

HIGHLIGHT

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Total synthesis of strictamine: a tutorial for novel and efficient synthesis†

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When teaching synthetic design in the classroom, the novelty and efficiency of the overall synthetic route are the key values being imparted to the younger generation of synthetic chemists. The creative synthetic design is of great interest to graduate students who are entering into this exciting field for the first time. Strictamine, an akuammiline alkaloid, bears a congested cage-like structure and has interesting biological activities. Numerous attempts have been made toward the total synthesis of this compound since its discovery five decades ago. The synthesis was not completed until last year when synthetic chemists began to apply novel ring construction strategies. In 2016, two research groups reported the total syntheses of strictamine and four reported the formal syntheses. In this Highlight, the synthesis of strictamine is used as a tutorial case to illustrate novelty and efficiency in total synthesis.

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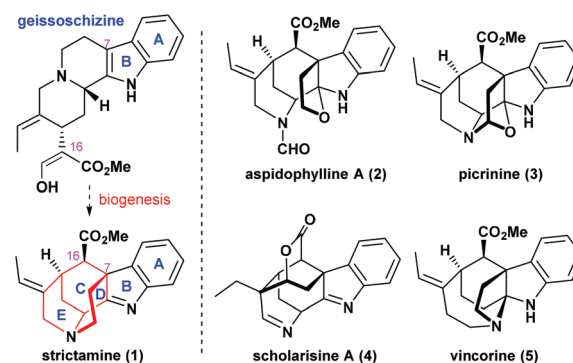
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Emerging drug resistance has become a growing public health concern.¹ Combating these resistances requires enormous efforts to accelerate the process of drug discovery for clinical use. Promising drug leads that are biologically active natural products are always plagued by limited natural resources. Recent advances in biotechnology may alleviate the problems of limited harvest;² however, the possibility of producing compounds of interest using microorganisms remains challenging. In contrast, combinatorial chemistry suffers from limited scaffolds, which clearly limits the possible chemical space that can be occupied by natural products.³ With this perspective in mind, in the past decade, the synthetic community was inspired to focus on a few particularly valuable synthetic targets and aimed to quickly provide synthetic natural products in large enough quantities to fulfill the requirements for derivatization.⁴ Step economy, redox economy, and atom economy emerged as the guiding tenets for synthesis and motivated the synthetic chemists to search for new answers.^{4b}

The primary intention of this study was to use the preparation of these materials to illustrate synthetic design in the classroom.⁵ When considering the synthetic route to a specific

natural product, efficiency is always the main concern. Prior to initiating any synthetic project, we may ask “are we able to reach this target in a more efficient way than that in the existing synthetic studies?” Moreover, the novelty of the chemical transformations, like the application of newly developed powerful synthetic methods as well as employing rarely utilized reactions on intricate substrates, is also taken into consideration. Ideally, these two facets can be combined into one synthetic work to elevate the current synthesis to a high level of chemical knowledge. The synthesis of strictamine is thus selected as an excellent example for illustration.

Strictamine, first isolated by Ganguli and co-workers in 1966 from *Rhazya stricta*,⁶ is a representative congener of the akuammiline alkaloids which have indolenine-fused cage-like structures (Scheme 1). It shows *in vitro* inhibitory activities against nuclear factor- κ B, which plays an important role in



Scheme 1 Strictamine, its biosynthetic precursor (geissoschizine), and other representative akuammiline alkaloids.

