Chemical Science



EDGE ARTICLE

View Article Online
View Journal | View Issue



Cite this: Chem. Sci., 2019, 10, 2280

dll publication charges for this article have been paid for by the Royal Society of Chemistry

Received 22nd October 2018 Accepted 18th December 2018

DOI: 10.1039/c8sc04705g

rsc.li/chemical-science

Synergistic self-seeding in one-dimension: a route to patchy and block comicelles with uniform and controllable length?

Jiangping Xu, Dac Hang Zhou, Qing Yu, Gerald Guerin, Ian Manners and Mitchell A. Winnik (**)**

Self-seeding is a process unique to polymer crystals, which consist of regions of different chain packing order and different crystallinity. Here we report the synergistic self-seeding behaviour of pairs of corecrystalline block copolymer (BCP) micelle fragments and show how this strategy can be employed to control the morphology of these BCP comicelles. Each micelle fragment has a critical dissolution temperature (T_c), and unimers of each BCP have a characteristic epitaxial growth rate. The T_c value affects the dissolution sequence of the fragments upon heating, while the unimer growth rate affects the growth sequence upon cooling. By carefully choosing micelle fragments having different T_c values as well as growth rates, we could prepare patchy comicelles and block comicelles with uniform and controllable length. This synergistic self-seeding strategy is a simple yet effective route to control both length and morphology of core-crystalline comicelles

Introduction

Cylindrical micelles of block copolymers (BCPs) have important applications in drug delivery,^{1–3} epoxy resin toughening,⁴ and templates for hybrid materials.⁵ Other applications may be possible if one could control their core/corona structure as well as their length. Various groups have examined AB/AC diblock copolymer blends and ABC triblock copolymers as ways to obtain different core or corona structures.^{6–11} These micelle structures can be manipulated by changing the blend ratio, the block ratio or the self-assembly protocol. However, it is a challenge to control the micelle length, and cylindrical micelles with broad length distributions are usually obtained.¹²

Recently, crystallization-driven self-assembly (CDSA) has become a powerful strategy to control both the length and the structure of cylindrical micelles.^{13–24} Like an initiator of living chain polymerization of monomers, a seed micelle with a crystalline core can direct the epitaxial deposition of the newly added BCP unimers (seeded-growth). The length of the cylindrical micelles can be controlled precisely by changing the

unimer-to-seed ratio. 16,20 Moreover, one can manipulate the crystallization kinetics to control the morphology of cylindrical micelles. 25-28 For example, in the seeded-growth process, a slow crystallization rate leads to linear micelles, while a rapid crystallization rate can lead to branched micelles. 15 In addition, by sequential addition of unimers bearing different corona chains, block comicelles can be obtained, 19,20 while simultaneous addition of pairs of different unimers leads to patchy micelles characterized by phase-segregation into nano-sized local patches. 23,27,28

We are interested in developing methodologies to construct well-defined morphologies of cylindrical micelles by manipulating their crystallization behaviour. Several years ago, we discovered that "self-seeding" of BCP micelle crystallites was an effective means to control micelle length. 14,29,30 Self-seeding is a phenomenon unique to polymer crystals, which consist of regions of different chain packing order and different crystallinity, and thus a broad range of melting temperatures. 31,32 Upon heating, the least crystalline domains are the first to melt or dissolve, while the most crystalline domains are the last to survive. These surviving nuclei serve as initiators to direct the crystal growth upon cooling.32 Our initial examples involved annealing solutions of polyferrocenyldimethylsilane (PFS)containing BCP micelle fragments (length: ~50 nm) above an onset temperature, followed by cooling to room temperature.29,30 Uniform rod-like micelles were formed, of which the length was sensitive to the annealing temperature (T_a) . This strategy has also been extended to other BCPs with other crystallizable blocks, such as polycaprolactone, 21,33 poly(3-

^aDepartment of Chemistry, University of Toronto, 80 St. George Street, Toronto, ON, M5S 3H6, Canada. E-mail: mwinnik@chem.utoronto.ca

^bSchool of Chemistry, University of Bristol, Bristol, BS8 1TS, UK

Key Laboratory of Material Chemistry for Energy Conversion and Storage, Ministry of Education, School of Chemistry and Chemical Engineering, Huazhong University of Science and Technology (HUST), Wuhan 430074, China

