Simple bioconjugate chemistry serves great clinical advances: Albumin as a versatile platform for diagnosis and precision therapy

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Simple bioconjugate chemistry serves great clinical advances: Albumin as a versatile platform for diagnosis and precision therapy

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Human Serum Albumin (HSA)
Abstract

Albumin is the most abundant circulating protein in plasma and recently emerged as a versatile protein carrier for drug targeting and for improving the pharmacokinetic profile of peptide or protein based drugs. Three drug delivery technologies related to albumin have been developed, which include the coupling of low-molecular weight drugs to exogenous or endogenous albumin, conjugating bioactive proteins by albumin fusion technology (AFT), and encapsulation of drugs into albumin nanoparticles. This review article starts with a brief introduction of human serum albumin (HSA), and then summarizes the mainstream chemical strategies of developing HSA binding molecules for coupling with drug molecules. Moreover, we also concisely condense the recent progresses of the most important clinical applications of HSA-binding platforms, and specify the current challenges that need to be met for a bright future of HSA-binding.
1. Introduction

Human serum albumin (HSA) is the most abundant protein in plasma, and it can serve as a versatile carrier for drug delivery as well as for prolonging the active profile of fast-clearance drugs.\textsuperscript{1-3} Besides being a key drug-delivery protein in blood, it also undertakes the transportation of many essential biomolecules, such as fatty acids, hormones and amino acids.\textsuperscript{4,5} HSA has a notably long half-life (19 days) in blood circulation.\textsuperscript{6, 7} HSA is produced in the liver cells as preproalbumin, then modified by Golgi vesicles to give secreted albumin. Approximately 13-14 g of albumin is secreted into intravascular system each day, and the extravascular HSA will return to intravascular circulation through the lymphatic system.\textsuperscript{8, 9} The degradation of HSA is highly dependent on the interaction with albumin receptors, including gp18, gp30, megalin and the neonatal Fc receptor (FcRn).

HSA is widely used as a carrier for small molecule drugs and imaging probes.\textsuperscript{10-19} It is biodegradable, non-toxic and lack of immunogenicity, making it an excellent candidate as an excipient for vaccines and many other pharmaceuticals.\textsuperscript{20, 21} In addition, HSA is robust against chemical modifications and can be stable in the pH range of 4-9 at 60 °C for as long as 10 h. Therefore, the amino acid residues on albumin can be readily linked with therapeutic drugs, imaging reporters and targeting molecules through chemical conjugation. Albumin is also found to specifically target tumor regions because of its enhanced permeability and retention (EPR) effect as well as albumin receptor binding, which is a unique advantage as the carrier for tumor-targeted drug delivery.\textsuperscript{22, 23}

There have been long-standing interests to develop a general strategy that can effectively prolong the active profile of pharmaceuticals. Among the currently developed methods,
conjugating pharmaceuticals to albumin-binding molecules is one of the most commonly used approaches due to its high efficiency and minimum side effect.\textsuperscript{24}

In the past two decades, a number of advances on HSA-binding therapeutics have been approved by the Food and Drug Administration (FDA), and many more are under active clinical investigation (Table 1). These successful discoveries are of significance to a broad spectrum of healthcare, especially for cancer therapy and diabetes treatment. Overall, the development is often derived from a new understanding of HSA chemistry, followed by a smart application designed to solve an emerging clinical challenge. For instance, the development and market approval of Abraxane, a paclitaxel albumin nanoparticle, became a landmark for both nanomedicine and albumin-based drug delivery technology with annual sales of $850 million in 2014. Indeed, the thoughts and rationales of these successes are greatly inspiring, not only for the development of future HSA-binding therapeutics, but also to the most general audiences including chemists, biologists and clinical doctors who have interests in new drug development. For a better understanding of these exciting progresses, we would like to share our review to guide the biological design, chemical screening and clinical application of HSA-based drugs, with focus on the strategy of \textit{in vivo} binding that are most practical for clinical use.

\textbf{Table 1.} A selective summary of clinically relevant HSA-based imaging agents and therapeutics\textsuperscript{4, 10, 20, 24-27}
2. General strategy to develop HSA-conjugated drugs

2.1. In vitro covalent conjugation

To start with, amide coupling based on lysine residue is the most classical method for in vitro covalent HSA conjugation (Figure 1, A-D).\(^{28-34}\) At present, the functional moieties often contain p-isothiocyanate (p-SCN) or NHS ester (N-hydroxysuccinimide) that can be obtained by in situ activation. However, these methods are not site-specific and always lead to a mixture of mono and multiply modified HSA.\(^{18, 19, 35-39}\)

To meet the challenge, an optimized coupling method was developed to perform the conjugation on cysteine-34 instead of lysines on HSA, and has provided better-defined HSA-drug conjugates that have high purity with a constant drug-loading ratio, a minimal alteration of the three-dimensional protein structure and a preset breaking point. However, as the cysteine-34 position on commercially available HSA is largely blocked by cysteine, homocysteine as well as other sulphydryl containing compounds, the HSA is a mixture of mercaptalbumin and...
nonmercaptalbumin and only approximately 20–60% of them contain free sulfhydryl groups. To solve this problem, Mansour et al. developed a one-step procedure of selectively reducing HSA with dithiothreitol (Cleland's reagent), giving approximately one sulfhydryl group for each HSA molecule (Figure 1E). In the next step, the reduced HSA is directly coupled with the maleimide modified drugs such as doxorubicin maleimide. Compared with the free doxorubicin, the HSA-conjugated version was significantly better on curing murine renal carcinoma (RENCA) at equitoxic dose.\textsuperscript{40}

![Figure 1. General methods of covalently conjugating small molecules onto albumin. (A) The coupling molecule is activated \textit{in situ} by using classical coupling reagents such as N,N'-dicyclohexylcarbodiimide (DCC), 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC), hydroxybenzotriazole (HOBT) and 2-(1H-benzotriazol-1-yl)-1,1,3,3-]
tetramethyluronium hexafluorophosphate (HBTU), and then attached onto lysine residue of HSA under weakly basic conditions. (B) The coupling molecule is activated as an N-hydroxysuccinimide (NHS) ester prior to be conjugated onto lysine residue of HSA under weakly basic conditions. (C) The coupling molecule is activated as a tetrafluorophenyl (TFP) ester prior to being conjugated onto lysine residue of HSA under weakly basic conditions. (D) The coupling molecule is modified to contain p-isothiocyanate (p-SCN), and then attached onto lysine residue of HSA under weakly basic conditions. (E) The coupling molecule is modified to contain maleimide moieties, and then attached onto cysteine residue of HSA under weakly basic conditions.

