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Turn on of sky-blue thermally activated delayed fluorescence and circularly polarized luminescence (CPL) via increased torsion by a bulky carbazolophane donor†

The carbazolophane (Czp) donor unit (indolo[2.2]paracyclophane) is introduced to the design pool of donors in thermally activated delayed fluorescence emitters. The increased steric bulk of the annelated donor unit forces an increased torsion between the carbazole and the aryl bridge resulting in a decreased ΔE_{ST} and an enhancement of the thermally activated delayed fluorescence in the triazine-containing emitter CzpPhTrz. Further, the closely stacked carbazole and benzene units of the paracyclophane show through-space $\pi-\pi$ interactions, effectively increasing the spatial occupation for the HOMO orbital. The chiroptical properties of enantiomers of [2.2]paracyclophane reveal mirror image circular dichroism (CD) and circularly polarized luminescence (CPL) with g_{lum} of 1.3×10^{-3} . rac-CzpPhTrz is a sky-blue emitter with λ_{PL} of 480 nm, a very small ΔE_{ST} of 0.16 eV and high Φ_{PL} of 70% in 10 wt% doped DPEPO films. Sky blue-emitting OLEDs were fabricated with this new TADF emitter showing a high maximum EQE of 17% with CIE coordinates of (0.17, 0.25). A moderate EQE roll-off was also observed with an EQE of 12% at a display relevant luminance of 100 cd m^{-2} . Our results show that the Czp donor contributes to both a decreased ΔE_{ST} and an increased photoluminescence quantum yield, both advantageous in the molecular design of TADF emitters.

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Introduction

Organic Thermally Activated Delayed Fluorescence (TADF) materials have come to the fore as attractive alternative emitters in Organic Light-Emitting Diodes (OLEDs) as they, like current state-of-the-art phosphorescent emitters, can recruit 100% of the generated excitons into light. Since 2012, there are numerous examples in the literature of OLEDs using TADF emitters.

TADF in typical donor-acceptor (D-A) emitters is facilitated by a small singlet-triplet excited state energy difference, $\Delta E_{\rm ST}$, a consequence of the small exchange integral between spatially separated frontier molecular orbitals. The molecular design frequently relies on large torsions between the donor, where the highest occupied molecular orbital (HOMO) resides and the bridging arene and/or between the bridge and the acceptor, where the lowest unoccupied molecular orbital (LUMO) is located. It becomes more difficult to adhere to this design paradigm in blue TADF emitters due to the reliance on weak

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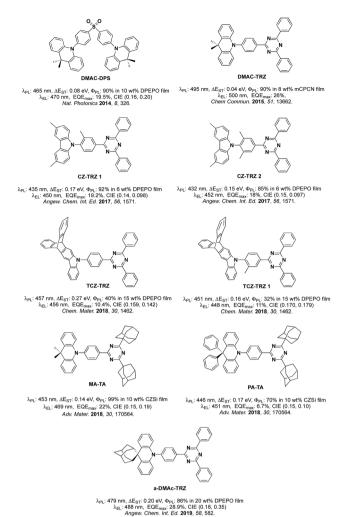


Fig. 1 Chemical structures and performance of known blue and triazine-based TADF emitters

carbazole (Cz) donors. The inherent size of Cz often results in much shallower torsions thereby leading to increased conjugation within the emitter and a larger ΔE_{ST} . Methylation of either Cz and/or the bridge has been explored as a strategy to combat the propensity for Cz to conjugate to the arene bridge.2 The state-of-the-art emitters based on the diphenyltriazine (cyaphenine) acceptor moiety in Fig. 1 illustrate these design principles. The methylation of bridging arene induced large torsions of between 71.2° and 82.3° in deep blue emitters Cz-Trz 1 and Cz-Trz 2, reported by Adachi et al. OLEDs using these emitters exhibited maximum external quantum efficiencies, EQE_{max} of 19.2 and 18.3% with CIE coordinates of (0.148, 0.09) and (0.15, 0.097) for Cz-TrZ 1 and Cz-Trz 2, respectively. Van Voorhis, Swager, Baldo, Buchwald et al.3 conducted a similar study using a triptycene extended carbazolyl donor unit (TCZ-TRZ), which showed a reduced overlap of the frontier orbitals than with the parent Cz donor-based emitter. Larger torsions were obtained through methyl substitution of the bridging arene (TCZ-TRZ 1). The photoluminescence quantum yield, $\Phi_{\rm PL}$, in these blue-emitting compounds is significantly decreased compared to those reported by Adachi et al.24 OLEDs showed

EQE_{max} values of 10.4 and 11% and CIE coordinates of (0.159, 0.142) and (0.170, 0.179) for TCZ-TRZ and TCZ-TRZ 1, respectively.

