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## Application of the QuEChERS method combined with UHPLC-QqQ-MS/MS for the determination of isoprocarb and carbaryl pesticides in Indonesian coffee†

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The performance of the QuEChERS method in this study, as indicated by a high percentage (>90%) of recovery observations falling within the range of 60-140% and a sample replicate deviation (% RSD) of <20%, for the routine analysis of isoprocarb and carbaryl pesticides, has been evaluated over a 14-month period for the export of Indonesian coffee. Following a seven-day observation of the stability of these pesticides in coffee extract, it was found that the added standard calibration solution remained stable and useable for seven days when stored at 4  $^{\circ}$ C and -20  $^{\circ}$ C. This validated method, with high sensitivity (a LOQ of 0.001 mg kg<sup>-1</sup> for isoprocarb and carbaryl), has been employed to monitor residues in Indonesian coffee exports to comply with maximum residue limits (MRLs). The samples with higher contamination levels were predominantly from robusta coffee (57.76%), followed by arabica coffee (6.17%). The detection rates for residues decreased by more than 90% in the last two months of the method's application. In the observation of coffee processing, it was found that isoprocarb residues in contaminated samples could be transferred to the processed coffee (roasted and its infusion) to a limited extent, while residues from the carcinogenic carbaryl were not detected due to evaporation. Additionally, chronic dietary risk assessment showed that contaminated samples of robusta and arabica coffees should not be considered a significant public health concern (hazard index HI < 1). However, continuous monitoring of pesticide residues in Indonesian coffee is still recommended, not only to conform to the MRLs of importing countries but also to ensure food trade.

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## Introduction

QuEChERS and modified QuEChERS with sample hydration and acetonitrile as the extraction solvent are frequently used methods

for determining pesticide residues in coffee beans. 1-6 The analysis of 117 pesticides via LC-MS/MS was possible using QuEChERS extraction without further dispersive-SPE clean-up<sup>6</sup> as opposed to previous methods that involved lengthy procedures to eliminate interference from coffee co-extractives.7 For example, interference was minimized by precipitating lipids and waxes through freezeout treatment, removing caffeine with dispersive liquid-liquid micro-extraction (DLLME)1 and incorporating the ChloroFiltr sorbent in dispersive-SPE.8 Despite the availability of QuEChERS methods for detecting pesticides in raw coffee, these methods have so far only been reported in the context of initial method development and/or applications with laboratory-fortified samples. 2,3,5-7,9 These conditions can differ significantly from those found in actual samples. Currently, the QuEChERS method is proving highly efficient in monitoring and controlling pesticides in coffee, as previously reported with flutriafol and pyraclostrobin, which were technical barriers to the export of Brazilian green coffee. 10 Due to these issues, the QuEChERS method will be applied to other pesticides such as carbaryl and isoprocarb, which are critical to monitor in Indonesian coffee following a rejected export status.

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An export rejection of Indonesian coffee was reported due to carbaryl (2009) and isoprocarb (2021) residues exceeding the Japanese standard, 0.01 mg kg<sup>-1</sup>.<sup>11,12</sup> The occurrence of carbaryl is probably from spraying pesticides used by the farmer (Fig. S1† captures a farmer applying insecticides to coffee plants immediately before harvest) for controlling plant pests and diseases. Also, isoprocarb was used to deter ants attracted to ripe coffee cherries due to their high sugar content. Unfortunately, this treatment was deemed illegal because isoprocarb was exclusively registered for controlling mealybug pests (*Planococcus citri*) on coffee plants.<sup>13</sup>

Carbaryl and isoprocarb are carbamate insecticides, having different properties. Carbaryl can only act as the parent compound and does not metabolize into an active intermediate. This pesticide is classified as class 3, possibly carcinogenic to humans. According to the USEPA, carbaryl is considered a moderate oral toxicant (category II) with an LD<sub>50</sub> (oral, rat) of 108–840 mg kg<sup>-1</sup> and is listed as a group C carcinogen, implying that it is "likely to be carcinogenic in humans". Isoprocarb residue may pose a risk to humans as it disrupts an enzyme that regulates acetylcholine, a neurotransmitter, because of its low toxicity to humans. As a carbamate insecticide, isoprocarb is thermally unstable and can decompose with increasing temperature to produce mono-aromatics and polycyclic aromatic hydrocarbons.

As mentioned previously, due to the frequent presence of the carbamate insecticides isoprocarb and carbaryl in Indonesian coffee, it is crucial to confirm that the QuEChERS method is effective for detecting these residues in contaminated samples within the complex or "unique" matrices of coffee beans. Therefore, this paper aims to validate the QuEChERS method and evaluate its efficacy in identifying isoprocarb and carbaryl in the exports of Indonesian coffee beans. These toxic carbamate pesticides are considered hazardous to human health following coffee consumption and are a cause of export difficulties. Consequently, additional findings related to the effects of coffee processing and dietary risk assessment of these residues will be presented in this study.

## Materials and methods

#### Chemicals

Multi-component pesticide analytical standards (LC pesticide kit and GC pesticide kit) with a concentration of 100 mg  $\rm L^{-1}$  in methanol were purchased from Restek (Bellefonte, USA). These multi-component standards were used in the initial monitoring of 320 pesticide residues in coffee beans produced in several Indonesian provinces. Meanwhile the individual standards of isoprocarb (purity > 98%) and carbaryl (purity > 98%) for application in the routine analysis were obtained from Sigma-Aldrich (St. Louis, Missouri, USA). The analytical standards were stored at -20 °C. Acetonitrile and formic acid were provided by Merck (Darmstadt, Germany). Anhydrous magnesium sulfate (MgSO<sub>4</sub>) and sodium chloride (NaCl) were purchased from Agilent (Santa Clara, California, USA). Bulk sorbents of primary-secondary amine (PSA), graphitized carbon black (GCB), C18, and sodium citrate dibasic sesqui-hydrate were also purchased from Agilent. The HPLC grade

water used for analysis and also for hydration of the samples was obtained using an ARIOSO water purification system (Korea).

#### Samples

**Surveyed samples.** Green coffee bean samples used for the initial monitoring of residues were sampled from 3 coffee producing regions, namely Lampung (35 samples), Aceh (14 samples), and East Java province (31 samples) between October and November 2021.

