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Pyridinium-substituted tetraphenylethylene salt-based photosensitizers by varying counter anions: a highly efficient photodynamic therapy for cancer cell ablation and bacterial inactivation†

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Cancer and bacterial infection seriously threaten the health of human beings. The development of an image-guided photosensitizer with a “Two-in-One” function that can be simultaneously used for both efficient cancer cell ablation and rapid bacterial inactivation is highly in demand. In this project, we designed and prepared two aggregation-induced emission luminogens (AIEgens) (called **TPEPy-I** and **TPEPy-PF6**) with a strong electron push–pull effect. They have a near-infrared (NIR) emission, a high ¹O₂ quantum yield up to 0.93 and a fluorescence turn-on effect in mitochondria. Upon white light irradiation, the two mitochondria-targeting AIEgens exhibit a highly efficient photodynamic ablation of HeLa cells as well as excellent photodynamic inactivation of both Gram-positive *S. aureus* and Gram-negative *E. coli*. The time-dependent density functional theory (TD-DFT) results indicate that compared to **TPEPy-PF6**, **TPEPy-I** can easily produce the triplet state that is a prerequisite for ¹O₂ formation. Moreover, the positive effect of iodide anions gives **TPEPy-I** a higher photodynamic efficacy in cancer cell ablation and bacterial inactivation as compared with **TPEPy-PF6**.

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1. Introduction

Cancer and bacterial infectious diseases threaten human health. Especially, cancer-associated bacteria seriously reduce the efficiency of cancer treatments.¹ It is important and urgent to develop an effective method enabling cancer therapy and killing of pathogenic bacteria simultaneously. Photodynamic therapy (PDT) has emerged as a promising method to combat cancer and pathogenic bacteria due to its distinct advantages, such as noninvasiveness, negligible drug resistance, localized treatment, high spatiotemporal precision and low side effects.² For instance, Photofrin is the first FDA-approved photosensitizer (PS) and is still widely employed in clinical cancer treatment. Its mechanism is associated with the generation of destructive singlet oxygen (¹O₂) or other reactive oxygen species (ROS) under light irradiation.³ Specifically, using

an emissive PS to realize image-guided PDT is important and attractive, since it conveniently offers diagnosis and treatment integration. However, traditional PSs suffer from quenched fluorescence and reduced ROS production in the aggregated state because of π – π stacking and the resulting aggregation-caused quenching (ACQ) effect, which makes the image-guided PDT unsatisfactory.

In contrast, aggregation-induced-emission PSs (AIE-PSs) show enhanced fluorescence and efficient photosensitizing characteristics in aggregate states, which can avoid the ACQ effect of PSs.^{4–16} In addition, some groups revealed that AIE luminogens (AIEgens) with an electron push–pull effect would favor ¹O₂ generation.¹⁷ To date, AIE-PSs have been widely employed in imaging-guided cancer cell ablation, and bacterial detection and inactivation.^{18–20} However, some AIE-PSs are ineffective in PDT mediated killing of Gram-negative bacteria.^{20fg} Particularly in solution, Gram-negative bacteria are protected from extracellularly produced singlet oxygen.²¹ More importantly, porins at the outer membranes of Gram-negative bacteria can work as ‘molecular sieves’, which hinder PSs of large size from going through the porin channels. These synergistic effects make PSs fail to kill Gram-negative species. Furthermore, bacterial infection as a cause of cancer has been studied, where the induction of chronic inflammation and the production of carcinogenic bacterial metabolites are regarded as possible mechanisms.²² The most

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Scheme 1 Synthetic routes to TPEPy-I and TPEPy-PF6.

specific example of the inflammatory mechanism of carcinogenesis is *Helicobacter pylori* infection. In addition, skin cancers are easily induced by bacterial infections. Therefore, the development of efficient AIE-PSs that can simultaneously kill cancer cells and bacteria would be appealing.

Recently, Chen's group reported a powerful combination of copper-cysteamine nanoparticles with potassium iodide to inactivate both Gram-positive MRSA and Gram-negative *E. coli*.²³ In that case, iodide anions showed a positive effect by the formation of triiodide ions to enhance the bactericidal ability. Inspired by this, a cationic AIE-PS containing an iodide anion (TPEPy-I, Scheme 1) was designed and synthesized by a simple and straightforward synthetic protocol. For comparison, another AIE-PS containing hexafluorophosphate (TPEPy-PF6, Scheme 1) was introduced for the same study. Both AIE-PSs have some interesting features, such as: (1) the AIEgens comprise a TPE segment (working as a donor), a thiophene fragment (π bridge), and a cationic pyridinium moiety (acceptor), enabling a broad absorption in the visible range and $^1\text{O}_2$ formation through a strong charge transfer. (2) Mitochondrion is an important site for energy conversion and the main source of cellular ROS, which plays a crucial role in mediating cell apoptosis. The presence of cationic pyridinium endows the two AIEgens with a function of targeting mitochondria that may improve PDT performance and imaging qualities. Meanwhile, the cationic pyridinium may help AIEgens penetrate the bacterial membranes by electrostatic interaction and improve their water solubility. (3) The D- π -A structure endows AIE-PSs with NIR emissions, which is highly desirable to achieve image-guided PDT with a low background noise. Our observations indicate that both TPEPy-I and TPEPy-PF6 have the "two in one" functions that can be used simultaneously for cancer cell ablation and bacterial inactivation. However, TPEPy-I is more efficient as an AIE-PS than TPEPy-PF6, in which the iodide ions may play some important roles in the photodynamic performance.

2. Results and discussion

2.1 Molecular design and synthesis

As shown in Scheme S1 (ESI[†]), compound 2 was prepared by a Suzuki reaction between TPE derivative 1 and 5-formyl-2-thiopheneboronic acid. TPEPy-I was obtained by the Knoevenagel reaction between 2 and pyridinium salt. The anion exchange between TPEPy-I and hexafluorophosphate generated TPEPy-PF6, which was expected to avoid the fluorescence quenching effect of

iodine atoms. The combination of a strong electron donor-acceptor (D-A) interaction with extended π -conjugation could facilitate intramolecular charge transfer (ICT), therefore resulting in low electronic bandgaps, broad absorption and long emission wavelengths. The two target products were characterized by NMR and HRMS (see Fig. S1-S6 in the ESI[†] for the details).

