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# Molecular jackhammers induce intracellular calcium release and skeletal muscle contraction by vibronic-driven action

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Intracellular calcium release (ICR) is of fundamental importance for numerous physiological and pathological processes, from cell differentiation to neurotransmitter release and muscle contraction. The ability to precisely control ICR is crucial for understanding cellular signaling mechanisms and developing therapeutic interventions for calcium-related disorders. Light-induced ICR offers numerous advantages over conventional approaches, including superior specificity, precise spatiotemporal control, excellent reproducibility, and minimal invasiveness. Here, we demonstrate that cyanine dyes function as molecular jackhammers (MJHs) to induce ICR in human embryonic kidney (HEK) 293 and immortalized mouse myoblast C2C12 cells through vibronic-driven action (VDA) following red light (640 nm) activation. VDA generates longitudinally and axially coordinated whole-molecule vibrations, enabling mechanical interactions with cells. Structure–activity relationship studies of twenty synthesized MJHs demonstrated that sulfonate-containing derivatives most effectively triggered intracellular calcium release while exhibiting minimal cytotoxicity. Mechanistic studies showed that MJH-induced calcium release operates through the inositol-triphosphate (IP<sub>3</sub>) pathway rather than reactive oxygen species generation. Finally, we demonstrated light-activated MJH-induced calcium waves *in vivo* using transgenic *Hydra vulgaris* expressing a fluorescent calcium indicator. This work establishes MJHs as molecular-scale devices for remote control of cellular signaling, expanding the utility of cyanine dyes for modulating physiological processes with potential therapeutic applications.

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## Introduction

Intracellular calcium release (ICR) is the process by which calcium ions stored in cellular organelles, such as the endoplasmic reticulum (ER) and sarcoplasmic reticulum (SR), are released into the cytoplasm.<sup>1</sup> ICR plays a fundamental role in regulating various physiological and pathological processes,<sup>2,3</sup> including signal transduction,<sup>4</sup> gene expression,<sup>5,6</sup> cell proliferation,<sup>7</sup> neurotransmitter release,<sup>8,9</sup> and muscle contraction.<sup>10,11</sup> Understanding ICR mechanisms is crucial for

elucidating cell signaling pathways and disease pathogenesis, with implications for drug development and clinical diagnostics.

Multiple channels, pumps, and signaling pathways regulate ICR. Traditional methods for inducing ICR include: (1) chemical agonists such as ionomycin<sup>12</sup> and thapsigargin,<sup>13,14</sup> or hormones such as parathyroid<sup>15</sup> and acetylcholine;<sup>16</sup> (2) electrical stimulation *via* membrane depolarization, which opens voltage-gated calcium channels in specific tissues to increase intracellular calcium levels;<sup>17–19</sup> and (3) mechanical stimuli such as cell compression and stretching.<sup>20,21</sup> However, these approaches have limitations, including potential cytotoxicity, poor specificity, limited temporal and spatial resolution, and low reproducibility. In contrast, light-activated molecular strategies offer several advantages: they are minimally invasive, highly specific and reproducible, precisely controllable, and fully reversible, enabling dynamic regulation of calcium release.<sup>22–26</sup>

Our laboratory previously developed the biological applications of Feringa-type molecular motors (MMs), overcrowded alkenes composed of a rotor and stator that perform

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unidirectional mechanical work upon activation with 400 nm light.<sup>27–30</sup> These motors can stimulate intercellular calcium waves<sup>31</sup> through the inositol-triphosphate (IP<sub>3</sub>) signaling pathway.<sup>32–34</sup> However, their requirement for 400 nm light activation limits broader applications due to potential tissue damage and poor penetration depth.<sup>35</sup>

Cyanine dyes are fluorescent compounds characterized by two nitrogen-containing heterocycles connected by a conjugated odd-numbered carbon chain, with one heterocycle carrying a positive charge.<sup>36,37</sup> These dyes exhibit high molar extinction coefficients, narrow absorption and emission bands, tuneable fluorescence, and excellent biocompatibility.<sup>36–38</sup> We recently demonstrated that cyanine dyes activated by low-intensity near-infrared (NIR) light (730 nm, 80 mW cm<sup>-2</sup>) for 10 min can function as molecular jackhammers (MJHs) to eradicate cancer cells through vibronic-driven action (VDA).<sup>39–42</sup> VDA emerges from the coupling between the electronic and vibrational modes of a molecule, creating a molecular plasmon that induces concerted whole-molecule vibrations on picosecond timescales when activated by light. These vibrations cause the entire molecule to stretch both longitudinally and axially, generating mechanical forces capable of disrupting cell membranes.<sup>39,42</sup> In giant unilamellar vesicles (GUVs), light activation of MJHs has been shown to induce vesicle shrinkage and collapse, which provides direct evidence of mechanically driven effects on lipid membranes. This mechanically driven process represents a new therapeutic approach distinct from traditional photodynamic and photothermal therapies, as evidenced by the fact that it is neither inhibited by reactive oxygen species (ROS) scavengers nor accompanied by media temperature increases.<sup>39,41,42</sup>

Different from inducing cancer cell death, here we demonstrate that MJHs activated by red light (640 nm) for only 250 ms effectively stimulate ICR *via* VDA by mechanically activating the IP<sub>3</sub> pathway. We show the efficacy of MJH-induced ICR both *in vitro* using HEK293 cells,<sup>43</sup> C2C12 cells<sup>44</sup> and *in vivo* using *Hydra vulgaris*.<sup>45,46</sup> Unlike our previous work on cancer cell eradication, brief MJH activation efficiently induces ICR while maintaining cell viability. The use of red light provides reduced autofluorescence, enhanced tissue penetration, and minimal tissue damage.<sup>35</sup> During the past decades, cyanine dyes have been found to have numerous applications in biomedical imaging, drug delivery and *in vivo* sensing.<sup>47–49</sup> However, this study introduces their capability to control intracellular calcium release through mechanical action, expanding its application as precise molecular tools for manipulating cellular processes in living systems.

## Results and discussion

Cyanine 5.5 is a versatile far-red fluorescent dye comprising two benzoindeole chromophores connected by five-methine units.<sup>39,42</sup> It exhibits both longitudinal molecular plasmons (LMP) and transversal molecular plasmons (TMP). Time-dependent density functional theory (TD-DFT) calculations confirmed the presence of a vibronic mode (coupled vibrational

oscillations and electronic states) at 640 nm,<sup>39,42</sup> as shown by the induced charge density of the molecular plasmon in Fig. 1a.<sup>50–52</sup>

The synthetic approach for both symmetrical and non-symmetrical MJHs is shown in Fig. 1b. Symmetrical MJHs **3** were synthesized from heterocyclic salts **1** and commercially available malondianil or glutacondianil **2** in the presence of NaOAc and Ac<sub>2</sub>O, affording good yields. Furthermore, non-symmetrical MJHs **3** required a two-step process: first, heterocyclic salts **1** and **2** were reacted in Ac<sub>2</sub>O to form acetylated precursors, that were then condensed with equimolar amounts of differently substituted heterocyclic salts **1'**, providing satisfactory yields (Schemes S1 and S2).<sup>44</sup> The photophysical properties including absorption and emission spectra of all synthesized MJHs were determined (Table S1 and Fig. S1–S21).

To investigate structure–activity relationships and the importance of VDA, we selected four representative MJHs (Fig. 1c). The non-symmetrical Cy5.5-type molecule **3m** incorporates one heterocyclic moiety with a methyl group and another with a six-carbon alkyl chain dimethylamino group. The Cy5-type molecule **3r** shares identical functional groups with **3m** but lacks one phenyl ring on each heterocyclic moiety. The Cy7-type molecule **3s** features two indole-based nitrogen atoms linked by seven methyl units. Molecule **3p** differs from **3m** by incorporating a three-carbon alkyl chain sulfonate group on one heterocyclic moiety. As shown in Fig. 1d, all molecules display a sharp, intense absorption band with a high-energy shoulder. At 640 nm, Cy5 (**3r**) shows maximum absorption at its main peak, while for Cy5.5 (**3m**) this wavelength matches its vibronic shoulder. In contrast, Cy7 (**3s**) shows minimal absorption at 640 nm as this wavelength is far from both its main peak and vibronic shoulder.