An alternative approach to functionalizing the weak Cz donor and arene bridge moieties of blue TADF compounds to control the balance of small ΔE_{ST} and high S_1 energy is to employ a stronger, bulkier donor such as dimethylacridine (DMAC).4 For example, employing DMAC in combination with a diphenylsulfone acceptor (DMAC-DPS, Fig. 1) resulted in skyblue emission with EQE_{max} and CIE coordinates of 19.5% and (0.16, 0.20).5 Upon changing the acceptor to diphenyltriazine in **DMAC-TRZ**, green emission ($\lambda_{EL} = 500 \text{ nm}$) was obtained with EQE_{max} as high as 26%.6 Incorporation of adamantyl groups onto the cyaphenine acceptor (MA-TA, Fig. 1) resulted in a significant blue-shift in the emission and OLEDs with this emitter showed high EQEmax of 22% and CIE coordinates of (0.15, 0.19). When the donor was modified to 9,9-diphenyl-9,10dihydroacridine (PA-TA), emission was further blue-shifted with CIE coordinates of (0.15, 0.10). However, the efficiency of the device suffered and the EQE_{max} was limited to 6.7%. Recently, an adamantyl-substituted acridine donor was introduced by Su et al. and coupled with a triazine acceptor. The rigid backbone of the donor in a-DMAc-TRZ suppressed nonradiative decay processes. Dual emission was observed from two conformers with an EQE_{max} of 28.9% and CIE of (0.18, 0.35).

Circularly polarized luminescence (CPL) enabled chiral molecules have been widely investigated for their potential applications in optical data storage⁸ and optical spintronics.⁹ In terms of display applications, such molecules are appealing especially in the development of circularly polarized OLEDs (CPOLEDs). At present, flat panel OLED-based display technologies require a film composed of a polarizer and a quarter wave plate to reduce the reflectance from the surroundings to a minimum in order to achieve high signal-to-noise contrast ratio. 10 Such a solution is however inefficient as the polarizer absorbs half of the emission from an OLED. Such losses can be mitigated if there is CPL. CPL-active small organic molecules (CPL-SOMs), mainly fluorescent compounds, have been investigated in this context; however, currently only a small number of CPL-SOMs are efficient both in terms of high photoluminescence quantum yield (Φ_{PL}) and luminescence dissymmetry factor (g_{lum}) . 11 Recently, a number of TADF-based CPL-SOMs have been reported (Fig. S1†). For example, Hirata et al. described a TADF CPL emitter, DPHN,12 by introducing a stereogenic carbon center between the donor and acceptor. This g_{lum} for this compound was 1.1 imes 10 $^{-3}$, but there was a very low Φ_{PL} of 4% in toluene. Pieters et al.13 enabled CPL emission in the TADF emitter R-1 through the incorporation of a BINOL unit that confers axial chirality to the molecule. The resulting compound exhibited $g_{\rm lum}$ of 1.3 imes 10⁻³, a high $\Phi_{\rm PL}$ of 53% in toluene and OLEDs displayed EQEmax of 9.1%. A chiral trans-1,2-diaminocyclohexane was used to link two imide-based TADF emitters in CAI-Cz. ¹⁴ Each enantiomer showed a $g_{\rm lum}$ value of 1.1 imes 10^{-3} and $\Phi_{\rm PL}$ as high as 98%. The resulting CPOLEDs exhibited excellent performances with an EQE_{max} of 19.8% with g_{EL} values of -1.7×10^{-3} and 2.3×10^{-3} for the two enantiomers. Tang et al. 15 used a similar design to that of Pieters et al. However, in

their design they combine in a single emitter BN-CF TADF, AIE (aggregation-induced emission) and CPL properties. BN-CF showed a $g_{\rm lum}$ of 1.2×10^{-2} , $\Phi_{\rm PL}$ of 32% in toluene, and the CPOLEDs exhibited an EQE_{max} of 9.3% and $g_{\rm EL}$ of 2.6×10^{-2} . Recently, Zhao $et~al.^{16}$ explored the inherent planar chirality of a [2.2]paracyclophane moiety. Emitter g-BNMe₂-C_p exhibited $g_{\rm lum}$ of 4.24×10^{-3} and $\Phi_{\rm PL}$ of 46% in toluene.

Results and discussion

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We recently demonstrated the first use of a [2.2]paracyclophane bridge within the design of TADF emitters where we exploited the π -stacking interactions of the two arene decks to moderate the electronic coupling of donor and acceptor groups.¹⁷ In this report, we demonstrate an effective design strategy to obtain both TADF and CPL in [2.2] paracyclophanes (PCP). We exploit the PCP to modulate the torsional angle of the donor with respect to the bridge and its inherent strength and thereby enhance the reverse intersystem crossing (RISC) efficiency in CzpPhTrz compared to the reference emitter CzPhTrz (Fig. 2). Further, the π - π stacking within the **PCP** enhances HOMO delocalization, thereby increasing the donor strength. Our donor design concept is based on an indole-annelated [2.2] paracyclophane, providing the carbazolophane (Czp, [2]paracyclo[2](1,4)carbazolophane). This compact skeleton features a closely coplanar stacked benzene unit on top of a carbazole bridged by two ethylene linkers that slightly bend the arenes from planarity. The short deck distance enables a throughspace π - π interaction thereby delocalizing the HOMO and perturbing the electronics and strengthening the donor. We further exploited the inherent planar chirality of the PCP scaffold to develop enantiopure TADF chromophores, (R_P) -CzpPhTrz and (S_P) -CzpPhTrz. We report the chiroptical properties of the enantiomers both in the ground and excited states with circular dichroism (CD) and CPL spectroscopy, respectively. The photophysical investigations of (rac)-CzpPhTrz revealed its excellent TADF properties with a small $\Delta E_{\rm ST}$ of 0.16 eV, a value smaller than those previously reported in Fig. 1, and a high photoluminescence quantum yield, Φ_{PL} , of 69% in 10 wt% **DPEPO** doped films. OLEDs fabricated using (rac)-CzpPhTrz showed sky blue emission with EQEmax as high as

