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A human exposome framework for guiding risk management and holistic assessment of recycled water quality

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Challenges associated with water scarcity and increasing water demand are leading many cities around the globe to consider water reuse as a step towards water sustainability. Recycled water may be used in a spectrum of applications, from irrigation or industrial use to direct potable reuse, and thus presents a challenge to regulators as not all applications require the same level of treatment. We propose that traditional drinking water standards identifying “safe” water quality are insufficient for recycled water and that using the “human exposome” as a framework to guide development of a risk management strategy offers a holistic means by which to base decisions impacting water quality. A successful and comprehensive plan for water reuse must consider 1) health impacts associated with both acute and chronic exposures, 2) all routes of exposure by which individuals may encounter recycled water, and 3) water quality at the true point of use after storage and transport through pipe networks, rather than at the point of treatment. Based on these principles we explore key chemical differences between recycled and traditional potable water, implications for distribution systems with respect to design and operation, occurrence of chronic contaminants, and the presence of emerging and often underappreciated microbial contaminants. The unique nature of recycled water has the potential to provide rapid regrowth conditions for certain microbial contaminants in these systems, which must be considered to achieve safe water quality at the point of use.

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Water impact

Here, we explore the inherent chemical differences between recycled and potable water and their implications for microbial regrowth in recycled water distribution systems. We propose that the concept of the “human exposome” – the summation of environmental exposures over the course of a lifetime – provides ideal guiding principles for the development of appropriate water reuse risk management strategies.

Introduction

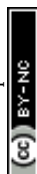
Water reuse is essential for satisfying domestic and industrial water demand worldwide and achieving water sustainability.¹ Domestic wastewater can be treated to the necessary level of quality and reused to reduce loss of treated effluent *via* discharge, relieve pressures on depleting groundwater aquifers, and minimize extraction of water from fragile environments such as drought-stricken surface waters. Treatment of wastewater for reuse is also cost effective compared to alternative approaches, such as obtaining freshwater from desalination.² Particularly when wastewater is treated for direct or indirect potable reuse, a “multi-barrier” treatment framework is typi-

cally used to ensure that multiple means of removing pathogens or harmful chemicals will protect public health in the case of a process failure or other unexpected event that could compromise water quality.³ However, while this approach is logical for controlling acute health threats associated with water as it leaves the treatment facility, it does not address concerns with respect to low-level chronic exposures or changes in water quality during distribution to the point of use.⁴

A major concern for water reuse in general is the lack of federal regulations.⁵ Further, the few nascent recycled water quality regulations and guidelines available, typically at the state and local level, have been narrowly focused on fecal indicator bacteria (*i.e.*, total and fecal coliforms).^{6–8} This approach addresses traditional concerns regarding fecal-associated pathogens, but does not necessarily provide insight into safeguarding microbial water quality during distribution.^{9,10} In particular, microbial contaminants that are of

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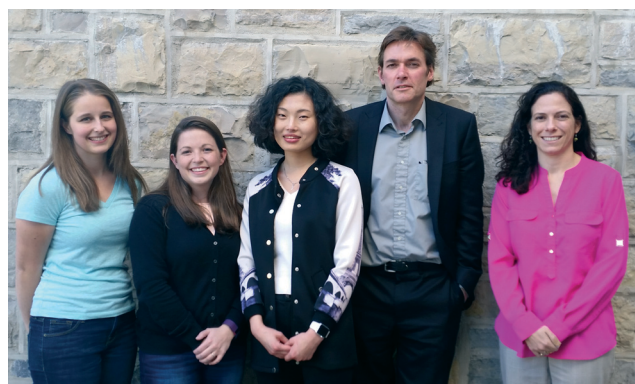
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concern due to their ability to grow within distribution systems, such as opportunistic pathogens (OP), antibiotic resistant bacteria and their associated antibiotic resistance genes (ARG), and free-living amoebae (FLA), have little or no relationship with fecal indicator bacteria based standards. Alternative frameworks for assessing and managing recycled water quality more holistically are emerging. For example, adaptations of the hazard analysis and critical control point (HACCP)¹¹ paradigm, which originated in the food safety industry, and the World Health Organization's (WHO) water safety plan (WSP)^{2,12} have been proposed for use as risk management frameworks for recycled water. Application of these adaptable frameworks could have the advantage of drawing attention to the entire treatment process, rather than focusing solely on absence of indicator organisms as a proxy for safe water. Still, we identify three key elements that should be taken into account for a truly comprehensive consideration of public health concerns: 1) evaluation of health impacts associated with both acute and chronic exposures; 2) accounting for all routes of exposure by which individuals may encounter contaminants in recycled water; and 3) con-

sideration of water quality at the true point of use after storage and transport through pipe networks, rather than at the point of treatment.

A more holistic approach to characterizing the physical, chemical, and microbial characteristics of recycled water, as well as the routes by which humans are exposed, can be derived from the emerging concept of the "human exposome." The exposome has been defined as "the cumulative measure of environmental influences and associated biological responses throughout the lifespan, including exposures from the environment, diet, behavior, and endogenous processes."¹³ The exposome includes general (e.g., climate, urban environment), specific (e.g., water, food, air), and internal (e.g., metabolism, gut/lung microbes) factors and their role in disease.¹⁴ Clearly, water, and its corresponding chemical and microbial properties is a fundamental component of the exposome. Water is fundamental to human health, survival, and hygiene and is an integral part of daily life, including direct contact (i.e., drinking, cooking, and showering) and indirect exposures *via* bioaerosols (i.e., cooling water, flush toilets, or lawn irrigation). We propose that adopting the



From left to right, the authors in the photograph are Laurel Strom, Emily Garner, Ni Zhu, Marc Edwards, and Amy Pruden

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water chemistry on the attenuation of emerging microbial contaminants and antibiotic resistant gene/bacteria in recycled water distribution systems, using the next-generation sequencing technology.

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Marc Edwards is the Charles P. Lunsford Professor of Civil and Environmental Engineering at Virginia Tech and a MacArthur Foundation Fellow. He is widely considered the foremost authority on lead in drinking water and recently revealed widespread lead contamination in Flint, MI tap water, along with an array of government agency failures that led to the crisis. His current research explores how water chemistry shapes the tap water microbiome and implications for design of water and energy conserving systems in green buildings.

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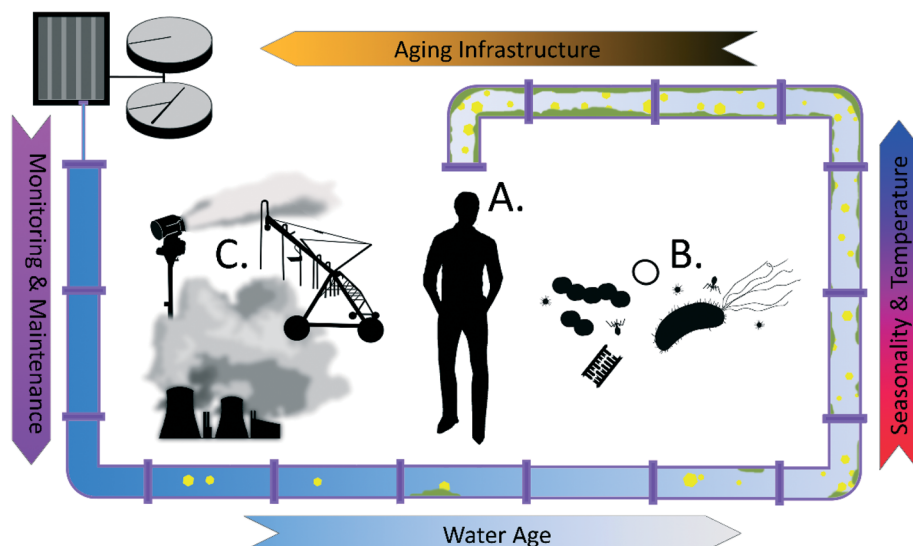


Fig. 1 Key aspects of the exposome paradigm for managing RWDS emphasize holistic consideration of potential exposures to recycled water, including A) chemical distinctions of recycled vs. traditional potable water, such as enriched organic matter/nutrients, disinfectant decay, critical reactive zones and chronic contaminants; B) emerging concerns about ARGs, OPs and other microbial contaminants; C) nontraditional routes of exposure, including inhalation, dermal contact.

exposome paradigm as a model for recycled water quality assessment can be used to guide development of an HACCP, WSP, or other comprehensive risk management strategy that more accurately reflects the true risks and exposures associated with water reuse and can strengthen implementation of existing risk management strategies.

In terms of safeguarding recycled water from the point of treatment to the point of use, we note that there are important distinctions between potable and recycled water that should not be ignored, particularly with respect to design, operation, and maintenance of recycled water distribution systems (RWDS). In this critical review, we note key chemical differences between recycled and potable waters, with a particular emphasis on organic matter, and explore implications for RWDSs with respect to design, operation, and intended application. We also discuss how these key differences impact the presence of chronic contaminants and emerging concerns about the presence of OPs, ARGs, and other microbial contaminants (Fig. 1). Our goal is to proactively address plausible public health risks associated with practical realities of recycled water use.

Unique aspects of RWDS design, operation, and water use

There is a broad continuum of applications for water reuse, ranging from unintended *de facto* reuse to direct potable reuse (DPR) produced by advanced treatment processes. *De facto* reuse refers to a situation where reuse of treated wastewater occurs but is not planned, for example, when a drinking water treatment plant intake is located downstream from a wastewater discharge.⁶ In the U.S., *de facto* reuse is widespread and becoming increasingly common in recent de-

cades. Rice *et al.* found that of the top 25 drinking water treatment plants most impacted by upstream wastewater treatment plant (WWTP) discharges in the U.S., the fraction of their water source comprised of WWTP discharges increased from between 2 to 16% in 1980 to an average of 68% under typical streamflow conditions in 2008.¹⁵ Some treatment plants received as much as 100% WWTP effluent under low flow conditions. Indirect potable reuse refers to the use of treated wastewater to augment other potable source waters following retention in an environmental buffer.⁶ Common environmental buffers include groundwater aquifer recharge and subsequent withdrawal prior to drinking water treatment or intentional discharge of wastewater effluent upstream or into a reservoir from which water is withdrawn for drinking water treatment. DPR consists of treating wastewater for direct use as a source water for drinking water treatment. DPR is currently limited in full scale application, with Windhoek, Namibia¹⁶ and Big Spring, Texas, USA¹⁷ serving as prime examples, but there is growing interest in expanding DPR infrastructure in the U.S.

Wastewater may also be treated for non-potable reuse when it can offset water demand associated with landscape or recreation area irrigation, agricultural and food crop irrigation, snowmaking, industrial use such as in cooling towers or in natural gas production, or to augment environmental waters such as in groundwater aquifer recharge, river or stream flow augmentation, or in wetlands.⁶ Though treatment requirements may reasonably be lower in these cases, all of these non-potable reuse scenarios have relevant human exposures that should be taken into consideration.

Table 1 illustrates that there is a wide range of observed recycled water quality characteristics as a function of increasing levels of treatment. Unlike drinking water, where



Table 1 Water quality as a function of treatment processes

Water quality parameter (units)	Treatment processes								
	Conventional activated sludge (CAS) effluent	CAS with filtration	CAS with biological nutrient removal	CAS with biological nutrient removal and filtration	Membrane bioreactor	Filtration with chlorine	CAS with filtration and chlorine	Membrane bioreactor with chlorine	Membrane bioreactor with UV/ozone
Turbidity (NTU)	2–15	0.5–4	2–8	0.3–2	≤1	1.5–8.7	3.6–6.3	1.7–4.3	0.2–0.5
Total suspended solids (mg L ⁻¹)	5–25	2–8	5–20	1–4	<2				
TOC (mg-C L ⁻¹)	10–40	8–30	8–20	1–5	0.5–5	12–16	6–8	3–5	2–3
BDOC (mg-C L ⁻¹)						5–7	1–2	1–2	<1
AOC (mg-C L ⁻¹)					0.2–1.4	1–2	1–2	1–2	<1
Total nitrogen (mg-N L ⁻¹)	15–35	15–35	3–8	2–5	<10	5–10	<1	1–3	5–6
Total phosphorus (mg-P L ⁻¹)	4–10	4–8	1–2	≤2	<0.3–5	3–5	1–2	5–9	<1
Volatile organic compounds (μg L ⁻¹)	10–40	10–40	10–20	10–20	10–20				
Total coliforms (CFU per 100 mL)	10 ⁴ –10 ⁵	10 ³ –10 ⁵	10 ⁴ –10 ⁵	10 ⁴ –10 ⁵	<100	<1	1–10	1–10	1–10
Protozoa and cysts (CFU per 100 mL)	10–10 ²	0–10	0–10	0–1	0–1	1–10	10–100	<1	≤1
Viruses (CFU per 100 mL)	10–10 ³	10–10 ³	10 ¹ –10 ³	10–10 ³	1–10 ³	Present	Present	Present	Negative

Source ref. 29 and 192.

consistent regulations are applied, recycled water is used for a wide spectrum of applications with required treatment varying based on intended use. For example, the most stringent requirements are applied to DPR, whereas residual nutrients can be viewed as a beneficial fertilizer in non-potable reuse scenarios. Hence, efficient reuse treatments ideally match the intended purpose of the recycled water. A number of U.S. states pioneering water reuse have recognized this concept of “fit-for-purpose,” tailoring recycled water treatment regulatory guidelines based on end uses. Determining the ideal configuration of treatment processes for different reuse scenarios would greatly benefit from research integrating water treatment outcomes with the exposome paradigm for more comprehensively considering chemical and microbial risks.

