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Chemical ADP-ribosylation: mono-ADPr-peptides and oligo-ADP-ribose

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ADP-ribosylation is an important post-translational modification that plays a pivotal role in many cellular processes, including cell signaling, DNA repair, gene regulation and apoptosis. Although chemical synthesis of mono- or poly-ADP-ribosylated biomolecules is extremely difficult due to the challenges in regio- and stereoselective glycosylation, suitable protective group manipulations and pyrophosphate construction, synthetic procedures towards these bio-related targets have been reported in recent years. Chemically synthesized well-defined ADP-ribose derivatives serve as useful tools in biological experiments aimed to further elucidate native ADP-ribosylation. In this review, we will discuss the synthetic studies on mono-ADP-ribosylated proteins and oligo-ADP-ribose chains. Future possible synthetic targets and upcoming new methods for the synthesis of these molecules are also included.

1. Introduction

ADP-ribosylation is a post-translation modification of proteins that occurs upon enzymatic transfer of the ADP-ribosyl moiety from NAD^+ to a nucleophilic side chain of an amino acid of a protein.^{1–3} As a result either mono-ADP-ribose (MAR) or oligo-ADP-ribose (PAR) becomes grafted to the protein (Fig. 1). Both modifications play an important regulatory role in various physiological and pathological processes.⁴ The transfer of PAR to amino acids on protein substrates is catalyzed by four enzymes of the PARP family: PARP1, PARP2, and PARP5a, and PARP5b. PAR can exist as a linear or branched polymer. Other PARP family members (PARP3, 4, 6–12, and 14–16) transfer only MAR to amino acids on protein substrates. Upon ADP-ribosylation of cellular proteins, either mono- or polymers, the posttranslational modification becomes subject to further recognition and processing by proteins that are capable of removing or binding PAR or MAR (Fig. 1).¹ Such variations of the ADP-ribosylation state result in a change of the intracellular signaling. Hydrolases such as PARG and enzymes from the ARH-family are responsible for the breakdown of PAR and MAR and thus for the reversal of ADP-ribosylation.^{5,6}

From the point of view of a bioorganic chemist, both mono-ADP-ribosylated (MARylated) and poly-ADP-ribosylated (PARylated) biopolymers (Fig. 1) present a significant challenge. Nevertheless, synthetic well-defined ADP-ribosylated proteins or their substructures are useful for the studies that are aimed at elucidation of the biological role of ADP-ribosyla-

tion. This review describes the synthetic advances towards the synthesis of mono-ADP-ribosylated proteins and oligo-ADP-ribose chains.

2. Chemical synthesis of mono-ADPr-peptides and ADPr-oligomers

The organic synthesis of ADP-ribosylated biomolecules is challenging as the construction of these hybrid structures requires the use of elements from the synthetic chemistry of nucleic acids, oligosaccharides and oligopeptides that are sometimes incompatible. The synthetic challenge is augmented by the necessity to introduce one or even multiple pyrophosphate linkages, which are notoriously difficult to construct efficiently. In the following sections we describe synthetic approaches to the primary challenges of chemical ADP-ribosylation, that is, ribosylation of side chains of various amino acids (Fig. 2, feature A) that culminates in the synthesis of mono-ADP-ribosylated proteins, stereoselective glycosylation of the 2'-OH of adenosine (Fig. 2, feature B) and assembling the oligo-sugar pyrophosphate chain of oligo-ADPr (Fig. 2, feature C).

2.1 Synthesis of mono-ADP-ribosylated peptides

Mono-ADP ribosylated proteins play intriguing roles in many cellular processes.⁷ An approach to deepen the insight into these processes to a molecular level comprises the design, synthesis and biological evaluation of well-defined synthetic mono-ADP-ribosylated derivatives. Relevant examples of such compounds are mono-ADP-ribosylated oligopeptides,^{8–10} as fragments of the naturally occurring proteins. The main challenges in the assembly of these ADP-ribosylated oligopeptides

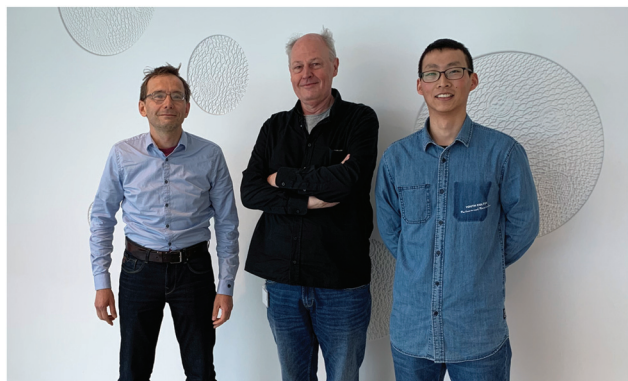
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are an efficient procedure for the introduction of the pyrophosphate function and a method for the stereoselective α -ribosylation of the nucleophilic side chains of amino acids. In this section we will focus on the construction of the α -glycosidic bond that joins the “distal” ribose of the ADPr-moiety and an amino acid side chain in the context of the mono-ADPr-peptide synthesis. The methods that have been developed for the introduction of one pyrophosphate linkage in mono-ADP-ribosylated peptides will be discussed in this section while the introduction of multiple pyrophosphates in short fragments of poly-ADPr is the subject of discussion in section 2.2.2. Application of a solid phase approach to mono-ADP-ribosylated oligopeptides is most obvious as a solution phase synthesis would be restricted in terms of the length and composition of the oligopeptide. While the introduction of the pyrophosphate moiety is feasible on a solid support,^{8,9} ribosylation of partially protected and immobilized oligopeptides with a protected ribose donor is almost impossible in terms of stereoselectivity and yield. Therefore, attention has been focused on the synthesis of suitably protected ribosylated amino acid building blocks that can be applied in solid phase peptide synthesis (SPPS). The main hurdle in the synthesis of these ribosylated amino acid building blocks is the difficulty to control the 1,2-*cis* configuration of the ribosyl anomeric linkage at the glycosylation stage. The sensitivity of this O-glycosidic bond to an acid adds another layer of complexity.

The first reported synthesis of suitably protected α -ribosylated amino acid building blocks and their application in a SPPS assembly of relevant ADP-ribosylated oligopeptides is by van der Heden van Noort *et al.*⁸ The choice for Fmoc-based peptide synthesis led to the synthesis of protected α -ribosylated asparagine (Asn) **5** and glutamine (Gln) **6** build-

ing blocks (Scheme 1). The route of synthesis started with the reduction of fully protected β -D-ribosylated azide **1** to an epimeric hemiaminal mixture **2**. Subsequently, EDC-mediated coupling with Z-Glu-OBn and Z-Asp-OBn, respectively and silica gel purification gave the individual anomers **3** and **4**. Protective group manipulation provided α -ribosylated Asn (**5**) and Gln (**6**) building blocks with the mutually orthogonal TBDPS and Fmoc protecting groups. Guided by the outcome of a solution phase study, SPPS was undertaken in which two procedures for the installation of the adenosine diphosphate function were explored. For that purpose, native¹¹ model peptide **11** containing an ADP-ribosylated Asn residue and peptide **15** originating from the N-terminus of human histone H2B containing an ADP-ribosylated Gln residue were selected. In the latter case Gln was chosen as a stabilized isostere of Glu that was reported to be the natural ADP-ribosylation site.¹² Hexapeptide **7** was obtained *via* SPPS using a BOP/HOBT Fmoc-based synthesis executed on Tentagel resin equipped with an HMBA linker. Upon removal of the TBDPS group at the 5-OH of the ribose, the immobilized peptide **7** (R = H) was phosphitylated with phosphoramidite **8** under the influence of the activator DCI, followed by oxidation using iodine in pyridine to give the activated phosphorimidazolite **9**. A reaction with the protected adenosine phosphate **10** led to the formation of the protected and immobilized target ADP-ribosylated peptide. Removal of the Dmb group on Glu and subsequent treatment with ammonia methanol, to affect both the removal of the remaining protecting groups and cleavage from the resin, gave after HPLC purification ADP-ribosylated hexapeptide **11**. With the aid of LC-MS analysis of the crude product the C-terminal carboxamide, the ribosyl 5-phospho-monoester and the corresponding H-phosphonate could be



From left to right: Dmitri Filippov, Gijs van der Marel and Qiang Liu

Qiang Liu is a PhD student at Leiden University where he works under the supervision of Prof. Gijs van der Marel and Dmitri Filippov. He earned his B.Sc in pharmaceuticals from Jining Medical University in 2012 and M.Sc in medicinal chemistry from Sichuan University in 2015 in China. Qiang focuses his research on the organic synthesis of bio-related ADP-ribosylated molecules such as

linear or branched ADPr-oligomers and mono-ADP-ribosylated peptides and their derivatives.

Gijs van der Marel is a Full Professor in Organic Chemistry at Leiden University. He trained with Prof. Jacques van Boom on the development of synthetic methods for oligonucleotides and obtained his PhD degree in 1981. His research has been directed at the development of synthetic chemistry to assemble all types of biopolymers: nucleic acids, peptides and carbohydrates as well as hybrids and analogues thereof to study their role in biology. He has supervised >100 PhD students and his research interests include synthetic organic chemistry methodology, biopolymer synthesis, glycobiology and chemical biology and immunology.

Dmitri Filippov obtained his Ph.D. degree from Leiden University (1998) under the guidance of Jacques van Boom and Gijs van der Marel while he was investigating the application of phosphoramidite reagents to the synthesis of peptide-nucleic acid hybrid biopolymers. Dmitri has remained in Leiden ever since where he currently is an Assistant Professor. His research interests include synthetic bioorganic chemistry with the focus on the chemistry of ADP-ribosylated biomolecules and other hypermodified peptides and nucleic acids, both native and artificial.