In vitro covalent conjugation has been widely used in preparing HSA-based drugs (Figure 2). For instance, to radiolabel HSA with radiometals for diagnostic imaging, radiometal chelators will be linked on HSA by incubating NHS-activated ester of DOTA (1,4,7,10-tetraazacyclododecan-1,4,7,10-tetraacetic acid) together with HSA under weakly basic condition (pH = 8-9).\textsuperscript{41, 42} \textsuperscript{18}F-HSA, \textsuperscript{68}Ga-DOTA-HSA, \textsuperscript{111}In-DTPA-HSA (DTPA is the abbreviation for diethylene triamine pentaacetic acid) and Gd-DTPA-HSA have been as blood-pool imaging reagents by using positron emission tomography (PET), single-photon emission computed tomography (SPECT) and magnetic resonance image (MRI), respectively.\textsuperscript{28, 43-45} In convention, radioactive HSA is prepared by multiple-step radiosynthesis. Nevertheless, by taking the advantage of the development of milder and more efficient radiolabeling strategies,\textsuperscript{46-50} one-step HSA labeling is likely to be practical in the near future. In addition, peptide and small molecular drugs have also been conjugated on the lysine residues of HSA to develop advanced HSA-binding imaging probe and therapeutics.\textsuperscript{28, 43-45}
Figure 2. Representative HSA drugs based on in vitro conjugation: (a) $^{111}$In-labeled HSA for single-photon emission computed tomography (SPECT); (b) Gd-labeled HSA for magnetic resonance imaging (MRI); (c) $^{68}$Ga-labeled HSA for positron emission tomography (PET); (d) Doxorubicin HSA conjugates for cancer chemotherapy with less side effect; (e) $^{18}$F-labeled HSA conjugates for PET; (f) CysCOOH HSA conjugates for photothermal therapy; (g) 3,5-Diiodothyronine HSA conjugates for antibody production in animals; (h) Exendin-4 peptide HSA conjugates for the treatment of type 2 diabetes.

Fusion protein technology (FPT) is a special way of in vitro conjugating HSA with functional moieties yet has been broadly used in preparing recombinant HSA/protein. By doing so, albumin protein conjugates are genetically engineered by putting together the genes of the two molecules and expressing the albumin fusion proteins in yeast strains (Figure 3).
Clinically, one such albumin fusion protein is Albuferon, a fusion protein of albumin and interferon α-2b for the treatment of hepatitis C. A number of other albumin fusion proteins have entered early clinical trials. These include fusion proteins with low-molecular weight peptides such as β-natriuretic peptide and glucagon-like peptide 1, as well as fusion proteins with cytokines. Albuleukin, an albumin fusion protein with recombinant interleukin-2 that has shown promising antitumor efficacy against murine renal cell carcinoma and melanoma.

2.2. In vivo covalent conjugation

Kratz et al. established a strategy that exploits endogenous HSA as a drug carrier. In this therapeutic strategy, the prodrug binds rapidly and selectively to the cysteine-34 position of circulating serum albumin after intravenous administration thereby generating a macromolecular transport form of the drug in situ in the blood. Indeed, the strategy of in vivo HSA conjugation would have several advantages over in vitro synthesized drug albumin conjugates: (a) The use of
commercial and possibly pathogenic albumin is avoided; (b) Easy to use and inexpensive to manufacture; (c) The related quality control is simple, which is comparable to any other low-molecular weight drug candidates.

The macromolecular prodrug approach targets the cysteine-34 position of albumin. A HPLC analysis demonstrates that approximately 70% of circulating albumin in the blood stream is mercaptalbumin (HMA) that contains an accessible cysteine-34.\textsuperscript{4, 61, 62} Moreover, the free thiol group of cysteine-34 of HSA is an unusual feature of an extracellular protein. As known, only three other major proteins that contain free cysteine residues in human plasma: (1) apolipoprotein B-100 of low-density lipoprotein (LDL) which has two cysteine residues (Cys-3734 and Cys-4190) located at the C-terminal end of the protein,\textsuperscript{63-65} (2) fibronectin which has two cryptic, free sulfhydryl groups,\textsuperscript{66} and (3) R1-antitrypsin which has a single cysteine residue (Cys-232).\textsuperscript{66-68} However, the sulfhydryl groups in these proteins do not react readily with sulfhydryl reagents under physiological conditions and are normally linked to either cysteine or glutathione in blood circulation. Therefore, the free thiol group on HSA, cysteine-34 of endogenous albumin, is a unique amino acid on the surface of a circulating protein, which is capable of further conjugation.\textsuperscript{69}
Proof of concept was obtained with the (6-maleimidocaproyl) hydrazone derivative of doxorubicin (DOXO-EMCH) that rapidly and selectively binds to circulating albumin within a few minutes (Figure 4). Inspired by translational research with DOXO-EMCH, many albumin-binding prodrugs have been developed (Figure 5). These prodrugs often consist of an anticancer drug, the maleimide group as the thiol-binding moiety and an enzymatically cleavable peptide linker. Examples include doxorubicin prodrugs that are cleaved by matrix metalloproteases 2 and 9,\textsuperscript{40} cathepsin B,\textsuperscript{70} urokinase plasminogen (uPA) or prostate-specific antigen (PSA),\textsuperscript{71, 72} methotrexate prodrugs that are cleaved by cathepsin B or plasmin,\textsuperscript{73} and camptothecin prodrugs that are cleaved by cathepsin B or unidentified proteases.\textsuperscript{74} In addition, maleimide derivatives with 5-fluorouracil analogues and platinum (II) complexes have been developed.\textsuperscript{75}
Figure 5. Structures of selected albumin-binding maleimide modified prodrugs. (a) Doxorubicin prodrug that is cleaved by cathepsin B; 76 (b) and (c) Albumin-binding prodrugs with Pt(II) complexes; 75 (d) Camptothecin prodrug that is cleaved by cathepsin B; 73 (e) Doxorubicin prodrug that is cleaved by prostate-specific antigen (PSA). 40, 60, 77, 78

2.3. In vitro non-covalent HSA binding

Besides covalently connecting HSA with small functional molecules, non-covalent van der Waals force or electronic interaction is another approach that can be used for HSA binding. 79, 80 For instance, certain radiometals can form robust conjugates with macro-aggregated albumin (MAA) without using any chelators, the resulting complexes (\(^{111}\text{In-MAA}\) and \(^{99m}\text{Tc-MAA}\), Figure 6 A-B) have been widely used in clinical diagnosis, especially for lung perfusion and for
detecting gastrointestinal bleeding by SPECT.\textsuperscript{20, 51, 81-83} To form this self-assembled capsule, firstly the intramolecular disulfide bonds of HSA are partially reduced by using glutathione (GSH) to give free sulfhydryl groups. Then, the pretreated HSA/water solution is mixed with small drugs in triaryl butyl alcohol (TBA). Here, TBA is used as the anti-solvent for albumin and water is used as the anti-solvent for the small molecular drugs. In the mixed solution, HSA and small molecular drugs would precipitate out because of the decreased solubility of both HSA and small molecular drugs. At last, this suspension is further incubated at 37 °C to form intermolecular disulfide to give small molecular drugs loaded HSA nanoparticles (\textbf{Figure 6C}).\textsuperscript{20, 84}

In addition, \textit{in vitro} non-covalent HSA binding is also commonly used in preparing HSA-nanoparticle complexes, especially for the purpose of imaging and therapy.\textsuperscript{2, 26-28, 85-90} For instance, IONPs (iron oxide nanoparticles) were incubated with dopamine to become moderately hydrophilic before being doped into HSA matrices \textit{via} non-covalent binding. In this case, a physical capsule is formed between HSA and IONP that can load small molecular drugs with high efficiency. Additionally, as shown in \textbf{Figure 6D}, the HSA matrix is capable of carrying fluorophores and radioactive reporters, therefore this type of HSA-binding nanoparticle can serve as a multiple functional platform for the purpose of both \textit{in vivo} imaging and drug delivery.
Figure 6. Representative strategies of *in vitro* non-covalent HSA binding. (A) $^{111}$In-labeled aggregated HSA for SPECT. (B) $^{99m}$Tc-labeled aggregated HSA for SPECT. (C) Schematic description of the preparation of self-cross linked HSA nanoparticles. Reprinted with the kind permission from refs 84. Copyright 2007 John Wiley & Sons, Ltd. (D) HSA coated iron oxide nanoparticles as multiple functional theranostic platform. Reprinted with permission from refs 91. Copyright 2010 Elsevier B.V.

