoles.

$$F_{3}C$$

$$CI$$

$$R = (Hetero)AryI, AlkenyI$$

$$Ar_{N}$$

$$R_{3}C$$

$$R = (Hetero)AryI, AlkenyI$$

$$Ar_{N}$$

$$R_{3}C$$

$$R_{$$

Scheme 53 Synthesis of 5-trifluoromethyl-1,2,4-triazoles.

different trifluoromethyl-substituted solvent and base, N-heterocycle products were obtained. The reaction was considered to be initiated by the activation of imidoyl chlorides by TiO2-NPs to result in a TiO2-imidoyl complex, which was attacked by the nucleophilic oxygen atom of amidoxime. The following intramolecular attack of the -NH anion on the imino group afforded 5-(trifluoromethyl)-4,5-dihydro-1,2,4-oxadiazol-5-amine 52-A. The elimination of aryl amine from 52-A led to trifluoromethyl-1,2,4-oxadiazole 52-B.

Our group developed an efficient approach for the assembly of 5-trifluoromethyl-1,2,4-triazoles through iodine-mediated annulation of trifluoroacetimidoyl chlorides and hydrazones under mild reaction conditions (Scheme 53).70 The scope of the protocol was broad, as illustrated by more than 40 successful examples with moderate to high efficiency, including other fluorinated imidoyl chlorides. The reaction proceeded in a consecutive one-pot manner, which presumably involved basepromoted intermolecular C-N bond formation via coupling of trifluoroacetimidoyl chlorides and hydrazones to give the trifluoroacetamidine derivative 53-A. Then, the base-promoted oxidative iodination of 53-A furnished an iodo species 53-B and subsequent intramolecular nucleophilic attacks of nitrogen on the iodo-substituted carbon atom afforded the intermediate 53-C. The final base-promoted deprotonation and rearomatization of 53-C led to the desired 1,2,4-triazole product.

4.2. Synthesis of 2-fluoroalkyl benzimidazoles

An approach to produce N-substituted 2-(trifluoromethyl)benzimidazoles via electrochemical oxidation of N,N'-disubstituted trifluoroethanimidamides was disclosed by Uneyama and co-workers (Scheme 54).71 The benzimidazoles 54-A could be smoothly formed under electrolysis conditions in dry acetonitrile with respect to N,N'-diaryl derivatives, whereas complex mixtures of trifluoroethanimidamides, p-benzoquinone imine derivatives 54-B and polymeric compounds were observed with regard to N-alkyl-N'-(4-methoxyphenyl) derivatives. When the reaction was conducted in wet acetonitrile, p-benzoquinone imine derivative 54-B was the major product. A two-electron-oxidation process was considered to be involved in the reaction.

Scheme 54 Synthesis of N-substituted 2-(trifluoromethyl)benzimidazoles.

Scheme 55 Synthesis of trifluoromethylated 6-hydroxybenzimidazoles.

Based on the previously reported synthesis of 2-(trifluoromethyl)benzimidazoles, Uneyama and co-workers further investigated the conversion of p-benzoquinone imines into trifluoromethylated 6-hydroxybenzimidazoles (Scheme 55).72 The p-benzoguinone imines 55-A were obtained by electrochemical oxidation of 2,2,2-trifluoroethanimidamides. Lewis acid-catalyzed intramolecular cyclization of 55-A gave rise to the desired trifluoromethylated 6-hydroxybenzimidazole products, and BF₃·OEt₂ was the best choice. The scope of the reaction was relatively broad and different substituted p-benzoquinone imines were applicable to afford the desired products in reasonable yields.

One year later, a more detailed investigation on the electrochemical and the oxidant-promoted reactions of perfluoroalkyl/aryl imidamides for the preparation of 2-(perfluoroalkyl/ aryl)benzimidazoles was presented by the same group. 73 The protocol provided a deeper understanding of the transformation of imidamides into benzimidazoles.

Wu and co-workers developed a copper(1)-catalyzed tandem reaction for the synthesis of 2-fluoroalkylbenzimidazoles starting from N-aryl trifluoroacetimidoyl (or bromodifluoroacetimidoyl) chlorides and primary amines (Scheme 56).74 In the transformation, amidines could be easily formed via the coupling of imidoyl chlorides and amines. The subsequent intramolecular C-N bond formation was enabled in the presence of copper catalyst to deliver the 2-fluoroalkylbenzimidazole products. The employment of K₂CO₃ or K₃PO₄ was essential for the success of the reaction and ligands were unnecessary for improving the yield.

Review

$$\begin{array}{c} R^{1} \\ R_{F} \\ CI \\ R_{F} = CF_{3}, CF_{2}Br \\ R^{1} = CI, Me, NO_{2}, R^{2} = Aryl, Alkyl, Allyl \\ \hline \\ R_{N} \\ CF_{2}Br \\ \hline \\ R_{N} \\ CF_{2}Br \\ \hline \\ R_{N} \\ CF_{3} \\ R_{1} \\ R_{2} \\ R_{3} \\ R_{1} \\ R_{2} \\ R_{3} \\ R_{3} \\ R_{4} \\ R_{5} \\ R_{1} \\ R_{5} \\ R_{1} \\ R_{5} \\ R_{1} \\ R_{2} \\ R_{3} \\ R_{2} \\ R_{3} \\ R_{3} \\ R_{4} \\ R_{5} \\ R_{1} \\ R_{5} \\ R_{5} \\ R_{1} \\ R_{2} \\ R_{3} \\ R_{4} \\ R_{5} \\$$

Scheme 56 Synthesis of 2-fluoroalkylbenzimidazoles.

Scheme 57 Synthesis of 2-(trifluoromethyl)benzimidazoles.