Routes of exposure

Stemming from the continuum of recycled water uses described above, there is also a range of relevant exposure scenarios (Table 2). Use of a traditional water quality paradigm based on monitoring of fecal indicator bacteria as a sole benchmark for establishing safe water quality neglects risk associated with non-fecal pathogens and routes of exposure other than ingestion. Though ingestion and aspiration are potential modes of exposure associated with DPR, inhalation and dermal contact are important, yet often overlooked, exposure routes for potable water and even more so for recycled water. Recycled water is commonly used for purposes that

generate inhalable aerosols, including use in cooling towers,¹⁸ spray irrigation,^{19,20} fire-fighting,²¹ toilet flushing,²² as well as for aesthetic purposes, such as in decorative fountains.^{23–25} Importantly, relevant exposure zones may be vast, with Legionnaire's disease infection associated with inhalation of aerosols from cooling towers located more than a mile away.¹⁸ There is also a strong likelihood of dermal contact when recycled water is used for irrigation in public recreation areas. This may be particularly relevant in irrigated parks, athletic fields, and snowmaking for recreational purposes. Dermal abrasions or other lacerations that may be pre-existing or occur during use of these facilities create an additional route of exposure for infection. When recycled water is used for food crop irrigation, chemical and microbial constituents may also be transmitted to humans on or within crops *via* ingestion.²⁶ When recycled water is used for DPR, a number of often overlooked routes of exposure should also be considered, for example, including use in humidifiers, ice machines, and decorative water features. A more holistic characterization of human exposures to recycled water constituents *via* such non-conventional routes is important for accurate assessment of health risk associated with use of recycled water.

Physical and operational issues

Water and wastewater infrastructure degradation has become one of the leading threats to public health and water



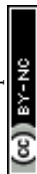
Table 2 Overview of non-traditional routes of exposure for recycled water and putative risk of infection or exposure

Route of exposure	Recycled water application (in addition to other potable uses)	Putative hazard	Infectious dose	Documented concentrations in recycled water [Percent samples positive (positive concentration range)]	Documented concentrations in drinking water
Dermal contact	Snowmaking, irrigation of athletic and recreation facilities, toilet flushing ²²	Opportunistic pathogen infection of wounds <i>Staphylococcus aureus</i> ¹⁰⁷ <i>Pseudomonas aeruginosa</i> ¹⁰⁸ <i>Acanthamoeba</i> spp. ¹⁶⁹ Cyanobacteria toxicity ¹⁸¹ Antibiotic resistant infections ¹⁹⁸	10^8 – 10^9 colony forming units (CFU) ^{a,195} 10^4 trophozoites ^{c,196}	17% ¹⁹³ 88% (1 ± 1 – 9 ± 10 CFU per 100 mL) ^{b,29} (up to $\sim 140 \mu\text{g mL}^{-1}$ chlorophyll) ¹⁹⁷ 8% for MRSA of 17% for susceptible <i>S. aureus</i> ^{d,193}	6.25% ¹⁹⁴ 5.6% (up to 700.3 $\pm$ 158.7 gene copies per mL) ¹¹¹ Not detected ¹¹¹
Inhalation and aspiration	Cooling towers, spray irrigation, toilet flushing, ²² fire suppression, ²¹ car washing ¹⁹⁹	Opportunistic pathogen infection of lungs <i>Legionella pneumophila</i> ^{e,20} <i>Mycobacterium</i> spp. ²⁰¹ <i>Pseudomonas aeruginosa</i> ²⁰² <i>Staphylococcus aureus</i> ¹⁹ <i>Naegleria fowleri</i> ^{168,169} Cyanobacteria toxicity ¹⁸¹ Antibiotic resistant infections ¹⁹ Fecal pathogens Disinfection byproducts ⁷⁹	10^3 – 10^6 CFU ²⁰⁰ 10^4 – 10^7 CFU ^{a,g,195} 10^8 – 10^9 CFU ^{a,195} 10^3 – 10^5 trophozoites ^{168,203}	81% ($0.4 \times 10^3 \pm 0.2 \times 10^3$ – $3.5 \times 10^3 \pm 16 \times 10^3$ CFU per 100 mL) ^{f,29} 88% (1 ± 1 – 9 ± 10 CFU per 100 mL) ^{b,29} 17% ¹⁹³ (up to $\sim 140 \mu\text{g mL}^{-1}$ chlorophyll) ¹⁹⁷ 8% for MRSA of 17% for susceptible <i>S. aureus</i> ^{d,193} (9.70–399.37 ^h $\mu\text{g TTHM L}^{-1}$) ⁷⁹	5.6% (up to 219.4 $\pm$ 23.8 gene copies per mL) ¹¹¹ 98.1% (2.1 ± 104 – 4.2 ± 103 gene copies per mL) ¹¹¹ 5.6% (up to 700.3 $\pm$ 158.7 gene copies per mL) ¹¹¹ 6.25% ¹⁹⁴ 8–27% ¹⁶⁸
Nasal aspiration	Recreation, Potable reuse (Sinus irrigation)	Opportunistic pathogen infection <i>Naegleria fowleri</i> ¹⁶⁸			8–27% ¹⁶⁸
Eye and ear contact	Recreation, direct potable reuse, indirect potable reuse	Opportunistic pathogen infection <i>Acanthamoeba keratitis</i> ¹⁶⁹ <i>Pseudomonas aeruginosa</i> ¹⁰⁸	10^4 trophozoites ¹⁹⁶ 10^8 – 10^9 CFU ^{a,195}	88% (1 ± 1 – 9 ± 10 CFU per 100 mL) ^{b,29}	5.6% (up to 700.3 $\pm$ 158.7 gene copies per mL) ¹¹¹
Colonization and delayed infection	Various	Antibiotic resistant infections ^{119–121} Opportunistic pathogen infection <i>Pseudomonas aeruginosa</i> ¹⁰⁸ <i>Staphylococcus aureus</i> ¹⁰⁷	10^8 – 10^9 CFU ^{a,195}	8% for MRSA of 17% for susceptible <i>S. aureus</i> ^{d,193} 88% (1 ± 1 – 9 ± 10 CFU per 100 mL) ^{b,29} 17% ¹⁹³	6.25% ¹⁹⁴

^a Oral route of infection. ^b Based on *Pseudomonas* spp. ^c Based on *Acanthamoeba keratitis*. ^d MRSA = methicillin resistant *Staphylococcus aureus*. ^e Putative hazards consider both *Legionella pneumophila* and other pathogenic *Legionella*. ^f Based on detection of *Legionella* spp. ^g Based on *Mycobacterium avium*. ^h TTHM = total trihalomethanes.

security.²⁷ Aging and poorly managed pipes can lead to a drastic decrease in water quality of the transported water,

arising from complex interactions among chemical, physical, and microbial constituents. New design considerations are



needed to ensure sustainable management of recycled water infrastructure, considerate of their distinct physical and operational characteristics relative to potable water systems.

In a recent survey of 71 recycled water systems in the US and Australia, Jjemba *et al.* identified infrastructure issues as the most prevalent problem associated with managing and maintaining water quality in RWDSSs. Over 20% of recycled water facilities listed infrastructure integrity as a water quality concern. The extensive list of infrastructure challenges revealed by the survey includes infrastructure deterioration from high chlorine residual, maintenance of desired pressure

and flow during low and inconsistent usage, lack of redundant design and storage, complicated branched distribution systems designed to supply multiple recycled applications from a centralized treatment plant, high corrosivity of water damaging metal pipes, and effective monitoring of the chemical and microbial quality.⁴ For example, to control microbial activity resulting from nutrient-rich recycled water, up to 40 mg L⁻¹ chlorine has been used in some systems, which can potentially result in widespread damage to water infrastructure.²⁸ Use of reservoirs as a way to satisfy on-demand recycled water applications has also been observed to be a

Table 3 Comparing water quality of typical drinking water vs. different recycled water applications

Parameter	Drinking water	Recycled water applications				Implications for distribution
		Private, urban and irrigation	Direct environmental reuse	Indirect potable recharge	Industrial applications	
Carbon source	500					-Provides the most limiting nutrient source for bacterial regrowth in distribution systems; -Consumption of carbons in the distribution system is observed to relate with increased bacterial activity ^{29,204}
Total dissolved solids (mg L ⁻¹)						
Chemical oxygen demand (mg L ⁻¹)		100	70–100	70–100	70	
Biochemical oxygen demand (mg L ⁻¹)		10–20	10–20		10	
pH	6.5–8.5	6–9	6–9	7–9	7–8.5	-Control bacterial growth in the distribution system; -Excess chlorine can cause carbon fragmentation and DBPs formation -Chlorine may exacerbate antibiotic resistance ^{147,148} -Concerns for nitrification and denitrification -Concern for nitrification -Eutrophication and degradation of water quality -Caused “red water” -Promote growth of corrosion bacteria and damage pipe integrity -May select for antibiotic resistant bacteria ²⁰⁵ -Promote growth of certain corrosion bacteria -Toxic to certain bacterial and aquatic species at elevated levels -May select for antibiotic resistant bacteria ^{143,206} -May select for antibiotic resistant bacteria ^{143,206} -May select for antibiotic resistant bacteria ²⁰⁷ Indicator bacteria for pathogenicity of water
DO (mg L ⁻¹)	Near saturation	>0.5	>3	>8	>3	
Total suspended solids (mg L ⁻¹)	N/A	10	10		10	
Chlorine residual (mg L ⁻¹)	<4	0.2–1.0	0.05		0.05	
Total Kjeldahl N (mg L ⁻¹)	<10	15–20	10–20		10	-Concerns for nitrification and denitrification -Concern for nitrification -Eutrophication and degradation of water quality -Caused “red water” -Promote growth of corrosion bacteria and damage pipe integrity -May select for antibiotic resistant bacteria ²⁰⁵ -Promote growth of certain corrosion bacteria -Toxic to certain bacterial and aquatic species at elevated levels -May select for antibiotic resistant bacteria ^{143,206} -May select for antibiotic resistant bacteria ^{143,206} -May select for antibiotic resistant bacteria ²⁰⁷ Indicator bacteria for pathogenicity of water
Ammonia-N (mg L ⁻¹)	< 0.2 ^a	2–20	1.5	0.2	1.5	
Total phosphorus (mg L ⁻¹)		2–5	0.2		0.2	
Iron (mg L ⁻¹)	0.3	2	2			
Copper (mg L ⁻¹)	1.0	0.2–1.0	0.2–1.0			-Concerns for nitrification and denitrification -Concern for nitrification -Eutrophication and degradation of water quality -Caused “red water” -Promote growth of corrosion bacteria and damage pipe integrity -May select for antibiotic resistant bacteria ²⁰⁵ -Promote growth of certain corrosion bacteria -Toxic to certain bacterial and aquatic species at elevated levels -May select for antibiotic resistant bacteria ^{143,206} -May select for antibiotic resistant bacteria ^{143,206} -May select for antibiotic resistant bacteria ²⁰⁷ Indicator bacteria for pathogenicity of water
Zinc (mg L ⁻¹)	5	0.5–2	0.5–2			
Pesticide (mg L ⁻¹)		0.05	0.05			
Fecal coliforms (CFU per 100 mL)	Zero	Zero	<200 ^b	Zero ^c	<200 ^a	

Source: ref. 6, 208 and 209. ^a WHO guidelines for drinking water quality. ^b Based on 7 day median with none >800 per 100 mL. ^c Based on 7 day median with non >14 per 100 mL.



challenge, with impaired water quality resulting from long stagnation time and proliferation of algae and aquatic vegetation.²⁹ Experience from potable water systems has also shown that interaction of iron pipes with water containing high organic content and oxygen tends to promote iron release, producing unacceptable discolored water following stagnation.³⁰ It is important to bear in mind that a shift in water chemistry can have disastrous unintended consequences for corrosion,^{31–33} and given the unique chemistry of recycled water (Table 3), it will be especially critical to bear this in mind. For example, previous studies have documented cases where switching potable water pipes to DPR pipes resulted in destabilization of the of existing corrosion scales and biofilms and an undesirable degradation of water quality at the point of use.³⁴ Despite the unresolved challenges associated with transporting recycled water, this alternative type of water also presents a creative opportunity for solving the challenging issue of leaking pipes. Tang *et al.* have successfully demonstrated the autogenous repair phenomenon in copper and iron pipes in drinking water distribution systems (DWDS) *via* beneficial corrosion deposition.³⁵ Optimistically, with a more diverse water chemistry profile, recycled water may be an even better candidate for protecting aging pipes.

