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Catalyst-free synthesis of novel 1,5benzodiazepines and 3,4-dihydroquinoxalines using isocyanide-based one-pot, three- and fourcomponent reactions†

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Reaction of benzimidazolone derivatives, or their thio- or aza-counterparts, with an isocyanide in the presence of acetone unexpectedly gave rise to novel tricyclic benzodiazepine derivatives in good yield by means of a four-component reaction incorporating two moles of acetone. Benzimidazole starting substrates bearing an electron-withdrawing group gave rise instead to dihydroquinoxaline derivatives by means of a three-component reaction. Use of deuterated acetone instead of acetone in the reactions significantly affected yield and reactivity in the four-component reaction but not in the three-component reaction.

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Introduction

One of the vital goals in modern synthetic chemistry approaches is adhering to the practice of green and sustainable chemistry by championing environmentally safe and atom-efficient, economical chemical processes.1 Recently, there has been increased interest from academia and industry in the use of multicomponent reactions (MCRs) as a result of their potential contribution to sustainable chemistry.2 MCRs also continue to attract attention in synthetic chemistry due to their ability to generate diversity and structural complexity in a single step.3 MCRs possess multiple advantages compared to conventional linear-step syntheses, for example: by reducing reaction times, using commercially accessible and cheaper reagents, their high atom economy and their easy-to-use, green experimental methods where the use of hazardous solvents and catalysts can be avoided. This results in both economic and environmental benefits.4-8

One of the benefits of the isocyanide functional group in MCRs is its unique reactivity; thus isocyanide-based MCRs (IMCRs) are considered amongst the most versatile MCRs to generate diversity and structural complexity. The use of isocyanides for the synthesis of nitrogen-containing heterocyclic compounds such as benzodiazepines and quinoxalines 15-18

is well exploited. For example, the reaction of *o*-phenylenediamine **1** and isocyanide **2** in the presence of two equivalents of acetone **3a** using ammonium chloride as a catalyst was reported to give benzodiazepine-2-carboxamides **4** as shown in Scheme **1a**. It was also reported that the reaction of *o*-phenylenediamine **1** and isocyanide **2** in the presence of one equivalent of acetone **3a**, using *p*-TsOH as a catalyst and ethanol as a solvent at room temperature gave rise to 3,4-dihydroquinoxalin-2-amines **5** (Scheme **1b**). Is

Benzodiazepines and quinoxaline are accessible, easy to functionalise and have a wide range of potential pharmacological applications. For example, benzodiazepines have been found to have antimicrobial and anthelmintic²⁰ activity. They also have antitumour²¹ activity and can act as calcium channel blockers,²² anti-HIV²³ agents and also have cytotoxic²⁴ properties. On the other hand quinoxalines show pharmacological properties such as antibiotic,²⁵ antimicrobial,²⁶ and antidiabetic²⁷ activity. They have also been shown to be pesticidal²⁸ and to inhibit aldose reductase^{29,30} There are several procedures

Scheme 1 Reported methods for the synthesis of (a) 1,5-benzodiazepine-2-carboxamides 4 and (b) 3,4-dihydroquinoxalin-2-amines 5.

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reported for the synthesis of benzodiazepines which mainly include the use of catalysts. The catalysts or stoichiometric reagents employed include zeolites, 31 BF3 · OEt2, 32 Sc(OTf)3, 33 MgO/POCl₃,³⁴ H-MCM-22,³⁵ FeAlP-550,³⁶ H₃PMo₁₂O₄₀ and H₃PW₁₂O₄₀,³⁷ CH₃COOH³⁸ and ZrOCl₂,³⁹ amongst others. These methods pose drawbacks such as high temperatures, generation of heavy metal waste, hazardous solvents, long reaction times and relatively expensive reagents.