To investigate ICR responses to MJH stimulation, we incubated HEK293 cells with MJHs (2 μM) and Fluo-4 (2 μM), which fluoresces green upon calcium binding (Fig. S22 and S23) for 45 min. Fig. 1e and SI Video 1 shows calcium response localization in cells treated with the Cy5.5-type compound **3p** that exhibits the strongest calcium release efficacy. Both MJH and Fluo-4 penetrated the cell membrane during incubation. Stimulation of a single **3p**-treated cell with a 640 nm laser at a power of 0.73 mW, scanning a 5 μm diameter circular area for 250 ms resulted in a significant increase in Fluo-4 fluorescence intensity in the targeted cell. The observed calcium release was laser-power dependent (Fig. S24), repeatable (Fig. S25), and did not cause cell death (Fig. S26 and S27). The time of 250 ms was chosen as the shortest effective irradiation duration that reliably induces robust calcium release without additional benefit at longer exposure times (Fig. S28).

Calcium responses in cells treated with 2 μM of each representative MJH and irradiated with 640 nm light revealed distinct trends (Fig. 1f). In all cases, cellular calcium intensity peaked approximately 20 s post-stimulation before gradually declining. Despite 640 nm corresponding to its absorption maximum, Cy5-type **3r** induced only weak calcium release. Cy7-type **3s** showed modestly higher amplitude than **3r**, while Cy5.5-type **3m** demonstrated enhanced amplitude, underscoring the importance of VDA for the observed calcium responses. The sulfonate-containing molecule **3p** exhibited the highest amplitude in





**Fig. 1** Design, synthesis, and cellular activity of molecular jackhammers. (a) Chemical structure of Cy5.5 showing longitudinal molecular plasmon (LMP, red horizontal arrow) and transversal molecular plasmon (TMP, blue vertical arrow), with calculated induced charge density from time-dependent density functional theory (TD-DFT).<sup>42</sup> (b) General synthetic pathway to molecule **3**. Reaction conditions (a) 1)  $\text{Ac}_2\text{O}$ , reflux; (2)  $\text{NaOAc}$ ,  $\text{Ac}_2\text{O}$ ,  $80\text{ }^\circ\text{C}$ ; reaction conditions (b)  $\text{NaOAc}$ ,  $\text{Ac}_2\text{O}$ ,  $80\text{ }^\circ\text{C}$ . (c) Representative molecular structures **3r**, **3m**, **3s** and **3p**. (d) UV-vis absorption spectra of **3r**, **3m**, and **3s**, highlighting the vibronic mode at the shoulder position (640 nm). (e) Confocal microscopy images showing rapid intracellular calcium release in HEK293 cells treated with molecular jackhammer **3p** ( $2\text{ }\mu\text{M}$ ) and calcium indicator Fluo-4 ( $2\text{ }\mu\text{M}$ ) after 640 nm laser stimulation. Red arrows indicate cells before and after stimulation. Scale bar =  $50\text{ }\mu\text{m}$  (applies to all images). (f) Normalized Fluo-4 fluorescence intensity traces HEK293 cells treated with each MJHs. The solid line represents the average responses of  $n = 6$  independent cells and the shaded region represents the standard error of the mean. **3r**, **3m**, **3s** and **3p** were treated to cells at  $2\text{ }\mu\text{M}$ . All the stimuli were delivered with a 640 nm laser of 250 ms with a power of 0.73 mW to a circular area of diameter  $5\text{ }\mu\text{m}$ . For all the plots, the cyan line indicates the time of stimulus presentation.

Fluo-4 fluorescence, revealing the importance of different ionic substituents to MJH-induced calcium release. Activation of **3p** at both 640 and 680 nm supports that the shoulder peak excitation stimulates greater calcium release than the main absorption peak, further suggesting VDA as the primary mode of response (Fig. S29). Moreover, we employed a control molecule **4** which lacks a positive charge in the conjugated structure and consequently does not support vibronic coupling.<sup>53</sup> Under activation of light with wavelengths from 400 to 640 nm, molecule **4** did not induce any detectable calcium release (Fig. S30). Control experiments using only DMSO solvent showed no detectable calcium responses under identical conditions (Fig. 1f).

To thoroughly investigate the relationship between their molecular structure and calcium release activity, we synthesized 20 symmetrical and non-symmetrical MJHs with diverse functional groups, including benzyl (**3b**), morpholino (**3e**), ester (**3f**), amide (**3g**), and carboxylic acid (**3n**) (Scheme S2).<sup>41</sup> The MJH structures and their normalized calcium release intensities are shown in Fig. 2a and b, respectively. Non-symmetrical Cy5.5 MJHs with sulfonate groups (**3p**, **3q**) demonstrated the highest

calcium release activity. MJHs with dimethylamine groups (**3c**, **3d**, **3l**, **3m**) showed intermediate activity, performing better than those with ester (**3f**), carboxylic acid (**3n**), or other groups (**3b**, **3g**, **3o**, **3i**). As expected, Cy5 and Cy7 derivatives (**3j**, **3k**, **3s**) showed relatively poor performance compared to Cy5.5 derivatives with the same groups, highlighting the importance of wavelength-specific activation of the molecular vibronic shoulder and, therefore, confirming the significance of VDA for the observed calcium responses.

To gain deeper insight into these structure–activity relationships, we calculated the experimental plasmonicity index (EPI) for selected MJHs using our previously established methodology.<sup>53</sup> The EPI was evaluated by monitoring solvent-dependent variations in the absorption maxima of MJHs across environments with different dielectric properties. The magnitude of the solvatochromic response serves as an indicator of collective charge oscillation within the molecular framework. Increased dielectric sensitivity is associated with stronger plasmonic character, allowing distinction between delocalized plasmon-like excitations and localized electronic transitions. This method provides a practical experimental





**Fig. 2** Structure–activity relationships of MJHs and their ability to induce intracellular calcium release. (a) Chemical structures of MJHs used in this study. (b) Normalized Fluo-4 fluorescence intensity in HEK293 cells treated with each MJH (2  $\mu\text{M}$ ). Error bars in the bar graphs represent the standard error of mean of  $n = 3$  experiments (six stimulated cells per experiment). (c) Normalized fluorescence intensity traces of Fluo-4 in C2C12 cells treated with 3p and DMSO with the same volume. The solid line represents the average responses of  $n = 6$  independent cells and the shaded region represents the standard error of the mean. 3p was treated to cells at 2  $\mu\text{M}$ . All the stimuli were delivered with a 640 nm laser of 500 ms with a power of 0.73 mW to a circular area of diameter 5  $\mu\text{m}$ . The cyan line indicates the time of stimulus presentation. The error bars in the bar graphs represent the standard error of mean of  $n = 3$  experiments (six stimulated cells per experiment). Statistical analyses were performed using a two-tailed  $t$ -test. \* $P$  value < 0.05, \*\* $P$  value < 0.01, \*\*\* $P$  value < 0.001. (d and e) Viability of HEK293 cells after incubation with increasing concentrations of MJHs assessed using the CCK-8 assay without light. Error bars represent the standard error of mean of  $n = 3$  experiments. (f)  $\text{IC}_{50}$  values of representative MJHs in the absence of light interpolated from dose–response curves in (d and e). Error bars represent the standard error of mean of  $n = 3$  experiments.

descriptor of molecular plasmonicity. EPI values for representative MJHs were measured (Fig. S31 and S32), and 3p exhibits the highest EPI among the tested compounds. We also noticed that in certain cases, the extent of calcium release induced by MJHs does not directly correlate with the measured EPI. For instance, although 3n exhibits high EPI among the molecules, it induces only a limited calcium release. These results suggest that, although EPI captures the molecular intrinsic properties,

extrinsic factors such as membrane insertion and functional group-dependent interactions with specific cellular components also play critical roles in modulating intracellular calcium release in cellular contexts. Recognizing the limitations of EPI measurements, we anticipate that more rigorous physico-chemical characterization, such as ultrafast spectroscopic approaches in future studies, will refine the mechanism.