Temperature, as an overarching parameter, is another critical factor that could have profound implications in designing and monitoring water reuse systems.³⁶ Not only is temperature directly related to microbial activity, disinfectant residual decay, corrosion rate, and dissolved oxygen levels, it is also indirectly linked to consumption patterns, flow patterns and velocity, and bulk water and biofilm interactions. Elevated recycled water temperatures may stem from extended stagnation times, particularly during the day in cases where irrigation is conducted at night to limit evaporation, as well as from use of above-ground pipelines, which facilitate transport of recycled water over long distances.

For on-demand non-potable water reuse applications, such as agricultural irrigation, landscaping, and toilet flushing, many studies have observed distinct consumption variations in daily and seasonal demand patterns.^{37,38} For example, on a daily scale, the generation of wastewater effluent usually peaks in the daytime when people are active, but the demand of irrigation water usually occurs at night with an offset time of approximately 12 hours. Discrepancies in user patterns makes water stagnation and storage, along with associated water quality deterioration, a prominent concern in design and maintenance of recycled systems. Multiple studies have also documented water quality deterioration during winter or high rainfall periods in systems largely used for irrigation, due to low user frequency.³⁹

True water age may differ substantially from the designed hydraulic residence time of the recycled water systems based on the actual end-use applications. Emerging work in premise (*i.e.*, building) plumbing systems, *i.e.* the water pipe networks within homes and buildings, has highlighted unique systematic features in terms of longer stagnation time, elevated temperature, and loss of disinfectant residual, which

serve to stimulate microbial proliferation precisely at the point of use, thus amplifying any potential exposure risk to end users.^{40,41} Similar investigations are needed to quantify the risk of exposure associated with user-driven demand patterns in non-potable reuse systems.

Important chemical differences anticipated between recycled and potable water distribution systems

Organic matter

One of the most distinctive characteristics of recycled water is the nature of dissolved organic matter (DOM) and its occurrence at elevated levels. The organic matter in typical potable waters consists of natural organic matter (NOM) derived mostly from oil, planktonic and vegetative matter, and decay by-products in natural water sources. However, it is important to note that DOM present in recycled waters may be quite distinct from that of potable water due to different sources and treatment processes. In a recent review comparing organic matter data published in the last 15 years for drinking water and recycled water systems, Hu *et al.* identified four distinct classes of NOM in recycled water: recalcitrant DOM, soluble microbial products from biological wastewater treatment units, transformation products from advanced treatment, and emerging contaminants associated with anthropogenic activities.⁴² It was concluded that DOM composition differed significantly between recycled water

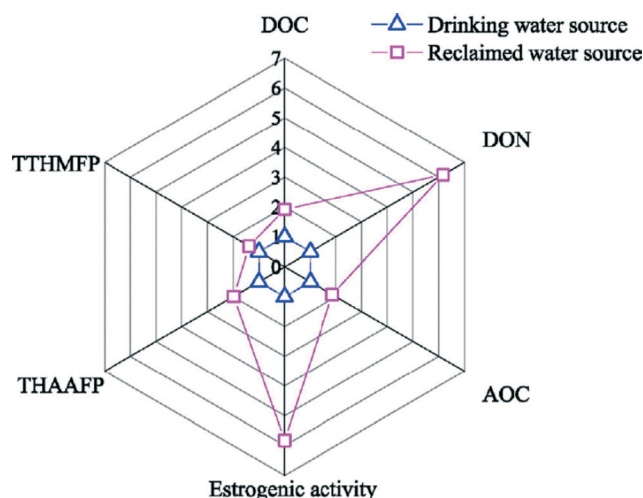


Fig. 2 Overview of typical normalized composition and potential magnitude of dissolved organic matter (DOM) in drinking water sources compared to recycled water sources, presented in terms of dissolved organic carbon (DOC), dissolved organic nitrogen (DON), assimilable organic carbon (AOC), estrogenic activity, total haloacetic acid formation potential (THAAFP), and total trihalomethane formation potential (TTHMFP). (Reprinted from *Sci. Total Environ.*, 551–552, Hong-Ying Hu, Ye Du, Qian-Yuan Wu, Xin Zhao, Xin Tang, Zhuo Chen, differences in dissolved organic matter between reclaimed water source and drinking water source, page 133–142, 2016, with permission from Elsevier.)



and drinking water evaluated against five critical chemical indicators: dissolved organic carbon (DOC), dissolved organic nitrogen, assimilable organic carbon (AOC), estrogenic activity, and disinfection byproduct (DBP) formation potential (Fig. 2). DOC in drinking water ranged from 1.5–11.2 mg L⁻¹ with a median of 3.9 mg L⁻¹ while that in recycled water ranged from 3.6–14.6 mg L⁻¹ with a median of 7.5 mg L⁻¹, indicating recycled water as a much more nutrient rich environment for microbial regrowth and byproduct formation. The heightened levels of biotoxicity, in terms of estrogenic levels, is also widely reported in studies examining effluent organic matter compositions, suggesting a potential health risk when used for recycling applications.⁴³

Biological stability, *i.e.*, the ability of drinking water to suppress microbial growth in the absence of disinfectants,⁴⁴ is especially of concern for safe transport and storage of treated water. Ideally, low nutrient water will limit growth in the distribution system, a strategy applied successfully in some European countries for eliminating the need for secondary disinfectant in DWDSs.⁴⁵ The proportion of DOC that facilitates bacterial regrowth is typically measured by either biodegradable dissolved organic carbon (BDOC) or AOC assays. An array of methods have been used to best evaluate the bacterial growth potential of various types of water samples, with established approaches generally being to measure the decrease in measured DOC over time or an increase in indicator bacteria counts as a proxy for biologically available DOC. To date, there is no widely accepted standardized method to quantify biostability. Reported threshold BDOC and AOC values to achieve biostability in drinking water systems using different methods are documented in Table 4. Existing surveys of recycled water systems have indicated orders of magnitude higher levels of organic carbon than in typical U.S. drinking water systems.^{46,47} In particular, biodegradable organic matter has been observed to be four or five times higher than that of drinking water,²⁹ while AOC can range from 505 to 918 µg L⁻¹ in moderately treated recycled water,⁴⁸ compared to 18 to 189 µg L⁻¹ in drinking water.⁴⁹

The abundance and type of biodegradable carbon in recycled water calls into question the extent to which the science of potable water delivery is directly translatable to recycled water distribution. In the only available study of its

kind, Jjemba *et al.* examined four RWDSs in the U.S. and observed a trend of AOC and BDOC consumption with increasing residence time, with an average reduction of 475 µg L⁻¹ AOC and 370 µg L⁻¹ BDOC from the distribution system point of entry to the point of use.²⁹ They concluded that the change in AOC and BDOC was due to enhanced microbial activity, indicating significant changes in both the quantity and the quality of the available carbon in the RWDSs. In parallel simulated RWDS loop studies, high organic carbon was also observed to result in rapid consumption of disinfectant residuals in the distribution systems.²⁹ Up to 6 mg L⁻¹ of chlorine was completely consumed within minutes in all systems, leaving the remainder of the distribution system vulnerable to bacterial growth.²⁹

Given the unique nature of organic matter and microbial composition of recycled water, existing assays such as those for AOC or BDOC, may not be suitable. Only one study could be found specifically aimed at adapting the AOC assay to recycled water.⁵⁰ By including test strains that are more ecologically representative of the sample waters, Zhao *et al.* concluded that the standard P17 and NOX strains applied in the AOC assay largely underestimate levels in recycled water.⁵⁰ Khan *et al.* have similarly highlighted the need to optimize the BDOC method for recycled water with their modified protocol improving repeatability and precision of results as verified by independent biochemical oxygen demand and chemical oxygen demand measurements.⁵¹

Another negative consequence of NOM in distributed water is that it can accelerate biocorrosion of pipes, which in turn can further stimulate AOC generation.^{39,52,53} BDOC is also believed to play an important role in microbiologically induced corrosion.^{54,55} Recycled water, as an abundant source of sulfate and nitrogen species, is likely to provide a nutrient-rich environment for iron-oxidizing/reducing bacteria^{28,56,57} and sulfate-reducing bacteria³⁴ to thrive in the DS, further raising concerns about the potential for recycled water to accelerate damage to pipe networks.

Redox zones and degradation of water quality

The distribution system can be thought of as a complex reactor with interrelated chemical and biological reactions occurring spatially and temporally as the water passes through the pipes.⁵⁸ The chemistry of treated potable water changes significantly during transport, with deteriorating DWDS water quality documented since the early 1920s.^{59–64} Masters *et al.* illustrated the water distribution system reaction phenomenon by demonstrating the formation of sequential redox zones as a function of water age in simulated DWDS.⁶⁵ Given greater physiochemical and microbial complexity in RWDS, we speculate that they would foster development of even more dramatic reactive zones, as a function of key physical and hydraulic design parameters such as residence time, flow pattern, hydraulic surface to volume ratio, and pipe layout. Consistent with this expectation, studies in lab and field-scale RWDS recently demonstrated elevated microbial activity

Table 4 Proposed threshold values to achieve biostability in drinking water distribution systems

Carbon source	Threshold values	Criteria	Ref.
BDOC	≤ 0.15 mg-C L ⁻¹	Stable BDOC values	210–212
	≤ 0.25 mg-C L ⁻¹		213
	≤ 0.30 mg-C L ⁻¹ at 15 °C		210
	≤ 0.15 mg-C L ⁻¹	No coliform growth	214
AOC	10 µg-C L ⁻¹	No heterotrophic plate count growth	215
	50 µg-C L ⁻¹	No coliform growth	216
	50 µg-C L ⁻¹	No <i>V. cholerae</i> growth	217
	100 µg-C L ⁻¹	No <i>E. Coli</i> growth	49



as indicated by rapid AOC/BDOC consumption, even at the earliest water age, and attenuated organic carbon at higher water ages.²⁹ Similarly in a 15 month monitoring study of RWDSs, the general pattern observed was an initial reactive zone where rapid microbial regrowth and chemical reactions occurred followed by relatively constant microbial and chemical reactivity along the length of the pipes.⁶⁶ Recognizing the reality of reactive zones in distribution systems and more deliberately monitoring them may provide valuable insight into predicting and preempting potential problems resulting from issues related to water chemistry.

Disinfectant residual

The intricate relationship between chlorine-based disinfectants and microbial and chemical stability has been intensely studied in DWDSs. Due to its strong oxidizing power, chlorination is generally the disinfectant of choice for microbial control in drinking water treatment. Chlorination can greatly reduce general bacterial counts and help satisfy drinking water microbial regulations. However, as a strong oxidizing agent, it is also known to interact with reductive species, metals, organic matter and pipe materials and, as a result, significantly impact the downstream water chemistry.⁶⁷ The most widely noted issue with disinfectants is the fragmentation of complex carbon compounds, thus increasing the fraction of biologically available carbon when high concentrations of chlorine are used.^{29,46,47} Given the tendency to use fecal indicators as a benchmark for assessing recycled water quality, it can be tempting for utilities to dose high concentrations of chlorine. Due to a higher chlorine demand typical of recycled water, disinfectant residual may be rapidly lost, leaving the rest of the RWDS vulnerable to microbial instability.⁶⁶ Also important to note is that there is growing concern regarding the efficacy of chlorine-based disinfectants against emerging resistant pathogens, which might be more abundant in recycled water than traditional potable water.^{68–71} The potential for indiscriminate use of disinfectants to inadvertently select disinfectant-resistant bacteria in the RWDS is worthy of exploring in future research. Further, the ability of bacteria to repair and recover in the distribution system following the shock of ultraviolet irradiation (UV), chlorine, or other disinfectant should be considered, as exemplified by recovery of viable but non-culturable bacteria.^{72–74}

Another issue worthy of consideration is the potential for enhanced DBP formation in recycled waters.^{75,76} In a study comparing DBP formation between wastewater effluent and surface water, Sirivedhin and Gray found that effluent-derived organic matter stimulated formation of higher proportions of brominated DBPs.⁷⁷ Nurizzo *et al.* evaluated the DBP formation potential with various disinfection agents and concluded that hypochlorite yielded the greatest total trihalomethanes, exceeding the Italian regulation for agricultural reuse, even when starting with high quality recycled water.⁷⁸ While DBPs tend to be ignored in recycled waters, particularly for non-potable applications, it is important to recognize

that inhalation is also a relevant exposure route to consider, with one model characterizing the inhalation exposure to trihalomethanes of irrigation workers using recycled water suggesting that there was a 13% risk of exceeding acceptable exposure levels for cancer risk.⁷⁹ The DBP issue illustrates that there can be tradeoffs between microbial control and chemical risks and that clearer guidance and alternative approaches are needed for recycled water to avoid negative consequences of blindly over-chlorinating.