**Routine sample analysis of coffee exports.** Furthermore, pesticide residue analyses of isoprocarb and carbaryl were continuously conducted for the coffee export (1385 samples) from November 2021 to December 2022, receiving rejection notifications.

#### Preparation of the standard and spiked samples

The LC and GC pesticide kit stock solutions were dissolved in acetonitrile to generate a multi-component mix solution containing 320 pesticide residues at a concentration of 5 mg  $L^{-1}$ . Then, employing appropriate fortification on each of 5 g of homogenized milling samples, standard additions (matrix based standards) were made at concentrations of 10, 20, 50, 100, and 200  $\mu g \ kg^{-1}$  and prepared following the sample preparation procedure. This standard series was used to quantify the residues of the initial monitoring sample test.

Individual stock solutions of isoprocarb and carbaryl at  $1000 \text{ mg L}^{-1}$  were made by weighing approximately 5 mg into a 10 mL flask and diluting with acetonitrile. The preparation, storage conditions, and storage time of the stock standard followed procedures from EU guidance18 to ensure the stability and purity of the standard. The stock solutions were kept in a freezer at -20 °C. A working solution mixture of isoprocarb and carbaryl with a concentration of 5 mg  $L^{-1}$  was prepared by appropriate dilution in acetonitrile. Then, spiked samples for the recovery study and standard additions at the concentration levels of 1, 5, 10, 50, and 100 μg kg<sup>-1</sup> were made by spiking working solution in 5 g of the sample in 5 g of finely ground green coffee sample. After vortexing, the solvent of working solution has definitely evaporated, cold water was added and the spiked sample was then stored at - 20 °C for 1 h to make a contaminated sample comprising the incurred real sample prior to the sample preparation procedure as we did in a previous study in another part.19 This addition standard series was used for the quantitation of routine sample analysis.

The use of standard addition could eliminate the matrix effect and improve the test result accuracy, but it will waste time and chemicals in its preparation. Since pesticides could remain stable in sample extracts used as calibrants for several days,  $^{20,21}$  and the stability study of isoprocarb and carbaryl in sample extracts of coffee beans was performed. The storage stability of the spike concentrations of 10, 50, and 100  $\mu$ g kg $^{-1}$ , which were stored at 25, 4, and -20 °C in clear and amber vials, respectively, was observed for 7 days.

## Sample preparation

Green coffee beans and roasted coffee. Immediately after the green coffee samples were received at the laboratory, 500 g of

the legal sample obtained by sampling was homogenized in a plastic bag and divided into two parts. The first part was stored overnight at -60 °C and ground using a Retsch mill (Germany) and the second part was stored at -20 °C for control purposes. It was ensured that the temperature didn't increase while the grinding was being conducted. The QuEChERS method, which had previously been used to analyze flutriafol and pyraclostrobin in coffee beans, 10 was used to extract the samples. In the beginning, the effect of GCB on the recovery of carbaryl, which has a planar structure as highlighted by Wang et al.,22 was investigated using a standard solution of  $100 \text{ mg L}^{-1}$ .

In a 50 mL centrifuge tube, 5 g of milled samples were weighed, and 10 mL of cold water was added. To make the sample hydrated, the mixture was left out for roughly 15 min. For extraction, 15 mL of acetonitrile was added and vortexed for 30 s, and extraction was continued for 30 min using an automated agitator. Salt (4 g MgSO<sub>4</sub>, 1 g NaCl, and 0.5 g sodium citrate dibasic sesquihydrate) was added, immediately shaken by hand, and continuously shaken in an automated agitator for 15 min. The tube was centrifuged at 7000 rpm for 5 min to obtain a partition between the water and organic layers. The organic phase (6 mL) was transferred to a 15 mL tube containing 4 g MgSO<sub>4</sub>, 0.4 g C18, 0.4 g PSA, and 0.2 g GCB. The tube was vigorously shaken for 1 min and centrifuged again. For initial monitoring of residues, 1 mL of clean acetonitrile extract was passed through a 0.22 µm PTFE syringe filter and placed in a vial for further measurement by LC-MS/MS and GC-MS/MS. Meanwhile, 3 mL of the extract was evaporated using a nitrogen evaporator until dry and then redissolved in 0.5 mL of acetonitrile for continuous specific measurements of isoprocarb and carbaryl residues. This evaporation or concentration step was used to improve the detection capability of LC-MS/ MS. The same procedure was employed for roasted coffee.

Brewed coffee. 15 g of finely roasted coffee powder was brewed with 250 mL boiling water (distilled) in an Erlenmeyer flask for 10-12 min. After brewing, the flask was immersed in cold water for 10 min for cooling, followed by separating the infusion from the coffee ground by passing it through filter paper (Whatman No. 1). For pesticide determination, 10 mL of brewed coffee solution was extracted using ethyl acetate as described by R. E. Kartasasmita.23 The final residue was then redissolved using acetonitrile. For quantitation, standard addition with the concentration levels of 1.0, 2.5, 10, 20, 30, and 50  $\mu g L^{-1}$  was used.

#### Instrumentation

LC-MS/MS instrument. A Triple Quadrupole 3500 LC-MS/MS system (AB Sciex, USA) coupled with a 1290 Infinity Liquid Chromatography system (Agilent) was used to analyze pesticides amenable to LC. The multiple reaction monitoring (MRM) operation of the target list of the analytes including isoprocarb and carbaryl is summarized (Table S1†). Water and acetonitrile containing 0.1% formic acid were used as mobile phases A and B respectively. For the separation of 210 pesticides in initial monitoring, a Merck Purospher STAR RP-18 Endcapped Hibar HPLC Column ( $L \times ID$ , 150 × 4.6 mm) was used with a flow rate

of 0.5 mL min<sup>-1</sup> under the following gradient conditions: 5% B (0-2 min); 5-75% B (2-5 min); 100% B (5-9 min); 100-0% B (9-12.5 min); 5% B (12-12.5 min); 5% B (12.5-16 min). Meanwhile for continuous analysis of isoprocarb and carbaryl, a Chromolith® RP-18e End-capped HPLC Column ( $L \times ID$ , 100  $\times$  3 mm) was used with a flow rate of 0.4 mL min<sup>-1</sup> and a shorter gradient elution: 10% B (0-1 min); 10-90% B (1-4 min); 90% B (4-6 min); 90-10% B (6-7 min); 10% B (7-7.5 min).