In our ongoing studies on the use of multicomponent reactions for the synthesis of diverse heterocycles44 we initiated an investigation of the reaction of benzimidazole derivatives with dimethyl acetylenedicarboxylate (DMAD) and isocyanides and unexpectedly prepared a range of diazepine and quinoxaline derivatives.

Results and discussion

Our investigation began with the reaction of tert-butyl isocyanide **2a** ($\mathbb{R}^3 = t$ -butyl), 1*H*-benzo[*d*]imidazol-2(3*H*)-one **6a** and dimethyl acetylenedicarboxylate (DMAD) 7a in acetone 3a, as the solvent. We expected to obtain one or both of two likely

products: compound 8, resulting from a three-component reaction or compound 9, from a two-component reaction (Scheme 2). This expectation was based on literature precedent as Zeng and co-workers40 reacted 1H-benzo[d]imidazole-2(3H)thione 6g (X = S) and DMAD using methanol as solvent to obtain compounds 9. Adib and co-workers, 41 on the other hand, treated imidazole derivative 4,5-diphenyl-1,3-dihydro-2Himidazol-2-one, with an isocyanide and DMAD 7a to produce the expected 5H-imidazo-[2,1-b][1,3]oxazine derivatives using acetone as solvent. Much to our surprise, reaction of benzimidazolone 6a, isocyanide 2a and DMAD 7a in acetone as solvent yielded only compound 10 in racemic form (Scheme 2). The reaction had unexpectedly included the solvent acetone 3a to form a diazepine-based scaffold, while DMAD 7 did not participate in the reaction. Interestingly, when the reaction was repeated using benzimidazoles 6g and 6h containing either nitrogen or sulfur at position X these reactions were also able to produce compounds 10 (Scheme 2). Reaction of unsubstituted benzimidazole derivatives $(R^1, R^2 = H)$ or those bearing the electron-releasing methyl group $(R^1, R^2 = CH_3)$ gave compounds 10 in generally good yields (67-84%) with a range of

Scheme 2 Synthesis of benzodiazepines and dihydroquinoxalines.

isocyanides, including *t*-Bu **2a**, tetramethylbutyl **2b**, cyclohexyl **2c** and tosylmethyl **2d** isocyanides. Reaction of benzimidazole derivatives bearing weakly deactivating groups such as carboxylate ester, fluoro or chloro under the same conditions gave no product. Replacing acetone **3a** with 3-pentanone **3b** in reaction with benzimidazoles **6b-f** and isocyanides **2a-c** also resulted in no conversion. Thus it appears that deactivating groups on **6c-d** on the aromatic ring strongly disfavoured formation of product **10**. In addition, the terminal methyl or the methylene group of 3-pentanone **3b** possibly sterically hindered the cyclisation reaction to form diazepine **10**.

To examine the role of DMAD, the reactions were repeated without addition of this reagent and the same products 10a-f were isolated (Table 1), with no impact on yield when compared to the initial method. Interestingly, the omission of DMAD from the reactions of benzimidazoles 6 bearing electron-withdrawing substituents resulted in formation of dihydroquinoxaline products 11a-g (Table 1). Synthesis of these dihydroquinoxalines 11 was achieved without the use of a catalyst, whereas previous reports on the preparation of similar quinoxaline compounds required the presence of various catalysts. 15-18 Several reactions were tested using 3-pentanone 3b in place of acetone 3a and only product 11h was obtained in 76% yield. Attempts to obtain these dihydroquinoxalines 11 directly from 1,2-phenylenediamine derivatives 1 instead of benzimidazoles 6 without the use of a catalyst were not successful and thus starting with benzimidazole 6 is key to the success of the reaction, although the important role played by the C=O, C=N or C=S moiety is not yet understood. The best dihydroquinoxaline 11 yields were obtained for di-substituted substrates. For example, reaction of 5,6-dichloro-1*H*-benzo[d]imidazol-2(3*H*)one 6c resulted in good yields of product 11f (74%) and 11g (76%) from *tert*-butyl and tetramethylbutyl isocyanides, respectively, using acetone **3a** (Table 1). Mono-substituted deactivating groups at either position 6 or 7 such as chlorine, carboxylate ester or fluorine resulted in low to average yields (22–52%). It is worth mentioning that unsymmetrical starting materials **6** may give rise to the possibility of regioisomers. We obtained only one of the two possible regioisomers of products **11a**, **11b** and **11e** which contain one H and either an ester or Cl group at positions R¹ and R². However, when using **6e** containing an F and H at positions R¹ and R², we were able to isolate two regioisomers after the reaction, but we were unable to distinguish them (**11c** and **11d**).