In the meantime, Zhang and co-workers reported a similar reaction for the construction of 2-(trifluoromethyl)benzimidazoles by Cu(1)/TMEDA-catalyzed cross coupling reaction of N-(2-haloaryl)trifluoroacetimidoyl chlorides with primary amines (Scheme 57).⁷⁵ Different from the Wu's reaction, the protocol required suitable ligand to promote the reaction and the Cl, Br, or I substituent at the 2-position of the phenyl group of imidoyl chlorides were all tolerated under the reaction conditions, albeit the slightly inferior reactivity of the aryl bromides and arvl chlorides.

A practical route to the synthesis of 2-fluoroalkylbenzimidazoles via PIDA-mediated oxidative intramolecular cyclization of the fluorinated amidines was presented by Wu and co-workers (Scheme 58).⁷⁶ The fluorinated amidines were generated from the coupling of fluorinated imidoyl chlorides with gas of ammonia at room temperature. Compared with the previous reported methods, the current protocol avoided the use of ortho-aminoanilines and ortho-haloanilines as the starting materials.

4.3. Synthesis of 2-fluoroalkyl benzothiazoles

A copper-catalyzed thiolation annulation reaction of 1,4-dihalides with NaHS for the synthesis of biologically active 2-trifluoromethyl benzothiazoles was disclosed by Zhang and coworkers (Scheme 59).⁷⁷ In this Ullmann reaction, the Cu(1)catalyzed formation of double carbon-sulfur bonds was realized. Less active N-(2-chlorophenyl)-trifluoroacetimidoyl chlorides were also applicable to the reaction with unsatisfactory reactivity. Notably, the 2-trifluoromethyl benzothiophenes

Scheme 58 Synthesis of 2-fluoroalkylbenzimidazoles by PIDA oxidation of amidines

Scheme 59 Synthesis of 2-trifluoromethyl benzothiazoles.

Synthesis of 2-trifluoromethyl benzothiazoles.

could be afforded through the Cu(1)-catalyzed thiolation annulation of 2-chloro-1-(2-haloaryl)-3,3,3-trifluoropropylenes with Na₂S.

An improved method for the synthesis of 2-trifluoromethylbenzothiazoles was described by Wu and co-workers (Scheme 60).⁷⁸ The reaction proceeded with common trifluoroacetimidoyl chlorides and sodium hydrosulfide hydrate by using PdCl2 as catalyst. A thiolation/C-H bond functionalization/C-S bond formation sequence was involved in the transformation. It was envisaged that 2,2,2-trifluoro-N-phenylethanethioamide was the intermediate for the reaction, which could be in situ generated from the coupling reaction of trifluoroacetimidoyl chloride with sodium hydrosulfide hydrate.

4.4. Synthesis of 2-fluoroalkyl indoles

Uneyama and co-workers developed an electrochemical oxidation of 2-substituted 3-(N-arylamino)-4,4,4-trifluoro-2butenoate for the production of 2-(trifluoromethyl)-indoles

Scheme 61 Synthesis of 2-(trifluoromethyl)-3H-indoles.

(Scheme 61).⁷⁹ The butanoate derivatives **61-A** could be readily prepared by the nucleophilic reaction of active methylene compounds with trifluoroacetimidoyl chlorides. The 61-A underwent a sequential reaction of single-electron-oxidation, deprotonation, a further single-electron-oxidation, addition of water and demethoxylation process to afford the p-benzoquinone imines 61-B. With respect to different substitutions attaching on 61-B, different reaction conditions were implemented to convert p-benzoquinone imines 61-B into 2-(trifluoromethyl)-3H-indoles.

Hao and co-workers established a facile strategy for the assembly of biologically important 2-fluoroalkyl-substituted indoles *via* the Grignard cyclization reaction of *N*-[2-(bromoalkyl) phenyl]imidoyl chlorides (Scheme 62).80 The α -bromination at the benzylic position of N-[2-(alkyl)phenyl]imidoyl chlorides 62-A with N-bromosuccinimide (NBS)/benzoyl peroxide (BPO) vielded 62-B. The Grignard cyclization reaction proceeded well in the presence of magnesium and was greatly influenced by the electronic effect of the substituent group on the benzene ring. The synthesized 2-fluoroalkyl indoles could undergo

$$R^{2} \stackrel{\text{II}}{ \sqcup} \qquad R^{1} \qquad R^{1} \qquad R^{2} \stackrel{\text{II}}{ \sqcup} \qquad R^{2} \stackrel{\text{II$$

Scheme 62 Synthesis of 2-fluoroalkyl indoles from N-[2-(bromoalkyl) phenyl]imidoyl chlorides.

Scheme 63 Synthesis of 2-fluoroalkyl indoles and their application.

further derivatization to construct more complicated fluorinecontaining indoles, which may constitute the precursor of the valuable biologically active molecules.

The same group further investigated the Grignard cyclization reaction to synthesize 2-fluoroalkyl-substituted indole derivatives and extended their application in the synthesis of biologically interested indoles (Scheme 63).81 The Grignard reagents were in situ generated and transformed into 2-fluoroalkyl indole products. N-[2-(Chloroalkyl)phenyl]imidoyl chlorides were also suitable for the reaction. The application of 3-methyl-2-trifluoromethyl indole was implemented. The protection of nitrogen groups in 63-B with 4-chlorobenzoyl chloride gave 63-C, which underwent bromination at 3-methyl group with NBS/AIBN to deliver 63-D. The nucleophilic substitution of 63-D with NaCN enabled the formation of N-deprotected product 63-E, followed by the subsequent hydrolysis of the cyanide group to yield 2-trifluoromethylated heteroauxin 63-F. The nucleophilic substitution of 63-D with malonate carboanion resulted in the formation of 63-G in good yields.

