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PEGylated surfaces for the study of DNA–protein interactions by atomic force microscopy†

 Bernice Akpinar,^{a,b} Philip J. Haynes,^{a,b} Nicholas A. W. Bell,^c Katharina Brunner,^{c,d}
 Alice L. B. Pyne ^{*a,e} and Bart W. Hoogenboom ^{*a,f}

DNA–protein interactions are vital to cellular function, with key roles in the regulation of gene expression and genome maintenance. Atomic force microscopy (AFM) offers the ability to visualize DNA–protein interactions at nanometre resolution in near-physiological buffers, but it requires that the DNA be adhered to the surface of a solid substrate. This presents a problem when working in biologically relevant protein concentrations, where proteins may be present in large excess in solution; much of the biophysically relevant information can therefore be occluded by non-specific protein binding to the underlying substrate. Here we explore the use of PLL-*b*-PEG_y block copolymers to achieve selective adsorption of DNA on a mica surface for AFM studies. Through varying both the number of lysine and ethylene glycol residues in the block copolymers, we show selective adsorption of DNA on mica that is functionalized with a PLL₁₀-*b*-PEG₁₁₃/PLL₁₀₀₀–2000 mixture as viewed by AFM imaging in a solution containing high concentrations of streptavidin. We show – through the use of biotinylated DNA and streptavidin – that this selective adsorption extends to DNA–protein complexes and that DNA-bound streptavidin can be unambiguously distinguished in spite of an excess of unbound streptavidin in solution. Finally, we apply this to the nuclear enzyme PARP1, resolving the binding of individual PARP1 molecules to DNA by in-liquid AFM.

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Introduction

Interactions between DNA and proteins regulate a number of processes crucial to cellular function that include transcription, chromosome maintenance, DNA replication and repair. DNA-binding proteins employ a range of different mechanisms to interact with both select and non-select sites on DNA.¹ Key mechanistic insights have been revealed using biophysical techniques such as fluorescence microscopy,^{2–4} optical tweezers,^{5,6} surface plasmon resonance,⁷ and atomic force microscopy (AFM).^{8,9}

AFM has been established as a powerful single-molecule technique to probe DNA–protein interactions, due to its ability

to directly image DNA at nanometre resolution under physiologically relevant conditions without the need for labelling.^{10–12} However, to obtain high-resolution images of biomolecules in liquids, the sample must be adhered to an underlying solid support. Muscovite mica is the substrate of choice for AFM imaging of DNA, due to the ease of preparing an atomically flat mica surface *via* cleavage along the basal plane, and due to the polar, hydrophilic nature of the cleaved surface, facilitating the adsorption and retention of biomolecules in aqueous solution. When mica is hydrated, K⁺ ions dissociate from interstitial sites within mica's aluminium phyllosilicate lattice, resulting in a net negative charge on the surface. To permit the adsorption of DNA onto its highly negatively charged phosphate backbone, the negative surface charge needs to be screened or compensated for.^{13,14} Many adsorption protocols have been established for the 2D confinement of DNA to pre-treated mica,¹⁵ one of the most commonly adopted being the use of transition metal cations such as Ni²⁺, Co²⁺ and Zn²⁺ that can substitute into vacant sites within the mica lattice, yielding positively charged patches for the adsorption of DNA.¹⁶ The strength of the electrostatic attraction can be modified by the presence of additional ions and chelating agents within the imaging buffer.^{17,18} Other methods to facilitate DNA adsorption include the modification of surface chemistry using silanes,¹⁵ the formation of partially positively charged lipid bilayers^{9,19} and the electrostatic

^aLondon Centre for Nanotechnology, University College London, 17-19 Gordon Street, London WC1H 0AH, UK. E-mail: a.l.pyne@sheffield.ac.uk, b.hoogenboom@ucl.ac.uk

^bDepartment of Chemistry, Imperial College London, SW7 2AZ, UK

^cThe Francis Crick Institute, 1 Midland Road, London, NW1 1AT, UK

^dDiscovery Biology, Discovery Sciences, R&D, AstraZeneca, 50F49, Mereside, Alderley Park, Macclesfield, Cheshire, SK10 4TG, UK

^eDepartment of Materials Science and Engineering, University of Sheffield, S1 3JD, UK

^fDepartment of Physics and Astronomy, University College London, Gower Street, London WC1E 6BT, UK

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adsorption of positively charged polymers such as poly-L-lysine (PLL).^{20,21}

The aforementioned approaches are often adopted for the study of DNA–protein binding using AFM. However, they can result in non-specific protein–surface interactions, which are non-trivial to deconvolute from specific DNA–protein interactions. The problem of non-specific adsorption can be addressed by the use of surface coatings that are protein repellent, for example polymer brushes that suppress protein binding by steric repulsion.²² One approach to suppress protein binding is to create an interfacial layer of polyethylene glycol (PEG) brushes. The high degree of hydration and flexibility of these brushes causes surface passivation when the chains are of sufficient length and grafted at high density.²³ Facile preparation of PEGylated surfaces is achieved using multifunctional copolymers comprising both surface binding domains and surface passivating PEG domains. Graft-copolymers with a cationic PLL backbone and PEG side chains (PLL-*g*-PEG) have proven particularly effective at self-assembling into densely packed polymeric brushes to form non-fouling surfaces.^{23–28} In addition, bio-recognition sites, such as RGD-peptides, have been incorporated into these films to promote cell adhesion whilst suppressing the non-specific adsorption of serum proteins.^{29,30} Similarly, the incorporation of biotin-terminated PEG chains has been used to form small molecule biosensors that selectively bind streptavidin, neutravidin and avidin.³¹ Unmodified PLL-*g*-PEG films have also shown the ability to selectively adsorb DNA polyelectrolytes onto the underlying positively charged PLL layer, whilst the PEG layer remains impervious to other proteins, as confirmed by fluorescence imaging.²⁸

The well-studied graft copolymer (PLL-*g*-PEG) adopts a comb-like conformation in solution comprising a long PLL backbone with randomly distributed PEG side chains, whilst the block copolymer (PLL-*b*-PEG) exhibits a linear worm-like conformation comprising regions of lysine repeats followed by regions of ethylene glycol repeats. Both copolymers can form protein repellent PEG brushes on a variety of substrates through the spontaneous electrostatic attachment of their lysine residues. In the case of the graft copolymer, the length of the PEG block and the grafting ratio affect the density and hence the efficacy of the anti-fouling brushes.²⁶ The diblock copolymer has been less widely employed for surface passivation but has been shown to be effective at inhibiting cell adhesion on glass surfaces micro-patterned with PLL₁₀₀-*b*-PEG₂₂.³² The passivation properties of the diblock copolymer (PLL_{*x*}-*b*-PEG_{*y*}) can be tuned by varying the degree of polymerization of both the PLL (*x*) and PEG (*y*) chains which would affect the packing density onto the underlying substrate, although as of yet variations of these have not been explored. We here set out to determine whether linear PLL_{*x*}-*b*-PEG_{*y*} diblock copolymers can be used in the functionalization of mica to yield a surface that selectively adsorbs DNA and allows the characterization of DNA–protein complexes by AFM. Through the optimization of the composition of the diblock copolymer, we have developed biphasic films which promote the adsorption of negatively charged DNA,

whilst passivating against non-specific protein adsorption. Specifically, we perform mica surface functionalization that allows high-resolution AFM imaging of DNA and of DNA–protein complexes in solution whilst resisting non-specific protein adsorption.