While the relative effectiveness of symmetrical *versus* non-symmetrical MJHs remains unclear for most functional groups, a notable observation emerged with compound **3h**. This symmetrical MJH, containing the same functional group as **3p**, showed remarkably poor activity, lower than Cy5 and Cy7 derivatives, likely due to its two negatively charged sulfonate groups impairing cell membrane binding (Fig. S33). Importantly, we demonstrated that MJH **3p**-induced calcium release could be achieved not only in HEK293 cells but also in skeletal muscle C2C12 cells (Fig. 2c and S34),<sup>44</sup> suggesting its broader potential applications across diverse cell types with distinct basal calcium levels. Also, the calcium release in C2C12 cells is repeatable (Fig. S35).

Beyond their ability to elicit calcium release, we also evaluated cytotoxicity as a critical factor for molecule optimization. We selected ten representative MJHs to investigate their impact on cell viability using the CCK-8 assay<sup>54</sup> in HEK293 cells. As shown in Fig. 2d and e, most MJHs exhibited high toxicity at 20  $\mu$ M. However, sulfonate-containing compounds **3p** and **3q** displayed lower toxicity even at higher concentrations. IC<sub>50</sub> values were determined for each MJH (Fig. 2f). The methyl-substituted MJH **3a** showed the highest toxicity. Amine-containing compounds **3l** and **3t** showed decreased yet significant toxicity. Compound **3n** with carboxylic acid showed improved but still suboptimal toxicity profiles. Notably, sulfonate-containing **3p** and **3q** not only exhibited the highest calcium release activity but also the lowest toxicity among all MJHs tested. For instance, MJH **3p** showed a 4.7-fold higher IC<sub>50</sub> value than **3n**. These results establish **3p** as the optimal MJH for inducing intracellular calcium release.

Next, we investigated the biological mechanisms of MJH-induced calcium release. Our previous work showed that MMs induce intracellular calcium release through the IP<sub>3</sub> pathway.<sup>32–34</sup> Given that both MJH and MMs generate mechanical work, we examined whether MJHs operate through the same pathway. The IP<sub>3</sub> pathway is initiated by the activation of G protein-coupled receptors (GPCRs),<sup>55</sup> which triggers the G protein subunit G $_{\alpha q/11}$  to separate from the G $_{\beta\gamma}$  subunit. Activated G $_{\alpha q/11}$  stimulates phospholipase C (PLC),<sup>56</sup> an enzyme responsible for breaking down phosphatidylinositol 4,5-bisphosphate (PIP<sub>2</sub>) into diacylglycerol (DAG) and inositol 1,4,5-trisphosphate (IP<sub>3</sub>). IP<sub>3</sub> then binds to IP<sub>3</sub> receptors (IP<sub>3</sub>R) on the ER membrane, triggering calcium release into the cytosol (Fig. 3a).

To systematically explore these mechanisms, we first examined whether MJH molecules colocalize with cellular membranes system, including plasma membrane, ER membrane and mitochondrial membrane.<sup>57</sup> Colocalization images show the distribution of MitoTracker Green (in green, mitochondria) or ER-Tracker Green (in green, ER) relative to **3p** (in red, Fig. 3b, and S36). The white color denotes the colocalization of green and red signals. Quantitative analysis from nine independent experiments (Fig. 3c) was conducted using Pearson's correlation coefficient (PCC,  $R_{\text{total}}$ ) to assess colocalization between red and green channels (see SI for calculation details). PCC results showed moderate colocalization between **3p** and ER ( $R_{\text{total}} = 0.68 \pm 0.02$ ) and slightly lower colocalization with mitochondria ( $R_{\text{total}} = 0.62 \pm 0.03$ ). Both tM1 and tM2

(calculated by Mander's overlap coefficient, MOC, see SI for calculation details) values were approximately 90% for **3p** with either ER or mitochondria, confirming decent colocalization. While these results suggest that MJHs are broadly distributed within cellular membranes rather than specifically enriched in a single organelle, the presence of MJHs in the ER membrane, where IP<sub>3</sub>R is located, is consistent with a potential role of the IP<sub>3</sub> pathway in the observed calcium response.

To identify the calcium source, we compared the responses of HEK293 cells treated with **3p** and stimulated by light in the presence and absence of extracellular calcium. Identical calcium release magnitudes under both conditions (Fig. 3d) indicated an intracellular source. We then used specific inhibitors to probe the role of the IP<sub>3</sub> pathway in the intracellular calcium release elicited by light-activated **3p** (Fig. 3e–h). Treatment with either FR900359 (ref. 58) (G $\alpha$  inhibitor) or gallein<sup>59</sup> (G $\beta\gamma$  inhibitor), alone or in combination, reduced calcium release by ~50%. The PLC antagonist U-73122 (ref. 60) almost completely blocked calcium release, strongly implicating IP<sub>3</sub> signaling as the primary mechanism. However, the partial inhibition of MJH-induced calcium release by G protein inhibitors compared to complete inhibition by PLC antagonists, suggests that this effect may reflect incomplete pharmacological suppression of GPCR signaling or the contribution of additional calcium-dependent mechanisms, such as calcium influx (*e.g.*, store-operated calcium entry, SOCE).<sup>58,61–63</sup> Consistent with previous reports that certain GPCRs can respond to mechanical stimulation,<sup>64,65</sup> MJH-induced mechanical perturbations may engage GPCR-associated signaling and thereby contribute to the observed calcium responses. These results highlight the potential of MJHs as a unique tool for modulating GPCR-associated calcium signaling.

To investigate potential photodynamic effects, we examined the role of ROS in MJH-induced calcium release. HEK293 cells were treated with ROS scavengers (10 mM thiourea and 0.25 mM sodium azide or MnTBAP up to 50  $\mu$ M) for 45 min,<sup>39,41,42,66</sup> and ROS levels were measured using CellROX Green Reagent.<sup>67</sup> The fluorescence of CellROX increases when oxidized by ROS and is widely used to measure ROS generation in living cells. Although scavengers reduced cellular ROS levels, calcium release remained unchanged compared to untreated controls (Fig. 3i, j, S37 and S38), indicating that photodynamic effects do not drive MJH-induced calcium release. Moreover, while **3p** induced the strongest calcium release among tested molecules (**3a**, **3d**, **3p**, and **3n**), it did not generate higher ROS levels (Fig. 3k), further confirming that ROS generation is not the primary mechanism underlying MJH-induced calcium release.

