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Waterborne nanoplastics and microplastics: analytical advances, modelling, and future directions

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Plastics' persistence throughout their life cycle has imposed a global burden of nano- and microplastics in aquatic systems. This frontier review consolidates recent advances in analytics, machine learning, and fate and transport modelling, and sets a practical agenda for decision-ready measurements. Analytical breakthroughs enable chemically specific imaging at tens-of-nanometre resolution using sub-micron vibrational methods, while hyperspectral stimulated Raman scattering delivers rapid single-particle chemical mapping. Emerging holography techniques provide *in situ*, real-time physicochemical characterization, capturing 3D size, shape, surface coatings, and other features and can distinguish nano- and microplastics from other particles within milliseconds. Complementary innovations, including label-free photonic and electrochemical sensors and separation workflows coupled with mass spectrometry, extend polymer specificity and quantification in complex waters. Across these platforms, machine learning accelerates denoising, feature extraction, automated classification, and imaging throughput. Yet, performance remains constrained by biased or limited datasets, label noise, and domain shifts across instruments, matrices, and weathering states. Modelling frontiers require adaptation for nanoplastics, where Brownian diffusion, rapid aggregation, and dynamic eco-coronas govern transport and water-sediment exchange. Existing nanomaterial models offer transferable scaffolds when re-parameterized for nanoplastic behaviour. To enable interoperable, validated, and scalable systems, we recommend: (a) a universal checklist for reporting nano- and microplastic analytics aligned with International Organization for Standardization (ISO) guidance for diverse media; (b) standard reference materials and blinded interlaboratory trials; (c) open, versioned datasets; (d) machine learning tasks with fixed splits and uncertainty reporting; (e) routine, end-to-end uncertainty quantification and traceability; (f) field testbeds integrating sensors, analytics, and fate and transport models to deliver policy-relevant indicators with calibrated confidence. Together, these steps will transform fragmented efforts into robust, decision-ready frameworks for safeguarding water quality in the age of nanoplastics.

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Environmental significance

Nanoplastics are an emerging contaminant of global concern, yet their detection in water remains hindered by their nanoscale dimensions, diverse chemistries, and interactions with natural matrices. While microplastics have been more extensively studied, this review places particular emphasis on nanoplastics, whose environmental monitoring remains scarce. We examine recent advances in analytical methods, spanning spectroscopy, mass spectrometry, laser-based imaging, holographic microscopy, and microfluidic platforms, and highlight the growing use of artificial intelligence for automated detection and classification. We further explore how integrating analytical approaches and field measurements with modelling can enhance predictive capacity and feedback into method development. Together, these advances enable more accurate exposure assessments, strengthen risk evaluation, and provide the foundation for effective policy responses to plastic pollution.

1. Introduction

Plastics originated in 1862 when Alexandre Parkes patented the first artificial plastic, derived from cellulose, a natural polymer.¹ Within a few years, numerous scientists and inventors became interested in this material, sparking a revolution that transformed everyday life, from food packaging and medicine to electronics. Plastics enabled the

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mass production of lightweight, durable products across diverse sectors. While plastics have undeniably reshaped modern living, their persistence throughout production, use, and disposal has created a long-lasting pollution burden that now affects nearly every environmental compartment.

Every year, an estimated 19 to 23 million tonnes of plastic waste leak into aquatic ecosystems, polluting lakes, rivers, and seas.² Global plastic production and waste are projected to increase through mid-century in the absence of stronger policy action, with leakage to aquatic systems expected to continue even under ambitious mitigation scenarios.^{3,4} These trends align with ongoing negotiations on a legally binding treaty to end plastic pollution, highlighting the need for actionable science to guide mitigation across the entire life cycle of plastics.⁵

While microplastics have received considerable attention, research is increasingly shifting toward smaller and analytically more challenging nanoplastics. Owing to their substantially higher surface area, distinct colloidal behaviour, and ability to traverse biological barriers more readily than larger particles, nanoplastics necessitate specialized analytical and modelling approaches. Recent standardization efforts (*e.g.*, ISO 24187:2023) have begun harmonizing microplastic terminology, facilitating cross-study comparisons.^{6–8} Consequently, most current studies adopt an operational definition of nanoplastics as particles $<1\ \mu\text{m}$, consistent with instrumental detection limits and colloidal properties.^{6,9,10}

Analytical advances over the past few years are reshaping what can be measured in laboratory environmental matrices. Submicron techniques such as optical photothermal infrared (O-PTIR) and atomic force microscopy-infrared spectroscopy (AFM-IR) have extended chemically specific imaging into the hundreds of nanometres regime, and in some cases down to tens of nanometres.^{11,12} Field-relevant demonstrations show that coupled O-PTIR and AFM-IR can detect and map nanoplastics within snow and water samples.^{13,14} In parallel, hyperspectral stimulated Raman scattering (SRS) spectroscopy has emerged as a rapid single-particle platform for identifying nano- and microplastics. Analyses of bottled water have measured $\sim 10^5$ plastics per L, $\sim 90\%$ of which are nanoplastics, markedly elevating prior intake estimates.¹⁵ Table 1 summarizes representative analytical techniques commonly used for measuring waterborne nano- and microplastics in environmental samples.

The fast pace of artificial intelligence development has accelerated the integration of machine learning into both analytical detection and data interpretation in nano- and microplastic research. Across IR and Raman measurements, machine learning has improved denoising, baseline correction, feature extraction, and automated classification of polymer types, especially under low signal-to-noise or overlapping signatures.^{34,35} Recent systematic reviews highlight rapid growth in machine learning assisted analytics for nano- and microplastics in diverse water matrices, while also flagging typical failure modes such as small or biased

training sets, label noise, and domain shift across instruments and matrices.^{34–36} Beyond spectroscopy, machine learning enabled imaging and signal processing are beginning to support real-time, *in situ* detection and quality assured quantification, pointing toward monitoring systems that connect sensors, algorithms, and models.²⁰

Nano- and microplastic measurement is only the beginning. Numerical models connect measurement to prediction, resolving the transport and fate of nano- and microplastic in natural waters. In aquatic systems, the fate and transport of these particles are controlled by multiscale processes such as advection–dispersion in bulk flow, Brownian-driven microscale collisions, and rapid physicochemical transformations including aging, eco-corona formation, and hetero-aggregation with natural colloids. These transformations dynamically alter effective size, density, and surface charge, shifting particle transport regimes, attachment efficiencies, and water–sediment exchange, thereby complicating modelling efforts.^{37–39} For instance, eco-coronas assembled from natural organic matter, proteins, and polysaccharides can stabilize or destabilize suspensions and alter biological interactions.⁴⁰ Fig. 1 provides an overview of the transformation and transport of nano- and microplastics in aquatic systems.

In this frontier review, we first survey recent analytical advances in detecting nanoplastic particles (section 2), assessing detection limits, size coverage, polymer specificity, and matrix tolerance. Next, we examine how machine learning is transforming spectroscopy, imaging, and signal processing in nano- and microplastic analysis (section 3), highlighting improvements in throughput and automation alongside persistent failure problems. We then assess microplastic fate and transport models in aquatic systems and discuss how existing model efforts can be adapted for nanoplastic particles (section 4). Finally, we outline future directions (section 5) aimed at developing robust, operational methods that translate particle-resolved observations into actionable guidance for monitoring, mitigation, and regulation.

2. Advances in techniques for physicochemical characterization and quantification of nanoplastics in environmental waters

Methods for detecting waterborne nanoplastics are rapidly diversifying spanning electrochemical and electroluminescent detectors, spectroscopies with active preconcentration, label-free photonic sensors, and separation techniques coupled to mass spectrometry. Together, these approaches pursue complementary goals: polymer-specific identification, number- or mass-based quantification, and reliable operation in complex matrices. Below, we review selected recent studies organized into four themes, highlighting each method's contributions, points of convergence and divergence, and



Table 1 A brief overview of commonly used technologies for nano- and microplastic detection in environmental waters. Selected mass-based and particle-based techniques for identification and quantification are tabulated, and their typical applications, key advantages, and main limitations are summarized for comparison

| Technologies | Major applications | Key strengths | Main limitations | Examples | Ref. |
|-----------------------|--|---|--|---|--------|
| Mass spectrometry | <i>Py-GC-MS</i> • Polymer identification • Additives and oligomers • Mass-based quantification | • High chemical specificity (GC separation) • Complex matrices (with cleanup) • Relatively mature, widely adopted | • Destructive • Preparation and analysis can be slow • Marker, library, and matrix effects can bias | Identifying and quantifying PE, PP, PS, PET, PVC, and PMMA in surface water and riverside groundwater | 16, 17 |
| | <i>TD-PTR-MS</i> • Rapid screening of thermal desorption markers • High throughput comparisons • Process or aging proxies | • Relatively fast • Sensitive to VOC/SVOC markers • Time-resolved signal during heating | • Lower specificity (overlapping markers/isobars) • Quantification is harder (calibration/transfer efficiency) • Destructive • Limited for low-volatility fractions | Identifying and quantifying PE, PP, PET, and PVC in lakes and streams | 18, 19 |
| | <i>MALDI-TOF-MS</i> • Oligomer or repeat unit patterns • Polymer-dependent MW distribution | • Fast spectra • Insight into MW and oligomer series • Not inherently limited by particle size | • Typically low throughput • Preparation sensitive (matrix and cation) • Molecular weight range constraints • Polymer-dependent ionization bias | Detecting PE and PTHF in lake and river waters | 20, 21 |
| Infrared spectroscopy | <i>ATR-FTIR</i> • Identification of larger plastic particles/fragments • QA/QC checks for reference materials | • Simple, widely available • Relatively easy sample preparation • Polymer specificity for many common plastics • Easy to combine with complementary techniques | • Low throughput, one-at-a-time • Particle size limit • Surface contamination (<i>e.g.</i> , biofilms) can interfere | Analyzing small microplastics in marine water samples | 22, 23 |
| | <i>μ-FTIR (micro-FTIR imaging/microspectroscopy)</i> • Automated mapping and identification of particles on filters • Size distribution and counts (down to ~10–20 μm typical) | • Higher throughput than conventional ATR • Produces chemical maps and particle metrics • Standardizable filter-based workflows | • Still limited for nano range • Long scan times at high resolution • Spectral mixing, weathering/additives/biofilms can reduce match quality | Detection of 12 polymer types in 22 urban water samples | 24, 25 |
| | <i>LDIR (laser direct infrared/QCL-based IR imaging)</i> • Rapid chemical imaging and identification of plastic particles on substrates • High throughput particle surveys | • Fast imaging (high throughput) • Automated particle detection and identification | • Detection limit depends on optics/substrates • Susceptible to low SNR, thickness/substrate artifacts | Detection of PE, PP, PS, and PVC in 7 groundwater bores | 26, 27 |



Table 1 (continued)

| Technologies | Major applications | Key strengths | Main limitations | Examples | Ref. |
|--------------------------------------|---|---|--|---|-----------------------------|
| | | <ul style="list-style-type: none"> • Screening of particles | <ul style="list-style-type: none"> • Domain shift across instruments and matrices • Library and machine learning model dependence for classification | | |
| Raman spectroscopy | <p><i>SERS</i></p> <ul style="list-style-type: none"> • Trace level polymer/additive detection • Nanoplastic focused assays • Hotspot-based mapping <p><i>μ-Raman (micro-Raman microscopy/mapping)</i></p> <ul style="list-style-type: none"> • Polymer ID • Particle counts/size on filters • Pigment and additive information, weathering signatures • Detailed single particle verification | <ul style="list-style-type: none"> • High sensitivity (signal enhancement) • Can work for very small/low-mass targets when adsorption is effective • Fast spectra once captured • High chemical specificity with little sample preparation • Polymer identification at smaller particle sizes • Flexible point identification and mapping • Widely available | <ul style="list-style-type: none"> • Strong dependence on substrate quality, hotspots, and particle–substrate contact (reproducibility) • Quantification is difficult, signal varies spatially • Interference from fluorescence, organic coatings, and salts • Fluorescence can overwhelm spectra (biofilms/organics/dyes) • Slower throughput for large-area scans • Laser heating/photodegradation risk for some polymers • Focusing and substrate effects can bias results | <p>Measuring the concentration of PS nanoplastics in a river water sample</p> <p>Testing microplastic content in drinking water samples</p> | <p>28, 29</p> <p>30, 31</p> |
| Fluorescence spectroscopy/microscopy | <ul style="list-style-type: none"> • Rapid screening and counting of suspected plastic particles • Size and shape mapping on filters • Bulk/spot fluorescence tracking (relative changes) • Pre-selection of particles for FTIR/Raman | <ul style="list-style-type: none"> • Fast, high throughput • Low cost, widely accessible • Good for locating small/faint particles | <ul style="list-style-type: none"> • Not polymer specific (needs FTIR/Raman/MS to confirm) • False positives from organic matter/biofilms/additives, dye uptake varies • Quantification biased by staining efficiency, bleaching, and imaging thresholds | <p>Detecting microfibers spiked in estuarine and sea water samples</p> | <p>32, 33</p> |

remaining gaps. Fig. 2 provides a conceptual overview of this section, presenting a landscape of surveyed analytical approaches and a recommended fit-for-purpose workflow.

