### Chemical Science



### **EDGE ARTICLE**

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## Double quick, double click reversible peptide "stapling"†

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The development of constrained peptides for inhibition of protein–protein interactions is an emerging strategy in chemical biology and drug discovery. This manuscript introduces a versatile, rapid and reversible approach to constrain peptides in a bioactive helical conformation using BID and RNase S peptides as models. Dibromomaleimide is used to constrain BID and RNase S peptide sequence variants bearing cysteine (Cys) or homocysteine (hCys) amino acids spaced at i and i+4 positions by double substitution. The constraint can be readily removed by displacement of the maleimide using excess thiol. This new constraining methodology results in enhanced  $\alpha$ -helical conformation (BID and RNase S peptide) as demonstrated by circular dichroism and molecular dynamics simulations, resistance to proteolysis (BID) as demonstrated by trypsin proteolysis experiments and retained or enhanced potency of inhibition for Bcl-2 family protein–protein interactions (BID), or greater capability to restore the hydrolytic activity of the RNAse S protein (RNase S peptide). Finally, use of a dibromomaleimide functionalized with an alkyne permits further divergent functionalization through alkyne–azide cycloaddition chemistry on the constrained peptide with fluorescein, oligoethylene glycol or biotin groups to facilitate biophysical and cellular analyses. Hence this methodology may extend the scope and accessibility of peptide stapling.

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### Introduction

Inhibition of α-helix mediated protein-protein interactions (PPIs)<sup>1-4</sup> represents an area of intense interest, due to their role in intracellular signalling processes<sup>5</sup> and demonstrated tractability for drug discovery.6,7 The last decade has seen an increasing emphasis on the development of constrained peptides8-10 for PPI inhibition. Constraint of a peptide in a bioactive conformation is proposed to improve target binding affinity (by preorganization),11,12 increase stability (e.g. by protecting against proteolysis)13-15 and enhance cell-uptake,16,17 although a number of recent studies have provided conflicting evidence for the enhanced cell permeability and cellular potency that is conferred through stapling. 18-20 Available strategies 9,13,21-25 to constrain a peptide in a helical conformation (Fig. 1a-e), include hydrocarbon "staples",26-28 lactam bridges,29-31 hydrogen-bond surrogates, 32 photoswitches 11,33,34 and triazoles introduced by "click" chemistry.35-38 The nature of the linker, including any

stereocenters, plays a role in regulating peptide conformation

Previously, we<sup>14</sup> and others<sup>61</sup> introduced  $\alpha$ -pentenylglycine as an alternative to the widely used  $\alpha,\alpha'$ -pentenylalanine for crosslinking the i and i + 4 residues within a peptide (Fig. 1c and d).<sup>26</sup> Production of this *des*-methyl constraint requires a more easily

and target protein affinity.31,39 Introduction of constraints may be achieved either by intramolecular reaction (Fig. 1a)28,31 or by chemoselective intermolecular reaction (Fig. 1b) between suitably disposed residues within the peptide sequence and an appropriate coupling partner. 36,40 Introduction of a constraint by intermolecular reaction, whilst synthetically more demanding, may be advantageous in that it provides access to functionalized constraints<sup>36</sup> and thus to peptides tailored for biophysical, cellular and chemical proteomics analyses. Herein, we use BID and RNase S variant peptides (1a-d) which respectively bind Bclx<sub>L</sub> proteins<sup>41</sup> and RNase S protein<sup>42</sup> (Fig. 1f and g), to demonstrate that dibromomaleimides<sup>43-51</sup> serve as versatile constraints that can be introduced and removed.52-57 These reagents have not previously been studied for their ability to control peptide helicity. Introduction of an alkyne onto the dibromomaleimide permits further divergent functionalization by "click" chemistry.58 This approach may therefore complement recently reported methods to constrain peptides by reaction of dichloroacetone<sup>59</sup> with side chain thiols and photo-reversible constraints using S,S'-tetrazine.60