A simple approach for the formation of 2-fluoroalkyl indoles from 2-aminobenzyl alcohols with fluoroalkanoic acids in the presence of Ph3P, CCl4, and NEt3 was developed by Wang and co-workers (Scheme 64).82 It was found that N-[2-(hydroxymethyl)phenyl]-2,2,2-trifluoroacetimidoyl chloride 64-

Scheme 64 Synthesis of 2-fluoroalkyl indoles and the plausible mechanism.

A was formed as an intermediate in the reaction. [Ph₃P⁺Cl] CCl₃⁻, in situ generated from the reaction of Ph₃P with a large excess amount of CCl4, could enable the transformation of a hydroxyl group into a chlorine atom, thereby giving the compound 64-B. The reaction of 64-B with Ph₃P formed a phosphonium salt 64-C, which was converted into active and unstable ylide 64-D. The final intramolecular cyclization of 64-**D** could lead to the indole product.

A visible-light-induced intramolecular radical cyclization of N-[2-(alkynyl)phenyl]trifluoroacetimidoyl chlorides for the assembly of 2-trifluoromethyl indoles was established by Zhou and co-workers (Scheme 65).83 In this transformation, imidoyl radical was generated through the irradiation with a 5 W blue LED in the presence of a visible-light photoredox catalyst, and a sequential C-C and C-O bond formation was involved. Triarylamine was employed as an effective additive to initiate a radical reaction. A wide range of substituents on the phenyl group attached to alkyne or nitrogen and functional groups were all smoothly tolerated, representing the good compatibility of the reaction.

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TMS$$

$$R^{1} = H, Me, CI \\ R^{2} = Aryl, Alkyl, TM$$

Scheme 65 Visible-light-induced synthesis 2-trifluoromethyl indoles

Scheme 66 Plausible reaction mechanism.

A plausible reaction mechanism was proposed as outlined in Scheme 66. At first, the excited [Ru(phen)₃²⁺]* was generated from photoexcitation of Ru(phen)32+, which accepted single-electron transfer from tris(4-anisyl)amine to yield Ru (phen)₃ and NAr₃ radical cations. The imidoyl radical **66-B** was formed by the reductive cleavage of the C-Cl bond of 66-A with the assistance of Ru(phen)₃⁺. Then, intramolecular radical addition of 66-B to alkyne gave vinyl radical 66-C, which underwent the propagation process to afford vinyl chloride 66-D. The hydrolysis of 66-D resulted in the formation of 66-F and herein isomerized to 2-trifluoromethyl indole product. Another pathway involved the single-electron oxidation of 66-C by Ru(phen)₃²⁺ or NAr₃ radical cation to deliver vinyl cation 66-E, followed by the hydrolysis and isomerization to give rise to the indole product. The additional mechanistic experiments indicated that the latter pathway was less likely involved in the reaction.

A Pd(0)-catalyzed $C(sp^3)$ -H functionalization of trifluoroacetimidoyl chlorides for the synthesis of 2-trifluoromethyl indoles was reported by Cramer and co-workers (Scheme 67).84 The trifluoroacetimidoyl chlorides were derived from o-methylanilines, which attached the adjacent methyl group to be activated in the reaction. In the optimization of the conditions, the choice of the ligand and additive was important. PCy3 and S-IPr had a positive effect on the reaction with the addition of the suitable additive. The reaction pathway was proposed. Oxidative addition of the Pd(0) catalyst to the C-Cl bond of imidoyl chloride afforded the complex 67-A, which underwent ligand exchange to give the intermediate 67-B. Subsequently, palladacycle 67-C was formed by the carboxylate-assisted concerted metalation-deprotonation (CMD) process. The reductive elimination of 67-C with the intramolecular C-C bond formation and tautomerization led to the final 2-trifluoromethyl indole product.

Scheme 67 Synthesis of 2-trifluoromethyl indoles Pd(0)-catalyzed C(sp3)-H functionalization.

Scheme 68 Synthesis of 3-arylsulfonyl-2-trifluoromethyl-indoles.

Dolbier Jr. and co-workers disclosed a synthetic approach for the preparation of 3-arylsulfonyl-2-trifluoromethyl-indoles in two steps from methyl-arylsulfones and trifluoroacetimidoyl chlorides (Scheme 68).85 The coupling reaction of methylsulfones with trifluoroacetimidoyl chlorides in the presence of LDA led to trifluoromethylimino/enaminoβ-sulfonylaryl intermediates 68A/68B as a mixture of imine and enamine tautomers. After the optimization of the copper-catalyzed cyclization reaction conditions, 3-arylsulfonyl-2-trifluoromethyl-indoles were obtained in moderate to good yields with Cu(OAc)₂·xH₂O as the catalyst and Cs₂CO₃ as the base. A Cu(1)/Cu(111) mechanism was probably involved in the transformation.

4.5. Synthesis of 2-fluoroalkyl quinolines

A Rh(I)-catalyzed coupling-cyclization reaction of trifluoroacetimidoyl chlorides with alkynes to construct 2-trifluoromethylated quinolines was presented by Uneyama and co-workers (Scheme 69). 86 The phosphine ligand TFP (P(2-furyl)₃) or dppe

$$\begin{array}{c} R^1 \\ R^3 \\ R^4 \\ R^3 \\ R^4 \\ R^6 \\ R^6$$

Scheme 69 Synthesis of 2-trifluoromethylated quinolines.

(Ph₂P(CH₂)₂PPh₂) was found to be beneficial for the reaction. The electron effect of the reaction was obvious, which illustrated that the imidoyl chlorides possessing an electron-withdrawing group enabled a sluggish yield. With regard to unsymmetrical alkynes, a bulkier group was prone to be attached to the quinoline 4-position and an electron-withdrawing group was attached to the quinoline 3-position.