2.1 Interface-engineered electrical and colorimetric sensors for selective nanoplastic detection

Researchers leverage interfacial chemistry, such as hydrophobicity, π - π interactions, corona dynamics, and charge, to convert nanoplastic capture into electrochemiluminescence, electrochemical, microwave, piezoelectric, or colorimetric signals.

2.1.1 Chromogenic and array-based colorimetry. These assays convert nanoplastic sorption to a chemically cross-reactive surface into a visible signal. Grumelot *et al.* developed a colorimetric sensor array to identify pristine, and protein corona-coated PS plastic particles (50 nm, 100 nm, and 2 μ m).⁴¹ Five different cross-reactive chemo-responsive dyes were

employed to fingerprint PS in buffered water, generating colour-difference maps and classifying patterns *via* hierarchical cluster analysis (HCA). Using standard chemometric analysis, the array distinguished PS from controls down to 10 ng mL⁻¹. The platform also differentiated pristine from protein corona-coated PS at 500–1000 ng mL⁻¹.⁴¹

Complementing this, Hu *et al.* introduced a single-atom Fe nanozyme confined in a zeolitic imidazolate framework (ZIF) with a colorimetric assay to quantify PS nanoplastics in water.⁴² In this system, PS nanoplastics adsorbed onto the nanozyme through electrostatic and π - π interactions, shielding Fe active sites and suppressing peroxidase-like activity toward substrates. The particle size dependent LODs of the colorimetric sensor were 0.212 (20 nm), 4.544 (30 nm), 7.624 (50 nm), 16.955 (100 nm), and 24.171 mg L⁻¹ (150 nm). A smartphone-assisted ZIF-FeSAN visual detection platform was also examined, with LODs of 0.851, 17.564, 34.554, 67.254, and 76.124 mg L⁻¹. In spiked





Fig. 1 Schematic overview of major transport (left) and transformation (right) pathways of plastic particles in aquatic systems.

tap, lake, and drinking waters, recoveries were ~ 91.5 – 108.1% for the colorimetric sensor and ~ 90.6 – 109.0% for the smartphone platform. Selectivity tests showed a strong response to 20 nm PS but little to PE, PP, and PET, and common ions produced negligible interference.⁴²

Both studies exhibit significant limitations: they focused exclusively on PS, relied on buffered water rather than complex environmental matrices, and were restricted to spiked samples instead of environmental samples. These constraints limit the application of the findings to real-world conditions.

2.1.2 Charge- and mass-loading electrical detectors. Researchers use label-free devices to sense interfacial dielectric or mass changes upon particle capture. Naik *et al.* developed a biomimetic electrochemical impedimetric sensor for selective detection of spiked PS nanoplastics (70 nm) in tap water.⁴³ Drawing inspiration from mussel adhesion proteins for surface binding and electron transfer, the authors modified a carbon paste electrode (CPE) with iron-doped polydopamine (FePDA), leveraging hydrophobic π - π interactions between PS aromatic rings and FePDA for selectivity over other polymers. A lower LOD of 1 mg L^{-1} was reached within a linear calibration range from 1 to 5 mg L^{-1} , and the recovery trails of PS spiked tap water yield 91–96% detection efficiency. The method's selectivity was evaluated against water-soluble, non-aromatic polymers (*e.g.*, PEG and PVA), yet broader cross-reactivity against other common plastics and performance in complex natural matrices or with weathered particles were not reported.⁴³

Researchers coupled electrophoresis with a quartz crystal microbalance (QCM) to detect PE nanoplastics in water.⁴⁴ The Ti-Au QCM was operated as the anode to attract

negatively charged PE, where adhesion caused resonance-frequency downshifts. A photoresist mask exposing a 1.5 mm central window concentrated deposition and enhanced sensitivity. At 10 V for 30 s per iteration, PE-spiked ultrapure water showed progressively larger shifts across repeated runs, whereas blanks remained small. The authors recommend ≤ 6 iterations to limit baseline drift from surface roughening. Applied to waters spiked with PE, sixth-iteration shifts were larger than corresponding blanks (*e.g.*, mineral water at 229 Hz with nanoplastics *vs.* 94 Hz without), and repeatability across five masked chips gave coefficients of variation of 8% in ultrapure and drinking water and 17% in mineral water. The authors indicated that the study was limited to a single polymer in spiked samples with no explicit LOD, mineral-water matrices increased mass loading, photolithographic masking was needed to gain sensitivity, and extended reuse caused baseline drift.⁴⁴

A microwave resonator approach to detect positively charged PS-NH₂ and negatively charged PS-SO₃H was proposed by Wang *et al.*⁴⁵ The platform included a printed circuit board (PCB) complementary split ring resonator (CSRR) and microfabricated GaAs integrated passive device (IPD), inferring concentration from resonance frequency shifts with $\sim 1.5 \mu\text{L}$ droplets. The authors synthesized PS-NH₂ and PS-SO₃H spheres in three size bins ($< 50 \text{ nm}$, 100 – 200 nm , and $> 200 \text{ nm}$) and measured linear responses over 0.1 – $10.0 \mu\text{g mL}^{-1}$. The lowest LODs were achieved with $< 50 \text{ nm}$ particles, at 9.78 ng mL^{-1} for PS-NH₂ and 27.57 ng mL^{-1} for PS-SO₃H. Measurements remained reliable in the presence of methyl orange dye, indicating tolerance to sample colour and some organic matter, yet demonstrations were performed in





Fig. 2 Conceptual summary of section 2, with an analytical landscape (the top panel) and a fit-for-purpose workflow (the bottom panel) for waterborne nanoplastics. The top panel approaches are grouped into four classes and each is linked to its dominant information yield. The bottom panel outlines a recommended fit-for-purpose workflow, followed by reporting minimum metadata (e.g., particle size range, LOD/Q, matrix realism, blanks, matching criteria, recovery, and uncertainty), to support QA/QC, comparability, and reproducibility across studies.

deionized water rather than environmental matrices. In addition, practical variables such as temperature or humidity drift, droplet handling, and chip cleaning can influence precision, while the higher-sensitivity IPD requires precise fabrication.⁴⁵

2.1.3 Electron-driven luminescence and photoelectrochemistry. Here, binding modulates charge transfer driven electrically or optically. Researchers presented an electrochemiluminescence assay to identify and quantify waterborne PP nanoparticles (<220 nm), validated on leachates from disposable PP food containers and spiked tap water, with a LOD of 0.948 mg L⁻¹.⁴⁶ The assay employed an amphiphilic perylene diimide (PDI-NH₂) probe with a persulfate coreactant, where strong probe-PP interactions yielded a selective ECL enhancement with an emission

maximum at ~547 nm, enabling fluorescence-matched output in a standard three-electrode cell. Selectivity tests showed clear discrimination of PP, attributed to its high surface hydrophobicity, against common plastics including PE, PS, PVC, PMMA, and PLA. The authors noted that performance was demonstrated for PP only and largely in filtered or otherwise simple matrices. The method's robustness in turbid, organic-rich environmental waters, as well as generalization across weathered particle states and broader size distributions, remained to be established.⁴⁶

A smartphone-coupled photoelectrochemical-electrochemical (PEC-EC) dual-mode sensor was built on a CdS/CeO₂ heterojunction for detecting PS nanoplastics in water.⁴⁷ The system relied on the interaction between proteins and nanoplastics, where the PS nanoparticles bound to the bovine



serum albumin (BSA) surface and led to the detachment of the protein crowns from the electrodes during the aggregation process, lowering the charge-transfer resistance and enhancing the sensing signals. The sensor showed a log-linear response from 0.5–800 $\mu\text{g mL}^{-1}$ with LODs of 0.38 ng mL^{-1} (PEC) and 9.77 ng mL^{-1} (EC) for irregularly shaped PS nanoparticles (200 and 400 nm). The sensor's recoveries reached 100.37–103.34% in PS spiked river water, yet the demonstration was PS-specific and relied on spiked samples, broader polymer scope and native field matrices were underexplored.⁴⁷

2.1.4 Bioadhesion and active-preconcentration electrochemical detection. These devices integrate biorecognition elements or mobile collectors to enhance the encounter rate prior to electrical measurement. Kim *et al.* introduced an epizoochory-inspired electrochemical sensor to quantify PS nanoplastics (460 nm) spiked in seawater, reaching a LOD of 0.772 ng mL^{-1} .⁴⁸ In this scheme, lysozyme amyloid oligomers were drop-cast onto roughened gold electrodes, the oligomers then detached from the electrodes due to their stronger affinity with contacting nanoplastics, re-exposing the gold surface and restoring the $[\text{Fe}(\text{CN})_6]^{3-/4-}$ redox current measured by cyclic voltammetry. The assay was evaluated across polymer types, shapes, and sizes, including irregular fragments ground from consumer PS, PE, PP, and PET re-dispersed in distilled water, and the sensor detected all four plastics while giving minimal signals for non-plastics (*e.g.*, wood, glass, and metal). The authors noted reduced performance at the smallest sizes (*e.g.*, 100 nm PS) and the need for matrix pretreatment to protect the sensor and mitigate interference, indicating that universality at the nanoscale and across naturally weathered, field-collected nanoplastics requires further validation.⁴⁸

Microrobots were used to preconcentrate 50 nm carboxylated PS nanoplastics in water and then detect them using electrochemical impedance spectroscopy (EIS).⁴⁹ The $\gamma\text{-Fe}_2\text{O}_3/\text{Pt}/\text{TiO}_2$ microrobots self-propelled under UV light and became positively charged, electrostatically capturing negatively charged carboxylated PS nanoparticles. The microrobots loaded with nanoplastics were then magnetically collected and deposited on screen printed electrodes, where the polymer layer restricted access of the $[\text{Fe}(\text{CN})_6]^{3-/4-}$ redox couple and increased the charge transfer resistance in EIS, yielding a capture signal. The authors pointed out that the approach required ultraviolet (365 nm) illumination and operation at acidic pH around 3 to maximize capture. The evaluation was restricted to 50 nm carboxylated PS in prepared water rather than environmental samples, and a discernible EIS response was obtained only after microrobot preconcentration at particle concentrations on the order of 10^6 particles per mL. Quantitative calibration for complex matrices was not established.⁴⁹

2.1.5 Enabling chemo(bio)sensors for dissolved plastic residues. While not particle-resolving, residue-focused sensors inform sources and co-occurring contaminants. Jebiril *et al.* reviewed nanomaterial-enabled electrochemical chemo(bio)sensors for detecting nanoplastic residues, such

as catechol, hydroquinone, bisphenol A, and resorcinol.⁵⁰ The authors surveyed methods and materials, covering voltammetric, amperometric, and impedance-based measurements with carbon, noble-metal, metal-oxide, and polymer-film modifiers, alongside enzyme, antibody, and aptamer recognition elements. Reported performance typically spans nM– μM LODs for phenolic residues, with demonstrations in tap, river, lake, dam, swamp, and mineral waters. Key limitations include matrix interferences in real waters, limited calibration standards and reference materials, biorecognition stability and electrode fouling, and the persistent gap between proof-of-concept devices and portable on-site monitoring.⁵⁰

Across interface-engineered sensors, reported LODs span from mg L^{-1} for nanozyme systems and some impedimetric formats to ng mL^{-1} for microwave and PEC–EC platforms. Most demonstrations remain PS-centric owing to the wide selection of PS standards available in the market. Meanwhile, mostly produced PE and PP are less investigated due to their limited commercial availability and inert reactivity comparing to PS. Moreover, most research rely on spiked, relatively simple water systems only. Such patterns consistently reflected in the studies reviewed in section 2 (Fig. 3).

