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Galen T. Craven^{‡a} and Rigoberto Hernandez^{*a}

The time-dependent geometrical separatrices governing state transitions in field-induced ketene isomerization are constructed using the method of Lagrangian descriptors. We obtain the stable and unstable manifolds of time-varying transition states as dynamic phase space objects governing configurational changes when the ketene molecule is subjected to an oscillating electric field. The dynamics of the isomerization reaction are modeled through classical trajectory studies on the Gezelter-Miller potential energy surface and an approximate dipole moment model which is coupled to a time-dependent electric field. We obtain a representation of the reaction geometry, over varying field strengths and oscillation frequencies, by partitioning an initial phase space into basins labeled according to which product state is reached at a given time. The borders between these basins are in agreement with those obtained using Lagrangian descriptors, even in regimes exhibiting chaotic dynamics. Major outcomes of this work are: validation and extension of a transition state theory framework built from Lagrangian descriptors, elaboration of the applicability for this theory to periodically- and aperiodically-driven molecular systems, and prediction of regimes in which isomerization of ketene and its derivatives may be controlled using an external field.

A purely geometrical picture of a chemical reaction, when available, can represent the most pertinent current understanding of the dynamics governing state transitions in activated events. For example, transition state theory (TST) and its variants constitute the most popular of the modern methods for prediction of chemical reaction rates. These theories rely on geometric information alone, requiring only knowledge of the respective potential energy surface to allow calculation of the rates of dynamical processes. As such, TST is a pillar of reaction dynamics and kinetics.^{1–8} In activated events, the interplay between rates and mechanisms of conversion from reactants to products can be controlled so as to bias the reactivity and selectivity of a reaction toward products of interest. The principal geometric structures needed to understand these conversions are invariant manifolds (reaction conduits) which form separatrices that distinguish between states.

In isolated Hamiltonian systems with two degrees of freedom, the relevant reaction dynamics can be understood through the unstable periodic orbit dividing surface (PODS) that separates reactant and product states.^{9–12} At higher dimensionality, the PODS can be generalized to a normally hyperbolic invariant manifold (NHIM).^{8,13–21} A NHIM has associated reaction cylinders (sta-

[‡] E-mail: gtcraven@gatech.edu

ble and unstable manifolds) in isolated systems at energies below bifurcation thresholds.^{22–25} Moreover, reactive events are mediated by these manifolds²⁶ on (sometimes bifurcating²⁷) reaction pathways attached to the NHIM. Thus, NHIMs are a powerful methodological tool for characterizing reactive events in Hamiltonian systems that are autonomous, that is conservative systems that are not subject to time-dependent forces.

In the case of thermally-activated ^{28–30} and field-induced reactions: ^{31–35} time-dependent normal form theory, ^{36,37} scattering theories, ^{38,39} and high-order perturbation theories ^{40,41} have provided insight into the geometric structure of externallyactivated processes. Control of state transitions in dynamical systems ^{42–44} can be achieved in nonstatistical processes, ⁴⁵ including those with fluctuating rates. ^{33,34,46,47} To overcome the intrinsic complexity in these mechanisms, theories built on stability analysis ^{48–51} of guiding phase space objects have been developed. ^{33,34,52,53} Of specific interest in this class of driven reactions are recent advances in the control of protein dynamics using lasers ^{54,55} and mechanical stress. ⁵⁶ Theories that describe the time-dependent geometry of such systems provide a method to obtain and manipulate the rate of transition in these, and other, time-dependent processes.