Chronic contaminants

The exposome highlights the importance of considering exposures over the course of one's lifetime, and thus, chronic contaminants are an important hazard worthy of consideration during risk assessment of recycled water. WWTPs are generally not designed with the intention of removing micro-constituents, such as pharmaceuticals and personal care products, recalcitrant organic compounds, heavy metals, nanomaterials, and industrial agricultural additives.^{80–84} Jelic *et al.* was able to detect 29 pharmaceutical products in the final effluent of one WWTP, *versus* 32 in the influent.⁸³ Even when discharged at micro-concentrations, up to hundreds of nanograms per liter of targeted micropollutants can still be consistently detected in receiving water bodies and levels can accumulate.⁸⁰ In a study that monitored 15 different WWTPs generating recycled water for groundwater recharge, detectable levels of all 20 most commonly used antibiotics were still found at elevated concentrations of 212–4035 ng L⁻¹ in recycled water and 19–1,270 ng L⁻¹ in groundwater.⁸⁵ Several studies have also observed seasonal patterns of higher discharge of pharmaceuticals and personal care products in wastewater effluent during low flow and less during high flow periods.^{86,87}

While increasing research attention and regulatory efforts have been devoted to understanding prevalence of non-conventional chemical constituents in WWTPs and in receiving environments,⁸⁸ studies specifically focusing on characterization and risk assessment of emerging chemical constituents of concern in the context of recycled water applications are limited.^{89–91} Advanced oxidation processes are particularly promising for removal of these pharmaceuticals and other organic compounds (Table 5).^{92,93} Negative ecological effects of chemical constituents on the aquatic environment have received much attention.^{94,95} Although a multi-barrier approach consisting of sequential treatment processes has promise, questions remain regarding the ideal treatment for various contaminant types and reasonable end point concentrations that are protective of human and ecological health. Given the diverse applications of recycled water, relevant, accurate and comprehensive risk models are needed considerate of the various environmental spheres of influence. Wastewater effluent discharged to surface water has resulted in detection of emerging pharmaceutical products in 80% of surface water samples.⁹⁶ Thus, the science and practice of distributing recycled water should proceed with a comprehensive



Table 5 Case studies of existing application of advanced treatment processes for intended reuse purposes. Treatment trains rely on use of biological activated carbon (BAC), reverse osmosis (RO), ultrafiltration, and UV

Advanced treatment processes	Intended reuse	Key results	Ref.
Ozone/H ₂ O ₂ + BAC	Piloted indirect potable reuse	• H₂O₂/ozone process demonstrated higher than 90% average removal rate in 21 of 31 targeted trace organic contaminants and hormonal products • BAC unit achieved higher than 95% removal for all targeted contaminant except benzophenone • High degree microbial inactivation • Raised concerns on elevated AOCs and microbial regrowth potential after H₂O₂/ozone treatment and • Fluorescence excitation-emission matrix showed distinctively transformed organic matter footprints after treatment	218
Standalone BAC	DOC and nitrogen removal	• Diminishing DOC removal rate after breakthrough is reached • More than 50% of total nitrogen removal rate	219
Ozone/peroxide + RO	General reuse applications	• Ozone and ozone/peroxide showed similar trace organic contaminant removal performance, likely due to inherently high hydroxyl radicals in wastewater effluent • Formation of up to 48 ng L⁻¹ NDMA is observed in wastewater effluent ozone systems, raising concern for future reuse applications	220
Ultrafiltration + RO + UV	Groundwater recharge	• 13 out of 291 targeted compounds are detected in post-UV and post-RO water • Calculated risk quotient for detected chemicals indicates safe reuse	89

approach to understanding of the fate and impacts of these emerging contaminants in relevant environments.

ARGs, OPs, and other emerging microbial concerns

It is important to recognize the complexity of RWDSs as an ecological habitat and that microbial concerns reach beyond traditional indicator organism paradigms. Here we consider these emerging microbial concerns within a comprehensive microbiome/exposome framework. Several recent studies have utilized DNA sequencing to provide new insight into the composition of the drinking water microbiome, but few have attempted to characterize the recycled water microbiome. Recycled water, and even potable water, both represent surprisingly complex microbial niches, housing a vast array of microbial species about which little is known. Normal fecal indicator bacterial monitoring fails to provide information about the broader microbiome, particularly with respect to oligotrophic organisms residing in distribution systems. Thus, a more holistic approach for characterizing water quality is needed to accurately describe the water quality at the point of use. Here we elaborate on microbial aspects of the exposome that are generally unrecognized in the regulatory landscape and are particularly relevant to RWDSs. While occurrence of fecal-associated pathogens is also of importance in recycled water systems, we have limited the scope of this review to emerging microbial concerns.

Epidemiological studies examining associations between recycled water exposure and disease have been limited and are crucial to identifying potential for disease transmission, determining suitability for public use, and informing effective risk mitigation strategies. For example, Durand and Schwebach did not find an association of gastrointestinal illness when irrigating public parks with non-potable recycled water *versus* potable water (6% *versus* 7% of park users

reporting symptoms associated with recycled wastewater irrigation *versus* potable water irrigation, respectively), though wet grass conditions during park usage were associated with an increased rate of illness.⁹⁷ A study of food crop irrigation with recycled water over a five year period found no undesirable consequences to the quality of vegetables or soil, thus exposure restrictions for farm workers were not deemed necessary.⁹⁸ In one study conducted in the U.S., even irrigation using trickling filter effluent wastewater was not associated with an increased rate of infection of rotavirus for residents of surrounding areas.⁹⁹ A study that examined occurrence of methicillin resistant *Staphylococcus aureus* (MRSA) in spray irrigation workers using recycled water did not find the presence of the resistant organism in nasal swabs from any workers tested, though the odds of carrying a non-resistant strain of the organism were slightly higher among spray irrigation workers than among office workers.¹⁹ While isolated reports of disease stemming from exposure to recycled water are helpful, rigorous, long-term epidemiological studies are needed to more precisely determine sources of disease and accurately characterize risk and to address emerging concerns.

Of rising interest is the influence of the distinct physiochemical nature of recycled water on the regrowth or attenuation of emerging pathogens and contaminants, particularly considering exposures relevant to non-conventional water reuse applications. Especially when organic carbon is no longer a growth-limiting resource, conventional fecal bacterial indicators are likely to be even less relevant to shifts in microbial ecology during distribution and the associated health risk. Efforts are underway to recognize the importance of microbial ecological interactions in distribution systems and the potential to harness them to foster a distribution system that favors the growth of non-pathogenic bacteria over pathogenic ones.⁶³ For example, Egli has identified the survival and growth strategies of various microbes in low-nutrient and stressed environments and competition between



pathogens and the indigenous microbiota.¹⁰⁰ A strategy of capitalizing upon specific ecological interactions, such as nutritional competition, antagonist growth, and symbiotic relationships for improved water quality and human health has been previously proposed for drinking water.¹⁰¹ This presents a potentially transformative and highly relevant approach for guiding RWDS management.

Opportunistic pathogens

RWDSs offer several unique characteristics that make them particularly well-suited for supporting regrowth of OPs. OPs in DWDSs are thought to be the primary source of waterborne disease in developed countries, including the U.S.^{102,103} Unlike most fecal pathogens, OPs do not typically impact the gastrointestinal system but rather they infect *via* alternative routes. To name a few, *Legionella pneumophila*, *Acinetobacter baumannii*, *Mycobacterium avium*, *Burkholderia pseudomallei*, *Stenotrophomonas maltophilia*, and *Aspergillus fumigatus* can infect hosts' lungs *via* inhalation;^{104–106} *S. aureus* infects *via* broken skin or mucus membranes;¹⁰⁷ *Pseudomonas aeruginosa* can infect hosts *via* the bloodstream, eyes, ears, skin, or lungs;¹⁰⁸ and *Acanthamoeba* spp. can cause infection of the eyes or central nervous system when inhaled or upon penetration of skin lesions.¹⁰⁹ These alternative routes of infection make OPs of particular interest for recycled water, where exposure routes other than simple ingestion are more relevant. Inhalation of aerosols from cooling towers or spray irrigation and dermal contact with irrigated surfaces, are important routes of exposure that should be accounted for when considering risks associated with OPs in recycled water.

OPs possess several distinct properties that make them particularly well suited for growth in RWDSs (Fig. 3). OPs tend to be resistant to disinfection, ranging from 21–658 times as resistant to chlorine as *Escherichia coli*, as in the cases of *P. aeruginosa* and *A. baumannii*, respectively.¹¹⁰ Many OPs are also resistant to phagocytosis by amoebae, becoming enclosed within an amoebic cyst, where they can be further protected from disinfectants and other harsh environmental conditions. Biofilms, where OPs tend to reside, offer protection from similar environmental assaults, in addition to acidic and alkaline conditions and shear force from high flow velocities.^{101,110} OPs also tend to grow at low organic carbon concentrations, which is pertinent to both DWDSs and RWDSs.¹¹⁰ Stagnation is a notorious risk factor for OP outbreak, and is common in RWDSs due to intermittent demand and seasonal shutdown.¹¹¹

Antibiotic resistance genes

Antibiotic resistance among human pathogens is a major public health concern. In the U.S., the Centers for Disease Control has estimated that antibiotic resistant bacteria cause at least two million infections and 23 000 deaths each year.¹¹² ARGs are now well-known to be elevated in human-contaminated surface waters,^{113–116} however with respect to human pathogens, specifically, there is reasonable evidence that they can gain ARGs from environmental bacteria *via* horizontal gene transfer (Fig. 3).^{117,118} Therefore, all members of the microbiome carrying ARGs are potentially of concern, particularly those that are common in human pathogens. In addition to the possibility of infection by antibiotic resistant

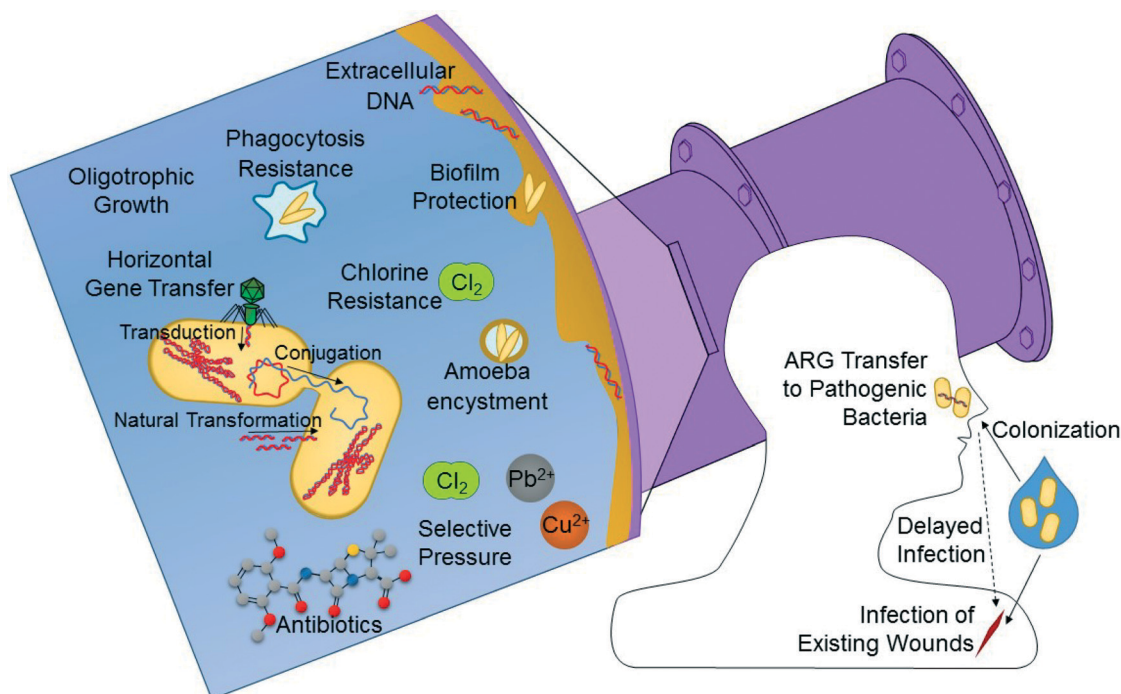
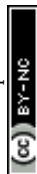


Fig. 3 Processes by which antibiotic resistant bacteria and opportunistic pathogens (OPs) can re-grow in RWDSs and relevant exposure routes.



bacteria upon exposure, human hosts may also become colonized and infected later.^{119–121} Similarly, it is possible that horizontal gene transfer may occur from colonized non-pathogenic bacteria to pathogenic ones, leading to antibiotic resistant infection. Thus, infection by antibiotic resistant bacteria may occur at a time and place separate from that of the initial exposure, which complicates traditional dose–response risk assumptions. ARGs or bacteria expressing ARGs corresponding to resistance to aminoglycoside, beta-lactam, chloramphenicol, fluoroquinolone, lincomycide, linezolid, lipopeptide, macrolide, sulfonamide, tetracyclines, and vancomycin antibiotics have been previously identified in recycled water or environments directly impacted by irrigation, infiltration, or groundwater recharge using recycled water.^{122–128} Since antibiotic resistance is a natural phenomenon inherent among many bacteria, studies that compare these abundances to relevant control environments, such as corresponding potable water or environments unimpacted by recycled water are of particular value. While the nature of reusing human wastewater means that prior to treatment, human pathogens or other bacteria carrying ARGs will be enriched compared to other source waters, multiple studies have demonstrated that ARGs are often not removed during treatment, and in some cases, are even amplified.^{129–132} Additionally, a study by Fahrenfeld *et al.* found that ARGs may also increase during distribution of recycled water as a broader range of monitored ARGs were present in point of use samples than in samples leaving the treatment plant.¹²²