2.4. *In vivo* non-covalent HSA targeting

The three-dimensional crystal structure of HSA was solved in early 1990s (Figure 7). It is a heart-shaped protein with three homogeneous domains, and each domain is composed of two subdomains that own the same structural motifs. Notably, HSA is one of the smallest proteins in human plasma. Both size and abundance explain the fact that the transportation of many metabolic compounds and therapeutic drugs is related to HSA by non-covalent binding. These HSA ligand-binding pockets are a series of hydrophobic cavities in subdomains II and III.
Indeed, the design of HSA-binding molecules is mainly based on the structures of binding pockets, which is also the key to determine the physical performance of HSA.

Figure 7. Crystal structure of albumin illustrating (A) small molecule binding site 1 and site 2 and (B) fatty acid (FA) binding site. Reprinted with the kind permission from refs. Copyright 2005 Elsevier Ltd.

Little was known about the variety of binding sites of HSA until an interesting study was reported in 1975, which was about the surprisingly different binding affinities of a number of fluorescent molecules for HSA. Changing the side chain on amino acid moiety of the darisylamino acids was found to substantially affect the binding of these compounds to HSA. In fact, the binding of the darisylamino acids to HSA varied both in the number of binding sites and in the binding tightness to these sites, suggesting that electrostatic and dipolar forces as well as steric factors play a role in both strength and specificity of binding. This study corroborates with the results of Chignell et al., who figured out based on circular dichroism measurements that the aromatic portion of flufenamic acid was inserted into a hydrophobic crevice on albumin while the carboxylate anion was associated with a cation that is around the gate of binding pocket. Overall, there are two high affinity binding sites for small heterocyclic or aromatic compounds
two to three dominant long-chain fatty acid binding sites (located on subdomains IB and IIIB), and two distinct metal-binding sites, making a total of six dominant areas of ligand association to albumin. In this part, we will elaborate on the chemistry of design, synthesis and screening of the small organic albumin-binding entities according to the specific binding sites.

2.4.1. HSA binding site 1

Binding site 1 is an essential pocket of HSA to carry and deliver small molecules in blood circulation. The interior environment of the pocket is predominantly apolar but is composed of two polar residues: an inner one towards the bottom and an outer polar residue near the entrance (Figure 8). Therefore, the molecules bind to pocket 1 generally contain a lipophilic aromatic structure in the middle and spherically surrounded by negative charges. Many dye molecules bind to domain II with high binding affinities (Table 2).

Figure 8 (A) Drug binding to site 1 in HSA (defatted). The detailed binding conformations is shown for 3-carboxy-4-methyl-5-propyl-2-furanpropanoic acid (CMPF), in which the drug is shown in a stick representation with a semi-transparent van der Waals surface. Sticks color-coded by atom type indicates selected side-chains; hydrogen bonds are shown as yellow dashed lines. (B) Top view of the superposition of CMPF bound to site 1 in defatted HSA.
Drugs are presented as a stick model with carbon atoms colored orange, nitrogen atoms in blue and oxygen atoms in red.\textsuperscript{85} Reprinted with permission from refs \textsuperscript{93}. Copyright 2005 Elsevier Ltd.

Table 2 Structure, binding affinity, number of binding for some classical binders to HSA binding site 1.\textsuperscript{93, 96, 97}

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Among them, Evans blue (EB) dye, as a good example, exhibits high affinity for binding site 1 on serum albumin. EB is an important tool in many physiologic and clinical investigations because of its high affinity for serum albumin, and has been used in clinical practice for almost 90 years as a way of determining patient plasma volume.\textsuperscript{98} By taking advantage of the high \textit{in vivo} binding affinity of EB to albumin, Niu \textit{et al.} developed a NOTA (1,4,7-triazacyclononane-N, N-triacetic acid) conjugate of a truncated form of Evans blue (NEB) for \textit{in vivo} albumin labeling. $^{18}$F-labeling was achieved by complexing with $^{18}$F-aluminum fluoride ($^{18}$F-AIF), and $^{68}$Ga and $^{64}$Cu labeling was accomplished through standard chelation chemistry (\textbf{Figure 9}).\textsuperscript{99-101}
Figure 9. (A) Schematic structure of supramolecular system of Evans Blue that binds to the site 1 on HSA. As shown, Evans blue dye exhibits strong tendency towards self-assembly to form stable, continuous, ribbon-like supramolecules when it binds to HSA. This self-assembling capability is also found to essentially correlate with the capacity of protein binding.\textsuperscript{102} Reprinted with permission from ref\textsuperscript{102}. Copyright 2000 John Wiley & Sons, Ltd. (B) Synthesis and \textsuperscript{18}F-AIF radiolabeling of NOTA(1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic acid)-trucated Evans blue conjugate (NEB).\textsuperscript{99}

2.4.2. HSA binding site 2

Different from site 1, binding site 2 has a single main polar patch, located close to one side of the entrance of the binding pocket (Figure 10). Based on the protein docking study, the hydrophobic binding cleft is about 16 Å deep and about 8 Å wide in the albumin molecule with a cationic group located near the surface. Therefore, as shown in Table 3, most of the binders to site 2 are lipophilic carboxylate derivatives. Nevertheless, a negative charge is not required for the molecule that binds to site 2. For example, diazepam, a basic drug molecule that exists mainly in the un-ionized form at neutral pH, also binds with high affinity for site 2. The presence of a positive charge often precludes binding to site 2. As shown in Table 3, aliphatic amines with
chain lengths C-3 to C-12 do not have measurable binding to site 2 although fatty acids with the same side chains are micromolar binders to the same binding pocket.\textsuperscript{93}

\textbf{Figure 10}. (A) Binding of indoxyl sulphate to site 2 in HSA. Indoxyl sulphate is shown in a stick representation with a semi-transparent van der Waals surface. Color-coding is the same as in Figure 8. (B) Top view of the superposition of indoxyl sulphate bound to site 2 in HSA along with a semi-transparent surface.\textsuperscript{85} Reprinted with permission from ref \textsuperscript{93}. Copyright 2005 Elsevier Ltd.
Recently, Neri *et al.* reported a class of 4-(p-iodophenyl)butyric acid derivatives that display stable non-covalent interaction with binding site 2. These HSA-binding tags were selected based on the strategy of Systematic Evolution of Ligands by Exponential Enrichment (SELEX). The candidate pool is a DNA-encoded chemical library with more than six hundred oligonucleotide-compound conjugates. After selection, the DNA sequences of stronger albumin binders were amplified by PCR and decoded on oligonucleotide microarrays. The corresponding signal intensities were normalized after selection against the intensities of compounds selected on empty resin (*Figure 11A-B*). The selected HSA-binding molecules are listed in *Figure 11C*. Interestingly, some of the selected HSA-binding molecules are structurally similar and featured by the basic structure of a 4-phenylbutanoic acid moiety, with different hydrophobic substituents on the phenyl ring (*Figure 11C*). Notably, one of these HSA-binding tags has been applied into

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several pharmaceutical systems to tune their clearance from blood circulation, such as elongation of the pharmaceutical profile of fast clearing drugs (Figure 11D), improved performance of MRI contrast agents (Figure 11E), and reduced kidney uptake of radiotherapeutic drugs (Figure 21).103

Figure 11. (A) Microarray readout of the selections performed against inactivated resin and resin displaying HSA (right panel). The spots corresponding to the enriched compounds 428 and 539 are enlarged (center); (B)
Enrichment of compounds in selections for HSA binding (compound numbers are indicated). (C) Structures of the molecules identified as potential binders; (D) Pharmacokinetic studies of fluorescein (black), 428-d-Lys-FAM (blue), 622-d-Lys-FAM (red), and phenethylamine-FAM (green) after injection in two mice each. As shown, the plasma concentration time course of $^{177}$Lu-labeled MSA is listed here for comparison; (E) Fluorescein angiography images in mice were recorded over 1 h after injection of 50 nmol of fluorescein (top row) and 428-d-Lys-FAM (bottom row). Reprinted with permission from ref. Copyright 2008 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim.