The photothermal effect was also studied. HEK293 cells were transfected to be enabled to express a thermosensitive fluorescent protein ELP-TEMP. ELP-TEMP fluoresces both blue and green, and the intensities decrease when the temperature increases.<sup>68</sup> Since the MJH molecules exhibit enhanced blue fluorescence upon light irradiation, we used the green channel to characterize the temperature changes during the experiment process. The result showed that despite of different abilities to induce calcium release, all the MJHs tested showed identical and negligible thermal effects (Fig. S39), indicating that the





**Fig. 3** Mechanistic studies of MJH-induced calcium release. (a) Schematic diagram of IP<sub>3</sub> pathway for calcium release. Prepared with Biorender.com (<http://Biorender.com>). (b) Super-resolution microscopy of HEK293 cells treated with 3p (2 μM) and either MitoTracker Green (2 μM) or ER-Tracker Green (2 μM). Green, MitoTracker Green or ER-Tracker Green; red, 3p; white, colocalization of green and red signals. Scale bar = 20 μm and is the same for both figures. (c) Colocalization analysis results from nine independent experiments. R<sub>total</sub>, Pearson's correlation coefficient; tM1, Manders split coefficients of green channel with threshold; tM2, Manders split coefficients of red channel with threshold. (d–i) Calcium release in 3p-treated HEK293 cells under various conditions: calcium-free PBS (d), FR900359 (10 μM) (e), Gallein (100 μM) (f), FR900359 (10 μM) + Gallein (100 μM) (g), U-73122 (10 μM) (h), and ROS scavengers (10 mM Thiourea + 0.25 mM Sodium azide) (i). (j) Normalized CellROX Green Reagent (2 μM) fluorescence in HEK293 cells treated with 3p (2 μM) with or without ROS scavengers. (k) Normalized CellROX Green Reagent (2 μM) fluorescence in HEK293 cells treated with 3a, 3d, 3p, or 3n (all at 2 μM). Statistical analyses were performed on 3a and 3p groups. Solid lines represent the average responses of n = 6 independent cells and the shaded region represents the standard error of the mean. Stimuli were presented after 30 s of imaging in each case. All the stimuli were delivered with a 640 nm laser of 250 ms with a power of 0.73 mW to a circular area of diameter 5 μm. For all the plots, the cyan line indicates the time of stimulus presentation. The control and experimental groups were imaged and stimulated on the same day using the same conditions and with the same batch of cells. The error bars in the bar graphs represent the standard error of the mean of n = 3 experiments (at least six stimulated cells per experiment). Statistical analyses were performed using a two-tailed t-test. \*P value < 0.05, \*\*P value < 0.01, \*\*\*P value < 0.001.

photothermal effect is also unlikely to be the reason for MJHs-induced calcium release.

To understand how the presence of different functional groups in MJHs affects their calcium release activity, we

performed molecular dynamics simulations examining MJH interactions with cell membranes. MJHs with different functional groups showed varying depths of center of mass (COM) in the cell membrane (Fig. 4a–e and S40), with a clear negative



correlation between COM depth and the efficiency of ICR induction. This relationship can be explained by the interactions of MJHs and the components of cell membranes.

Cell membranes consist primarily of phospholipids, with hydrophobic hydrocarbon tails oriented inward and hydrophilic phosphoryl head groups exposed outward. The hydrophobic core of MJHs preferentially inserts into the membrane's interior, while the polar or charged functional groups on the arm tend to interact with the phosphoryl heads of phospholipids. For MJH **3a**, which lacks functional groups on its arm, exhibits the deepest COM depth as the entire molecule embeds within the membrane. In contrast, **3m**, featuring a positively charged amine group on its arm, forms electrostatic interactions with the negatively charged phosphoryl head groups, limiting its insertion depth. The shallowest insertion is observed for MJHs **3p** and **3q**, which contain negatively charged sulfonate groups on their arms. Electrostatic repulsion between these charged sulfonate groups and the phosphoryl group prevents the arms of **3p** and **3q** from embedding into the membrane. As a result, only the hydrophobic core of these molecules inserts into the membrane, leading to the shallowest COM depths for **3p** and **3q**.

The observed negative correlation between COM depth and ICR induction efficiency could be rationalized by considering the spatial relationship between MJHs and GPCRs in the membrane. The key mechanosensitive region of GPCRs, helix 8,<sup>69</sup> is positioned near the membrane boundary. We hypothesize that MJHs with shallower COM depths are situated closer to helix 8, which enhances their ability to interact with this

mechanosensitive region (Fig. 4f). While our simulations do not include GPCR proteins and therefore do not directly capture protein–molecule interactions, this spatial proximity may increase the likelihood of mechanical perturbation of membrane-associated domains such as helix 8. This model provides a plausible explanation for the enhanced activity observed for molecules with shallower insertion depths.

Having elucidated the mechanism, we next investigated whether MJH-induced ICR can promote muscle contraction in C2C12 myotubes, differentiated muscle cells capable of contraction in response to increased intracellular calcium.<sup>70</sup> In resting skeletal muscle (Fig. 5a), tropomyosin blocks myosin-binding sites on actin, preventing the interaction between myosin heads and actin filaments. When an action potential triggers calcium release from the sarcoplasmic reticulum, calcium binds to troponin, causing tropomyosin to shift and expose these binding sites. Myosin heads then form cross-bridges with actin and perform a power stroke, pulling actin filaments toward the sarcomere center, resulting in contraction. ATP binds to myosin, causing it to detach from actin, and its hydrolysis resets the myosin head for the next cycle. Muscle relaxation occurs as calcium is actively pumped back into the sarcoplasmic reticulum, restoring the resting state.<sup>71–73</sup>

As predicted, MJH **3p** induced localized calcium release in C2C12 myotubes (Fig. 5b and SI Video 2), leading to myotube contractions (Fig. 5c and SI Videos 3 and 4). No calcium release or contraction occurred in myotubes treated with DMSO, **3p** combined with U-73122, or **3n** alone (Fig. 5c–g and S41). We quantified cellular contraction using image subtraction

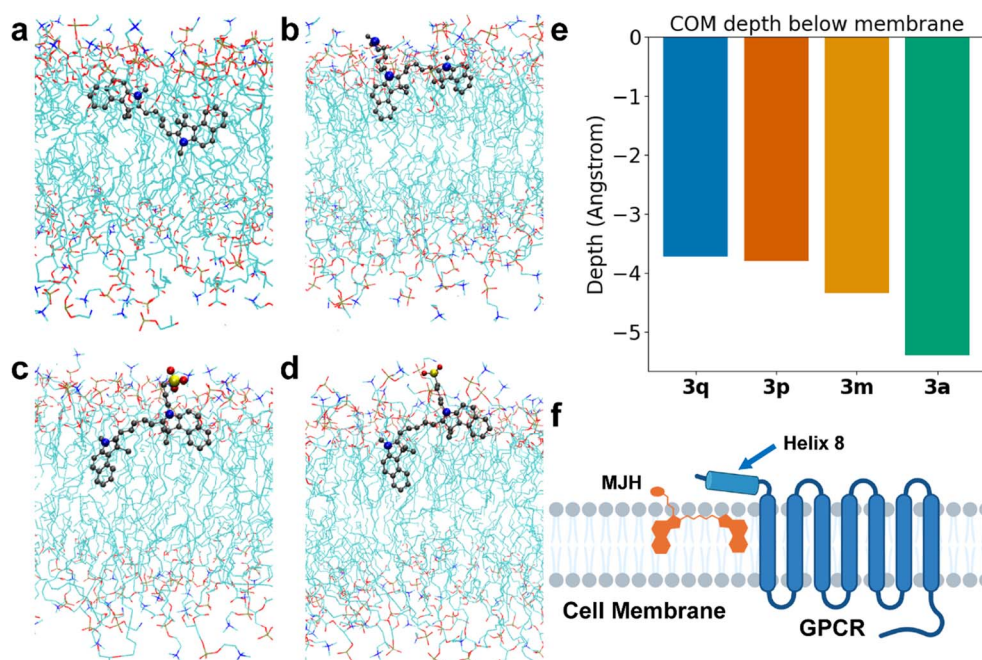


Fig. 4 Positioning of different MJHs in the cell membrane and its relation to their biological activity. (a–d) Molecular dynamics simulation results showing the positioning of MJHs **3a** (a), **3m** (b), **3p** (c), and **3q** (d) within the cell membrane. (e) Quantitative analysis of center-of-mass (COM) depths for the four MJH molecules, showing deepest insertion for **3a** (no functional groups), intermediate for **3m** (single amine group), and shallowest for **3p** and **3q** (sulfonate groups). (f) Schematic representation of the positioning of GPCR proteins and MJH molecules within the cell membrane.