Based on the proposed reaction mechanisms, rationalisation of the formation of the two different products is possible. For benzimidazole substrates 6 that are unsubstituted or with

Scheme 3 Proposed reaction mechanism for the formation of 10.

Table 1 Catalyst-free synthesis of 1,5-benzodiazepines and 3.4-dihydroquinoxalines excluding DMAD^b

Starting material	X	Product	R^1	\mathbb{R}^2	\mathbb{R}^3	\mathbb{R}^4	Yield%
6a	О	10a	Н	Н	<i>t</i> -Bu	CH_3	81
6g	S	10a	H	H	<i>t</i> -Bu	CH_3	83
6h	N	10a	H	H	<i>t</i> -Bu	CH_3	84
6a	О	10b	Н	H	Tetramethylbutyl	CH_3	77
6g	S	10b	H	H	Tetramethylbutyl	CH_3	79
6h	N	10b	Н	H	Tetramethylbutyl	CH_3	80
6a	О	10c	Н	H	Cyclohexyl	CH_3	80
6g	S	10c	Н	Н	Cyclohexyl	CH_3	82
6h	N	10c	Н	Н	Cyclohexyl	CH_3	84
6h	N	10 d	Н	Н	TosMe	CH_3	71
6b	O	10e	CH_3	CH_3	<i>t</i> -Bu	CH_3	74
6b	О	10f	CH_3	CH_3	Tetramethylbutyl	CH_3	67
6d	O	11a	CO_2CH_3	Н	<i>t</i> -Bu	CH_3	46
6d	О	11b	CO_2CH_3	Н	Tetramethylbutyl	CH_3	52
6e	O	$\mathbf{11c}^{a}$	F	Н	Tetramethylbutyl	CH_3	30
6e	О	$\mathbf{11d}^a$	Н	\mathbf{F}	Tetramethylbutyl	CH_3	22
6f	О	11e	Cl	Н	Tetramethylbutyl	CH ₃	46
6c	O	11f	Cl	Cl	<i>t</i> -Bu	CH_3	74
6c	O	11g	Cl	Cl	Tetramethylbutyl	CH_3	76
6h	N	11h	Н	Н	<i>t</i> -Bu	CH_3CH_2	76

^a Compounds **11c** and **11d** were isolated from the same reaction. ^b Reaction conditions: isocyanide **2** (1 eq.), ketone **3** (2 eq.) and benzimidazole derivative **6** (1 eq.) were stirred at room temperature for 24 h.

R1
$$R^{1}$$
 R^{2} R^{3} R^{3} R^{3} R^{4} R^{2} R^{2} R^{2} R^{2} R^{3} R^{4} R^{2} R^{2} R^{4} R

Scheme 4 Benzodiazepine 13 obtained from imine 12.

