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We report an expeditious catalytic asymmetric approach to clavine alkaloids via a key Heck cyclization. This reaction sets the formation of vicinal stereocenters with excellent diastereoselectivity. Utilizing the aforementioned strategy, the formal total synthesis of cycloclavine (1) has been achieved via another key late-stage esteraminolysis of 6.

Clavine alkaloids (1 and 2; Fig. 1) are a subclass of the ergot family of indole-containing alkaloids produced by several members of the Clavicipitaceae and Trichocomaceae families of filamentous fungi.^{1,2} They have also been identified in plants of the families Convolvulaceae, Poaceae and Polygalaceae.³ Ergot alkaloids (1–4; Fig. 1) primarily target serotonin (5-HT) receptors^{4a} and α -adrenergic and dopamine receptors. Reportedly, some natural or semisynthetic ergoline derivatives are used as **COMMUNICATION**
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Fig. 1 Clavine alkaloids (1 and 2) and lysergic acid (3 and 4) family.

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Catalytic asymmetric formal total syntheses of (+)- and (-)-cycloclavine†

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drugs, such as pergolide (2d) used as an anti-prolactin and anti-Parkinson's disease drug.^{4,5}

Therefore, significant progress has been made in the identification and characterization of genes responsible for the biosynthesis of clavine alkaloids (Fig. 1).⁶ Structurally, clavine alkaloids can exist in pentacyclic [such as cycloclavine (1)] and tetracyclic [such as festuclavine $(2a-c)$] forms.^{4b} Lysergic acid $(3a)$ (and its derivatives such as ergometrine 3b and ergopeptam alkaloids 4a–b) differs from clavine alkaloids 2a–c only in the oxidation state [see, $3a$].^{7a,b}

Cycloclavine (1) was isolated from the seeds of the African morning glory shrub Ipomoea hildebrandtii, and later from a species of filamentous fungus, Aspergillus japonicus. 8a,b Although smaller in size, structurally cycloclavine (1) poses a formidable challenge because of its complex architecture with a pyrrolidine ring linked with a strained cyclopropane ring with three contiguous stereocenters, out of which two are vicinal all-carbon quaternary stereocenters.⁹ Despite the encouraging medicinal value of select clavine congeners, a comprehensive biological evaluation for the majority of these naturally occurring alkaloids has yet to be undertaken. From 2008 till 2016, only racemic syntheses of cycloclavine (1) have been reported, out of which three total syntheses^{10–12} and two formal total syntheses^{13,14} are reported. Interestingly, two consecutive coupling reactions such as selective alkylation of a dienolate and an intramolecular Heck reaction are utilized by Opatz and Netz for a racemic formal synthesis of cycloclavine $(1).¹⁴$

Recently, the first catalytic enantioselective total synthesis of unnatural $(-)$ -cycloclavine $(ent-1)$ has been achieved by Wipf and McCabe^{15a} via Rh-catalyzed enantioselective cyclopropanation (up to 74% ee) of an unsubstituted allene to access a methylenecyclopropane derivative. Very recently, Cao and co-workers have reported an elegant formal total synthesis of naturally occurring $(+)$ -cycloclavine $(1)^{15b}$ while our manuscript was under preparation. This synthesis features a Zn-mediated asymmetrical nucleophilic addition of N-tert-butanesulfinimine, an intramolecular esteraminolysis reaction followed by isomerization of an exocyclic double bond and a late-stage intramolecular Heck coupling reaction.^{15b}

In this context, a unified strategy for the synthesis of 1 and 2 in enantioenriched form would present opportunities to provide access to significant quantities of the natural products and

related analogues. Retrosynthetically, we envisioned that cycloclavine (1) can be synthesized from an advanced enantiopure intermediate α , β -unsaturated amide 5 (Scheme 1) via isomerization followed by reduction of the amide functionality and cyclopropanation.¹⁰ Compound 5 can be accessed from α , β -unsaturated ester 6, which could in fact be the advanced intermediate for clavine alkaloids 2a–d sharing vicinal stereocenters (Fig. 1). We reasoned that ester 6 has the potential to afford two different tetracyclic intermediates such as 5 and 13 (Scheme 2). An ester-aminolysis of 6 can provide access to 5, on the other hand 6 can also afford ester 13 following a 5-endo-trig cyclization (Scheme 2). We argued that as the secondary amine (HOMO) and C= $O \pi^*$ (LUMO) are in proper alignment (see the orbital representation in 11), an ester-aminolysis of 6 would be facile to afford tetracyclic amide 5. However, a 5-endo-trig cyclization of 6 would not be possible because of bad alignment of the secondary amine (HOMO) and C=C π^* (LUMO) (see the orbital representation of intermediates 12a and 12b).

Further, we thought that α , β -unsaturated ester 6 with an *exo*double bond can be synthesized via a key intramolecular Heck cyclization of allylamine 7 (Scheme 1). Enantioenriched allylic amine 7 can be synthesized from allylic alcohol 8 via Mitsunobu type inversion using an azide nucleophile followed by synthetic manipulations. Non-racemic allyl alcohol 8 can be accessed from aldehyde 10 via a D-proline catalysed α -aminoxylation reaction with nitrosobenzene through the intermediate aldehyde 9 (Scheme 1). Importantly, since both enantiomers of proline are commercially available, one can synthesize both antipodes of allylic alcohols, i.e. 8 and ent-8.