Methods

Materials

Relaxed plasmid pBR322 DNA was purchased from Inspiralis Ltd. For AFM studies of streptavidin binding, a 672 bp length of DNA was prepared by PCR amplification of a section of lambda DNA (New England Biolabs) using a forward primer 5'-CGATGTGGTCTCACAGTTTGTAGTCTGGTCTCG-3' and reverse primer 5'-GGAAGAGGTCTCTTAGCGGTACGTTTCCGTC-3' purchased from Integrated DNA Technologies. Each primer was labelled at its 5' end with a single biotin thereby resulting in a double-stranded DNA PCR product labelled at both ends with biotin. The PCR product was purified using a QIAquick PCR purification kit (Qiagen). For AFM studies of PARP1 binding, a 496 bp section of DNA was prepared using PCR amplification of a section of lambda DNA with forward primer 5'-TGAAATTGCCGCGTATTACGC-3' and reverse primer 5'-TTTCTCGTAGGTACTCAGTCCG-3'. The PCR product was digested with Nt.BsmAI (New England Biolabs) according to the manufacturer's protocol. This produces a single nick that is located at 172 bp from one end of the DNA, *i.e.*, at ~1/3 of the DNA length. The DNA was then purified using a QIAquick purification kit.

Monovalent streptavidin was produced by the Howarth lab.³³ Block copolymers methoxy-poly(ethylene glycol)-block-poly(L-lysine hydrochloride) with varying degrees of polymerization of the poly-L-lysine and polyethylene glycol blocks were purchased as lyophilized powders from Alamanda Polymers. The polymers used for this study were PLL₁₀-PEG₂₂, PLL₁₀-PEG₁₁₃, PLL₁₀₀-PEG₁₁₃ and PLL₁₀-PEG₄₅₄ where the subscript refers to the degree of polymerization, *i.e.*, the number of monomer repeats. ESI Table 1† details the corresponding molecular weights for each of the polymers used. A 0.01% w/v solution of poly-L-lysine (PLL_{1000–2000}) with approximately one HBr molecule per lysine residue, along with all other reagents, were purchased from Sigma-Aldrich.

Agarose gel electrophoresis

Biotinylated DNA binding to mono and tetravalent streptavidin was verified by AGE (1% agarose, 1 × TBE) using a BioRad Wide Mini-Sub Cell GT electrophoresis system. 5 µl of pre-incubated samples were mixed with 1 µl of 6× loading buffer before loading onto the agarose gel. The samples were allowed to migrate for 90 minutes (running buffer: 1× TBE; 90 V). The gel was stained for 40 minutes in a solution 3× GelRed (Biotium) and visualized using UV light.

Mica modification and DNA deposition

For the preparation of copolymer films, freshly cleaved mica discs (diameter: 5 mm) were covered in 10 µl of solution com-



prising only PLL-*x*-*b*-PEG_y (1 mg ml⁻¹ in MilliQ water) or a mixture of 5 μl of PLL-*x*-*b*-PEG_y solution and 5 μl PLL_{1000–2000} (0.01% w/v). Mica discs were incubated with these solutions for 45 minutes in a humid environment, before washing 5 times with double-deionised water (Milli-Q, Merck Millipore) and 5 times with imaging buffer (10 mM phosphate buffer pH 7.4). 5 μl of DNA plasmid (~1.5 ng μl⁻¹) or 3 μl biotinylated DNA (~3.5 nM) was immediately added to the disc and allowed to equilibrate for approximately 10 minutes prior to imaging. A similar protocol was followed for functionalization with PLL alone but the PLL incubation time was reduced to 1 minute before thoroughly rinsing to minimize surface contamination. A solution of PLL, either 0.01% (Fig. 3(a)) or 0.001% (Fig. 2) was used to form full or partial monolayers, onto which DNA could be adsorbed.

PARP1 binding studies

Recombinant PARP1 carrying an N-terminal hexahistidine AviTag was produced using a pFastBac vector based Baculo virus expression system for expression in Sf21 insect cells. The cells were harvested by centrifugation and pellets were solubilized in binding buffer (25 mM Tris-HCl, pH 7.4, 150 mM NaCl, 1 mM TCEP, 10 mM imidazole) prior to sonication in the presence of DNase I (Sigma D4527) and EDTA-free protease inhibitor cocktail (Roche 37378900). The lysed pellet was clarified by centrifugation, the supernatant was mixed with 3 mL of Ni-NTA resin (Thermo Scientific 88222) and incubated for 1 hour at 4 °C. Ni-NTA beads were loaded onto a gravity flow column and washed with binding buffer. PARP1 elution was achieved by employing an imidazole gradient in binding buffer. Fractions containing PARP1 were further purified by ion exchange chromatography on a HiTrap heparin HP (GE Healthcare) column followed by size exclusion chromatography on a Hi load 16/60 Superdex 200 (Amersham Biosciences) prep grade column. Purification of PARP1 was monitored by SDS-PAGE, and the fractions shown in Fig. S1† were pooled and used for further AFM experiments.

For prior adsorption of DNA on the mica, the cleaved mica was incubated with in total 20 μl of the 1 : 1 PLL₁₀-*b*-PEG₁₁₃/PLL_{1000–2000} mixture and left to incubate in a humidified Petri dish for 45 minutes. This was then washed 5 times with double-deionised water and a further 5 times with imaging buffer (12.5 mM NaCl, 12.5 mM HEPES, 0.5 mM TCEP, pH 7.8, filtered by passage through a 0.2 μm syringe followed by a 10 kDa cutoff centrifugal filter (Amicon Ultra, Millipore)). 20 μL of 496 bp linear DNA with an ss break (1.5 ng μl⁻¹, 7.8 nM) was then added to the disk and gently mixed. After a 30 minute adsorption, the sample was then washed 5 times and made up to 30 μL with imaging buffer.