2.2 Solvatochromism and AIE properties

As shown in Fig. S7 and S8 (ESI[†]), TPEPy-I and TPEPy-PF6 show obvious solvatochromism, where their absorption and emission spectra are remarkably affected by the polarity of solvents. For example, TPEPy-I in THF has a broad absorption from 350 to 550 nm with the absorption maximum at 453 nm (with a $40\,000\text{ M}^{-1}\text{ cm}^{-1}$ molar extinction coefficient) and an emission at 744 nm with a large Stokes shift of 191 nm. Similarly, TPEPy-PF6 in THF show a large Stokes shift of 169 nm with a maximum absorption at 464 nm (with a $44\,000\text{ M}^{-1}\text{ cm}^{-1}$ molar extinction coefficient) and an emission maximum at 733 nm. In addition, the emission intensity of the two AIEgens is greatly reduced or even quenched in polar solvents such as DMSO and DMF, suggesting a strong ICT effect in polar media. As compared with TPEPy-PF6, the heavy atom effect of TPEPy-I slightly weakens the emission intensity. The Stokes shift for TPEPy-I and TPEPy-PF6 is much larger than that of the commercial NIR fluorophores (less than 50 nm),¹⁹ which can avoid light interference from the excitation light and the self-absorption of emission during biomedical imaging.

The investigation of the AIE feature in DMSO/water mixtures with different water fractions was carried out and it was found that their emissions were almost totally quenched in DMSO. Their photoluminescence (PL) intensities are gradually enhanced with increasing the fraction of water (Fig. S9, ESI[†]). The strongest PL intensity was observed at a 90% fraction of water, in which the PL intensities of TPEPy-I and TPEPy-PF6 were enhanced to about 6.5- and 5.9-fold, respectively, as compared with that of the DMSO solutions. DLS revealed that the average hydrodynamic diameters of nanoaggregates that formed in DMSO/water (1/9, v/v) are around 179 and 301 nm for TPEPy-I and TPEPy-PF6, respectively (Fig. S10, ESI[†]). Meanwhile, TPEPy-I and TPEPy-PF6 also have strong fluorescence in the solid state, and their powder samples show emission peaks at 720 and 718 nm, respectively, (Fig. S11a, ESI[†]). The emission spectra of TPEPy-I and TPEPy-PF6 solids in PMMA films are shown in Fig. S11b (ESI[†]) and their absolute PL quantum yields were found to be 16% and 45%, respectively. All these results demonstrate that both TPEPy-I and TPEPy-PF6 are typical AIE-active molecules.

2.3 ROS and singlet oxygen generation

Because the two AIEgens have a strong absorption in the visible light region, they can be activated by visible light for PDT. The ROS generation was evaluated under white light excitation using a commercial ROS indicator, 2',7'-dichlorodihydrofluorescein diacetate ($\text{H}_2\text{DCF-DA}$), which is capable of emitting a green fluorescence at around 525 nm when oxidized by ROS. As shown in Fig. 1, for the mixture of each AIEgen and $\text{H}_2\text{DCF-DA}$ in aqueous solution, the emission at 525 nm was gradually intensified with



Fig. 1 Change of PL intensity at 525 nm of H₂DCF-DA in the presence or absence of (a) **TPEPy-I** (10 μM) and (b) **TPEPy-PF6** (10 μM). (c) The fluorescence enhancement ratio of 525 nm of H₂DCF-DA in the presence of the two AIEgens or RB upon light irradiation for different time durations.

increasing irradiation time duration. The 525 nm emission reaches its maximum value after 5 min activation with a 193-fold fluorescence enhancement, suggesting that the two AIEgens have an extremely high capability for ROS generation. Such a change, however, was not observed for each AIEgen or H₂DCF-DA alone under the same conditions.

It is well known that many types of ROS including type I (radicals or radical ions such as O₂^{•-}, OH[•], and O₂^{2•-}) and type II (¹O₂) are involved in the PDT process.²⁴ ¹O₂ is generally identified as the primary species for PDT due to its higher diffusion and stronger interaction with tissues than free radicals.²⁵ Thus, we measured ¹O₂ production by **TPEPy-I** and **TPEPy-PF6** upon white light irradiation using 9,10-anthracenediylbis(methylene)dimalonic acid (ABDA) as a ¹O₂ probe. ABDA can selectively undergo oxidation by ¹O₂ to yield endoperoxid, resulting in a corresponding decrease in its absorbance. As shown in Fig. 2a and b, the characteristic absorption peaks of ABDA decrease promptly with prolonged light irradiation. After 3 minutes exposure to white light, only 40% ABDA degraded in the presence of Rose Bengal (the most widely used PS in PDT), compared to 87% and 85% ABDA consumption in the presence of **TPEPy-I** and **TPEPy-PF6** aggregates, respectively (Fig. 2c and d). When Rose Bengal (RB) was employed as the standard PS (0.75 in water), the ¹O₂ quantum yields of **TPEPy-I** and **TPEPy-PF6** were calculated to be 0.89 and 0.93, respectively, indicating a high ¹O₂ generation efficiency by the two AIEgens (Fig. S12, ESI†). Their ¹O₂ efficiencies are even higher than those of clinically used PSs such as Photofrin (0.28) or Laserphyrin (0.48).²⁶

We further detected ROS formation by **TPEPy-I** and **TPEPy-PF6** inside HeLa cells upon white-light activation using H₂DCF-DA. As shown in Fig. S13 (ESI†), an obvious green fluorescence signal was observed inside the cells from H₂DCF-DA when it was incubated into the cells with either of the two AIEgens, indicating the

formation of ROS inside the cells. The superior ROS and ¹O₂ generation efficacy suggests that the two AIEgens are ideal PS candidates for PDT applications.

2.4 TD-DFT calculation

As we know, there are inherent relationships between ROS generation and intersystem crossing (ISC). ISC between the singlet and triplet excited states of molecules is a prerequisite for ¹O₂ generation. It has been recognized that small singlet-triplet gaps appreciably boost ISC rates. The ISC rate (k_{ISC}) can be estimated from an empirical formula based on perturbation theory: $k_{ISC} \propto |\langle S_m | H_{SO} | T_n \rangle|^2 / (\Delta E_{S-T})^2$. H_{SO} is the spin-orbit perturbation Hamiltonian and $\langle S_m | H_{SO} | T_n \rangle$ is the spin-orbit matrix element (SOCME) between the m th singlet excited state and the adjacent n th triplet excited state. ΔE_{S-T} is the singlet-triplet energy gap of the ISC channel of interest, which can play a crucial role in enhancing the ISC rate, as well as the high ¹O₂ generation efficiency. To gain a mechanistic understanding of the extremely high ¹O₂ quantum yield of the two AIEgens, we carried out time-dependent density functional theory (TD-DFT) calculations to optimize their excited states structures and calculate ΔE_{S-T} using the Gaussian 09 program and compute SOCME under scalar relativistic effects at the level of the zero-order regular approximation (ZORA) as implemented in the ORCA 4.2.1 program (more detail is provided in the ESI†). The conductor-like polarizable continuum model (C-PCM) for water was employed to account for the implicit solvent effect of water in experiments.

