

Cite this: *Chem. Commun.*, 2018, 54, 13018Received 24th September 2018,
Accepted 10th October 2018

DOI: 10.1039/c8cc07667g

rsc.li/chemcomm

A desymmetrization-based approach to morphinans: application in the total synthesis of oxycodone†‡

Kun Ho (Kenny) Park and David Y.-K. Chen[†] *

Here we report a total synthesis of the pharmacologically significant morphinan alkaloid, oxycodone. The centerpiece of the developed strategy features the first application of the Rovis desymmetrization of peroxyquinol in target-oriented total synthesis to access an optically active phenanthrene framework shared by the morphinans. A Stork–Ueno radical cyclization under photoredox conditions installed the all-carbon quaternary stereocenter, and a late-stage reductive desilylation with concomitant piperidine formation secured the core structure of the target molecule.

The ready availability of symmetrical compounds, either naturally occurring or synthetic, has had a profound impact on synthetic organic chemistry.¹ In the retrosynthetic analysis of molecules which do not possess obvious symmetry element(s), unveiling a symmetrical sub-structure or a potential synthetic precursor is often non-trivial. Even if a potential synthetic precursor can be identified, the choice of desymmetrization method in the forward synthesis remains highly challenging especially if the process is also intended to deliver enantioselectivity. Our laboratory has a long-standing interest in desymmetrization-based synthetic approaches to access architecturally complex natural products, and during the course of our studies has successfully demonstrated conceptually novel solutions to challenging synthetic problems.²

Naturally occurring and synthetic morphinans are well-recognized for their therapeutic value as well as illicit usage, and continue to be the center of chemical³ and biological investigations⁴ more than 90 years after Robinson's landmark structural elucidation of morphine.⁵ In accordance to the proposed desymmetrization approach to access the generic morphinan core structure shown in Scheme 1, our synthesis commenced with the preparation of biaryl phenol intermediate **1**.



Scheme 1 Proposed desymmetrization approach to morphinans and oxycodone (**2**). MOM = methoxymethyl.

As shown in Scheme 2, a Wittig-olefination⁶ between benzaldehyde derivative **3**⁷ and phosphonium salt **4**⁸ followed by a controlled hydrogenation (Pd/C, H₂) readily delivered biaryl phenol **1** on a multi-gram scale, provided that hydrogenation was performed under carefully monitored reaction conditions to avoid any undesired reductive debromination. Oxidative dearomatization of phenol **1** under hypervalent iodine conditions (PIDA) afforded hydroxy dienone **5** uneventfully,⁹ which underwent an intramolecular Heck cyclization [Pd(OAc)₂, PPh₃] followed by a chemoselective hydrogenation [Rh(PPh₃)₃Cl, H₂] to furnish hydroxy enone **7**. Hydroxy enone **7** could also be prepared through a Rh(PPh₃)₃Cl catalyzed hydrosilylation of dienone **5** followed by an intramolecular Heck reaction of the intermediate hydroxy enone **8**. Interestingly, the intramolecular Heck condition developed for hydroxy dienone **5** was totally ineffective for hydroxy enone **8**, therefore, after extensive studies, an alternative protocol involving PdCl₂(PPh₃)₂, X-Phos, and *t*Bu₃PH·BF₄ was implemented to realize the desired transformation.¹⁰ With phenanthrene system **7** in hand, installation of the all-carbon quaternary stereocenter in the morphinans was accomplished using both conventional (*n*Bu₃SnH, Et₃B) and a photoredox variant¹¹ of the Stork–Ueno radical cyclization¹² through the intermediacy of iodoacetal **9**, to deliver tetracyclic acetal **11** as an inconsequential mixture of diastereoisomers in a 57–62% overall yield. Evidently, the operationally benign photoredox protocol circumvented extensive chromatographic removal of organotin residues which proved synthetically more attractive. Treatment of **11** with *m*CPBA in the presence of BF₃·OEt₂¹³ smoothly oxidized its acetal moiety to the corresponding lactone,

Department of Chemistry, Seoul National University, Gwanak-1 Gwanak-ro, Gwanak-gu, Seoul 08826, South Korea. E-mail: davidchen@snu.ac.kr

† This paper is dedicated to Professor Chi-Huey Wong on the occasion of his 70th birthday.