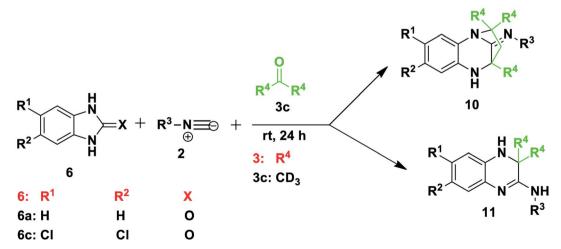
Scheme 5 Possible mechanism for the formation of dihydroquino-

electron-releasing substituents *para* to the reacting nitrogen atoms, both nitrogen atoms are sufficiently nucleophilic to participate in imine formation with acetone and give rise to compound **10** (Scheme 3). Initially, reaction of one of the nitrogen atoms takes place to give the imine, which is followed by ring-opening of the five-membered ring. The resulting isocyanate reacts with water generated *in situ* during imine formation, leading to decomposition of the isocyanate to give an amine and CO₂. Reaction of the this amine with acetone leads to intermediate **A** that then cyclises internally to give intermediate **B**. We are suggesting that some kind of concerted process takes place where the iminium ion moiety of **B** is attacked by the isocyanide **2** while an internal nitrogen nucleophile attacks the developing nitrilium ion to give the final

product **10**. The reason for suggesting a concerted mechanism is that when we prepared and isolated imines **12** (Scheme 4) as possible intermediates in the mechanism, and subjected them to reaction with an isocyanide **2**, no reaction occurred. When we added an acid catalyst such as Montmorillonite K-10 clay to the reaction to first generate the iminium ion of **12**, we were only able to isolate products **13**, with no product **10** being observed at all.

In contrast, for benzimidazole derivatives bearing an electron-withdrawing group only one of the nitrogen atoms reacts with acetone and results in the formation of compound 11 (Scheme 5). Kolla and Lee¹⁵ describe a similar mechanism for formation of 11 starting from o-phenylenediamines, using EDTA as a catalyst. The reaction of compound 6 proceeds as described above for Scheme 3 until formation of imine C that is then directly attacked by the isocyanide to give a nitrilium ion D (Scheme 5). This is then intercepted by an internal nitrogen nucleophile to give the final dihydroquinoxaline 11. Based on this mechanism, it is probable that the initial nucleophilic attack is made by the nitrogen that is meta to the electronwithdrawing group. This makes it likely that the single regioisomer isolated for products, 11a, 11b and 11e, had $R^1 = H$ and R^2 = ester or Cl. This is supported by results obtained by Shaabani et al. for 1,2-phenylenediamines.19 When these workers reacted 1,2-phenylenediamines with an isocyanide and one equivalent of acetone, using NH₄Cl as a catalyst, they isolated dihydroquinoxalines such as 11. Interestingly, when they reacted 1,2-phenylenediamines with an isocyanide and two equivalents of acetone in the presence of NH₄Cl, they isolated benzodiazepine derivatives such as 13, with no tricyclic benzodiazepines 10 being reported from their reaction. This clearly indicates a unique mechanism when starting with benzimidazole derivatives (6) as we have done: no catalyst is required for these reactions and use of certain derivatives gives unique products 10. It is interesting to note that none of our reactions starting from 6 gave alternative benzodiazepine product 13.

The role and involvement of acetone in these reactions was further elucidated by replacing normal acetone 3a with



Scheme 6 Catalyst-free synthesis of benzodiazepines 10 and dihydroquinoxalines 11 using deuterated acetone.

Table 2 Yields obtained in the synthesis of 1,5-benzodiazepines and 3.4-dihydroquinoxalines using deuterated acetone

Starting material	X	Product	R^1	R^2	R^3	R^4	Acetone- d_6 yield% (time)
6a	0	10g	Н	Н	Tetramethylbutyl	CD_3	57 (24 h)
6a	O	10h	Н	Н	Cyclohexyl	CD_3	61 (24 h)
6a	О	11i	H	Н	<i>t</i> -Bu	CD_3	70 (24 h)
6c	О	11j	Cl	Cl	<i>t</i> -Bu	CD_3	72 (24 h)
6c	O	11k	Cl	Cl	Tetramethylbutyl	CD_2	80 (24 h)

^a Reaction conditions: isocyanide 2 (1 eq.), deuterated acetone 3c (2 eq.) and benzimidazole derivative 6 (1 eq.) were stirred at room temperature for 24 h.