Various features of recycled water potentially make it a prime medium for the growth of antibiotic resistant bacteria and propagation of their associated ARGs during distribution. In particular, residual antibiotics that escape removal during treatment can exert selective pressure and encourage persistence of ARG-carrying bacteria. Though antibiotics will likely be found in recycled water at sub-lethal concentrations, this low level exposure has actually been shown to encourage the persistence of bacteria that carry ARGs *via* several mechanisms.^{133–135} Gullberg *et al.* found that bacteria maintained plasmids carrying beta-lactam resistance genes even at concentrations of antibiotics and heavy metals nearly 140 times below the compound's minimum inhibitory concentration.¹³⁶ Other studies have also demonstrated that sub-lethal antibiotics can stimulate propagation of ARGs by activating horizontal gene transfer.^{137–142} Prudhomme *et al.* demonstrated that intermediate concentrations of streptomycin induced genetic transformation in *Streptococcus pneumoniae*.¹⁴⁰ Beaber *et al.* demonstrated that 'SOS response' among *Vibrio cholerae* induced by the presence of ciprofloxacin enhances transfer of resistance genes *via* conjugation.¹⁴² Low levels of antibiotics or other selective agents also act to encourage adaptive evolution including development of resistance mutants.¹³⁵

Antibiotics are not the only antimicrobials with potential to select for ARGs in recycled water systems. Heavy metals, such as copper and iron (which are commonly used in distribution systems), have long been suspected to select for ARGs

in a variety of environments.¹⁴³ Metal-driven selection of ARGs is also of concern due to the presence of various heavy metals capable of ARG selection common in many wastewaters, such as copper, zinc, nickel, mercury, and even nanosilver.^{144,145} Disinfectants have also been known to select for ARGs.^{146–148} Following chlorination, *E. coli* carrying the *tetA* tetracycline resistance gene were found to be even more tolerant to tetracycline than non-chlorinated *E. coli*.¹⁴⁸ Chlorination has also been reported to concentrate a variety of ARGs in potable water.¹⁴⁷

In addition to increasing ARGs *via* mutations, natural selection, and horizontal gene transfer, presence of residual antibiotics can enhance biofilm formation.¹⁴⁹ Studies of *Staphylococcus aureus*, *E. coli*, and *P. aeruginosa* have all indicated that sub-inhibitory concentrations of various antibiotics induce biofilm formation.^{150–152} Extensive biofilm formation provides a fertile environment for the transfer of ARGs *via* horizontal gene transfer. Dense microbial communities existing in biofilms with extensive cell to cell contact facilitate transfer *via* conjugation.^{153,154} Notably, a key component of biofilms, extracellular polymeric substances, is partially comprised of DNA expelled from cells.¹⁵⁵ This may provide a reservoir of free DNA-based ARGs, which have been shown to be available for uptake into cells *via* transformation.^{156,157} Biofilms themselves offer protection from antibiotics or other antimicrobial agents *via* the principle of collective resistance, where cells are physically shielded from exposure to the antimicrobial.¹⁵⁸

While transmission of antibiotic resistant bacteria is an acute public health threat, the possibility that water reuse may exacerbate the overall spread of antibiotic resistance has been suggested.¹⁵⁹ The water cycle as a whole has recently been subject to scrutiny as a potentially important, yet understudied, route for the spread of antibiotic resistance.^{160–162} Given the gravity of the antibiotic resistance problem and several lines of reasoning that water reuse can contribute to its spread, additional research is urgently needed to determine whether consideration of antibiotic resistance should be of central concern to comprehensive long-term risk management strategies.

Viruses

Though removal of viruses in recycled water is of great importance, the presence of viruses in recycled water is rarely monitored. Treatment goals and regulations regarding virus removal are typically presented in the form of expected log-removal achieved through treatment such as disinfection, largely due to the analytical difficulty of direct virus detection.⁶ Low recovery rates, complex and time-consuming laboratory culture procedures, slow turn-around time for culture results, and inability of molecular techniques to differentiate viable from non-viable viruses are major challenges. Problems with the indirect monitoring paradigm may arise, however, because viruses may be resistant to some modes of disinfection. For example, adenoviruses are known to resist ultraviolet irradiation.⁶



A recent study of the viral metagenome (*i.e.*, total DNA extracted from viral component) revealed approximately 10^8 virus-like particles (VLP) per mL in non-potable recycled water, 1000 times more than that measured in potable water.¹⁶³ Further, genetic markers corresponding to viruses targeting eukaryotes were non-detectable in potable water, while two percent of the viruses in recycled water corresponded to eukaryotic hosts. This is logical, indicating that recycled water is more susceptible to carrying viruses associated with humans than traditional potable water.

Bacteriophages, which represent the vast majority of viruses in both potable and recycled waters, have their own relevance to human health as they act as agents of transfer for ARGs among bacteria *via* transduction. Bacteriophages have been largely neglected as constituents in potable and recycled water, though they have been found to be highly abundant in both raw wastewater (10^8 VLP mL⁻¹) and in potable water (10^5 – 10^6 VLP mL⁻¹).¹⁶³ Though transduction is generally considered a rare transfer event, occurring only once in every 10^7 – 10^9 phage infections,¹⁶⁴ the sheer abundance of bacteriophages documented in wastewater and potable water suggests that it is likely a significant phenomenon in recycled water.

Amoebae

FLA are of growing concern in drinking water plumbing. Many FLA, such as *Acanthamoeba* spp. and *Vermameoba* spp., graze on bacterial biofilms, and in doing so, can serve as an important vector for amplifying and disseminating OPs.¹⁶⁵ For example, *Legionella*, *Mycobacterium*, and *Pseudomonas* spp. can amplify within FLA when grazed upon, which enhances their dissemination and virulence.¹⁶⁵

FLA themselves can sometimes be pathogenic, as is the case with keratitis or primary amoebic meningoencephalitis (PAM).¹⁶⁵ Similar to other OPs, non-ingestion routes of exposure are important for pathogenic FLA. PAM is contracted when *N. fowleri* is forced into the nasal cavity and migrates to the brain, while keratitis occurs when pathogenic *Acanthamoeba* spp. infect the eye.¹⁶⁶ Such exposures have been documented both in recreational and drinking water.^{109,167,168} However, relevant to recycled water, inhalation is under investigation as a primary transmission method.^{168,169}

The design of RWDSs can instigate the growth of biofilm, providing a reservoir for FLA and an environment to promote interactions with amoeba-resisting bacteria. Recycled waters have complex microbial communities and high availability of nutrients and other organic matter, creating optimal conditions for biofilm establishment.¹⁷⁰ Recent studies have also shown increased chlorine resistance of FLA in the presence of naturally established biofilm^{171,172} and even non-biofilm *Vermameoba* spp. have been observed to resist chlorination.¹⁷³

The relationship between *L. pneumophila* and FLA has been the most closely studied. Resistance to amoebae can provide protection from disinfection, competition, environ-

mental stress and predation for *L. pneumophila*.^{174–176} Additionally, different FLA have been shown to survive a wide range of temperatures, from 10–45 °C, with some cases indicating survival near 0 °C, potentially allowing for protection of *Legionella* spp. and other amoeba-resisting bacteria during winter, while amoeba are encysted.^{165,177–179} Little is still known about the diversity and abundance of amoebae and their interactions with amoeba-resisting bacteria in drinking water, let alone recycled water. Gaining a better understanding of the interactions between these microorganisms and the ways in which this may aid the growth of pathogenic bacteria is essential for better understanding the exposome associated with recycled water.

Algae

Algae are a common nuisance in recycled water systems. Though algal growth frequently occurs in systems that use open storage rather than in distribution system pipes, algal cells can persist throughout distribution systems where they have been found to correlate with AOC and BDOC.²⁹ Decaying algal cells can even be a source of BDOC, contributing to the regrowth of other microbial constituents. Increased regrowth resulting from organic carbon made available from decaying algal cells has also been linked with a loss of oxygen and dissipation of chlorine residual.²⁹ Elevated concentrations of algae may carry potential for the production of harmful algal toxins. Cyanobacteria toxins have been linked to liver damage, neurotoxicity, gastroenteritis, pneumonia, and even death.¹⁸⁰ Though these symptoms primarily arise from ingestion of the toxins, skin irritations and allergic reactions have been noted following dermal contact with cyanobacteria toxins and respiratory disease has been documented following suspected inhalation of the toxins.¹⁸¹ In cases where non-potable reuse occurs, these problems may be particularly challenging to identify as taste and odor complaints from consumers will be unlikely.

Conclusion

Given the increasing trend of water reuse across the globe, it is important that all aspects, including end-users, treatment plant management and operation, regulation, and public health protection are taken into consideration in the planning and implementation of water reuse risk management strategies. In this paper, we summarized the inherently different biochemistry of recycled water in the distribution system as a function of various usage and operational factors. We also discussed acute and long-term risks from the chemical and microbial contaminants that may result from multi-dimensional usage routes and the associated exposure risks associated with various end use of the recycled waters.

Increased awareness of traditionally underappreciated routes of exposure is key to the safe use of recycled water. The history of drinking water epidemiology provides numerous examples of infection *via* atypical routes of exposure. For example, in 2015 an outbreak of Legionnaire's disease that



killed 12 people in New York City was linked to infection *via* aerosolized bacteria from cooling towers sourced from potable drinking water.¹⁸² *N. fowleri* infection from drinking water has occurred from use for nasal irrigation and in children *via* bathing or playing on an outdoor water slide.¹⁶⁸ Prevention of infection as a result of unintended exposures with recycled water requires proactive action when planning for treatment and distribution of recycled water. One key “critical control point” that must be considered in this planning is the distribution system to avoid degradation of water quality during distribution. Treating recycled water to remove nutrients and achieve biostability is one promising approach to help ensure safe water at the point of use, but additional treatment at the point of use may also be necessary in some cases, as both the physiochemical and microbial water quality change significantly during distribution and in premise plumbing systems. A key research gap exists regarding the most effective approaches for achieving biostability of recycled water during distribution. Specific and cost-effective engineering controls for nutrient recycling and limiting regrowth during distribution must be identified for respective intended uses. Identification of emerging chronic contaminants and microbial contaminants is also important in minimizing potentially harmful exposures. Rigorous studies that examine the health implications of non-traditional routes of exposure are quite limited and are challenging to design given the lack of available knowledge about infectious doses (particularly based on non-oral routes of exposure), magnitude of exposure *via* non-traditional routes, and concentrations of emerging contaminants that are typical of recycled water. In addition, virulence and individual susceptibility varies widely for many of the microbial constituents discussed, making it important to consider exposure of immunocompromised populations when assessing risk. In addition to these research gaps, development of quantitative microbiological risk assessments (QMRA) would be extremely valuable for assessing the risk associated with the presence OPs, ARGs, FLA, and viruses in recycled water. Epidemiological studies are also critical for linking actual human illness and associated microbial sources with recycled water. Finally, research is needed to tailor treatment processes to serve specific intended end uses (*e.g.* Table 1), along with addressing emerging concerns identified here, while also developing best management practices for distribution systems and premise plumbing for preventing re-growth and deterioration of water quality in RWDSSs.

While overarching regulations that consider the comprehensive implications and scope of water recycling are currently lacking in many places, there are also practical lessons we can learn from international leaders on adopting a comprehensive risk management approach towards water reuse, notably the Australian Guidelines for Water Recycling, the WHO's framework of WSP, and the HACCP paradigm. Complementing these strategies with a holistic approach focused on the human exposome creates a framework in which consideration of user exposures drives establishment of water

quality standards. Though a regulatory framework that addresses the exposure risks and potential for regrowth associated with use of recycled water is an ideal long term goal, an interim approach of more basic best management practices, as suggested by Jjemba *et al.*, are a reasonable starting place to enable municipalities utilizing recycled water to proactively act to limit bacterial regrowth and preserve water quality at the point of use.²⁹ Best management practices should continually be revised as knowledge gaps are addressed, to ensure that the most meaningful water quality indicators are targeted.

Adoption of the human exposome paradigm aims to ensure comprehensive understanding of the risks and uncertainties regarding alternative recycled water sources. Enhanced knowledge could provide critical guidance on safe management and inform much-needed regulations as the use of recycled water expands. However, while this exposome approach highlights the multi-dimensional risks and uncertainties regarding use of recycled water, it also must be recognized that water reuse plays an integral role in addressing the grand challenge of water scarcity. It is estimated that a third of the world's population is currently living with moderate to high levels of water stress¹⁸³ and approximately 50% will suffer water shortages by 2025.¹⁸⁴ Implementing water reuse projects is imperative to meet water needs in drought-stricken areas, despite potential risks and concerns. As estimated by Brown, current groundwater sources, serving more than half of the world's population, are largely overdrafted.¹⁸⁵ Lack of new alternative water supplies, compounded with increasing water demand, would further intensify water scarcity stress. Schoreder *et al.* has estimated that the potential benefits of reuse could offset water supply for a community of 1 million people by 75 million gallons per day.¹⁸⁶ In cases where water scarcity lends to the likelihood of *de facto* reuse, then it is better to have intentional reuse guided by best management practices to minimize risks.