Recently, a method relying on the construction of a Lagrangian descriptor (LD) 57 has been proposed to reveal phase objects such as hyperbolic trajectories 58,59 and invariant manifolds. This methodology relies on computation and comparison of the arc

^a Center for Computational Molecular Science and Technology, School of Chemistry and Biochemistry, Georgia Institute of Technology, Atlanta, GA, USA;

^{*} E-mail: hernandez@gatech.edu

lengths in forward and backward time for sets of trajectories with initial conditions in regions of phase space that are relevant to the system dynamics. The measure of these LDs can vary greatly on each side of a phase space boundary and their application to time-dependent systems has provided insight into the structures governing dynamical evolution in ocean flow patterns⁵⁷ and other systems modulated by external fields.⁶⁰ In thermal environments, the mechanisms and intrinsic reaction conduits that drive time-dependent reactive phenomena can also be described by this methodology.⁶¹

Our focus here is to develop a theory for reaction dynamics in field-induced systems through application of LDs using isomerization of ketene as a model system. Specifically, a timedependent field is applied to the ketene molecule for which the time-dependent manifolds and hyperbolic trajectories governing reactivity are revealed through LDs. The gas-phase (isolated) ketene isomerization reaction has received attention both as a model to investigate geometrical properties 62-65 and for its physical importance to chemical processes such as the Wolff rearrangement.^{66–68} It is also a paradigmatic example reaction for the chemical control of interest to this work because fieldmodified reactivity of ketene has been observed.^{69,70} To illustrate the methodology, we construct the network of manifolds over varying field strengths and oscillation frequencies. At select field parameters we observe the onset of chaos, as characterized by entanglement of these manifolds. The results provide insight into the treatment of field-induced reaction dynamics and other activated events driven under dynamical load.

1 Model and Simulation Details

The isomerization of ketene:

$$H_2C^1 = C^2 = O \Longrightarrow O = C^1 = C^2 H_2$$
 (1)

proceeds through metastable oxirene and formylmethylene intermediates which participate in the generally accepted mechanism for the Wolff rearrangement.^{68,71} Using the structural calculations of Scott *et al.*,⁶⁸ Gezelter and Miller (GM) constructed a model potential energy surface for this unimolecular reaction.⁷² They identified normal and local modes associated with the coupled in-plane motion of O and H atoms as the principal reaction coordinate (q_1), and the out-of-plane vibration of the H atom in oxirene as a secondary coordinate (mode q_2).^{62,63,72} The potential of the multi-dimensional GM surface is

$$V(q_1, q_2) = V_1(q_1) + V_{\text{coup}}(q_1, q_2),$$
(2)

which contains a one-dimensional term (see Fig. 1) along the fundamental reaction coordinate q_1

$$V_1(q_1) = a_2 q_1^2 + a_4 q_1^4 + a_6 q_1^6 + c q_1^2 e^{-dq_1^2} , \qquad (3)$$

and a coupling term

$$V_{\rm coup}(q_1, q_2) = \frac{k_1}{2} \left(q_2 + \frac{d_1 q_1^4}{k_1} \right)^2.$$
 (4)

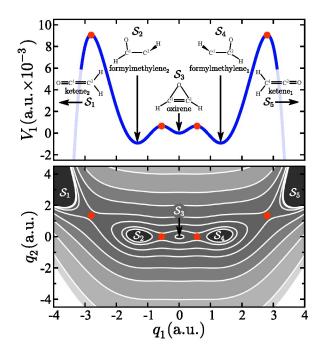


Fig. 1 The GM potential and corresponding structures for ketene isomerization along the principal reaction coordinate q_1 are shown in the top panel and a contour plot of the multidimensional surface in the $q_1 \times q_2$ plane is shown below. The position of each TS is shown as a circular marker (red) and each stable-state basin S_k is labeled.

Table 1 Parameters of the GM potential

Parameter	Value	
a_2	$-2.3597 \times 10^{-3} (E_{\rm h} a_0^{-2})$	
<i>a</i> ₄	$1.0408 \times 10^{-3} (E_{\rm h} a_0^{-4})$	
<i>a</i> ₆	-7.5496×10^{-5} ($E_{\rm h}a_0^{-6}$) 7.7569×10^{-3} ($E_{\rm h}a_0^{-2}$)	
с		
d	1.9769 (a_0^{-2})	
k1 (Model I)	$0 (E_h a_0^{-2})^{\circ}$	
k1 (Model II)	$1.0074 \times 10^{-2} \ (E_{\rm h} a_0^{-2})$	
d_1	$-2.45182 \times 10^{-4} (E_h a_0^{-5})$	

A projection of the GM potential onto the $q_1 \times q_2$ plane is shown in Fig. 1 and the system parameters are given in Table 1. Using the GM potential, we develop two separate models (I and II) for field-induced ketene isomerization: In Model I, the coupling term V_{coup} is ignored and an effective two-dimensional phase space is constructed by setting $k_1 = 0$. Model II is a four dimensional phase space model, constructed by including the coupling term and generalized velocity of mode q_2 .