For adsorption using PLL, 10 μL of PLL (0.01%) was added to a freshly cleaved mica surface and left to incubate for 1 minute. The disk was then held at an angle and thoroughly rinsed under a stream of double-deionised water. The disk was then blotted and 20 μL 494 bp linear DNA with an ss break (0.3 ng μl⁻¹, 1.6 nM) was added. After incubating for 10 minutes, the sample was washed 5 times with imaging

buffer. For adsorption using NiCl₂, the freshly cleaved mica was incubated for 20 minutes with the DNA sample (0.3 ng μl⁻¹, 1.6 nM) and 4 mM NiCl₂. The sample was then washed 5 times with imaging buffer (the same as above but containing 3 mM NiCl₂). For the PARP exchange assays, a buffer exchange for PARP1 in imaging buffer was carried out. Imaging was resumed after 5 minutes of incubation.

AFM imaging

All AFM imaging was carried out at room temperature with the samples hydrated in imaging buffer. Data were recorded using a Dimension FastScan Bio AFM (Bruker, Santa Barbara, USA), using force–distance–curve based imaging (PeakForce Tapping mode). Force–distance curves were recorded over 10–40 nm (PeakForce Tapping amplitude of 5–20 nm), at a frequency of 8 kHz. FastScan D (Bruker) cantilevers were used for all imaging (resonance frequency of ~110 kHz, nominal spring constant ~0.25 Nm⁻¹). Images were processed using first-order line-by-line flattening, median line-by-line flattening and zeroth order plane fitting to remove the sample tilt and background using Gwyddion.

Quantification of protein binding by AFM

To quantify the amount of background protein in the experiments using DNA plasmids, AFM images were thresholded using Gwyddion, to differentiate streptavidin molecules (~4 nm height), DNA molecules (~2 nm height), and the substrate (~0 nm height), see Fig. S2.† The height thresholds were adjusted accordingly for each image to give the best identification of streptavidin and DNA, as determined by visual inspection; Fig. S2† also shows an example of a mask fitted on an image where DNA and streptavidin are bound to DNA. For each quantification, a total area of at least 74 μm² was analyzed to give the percentage surface coverage of streptavidin as shown in Fig. 4(c). In the studies of the biotinylated DNA, individual streptavidin molecules were counted using ImageJ to obtain the number of streptavidin molecules that were not bound to the ends of DNA, with results shown in Fig. S5.†

Results and discussion

Poly-L-lysine (PLL) has been used extensively to immobilize DNA on a mica surface. This immobilization depends on an interfacial layer of positively charged long-chain (PLL) polymers that bind DNA through electrostatic attraction. However, this surface also facilitates the non-specific adsorption of proteins, thus complicating the identification of targeted DNA–protein interactions. Copolymers comprising both PLL and PEG repeats achieve reduced non-specific protein adsorption through an additional PEG component, which acts as a steric barrier to protein binding (Fig. 1). To compare, on the one hand, the efficiency of diblock PLL-*x*-*b*-PEG_y copolymers for specific DNA and DNA–protein immobilization, and, on the other hand, immobilization using traditional PLL protocols, we first characterized DNA adsorption on a mica functiona-



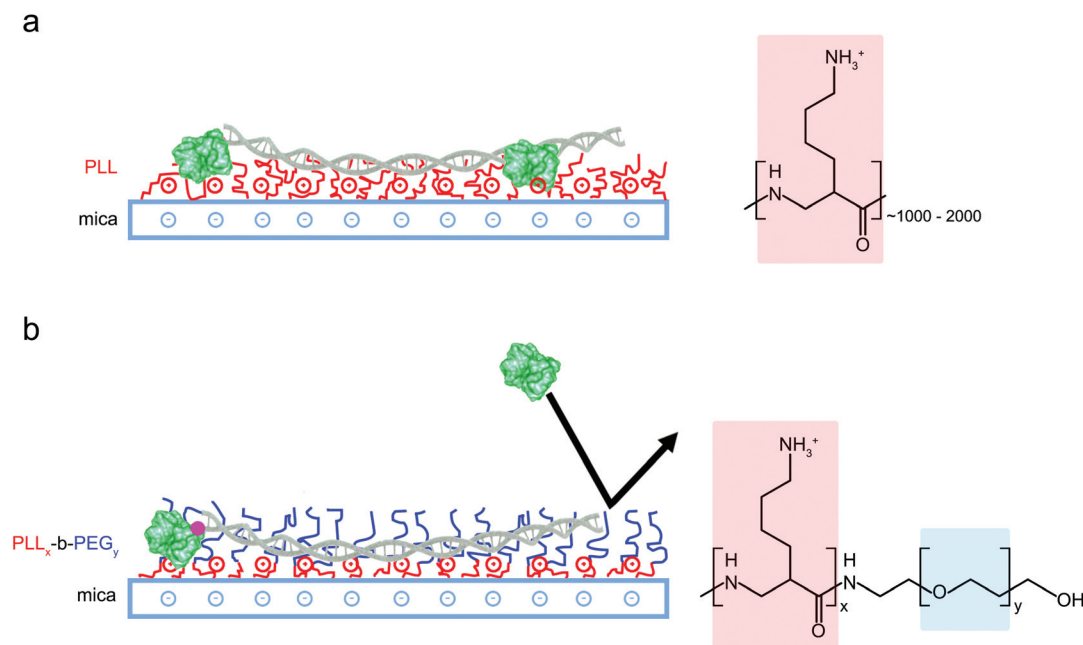


Fig. 1 Schematic representation of different DNA surface adsorption methods. (a) Adsorption of DNA and proteins is promoted by modifying mica substrates with positively charged long-chain PLL₁₀₀₀₋₂₀₀₀. (b) PLL_x-b-PEG_y block copolymers form films of densely packed PEG chains that repel proteins whilst the accessible lysine residues promote electrostatic adsorption of the highly charged DNA only.

lized with PLL only. PLL spontaneously attaches to the negatively charged mica surface *via* its highly charged lysine residues ($\text{pK}_a \sim 10.5$) to yield homopolymer films stable over a range of pH and salt concentrations.³⁴ The PLL used here was PLL₁₀₀₀₋₂₀₀₀ where the subscript refers to the number of lysine repeats in the homopolymer, corresponding to a molecular weight of 150 000–300 000 g mol^{-1} . See the ESI (Table S1†) for molecular weights of the other polymers used in this study.