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intramolecular (a)  $\alpha.\alpha$ '-disubstituted cross-link (b) isosterio intermolecular S,S'-maleimide crosslink Ac-EDIIRNIARHLACVGDCNLDRSIW-NH2 Ac-EDIIRNIARHLA CVGD, CN, DRSIW-NH2 RED-BID-hCys 1b (g) SH HS Bcl-x, / BID Ac-KETAA CKFECQHMDS-NHa RED-RNase-S-Cys 1c Ac-KETAA, CKFE, CQHMDS-NH2 RED-RNase-S-hCys RNase S

# Fig. 1 Overview of peptide stapling strategies and linkers for constraining peptides in an $\alpha$ -helical conformation together with structures for the model protein–protein interactions used in this study; (a) schematic for intramolecular crosslinking; (b) schematic for intermolecular crosslinking; (c) S,S- $\alpha,\alpha'$ -disubstituted $i,\ i+4$ hydrocarbon "staple"; (d) S,S-des-methyl $i,\ i+4$ hydrocarbon constraint; (e) $i,\ i+4$ S,S'-maleimide staple reported in this work; (f) and (g) model peptide sequences 1a-d for BID (1a-d) and RNase S (1c-d) and protein–protein interactions in this study (Bcl- $x_L$ (green)/BID (blue) PDB ID: 4QVE; RNase S protein (green)/S peptide (blue) PDB ID: 1CJQ; hotspot residues (red)).

S protein / S peptide

synthesized amino acid which is less sterically hindered for peptide coupling, and, as shown by us, has no deficiencies in terms of biophysical properties when applied to BH3 peptides BID and BIM.  $^{62}$  We hypothesized that this all hydrocarbon 8-atom constraint could be replaced with an isoatomic S, S'-maleimide crosslink by reaction of suitably disposed homocysteine (hCys) residues with dibromomaleimide (or for cysteine (Cys), a 6-atom constraint) (Fig. 1e).

Peptides 1a–d bearing Cys or hCys amino acids in positions used previously<sup>26,62</sup> for  $i \rightarrow i+4$  constraints were independently oxidised to the constrained disulfide 2a–d or cross-linked with dibromomaleimide 4 to generate 3a–d with rapid and complete conversion. Alternatively, this sequence 1a–d to 3a–d could be performed in one pot and peptides 1a–d fully regenerated by addition of an excess of thiol (Fig. 2 and S4–7†). It is noteworthy that conversion of the Cys-derived sequences 1a and 1c to their constrained derivatives 1a and 1c to their corresponding aCys derivatives 1c and 1c to the noted that these studies were carried out in unbuffered aqueous solution and future studies will focus on a more detailed analysis of the reaction rates, their pH dependence and comparison to other rapid constraining reactions.

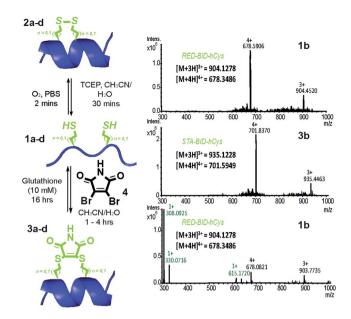


Fig. 2 Reversible constraining reaction sequence for peptides 1–3a–d: conditions for conversion from the oxidised state (OX) through the intermediate reduced state (RED) to the constrained peptide (STA) and *vice versa* (LHS); HRMS analyses of reaction mixture for reversible interconversion between 1b and 3b (relevant masses are indicated in black for the peptide and in green for excess glutathione used during unconstraining) (RHS).

fact that a range of thiols were used including a 10 mM aqueous solution of glutathione, which corresponds to the average concentration present in cancer cells.<sup>64,65</sup> It may thus be feasible to harness this approach for controlled delivery of a peptide to the diseased cell; a focus of future studies. Interestingly, removal of the maleimide cross-link could be abrogated through addition of an excess of the reducing agent TCEP, resulting in a permanent succinimide cross-link (Fig. S9†). Succinimide linkages of a similar nature have been shown to undergo thiol mediated cleavage;<sup>48</sup> future efforts to "lock" the constraint may therefore be more appropriately served by hydrolysis of the imide in the maleimide crosslink.<sup>66</sup>