Four transition states (TS) dividing surfaces exist along the fundamental reaction coordinate q_1 : the inner TSs ($q_1 = \pm 0.5467 a_0$) which separate the oxirene-formylmethlyene stable basins, and the outer TSs ($q_1 = \pm 2.8047 a_0$) which separate the formylmethylene-ketene basins.

The indices of these TSs used herein is given in Table 2. Further stability analysis of the TSs can be found in Ref. 65. The barrier separating oxirene and the inner TS has an activation energy $\Delta V = 0.00067$ a.u. and the formylmethylene to outer TS barrier has a larger activation barrier of $\Delta V = 0.010$ a.u. and thus formylmethylene is correspondingly the most stable of the the

Table 3 Dipole moments of various structures in ketene isomerization. All values are reported in atomic units

Structure	Computation	$\mu_{ m m}$	Experiment
oxirene	1.01	1.00	-
inner TS	1.05	1.05	-
formylmethylene	1.09	1.14	-
outer TS	1.18	1.13	-
ketene ($q_1 = 4.5 a_0$)	0.602	0.630	-
ketene $(\lim q_1 \to \infty)$	0.602	0.602	0.570 (Ref. 73), 0.560 (Ref. 74)

Table 2 Positions of TSs along the fundamental reaction coordinate q_1

Transition State	Position
TS ₁ (outer)	$-2.8047(a_0)$
TS ₂ (inner)	$-0.5467(a_0)$
TS ₃ (inner)	$0.5467(a_0)$
TS ₄ (outer)	$2.8047(a_0)$

two metastable structures.

Ketene is active in electric fields and susceptible to fieldinduced reactivity studies.^{69,70} We will approximate the fieldmatter interaction of ketene with a classical dipole approximation. To construct a dipole function, the geometry of each TS and stable-basin structure was taken from Ref. 68 in which Scott *et al.* performed electronic structure calculations at the CCSD(T)/6-311G(df, p) level. Using these geometries, the dipole moment of each was computed by employing a B3LYP/6-311+G** scheme. The results of these calculations are shown in Table. 3. The dipole moment along q_1 is modeled through an approximation of the form

$$\mu_{\rm m}(q_1) = \mu_0 \left[\exp\left(-\alpha (q_1 - q_0)^4 \right) \right] + \mu_0 \left[\exp\left(-\alpha (q_1 + q_0)^4 \right) \right] + \mu_{\rm ketene}.$$
(5)

where we assume that out-of-plane hydrogen motion in oxirene (q_2) has a negligible effect on the dipole moment in the pertinent dynamical regions and about the minimum energy pathway. The best-fit parameters for eqn (5) were obtained from least-squares regression to the calculated dipole moments, yielding $\mu_0 = 0.546 \ ea_0$, $\alpha = 0.0701 \ a_0^{-4}$, $q_0 = 1.95 \ a_0$, and $\mu_{\text{ketene}} = 0.602 \ ea_0$. A comparison between the results given by the model dipole moment surface μ and the computational results can be found in Table. 3.