‡ Electronic supplementary information (ESI) available. See DOI: 10.1039/c8cc07667g



Scheme 2 Synthesis of pentacyclic ketolactone **13**. *Reagents and conditions:* (a) NaH (3.5 equiv.), THF, 0 °C to 23 °C, 2 h, 85%; (b) Pd/C (10% wt/wt, 0.05 equiv.), H₂, EtOAc/MeOH (4 : 1), 23 °C, 1 h, 70%; (c) PIDA (1.2 equiv.), MeCN/H₂O (1 : 1), 0 °C, 2 h, 66%; (d) Pd(OAc)₂ (0.15 equiv.), K₂CO₃ (3.0 equiv.), PPh₃ (0.3 equiv.), toluene, 110 °C, 24 h, 80%; (e) Rh(PPh₃)₃Cl (0.05 equiv.), H₂, benzene, 23 °C, 12 h, 78%; (f) Rh(PPh₃)₃Cl (0.05 equiv.), Et₃SiH (2.0 equiv.), benzene, 50 °C, 10 h, recovered **5**: 16%; then TBAF (1.0 M in THF, 1.8 equiv.), THF, 0 °C to 23 °C, 4 h, **8**: 42% for two steps; (g) PdCl₂(PPh₃)₂ (0.15 equiv.), Na₂CO₃ (3.0 equiv.), X-Phos (0.4 equiv.), tBu₃PH•BF₄ (0.8 equiv.), DMSO, 100 °C, 40 h, 69%; (h) NIS (6.0 equiv.), ethyl vinyl ether (8.0 equiv.), CH₂Cl₂, -20 °C to 23 °C, 12 h; (i) [Ir(ppy)₂(dtbbpy)]PF₆ (0.03 equiv.), iPr₂NEt (10 equiv.), 12 W lamp, MeCN, 36 h, 57% for two steps; or Et₃B (1.3 equiv.), nBu₃SnH (1.3 equiv.), air, benzene, 23 °C, 12 h, 62% for two steps; (j) mCPBA (1.8 equiv.), BF₃•OEt₂ (2.0 equiv.), CH₂Cl₂, 0 °C, 1 h, 75%; (k) PyH•Br₃ (1.1 equiv.), CH₂Cl₂/AcOH (1 : 3.4), 23 °C, 30 min; then LiBr (6.1 equiv.), Et₃N (1.6 equiv.), CH₃CN, 60 °C, 10 min, 64%. Bn = benzyl; mCPBA = *meta*-chloroperoxy benzoic acid; Cy = cyclohexyl; [Ir(ppy)₂(dtbbpy)]PF₆ = [4,4'-bis(1,1-dimethylethyl)-2,2'-bipyridine-*N*1,*N*1']bis[2-(2-pyridinyl-*N*)phenyl-*C*]iridium(III) hexafluorophosphate; NIS = *N*-iodosuccinimide; PIDA = [bis(acetoxy)iodo]benzene; Pd(OAc)₂ = palladium(II) acetate; PyH•Br₃ = pyridinium tribromide; TBAF = tetra-*n*-butylammonium fluoride; and X-Phos = 2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl.

which also fortuitously removed the MOM ether to afford phenol **12**. Advancing tetracyclic ketone **12** next called for the formation of the five-membered oxacycle that resides in several flagship morphinans.⁴ To this end, application of the phenol-directed bromination (PyH•Br₃) followed by a thermally induced cyclization developed by the Fukuyama laboratory¹⁴ completed the synthesis of pentacyclic key intermediate **13** in nine-steps from **3** and **4**.