Peptides 1a,b and 3a,b were analysed by CD spectroscopy and subjected to proteolysis assays and fluorescence anisotropy competition assays for inhibition of the BID/Mcl-1, BAK/Bcl-x<sub>L</sub> and NOXA-B/Mcl-1 interactions (Fig. 3, S19-23 and S29-30†). Constrained peptides 3a,b showed no significant difference in helicity (42% and 43%) to each other as a result of the different cross-link length but both were more helical than the corresponding reduced peptide 1a or 1b, (24% and 29%). Molecular dynamics calculations were conducted on 1b and 3b - these further support the conclusion that the helical conformation predominates when constrained by the maleimide linker (Fig. S25 and S26†). Tryptic proteolysis as monitored by LC-MS<sup>14</sup> established that the conformationally constrained peptides 3a,b exhibit improved proteolytic resistance in comparison to the reduced peptides **1a,b**; proteolysis was retarded up to 7 residues removed from either side of the thiol residues (Fig. S19-23†). Taken together these data indicate that the maleimide constraint confers comparable enhancement of helicity and

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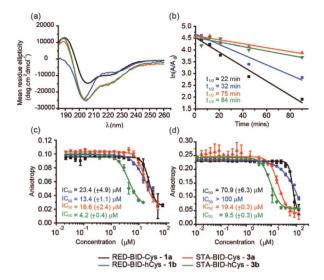


Fig. 3 Analyses on BID derived peptides 1a,b and 3a,b. (a) Circular dichroism spectra (water/30% acetonitrile); (b) trypsin proteolysis in a 1:10 000 enzyme/substrate ratio (50 mM phosphate buffer, 200 mM NaCl, pH 7.50); (c) fluorescence anisotropy competition assay for inhibition of the FITC-BID/Mcl-1 interaction (25 nM tracer, 150 nM protein, 50 mM Tris buffer, 150 mM NaCl, 0.61% of Triton X-100, pH 7.40); (d) fluorescence anisotropy competition assay for inhibition of the BODIPY-BAK/Bcl-x<sub>L</sub> interaction (25 nM tracer, 150 nM protein, 50 mM phosphate buffer, 200 mM NaCl, 0.02 mg mL<sup>-1</sup> BSA, pH 7.50).

proteolytic resistance to the *des*-methyl hydrocarbon constraint reported by us earlier. <sup>62</sup> Peptides **1a,b** and **3a,b** inhibited the BID/Mcl-1 interaction with comparable potency to the native **WT-BID 8** sequence (IC $_{50}\sim15$ –25  $\mu M$ ) except for the constrained peptide **3b** which was slightly more potent (IC $_{50}\sim4.2$   $\mu M$ ), thus the constraining/unconstraining reagents have either a weak or no effect on the inhibitory potency for this PPI. Although inconsistent with the concept that preorganization should promote higher binding affinity, the result is consistent with our own recent studies on hydrocarbon-constrained

peptides, which illustrated enthalpy-entropy compensation operates for these peptides which bind through an induced-fit mechanism.  $^{62}$  For inhibition of BAK/Bcl- $x_L$  both constrained peptides 3a,b were shown to be significantly superior inhibitors compared to the reduced peptides 1a,b. The differential effects upon potency for the same peptide against Bcl- $x_L$  and Mcl-1 may arise as a consequence of favourable non-covalent contacts between the maleimide and Bcl- $x_L$ ; it is anticipated that future structural analyses will shed light on these differences.

The synthetic method was further adapted to allow the functionalization of the BID-Cys peptides 7a-c. Using Mitsunobu alkylation,67 alkyne-dibromomaleimide 5 was prepared and conjugated to peptide 1a to generate the constrained peptide 6. Without isolation or purification, subsequent "click" reaction with biotin, fluorescein and PEG azides38 afforded complete conversion to click-conjugated constrained peptides 7a-c (Fig. 4a and b). Related "clickable maleimides" have previously been used to conjugate small molecule drugs to polymers,68 whilst "clicked" dibromomaleimides have been used to conjugate polymers to oxytocin so as to improve solubility.51 Here we have expanded this methodology, demonstrating that this functionality can be used to constrain peptide conformation and as a divergent method for peptide functionalization. The utility of these modifications was demonstrated through a direct fluorescence anisotropy titration with the fluorescein linked peptide 7b (Fig. 4c). The observed potencies are consistent with the EC50 values determined in the competition assays for 3a,b indicating no detrimental consequence from the introduction of a fluorophore in this position. Such a modification could thus facilitate cellular studies whilst leaving the N-terminus of the peptide free to allow further orthogonal derivatization.