PLL surface functionalization is obtained by the incubation of a cleaved mica surface in PLL solution. Deposition at low concentrations (0.001% w/v) allows relaxation of the lysine chains and adsorption in flattened, stretched out conformations where individual poly-L-lysine molecules are resolved (Fig. 2).³⁴ High resolution on the individual lysine chains was achieved in 10 mM phosphate buffer, a relatively poor solvent for poly-L-lysine, reducing the repulsion of the AFM tip which can then come into contact with the collapsed chains (Fig. 2(b)).³⁵ The PLL chains are seen to preferentially align along three orientations, with an angular difference of about 60° (Fig. 2(d)). PLL chains appear to be better resolved when aligned at larger angles with respect to the fast scan axis (left to right in these images). When aligned along the fast scan axis itself, PLL chains are more difficult to resolve as their width is approximately equal to the width of one scan line (0.5 nm) and therefore more sensitive to the precise position of consecutive lines. While the underlying atomic lattice of the mica substrate cannot be resolved in these images, the observed orientations are consistent with a molecular arrangement in which the lysine residues occupy interstitial sites on the mica lattice vacated by K^+ ions (Fig. 2(e)).³⁶

When deposited from the stock solution at high PLL concentration (0.01% w/v), the lysine chains adopt more globular forms, resulting in an apparently more homogeneous surface functionalization (Fig. 3(a), see Fig. S3† for comparison of PLL deposition at low (a) and high concentrations (b)). PLL functionalized mica enhances the adsorption of DNA, but also of other biomolecules that may be present in solution, including those of reduced charge. This is demonstrated by the immobilization of both the highly negatively charged plasmid DNA and the slightly negatively charged streptavidin ($\text{pK}_a \sim 5.0\text{--}6.0$, at pH 7.4) (Fig. 3(a)).³⁷ We show that the surface can be modified to achieve a more preferential, selective adsorption of DNA by functionalization with PLL₁₀-b-PEG₂₂ alone or by a combination of PLL₁₀-b-PEG₂₂ and long-chain PLL₁₀₀₀₋₂₀₀₀. In the presence of the block copolymer streptavidin adheres as sparse clusters (Fig. 3(b) and (c)), perhaps due to the heterogeneous surface coverage of the protein-repellent PEG. In both cases, large areas of functionalized mica are visible without the protein adsorbed. This can be explained by the effective repulsion that arises when the polyethylene-glycol chains form a sufficiently dense steric barrier.

To achieve a homogeneous surface that resists protein adsorption across the entire sample, the PEG molecules should be grafted at a density that is large enough to facilitate the overlap between different chains, resulting in the formation of a dense polymer brush.³⁸ This requires the radius of gyration R_G for the polymer to be comparable or larger than the mean distance between grafting sites. It follows that longer PEG chains are more effective in passivating a surface against protein binding, provided that they are grafted at sufficient



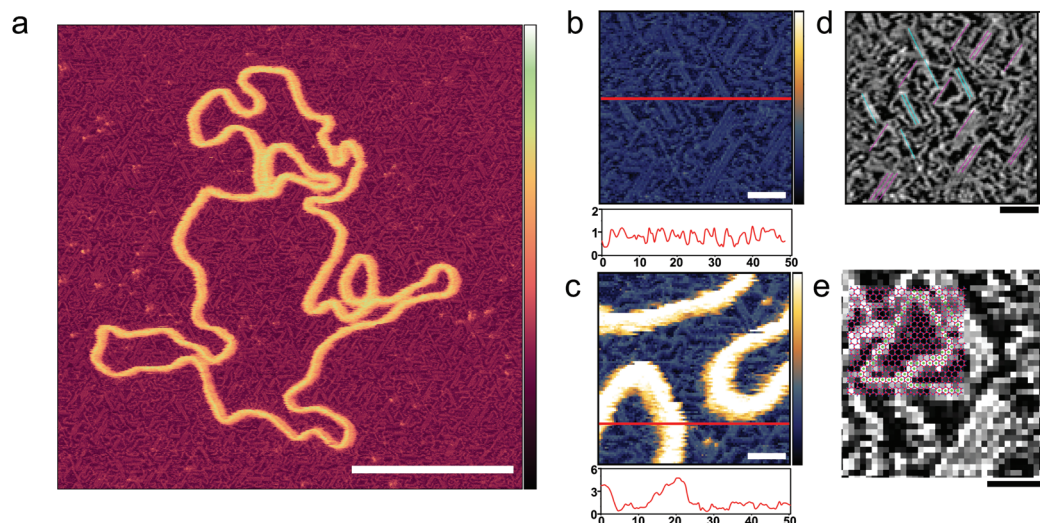


Fig. 2 Ordering of poly-L-lysine chains on a mica substrate. (a) AFM image taken in solution with tip sampling every ~ 0.5 nm showing a DNA plasmid adsorbed onto PLL₁₀₀₀₋₂₀₀₀-functionalized mica. (b and c) At higher magnification, PLL chains are unambiguously resolved. Height profiles underneath provide an indication of the respective protrusions of the PLL chains and of the DNA. (d) The axis of alignment observed in (b) is high-lighted; (e) the mica lattice geometry³⁶ is here aligned and overlaid with the resolved lysine chains, and their corresponding overlap with vacancies is observed. Inset colour scale for (a) is 8 nm and inset colour scales for (b) and (c) 4 nm, for (d) and (e) 0.8 nm. Scale bar for (a) is 100 nm, 10 nm for (b–d) and 5 nm for (e).

