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Aggregation-induced emission polymers for high performance PLEDs with low efficiency roll-off[†]

JIABAO GU,^a ZENG XU,^a DONGGE MA,^a ANJUN QIN^a and BEN ZHONG TANG^a

As the congener of organic light-emitting diodes, polymeric light-emitting diodes (PLEDs) possess a number of distinct merits, such as a low-cost wet fabrication process, which make them applicable in large-area flexible display and lighting fields. However, most emissive polymers used in PLEDs suffer from the aggregation-caused quenching (ACQ) effect, which makes the device show large efficiency roll-off. In this work, two polymers, pTPE-DPA-Cz and pTPE-DPA-Flu, featuring aggregation-induced emission (AIE) characteristics were facilely synthesized through Suzuki–Miyaura polycoupling reaction by incorporating the AIE unit of TPE-DPA in their main chains. The resultant polymers possess good film-forming ability, excellent thermal stability and high photoluminescence quantum yields (PLQYs) in their film states, facilitating the fabrication of PLEDs through a solution process. Indeed, the PLEDs using pTPE-DPA-Cz and pTPE-DPA-Flu as emitting layers (EMLs) could achieve a maximum external quantum efficiency (EQE) of 3.26% (doped EML) and current efficiency of 3.69 cd A⁻¹ (non-doped EML). Notably, all the devices exhibit a quite low efficiency roll-off. This work indicates that AIE polymers are ideal candidates for the construction of high performance PLEDs with low efficiency roll-off.

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Introduction

Organic light-emitting diodes (OLEDs) have been widely applied in the fields of display and lighting since the first report by Tang and VanSlyke in 1987 because they possess a lot of distinctive merits such as excellent image quality, broad viewing angle, and being light, thin and flexible.¹ In recent years, with the trend in OLED applications shifting to large-area lighting and flexible display fields, the low-cost wet fabrication process has attracted much attention, which is more economical than the dry process.² In order to meet the requirements for wet fabrication processes, polymeric materials showing excellent film-forming ability were introduced into OLEDs, and therefore, extensive research on polymeric LEDs (PLEDs) has been conducted.³

The first PLED was reported in 1990, in which poly(*p*-phenylene vinylene)s were used as the emitter. However, the external quantum efficiency (EQE) was only 0.05%.⁴ Although high performance

PLEDs have been reported in recent years, the polymers used for the emitting layers (EMLs) mostly suffer from the aggregation-caused quenching (ACQ) effect, which greatly hampers the development of PLEDs. Therefore, to avoid the ACQ effect, the proportion of luminous groups in polymers for PLEDs is controlled at a low degree, typically lower than 20%. Moreover, most of the PLEDs with ACQ polymers as EMLs encountered the problem of high efficiency roll-off, which further limits their applications.⁵

To construct practically applicable PLEDs, the following requirements should be fulfilled. First, the emissive polymers utilized as the EMLs of PLEDs should be easily synthesized with good solubility and high photoluminescence quantum yield (PLQY); second, the PLEDs should have high performance with low efficiency roll-off;⁶ third, the electroluminescent (EL) spectrum of the PLEDs should be stable. Thus, the polymers featuring aggregation-induced emission (AIE) characteristics should be ideal candidates for the construction of PLEDs.

AIE, conceptually coined by Tang *et al.* in 2001, refers to a unique phenomenon that a kind of non- or weakly emissive molecule is induced to emit intensely upon aggregation or being fabricated into thin films. AIE is exactly opposite to ACQ, thus, the AIE luminogens (AIEgens) have attracted much attention and a large variety of them with great structural diversity have been developed, such as tetraphenylethylene (TPE), hexaphenylsilole (HPS), distyreneanthracene (DSA), tetraphenylbenzene (TPB)⁷ and tetraphenylpyrazine (TPP),⁸ and their derivatives.