The use of dibromomaleimide-based constraints for promotion of  $\alpha$ -helicity was then further demonstrated using the RNase S-protein/S-peptide interaction with the choice of amino acid positions at which to introduce the constraint

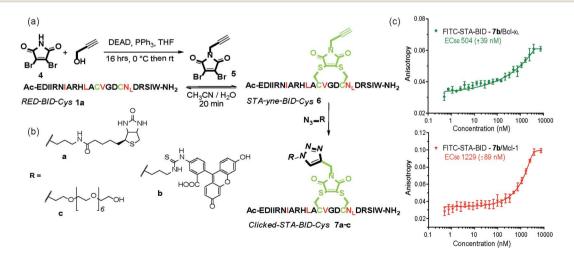


Fig. 4 (a) Scheme illustrating double click chemistry for generating functionalized constrained peptides 7a-c. An alkyne-maleimide was used to generate yne-STA-BID-Cys 6, which was further functionalized by click reaction with (b) various azide derivatives. (c) Fluorescence anisotropy direct binding assay for interaction of FITC-STA-BID 7b with Mcl-1 and Bcl- $x_L$  (75 nM tracer, 15  $\mu$ M protein, 50 mM Tris buffer, 150 mM NaCl, 0.61% of Triton X-100, pH 7.40).

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(a) (b) S protein RNase A WT-S - 2c - OX-S-Cys - 2c 100% STA-S-hCys - 3c - STA-S-hCys - 3d 80% WT-S - 8 60% W

Fig. 5 Biophysical and bioactivity analyses of WT-S and S derived peptides 3–4a,c. (a) Circular dichroism spectra recorded in water (30% TFE); (b) RNA degradation assay showing the restoration of S protein activity. S protein was used as a negative control, and RNase A as a positive control. (50 mM Tris, 100 mM NaCl, pH 7.50).

informed by previous studies by the Verdine,26 Hamachi69 and Dawson<sup>59</sup> groups. S peptides **1c** and **1d** bearing Cys and hCys residues at the i and i + 4 positions were synthesized using standard solid-phase synthesis in good yield and purified by HPLC (Fig. S2 and S3†). Peptides 1c and 1d could not readily be isolated as reduced bis-thiols; one-pot reduction followed by the reaction with dibromomaleimide 4 afforded peptides 3c and 3d (Fig. 2). Both peptides were then analysed by CD spectroscopy and used in an RNA degradation assay to determine the restoration of enzymatic activity to S-protein in place of the S-peptide (Fig. 5a and b). In this instance, different and more pronounced behaviour for the Cys and hCys variants was observed, highlighting that the effects of introducing a constraint are contextdependent. The Cys-derived RNase S peptides 2-3c exhibit poor helicity (e.g. 3c = 9% versus wild-type 8 = 30%) and corresponding low restoration of enzymatic activity (Fig. S31†). Molecular modelling suggests that the introduction of a disulfide or a maleimide bridge does not support a helical conformation (Fig. S27 and S28 $\dagger$ ). In contrast, for the hCys **1–3d** series introduction of a disulfide or a maleimide crosslink supports a helical structure comparable to the WT sequence and consistent with the moderate effects on helicity for an 8-carbon constraint in these positions. (Furthermore, and fully consistent with the work of Hamachi and co-workers,69 upon addition of trifluoroethanol (TFE), the constrained peptides reach a greater overall α-helical content). Here, introduction of a disulfide (2d) or a maleimide (3d) crosslink results in enhanced helicity in comparison to the WT sequence (2d = 56%, 3d = 52% and wild-type 8 = 30%). As a result 3d is much better able to restore RNase S protein activity than is 3c.