GC-MS/MS instrument. A Shimadzu GC-MS-TQ8050 (Shimadzu Scientific Instruments, Japan) was used to analyze 200 pesticides amenable to GC. The chromatographic separation was done using a capillary HP-5MS column. The GC operation conditions and MRM parameters for target pesticides are shown respectively in Tables S2 and S3.†

#### Method validation

The validation of the isoprocarb and carbaryl analysis methods in coffee beans was carried out following the procedures and performance criteria of the European Commission, SANTE 11312/2021.18 Method performance was evaluated through validation parameters: linearity, precision, recovery, and limit of quantification (LOQ). The matrix effect was investigated using standards prepared in solvents and sample extracts and then calculated using the following formula.

$$ME(\%) = \left(\frac{\text{slope of calibration curve in matrix}}{\text{slope of calibration curve in solvent}} - 1\right) \times 100$$
(1)

## Coffee processing

The contaminated green coffee samples (4 samples of robusta) with isoprocarb and carbaryl were roasted using a Probat Probatino coffee roaster at 3 different roasting levels (light, medium, and dark) based on the Agtron System/Specialty Coffee Association of America (SCAA) classification. The light-roast bean was exposed to 200 °C for 8 min or right at the first crack; the medium-roast bean was exposed to 220 °C for 12 min or to the end of the first crack or the beginning of the second crack; the dark-roast bean was exposed to 230 °C for 14 min or at the end of the second crack. The residue at each roasting level and its infusion were analyzed in triplicate. The processing factor (PF) that indicates the effect of coffee processing on the level of pesticide residue was evaluated using the formula:

$$PF = \frac{\text{concentration of pesticides in roasted coffee}(\mu g \ kg^{-1})}{\text{concentration of pesticides in green bean}(\mu g \ kg^{-1})}$$
(2)

The transfer rate of pesticide residue from roasted coffee to its infusion was calculated using the following equation.24

$$R = C_{\rm in} \times V_{\rm in}/C \times M \times 100 \tag{3}$$

where R is the transfer rate (%), Cin is the concentration of pesticide in infusion coffee ( $\mu g L^{-1}$ ),  $V_{in}$  is the volume of coffee infusion (mL), C is the concentration of pesticide in roasted coffee ( $\mu g \ kg^{-1}$ ) and M is the weight of roasted coffee (g).

#### Deterministic risk assessment

Chronic exposure to isoprocarb and carbaryl residues from coffee was calculated by comparing consumption data by age group from Indonesia<sup>25</sup> and the United States,<sup>26</sup> the country with the highest consumption. Risk as the chronic hazard quotient (HQ) was obtained from the quotient between exposure or estimated daily intake (EDI, mg per kg BW) and acceptable daily intake (ADI, mg per kg BW). The ADI of isoprocarb (0.002 mg per kg BW) and carbaryl (0.008 mg per kg BW) was obtained from the report of the Joint FAO/WHO Meeting on Pesticide Residues (JMPR) and other references.<sup>27,28</sup> The calculation formula are eqn (4) and (5).

$$EDI = \frac{C_i \times F}{BW} \tag{4}$$

$$HQ = \frac{EDI}{ADI} \tag{5}$$

The cumulative risk due to isoprocarb and carbaryl in coffee was assessed as a hazard index (HI) using eqn (6).

$$HI = HQ_{isoprocarb} + HQ_{carbaryl}$$
 (6)

Pesticide exposure poses no risk to the consumer if the HI and/or HQ value is less than or equal to 1. In contrast, if the HI and/or HQ value is greater than 1, the consumer's health risk is unacceptable.

## Results and discussion

#### Method validation

The evaluation of the validation results (Table 1) can be declared to meet the requirements in the European guideline;18 therefore, the method used in this study is valid for carbaryl and isoprocarb determination. Good repeatability and reproducibility precision were obtained from fortified coffee samples at concentration levels of 0.001 mg kg<sup>-1</sup>, 0.01 mg kg<sup>-1</sup>, and  $0.05 \text{ mg kg}^{-1}$ , i.e. in the range of 2.0-10.6% and 5.2-15.2%, respectively. All fortified levels showed excellent recoveries (ranging from 85.6 to 102.6%). In addition, after being purified with GCB, the fortified sample resulted in isoprocarb and carbaryl recoveries of 102.6  $\pm$  1.9% and 83.2  $\pm$  1.9%, respectively. These results confirm that planar pesticides can be strongly absorbed by GCB as previously reported by Li et al.29 The recovery reduction of carbaryl was not particularly considerable (<20%) and remains within the performance criteria of the method validation guidelines in SANTE 11312/2021.18 Since the extract was concentrated six times through evaporation, high sensitivity was attained (a LOQ of 0.001 mg kg<sup>-1</sup> in green and roasted coffee). This LOQ value indicated that the test method's detection sensitivity was up to 10 times below the standard (Japanese regulation). In brewed coffee, achieving a lower LOQ (0.0001 mg L<sup>-1</sup>) was made possible by preconcentrating the sample 10 times before injection. This high factor was feasible due to the very low matrix effects, a consequence of the dilution that occurs during brewing.30 Further

 Table 1
 Validation result and uncertainty estimation of isoprocarb and carbaryl in green coffee beans

	Value		
Analytical parameter	Isoprocarb	Carbaryl	
Recovery $(n = 6)$ , $(\%)$			
Level 0.001 (mg kg <sup>-1</sup> )	92.4	88.1	
Level 0.01 (mg kg <sup>-1</sup> )	87.3	102.6	
Level 0.05 (mg kg <sup>-1</sup> )	85.6	100.8	
Repeatability $(n = 6)$ , RSD <sub>r</sub> (%)			
Level 0.001 (mg kg <sup>-1</sup> )	10.6	5.5	
Level 0.01 (mg kg <sup>-1</sup> )	3.1	2.0	
Level $0.05 \text{ (mg kg}^{-1}\text{)}$	5.2	5.3	
Within-laboratory reproducibility, RSD <sub>R</sub> (%)			
Level 0.001 (mg kg <sup>-1</sup> )	11.3	15.2	
Level 0.01 (mg kg <sup>-1</sup> )	6.9	5.3	
Level $0.05 \text{ (mg kg}^{-1)}$	9.4	5.2	
Linear range (mg kg <sup>-1</sup> )	0.001 – 0.100	0.001-0.100	
Correlation coefficients (r)	1.0000	0.9994	
Limit of quantification (LOQ), (mg kg <sup>-1</sup> )	0.001	0.001	
Expanded measurement uncertainty, $U'$ (%)	33	35	

information on the method validation result of roasted coffee and brewed coffee is shown in Table S4.† The QuEChERS method had a low matrix effect for isoprocarb (ME = -15%) and a moderate matrix effect for carbaryl (ME = 28%). Therefore, a matrix matched standard was applied to reduce the error and increase the accuracy.