We firstly found through natural transition orbital (NTO) analysis (Fig. S14, ESI†) that the lowest singlet excited state (S_1) of **TPEPy**⁺ is mainly a local transition (LT) in the acceptor with a minor participation in the donor. It has a strong light adsorption,



Fig. 2 (a) UV-vis absorption spectra of ABDA (100×10^{-6} M) in (a) **TPEPy-I** and (b) **TPEPy-PF6** solutions (5×10^{-6} M) irradiated for different durations with white light irradiation (30 mW cm^{-2}). (c) UV-vis absorption spectra of ABDA (100×10^{-6} M) irradiated for different durations with white light irradiation (30 mW cm^{-2}). (d) Absorbance of ABDA at 380 nm in **TPEPy-I**, **TPEPy-PF6** and Rose Bengal (RB) aqueous solutions for different durations with white light irradiation.

indicated by an oscillator strength (f) of 1.94. Similarly, the lowest spin-forbidden triplet state (T_1) is also a LT in the acceptor part. Considering Kasha's rule, we assume that all types of transitions (fluorescence, thermal relaxation, and ISC) after excitation of this molecule will start from S_1 . We found that the third lowest triplet state (T_3) is the most adjacent triplet state to S_1 . They can form effective ISC channels, thanks to the small ΔE_{S-T} ($S_1 \rightarrow T_3$: -0.22 eV), albeit the SOCME is small ($S_1 \rightarrow T_3$: 0.33 cm^{-1}). The intrinsically small spin-orbit coupling is typical for pure organic chromophores. To further strengthen the ISC, we replace PF_6^- with an iodide ion (I^-), based on well-accepted experience that halogen atoms can effectively boost ISC of organic molecules because of their prominent relativistic effect. I^- and **TPEPy**⁺ form a weakly bound complex, with I^- being *ca.* $3.7 \text{ \AA}$ away from the center of the pyridine ring. As clearly shown in Table 1, the introduction of I^- greatly increases the SOCME of the $S_1 \rightarrow T_3$ ISC channel by two orders of magnitude. This is because of the contribution of the 5p orbital of I to S_1 and T_3 (Fig. S14, ESI†), which enables efficient spin-orbit coupling. The reduction of ΔE_{S-T} also helps to boost the ISC from S_1 to T_3 . Furthermore, the T_1-S_0 gaps of both **TPEPy-PF6** and **TPEPy-I** are sufficiently large (1.17 eV and 1.62 eV) to supply the energy needed to activate $^3\text{O}_2$, because previous studies revealed that the molecule should have a T_1 state with an energy higher than 0.98 eV to convert ground-state molecular oxygen into excited state singlet oxygen.²⁷ Therefore, compared to **TPEPy-PF6**, **TPEPy-I** can more easily generate a triplet state that is a prerequisite for $^3\text{O}_2$ sensitization to produce $^1\text{O}_2$.

2.5 Mitochondria-specific targeting bioimaging

The intracellular distribution of **TPEPy-I** and **TPEPy-PF6** in living HeLa cells was studied by using confocal laser scanning

Table 1 The singlet-triplet energy gap (ΔE_{S-T}) between S_1 and T_1 through T_3 and their corresponding spin-orbit coupling matrix elements (SOCMEs) for **TPEPy-PF6** and **TPEPy-I**

Species	Transition	$\Delta E_{S-T}/\text{eV}$	SOCME/ cm^{-1}
TPEPy ⁺	$S_1 \rightarrow T_1$	1.27	0.04
	$S_1 \rightarrow T_2$	0.76	0.32
	$S_1 \rightarrow T_3$	-0.22	0.33
	$S_0 \rightarrow T_1$	1.17	0.19
	$S_1 \rightarrow T_1$	1.30	5.45
TPEPy-I	$S_1 \rightarrow T_2$	0.47	5.21
	$S_1 \rightarrow T_3$	-0.03	33.18
	$S_0 \rightarrow T_1$	1.62	0.98
	$S_1 \rightarrow T_1$		

Note: for simplicity, **TPEPy-PF6** is modeled by its cation part, *i.e.* **TPEPy**⁺, in TD-DFT calculations. All energies are computed using the ωB97XD functional in conjunction with the TZVP basis set for all light elements and SDD pseudopotential for iodine. The implicit solvent effect of water is treated by the C-PCM solvation model.

microscopy (CLSM). After treatment with **TPEPy-I** and **TPEPy-PF6** for 1 h, HeLa cells exhibited a strong red fluorescence in the cytoplasm, suggesting a fast permeability of AIEgens into living cells. To realize the specificity of the AIEgens for cell imaging, co-localization experiments were then carried out with MitoTracker Deep Red, which is a commercial probe for mitochondrial imaging. It is interesting to find that **TPEPy-I** and **TPEPy-PF6** can specifically stain the mitochondria. The perfect overlapping of their images with that of MitoTracker Deep Red gives rise to high overlap coefficients (0.97) (Fig. 3), indicating their superior specificities for staining mitochondria. The excellent mitochondria-targeting specificity could be attributed to the suitable lipophilicity and cationic property of AIEgens.²⁸



Fig. 3 Co-localization imaging of HeLa cells stained with MitoTracker Deep Red (MTDR), and **TPEPy-I** and **TPEPy-PF6**. (A and B) Confocal images of HeLa cells stained with (A1) **TPEPy-I**, (A2) **TPEPy-PF6**, and (B1 and B2) pseudo-color of MitoTracker Deep Red. (C1 and C2) Bright field. (D1 and D2) Merged images of panels (A)–(C). (E1 and E2) Scatter plot indicating the correlation coefficient between panel (A) and (B). λ_{ex} : 405 nm. Concentration: 10 μM (**TPEPy-I** and **TPEPy-PF6**), 50 nM (MTDR). MitoTracker Red, λ_{ex} = 543 nm, λ_{em} = 548–683 nm, **TPEPy-I** and **TPEPy-PF6**, λ_{ex} = 488 nm, λ_{em} = 600–650 nm.

2.6 PDT in cancer cell ablation

TPEPy-I and **TPEPy-PF6** exhibit a negligible dark cytotoxicity at concentrations up to 50 μM with 90% viability of HeLa cells (Fig. S15, ESI[†]), which means they have a good biocompatibility.

In contrast with their low dark cytotoxicity, **TPEPy-I** and **TPEPy-PF6** exhibit a high phototoxicity (Fig. 4). For example, the cell viabilities are found to be 88.8% vs. 93.6%, 70.4% vs. 92.6%, and 58.2% vs. 66.6% at 0.05, 0.1 and 0.5 μM for **TPEPy-I** and **TPEPy-PF6**, respectively. This indicates that **TPEPy-I** is more efficient in cancer cell ablation than **TPEPy-PF6** at low concentrations less than 0.5 μM . As the concentration of the two AIEgens reaches 1 μM , there is no actual difference between them; both are effective in cancer cell ablation with the cell viability lower than 5% upon white light irradiation. The dependence of cell viability on the PS dose is in line with the characteristics of PDT. The high potency at low concentration indicates that **TPEPy-I** and **TPEPy-PF6** outperform most conventional PSs, such as porphyrin, chlorin, BODIPY, or their respective derivatives.² As we know, mitochondrion is the main organelle targeted by PDT, and the mitochondria-specific targeting capability²⁹ and rapid $^1\text{O}_2$ generation of **TPEPy-I** and **TPEPy-PF6**

make them ideal for photodynamic applications, which cause damage *in situ* to exert excellent therapeutic efficiency.