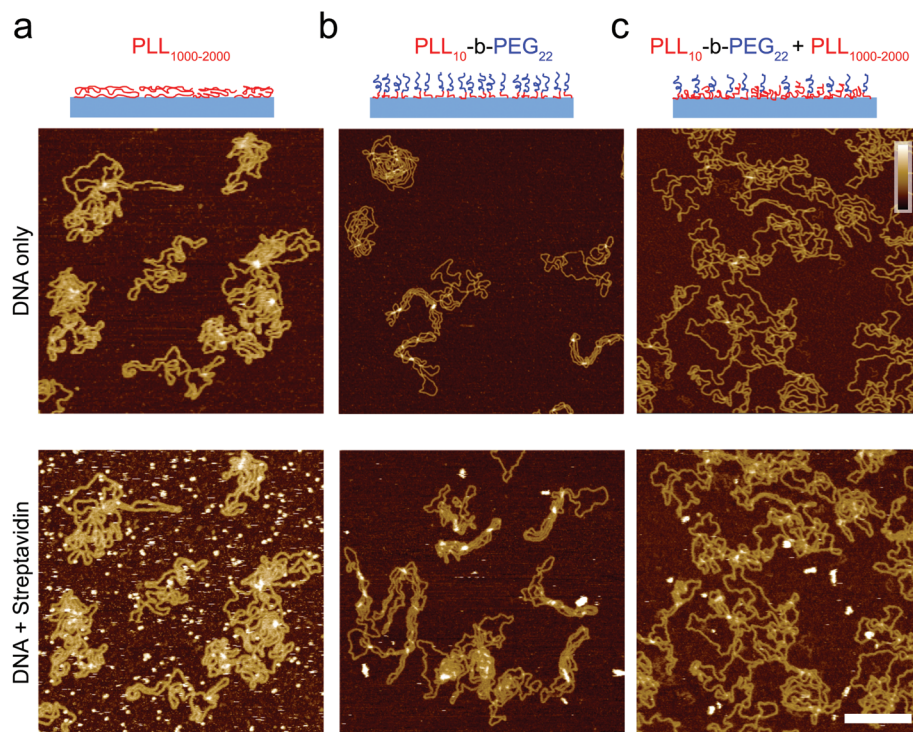


Fig. 3 Characterization of the adsorption of DNA plasmid and streptavidin on functionalized mica. Streptavidin (160 nM) was added after DNA immobilization, DNA was incubated for 10 minutes prior to imaging and streptavidin was left to equilibrate for 10 minutes prior to imaging on (a) PLL₁₀₀₀₋₂₀₀₀ only surface, (b) PLL₁₀-*b*-PEG₂₂ and (c) a mixed PLL₁₀-*b*-PEG₂₂/PLL₁₀₀₀₋₂₀₀₀ surface. Colour scale (inset in c) 10 nm; scale bar 200 nm. Images taken in solution.

densities.³⁹ By increasing the PEG block length (y) in the PLL_{*x*}-*b*-PEG_{*y*} diblock copolymer, we generated an optimized surface functionalization which resisted non-specific protein adsorp-

tion in 160 nM streptavidin (Fig. 4). PLL₁₀-*b*-PEG₁₁₃ constructs were more effective than PLL₁₀-*b*-PEG₂₂ in preventing streptavidin binding, however co-functionalization with PLL₁₀₀₀₋₂₀₀₀



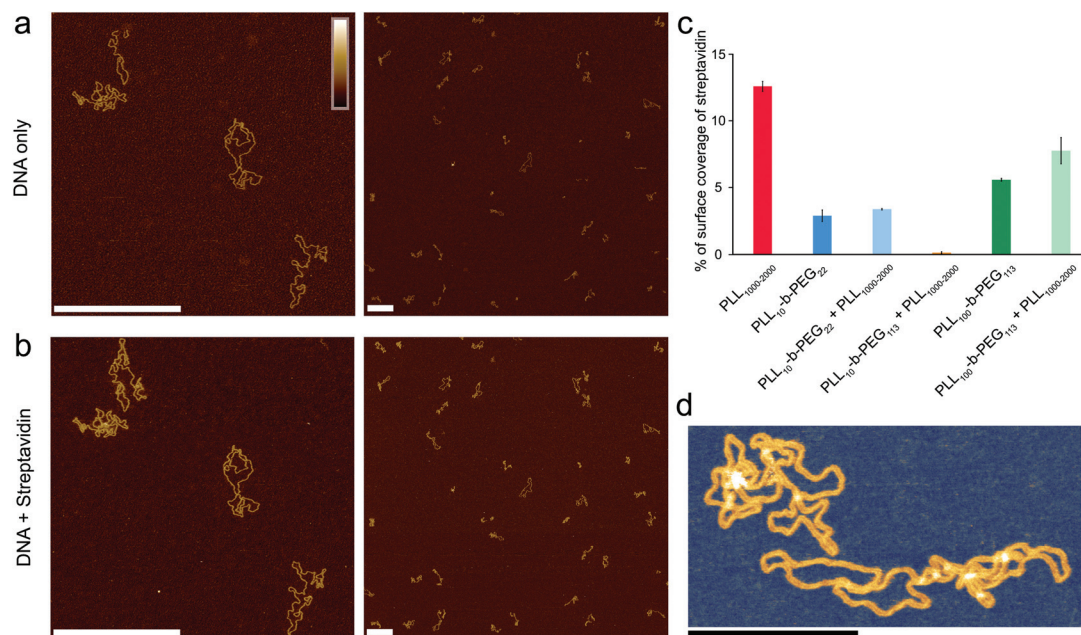


Fig. 4 Optimized poly(L-lysine)-*b*-poly(ethylene glycol) surfaces for exclusive DNA adsorption. AFM images taken in solution showing selective DNA adsorption on PLL₁₀-*b*-PEG₁₁₃/PLL_{1000–2000} surfaces. (a) DNA plasmid only. (b) The same area following the addition of 160 nM streptavidin. (c) Percentage background streptavidin coverage at 160 nM for functionalization protocols using different PEG chain lengths, and error bars correspond to the minimum and maximum values as determined from two different areas. (d) A higher resolution image of DNA on the PLL₁₀-*b*-PEG₁₁₃/PLL_{1000–2000} surface. Colour scales (see inset top left) for (a) and (b) are 7 nm, (d) with inset colour scale is 9 nm. Scale bars in (a) and (b) are 500 nm and in (d) 200 nm.

was required to facilitate the adsorption and imaging of DNA; functionalization with the block copolymer alone yielded a densely packed surface that did not appear to bind DNA (data not shown). Finally, for even longer PEG chains (PLL₁₀-*b*-PEG₄₅₄), DNA adsorption appeared to be prevented altogether, even in the additional presence of PLL_{1000–2000} chains (Fig. S4†). This comes with the caveat that we cannot fully exclude that DNA adsorption onto the underlying PLL layer is obscured by the PEG layer: the hydrodynamic radius of PEG₄₅₄ is ~13.7 nm and therefore the film thickness is expected to be much greater than the height of the DNA.

In addition to varying the PEG block length, we studied the effect of changing the PLL chain length *x* in the diblock copolymer PLL_{*x*}-*b*-PEG₁₁₃ (Fig. S4†). In contrast to PLL₁₀-*b*-PEG₁₁₃, PLL₁₀₀-*b*-PEG₁₁₃ facilitated DNA adsorption without additional long chain PLL, however this surface was less selective, binding increased streptavidin. This implies that the longer lysine block (in PLL_{*x*}-*b*-PEG₁₁₃) increases the spacing between the passivating PEG moieties. In this case the effective grafting distance between these moieties becomes larger than their extension (radius of gyration), such that they adopted collapsed coil conformations and no longer formed an effective steric barrier.²² We also note that with the increased length PLL in the block copolymer, the DNA plasmids appear more condensed than on PLL₁₀-*b*-PEG_{*y*}, forming toroid and rod-like structures, as seen in Fig S4(a) and (b).†⁴⁰