Table 3 Yields from catalyst-free synthesis of benzo-1,5-diazepines^a

Product	R^1	R^2	R^3	Yield% A	Yield% B
14a	Н	Н	CH ₃	83	72
14b	Н	Н	CH ₂ CH ₃	79	68
14c	Н	Н	t-Bu	75	62
14d	NO_2	Н	CH_3	92	_
14e	NO_2	Н	CH ₂ CH ₃	85	_
14f	NO_2	Н	t-Bu	80	_
14g	Cl	Cl	CH_3	83	71
14h	Cl	Cl	CH ₂ CH ₃	76	64
14i	Cl	Cl	t-Bu	75	66
14j	CH_3	CH_3	CH_3	62	60

 $[^]a$ Reaction conditions: o-phenylenediamine 1 (1 eq.) or benzimidazole derivative 6 (1 eq.) and acetylenedicarboxylate 7 (1 eq.) were refluxed in ethanol (30 ml) for 12 h.

deuterated acetone (acetone- d_6) **3c** to probe reactivity and trace atom involvement in the formation of benzodiazepine **10** and dihydroquinoxaline **11** products (Scheme 6). The expense of acetone- d_6 restricted us to fewer reactions and a total of 5 reactions were tested. The reactions were set up in the same way as described for acetone **3a** (Table 1). When 1H-benzo[d]imidazol-2(3H)-one **6a** was reacted with acetone- d_6 **3c** and either t-Bu-, tetramethylbutyl- or cyclohexyl isocyanides, two deuterated benzodiazepines **10g**- \mathbf{h} and one dihydroquinoxaline **11i** were isolated (Table 2). Interestingly, reaction of 1H-benzo[d]

imidazol-2(3H)-one 6a and tert-butyl isocyanide 2a with acetone d_6 3c did not give rise to the benzodiazepine product, instead giving only the deuterated dihydroquinoxaline N-(3,3-dimethyl-3,4-dihydroquinoxalin-2(1H)-ylidene)-2-methylpropan-2-amine 11i, in contrast to the result seen for acetone 3a. When 5,6dichloro-1*H*-benzo[d]imidazol-2(3*H*)-one **6c** was reacted with tert-butyl isocyanide 2a or tetramethylbutyl isocyanide 2b in acetone- d_6 3c two dihydroquinoxalines 11j-k were obtained. For the synthesis of partially deuterated benzodiazepines 10g-h the yield obtained was average, hence it appears that the acetone- d_6 was relatively less reactive compared to normal acetone for synthesis of benzodiazepines. However, for the synthesis of the deuterated dihydroquinoxaline derivatives 11i-k acetone-d₆ was relatively reactive and resulted in comparable yields to those obtained using normal acetone. The highest yield was of 10k from reaction of deactivating di-substituted 5,6-dichloro-1Hbenzo[d]imidazol-2(3H)-one 6 \mathbf{c} .

We then moved on to investigate the reactions of benzimidazole derivatives **6** with dialkyl acetylenedicarboxylates 7 as Michael acceptors under reflux conditions and comparing this to the reactions of 1,2-phenylenediamines **1** under the same conditions. In this case, unlike the reactions described earlier, the same results were observed for reaction of the benzimidazole **6** and 1,2-phenylenediamine **1**. In both cases the reactions in refluxing ethanol gave rise to 4-oxo-4,5-dihydro-1*H*-benzo-1,5diazepine-2-carboxylate derivatives **14** (Table 3 and Scheme 7). Solomko and co-workers⁴³ reported the synthesis of a similar product, ethyl 2,3-dihydro-2-oxo-1*H*-1,5-benzodiazepine-4-

1:
$$R^1$$
 R^2 R^1 R^2 R^3 R

Scheme 7 Catalyst-free synthesis of benzodiazepines 14 under ethanol reflux.