From the stability observation, isoprocarb and carbaryl in coffee bean extracts can be stable for 7 days when stored at 4 °C and -20 °C, respectively (Table 2). Meanwhile, the effect of solvent evaporation may cause a trend of increasing residue for the standard stored at room temperature, 25 °C, causing a higher uncertainty result. Therefore, it can be inferred that daily preparation of the addition standard is unnecessary and it can be stored for 7 days at 4 °C and -20 °C before being measured for pesticide determination. This finding was triggered by the fact that other pesticide standards in plant extracts of ethyl acetate were stable for more than 40 days when stored at low temperatures.<sup>20</sup>

#### On-going method validation

The on-going method validation was necessary to be observed for long-term purposes through the recovery of fortified samples and the deviation of sample replicas that were analyzed by internal quality control (IQC) as shown Fig. 1. A blank coffee sample previously tested to be free of pesticides was fortified at concentration levels of  $0.01~\rm mg~kg^{-1}$  and  $0.05~\rm mg~kg^{-1}$  alternately during routine daily analysis. Outstanding performance was indicated by the high percentage (>90%) of internal quality assurance of recovery observations, which are in the range of 60–140% and % RSD < 20% for 14 months. As a result, all recoveries still meet the acceptance criteria for routine recovery tests (60–140%). Testing results with % RSD > 20 have been investigated and re-analyzed. Thus, this observation of the acceptability of IQC is very important to ensure the accuracy of the method applied in this study for long routine analysis periods.

Table 2 Isoprocarb and carbaryl stability in coffee bean extract stored at 25 °C, 4 °C, and -20 °C in clear and amber vials

	Recovery (%)											
Isoprocarb							Carbaryl					
	Storage conditions				Storage conditions							
	25 °C		4 °C			4 °C						
Storage period (day)	Clear vial	Amber vial	Clear vial	Amber vial	Clear vial	Amber vial	Clear vial	Amber vial	Clear vial	Amber vial	Clear vial	Amber vial
Spike level: 0.01 m	ig kg <sup>-1</sup>											
1	$93\pm1$	$98\pm1$	$93 \pm 5$	$105\pm1$	$92 \pm 2$	$102\pm4$	$92\pm12$	$86\pm2$	$90 \pm 6$	$88 \pm 9$	$82\pm6$	$94\pm3$
2	$116\pm7$	$122\pm5$	$114\pm2$	$116\pm 6$	$116\pm4$	$118\pm2$	$97\pm15$	$112\pm 8$	$108\pm2$	$92\pm1$	$97 \pm 2$	$94 \pm 5$
3	$172\pm4$	$142\pm2$	$117\pm1$	$117\pm1$	$117\pm7$	$114\pm4$	$145\pm29$	$184\pm6$	$127\pm4$	$118\pm1$	$105\pm2$	$95\pm4$
4	$154\pm3$	$146\pm1$	$101\pm3$	$118\pm1$	$108 \pm 4$	$113\pm3$	$160 \pm 28$	$130\pm1$	$99\pm3$	$97\pm8$	$87\pm7$	$88\pm2$
7	$193\pm4$	$187\pm1$	$88 \pm 3$	$96 \pm 0$	$88 \pm 1$	$96\pm16$	139 ± 18	$161 \pm 1$	$77 \pm 5$	$74 \pm 0$	$78 \pm 4$	$81 \pm 6$
Spike level: 0.05 m	ng kg <sup>-1</sup>											
1	$110 \pm 4$	$114\pm2$	$108\pm3$	$107\pm2$	$103\pm2$	$107\pm4$	$99\pm2$	$100\pm3$	$95\pm1$	$94\pm3$	$93\pm3$	$92\pm 5$
2	$119\pm2$	$113\pm2$	$113\pm4$	$114\pm7$	$113\pm 6$	$109\pm2$	$115\pm4$	$115\pm1$	$108 \pm 6$	$107 \pm 4$	$120\pm7$	$106\pm2$
3	$137\pm1$	$108\pm2$	$119\pm4$	$120\pm4$	$106\pm0$	$110\pm2$	$106\pm1$	$86\pm2$	$95\pm4$	$96\pm2$	$104\pm1$	$89 \pm 1$
4	$160\pm1$	$151\pm4$	$114\pm1$	$127\pm4$	$116\pm1$	$116\pm0$	$136\pm2$	$129\pm4$	$96\pm1$	$110\pm1$	$98\pm1$	$93\pm0$
7	$151\pm 6$	$162\pm2$	$91 \pm 1$	$115\pm5$	$84 \pm 2$	$109\pm5$	$134\pm3$	$152\pm 6$	$77 \pm 4$	$102\pm5$	$73 \pm 0$	$101\pm2$
Spike level: 0.100 i	ng kg <sup>-1</sup>											
1		$105\pm5$	$110\pm2$	$115\pm1$	$105\pm2$	$100\pm1$	$81\pm2$	$85\pm7$	$84\pm2$	$89\pm0$	$84\pm3$	$113\pm5$
2	$117\pm1$	$111\pm4$	$104 \pm 4$	$110\pm3$	$110\pm3$	$108\pm4$	$104 \pm 6$	$101\pm3$	$92\pm4$	$100\pm3$	$99 \pm 4$	$100\pm2$
3	$141\pm0$	$134\pm2$	$110\pm1$	$122\pm 6$	$114\pm1$	$112\pm4$	$101\pm1$	$94\pm2$	$100\pm2$	$97\pm3$	$88\pm2$	$86\pm3$
4	$123\pm7$	$130\pm 8$	$125\pm 6$	$124\pm4$	$109 \pm 4$	$111\pm 9$	$98 \pm 9$	$131\pm4$	$99\pm2$	$100\pm5$	$86\pm3$	$88\pm7$
7	$183\pm1$	$144\pm 6$	$100\pm3$	$91 \pm 4$	$92\pm1$	$84\pm1$	$142\pm 6$	$184\pm5$	$82\pm4$	$72\pm4$	$121 \pm 4$	$81\pm7$