2.7 Observation of cell death and morphological changes during PDT

The live- and dead-cell staining experiments were used to further justify the photodynamic efficacy of **TPEPy-I** and **TPEPy-PF6**. The representative images are presented in Fig. 5 and Fig. S16 (ESI[†]), where live and dead cells were stained by calcein AM (green fluorescence) and propidium iodide (red fluorescence), respectively. Under white light irradiation for different time durations, the population of dead cells, as indicated by red fluorescence, increased while the live cell population as indicated by green emission decreased. More cells were destroyed in both cases with the increase of the irradiation duration. After white light irradiation for 5 min, almost all HeLa cells were killed in both cases, indicating the high PDT efficacy of **TPEPy-I** and **TPEPy-PF6**.

In addition, we found that laser irradiation induced a dramatic change in the cell morphology. Interestingly, bubble generation was present during the CLSM observations, where calcein AM was used to detect cell survival for labeling.



Fig. 4 Effect of (a) **TPEPy-I** and (b) **TPEPy-PF6** with and without room-light irradiation on cell proliferation of HeLa cells evaluated by the MTT assay. Light power: 30 mW cm^{-2} .



Fig. 5 Live/dead staining of **TPEPy-I** (10 μ M) treated HeLa cells with light irradiation for 30 s, 1 min, 2 min, 3 min, 4 min and 5 min. The live cells were stained by calcein-AM (green), whereas dead cells were stained by PI (red).

Before irradiation, regular and normal cell morphology was observed. However, upon laser stimulation, swelling and blebbing appeared on the cells. More bubbles formed with increasing laser duration (Fig. 6a). As we know, cell morphological changes, such as blebbing, are a sign of cell death.³⁰ When the cells were irradiated by laser for desired time (20 scanning cycles) and then left for 5 min for PI staining, the red emission from PI in the cell nuclei was obviously present due to the PI intercalating with the DNA from the nuclei. These changes can be attributed to the fact that the ROS generated from the AIEgen considerably disrupted the rigidity and permeability of the plasma membrane, causing cancer cell apoptosis and necrosis. The bright field images also confirmed that the shape of the cell changed dramatically after HeLa cells were stained by AIEgens under white light irradiation (Fig. 6b). The shrinkage and fusion of cell walls as well as the appearance of multiple protuberances were found, providing direct and strong evidence for the toxicity on HeLa cells.

2.8 Bacterial elimination

Since **TPEPy-I** and **TPEPy-PF6** are positively charged, it is expected that they can target bacteria through electrostatic

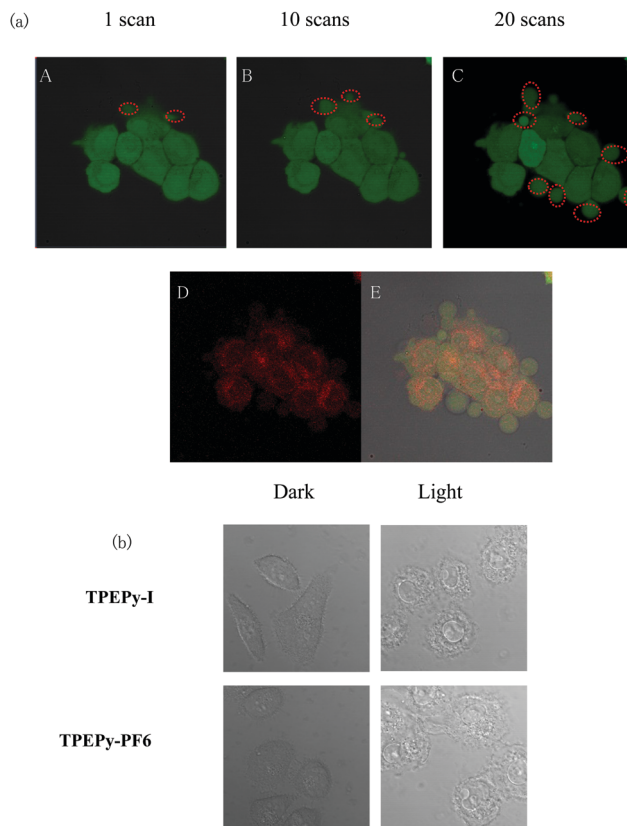


Fig. 6 (a) The CLSM images of **TPEPy-I**-pretreated HeLa cells stained by calcein-AM after (A) 1, (B) 10, and (C) 20 scans. (D) The CLSM images of **TPEPy-I**-pretreated HeLa cells stained by PI after 20 scans. (E) Merged image of (C) and (D). (b) The bright field images of irradiated **TPEPy-I** and **TPEPy-PF6**-pretreated living HeLa cells. [**TPEPy-I**] and [**TPEPy-PF6**] = 10 μ M, irradiation for 5 min.

interactions.³¹ The bacterial staining performance of the two AIEgens on various bacteria was observed by using CLSM. Gram-positive *S. aureus* and Gram-negative *E. coli* were selected as representatives, because they are negatively charged and facilitate the positively charged AIEgens to stain bacteria. The bright red fluorescence in *S. aureus* incubated with **TPEPy-I** and **TPEPy-PF6** was readily observed with high contrast relative to the background (Fig. S17, ESI[†]), indicating that both AIEgens can bind to Gram-positive bacteria efficiently. However, the two AIEgens failed to stain *E. coli*, where scarcely any fluorescence signals could be detected under CLSM (Fig. S18a, ESI[†]). The selective imaging behavior of Gram-positive bacteria *versus* Gram-negative bacteria could be ascribed to their differences in surface structures and chemical components.

As shown in Fig. S18b (ESI[†]), G^+ bacteria only have a cytoplasmic membrane covered by a loose and poriferous cell wall, and a crosslinked and thick peptidoglycan layer about 20–80 nm in size with acidic residues in the outer walls. In contrast, G^- bacteria have a thinner peptidoglycan layer, which is embedded in the phospholipid bilayer. Meanwhile, G^- bacteria possess an additional outer membrane, which performs a barrier function.³² These differences in the two types of pathogens allow AIEgens to penetrate their cell membrane and thus localize in them in

different microenvironments. No fluorescence signals detected under CLSM were observed for *E. coli*, while *S. aureus* presented bright red fluorescence with high labeling efficiency, indicating the distinct binding affinities of the two AIEgens to the two pathogens. Lacking the protection of an outer membrane, **TPEPy-I** and **TPEPy-PF6** can readily penetrate the cell membrane and enter the inside of *S. aureus*. The two AIEgens are restricted effectively by the internal environment, which turns on their emission. For *E. coli*, it is possible that the outer membrane (phospholipid layer) inhibits the insertion of the two AIEgens or the inserted part is easily removed due to weak interaction between *E. coli* and the two AIEgens. Several references reported selective imaging of G^+ over G^- with AIEgens.³³ The bacterial imaging results indicate that **TPEPy-I** and **TPEPy-PF6** have the potential to selectively recognize Gram-positive bacteria over Gram-negative ones.