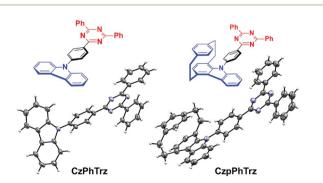


Fig. 2 Carbazol(ophan)yl triazine TADF emitters with molecular structures of CzPhTrz (left) and (rac)-CzpPhTrz (right) drawn at 50% probability level.

17% and CIE coordinates of (0.17, 0.25). The impact of this donor is clear when compared to the OLED using (rac)-CzPhTrz, which exhibited a blue-shifted emission with CIE coordinates of (0.14, 0.12) along with a much lower EQE_{max} of 5.8%.

Synthesis

The Czp donor was synthesized according to the protocol described by Bolm et *al.*¹⁸ (see ESI†). Emitters CzPhTrz and (*rac*)-CzpPhTrz were obtained in fair and good yields of 42% and 84%, respectively following nucleophilic aromatic substitution of the respective donors with fluorophenyltriazine. Both emitters were purified using a combination of column chromatography, recrystallization and gradient sublimation.§

The structures of both emitters were unequivocally determined by single crystal X-ray diffractometry (Fig. 2). The presence of the cyclophane in (*rac*)-CzpPhTrz induces a torsion of 55.86° between the planes of the Czp donor and the phenyl bridge while the corresponding torsion in CzPhTrz is significantly shallower at 45.11°. The distance between decks in the PCP core is 3.07 Å at the N-connected carbon atom, which is consistent with distances in both the free carbazolophane¹⁸ (3.06 Å) and nonsubstituted [2.2]paracyclophanes (3.08–3.10 Å).¹⁹

Thermal gravimetric analysis (TGA) and differential scanning calorimetry (DSC) were employed to assess the thermal stability of (rac)-CzpPhTrz (Fig. S8†). (rac)-CzpPhTrz exhibited excellent thermal stability with a decomposition temperature, $T_{\rm d}$, corresponding to a 5% weight loss, of 360 °C, which is a little higher than 330 °C measured for CzPhTrz. CzPhTrz and (rac)-CzpPhTrz exhibited melting temperatures, $T_{\rm m}$ of 270 °C and 263 °C, respectively, ensuring the formation of stable films of both emitters by vacuum deposition. The presence of a PCP structural unit introduces planar chirality to the emitter. Enantiomeric emitters ($R_{\rm P}$)-CzpPhTrz and ($S_{\rm P}$)-CzpPhTrz were also obtained. These were synthesized from the chiral Czp intermediates, which

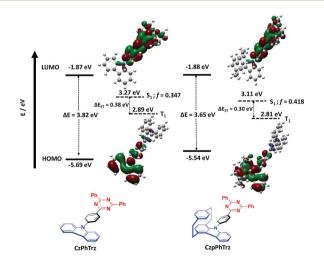


Fig. 3 DFT [PBE0/6-31G(d,p)] calculated ground and TDA-based excited state energies and electron density distributions of the HOMOs and LUMOs of CzPhTrz (left) and CzpPhTrz (right), f is the oscillator strength.

Table 1 Photophysical and electrochemical properties of CzPhTrz and CzpPhTrz

	$\lambda_{\mathrm{abs}}{}^{a}, \\ [\mathrm{nm}]$	$\lambda_{\mathrm{PL}}{}^{a}, \ [\mathrm{nm}]$	$\Phi_{\mathrm{PL}}{}^{b}, \ [\%]$	$\lambda_{\mathrm{PL}}^{}c},$ [nm]	$\Phi_{\mathrm{PL}}{}^{c}, \ [\%]$	$\tau_{\mathrm{p}}^{}c},$ [ns]	τ _d ^c , [μs]	HOMO ^{d,e} , [eV]	LUMO ^{d,e} , [eV]	$\Delta E_{ m redox}^f$, [eV]	S_1/T_1^g , [eV]	$\Delta E_{\rm ST}^{g}$, [eV]
CzPhTrz CzpPhTrz	363 375	446 470	60 (58) 70 (55)	448 482	50 69	4.8 7	— 65	-6.11 -5.80	-3.26 -3.23	2.86 2.58	3.14/2.82 2.89/2.73	0.32 0.16