[†] Electronic supplementary information (ESI) available: Experimental details, self-seeding kinetics, TEM images of stained micelles, additional figures of the patchy and block comicelles (Fig. S1–S16), and fraction of surviving fragments at different temperatures (Table S1). See DOI: 10.1039/c8sc04705g

hexylthiophene)¹⁴ and oligo(*p*-phenylenevinylene),³⁴ and BCPs with a liquid crystalline block.³⁵

Although self-seeding is effective for controlling the length of cylindrical micelles, it has not been used for morphology control. Control over morphology in a self-seeding experiment requires an understanding of the factors that affect both the dissolution step upon heating and the growth rate upon cooling. Here we examine the self-seeding behaviour of mixtures of micelle fragments and use kinetic studies of seeded growth to model the regrowth step. Based upon this understanding of the dissolution and regrowth steps, we designed experiments on mixtures of micelle fragments that allow us to control both the length and morphology of the micelles obtained.

Results and discussion

Edge Article

We began by examining the self-seeding behaviour of each type of micelle fragment (Fig. 1). These include three PFS-b-P2VP samples (P2VP = poly(2-vinylpyridine)): PFS_{17} -b-P2VP₁₇₀ (L_0 = 43 nm), PFS₂₅-b-P2VP₃₃₀ ($L_0 = 49$ nm), and PFS₃₅-b-P2VP₄₀₀ ($L_0 =$ 52 nm) (Chart S1 and Fig. S1 in ESI†). Typically, aliquots (0.5 mL) of a micelle fragment solution (0.02 mg mL⁻¹) in 2-propanol were heated at different temperatures for 30 min and then cooled to room temperature and aged for 1 day (see ESI†). We recently showed that self-seeding is characterized by a critical dissolution temperature (T_c) corresponding to the temperature at which the initial average micelle length (L_0) doubled upon cooling (to $2L_0$).³⁶ When the annealing temperature $T_a > T_c$, the average length of the micelles (L_n) increased sharply as a function of temperature (Fig. 1 and S2-S6†). These PFS-b-P2VP samples showed very different T_c values (Fig. 1b): for PFS₁₇-b-P2VP₁₇₀, $T_c = 35$ °C; for PFS₂₅-b-P2VP₃₃₀, $T_c = 55$ °C; and for PFS₃₅-b-P2VP₄₀₀, $T_c = 67$ °C. The difference in T_c may be attributed to the difference in PFS length, as the increase in PFS length will decrease its solubility upon heating.

Then we examined the self-seeding behaviour of PFS₂₆-b-PNIPAM₁₉₀ (PNIPAM = poly(N-isopropylacrylamide), $L_0 = 52$ nm) and PFS₂₆-b-PNIPAM₅₂₀ ($L_0 = 50$ nm) fragments. As shown in Fig. 1c, for PFS₂₆-b-PNIPAM₁₉₀, $T_c = 61$ °C; while for PFS₂₆-b-PNIPAM₅₂₀, $T_c = 65$ °C. Although the lengths of PNIPAM block

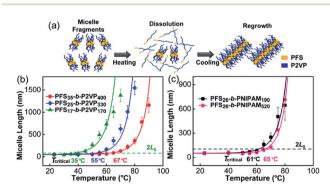


Fig. 1 Self-seeding behaviour of single kinds of micelle fragments. (a) A schematic diagram of the self-seeding process. (b) and (c) show the increase in micelle length as a function of annealing temperature. The fitted curves were calculated as described in ref. 36.

are very different, the $T_{\rm c}$ of these two fragments are similar, confirming that the length of PFS block plays a more important role in determining $T_{\rm c}$. It is worth noting that there is an upper limit of the annealing temperature for each self-seeding experiment. When the fragments were heated at a temperature above this limit, branched micelles were formed as the solution cooled (Fig. S2–S6†). The formation of branched structures under these conditions suggests a different dissolution-regrowth mechanism, a topic now under investigation.