Next, we examined the antibacterial activity of **TPEPy-I** and **TPEPy-PF6** upon white light irradiation. The standard plate colony-counting method was used to determine the percentage of live bacteria. The two AIEgens display a dose-dependent antibacterial efficiency toward both *S. aureus* and *E. coli*. Regarding the ability of **TPEPy-I** and **TPEPy-PF6** to kill *S. aureus*, the inhibition percentage is more than 92% and 97% at 0.5 μM AIEgens, respectively (Fig. 7a and b). When the concentration of the two

AIEgens reached 1 μM (Fig. 7c), the inhibition percentage reached nearly 100%. For commercial PSSs, the concentration of Toluidine blue O used to kill more than 99.0% *S. aureus* is up to 80 μM .³⁴ For chlorin e6 (Ce6), the concentration required to kill 99.9% of *S. aureus* is 40 μM .³⁵ This comparison strongly suggests that **TPEPy-I** and **TPEPy-PF6** are excellent antibacterial PDT agents for *S. aureus*.

Meanwhile, **TPEPy-I** exhibits a better antibacterial activity toward *E. coli* than **TPEPy-PF6** at low concentrations less than 1 μM . About 97% *E. coli* was eliminated at the concentration of 5 μM AIEgens (Fig. 8). The killing effect of **TPEPy-I** and **TPEPy-PF6** on *S. aureus* is more efficient than that on *E. coli* because Gram-negative bacteria have an additional protecting layer in the outer membrane. Similar findings have been reported in previous studies.³⁶ Photographs of bacteria cultured on agar plates further confirmed the results mentioned above. Almost no bacterial colony was observed on the agar for *S. aureus* and *E. coli* in the presence of 1 and 5 μM AIEgens, respectively, which showed better antibacterial activity than that of commercial Ce6 (Fig. 8c).

Some control experiments were carried out. In the dark or AIEgens alone or light alone, no obvious drop in the survival rate of bacteria was found, which suggests that the potent antimicrobial activity is entirely a consequence of the intrinsic ROS generation induced by AIEgens and white-light irradiation



Fig. 7 The killing efficiency of (a) **TPEPy-I** and (b) **TPEPy-PF6** on *S. aureus*. (c) Photographs of *S. aureus* cultured on an agar plate supplemented with **TPEPy-I**, **TPEPy-PF6** and Ce6 at different concentrations.

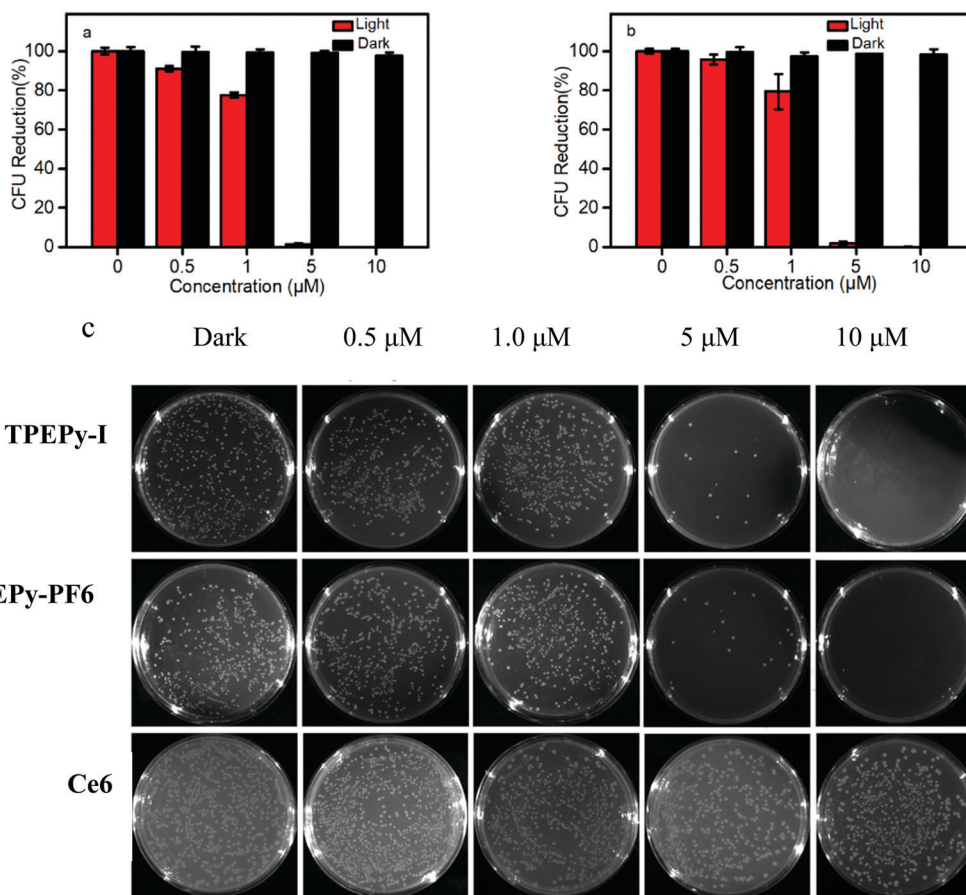


Fig. 8 The killing efficiency of (a) **TPEPy-I** and (b) **TPEPy-PF6** on *E. coli*. (c) Photographs of *E. coli* cultured on an agar plate supplemented with **TPEPy-I**, **TPEPy-PF6** and Ce6 at different concentrations.

(Fig. S19, ESI†). All results indicate that **TPEPy-I** and **TPEPy-PF6** possess excellent antibacterial activity towards both Gram-positive and Gram-negative bacteria at low PS concentrations compared to reported data (Table S1, ESI†), suggesting that they are excellent broad-spectrum antibacterial agents.

PDT can target both external and internal structures of bacteria, and it does not really require the PSs to enter the bacteria, thus the sterilization mechanism of PDT is different from traditional antibiotics. It has been proposed that the cationic PS can penetrate the outer membrane of Gram-negative bacteria by the “self-promoted uptake pathway” in which the divalent metal cations Ca^{2+} and Mg^{2+} are gradually displaced by the PS, and the lipopolysaccharide in the outer membrane permeability barrier is destabilized.³⁷ In our case, the two AIEgens exhibited remarkable $^1\text{O}_2$ generation ability (up 93%) under illumination of white light. So, the proposed molecules can provide an efficient bacterial inactivation route toward *E. coli*. Some white light triggered PDT methods using AIEgens with highly efficient cancer cell ablation and bacterial inactivation have been reported.³⁸

2.9 The mechanism of cancer cell ablation and bacterial elimination

From the results discussed above, **TPEPy-I** is more efficient in cancer cell ablation and bacterial elimination than **TPEPy-PF6**.