Apparent selectivity here in section 2.1 typically derives from non-covalent hydrophobic and π – π interactions favouring aromatic polymers like PS, protein-corona dynamics, or surface charge. Consequently, performance often degrades for aliphatic, non-aromatic polymers such as PE and for strongly weathered plastic particles. Few studies document operational durability like fouling and baseline drift, inter-instrument reproducibility, or quantitative response functions with respect to ionic strength, pH, and dissolved organic matter. Portable units, such as smartphone-integrated PEC–EC and planar microwave and QCM devices, are promising for *in situ* screening, but calibration transfer and blinded validation in real environmental waters are still needed for fit-for-purpose deployment.

2.2 Raman-based and microprobe vibrational fingerprinting

Raman and microprobe vibrational techniques deliver polymer-specific fingerprints at or near the single-particle level. To overcome weak scattering and diffusion limits, recent studies combine surface-enhanced Raman spectroscopy (SERS) with active capture or interface-based enrichment, while resonance Raman and atomic force microscopy-infrared (AFM-IR) provide intrinsic chemical signatures without plasmonic amplification.

2.2.1 SERS with active or interface-based preconcentration. SERS platforms amplify weak Raman signals by concentrating particles into electromagnetic hot spots or enriching them at engineered interfaces. Recent implementations include electro-photonic tweezers, light-driven microrobots, optical tweezers paired with SERS collectors, shrinking surface-bubble deposition, self-assembled Ag-nanowire membranes,





Fig. 3 (a) Polymer coverage across studies in section 2 vs. 10 years' production estimation of the corresponding polymer. (b) Best mass LOD (ng mL⁻¹) vs. matrix realism of studies reviewed in section 2. Matrix realism scale: 0 = distilled water, buffer, simulated water, or other simple laboratory matrices; 1 = spiked real-world water samples, including environmental waters, tap water, and drinking water; 2 = native field waters.

metal-phenolic aggregation, and electrochemical SERS concepts.^{28,51-56} Across these strategies, researchers aim to overcome mass transfer limits to enable polymer-specific detection and, where feasible, quantification in environmental waters.

An electro-photonic tweezers platform was used to identify 200 nm PS nanoplastics in water.⁵¹ In this system, long-range alternating current electro-osmosis flows and dielectrophoresis (DEP) forces drove suspended nanoparticles into the Raman spot, while co-introduced gold nanorods created SERS hotspots that compensated for weak cross-sections at sub-wavelength sizes. LODs of 4.66 mg L⁻¹ and 1.17 µg L⁻¹ were reported for 200 nm and 30 nm PS nanoparticles in distilled water. The authors further demonstrated that PS and PMMA model nanoparticles exhibit distinct DEP trapping optima enabling target-specified separation, and showed operation in higher-conductivity tap water. They noted constraints including reliance on commercially available, monodisperse spheres and simple laboratory matrices, underscoring the need for broader polymer scope, weathered particles, and native environmental waters.⁵¹

A light-sensitive microrobot-based approach was introduced for capturing and detecting nanoplastics spiked

in water.⁵² The ATAR microrobots (*i.e.*, Au/TiO₂/Au/R-Fe₃O₄) self-propelled under 400 nm light using H₂O as fuel, sweeping nanoplastics into their multilayer TiO₂/Au cages *via* electrostatic attraction and physical trapping. The robots were then magnetically separated, and the captured particles were analysed by SERS at the robots' Au hot spots. LODs of 1.27 µg mL⁻¹ for PS (500 nm) and 0.61 µg mL⁻¹ for PMMA (500 nm) in pure dispersions were reported. SERS confirmed recoveries of >80% for both PS and PMMA standards spiked in river water samples. The authors noted several constraints, including reliance on monodisperse 500 nm spheres and spiked or simple matrices, scalability limited by Au cost, and the need to assess potential TiO₂ toxicity and broader biocompatibility, pointing to biodegradable alternatives and validation on weathered, low-abundance field samples.⁵²

An optical-manipulation and SERS platform was presented that simultaneously concentrated and quantified environmental nanoplastics in river and seawater.²⁸ The setup used 20 µm gold nanoparticle stacks coated with PLA as optical tweezers to drive nanoplastics under laser irradiation, and then 80 µm stacks to enrich and detect them. A brief cleaning rinse mitigated humic-acid interference that otherwise suppressed PS SERS signals, while black carbon at 10 µg L⁻¹ had negligible impact. Applied to



natural waters, the platform detected PS at 6.5–8.5 $\mu\text{g L}^{-1}$ and PET nanoplastics at 66 $\mu\text{g L}^{-1}$ in river water, and PS at 0.7–1.0 $\mu\text{g L}^{-1}$ (beach) and 1.4–1.8 $\mu\text{g L}^{-1}$ (mariculture) in seawater. The authors noted potential eco-corona effects and recommended Fenton digestion for accurate quantitation.²⁸

A shrinking surface bubble deposition (SSBD) method was employed in combination with SERS for identification and scanning electron microscopy (SEM) for morphology to study nanoplastics in ocean waters.⁵³ In SSBD, seawater was mixed with 10 nm Ag nanoparticles, and laser heating generated a surface bubble whose Marangoni flow concentrated suspended particles at the three-phase contact line. The co-deposited Ag nanoparticles both aided deposition and enabled SERS for chemical identification. In water samples collected across seven locations in two oceans, the authors reported nylon nanofibers, PS flakes, and PET ball-stick nanostructures. The approach is not yet quantitative, further fluid-mechanics studies are needed to relate deposited spot density to bulk concentrations.⁵³

Membrane filtration was coupled with SERS by self-assembling silver nanowire membranes that both preconcentrate nanoplastics and enhance their Raman signals.⁵⁴ Under controlled conditions, the membranes retained 86.7% of 50 nm standard PS particles and ~93.1–98.0% of 100–1000 nm PS particles. Raman mapping resolved particle distributions across 10^{-1} – 10^{-7} g L^{-1} . The method was tested on river and seawater samples, where no nanoplastics were detected. The authors then spiked 500 nm PS standards into seawater and were able to detect them, yet the spiked river water produced no signal. The authors noted matrix effects that hindered detection in environmental waters, and the absence of a full quantitative calibration for complex matrices, despite the counting potential of Raman mapping.⁵⁴

Metal–phenolic network-mediated aggregation was combined with gold nanoparticle-based SERS and customized machine learning to enrich, identify, and quantify PS (50 nm, 500 nm, and 1 μm), PMMA (500 nm), PE (740 nm to ~5 μm), and PLA (250 nm) nanoplastics.⁵⁵ Applied to spiked lake water, a LOD of 10 mg L^{-1} was reported for all four types of plastics, with classification accuracies >95% for PE, PS, and PMMA, and ~74% for PLA. Recoveries in spiked matrices were ~80–120% for PS and PMMA, while PLA at low concentrations and PE in lake water showed lower recoveries, reflecting matrix interferences (e.g., metal-ion competition with Zr^{4+} for tannic acid) and a weaker PE Raman band at 1297 cm^{-1} . Limitations noted by the authors include higher LODs in complex waters, challenges in PLA identification at low concentrations due to potential biodegradation, the need to re-establish PE calibration in lake water, and evaluation constrained by the limited set of commercially available nanoplastic types and sizes.⁵⁵

A perspective was presented by Yang *et al.* on coupling electrochemistry with SERS as a dual-mode nanoplastic sensing system, leveraging electrochemistry's quantitative capability together with SERS's polymer-specific fingerprints.⁵⁶ The authors highlighted potential-controlled electrosorption to

preconcentrate nanoplastic particles at SERS hot spots, and electrochemical deposition and activation to build and refresh hot-spot-rich conductive substrates. They also described SERS-assisted electrochemical workflows that identify polymer type and profile surface adsorbates, while the electrochemical channel provides concentration and kinetic information.⁵⁶

SERS methods succeed by concentrating analytes, but the same capture physics can bias results by size, charge, and shape, while organic matter and salts in the water can reduce the signal. Robust use therefore requires reporting and correcting for preconcentration factors, capture efficiencies, and recoveries, along with strategies to refresh hot spots and mitigate fouling.

2.2.2 Resonance Raman and AFM-IR fingerprinting. Here we review selected intrinsic fingerprinting methods that do not rely on plasmonic enhancement. Cărdan *et al.* revealed that the copper-phthalocyanine blue pigment (PB15) embedded in aged blue plastics produced a strong resonance-Raman signal that enabled confocal micro-Raman detection of environmentally derived blue nanoplastics down to ~500–800 nm, even when polymer bands were absent. Surveying naturally weathered plastics including LDPE, HDPE, PP, PS, and PET, PB15 occurred in 81% of blue/green macroplastics and 51% of blue microplastics. The authors also noted potential spectral interferences from plant- and animal-derived, blue-dyed fibres but highlighted their distinct coloration chemistry and Raman profiles. The study framed PB15 as a double-edged marker, *i.e.*, a Trojan-horse risk coupling pigment and plastic, and an opportunity to target blue nanoplastics in complex environmental matrices.⁵⁷

Dielectrophoresis was combined with Raman spectroscopy to trap and identify nanoplastic particles in suspension.⁵⁸ The electric field generated negative dielectrophoresis, concentrating particles into the Raman confocal volume for 30 s before acquisition. In bottled drinking water spiked separately with each polymer, 200 nm PS, 180 nm PP, and 100 nm PET were identified at ~20 $\mu\text{g mL}^{-1}$. The authors indicated that direct injection into the dielectrophoresis cell minimizes interference from dissolved salts, but more complex matrices like tap or environmental water will require pretreatment such as cascade filtration, enzymatic or chemical digestion, and preconcentration.⁵⁸

AFM-IR was used to identify nanoplastics in surface seawater (0.1–1.0 m depth).¹⁴ After sequential filtration and ethanol-assisted deposition to suppress coffee-ring effects, particles were screened by optical microscopy and SEM and energy dispersive X-ray (EDX), and then chemically assigned by AFM-IR. The authors identified nanoplastics including highly crystalline poly(3-hydroxybutyrate) (~700 nm) and a bisphenol-A based epoxy (~860 nm), alongside polyester microplastics. They also noted the method's low throughput and the value of pairing it with mass spectrometry methods to provide quantitative context.¹⁴

2.2.3 Plasmonic binding sensors without Raman enhancement. A nano-plasmonic biosensor was developed to



detect and quantify nanoplastic particles in seawater.⁵⁹ PMMA (100 nm) was used as a nanoplastic model for control tests. The platform employed estrogen receptors grafted onto a polymer-based gold nanograting (ER-GNG). Binding of PMMA nanoplastics to the receptor modulated the hybrid surface plasmon resonance (SPR) and localized surface plasmon resonance (LSPR) modes, resulting in a characteristic blue shift in the plasmonic resonance. The method was applied to real seawater, where 1.5 ng mL⁻¹ in the 1:20 dilution of nanoplastics was detected without further sample pretreatment, while the polymer type was unspecified.⁵⁹

Across the methods surveyed, SERS platforms offer the highest sensitivity but require careful control of capture bias, hot-spot stability, and matrix effects. Resonance Raman and AFM-IR provide chemical assignments at lower throughput. Plasmonic binding sensors enable simple, label-free monitoring but lack spectral identification. Priority next steps are to adopt matrix-matched calibrations, develop certified reference materials, routinely report preconcentration factors, capture efficiencies and recoveries, conduct blinded tests on environmental waters, and evaluate performance with weathered plastics.

