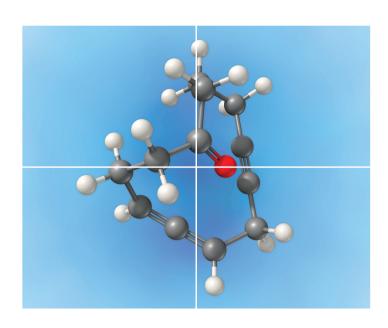
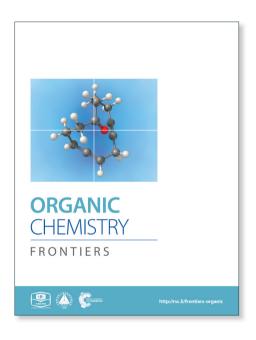
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Enantioselective Nickel-Catalyzed Alkylative Alkyne- Aldehyde Cross-Couplings

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An efficient asymmetric nickel-catalyzed alkylative alkynealdhehyde cross-coupling is developed by employing a P-chiral monophosphorus ligand-BI-DIME, allowing rapid access to a series of chiral tetra-substituted olefinic allylic alcohols in high yields and good to excellent ee's. The three-component reactions enjoy excellent regio- and enantioselectivities, and a broad substrate scope from readily available starting materials.

Development of efficient multicomponent reactions¹ for rapid construction of molecules with complexity from simple feedstock in high yields and excellent selectivities has become a field of significant interest to synthetic organic chemists. Such reactions are often more environmentally beneign, energysaving, and with high atom and step economy. Chiral tetrasubstituted olefinic allylic alcohols² have increasingly become a class of important intermediates in organic synthesis. Their syntheses are often tedious, with requirement of multi synthetic steps. The single-step, three-component, nickel-catalyzed alkylative alkyne-aldehyde cross-coupling^{3,4} from readily available starting materials under mild conditions has provided an appealing method in constructing such structures (Figure 1). Since Montgomery first reported⁵ this transformation, enantioselective nickel-catalyzed alkylative alkyne-aldehyde coupling has attracted considerable attention. Due to lack of efficient chiral nickel catalysts,6,7 only a few highly enantioselective examples were reported with limited substrate scope. Excellent enantioselectivities were achieved by Zhou and coworkers3f by employing a chiral 6,6'-disubstituted spiro phosphoramidite ligand on a series of aryl aldehydes as substrates. In contrast, only one example with alkyl aldehyde

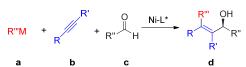


Figure 1. Nickel-catalyzed alkylative alkyne-aldehyde cross-coupling

was reported at 10 mol % nickel catalyst loading. We herein report an efficient asymmetric nickel-catalyzed alkylative alkyne-aldehyde coupling that has provided high yields and good to excellent enantioselectivities for a series of chiral tetrasubstituted olefinic allylic alcohols with the employment of a P-chiral monophosphorus ligand BI-DIME.

We recently reported⁸ an asymmetric nickel-catalyzed reductive cyclization of alkynones that have led to a series of tertiary allylic alcohols bearing furan/pyran rings in excellent yields and enantioselectivities with AntPhos/BI-DIME⁹ as the chiral ligand. Its broad substrate scope for both alkyl and aryl substrates prompted us to investigate the intermolecular alkylative alkyne-aldehyde couplings with the P-chiral phosphorus ligands developed in our laboratory¹⁰. Thus, the ternary reactions between 1-phenyl-1-propyne benzaldehyde (2a), and dimethylzinc were carried out at room temperature in toluene for 16 h under nitrogen in the presence of Ni(COD)₂ (5 mol %) and a P-chiral biaryl monophosphorus ligand (6 mol %). As shown in Table 1, the ligand structure significantly impact its reactivity and enantioselectivity. When ligand **L1** with no substituents on the low aryl ring was employed, the desired coupling product 3a was isolated in 67% yield and 50% ee (entry 1). Nearly no formation of its regio-isomer 3a' was observed according to ¹H NMR. When BI-DIME with two methoxy substituents on the low aryl ring was used as the ligand, a high yield (95%) and excellent enantioselectivity (90%) were observed (entry 2). However, ligands with aryloxy substituents on the low aryl ring provided little or no formation of 3a (entries 3-4). Ligands with alkyl groups at the 2 position of the oxophosphole ring ring also provided diminished reactivities, as both **L5** and **L6** were less effective (entries 5-6). AntPhos, which was highly effective for reductive cyclization of alkynones⁸, also proved to be less efficient (entry 7). Solvent also played an important role for the reactivity and enantioselectivity, and toluene was found to be best among five solvents studied. We believed that the strong solvation of polar solvents with the nickel catalyst inhibited the cycloaddition step of the catalytic cycle.