The chromatogram peaks of a positive sample of isoprocarb (248  $\mu$ g kg<sup>-1</sup> with 5× dilution) and carbaryl (57  $\mu$ g kg<sup>-1</sup>) in green beans are presented in Fig. 2. For early eluting chemicals, on the other hand, interference from the simultaneous extraction of the coffee matrix can have a considerable impact on the analyte's signal response. Contrarily, interference caused by the coextractive compounds of the coffee matrix (e.g. caffeine, theobromine, etc.) has not considerably influenced the signal

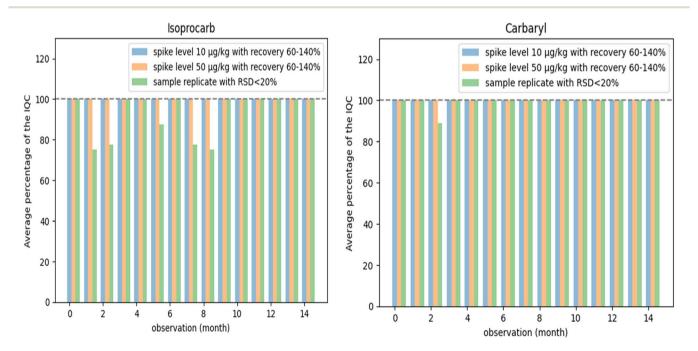
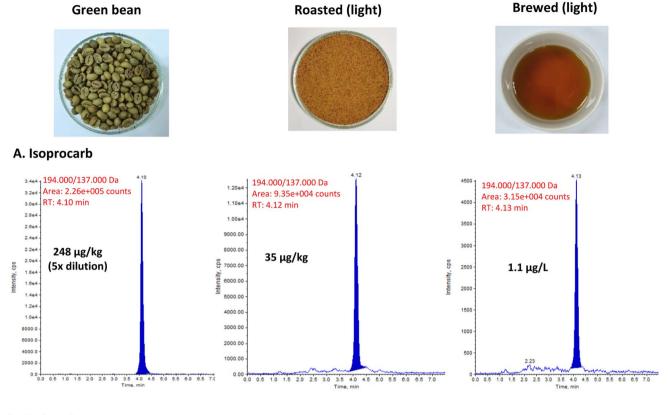


Fig. 1 The percentage average of internal quality control (IQC) in routine analysis (7 to 15 analyses every month) with recovery within the range 60-140% and % RSD < 20.



## **B.** Carbaryl

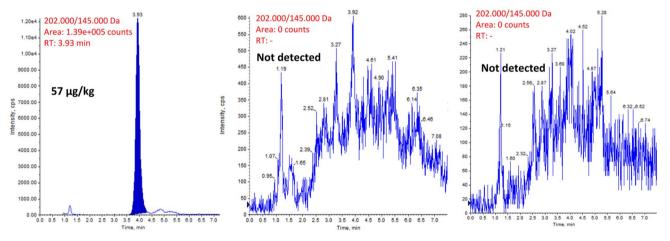


Fig. 2 LC-MS/MS chromatogram of (a) isoprocarb and (b) carbaryl in contaminated robusta coffee during coffee processing

response with our target compounds as previously reported.<sup>2</sup> Isoprocarb was still detected down to 1.1  $\mu$ g L<sup>-1</sup> in brewed coffee. Meanwhile, carbaryl was undetectable after the roasting process.

#### Investigation of isoprocarb and carbaryl in Indonesian coffee

Occurrence of pesticides in surveyed coffee samples. Carbamate insecticides of isoprocarb and carbaryl were detected in around 34% of all samples (80 samples) and all detected samples were originally from Lampung province (robusta coffee). High concentrations of carbaryl and isoprocarb were detectable in one and three samples, respectively. Following

these findings, continuous monitoring is required to control residue levels and evaluate compliance with the safety standards of importing countries. Insecticides isoprocarb and carbaryl were used extravagantly to repel ants before harvest, generating residual problems on green coffee beans. In addition, a few other pesticides (320 pesticides, Table S5†) such as chlorpyrifos were detected but below the LOQ. Therefore, addressing the presence of carbaryl and isoprocarb was key to eliminating rejection notifications during exportation.

**Isoprocarb and carbaryl in routine sample analysis of coffee exports.** Robusta (57.76%) coffee has more contaminated

samples than arabica coffee (6.17%), as indicated by the distribution of pesticides shown in Fig. 3. This method was appraised as applicable for the evaluation of isoprocarb and carbaryl in positive samples of robusta coffee in the range of  $0.001 \text{ mg kg}^{-1}$  to  $0.239 \text{ mg kg}^{-1}$  and  $0.001 \text{ mg kg}^{-1}$  to 0.249 mgkg<sup>-1</sup>, respectively, and positive samples of arabica coffee, each in the range of 0.001 mg  $\rm kg^{-1}$  to 0.022 mg  $\rm kg^{-1}$  and 0.001 mg kg<sup>-1</sup> to 0.020 mg kg<sup>-1</sup>. Purified extract from a blank coffee sample was used to dilute the sample extract, which had a higher pesticide content than the addition calibration standard's highest level (0.100 mg kg<sup>-1</sup>), during sample analysis. From these results, the method in this study can be applied to determine isoprocarb and carbaryl for long-term routine analysis from November 2021 to December 2022 (1385 samples). Surprisingly, the detected samples in this study were reduced by more than 90% in the last two months (November to December 2022). These results can be considered as a benefit derived from the Ministry of Agriculture's official dissemination of good agricultural practices (GAP) to farmers.<sup>13</sup> Mechanical trapping was found to be an effective technique for controlling black ants (Dolichoderus thoracicus) on Indonesian coffee plants.31