2.3 Label-free photonic and refractometric sensors for number-focused quantification

Label-free photonic and refractometric platforms quantify nanoplastic particles by converting changes in light scattering, optical interference, laser-induced breakdown, or fluorescence lifetime into concentration measurements.

A laser-backscattered fibre-embedded optofluidic chip (LFOC) was developed to quantify nanoplastics spiked in water.⁶⁰ In the device, a 635 nm beam was launched and collected through a single-multimode fibre coupler, and the 180° back-scattered signal scaled linearly with both mass concentration and particle number. Across 20–500 nm PS, LODs were 60.00, 19.75, 1.22, 0.23, and 0.39 µg mL⁻¹, respectively. The quantitative approach also extended to 200 nm PE, PET, PMMA, and PP using a universal calibration with a LOD of 1.08 µg mL⁻¹. Spiked river water samples yield 95.56–114.47% recoveries. The authors noted several limitations: the platform did not provide polymer-specific chemical identification; performance depended on particle size and optical conditions (*e.g.*, refractive-index effects on Fresnel reflections); number concentrations were inferred from size and density rather than directly counted; and real-world evaluation relied on spike-and-recovery rather than native field or weathered particles.⁶⁰

Laser-induced breakdown detection (LIBD) was evaluated as a particle counting approach for nanoplastics in water.⁶¹ In LIBD, a focused laser irradiated the suspension, and then multiphoton ionization produced seed electrons that absorbed energy *via* inverse Bremsstrahlung and avalanched to a dense plasma, launching a shockwave recorded as optical plumes or acoustic spikes. Irregular, polydisperse PS, PP, and PET nanoplastics (~100 nm) generated from single-use plastic

products and dispersed in a simple saline electrolyte yielded LODs of 1 × 10⁴–3 × 10⁵ particles per mL. Sensitivity depended on particle size, concentration, and material properties (*i.e.*, density, ionization energy, optical attenuation, and aggregation), generally favouring hard over soft particles. The authors noted that water chemistry and multimodal size distributions could shift breakdown probability and mask smaller particles, and they recommended coupling LIBD with size-separation approaches and exercising caution when extrapolating from PS-based calibrations.⁶¹

A fluorescence lifetime analysis (FLA) system was presented that uses label-free fluorescence with fit-free phasor analysis of time-correlated single-photon counting to detect model PS nanoplastics in water.⁶² The method detected unmodified PS suspensions with a LOD of 0.01 mg mL⁻¹. Validation used commercial PS models, including PS-nano (121 nm) and PS-micro (1.35 µm), with fluorescent COOH-PS (35.8 nm) and NH₂-PS (140 nm) as references. Phasor modulation scaled with concentration, enabling calibration-style quantification, while the phase lifetime remained characteristic of the particle type. The authors noted current constraints where evaluation is limited to PS in water and polymer differentiation is not yet demonstrated, highlighting the need to lower the LOD further and to distinguish among plastics in more complex matrices.⁶²

An optical microfiber Mach–Zehnder (M–Z) interferometer sensor was proposed by Li *et al.* for detecting PS nanoparticles (100 and 150 nm) in water.⁶³ A tapered microfiber supported two-mode interference, and adsorption-induced refractive-index changes in the evanescent field shifted the transmission spectrum. An aminated surface decorated with L-phenylalanine (L-PHA) captured PS *via* π–π stacking for selective monitoring. In simulated environmental water samples, LODs were 2.31 × 10⁻⁶ mg mL⁻¹ and 2.96 × 10⁻⁶ mg mL⁻¹ for 150 nm and 100 nm PS, respectively. The authors reported measurable wavelength shifts in nine PS spiked environmental waters including lakes, seawater, and wastewater, without providing recoveries or other quantitative results.⁶³

In a model simulation study, a plasmonic refractive index sensor for potential nanoplastic detection in water was proposed by Guchhait *et al.*⁶⁴ The device comprised a metal–insulator–metal (*i.e.*, Ag–air–Ag) waveguide side-coupled to a concentric square-ring resonator and was simulated in 2D using COMSOL finite-element optics. Surface plasmon modes in the waveguide produced transmission resonances that shifted with the local refractive index (*n*) inside the resonator as the sample filled it. In an initial nanoplastic scenario with *n*_{water} ≈ 1.34 and *n*_{nanoplastics} ≈ 1.5, the system exhibited a linear resonance shift for Δ*n* = 0.00025–0.001, corresponding to 0.15625–0.625% plastics in water (v/v). The authors cautioned that temperature, salinity, and co-present particulates can also shift refractive index, and suggested adding a plastic-binding peptide to enhance specificity.⁶⁴

Label-free photonic sensors provide rapid, compact measurements with competitive LODs, but their signals are



generally not specific to polymer type and are sensitive to matrix factors such as refractive index, salinity, and dissolved organic matter. Calibrations can misestimate concentrations when particle sizes or shapes vary, and number concentration is often inferred rather than directly counted unless sizing is built in. In practice, these platforms can be potentially used as front-end screens or counters that access samples for follow-up, chemically specific confirmation.

2.4 Separation-based workflows and mass spectrometry routes for chemically resolved quantification

Separation-centric approaches decouple particle size, composition, and matrix effects, enabling polymer identity and abundance to be quantified with reduced ambiguity. Coupling these separations to mass spectrometry provides a bridge between particle-number measurements and polymer-specific mass detection.

Capillary electrophoresis with an ultraviolet-visible (UV-vis) spectrophotometric diode-array detector (DAD) was developed for size separation and detection of PS and PMMA nanoplastic spheres under alkaline conditions.⁶⁵ Experiments were run at pH 8.9 for PS (30–300 nm) and at pH 11.9 for PMMA (50–200 nm), with LODs on the order of 10^{11} particles per mL for PS and $\sim 5 \times 10^{11}$ particles per mL⁻¹ for PMMA. The study also quantified nonlinear electrophoresis at high fields and reported size-dependent effective mobility, with surface charge density decreasing as size increased. Demonstrations were limited to buffered suspensions, while no environmental waters were tested. Limitations noted by the authors also include low sensitivity for real samples, potential capillary clogging or tailing above ~ 300 nm, UV interferences and peak overlap for mixtures, and a need for preconcentration. The authors further suggested downstream MS coupling for a more comprehensive analysis.⁶⁵

A pyrolysis-gas chromatography mass spectrometry (Py-GC-MS) workflow was developed for nanoplastic particles and agglomerates (10–1000 nm) in environmental waters.⁶⁶ Following H₂O₂ oxidative cleanup, samples were firstly passed through a 1 μ m filter and then concentrated with a 100 kDa stirred-cell ultrafiltration step prior to pyrolysis. Polymer identification and quantification relied on selected indicator ions, for example, styrene dimer and trimer for PS, C₁₈–C₂₁ *n*-alkadienes for PE, and vinyl-benzoate species for PET. The authors reported < LOD – 0.76 μ g L⁻¹ PE, 0.18–0.25 μ g L⁻¹ PET, 0.32–0.51 μ g L⁻¹ PS, < LOD – 0.59 μ g L⁻¹ PP, and < LOD – 0.04 μ g L⁻¹ Nylon 66 nanoplastics in surface water, reservoir water, and stormwater.⁶⁶

Li *et al.* combined asymmetric flow field-flow fractionation (AF4) coupled with multi-angle light scattering (MALS) and Py-GC-MS to obtain size-resolved particle counts and polymer-specific mass quantification of nanoplastics in water.⁶⁷ AF4-MALS provided calibrated models to infer concentration and particle number from MALS peak area and size, while Py-GC-MS identified PS and PMMA *via* styrene and methyl methacrylate pyrolysates and quantified mass. For PS

(60–300 nm) without preconcentration, AF4-MALS showed size-dependent LODs of 0.5–5 ppm. Py-GC-MS quantification used indicator ions with linear calibrations for styrene (0.02–8 μ g, PS) and methyl methacrylate (0.03–10 μ g, PMMA), and an estimated LOD of 0.01 μ g for both markers. In bottled water spiked with 200 nm PS and 100 nm PMMA, Py-GC-MS confirmed polymer identity and yielded mass-based overall recoveries of 81.5% (PS) and 69.4% (PMMA), complementing AF4-derived fractional recoveries of 57.2–61.0%. The authors further noted that preconcentration is essential yet risks pore clogging and morphology change, which can depress AF4 recovery and underestimate concentrations.⁶⁷

Nanoplastics were identified and quantified across the North Atlantic using thermal-desorption proton-transfer-reaction mass spectrometry (TD-PTR-MS).¹⁹ Along a transect from the subtropical gyre to the European shelf, the authors measured ~ 1.5 –32.0 mg m⁻³ of PET, PS, and PVC throughout the water column. The reported mixed-layer (10 m below sea level) totals were ~ 1.4 times higher than at intermediate depth, with the highest values near Europe. Intermediate depth (1000 m below sea level) totals were ~ 1.8 times higher inside the gyre. Bottom waters (30 m above the seafloor offshore and 5–10 m above the seafloor at coastal sites) averaged ~ 5.5 mg m⁻³ and dominated by PET. The mixed layer of the temperate-subtropical North Atlantic was extrapolated to ~ 27 million tonnes. The authors noted that reported concentrations were lower limit estimates because thermal desorption and ionization convert only some of the plastic into detectable ions. Spike-and-recovery tests for PS gave about $\sim 7\%$ recovery, indicating that true PS levels may be higher. Monte Carlo simulations showed that any overestimation driven by organic matter would fail before exceeding $\sim 31\%$. PE and PP were not detected, which the authors attributed to possible chemical alteration in seawater, masking by organic matter, or concentrations below LOD. The one micrometre prefilter also removed most marine snow, so larger aggregates were excluded.¹⁹