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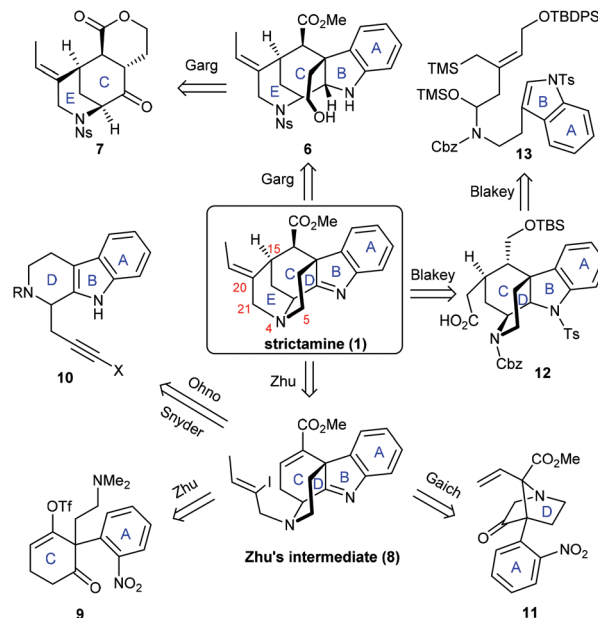
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Highlight

regulating gene expression in immune and inflammatory responses.⁷ The intriguing polycyclic chemical structure includes a highly congested methanoquinolizidine core (highlighted in red). The C–D–E rings of the core are connected *via* the C7–C16 bond and four stereocenters including a challenging C7 quaternary carbon center. Although, since the early 1970s, numerous attempts have been made to complete the total synthesis,⁸ the synthesis was not achieved until very recently when several research groups solved the long-standing puzzle of this molecule. Six synthetic groups have accomplished the total or formal synthesis of strictamine, making this molecule a hot synthetic target. Herein, we summarize the innovative strategies and methods developed during the chemical synthesis of strictamine.

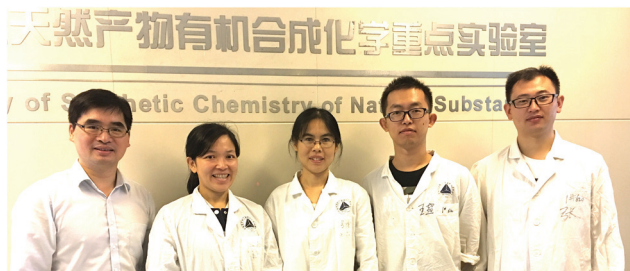
To synthesize this complex structure, different strategies and methodologies have been adopted for building the congested ring system (Scheme 2). In Garg's work, the C5–N4 bond was chosen to complete the D-ring *via* an intramolecular substitution.⁹ Intermediate 6 was built up *via* a Fischer indolization of ketone 7 bearing the C and E rings. The Zhu group disconnected the C15–C20 bond in the E-ring. This simplified strictamine to vinyl iodide intermediate 8, in which the D-ring could be constructed by an intramolecular nucleophilic substitution of brominated 9 (*vide infra*) bearing the A and C rings.¹⁰ Interestingly, the Ohno, Snyder, and Gaich groups also selected compound 8 as their synthetic target. In the Ohno and Snyder syntheses, they used Au(I)-catalyzed 6-*endo*-dig cyclizations of dihydro- β -carboline derivatives 10 (X varied between groups) to construct the C ring.¹¹ In Gaich's work, the C ring was established through a [2,3]-Stevens rearrangement of precursor 11 bearing the A and D rings.¹² The Blakey group



Scheme 2 Strategies and tactics in the total synthesis of strictamine.

completed the synthesis of the strictamine pentacyclic core with a C21-carboxylic acid group and closed the E-ring *via* amidation.¹³ A cascade annulation of compound 13 bearing the A and B rings furnished the C and D rings in one step. In the following section, we will highlight the novelty in each synthetic route.