### Conclusions

In conclusion, the attractiveness of thiol-based "stapling" approaches has been recognized for some time, <sup>24</sup> but no general approach has been widely adopted; we have developed a new method to rapidly and reversibly introduce a functionalized constraint into two model peptides that enhances a range of biophysical and biochemical properties. Moreover, the constraint may be further functionalized in a divergent manner with motifs useful in chemical biology and proteomics

applications. The approach can be carried out using natural readily available amino acid side chains on unprotected peptides, potentially bringing peptide stapling to a larger community than hitherto have had access and extends the applications of dibromomaleimide and related reagents. Future efforts will apply this approach to a wider range of helix mediated PPIs with the aim of establishing the generality of the method. The approach may also be useful to identify transient binding pockets proximal to a native ligand binding site using the template directed approach introduced by Sharpless.70 Similarly, the reversibility of the constraint, may permit more facile delivery of a peptide-based reagent to the cell, where it can become unconstrained and less easily transported back out of the cell. Together with the additional functional handle introduced on the maleimide constraint (which is orthogonal in terms of its reactivity), this could facilitate detailed cellular and biochemical analyses e.g. identification of additional targets of the constrained peptide by pull-down. These objectives will represent the focus of our future studies.

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#### References

- 1 V. Azzarito, K. Long, N. S. Murphy and A. J. Wilson, *Nat. Chem.*, 2013, 5, 161–173.
- L.-G. Milroy, T. N. Grossmann, S. Hennig, L. Brunsveld and C. Ottmann, *Chem. Rev.*, 2014, 114, 4695–4748.
- 3 M. R. Arkin, Y. Tang and J. A. Wells, *Chem. Biol.*, 2014, 21, 1102–1114.
- 4 M. Pelay-Gimeno, A. Glas, O. Koch and T. N. Grossmann, *Angew. Chem., Int. Ed.*, 2015, **54**, 8896–8927.
- 5 A. Vinayagam, U. Stelzl, R. Foulle, S. Plassmann, M. Zenkner, J. Timm, H. E. Assmus, M. A. Andrade-Navarro and E. E. Wanker, *Sci. Signaling*, 2011, 4, rs8.
- 6 W. H. Wilson, O. A. O'Connor, M. S. Czuczman, A. S. LaCasce, J. F. Gerecitano, J. P. Leonard, A. Tulpule, K. Dunleavy, H. Xiong, Y.-L. Chiu, Y. Cui, T. Busman, S. W. Elmore, S. H. Rosenberg, A. P. Krivoshik, S. H. Enschede and R. A. Humerickhouse, *Lancet Oncol.*, 2010, 11, 1149–1159.
- 7 I. Ray-Coquard, J.-Y. Blay, A. Italiano, A. Le Cesne, N. Penel, J. Zhi, F. Heil, R. Rueger, B. Graves, M. Ding, D. Geho, S. A. Middleton, L. T. Vassilev, G. L. Nichols and B. N. Bui, *Lancet Oncol.*, 2012, 13, 1133–1140.
- 8 P. M. Cromm, J. Spiegel and T. N. Grossmann, *ACS Chem. Biol.*, 2015, **10**, 1362–1375.
- 9 Y. H. Lau, P. de Andrade, Y. Wu and D. R. Spring, *Chem. Soc. Rev.*, 2015, 44, 91–102.