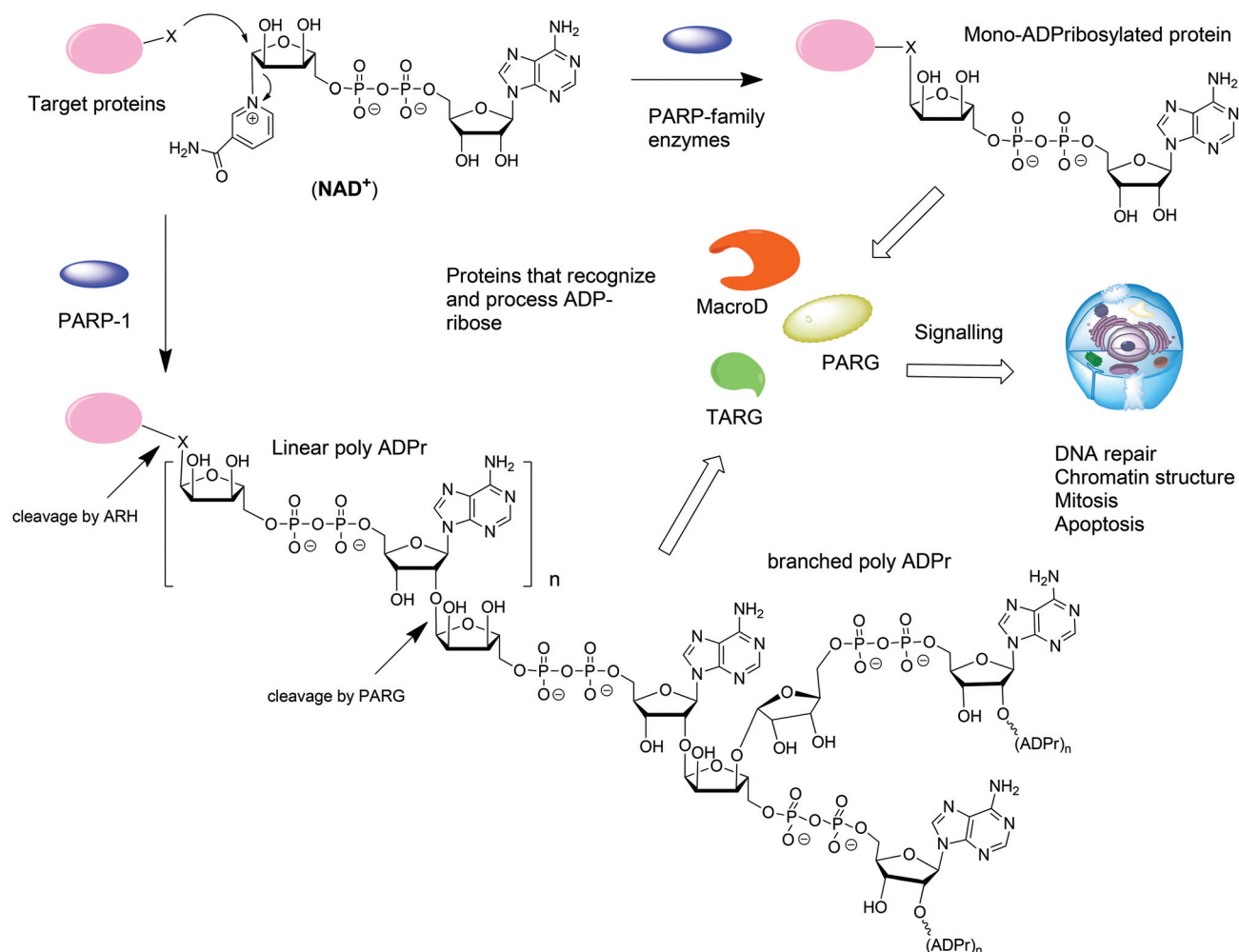


Fig. 1 Biosynthesis and metabolism of mono- and poly-ADP-ribosylated proteins.

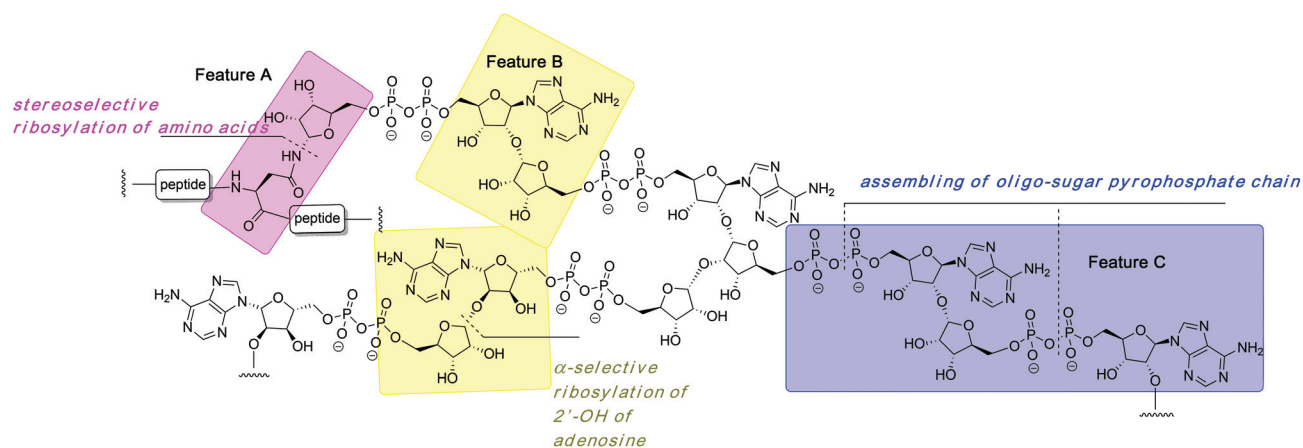


Fig. 2 Structure of poly-ADP-ribose with its most conspicuous synthetically challenging features.

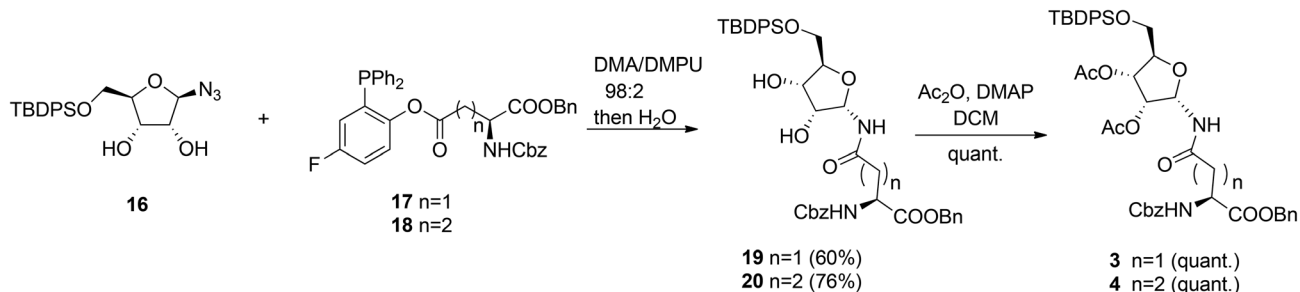
identified as side products. It was reasoned that the formation of H-phosphonate could be suppressed by the reversal of the procedure for pyrophosphate formation. To this end the phos-

phate was installed on the immobilized peptide (*i.e.* 13), while activated phosphorimidazolate (*i.e.* 14) was prepared in solution. The assembly of ADP-ribosylated peptide 15 started with





Fluorinated triphenylphosphines functionalized with Z-Asp-OBn (**17**) and Z-Glu-OBn (**18**) were used in ligation reactions with differently protected β -D-ribofuranosyl azides. It turned out that both stereochemistry and productivity of these reactions were dependent on the protection of the hydroxyl groups in the ribose moiety. Protection of the primary 5-OH with the TBDPS group (**16**) produced (Asn) **19** and (Gln) **20** in good yields. Subsequent acetylation of **19** and **20** gave known⁸ SPPS building blocks **3** and **4**.

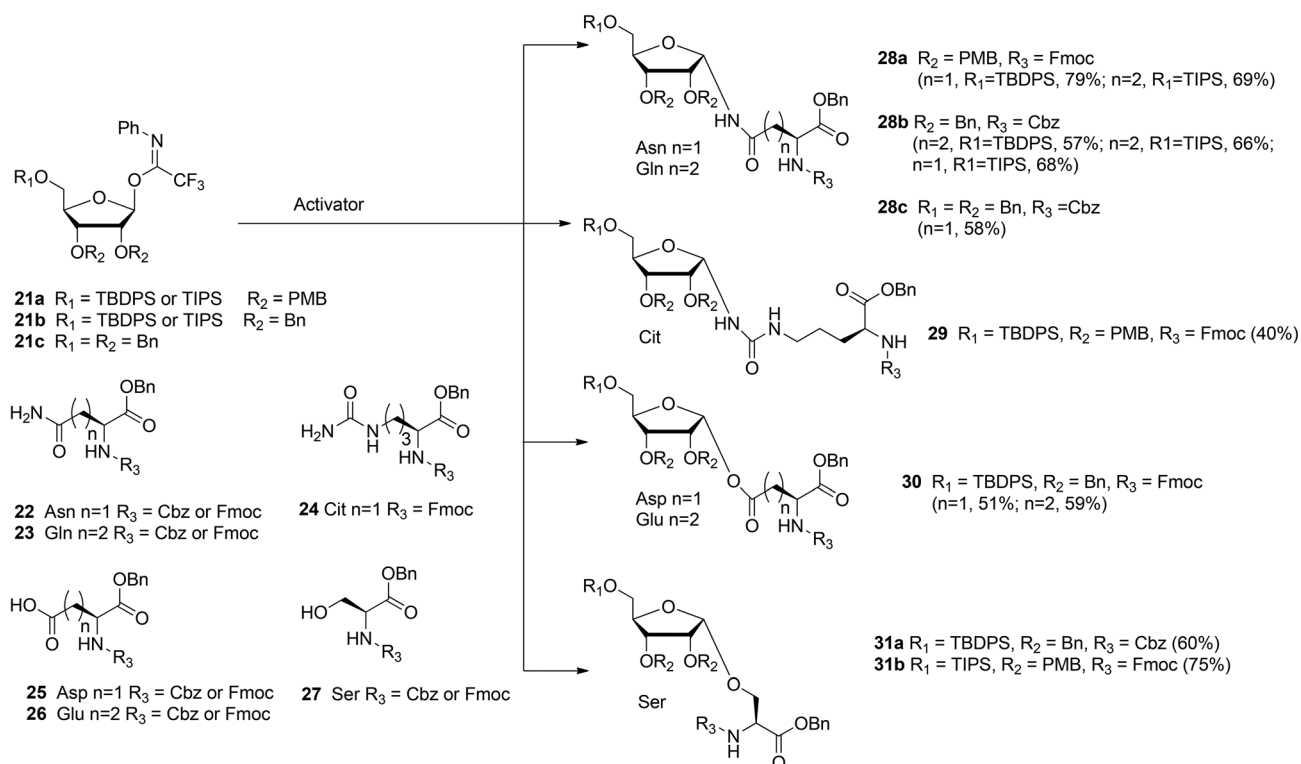


Scheme 2 Synthesis of α -ribofuranosyl amides using fluorinated phosphines.