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Table 1. Enantioselective Nickel-Catalyzed Alkylative Couplings of Alkynes with Aldehydes: Optimization of Conditions

L1: R₁= H L2: R₁= OMe (BI-DIME) L3: R1= OPh

L4: $R_1 = O(3,5-Me_2C_6H_3)$

′tBu

L5: R = Me

L6: R = Bn L7 (AntPhos)

Entries ^[a]	Ligand	solvent	Yield of 3a	ee (%) [c]
			(%) ^[b]	
1	L1	toluene	67	50
2	L2	toluene	95	90
3	L3	toluene	12	74
4	L4	toluene	<5	nd
5	L5	toluene	36	76
6	L6	toluene	<5	nd
7	L7	toluene	46	42
8	L2	dioxane	78	79
9	L2	THF	25	70
10	L2	DCM	19	82
11	L2	DME	0	/

[a] The reactions were carried out in a solvent (0.1 M) at rt under nitrogen for 16 h with 1a (0.20 mmol), 2a (0.40 mmol), Ni(COD)₂ (5 mol %), ligand (6 mol %), and Me₂Zn (0.60 mmol). The absolute configuration of 2a was determined by comparison of its optical rotation to reported data. ^{3f} [b] Isolated yields, ratio of **3a/3a** > 20/1 in all cases. [c] Determined by chiral HPLC on a Chiralcel OD-H column.

We then looked into the substrate scope of the reductive alkynealdehyde cross-coupling. As can be seen in Table 2, a series of chiral allylic alcohols with tetrasubstituted olefin functionality were prepared in high yields (62-99%), high regioselectivities (3b-ae/3'bae >10/1), and good to excellent enantioselectivities (78-96% ee). A range of aryl aldehydes were successfully employed regardless of their electronic properties and substitution patterns. No significant electronic effect of the aryl aldehyde was observed on the enantioselectivity of the coupling (3b vs 3e-h). Besides para- or metasubstituted aryl aldehydes, ortho-substituted aryl aldehydes were also successfully employed for the first time in reductive alkyne-aldehyde cross-couplings to provide excellent ees (3m-p). Both 1- and 2naphthaldehydes (3q-r) were applicable for this transformation. An array of aliphatic aldehydes (3s-y) were also employed to provide corresponding chiral allylic alcohols in good yields and excellent ees. The enantioselectivities increased with the bulkiness of the aldehyde, as the reaction with pivalaldehyde yielded the alcohol 3y in 95% ee albeit with a moderate yield (62%). 1-Phenyl-1-butyne and 1-phenyl-1-hexyne were also successfully employed to provide the corresponding products in excellent regioselectivities. This was in contrast to Zhou's system3f where a significant amount of regioisomers were formed. Finally, heteroaryls such as thiophene and furan were well tolerated. Both compounds 3ab and 3ac with thiophene moieties were successfully prepared in excellent ees. A

moderated ee (79% or 84%) was achieved when 2- or 3-furanaldehyde was employed as the substrates.

Table 2. Substrate scope

[a] Reaction conditions: alkynes (0.20 mmol), aldehydes (0.40 mmol), Ni(COD)₂ (5 mol%), **L2** (6 mol%), and Me₂Zn (0.60 mmol), toluene (0.10 M), rt under N_2 , 16 h, isolated yield. **3b-ae/3'b-ae** > 19:1 unless

84% ee, 91% yield

93% ee, 80% yield

79% ee, 90% yield

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otherwise specified; The absolute configurations were determined by comparison of their optical rotations to reported data or by analogy.