The possible mechanism was also depicted in Scheme 69. The oxidative addition of rhodium(1) complex to the imidoyl chloride formed trifluoroacetimidoyl rhodium(III) species 69-A. Then, the alkyne insertion into the Rh-C bond of 69-A delivered the vinylrhodium 69-B, which underwent intramolecular electrophilic substitution to release the quinoline product (Path A). Another pathway involved intramolecular thermal 6π-electrocyclization to generate intermediate 69-C, followed by the subsequent β-hydride elimination to provide the quinoline product (Path B). The experimental results supported the electrophilic substitution pathway.

Yuan and co-workers developed a palladium-catalyzed coupling reaction of trifluoroacetimidoyl iodide with dialkyl vinylphosphonates and a sequential cyclization reaction for the preparation of 2-trifluoromethyl-quinoline derivatives (Scheme 70).87 The transformation proceeded via the Heck coupling reaction to give the corresponding α,β-unsaturated imines in good yields, which were converted into quinoline derivatives with Pd/C in nitrobenzene. The protocol realized the simultaneous introduction of the trifluoromethyl group and the phosphoryl moiety into heterocycle molecules.

Uneyama, Katagiri and co-workers disclosed a palladiumcatalyzed chloroimination of fluorinated imidoyl chlorides to a triple bond to build 4-chloro-2-perfluoroalkyl quinolines in high yields (Scheme 71).88 Noteworthy was that chloroimination of a multiple bond was recognized as a useful tool to construct nitrogen-containing heterocycles. Moreover, additional chloride source was required to accelerate the reaction and TBAC was the best choice. Besides CF₃ group, other

$$F_{3}C$$

$$Pd(dba)_{2} (5 \text{ mol}\%) \\
K_{2}CO_{3} (2.0 \text{ equiv}) \\
\hline
F_{3}C$$

$$R$$

$$R_{3}C$$

$$R$$

$$F_{3}C$$

$$R$$

$$R_{3}C$$

$$R$$

$$R_{3}C$$

$$R_{43}$$

$$R_{5}C$$

$$R_{3}C$$

$$R_{43}$$

$$R_{5}C$$

$$R_{5}C$$

$$R_{7}C$$

Scheme 70 Synthesis 2-trifluoromethyl-4-(O,O-dialkyl)-phosphoryl-quinoline

$$\begin{array}{c} R = H, \ F, \ CI, \ CO_2Et \\ R_F = CF_3, \ CF_2H, \ CF_2CI, \ C_3F_7 \\ \end{array}$$

Scheme 71 Synthesis of 4-chloro-2-perfluoroalkyl quinolines.

fluorinated groups, such as C₃F₇, CF₂H, and CF₂Cl, were all tolerated in the reaction.

The reaction was initiated by the similar oxidative addition of the Pd(0) species into the C-Cl bond of imidoyl chloride to give the Pd(II) complex 71-A. The triple bond was activated by the Pd(II) catalyst and attacked by an exogenous chloride to yield the seven-membered trans-chloropalladation palladacycle intermediate 71-C. Reductive elimination of 71-C could deliver the desired quinoline product.

A library of highly substituted 2-perfluoroalkyl-3-iodoquinolines were prepared via electrophilic iodocyclization of perfluoroalkyl propargyl imines/amines (Scheme 72).89 In the presence of I2 and CAN, iodocyclization of perfluoroalkyl propargyl imines occurred to provide the corresponding quinoline products in high yields. As for para-substituted methoxy perfluoroalkyl propargyl imines, the azaspiro compound was obtained. In the reaction, iodine cation was generated from

$$R^{1} = H, \text{ Br, Me} \\ R^{2} = \text{Aryl, Alkenyl, R}_{F} = \text{CF}_{3}, \text{C}_{2}\text{F}_{9}$$

$$R^{1} = H, \text{ Br, Me} \\ R^{2} = \text{Aryl, Alkenyl, R}_{F} = \text{CF}_{3}, \text{C}_{4}\text{F}_{9}$$

$$R^{2} = \text{Aryl, Alkenyl, R}_{F} = \text{CF}_{3}, \text{C}_{2}\text{F}_{9}$$

$$R^{2} = \text{Aryl, Alkyl, Alkenyl, R}_{F} = \text{CF}_{3}, \text{C}_{2}\text{F}_{5}, \text{C}_{4}\text{F}_{9}$$

$$R^{2} = \text{Aryl, Alkyl, Alkenyl, R}_{F} = \text{CF}_{3}, \text{C}_{2}\text{F}_{5}, \text{C}_{4}\text{F}_{9}$$

$$R^{3} = \text{H, F, Br, Me, OMe}$$

$$R^{2} = \text{Aryl, Alkyl, Alkenyl, R}_{F} = \text{CF}_{3}, \text{C}_{2}\text{F}_{5}, \text{C}_{4}\text{F}_{9}$$

Scheme 72 lodocyclization of perfluoroalkyl propargyl imines/amines.

the activation of CAN, which attacked the carbon-carbon triple bond of the propargylic imine to result in the formation of the iodonium intermediate 72-A. Next, intramolecular cyclization of the aromatic ring with the activated triple bond led to the desired quinoline product. Perfluoroalkyl propargyl amines could be synthesized via Sonogashira coupling of perfluoroalkyl imidoyl iodides with alkynes followed by reduction with NaBH₃CN. The combination of I₂ and NaHCO₃ or the solely stronger electrophile ICl enabled the transformation of perfluoroalkyl propargyl amines into 2-perfluoroalkyl-3-iodoquinolines.