Compared with low mass AIEgens, AIE polymers, however, have been less explored.⁹ According to the restriction of intramolecular

^a State Key Laboratory of Luminescent Materials and Devices, Guangdong Provincial Key Laboratory of Luminescence from Molecular Aggregates, Center for Aggregation-Induced Emission, South China University of Technology, Guangzhou 510640, China. E-mail: msqinaj@scut.edu.cn

^b Department of Chemistry, Hong Kong Branch of Chinese National Engineering Research Centre for Tissue Restoration and Reconstruction, Institute for Advanced Study, and Department of Chemical and Biological Engineering, The Hong Kong University of Science & Technology, Clear Water Bay, Kowloon, Hong Kong, China

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motion (RIM), the mechanism of AIE, the intramolecular motion of the AIE units in polymers is more easily restricted than in low mass molecules, leading to a stronger fluorescence in the aggregate and/or film state.¹⁰ Furthermore, AIE polymers enjoy many outstanding features such as good processability, facile functionalization, structural diversity and high thermal stability, which enable them to be applied in diverse areas.^{11–14}

However, the application of AIE polymers in PLEDs is rarely reported. In 2012, Li *et al.* constructed linear and hyperbranched AIE polymers bearing the AIE unit of TPE and they were used to fabricate PLEDs as non-doped EMLs.¹² The devices showed a current efficiency (CE) in the range of 0.093 to 2.13 cd A^{−1}. In 2018, they further synthesized hyperbranched polymers based on tetraphenylcyclopentadiene and fabricated PLEDs using them as doped EMLs but with high efficiency roll-off.¹³ Thus, PLEDs with higher performance and low efficiency roll-off with AIE polymers as EMLs should be further developed.

In this work, we synthesized two TPE-based AIE polymers, pTPE-DPA-Cz and pTPE-DPA-Flu, through a Suzuki–Miyaura polycoupling reaction (Scheme 1). In addition to the excellent thermal stability, the resultant polymers exhibit PLQYs up to 63.3% in the doped thin films. By utilizing these AIE polymers as EMLs, PLEDs were facilely fabricated *via* a solution-process. The results show that the device with 5% pTPE-DPA-Flu in a 4,4'-di(9H-carbazol-9-yl)-1,1'-biphenyl (CBP) host as EML exhibits a maximum EQE of 3.26% and the non-doped device using pTPE-DPA-Cz as the EML could perform with CE as high as 3.69 cd A^{−1}. More excitingly, low efficiency roll-off of the devices was realized, providing an effective molecular design strategy for AIE polymer based PLEDs.

Results and discussion

Synthesis

The synthetic routes to the monomers are shown in Scheme S1 (ESI†) and those to the AIE polymers pTPE-DPA-Cz and pTPE-DPA-Flu are presented in Scheme 1. The Suzuki–Miyaura polycoupling of the AIE unit containing dibromides and diborates bearing carbazole and fluorene moieties in the presence of the catalytic system of Pd(PPh₃)₄, tetraethyl ammonium hydroxide propagated smoothly in toluene at 85 °C. To eliminate the possible terminal defect, the phenylboronic acid and bromobenzene were added successively to react with the bromo- and borate end groups, respectively. Finally, the polymers were

collected by precipitating the reaction solution in an excess amount of methanol and the polymers were further purified by washing thoroughly with acetone by Soxhlet extraction.

The structures of the intermediates, monomers and polymers were characterized by ¹H and ¹³C NMR spectra and satisfactory results were obtained (Fig. S1–S10, ESI†). The resultant pTPE-DPA-Cz and pTPE-DPA-Flu are soluble in commonly used organic solvents, such as tetrahydrofuran (THF), dichloromethane (DCM), chloroform, and toluene, which facilitate the solution process for PLED fabrication. The weight-average molecular weights of the resultant pTPE-DPA-Cz and pTPE-DPA-Flu were determined by advanced polymer chromatography (APC) in THF on the basis of a linear polystyrene calibration, and the values of 1.9×10^4 and 2.4×10^4 g mol^{−1} were obtained, respectively (Table S1, ESI†), which are high enough for forming films with good quality. The resultant polymers are also thermally stable as revealed by the thermogravimetric analysis (TGA). The 5% weight-loss temperatures (T_d) of pTPE-DPA-Cz and pTPE-DPA-Flu are as high as 510 and 442 °C, respectively (Fig. S11A, ESI†). Moreover, their glass transition temperatures (T_g) are also as high as 193 and 152 °C as measured by differential scanning calorimetry (DSC) (Fig. S11B, ESI†). These results indicate that our resultant polymers are stable enough for PLED applications as EMLs.