To demonstrate the practicality of this transformation, the reaction of 1-phenyl-1-propyne (1a), 2,6-dimethoxybenzaladehyde (2af), and dimethylzinc with the Ni-(R)-BI-DIME catalyst was carried out in toluene at 1 gram scale. The desired product 3af was obtained in 97% yield and 98% ee, demonstrating a highly efficient and practical example of enantioselective alkylative alkyne-aldehyde cross-coupling with a di-*ortho*-substituted aldehyde as the substrate.

Scheme 1. Gram-scale asymmetric alkylative alkyne-aldehyde cross-coupling

Previous mechanistic studies by Montgomery^{3b,11,13a}, Jamison^{7a,12,13b,13c}, & Houk^{3b,13} as well as by us⁸ on nickelcatalyzed reductive/alkylative alkyne-carbonyl cross-coupling has suggested that the stereoselectivity of this three-component transformation is likely to be determined at the cycloaddition step of alkyne-aldehyde with a Ni(0)-(R)-BI-DIME species. To shed light on the stereochemical translation of this reaction, the two conformers I and II of the cycloaddition intermediates derived from 1a, 2a and the Ni-(R)-BI-DIME complex were subjected to DFT calculations (Figure 2).¹⁴ The calculated energy of conformer I was ~0.5 kcal/mol higher than that of conformer \mathbf{II} , which was very likely due to a greater steric influence of the *tert*-butyl group of (*R*)-BI-DIME over the phenyl group in 1a. The more favorable conformer II led to the chiral alcohol product with the observed stereochemistry. A more sterically crowded Ni(II) structure would provide an even greater difference between two conformers. enantioselectivities were therefore achieved when more hindered alkyl aldehydes or di-ortho-subsituted aryl aldehydes were employed as the reagents.

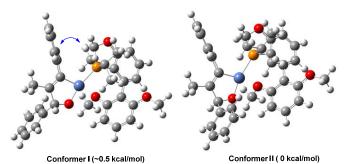


Figure 2. The DFT calculated conformers \mathbf{I} and \mathbf{II} of the cycloaddition intermediates derived from $\mathbf{1a}$, $\mathbf{2a}$, and the Ni-(R)-BI-DIME complex

In summary, we have developed an efficient asymmetric nickelcatalyzed alkylative alkyne-aldehyde cross-coupling with the employment of a P-chiral phosphorus ligand-BI-DIME. The threecomponent reaction enjoys good to excellent regio- and enantioselectivities, and provides a broad substrate scope with good functional group compatibility. A series of chiral allylic alcohols with tetra-substituted olefin moieties were thus formed in high yields and ee's. Further study is focusing on expanding its substrate scope and applications in medicinal chemistry and natural product synthesis.