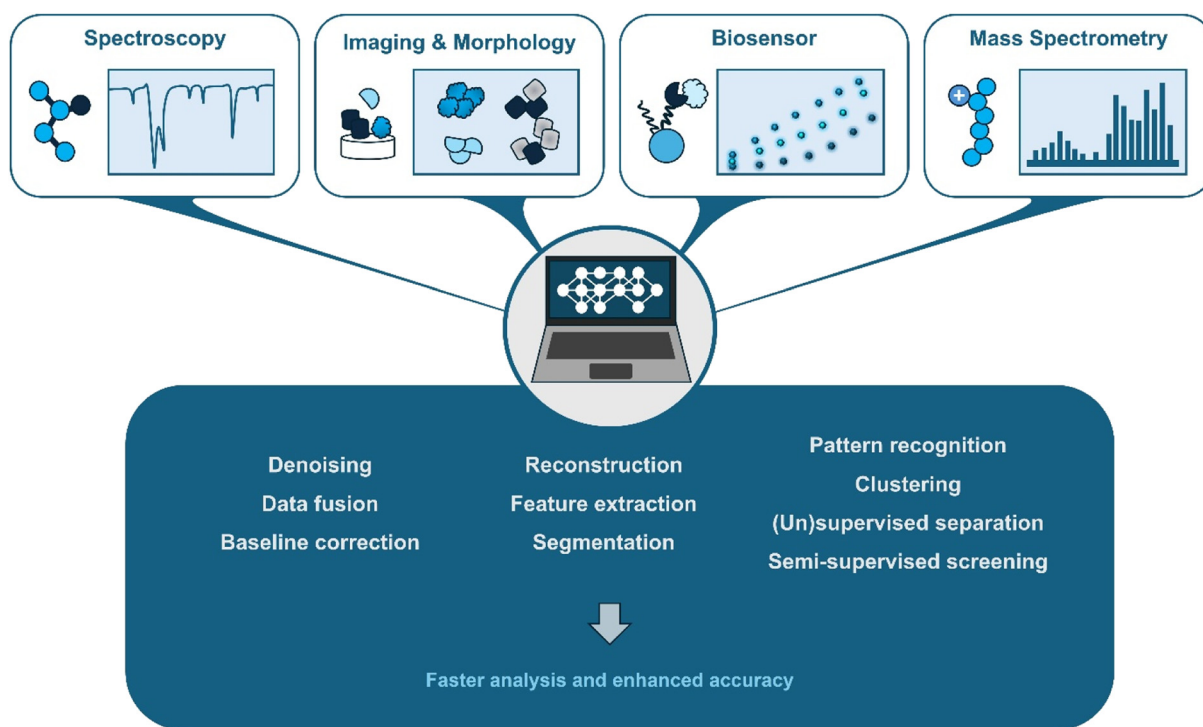
Mass spectrometry approaches provide polymer-specific mass measurements and are generally less affected by additive composition, but they often trade throughput for selectivity and sample preparation steps can alter particles. AF4 links particle sizing with mass quantification, yet results depend strongly on membrane chemistry and on how the sample was collected, stored, and pretreated. Py-GC-MS studies have shown that false positives and associated overestimation can arise in complex, organic-rich matrices because endogenous constituents, notably lipids, can generate non-specific pyrolysis products that overlap with polymer marker profiles, particularly for PE and PVC.^{68,69} Accordingly, Py-GC-MS quantification in biotic or lipid-rich environmental samples should be supported by strict cleanup, procedural blanks/controls, careful marker selection, and orthogonal confirmation to minimize misassignment and overreporting. Environmental samples analysed by TD-PTR-MS in the lab have yielded polymer-related ion signals, though recoveries are often low and some



polymers are missed, underscoring the need for certified reference materials, inter-laboratory harmonization, and careful spike-recovery evaluation in real matrices.

Across nano- and microplastic research, translating quantification between particle number-based metrics (*e.g.*, FTIR and Raman particle counting) and mass-based metrics (*e.g.*, thermal and MS approaches) remains a challenge. Conversions require assumptions about particle size distributions, morphology (*e.g.*, fibres *vs.* fragments), and polymer density, yet these parameters are often

incompletely reported and can vary substantially across matrices and weathering states. Method-specific detection windows further bias which size fractions are captured, making back-calculation non-unique and limiting direct comparability across studies that quantify on different bases.^{70,71} Recent reporting and standardization efforts therefore emphasize transparent documentation of size bins, shape classes, and quantification basis to enable more defensible inter-study comparisons and, where attempted, conversions.



Cross-Cutting Challenges



Fig. 4 Machine learning enabled integration of analytical modalities for nano- and microplastic analysis and associated cross-cutting challenges. Top panels illustrate representative analytical inputs feeding into machine learning workflows. Central processing steps collectively enable faster analysis and improved classification accuracy. Bottom panels summarize cross-cutting challenges that limit robust deployment across matrices and scales.



3. The incorporation of machine learning with analytical approaches

The rapid advancement of machine learning has opened new opportunities for detecting nano- and microplastics in aquatic systems. Conventional spectroscopic, spectrometric, and imaging-based techniques are often limited by labour-intensive workflows, operator bias, and the inherent difficulty of distinguishing plastics from natural colloids and other nanoscale or microscale particles.^{34,72,73} By applying data-driven algorithms, machine learning offers a path to improved signal processing, extraction of meaningful patterns from complex datasets, and automated classification.^{34,72–74} These capabilities hold promise for streamlining workflows and achieving more consistent results, while accelerating the transition of laboratory techniques into scalable, field-deployable monitoring systems. In what follows, we organize recent progresses before analysing cross-cutting needs and outlook. Fig. 4 summarizes how machine learning methods are integrated across diverse analytical techniques for nano- and microplastic analysis, while highlighting the key challenges that constrain robust performance across matrices, size regimes, and deployment contexts.

3.1 Spectroscopy-based detection

Spectroscopy provides chemically specific fingerprints. Machine learning is predominantly leveraged in this field for denoising, reconstruction, feature extraction, and classification under low signal-to-noise.

3.1.1 Infrared spectroscopy. A machine learning-based workflow for polymer identification from Fourier transform infrared (FTIR) spectra of microplastics was proposed by Liu *et al.*⁷⁵ To address baseline shifts, noise, and interfering peaks that often compromise FTIR analysis, the authors employed two deep learning reconstruction models, including an autoencoder (AE) and a V-like convolutional neural network (VCNN), to enhance spectral quality prior to classification. Four classification models were trained, including decision trees, random forest, linear support vector machines, and one-dimensional convolutional neural networks, for microplastic identification. The workflow retained practical value with a top-1 accuracy of 71.43% and top-3 accuracy above 90% when applied to environmental datasets, while the authors noted that the approach was limited by an inadequate and biased environmental dataset, augmentation methods that did not capture real-world weathering and oxidation, and the potential for deep learning-based reconstruction to introduce additional noise and spectral distortions.⁷⁵

A dense feed-forward neural network (DNN) was assessed for automated classification of FTIR spectra of environmental microplastics, benchmarking its performance against other machine learning models and human annotations.⁷⁶ Trained on an aggregated dataset spanning 16 polymer categories

from diverse marine and freshwater sources, the DNN delivered higher precision and F1 scores than alternative approaches. The model also revealed systematic mislabelling in public FTIR datasets and correctly re-assigned spectra that human annotators had misclassified.⁷⁶

A deep learning-assisted spectroscopy fusion framework was introduced for the identification of microplastics using Raman and attenuated total reflection Fourier transform infrared (ATR-FTIR) spectra.⁷⁷ A one-dimensional convolutional neural network (1D-CNN) with an embedded multi-head attention mechanism was trained on eight polymer types, reaching 73% and 75% accuracy with ATR-FTIR and Raman spectra, respectively. The authors also proposed a three-level fusion strategy that combined complementary information from ATR-FTIR and Raman to enhance the recognition performance. The model yielded progressively higher accuracies with an increased level of fusion data, and was validated in detecting PMMA microplastics spiked in tap water.⁷⁷

Tian *et al.* paired quantum cascade laser-based IR (LDIR) with a hybrid machine learning workflow to identify weathered microplastics in surface and drinking water. The authors trained ensemble supervised classifiers, *i.e.*, subspace *k*-nearest neighbour (Sub-*k*NN) and boosted decision tree (BDT), on spectrum characteristics of a labelled subset and applied confidence thresholds to scale predictions to the full dataset. Low confidence spectra were then clustered with a density-based spatial clustering of applications with a noise (DBSCAN) model guided by a t-distributed stochastic neighbour embedding model, followed by domain experts labelling efforts. The authors noted that the methodology can be constrained by the availability of quality labelled spectra for training, and that classification is bounded by the set of predefined categories, with other polymers treated as outliers. They also noted that the heterogeneous non-plastic fraction can be difficult to group cleanly and may be falsely labelled as plastic, and that DBSCAN outcomes depend on pragmatic, subjective choices of parameters (*e.g.*, the minimum number of points and the neighbourhood searching radius) that affect the balance between clusters and outliers.⁷⁸

3.1.2 Raman spectroscopy. An automatic workflow for the identification of PS and PVC nanoplastics in rainwater was implemented by combining highly reflective aluminium substrates with Raman spectroscopy and machine learning.⁷⁹ To overcome weak Raman signals and noise interference typical of nanoscale particles, the authors applied a peak extraction and retention (PEER) algorithm for spectral pre-processing, followed by training a random forest model comprising 100 uncorrelated decision trees. The model was developed on five polymers, including PE, PTFE, PS, PMMA, and PVC, and reached an average classification accuracy of 98.8%, despite the inherent difficulty of separating PE from PTFE and PMMA from PVC because of overlapping spectral features. Compared with support vector machine (SVM) and back propagation (BP) neural networks, the random forest



outperformed both, particularly in handling low signal-to-noise spectra.⁷⁹

Qian *et al.* introduced a hyperspectral stimulated Raman scattering (SRS) microscopy platform integrated with a data-driven spectral matching algorithm to achieve single-particle chemical imaging of nanoplastics.¹⁵ By employing a narrowband SRS imaging scheme, the system reached LODs down to below 100 nm, while the tailored algorithm enabled polymer identification across seven common plastics, including polyamide (PA), PP, PE, PMMA, PVC, PS, and PET. Beyond quantification, multidimensional profiling of particle size, morphology, and composition uncovered noticeable heterogeneity and non-orthogonality between different plastics. The authors applied the method to bottled water as a model system and reported average concentrations of $2.4 \pm 1.3 \times 10^5$ nano- and microplastic particles per litre.¹⁵

An agarose-based microfiltration device was paired with Raman spectroscopy and a convolutional neural network (CNN) to identify 100 nm PS nanoplastics in deionized water and filtered seawater.⁸⁰ Raman spectra were pre-processed and augmented before being analysed with a fine-tuned CNN focused on the polymer fingerprint region. The approach achieved a LOD of $6.25 \mu\text{g mL}^{-1}$ while reducing mapping time by 50%. The CNN also outperformed true component analysis (TCA) at shortened acquisition times (*i.e.*, 0.2 and 0.1 s per point), maintaining $\geq 75\%$ prediction confidence, whereas TCA failed due to low signal-to-noise ratios. Occasional false positives occurred under weak signal intensities, pointing to the need for further noise-reduction and signal-enhancement strategies.⁸⁰

3.1.3 Fluorescence labelling and imaging. Image-based quantification of microplastics in bottled water was advanced by introducing an Ilastik-trained random forest classifier for semantic segmentation of Nile red-stained fluorescence microscopy images.⁸¹ The classifier was trained on images of blanks, bottled water samples, and stained microplastic standards including PE, polyamide-6 (PA 6), PET, and polyvinylidene chloride (PVDC). This supervised learning workflow replaced conventional thresholding approaches by generating binarized masks, which were then refined in ImageJ using watershed and noise-reduction functions. The method demonstrated a minimum detectable particle size of 10 μm , with particle count detection and quantification limits of 28 and 85 particles per 500 mL, respectively. When applied to 20 bottled water samples, the method reported microplastic abundances ranging from below detection to 7237 items per 500 mL.⁸¹