^a In PhMe at 298 K. ^b Quinine sulfate (0.5 m) in H₂SO₄ (aq) was used as the reference ($Φ_{PL}$: 54.6%, $λ_{exc}$ = 360 nm). ¹⁵ Values quoted are in degassed solutions, which were prepared by three freeze-pump-thaw cycles. Values in parentheses are for aerated solutions, which were prepared by bubbling air for 10 min. ^c Thin films were prepared by vacuum depositing 10 wt% doped samples in **DPEPO** and values were determined using an integrating sphere ($λ_{exc}$ = 360 nm); degassing was done by N₂ purge for 10 minutes. ^d In DCM with 0.1 m [nBu₄N]PF₆ as the supporting electrolyte and Fc/Fc⁺ as the internal reference (0.46 V vs. SCE). ^{16 e} The HOMO and LUMO energies were determined using $E_{HOMO/LUMO}$ = $-(E_{pa,1}^{ox}/E_{pc,1}^{red} + 4.8)$ eV (ref. 17) where E_{pa}^{ox} and E_{pc}^{red} are anodic and cathodic peak potentials, respectively. ^f ΔE_{redox} = $|E_{HOMO} - E_{LUMO}|$. ^g Determined from the onset of prompt and delayed spectra of 10 wt% doped films in **DPEPO**, measured at 77 K ($λ_{exc}$ = 355 nm).

were separated by chiral HPLC. The absolute configuration could be unambiguously assigned by CD (see ESI† Section 5).

Theoretical properties

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Density Functional Theory (DFT) calculations were conducted on both CzPhTrz and CzpPhTrz (Fig. 3) in order to gain insight into the impact of the use of the Czp donor on the optoelectronic properties of the emitters. In both emitters, the HOMO is localized on the carbazole-based donor with some contribution extending onto the phenyl bridge while the LUMO is distributed over both the phenyl bridge and the triazine acceptor. For CzpPhTrz, the HOMO is extended significantly onto the PCP core owing to the strong π - π through-space interactions within the Czp donor. CzPhTrz has been previously studied where it has been reported20 to possess a rod-like structure coupled with a $\Delta E_{\rm ST}$ of 0.36 eV and a calculated dihedral angle between the Cz and the phenyl bridge of 51.8°. The large $\Delta E_{\rm ST}$ for this molecule leads to inefficient TADF. Lee et al.20 further showed CzPhTrz to be isotropic in the **DPEPO** film and so without the horizontal orientation desired to improve light outcoupling. OLEDs fabricated with this emitter ($\lambda_{EL} = 449 \text{ nm}$) showed an EQE_{max} of only 4.2% (EQE₁₀₀₀ = 0.6%) despite a photoluminescence quantum yield, Φ_{PL} , of 71%. We similarly calculated a small twist angle of 51° between the donor and phenyl bridge, in a good agreement with the crystal structure analysis (45.11°, Fig. 2), and a large $\Delta E_{\rm ST}$ of 0.38 eV in CzPhTrz (Fig. 3). Introduction of the rigid and sterically bulky PCP onto the donor in CzpPhTrz not only induced a larger torsion between the donor and the bridge but also extended the conjugation length. Thus, Czp is a stronger donor, the impact of which is a lower S_1 energy, a smaller $\Delta E_{\rm ST}$ and a more efficient TADF process in CzpPhTrz. Indeed, DFT calculations predict an increased torsion of 58° (55.86° in the crystal structure), a smaller $\Delta E_{\rm ST}$ of 0.30 eV and a lower S_1 energy of 3.11 eV compared to CzPhTrz $[E(S_1) = 3.27 \text{ eV}]$. The S_1 energy is nevertheless predicted to be sufficiently high such that CzpPhTrz should remain a good blue TADF emitter candidate.

Electrochemical properties

The HOMO and LUMO levels of CzPhTrz and CzpPhTrz were ascertained by cyclic voltammetry (CV) and differential pulse voltammetry (DPV) measurements, respectively, in

dichloromethane. The CVs and DPVs are shown in Fig. S7† and data are summarized in Table 1. Both emitters possess similar reduction potentials at -1.78 V, which were assigned to the reduction of the cyaphenine acceptor. However, and in line with the theoretical studies, the oxidation potential of CzpPhTrz (1.14 V) is significantly cathodically shifted to that of CzPhTrz (1.43 V). Thus, the introduction of the PCP moiety appreciably reduces the ionization potential (IP) of the donor and thus the electrochemical gap. Expectedly, the oxidation waves of both compounds were found to be irreversible as the carbazole radical cation is known to be electrochemically unstable and subsequently can undergo dimerization.²¹

Photophysical properties

Fig. 4A shows the UV-Vis absorption and photoluminescence (PL) spectra of CzPhTrz and CzpPhTrz in toluene and the data

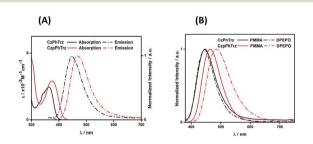


Fig. 4 (A) UV-vis absorption and PL spectra in PhMe (B) PL spectra of thin films of CzPhTrz and CzpPhTrz ($\lambda_{\rm exc}=360$ nm).