2.4.3. Fatty acid modification for HSA binding

When Kendall accomplished HSA crystallization in 1941, he found that the product contained a small amount of free fatty acid (FA). In addition, other researchers noted that the lipids extracted from blood plasma contained small quantities of FA as well. In the following decades, multiple binding sites were found for FA, and the binding affinity of fatty acid for HSA is mildly strong with an association constant in the range of $10^{-4}$-$10^{-6}$ M$^{-1}$. As fatty acids are commercially inexpensive and can be readily attached to other pharmaceutical moieties, conjugating FA onto GLP (glucagon-like peptide) or insulin has been an effective way to develop long-acting antidiabetic therapeutics. For example, GLP-1 analog exendin-4 has been modified by two fatty acids: lauric acid (LUA, C12) and palmitic acid (PAA, C16) at its two lysine residues. The resulting FA-exendin-4 conjugates were tested as regulators of blood glucose to cure type 2 diabetes, and showed notably longer blood circulation profile over exendin-4 (Table 4). Additionally, the FA acylated insulin has also been developed as a long-circulating anti-diabetic drug. It binds at the long-chain fatty acid binding sites, but the binding affinity is lower than that of the free fatty acids and depends to a relatively small degree on the number of carbon atoms in the fatty acid. This FA insulin conjugate showed prolonged...
circulatory half-life, but FA modification is not applicable to a broad set of molecules because of its negative effect on solubility.

3. Medical applications based on HSA-conjugates or HSA-binding moieties

3.1. Blood pool imaging agents

3.1.1. The efficacy of MRI contrast agents is improved by HSA-binding

The interest in the investigation of the binding ability towards HSA of paramagnetic complexes based on Gd(III), the most used $T_1$ contrast agents, is driven by two main reasons. First, the pharmacokinetic and pharmacodynamic properties of a HSA-binding contrast agent can be essentially effected by HSA, as the contrast agent is usually administered intravenously while HSA is the most predominant protein in the blood. After binding, the blood clearance of the contrast agent will be slowed down, and consequently the blood half-life and intravascular retention, will be increased. Thus, HSA-binding has been primarily considered for the visualization of vascular structures and for detecting regions with abnormal vascular permeability. Also, HSA binding can significantly improve the efficacy of these agents because the water proton relaxation time is strongly dependent on the tumbling motion of the metal complex.

As described previously, the presence of hydrophobic moieties as well as hydrophilic negatively charged groups are the basic structural requirements for binding pocket 2 of HSA, most of the work in this field has been focused on the design of metal complexes matching such features.
Gadofosveset or MS-325 (trade name: Ablavar, (trisodium 2-(R)-[(4,4-diaryl cyclohexyl) phosphonooxymethyl]diethylene-triaminepentaacetatoaquogadolinium)) is a clinically approved gadolinium (Gd) based blood-pool MRI contrast agent (Figure 12A) as an aid in diagnosing aortoiliac occlusive disease in patients with known or suspected peripheral vascular disease (PVD) or abdominal aortic aneurysm (AAA). As a result of transient binding to HSA, Gadofosveset has ten times the signal-enhancing power of existing contrast agents as well as prolonged retention in the blood (Figure 12B and 12C). This enables rapid acquisition of high-resolution magnetic resonance angiography (MRA) using standard MRI machines. Moreover, HSA binding offers an additional benefit beyond localization in the blood pool. The contrast agent begins to spin much more slowly, at the rate albumin spins, causing a relaxivity gain that produces a substantially brighter signal than would be possible with freely circulating gadolinium (Table 4).

Table 4 Relaxivity for Ablavar and other Gadolinium (III) complexes

<table>
<thead>
<tr>
<th>Name</th>
<th>( r_1 ) (mM(^{-1}) s(^{-1}))</th>
<th>Temperature (°C)</th>
<th>pH</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>TREN-1-Me-3,2-</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>HOPO</td>
<td>10.5</td>
<td>25</td>
<td>7.4</td>
<td>130</td>
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<tr>
<td>DTPA</td>
<td>4.3</td>
<td>25</td>
<td>7.4</td>
<td>131</td>
</tr>
<tr>
<td>DOTA</td>
<td>4.2</td>
<td>25</td>
<td>7.4</td>
<td>131</td>
</tr>
<tr>
<td>DTPA-Bisamide</td>
<td>4.58</td>
<td>25</td>
<td>7.4</td>
<td>131</td>
</tr>
<tr>
<td>DO3A</td>
<td>4.8</td>
<td>25</td>
<td>7.4</td>
<td>131</td>
</tr>
<tr>
<td>MP-2269</td>
<td>6.2</td>
<td>40</td>
<td>7.4</td>
<td>131</td>
</tr>
<tr>
<td>Ablavar</td>
<td>6.6</td>
<td>25</td>
<td>7.4</td>
<td>127</td>
</tr>
</tbody>
</table>
Figure 12. (A) Chemical structure of Gadofosveset. (B) Observed longitudinal ($r_1^{\text{obs}}$, circles) and transverse ($r_2^{\text{obs}}$, squares) relaxivity for 0.1 mM Gadofosveset in the presence (filled symbols) and in the absence (open symbols) of 22.5% (w/v) HSA at 37 °C, phosphate-buffered saline, pH 7.4. (C) Observed longitudinal ($r_1^{\text{obs}}$, circles) and transverse ($r_2^{\text{obs}}$, squares) relaxivity for 0.1 mM Gadofosveset in the presence (filled symbols) and in the absence (open symbols) of 22.5% (w/v) HSA at 37 °C, phosphate-buffered saline, pH 7.4. Reprinted with permission from ref\textsuperscript{127}. Copyright 2002 American Chemical Society (D-G) Comparable coronal projections of (D) conventional angiography, (E) gadofosveset enhanced MR angiography, (F) two-dimensional TOF MR angiography, and (G) a transverse reconstruction of a steady-state gadofosveset dataset showing stenoses (arrows) in both right and left common iliac arteries.\textsuperscript{132} Reprinted with permission from ref\textsuperscript{126}. Copyright 2007 Radiological Society of North America.