With the aim of broadening the range of the synthetically accessible ribosylated amino acids Kistemaker *et al.*¹⁷ developed an alternative ribosylation method that employed ribosyl donors **21a**, **b**, and **c** (Scheme 3) with the *N*-phenyl trifluoroacetimidate leaving group and with non-participating ether protecting groups at 3- and 2-OH. The latter feature allows the formation of both *O*- and *N*-glycosidic linkages *via* highly α -selective acid catalyzed glycosylation. Condensation of per-benzylated donor **21c** with Asn acceptor **22** (R_3 = Cbz) under various conditions led to α -product **28c** (n = 1). However, these conditions were not transferable to other acceptors (*e.g.* Glu acceptor **23** (R_3 = Cbz)). It was reasoned that the selectivity of the ribosylation could be improved by replacing the benzyl group at the 5-OH in the ribose by the bulkier TBDPS or TIPS protecting groups to give donor **21b** (R_1 = TBDPS or TIPS).

Several activator systems were tested and the results of these tests indicated that TMSOTf and HClO₄-SiO₂ were the most favorable activators. A reaction of donor **21b** (R_1 = TBDPS or TIPS) with Asn acceptor **22** (R_3 = Cbz) and Glu acceptor **23** (R_3 = Cbz) gave good to excellent yields of α -products **28b** (n = 1, 2, respectively). Next, this glycosylation protocol was applied to Cbz- and Fmoc-protected glutamic acid (Glu, **26**), aspartic acid (Asp, **25**) and serine (Ser, **27**). Using TMSOTf as activator protected derivatives of ribosylated Asp **30** (n = 1, α/β = 98:2, 51%), ribosylated Glu **30** (n = 2, α/β = 98:2, 59%) and ribosylated Ser **31a** (R_1 = TBDPS, α/β = 1:0, 60%) were obtained.

In order to minimize protecting group manipulations towards the ribosylated amino acid building blocks suitable for SPPS, the benzyl groups at the 2-OH and 3-OH in the ribosyl donor were replaced with acid labile PMB ethers and



Scheme 3 Trifluoroacetimidate ribosylation of partially protected amino acids.



the Cbz group in the amino acid acceptors was replaced with the Fmoc group. Condensation of imidate donor **21a** (R_1 = TBDPS) with Asn acceptor **22** (R_3 = Fmoc) in DCM under the influence of TMSOTf furnished **28a** (n = 1, α/β = 97 : 3, 79%). A similar condensation using the less nucleophilic citrulline (Cit) acceptor **24** proceeded in a less α -selective manner to give **29** (α/β = 78 : 22, 40%). The insolubility of Gln acceptor **23** (R_3 = Fmoc) required a change to dioxane/DCM as the solvent system and $\text{HClO}_4\text{-SiO}_2$ as the activator to give **28a** (n = 2, α/β = 93 : 7, 69%). Finally, condensation of Ser acceptor **27** (R_3 = Fmoc) with donor **21a** (R_1 = TIPS) furnished ribosylated Ser **31b** (α/β = 1 : 0, 75%).

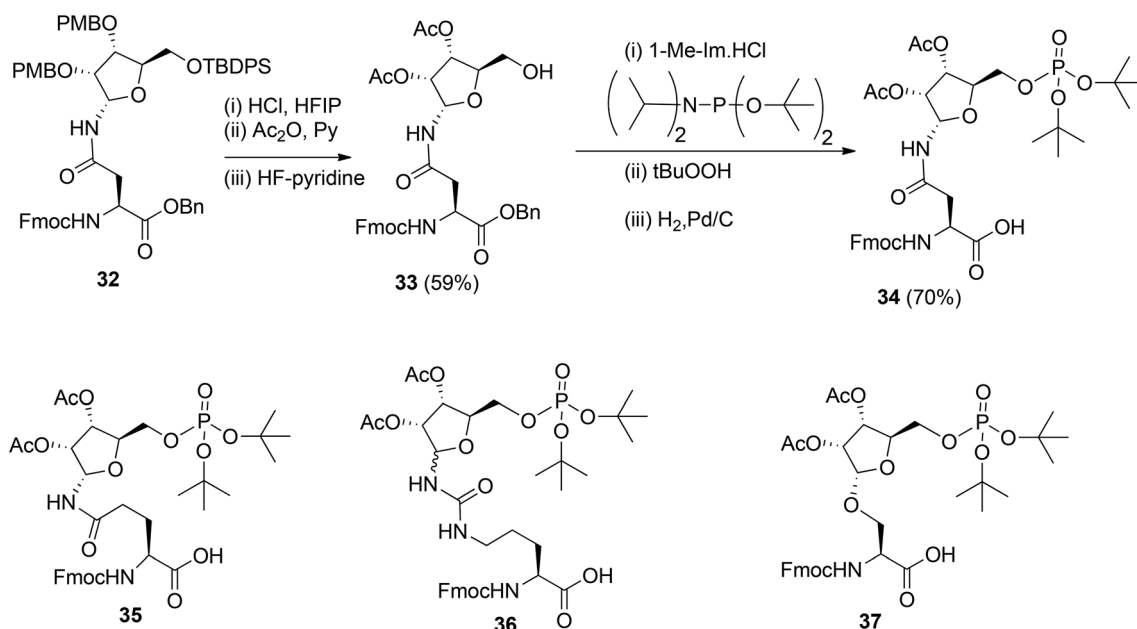
To obtain relevant ADP-ribosylated oligopeptides by SPPS, Kistemaker *et al.*^{9,18} also searched for another procedure for pyrophosphate formation. To this end, the solution phase method for the synthesis of sugar nucleotides, reported by Gold *et al.*,¹⁹ was adopted. This procedure⁹ combines phosphoramidite (P^{III}) with phosphate (P^{V}) chemistry and the adaptation to a solid phase procedure required the on-resin formation of a phosphomonoester. It was reasoned that this could be circumvented by the development of the protected pre-phosphorylated amino acid building blocks **34–37** (Scheme 4). The synthesis of these phosphorylated amino acids is illustrated by the preparation of Asn building block **34**. The PMB groups in fully protected ribosylated Asn **32** were replaced by acetyl groups by acidolysis, followed by acetylation while 5-OH was unmasked by desilylation to afford **33**. The *tert*-butyl group was selected as an orthogonal phosphate protecting group.

The di-*tert*-butyl phosphate triester was installed with di-*tert*-butyl *N,N*-diisopropylphosphoramidite and subsequent oxidation of the intermediate phosphite triester. Finally,

hydrogenolysis of the benzyl ester gave the α -configured Asn building block **34** suitable for SPPS. Transferring this procedure to other amino acids showed that the anomeric integrity of Gln **35** and Ser **37** remained intact while Cit **36** was obtained as an anomeric mixture, which could be separated by column chromatography.

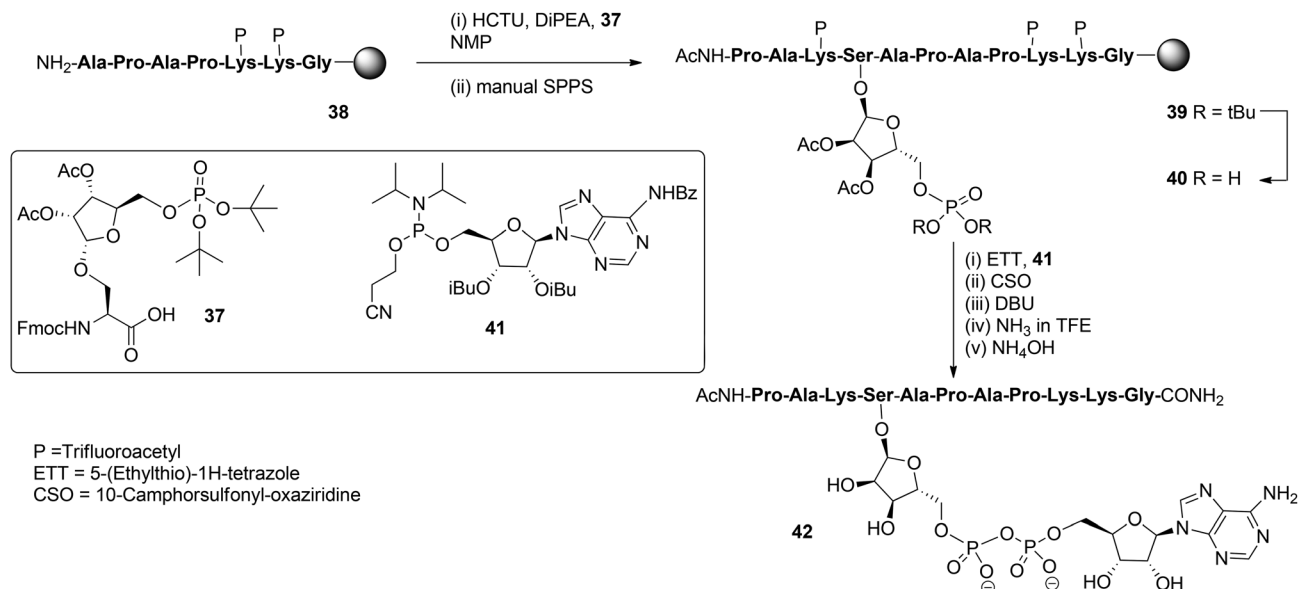
With these building blocks available SPPS could be undertaken and relevant ADP-ribosylated oligopeptide fragments from Histone H2B, RhoA protein and HNP-1 defensin were obtained.⁹ The synthesis of Ser-ADPr H2B peptide **42** (Scheme 5) serves as a representative example of the usefulness of this methodology for the preparation of ADP-ribosylated peptides with a native ADP-ribosylation site.¹⁰ The SPPS of hendecapeptide **42** was carried on Tentagel resin, equipped with an HMBA-linker. First, intermediate immobilized heptapeptide **38** was produced with automated SPPS utilizing Fmoc chemistry and trifluoroacetyl protected lysine residues. Subsequent elongation to phosphoribosylated peptide **39** was done manually using serine phosphotriester **37** and commercially available protected amino acids.

To allow pyrophosphate introduction, di-*tert*-butyl phosphate protective groups in the immobilized **39** were removed with TFA and DCM, followed by neutralizing with pyridine. A reaction of the obtained phosphate monoester **40** with phosphoramidite **41** under the influence of activator ETT was followed by the oxidation of the $\text{P}^{\text{III}}\text{-P}^{\text{V}}$ intermediate with CSO and, finally, the cyanoethyl group in the pyrophosphate was removed with DBU. Target pyrophosphate **42** was obtained by a two-step procedure. First, cleavage from the resin was attained by treatment with a saturated NH_3 solution in trifluoroethanol. In this way the formation of a carboxylic acid at the C-terminus of the peptide is circumvented and only carboxa-



Scheme 4 Ribosylated amino acids with a phosphotriester at 5-OH.





Scheme 5 SPPS of Ser-ADPr hendecapeptide originating from histone H2B.