carboxylate, from o-phenylenediamine and oxaloacetic ester in o-xylene under reflux. This kind of reaction involves Michael addition/intramolecular cyclization leading to the formation of interesting fused [6-7] diazepine scaffolds. The benzodiazepine-2-carboxylate derivatives 14 were successfully synthesised in various yields according to the different nature of their functional groups from 1,2-phenylenediamines 1 (yield A) and benzimidazoles 6 (yield B) (Table 3). When comparing the reactivity of benzimidazole 6 and 1,2-phenylenediamines 1 towards product formation it was observed that 1,2-phenylenediamines 1 containing free primary amines gave higher yields compared to the benzimidazoles 6. When using 1,2phenylenediamines 1 the highest yield obtained was that of product 14d with a strongly deactivating and electronwithdrawing nitro group at position 8. This might be attributable to the fact that the nitro-products 14d-f readily precipitated out of the reaction mixture as yellow solids, making isolation highly efficient. The lowest yield was obtained for product 14j, di-substituted with the activating and electron-donating methyl group at positions 7 and 8. Substrates di-substituted with the weakly deactivating chlorine atoms at position 7 and 8 gave good yields when using 1,2-phenylenediamine starting materials 1. The reaction of substrates 1 or 6 containing monosubstituted deactivating groups (F, Cl and CO₂CH₃) were not successful, and neither was the reaction of the substrate monosubstituted with the activating CH₃ group. From the outcome it was observed that the reactions were more favoured with a strongly deactivating mono-substituent (nitro), rather than an activating group (methyl) at position 7 or 8. As expected, the unsymmetrical starting material 1i may give rise to the possibility of two regioisomeric products 14d and 14f, however, in each case we obtained only one regioisomer but we were unable to establish unequivocally which one was present.

The structures of the benzodiazepines and dihydroquinoxalines obtained (**10a-h**, **11a-k** and **14a-j**) were elucidated by FTIR spectroscopy, NMR spectroscopy, HRMS and in certain instances these were confirmed using single-crystal X-ray crystallographic analysis (for compounds **10a**, **10e**, **11f** and **11g**). By way of a representative example the spectroscopic data for compound **10a** (Table 1) is given: the ¹H NMR spectrum showed multiplet peaks corresponding to four aromatic CH groups (positions 6–9) at $\delta = 6.69-6.52$, a broad NH singlet at $\delta = 3.90$, two doublets at $\delta = 2.04-2.00$ ppm and 1.79-1.76 ppm corresponding to the CH₂ at position 3, a singlet from CH₃ at

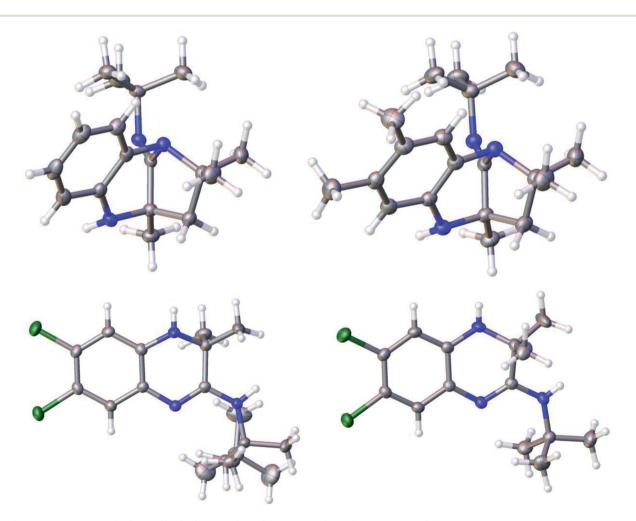


Fig. 1 Crystal structures of 10a (top left), 10e (top right), 11f (bottom right) and (11g bottom left). Carbon atoms are represented as grey spheres, nitrogen atoms as blue spheres, hydrogen atoms as white spheres and chlorine atoms as green spheres.