Equally important to the exposure risk from recycled water is lack of access to traditional potable water sources and poor water quality due to degraded source water. Globally, there are over five million deaths associated with poor water quality every year.¹⁸⁷ Achieving an environmentally sustainable and socially beneficially water demand management plan requires proactive evaluation of the highest priority needs and identification of the key drivers and barriers to the implementation of water recycling projects. Positive associations between information availability and the acceptance of water reuse have been noted.^{188–190} As end users become more educated about this alternative water source, their willingness to use recycled water increases.¹⁹¹ Nonetheless, comprehensively addressing all possible public health concerns will be an essential pillar to advancing water sustainability.

Glossary

AOC	Assimilable organic carbon
ARG	Antibiotic resistance gene



BAC	Biological activated carbon
BDOD	Biodegradable dissolved organic carbon
CAS	Conventional activated sludge
CFU	Colony forming unit
DBP	Disinfection byproduct
DOC	Dissolved organic carbon
DOM	Dissolved organic matter
DON	Dissolved organic nitrogen
DPR	Direct potable reuse
DWDS	Drinking water distribution system
FLA	Free-living amoebae
HACCP	Hazard analysis and critical control point
MRSA	Methicillin resistant <i>Staphylococcus aureus</i>
NOM	Natural organic matter
OP	Opportunistic pathogen
PAM	Primary amoebic meningoencephalitis
QMRA	Quantitative microbiological risk assessment
RO	Reverse osmosis
RWDS	Recycled water distribution system
THAAFP	Total haloacetic acid formation potential
TTHM	Total trihalomethanes
TTHMFP	Total trihalomethane formation potential
UV	Ultraviolet irradiation
VLP	Virus-like particle
WHO	World Health Organization
WSP	Water safety plan
WWTP	Wastewater treatment plant

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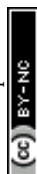
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References

- 1 J. Anderson, *Water Sci. Technol.: Water Supply*, 2003, 3, 1–10.
- 2 D. Goodwin, M. Raffin, P. Jeffrey and H. M. Smith, *Environ. Sci.: Water Res. Technol.*, 2015, 1, 709–722.
- 3 D. Gerrity, B. Pecson, R. S. Trussell and R. R. Trussell, *J. Water Supply: Res. Technol.-AQUA*, 2013, 62, 321–338.
- 4 P. Jjemba, W. Johnson, Z. Bukhari and M. LeChevallier, *J. Water Reuse Desalin.*, 2014, 4, 209.
- 5 F. Digiano and M. Aitkin, Workshop on Water Reuse and Dual Systems for Potable and Non-Potable Water Supply. U.S. Environmental Protection Agency, Final Report for Cooperative Agreement: 83422201-0, 2010, 51 pp.
- 6 USEPA, *Guidelines for Water Reuse*, United States Environmental Protection Agency, 2012.
- 7 NRMCC EPHC & AHMC, Australian Guidelines for Water Recycling: Managing Health and Environmental Risks (Phase 1), Natural Resource Ministerial Management Council, Environment Protection and Heritage Council and Australian Health Ministers, 2006.
- 8 J. Lahnsteiner and G. Lempert, *Water Sci. Technol.*, 2007, 55, 441–448.
- 9 M. Agulló-Barceló, F. Oliva and F. Lucena, *Environ. Sci. Pollut. Res.*, 2013, 20, 4448–4454.
- 10 V. J. Harwood, A. D. Levine, T. M. Scott, V. Chivukula, J. Lukasik, S. R. Farrah and J. B. Rose, *Appl. Environ. Microbiol.*, 2005, 71, 3163–3170.
- 11 J. Swierc, D. Page, J. Van Leeuwen and P. J. Dillon, *CSIRO L. Water Tech. Rep.*, 2005, vol. 20.
- 12 J. Bartram, L. Corrales, A. Davison, D. Deere, D. Drury, B. Gordon, G. Howard, A. Rinehold and M. Stevens, *Water safety plan manual: step-by-step risk management for drinking-water suppliers*, World Health Organization, Geneva, 2009.
- 13 G. W. Miller and D. P. Jones, *Toxicol. Sci.*, 2014, 137, 1–2.
- 14 C. P. Wild, *Cancer Epidemiol., Biomarkers Prev.*, 2005, 14, 1847–1850.
- 15 J. Rice, A. Wutich and P. Westerhoff, *Environ. Sci. Technol.*, 2013, 47, 11099–11105.
- 16 M. Lafforgue and V. Lenouvel, *Environ. Sci.: Water Res. Technol.*, 2015, 1, 622–631.
- 17 J. Sutherland, E. Steinle-Darling, A. Salveson and E. Mancha, *presented in part at Water Quality Technology Conference*, Fort Worth, July, 2015.
- 18 D. G. Addiss, J. P. Davis, M. LaVenture, P. J. Wand, M. A. Hutchinson and R. M. McKinney, *Am. J. Epidemiol.*, 1989, 130, 557–568.
- 19 R. E. Rosenberg Goldstein, S. a Micallef, S. G. Gibbs, X. He, A. George, A. A. R. Sapkota, S. W. Joseph and A. A. R. Sapkota, *Int. J. Environ. Res. Public Health*, 2014, 11, 4340–4355.
- 20 F. Brissaud, E. Blin, S. Hemous and L. Garrelly, *Water Sci. Technol.*, 2008, 57, 781–787.
- 21 K. E. Borboudaki, N. V. Paranychianakis and K. P. Tsagarakis, *Environ. Manage.*, 2005, 36, 610–623.
- 22 C. P. Gerba, C. Wallis and J. L. Melnick, *Appl. Microbiol.*, 1975, 30, 229–237.
- 23 W. G. Hlady, R. C. Mullen, C. S. Mintz, B. G. Shelton, R. S. Hopkins and G. L. Daikos, *Am. J. Epidemiol.*, 1993, 138, 555–562.
- 24 T. E. Haupt, R. T. Heffernan, J. J. Kazmierczak, H. Nehls-Lowe, B. Rheineck, C. Powell, K. K. Leonhardt, A. S. Chitnis and J. P. Davis, *Infect. Control Hosp. Epidemiol.*, 2012, 33, 185–191.
- 25 T. N. Palmore, F. Stock, M. White, M. Bordner, A. Michelin, J. E. Bennett, P. R. Murray and D. K. Henderson, *Infect Control Hosp Epidemiol*, 2009, 30, 764–768.
- 26 J. B. Rose and C. S. Clark, *Crit. Rev. Environ. Control*, 1986, 16, 231–256.
- 27 ASCE 2013 Report Card for America's Infrastructure, <http://infrastructurereportcard.org/a/#p/home>, (accessed December 2015).
- 28 H. Wang, C. Hu, X. Hu, M. Yang and J. Qu, *Water Res.*, 2012, 46, 1070–1078.



- 29 P. K. Jjemba, L. a. Weinrich, W. Cheng, E. Giraldo and M. W. LeChevallier, *Appl. Environ. Microbiol.*, 2010, **76**, 4169–4178.
- 30 P. Sarin, V. L. Snoeyink, J. Bebee, K. K. Jim, M. A. Beckett, W. M. Kriven and J. A. Clement, *Water Res.*, 2004, **38**, 1259–1269.
- 31 USEPA, <http://archive.epa.gov/region03/dclead/web/html/index-3.html> (accessed January 2016).
- 32 Mlive, http://www.mlive.com/news/flint/index.ssf/2015/09/new_testing_shows_flint_water.html (accessed January 2016).
- 33 MATCOR, <http://www.matcor.com/water-line-break-corrosion-a-headache-for-tucson-water/> (accessed January 2016).
- 34 F. Yang, B. Shi, Y. Bai, H. Sun, D. a. Lytle and D. Wang, *Water Res.*, 2014, **59**, 46–57.
- 35 M. Tang, S. Triantafyllidou and M. A. Edwards, *Environ. Sci. Technol.*, 2015, **49**, 8697–8703.
- 36 M. V. Story and C. E. Kaucner, *Water Quality Research Australia: Research Report 79*, 2009, p. 122.
- 37 H. Rufenacht and H. Guibentif, *J. Water Supply: Res. Technol.-AQUA*, 1997, **46**, 196–201.
- 38 R. M. Willis, R. A. Stewart, P. R. Williams, C. H. Hacker, S. C. Emmonds and G. Capati, *Desalination*, 2011, **272**, 201–211.
- 39 J. Nawrocki, U. Raczky-Stanislawiak, J. Świetlik, A. Olejnik and M. J. Sroka, *Water Res.*, 2010, **44**, 1863–1872.
- 40 P. Ji, J. Parks, M. A. Edwards and A. Pruden, *PLoS One*, 2015, **10**, e0141087.
- 41 W. J. Rhoads, A. Pruden and M. A. Edwards, *Environ. Sci.: Water Res. Technol.*, 2016, **2**, 164–173.
- 42 H.-Y. Hu, Y. Du, Q.-Y. Wu, X. Zhao, X. Tang and Z. Chen, *Sci. Total Environ.*, 2016, 551–552, 133–142.
- 43 H. K. Shon, S. Vigneswaran and S. A. Snyder, *Crit. Rev. Environ. Sci. Technol.*, 2006, **36**, 327–374.
- 44 B. Rittman and V. Snoeyink, *J. - Am. Water Works Assoc.*, 1984, **76**, 106–114.
- 45 P. W. J. J. van der Wielen and D. van der Kooij, *Water Res.*, 2010, **44**, 4860–4867.
- 46 P. Thayanukul, F. Kurisu, I. Kasuga and H. Furumai, *Water Res.*, 2013, **47**, 225–232.
- 47 L. A. Weinrich, P. K. Jjemba, E. Giraldo and M. W. LeChevallier, *Water Res.*, 2010, **44**, 5367–5375.
- 48 M. Karim and M. W. LeChevallier, *Microbiological quality of reuse water*, Internal Report, Voorhees, NJ, 2005.
- 49 M. W. LeChevallier, N. J. Welsh and D. B. Smith, *Appl. Environ. Microbiol.*, 1996, **62**, 2201–2221.
- 50 X. Zhao, H. Hu, S. Liu, F. Jiang, X. Shi, M. Li and X. Xu, *Front. Environ. Sci. Eng.*, 2013, **7**, 483–491.
- 51 E. Khan, R. W. Babcock Jr., I. H. Suffet and M. K. Stenstrom, *Water Environ. Res.*, 1998, **70**, 1025–1032.
- 52 L. S. McNeill and M. Edwards, *J. - Am. Water Works Assoc.*, 2014, **93**, 88–100.
- 53 M. Edwards and N. Sprague, *Corros. Sci.*, 2001, **43**, 1–18.
- 54 A. K. Camper, *Int. J. Food Microbiol.*, 2004, **92**, 355–364.
- 55 M. W. LeChevallier, C. D. Lowry, R. G. Lee and D. L. Gibbon, *J. - Am. Water Works Assoc.*, 1993, **85**, 111–123.
- 56 H. Wang, C. Hu, L. Zhang, X. Li, Y. Zhang and M. Yang, *Water Res.*, 2014, **65**, 362–370.
- 57 Y. Zhu, H. Wang, X. Li, C. Hu, M. Yang and J. Qu, *Water Res.*, 2014, **60**, 174–181.
- 58 A. K. Camper and P. Dircks, *Distribution system as reactor*, 1996, <https://www.biofilm.montana.edu/resources/images/industrial-systems-processes/distribution-system-reactor.html>, (accessed January 2016).
- 59 S. T. Powell, *J. - Am. Water Works Assoc.*, 1921, **8**, 522–529.
- 60 A. M. Buswell, *J. - Am. Water Works Assoc.*, 1938, **30**, 1651–1654.
- 61 N. J. Howard, *J. - Am. Water Works Assoc.*, 1940, **32**, 1501–1506.
- 62 J. Hall, A. D. Zaffiro, R. B. Marx, P. C. Kefauver, E. R. Krishnan, R. C. Haught and J. G. Herrmann, *J. - Am. Water Works Assoc.*, 2007, **99**, 66–77.
- 63 D. Berry, C. Xi and L. Raskin, *Curr. Opin. Biotechnol.*, 2006, **17**, 297–302.
- 64 P. Piriou, B. E. Rittmann and J. E. Wooschlager, *Urban Water J.*, 2005, **2**, 69–79.
- 65 S. Masters, H. Wang, A. Pruden and M. A. Edwards, *Water Res.*, 2015, **68**, 140–149.
- 66 O. M. Ajibode, C. Rock, K. Bright, J. E. T. McLain, C. P. Gerba and I. L. Pepper, *J. Water Reuse Desalin.*, 2013, **3**, 185–196.
- 67 M. K. Ramseier, A. Peter, J. Traber and U. von Gunten, *Water Res.*, 2011, **45**, 2002–2010.
- 68 K. A. Steed and J. O. Falkinham, *Appl. Environ. Microbiol.*, 2006, **72**, 4007–4011.
- 69 R. H. Taylor, J. O. Falkinham, C. D. Norton and M. W. LeChevallier, *Appl. Environ. Microbiol.*, 2000, **66**, 1702–1705.
- 70 I. Marchesi, P. Marchegiano, A. Bargellini, S. Cencetti, G. Frezza, M. Miselli and P. Borella, *J. Hosp. Infect.*, 2011, **77**, 47–51.
- 71 M. Dupuy, S. Mazoua, F. Berne, C. Bodet, N. Garrec, P. Herbelin, F. Ménard-Szczebara, S. Oberti, M.-H. Rodier, S. Soreau, F. Wallet and Y. Héchar, *Water Res.*, 2011, **45**, 1087–1094.
- 72 J.-J. Huang, H.-Y. Hu, F. Tang, Y. Li, S.-Q. Lu and Y. Lu, *Water Res.*, 2011, **45**, 2775–2781.
- 73 M. Guo, J. Huang, H. Hu and W. Liu, *Biomed. Environ. Sci.*, 2011, **24**, 400–407.
- 74 J. Süss, S. Volz, U. Obst and T. Schwartz, *Water Res.*, 2009, **43**, 3705–3716.
- 75 R. M. Clark and M. Sivaganesan, *J. Environ. Eng.*, 1998, **124**, 1203–1210.
- 76 S. W. Krasner, *Philos. Trans. R. Soc., A*, 2009, **367**, 4077–4095.
- 77 T. Sirivedhin and K. A. Gray, *Water Res.*, 2005, **39**, 1025–1036.
- 78 C. Nurizzo, M. Antonelli, M. Profaizer and L. Romele, *Desalination*, 2005, **176**, 241–253.
- 79 C. C. Wang, Z. G. Niu and Y. Zhang, *J. Hazard. Mater.*, 2013, **262**, 179–188.
- 80 C. M. Coetsier, S. Spinelli, L. Lin, B. Roig and E. Touraud, *Environ. Int.*, 2009, **35**, 787–792.