With the potential energy and dipole moment surfaces now defined, the molecular Lagrangian for ketene can be written as

$$\mathscr{L} = \frac{1}{2}m_{\rm f}\dot{q}_1^2 + \frac{1}{2}m_{\rm H}\dot{q}_2^2 - V_1(q_1) - V_{\rm coup}(q_1, q_2) + \mathscr{E}(t)\mu_{\rm m}(q_1), \quad (6)$$

leading to the equations of motion

$$m_{\rm f}\ddot{q}_1 = -\frac{dV_1}{dq_1} - \frac{\partial V_{\rm coup}}{\partial q_1} + \mathscr{E}(t)\frac{d\mu_{\rm m}}{dq_1}, \tag{7a}$$

$$n_{\rm H}\ddot{q}_2 = -\frac{\partial V_{\rm coup}}{\partial q_2},\tag{7b}$$

where $\mathscr{E}(t)$ is the applied electric field. Associated with each coordinate (q_1, q_2) is the corresponding generalized velocity and thus the dynamics is either two dimensional (Model I) or four dimensional (Model II) with each phase space point represented by $\mathbf{q} = (q_1, \dot{q}_1)$ or $\mathbf{q} = (q_1, \dot{q}_1, q_2, \dot{q}_2)$, respectively. A constant-mass approximation associated with the corresponding normal mode of the oxirene intermediate has been calculated as $m_{\rm f} = 9581$ in atomic units (a.u.).⁶³ The mass of the out-of-plane oxirene stretch mode is $m_{\rm H} = 1837$ a.u., 63,65,72 the mass of hydrogen. The chosen field takes the sinusoidal form $\mathscr{E}(t) = \mathscr{E}_0 \sin(\omega t + \phi)$ where \mathscr{E}_0 is the field strength, ω is the oscillation frequency, and ϕ is the phase which is held constant ($\phi = \pi$). We investigate field strengths in range of those used previously in studies of field-induced diatomic chemical reactions,⁷⁵ and also strengths in range of ionization thresholds. The frequencies studied are below $\omega = 0.2$ a.u. (230) nm), a value previously used in experimental analysis of the decomposition of ketene.⁷⁰ Throughout this study, the integration of eqn (7a) and (7b) is performed using a Runga-Kutta fourthorder scheme with a time-step $\Delta t = 0.1$ (atomic units).

2 Lagrangian Descriptors

The application of LDs to the study of time-dependent flows has ranged over systems as diverse as thermal chemical reactions, ⁶¹ human biomechanics, ⁷⁶ and ocean flow patterns. ⁵⁷ In the general formulation, ^{57,60} a LD can be expressed as

$$M(\mathbf{q}_0, t_0)_{\tau} = \int_{t_0-\tau}^{t_0+\tau} \mathscr{P}(\mathbf{q}(t)) dt, \qquad (8)$$

where \mathscr{P} is a bounded positive quantity and is a functional of the trajectory $\mathbf{q}(t)$ with *initial* condition \mathbf{q}_0 at time t_0 . The integration of this quantity is taken over the time interval $[t_0 - \tau, t_0 + \tau]$ and thereby includes information from both forward- and backward-time. Thus, a LD is a mapping from the phase space point \mathbf{q}_0 to the scalar value M using dynamical information obtained along a corresponding trajectory. The LDs corresponding to arc length measures in phase space and configuration space are useful for constructing the manifolds associated with the TS.^{57,60,61} The values of these LDs diverge on either side of a phase space boundary separating distinct dynamical basins. These special boundaries are manifolds and they are distinguished by drastic changes in the derivative of M and correspond to local minima on the M surface.

The LD corresponding to the arc length projection in configuration space is

$$L(\mathbf{q}_{0},t_{0})_{\tau} = \int_{t_{0}-\tau}^{t_{0}+\tau} \left\| \dot{\mathbf{q}}_{c}(\mathbf{q}_{0},t_{0},t) \right\| dt,$$
(9)

where \mathbf{q}_c is a set of generalized coordinates and $\|\cdot\|$ is the norm. Thus, *L* is a measure of the distance traveled in configuration space over the time interval $[t_0 - \tau, t_0 + \tau]$ by the trajectory that passes through the phase space point \mathbf{q}_0 at time t_0 . This LD can be separated into forward-time