On the basis of the above results, we designed self-seeding experiments on micelle fragment mixtures. We initially investigated the self-seeding behaviour of pairs of micelle fragments with similar T_c values, i.e., PFS₂₆-b-PNIPAM₁₉₀ ($T_c = 61$ °C) and PFS_{35} -b- $P2VP_{400}$ ($T_c = 67$ °C). After annealing this mixture at 40 °C (i.e., $T_a < T_c$), both fragments could be clearly distinguished in TEM images (Fig. 2a and S7a†), as the PFS35-b-P2VP₄₀₀ micelles look darker and thicker than PFS₂₆-b-PNIPAM₁₉₀ micelles. After annealing at 60 °C, some tadpole-like micelles could be observed (Fig. 2b and S7b†) with a thicker PFS₃₅-b-P2VP₄₀₀ head and a thin PFS₂₆-b-PNIPAM₁₉₀ tail. When $T_{\rm a} = 70$ °C, patchy micelles were obtained, characterized by local microphase separation between P2VP and PNIPAM chains (Fig. 2c and S7c \dagger). Further increase in T_a resulted in longer patchy comicelles (Fig. 2d-f, S7d and e†). Therefore, the length of the patchy comicelles could be controlled by T_a in the range of 70 to 90 °C (Fig. 2f and S8†).

Since the $T_{\rm c}$ values of PFS₂₆-b-PNIPAM₁₉₀ and PFS₃₅-b-P2VP₄₀₀ fragments are similar, about half of both fragments

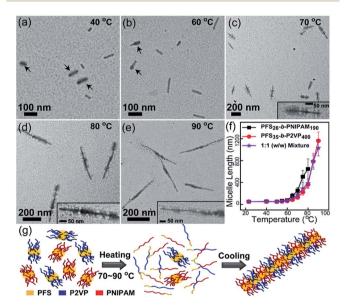


Fig. 2 Self-seeding behaviour of a PFS₂₆-b-PNIPAM₁₉₀ and PFS₃₅-b-P2VP₄₀₀ micelle fragment mixture (mass ratio 1 : 1). (a–e) TEM images of the micelles obtained by annealing the mixture at 40 to 90 °C. Arrows in (a) indicate PFS₃₅-b-P2VP₄₀₀ fragments. Arrows in (b) indicate tadpole-like comicelles. The insets in (c–e) show magnified TEM images. (f) shows the change in comicelle length as a function of T_a . (g) is a schematic diagram of the self-seeding process of the mixture. These structures are further confirmed by selectively staining the P2VP chains with PtNPs (see Fig. S7†).

Chemical Science

simultaneously dissolve into unimers upon heating to $T_{\rm c}$, while the rest survive (Fig. S9 and Table S1†). Upon cooling to room temperature, both unimers add competitively to the surviving fragments. In this step, the growth rates of unimers dominate the morphology of the comicelles. In our previous work, we demonstrated that similar epitaxial growth rates resulted in the formation of patchy comicelles, while significantly dissimilar growth rates led to block comicelles. We also showed that the

self-seeding process for this mixture, both unimers, which come from the dissolution of micelle fragments, add to the surviving seeds at similar rates, resulting in the formation of patchy comicelles (Fig. 2c–e). At higher $T_{\rm a}$, more micelle fragments dissolved, leading to fewer seeds and more unimers, and thus longer patchy comicelles.

epitaxial growth rates of PFS26-b-PNIPAM190 and PFS35-b-

P2VP₄₀₀ were similar.²⁸ Therefore, in the cooling stage of the

We then examined the self-seeding behaviour of fragments having similar $T_{\rm c}$ values but very different unimer epitaxial growth rates (Fig. 3). PFS₂₆-b-PNIPAM₅₂₀ ($T_{\rm c}=65\,^{\circ}{\rm C}$) and PFS₃₅-b-P2VP₄₀₀ ($T_{\rm c}=67\,^{\circ}{\rm C}$) micelle fragments have similar $T_{\rm c}$ values, but the growth rate of PFS₂₆-b-PNIPAM₅₂₀ unimer is much slower than that of PFS₃₅-b-P2VP₄₀₀ unimer.²⁸ After annealing the solution at 60 $^{\circ}{\rm C}$, both micelle fragments could still be distinguished, and their average length ($L_{\rm n}=56$ nm) did not change significantly (Fig. S10a†). However, when the mixture was annealed at 70 to 90 $^{\circ}{\rm C}$, comicelles with a thick middle block flanked by thin tips could be observed (70 $^{\circ}{\rm C}$: $L_{\rm n}=81$ nm, Fig. S10b;† 80 $^{\circ}{\rm C}$: $L_{\rm n}=210$ nm, Fig. 3a; 90 $^{\circ}{\rm C}$: $L_{\rm n}=731$ nm, Fig. 3b). After selectively staining the P2VP chains with platinum nanoparticles (PtNPs), symmetric triblock comicelles with