The extended blood half-life of Gadofosveset also results in longer time period for imaging, which allows the radiologist to perform multiple imaging experiments and to image under steady-state conditions (Figure 12G as an example).\textsuperscript{58, 133-135} In addition to imaging peripheral vascular disease and coronary artery disease (Figure 12D-12F), current trials are being
conducted to evaluate Gadofosveset as an aid in diagnosing breast cancer and to identify myocardial perfusion defects with delayed high-resolution imaging. \(^{136-138}\)

Another promising case of developing HSA-binding MRI contrast agent, 428-d-Lys-\(\beta\)-Ala-DTPA-Gd (Figure 13A) was contributed by the Neri group, which also targets binding site 2. \(^{104}\) The dissociation constant of 428-d-Lys-\(\beta\)-Ala-DTPA-Gd to HSA was determined by ITC at 37 \(^\circ\)C (\(K_d = 3.3 \mu\)M, Figure 13B), while Gd-DTPA had negligible binding to HSA. Pharmacokinetic profiles were studied in mice by injecting DTPA and 428-d-Lys-\(\beta\)-Ala-DTPA complexed with \(^{177}\)Lu, thus allowing quantification by gamma-counting. Similar to the situation encountered with the fluorescein derivatives, the plasma concentration of DTPA-\(^{177}\)Lu decreased rapidly and was no longer detectable at 60 min after injection, whereas 428-d-Lys-\(\beta\)-Ala-DTPA-\(^{177}\)Lu displayed a substantially slower biphasic pharmacokinetic profile (Figure 13C; DTPA-\(^{177}\)Lu: \(t_{1/2} = 8.6\) min vs. 428-d-Lys-bAla-DTPA-\(^{177}\)Lu: \(t_{1/2} = 408\) min). The rapid extravasation of DTPA-Gd in comparison to 428-d-Lys-\(\beta\)-Ala-DTPA-Gd was also observed by MRI procedures following intravenous injection of the contrast agents. MRI analysis of major blood vessels of the brain revealed a slower decrease of signal intensities in those injected with 428-d-Lys-\(\beta\)-Ala-DTPA-Gd (Figure 13C-13E).
Figure 13. (A) Chemical structure of 428-d-Lys-β-Ala-DTPA-Gd. (B) $K_d$ value of 428-d-Lys-DTPA-Gd to HSA determined by isothermal titration calorimetry (ITC) at 37 °C. (C) Pharmacokinetic studies of DTPA-\textsuperscript{177}Lu (filled symbols) and 428-d-Lys-bAla-DTPA-\textsuperscript{177}Lu (empty symbols) after i.v. injection in mice. The plasma concentration time course of \textsuperscript{177}Lu-labeled mouse serum albumin is given for comparison. (D) Transverse MR images of the mouse head indicating the region of interest (ROI) used to select the blood vessel. (E) Time course of the MR signal intensity in the ROI after injection of Gd-DTPA (left panels) and 428-d-Lys-b-Ala-DTPA-Gd (right panels).\textsuperscript{104} Reprinted with permission from ref\textsuperscript{104}. Copyright 2008 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim.

3.1.2. Radiolabeled HSA as the blood pool imaging agents

Although many radiolabeled HSA derivatives have been developed as blood pool agents for radionuclide imaging,\textsuperscript{139-144} the true revolution came from the recent report of $^{18}$F-NEB (Figure
14A, NOTA conjugated truncated Evans Blue), of which the preparation has been described previously. Within a few minutes after tracer injection, $^{18}$F-NEB reached the highest SUV value in the blood. Afterwards, a slow but steady clearance of the radioactivity was observed from the blood, due to the turnover of albumin from blood circulation and slight dissociation of $^{18}$F-NEB from albumin. As shown in Figure 14, this in vivo labeling strategy can be applied to blood-pool imaging to evaluate cardiac function under both physiologic and pathologic conditions (Figure 14B, 14C and 14D). This method can also be used to evaluate vascular permeability in tumors, inflammatory diseases, and ischemic or infarcted lesions.

Soon after the establishment of $^{18}$F-NEB, a first-in-human study was successfully performed with $^{68}$Ga-labeled NEB (Figure 14A). After intravenous injection, majority of the radioactivity was retained in the blood circulation due to the stable interaction of $^{68}$Ga-NEB with serum albumin (Figure 14B). Dosimetry study confirmed the safety with acceptable absorbed doses by critical organs even with multiple injections for one patient.

Overall, as a blood pool imaging agent, the preliminary clinical studies of $^{68}$Ga-NEB demonstrate the value of differentiating hepatic hemangioma from other benign or malignant focal hepatic lesions. In addition, NEB can be easily labeled with different positron emitters of various half-lives and demonstrates promising pharmacokinetics in human, warranting further clinical applications of NEB-based PET tracers.
Figure 14. (A) Chemical structure of \( ^{18}\text{F}-\text{NEB} \) and \( ^{68}\text{Ga-NEB} \). (B) Series of maximum-intensity-projection PET images in normal mice after intravenous injection of either \(^{18}\text{F}-\text{ALF-NEB} \) or \(^{18}\text{F-FB-MSA} \). Each mouse received around 3.7 MBq of radioactivity. Images were reconstructed from a 60-min dynamic scan. (C) Time–activity curves of ROIs outlined over muscle, heart, liver, and bladder regions on \(^{18}\text{F}-\text{ALF-NEB} \) PET images. (D) Time–activity curves of ROIs outlined over muscle, heart, liver, and bladder regions on \(^{18}\text{F-FB-MSA} \) PET images. Reprinted with permission from refs \(^{99}\). Copyright 2014 Society of Nuclear Medicine and Molecular Imaging. (E) Multiple time-point whole-body maximum intensity projection PET images of a female healthy volunteer at 5, 10, 15, 30, 45, 60, 75, and 90 min after intravenous administration of \(^{68}\text{Ga-NEB} \). Reprinted with permission from ref\(^{101}\). Copyright 2014 Society of Nuclear Medicine and Molecular Imaging.
3.1.3. Labeled HSA for Lymph Node Mapping

Besides being a blood pool imaging agent, radiolabeled or fluorophore attached HSA is often used to noninvasively identify the lymph nodes for cancer diagnosis or guiding surgery.\textsuperscript{145-147} For instance,\textsuperscript{99m}Tc-HSA has been successfully applied to map sentinel lymph nodes for identifying the patients with melanoma and regional nodal micrometastasis, and exhibits statistically better concordance rate than the radiotracers without HSA-conjugation.\textsuperscript{148, 149}

As another good example, \textsuperscript{18}F-NEB (\textbf{Figure 14A}) has also been applied to accurately locate sentinel lymph nodes.\textsuperscript{100} After local injection, both \textsuperscript{18}F-NEB and EB form complexes with endogenous albumin in the interstitial fluid and allow for visualizing the lymphatic system. Positron emission tomography (PET) and/or optical imaging of LNs was performed in three different animal models including a hind limb inflammation model, an orthotropic breast cancer model, and a metastatic breast cancer model (\textbf{Figure 15}). In these three models, the LNs can be distinguished clearly by using the blue color and fluorescence signal from EB as well as the PET signal from \textsuperscript{18}F-NEB, suggesting that this combination of \textsuperscript{18}F-NEB and EB is potentially useful for mapping sentinel LNs and provide intraoperative guidance for clinical diagnosis.
Figure 15. (A) Longitudinal fluorescence imaging of the lymphatic system after hock injection of $^{18}$F-AIF-NEB/EB. LNs and lymphatic vessels are clearly visible. (B) Ex vivo optical imaging of LNs without skin. (C) Photograph of the same mice to show the blue color within the LNs. (D) Co-registration of optical image (Left) and PET image (Middle) to present the popliteal LNs, indicated by a white arrow. (E) Co-registration of optical image (Left) and PET image (Middle) to present the sciatic LNs, indicated by a white arrow. The mice were euthanized at 90 min after hock injection of $^{18}$F-AIF-NEB/EB and the skin was removed. Reprinted with permission from ref. Copyright 2015 National Academy of Sciences.