Wu and co-workers demonstrated a palladium-catalyzed tandem Sonogashira-alkyne carbocyclization reaction of β-trifluoromethyl β-enaminoketones with alkynes for the construction of 2-trifluoromethylquinolines (Scheme 73). 90 The β-trifluoromethyl β-enaminoketones were readily prepared from the LDA-mediated condensation of methyl ketones with trifluoroacetimidoyl chlorides. Various β-enaminoketones and terminal alkynes were applicable to the reaction, whereas aliphatic terminal alkynes failed to give the desired product under the standard conditions. Importantly, the presence of an ortho-β-acyl-enamine group probably had a strong orthosubstituent chelating effect to assist the oxidative addition of Pd(0) with aryl halide and stabilize the formed arylalkynyl-

Scheme 73 Palladium-catalyzed synthesis of 2-trifluoromethylquinolines.

Scheme 74 Synthesis of 2-fluoromethylated guinolines.

palladium species, thereby promoting the Sonogashira C-C bond formation process.

A Cu(1)-catalyzed coupling reaction and subsequent intramolecular cyclization from N-aryl-fluorinated imidoyl iodides and terminal alkynes for the construction of 2-fluoromethylated quinolines was achieved by Wu and co-workers (Scheme 74). 91 NEt₃ was considered to play a fundamental role to increase the solubility of the copper salt, whereas K₃PO₄ only gave the fluorinated alkynyl imine product. When phenylacetylene was used as a substrate, no quinoline product was observed, illustrating the importance of the terminal alkyne bearing α -H.

The reaction mechanism was proposed as depicted in Scheme 74. The insertion of copper alkynide into the C-I bond of imidoyl iodides by oxidant addition generated copper complex 74-A. Then the reductive elimination of 74-A led to intermediate 74-B, which was formed in the presence of Cu(1) with the complexation effect of Et₃N. A base-induced propargyl-allenyl isomerization occurred to give the allene intermediate 74-C, which undergoes a nucleophilic attack by the benzene ring to the central carbon of the allene in a Friedel-Crafts reaction manner to afford 74-D. The presence of the terminal alkyne α-H promoted the formation of an allene moiety. Finally, the protonation and isomerization of 74-D gave rise to the quinoline product 74-F.

The similar synthetic method to access 2-fluoromethylated quinoline derivatives was successfully applied to the total synthesis of all four stereoisomers of mefloquine by Leonov, Griesinger and co-workers. 92 In the protocol, a domino Sonogashira-6π-electrocyclization reaction was identified as the key step.

A Cu-catalyzed tandem reaction of fluorinated terminal alkynes with sulfonyl azides for the synthesis of 2-trifluoromethylquinolines was reported by Wu and co-workers (Scheme 75).93 The azide-alkyne cycloaddition (CuAAC) reac-

$$R_{F}$$

Scheme 75 Synthesis of 2-fluoromethylated quinolines from fluorinated alkynes and sulfonyl azides.

tion to construct a 1,2,3-triazole core was involved in the reaction. The electronic effect of substituents on the N-aromatic moiety of fluorinated terminal alkynes exerted an obvious influence on the reaction and the steric hinderance had a marginal effect on the reaction efficiency. The reaction pathway was proposed to be initiated by the copper-catalyzed click reaction of fluorinated terminal alkynes and sulfonyl azides to deliver the triazolo-copper intermediate 75-A, which was converted into ketenimine species 75-B with the release of N₂. The subsequent cyclization and isomerization process could provide the quinoline product.

A visible-light-induced radical cyclization of trifluoroacetimidoyl chlorides and alkynes for the facile preparation of 2-trifluoromethyl quinolines was developed by Zhou and coworkers (Scheme 76).94 The generation of trifluoroacetimidoyl radicals via visible-light-irradiated activation of C(sp²)-Cl bonds was involved in the reaction, which was readily interrupted by the addition of alkynes. Besides photoredox catalyst, the addition of $(nBu)_3N$ having a stronger electron-donating ability was inevitable for the high efficiency of the reaction. The reaction exhibited a broad substrate scope and tolerated a variety of functional groups. With regard to the imidoyl chlorides with a para-substitution on the aromatic ring, two products were observed as isomers in an appropriate ratio.

A plausible mechanism was proposed as outlined in Scheme 77. The excited $[Ru(bpy)_3]^{2+*}$ was generated by the

$$R^{1} \longrightarrow CF_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

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$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

$$R^{2} \longrightarrow F_{3} = [Ru(bpy)_{3}]Cl_{2} \cdot 6H_{2}O (2 \text{ mol}\%)$$

Scheme 76 Visible-light-induced synthesis of 2-fluoromethylated quinolines.

Path A:

Scheme 77 Proposed mechanistic pathways.

visible-light-induced photoexcitation of [Ru(bpy)₃]²⁺. The formation of [Ru(bpy)₃]⁺ and the radical cation of tributylamine was achieved through single-electron transfer from (n-Bu)₃N to $[Ru(bpy)_3]^{2+*}$. The reductive cleavage of the $C(sp^2)$ -Cl bond by [Ru(bpy)₃]⁺ furnished the radical intermediate 77-A, which underwent radical addition of alkynes to give vinyl radical 77-B. A homolytic aromatic substitution occurred to deliver cyclohexadienyl radical intermediate 77-C (Path A), followed by a single electron transfer from the photoexcited [Ru(bpy)₃]²⁺* catalyst or the radical cation $(n-Bu)_3N$ to yield cation 77-D. The deprotonation of 77-D could afford the quinoline product 77-E. The isomer 77-E' was considered to be formed through Path B, in which the reaction proceeded through ipso cyclization, C-N bond cleavage and aromatization sequence to generate cyclohexadienyl radical intermediate 77-C'.