Acknowledgements

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Notes and references

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- † Electronic Supplementary Information (ESI) available: [detailed procedures of cross-coupling reactions, characterization data, and NMR spectra]. See DOI: 10.1039/c000000x/
 - Selected reviews: a) D. J. Ramon, M. Yus, Angew. Chem. Int. Ed. 2005, 44, 1602; b) B. B. Toure, D. G. Hall, Chem. Rev. 2009, 109, 4439; c) B. Ganem, Acc. Chem. Res. 2009, 42, 463; d) A. Domling, W. Wang, K. Wang, Chem. Rev. 2012, 112, 3083; e) G. Guillena, D. J. Ramon, M. Yus, Tetrahedron: Asymmetry. 2007, 18, 693; f) C. de Graaff, E. Ruijter, R. V. A. Orru, Chem. Soc. Res. 2012, 41, 3969; g) J. E. Biggs-Houck, A. Younai, J. T. Shaw, Curr. Opin. Chem. Biol. 2010, 14, 371; h) N. R. Candeias, F. Montalbano, P. Cal, P. Gois, Chem. Rev. 2010, 110, 6169.
 - a) Comprehensive Organic Synthesis, B. M. Trost, Ed.;
 Pergamon Press: Oxford, 1991; b) A. H. Hoveyda, D. A. Evans,
 G. C. Fu, Chem. Rev. 1993, 93, 1307; c) D. Banerjee, K. Junge,
 M. Beller, Angew. Chem. Int. Ed. 2014, 53, 13049; d) W. Zi, Y.
 -M. Wang, D. F. Toste, J. Am. Chem. Soc. 2014, 136, 12864; e)
 E. Vedajs, J. A. Mackay, Org. Lett. 2001, 3, 535; f) M. I. Calaza,
 E. Hupe, P. Knochel, Org. Lett. 2003, 5, 1059.
 - 3 For recent papers on nickel-catalyzed alkylative/reductive alkyne-aldehyde cross-coupling: a) E. P. Jackson, J. Montgomery, J. Am. Chem. Soc. 2015, 137, 958; b) M. T. Haynes, P. Liu, R. D. Baxter, A. J. Nett, K. N. Houk, J. Montgomery, J. Am. Chem. Soc. 2014, 136, 17495; c) S. Z. Tasker, E. A. Standley, T. F. Jamison Nature. 2014, 509, 299; d) K. Tanaka, Y. Taiima, Eur. J. Org. Chem. 2012, 3715; e) J. Montgomery, Acc. Chem. Res. 2000, 33, 467; f) Y. Yang, S. -F. Zhu, C. -Y. Zhou, Q. -L. Zhou, J. Am. Chem. Soc. 2008, 130, 14052.
 - 4 For recent papers on catalytic reductive coupling with other metals: a) H. A. Reichard, M. Mclaughlin, M. Z. Chen, G. C. Micalizio, Eur. J. Org. Chem. 2010. 391; b) J. F. Bower, I. S. Kim, R. L. Patman, M. J. Krische, Angew. Chem. Int. Ed. 2009, 48, 34; c) E. Skucas, M.-Y. Ngai. V. Komanduri, M. J. Krische, Acc. Chem. Res. 2007, 40, 1394; d) A. B. Bahadoor, A. Flyer, G. C. Micalizio, J. Am. Chem. Soc. 2005, 127, 3694; e) I. Ojima, M. Tzamarioudaki, C. Y. Tsai, J. Am. Chem. Soc. 1994, 116, 3643.
 - a) E. Oblinger, J. Montgomery, J. Am. Chem. Soc. 1997, 119,
 9065; b) X. Qi, J. Montgomery, J. Org. Chem. 1999, 64, 9310;