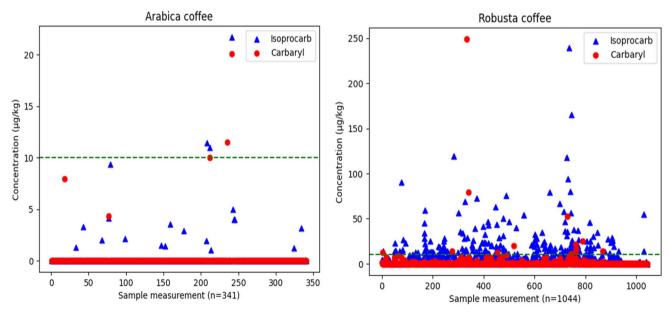
Isoprocarb and carbaryl have been detected worldwide, not only in coffee and agricultural products32-37 but also in other products, including animal products, 38,39 environmental materials,40-47 and biological samples,48 as summarized in Table 3. Beyond being beneficial in controlling insecticidetarget organisms, the use of carbamate pesticides has a negative impact on non-target organisms and the environment. Therefore, the evaluation of its occurrence and its effect on humans and the environment is essential as a consequence of applying this carbamate pesticide.

#### Effect of coffee processing

The roasting of coffee can reduce or even remove pesticide residues. Previous studies investigated some organochlorine,

pyrethroid, and organophosphate pesticides using spiked/ fortified samples rather than those using incurred pesticide or naturally contaminated samples. As a result, residues were still detected in roasted coffee for stable organochlorine pesticides (reduction > 80%) and deltamethrin (reduction > 70%). Meanwhile for other organophosphates and pyrethroids, the reduction was 100%.3,49 In comparison, the field trial's contaminated sample of the organonitrogen compound dinotefuran was reduced by 62.2% to 100% as a result of the roasting process, and residues could still be detected in the brew. 50 As revealed in Table 4, the coffee roasting process can reduce the residues by up to 100% for isoprocarb at medium and dark roasting degrees and for carbaryl at all roasting degrees in contaminated robusta coffee. Isoprocarb reduction ranges from 83.60 to 93.55% at a light degree.

Previous studies on coffee roasting mentioned that physicochemical properties, such as solubility and vapor pressure, might be responsible factors for the pesticide loss caused by evaporation and thermal degradation.3,49,50 Pesticides characterized as volatile with low water solubility and high vapor pressure could considerably contribute to their residue reduction.51 Pesticides with low vapor pressure cannot be easily removed by heating. From their physicochemical properties, isoprocarb (solubility =  $504.1 \text{ mg L}^{-1}$  at 25 °C and vapor pressure =  $1.01 \times 10^{-4}$  mmHg at 25 °C) and carbaryl (solubility = 416.2 mg L<sup>-1</sup> at 25 °C and vapor pressure =  $2.09 \times 10^{-5}$  mmHg at 25 °C) as an insecticide carbamate derived from a carbamic acid have moderate solubility and low volatility. However, both are more volatile than organochlorine, organophosphate, and pyrethroid pesticides, the residues of which can be lost during coffee bean roasting, as previously observed.3,49 Additionally, other physical properties, such as molecular weight, are related to the phenomenon of residue loss as a result of thermal processes.<sup>52</sup> The small molecular size of isoprocarb (193.242 g mol<sup>-1</sup>) can cause a higher penetration rate or diffusion into the



Isoprocarb and carbaryl distribution in arabica coffee (n = 341) and robusta coffee (n = 1044) detected by the method.

Table 3 Comparison of isoprocarb and carbaryl concentrations in coffee and other products worldwide

Country	Product	Pesticides	No. (%) of positive samples	Concentration	Ref.
Indonesia	Green coffee	Isoprocarb	678 (49)	0.001-0.239 mg kg <sup>-1</sup>	This study
		Carbaryl	200 (14)	$0.001$ – $0.249 \text{ mg kg}^{-1}$	
United Arab Emirates	Milk	Carbaryl	9 (53)	$0.606-7.771 \ \mu g \ kg^{-1}$	38
China	Greenhouse cucumbers	Isoprocarb	16 (9)	0.03-1.13 mg kg <sup>-1</sup>	32
Ethiopia	Fish	Carbaryl	_ ` `	0.2-56.5 μg kg <sup>-1</sup>	39
Brazil	Sweet pepper	Carbaryl	_	$5 \text{ mg kg}^{-1}$	33
Malaysia	Cocoa bean	Isoprocarb	2 (1)	$0.010$ – $0.017 \text{ mg kg}^{-1}$	34
South Africa	Air	Carbaryl	54 (100)	$Median = 0.02 \text{ ng m}^{-3}$	40
		•	,	$Max = 1.3 \text{ ng m}^{-3}$	
USA	Bat hair	Carbaryl	_	41.4-216.7 pg mg <sup>-1</sup>	48
China	Surface water from a lake	Isoprocarb	208 (100)	Average = $17 \text{ ng L}^{-1}$	41
		•		$Max = 406 \text{ ng L}^{-1}$	
USA	Dust	Carbaryl	163 (64.7)	$Median = 22.4 \text{ ng g}^{-1}$	42
Pakistan	Date palm fruit	Carbaryl	1 (5)	2.8 μg kg <sup>-1</sup>	35
Poland	Soil	Carbaryl	45 (20)	$< 0.01-28.07 \ \mu g \ kg^{-1}$	43
Africa	Air	Carbaryl	68 (41)	_	44
Malawi	Surface water	Carbaryl		$0.083  0.254 \text{ mg L}^{-1}$	45
	Groundwater	Carbaryl	_	$0.07$ – $0.492 \text{ mg L}^{-1}$	
	Soil	Carbaryl	_	$1.154-1.305~{ m mg~L^{-1}}$	
China	Surface watersheds	Isoprocarb	25 (100)	$0.47$ –39.06 ng $L^{-1}$	46
Spain	Citrus fruit	Carbaryl	94 (1)	$0.03 \text{ mg kg}^{-1}$	36
Bangladesh	Water from paddy	Carbaryl	2 (7)	$14.1$ – $18.1~\mu g~L^{-1}$	47
	and vegetable fields	•	. ,	. 0	
Saudi Arabia	Tomato	Carbaryl	1 (0.6)	$0.390 \text{ mg kg}^{-1}$	37
	Squash	•	3 (1.9)	$0.209-1.148 \text{ mg kg}^{-1}$	
	Cucumber		3 (1.9)	$0.384-1.457 \text{ mg kg}^{-1}$	
	Egg-plant		2 (1.3)	$1.686-1.917 \text{ mg kg}^{-1}$	
	Green pepper		2 (1.3)	$0.653-2.228 \text{ mg kg}^{-1}$	
	Lettuce		3 (1.9)	$0.538-1.641 \text{ mg kg}^{-1}$	
	Carrot		1 (0.6)	$0.891 \text{ mg kg}^{-1}$	
	Cabbage		3 (1.9)	$0.069-1.765 \text{ mg kg}^{-1}$	