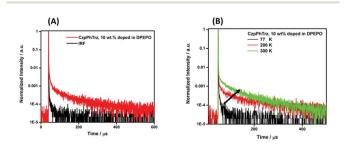


Fig. 5 (A) Time-resolved PL decay curve and (B) temperature dependence of time-resolved PL lifetime of CzpPhTrz in 10 wt% DPEPO film ($\lambda_{\rm exc}=378$ nm), IRF = instrument response function.

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are summarized in Table 1. Both compounds possess an intramolecular charge transfer (ICT) absorption band from the carbazole-based donor to the triazine acceptor at 363 and 375 nm, respectively, for CzPhTrz and CzpPhTrz. The more intense and red-shifted band in CzpPhTrz agrees with the calculations shown in Fig. 3.

Both compounds exhibited unstructured PL spectra in PhMe, indicative of an excited state with strong intramolecular charge transfer (ICT) character. Analogous to their relative behavior in the ground state, the photoluminescence maximum, λ_{PL} , of CzpPhTrz at 470 nm is red-shifted compared to that of CzPhTrz ($\lambda_{PL}=446$ nm). The Φ_{PL} values in PhMe for CzPhTrz and CzpPhTrz are 60% and 70%, which decreased to 58% and 55%, respectively, upon exposure to air. The enhanced $\Phi_{\rm PL}$ of CzpPhTrz is consistent with the higher predicted oscillator strength. The time-resolved PL spectra in PhMe for both emitters (Fig. S9†), however, do not show any delayed emission,

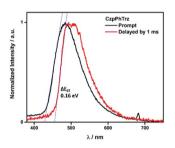


Fig. 6 Prompt and delayed spectra (at 77 K) of CzpPhTrz in 10 wt% **DPEPO** film ($\lambda_{\text{exc}} = 355 \text{ nm}$).

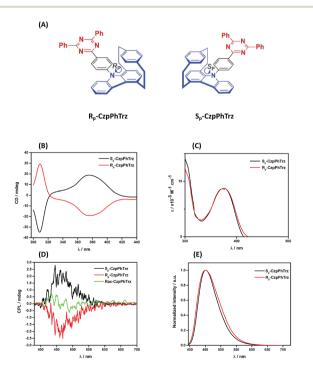


Fig. 7 (A) Chemical structures (B) CD (C) UV-vis (D) CPL and (E) PL spectra of (R_P)-CzpPhTrz and (S_P)-CzpPhTrz in degassed PhMe (λ_{exc} = 375 nm).

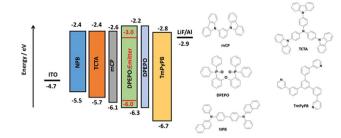


Fig. 8 Chemical structures and energy levels of materials used for the device fabrication

which would suggest that the TADF mechanism in solution is very weak despite the oxygen sensitivity noted for Φ_{PI} .

We next investigated the PL behavior in the solid-state as doped thin films (Fig. 4B). In both PMMA and **DPEPO** matrices, both compounds showed unstructured ICT-based emission. In 10 wt% doped solution-processed PMMA films, CzPhTrz shows deep blue emission with λ_{PL} of 446 nm; the emission energy is essentially the same in vacuum-deposited 10 wt% doped **DPEPO** films at 448 nm. The emission of CzpPhTrz is expectedly red-shifted at 464 nm in PMMA, and this emission is further red-shifted to 482 nm in the more polar DPEPO. The $\Phi_{\rm PL}$ values of the PMMA films of CzPhTrz and CzpPhTrz are 25% and 45%, respectively. Further, both compounds exhibited a mono exponential decay with PL lifetimes, τ of 5 ns for CzPhTrz and 8 ns for CzpPhTrz, suggesting that TADF mechanism is absent in a non-polar host like PMMA, results in line with those observed in PhMe. The much enhanced $\Phi_{\rm PL}$ of CzpPhTrz is 69% in DPEPO, which is significantly higher than that of CzPhTrz ($\Phi_{PL} = 50\%$). Reducing the doping concentration to 1 wt% resulted in a lower Φ_{PL} of 43% for CzpPhTrz (Table S2†). Time-resolved PL measurements of the 10 wt% DPEPO films revealed divergent behavior between the two

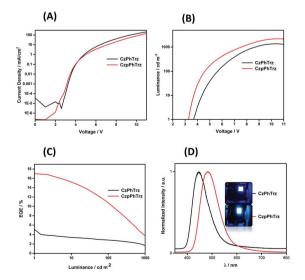


Fig. 9 (A) Current density-voltage characteristics (B) luminance vs. voltage. (C) EQE vs. luminance (D) normalized EL spectra of CzPhTrz and CzpPhTrz.