An approach for the construction of 4-amino-2-(trifluoromethyl)quinolines by the intramolecular Friedel-Crafts reaction was presented by Darehkordi and coworkers (Scheme 78). 95 At first, 2-(1-(arylamino)-2,2,2-trifluoroethylidene)malononitrile derivatives 78-A were afforded by the reaction of trifluoroacetimidoyl chlorides with malononitrile in the presence of sodium hydride at ambient temperature or under microwave irradiation. Then, an intramolecular Friedel-Crafts reaction occurred with the promotion of AlCl₃ to give rise to intermediate 78-C, which underwent 1,3-H shift to yield 4-amino-2-(trifluoromethyl)quinolines. It should be noted that the protocol allowed for the preparation of a variety of 4-amino-2-(trifluoromethyl)quinolines in good yields without further

A Pd⁰-catalyzed intramolecular 6-endo-trig Heck cyclization reaction of fluorinated imidoyl chlorides with a double bond for the assembly of 2-fluoroalkyl quinolines was developed by

Scheme 78 Synthesis of 4-amino-2-(trifluoromethyl)guinolines.

Scheme 79 Synthesis of 2-fluoroalkyl quinolines.

Chen, Liu and co-workers (Scheme 79).96 The scope of the reaction was broad, as verified by the good tolerance of diverse substitutions and functional groups. The transformation proceeded through a classical Heck-type reaction mechanism. Oxidative addition of the Pd⁰ catalyst into the C-Cl bond of the imidoyl chloride gave palladium complex 79-A, which underwent intramolecular insertion of the alkene into the C-Pd^{II} bond to furnish the intermediate **79-B.** A *syn*-selective β -hydride elimination of the intermediate **79-B** yielded the 2-fluoroalkyl quinoline product. The Pd^{II} species **79-B'** could not be formed because of the difficulty of the C-Pd^{II}-Cl reductive elimination process.

4.6. Synthesis of 6-fluoroalkyl-phenanthridines

Zhang and co-workers described a palladium-catalyzed tandem Suzuki/C-H arylation reaction of N-(2-bromophenyl)trifluoroacetimidoyl chlorides with arylboronic acids for the synthesis of 6-(trifluoromethyl)phenanthridines (Scheme 80).⁹⁷ A Pd(0)/bulky phosphine ligand catalytic system and a nonpolar solvent enabled the high efficiency of the transformation. The reaction was suggested to be initiated by the Suzuki reaction to provide the coupling product 80-A. The oxidative addition of 80-A to the Pd(0) catalyst yielded the Pd(11) complex 80-B, which underwent a C-H activation process to deliver palladacycle 80-C and a molecule of HBr. Finally, the reductive elimination of 80-C gave a phenanthridine product (Path a). In an alternative Pd(II)/Pd(IV) pathway (Path b), imine-directing C-H activation occurred to produce the Pd(II) complex 80-D, followed by the intramolecular oxidative addition into phenyl bromide to generate the seven-membered palladacycle 80-E. The following reductive elimination resulted in the formation of desired products and released the Pd(II) species.

Rh-Catalyzed alkyne [2 + 2 + 2] cycloaddition reactions of diynes with various alkynes for the preparation of fluorine-containing multisubstituted phenanthridines was achieved by Wu, Gong and co-workers (Scheme 81). The diynes were obtained by the Sonogashira reactions between trifluoroacetimidoyl chlorides bearing an alkyne moiety and terminal alkynes. The reaction showed good functional group tolerance, and good yields were obtained with regard to most of the substrates, except for electron-deficient alkyne dimethyl acetylene-dicarboxylate (DMAD). It should be noted that when a bromodifluoromethyl group was attached on the diynes, a tandem Rh-catalyzed cycloaddition/C-H difluoromethylenation process

$$R^{1} = H, \text{ Me, F, Cl, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{1} = H, \text{ Me, F, Cl, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{2} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{2}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl, Ph, CN, CF}_{3}, \text{ NO}_{4}$$

$$R^{3} = H, \text{ Me, OMe, F, Cl,$$

Scheme 80 Synthesis of 6-(trifluoromethyl)phenanthridines.

$$\begin{array}{c} R^{1} \\ R^{2} \\ R^{3} \\ R^{2} \\ R^{3} \\ R^{4} \\ R^{2} \\ R^{3} \\ R^{2} \\ R^{3} \\ R^{4} \\ R^{1} \\ R^{1} \\ R^{1} \\ R^{1} \\ R^{1} \\ R^{2} \\ R^{3} \\ R^{2} \\ R^{3} \\ R^{2} \\ R^{3} \\ R^{4} \\ R^{1} \\ R^{1} \\ R^{1} \\ R^{1} \\ R^{1} \\ R^{2} \\ R^{3} \\ R^{2} \\ R^{3} \\ R^{2} \\ R^{3} \\ R^{4} \\ R^{2} \\ R^{3} \\ R^{4} \\ R^{5} \\$$

Scheme 81 Synthesis of fluorine-containing multisubstituted

occurred to give rise to more-complicated fluorine-containing polycyclic compounds.

A visible-light-mediated intramolecular radical cyclization reaction of trifluoroacetimidoyl chlorides for the construction of 6-(trifluoromethyl)phenanthridines was presented by Fu and co-workers (Scheme 82). 99 In the protocol, an imidoyl radical was *in situ* generated from the cleavage of $C(sp^2)$ -Cl bond induced by visible light. The plausible reaction pathway was similar to that reported in Zhou's work, 94 which involved the formation of a radical intermediate and a cation intermediate via a single-electron transfer process (SET).