coffee bean matrix, which may indicate why the residue still remains in light-roasted coffee.

The transfer rate of isoprocarb from the light degree of roasted coffee to the brew is shown in Table 4, ranging from 0 to 69.58%. Isoprocarb can be transformed into their metabolites through thermal processing.<sup>17</sup> It implied that exposure to isoprocarb and other possible metabolites is feasible because the residue can enter the human body through drinking coffee if it is roasted to brightness. Conversely, carbaryl will remain as a precursor when it evaporates during thermal processing.<sup>14</sup> Consequently, it can be concluded that the presence of the probable carcinogenic carbaryl in coffee beans does not correlate with health risks due to coffee consumption.

#### Risk assessment

Using mean residue and consumption data, a simple calculation of chronic dietary exposure was completed using a deterministic approach. Based on age group comparisons between populations in Indonesia and the United States, the risk was calculated and differentiated for robusta and arabica coffee, as shown in Table 5. These results were calculated using the worst-case upper-bound (UB) scenario, where test results below the LOQ were replaced with the LOQ value. The estimated body weights of the Indonesian population used in the calculation

are 11.67 kg (0–59 months), 27.5 kg (5–12 years), 46.33 kg (13–18 years), 57.87 kg (19–55 years), 52.3 kg (>55 years), and 50.78 (all ages). Meanwhile for the US population, an average body weight of 60 kg was used for all groups of adult age.  $^{54}$ 

The hazard quotient (HQ) for isoprocarb (5.24  $\times$  10<sup>-6</sup> to  $2.18 \times 10^{-2}$ ) was higher than that of carbaryl (1.22  $\times$  10<sup>-6</sup> to  $1.33 \times 10^{-3}$ ). Robusta coffee has a much higher HQ value than arabica coffee because it was more contaminated with isoprocarb and carbaryl. When comparing the populations of Indonesia and the US, the risk is lower for Indonesian consumers in comparison to US consumers. Children (0-59 months) in Indonesia have the lowest HQ values, which can be associated with comparatively low consumption. Remarkably, comparable findings were observed when an evaluation was executed to identify the highest risk by age group, revealing that the highest risk was found for Indonesian customers aged over 55 years and US consumers aged 51-70 years. Overall, the potential risk of pesticide exposure through Indonesian coffee consumption did not pose a serious concern to human health, as confirmed by HQ and/or HI values of less than 1 for all age groups for both robusta and arabica coffee. Similarly, a low risk was also found for exposure to other pesticides in coffee.55

**Table 4** Effect of coffee processing on pesticide residues in robusta coffee reported with standard deviation from replicated samples  $(n = 3)^{a,b}$ 

		Concentration (μ	g kg <sup>-1</sup> )				
Sample code	Pesticide	Green bean	Roasted coffee	Brewed coffee	Processing factor (PF)	Transfer rate (%)	
Robusta A	Isoprocarb	$35.20 \pm 4.33$	$Light = 2.27 \pm 0.16$	ND	0.064	N/A	
	•		Medium = ND	N/A	N/A	N/A	
			Dark = ND	N/A	N/A	N/A	
	Carbaryl	$57.20 \pm 10.55$	Light = ND	N/A	N/A	N/A	
	·		Medium = ND	N/A	N/A	N/A	
			Dark = ND	N/A	N/A	N/A	
Robusta B	Isoprocarb	$248.33 \pm 21.57$	Light = $35.43 \pm 5.53$	$1.117\pm0.061$	0.143	L = 56.75	
	•		Medium = ND	N/A	N/A	N/A	
			Dark = ND	N/A	N/A	N/A	
	Carbaryl	$9.05\pm0.74$	Light = ND	N/A	N/A	N/A	
			Medium = ND	N/A	N/A	N/A	
			Dark = ND	N/A	N/A	N/A	
Robusta C	Isoprocarb	$21.90 \pm 1.97$	$Light = 3.57 \pm 0.23$	$0.138 \pm 0.007$	L = 0.163	L = 69.58	
	•		Medium = ND	N/A	N/A	N/A	
			Dark = ND	N/A	N/A	N/A	
	Carbaryl	$12.57\pm0.97$	Light = ND	N/A	N/A	N/A	
	•		Medium = ND	N/A	N/A	N/A	
			Dark = ND	N/A	N/A	N/A	
Robusta D	Isoprocarb	$10.55 \pm 0.78$	$Light = 1.73 \pm 0.24$	N/A	L = 0.164	N/A	
	•		Medium = ND	N/A	N/A	N/A	
			Dark = MD	N/A	N/A	N/A	
	Carbaryl	$13.07\pm1.51$	Light = ND	N/A	N/A	N/A	
	•		Medium = ND	N/A	N/A	N/A	
			Dark = ND	N/A	N/A	N/A	

<sup>&</sup>lt;sup>a</sup> ND = not detected (below the LOQ). <sup>b</sup> N/A = not applicable to be analyzed and/or calculated because the residue was not detected after the roasting process.