Extending this concept, Meyers *et al.* evaluated decision tree and random forest algorithms for automated Nile red-based identification of pristine microplastics, and plastic particles weathered in semi-controlled marine environments.⁸² Five types of uncoloured microplastics with heterogeneous shapes, including PE, PET, PP, PS, and PVC, were deployed for 12 months in subsurface coastal waters and at deep-sea sites to undergo natural weathering. Both models reported $>90\%$ accuracy for detecting and identifying

pristine particles, while random forest classifiers outperformed decision tree models in identifying weathered plastics. A lower LOD of 2–4 μm was reached.⁸²

A laser-induced fluorescence (LIF) system was coupled with machine learning algorithms for the identification of predominantly uncoloured, known microplastics in simulated marine water.⁸³ Using 405 nm laser excitation, fluorescence spectra of plastics, including PE, PP, PS, and PET, together with common marine materials, were recorded and pre-processed with principal component analysis before classification. A two-step machine learning workflow first distinguished plastics from non-plastic organics with 97.6% accuracy and then identified polymer types with 88.3% accuracy. The authors noted challenges such as spectral overlaps from pigments or biofilms, which may obscure plastic signals, and emphasized the need for further validation with more diverse and coloured samples.⁸³

A wireless portable device was engineered for quantifying nano- and microplastics by integrating machine learning algorithms with fluorescence imaging.⁸⁴ PS nanoparticles (50 nm) spiked into tap water were labelled and concentrated using a luminescent metal phenolic network (L-MPN) supramolecular labelling strategy for fluorescence-based detection. A customized MATLAB code executed on a smartphone converted raw fluorescence microscopic images into quantitative outputs, while a decision tree model embedded in the system facilitated the analysis across diverse particle sizes and concentrations. The lower boundaries of the size-dependent LODs ranged from 330 particles for 10 μm microplastics to 2.58×10^8 particles for 50 nm nanoplastics under optimized labelling conditions.⁸⁴

3.2 Imaging and morphological analysis

Unlike spectroscopy, morphology-centric pipelines classify particles using shape, phase, and dynamical features, even when polymer chemistry is not explicitly available, offering label-free throughput but requiring careful validation against chemical identities.

3.2.1 Optical and light-based imaging. Photoacoustic imaging-assisted deep learning was used to characterize microplastics sourced from environmental water samples.⁸⁵ After laboratory screening and FTIR confirmation, six representative morphologies, including flakes of PVC, granules of PP, loops of PVC, ovals of PVC, rods of PP, and fragments of PE were selected, alongside non-plastic controls. To address limited datasets, the authors developed an automated pre-processing pipeline using the segment anything model (SAM), thresholding, and contour-based anchor box slicing to generate segmentation masks and extract individual sample images, thereby expanding the dataset without extensive manual labelling. These data were then used to train a generative deep learning model with dual proxy tasks, which enabled unsupervised segmentation, target detection, and classification of microplastic morphologies independent of polymer type.⁸⁵



A lensless shadow microscopy (LSM) strategy combined with deep learning-based object detection algorithms (*i.e.*, You Only Look Once (YOLO)) was introduced for quantifying abraded sponge microplastic fibres spiked into natural waters.⁸⁶ The custom LSM system offered a field-of-view $>1\text{ cm}^2$ with submicron-pixel resolution ($\sim 500\text{ nm}$ per pixel), while the algorithms automatically classified and measured linear and branched fibres in projection images. The method reported a LOD of 10 items per mL and was validated across dispersed and aggregated fibres as well as natural water matrices. The authors noted that sample transfer losses and the lack of chemical characterization remain as limitations, highlighting the need for complementary spectroscopic methods and optimized handling procedures.⁸⁶

A machine learning-integrated droplet microfluidic system, named MiDREAM, was presented for the quantification and size classification of spiked PS microspheres in water.⁸⁷ The system encapsulated microplastics into uniform droplets ($\sim 142\text{ }\mu\text{m}$ diameter) and combined phase-contrast imaging with an optimized You Only Look Once (YOLO) v8 convolutional neural network for automated detection. The method differentiated PS microspheres across size classes from 3 to $50\text{ }\mu\text{m}$ and outperformed hemocytometer counting and surface enhanced Raman spectroscopy (SERS) in accuracy. The approach was demonstrated using spiked PS particles in environmental water matrices, while distinguishing plastics from non-plastic particulates in real-world samples requires complementary chemical confirmation.⁸⁷

An artificial intelligence-assisted nano-digital inline holographic microscopy platform was introduced for the automated characterization and classification of nano- and microplastics in natural waters.²⁰ From raw holograms, over 20 morphological parameters, including particle size, shape, perimeter, optical phase, surface roughness, area, and edge gradient, were extracted to train deep neural network classifiers under supervised learning. The system reached characterization and classification rates of 1.4 and 25 particles per second, respectively, and identified $\sim 2\%$ and $\sim 1\%$ of suspended particles as plastics in lake and river waters. The authors applied size correction approaches to address the overestimation observed at the nanoscale and further highlighted the need for optimized setups and improved numerical reconstructions to extend the LOD below 100 nm .²⁰

3.2.2 Electron microscopy-based imaging. Researchers sought to couple scanning electron microscopy (SEM) with a UNet-based computer vision algorithm to differentiate plastic particles ($\sim 1\text{ }\mu\text{m}$) from other materials by tracking spatiotemporal deformation behaviour under electron beam irradiation.⁸⁸ Lower-crystallinity polymers such as PVC and PET exhibited distinct shrinkage patterns compared to HDPE and non-plastic media, and blinded sample tests confirmed that the approach could identify plastics within laboratory-prepared mixed matrices. The authors noted that the

integration of deep learning enabled automated segmentation and deformation analysis, reducing reliance on manual measurements, while constraints remained, including variability from particle size and shape, deviations between manual and computational results, and noise in the automated dataset.⁸⁸

3.3 Biosensor platforms

Beyond optical contrast, receptor-based sensors encode binding kinetics and selectivity in electrical and optical transduction signals that machine learning can disentangle across particle size and chemistry.

Researchers developed a peptide sensor-based strategy combined with machine learning for the identification of acrylate- and methacrylate-type polymeric nanoparticles in water.⁸⁹ Short peptides conjugated with a microenvironment-sensitive fluorophore produced distinct fluorescence spectra upon interacting with nanoparticles of different polymeric compositions. Both supervised (*i.e.*, linear discriminant analysis) and unsupervised (*i.e.*, principal component and hierarchical cluster analyses) machine learning algorithms were applied to these signal patterns, enabling discrimination between homopolymeric and copolymeric nanoparticles. The authors noted that while the method could differentiate nanoparticles with slightly different chemical structures, the study was conducted on as-synthesized polymeric nanoparticles with controlled sizes and compositions, which may not fully capture the heterogeneity and weathering of environmental nanoplastics, and that the relatively simple linear algorithms employed may face limitations when applied to more complex or noisy datasets.⁸⁹

A proof-of-concept sensor was presented for the selective identification of PS and PMMA particles at both the nanoscale (100 nm) and microscale ($20\text{ }\mu\text{m}$) by coupling a plasmonic probe with supervised machine learning.⁹⁰ The sensor was based on an ER-SPR-POF platform, in which estrogen receptors (ER) were immobilized on a gold-coated plastic optical fibre (POF) to induce a surface plasmon resonance (SPR) response. Shifts in the plasmonic resonance wavelength, arising from binding interactions between plastic particles and the receptor, were analysed as input features for a predictive machine learning model built with MATLAB's Statistics and Machine Learning Toolbox. This approach enabled classification of both polymer type and particle size. To demonstrate environmental applicability, the system was further tested in simulated seawater spiked with PMMA nanoplastics.⁹⁰

3.4 Mass spectrometry

Mass spectrometry converts particles to diagnostic fragments and time-profiles, where machine learning organizes these high-dimensional signatures for class discovery and screening.

Researchers showcased fast chemical screening of plastic particles ranging from 500 nm to $20\text{ }\mu\text{m}$ in aerodynamic diameter by combining chromatography-free thermal



desorption and pyrolysis mass spectrometry with unsupervised and semi-supervised machine learning.⁹¹ After acquiring polymer-specific fingerprints and time-resolved release profiles, the authors first applied principal component analysis (PCA) for data reduction and projected the spectra onto the resulting components. They then employed Gaussian mixture models (GMM) and fuzzy *c*-means (FCM) clustering to differentiate, cluster, and interpret the mass spectral data. The study was conducted on controlled nano- and microplastic suspensions generated in

the laboratory, without direct extension to environmental samples.⁹¹

3.5 Cross-cutting themes, limitations, and outlook

Machine learning is improving consistency and throughput across spectroscopy, imaging, biosensing, and mass spectrometry, but several shared bottlenecks determine real-world performance. Synthesizing insights from selected studies together with recent reviews, we discuss existing

| | | | | | | |
|----------------------------|------------------------|----------------------------|--|--------|-----------------------------|------------------------------|
| Liu et al., 2024 [75] | 1 | 1 | 1 | 0 | 0 | 1 |
| Herb et al., 2025 [76] | 0 | 1 | 0 | 0 | 0 | 0 |
| Li et al., 2025 [77] | 0 | 0 | 0 | 1 | 1 | 0 |
| Tian et al., 2022 [78] | 1 | 1 | 0 | 0 | 0 | 0 |
| Xie et al., 2023 [79] | 0 | 0 | 1 | 0 | 0 | 0 |
| Qian et al., 2024 [15] | 0 | 0 | 0 | 0 | 1 | 1 |
| Gong et al., 2025 [80] | 0 | 1 | 1 | 1 | 0 | 1 |
| Vitali et al., 2024 [81] | 0 | 0 | 0 | 0 | 1 | 0 |
| Meyers et al., 2024 [82] | 0 | 0 | 1 | 0 | 0 | 1 |
| Merlemis et al., 2024 [83] | 0 | 0 | 0 | 1 | 1 | 1 |
| Ye et al., 2024 [84] | 0 | 0 | 0 | 1 | 1 | 0 |
| Han et al., 2024 [85] | 0 | 0 | 0 | 1 | 1 | 0 |
| Su et al., 2024 [86] | 1 | 1 | 0 | 1 | 0 | 0 |
| Jeon et al., 2025 [87] | 0 | 0 | 1 | 1 | 0 | 0 |
| Wang et al., 2024 [20] | 0 | 0 | 1 | 0 | 0 | 0 |
| Stine et al., 2023 [88] | 1 | 1 | 0 | 1 | 1 | 1 |
| Hasegawa et al., 2025 [89] | 0 | 0 | 0 | 1 | 1 | 1 |
| Seggio et al., 2024 [90] | 0 | 0 | 0 | 1 | 1 | 1 |
| Forbes et al., 2023 [91] | 0 | 0 | 0 | 1 | 1 | 0 |
| | Biased / Small dataset | Label noise / Mislabelling | Low signal-to-noise / Confusable polymer types | Spiked | Simple / Simulated matrices | Domain shift / Heterogeneity |

Fig. 5 Risk landscape summarizing limitations and deployment context across machine learning assisted analytical studies of waterborne nano- and microplastics (section 3). Absence of a flag (1 = flagged; 0 = not flagged) does not imply absence of risk, it may indicate the issue was untested or unreported. Categories: biased/small dataset: training or test sets are small, imbalanced, or skewed by polymer, size, weathering, or matrix; label noise/mislabelling: known or suspected mis-annotations by human or database errors; low signal-to-noise/confusable polymer types: weak signals or overlapping signatures leading to class confusion (e.g., PE vs. PTFE; plastics vs. organics); spiked: performance demonstrated on standard or lab-generated plastics rather than original environmental nano- and microplastics; simple/simulated matrices: no validation in real environmental waters (e.g., only spiked tap waters, artificial seawater, and lab suspensions); domain shift/heterogeneity: performance sensitive to real-world variabilities (e.g., weathering, additives, and biofilms) and to shifts across instruments and matrices.



challenges and propose a forward-looking agenda grounded in demonstrated practice.