- 81 S. Mompelat, B. Le Bot and O. Thomas, *Environ. Int.*, 2009, **35**, 803–814.
- 82 M. Gros, M. Petrović, A. Ginebreda and D. Barceló, *Environ. Int.*, 2010, **36**, 15–26.
- 83 A. Jelic, M. Gros, A. Ginebreda, R. Cespedes-Sánchez, F. Ventura, M. Petrovic and D. Barcelo, *Water Res.*, 2011, **45**, 1165–1176.
- 84 M. Huerta-Fontela, M. T. Galceran and F. Ventura, *Water Res.*, 2011, **45**, 1432–1442.
- 85 Y. Ma, M. Li, M. Wu, Z. Li and X. Liu, *Sci. Total Environ.*, 2015, **518–519**, 498–506.
- 86 D. W. Kolpin, M. Skopec, M. T. Meyer, E. T. Furlong and S. D. Zaugg, *Sci. Total Environ.*, 2004, **328**, 119–130.
- 87 G. A. Loraine and M. E. Pettigrove, *Environ. Sci. Technol.*, 2006, **40**, 687–695.
- 88 J. D. Peterson, R. R. Murphy, Y. Jin, L. Wang, M. B. Nessler and K. Ikehata, *Water Environ. Res.*, 2011, **83**, 1853–1875.
- 89 F. Busetti, M. Ruff and K. L. Linge, *Environ. Sci.: Water Res. Technol.*, 2015, **1**, 659–667.
- 90 S. Weber, S. Khan and J. Hollender, *Desalination*, 2006, **187**, 53–64.
- 91 S. Khan, in *Sustainable Water*, ed. R. Hester and R. Harrison, The Royal Society of Chemistry, London, 2011, ch. 6, pp. 114–137.
- 92 M. M. Huber, S. Canonica, G. Y. Park and U. Von Gunten, *Environ. Sci. Technol.*, 2003, **37**, 1016–1024.
- 93 I. H. Kim, N. Yamashita, Y. Kato and H. Tanaka, *Water Sci. Technol.*, 2009, **59**, 945–955.
- 94 L. H. M. L. M. Santos, A. N. Araújo, A. Fachini, A. Pena, C. Delerue-Matos and M. C. B. S. M. Montenegro, *J. Hazard. Mater.*, 2010, **175**, 45–95.
- 95 C. R. Tyler, S. Jobling and J. P. Sumpter, *Crit. Rev. Toxicol.*, 2008, **28**, 319–361.
- 96 D. W. Kolpin, E. T. Furlong, M. T. Meyer, E. M. Thurman, S. D. Zaugg, L. B. Barber and H. T. Buxton, *Environ. Sci. Technol.*, 2002, **36**, 1202–1211.
- 97 R. Durand and G. Schwebach, *Am. J. Public Health*, 1989, **79**, 1659–1660.
- 98 B. Sheikh, R. P. Cort, W. R. Kirkpatrick, R. S. Jaques and T. Asano, *Res. J. Water Pollut. Control Fed.*, 1990, **62**, 216–226.
- 99 R. L. Ward, D. R. Knowlton, J. Stober, W. Jakubowski, T. Mills, P. Graham and D. E. Camann, *Water Res.*, 1989, **23**, 1503–1509.
- 100 T. Egli, *Water Res.*, 2010, **44**, 4826–4837.
- 101 H. Wang, M. A. Edwards, J. O. Falkinham 3rd and A. Pruden, *Environ. Sci. Technol.*, 2013, **47**, 10117–10128.
- 102 A. Pruden, M. A. Edwards, J. O. Falkinham III, M. Arduino, J. Bird, R. Birdnow, E. Bedard, A. Camper, J. Clancy, E. Hilborn, V. Hill, A. Martin, S. Masters, N. R. Pace, M. Prevost, A. Rosenblatt, W. Rhoads, J. E. Stout and Y. Zhang, *State of the Science and Research Needs for Opportunistic Pathogens in Premise Plumbing: Methodology, microbial ecology, and epidemiology*, Water Research Foundation, Denver, CO, 2013.
- 103 J. M. Brunkard, E. Ailes, V. A. Roberts, V. Hill, E. D. Hilborn, G. F. Craun, A. Rajasingham, A. Kahler, L. Garrison, L. Hicks, J. Carpenter, T. J. Wade, M. J. Beach and J. S. Yoder, *MMWR Morb Mortal Wkly Rep.*, 2008, **57**, 1–38.
- 104 D. W. Fraser, T. R. Tsai, W. Orenstein and W. E. Parkin, *N. Engl. J. Med.*, 1989, **297**, 1189–1197.
- 105 A. Y. Peleg, H. Seifert and D. L. Paterson, *Clin. Microbiol. Rev.*, 2008, **21**, 538–582.
- 106 C. R. Horsburgh, *N. Engl. J. Med.*, 1991, **324**, 1332–1338.
- 107 F. D. Lowy, *N. Engl. J. Med.*, 1998, **339**, 520–532.
- 108 G. P. Bodey, R. Bolivar, V. Fainstein and L. Jadeja, *Rev. Infect. Dis.*, 1983, **5**, 279–313.
- 109 F. Marciano-Cabral and G. Cabral, *Clin. Microbiol. Rev.*, 2003, **16**, 273–307.
- 110 J. Falkinham, *Int. J. Environ. Res. Public Health*, 2015, **12**, 4533–4545.
- 111 H. Wang, M. Edwards, J. O. Falkinham and A. Pruden, *Appl. Environ. Microbiol.*, 2012, **78**, 6285–6294.
- 112 US CDC, *Antibiotic Resistance Threats in the United States*, 2013.
- 113 A. Pruden, M. Arabi and H. N. Storteboom, *Environ. Sci. Technol.*, 2012, **46**, 11541–11549.
- 114 D. W. Graham, S. Olivares-Rieumont, C. W. Knapp, L. Lima, D. Werner and E. Bowen, *Environ. Sci. Technol.*, 2011, **45**, 418–424.
- 115 G. C. A. Amos, P. M. Hawkey, W. H. Gaze and E. M. Wellington, *J. Antimicrob. Chemother.*, 2014, **69**, 1785–1791.
- 116 E. Marti, J. Jofre and J. L. Balcazar, *PLoS One*, 2013, **8**, e78906.
- 117 R. Benveniste and J. Davies, *Proc. Natl. Acad. Sci. U. S. A.*, 1973, **70**, 2276–2280.
- 118 K. J. Forsberg, A. Reyes, B. Wang, E. M. Selleck, M. O. A. Sommer and G. Dantas, *Science*, 2012, **337**, 1107–1111.
- 119 K. A. Davis, J. J. Stewart, H. K. Crouch, C. E. Florez and D. R. Hoshpenthal, *Clin. Infect. Dis.*, 2004, **39**, 776–782.
- 120 C. von Eiff, K. Becker, K. Machka, H. Stammer and G. Peters, *N. Engl. J. Med.*, 2001, **344**, 11–16.
- 121 D. Bogaert, R. De Groot and P. W. M. Hermans, *Lancet Infect. Dis.*, 2004, **4**, 144–154.
- 122 N. L. Fahrenfeld, Y. J. Ma, M. O'Brien and A. Pruden, *Front. Microbiol.*, 2013, **4**, 1–10.
- 123 F. Barker-Reid, E. M. Fox and R. Faggian, *J. Water Health*, 2010, **8**, 521–531.
- 124 J. McLain and C. Williams, *Sustainability*, 2014, **6**, 1313–1327.
- 125 R. E. R. Goldstein, S. A. Micallef, S. G. Gibbs, A. George, E. Claye, A. Sapkota, S. W. Joseph and A. R. Sapkota, *Sci. Total Environ.*, 2014, **466**, 404–411.
- 126 F. H. Wang, M. Qiao, Z. E. Lv, G. X. Guo, Y. Jia, Y. H. Su and Y. G. Zhu, *Environ. Pollut.*, 2014, **184**, 247–253.
- 127 F. Wang, M. Qiao, J. Su, Z. Chen, X. Zhou and Y. Zhu, *Environ. Sci. Technol.*, 2014, **48**, 9079–9085.
- 128 U. Bockelmann, H. H. Dorries, M. N. Ayuso-Gabella, M. S. de Marçay, V. Tandoi, C. Levantesi, C. Masciopinto, E. Van Houtte, U. Szewzyk, T. Wintgens and E. Grohmann, *Appl. Environ. Microbiol.*, 2009, **75**, 154–163.