Organic Chemistry Frontiers Accepted Manuscript

- c) M. Lozanov, J. Montgomery, J. Am. Chem. Soc. 2002, 124, 2106, d) Y. Ni, K. K. Amarasinghe, J. Montgomery, Org. Lett. 2002, 4, 1743.
- 6 For recent papers on non-enantioselective nickel-catalyzed reductive/alkylative coupling: a) X. -Q. Tang, J. Montgomery, J. Am. Chem. Soc. 1999, 121, 6098; b) W. -S. Huang, J. Chan, T. F. Jamison, Org. Lett, 2000, 2, 4221, c) K. Takai, S. Sakamoto, T. Isshiki, Org. Lett, 2003, 5, 653; d) K. M. Miller, T. Luanphaisarnnont, C. Molinaro, T. F. Jamison, J. Am. Chem. Soc. 2004, 126, 4130; e) K. M. Miller, T. F. Jamison, J. Am. Chem. Soc. 2004, 126, 15342;
- For recent papers on asymmetric nickel-catalyzed reductive/alkylative coupling: a) K. M. Miller, W. -S. Huang, T. F. Jamison, J. Am. Chem. Soc. 2003, 125, 3442; b) K. M. Miller, T. F. Jamison, Org. Lett, 2005, 7, 3077; c) Y. Sato, N. Saito, M. Mori, J. Am. Chem. Soc. 2000, 122, 2371; d) Y. Sato, N. Saito, M. Mori, J. Org. Chem. 2002, 67, 9310; e) Y. Yang, S.-F. Zhu, H.-F. Duan, C.-Y. Zhou, L.-X. Wang, Q. -L. Zhou, J. Am. Chem. Soc. 2007, 129, 2248; f) M. R. Chaulagain, G. J. Sormunen, J. Montgomery, J. Am. Chem. Soc. 2007, 129, 9568; g) C. T. Check, K. P. Jang, C. B. Schwamb, A. S. Wong, M. H. Wang, K. A. Scheidt, Angew. Chem. Int. Ed. 2015, 54, 4264.
- W. Fu, M. Nie, A. Wang, W. Tang, Angew. Chem. Int. Ed. 2015, 54, 2520.
- a) W. Tang, A. G. Capacci, X. Wei, W. Li, A. White, N. D. Patel, J. Savoie, J. J. Gao, S. Rodriguez, B. Qu, N. Haddad, B. Z. Lu, D. Krishnamurthy, N. K. Yee, C. H. Senanayake, Angew. Chem. Int. Ed. 2010, 49, 5879; b) W. Tang, S.Keshipeddy, Y. Zhang, X. Wei, J. Savoie, N. D. Patel, N. K. Yee, C. H. Senanayake, Org. Lett. 2011, 13, 1366; c) S. Rodriguez, B. Qu, N. Haddad, D. C. Reeves, W. Tang, H. Lee, D. Krishnamurthy, C. H. Senanayake, Adv. Synth. Catal. 2011, 353, 533; d) Q. Zhao, C. Li, C. H. Senanayake, W. Tang, Chem. Eur. J. 2013, 19, 2261; e) C. Li, G. Xiao, Q. Zhao, H. Liu, T. Wang, W. Tang, Org. Chem. Front. 2014, 1, 225.
- a) W. Tang, N. D. Patel, G. Xu, X. Xu, J. Savoie, S. Ma, M.-H. Hao, S. Keshipeddy, A. G. Capacci, X. Wei, Y. Zhang, J. J. Gao, W. Li, S. Rodriguez, B. Z. Lu, N. K. Yee, C. H. Senanayake, Org. Lett. 2012, 14, 2258; b) K. Li, N. Hu, R. Luo, W. Yuan, W. Tang, J. Org. Chem. 2013, 78, 6350; c) G. Xu, W. Fu, G. Liu, C. H. Senanayake, W. Tang, J. Am. Chem. Soc. 2014, 136, 570; d) G. Xu, Q. Zhao, W. Tang, Chin. J. Org. Chem. 2014, 34, 1919; e) C. Li, T. Chen, B. Li, G. Xiao, W. Tang, Angew. Chem. Int. Ed. 2015, 54, 3792; f) K. Du, P. Guo, Y. Chen, Z. Cao, Z. Wang, W. Tang, Angew. Chem. Int. Ed. 2015, 54, 3033.
- a) G. M. Mahandru, G. Liu, J. Montgomery, J. Am. Chem. Soc. 2004, 126, 3698; b) G. M. Mahandru, G. Liu, J. Montgomery, J. Am. Chem. Soc. 2004, 126, 15316; c) H. A. Malik, G. J. Sormunen, J. Montgomery, J. Am. Chem. Soc. 2010, 132, 6304.
- 12 S. J. Patel, T. F. Jamison, Angew. Chem. Int. Ed. 2003, 42, 1364.
- 13 a) P. Liu, J. Montgomery, K. N. Houk, J. Am. Chem. Soc. 2011, 133, 6956; b) P. Liu, P. McCarren, P. H. -Y. Cheong, T. F. Jamison, K. N. Houk, J. Am. Chem. Soc. 2010, 132, 2050; c) P. McCarren, P. Liu, P. H. -Y. Cheong, T. F. Jamison, K. N. Houk, J. Am. Chem. Soc. 2009, 131, 6654.

14 DFT calculations were performed with Gaussian 03 package and the geometries were optimized with B3LYP and a mixed basis set of 6-31G(d) for all atoms.