Full quantification of streptavidin binding is complicated when considering images with complex arrangements of DNA

and streptavidin on the surface. However, PLL₁₀-*b*-PEG₁₁₃/PLL_{1000–2000} functionalization emerges as the most effective in suppressing streptavidin binding whilst allowing visualization of DNA by AFM, both by qualitative comparison of the images and by tentative quantification (Fig. 4c). To determine if this functionalization is also effective at studying DNA–protein interactions, we created a 672 bp length of dsDNA with a single biotin at each end by using PCR amplification with biotinylated primers. Biotin binds to streptavidin with an extremely high affinity with *K_d* on the order of femtomolar. These binding partners were chosen for the strong binding affinity of their interaction and relatively low dissociation rate (less than 10% of molecules dissociated after 12 hours at 37 °C).³³ Two streptavidin variants were considered: tetravalent streptavidin and monovalent streptavidin. Although both exhibit a similar binding affinity, monovalent streptavidin has only one functional biotin binding subunit compared to four in wild-type streptavidin. This prevents end-tailing of biotin labelled DNA. The binding of both proteins to the dsDNA construct was confirmed by electrophoretic band shift assay (Fig. S5†). DNA incubated with a 50× excess of monovalent streptavidin was adsorbed on the PLL₁₀-*b*-PEG₁₁₃/PLL_{1000–2000} functionalized mica surface (Fig. 5(a)). DNA molecules with streptavidin bound to both biotinylated ends were specifically adsorbed to the surface (Fig. 5(b)). The excess monovalent streptavidin in solution was not observed at high concentration on the surface, implying good non-specific protein passivation. 40% of the adsorbed streptavidin molecules were found at the ends



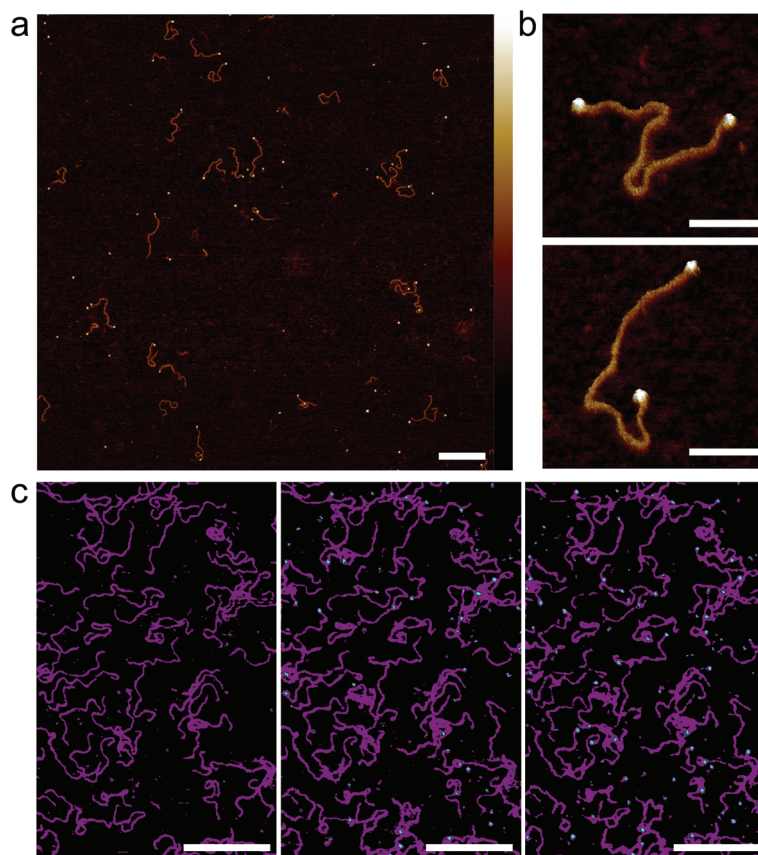


Fig. 5 Streptavidin binding to dual-end biotinylated 672 bp DNA on mica treated with PLL₁₀-*b*-PEG₁₁₃/PLL_{1000–2000}. (a) AFM image taken in solution of 672 bp DNA after pre-incubation with ~50 molar excess of monovalent mono-streptavidin over biotin tag. (b) High-resolution images showing mono-streptavidin bound to both ends of dual-biotin DNA. (c) An AFM image with the colour scale adjusted to highlight immobilized DNA (magenta) and added tetraivalent streptavidin (cyan) for streptavidin concentrations of 0 nM, 150 nM and 755 nM (from left to right). Colour scale for (a) and (b) is 5 nm. Scale bars for (a) and (c) are 200 nm and for (b) 50 nm.

of DNA ($n = 531$). The advantages of PLL₁₀-*b*-PEG₁₁₃/PLL_{1000–2000} functionalization were further demonstrated by comparison with the traditional PLL_{1000–2000} only functionalization which yielded increased adsorption of non-DNA-bound streptavidin on the surface (Fig. S6†).

To determine next whether this method can be used to study the binding of proteins to DNA *in situ*, tetraivalent streptavidin was flowed over biotinylated DNA that had already been immobilized on PLL₁₀-*b*-PEG₁₁₃/PLL_{1000–2000} functionalized mica (Fig. 5(c)). Binding is observed as the formation of cyan protrusions at the ends of the immobilized biotinylated DNA substrates (magenta) which increase from 150 nM to 750 nM. A higher concentration of streptavidin is required for immobilized biotinylated DNA as compared to biotinylated DNA in solution. This suggests limited accessibility of the biotin binding site which is attached to the end of a large DNA molecule and hidden underneath the PEG layer. In this instance tetraivalent streptavidin was used as opposed to monovalent streptavidin to increase the surface area for binding and thus reduce steric hindrance effects.⁴¹ High-resolution AFM imaging requires the surface immobilization of DNA, which can result in the masking of binding sites, and

consequently we found it best, in this case, to pre-incubate the DNA with the streptavidin prior to depositing the DNA on the surface.