Table 2 Electroluminescence properties of CzPhTrz and CzpPhTrz

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	V_{on}^{a} , [V]	$\lambda_{\mathrm{EL}}{}^{b}$, [nm]	EQE_{max}^{c} , EQE_{100}^{d} , [%]	CE_{\max}^{c} , [cd A ⁻¹]	$PE_{max}^{ c}$, $[lm W^{-1}]$	CIE^e , (x, y)
CzPhTrz	3.6	446	5.8; 2.8	10.5	8.6	(0.14, 0.12)
CzpPhTrz	3.2	480	17.0; 12.0	34.8	32.5	(0.17, 0.25)

 $[^]a$ Measured at 1 cd m $^{-2}$. b Emission maxima at 1 mA cm $^{-2}$. c Maximum efficiencies at 1 cd m $^{-2}$. d EQE at 100 cd m $^{-2}$. e Commission Internationale de l'Eclairage coordinates at 1 mA cm $^{-2}$.

emitters. Only a nanosecond ($\tau=4.8$ ns) monoexponential fluorescence decay was observed for CzPhTrz (Fig. S10†), results that agree with the literature. For CzpPhTrz, both prompt ($\tau_p=7$ ns) and delayed ($\tau_d=65~\mu s$) emission was observed (Fig. 5A). We next studied the temperature dependence of the delayed lifetime of CzpPhTrz (Fig. 5B). An increase in the intensity of the delayed emission as a function of increasing temperature provides corroborating evidence to support the TADF character of CzpPhTrz.

The $\Delta E_{\rm ST}$ of CzPhTrz and CzpPhTrz were determined from the singlet and triplet energies estimated from the onset of the prompt and delayed emission spectra, respectively, measured in 10 wt% DPEPO doped films at 77 K (Fig. 6). CzpPhTrz possesses a high S_1 energy of 2.90 eV $[E(S_1) = 2.87 \text{ eV}]$ at room temperature] coupled with a small $\Delta E_{\rm ST}$ value of 0.16 eV, confirming its strong potential as a blue TADF emitter. The turn-on of the TADF in DPEPO is due to preferential stabilization of the 1 CT state and the associated decrease in $\Delta E_{\rm ST}$. On the other hand, a rather large $\Delta E_{\rm ST}$ of 0.32 eV (Fig. S11†) was determined for CzPhTrz, implying a much weaker TADF character. These results clearly demonstrate that the introduction of the bulky PCP moiety with the carbazole donor to form the carbazolophane (Czp) not only turns on the TADF channel in the emission mechanism in polar media, but the high Φ_{PL} is maintained.

Chiroptical properties

The ground and excited state chiroptical properties of (R, S)-CzpPhTrz (Fig. 7a) were studied by CD and CPL spectroscopy, respectively in toluene. As shown in Fig. 7B, there is the expected mirror image spectra corresponding to each of the two enantiomers. Using the molar extinction coefficients $\varepsilon(R)$ and $\varepsilon(S)$ (Fig. 7C) of the CT band at 375 nm, the CD dissymmetry factor was estimated to be $+5.0 \times 10^{-3}$ and -7.0×10^{-3} for the $R_{\rm P}$ and $S_{\rm P}$ enantiomers, respectively. Similarly, mirror image CPL spectra (Fig. 7D) were obtained with $g_{\rm lum}$ values for both $R_{\rm P}$ and $S_{\rm P}$ enantiomers at their emission maxima at 1.2×10^{-3} and 1.3×10^{-3} . These are typical values for CPL-SOMs.

Device fabrication

Based on the promising photophysical properties of $\it rac$ -CzpPhTrz in DPEPO films, we fabricated vacuum-deposited multilayer OLEDs. A deep HOMO level of -6.06 eV for $\it rac$ -CzpPhTrz was measured by UV ambient pressure photoemission spectroscopy (APS), which is deeper than that inferred from cyclic voltammetry measurements ($\it E_{HOMO} = 5.80$ eV,

Table 1) by 0.26 eV, due in part to the different measurement environments (neat film for APS and DCM for DPV). Therefore, to confine the excitons in the emissive layer, DPEPO, with a HOMO energy of -6.3 eV was chosen as a host material in the emissive layer.22 Optimum host and doping concentration were assessed by absolute $\Phi_{\rm PL}$ measurements (Table S2†). An energy level diagram of the OLED stack and chemical structures of the materials used are shown in Fig. 8. The OLED architecture employed consisted of: ITO/NPB (30 nm)/TCTA (20 nm)/mCP (10 nm)/rac-CzpPhTrz : DPEPO (10 wt%, 20 nm)/rac-CzpPhTrz : DPEPO (10 mm)/rac-CzpPhTrz : DPEPO (10 mm)/racDPEPO (10 nm)/TmPyPB (40 nm)/LiF (1 nm)/Al (100 nm), where N,N'-bis(naphthalen-1-yl)-N,N'-bis(phenyl)benzidine (NPB) was used as a hole injection layer (HIL), tris(4-carbazoyl-9-ylphenyl)amine (TCTA) was used as a hole transporting layer (HTL), mCP (1,3-bis(N-carbazolyl)benzene) and DPEPO were used as an electron and hole blockers, respectively. The electron transporting layer (ETL) was comprised of 1,3,5-tris(3pyridyl-3-phenyl)benzene (TmPyPB), which possesses a high electron mobility of 10^{-4} cm² V⁻¹ s⁻¹ and a high triplet energy of 2.75 eV along with a deep HOMO energy of 6.7 eV.23 We also fabricated the reference OLED using CzPhTrz as the emitter using the same device architecture.