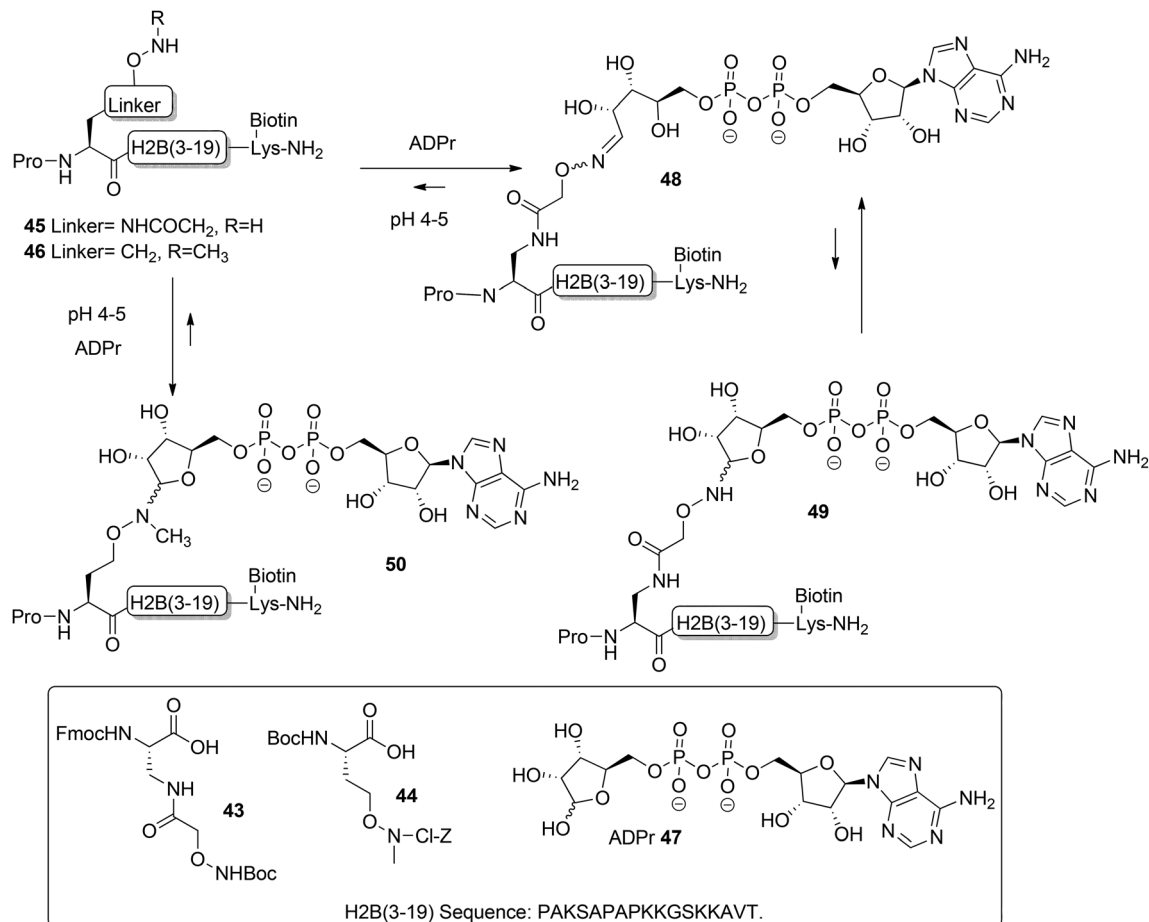
amide is produced. Addition of NH_4OH effected the removal of all remaining protective groups giving hendecapeptide **42**. This is the first reported synthesis of an ADP-ribosylated peptide on serine,¹⁰ which has been found to be one of the most widespread ADPr-modification sites in histones, upon DNA damage.^{20–23} It is also the first real, naturally occurring ADP-ribosylated oligopeptide as all others are provided with stabilized anomeric ribosyl linkages, that is, asparagine (Asn) and glutamine (Gln)^{8,9} as stabilized analogues of aspartic acid (Asp) and glutamic acid (Glu), respectively.

Moyle and Muir²⁴ reported the synthesis and a biochemical evaluation of stabilized and artificial mono-ADP-ribose conjugated peptides. An N-terminal (3–19) oligopeptide²⁴ of histone H2B protein with the glutamate residue mono-ADP-ribosylated was selected as a model (Scheme 6). With the aid of manual SPPS on MBHA resin using HBTU/DIPEA, oligopeptides **45** and **46** having either an aminoxy or *N*-methyl aminoxy functionality were assembled. The aminoxy-containing building block **43**²⁵ or *N*-methyl aminoxy containing amino acid **44**²⁶ was incorporated instead of the glutamic acid at the N-terminus. After cleavage from the resin the (*N*-methyl) aminoxy groups in the oligopeptides **45** and **46** were reacted with the hemiacetal of the ribose moiety in free ADP-ribose **47** producing an ADPr appendage. The aminoxy group in **45** led mainly to ring-opened ADPr peptide **48** and a small amount of the ring-closed form **49**, while the *N*-methyl aminoxy group in **46** gave the ring-closed ADPr peptide **50** exclusively. By executing the ligation procedure at pH 4.5 the oxime formation becomes selective, leaving all natural amino acid side-chain functionalities, including those of lysine and arginine residues, intact.

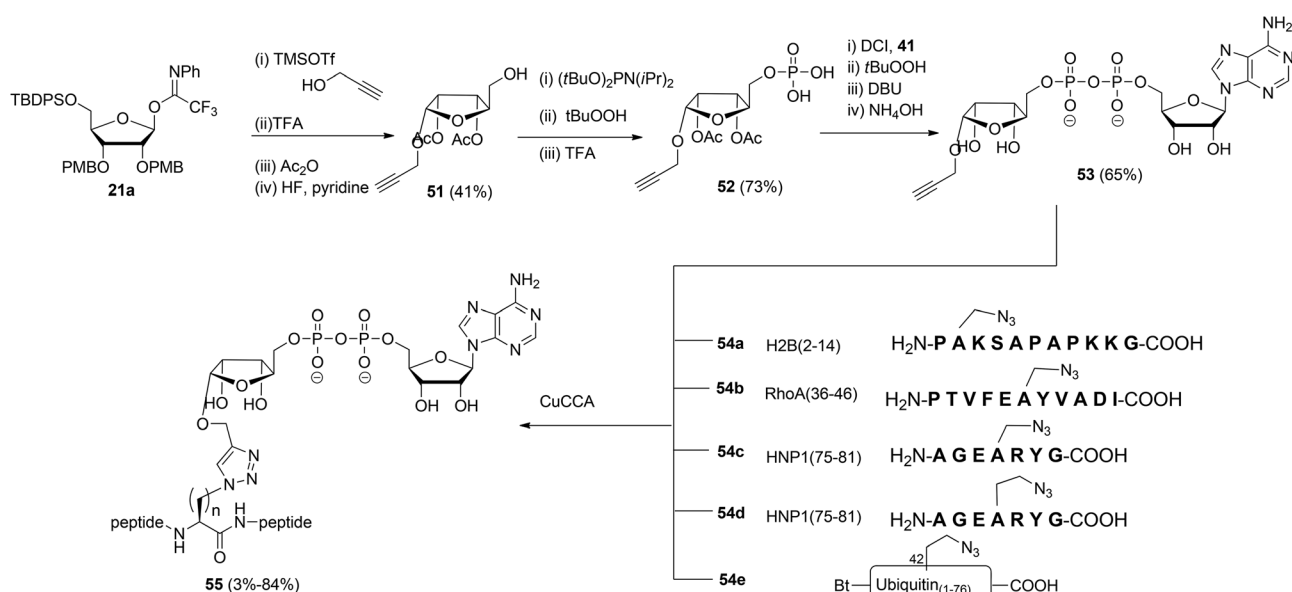
Liu *et al.*²⁷ reported another type of ADPr analogue that can be easily introduced in oligopeptides with the aid of a

copper-catalyzed azide–alkyne cycloaddition (CuAAC) to give triazole-linked ADP-ribosylated peptides (Scheme 7). With the assistance of SPPS and standard amino acid building blocks, either azido-alanine or azido-homoalanine was incorporated at specific positions of peptides, the sequences of which were based on those of biologically relevant ADP-ribosylated proteins. Both azido functionalized oligopeptides (**54a–d**) and ubiquitin protein **54e** in which Arg42 was replaced by azido-homoalanine were assembled by standard SPPS. After cleavage from the resin and concomitant removal of all protecting groups, followed by purification with RP-HPLC, peptides **54a–e** were subjected to the critical CuAAC reaction with alkyne **53**. This ADPr building block was efficiently prepared by the reaction of imidate donor **21a** with propargyl alcohol. Isolation of the pure α -anomer, followed by protective group manipulation gave glycoside **51**. Next, phosphorylation, oxidation to the phosphodiester and removal of the *t*Bu protecting groups gave phosphomonoester **52** that was used in combination with phosphoramidite **41** to provide pyrophosphate **53** via the same $\text{P}^{\text{III}}\text{--P}^{\text{V}}$ procedure¹⁹ as described for the synthesis of the Ser-ADPr hendecapeptide in Scheme 5. The CuAAC reaction of **53** with **54a–e** in Tris buffer at pH 7.6 under the influence of CuSO_4 , sodium ascorbate and a tris-triazole ligand proceeded efficiently to give triazole-linked adenosine diphosphate ribosylated peptides **55**. The CuAAC reaction of **53** proved to be successful for small proteins as illustrated by the total chemical synthesis of a biologically active ADP-ribosylated ubiquitin derivative, demonstrating that triazole linked ADPr can be employed as bio-isosteres of ADPr-Arg in peptides or proteins (Scheme 7).²⁷ A different CuAAC mediated approach to the synthesis of triazole linked peptides has been reported by Li *et al.*²⁸





Scheme 6 The aminoxy and *N*-methyl aminoxy functionalized peptides **48–50** of the group of Muir.



Scheme 7 Synthesis of triazole linked ADPr-peptides by CuAAC chemistry.