10 L. Nevola and E. Giralt, *Chem. Commun.*, 2015, **51**, 3302–3315.

**Chemical Science** 

- 11 P. Wysoczanski, R. J. Mart, E. J. Loveridge, C. Williams, S. B. M. Whittaker, M. P. Crump and R. K. Allemann, *J. Am. Chem. Soc.*, 2012, **134**, 7644–7647.
- 12 Y. S. Chang, B. Graves, V. Guerlavais, C. Tovar, K. Packman, K.-H. To, K. A. Olson, K. Kesavan, P. Gangurde, A. Mukherjee, T. Baker, K. Darlak, C. Elkin, Z. Filipovic, F. Z. Qureshi, H. Cai, P. Berry, E. Feyfant, X. E. Shi, J. Horstick, D. A. Annis, A. M. Manning, N. Fotouhi, H. Nash, L. T. Vassilev and T. K. Sawyer, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, 110, E3445–E3454.
- 13 D. Wang, W. Liao and P. S. Arora, *Angew. Chem., Int. Ed.*, 2005, 44, 6525-6529.
- 14 D. J. Yeo, S. L. Warriner and A. J. Wilson, *Chem. Commun.*, 2013, 49, 9131–9133.
- 15 P. M. Cromm, J. Spiegel, P. Küchler, L. Dietrich, J. Kriegesmann, M. Wendt, R. S. Goody, H. Waldmann and T. N. Grossmann, ACS Chem. Biol., 2016, 11, 2375–2382.
- 16 Q. Chu, R. E. Moellering, G. J. Hilinski, Y.-W. Kim, T. N. Grossmann, J. T. H. Yeh and G. L. Verdine, MedChemComm, 2015, 6, 111–119.
- 17 G. H. Bird, E. Mazzola, K. Opoku-Nsiah, M. A. Lammert, M. Godes, D. S. Neuberg and L. D. Walensky, *Nat. Chem. Biol.*, 2016, 12, 845–852.
- 18 T. Okamoto, K. Zobel, A. Fedorova, C. Quan, H. Yang, W. J. Fairbrother, D. C. S. Huang, B. J. Smith, K. Deshayes and P. E. Czabotar, *ACS Chem. Biol.*, 2012, **8**, 297–302.
- 19 Y.-C. Li, L. W. Rodewald, C. Hoppmann, E. T. Wong, S. Lebreton, P. Safar, M. Patek, L. Wang, K. F. Wertman and G. M. Wahl, *Cell Rep.*, 2014, 9, 1946–1958.
- 20 T. Okamoto, D. Segal, K. Zobel, A. Fedorova, H. Yang, W. J. Fairbrother, D. C. S. Huang, B. J. Smith, K. Deshayes and P. E. Czabotar, ACS Chem. Biol., 2014, 9, 838–839.
- 21 M. M. Madden, A. Muppidi, Z. Li, X. Li, J. Chen and Q. Lin, *Bioorg. Med. Chem. Lett.*, 2011, 21, 1472–1475.
- 22 H. Jo, N. Meinhardt, Y. Wu, S. Kulkarni, X. Hu, K. E. Low, P. L. Davies, W. F. DeGrado and D. C. Greenbaum, *J. Am. Chem. Soc.*, 2012, 134, 17704–17713.
- 23 A. Muppidi, K. Doi, S. Edwardraja, E. J. Drake, A. M. Gulick, H.-G. Wang and Q. Lin, *J. Am. Chem. Soc.*, 2012, **134**, 14734– 14737.
- 24 D. P. Fairlie and A. Dantas de Araujo, *Pept. Sci.*, 2016, **106**, 843–852.
- 25 A. J. Rojas, C. Zhang, E. V. Vinogradova, N. Buchwald, J. Reilly, B. Pentelute and S. Buchwald, *Chem. Sci.*, 2017, DOI: 10.1039/C6SC05454D.
- 26 C. E. Schafmeister, J. Po and G. L. Verdine, *J. Am. Chem. Soc.*, 2000, **122**, 5891–5892.
- 27 Y.-W. Kim, P. S. Kutchukian and G. L. Verdine, *Org. Lett.*, 2010, **12**, 3046–3049.
- 28 Y.-W. Kim, T. N. Grossmann and G. L. Verdine, *Nat. Protoc.*, 2011, **6**, 761–771.
- 29 C. Bracken, J. Gulyás, J. W. Taylor and J. Baum, J. Am. Chem. Soc., 1994, 116, 6431–6432.
- 30 R. S. Harrison, G. Ruiz-Gómez, T. A. Hill, S. Y. Chow, N. E. Shepherd, R.-J. Lohman, G. Abbenante, H. N. Hoang and D. P. Fairlie, *J. Med. Chem.*, 2010, 53, 8400–8408.