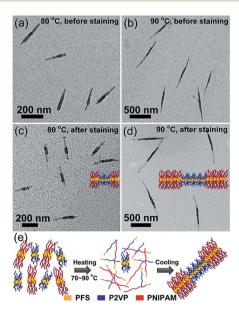


Fig. 3 Self-seeding behaviour of PFS $_{26}$ -b-PNIPAM $_{520}$ and PFS $_{35}$ -b-P2VP $_{400}$ fragment mixtures (mass ratio 1 : 1). (a–d) TEM images of the micelles obtained by annealing the mixture for 30 min at (a and c) 80 °C, and (b and d) 90 °C. (a) and (b) show the comicelles before staining, while (c) and (d) show the comicelles after selectively staining the P2VP block with PtNPs. (e) Schematic diagram of the self-seeding process of the mixture.

 PFS_{35} -b- $P2VP_{400}$ as the middle block and PFS_{26} -b- $PNIPAM_{520}$ as the end blocks could be clearly distinguished (Fig. 3c–d and S10f–h†).

Upon heating above $T_{\rm c}$, both PFS $_{35}$ -b-P2VP $_{400}$ and PFS $_{26}$ -b-PNIPAM $_{520}$ fragments dissolved into unimers, which then regrew onto the surviving seeds upon cooling. Since the epitaxial growth rate of PFS $_{26}$ -b-PNIPAM $_{520}$ is much slower than that of PFS $_{35}$ -b-P2VP $_{400}$, 28 most of PFS $_{35}$ -b-P2VP $_{400}$ added onto the seed micelles before PFS $_{26}$ -b-PNIPAM $_{520}$. After consumption of PFS $_{35}$ -b-P2VP $_{400}$, the growth of PFS $_{26}$ -b-PNIPAM $_{520}$ dominated the growth process, resulting in symmetric BAB block comicelles with PFS $_{35}$ -b-P2VP $_{400}$ as the A blocks and PFS $_{26}$ -b-PNIPAM $_{520}$ as the B blocks.

Then we carried out self-seeding experiments with mixed fragments having very different T_c , i.e., PFS₁₇-b-P2VP₁₇₀ ($L_0 =$ 43 nm, $T_{\rm c}=35$ °C) and PFS₂₆-b-PNIPAM₁₉₀ fragments ($L_0=$ 52 nm, $T_c = 61$ °C). These two BCP unimers have similar epitaxial growth rates (Fig. S11–S13 \dagger). The large difference in T_c values ($\Delta T_c = 26$ °C) makes it possible to manipulate the dissolution sequence of fragments by carefully controlling T_a . Fig. 4 shows the morphology evolution of comicelles as T_a was increased. No changes could be observed at 30 °C (Fig. 4a). When $T_a = 40$ °C, there are a small fraction of matchstick-like comicelles with a PFS₁₇-b-P2VP₁₇₀ head (~28 nm) and a PFS₂₆b-PNIPAM₁₉₀ stick (~55 nm) (Fig. 4b), along with PFS₁₇-b-P2VP₁₇₀ and PFS₂₆-b-PNIPAM₁₉₀ homomicelles. After annealing at 50 and 60 $^{\circ}$ C, ABA triblock comicelles were formed with PFS₂₆-b-PNIPAM₁₉₀ as the central B block, flanked by PFS₁₇-b-P2VP₁₇₀ A blocks (Fig. 4c and d). Interestingly, further increasing T_a to 70–80 °C led to patchy comicelles rather than block comicelles (Fig. 4e and f).