In 2016, the Garg group at UCLA reported the first enantioselective total synthesis of (+)-strictamine featuring an inter-



From right to left: Shiju Zhang, Chao Wang, Yan Wang, Sha-Hua Huang and Ran Hong

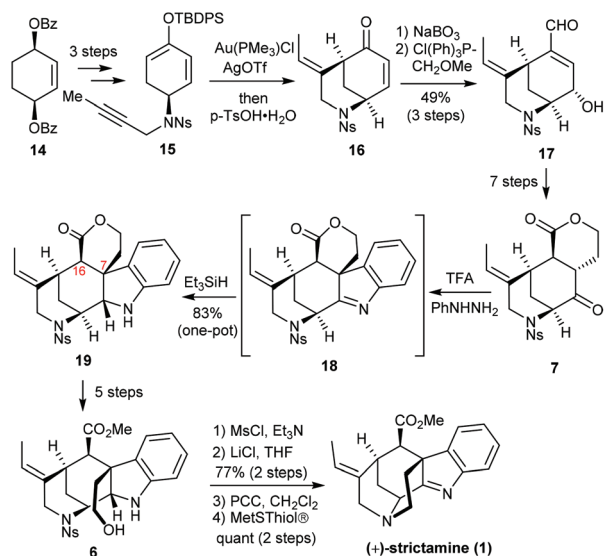
Chao Wang received his bachelor's degree in pharmaceuticals (2011) and master's degree in medicinal chemistry (2015) from the China Pharmaceutical University. After working in Professor Wei-Sheng Tian's group (SIOC) as a research assistant, he then joined the Hong group in 2016 to pursue his PhD on the total synthesis of natural products.

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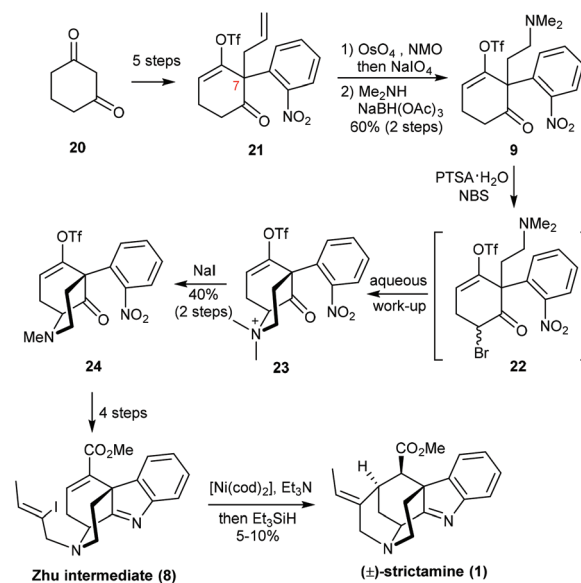
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Ran Hong graduated from Sichuan University (1996) and obtained his PhD (2001) under the supervision of Prof. Guo-Qiang Lin at SIOC. After conducting postdoctoral research in the laboratories of late Prof. Jay K. Kochi at the University of Houston and Prof. Li Deng at Brandeis University, he was appointed as a Group Leader at SIOC in 2007 and initiated the research program to apply the biomimetic approach in natural product synthesis as well as synthetic method development.



Scheme 3 Garg's total synthesis of (+)-strictamine.



Scheme 4 Zhu's total synthesis of (±)-strictamine.

rupted Fischer indolization approach (Scheme 3).⁹ Beginning with dibenzoate **14**, desymmetrization, saponification, oxidation and enolization were adopted to prepare silyl enol ether **15**. The Toste Au(I)-catalyzed cycloisomerization delivered [3.3.1]-azabicyclic enone **16**, which was further elaborated to enal **17** through an epoxidation/Wittig olefination sequence. The full carbon framework in ketolactone **7** was established in seven steps. The subsequent Fischer indolization through intermediate **18** proceeded smoothly, and reduction of the intermediary imine was achieved using a hydride source and afforded **19**. The challenging C7 quaternary stereocenter was installed during this step. With **19** in hand, the virtually feasible formation of hydroxyester **6** proved difficult. The C16 stereocenter was readily epimerized under a variety of direct lactone ring-opening conditions. A five-step work-around sequence was eventually developed. It entailed reduction of the lactone, selective silylation of the primary alcohol, and an oxidation/esterification with an *in situ* deprotection to deliver the requisite hydroxyester **6**. Finally, the hydroxyl group was activated by mesylation and chlorination. After PCC oxidation to reveal a ketimine functional group, denosylation and ring closure by the liberated amine completed the total synthesis of strictamine.