Fig. 5 MJHs cause skeletal muscle cell contraction by inducing intracellular calcium release. (a) Schematic of the skeletal muscle cell contraction. Prepared with Biorender.com (<http://Biorender.com>). (b, d and f) Confocal microscopy images of C2C12 myotubes treated with Fluo-4 (2  $\mu$ M) and 3p (2  $\mu$ M) (b), DMSO (d) or 3p (2  $\mu$ M) plus U-73122 (10  $\mu$ M) (f). Images captured at  $t = 0, 35$ , and  $140$  s; laser stimulation at  $t = 30$  s. Red arrows indicate stimulated myotubes. Scale bar = 20  $\mu$ m and is the same for all images. (c, e and g) Brightfield images after image-subtraction process of C2C12 myotubes treated with Fluo-4 (2  $\mu$ M) and 3p (2  $\mu$ M) (c), DMSO (e) and 3p (2  $\mu$ M) and U-73122 (10  $\mu$ M) (g). Scale bar = 20  $\mu$ m and is the same for all images. (h) The magnitude of C2C12 myotubes contraction treated with 3p (2  $\mu$ M), 3n (2  $\mu$ M), DMSO and 3p (2  $\mu$ M) plus U-73122 (10  $\mu$ M). In each experiment, 20 myotubes were stimulated, and the average magnitude of contraction is shown in the bar graph. Stimulated cells were irradiated with a 640 nm laser for 1.5 s with a power of 0.73 mW to a circular area of diameter 5  $\mu$ m. Statistical analyses were performed using a two-tailed  $t$ -test. \* $P$  value < 0.05, \*\* $P$  value < 0.01, \*\*\* $P$  value < 0.001.

method, where pixel brightness corresponds to movement magnitude (see SI for calculation details). This analysis confirmed that 3p induced significant myotube contraction, while cells treated with DMSO, 3n, or 3p + U-73122 showed no contraction (Fig. 5h). These results demonstrate that MJHs can induce skeletal muscle contraction through intracellular calcium release, suggesting potential applications for muscle-related study or therapy.

To investigate whether calcium release could be induced *in vivo*, we used the model organism *Hydra vulgaris*, which is well-suited for this study due to its optical transparency, simple

multicellular organization, and calcium-dependent physiological behavior that can be directly visualized using genetically encoded calcium indicators. *Hydra* is a millimeter-sized freshwater organism with two layers of tissue separated by an extracellular matrix, exhibiting radial symmetry with a cylindrical body structure. One end of its body connects to an oral region and tentacles, while the opposite end is connected to the foot (Fig. 6a).<sup>45,46</sup> The transgenic *Hydra* line used in this study were genetically engineered to express GCaMP7b, enabling visualization of calcium dynamics in the endothelial epitheliomuscular cells (see Methods). *Hydra* were treated with





**Fig. 6** MJH induces regional calcium release *in vivo* in *Hydra*. (a) Image of a *Hydra* positioned within a microfluidic chamber, with its anatomical regions clearly labeled. Scale bar = 500  $\mu\text{m}$ . (b) Distribution of **3p** (6  $\mu\text{M}$ ) in *Hydra* after 24 h incubation, scale bar = 200  $\mu\text{m}$ . (c) Representative confocal microscopy images of *Hydra* treated with **3p** (6  $\mu\text{M}$ ), **3n** (1  $\mu\text{M}$ ), or DMSO. Images captured pre-stimulation and at  $t = 5$  s, and  $t = 15$  s; laser stimulation at  $t = 0$  s. Red arrows indicate stimulated areas. Scale bar = 200  $\mu\text{m}$  and is the same for all images. (d) Normalized GCaMP7b fluorescence intensity traces of *Hydra* treated with MJH **3p**, **3n**, DMSO or **3p** without light (dark control). The bold, black-colored traces represent the average trace ( $n = 25$  across five *Hydra*). For all the plots, the cyan lines indicate the time of stimulus with 640 nm light ( $9.0 \times 10^2 \text{ W cm}^{-2}$ , 3 s).



each MJH (**3p** as the most potent MJH and **3m** and **3n** serve as control groups) and incubated for 24 h prior to experiments. After incubation, the MJH molecules effectively penetrated *Hydra* and distributed uniformly throughout its tissues (Fig. 6b).

Upon 640 nm laser stimulation, localized calcium release, indicated by green color fluorescence, was observed exclusively in *Hydra* treated with **3p**. In contrast, *Hydra* treated with **3m**, **3n**, or DMSO exhibited minimal or undetectable responses (Fig. 6c, S42, SI Videos 5 and 6). Quantitative analysis of calcium release confirmed that only the **3p**-treated group exhibited significant calcium elevation following stimulation (Fig. 6d and S43). The gradual increases in GCaMP7b fluorescence of **3n**-, **3m**- and DMSO-treated groups during the latter half of the experiment can be attributed to the spontaneous movements of *Hydra*, which exhibits calcium waves.<sup>74</sup> These spontaneous increases were distinguished from MJH-induced responses by their full-body fluorescence accompanying *Hydra* contraction uncorrelated to the stimulation, whereas MJH-induced responses remained localized to the laser-irradiated area.

Notably, **3p** demonstrated remarkable biocompatibility in *Hydra* compared to other MJHs. While *Hydra* treated with **3n** and **3m** at concentrations of 3  $\mu\text{M}$  or higher all disintegrated and died, limiting their usable concentration to 1  $\mu\text{M}$ , *Hydra* treated with up to 12  $\mu\text{M}$  **3p** remained healthy and continued normal growth post-experiments (Fig. S44). These results demonstrate that MJH **3p** effectively induces calcium release *in vivo* with minimal toxicity, suggesting potential applications in biological systems.

## Conclusion

Here, we demonstrate that cyanine dyes can function as MJHs to modulate intracellular calcium signaling with high spatio-temporal resolution. These light-activated molecules generate mechanical forces through vibronic-driven action (VDA), activating mechanosensitive  $\text{IP}_3$  pathways to trigger intracellular calcium release in skeletal muscle cells and *in vivo* in *Hydra*. Structure–activity relationship studies across 20 synthesized MJHs revealed that sulfonate-containing derivatives exhibited optimal calcium release efficiency with minimal toxicity. Molecular dynamics simulations provided mechanistic insights into the advantages conferred by the sulfonate group. The combination of intrinsic fluorescence for precise detection and red-light (640 nm) activation for deep tissue penetration and reduced phototoxicity, and the MJHs have the potential to become biologically safe and practically useful tools in research and, eventually, broader biomedical contexts. These properties, together with their demonstrated efficacy *in vitro* and *in vivo*, establish MJHs as a new class of molecular tools for controlling cellular signaling with potential therapeutic applications.

## Methods

### General information of synthesis

All glassware was oven-dried overnight prior to use. All chemicals were purchased from commercial suppliers and used

without further purification. All reactions were carried out under Argon atmosphere unless otherwise noted. Flash column chromatography was performed using 230–400 mesh silica gel. Thin layer chromatography (TLC) was performed using glass plates pre-coated with silica gel F<sub>254</sub> 0.25 mm layer thickness purchased from EM Science. <sup>1</sup>H-NMR spectra were recorded at 500 MHz and 600 MHz, <sup>13</sup>C-NMR spectra were recorded at 126 MHz and 151 MHz. Chemical shifts ( $\delta$ ) are reported in ppm relative to tetramethylsilane (TMS). HRMS (ESI) were conducted on Agilent Quadrupole-TOF LC/MS spectrometer. Synthesis and molecular characterization data are in the SI.

### Synthesis of symmetric MJHs (3a–3k)

**General procedure a.** A screw-capped vial was charged with *N*-heterocyclic salt **1** (2 equiv.), **2** (1 equiv.) and NaOAc (3 equiv.) dissolved in anhydrous  $\text{Ac}_2\text{O}$ . The mixture was heated at 80 °C under argon atmosphere until the reaction was completed. The solvent was removed under reduced pressure and the crude product was purified by silica gel chromatography with DCM/MeOH as eluent to afford MJHs **3**.