The electroluminescence (EL) properties are shown in Fig. 9 and the data are summarized in Table 2. OLEDs based on CzPhTrz exhibited a deep blue emission with a $\lambda_{\rm EL}$ of 446 nm and CIE coordinates of (0.16, 0.12) whereas devices employing *rac*-CzpPhTrz showed a slightly red-shifted emission with a $\lambda_{\rm EL}$ of 480 nm and CIE coordinates of (0.17, 0.25). The EL spectra matched very well to the corresponding PL spectra. Both devices exhibited steep current-voltage-luminance behavior (Fig. 9A and B) and low turn-on voltages ($V_{\rm on}$) of 3.6 V and 3.2 V for the OLEDs based on CzPhTrz and *rac*-CzpPhTrz, respectively. The lower $V_{\rm on}$ for the OLED with *rac*-CzpPhTrz is attributed to the lower energy gap of this emitter.

Fig. 9C shows the EQE *versus* luminance of the two OLEDs. The device with rac-CzpPhTrz exhibited a high EQE_{max} of 17% at low brightness (at 1 cd m⁻²), indicating an efficient triplet harvesting in the OLED mediated by the TADF channel. Further the device showed a moderate efficiency roll-off to 12% at a display relevant brightness of 100 cd m⁻². Assuming the recombination efficiency to be unity and taking the Φ_{PL} of 70%, the device with rac-CzpPhTrz possessed an outcoupling efficiency of 25%. On the other hand, devices based on CzPhTrz showed an EQE_{max} of 5.8% at low brightness, which reduced to 2.8% at 100 cd m⁻². The significantly poorer performance is due in part to the larger ΔE_{ST} of the emitter, which renders TADF very inefficient in the device. To summarize, we observed a three-fold enhancement in the

EQE_{max} as a function of the use of the Czp donor in *rac*-CzpPhTrz. To the best of our knowledge, the OLED with *rac*-CzpPhTrz is the first example of an EL device incorporating a [2.2] paracyclophane moiety. *rac*-CzpPhTrz indeed is amongst the state-of-the art blue to sky-blue TADF triazine-based emitters in terms of EQE_{max} though at a cost of a red-shifted EL spectrum compared to OLEDs based on emitters shown in Fig. 1.

Conclusion

Edge Article

In conclusion, we have demonstrated how TADF and CPL can be turned on in triazine emitters through incorporation of a [2.2] paracyclophane moiety within the carbazole donor structure. The resulting carbazolophane (Czp) donor unit both provides a through-space π - π interaction and an increased torsion as a function of the increased inherent bulkiness of the donor. This enhances the donor strength and forces a larger torsion between the donor and the phenyl bridge, leading to a much more efficient TADF process. Moreover, the enantiomers possessed mirror image CD and CPL spectra with g_{lum} in line with other CPL-SOMs. The resulting sky-blue OLED showed a three-fold enhancement in performance compared to the reference device with EQEmax of 17%. These results demonstrate clearly the value of the Czp donor, which we believe can be applied to a whole range of TADF emitters to enhance OLED and CPL performance.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

§ Synthesis, NMR spectra, elemental analysis data, TGA and DSC, cyclic voltammetry data, supplementary photophysical measurements, computational data, electroluminescence data and crystallographic data (CCDC-1871934 and CCDC-1871935). During the proofs stage it has been pointed out to us that the structure and synthesis of CzpPhTRZ has appeared in a patent: Buchwald, S. L.; Huang, W., "[2.2]Paracyclophane-Derived Donor/Acceptor-Type Molecules for OLED Applications", Massachusetts Institute of Technology, International Patent application number PCT/US2016/035655, published 8 December 2016.