Photophysical properties

The UV-vis absorption and photoluminescence (PL) spectra of pTPE-DPA-Cz and pTPE-DPA-Flu in neat and doped films are shown in Fig. 1 and the data are summarized in Table 1. In THF solutions, THF/water mixtures with water fraction (f_w) of 99%, neat films and doped films, pTPE-DPA-Cz and pTPE-DPA-Flu all show close absorption maxima at *ca.* 385 nm. In THF solutions,

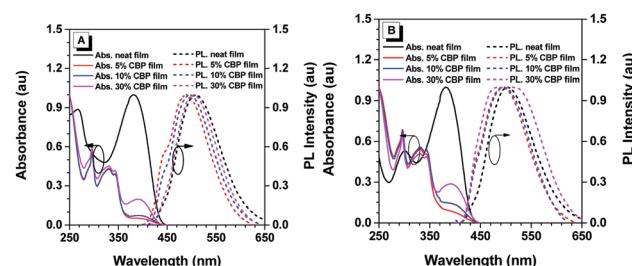
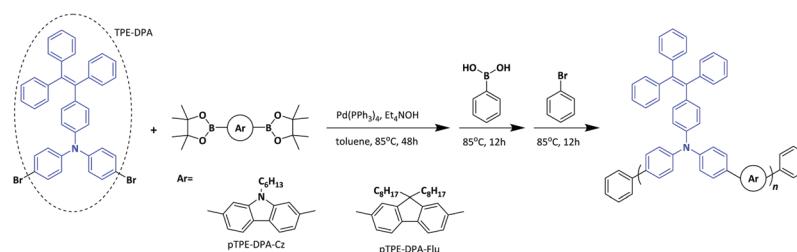


Fig. 1 The UV and photoluminescence (PL) spectra of (A) pTPE-DPA-Cz and (B) pTPE-DPA-Flu. The neat and doped films of polymers with weight ratios of 5, 10 and 30% in CBP were deposited on a quartz substrates.



Scheme 1 Synthetic routes to pTPE-DPA-Cz and pTPE-DPA-Flu.

Table 1 Photophysical properties of pTPE-DPA-Cz and pTPE-DPA-Flu

Soln ^a	Neat film ^b				Doped film ^c							
	λ_{abs} (nm)	λ_{em} (nm)	Φ_F (%)	λ_{em} (nm)	Φ_F (%)	τ (ns)	λ_{em} (nm)	Φ_{PL} (%)	τ (ns)	λ_{em} (nm)	Φ_{PL} (%)	τ (ns)
pTPE-DPA-Cz	383	500	0.4	508	28.7	2.5	486	58.3	2.9	496	63.3	3.0
pTPE-DPA-Flu	384	495	1.5	504	31.4	2.3	493	59.7	2.6	495	56.1	2.7

^a Measured in THF solution (2×10^{-3} mg mL⁻¹) at room temperature. ^b Deposited on a quartz substrate. ^c In CBP matrix (5 wt%, 10 wt%, 30 wt% respectively) deposited on quartz substrates.

pTPE-DPA-Cz and pTPE-DPA-Flu exhibit weak PL peaks at 500 and 495 nm respectively. While, with addition of the poor solvent of water into their THF solutions, the emission gradually intensified especially at high f_w , demonstrating a typical AIE feature (Fig. 2). The PLQY measurement further confirms the AIE feature of these two polymers. The values of pTPE-DPA-Cz and pTPE-DPA-Flu were recorded to be 0.4% and 1.5% in THF respectively, which reach up to 63.3% in their film states (Table 1).

Furthermore, compared with the neat films, the emission peaks of their doped films in a CBP host are blue-shifted slightly, which can be ascribed to the suppressed intermolecular dipole–dipole interaction and weakened intramolecular charge transfer effect from more polar neat films to less polar doped films,¹⁵ whereas, a slightly red-shifted emission of the doped film with 30 wt% pTPE-DPA-Flu might be attributed to the interaction between its functional groups and the host molecules. The room-temperature transient spectra of pTPE-DPA-Cz and pTPE-DPA-Flu in neat and doped films were measured and the lifetimes of prompt fluorescence in the range from 2.3 to 3.0 ns were obtained, indicating that the polymers emit fluorescence (Table 1 and Fig. S12, ESI[†]).