3.2. HSA as a regulating platform of managing blood sugar level

One of the most important clinical applications of HSA-binding strategy is to elongate the blood circulation of anti-diabetic drugs. Up to now, three HSA-binding anti-diabetic drugs have been approved by the U.S. FDA, and at least ten more candidates are under clinical tests.\textsuperscript{150-154}

Glucagon-like peptide (GLP)-1 is a 30–amino acid peptide hormone secreted from gut endocrine cells in response to nutrient ingestion that promotes nutrient assimilation through regulation of gastrointestinal motility and islet hormone secretion.\textsuperscript{155} Infusion of GLP-1 into
normal or diabetic human subjects stimulates insulin and inhibits glucagon secretion, thereby indirectly modulating peripheral glucose uptake and control of hepatic glucose production, therefore can enhance GLP-1 action for the treatment of type 2 diabetes.\textsuperscript{156}

A major challenge for the therapeutic use of regulatory peptides, including native GLP-1, is a short circulating $t_{1/2}$, due principally to rapid enzymatic inactivation and/or renal clearance. Although infusion of native GLP-1 is highly effective in lowering blood glucose in subjects with type 2 diabetes, a single subcutaneous injection of the native peptide is quickly degraded and disappears from the circulation within minutes.\textsuperscript{157} Hence, the majority of pharmaceutical approaches to the development of GLP-1 mimetic agents have focused on the development of long-acting degradation-resistant peptides, such as Albugon, monoExendin-4 HSA (E1HSA), bisExendin-4 HSA (E2HSA), and so on. The pharmaceutical characteristics and pharmacokinetic properties of these HSA-binding or HSA containing anti-diabetic drugs are summarized in \textbf{Table 5}.

Albugon, or E1HSA, is a recombinant exendin-4-human serum albumin (HSA) fusion protein which retains the GLP-1 receptor binding activity of exendin-4 and as such is expected to exert glucose lowering effects with a prolonged duration (\textbf{Figure 16A}).\textsuperscript{158} In order to effectively bind to GLP-1 receptor, HSA was fused at the C-terminus of Ex4 and there is a 5-aa linker (GGGGS) was inserted between them.\textsuperscript{159} To determine the \textit{in vivo} bioactivity of E1HSA, an oral glucose tolerance test (OGTT) was performed in diabetic db/db mice by a single injection of E1HSA. As shown in \textbf{Figure 16B-16D}, glucose tolerance in diabetic db/db mice was effectively improved by E1HSA over the control group. In addition, an obvious dose–effect relationship was observed between the postdose serum glucose concentration and injection dose. Moreover,
postdose time-course observation indicated that the glucose-lowering effect of E1HSA lasted for at least 24 hours.\textsuperscript{57}

Table 5. Summary of HSA-based blood glucose regulatory drugs. AUCs were extrapolated by the formula of $AUC = \frac{1}{2} \times [(2 \times \text{biological half-life value} + \text{peak time value}) \times \text{peak effect value}]^{20, 57, 113, 152, 160-162, 211}$

<table>
<thead>
<tr>
<th>Sample</th>
<th>Chemical Construction</th>
<th>$t_{1/2}$ (h)</th>
<th>CL/F (mL/h\textsuperscript{-1}•kg\textsuperscript{-1})</th>
<th>Vd (mL•kg\textsuperscript{-1})</th>
<th>C\textsubscript{max} (ng/mL)</th>
<th>AUC (ng•h•mL\textsuperscript{-1})</th>
<th>MRT (h)</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>Exenatide</td>
<td>Peptide</td>
<td>0.58 ± 0.09</td>
<td>861.2 ± 164.0</td>
<td>1534.8 ± 415.9</td>
<td>0.20 ± 0.06</td>
<td>0.39 ± 0.12</td>
<td>1.39 ± 0.15</td>
<td>20</td>
</tr>
<tr>
<td>Albugon</td>
<td>Peptide HSA perfusion</td>
<td>108 ± 13</td>
<td>14.3 ± 2.8</td>
<td>91.7 ± 13.6</td>
<td>614 ± 136</td>
<td>175000 ± 37000</td>
<td>94.8 ± 24.0</td>
<td>152, 160, 161</td>
</tr>
<tr>
<td>E1HSA</td>
<td>Peptide HSA perfusion</td>
<td>56.7</td>
<td>2.1</td>
<td>102</td>
<td>N.A.</td>
<td>206815.9</td>
<td>66.7</td>
<td>162</td>
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<tr>
<td>E2HSA</td>
<td>Double peptide HSA perfusion</td>
<td>53.4 ± 8.0</td>
<td>1.66 ± 0.27</td>
<td>125.5 ± 8.8</td>
<td>1810.1 ± 198.7</td>
<td>179182 ± 27148</td>
<td>78.3 ± 6.2</td>
<td>57</td>
</tr>
<tr>
<td>LUA-MI</td>
<td>Peptide fatty acid conjugates</td>
<td>4.0 ± 0.5</td>
<td>6.61 ± 2.06</td>
<td>38.87 ± 15.30</td>
<td>892.66 ± 249.31</td>
<td>6940.4 ± 2571.5</td>
<td>7.85 ± 0.32</td>
<td>113</td>
</tr>
<tr>
<td>Ex-PEG (5kDa)</td>
<td>Peptide PEG conjugates</td>
<td>6.1 ± 0.8</td>
<td>37.5 ± 13.5</td>
<td>310 ± 79.1</td>
<td>76.1 ± 14.5</td>
<td>101.0 ± 25.8</td>
<td>7.9 ± 0.3</td>
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<tr>
<td>Ex-PEG (20kDa)</td>
<td>Peptide PEG conjugates</td>
<td>49.4 ± 7.7</td>
<td>4.4 ± 0.4</td>
<td>272.9 ± 63.2</td>
<td>154.0 ± 5.3</td>
<td>893.5 ± 119.3</td>
<td>53.1 ± 0.6</td>
<td>211</td>
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<td>Ex-PEG (40kDa)</td>
<td>Peptide PEG conjugates</td>
<td>76.4 ± 7.4</td>
<td>2.3 ± 0.3</td>
<td>259.0 ± 61.6</td>
<td>148.1 ± 24.5</td>
<td>1780.7 ± 279.9</td>
<td>78.9 ± 0.4</td>
<td>211</td>
</tr>
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</table>

Figure 16 (A) Schematic structure of E1HSA; (B) E1HSA lowers the blood glucose in db/db mice by an oral glucose tolerance test (OGTT). Single dose of E1HSA (0.3, 1, and 3 mg/kg) or HSA (3 mg/kg) were injected
intraperitoneally in mice. OGTT was carried out at various times postdose to evaluate the duration of E1HSA action (B): 12 h; (C): 24 h; (D): 36 h. Values are expressed as means ± SE; n = 5 mice/group. Reprinted with permission from ref 58. Copyright 2007 European Peptide Society and John Wiley & Sons, Ltd.

3.3. HSA as carrier for precision cancer therapy

HSA has long been a versatile drug carrier for developing effective anti-cancer agents. Upon binding to HSA, both the pharmacokinetics and pharmaceutical profiles of chemotherapeutic drugs may be changed to give better drug delivery efficiency as well as less side effect. In this section, we summarize some of the recent developments in the field of HSA-drug conjugates, with the focus on chemotherapy and radiotherapy for cancers.