Due to the large difference in T_c value, the PFS₁₇-b-P2VP₁₇₀ fragments begin to dissolve at lower temperatures. For example at 40 °C, only 57% of PFS₁₇-b-P2VP₁₇₀ fragments survive, but 96% of PFS₂₆-b-PNIPAM₁₉₀ fragments survive (Fig. S9† and Table S1†). Upon cooling, the PFS₁₇-b-P2VP₁₇₀ unimer adds to both PFS₁₇-b-P2VP₁₇₀ and PFS₂₆-b-PNIPAM₁₉₀ fragments, leading to the formation of longer PFS₁₇-b-P2VP₁₇₀ homomicelles (~79 nm) and heterogeneous PFS₁₇-b-P2VP₁₇₀/PFS₂₆-b-PNIPAM₁₉₀ matchstick-like comicelles (Fig. 4b). Although the length of PFS₁₇-b-P2VP₁₇₀ homomicelles increased from 43 nm to 79 nm, the length of PFS₂₆-b-PNIPAM₁₉₀ micelles did not change (\sim 54 nm, Fig. S14†). For $T_a = 50$ °C, 96% of PFS₂₆-b-PNIPAM₁₉₀ fragments survive, but only 16% of PFS₁₇-b-P2VP₁₇₀ fragments persist; whereas at 60 °C, 66% of PFS₂₆-b-PNIPAM₁₉₀ fragments survive, but only 6% of PFS₁₇-b-P2VP₁₇₀ fragments remain (Fig. S9† and Table S1†). Therefore, most PFS₁₇-b-P2VP₁₇₀ fragments dissolve to give unimers at 50–60 °C, which then add to the surviving PFS₂₆-b-PNIPAM₁₉₀ fragments upon cooling, resulting in the formation of block comicelles (Fig. 4g). At higher T_a , both PFS₁₇-b-P2VP₁₇₀ and PFS₂₆-b-PNIPAM₁₉₀ fragments dissolve to form unimers. For example at 80 °C, all of the PFS₁₇-b-P2VP₁₇₀ fragments dissolve while 8% of PFS₂₆-b-PNIPAM₁₉₀ fragments survive. After cooling to room temperature, both unimers, which have similar growth rates, simultaneously add to the surviving PFS26-b-PNIPAM190 seeds, resulting in the formation of patchy comicelles (Fig. 4g).

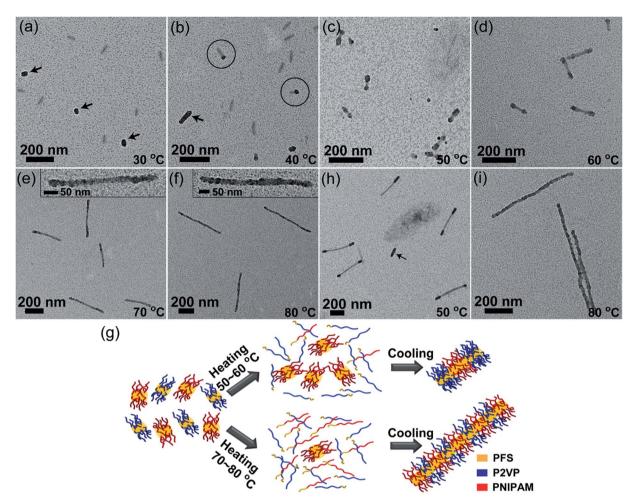


Fig. 4 Self-seeding of a PFS_{26} -b-PNIPAM₁₉₀ and PFS_{17} -b-P2VP₁₇₀ fragment mixture (mass ratio 1 : 1). (a–f) TEM images of the micelles obtained by annealing the mixture at 30 to 80 °C. The arrows in (a) indicate the PFS_{17} -b-P2VP₁₇₀ fragments. The arrow in (b) indicates a PFS_{17} -b-P2VP₁₇₀ homomicelle. The circles in (b) indicate matchstick-like micelles. The insets in (e and f) show magnified TEM images of patchy comicelles. (g) Schematic diagram of the self-seeding process of the fragment mixture. (h and i) TEM images of the micelles obtained by annealing a mixture of PFS_{26} -b-PNIPAM₁₉₀ micelles (L_n = 290 nm) and PFS_{17} -b-P2VP₁₇₀ fragments (L_0 = 43 nm) at (h) 50 °C and (i) 80 °C. The arrow in (h) indicates a PFS_{17} -b-P2VP₁₇₀ homomicelle. The P2VP chains were selectively stained with PtNPs.