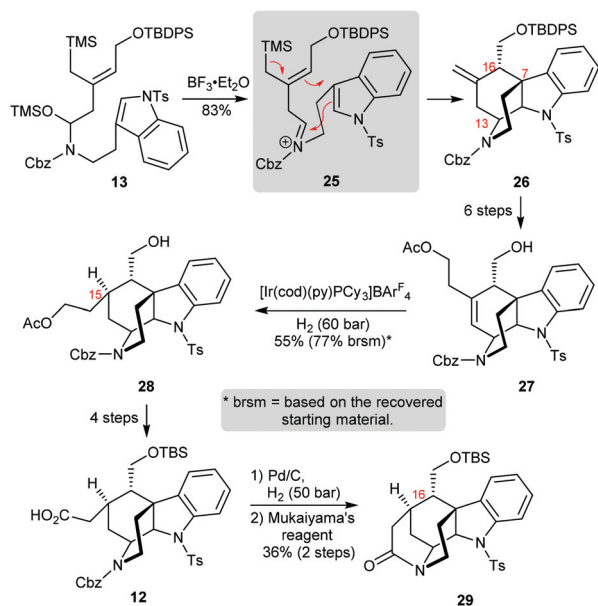
The interrupted Fischer indole synthesis with the appropriate precursor smoothly afforded the desired stereochemistry of the quaternary C7 carbon center in the total synthesis of the akuammiline alkaloid. However, the preparation of precursor **7** and elaboration of advanced intermediate **19** to strictamine still required a series of transformations, which attenuates the overall efficiency of this strategy.

The Zhu group at EPFL reported a racemic synthesis of strictamine (Scheme 4).¹⁰ In contrast to Garg's approach, they began with the elaboration of **21**, which already contained the C7 quaternary stereocenter. **21** was readily prepared from 1,3-cyclohexanedione **20** in five steps based on their previous work.¹⁴ The terminal alkene was converted to a tertiary amine

via a conventional oxidative cleavage protocol and reductive amination with dimethylamine. The chemoselectivity at this step was essential for preserving the ketone functional group. It was found that if a primary amine was used, condensation with the ketone would compete with the proposed transformation, and the resulting hexahydropyrrole would be a dead-end for the synthesis. After the introduction of bromine at the α-position of ketone **9**, a basic aqueous work-up provided the quaternary ammonium salt **23**, which was subjected to demethylation for which sodium iodide was found to be the most effective reagent. After four straightforward transformations to set up all the functional groups, the final cyclization of **8** to strictamine turned out to be extremely difficult. An intramolecular Michael addition mediated by lithium-halogen exchange, a Pd-catalyzed reductive Heck reaction, and a radical cyclization were all attempted. The desired product was only obtained when using Ni(cod)₂ and Et₃N with Et₃SiH; under these conditions, vinyl iodide **8** was converted to strictamine in 5–10% yield.

Because Zhu's strategy involved the early-stage creation of the C7 quaternary center and rapid construction of the D-ring, the precursor for the formation of the indolenine unit did not require the tedious redox manipulations seen in other syntheses. Although the yield of the final nickel-catalyzed cyclization step is rather poor, it is still an efficient tactic for constructing the framework and will no doubt be adopted by other groups to enhance the novelty of their own synthetic strategies.