$$L_{\rm f}(\mathbf{q}_0, t_0)_{\tau} = \int_{t_0}^{t_0 + \tau} \left\| \dot{\mathbf{q}}_c(\mathbf{q}_0, t_0, t) \right\| dt, \tag{10}$$

and backward-time

$$L_{\mathsf{b}}(\mathbf{q}_{0},t_{0})_{\tau} = \int_{t_{0}-\tau}^{t_{0}} \left\| \dot{\mathbf{q}}_{c}(\mathbf{q}_{0},t_{0},t) \right\| dt,$$
(11)

components by integrating in the respective time direction. The value of τ is chosen to be long enough such that distinct features are observed on the LD surfaces. Note that the initial conditions of each trajectory $\mathbf{q}_0(t_0) = (q_1(t_0), \dot{q}_1(t_0), q_2(t_0))$ are phase space coordinates, while the integration of these LDs is performed over projections into configuration space. We use eqn (9)-(11) below to describe the geometry of ketene isomerization.

With the inclusion of field-dipole coupling, the system of equations (7a) and (7b) becomes time-dependent. The stationary transition states (See Table 2) now become time-dependent and generate a set of stable manifolds W^s , unstable manifolds W^u , and corresponding hyperbolic trajectories. The manifolds associated with each moving TS are also time-dependent. The direct construction of these objects can provide physical insight into the mechanism and nature of reactions in which moving energy barriers separate reactant and product states. The LD procedure, requiring forward and backward integration in time, thus provides information on how the past reaction geometry (described by the manifold network) evolves into the future geometry in these timedependent flows.

The two models summarized in Sec. 1 contain distinct topology due to differences in dimensionality. The complexity of the dynamics that occur in these topological spaces can be reduced by considering different sets of phase spaces and phase planes in which to perform the LD analysis. Model I has a single degree of freedom and its dynamics occur solely on the principal reaction coordinate q_1 and the corresponding $q_1 \times \dot{q}_1$ phase space. The measure of eqn (9)-(11) in Model I is taken over q_1 as it is the only configuration space component. The addition of a a second degree of freedom q_2 in Model II leads to the generation of additional planes in phase space, e.g., $q_2 \times \dot{q}_2$ and configuration space $q_1 \times q_2$. While other spaces exist, we find these planes as the most intuitive and pertinent in which to perform analysis. Alternatively, the complex reaction dynamics in systems with high dimensionality is often simplified by identifying suitable collective variable reaction coordinates and performing analysis within a reduced space. In Model II, we therefore consider LDs constructed using the arc length measure over q_1 space alone, as it is the putative reaction coordinate. This reductionist approach is particularly advantageous for chemical reactions addressed through LD analysis because the dynamics in the nonreactive modes play only a secondary role in activated events.

3 Dynamics

We first consider an autonomous model of ketene isomerization by removing the driving field (i.e. $\mathcal{E}_0 = 0.0$). In this system, the TS saddle points are connected by a network of heteroclinic and ho-

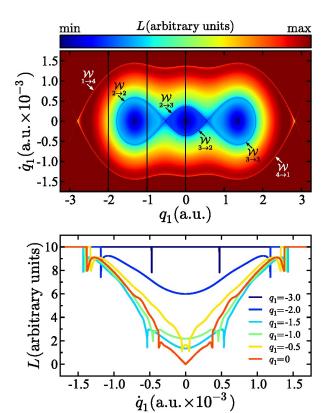
Fig. 2 (top) A contour plot of the Lagrangian surface *L* for autonomous $(\mathscr{E}_0 = 0.0)$ ketene isomerization in $q_1 \times \dot{q}_1$ space. The heteroclinic and homoclinic loops are marked as discussed in the text and correspond to distinctive features on *L*. (bottom) The values of *L* along the constant values of q_1 given in the legend. The location of the slices corresponding to $q_1 \in \{-2.0, -1.0, 0\}$ are shown as solid black lines in the top panel. All parameter values are given in atomic units. For visual clarity in this and all other figures illustrating LD surfaces, a threshold value limits the upper bound of *L* and corresponds to the "max" value shown in the deepest red.