A palladium-catalyzed intramolecular C(sp²)–H functionalization of trifluoroacetimidoyl chlorides to access 6-trifluoromethyl-phenanthridines was achieved by the same group (Scheme 83).¹⁰⁰ Control experiment ruled out the involvement of the radical pathway. The oxidative addition of the C–Cl bond of trifluoroacetimidoyl chloride to Pd(0) catalyst led to Pd(II) species 83-A, which occurred the intramolecular electrophilic palladation with C–H activation process to give the seven-member palladacycle 83-B. The reductive elimination of 83-B provided the phenanthridine product.

$$R^{1} = H, Me, F, Cl, CF_{3}$$

$$R^{2} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{3} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{4} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{5} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{5} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{6} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me, Et, n-Pr, OMe, Ph, F, Cl, CF_{3}$$

$$R^{7} = H, Me$$

Scheme 82 Visible-light-mediated synthesis of 6-(trifluoromethyl) phenanthridines.

$$R^{1} \stackrel{\text{il}}{=} R^{2} \qquad Pd(PPh_{3})_{4} \ (10 \text{ mol}\%) \\ PCy_{3} \ (20 \text{ mol}\%) \\ R_{2}CO_{3} \ (2.0 \text{ equiv}) \\ \hline CF_{3} \\ R^{1} = H, \text{ Me, F, Cl, CF}_{3} \\ R^{2} = H, \text{ Me, Et, OMe, Ph, F, Cl, CF}_{3} \\ \hline Pd(0) \\ R^{1} \stackrel{\text{il}}{=} R^{2} \\ \hline R^{3} = R^{2} \\ \hline Pd(0) \\ \hline R^{1} \stackrel{\text{il}}{=} R^{2} \\ \hline R^{3} = R^{3} \\ \hline R^{3} = R^$$

Scheme 83 Palladium-catalyzed 6-(trifluoromethyl) synthesis of phenanthridines.

Chen, Zhu and co-workers developed a palladium-catalyzed, Catellani-type dehydrogenation annulation reaction from fluorinated imidoyl chlorides and aryl iodines for the synthesis biologically relevant 6-fluoroalkyl phenanthridines (Scheme 84).101 The norbornene (NBE) was utilized to mediate the reaction. The transformation showed broad substrate scope with respect to fluorinated imidoyl chlorides and aryl iodines. Various CF₂X (X = H, Cl, Br, CF₃) groups could be incorporated into phenanthridines under the optimal conditions.

Several control experiments were performed to shed light on the reaction mechanism and the possible reaction pathway was depicted in Scheme 84. The oxidative addition of Pd(0) with aryl iodine and the subsequent insertion of norbornene and C-H activation afforded the Pd(II) complex 84-A. Another oxidative addition of imidovl chloride with 84-A could vield the Pd(IV) intermediate, which underwent the reductive elimination to realize the C-C bond formation, with the release of Pd (II) species 84-C. Then, another intramolecular C-H activation process occurred with the removal of norbornene to give rise to Pd(II) intermediate 84-D, followed by the reductive elimination to produce the desired phenanthridines.

Scheme 84 Norbornene-mediated synthesis of 6-fluoroalkylphenanthridines

$$\begin{array}{c} R_{F} \\ R_{I} \\$$

Scheme 85 Synthesis of 6-fluoroalkyl-phenanthridines.

The same group also reported an approach for the preparation of 6-fluoroalkyl-phenanthridines via a palladium-catalyzed tandem reaction of fluorinated imidoyl chlorides and 2-bromophenylboronic acid (Scheme 85). 102 Good compatibility was obtained for the reaction, but non-fluorinated imidoyl chlorides and the styrene derivatives were not suitable for the reaction, highlighting the key role of the CF₂X group and the imine group for the success of the transformation. The possible mechanism was similar to the reactions previously mentioned. The palladium-catalyzed Suzuki reaction was initiated, followed by the intramolecular C-H activation to furnish the phenanthridine product.

4.7. Synthesis of other trifluoromethyl N-heterocycles

A copper(1)-catalyzed tandem reaction of N-(o-haloaryl)alkynylimines and sodium azide was developed for the preparation of biologically active 4-(trifluoromethyl)-[1,2,3]triazolo[1,5-a]quinoxalines (Scheme 86). 103 The reaction proceeded well only in the presence of catalytic Cu(I) and L-proline ligand in DMSO at room temperature. Good functional group tolerance of the reaction allowed for the variation of the phenyl ring or alkyne moiety of N-(o-haloaryl)alkynylimines. A copper-catalyzed cascade [3 + 2] cycloaddition/intramolecular C-N coupling reaction sequence was involved in the reaction, providing various fluorine-containing 1,2,3-triazolo tricyclic compounds.

A two-step reaction for the construction of 6-trifluoromethylindolo[1,2-c]quinazolines and other related trifluoro-

4-(trifluoromethyl)-[1,2,3]triazolo[1,5-a] Scheme 86 Synthesis of quinoxalines.

Scheme 87 Synthesis of 6-trifluoromethylindolo[1,2-c]quinazolines and related heterocycles.

methylated heterocycles by the use of trifluoroacetimidoyl chlorides and indoles as starting materials was presented by Wu and co-workers (Scheme 87). In the protocol, 1-(*N*-arylimino)indole derivatives were formed by the reaction of trifluoroacetimidoyl chlorides with indoles *via* the addition/elimination process, which underwent palladium-catalyzed intramolecular cyclization to afford 6-trifluoromethylindolo[1,2-*c*] quinazolines. The scope of the reaction could also be extended to pyrrole, pyrazole, *N*-substituted indoles and benzene to assemble the corresponding trifluoromethylated heterocycles.