The Blakey group at Emory University published a synthesis of the strictamine pentacyclic core *via* a creative iminium ion cascade annulation (Scheme 5).¹³ Starting with hemiaminal **13**, an *N*-acyl iminium ion was first produced in the presence of boron trifluoride diethyl etherate, and then it was attacked by the 2-position of the indole. The corresponding benzylic carbocation was further trapped by the allylsilane, thus estab-



Scheme 5 Blakey's synthesis of the strictamine pentacyclic core structure.

lishing the C and D rings with the correct stereochemistry at C7 and C13 in one step. With the key intermediate **26** in hand, the chain elongation of the exocyclic olefin required six steps. To secure the stereocenter at C15, Crabtree's catalyst ($[\text{Ir}(\text{cod})(\text{py})(\text{PCy}_3)_3]\text{BARF}_4$) was found to be the best choice for hydrogenation with good facial selectivity. Subsequent manipulation of **28** provided carboxylic acid **12** in four steps since the primary alcohol at C16 had to be protected. After removal of the Cbz group and amidation with Mukaiyama's reagent, pentacyclic skeleton **29** was obtained in a moderate yield.

Unlike the stepwise annulation strategies used by other groups, this work invented an intriguing cascade reaction to build two rings in a single step. Interestingly, when the Cbz group in **13** was replaced by an electron-donating group (Bn), the corresponding iminium ion would be attacked by the C3-position of indole, leading to the core structure of a different alkaloid.¹⁵ However, in this synthesis, the conversion of tetra-cycle **26** to pentacyclic core **29** required a long synthetic sequence (11 steps). Moreover, the inversion of the C16 stereocenter and installation of the quaternary C7 center as well as the development of an enantioselective variant of the cascade reaction which would be required to complete the synthesis remain a long way off.

The Ohno group at Kyoto University devised a racemic formal synthesis of strictamine based on a novel gold-catalyzed nucleophilic attack of the indole ring on a carbon-carbon triple bond (Scheme 6A).^{11a} The addition of a Grignard reagent to 3,4-dihydro- β -carboline **30** delivered propargyl derivative **31**. After some experimentation it was found that a combination of SPhosAuCl and AgNTf₂ was the most efficient catalytic system for achieving the 6-endo-dig cyclization of nosylamide **32**. To reach Zhu's intermediate **8**, they managed to introduce additional 9 steps to install an appropriate ester group at C16, although advanced intermediate **33** was virtually identical to **8**.

This strategy was elegant in regard to construction of the framework by the Au-catalyzed cyclization to furnish the C ring and the C7-quaternary carbon center as well as the double bond between C15 and C16 that was required for Zhu's final ring closure step. However, the series of functional group modifications from **31** to **32** and **33** to **8** diminished the efficiency of this work.

In 2017, the Snyder group at the University of Chicago presented an impressive 7-step asymmetric formal synthesis of



Scheme 6 (A) Ohno's formal synthesis of (±)-strictamine. (B) Snyder's formal synthesis of (+)-strictamine. (C) Gaich's formal synthesis of (±)-strictamine.

strictamine *via* an asymmetric propargylation and metal-mediated cyclization (Scheme 6B).^{11b} They started with the same β -carboline **30** as Ohno. In contrast to Ohno's method, they used allenylboronate to complete the asymmetric propargylation using a Cu(I)-(R)-DTBM-SEGPhos complex as the catalyst to deliver the adduct in good yield and enantioselectivity (as shown for **34**). The subsequent CO-insertion into the terminal alkyne proceeded smoothly. After Boc deprotection by trifluoroacetic acid, the resulting ammonium TFA salt bearing an electron-deficient alkyne, **35**, was effectively transformed *via* the key 6-*endo*-dig cyclization with stoichiometric AgSbF₆ and catalytic Ph₃PAuCl. From there, only one more step was required to reach Zhu's intermediate **8**.

The independent development of the Au(I)-catalyzed 6-*endo*-dig cyclization in Snyder's synthesis offered several advantages over Ohno's work. This synthetic design allows the introduction of the appropriate functional groups in well-planned steps, which makes this an appealing strategy. Although the final Ni-catalyzed cyclization needs to be further optimized, based on the number of synthetic steps, it clearly merges both novelty and efficiency in one classic synthesis.