- 129 L. Rizzo, C. Manaia, C. Merlin, T. Schwartz, C. Dagot, M. C. Ploy, I. Michael and D. Fatta-Kassinos, *Sci. Total Environ.*, 2013, **447**, 345–360.
- 130 Y. L. Zhang, C. F. Marrs, C. Simon and C. W. Xi, *Sci. Total Environ.*, 2009, **407**, 3702–3706.
- 131 A. Schlüter, R. Szczepanowski, A. Pühler and E. M. Top, *FEMS Microbiol. Rev.*, 2007, **31**, 449–477.
- 132 M. Munir, K. Wong and I. Xagorarakis, *Water Res.*, 2011, **45**, 681–693.
- 133 A. Liu, A. Fong, E. Becket, J. Yuan, C. Tamae, L. Medrano, M. Maiz, C. Wahba, C. Lee, K. Lee, K. P. Tran, H. Yang, R. M. Hoffman, A. Salih and J. H. Miller, *Antimicrob. Agents Chemother.*, 2011, **55**, 1204–1210.
- 134 G. McVicker, T. K. Prajsnar, A. Williams, N. L. Wagner, M. Boots, S. A. Renshaw and S. J. Foster, *PLoS Pathog.*, 2014, **10**, e1003959.
- 135 D. I. Andersson and D. Hughes, *Nat. Rev. Microbiol.*, 2014, **12**, 465–478.
- 136 E. Gullberg, L. M. Albrecht, C. Karlsson, L. Sandegren and D. I. Andersson, *Ambio*, 2014, **5**, 19–23.
- 137 P. Y. Zhang, P. P. Xu, Z. J. Xia, J. Wang, J. Xiong and Y. Z. Li, *FEMS Microbiol. Lett.*, 2013, **348**, 149–156.
- 138 Z. J. Xia, J. Wang, W. Hu, H. Liu, X. Z. Gao, Z. H. Wu, P. Y. Zhang and Y. Z. Li, *J. Ind. Microbiol. Biotechnol.*, 2008, **35**, 1157–1163.
- 139 B. Song, G. R. Wang, N. B. Shoemaker and A. A. Salyers, *J. Bacteriol.*, 2009, **191**, 1078–1082.
- 140 M. Prudhomme, L. Attaiech, G. Sanchez, B. Martin and J. P. Claverys, *Science*, 2006, **313**, 89–92.
- 141 C. Úbeda, E. Maiques, E. Knecht, Í. Lasa, R. P. Novick and J. R. Penadés, *Mol. Microbiol.*, 2005, **56**, 836–844.
- 142 J. W. Beaber, B. Hochhut and M. K. Waldor, *Nature*, 2004, **427**, 72–74.
- 143 C. Baker-Austin, M. S. Wright, R. Stepanauskas and J. V. McArthur, *Trends Microbiol.*, 2006, **14**, 176–182.
- 144 L. Sörme and R. Lagerkvist, *Sci. Total Environ.*, 2002, **298**, 131–145.
- 145 S. K. Brar, M. Verma, R. D. Tyagi and R. Y. Surampalli, *Waste Manage.*, 2010, **30**, 504–520.
- 146 D. P. Karumathil, H.-B. Yin, A. Kollanoor-Johny and K. Venkitanarayanan, *Int. J. Environ. Res. Public Health*, 2014, **11**, 1844–1854.
- 147 P. Shi, S. Jia, X. X. Zhang, T. Zhang, S. Cheng and A. Li, *Water Res.*, 2013, **47**, 111–120.
- 148 J. J. Huang, H.-Y. Hu, Y.-H. Wu, B. Wei and Y. Lu, *Chemosphere*, 2013, **90**, 2247–2253.
- 149 A. Boehm, S. Steiner, F. Zaehring, A. Casanova, F. Hamburger, D. Ritz, W. Keck, M. Ackermann, T. Schirmer and U. Jenal, *Mol. Microbiol.*, 2009, **72**, 1500–1516.
- 150 M. Blickwede, R. Goethe, C. Wolz, P. Valentin-Weigand and S. Schwarz, *J. Antimicrob. Chemother.*, 2005, **56**, 315–323.
- 151 D. Li, A. Renzoni, T. Estoppey, C. Bisognano, P. Francois, W. L. Kelley, D. P. Lew, J. Schrenzel and P. Vaudaux, *Antimicrob. Agents Chemother.*, 2005, **49**, 916–924.
- 152 L. R. Hoffman, D. A. D'Argenio, M. A. MacCoss, Z. Zhang, R. A. Jones and S. I. Miller, *Nature*, 2005, **436**, 1171–1175.
- 153 M. Hausner and S. Wuertz, *Appl. Environ. Microbiol.*, 1999, **65**, 3710–3713.
- 154 S. Molin and T. Tolker-Nielsen, *Curr. Opin. Biotechnol.*, 2003, **14**, 255–261.
- 155 T. Das, S. Sehar and M. Manefield, *Environ. Microbiol. Rep.*, 2013, **5**, 778–786.
- 156 Y. Li, P. C. Y. Lau, J. H. Lee, R. P. Ellen and D. G. Cvitkovitch, *J. Bacteriol.*, 2001, **183**, 897–908.
- 157 S. Hannan, D. Ready, A. S. Jasni, M. Rogers, J. Pratten and A. P. Roberts, *FEMS Immunol. Med. Microbiol.*, 2010, **59**, 345–349.
- 158 N. M. Vega and J. Gore, *Curr. Opin. Microbiol.*, 2014, **21**, 28–34.
- 159 A. Pruden, *Environ. Sci. Technol.*, 2014, **48**, 5–14.
- 160 N. J. Ashbolt, A. Amezcua, T. Backhaus, P. Borriello, K. K. Brandt, P. Collignon, A. Coors, R. Finley, W. H. Gaze, T. Heberer, J. R. Lawrence, D. G. J. Larsson, S. A. McEwen, J. J. Ryan, J. Schonfeld, P. Silley, J. R. Snape, C. Van den Eede and E. Topp, *Environ. Health Perspect.*, 2013, **121**, 993–1001.
- 161 E. M. H. Wellington, A. B. Boxall, P. Cross, E. J. Feil, W. H. Gaze, P. M. Hawkey, A. S. Johnson-Rollings, D. L. Jones, N. M. Lee, W. Otten, C. M. Thomas and A. P. Williams, *Lancet Infect. Dis.*, 2013, **13**, 155–165.
- 162 H. K. Allen, J. Donato, H. H. Wang, K. A. Cloud-Hansen, J. Davies and J. Handelsman, *Nat. Rev. Microbiol.*, 2010, **8**, 251–259.
- 163 K. Rosario, C. Nilsson, Y. W. Lim, Y. Ruan and M. Breitbart, *Environ. Microbiol.*, 2009, **11**, 2806–2820.
- 164 M. Muniesa, M. Colomer-Illuch and J. Jofre, *Future Microbiol.*, 2013, **8**, 739–752.
- 165 J. M. Thomas, N. J. Ashbolt, N. Exposure, U. States, E. P. Agency, J. M. Thomas and N. J. Ashbolt, *Environ. Sci. Technol.*, 2011, **45**, 860–869.
- 166 US CDC, *Antibiotic Resistance Threats in the United States*, 2013, p. 114.
- 167 B. Blair, P. Sarkar, K. R. Bright, F. Marciano-Cabral and C. P. Gerba, *Emerging Infect. Dis.*, 2008, **14**, 1498–1499.
- 168 T. A. Bartrand, J. J. Causey and J. L. Clancy, *J. - Am. Water Works Assoc.*, 2014, **106**, E418–E432.
- 169 A. J. Martinez and G. S. Visvesvara, *Brain Pathol.*, 1997, **7**, 583–598.
- 170 M. C. Liang, Z. G. Ning and Y. K. Li, *Environ. Earth Sci.*, 2015, **73**, 4325–4338.
- 171 V. Thomas, J. Loret, M. Jousset and G. Greub, *Environ. Microbiol.*, 2008, **10**, 2728–2745.
- 172 H. C. Miller, J. Wylie, G. Dejean, A. H. Kaksonen, D. Sutton, K. Braun and G. J. Puzon, *Environ. Sci. Technol.*, 2015, **49**, 11125–11131.
- 173 J. M. Thomas and N. J. Ashbolt, *Environ. Sci. Technol.*, 2011, **45**, 860–869.
- 174 J. Lu, H. Y. Buse, V. Gomez-Alvarez, I. Struewing, J. Santo Domingo and N. J. Ashbolt, *J. Appl. Microbiol.*, 2014, **117**, 905–918.
- 175 M. T. García, S. Jones, C. Pelaz, R. D. Millar and Y. Abu Kwaik, *Environ. Microbiol.*, 2007, **9**, 1267–1277.
- 176 P. K. Jjemba, W. Johnson, Z. Bukhari and M. W. LeChevallier, *Pathogens*, 2015, **4**, 470–502.
- 177 J. L. Griffin, *Science*, 1972, **178**, 869–870.



- 178 H. Y. Buse and N. J. Ashbolt, *Lett. Appl. Microbiol.*, 2011, **53**, 217–224.
- 179 M. A. Söderberg, J. Dao, S. R. Starkenburg and N. P. Cianciotto, *Appl. Environ. Microbiol.*, 2008, **74**, 5583–5588.
- 180 B. C. Hitzfeld, S. J. Höger and D. R. Dietrich, *Environ. Health Perspect.*, 2000, **108**, 113–122.
- 181 G. Codd, S. Bell, K. Kaya, C. Ward, K. Beattie and J. Metcalf, *Eur. J. Phycol.*, 1999, **34**, 405–415.
- 182 B. Mueller, New York Times, http://www.nytimes.com/2015/10/02/nyregion/legionnaires-bacteria-regrew-in-bronx-cooling-towers-that-were-disinfected.html?_r=0, (accessed January 2016).
- 183 B. E. Hatt, A. Deletic and T. D. Fletcher, *J. Environ. Manage.*, 2006, **79**, 102–113.
- 184 Australian Government National Health and Medical Research Council, 2004.
- 185 L. R. Brown, *Full Planet, Empty Plates: The New Geopolitics of Food Scarcity*, 2012.
- 186 E. Schroeder, G. Tchobanoglous, H. L. Leverenz and T. Asano, *Natl. Water Reseach Inst. White Pap.*, 2012, pp. 1–20.
- 187 S. Adapa, N. Bhullar and S. Desouza, *J. Cleaner Prod.*, 2016, DOI: 10.1016/j.jclepro.2016.02.083.
- 188 D. Baumann, *Appl. Geogr.*, 1983, **3**, 79–84.
- 189 J. G. Lohman and Milliken L.C., U.S. Dep. Inter. Denver, 1985.
- 190 S. Dolnicar, A. Hurliman and L. D. Nghiem, *J. Environ. Manage.*, 2010, **91**, 1288–1293.
- 191 K. P. Tsagarakis, A. N. Menegaki, K. Siarapi and F. Zacharopoulou, *J. Risk Res.*, 2013, **16**, 133–144.
- 192 T. Asano, F. L. Burton, H. L. Leverenz, R. Tsuchihashi and G. Tchobanoglous, *Water Reuse: Issues, Technologies, and Applications*, McGraw-Hill, New York City, 2016.
- 193 R. E. Rosenberg Goldstein, S. A. Micallef, S. G. Gibbs, J. A. Davis, X. He, A. George, L. M. Kleinfelter, N. A. Schreiber, S. Mukherjee, A. Sapkota, S. W. Joseph and A. R. Sapkota, *Environ. Health Perspect.*, 2012, **120**, 1551–1558.
- 194 M. W. LeChevallier and R. J. Seidler, *Appl. Environ. Microbiol.*, 1980, **30**, 739–742.
- 195 P. Rusin, J. Rose, C. Haas and C. Gerba, *Rev. Environ. Contam. Toxicol.*, 1997, **152**, 57–83.
- 196 F. Xianmin, Z. Wenyu, W. Yuehua, Z. Donghai, J. Xiaoming and L. Shijie, *Anat. Rec.*, 2015, **298**, 1509–1517.
- 197 P. K. Jjemba, L. Weinrich, W. Cheng, E. Giraldo and M. W. LeChevallier, *Guidance Document on the Microbiological Quality and Biostability of Reclaimed Water Following Storage and Distribution*, 2010.
- 198 L. Macmillian, New York Times Green Blog, http://green.blogs.nytimes.com/2012/10/10/do-not-publish-on-the-slopes-gliding-on-wastewater/?_r=0, (accessed January 2016).
- 199 Q. Y. Wu, Y. Y. Li and H. Y. Hu, *Huanjing Kexue Xuebao*, 2013, **33**, 844–849.
- 200 T. W. Armstrong and C. N. Haas, *Risk Anal.*, 2007, **27**, 1581–1596.
- 201 T. P. Primm, C. A. Lucero and J. O. Falkinham, *Clin. Microbiol. Rev.*, 2004, **17**, 98–106.
- 202 B. J. Williams, J. Dehnhostel and T. S. Blackwell, *Respirology*, 2010, **15**, 1037–1056.
- 203 Y. J. Lee, J. H. Kim, H. J. Sohn, J. Lee, S. Y. Jung, Y. J. Chwae, K. Kim, S. Park and H. J. Shin, *Parasite Immunol.*, 2011, **33**, 382–389.
- 204 H. Ryu, A. Alum and M. Abbaszadegan, *Environ. Sci. Technol.*, 2005, **39**, 8600–8605.
- 205 C. W. Knapp, S. M. McCluskey, B. K. Singh, C. D. Campbell, G. Hudson and D. W. Graham, *PLoS One*, 2011, **6**, e27300.
- 206 M. J. Mohammad Rusan, S. Hinnawi and L. Rousan, *Desalination*, 2007, **215**, 143–152.
- 207 B. Kurenbach, D. Marjoshi, C. F. Amabile-cuevas, G. C. Ferguson, W. Godsoe, P. Gibson and J. A. Heinemann, *mBio*, 2015, **6**, 1–9.
- 208 M. Salgot, E. Huertas, S. Weber, W. Dott and J. Hollender, *Desalination*, 2006, **187**, 29–40.
- 209 J. K. Edzwald, *Water Quality & Treatment: A Handbook on Drinking Water*, McGraw-Hill, New York City, 6th edn, 2011.
- 210 C. Volk, C. Renner, C. Robert and J. C. Joret, *Environ. Technol.*, 1994, **15**, 545–556.
- 211 P. Servais, P. Laurent and D. Gatel, presented in part at *Proc. of the Annual AWWA Water Qual. Technol. Conf.*, New Orleans, 1995.
- 212 P. Laurent, M. Prévost, J. Cigana, P. Niquette and P. Servais, *Water Res.*, 1999, **33**, 1387–1398.
- 213 P. Niquette, P. Servais and R. Savoir, *Water Res.*, 2001, **35**, 675–682.
- 214 C. J. Volk and M. W. LeChevallier, *J. Am. Water Works Assoc.*, 2000, **92**, 64–76.
- 215 D. van der Kooij, A. Visser and W. A. M. Hijnen, *J. Am. Water Works Assoc.*, 1982, **74**, 540–545.
- 216 M. W. LeChevallier, W. Schulz and R. G. Lee, *Appl. Environ. Microbiol.*, 1991, **57**, 857–862.
- 217 M. Vital, H. P. Fuchslin, F. Hammes and T. Egli, *Microbiology*, 2007, **153**, 1993–2001.
- 218 D. Gerrity, S. Gamage, J. C. Holady, D. B. Mawhinney, O. Quiñones, R. A. Trenholm and S. A. Snyder, *Water Res.*, 2011, **45**, 2155–2165.
- 219 Ç. Kalkan, K. Yapsakli, B. Mertoglu, D. Tufan and A. Saatci, *Desalination*, 2011, **265**, 266–273.
- 220 A. N. Pisarenko, B. D. Stanford, D. Yan, D. Gerrity and S. A. Snyder, *Water Res.*, 2012, **46**, 316–326.