- 31 A. D. de Araujo, H. N. Hoang, W. M. Kok, F. Diness, P. Gupta, T. A. Hill, R. W. Driver, D. A. Price, S. Liras and D. P. Fairlie, *Angew. Chem., Int. Ed.*, 2014, 53, 6965–6969.
- 32 A. Patgiri, A. L. Jochim and P. S. Arora, *Acc. Chem. Res.*, 2008, **41**, 1289–1300.
- 33 S. Kneissl, E. J. Loveridge, C. Williams, M. P. Crump and R. K. Allemann, *ChemBioChem*, 2008, 9, 3046–3054.
- 34 L. Nevola, A. Martín-Quirós, K. Eckelt, N. Camarero, S. Tosi, A. Llobet, E. Giralt and P. Gorostiza, *Angew. Chem., Int. Ed.*, 2013, 52, 7704–7708.
- 35 O. Torres, D. Yüksel, M. Bernardina, K. Kumar and D. Bong, ChemBioChem, 2008, 9, 1701–1705.
- 36 Y. H. Lau, P. de Andrade, S.-T. Quah, M. Rossmann, L. Laraia, N. Skold, T. J. Sum, P. J. E. Rowling, T. L. Joseph, C. Verma, M. Hyvonen, L. S. Itzhaki, A. R. Venkitaraman, C. J. Brown, D. P. Lane and D. R. Spring, *Chem. Sci.*, 2014, 5, 1804–1809.
- 37 Y. H. Lau, P. de Andrade, N. Skold, G. J. McKenzie, A. R. Venkitaraman, C. Verma, D. P. Lane and D. R. Spring, Org. Biomol. Chem., 2014, 12, 4074–4077.
- 38 P. T. Tran, C. Ø. Larsen, T. Røndbjerg, M. De Foresta, M. B. A. Kunze, A. Marek, J. H. Løper, L.-E. Boyhus, A. Knuhtsen, K. Lindorff-Larsen and D. S. Pedersen, Chem.-Eur. J., 2017, 23, 3490-3495.
- 39 K. Hu, H. Geng, Q. Zhang, Q. Liu, M. Xie, C. Sun, W. Li, H. Lin, F. Jiang, T. Wang, Y.-D. Wu and Z. Li, *Angew. Chem.*, *Int. Ed.*, 2016, 55, 8013–8017.
- 40 S. K. Sia, P. A. Carr, A. G. Cochran, V. N. Malashkevich and P. S. Kim, *Proc. Natl. Acad. Sci. U. S. A.*, 2002, **99**, 14664– 14669.
- 41 M. Kvansakul and M. Hinds, Apoptosis, 2015, 20, 136-150.
- 42 R. T. Raines, Chem. Rev., 1998, 98, 1045-1066.
- 43 M. E. B. Smith, F. F. Schumacher, C. P. Ryan, L. M. Tedaldi, D. Papaioannou, G. Waksman, S. Caddick and J. R. Baker, *J. Am. Chem. Soc.*, 2010, 132, 1960–1965.
- 44 P. Moody, M. E. B. Smith, C. P. Ryan, V. Chudasama, J. R. Baker, J. Molloy and S. Caddick, *ChemBioChem*, 2012, 13, 39–41.
- 45 M. W. Jones, R. A. Strickland, F. F. Schumacher, S. Caddick, J. R. Baker, M. I. Gibson and D. M. Haddleton, *J. Am. Chem. Soc.*, 2012, 134, 1847–1852.
- 46 F. F. Schumacher, V. A. Sanchania, B. Tolner, Z. V. F. Wright, C. P. Ryan, M. E. B. Smith, J. M. Ward, S. Caddick, C. W. M. Kay, G. Aeppli, K. A. Chester and J. R. Baker, *Sci. Rep.*, 2013, 3, 1525.
- 47 M. P. Robin, P. Wilson, A. B. Mabire, J. K. Kiviaho, J. E. Raymond, D. M. Haddleton and R. K. O'Reilly, *J. Am. Chem. Soc.*, 2013, 135, 2875–2878.
- 48 C. Marculescu, H. Kossen, R. E. Morgan, P. Mayer, S. A. Fletcher, B. Tolner, K. A. Chester, L. H. Jones and J. R. Baker, *Chem. Commun.*, 2014, 50, 7139–7142.
- 49 J. Youziel, A. R. Akhbar, Q. Aziz, M. E. B. Smith, S. Caddick, A. Tinker and J. R. Baker, *Org. Biomol. Chem.*, 2014, **12**, 557–560.
- 50 M. P. Robin and R. K. O'Reilly, *Chem. Sci.*, 2014, 5, 2717–2723.