3.5.1 Data realism, standardization, and model robustness. Dominant sources of error in machine learning-assisted analytical workflows are data scarcity, label noise, and domain shift. Models are sometimes trained on small, biased datasets with inconsistent annotations, and then face shifts between pristine and weathered particles and between laboratory and environmental matrices (Fig. 5). For instance, Liu *et al.* noted biased environmental spectra and unrealistic augmentations for FTIR;⁷⁵ Herb *et al.* uncovered systematic mislabels in public FTIR data sets;⁷⁶ Meyers *et al.* documented performance drops on weathered *versus* pristine particles;⁸² Gong *et al.* reported false positives at low signal-to-noise ratios in Raman mapping.⁸⁰ Researchers urge standard operating protocols, richer metadata, shared repositories, and multi-metric reporting (*e.g.*, precision and sensitivity), since accuracy alone varies widely and can be misleading under class imbalance.⁷⁴ Furthermore, there is a need of open, versioned libraries spanning weathering states, additives/pigments, biofilms, and matrices, with paired chemistry and morphology labels and clear guidance for calibration transfer.^{34,73}

In addition, pre-processing is decisive for model robustness. In spectral acquisition and downstream analysis, baseline correction, denoising/cosmic-ray removal, spectral windowing/derivatives, and dimensionality reduction (*e.g.*, PCA) determine both performance and runtime. When tuned properly, PCA can deliver large speedups with minimal information loss. For example, pairing PEER with a random forest stabilized low-SNR Raman spectra,⁷⁹ while fingerprint-windowed CNNs accelerated mapping at short dwell times.⁸⁰ Given pervasive domain shifts (*i.e.*, instrument, substrate, and matrix differences), safeguards such as transfer learning, calibrated uncertainty with abstention, and out-of-distribution checks are essential, particularly for confusable polymer pairs noted in spectroscopy studies.^{15,79,80} Finally, researchers emphasize interpretability and physics-aware cues, and note that with sound pre-processing, simpler spectral models can rival heavier architectures while being easier to validate and deploy.^{20,34,72,88,91}

3.5.2 Nanoscale frontier and field deployment. Nanoplastic detection remains the most challenging frontier compared with microplastic analysis. Beyond inherent analytical limitations, weak or shifted peaks, small cross-sections, and background interference hinder single-particle identification, underscoring the need for tailored pre-processing, denoising, and data fusion.³⁴ The SRS platform achieved detection below 100 nm yet revealed compositional and morphological heterogeneity that complicates spectral matching,¹⁵ while AI-assisted nano-DIHM required size corrections as features approached diffraction and reconstruction limits.²⁰ Accordingly, researchers call for nanoplastic focused databases, improved signal-to-noise approaches, and rigorous validation of substrate/

enhancement strategies.³⁴ In parallel, edge deployment (*e.g.*, smartphone fluorescence with on-device models⁸⁴) is becoming feasible. However, researchers caution that calibration transfer, drift management, and environmental interferences must be engineered alongside the analytics, and human oversight remains essential as systems transition to routine monitoring.^{72,73}

4. Fate and transport models of waterborne nano- and microplastics

The numerical modelling of nanoplastic fate and transport remains in its early stages compared with that of microplastics, primarily because the dominant governing processes differ across scales. At the nanoscale, particle behaviour is controlled by Brownian motion, molecular diffusion, and surface-related interactions.⁹² In contrast, at the microscale, transport is dominated by physical processes such as gravitational settling, hydrodynamic drag, shear-induced transport, and inertial dynamics, which are strongly influenced by particle size, shape, and density, as well as interactions with the surrounding fluid and other particles. These mechanisms result in distinct transport behaviours across different particle size classes.^{93,94}

4.1 Existing models for microplastic studies

The transport of microplastics in water is primarily described by the advection–diffusion equation, which forms the mathematical foundation of most fate and transport models. Numerical solutions to this equation enable the prediction of the spatial and temporal evolution of particle concentrations and trajectories. Because water acts as the carrier medium, hydrodynamic parameters, including flow velocity, water depth, turbulence intensity, and density stratification, must be prescribed. Consequently, fate and transport models are typically coupled with hydrodynamic solvers such as Delft3D,⁹⁵ MIKE 21,⁹⁶ TELEMAC,⁹⁷ or computational fluid dynamics (CFD) models such as OpenFOAM,⁹⁴ which compute the detailed flow fields and turbulence characteristics required for particle transport simulations.^{98–103}

In the context of numerical modelling, the advection–diffusion equation can be solved using either Eulerian or Lagrangian approaches. Eulerian models represent particle concentrations as continuous fields that evolve over space and time. They are particularly suitable for large-scale or long-term simulations, such as those of coastal, estuarine, or riverine systems, where the objective is to capture overall concentration patterns rather than the trajectories of individual particles. Processes such as aggregation, degradation, fragmentation, and sedimentation can be incorporated through source–sink or reaction terms in the governing equations.^{94,102,104}

Lagrangian models, in contrast, track the trajectories of individual particles or particle parcels as they move through



the flow field. This approach explicitly accounts for variability in particle properties, such as size, density, and shape, and captures stochastic particle–flow interactions. The Lagrangian framework is particularly effective for resolving three-dimensional transport, deposition, and resuspension processes in dynamic flow environments.^{105,106}

Because microplastics behave as discrete particles with negligible molecular diffusion and are insoluble, and because modelling objectives often focus on particle-scale metrics, including residence time, accumulation zones, and transport pathways, the Lagrangian or particle-tracking approach has become the dominant framework in the literature. Recent advances in parallel and graphics processing unit (GPU)-based computing have further reinforced this trend by significantly improving computational efficiency and enabling the simulation of millions of particles.

Eulerian models however are generally impractical for simulating large domains or long time periods because the governing equations must be solved for all computational mesh nodes at every time step, even in regions where particles are absent, making them computationally inefficient for sparse particle distributions. The Lagrangian framework, on the other hand, allows direct computation of integral transport parameters that are critical for understanding environmental behaviour, such as settling, residence time, accumulation zones, and source–sink connectivity.^{100,107}

The modelling capabilities and methodologies, particularly for kinematic or advection–diffusion based approaches, which remain the most widely used, have been comprehensively reviewed in several recent studies.^{94,103,108} Despite variations in the employed hydrodynamic models or minor differences in numerical formulations, these models are fundamentally governed by the same physical principles described above.

Beyond kinematic models, more advanced dynamic approaches solve the full particle equation of motion to represent complex particle dynamics. These models explicitly incorporate forces such as drag, gravity, buoyancy, and lift, thereby resolving behaviours such as settling, resuspension, inertial lag, particle transformation, and inter-particle interactions.^{104,109} Dynamic models solve a force balance for each individual particle, enabling explicit representation of a wide range of physical and physicochemical forces acting on microplastics, including inter-particle collisions, particle–bed interactions, electrostatic and surface forces, and changes in particle properties such as size, density, and surface roughness based on ambient environment properties.¹¹⁰

Such detailed models provide valuable insights into mechanisms governing microplastic transport, aggregation, and retention, particularly in near-bed regions and sediment–water interfaces.^{104,111,112} For instance, vertical transport plays a crucial role in determining microplastic residence time and spatial distribution. In kinematic models, this effect can only be incorporated indirectly through empirical settling formulations derived from laboratory experiments. In two-dimensional frameworks, settling velocity is often estimated

from empirical relationships based on particle density and shape, whereas in three-dimensional models, it is represented as the resultant of the water's vertical velocity and the particle's terminal settling velocity. These formulations provide an approximate description of vertical motion.^{94,109,113}

The main challenge of computer models lies in the representation of complex environmental processes, including transformation (*e.g.*, biofouling, UV weathering, and aggregation^{114–118}), turbulent dispersion,^{119–121} settling velocity,^{122–124} and sedimentation/resuspension.^{125–127} Representing these processes in simplified yet physically meaningful ways, so they can be feasibly incorporated into numerical models without excessive computational cost, remains a major modelling limitation.

In this regard, progresses in analytical advances play a critical role. High-quality observational data obtained from these techniques provide essential information for parameterizing key processes and improving model calibration and validation, ultimately enhancing the predictive accuracy of numerical simulations.

4.2 Opportunities and challenges in modelling nanoplastic particles

Fate and transport models that explicitly represent the underlying physical and chemical mechanisms for nanoplastics in natural waters are still sparse. Compared with microplastics, nanoscale plastic particles are governed by Brownian diffusion, rapid homo- and hetero-aggregation, and fast-evolving surface chemistry such as eco-coronas and photo or thermal aging. Mesocosm studies already point to strong retention and sediment coupling of gold-doped PS nanoparticles in relatively simple freshwater systems.¹²⁸ Meanwhile, electrolyte composition and natural organic matter can either stabilize suspensions or promote aggregation, depending on ionic conditions and particle aging.^{129,130} These processes modify particle size, density, and surface charge, which in turn control advection–diffusion, water–sediment exchange, and bioavailability.³⁸ Improving model feasibility therefore requires careful treatment of these coupled transformations (Fig. 6).

For bulk transport of primary nanoplastics, longitudinal displacement is governed by advection, while cross-sectional mixing is typically set by shear and turbulence. Within this hydrodynamic regime, Brownian motion dominates the microscale collisions that initiate aggregation. A standard 2D/3D advection–diffusion–reaction (ADR) framework is thus appropriate, with advection and turbulent diffusion resolving transport, and Stokes–Einstein Brownian diffusion informing both collision kernels and near-interface mass transfer.¹³¹ Model realism depends on accurate hydrological and geochemical inputs and on regularly updating particle physicochemical parameters as they evolve. Interactions with natural colloids (*e.g.*, clays, metal oxides, and biogenic particles) and weathering from light-driven oxidation shift particle size and surface chemistry, often increasing particle





Fig. 6 Mechanistic factors on nanoplastic fate and transport in natural waters for model simulations. Hydrodynamics and Brownian collisions set encounter rates (left). Transformations, aggregation with natural colloids and weathering, together with geochemical factors update particle properties (middle). The evolving properties regulate advection–diffusion–reaction transport and near-interface mass transfer, water–sediment exchange, and bioavailability (right). Settling of primary nanoplastics is neglected initially and activated when aggregates reach the micrometre scale. Arrows indicate influence.

attachment efficiency. As attachment efficiency is also sensitive to salinity, divalent cations, and natural organic matter, these factors together determine nanoplastic fate and transport.^{132–134} Gravitational settling of primary nanoplastics can usually be neglected at first and activated dynamically once aggregates grow to microscale sizes.^{135,136}

Several tools from engineered nanomaterial modelling offer practical starting points. The spatially explicit NanoDUFLOW river model and the multimedia SimpleBox4Nano model both represent aggregation with natural solids and track particle size distributions.^{135,137} With appropriate re-parameterization for plastics, such as no dissolution and buoyancy and density that evolve with sorption and corona formation, these tools can be adapted to nanoplastics. As a screening complement across nano- and microplastics, SimpleBox4Plastic supports scenario analysis for nanoplastics in natural waters.¹³⁸

5. Future directions

Safeguarding water quality in the age of nanoplastics demands a decisive shift from isolated demonstrations to interoperable, rigorously validated, and scalable systems that seamlessly integrate measurements, models, and management. Building on the advances discussed in this review, we chart a practical near-term agenda anchored in

mutually reinforcing priorities designed to accelerate innovation and real-world impact.