As shown in Scheme 3a, pentacyclic ketone **13** was next protected as its dioxolane derivative followed by treatment of the resulting lactone **14** with methylamine to provide hydroxy amide **15**. Exhaustive reduction of amide **15** and subsequent tosylation of the resulting amine furnished sulfonamide **16**, which further underwent a benzylic bromination (NBS) and HBr elimination to afford styrene derivative **17**. Interestingly, switching the order of exhaustive amide reduction and styrene olefin introduction led to the formation of saturated tosylamide **16** (Scheme 3b), whereas reduction of styrene containing lactone **18/18a** afforded diol **21** uneventfully with complete retention of the olefin. These findings clearly suggested that the proximal nitrogen played an assisting role in the unexpected saturation of the styrene olefin in **19**. Furthermore, introduction of the morphinan nitrogen atom *via* amidation of lactone **14** proved uniquely effective, as conventional substitution reactions engaging diol **21** and reductive amination protocols on acetal or hemiacetal **24** were both unsuccessful in our studies (Scheme 3c).

Tosylamide **17** bearing a tertiary alcohol was poised for the completion of oxycodone (**2**, Scheme 3a), a pharmacologically

important semi-synthetic morphinan currently used in clinics.¹⁵ Accordingly, detosylation of **17** under Birch-type conditions took place with concomitant piperidine formation, and subsequently furnished oxycodone (**2**) upon dioxolane removal (HCl aq.). Related Birch-type reductive cyclization was featured in several previous syntheses of morphinans,³ however, the feasibility of allylic alcohol system **17** has never been demonstrated. All physical properties of synthetic oxycodone (**2**) were in perfect agreement with the reported data,¹⁴ and this developed sequence represented a significant reduction in step-count compared to the first and only reported total synthesis of oxycodone (**2**) to date by the Fukuyama laboratory.¹⁴

Finally, by accessing optically active tricyclic tertiary alcohol **7**, an asymmetric version of the aforementioned synthetic pathway is outlined in Scheme 4. Inspired by the work of Feringa,¹⁶ our initial efforts toward an optically active dienone **6** were based on the asymmetric Heck reaction of hydroxy dienone **5** in the presence of a chiral ligand. However, after screening numerous conditions only a minuscule level of asymmetric induction was observed (Scheme 4a). After contemplating several cyclohexadienone-based desymmetrization protocols¹⁷ including the enantioselective conjugate reduction of prochiral 2,5-cyclohexadione recently reported by Corey and co-workers (Scheme 4b),¹⁸ we were attracted to the effectiveness and novelty of the enantioselective desymmetrization of peroxyquinols developed by the Rovis laboratory (Scheme 4c).¹⁹ In this context, application of the Rovis technology took advantage of the earlier described phenol intermediate **1** and its oxidized peroxyquinol derivative **26**, which on treatment under the



Scheme 3 (a) Completion of total synthesis of oxycodone (2); (b) contrasting results in the lithium aluminium hydride mediated reduction of styrene-containing lactone **18** and amide **19**; (c) attempted introduction of morphinan nitrogen via substitution and reductive amination conditions. *Reagents and conditions:* (a) TMSCl (4.8 equiv.), CH₂Cl₂/ethylene glycol (1:1), 23 °C, 7 h, 76%; (b) MeNH₂ (10 equiv.), MeOH/CH₂Cl₂ (6:1), 23 °C, 12 h, 86%; (c) LiAlH₄ (10 equiv.), THF, reflux, 12 h; then TsCl (1.4 equiv.), Et₃N (2.6 equiv.), CH₂Cl₂, 23 °C, 4 h, 66% for two steps; (d) NBS (1.1 equiv.), benzoyl peroxide (0.1 equiv.), CCl₄, reflux, 45 min; then Et₃N (16.5 equiv.), reflux, 10 min, 70%; (e) Li (excess), NH₃, tBuOH/THF (1:10), -78 °C, 15 min, 70%; then HCl (2.0 N aq.)/THF (1:10), 80 °C, 12 h, 61%; (f) NBS (1.1 equiv.), benzoyl peroxide (0.2 equiv.), CCl₄, reflux, 45 min; then Et₃N (5.6 equiv.), reflux, 10 min, **18**: **18a** ca. 5:1, 64%; (g) LiAlH₄ (8.3 equiv.), THF, 50 °C, 5 h, 51%; (h) MeNH₂ (10 equiv.), MeOH/CH₂Cl₂ (6:1), 23 °C, 12 h, 92%; (i) LiAlH₄ (10 equiv.), THF, reflux, 12 h, 63%; then TsCl (1.2 equiv.), Et₃N (10.8 equiv.), CH₂Cl₂, 23 °C, 3 h, 76%. MsCl = methanesulfonyl chloride; NBS = *N*-bromosuccinimide; TMSCl = trimethylsilyl chloride; Ts = *para*-toluenesulfonyl; TsCl = *para*-toluenesulfonyl chloride.