**Edge Article** 

51 J. Collins, J. Tanaka, P. Wilson, K. Kempe, T. P. Davis, M. P. McIntosh, M. R. Whittaker and D. M. Haddleton, *Bioconjugate Chem.*, 2015, 26, 633-638.

- 52 Oxime and Hydrazone constraints employing non-natural amino acids introduced by Horne and co-workers [ref. 53 and 54] have been used to stabilize a helical conformation but not applied to PPI inhibition.
- 53 C. M. Haney, M. T. Loch and W. S. Horne, *Chem. Commun.*, 2011, 47, 10915–10917.
- 54 C. M. Haney and W. S. Horne, J. Pept. Sci., 2014, 20, 108-114.
- 55 Both iminoboronate formation<sup>56</sup> and reaction of Meldrums acid with an amine and thiol side chain<sup>57</sup> have recently shown to be amenable to reversible peptide cyclization and might be applied to stabilization of peptides for inhibiton of PPIs.
- 56 A. Bandyopadhyay and J. Gao, *J. Am. Chem. Soc.*, 2016, **138**, 2098–2101.
- 57 A. M. Johnson and E. V. Anslyn, *Org. Lett.*, 2017, **19**, 1654–1657.
- 58 H. C. Kolb, M. G. Finn and K. B. Sharpless, *Angew. Chem., Int. Ed.*, 2001, **40**, 2004–2021.
- 59 N. Assem, D. J. Ferreira, D. W. Wolan and P. E. Dawson, *Angew. Chem., Int. Ed.*, 2015, **54**, 8665–8668.
- 60 S. P. Brown and A. B. Smith, J. Am. Chem. Soc., 2015, 137, 4034–4037.
- 61 W. Nomura, H. Aikawa, N. Ohashi, E. Urano, M. Métifiot, M. Fujino, K. Maddali, T. Ozaki, A. Nozue, T. Narumi,

- C. Hashimoto, T. Tanaka, Y. Pommier, N. Yamamoto, J. A. Komano, T. Murakami and H. Tamamura, *ACS Chem. Biol.*, 2013, **8**, 2235–2244.
- 62 J. A. Miles, D. J. Yeo, P. Rowell, S. Rodriguez-Marin, C. M. Pask, S. L. Warriner, T. A. Edwards and A. J. Wilson, *Chem. Sci.*, 2016, 7, 3694–3702.
- 63 P. Diderich, D. Bertoldo, P. Dessen, M. M. Khan, I. Pizzitola, W. Held, J. Huelsken and C. Heinis, ACS Chem. Biol., 2016, 11, 1422–1427.
- 64 J. M. Estrela, A. Ortega and E. Obrador, Crit. Rev. Clin. Lab. Sci., 2006, 43, 143–181.
- 65 P. Monostori, G. Wittmann, E. Karg and S. Túri, J. Chromatogr. B: Anal. Technol. Biomed. Life Sci., 2009, 877, 3331–3346.
- 66 M. Morais, J. P. M. Nunes, K. Karu, N. Forte, I. Benni, M. E. B. Smith, S. Caddick, V. Chudasama and J. R. Baker, Org. Biomol. Chem., 2017, 15, 2947–2952.
- 67 F. F. Schumacher, M. Nobles, C. P. Ryan, M. E. B. Smith, A. Tinker, S. Caddick and J. R. Baker, *Bioconjugate Chem.*, 2011, 22, 132–136.
- 68 H. Wang, M. Xu, M. Xiong and J. Cheng, *Chem. Commun.*, 2015, **51**, 4807–4810.
- 69 I. Hamachi, Y. Yamada, T. Matsugi and S. Shinkai, *Chem.*–*Eur. J.*, 1999, 5, 1503–1511.
- 70 W. G. Lewis, L. G. Green, F. Grynszpan, Z. Radić, P. R. Carlier, P. Taylor, M. G. Finn and K. B. Sharpless, Angew. Chem., Int. Ed., 2002, 41, 1053–1057.