Electrochemical properties

The electrochemical properties of the pTPE-DPA-Cz and pTPE-DPA-Flu were investigated using cyclic voltammetry (CV) and their electrochemical data are summarized in Table S1 (ESI[†]). According to their scanned cyclic voltammograms (Fig. 3), the onset oxidation potential of pTPE-DPA-Cz and pTPE-DPA-Flu occurred at 0.76 and 0.69 V, respectively. Based on the equation of $E_{\text{HOMO}} = -(E_{\text{Onset}(\text{ox})} + 4.8)$ eV and the onset potential, the highest occupied molecular orbital (HOMO) energy levels of pTPE-DPA-Cz and pTPE-DPA-Flu were deduced

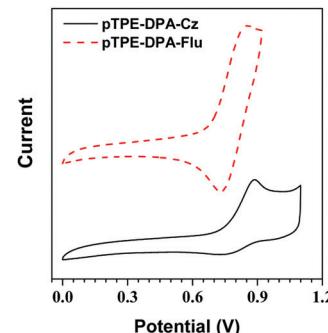


Fig. 3 Cyclic voltammograms of pTPE-DPA-Cz and pTPE-DPA-Flu measured in DCM containing 0.1 M tetra-*n*-butylammonium hexafluorophosphate. Scan rate: 100 mV s⁻¹.

to be -5.56 and -5.49 eV, respectively. As the onset reduction potential could not be observed clearly, the lowest unoccupied molecular orbital (LUMO) energy levels were alternatively calculated by the equation of $E_{\text{LUMO}} = (E_{\text{HOMO}} + E_g^{\text{opt}})$ eV, where E_g^{opt} was estimated by the absorption spectra of the polymer films. Therefore, the LUMO energy levels of the polymers were estimated to be -2.62 and -2.57 eV.

Theoretical calculations

To gain deep insight into the photo-physical properties of the polymers, the distributions of the HOMOs and LUMOs, and the models of their repeating units of TPE-DPA-Cz and TPE-DPA-Flu were calculated *via* density functional theory (DFT) at the B3LYP/6-31G(d) level. As illustrated in Fig. 4, the electron clouds of the HOMOs are located mainly on the triphenylamine units, whereas the electron clouds of the LUMOs are mostly localized on the TPE units, demonstrating that the fluorescence of the polymers mainly comes from the TPE-DPA units.

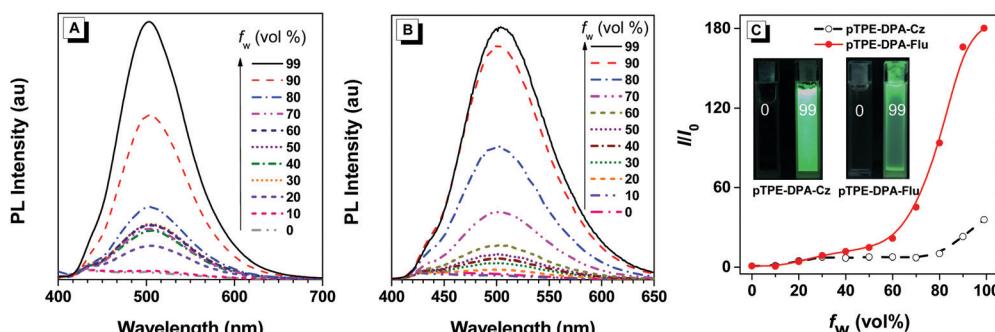


Fig. 2 PL spectra of (A) pTPE-DPA-Cz and (B) pTPE-DPA-Flu in THF/water mixtures with different water fractions (f_w). (C) Plots of I/I_0 values versus f_w . I_0 is the PL intensity in pure THF. Inset: The photographs of the polymers in THF and THF/water mixtures with f_w of 99%.

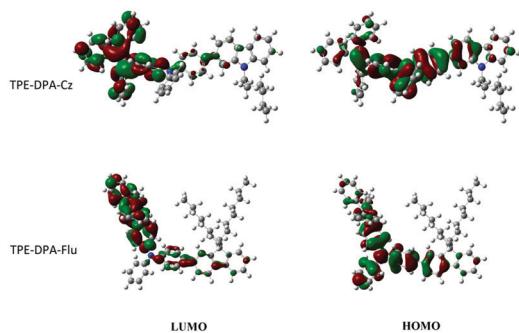


Fig. 4 The HOMO and LUMO distributions of the calculation models of TPE-DPA-Cz and TPE-DPA-Flu obtained by DFT calculations performed at the B3LYP/6-31G(d) level.