Finally, we consider the nuclear enzyme poly(ADP-ribose) polymerase-1 (PARP1) as an example of a DNA-binding protein in a biologically relevant context. Present in the nucleus at a concentration on the order of 10 μ M,^{42,43} PARP1 plays an important role in DNA break repair and as such has also been targeted by anticancer drugs.⁴⁴ Previous AFM studies on PARP-DNA binding, carried out on dried samples, have shown that the sample preparation for observing DNA-bound PARP is non-trivial, and have visualized PARP bound to DNA breaks/ends as well as to undamaged DNA.^{45–47} Here we use PARP1 to demonstrate the wider applicability of our method, imaging DNA before and after the addition of the enzyme in solution (Fig. 6). Importantly, we find that the added PARP predominantly binds to the solid support (Fig. 6(a and b)) when this mica support is functionalized using common protocols for AFM imaging in solution, such as the addition of Ni²⁺ ions⁴⁸ and PLL_{1000–2000}.²⁰ Specifically, we observe a corrugated background of surface-bound proteins over the whole image (Fig. 6 (a and b), bottom), precluding the identification of specific



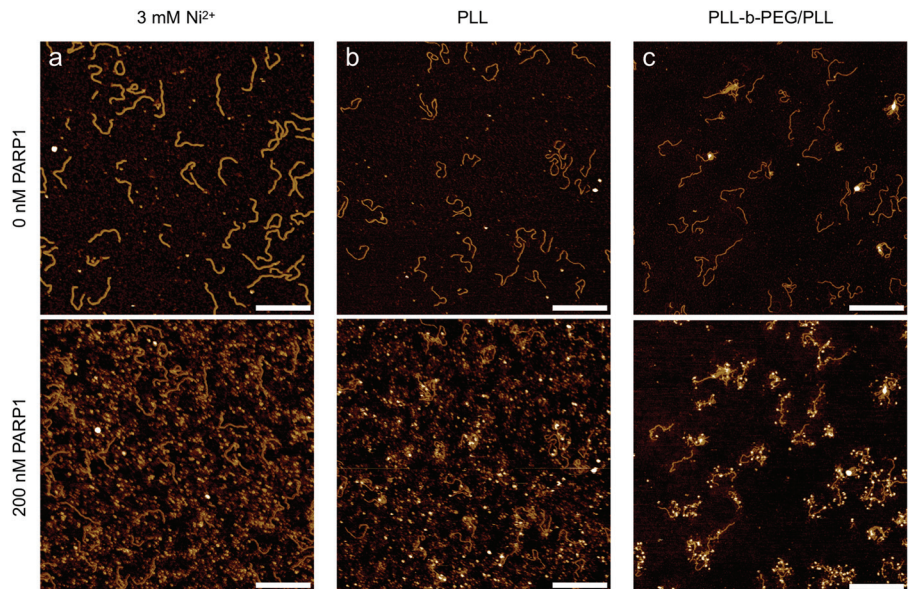


Fig. 6 PARP1 binding to 496 bp linear DNA on different mica surfaces. In-solution AFM images of the same area with surface-bound DNA molecules before (top) and after (bottom) the addition of 200 nM PARP1 (a concentration at which – under appropriate conditions – noticeable PARP1-DNA binding could be observed in solution), with the DNA adhered (a) to bare mica via Ni^{2+} ions, (b) to mica that has been functionalized with $\text{PLL}_{1000-2000}$ and (c) to mica that has been functionalized with $\text{PLL}_{10-b}\text{-PEG}_{113}/\text{PLL}_{1000-2000}$. Colour scale is 4 nm; scale bars 200 nm.

protein–DNA binding events. By contrast, in line with our results on streptavidin, $\text{PLL}_{10-b}\text{-PEG}_{113}/\text{PLL}_{1000-2000}$ adequately passivates the mica substrate against protein binding while still allowing DNA adhesion, and thus facilitates the single-molecule detection of DNA-bound PARP1, here shown as white, globular structures decorating the DNA molecules (Fig. 6(c), bottom).

Conclusions

We have demonstrated the use of hydrophilic diblock copolymers comprising both a cationic surface binding domain (PLL) along with a neutral protein repellent domain (PEG) for the formation of passivating films for the selective immobilization of DNA and DNA–protein complexes. The chain lengths of both blocks were optimized to repel the non-specific adsorption of streptavidin in solution whilst adsorbing highly charged DNA molecules. The surface-passivating properties of this PEG film are demonstrated through the selective binding of biotinylated DNA–streptavidin complexes, minimising non-specific streptavidin surface binding. Finally, by visualizing the binding of the nuclear enzyme PARP1 to surface-bound DNA molecules, we illustrate how this surface functionalization can facilitate AFM studies of DNA interactions with other, biologically relevant proteins.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 N. M. Luscombe, S. E. Austin, H. M. Berman and J. M. Thornton, *Genome Biol.*, 2000, **1**, 1–37.
- 2 N. Fili, C. P. Toseland, M. S. Dillingham, M. R. Webb and J. E. Molloy, in *Single Molecule Enzymology: Methods and Protocols*, ed. G. I. Mashanov and C. Batters, Humana Press, Totowa, NJ, 2011, pp. 193–214.
- 3 J. Hohlbein and A. N. Kapanidis, in *Single-molecule enzymology: fluorescence-based and high-throughput methods*, ed. M. Spies and Y. R. Chemla, Elsevier Academic Press Inc., San Diego, 2016, vol. 581, pp. 353–378.
- 4 M. Ganji, I. A. Shaltiel, S. Bisht, E. Kim, A. Kalichava, C. H. Haering and C. Dekker, *Science*, 2018, **360**, 102–105.