The proposed mechanism can be ascribed to several possibilities. First, **TPEPy-I** may produce more triplet states in favor of $^1\text{O}_2$ formation from the TD-DFT calculations. Second, more ROS can be generated through iodide anions involved a type-I electron-transfer photochemical mechanism. Third, the iodide anion can be oxidized to molecular iodine I_2 or I_3^- by both type-I and type-II ROS.³⁹ The higher generation of toxic species, such as H_2O_2 , triiodide ions, and singlet oxygen from **TPEPy-I** may be the major killing mechanism for its enhanced cancer cell ablation and bacterial elimination activity. Some groups have reported similar results that the combination of triiodide with H_2O_2 can promote cell and microbial killing.⁴⁰

The decrease of mitochondrial membrane potential (MMP) is a crucial indicator to assess the dysfunction of mitochondria and it plays an important role in the release of the pro-apoptotic proteins to trigger caspase activation and cell apoptosis.⁴¹ The change of mitochondrial membrane potential was determined using JC-10 dye as the indicator, which tends to aggregate with red fluorescence with a high MMP but becomes monomeric with green fluorescence with a low MMP. Therefore, the change of JC-10 fluorescence can be used to assess the status of mitochondria. As shown in Fig. S20 (ESI†), under irradiation from white light, HeLa cells incubated with **TPEPy-I** and **TPEPy-PF6** display enhanced green fluorescence. The ratio of $I_{\text{green}}/I_{\text{red}}$

in the presence of the two AIEgens was higher than that of light alone, proving that the generated ROS by the two AIEgens injured the cellular mitochondria and resulted in the loss of the MMP.

As we know, cationic fluorescent dyes prefer to stain mitochondria in living cells through electronic interaction due to the negative charge of the mitochondrial inner membrane.⁴² We assume that **TPEPy-I** and **TPEPy-PF6** could be taken up by cancer cells through endocytosis. Then, the two AIEgens selectively accumulated into the mitochondria, thereby inducing efficient mitochondrial dysfunction and intrinsic cancer cell apoptosis by PDT.

3. Conclusions

In summary, two NIR AIEgens (**TPEPy-I** and **TPEPy-PF6**) based on an electron push–pull framework are synthesized and characterized. Due to their AIE backbones with D–A structural units, both the AIEgens show broad absorption in the visible range, while a cationic group is introduced into the molecular design for mitochondria-specific targeting and bacterial membrane anchoring. They can generate ROS efficiently inside cancer cells and bacteria, causing oxidative damage to the mitochondria of cancer cells and cell walls of bacteria. It was demonstrated that low concentrations of AIEgens (1 μM) almost completely kill cancer cells upon white light irradiation. Moreover, benefitting from the enhanced membrane interactions and photosensitizing ability, **TPEPy-I** and **TPEPy-PF6** exhibited efficient antibacterial capability in destroying *S. aureus* and *E. coli* at a low concentration (0.5 μM). These results revealed the great potential of these AIEgens with NIR emission, a high yield of ROS generation, and low dark toxicity to serve in image guided PDT.

Conflicts of interest

There are no conflicts to declare.

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References

- B. D. Wallace, H. Wang, K. T. Lane, J. E. Scott, J. Orans, J. S. Koo, M. Venkatesh, C. Jobin, L. A. Yeh, S. Mani and M. R. Redinbo, *Science*, 2010, **330**, 831–835.
- (a) D. E. J. G. J. Dolmans, D. Fukumura and R. K. Jain, *Nat. Rev. Cancer*, 2003, **3**, 380–387; (b) A. P. Castano, P. Mroz and M. R. Hamblin, *Nat. Rev. Cancer*, 2006, **6**, 535–545; (c) M. H. Al-Afyouni, T. N. Rohrabough, K. F. Al-Afyouni and C. Turro, *Chem. Sci.*, 2018, **9**, 6711–6720; (d) H. Shi, X. Ma, Q. Zhao, B. Liu, Q. Qu, Z. An, Y. Zhao and W. Huang, *Adv. Funct. Mater.*, 2014, **24**, 4823–4830; (e) D. Tao, L. Feng, Y. Chao, C. Liang, X. Song, H. Wang, K. Yang and Z. Liu, *Adv. Funct. Mater.*, 2018, **28**, 1804901; (f) H. Yuan, H. Chong, B. Wang, C. Zhu, L. Liu, Q. Yang, F. Lv and S. Wang, *J. Am. Chem. Soc.*, 2012, **134**, 13184–13187; (g) Y. Li, Q. Wu, M. Kang, N. Song, D. Wang and B. Z. Tang, *Biomaterials*, 2020, **232**, 119749; (h) M. Gao and B. Z. Tang, *Coord. Chem. Rev.*, 2020, **402**, 213076; (i) C. Sun, H. Zhang, L. Yue, S. Li, Q. Cheng and R. Wang, *ACS Appl. Mater. Interfaces*, 2019, **11**, 22925–22931; (j) Y. Ding, S. Li, L. Liang, Q. Huang, L. Yuwen, W. Yang, R. Wang and L. Wang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 9980–9987.
- T. J. Dougherty, C. J. Gomer, B. W. Henderson, G. Jori, D. Kessel, M. Korbelik, J. Moan and Q. Peng, *J. Natl. Cancer Inst.*, 1998, **90**, 889–905.
- (a) X. Zhao, Y. J. Yang, Y. Yu, S. M. Guo, W. X. Wang and S. M. Zhu, *Chem. Commun.*, 2019, **55**, 13542–13545; (b) G. X. Feng and B. Liu, *Acc. Chem. Res.*, 2018, **51**, 1404–1414.
- Z. Y. Liu, H. Zou, Z. Zhao, P. F. Zhang, G. G. Shan, T. K. Kwok, Ryan, J. W. Y. Lam, L. Zheng and B. Z. Tang, *ACS Nano*, 2019, **13**, 11283–11293.
- N. Alifu, X. Dong, D. Li, X. Sun, A. Zebibula, D. Zhang, G. Zhang and J. Qian, *Mater. Chem. Front.*, 2017, **1**, 1746–1753.
- X. Sun, A. Zebibula, X. Dong, G. Zhang, D. Zhang, J. Qian and S. He, *ACS Appl. Mater. Interfaces*, 2018, **10**, 25037–25046.
- C. Chen, H. Qu, R. Liu and D. Ding, *Adv. Mater.*, 2019, 1806331.
- (a) Z. Zheng, T. Zhang, H. Liu, Y. Chen, R. T. K. Kwok, C. Ma, P. Zhang, H. H. Y. Sung, I. D. Williams, J. W. Y. Lam, K. S. Wong and B. Z. Tang, *ACS Nano*, 2018, **12**, 8145–8159; (b) B. Gu, W. Wu, G. Xu, G. Y. F. Feng, P. H. J. Chong, J. Qu, K. T. Yong and B. Liu, *Adv. Mater.*, 2017, **29**, 1701076; (c) Y. Yang, L. Wang, H. Q. Cao, Q. Li, Y. Li, M. J. Han, H. Wang and J. B. Li, *Nano Lett.*, 2019, **19**, 1821–1826.
- (a) J. Qi, C. Chen, D. Ding and B. Z. Tang, *Adv. Healthcare Mater.*, 2018, **7**, 1800477; (b) C. Chen, X. Ni, S. R. Jia, Y. Liang, X. L. Wu, D. L. Kong and D. Ding, *Adv. Mater.*, 2019, **31**, 1904914.
- H. Gao, X. Zhang, C. Chen, K. Li and D. Ding, *Adv. Biosyst.*, 2018, **2**, 1800074.
- X. Ni, X. Zhang, X. Duan, H. Zheng, X. Xue and D. Ding, *Nano Lett.*, 2019, **19**, 318–330.
- D. Wang, M. M. S. Lee, W. Xu, G. Shan, X. Zheng, R. T. K. Kwok, J. W. Y. Lam, X. Hu and B. Z. Tang, *Angew. Chem., Int. Ed.*, 2019, **58**, 5628–5632.
- W. Wu, D. Mao, S. Xu, S. Ji, F. Hu, D. Ding, D. Kong and B. Liu, *Mater. Horiz.*, 2017, **4**, 1110–1114.
- F. Hu, D. Mao, K. Kenry, X. Cai, W. Wu, D. Kong and B. Liu, *Angew. Chem., Int. Ed.*, 2018, **57**, 10182–10186.
- S. Wang, W. Wu, P. Manghnani, S. Xu, Y. Wang, C. C. Goh, L. G. Ng and B. Liu, *ACS Nano*, 2019, **13**, 4742–4751.
- W. B. Wu, D. Mao, F. Hu, S. D. Xu, C. Chen, C. J. Zhang, X. M. Cheng, Y. Y. Yuan, D. Ding, D. L. Kong and B. Liu, *Adv. Mater.*, 2017, **29**, 1700548.