conditions described by Rovis and co-workers in the presence of chiral phosphoric acid (*R*)-TRIP afforded peroxyacetal **27** in 66% yield as a single diastereoisomer. Interestingly, a racemic version of the same process mediated by diphenylphosphinic



Scheme 4 Preparation of optically active intermediate **7**. (a) Initial exploration of an asymmetric intramolecular Heck reaction of prochiral hydroxy dienone **5**; (b) attempted enantioselective desymmetrizing reduction of prochiral hydroxy dienone **5**; (c) synthesis of optically active hydroxy enone **7** using Rovis asymmetric peroxyquinol chemistry. *Reagents and conditions:* (a) TPP (0.05 equiv.), O₂, 27 W lamp, CHCl₃, 23 °C, 5 d, 46%; (b) (*R*)-TRIP (0.05 equiv.), isobutyraldehyde (1.2 equiv.), 4 Å MS, 1,2-dichloroethane, 45 °C, 21 h, 66%; (c) Rh/Al, PtO₂ (0.3 equiv.), H₂, EtOAc, 3 h; then DBU (2.6 equiv.), CH₂Cl₂, 45 °C, 4 h, 43% for two steps. MS = molecular sieves; TPP = 5,10,15,20-tetraphenyl-21*H*,23*H*-porphine; (*R*)-TRIP = (*R*)-3,3'-bis(2,4,6-triisopropylphenyl)-1,1'-binaphthyl-2,2'-diylhydrogenphosphate; DBU = 1,8-diazabicyclo[5.4.0]undec-7-ene.

acid resulted in a mixture of diastereoisomers (*ca.* 3:1), suggesting the influential role of chiral phosphoric acid in both enantio- and diastereocontrol (see the ESI†). Optically active peroxyacetal **27** further underwent a chemoselective hydrogenation (Rh/Al, PtO₂, and H₂) followed by a Kornblum-DeLamare type fragmentation²⁰ to furnish hydroxy enone **8** (95:5 er, determined by chiral HPLC analysis). Optically active hydroxy enone **8** also participated in an intramolecular Heck reaction under the previously described conditions for racemic **8** to afford optically active enone **7** with no erosion of optical purity, thereby establishing an asymmetric pathway to oxycodone (2).

In summary, a desymmetrization-based synthetic strategy has enabled a concise and asymmetric entry to oxycodone (2) from a readily accessible, reduced stilbene system **1**. A salient feature of the developed synthesis featured a Rovis desymmetrization of peroxyquinol **26** through chiral Bronsted acid catalysis, an intramolecular photoredox radical cyclization to forge the congested quaternary stereocenter, and a late-stage

reductive detosylation with concomitant piperidine formation. Further improvements of the developed synthetic sequence and application of the synthetic technologies discussed herein to access other naturally occurring and designed morphinans are currently under investigation in our laboratory.