2.2 Synthesis of poly-ADPr chain

The organic synthesis of fragments of poly-ADP-ribose (PAR, Fig. 1) comprises a repetitive introduction of both an α -glycosidic bond between the ribose and the 2'-OH of adenosine and a pyrophosphate linkage between the primary OHs of the adenosine and the ribose moiety. To acquire ADPr oligomers of a certain length both solution and solid phase approaches require the design and synthesis of suitably protected and functionalized building blocks. Monomeric building blocks could be envisaged, in which a pyrophosphate moiety is incorporated but these must then act as ADP-ribofuranosyl donors that are suitable for repetitive α -ribosylation of the 2'-OH of the terminal adenosine moiety of the growing PAR-chain. Although, such a method would resemble the biosynthesis of PAR, in which NAD^+ fulfills the role of the ADP-ribofuranosyl donor this approach should be rejected because the repetitive introduction of multiple α -ribosidic bonds in the presence of (anionic) pyrophosphates is almost impossible. Therefore the α -ribosidic bond should be preinstalled in the building block while the pyrophosphate moiety is then repetitively introduced during the assembly of the oligo-ADP-ribose chain. Both syntheses of ADP-ribose oligomers that are reported to date use the latter strategy.^{18,29}

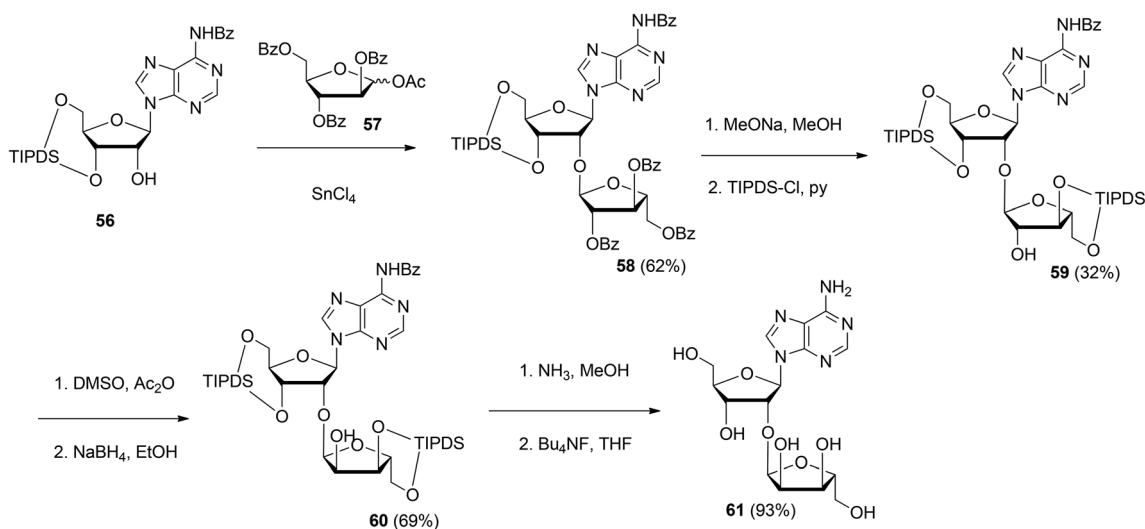
In the following sections, we first describe the methods that have been developed for the synthesis of 2'-O-ribosylated adenosine building blocks (section 2.2.1), and next, the methods of pyrophosphate formation in the framework of the assembly of fragments of poly-ADPr-ribose will be discussed (section 2.2.2).

2.2.1 Building block synthesis-ribosylated adenosine. In 2008 Mikhailov *et al.*³⁰ reported the first synthesis of a 2'-O- α -D-ribofuranosyladenosine building block (**60**, Scheme 8). The potentially problematic α -ribosylation was circumvented by the use of 1-O-acetyl-2,3,5-tri-O-benzoyl-D-arabinofuranose **57** as a donor. Condensation of donor **57** with adenosine acceptor **56**

under the influence of tin tetrachloride afforded, by neighboring group participation, trans-configured disaccharide nucleoside **58**. To obtain 2'-O- α -D-ribofuranosyladenosine building block **60**, the route of synthesis was continued by protective group manipulation and finally by inversion of 2'-OH to give the desired ribo-configuration *via* an oxidation-reduction sequence. The protective groups in building block **60** were removed to produce 2'-O- α -D-ribofuranosyladenosine **61**. The group of Marx³¹ applied this method for the preparation of ribosylated adenosine analogues to develop PARP inhibitors.

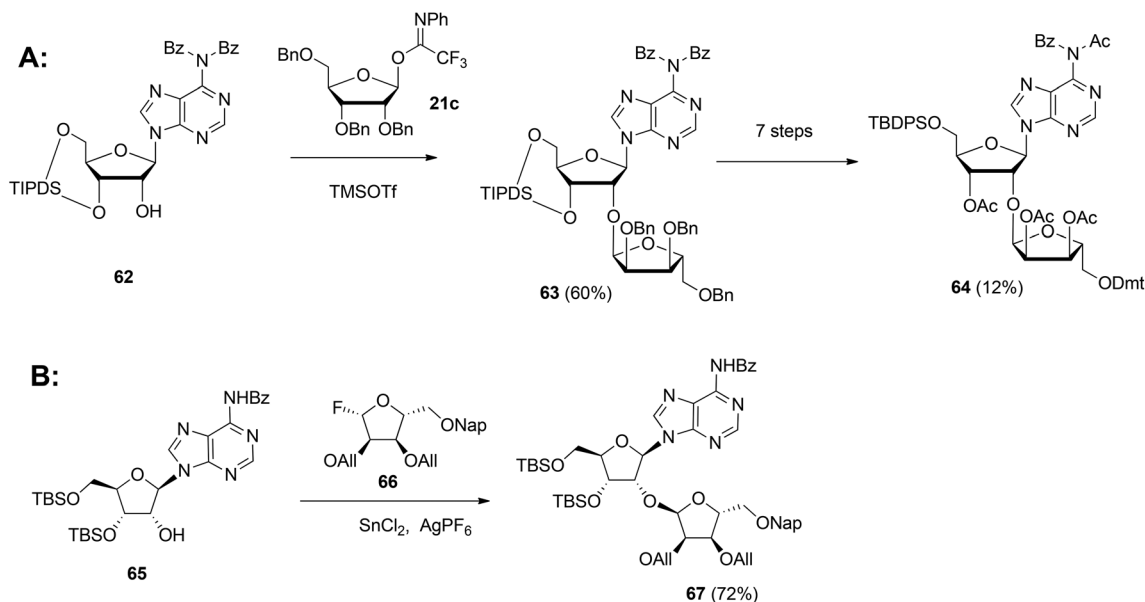
Although the method of Mikhailov *et al.*³⁰ is robust the route of synthesis to a monomeric building block suitable for the assembly of oligo-ADP-ribose is rather lengthy due to the necessity to invert the 2'-OH position of ribose and the subsequent introduction of orthogonal protective groups. For the synthesis of oligo-ADP-ribose, more direct approaches to attain α -selective glycosylation of adenosine were developed. Van der Heden van Noort *et al.*³³ reported the synthesis of 2'-O- α -D-ribofuranosyladenosine (**64**, Scheme 9A) with TBDPS and DMT as orthogonal protecting groups on the primary hydroxyl functions of the ribose moieties. The key step is the TMSOTf mediated condensation of (*N*-phenyl)-2,2,2-trifluoroacetimidate donor **21c** and adenosine acceptor **62** to furnish fully protected ribofuranosyladenosine **63** in an α -selective manner. Subsequent protecting group manipulation yielded **64**, amenable for the assembly of oligo-ADP-ribose. Recently, Shirinfar *et al.*³² reported the synthesis of a protected phosphorylated ribofuranosyladenosine building block, using the same glycosylation procedure.

In 2015 Lambrecht *et al.*²⁹ reported the synthesis of orthogonally protected ribosyl adenosine **67** (Scheme 9B) by the α -selective condensation of β -fluoride donor **66**, obtained in six steps from ribose with adenosine acceptor **65**. Crucial for the productivity of this reaction was the use of $\text{AgPF}_6/\text{SbCl}_2$ as an activator combination.



Scheme 8 Synthesis of 2'-O- α -ribosylated adenosine **61** using 1-O-acetyl-2,3,5-tri-O-benzoyl-D-arabinofuranose **57** as the donor.

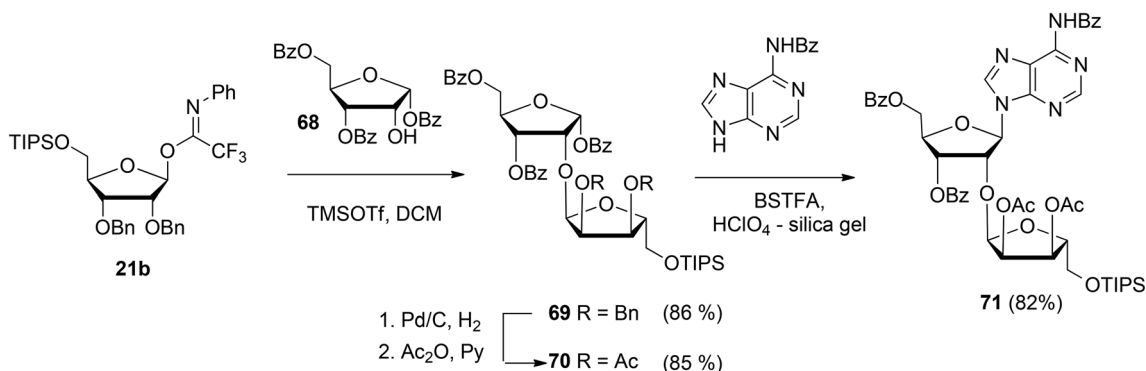




Scheme 9 Synthesis of orthogonally protected ribosylated adenosine with (A): 1-*O*-(*N*-phenyl)-2,2,2-trifluoroacetimido-2,3,5-tri-*O*-benzyl- β -ribofuranose **21c** and (B): glycosyl fluoride **66** as the donors.

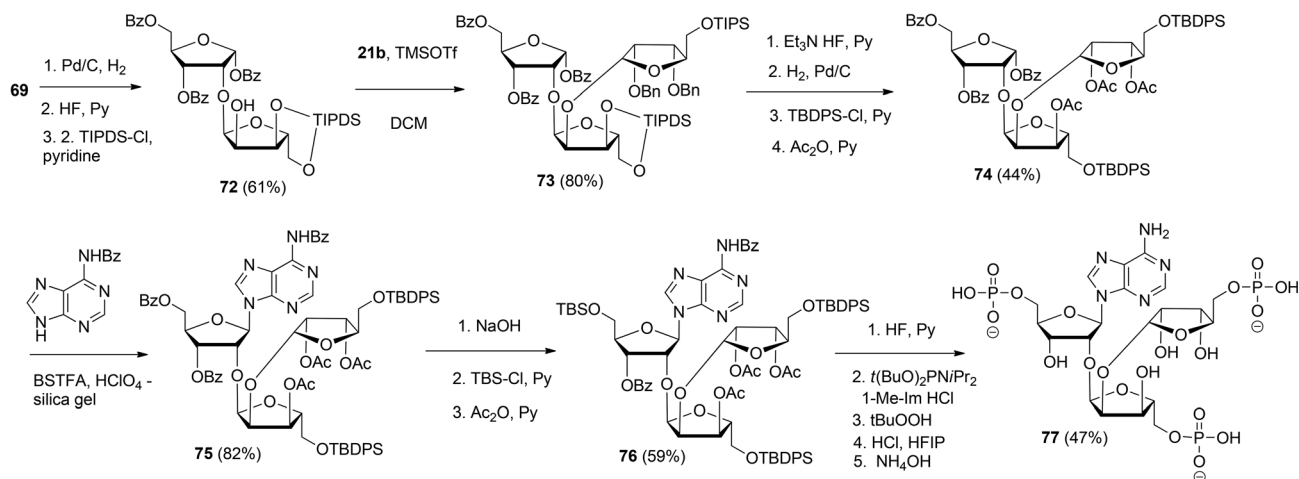
Guided by the need to scale up the process and to acquire sufficient quantities of a suitable ribosyl-adenosine building block Kistemaker *et al.*¹⁸ developed a new method. Side reactions on the nucleobase often accompany the glycosylation of protected nucleosides limiting the scalability of such synthetic strategies.³³ Therefore it was decided to install the adenine base by a Vorbrüggen reaction after the ribosylation event.^{34,35} As depicted in Scheme 10 a reaction of benzylated (*N*-phenyl)-2,2,2-trifluoroacetimidate donor **21b** and commercially available acceptor 1,3,5-tri-*O*-benzoylribose **68** led to the isolation of α -configured disaccharide **69**. After hydrogenolysis and acetylation, **70** was obtained. Vorbrüggen coupling of **70** and Bz-adenine under the influence of the immobilized acid (HClO₄-SiO₂), introduced the adenine base both regio- and β -stereoselectively. It is noteworthy that this approach gives access to a large amount of **71**, the precursor of a suitable 2-*O*-ribosylated adenosine building block.