### Synthesis of non-symmetric MJHs (3l–3s)

**General procedure b.** *N*-Heterocyclic salt **1** (1 equiv.) and **2** (1 equiv.) were added to a solution of  $\text{Ac}_2\text{O}$ . The reaction mixture was stirred at 110 °C for 1 h, then cooled to room temperature and a large volume of diethyl ether was added, the reaction mixture was filtered off to afford solid. Subsequently, to a screw-capped vial charged with the residue, another different salt **1'** (1 equiv.), anhydrous sodium acetate (3 equiv.) in anhydrous  $\text{Ac}_2\text{O}$  and stirred at 80 °C under the argon atmosphere. After the reaction was complete, the product was purified by silica gel column chromatography with DCM/MeOH as eluent to afford MJHs **3**.

### UV-vis and emission spectra measurements

UV-vis spectra were acquired in a 1-cm quartz cuvette using a Shimadzu UV-2450 spectrophotometer. Molecules were dissolved in solvent (water, DMSO, methanol, ethanol, and isopropanol at 1 mM). Steady-state photoluminescence emission spectra were obtained on a Horiba Jobin Yvon FluoroLog exciting with a 450 W Xe lamp in front face (FF) mode. Relative quantum yield measurements were determined using ICG in DMSO as a reference ( $\Phi = 0.13$ ) as a standard. Photoluminescence quantum yield of unknown sample can be calculated by this equation,  $\Phi_{\text{std}}$  is the quantum yield of standard,  $A_{\text{std}}$  is absorbance intensity,  $I_{\text{std}}$  is emission intensity,  $n_{\text{std}}$  is the refractive index of solvent.

$$\Phi_{\text{unk}} = \Phi_{\text{std}} \times \frac{A_{\text{std}}}{I_{\text{std}}} \times \frac{I_{\text{unk}}}{A_{\text{unk}}} \times \frac{n_{\text{unk}}^2}{n_{\text{std}}^2}$$

### Molecule preparation

MJH stock solutions were prepared in DMSO and diluted 1 : 1000 into the culture medium, resulting in a final DMSO concentration of 0.1% (v/v) in all MJH-treated conditions. The



DMSO control was defined as 0.1% (v/v) DMSO in the absence of MJHs.

### Cell lines and culture conditions

HEK293 cells and C2C12 cells were obtained from the American Type Culture Collection (ATCC) and cultured with DMEM Medium (Corning, MT10013CV), the medium was supplemented with 10% FBS (Fetal Bovine Serum, Corning, MT35010CV) and 1% penicillin-streptomycin (Gibco, 15-140-122) in a humidified incubation system containing 5% CO<sub>2</sub> at 37 °C. For the passage step, cells are detached with 0.05% trypsin-EDTA (Gibco, 25-300-054) and passaged every 2–3 days. For C2C12 differentiation, after the cells grew to near full confluency, media replacement was done with differentiation media (DMEM with 2% horse serum and 1% antimycin), and fresh differentiation media replacement was done every 1–2 d.

### Confocal microscopy

Confocal experiments were done with a Nikon A1-Rsi confocal microscope. Cells were plated in the  $\mu$ -Plate 24-well plate (ibidi) with a concentration of 120 000 per well (HEK293 cells) or 7000 per well (C2C12 cells). Before imaging, the media was replaced with ECB-JR imaging buffer (119 mM NaCl, 5 mM KCl, 10 mM HEPES, 2 mM CaCl<sub>2</sub>, 1 mM MgCl<sub>2</sub> (pH 7.2); 320 mOsm). Experiments were conducted 1–2 days after cells were plated except for the experiments with differentiated C2C12 cells, which took place after 4–9 days when differentiation media was replaced.

In experiments with IP<sub>3</sub> pathway inhibitors, cells were initially incubated with MJHs and Fluo-4 and irradiated with 640 nm light as a positive control. After that, the inhibitor was immediately added to the well, and the cells were again irradiated with 640 nm light.

In experiments of ROS generation measurements, MJHs and CellROX were incubated with cells for 45 min. In experiments that ROS scavengers were used, the scavengers (thiourea 10 mM, sodium azide 0.25 mM) were incubated with cells for 45 min, and after replacing ECB-JR, the same amount of ROS scavengers were also added into ECB-JR solution.

In experiments testing photothermal effects, HEK293 cells were transfected with TEMP/pcDNA3.1(–) using lipofectamine 3000. The plasmid was a gift from Takeharu Nagai (Addgene plasmid # 178434; <https://n2t.net/addgene:178434>; RRID:Addgene\_178434). The temperature was calibrated with a VAHeat Smart Substrate (part number SmS-SR) purchased from Interference. VAHeat provides precise control of real-time temperature on its glass coverslip. We plated HEK293 on its accompanying cell culture reservoir and transfected them with ELP-TEMP. We increased the temperature from room temperature (22 °C) to 40 °C incrementally by 1 °C, and imaged the green channel at each degree on the Nikon A1 Rsi confocal microscope. 6 Individual cells were measured at each experiment, and 3 replications were conducted. We normalized their fluorescence to the initial 22 °C fluorescence to get the curve in Fig. S39a.

### Super-resolution microscopy

Super-resolution microscope experiments were done with a Zeiss LSM980 Airyscan-2 microscope using Single Laser Point Scanning mode. Fluorescent dyes were added into the media with a 1 : 1000 dilution. After incubating with MJH molecule and the fluorescent dyes for 45 min, the media was replaced with ECB-JR, and the cells were imaged on the airyscan microscope. The dyes for organelle-labeling were obtained from Thermo Fisher. MitoTracker Green FM was used to label mitochondria and ER-Tracker Green was used to label ER.

### CCK-8 assay protocol on HEK293 cells

Dispense 100  $\mu$ L cell suspension (20000 cells per well) in a 96-well plate. Pre-incubate the plate overnight in incubator (37 °C, 5% CO<sub>2</sub>). Add 100  $\mu$ L media with various concentrations of MJHs to be tested to the plate (e.g., 1  $\mu$ M concentration preparation : 1  $\mu$ L of 1 mM of molecules dissolved in 0.5 mL media and add 100  $\mu$ L to each well of the plate without replacement of media) and incubate the plate for 3 h in the incubator. Add 10 vol% (20  $\mu$ L) of CCK-8 solution to each well of the plate and keep incubating the plate for 1.5 hours in the incubator. Finally, measure the absorbance at 450 nm using a microplate reader.

### Molecular dynamics simulation

We performed molecular dynamics (MD) simulations to determine the equilibrium distance of dyes below the membrane surface. The details are provided in the SI (Molecular dynamics simulation section).

### Cell contraction magnitude quantification

Contraction magnitude was quantified using the image subtraction method. The first frame of the brightfield image was subtracted from all consecutive brightfield images, resulting in a new series of images where the first frame is totally black, and the pixel intensity represents the difference in each frame compared to the very first frame. The movements are positively correlated to the brightness of the pixel.

In this new series of images, three unstimulated regions were chosen and averaged as background to rule out the effect of drift and normal cell activities during the course of the experiment. The background trace was subtracted from traces of each stimulated cells to get corrected traces. The corrected traces were processed in the same way as the fluorescence measurements ( $\Delta F/F_0$ ) and zeroed at the point of stimulation.

### Hydra transgenic line and maintenance

Hydra were maintained in a Hydra medium containing CaCl<sub>2</sub>·2H<sub>2</sub>O (1.00 mM), MgCl<sub>2</sub>·6H<sub>2</sub>O (0.10 mM), KNO<sub>3</sub> (0.03 mM), NaHCO<sub>3</sub> (0.50 mM) and MgSO<sub>4</sub> (0.08 mM) in deionized water in a light-cycled (12 h : 12 h light : dark) incubator at 18 °C. Hydra were fed with freshly hatched Artemia nauplii (Brine Shrimp Direct) three times a week. All the experiments were performed at room temperature after starving the animals for two days. The transgenic line expressing the calcium indicator GCaMP7b



under the Efl $\alpha$  promoter in the endodermal epitheliomuscular cells was used for all experiments.