3.3.1 Albumin-bound drug nanoparticle increases the therapeutic index of conventional chemotherapy drugs

Figure 17. (A) Chemical structure of paclitaxel, which is a hydrophobic small molecule with poor solubility in the blood. (B) and (C) Representative structures of small drugs loaded albumin nanoparticles, the diameters of this
complex is between 80 to 150 nm with a mean value of 130 nm. Reprinted with permission from ref\textsuperscript{163}. Copyright 2010 Elsevier B.V. (D) Process of gp60-mediated transcytosis of albumin across the vascular endothelium. The endothelial transcytosis of albumin is started by binding to the 60-kDa glycoprotein (gp60) receptor on the cell surface. This interaction induces caveolin and results in invagination and pinching off of the endothelial cell membrane, thereby concentrating and transporting albumin complex into vesicular structures denoted as caveolae ("little caves").

In general, Paclitaxel and other chemotherapeutic drugs are hydrophobic and thereby have poor solubility in blood circulation (Figure 17A).\textsuperscript{2, 164} To solve the problem, organic agents including polyethyalted castor oil (Cremophor® EL) and ethanol are required in their clinical formulations as their vehicles.\textsuperscript{165, 166} Nevertheless, these vehicles often cause severe toxicities, requiring prolonged infusion or premedication to reduce the risk of hypersensitivity reaction. Interestingly, albumin binds to many types of hydrophobic molecules in a reversible manner and consequently can help to transport the drugs in the body.\textsuperscript{10, 167} Moreover, as an intrinsic protein carrier in the blood, utilizing albumin as the drug vehicle avoids the risk of hypersensitivity reaction caused by the artificial formulation, is thus capable of serving as a clinically safer platform to deliver hydrophobic drugs in the body.

Besides the reduced toxicity and less immunogenicity, albumin also assists the transportation of plasma constituents through endothelial cells via albumin receptor binding. Traditionally, only the unbound drugs were thought to be able to penetrate vascular wall via junctional gaps between endothelial cells.\textsuperscript{168} Nonetheless, a selective transportation mechanism was disclosed recently that albumin-bound molecules can across vascular endothelium through albumin transcytosis.\textsuperscript{169-171} This process, illustrated in Figure 17D, is thought to play a key role in delivering proteins across the vascular endothelium in order to meet the nutritional needs of cells. Because of its
abnormal requirement of nutrition, tumors often take higher level of albumin than healthy tissues, and thereby albumin-bound drugs can be delivered to tumor with better selectivity.

SPARC, which is short for secreted protein that is acidic and rich in cysteine, is an extracellular matrix glycoprotein that is essentially related to tumor metastasis. It has been shown to be overexpressed on cancer cells and associated with poor prognosis in a number of tumors.\textsuperscript{172} Interestingly, recent evidence suggests that albumin exhibits high binding affinity to SPARC,\textsuperscript{173-176} and the tumor secretion of SPARC also plays a key role for the high tumor uptake of albumin.\textsuperscript{177, 178} Therefore, SPARC-inducing effect accumulates albumin to the areas of tumor that may further improve the delivery efficiency for albumin bound drugs (Figure 17B and 17C).

Small molecular drugs loaded albumin nanoparticles can be prepared in a number of ways, namely, desolvation,\textsuperscript{179, 180} emulsification,\textsuperscript{181, 182} thermal gelation,\textsuperscript{183} nano-spray drying,\textsuperscript{184} nab-technology,\textsuperscript{163} and self-assembly.\textsuperscript{185} Here, we will mainly focus on nab\textsuperscript{TM}-Technology, which is a biologically interactive delivery system that uses the biochemical properties of albumin to increase drug delivery to tumors. The first commercial product using this technology, Abraxane (nab-paclitaxel), is a solvent-free, 130-nm albumin particle form of paclitaxel. An \textit{in vitro} experiment conducted with human lung microvessel endothelial cells indicated that the transportation of fluorescently labelled paclitaxel is about 4.2-fold greater rate across an endothelial cell monolayer when formulated as nab-paclitaxel than CrEL-paclitaxel (CrEL: Cremophor\textsuperscript{®} EL, polyethylated castor oil). Because of its comparatively better efficacy for cancer treatment, Abraxane was approved by FDA in 2005 for the treatment of breast cancer cases where cancer did not respond to other chemotherapy (Figure 18). In 2012 and 2013, Abraxane received approval from FDA to be used for the treatment of non-small cell lung
cancer (NSCLC) as well as advanced prostate cancer because of its less toxicity during the treatment.

Figure 18. (A) Time to disease progression in a Phase III comparative trial of nab-paclitaxel versus CrEL-paclitaxel. Reprinted with permission from ref\textsuperscript{164}. Copyright 2005 American Society of Clinical Oncology. (B) Better efficacy of albumin-bound paclitaxel, compared with polyethylated castor oil-based paclitaxel in women with metastatic breast cancer. Reprinted with permission from ref\textsuperscript{163}. Copyright 2010 Elsevier B.V.

### 3.3.2. HSA-nanoparticle (NP) complex as a theranostic platform for diagnostic imaging and small molecular drugs delivery

In the past decade, HSA-NP complex has been developed as a common nanoplatform with both imaging and therapeutic functions, denoted as “nanotheranostics”\textsuperscript{2, 26-28, 85-90}. For instance, after coupling with targeting ligands and imaging moieties, iron oxide nanoparticles (IONPs) can provide many potential applications including multimodality imaging and therapy. In addition, HSA coated nanoparticles generally give reduced accumulation in mononuclear phagocytic system-related organs over the naked nanoparticles\textsuperscript{85, 87-91, 186}. In a pilot study, doxorubicin (Dox) was encapsulated into the HINPs (HSA coated iron oxide nanoparticle). About 0.5 mg of Dox and 1 mg of IONPs (iron oxide nanoparticles) could be loaded based on 10 mg of HSA matrices. The resulting D-HINPs (Dox loaded HINPs) could release Dox in a sustained fashion and
effectively suppressed tumor growth that was much better than free Dox on a 4T1 murine breast cancer xenografts model.\textsuperscript{186}

This strategy was then extended to load other types of small molecules and to build a multimodal-imaging platform (\textbf{Figure 19}). This combinational MRI/PET/NIRF theranostics nanosystem is capable of integrating the strengths of high anatomical resolution (MRI), \textit{in vitro} validation (NIRF), quantitative evaluation (PET) and cancer treatment, therefore can be a platform technology in theranostics.\textsuperscript{27, 89-91, 187}

\textbf{Figure 19.} (\textbf{A}) A brief scheme to describe the preparation of albumin-coated IONP. Reprinted with permission from ref\textsuperscript{87}. Copyright 2012 American Chemical Society (\textbf{B}) MR images taken before, and 1 and 4 h after the injection of NPs (6 mg of Fe per mL). As illustrated here, the contrast enhancement was decreased from 26.1\% to 5.2\% and then 4.3\% at 0 h, 1 h and 4 h p.i., which was the result of tumor accumulation of HINPs. Reprinted with permission from ref\textsuperscript{87}. Copyright 2011 American Chemical Society. (\textbf{C}) Schematic illustration of the multi-functional HSA-IONPs. (\textbf{D}) Representative \textit{in vivo} NIRF images of mouse injected with HSA-IONPs. Images were acquired 1 h, 4 h and 18
h post injection. (E) In vivo PET imaging results of mouse injected with HSA-IONPs. Images were acquired 1 h, 4 h and 18 hours post injection. (F) MRI images acquired before and 18 h post injection.\textsuperscript{87, 89, 90, 186} Reprinted with permission from ref\textsuperscript{91}. Copyright 2010 Elsevier B.V.