Table 5 Comparison of chronic risk assessment between Indonesia and the country with the highest consumption, the United States

					Risk index			
		Age group	EDI (mg per k	g BW per day)	Hazard quotient, HQ			
Commodity	Country/population		Isoprocarb	Carbaryl	Isoprocarb	Carbaryl	Hazard index, HI	
Arabica coffee	Indonesia United States	0–59 months 5–12 years 13–18 years 19–55 years >55 years All ages 20–30 31–50 51–70 ≥71	$1.05 \times 10^{-8}$ $1.78 \times 10^{-8}$ $5.01 \times 10^{-8}$ $1.65 \times 10^{-7}$ $1.68 \times 10^{-7}$ $1.44 \times 10^{-7}$ $2.86 \times 10^{-6}$ $5.38 \times 10^{-6}$ $7.51 \times 10^{-6}$ $6.31 \times 10^{-6}$	$9.80 \times 10^{-9}$ $1.66 \times 10^{-8}$ $4.69 \times 10^{-8}$ $1.54 \times 10^{-7}$ $1.57 \times 10^{-7}$ $1.35 \times 10^{-7}$ $2.68 \times 10^{-6}$ $5.03 \times 10^{-6}$ $7.02 \times 10^{-6}$ $5.90 \times 10^{-6}$	$5.24 \times 10^{-6}$ $8.89 \times 10^{-6}$ $2.51 \times 10^{-5}$ $8.24 \times 10^{-5}$ $8.42 \times 10^{-5}$ $7.22 \times 10^{-5}$ $1.43 \times 10^{-3}$ $2.69 \times 10^{-3}$ $3.76 \times 10^{-3}$ $3.15 \times 10^{-3}$	$\begin{array}{c} 1.22\times10^{-6}\\ 2.08\times10^{-6}\\ 5.86\times10^{-6}\\ 1.93\times10^{-5}\\ 1.97\times10^{-5}\\ 1.69\times10^{-5}\\ 3.35\times10^{-4}\\ 6.29\times10^{-4}\\ 8.78\times10^{-4}\\ 7.37\times10^{-4} \end{array}$	$6.46 \times 10^{-6}$ $1.10 \times 10^{-5}$ $3.09 \times 10^{-5}$ $1.02 \times 10^{-4}$ $1.04 \times 10^{-4}$ $8.91 \times 10^{-5}$ $1.77 \times 10^{-3}$ $3.32 \times 10^{-3}$ $4.63 \times 10^{-3}$ $3.89 \times 10^{-3}$	
Robusta coffee	Indonesia United States	All ages 0–59 months 5–12 years 13–18 years 19–55 years >55 years All ages 20–30 31–50 51–70 ≥71 All ages	$5.66 \times 10^{-6}$ $6.09 \times 10^{-8}$ $1.03 \times 10^{-7}$ $2.91 \times 10^{-7}$ $9.58 \times 10^{-7}$ $9.78 \times 10^{-7}$ $1.66 \times 10^{-5}$ $3.13 \times 10^{-5}$ $4.36 \times 10^{-5}$ $3.66 \times 10^{-5}$ $3.29 \times 10^{-5}$	$5.29 \times 10^{-6}$ $1.48 \times 10^{-8}$ $2.52 \times 10^{-8}$ $7.09 \times 10^{-8}$ $2.33 \times 10^{-7}$ $2.38 \times 10^{-7}$ $2.04 \times 10^{-6}$ $7.61 \times 10^{-6}$ $1.06 \times 10^{-5}$ $8.92 \times 10^{-6}$ $8.01 \times 10^{-6}$	$2.83 \times 10^{-3}$ $3.04 \times 10^{-5}$ $5.17 \times 10^{-5}$ $1.46 \times 10^{-4}$ $4.79 \times 10^{-4}$ $4.89 \times 10^{-4}$ $4.20 \times 10^{-4}$ $8.32 \times 10^{-3}$ $1.56 \times 10^{-2}$ $2.18 \times 10^{-2}$ $1.83 \times 10^{-2}$ $1.64 \times 10^{-2}$	$6.62 \times 10^{-4}$ $1.85 \times 10^{-6}$ $3.14 \times 10^{-6}$ $8.87 \times 10^{-6}$ $2.91 \times 10^{-5}$ $2.98 \times 10^{-5}$ $2.55 \times 10^{-5}$ $5.07 \times 10^{-4}$ $9.51 \times 10^{-4}$ $1.33 \times 10^{-3}$ $1.12 \times 10^{-3}$ $1.00 \times 10^{-3}$	$3.49 \times 10^{-3}$ $3.23 \times 10^{-5}$ $5.48 \times 10^{-5}$ $1.55 \times 10^{-4}$ $5.08 \times 10^{-4}$ $5.19 \times 10^{-4}$ $4.45 \times 10^{-4}$ $8.83 \times 10^{-3}$ $1.66 \times 10^{-2}$ $2.31 \times 10^{-2}$ $1.94 \times 10^{-2}$ $1.74 \times 10^{-2}$	

## Conclusion

The validated QuEChERS method for detecting isoprocarb and carbaryl insecticides in coffee beans has good performance, with a high percentage (>90%) of recovery in the range of 60–140% and % RSD of sample replication < 20%. Following the application of the method, isoprocarb and carbaryl in positive samples could be evaluated in robusta coffee in the ranges of 0.001 mg kg<sup>-1</sup> to 0.239 mg kg<sup>-1</sup> and 0.001 mg kg<sup>-1</sup> to 0.249 mg kg<sup>-1</sup>, respectively, and in arabica coffee in the ranges of 0.001 mg kg<sup>-1</sup> to 0.022 mg kg<sup>-1</sup> and 0.001 mg kg<sup>-1</sup> to 0.020 mg kg<sup>-1</sup>, respectively. Based on the additional data related to the effect of coffee processing and chronic dietary exposure conducted using a deterministic approach, it can be concluded that the consumption of coffee contaminated with isoprocarb and carbaryl does not pose a risk to human health.

## **Author contributions**

Harmoko Harmoko: conceptualization, methodology, formal analysis, visualization, investigation, writing – original draft, and writing – review & editing. Hasim Munawar: writing – original draft and writing – review & editing. Syaiful Bahri: validation. Nuri Andarwulan: supervision. Daryono Hadi Tjahjono: supervision. Rahmana E. Kartasasmita: supervision and writing – review & editing. Amadeo R. Fernández-Alba: conceptualization, data presentation, supervision and writing – review & editing.

## Conflicts of interest

There are no conflicts to declare.

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