- 18 S. Xu, Y. Yuan, X. Cai, C.-J. Zhang, F. Hu, J. Liang, G. Zhang, D. Zhang and B. Liu, *Chem. Sci.*, 2015, **6**, 5824–5830.
- 19 (a) T. Maisch, *Mini-Rev. Med. Chem.*, 2009, **9**, 974–983; (b) H. Chen, W. Zhang, G. Zhu, J. Xie and X. Chen, *Nat. Rev. Mater.*, 2017, **2**, 1–8.
- 20 (a) C. Zhou, W. Xu, P. Zhang, M. Jiang, Y. Chen, R. T. K. Kwok, M. M. S. Lee, G. Shan, R. Qi, X. Zhou, J. W. Y. Lam, S. Wang and B. Z. Tang, *Adv. Funct. Mater.*, 2019, **29**, 18059864; (b) R. Hu, F. Zhou, T. Zhou, J. Shen, Z. Wang, Z. Zhao, A. Qin and B. Z. Tang, *Biomaterials*, 2018, **187**, 47–54; (c) Y. Li, Z. Zhao, J. Zhang, R. T. K. Kwok, S. Xie, R. Tang, Y. Jia, J. Yang, L. Wang, J. W. Y. Lam, W. Zheng, X. Jiang and B. Z. Tang, *Adv. Funct. Mater.*, 2018, **28**, 180463242; (d) M. Jiang, X. Gu, R. T. K. Kwok, Y. Li, H. H. Y. Sung, X. Zheng, Y. Zhang, J. W. Y. Lam, I. D. Williams, X. Huang, K. S. Wong and B. Z. Tang, *Adv. Funct. Mater.*, 2018, **28**, 1704589; (e) E. Zhao, Y. Chen, S. Chen, H. Deng, C. Gui, C. W. T. Leung, Y. Hong, J. W. Y. Lam and B. Z. Tang, *Adv. Mater.*, 2015, **27**, 4931–4937; (f) M. M. Kang, C. C. Zhou, S. M. Wu, B. R. Yu, Z. J. Zhang, N. Song, M. M. S. Lee, W. H. Xu, F. J. Xu, D. Wang, L. Wang and B. Z. Tang, *J. Am. Chem. Soc.*, 2019, **141**, 16781–16789; (g) M. M. Kang, T. K. Kwok, Ryan, J. G. Wang, H. Zhang, W. Y. Lam, Jacky, Y. Li, P. F. Zhang, H. Zou, X. G. Gu, F. Li and B. Z. Tang, *J. Mater. Chem. B*, 2018, **6**, 3894–3903; (h) Q. Y. Li, Y. Li, T. L. Min, J. Y. Gong, L. L. Du, D. L. Phillips, J. K. Liu, J. W. Y. Lam, H. H. Y. Sung, I. D. Williams, R. T. K. Kwok, C. L. Ho, K. Li, J. G. Wang and B. Z. Tang, *Angew. Chem., Int. Ed.*, 2020, **59**, 10184–10188.
- 21 G. Valduga, G. Bertoloni, E. Reddi and G. Jori, *J. Photochem. Photobiol., B*, 1993, **21**, 81–86.
- 22 J. Parsonnet, *Environ. Health Perspect.*, 1995, **103**, 263–268.
- 23 X. M. Zhen, L. Chudal, N. K. Pandey, J. Phan, X. Ran, E. Amandor, X. J. Huang, O. Johnson, Y. P. Ran, W. Chen, M. R. Hamblin and L. Y. Huang, *Mater. Sci. Eng., C*, 2020, 110659.
- 24 (a) D. E. Dolmans, D. Fukumura and R. K. Jain, *Nat. Rev. Cancer*, 2003, **3**, 380–387; (b) J. Moan and Q. Peng, *Anticancer Res.*, 2002, **23**, 3591–3600; (c) N. Zhao, B. Wu, X. Hu and D. Xing, *Biomaterials*, 2017, **141**, 40–49; (d) S. Xu, Y. Yuan, X. Cai, C.-J. Zhang, F. Hu, J. Liang, G. Zhang, D. Zhang and B. Liu, *Chem. Sci.*, 2015, **6**, 5824–5830.
- 25 O. T. Fackler and R. Grosse, *J. Cell Biol.*, 2008, **181**, 879–884.
- 26 (a) E. Zhao, Y. Chen, H. Wang, S. Chen, J. W. Lam, C. W. Leung, Y. Hong and B. Z. Tang, *ACS Appl. Mater. Interfaces*, 2015, **7**, 7180–7188; (b) W. Zhang, Y. Huang, Y. Chen, E. Zhao, Y. Hong, S. Chen, J. W. Y. Lam, J. Hou and B. Z. Tang, *ACS Appl. Mater. Interfaces*, 2019, **11**, 10567–10577; (c) G. Feng, Y. Yuan, H. Fang, R. Zhang, B. Xing, G. Zhang, D. Zhang and B. Liu, *Chem. Commun.*, 2015, **51**, 12490–12493; (d) M. Gao, Q. Hu, G. Feng, N. Tomczak, R. Liu, B. Xing, B. Z. Tang and B. Liu, *Adv. Healthcare Mater.*, 2015, **4**, 659–663.
- 27 (a) C. Schweitzer and R. Schmidt, *Chem. Rev.*, 2003, **103**, 1685–1757; (b) L. P. Silva, A. Núñez-Montenegro, C. M. Magalhães, P. J. O. Ferreira, D. Duarte, P. Gonzalez-Berdullas, J. E. Rodríguez-Borges, N. Vale and J. C. G. Esteves da Silva, *Eur. J. Med. Chem.*, 2019, **183**, 111683.