5.1 Converging on interoperable protocols and reporting standards

Fragmented approaches to sampling, extraction, spectral preprocessing, and classification undermine comparability across studies and slow the adoption of robust methods. Without minimum reporting standards, uncertainty remains opaque, and meta-analyses lack reliability, creating barriers to evidence-based regulation.

To close this gap, the scientific community should establish a minimum information checklist for reporting nano- and microplastic analytics aligned with ISO 24187 and emerging guidance from the European Committee for Standardization (CEN) and ISO. This checklist should, where applicable, specify sample origin, pretreatment, size and shape bins, polymer classes, and quality control measures, including field and procedural blanks, recovery surrogates, and detection and quantification limits for each polymer and matrix. It should also include model architecture and training data for machine learning approaches, as well as methods for uncertainty quantification. Standardized reporting is critical for translating research into actionable policies, enabling regulators to set enforceable limits,



monitor compliance, and protect ecosystems and public health in the face of growing nanoplastic risks.

5.2 Environmentally realistic reference materials and interlaboratory trials

There are key challenges associated with developing environmentally realistic reference materials and conducting interlaboratory trials. For example, validation using pristine, monodisperse beads often overestimates analytical performance and obscures failure modes that arise with weathered polymers, additives, biofilms, and heterogeneous matrices. This discrepancy limits the reliability of current detection and quantification methods for nano- and microplastics in real-world environments. To address these challenges, we propose two systematic objectives: (a) develop and certify environmentally realistic reference materials that accurately reflect actual environmental conditions; and (b) conduct blinded interlaboratory comparisons to benchmark analytical methods under realistic scenarios.

For the production of reference materials, we recommend generating size-fractionated nano- and microplastics that incorporate controlled ultraviolet, thermal, and oxidative weathering, as well as representative additives and ecocoronas. These materials should be embedded in matrices representative of freshwater, estuarine, wastewater, and drinking water environments. Interlaboratory trials should implement blinded comparisons across spectroscopy, mass spectrometry, imaging, and electrochemical techniques. Key performance metrics to be reported include recovery rates, false positive and false negative rates, size-dependent bias, and robustness to matrix variability, among others. By pursuing these objectives, we anticipate several positive outcomes: (a) standardized, certified reference materials tailored to realistic environmental conditions; (b) comprehensive performance benchmarks for analytical methods; and (c) improved reliability and reproducibility in the detection of nano- and microplastics.

5.3 Shared datasets and comparisons for machine learning assisted studies

Comparisons across machine learning assisted studies remain challenging when groups rely on different datasets and scoring rules. A practical solution emphasizes shared datasets, shared tasks, and standardized reporting. Community datasets for IR, Raman, SERS, SRS, O-PTIR, and AFM-IR, along with particle-level images are needed. Each record should include accurate labels and a brief note on sample origin and preparation. Upon publication, associated data and code should be deposited in open repositories to enable reuse.

Common tasks should include polymer identification, size bin classification, weathering stage recognition, and out-of-scope detection, with fixed train/validation/test splits. Additional challenge subsets spanning instruments, matrices, and weathering states should be considered to assess

robustness under real-world variation. Baseline, reproducible workflows with unit tests and clear documentation should serve as reference points that others can rerun.

Reporting should cover both standard scores (*e.g.*, accuracy and precision) and calibrated uncertainty (*e.g.*, calibration curves and prediction intervals), allowing readers to evaluate predictions alongside their associated confidence levels.

5.4 Routine uncertainty quantification and traceability

Accurate characterization of microplastic contamination and polymeric material distribution is critical for informed decision-making in product design, regulatory permitting, and environmental remediation. These decisions increasingly require defensible, quantitative estimates of concentration, flux, and source attribution that are expressed through credible intervals rather than single-point values. Point estimates, while common, fail to capture the inherent variability and uncertainty associated with environmental sampling and analytical workflows, leading to potential misinterpretation of risk and exposure.

To address these challenges, comprehensive end-to-end uncertainty quantification frameworks are essential. Such frameworks must propagate error across all stages of analysis, beginning with sampling and extending through extraction losses, instrumental signal-to-noise limitations, and classification ambiguity. This propagation ensures that uncertainty is not underestimated when translating raw measurements into particle-based and mass-based concentration estimates. Sampling uncertainty, in particular, must explicitly account for spatial and temporal variability and recovery efficiency, given the heterogeneity of environmental matrices and the dynamic nature of contaminant distributions.

Standardization of reporting practices is equally critical for comparability and reproducibility across studies. Confidence intervals at defined levels should accompany particle counts and mass estimates, while polymer-resolved misclassification matrices should be incorporated to quantify classification error. Furthermore, sensitivity analyses evaluating the influence of preprocessing choices on final outcomes are necessary to identify methodological biases and improve robustness. Establishing these rigorous protocols will not only enhance scientific transparency but also strengthen regulatory frameworks and risk assessments, ultimately supporting evidence-based strategies for pollution mitigation and sustainable material management.

5.5 Field testbeds that connect sensors, analytics, and models to decisions

To move beyond laboratory demonstrations, we propose establishing real-world testbeds that integrate continuous monitoring, automated data processing, and catchment-scale modelling into a single operational pipeline. These



pilots will demonstrate how advanced sensing and analytics can deliver actionable insights for water safety and resource management. Field instruments will stream data through open interfaces into validated analysis workflows, which then feed hydrological and water-quality models. The combined system will produce decision-ready outputs, including particle number and mass fluxes, removal efficiency across treatment steps, threshold exceedance frequency, and event detection, accompanied by calibrated uncertainty and confidence levels.

This end-to-end integration moves beyond fragmented research toward deployable solutions. Testbeds will be implemented along critical pathways, from wastewater outfalls to downstream drinking-water intakes across diverse contexts, including low-resource settings. Each deployment will include side-by-side confirmation with reference methods, routine checks on false positives and false negatives, and stress testing under storms, maintenance upsets, and seasonal change. Performance will be judged on uptime, latency from sample to result, accuracy with quantified uncertainty, false-alarm rate, cost per decision, and ease of integration with utility workflows. Outputs will include shared data schemas, reference implementations of the analysis workflows, and concise playbooks that show how to choose methods for screening, compliance monitoring, incident response, and product stewardship. By proving the full measurement-to-action pathway in the field, these testbeds will convert scattered advances into tools that operators and regulators can deploy at scale, accelerating progress toward resilient water systems and safeguarding public health.

5.6 Translation of science to policy, procurement, and practice

Ongoing treaty negotiations and national or regional regulations require practical, standardized metrics and methods that authorities and industry can implement effectively. It is important to link analytical results to policy-relevant indicators, polymer-resolved particle counts and mass fluxes, removal efficiency with its uncertainty, the frequency of threshold exceedances, and the confidence in source attribution. Short guidance should be co-developed with regulators, utilities, and industry on choosing methods for screening, compliance monitoring, incident response, and product stewardship, with decision trees keyed to matrix, concentration range, and available resources. Build in equity and ethics by ensuring reference materials and test sets reflect low-resource contexts and by supporting capacity building and open tools to avoid dependence on a single technology.

Together these steps shift the field from disparate demonstrations to a coherent, reproducible, and scalable evidence base. By making methods work together through

common standards, the community can ensure that rapid advances in nanoscale analytical capability, machine learning, and modelling translate into timely and trusted guidance for protecting water quality across the plastic life cycle.

5.7 New technologies, improved methods, more accurate analyses, enhanced models, and the use of AI for predicting hot spots and advancing remediation strategies

Rapid advancements in technology are transforming environmental monitoring and remediation. To meet growing regulatory and sustainability demands, new tools and methods must deliver higher accuracy, predictive capabilities, and adaptive solutions. Innovations in nano-sensor design, sampling techniques, and laboratory instrumentation are enabling ultra-trace detection and real-time monitoring. These improvements reduce measurement uncertainty and provide more reliable data for compliance and risk assessment.

Next-generation models now integrate physical, chemical, and biological processes to better represent complex environmental systems. These models support scenario testing, improved forecasting, and more accurate estimation of pollutant transport and transformation.

Artificial intelligence (AI) is enhancing decision-making by leveraging high-resolution data and predictive analytics to enable proactive interventions. AI-driven prioritization ensures efficient allocation of monitoring and remediation resources. Open-source tools and capacity-building initiatives can help prevent technological monopolies and support low-resource regions.

Yet, significant challenges remain. Ethical and social concerns include bias and discrimination, which can emerge from demographic, geographic, and linguistic skews in training data and evaluation benchmarks, causing models to reflect dominant cultural norms and to underperform for underrepresented groups. Additional risks include privacy violations, limited transparency, and potential job displacement due to labour market shifts. Technical challenges involve robustness, reliability, security risks, and generalization limits, as many AI models struggle to perform consistently across diverse contexts without retraining. Environmental impacts are also notable, as the high energy consumption contributes to carbon emissions and ecological degradation.

Therefore, detection and remediation approaches must be carefully evaluated. It is worth remembering that technologies initially developed for sustainability, such as plastics or fluorochlorocarbons, later revealed unintended consequences, including ozone depletion. Similarly, the rapid expansion of industrial infrastructure can trigger unintended environmental burdens, such as water depletion, further exacerbating water scarcity in regions already under hydrologic stress. Anthropogenic activities continue to affect climate and planetary health.



AI must be applied responsibly and with purpose, not in vain.

Conflicts of interest

The authors declare that they have no known conflicts of interest that could influence this work.

Abbreviations

| | |
|-----------|---|
| ADR | Advection–diffusion–reaction |
| AF4 | Asymmetric flow field-flow fractionation |
| AFM | Atomic force microscopy |
| BSA | Bovine serum albumin |
| CEN | European Committee for Standardization |
| CPE | Carbon paste electrode |
| CSRR | Complementary split ring resonator |
| DAD | Diode-array detector |
| DEP | Dielectrophoresis |
| EC | Electrochemical |
| EDX | Energy dispersive X-ray |
| EIS | Electrochemical impedance spectroscopy |
| FePDA | Iron-doped polydopamine |
| FLA | Fluorescence lifetime analysis |
| HCA | Hierarchical cluster analysis |
| HDPE | High-density polyethylene |
| IPD | Integrated passive device |
| IR | Infrared |
| ISO | International Organization for Standardization |
| LDPE | Low-density polyethylene |
| LFOC | Laser-backscattered fiber-embedded optofluidic chip |
| LIBD | Laser-induced breakdown detection |
| LOD | Limit of detection |
| L-PHA | L-Phenylalanine |
| MALS | Multi-angle light scattering |
| PCB | Printed circuit board |
| PE | Polyethylene |
| PEC | Photoelectrochemical |
| PEG | Polyethylene glycol |
| PET | Polyethylene terephthalate |
| PLA | Polylactic acid |
| PMMA | Polymethyl methacrylate |
| PP | Polypropylene |
| PS | Polystyrene |
| PVA | Polyvinyl alcohol |
| PVC | Polyvinyl chloride |
| Py-GC-MS | Pyrolysis-gas chromatography mass spectrometry |
| QCM | Quartz crystal microbalance |
| SEM | Scanning electron microscopy |
| SERS | Surface-enhanced Raman spectroscopy |
| SPR | Surface plasmon resonance |
| SSBD | Shrinking surface bubble deposition |
| TD-PTR-MS | Thermal-desorption proton-transfer-reaction mass spectrometry |
| UV-vis | Ultraviolet-visible |
| ZIF | Zeolitic imidazolate framework |

Data availability

Data available upon reasonable request.

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