This glycosylation strategy was adopted towards the synthesis of the branching point of ADPr-chains, culminating in the construction of both unphosphorylated³⁶ and triphosphorylated³⁷ branched ADPr fragments. The assembly of *O*- α -D-ribofuranosyl-(1''' $\rightarrow$ 2'')-*O*- α -D-ribofuranosyl-(1'' $\rightarrow$ 2')-adenosine-5',5'',5'''-tris(phosphate) **77** is depicted in Scheme 11. Protected disaccharide **69**, termed parobiose,³⁷ was converted into **72**, in which the non-reducing ribose is provided with a free 2-OH group. Subsequent TMSOTf mediated condensation of **72** with benzylated (*N*-phenyl)-2,2,2-trifluoroacetimidate donor **21b** led to all alpha configured tri-riboside **73** (protected parotriose).³⁷ After protective group manipulation the adenine base was introduced through a similar Vorbrüggen type glycosylation as described above to afford protected *O*- α -D-ribofuranosyl-(1''' $\rightarrow$ 2'')-*O*- α -D-ribofuranosyl-(1'' $\rightarrow$ 2')-adenosine **75**. Transformation of **75** to **76** allowed the three-fold introduction of di-*tert*-butyl-phosphotriesters with the aid of phosphorami-



Scheme 10 Synthesis of ribosylated adenosine via Vorbrüggen type glycosylation.



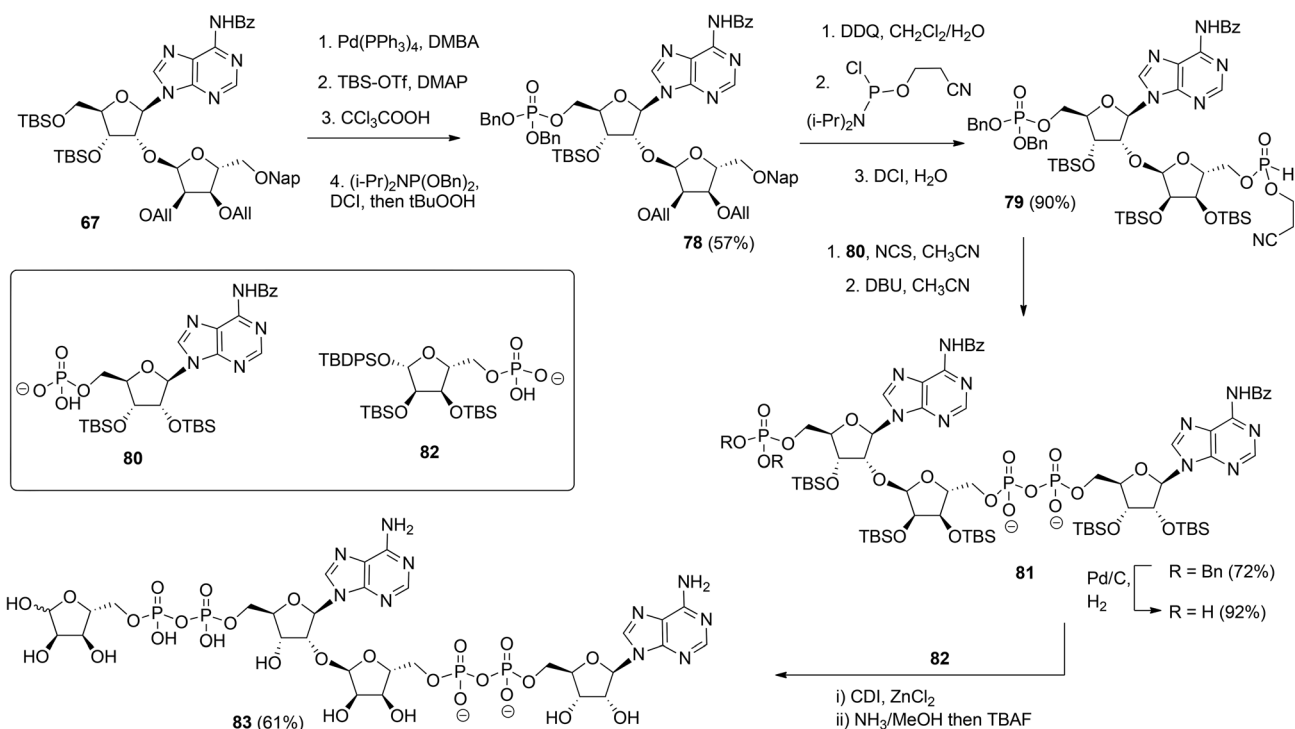


Scheme 11 Synthesis of *O*-α-D-ribofuranosyl-(1' → 2'')-*O*-α-D-ribofuranosyl-(1'' → 2')-adenosine-5',5'',5'''-tris(phosphate): the branching point of ADPr-chain.

dite chemistry. Treatment of **76** with 10 equivalents of di-*tert*-butyl-*N,N*-diisopropylphosphoramidite under the influence of 1-methylimidazole and 1-methylimidazolium chloride as the activator, followed by oxidation of the intermediate phosphite triesters and global deprotection yielded target triphosphate **77** which proved to be identical to the native structure^{38,39} as ascertained by NMR spectroscopy.

2.2.2 Synthetic approaches to oligo-ADPr. Pyrophosphates are important functional groups in a wide array of naturally occurring compounds and a lot of procedures for the synthesis

of pyrophosphates have been reported.^{19,40–48} However, the occurrence of multiple pyrophosphates in one molecule, such as in oligo-ADPr, is unprecedented and presents a special challenge. Only two syntheses of short fragments of oligo-ADPr have been published to date. Lambrecht *et al.*²⁹ (Scheme 12) reported a solution phase synthesis of an ADPr dimer, in which they relied on the classic Atherton–Todd chemistry to construct pyrophosphate bridges.²⁹ In their route of synthesis ribosylated adenosine building block **67**, obtained as described above (Scheme 9), was subjected to protective group



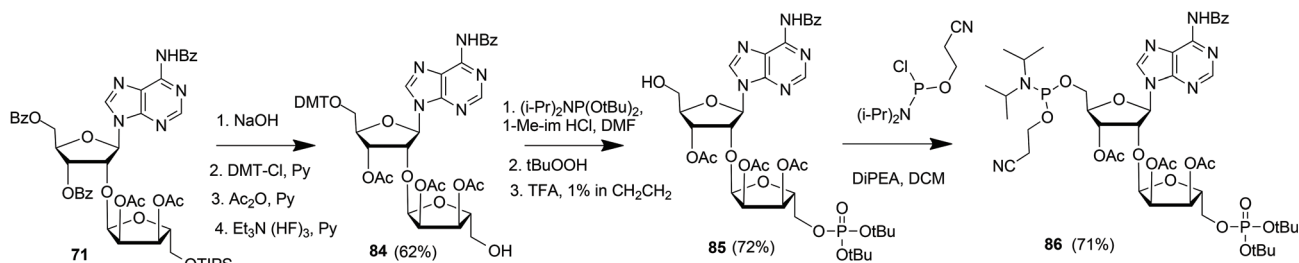
Scheme 12 Synthesis of ADPr dimer **83** by Lambrecht *et al.*



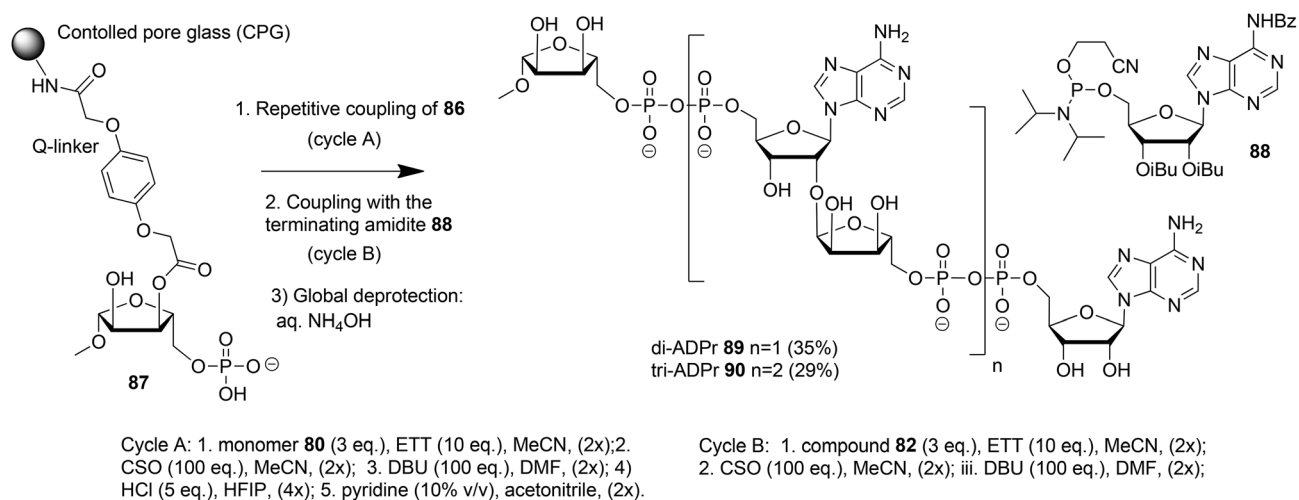
manipulation to allow the installation of a dibenzyl phosphotriester at the primary OH of adenosine with the aid of phosphoramidite chemistry and subsequent oxidation. The naphthyl ether at the 5'-OH of the ribose moiety in thus obtained **78** was selectively removed and a H-phosphonate diester was introduced with 2-cyanoethyl *N,N*-diisopropylchlorophosphoramidite and subsequent hydrolysis of the intermediate phosphoramidite to give **79**. Oxidative chlorination of H-phosphonate of **79** with NCS afforded an intermediate chlorophosphate which was condensed with adenosine monophosphate **80** to give after the removal of the cyanoethyl group pyrophosphate **81** (R = Bn) in good yield. Hydrogenolysis of the benzyl group afforded terminal phosphate **81** (R = H). Unfortunately, the introduction of the second pyrophosphate with the same method failed. Therefore silylated ribose monophosphate **82** was treated with CDI and condensation of the resulting phosphorimidazolide with **81** (R = H) gave after the removal of all protecting groups the target ADP dimer **83** in good yield.