In order to make the block comicelles shown in Fig. 4c and d more distinguishable, we conducted a self-seeding experiment at 50 °C with a 1 : 1 mixture of PFS_{26} -b-PNIPAM₁₉₀ micelles ($L_n = 290$ nm, Fig. S15†) and PFS_{17} -b-P2VP₁₇₀ fragments ($L_0 = 43$ nm). As shown in Fig. 4h and S16,† ABA triblock comicelles can be clearly visualized, accompanied by a few PFS_{17} -b-P2VP₁₇₀ homomicelles formed by epitaxial growth on surviving PFS_{17} -b-P2VP₁₇₀ seeds. Upon heating to 80 °C, most of the long and short micelles dissolved. The competitive growth of both unimers on the surviving seeds led to the formation of patchy comicelles (Fig. 4i).

Conclusions

In summary, we found that the morphology of core-crystalline cylindrical micelles could be controlled by manipulating the synergistic self-seeding behaviour of micelle fragment mixtures. The morphology of the comicelles is dependent on (1) the dissolution sequence of micelle fragments, and (2) the epitaxial growth rate of the corresponding unimers. By blending two

kinds of micelle fragments with similar $T_{\rm c}$ values and similar growth rates, patchy micelles with controllable length can be obtained by self-seeding. If the fragments have similar $T_{\rm c}$ values but very different growth rates, block comicelles are obtained. On the other hand, if micelle fragments with very different $T_{\rm c}$ are blended, the morphology of comicelles can be controlled by varying the annealing temperature. At lower temperatures, block comicelles are obtained; whereas at higher temperatures, patchy comicelles are formed. Thus, by carefully controlling the dissolution and growth sequence of micelle fragments, both the length and morphology of micelles can be varied. This synergistic self-seeding strategy not only offers a convenient route for preparing comicelles with controllable morphology and uniform length, but also provides a platform for studying the synergistic self-seeding behaviour of polymer seed crystallites.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

Chemical Science

The Toronto authors thank NSERC Canada for their support of this research.

Notes and references

- 1 Y. Geng, P. Dalhaimer, S. Cai, R. Tsai, M. Tewari, T. Minko and D. E. Discher, *Nat. Nanotechnol.*, 2007, 2, 249–255.
- 2 Z. Deng, S. Yuan, R. X. Xu, H. Liang and S. Liu, *Angew. Chem.*, Int. Ed., 2018, 57, 8896–8900.
- 3 X. Hu, S. Zhai, G. Liu, D. Xing, H. Liang and S. Liu, *Adv. Mater.*, 2018, **30**, 1706307.
- 4 J. M. Dean, N. E. Verghese, H. Q. Pham and F. S. Bates, *Macromolecules*, 2003, 36, 9267–9270.
- 5 J. Yuan and A. H. Müller, *Polymer*, 2010, **51**, 4015–4036.
- 6 D. J. Pochan, J. Zhu, K. Zhang, K. L. Wooley, C. Miesch and T. Emrick, *Soft Matter*, 2011, 7, 2500–2506.
- 7 J. Dupont, G. Liu, K. i. Niihara, R. Kimoto and H. Jinnai, *Angew. Chem., Int. Ed.*, 2009, **48**, 6144–6147.
- 8 H. Cui, Z. Chen, S. Zhong, K. L. Wooley and D. J. Pochan, *Science*, 2007, 317, 647–650.
- 9 H. Gröschel, A. Walther, T. I. Löbling, F. H. Schacher, H. Schmalz and A. H. Müller, *Nature*, 2013, **503**, 247–251.
- 10 A. M. Oliver, J. Gwyther, M. A. Winnik and I. Manners, *Macromolecules*, 2018, 51, 222-231.
- 11 D. J. Lunn, J. R. Finnegan and I. Manners, *Chem. Sci.*, 2015, **6**, 3663–3673.
- 12 Y. Mai and A. Eisenberg, *Chem. Soc. Rev.*, 2012, 41, 5969-5985.
- 13 N. Petzetakis, A. P. Dove and R. K. O'Reilly, *Chem. Sci.*, 2011, 2, 955–960.
- 14 J. Qian, X. Li, D. J. Lunn, J. Gwyther, Z. M. Hudson, E. Kynaston, P. A. Rupar, M. A. Winnik and I. Manners, *J. Am. Chem. Soc.*, 2014, 136, 4121–4124.
- 15 B. Fan, R.-Y. Wang, X.-Y. Wang, J.-T. Xu, B.-Y. Du and Z.-Q. Fan, *Macromolecules*, 2017, **50**, 2006–2015.
- 16 J. B. Gilroy, T. Gädt, G. R. Whittell, L. Chabanne, J. M. Mitchels, R. M. Richardson, M. A. Winnik and I. Manners, *Nat. Chem.*, 2010, 2, 566–570.
- 17 W. N. He and J. T. Xu, Prog. Polym. Sci., 2012, 37, 1350-1400.
- 18 H. Qiu, Y. Gao, C. E. Boott, O. E. Gould, R. L. Harniman, M. J. Miles, S. E. Webb, M. A. Winnik and I. Manners, *Science*, 2016, 352, 697–701.