Film quality

In consideration of the significant impact of film quality on the device performance, the quality and morphology of the spin-coated films of these two AIE polymers were tested by atomic force microscopy (AFM). The films were prepared on silicon substrates by a spin-coating process which will be used for the device fabrication. As shown in Fig. 5, the neat films of pTPE-DPA-Cz and pTPE-DPA-Flu exhibit smooth surfaces with small root mean square (RMS) values of 0.373 and 0.575 nm, respectively, indicating that uniform films can be produced through a solution process. Similarly, their doped films also display small RMS values of 0.368 and 0.566 nm. With the excellent film forming ability, the leakage current of the PLEDs could be greatly reduced, which benefits the injection of carriers and the uniformity of luminance.

Electroluminescence

To evaluate the EL performance of these two AIE polymers, we manufactured devices through a solution process with the configuration of ITO/PEDOT:PSS (50 nm)/EMLs/TmPyPB (40 nm)/LiF (1 nm)/Al, in which indium tin oxide (ITO) acts as the anode,

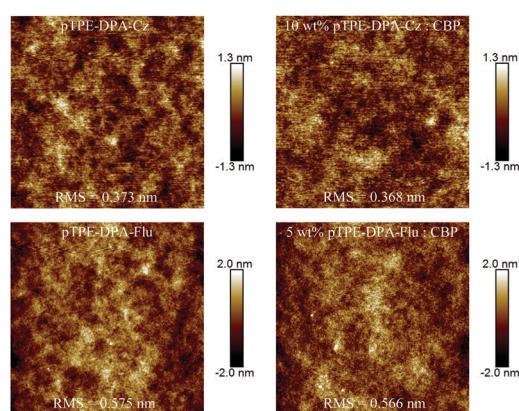


Fig. 5 AFM images of the neat and doped films of pTPE-DPA-Cz and pTPE-DPA-Flu. Neat films: Si/PEDOT:PSS (50 nm)/pTPE-DPA-Cz(60 nm); Si/PEDOT:PSS (50 nm)/pTPE-DPA-Flu(50 nm). Doped films: Si/PEDOT:PSS (50 nm)/10 wt% pTPE-DPA-Cz:CBP(55 nm); Si/PEDOT:PSS (50 nm)/5 wt% pTPE-DPA-Flu:CBP(60 nm).

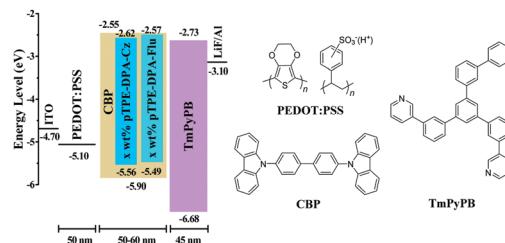


Fig. 6 Energy level diagram and molecular structures of the materials for PLEDs.

poly(3,4-ethylenedioxythiophene):poly(styrene-sulfonate) (PEDOT:PSS) and LiF are employed as hole- and electron-injecting layers, 1,3,5-tri(*m*-pyrid-3-yl-phenyl)benzene (TmPyPB) is selected as the electron-transporting layer, and Al is chosen for the cathode (Fig. 6). Devices 1–3 refer to the EMLs with 5, 10, and 30 wt% pTPE-DPA-Cz doped in CBP (55 nm), respectively, and device 4 for non-doped pTPE-DPA-Cz (60 nm) EML. Similarly, the devices 5–7 represent the EMLs of 5, 10, and 30 wt% pTPE-DPA-Flu doped in CBP (60 nm), respectively, and non-doped pTPE-DPA-Flu (50 nm) as the EML was assigned as device 8.

The device performance data are listed in Table 2 and the curves are shown in Fig. 7 and Fig. S13 and S14 (ESI[†]). For the devices 1–4, the EL peaks red shift from 474 to 517 nm, suggesting that there might be intermolecular interactions. Among the doped devices, device 2 shows a maximum CE, power efficiency (PE), and EQE of 6.47 cd A⁻¹, 4.73 lm W⁻¹, and 2.98%, respectively. It is worth noting that the values could remain to be 6.02 cd A⁻¹, 4.53 lm W⁻¹, and 2.77% at the luminance of 500 cd m⁻², respectively, demonstrative of a low efficiency roll-off. Furthermore, the non-doped device 4 exhibits a maximum CE, PE, and EQE of 3.69 cd A⁻¹, 2.28 lm W⁻¹ and 1.46%, respectively. The same as the doped devices, a smaller efficiency roll-off is achieved with CE, PE and EQE values of 3.67 cd A⁻¹, 2.13 lm W⁻¹ and 1.44% at the luminance of 500 cd m⁻², respectively. Quite similar behaviors were obtained for devices 5–8, and the efficiency roll-off of them is also very small. Therefore, it is feasible that highly efficient PLEDs with low efficiency roll-off can be fabricated by using AIE polymers as EMLs.