This work was supported by the National Research Foundation of Korea (NRF) grant funded by the Korean government (MSIP) (No. 2013R1A1A2057837; and 2014R1A5A1011165, Center for New Directions in Organic Synthesis), and Novartis. Kenny Park was supported by the BK21Plus Program, Ministry of Education. We thank Hyeonuk Woo and Shohee Park for preliminary synthetic studies.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- For reviews and selected recent examples of the use of desymmetrization in total synthesis, see: (a) S. R. Magnuson, *Tetrahedron*, 1995, **51**, 2167; (b) M. C. Willis, *J. Chem. Soc., Perkin Trans. 1*, 1999, 1765; (c) M. Wang, M. Feng, B. Tang and X. Jiang, *Tetrahedron Lett.*, 2014, **55**, 7147.
- Y. Yoshii, H. Tokuyama and D. Y.-K. Chen, *Angew. Chem., Int. Ed.*, 2017, **56**, 12277.
- For an extensive list of morphinan syntheses, see ESI†.
- For a recent book and review chapter, see: (a) R. B. Herbert, H. Venter and S. Pos, *Nat. Prod. Rep.*, 2000, **17**, 317; (b) H. Nagase, *Chemistry of Opioids; Top. Curr. Chem.*, Springer-Verlag, Berlin Heidelberg, 2011, vol. 299, pp. 1–312.
- J. M. Gulland and R. Robinson, *Mem. Proc. – Manchester Lit. Philos. Soc.*, 1925, **69**, 79.
- For a related Wittig reaction followed by chemoselective hydrogenation, see: J. C. C. Morales, A. G. Torres and E. Gonzalez-Zamora, *Eur. J. Org. Chem.*, 2011, 3165.
- J. Duchek, G. Piercy, J. Gilmet and T. Hudlicky, *Can. J. Chem.*, 2011, **89**, 709.
- F. Colobert, R. Des Mazery, G. Solladie and M. C. Carreno, *Org. Lett.*, 2002, **4**, 1723.
- For a review, see: R. M. Moriarty and O. Prakash, *Org. React.*, 2001, **57**, 327.
- Heck reaction of substrate **8** under the conditions developed for substrate **5** only afforded recovered starting material (**8**). For an example of the modified conditions, see: T. N. Gowala and J. Pabba, *Tetrahedron Lett.*, 2015, **56**, 1801.
- H. Kim and C. Lee, *Angew. Chem., Int. Ed.*, 2012, **51**, 12303.
- For a review, see: X. J. Salom-Roig, F. Denes and P. Renaud, *Synthesis*, 2004, 1903.
- M. Inoue, T. Sato and M. Hiram, *Angew. Chem., Int. Ed.*, 2006, **45**, 4843.
- A. Kimishima, H. Umihara, A. Mizoguchi, S. Yokoshima and T. Fukuyama, *Org. Lett.*, 2014, **16**, 6244.
- For chemical, biological and clinical studies, see: (a) T. Heiskanen and E. Kalso, *Pain*, 1997, **73**, 37; (b) S. Mercadante and E. Arcuri, *Cancer Treat. Rev.*, 1998, **24**, 425; (c) C. P. Watson and N. Babul, *Neurology*, 1998, **50**, 1837; (d) B. Silvestri, E. Bandieri, S. Del Prete, G. P. Ianniello, G. Micheletto, M. Dambrosio, G. Sabbatini, L. Endrizzi, A. Marra, E. Aitini, A. Calorio, F. Garetto, G. Nastasi, F. Piantedosi, V. Sidoti and P. Spanu, *Clin. Drug Invest.*, 2008, **28**, 399.
- R. Imbos, A. J. Minnard and B. L. Feringa, *J. Chem. Soc., Dalton Trans.*, 2003, 2017.
- For a review of desymmetrization methods of cyclohexadienones, see: K. A. Kalstabakken and A. M. Harned, *Tetrahedron*, 2014, **70**, 9571.
- Y. Han, S. Breitler, S.-L. Zheng and E. J. Corey, *Org. Lett.*, 2016, **18**, 6172.
- D. M. Rubush, M. A. Morges, B. J. Rose, D. H. Thamm and T. Rovis, *J. Am. Chem. Soc.*, 2012, **134**, 13554. Absolute stereochemical assignments of optically active intermediates depicted in Scheme 4c were inferred based on the work of Rovis and co-workers.
- For a related process, see: C. W. Jefford, J.-C. Rossier and J. Boukouvalas, *Chem. Commun.*, 1987, 1593. Furthermore, a E1cb-type mechanism cannot be excluded to account for this transformation.