Kistemaker *et al.* reported a solid phase synthesis of both an ADPr dimer and trimer (Scheme 14).¹⁸ To be able to introduce multiple pyrophosphates a method to access sugar-nucleotides which was based on the combination of P^{III} - P^V

chemistry was investigated.¹⁹ This methodology proved to be convenient and expedient not only for the synthesis of mono-ADP-ribosylated peptides (Scheme 5) but also for the synthesis of various bioorganic pyrophosphate derivatives both in solution^{47,49,50} and on the solid phase.^{51,52} It was expected that this P^{III} - P^V method would be uniquely suitable for the repeated pyrophosphorylation on the solid phase, not least due to its mild nature and fast kinetics. To be able to introduce multiple pyrophosphate functions building block **86**, provided with di-*tert*-butyl phosphotriester and 2-cyanoethyl *N,N*-diisopropylphosphoramidite, was designed. The synthesis of building block **86** (Scheme 13) started with 2-*O*-ribosyladenosine **71** which was obtained in sufficient quantities and good yield as described above (Scheme 10). Protective group manipulation of dimer **71** gave **84** in which the free primary OH in the ribose moiety was provided with a di-*tert*-butyl phosphotriester using standard phosphoramidite chemistry, followed by oxidation of the intermediate phosphite triester. Finally, removal of the DMT group to give **85** and a reaction with 2-cyanoethyl *N,N*-diisopropylchlorophosphoramidite resulted in the isolation of building block **86** that contains both a phosphoramidite and a protected precursor of the phosphate monoester.



Scheme 13 Synthesis of ribosylated adenosine building block **86** suitable for solid-phase preparation of oligo-ADPr fragments.



Scheme 14 Solid-phase synthesis of ADPr dimer **89** and trimer **90**.

The solid phase synthesis of an ADPr dimer **89** and trimer **90** using building block **86** and terminating building block **88** is shown in Scheme 14.

Guided by the state of the art in automated DNA synthesis, controlled pore glass (CPG)¹⁸ with long alkyl amine chains was used as the solid support while hydroquinone-*O,O'*-diacetic acid (Q-linker) was selected as a linker for its improved resistance to DBU that was used to cleave 2-cyanoethyl protection from the pyrophosphate. Functionalization of this solid support with protected ribose and introduction of the phosphate monoester at the primary position gave "initiator" **87**. At this stage, up to two coupling cycles with building block **86** were undertaken. The coupling cycle took one hour and involved 5-ethylthiotetrazole (ETT) mediated condensation of **80** with the immobilized monophosphate, oxidation of the obtained labile phosphite-phosphate (P^{III}-P^V) intermediate by (1*S*)-(+)-(10-camphorsulfonyl)oxaziridine (CSO), and removal of the 2-cyanoethyl group in the partially protected pyrophosphate intermediate with DBU. The final unmasking of the di-*tert*-butyl phosphotriester with HCl/HFIP followed by neutralization with pyridine allowed the next elongation with either **86** or terminating building block **88**. The immobilized and partially protected ADPr dimer and trimer were cleaved from the resin and completely deprotected by treatment with aqueous ammonia and purified to give milligram quantities of ADPr dimer **89** and trimer **90**.

3. Conclusions and outlook

It can be concluded that significant progress has been achieved in the last ten years towards the synthesis of both mono-ADPr peptides and oligo-ADPr.

For the synthesis of mono-ADPr-peptides most successes have been achieved in the preparation of the constructs, in which the ADPr-moiety is attached to the peptide backbone *via* isosteres of the natural amino acids either close, such as Gln⁹ instead of native Glu, or remote,²² for example, a triazole linkage²⁷ as a substitute for the arginine side chain. Native ADPr-Ser containing peptides¹⁰ have been synthesized nevertheless. A fully synthetic ADPr-protein has been prepared *via* a convergent approach based on copper-catalyzed click chemistry and has been shown to be biologically active.²⁷ The synthetic ADPr-peptides have been extensively applied in biochemical^{9,53} and proteomics studies.^{54–56} Thus, ADPr-peptides have been used as the essential calibration standard to enable quantification of ADP-ribosylation in the proteome.^{54,55} Another application of the synthetic ADPr-peptides is to validate the pull-down steps in the ELTA-methodology that has been developed for selective labeling of ADP-ribose and ADP-ribosylated proteins.⁵⁷ Concerning the synthetic ADPr-oligomers, the dimeric ADPr was cocrystallized with PARG to investigate the details of the mechanism of the catalytic action of this enzyme. The synthetic dimeric and trimeric ADPr fragments were essential for determining the minimum length of PAR necessary to bind ALC1, which is a chromatin remodeler involved in oncogenesis.⁵³ Looking towards the future appli-

cations, one can envisage the use of synthetic ADPr-oligomers in experiments for which less well-defined and, arguably, less homogeneous PAR-fragments of a biological origin have been applied up to now.^{58–61} Future synthetic studies will probably focus on the native ADP-ribosylated peptides, containing ADPr-Arg, ADPr-Cys, ADPr-Tyr and ADPr-His as the ADP-ribosylation sites. To make the synthesis of ADPr-peptides more practical, it is imperative to develop a synthetic methodology that does not require significant alterations in the standard SPSS protocols such as the use of alkali labile side-chain protection and linkers. Another important avenue of research is the development of more chemically robust derivatives, for example, mono-ADPr-peptides that contain carba-ribose⁶² residues instead of native ribofuranose and methylene bisphosphonate⁶³ as pyrophosphate isosteres. Such stabilized derivatives of ADP-ribosylated biomolecules may prove invaluable for structural studies on the native enzymes because one can then forego the use of catalytically inactive mutants.⁶⁴ Methylene bisphosphonates have been used in the past as substitutes of the native pyrophosphate in studies aimed to generate antibodies for mono-ADP-ribosylated proteins.⁶⁵ It has been found that rendering the native pyrophosphate nuclease resistant by converting it into methylene bisphosphonate was essential to raise antibodies against ADP-ribose.

Both solid-phase¹⁸ and solution²⁹ methodologies have been published for the synthesis of ADPr-dimers. It seems, however, that the solid-phase synthesis is more practical as evident from the fact that tri-ADPr was prepared using a solid-phase approach. Various methods toward the introduction of the key α -glycosidic bond that connects the monomers in the ADPr-chain have been explored. It appears that the methods that involved direct glycosylation of the 2'-OH of ribose, whether in the absence of the adenine nucleobase or having the nucleobase preinstalled, are the most useful ones for the monomer synthesis. In particular, glycosylation of the 2-OH of ribose followed by the introduction of adenine *via* Vorbrüggen type condensation allowed the preparation of the branching point of the ADPr-chain.^{36,37} Up to now, the P(III)-P(V) approach to pyrophosphate formation¹⁹ has been proven to be suitable for the synthesis of the ADPr-chains up to trimers on a solid phase, while the classic method of pyrophosphate formation allowed the synthesis of di-ADPr in solution. Such short fragments already proved to be useful in the biochemical⁵³ and structural studies.²⁹

Further developments in the methodology of chemical ADP-ribosylation can be envisaged. Improved temporal protective groups for phosphate monoesters should enable the synthesis of longer ADPr-chains, perhaps up to 15–25 ADPr units long. Various resins, such as polystyrene-based resins, could be explored for scale up and the preparation of oligo-ADPr fragments grafted to peptides. The synthesis of branched oligo-ADPr and ADPr-chains attached to peptides should also be possible.

Conflicts of interest

There are no conflicts to declare.