 $\delta=1.39$, one singlet for three *tert*-butyl CH₃ groups at $\delta=1.28$, and two CH₃ singlets at $\delta=1.16$ and $\delta=1.08$. The ¹³C NMR spectrum of **10a** showed 15 distinct resonances in accordance with the proposed structure. The mass spectrum of **10a** showed the molecular ion [M + H]⁺ peak at m/z 272.2119 which is consistent with the mass of the proposed product for which the calculated value is 272.2121. The FTIR spectrum displayed characteristic absorption bands for the imine group at 1696 cm⁻¹ and for the amine group at 2922 cm⁻¹. Unambiguous evidence for the structures of **10a** and **10e** was obtained from single-crystal X-ray analysis as shown in Fig. 1.

For **11a** the ¹H NMR spectrum showed a doublet at $\delta = 7.51$ – 7.49 with a J value of 8 Hz corresponding to an aromatic CH, a doublet between $\delta = 7.28$ –7.27 with a J value of 1.6 Hz corresponding to CH, a doublet appeared at $\delta = 7.08-7.06$ with a I value of 8.4 Hz corresponding to an aromatic CH, two singlets were found at $\delta = 4.46$ and $\delta = 3.60$ corresponding to two NH groups, a singlet at $\delta = 3.89$ was assigned to the methoxy group, a singlet from the *tert*-butyl group appeared at $\delta = 1.51$ and a singlet from two methyl groups was observed at $\delta = 1.31$. The ¹³C NMR spectrum of 11a showed 13 distinct resonances in accordance with the proposed structure. The mass spectrum of 11a showed the molecular ion $[M + H]^+$ peak at m/z 290.1865 which is consistent with the mass of the proposed product for which the calculated value is 290.1863. The FTIR spectrum displayed characteristic absorption bands for the carbonyl group at 1693 cm⁻¹, imine group at 1616 cm⁻¹ and for the amine group at 2901 cm⁻¹. The structures of 11f and 11g were confirmed by X-ray crystallography (Fig. 1).

For **14a** (Table 3) the ¹H NMR spectrum showed two singlets at δ 11.74 and 11.03 corresponding to two NH groups. Multiplet peaks were found around δ 7.40–7.39 and δ 7.05–7.03 corresponding to four aromatic CH protons. A singlet CH peak at δ 5.52 from position 3 and a singlet methyl peak at δ 3.68 were also observed. The ¹³C NMR spectrum of **14a** showed 11 distinct resonances in accordance with the proposed structure. The mass spectrum of **14a** showed the molecular ion [M + H]⁺ peak at m/z 219.0752 which is consistent with the mass of the proposed product. The FTIR spectrum displayed characteristic absorption bands for the carbonyl group at 1686 cm⁻¹ and 1614 cm⁻¹ and for the amine group at 3215 cm⁻¹ and 2902 cm⁻¹.

Conclusions

We have developed a novel, catalyst-free, solvent-free green method for the synthesis of benzodiazepines 10 and dihydroquinoxalines 11. Two different series of compounds, 10 and 11, were synthesised via a multicomponent reaction of benzimidazole 6, isocyanide 2 and acetone 3a. The generated benzodiazepine scaffolds 10 were only obtained when the benzimidazole substituents were electron-donating or where the benzimidazoles were unsubstituted. The omission of DMAD from the initial method resulted in dihydroquinoxaline derivatives 11 from benzimidazoles 6 with only deactivating groups such as Cl, F and CO_2CH_3 as substituents on R^1 and R^2 . When 1,2-phenylenediamine 1 or benzimidazole 6 were reacted with

electron deficient alkynes (DMAD, DEtAD and DTAD) 7 under ethanol reflux conditions as a two component reaction they resulted in formation of benzodiazepine derivatives 14. Reaction of benzimidazoles 6 gave lower yields of 14 when compared to reaction of 1,2-phenylenediamines 1.

Conflicts of interest

There are no conflicts of interest to declare.

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