### 3.3.3. Small molecule HSA conjugates for cancer chemotherapy

The first HSA-drug conjugate that was evaluated in clinical trials was a HSA-conjugated chemotherapeutic drug: methotrexate-HSA conjugate (MTX-HSA). A phase I study with 17 patients treated with weekly MTX-HSA\textsuperscript{77} found that two patients with renal cell carcinoma and one patient with mesothelioma responded to MTX-HSA therapy (one partial response, two minor responses). However, the clinical trial stopped at phase II as no objective response was seen with metastatic renal carcinoma.\textsuperscript{78} The failure was most likely attributed to the drawbacks of MTX-HSA, such as unclear chemical structure and unclear metabolic pathway of MTX-HSA.

By taking advantage of the in vivo maleimide-HSA conjugation strategy which was detailed in session 2.2, DOXO-EMCH was highly effective in preclinical tumor models (Figure 20). As expected, there was a pronounced difference between the levels of DOXO-EMCH and doxorubicin in the serum of MDA-MB-435 tumor mice. Good antitumor effect was achieved at 3 × 16 mg/kg doxorubicin equivalents and complete remission was found at 3×24 mg/kg. Notably, preliminary toxicity studies in nude mice showed that the maximum tolerated dose of DOXO-EMCH was approximately 4.5 times higher than that of free doxorubicin.\textsuperscript{188}
DOXO-EMCH entered clinical trial in 2007, and was renamed INNO-206 or Aldoxorubicin in 2008. The on-going clinical studies suggest that INNO-206 can be administered safely at higher doses in patients than free doxorubicin, resulting in better efficacy compared with the currently available anthracyclines to treat several types of cancer.\textsuperscript{188, 190, 191}

### 3.3.4. Radiolabeled HSA-conjugate for internal radiotherapy of cancer

In 2013, Shibili and his colleagues from ETH reported a strategy in which a DOTA-folate conjugate was coupled with a small molecule albumin binder, denoted as cm09. Radiolabeled
folic acid derivatives have been used for folate receptor (FR) targeted imaging and therapy.\textsuperscript{192-194} However, using folate-based radiopharmaceuticals for therapy has long been regarded as an unattainable goal because of the poor tumor-to-kidney uptake ratio. As known, the rapid clearance of DOTA-folate conjugates from the blood circulation is generally considered as an advantage over the other targeting strategies.\textsuperscript{195-197} It is because rapid clearance usually gives high tumor-to-background contrast therefore minimizes the exposure of major organs to the therapeutic probe.\textsuperscript{198-204} However, this pharmacokinetics is a double-bladed sword that is also responsible for the relatively low uptake of folate conjugates in tumor tissue and an extremely high accumulation of radioactivity in the kidneys.\textsuperscript{205-209} In addition, once folate conjugate is cleared from blood into renal system, most of them would be strongly trapped by folate-binding protein in the kidneys, and therefore the kidney uptake of folate will not decrease over time.\textsuperscript{209} To solve this problem, this group reasoned that a HSA-binding radiopharmaceutical could change this dissatisfying situation as prolonged blood circulation could improve the tumor uptake, and reduce the problematic renal accumulation of the DOTA-folate conjugates.\textsuperscript{182,183}

As shown in \textbf{Figure 21}, installation of an albumin-binding entity into the structure of a folate-based radioconjugate improved the overall tissue distribution significantly. Tumor uptake was doubled, and kidney retention was reduced to 30\% of the value obtained with folate conjugates without an albumin-binding entity. In addition, tumor growth inhibition was observed without radiotoxic side effects.\textsuperscript{103}
Figure 21. (A) Chemical structure of cm09. (B-C) SPECT/CT images of KB tumor–bearing mice injected with $^{177}\text{Lu}$-cm09 (B) and $^{177}\text{Lu}$-EC0800 (C). Accumulation of radioactivity was found in FR-positive tumors (white arrows) and kidneys (yellow arrows). Images show a significantly improved tumor-to-kidney ratio (1.0 vs. 0.2) at 1, 4, 24, and 72 h after injection in mice that received $^{177}\text{Lu}$-cm09, compared with mice that received $^{177}\text{Lu}$-EC0800. (D) Internal radiation therapy protocol. (E) Average relative tumor size over time under different treatment regimens. (F) Relative body weight of mice under different therapies. (G) Survival curves of mice from groups A–E. (A, dark blue) control group. (B, green) unlabeled cm09. (C, red) 1 x 20 MBq of $^{177}\text{Lu}$-cm09. (D, violet) 2 x 10 MBq of $^{177}\text{Lu}$-cm09. (E, light blue) 3 x 7
4. Conclusions and outlook

As the key circulating protein in the blood circulation, albumin has been an excellent delivery platform for a number of endogenous and exogenous compounds. It has also been used to extend blood half-life and reduce renal clearance of both imaging probes and therapeutic drugs.

An ideal albumin-binding imaging probe may not only have a slow clearance from the blood, but also truly reflect a clear biological pathway in the body, viz., the signal it provides needs to be correlated with circulation, metabolism and bioactivity of natural albumin. To accomplish this goal, the labeled albumin should be indistinguishable with natural albumin for in vivo bioactivity, thus the imaging tag should be small in size, free of charge and stable in vivo. In addition, detachment of imaging reporters from the imaging probe should be avoided as it often gives misleading information for clinical diagnosis, therefore the binding strategy to HSA has to be robust, covalent and irreversible, though some of the non-covalent binding strategies (e.g. Evans Blue NOTA derivatives) also gives promising results in the clinic. In addition, considering the clinical practice and operational simplicity, HSA-binding imaging probe would better be a small molecule with unambiguous definition of chemistry, consequently the in vivo targeting strategy will be the one of choices for the future development of HSA-binding imaging probes.

In addition, in order to develop a more convenient and possibly less expensive treatment for diabetes, a HSA-binding blood glucose regulator would ideally have the longest if possible glucose-lowering effect without apparent side effect to the patients. Meanwhile, to design a better HSA binding cancer therapeutic drug, the key here is to improve the tumor specificity,
meaning increasing the tumor uptake while reducing the unnecessary cytotoxicity on healthy tissues. If possible, on-site drug release would be preferred, as it may essentially reduce the side effect since lower therapeutic dose would be applied to the patients.

In conclusion, as \textit{in vitro} HSA conjugation chemistry has been well established, it is believed that the future of HSA-binding chemistry should focus on developing new \textit{in vivo} HSA binders, either covalent or non-covalent. In addition, when a functional moiety is covalently coupled to HSA, the nonspecific adsorption of small molecules onto HSA is hard to be removed or purified, which is always a concern but can be avoided by \textit{in vivo} targeting approaches. In the case of \textit{in vivo} covalent binding, a faster and more bio-orthogonal conjugation method is in great need to improve the efficiency and selectivity of the binding reaction. For \textit{in vivo} non-covalent binders, systematic chemical screening is necessary to develop a series of HSA binders toward different binding sites with various binding affinities. It is noteworthy that the strongest binder is not always in favor, as we may need a balance between blood retention and clearance in certain circumstances. Moreover, the space linker between HSA-binding moiety and the functional molecule also needs more comprehensive investigation, and the ultimate goal would be a linker design that does not compromise the function of the albumin binder as well as the molecules of interest.
References