- 19 X. Wang, G. Guerin, H. Wang, Y. Wang, I. Manners and M. A. Winnik, *Science*, 2007, 317, 644-647.
- 20 J. Schmelz, A. E. Schedl, C. Steinlein, I. Manners and H. Schmalz, *J. Am. Chem. Soc.*, 2012, **134**, 14217–14225.
- 21 M. C. Arno, M. Inam, Z. Coe, G. Cambridge, L. J. Macdougall, R. Keogh, A. P. Dove and R. K. O'Reilly, *J. Am. Chem. Soc.*, 2017, 139, 16980–16985.
- 22 D. Tao, C. Feng, Y. Cui, X. Yang, I. Manners, M. A. Winnik and X. Huang, *J. Am. Chem. Soc.*, 2017, **139**, 7136–7139.
- 23 A. Nazemi, X. He, L. R. MacFarlane, R. L. Harniman, M.-S. Hsiao, M. A. Winnik, C. F. Faul and I. Manners, *J. Am. Chem. Soc.*, 2017, 139, 4409–4417.
- 24 M. Inam, G. Cambridge, A. Pitto-Barry, Z. P. Laker, N. R. Wilson, R. T. Mathers, A. P. Dove and R. K. O'Reilly, *Chem. Sci.*, 2017, 8, 4223–4230.
- 25 H. Qiu, Y. Gao, V. A. Du, R. Harniman, M. A. Winnik and I. Manners, J. Am. Chem. Soc., 2015, 137, 2375–2385.
- 26 G. Cambridge, G. Guerin, I. Manners and M. A. Winnik, Macromol. Rapid Commun., 2010, 31, 934–938.
- 27 J. R. Finnegan, D. J. Lunn, O. E. C. Gould, Z. M. Hudson, G. R. Whittell, M. A. Winnik and I. Manners, *J. Am. Chem. Soc.*, 2014, **136**, 13835–13844.
- 28 J. Xu, H. Zhou, Q. Yu, I. Manners and M. A. Winnik, *J. Am. Chem. Soc.*, 2018, **140**, 2619–2628.
- 29 J. Qian, Y. Lu, A. Chia, M. Zhang, P. A. Rupar, N. Gunari, G. C. Walker, G. Cambridge, F. He, G. Guerin, I. Manners and M. A. Winnik, *ACS Nano*, 2013, 7, 3754–3766.
- 30 J. S. Qian, G. Guerin, Y. J. Lu, G. Cambridge, I. Manners and M. A. Winnik, *Angew. Chem., Int. Ed.*, 2011, 50, 1622–1625.
- 31 D. Blundell, A. Keller and A. Kovacs, *J. Polym. Sci., Part B: Polym. Lett.*, 1966, 4, 481–486.
- 32 J. Xu, Y. Ma, W. Hu, M. Rehahn and G. Reiter, *Nat. Mater.*, 2009, **8**, 348–353.
- 33 W. N. He, B. Zhou, J.-T. Xu, B.-Y. Du and Z.-Q. Fan, *Macromolecules*, 2012, **45**, 9768–9778.
- 34 D. Tao, C. Feng, Y. Lu, Y. Cui, X. Yang, I. Manners, M. A. Winnik and X. Huang, *Macromolecules*, 2018, 51, 2065–2075.
- 35 X. Li, B. Jin, Y. Gao, D. W. Hayward, M. A. Winnik, Y. Luo and I. Manners, *Angew. Chem., Int. Ed.*, 2016, 55, 11392–11396.
- 36 G. Guerin, P. A. Rupar, I. Manners and M. A. Winnik, *Nat. Commun.*, 2018, **9**, 1158.