### Hydra stimulation experiments

All *Hydra* experiments were performed two days after feeding. Before experiments, *Hydra* were incubated with MJH molecules for 24 h. Stimulation and imaging was done with a Nikon A1 confocal microscope and at room temperature. *Hydra* were irradiated by 640 nm laser with a power of  $9.0 \times 10^2 \text{ W cm}^{-2}$  at a circular region 10  $\mu\text{m}$  diameter for 3 s. Stimulations were performed when *Hydra* were relaxed to avoid their spontaneous contractions.

### Traces plot data analysis

The acquired images of confocal and super-resolution experiments were analyzed with Fiji. The fluorescence intensity was measured at each time point and processed using Python code. In the code,  $F_0$  was defined as the average fluorescence intensity in the first ten time points. The traces were normalized by  $F_0$  and then presented by  $\Delta F/F_0$ . Then the traces of each experiment were averaged and plotted with the gray regions as standard error of the mean.

### Statistical analysis

A minimum of three individual trials were conducted for each experiment and at least 6 cells were stimulated in each trial. The average maximum of each trial was calculated by a Python code to get the mean and standard error of the mean. Error bars and  $p$  values were calculated by GraphPad Prism.

## Author contributions

B. L., Y. R. and J. M. T. conceived the idea and designed the experiments. Y. R. conducted most of the biological experiments and characterizations with the help of B. L., D. A., J. L. B., C. A. O., A. S., G. C., S. X., T. S. and Z. W. B. L. conducted organic synthesis, characterizations and cytotoxicity tests. B. L., L. Z. and A. A. M. conducted UV-vis and spectroscopy measurement. V. V. and A. K. conducted the molecular dynamics simulations. Y. R., S. K. and J. T. R. conducted the *Hydra* experiments. B. L., Y. R. and J. M. T. wrote the original draft. J. M. T. was responsible for the funding acquisition, project administration, and overseeing of all phases of the project. All authors discussed the results and commented on the manuscript.

## Conflicts of interest

Rice University owns intellectual property on the use of MJH for permeabilization of cell membranes and use in calcium release. The intellectual property is presently unlicensed.

## Data availability

All data needed to support the conclusions in the paper are present in the manuscript and/or supplementary information

(SI). Supplementary information: experimental procedures and characterization data for new compounds (NMR spectra, photophysical properties). See DOI: <https://doi.org/10.1039/d5sc09717g>.

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## References

- 1 M. J. Berridge, M. D. Bootman and H. L. Roderick, *Nat. Rev. Mol. Cell Biol.*, 2003, **4**, 517–529.
- 2 M. J. Berridge, P. Lipp and M. D. Bootman, *Nat. Rev. Mol. Cell Biol.*, 2000, **1**, 11–21.
- 3 D. E. Clapham, *Cell*, 2007, **131**, 1047–1058.
- 4 L. Fedrizzi, D. Lim and E. Carafoli, *Biochem. Mol. Biol. Educ.*, 2008, **36**, 175–180.
- 5 A. E. West, W. G. Chen, M. B. Dalva, R. E. Dolmetsch, J. M. Kornhauser, A. J. Shaywitz, M. A. Takasu, X. Tao and M. E. Greenberg, *Proc. Natl. Acad. Sci. U. S. A.*, 2001, **98**, 11024–11031.
- 6 B. K. Puri, Calcium Signaling and Gene Expression, in *Calcium Signaling*, ed. M. Islam, Springer, Cham, 2020, vol. 1131, pp. 537–545.
- 7 M. J. Berridge, *BioEssays*, 1995, **17**, 491–500.
- 8 P. L. Greer and M. E. Greenberg, *Neuron*, 2008, **59**, 846–860.
- 9 A. C. Dolphin and A. Lee, *Nat. Rev. Neurosci.*, 2020, **21**, 213–229.
- 10 A. G. Szent-Györgyi, *Biophys. J.*, 1975, **15**, 707–723.
- 11 I. Y. Kuo and B. E. Ehrlich, *Cold Spring Harbor Perspect. Biol.*, 2015, **7**, a006023.
- 12 A. J. Morgan and R. Jacob, *Biochem. J.*, 1994, **300**, 665–672.
- 13 O. Thastrup, P. J. Cullen, B. K. Drøbak, M. R. Hanley and A. P. Dawson, *Proc. Natl. Acad. Sci. U. S. A.*, 1990, **87**, 2466–2470.
- 14 T. G. Favero and J. J. Abramson, *Cell Calcium*, 1994, **15**, 183–189.
- 15 K. D. Ryder and R. L. Duncan, *J. Bone Miner. Res.*, 2001, **16**, 240–248.
- 16 R. R. Resende, K. N. Gomes, A. Adhikari, L. R. G. Britto and H. Ulrich, *Cell Calcium*, 2008, **43**, 107–121.
- 17 L. Khatib, D. E. Golan and M. Cho, *FASEB J.*, 2004, **18**, 1903–1905.
- 18 S. Buvinic, G. Almarza, M. Bustamante, M. Casas, J. López, M. Riquelme, J. C. Sáez, J. P. Huidobro-Toro and E. Jaimovich, *J. Biol. Chem.*, 2009, **284**, 34490–34505.
- 19 S. Staehelke, M. Bielfeldt, J. Zimmermann, M. Gruening, I. Barke, T. Freitag, S. Speller, U. Van Rienen and B. Nebe, *Cells*, 2022, **11**, 2650.