- 5 A. Tempestini, C. Monico, L. Gardini, F. Vanzi, F. S. Pavone and M. Capitanio, *Nucleic Acids Res.*, 2018, **46**, 5001–5011.
- 6 M. D. Newton, B. J. Taylor, R. P. C. Driessen, L. Roos, N. Cvetic, S. Allyjaun, B. Lenhard, M. E. Cuomo and D. S. Rueda, *Nat. Struct. Mol. Biol.*, 2019, **26**, 185–192.
- 7 B. Dey, S. Thukral, S. Krishnan, M. Chakrobarty, S. Gupta, C. Manghani and V. Rani, *Mol. Cell. Biochem.*, 2012, **365**, 279–299.
- 8 Y. L. Lyubchenko, A. A. Gall and L. S. Shlyakhtenko, in *Electron Microscopy: Methods and Protocols*, ed. J. Kuo, Humana Press, Totowa, NJ, 2014, pp. 367–384.
- 9 M. Shibata, H. Nishimasu, N. Kodera, S. Hirano, T. Ando, T. Uchihashi and O. Nureki, *Nat. Commun.*, 2017, **8**, 1430.
- 10 R. Subramani, S. Juul, A. Rotaru, F. F. Andersen, K. V. Gothelf, W. Mamdouh, F. Besenbacher, M. Dong and B. R. Knudsen, *ACS Nano*, 2010, **4**, 5969–5977.
- 11 L. Alonso-Sarduy, C. Roduit, G. Dietler and S. Kasas, *FEBS Lett.*, 2011, **585**, 3139–3145.
- 12 A. Pyne, R. Thompson, C. Leung, D. Roy and B. W. Hoogenboom, *Small*, 2014, **10**, 3257–3261.
- 13 D. Pastre, O. Pietrement, S. Fusil, F. Landousy, J. Jeusset, M.-O. David, L. Hamon, E. Le Cam and A. Zozime, *Biophys. J.*, 2003, **85**, 2507–2518.
- 14 C. Leung, H. Kinns, B. W. Hoogenboom, S. Howorka and P. Mesquida, *Nano Lett.*, 2009, **9**, 2769–2773.
- 15 M. Bezanilla, S. Manne, D. E. Laney, Y. L. Lyubchenko and H. G. Hansma, *Langmuir*, 1995, **11**, 655–659.
- 16 H. G. Hansma and D. E. Laney, *Biophys. J.*, 1996, **70**, 1933–1939.
- 17 A. J. Lee, M. Szymonik, J. K. Hobbs and C. Wälti, *Nano Res.*, 2015, **8**, 1811–1821.
- 18 P. R. Heenan and T. T. Perkins, *ACS Nano*, 2019, **13**, 4220–4229.
- 19 J. Mou, D. M. Czajkowsky, Y. Zhang and Z. Shao, *FEBS Lett.*, 1995, **371**, 279–282.
- 20 M. Bussiek, N. Mücke and J. Langowski, *Nucleic Acids Res.*, 2003, **31**, e137–e137.
- 21 T. Brouns, H. De Keersmaecker, S. F. Konrad, N. Kodera, T. Ando, J. Lipfert, S. De Feyter and W. Vanderlinden, *ACS Nano*, 2018, **12**, 11907–11916.
- 22 M. Heuberger, T. Drobek and N. D. Spencer, *Biophys. J.*, 2005, **88**, 495–504.
- 23 P. Kingshott, H. Thissen and H. J. Griesser, *Biomaterials*, 2002, **23**, 2043–2056.
- 24 G. L. Kenausis, J. Vörös, D. L. Elbert, N. Huang, R. Hofer, L. Ruiz-Taylor, M. Textor, J. A. Hubbell and N. D. Spencer, *J. Phys. Chem. B*, 2000, **104**, 3298–3309.
- 25 N.-P. Huang, R. Michel, J. Voros, M. Textor, R. Hofer, A. Rossi, D. L. Elbert, J. A. Hubbell and N. D. Spencer, *Langmuir*, 2001, **17**, 489–498.
- 26 S. Pasche, S. M. De Paul, J. Vörös, N. D. Spencer and M. Textor, *Langmuir*, 2003, **19**, 9216–9225.
- 27 T. M. Blättler, S. Pasche, M. Textor and H. J. Griesser, *Langmuir*, 2006, **22**, 5760–5769.
- 28 R. Schlapak, D. Armitage, N. Saucedo-Zeni, W. Chrzanowski, M. Hohage, D. Caruana and S. Howorka, *Soft Matter*, 2009, **5**, 613–621.
- 29 S. VandeVondele, J. Vörös and J. A. Hubbell, *Biotechnol. Bioeng.*, 2003, **82**, 784–790.
- 30 S. Patel, R. G. Thakar, J. Wong, S. D. McLeod and S. Li, *Biomaterials*, 2006, **27**, 2890–2897.
- 31 L. A. Ruiz-Taylor, T. L. Martin, F. G. Zaugg, K. Witte, P. Indermuhle, S. Nock and P. Wagner, *Proc. Natl. Acad. Sci. U. S. A.*, 2001, **98**, 852–857.
- 32 E. Tkachenko, M. Sabouri-Ghomi, O. Pertz, C. Kim, E. Gutierrez, M. Machacek, A. Groisman, G. Danuser and M. H. Ginsberg, *Nat. Cell Biol.*, 2011, **13**, 660.
- 33 M. Howarth, D. J. F. Chinnapan, K. Gerrow, P. C. Dorrestein, M. R. Grandy, N. L. Kelleher, A. El-Husseini and A. Y. Ting, *Nat. Methods*, 2006, **3**, 267–273.
- 34 M. Porus, P. Maroni and M. Borkovec, *Langmuir*, 2012, **28**, 5642–5651.
- 35 X. Jin, L. Leclercq, N. Sisavath and H. Cottet, *Macromolecules*, 2014, **47**, 5320–5327.
- 36 T. Arai, K. Sato, A. Iida and M. Tomitori, *Sci. Rep.*, 2017, **7**, 4054.
- 37 N. M. Green, *Methods Enzymol.*, 1990, **184**, 51–67.
- 38 J. N. Israelachvili, *Intermolecular and Surface Forces*, 3rd edn, Academic Press, San Diego, CA, 2011.
- 39 N. Giamblanco, G. Marletta, A. Graillot, N. Bia, C. Loubat and J.-F. Berret, *ACS Omega*, 2017, **2**, 1309–1320.
- 40 D. Y. Kwoh, C. C. Coffin, C. P. Lollo, J. Jovenal, M. G. Banaszczyk, P. Mullen, A. Phillips, A. Amini, J. Fabrycki, R. M. Bartholomew, S. W. Brostoff and D. J. Carlo, *Biochim. Biophys. Acta, Gene Struct. Expression*, 1999, **1444**, 171–190.
- 41 G. V. Dubacheva, C. Araya-Callis, A. Geert Volbeda, M. Fairhead, J. Codée, M. Howarth and R. P. Richter, *J. Am. Chem. Soc.*, 2017, **139**, 4157–4167.
- 42 A. Ludwig, B. Behnke, J. Holtlund and H. Hilz, *J. Biol. Chem.*, 1988, **263**, 6993–6999.
- 43 H. Yamanaka, C. A. Penning, E. H. Willis, D. B. Wasson and D. A. Carson, *J. Biol. Chem.*, 1988, **263**, 3879–3883.
- 44 Y. Pommier, M. J. O'Connor and J. de Bono, *Sci. Transl. Med.*, 2016, **8**, 362ps17.
- 45 M. V. Sukhanova, S. Abrakhi, V. Joshi, D. Pastre, M. M. Kutuzov, R. O. Anarbaev, P. A. Curmi, L. Hamon and O. I. Lavrik, *Nucleic Acids Res.*, 2015, **44**, e60–e60.
- 46 L. Liu, M. Kong, N. R. Gassman, B. D. Freudenthal, R. Prasad, S. Zhen, S. C. Watkins, S. H. Wilson and B. Van Houten, *Nucleic Acids Res.*, 2017, **45**, 12834–12847.
- 47 M. V. Sukhanova, L. Hamon, M. M. Kutuzov, V. Joshi, S. Abrakhi, I. Dobra, P. A. Curmi, D. Pastre and O. I. Lavrik, *J. Mol. Biol.*, 2019, **431**, 2655–2673.
- 48 H. G. Hansma, D. E. Laney, M. Bezanilla, R. L. Sinsheimer and P. K. Hansma, *Biophys. J.*, 1995, **68**, 1672–1677.