On the basis of previous studies on the proline catalysed a-aminoxylation reaction of aliphatic aldehydes with nitrosobenzene,¹⁶ we decided to investigate the potential of this process in the catalytic asymmetric total synthesis of clavine alkaloids (Fig. 1). Towards this direction, we synthesized 3-allyl-4 bromoindole 15 from the Pd(0)-catalyzed reaction of 4-bromoindole 14 with allyl alcohol in the presence of triethylborane using Tamaru's report.¹⁷ This was then reacted with borane followed by oxidation with H_2O_2 in the presence of NaOH to afford a primary alcohol, which was then oxidized to obtain aldehyde 10 under Swern oxidation (Scheme 3). Having aldehyde 10 in hand, we then conducted a catalytic enantioselective a-aminoxylation reaction with nitrosobenzene in the presence of 10 mol% D-proline (Scheme 3). This reaction afforded an a-aminoxylated aldehyde, which was immediately reacted with a stabilized Wittig reagent prepared from 2-bromo ethylpropionate to afford compound E-ester 16 as the sole isomer in 85% yield over 2 steps with 96% enantioselectivity.¹⁸ Communication
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With compound 16 in hand, our effort was thereafter to elaborate to allylic amine 19 for key Heck cyclization

Scheme 2 An ester-aminolysis versus 5-endo-trig cyclization of α , β -unsaturated ester 6. Scheme 3 Asymmetric synthesis of key α , β -unsaturated ester 20.

(Scheme 3). Towards this, N–O bond cleavage was performed with anhydrous $\text{Cu(OAc)}_2, ^{19,20}$ followed by mesylation and azide formation, affording 17 in 82% yield over 3 steps. The azide functionality was reduced under Staudinger conditions, followed by Boc-protection leading to intermediate 18 in 87% yield over 2 steps. The latter was N-methylated using methyl iodide to afford allyl amine 19 (Scheme 3). The intramolecular Heck cyclization of 19 was performed with 5 mol% $Pd(OAc)$ ₂ and 10 mol% PPh₃. Gratifyingly, this reaction afforded a single diastereomer of 20 in 87% yield (Scheme 3). 21

We urged that the Heck cyclization of 19 can proceed through the intermediate $Pd(n)$ -species 21a (Scheme 4). However, in order to minimize the steric clash, 21a could immediately form 21b via a C–C bond rotation. The formation of the tetra-substituted α , β -unsaturated ester 22 from this intermediate is not possible since Pd (n) and β -hydride are *anti*-position to each other (Scheme 4). At this situation, a β-hydride transfer from an adjacent methyl group in 21c could afford 20 having vicinal stereogenic centers (Scheme 4).

Further, compound 20 was elaborated under a key cyclization in order to get the tetracyclic core of cycloclavine (1). Towards this, we deprotected the Boc group in the presence of trifluoroacetic acid at 25 \degree C to afford 6, which under refluxing toluene afforded the ester-aminolysis product 5 with an exocyclic double bond in 88% isolated yield over 2 steps (Scheme 5).²² To our delight, no trace of the 5-endo-trig cyclization (aza-Michael reaction) product was observed, as confirmed from ¹H-NMR analysis of the crude reaction mixture.

With the enantioenriched tetracyclic 5 in hand, we then isomerized to γ -lactam using cesium carbonate in THF to afford 23 in 94% yield (Scheme 5). 23 The latter was reduced using $LiALH₄$ in the presence of $AlCl₃$ to furnish the electron-rich tetrasubstituted double bonded product 24 in 89% yield.²⁴ Further, in order to access the antipode of 25, we performed the catalytic enantioselective α -aminoxylation reaction of 10 with nitrosobenzene in the presence of 10 mol% L-proline, which afforded the product ent-16 in 97% ee after a Wittig reaction (Scheme 6). This enantioenriched material was elaborated to ent-25 via a similar reaction sequence as shown in

Scheme 4 Rationale of highly diastereoselective Heck cyclization. There are no conflicts to declare.

Scheme 6 Asymmetric synthesis of $(-)$ -cycloclavine (ent-1).

Schemes 3 and 5. As the total synthesis of cycloclavine (1) from 25 is known, our effort culminated in the formal total synthesis of this alkaloid.

In summary, the catalytic enantioselective formal total synthesis of both antipodes of cycloclavine (1) has been achieved via a late stage ester-aminolysis of an α , β -unsaturated ester intermediate 6. The vicinal stereocenters of this advanced intermediate were established following an intramolecular Heck cyclization of an enantioenriched α , β -unsaturated ester having allylamine 19. Since both enantiomers of proline are inexpensive and commercially available, our strategy offers an expeditious approach to either enantiomer of cycloclavine (1). Further efforts for a rational extension of the strategy to other congeners of clavine alkaloids are underway and will be reported in due course.²⁵

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Conflicts of interest

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