The Gaich group at the University of Konstanz reported another formal synthesis of strictamine using [2,3]-Stevens rearrangements as the key transformations (Scheme 6C).¹² The synthesis began with α -cyanoacetate **37**, which was elaborated to highly substituted pyrrolidine **38** over 8 steps. The allyl group on the nitrogen was transferred to the 2-position of the pyrrolidine ring with excellent stereochemical control through the first Stevens rearrangement. Compound **39** was elaborated to bridged bicyclic system **11** over eight steps *via* N-H insertion of an α -diazo ketone intermediate. The crucial azabicyclic-[3.3.1]-nonane skeleton, **41**, was established through the second Stevens rearrangement *via* intermediate **40**, which was generated in the presence of silver triflate and proton sponge. That was followed by reduction of the nitro group using tin chloride with concomitant imine formation upon treatment with phosphorus tribromide.

Two Stevens rearrangements were the key features in the Gaich synthesis. The outstanding design of the molecular skeleton would be a textbook example for illustrating the application of a Stevens rearrangement in a total synthesis. While several steps in the synthesis of rearrangement precursors **38** and **11** need to be further optimized, this synthesis deserves more attention for its efficiency in an alkaloid synthesis.

Outlook for teaching and learning synthesis

In summary, distinct strategies and synthetic methods have been developed for the total or formal synthesis of strictamine. The Garg group utilized an interrupted Fischer indolization with an appropriate precursor to install the correct stereochemistry. The outstanding contribution of Zhu's synthesis was to introduce the C7-quaternary stereocenter at an early stage, which substantially simplified the synthetic design. The

only remaining challenge is further optimization of the nickel-catalyzed Michael-type vinylation to close the E-ring. The impressive cascade reaction to furnish the complex bridged ring system bearing the full carbon framework from the Blakey group relies on a deep appreciation of the molecular architecture; however, the development of an asymmetric variant would be a challenging task. The Ohno group utilized a gold-catalyzed 6-*endo*-dig cyclization to establish the framework of the target vinyl iodide in Zhu's work. The Snyder group further explored the chemistry stemming from the Ohno study, and their asymmetric propargylation of the imine is noteworthy. The Gaich group again targeted the Zhu's intermediate to complete a formal synthesis. They ingeniously employed [2,3]-Stevens rearrangements to construct the core structure. Even though some transformations and the yield of key reactions require further improvement, these studies exemplify the art of synthesis in terms of both efficiency and novelty in the total syntheses of natural products. The journey of pursuing the ideal synthesis of strictamine provides a good illustration for graduate students to better understand the principles of synthetic design. In the classroom, after a one semester course in synthetic methods, it is time for graduate students to apply the reactions they have learnt to more complex targets and evaluate the merits of various synthetic strategies. It is thus better to group several ingenious synthetic designs together to understand the bond disconnection approach as well as the strategic blueprint. The "art" inherent in synthesis will thus be visible. It is even more exciting to compare chemical synthesis with the elegance of biosynthesis in which strictamine may be constructed through an oxidative coupling between the α -carbon of the ester (C16) and the indole (C7) (Scheme 1). Such oxidative C-C bond forming reaction has been executed,^{8f,i} although without success, and it is believed that the biomimetic approach from naturally abundant geissoschizine would be a truly remarkable synthesis. Graduate students would not be satisfied with the one or two synthetic strategies they initially devised as the only take-home examples; they would be eager to compare different approaches to visualize the novel and efficient pathways to the final product. This creative learning process is valuable to train the next-generation of synthetic chemists.

Further reading

The Zu group at Tsinghua University recently reported a unified approach to access several congeners of akuammiline alkaloids and devised an elegant ring-reconstruction approach to complete the synthesis of strictamine.¹⁶ On comparing with previous studies, the E-ring formation with concomitant installation of the ester group at C16 with a requisite stereochemistry was outstanding. For the synthetic studies on akuammiline alkaloids in general, we encourage the readers to compare the creativity in individual synthetic designs with previous syntheses in this field.^{17,18} More generally, the central theme for synthesis could stem from the construction of the

morphan core. Several elegant reviews related to this topic are also highly recommended.¹⁹

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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