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References

- 1 B. Luscher, M. Butepage, L. Eckei, S. Krieg, P. Verheugd and B. H. Shilton, *Chem. Rev.*, 2018, **118**, 1092–1136.
- 2 P. O. Hassa, S. S. Haenni, M. Elser and M. O. Hottiger, *Microbiol. Mol. Biol. Rev.*, 2006, **70**, 789–829.
- 3 W. L. Kraus, *Mol. Cell*, 2015, **58**, 902–910.
- 4 M. S. Cohen and P. Chang, *Nat. Chem. Biol.*, 2018, **14**, 236–243.
- 5 D. Slade, M. S. Dunstan, E. Barkauskaite, R. Weston, P. Lafite, N. Dixon, M. Ahel, D. Leys and I. Ahel, *Nature*, 2011, **477**, 616–620.
- 6 P. Fontana, J. J. Bonfiglio, L. Palazzo, E. Bartlett, I. Matic and I. Ahel, *eLife*, 2017, **6**, e28533.
- 7 M. Butepage, L. Eckei, P. Verheugd and B. Luscher, *Cells*, 2015, **4**, 569–595.
- 8 G. J. van der Heden van Noort, M. G. van der Horst, H. S. Overkleeft, G. A. van der Marel and D. V. Filippov, *J. Am. Chem. Soc.*, 2010, **132**, 5236–5240.
- 9 H. A. Kistemaker, A. P. Nardozza, H. S. Overkleeft, G. A. van der Marel, A. G. Ladurner and D. V. Filippov, *Angew. Chem., Int. Ed.*, 2016, **55**, 10634–10638.
- 10 J. Voorneveld, J. G. M. Rack, I. Ahel, H. S. Overkleeft, G. A. van der Marel and D. V. Filippov, *Org. Lett.*, 2018, **20**, 4140–4143.
- 11 A. Sekine, M. Fujiwara and S. Narumiya, *J. Biol. Chem.*, 1989, **264**, 8602–8605.
- 12 N. Ogata, K. Ueda and O. Hayaishi, *J. Biol. Chem.*, 1980, **255**, 7610–7615.
- 13 M. A. Bonache, F. Nuti, A. Le Chevalier Isaad, F. Real-Fernández, M. Chelli, P. Rovero and A. M. Papini, *Tetrahedron Lett.*, 2009, **50**, 4151–4153.
- 14 F. Nisic, G. Speciale and A. Bernardi, *Chem. – Eur. J.*, 2012, **18**, 6895–6906.
- 15 F. Nisic and A. Bernardi, *Carbohydr. Res.*, 2011, **346**, 465–471.
- 16 G. Speciale, A. Bernardi and F. Nisic, *Molecules*, 2013, **18**, 8779–8785.
- 17 H. A. Kistemaker, G. J. van Noort, H. S. Overkleeft, G. A. van der Marel and D. V. Filippov, *Org. Lett.*, 2013, **15**, 2306–2309.
- 18 H. A. Kistemaker, L. N. Lameijer, N. J. Meeuwenoord, H. S. Overkleeft, G. A. van der Marel and D. V. Filippov, *Angew. Chem., Int. Ed.*, 2015, **54**, 4915–4918.
- 19 H. Gold, P. van Delft, N. Meeuwenoord, J. D. C. Codée, D. V. Filippov, G. Eggink, H. S. Overkleeft and G. A. van der Marel, *J. Org. Chem.*, 2008, **73**, 9458–9460.
- 20 O. Leidecker, J. J. Bonfiglio, T. Colby, Q. Zhang, I. Atanassov, R. Zaja, L. Palazzo, A. Stockum, I. Ahel and I. Matic, *Nat. Chem. Biol.*, 2016, **12**, 998–1000.
- 21 J. J. Bonfiglio, P. Fontana, Q. Zhang, T. Colby, I. Gibbs-Seymour, I. Atanassov, E. Bartlett, R. Zaja, I. Ahel and I. Matic, *Mol. Cell*, 2017, **65**, 932–940.
- 22 L. Palazzo, O. Leidecker, E. Prokhorova, H. Dauben, I. Matic and I. Ahel, *eLife*, 2018, **7**, e34334.
- 23 Q. Liu, B. I. Florea and D. V. Filippov, *Cell Chem. Biol.*, 2017, **24**, 431–432.
- 24 P. M. Moyle and T. W. Muir, *J. Am. Chem. Soc.*, 2010, **132**, 15878–15880.
- 25 F. Wahl and M. Mutter, *Tetrahedron Lett.*, 1996, **37**, 6861–6864.
- 26 M. R. Carrasco and R. T. Brown, *J. Org. Chem.*, 2003, **68**, 8853–8858.
- 27 Q. Liu, H. A. V. Kistemaker, S. Bhogaraju, I. Dikic, H. S. Overkleeft, G. A. van der Marel, H. Ova, G. J. van der Heden van Noort and D. V. Filippov, *Angew. Chem., Int. Ed.*, 2018, **57**, 1659–1662.
- 28 L. Li, Q. Li, S. Ding, P. Xin, Y. Zhang, S. Huang and G. Zhang, *Molecules*, 2017, **22**, 1346.
- 29 M. J. Lambrecht, M. Brichacek, E. Barkauskaite, A. Ariza, I. Ahel and P. J. Hergenrother, *J. Am. Chem. Soc.*, 2015, **137**, 3558–3564.
- 30 S. N. Mikhailov, I. V. Kulikova, K. Nauwelaerts and P. Herdewijn, *Tetrahedron*, 2008, **64**, 2871–2876.
- 31 M. Zheng, M. Mex, K. H. Gotz and A. Marx, *Org. Biomol. Chem.*, 2018, **16**, 8904–8907.
- 32 B. Shirinfar and N. Ahmed, *Helv. Chim. Acta*, 2018, **101**, e1700226.
- 33 G. J. van der Heden van Noort, H. S. Overkleeft, G. A. van der Marel and D. V. Filippov, *Org. Lett.*, 2011, **13**, 2920–2923.
- 34 H. Vorbrüggen and K. Krolikiewicz, *Angew. Chem., Int. Ed. Engl.*, 1975, **14**, 421–422.
- 35 H. Vorbrüggen, K. Krolikiewicz and B. Bennua, *Chem. Ber.*, 1981, **114**, 1234–1255.
- 36 H. A. Kistemaker, H. S. Overkleeft, G. A. van der Marel and D. V. Filippov, *Org. Lett.*, 2015, **17**, 4328–4331.
- 37 Q. Liu, H. A. V. Kistemaker, H. S. Overkleeft, G. A. van der Marel and D. V. Filippov, *Chem. Commun.*, 2017, **53**, 10255–10258.
- 38 M. Miwa, M. Ishihara, S. Takishima, N. Takasuka, M. Maeda, Z. Yamaizumi, T. Sugimura, S. Yokoyama and T. Miyazawa, *J. Biol. Chem.*, 1981, **256**, 2916–2921.
- 39 M. Miwa, N. Saikawa, Z. Yamaizumi, S. Nishimura and T. Sugimura, *Proc. Natl. Acad. Sci. U. S. A.*, 1979, **76**, 595–599.
- 40 M. Dong, T. Kirchberger, X. Huang, Z. J. Yang, L. R. Zhang, A. H. Guse and L. H. Zhang, *Org. Biomol. Chem.*, 2010, **8**, 4705–4715.
- 41 X. Xue, R. B. Zheng, A. Koizumi, L. Han, J. S. Klassen and T. L. Lowary, *Org. Biomol. Chem.*, 2018, **16**, 1939–1957.
- 42 F. Ravalico, I. Messina, M. V. Berberian, S. L. James, M. E. Migaud and J. S. Vyle, *Org. Biomol. Chem.*, 2011, **9**, 6496–6497.



- 43 A. R. Kore and G. Parmar, *Synth. Commun.*, 2006, **36**, 3393–3399.
- 44 I. B. Yanachkov, E. J. Dix, M. I. Yanachkova and G. E. Wright, *Org. Biomol. Chem.*, 2011, **9**, 730–738.
- 45 S. Wendicke, S. Warnecke and C. Meier, *Angew. Chem., Int. Ed.*, 2008, **47**, 1500–1502.
- 46 A. Hofer, G. S. Cremosnik, A. C. Muller, R. Giambruno, C. Trefzer, G. Superti-Furga, K. L. Bennett and H. J. Jessen, *Chem. – Eur. J.*, 2015, **21**, 10116–10122.
- 47 Q. Zhang, P. L. Howell, H. S. Overkleeft, D. V. Filippov, G. A. van der Marel and J. D. C. Codee, *Carbohydr. Res.*, 2017, **450**, 12–18.
- 48 N. Qi, K. Jung, M. Wang, L. X. Na, Z. J. Yang, L. R. Zhang, A. H. Guse and L. H. Zhang, *Chem. Commun.*, 2011, **47**, 9462–9464.
- 49 I. Pavlovic, D. T. Thakor, J. R. Vargas, C. J. McKinlay, S. Hauke, P. Anstaett, R. C. Camuna, L. Bigler, G. Gasser, C. Schultz, P. A. Wender and H. J. Jessen, *Nat. Commun.*, 2016, **7**, 10622.
- 50 J. G. M. Rack, A. Ariza, B. S. Drown, C. Henfrey, E. Bartlett, T. Shirai, P. J. Hergenrother and I. Ahel, *Cell Chem. Biol.*, 2018, **25**, 1533–1546.
- 51 G. S. Cremosnik, A. Hofer and H. J. Jessen, *Angew. Chem., Int. Ed.*, 2014, **53**, 286–289.
- 52 B. A. Anderson and R. Krishnamurthy, *Chem. – Eur. J.*, 2018, **24**, 6837–6842.
- 53 H. R. Singh, A. P. Nardoza, I. R. Moller, G. Knobloch, H. A. V. Kistemaker, M. Hassler, N. Harrer, C. Blessing, S. Eustermann, C. Kotthoff, S. Huet, F. Mueller-Planitz, D. V. Filippov, G. Timinszky, K. D. Rand and A. G. Ladurner, *Mol. Cell*, 2017, **68**, 860–871.
- 54 V. Bilan, N. Selevsek, H. A. V. Kistemaker, J. Abplanalp, R. Feurer, D. V. Filippov and M. O. Hottiger, *Mol. Cell. Proteomics*, 2017, **16**, 949–958.
- 55 J. Abplanalp, M. Leutert, E. Frugier, K. Nowak, R. Feurer, J. Kato, H. V. A. Kistemaker, D. V. Filippov, J. Moss, A. Caflisch and M. O. Hottiger, *Nat. Commun.*, 2017, **8**.
- 56 R. L. McPherson, R. Abraham, E. Sreekumar, S.-E. Ong, S.-J. Cheng, V. K. Baxter, H. A. V. Kistemaker, D. V. Filippov, D. E. Griffin and A. K. L. Leung, *Proc. Natl. Acad. Sci. U. S. A.*, 2017, **114**, 1666–1671.
- 57 Y. Ando, E. Elkayam, R. L. McPherson, M. Dasovich, S. J. Cheng, J. Voorneveld, D. V. Filippov, S. E. Ong, L. Joshua-Tor and A. K. L. Leung, *Mol. Cell*, 2019, **73**, 845–856.
- 58 J. Fahrner, R. Kranaster, M. Altmeyer, A. Marx and A. Burkle, *Nucleic Acids Res.*, 2007, **35**, e143.
- 59 Q. Chen, M. A. Kassab, F. Dantzer and X. Yu, *Nat. Commun.*, 2018, **9**, 3233.
- 60 T. I. Kam, X. Mao, H. Park, S. C. Chou, S. S. Karuppagounder, G. E. Umanah, S. P. Yun, S. Brahmachari, N. Panicker, R. Chen, S. A. Andrabi, C. Qi, G. G. Poirier, O. Pletnikova, J. C. Troncoso, L. M. Bekris, J. B. Leverenz, A. Pantelyat, H. S. Ko, L. S. Rosenthal, T. M. Dawson and V. L. Dawson, *Science*, 2018, 362.
- 61 K. A. Krukenberg, S. Kim, E. S. Tan, Z. Maliga and T. J. Mitchison, *Chem. Biol.*, 2015, **22**, 446–452.
- 62 S. Shuto, M. Fukuoka, A. Manikowsky, Y. Ueno, T. Nakano, R. Kuroda, H. Kuroda and A. Matsuda, *J. Am. Chem. Soc.*, 2001, **123**, 8750–8759.
- 63 S. B. Engelsma, N. J. Meeuwenoord, H. S. Overkleeft, G. A. van der Marel and D. V. Filippov, *Angew. Chem. Int. Ed.*, 2017, **56**, 2955–2959.
- 64 M. F. Langelier, L. Zandarashvili, P. M. Aguiar, B. E. Black and J. M. Pascal, *Nat. Commun.*, 2018, **9**, 844.
- 65 T. Meyer and H. Hilz, *Eur. J. Biochem.*, 1986, **155**, 157–165.