- (a) Z. Yang, Z. Mao, Z. Xie, Y. Zhang, S. Liu, J. Zhao, J. Xu,
 Z. Chi and M. P. Aldred, *Chem. Soc. Rev.*, 2017, 46, 915–1016; (b) M. Y. Wong and E. Zysman-Colman, *Adv. Mater.*,
 2017, 29, 1605444; (c) Y. Liu, C. Li, Z. Ren, S. Yan and
 M. R. Bryce, *Nat. Rev. Mater.*, 2018, 3, 18020.
- (a) L. S. Cui, H. Nomura, Y. Geng, J. U. Kim, H. Nakanotani and C. Adachi, *Angew. Chem., Int. Ed. Engl.*, 2017, 56, 1571–1575;
 (b) Y.-C. Duan, L.-L. Wen, Y. Gao, Y. Wu, L. Zhao, Y. Geng, G.-G. Shan, M. Zhang and Z.-M. Su, *J. Phys. Chem. C*, 2018, 122, 23091–23101.
- 3 W. Huang, M. Einzinger, T. Zhu, H. S. Chae, S. Jeon, S.-G. Ihn, M. Sim, S. Kim, M. Su, G. Teverovskiy, T. Wu, T. Van Voorhis, T. M. Swager, M. A. Baldo and S. L. Buchwald, *Chem. Mater.*, 2018, **30**, 1462–1466.
- 4 Y. Wada, S. Kubo and H. Kaji, Adv. Mater., 2018, 30, 1705641.
- 5 Q. Zhang, B. Li, S. Huang, H. Nomura, H. Tanaka and C. Adachi, *Nat. Photonics*, 2014, **8**, 326–332.
- 6 W.-L. Tsai, M.-H. Huang, W.-K. Lee, Y.-J. Hsu, K.-C. Pan, Y.-H. Huang, H.-C. Ting, M. Sarma, Y.-Y. Ho, H.-C. Hu, C.-C. Chen, M.-T. Lee, K.-T. Wong and C.-C. Wu, *Chem. Commun.*, 2015, 51, 13662–13665.
- 7 W. Li, X. Cai, B. Li, L. Gan, Y. He, K. Liu, D. Chen, Y.-C. Wu and S.-J. Su, *Angew. Chem., Int. Ed.*, 2019, **58**, 582–586.
- 8 C. Wang, H. Fei, Y. Qiu, Y. Yang, Z. Wei, Y. Tian, Y. Chen and Y. Zhao, *Appl. Phys. Lett.*, 1999, 74, 19–21.
- 9 T. Harada, H. Hayakawa, M. Watanabe and M. Takamoto, *Rev. Sci. Instrum.*, 2016, **87**, 075102.
- 10 B. C. Kim, Y. J. Lim, J. H. Song, J. H. Lee, K. U. Jeong, J. H. Lee, G. D. Lee and S. H. Lee, *Opt. Express*, 2014, 22(Suppl. 7), A1725–A1730.
- E. M. Sanchez-Carnerero, A. R. Agarrabeitia, F. Moreno,
 B. L. Maroto, G. Muller, M. J. Ortiz and S. de la Moya,
 Chemistry, 2015, 21, 13488–13500.
- 12 T. Imagawa, S. Hirata, K. Totani, T. Watanabe and M. Vacha, *Chem. Commun.*, 2015, **51**, 13268–13271.
- 13 S. Feuillastre, M. Pauton, L. Gao, A. Desmarchelier, A. J. Riives, D. Prim, D. Tondelier, B. Geffroy, G. Muller, G. Clavier and G. Pieters, *J. Am. Chem. Soc.*, 2016, 138, 3990–3993.
- 14 M. Li, S. H. Li, D. Zhang, M. Cai, L. Duan, M. K. Fung and C. F. Chen, *Angew. Chem., Int. Ed. Engl.*, 2018, 57, 2889–2893.
- 15 F. Song, Z. Xu, Q. Zhang, Z. Zhao, H. Zhang, W. Zhao, Z. Qiu, C. Qi, H. Zhang, H. H. Y. Sung, I. D. Williams, J. W. Y. Lam, Z. Zhao, A. Qin, D. Ma and B. Z. Tang, Adv. Funct. Mater., 2018, 28, 1800051.
- 16 M. Y. Zhang, Z. Y. Li, B. Lu, Y. Wang, Y. D. Ma and C. H. Zhao, *Org. Lett.*, 2018, **20**, 6868–6871.
- 17 E. Spuling, N. Sharma, I. D. W. Samuel, E. Zysman-Colman and S. Bräse, *Chem. Commun.*, 2018, **54**, 9278–9281.
- 18 P. Lennartz, G. Raabe and C. Bolm, *Isr. J. Chem.*, 2012, **52**, 171–179.
- 19 H. Wolf, D. Leusser, M. R. V. Jørgensen, R. Herbst-Irmer, Y.-S. Chen, E.-W. Scheidt, W. Scherer, B. B. Iversen and D. Stalke, *Chem.-Eur. J.*, 2014, 20, 7048–7053.
- 20 S. Y. Byeon, J. Kim, D. R. Lee, S. H. Han, S. R. Forrest and J. Y. Lee, *Adv. Opt. Mater.*, 2018, 6, 1701340.

21 (a) A. Tomkeviciene, J. V. Grazulevicius, D. Volyniuk, V. Jankauskas and G. Sini, *Phys. Chem. Chem. Phys.*, 2014,
16, 13932–13942; (b) T. Fukushima, J. Yamamoto, M. Fukuchi, S. Hirata, H. H. Jung, O. Hirata, Y. Shibano, C. Adachi and H. Kaji, *AIP Adv.*, 2015, 5, 087124.

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- 22 C. Han, Y. Zhao, H. Xu, J. Chen, Z. Deng, D. Ma, Q. Li and P. Yan, *Chem.-Eur. J.*, 2011, **17**, 5800–5803.
- 23 S.-J. Su, T. Chiba, T. Takeda and J. Kido, *Adv. Mater.*, 2008, **20**, 2125–2130.