- 20 M. J. Sanderson, A. C. Charles and E. R. Dirksen, *Cell Regul.*, 1990, **1**, 585–596.
- 21 M. Tsutsumi, K. Inoue, S. Denda, K. Ikeyama, M. Goto and M. Denda, *Cell Tissue Res.*, 2009, **338**, 99–106.
- 22 R. Zucker and J. Nolte, *Nature*, 1978, **274**, 78–80.
- 23 G. H. Gold and J. I. Korenbrot, *Proc. Natl. Acad. Sci. U. S. A.*, 1980, **77**, 5557–5561.
- 24 P. Gorostiza and E. Y. Isacoff, *Science*, 2008, **322**, 395–399.
- 25 T. Ishii, K. Sato, T. Kakumoto, S. Miura, K. Touhara, S. Takeuchi and T. Nakata, *Nat. Commun.*, 2015, **6**, 8021.
- 26 G. Ma, S. Wen, L. He, Y. Huang, Y. Wang and Y. Zhou, *Cell Calcium*, 2017, **64**, 36–46.
- 27 V. García-López, F. Chen, L. G. Nilewski, G. Duret, A. Aliyan, A. B. Kolomeisky, J. T. Robinson, G. Wang, R. Pal and J. M. Tour, *Nature*, 2017, **548**, 567–572.
- 28 V. García-López, D. Liu and J. M. Tour, *Chem. Rev.*, 2020, **120**, 79–124.
- 29 D. Arnold, B. Li, J. L. Beckham, K. Lopez-Jaime, C. Ayala-Orozco, Y. Rui, G. Chaudhry, J. M. Seminario and J. M. Tour, *J. Am. Chem. Soc.*, 2025, **147**, 22387–22401.
- 30 A. L. Santos, D. Liu, A. K. Reed, A. M. Wyderka, A. van Venrooy, J. T. Li, V. D. Li, M. Misiura, O. Samoylova, J. L. Beckham, C. Ayala-Orozco, A. B. Kolomeisky, L. B. Alemany, A. Oliver, G. P. Tegos and J. M. Tour, *Sci. Adv.*, 2022, **8**, eabm2055.
- 31 J. L. Beckham, A. R. van Venrooy, S. Kim, G. Li, B. Li, G. Duret, D. Arnold, X. Zhao, A. L. Santos, G. Chaudhry, J. T. Robinson and J. M. Tour, *Nat. Nanotechnol.*, 2023, **18**, 1051–1059.
- 32 M. J. Berridge, *Nature*, 1993, **361**, 315–325.
- 33 K. Mikoshiba, *J. Neurochem.*, 2007, **102**, 1426–1446.
- 34 E. Decrock, M. De Bock, N. Wang, A. K. Gadicherla, M. Bol, T. Delvaeye, P. Vandenabeele, M. Vinken, G. Bultynck, D. V. Krysko and L. Leybaert, *Biochim. Biophys. Acta, Mol. Cell Res.*, 2013, **1833**, 1772–1786.
- 35 C. Ash, M. Dubec, K. Donne and T. Bashford, *Laser Med. Sci.*, 2017, **32**, 1909–1918.
- 36 A. Mishra, R. K. Behera, P. K. Behera, B. K. Mishra and G. B. Behera, *Chem. Rev.*, 2000, **100**, 1973–2012.
- 37 W. Sun, S. Guo, C. Hu, J. Fan and X. Peng, *Chem. Rev.*, 2016, **116**, 7768–7817.
- 38 L. Štacková, E. Muchová, M. Russo, P. Slaviček, P. Štacko and P. Klán, *J. Org. Chem.*, 2020, **85**, 9776–9790.
- 39 C. Ayala-Orozco, D. Galvez-Aranda, A. Corona, J. M. Seminario, R. Rangel, J. N. Myers and J. M. Tour, *Nat. Chem.*, 2024, **16**, 456–465.
- 40 C. Ayala-Orozco, G. Li, B. Li, V. Vardanyan, A. B. Kolomeisky and J. M. Tour, *Adv. Mater.*, 2024, **36**, 2309910.
- 41 B. Li, C. Ayala-Orozco, T. Si, L. Zhou, Z. Wang, A. A. Martí and J. M. Tour, *Adv. Sci.*, 2024, **11**, 2405965.
- 42 C. Ayala-Orozco, V. Vardanyan, K. Lopez-Jaime, Z. Wang, J. M. Seminario, A. B. Kolomeisky and J. M. Tour, *RSC Mechanochem.*, 2025, **2**, 706–722.
- 43 G. Shaw, S. Morse, M. Ararat and F. L. Graham, *FASEB J.*, 2002, **16**, 869–871.
- 44 D. K. McMahon, P. A. Anderson, R. Nassar, J. B. Bunting, Z. Saba, A. E. Oakeley and N. N. Malouf, *Am. J. Physiol.: Cell Physiol.*, 1994, **266**, C1795–C1802.
- 45 T. Fujisawa, *Dev. Dyn.*, 2003, **226**, 182–189.
- 46 S. Ghaskadbi, *Resonance*, 2020, **25**, 1197–1213.
- 47 C. Shi, J. B. Wu and D. Pan, *J. Biomed. Opt.*, 2016, **21**, 050901.
- 48 A. P. Gorka, R. R. Nani and M. J. Schnermann, *Acc. Chem. Res.*, 2018, **51**, 3226–3235.
- 49 T. C. Pham, V.-N. Nguyen, Y. Choi, S. Lee and J. Yoon, *Chem. Rev.*, 2021, **121**, 13454–13619.
- 50 Y. Cui, A. Lauchner, A. Manjavacas, F. J. García de Abajo, N. J. Halas and P. Nordlander, *Nano Lett.*, 2016, **16**, 6390–6395.
- 51 G. Orlandi and W. Siebrand, *J. Chem. Phys.*, 1973, **58**, 4513–4523.
- 52 F.-F. Kong, X.-J. Tian, Y. Zhang, Y.-J. Yu, S.-H. Jing, Y. Zhang, G.-J. Tian, Y. Luo, J.-L. Yang, Z.-C. Dong and J. G. Hou, *Nat. Commun.*, 2021, **12**, 1280.
- 53 C. Ayala-Orozco, B. Li, G. Li and J. M. Tour, *Chem. Sci.*, 2025, **16**, 2718–2729.
- 54 K. Chamchoy, D. Pakotiprapha, P. Pumirat, U. Leartsakulpanich and U. Boonyuen, *BMC Biochem.*, 2019, **20**, 4.
- 55 V. Dhyani, S. Gare, R. K. Gupta, S. Swain, K. V. Venkatesh and L. Giri, *Cell. Signalling*, 2020, **74**, 109717.
- 56 C. A. Bill and C. M. Vines, *Adv. Exp. Med. Biol.*, 2020, **1131**, 215–242.
- 57 K. W. Dunn, M. M. Kamocka and J. H. McDonald, *Am. J. Physiol.: Cell Physiol.*, 2011, **300**, C723–C742.
- 58 R. Schrage, A.-L. Schmitz, E. Gaffal, S. Annala, S. Kehraus, D. Wenzel, K. M. Büllsbach, T. Bald, A. Inoue, Y. Shinjo, S. Galandrin, N. Shridhar, M. Hesse, M. Grundmann, N. Merten, T. H. Charpentier, M. Martz, A. J. Butcher, T. Slodczyk, S. Armando, M. Efferen, Y. Namkung, L. Jenkins, V. Horn, A. Stöfel, H. Dargatz, D. Tietze, D. Imhof, C. Galés, C. Drewke, C. E. Müller, M. Hölzel, G. Milligan, A. B. Tobin, J. Gomeza, H. G. Dohlman, J. Sondek, T. K. Harden, M. Bouvier, S. A. Laporte, J. Aoki, B. K. Fleischmann, K. Mohr, G. M. König, T. Tüting and E. Kostenis, *Nat. Commun.*, 2015, **6**, 10156.
- 59 G. Sanz, I. Leray, A. Muscat, A. Acquistapace, T. Cui, J. Rivière, S. Vincent-Naulleau, V. Giandomenico and L. M. Mir, *BMC Res. Notes*, 2017, **10**, 541.
- 60 D. MacMillan and J. McCarron, *Br. J. Pharmacol.*, 2010, **160**, 1295–1301.
- 61 J. G. Schlegel, M. Tahoun, A. Seidinger, J. H. Voss, M. Kuschak, S. Kehraus, M. Schneider, M. Matthey, B. K. Fleischmann, G. M. König, D. Wenzel and C. E. Müller, *ACS Pharmacol. Transl. Sci.*, 2021, **4**, 888–897.
- 62 G. A. Sanchez, A. V. Smrcka and E. M. Jutkiewicz, *Mol. Pharmacol.*, 2024, **106**, 47–55.
- 63 A. B. Parekh and J. W. Putney, *Physiol. Rev.*, 2005, **85**, 757–810.
- 64 K. Hardman, A. Goldman and C. Pliotas, *Curr. Opin. Physiol.*, 2023, **35**, 100689.
- 65 A. J. Shetty, A. Sirbu and P. Annibale, *J. Mol. Endocrinol.*, 2025, **75**, e240147.



- 66 Q. Zhou, C. Gensch, C. Keller, H. Schmitt, J. Esser, M. Moser and J. K. Liao, *Vasc. Pharmacol.*, 2015, **72**, 163–171.
- 67 E. Lozoya-Gloria, I. Cornejo-Corona, H. R. Thapa, D. R. Browne and T. P. Devarenne, *Bio-Protoc.*, 2017, **7**, e2508.
- 68 C. Q. Vu, S. Fukushima, T. Wazawa and T. Nagai, *Sci. Rep.*, 2021, **11**, 16519.
- 69 S. Erdogmus, U. Storch, L. Danner, J. Becker, M. Winter, N. Ziegler, A. Wirth, S. Offermanns, C. Hoffmann, T. Gudermann and M. Mederos y Schnitzler, *Nat. Commun.*, 2019, **10**, 5784.
- 70 D. Yaffe and O. Saxel, *Nature*, 1977, **270**, 725–727.
- 71 A. F. Huxley and R. Niedergerke, *Nature*, 1954, **173**, 971–973.
- 72 W. Lehman, R. Craig and P. Vibert, *Nature*, 1994, **368**, 65–67.
- 73 M. A. Geeves and K. C. Holmes, *Adv. Protein Chem.*, 2005, **71**, 161–193.
- 74 J. R. Szymanski and R. Yuste, *Curr. Biol.*, 2019, **29**, 1807–1817e3.