Conclusions

In summary, two AIE polymers, pTPE-DPA-Cz and pTPE-DPA-Flu, bearing the AIE unit of TPE-DPA were designed and synthesized through Suzuki–Miyaura polycoupling. These two polymers possess good thermal stability and show high PLQY values in their solid and film states. Moreover, they also possess excellent film forming ability and can be fabricated into high quality films by a simple spin-coating process, which greatly benefits the device performance. Thus, doped and non-doped solution-processed PLEDs were fabricated using the AIE polymers as emitting layers. The results show that the doped device of pTPE-DPA-Flu with 5% doping concentration in the CBP host

Table 2 EL performances of PLEDs based on pTPE-DPA-Cz (devices 1–4) and pTPE-DPA-Flu (devices 5–8)

Device	V_{on} (V)	L_{max} (cd m $^{-2}$)	CE a (cd A $^{-1}$)	PE a (lm W $^{-1}$)	EQE a (%)	λ_{max}^b (nm)	CIE b (x, y)
1	3.5	5350	4.55/4.50	3.45/2.85	2.63/2.60	474	(0.204, 0.247)
2	3.2	5967	6.47/6.02	4.73/4.53	2.98/2.77	482	(0.209, 0.316)
3	3.1	6390	6.06/5.48	4.43/4.14	2.57/2.33	487	(0.225, 0.364)
4	3.2	2866	3.69/3.67	2.28/2.13	1.46/1.44	517	(0.273, 0.431)
5	3.6	4578	6.32/5.60	4.62/3.35	3.26/2.88	480	(0.208, 0.275)
6	3.5	3308	4.20/3.96	2.95/2.62	2.06/1.94	482	(0.209, 0.294)
7	3.3	3549	5.28/4.96	3.96/3.46	2.32/2.18	484	(0.230, 0.350)
8	3.7	2573	3.26/3.19	2.02/1.83	1.47/1.44	486	(0.242, 0.370)

^a Order of maximum, then values at 500 cd m $^{-2}$. ^b Measured at 500 cd m $^{-2}$.

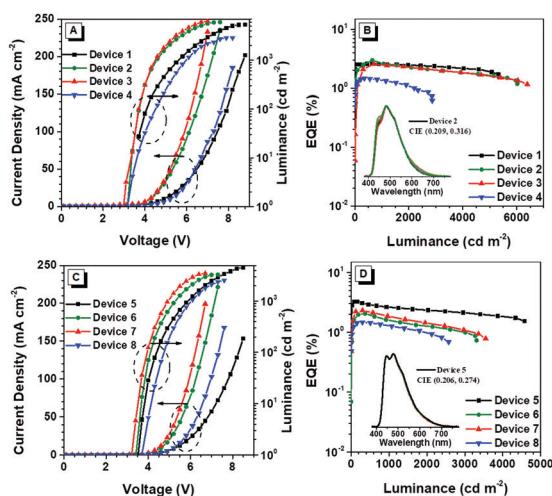


Fig. 7 (A and C) Current density–luminance–voltage curves of devices 1–8. (B) and (D) External quantum efficiency as a function of luminance for devices 1–8. Inset: Normalized EL spectra of devices 2 (B) and 5 (D).

exhibits a maximum EQE of 3.26% and the non-doped PLED using pTPE-DPA-Cz as EML could perform with CE as high as 3.69 cd A $^{-1}$. More importantly, unlike most PLEDs based on ACQ polymers, which suffer from serious efficiency roll-off, these PLEDs show quite low efficiency roll-off, demonstrating their outstanding efficiency stability. These results indicate that the AIE polymers can be ideal candidates for the construction of high-performance solution-processed PLEDs with low efficiency roll-off.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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