- 28 (a) N. Li, Y. Y. Liu, Y. Li, J. B. Zhuang, R. R. Cui, Q. Gong, N. Zhao and B. Z. Tang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 24249–24257; (b) N. Zhao, S. Chen, Y. Hong and B. Z. Tang, *Chem. Commun.*, 2015, **51**, 13599–13602; (c) D. Wang, M. M. S. Lee, G. Shan, R. T. Kwok, J. W. Y. Lam, H. Su, Y. Cai and B. Z. Tang, *Adv. Mater.*, 2018, **30**, 1802105.
- 29 D. Wang, H. Su, R. T. K. Kwok, X. Hu, H. Zou, Q. Luo, M. M. S. Lee, W. Xu, J. W. Y. Lam and B. Z. Tang, *Chem. Sci.*, 2018, **9**, 3685–3693.
- 30 C. Zhu, Q. Yang, L. Liu, F. Lv, S. Li, G. Yang and S. Wang, *Adv. Mater.*, 2011, **23**, 4805–4810.
- 31 (a) X. Li, H. Bai, Y. Yang, J. Yoon, S. Wang and X. Zhang, *Adv. Mater.*, 2019, **31**, 1805092; (b) F. Hu, S. Xu and B. Liu, *Adv. Mater.*, 2018, **30**, 1801350; (c) H. Chen, S. L. Li, M. Wu, K. Kenry, Z. M. Huang, C. S. Lee and B. Liu, *Angew. Chem., Int. Ed.*, 2019, **58**, 1–6.
- 32 (a) H. Bai, H. Chen, R. Hu, M. Li, F. Lv, L. Liu and S. Wang, *ACS Appl. Mater. Interfaces*, 2016, **8**, 31550–31557; (b) Y. Wang, T. S. Corbitt, S. D. Jett, Y. Tang, K. S. Schanze, E. Y. Chi and D. G. Whitten, *Langmuir*, 2012, **28**, 65–70.
- 33 (a) C. Zhou, M. Jiang, J. Du, H. Bai, G. Shan, R. T. K. Kwok, J. H. C. Chau, J. Zhang, J. W. Y. Lam, P. Huang and B. Z. Tang, *Chem. Sci.*, 2020, **11**, 4730–4740; (b) R. Hu, F. Zhou, T. Zhou, J. Shen, Z. Wang, Z. Zhao, A. Qina and B. Z. Tang, *Biomaterials*, 2018, **187**, 47–54; (c) M. Kang, R. T. K. Kwok, J. Wang, H. Zhang, J. W. Y. Lam, Y. Li, P. Zhang, H. Zou, X. Gu, F. Li and B. Z. Tang, *J. Mater. Chem. B*, 2018, **6**, 3894–3903.
- 34 M. Wainwright and K. B. Crossley, *J. Chemother.*, 2002, **14**, 431–443.
- 35 (a) X. Li, H. Bai, Y. Yang, J. Yoon, S. Wang and X. Zhang, *Adv. Mater.*, 2019, **31**, 1805092; (b) F. Hu, S. Xu and B. Liu, *Adv. Mater.*, 2018, **30**, 1801350.
- 36 (a) M. A. Pereira, M. A. F. Faustino, J. P. C. Tome, M. G. P. M. S. Neves, A. C. Tome, J. A. S. Cavaleiro, A. Cunha and A. Almeida, *Photochem. Photobiol. Sci.*, 2014, **13**, 680–690; (b) T. Wu, A. C. McCandlish, L. S. Gronenberg, S. S. Chng, T. J. Silhavy and D. Kahne, *Proc. Natl. Acad. Sci. U. S. A.*, 2006, **103**, 11754–11759; (c) H. Ma, Y. Ma, L. Lei, W. Yin, Y. Yang, T. Wang, P. Yin, Z. Lei, M. Yang, Y. Qin and S. Zhang, *ACS Sustainable Chem. Eng.*, 2018, **6**, 15064–15071.
- 37 A. Minnock, D. I. Vernon, J. Schofield, J. Griffiths, J. H. Parish and S. B. Brown, *Antimicrob. Agents Chemother.*, 2000, **44**, 522–527.
- 38 (a) N. Zhao, P. Li, J. Zhuang, Y. Liu, Y. Xiao, R. Qin and N. Li, *ACS Appl. Mater. Interfaces*, 2019, **11**, 11227–11237; (b) Y. Zheng, H. Lu, Z. Jiang, Y. Guan, J. Zou, X. Wang, R. Cheng and H. Gao, *J. Mater. Chem. B*, 2017, **5**, 6277–6281.
- 39 (a) X. Zhen, L. Chudal, N. K. Pandey, J. Phan, X. Ran, E. Amandor, X. Huang, O. Johnson, Y. Ran, W. Chen, M. R. Hamblin and L. Huang, *Mater. Sci. Eng., C*, 2020, **110**, 110659; (b) L. Huang, G. Szweczyk, T. Sarna and M. R. Hamblin, *ACS Infect. Dis.*, 2017, **3**, 320–328.

- 40 (a) J. Mosinger and B. Mosinger, *Experientia*, 1995, **51**, 106–109; (b) L. Y. Huang, L. Ma, W. J. Xuan, X. M. Zhen, H. Zheng, W. Chen and M. R. Hamblin, *J. Biomed. Nanotechnol.*, 2019, **15**, 2142–2148.
- 41 (a) B. Purushothaman, J. Choi, S. Park, J. Lee, A. A. S. Samson, S. Hong and J. M. Song, *J. Mater. Chem. B*, 2019, **7**, 65–79; (b) T. Qi, B. Chen, Z. Wang, H. Du, D. Liu, Q. Yin, B. Liu, Q. Zhang and Y. Wang, *Biomaterials*, 2019, **213**, 119219.
- 42 (a) J. Zielonka, J. Joseph, A. Sikora, M. Hardy, O. Ouari, J. Vasquez-Vivar, G. Cheng, M. Lopez and B. Kalyanaraman, *Chem. Rev.*, 2017, **117**, 10043–10120; (b) Z. Xu and L. Xu, *Chem. Commun.*, 2016, **52**, 1094–1119.