The intramolecular hydroamination of fluorinated N-(ortho-alkynylaryl)amidines under different conditions for the synthesis of fluorinated N-heterocycles was explored by Wu and co-workers (Scheme 88). 105 The indole derivative 88-A was smoothly produced with NaAuCl₄·2H₂O as catalyst through a 5-endo-dig cyclization. As for the amidine with an ethynyl substituent, various Lewis acid catalysts, including ZnCl₂, BF₃·OEt₂, InCl₃·3H₂O, CuSO₄ and Cu(OTf)₂, could promote the reaction to deliver quinazoline 88-B as a major product with high efficiency, whereas the indole product 88-C was observed in the presence of Cu(OAc)2. Under basemediated conditions, a 6-exo-dig pathway occurred to furnish quinazoline products 88-D, which underwent the ozonolysis process to yield quinazolone 88-E. The electrophilic cyclization of the amidines proceeded well under the I2/NaHCO3 system, offering 3-iodo-substituted indoles 88-F in good yields.

A series of structurally different C(3)-substituted benzo [e][1,2,4]triazine derivatives were synthesized by Kaszynski and co-workers (Scheme 89). Among these compounds, 3-(tri-fluoromethyl)-benzo [e][1,2,4]triazines were constructed by the Ruppert reaction of 3-iodobenzo [e][1,2,4]triazine with TMSCF₃. An alternative synthetic pathway was the oxidative cyclization of amidrazone, which was obtained from the reaction of trifluoroacetimidoyl chlorides with 1,2-di(propan-2-ylidene) hydrazine and hydrazine hydrate. A similar approach had been

Scheme 88 Intramolecular hydroamination of fluorinated N-(orthoalkynylaryl)amidines.

$$F_{3}C \qquad CI \qquad \frac{\text{Me}_{2}C=\text{NN}=\text{CMe}_{2}}{\text{NH}_{2}\text{NH}_{2}\text{H}_{2}\text{O}} \\ \text{DMF, rt, 5 h} \\ \text{NN} \\ \text{NN} \\ \text{NN} \\ \text{CF}_{3} \\ \text{L-BuOCI/CH}_{2}\text{CI}_{2} \\ \text{-70 °C to rt, 4 h} \\ \text{NN} \\ \text{NCF}_{3} \\ \text{NCF}_{4} \\ \text{NCF}_{3} \\ \text{NCF}_{4} \\ \text{NCF}_{3} \\ \text{NCF}_{4} \\ \text{NCF}_{4} \\ \text{NCF}_{5} \\ \text{NCF$$

Scheme 89 Synthesis of 3-(trifluoromethyl)-benzo[e][1,2,4]triazine.

reported by Uneyama and co-workers for the synthesis of diverse trifluoromethylated N-heterocycles. 107

Pace and co-workers realized a conceptually novel and divergent lithium halocarbenoid-mediated mono- or bishomologation reaction to enable the formation of quaternary trifluoromethyl aziridines (Scheme 90). Trifluoroacetimidoyl chlorides were applied as electrophilic imine surrogates to receive the addition of one or two homologating agents. By simply modulating the stoichiometric amounts of lithium carbenoids (LiCH2Cl), the protocol could render the divergent preparation of α -halo- α -trifluoromethyl aziridines and α -halomethyl- α -trifluoromethyl aziridines. Due to the strong electrophilicity of trifluoroacetimidoyl chlorides, several functional groups sensitive to organolithium reagents could be well tolerated in the transformation.

The reaction was recognized to be initiated by the addition of the first equivalent of LiCH₂Cl to trifluoroacetimidoyl chlorides to afford the tetrahedral intermediate **90-A**. An intramolecular nucleophilic displacement of **90-A** generated the

Scheme 90 Synthesis of quaternary chloro- and halomethyl-trifluoromethyl aziridines

Scheme 91 Synthesis of perfluoroalkylated benzazepines.

mono-homologated chloroaziridine 90-B. The chloroaziridine could be spontaneously transformed into a highly electrophilic azirinium ion, 90-C, in the presence of an excess amount of the homologating agent. The second homologation occurred to yield the chloromethylaziridine product 90-D.

A phosphine-catalyzed divergent [4 + 3] annulation reaction of fluorinated imidoyl chlorides with MBH carbonates for the construction of perfluoroalkylated benzazepines was disclosed by Huang and co-workers (Scheme 91). 109 In the protocol, the fluorinated imidoyl chlorides were first applied as an electrophilic four-atom building block to participate in the phosphine-catalyzed [4 + 3] cycloaddition reactions, providing various seven-membered N-containing heterocycles in moderate to good yields. By the employment of PPh3 or P(p-OCH₃Ph)₃ as the catalyst, different perfluoroalkylated benzazepines were produced with good efficiency. The reaction began with the formation of phosphorus ylides and the subsequent nucleophilic attack on imidoyl chlorides. Then, carbanion intermediates were generated upon the delocalization process in the presence of bases, which were captured by the electrophile to afford perfluoroalkylated benzazepine products via a dearomatization/rearomatization process.

Conclusions

In conclusion, abundant and fruitful progresses in the applications of fluorinated imidoyl halides in the past few decades have been summarized and discussed. This compound provides a facile and rapid pathway to directly build valuable fluorine-containing molecules. Nevertheless, there is still ample room to explore more reaction patterns and potential applications in this area. In consideration of the unique characters of fluorinated imidovl halides, they have been applied as versatile electrophilic synthons to participate in a wide range of reactions, offering the possibility of constructing more and more structurally diverse fluorine-containing scaffolds. Apart from the regular involvement in coupling and annulation reactions, the exploitation of more in-depth applications of fluorinated imidovl halides to construct structurally complicated fluorine-containing molecules and natural products is highly desirable in the future.

Conflicts of interest

There are no conflicts of interest to declare.

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