moclinic loops. The heteroclinic loops consist of trajectories that approach different TSs in the infinite past and future. Homoclinic loops are made up of trajectories that approach the same TS in forward and backward time. These loops correspond to the stable (unstable in backward time) manifold for a given saddle point TS_i and are denoted as

$$\underset{i \to j}{W}(\mathrm{TS}_i)$$
 and $\underset{i \to i}{W}(\mathrm{TS}_i)$

respectively. The notation $i \rightarrow j$ signifies the heteroclinic connection from TS_i to TS_j and $i \rightarrow i$ signifies a homoclinic connection. Note that in backward time, the direction of the $i \rightarrow j$ connection is $j \rightarrow i$. These manifolds can be constructed by holding q_1 constant while minimizing the LD with respect to \dot{q}_1 , and/or finding the corresponding points where the derivative of the LD changes "abruptly."^{57,60} The manifolds are clearly visible in the top panel of Fig. 2 as strong local minimia on the *L* surface and the power of LDs is evident as the manifold structure in forward and backward time are revealed.

In this autonomous system, trajectories with initial conditions on hyperbolic fixed points result in $L_f = 0$, $L_b = 0$, and L = 0 by



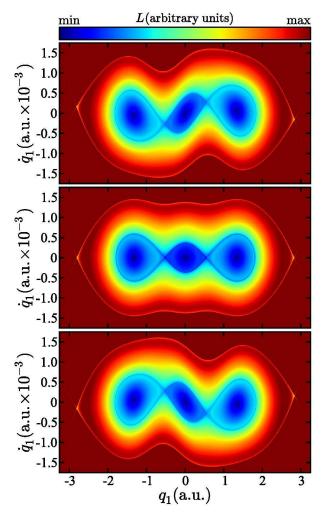


Fig. 3 Contour plots of the Lagrangian surface *L* for field-induced ketene isomerization (Model I) in $q_1 \times \dot{q}_1$ space with $\mathscr{E}_0 = 0.5$, $\omega = 0.0315$, $\tau = 10^3$, for $t_0 = 0$ (top), $t_0 = 75$ (middle), and and $t_0 = 100$ (bottom). All parameter values are given in atomic units.

definition, and these points correspond to TS saddle points. All other trajectories with initial conditions not on TS_i saddle points have LDs with positive values such that finding the minima allows locating the TSs. This behavior is illustrated by the $q_1 = 0$ curve in Fig. 2. The stable manifolds of these TSs can be constructed by considering forward-time integration only. For a chosen initial position (that is held constant), varying the initial velocity leads to trajectories with two different behaviors: those that are not on the stable manifold move away from the TS (sliding down the barrier) and thus have large values of $L_{\rm f}$ over sufficient propagation. However, on this line of constant initial position, there is one initial velocity that gives a trajectory that approaches the TS, remaining bounded for all time. This trajectory belongs to the stable manifold and will have the minimum value of $L_{\rm f}$, as it never descends from the region of the barrier top into the metastable energy wells. The unstable manifold can be constructed using similar arguments on the L_b surface, and backwards-time integration. In the bottom panel of Fig. 2, it can be seen that the stable (unstable) manifold of each TS exhibits distinctive features on the *L* surface.

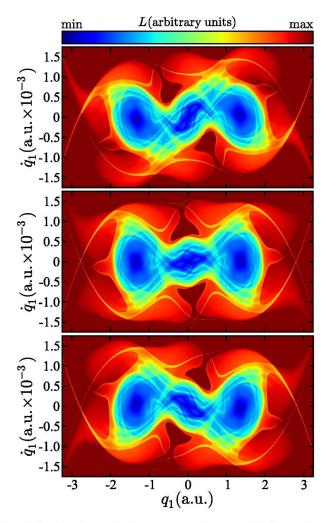


Fig. 4 Contour plots of the Lagrangian surface *L* for field-induced ketene isomerization (Model I) in $q_1 \times \dot{q}_1$ space with $\mathscr{E}_0 = 0.03$, $\omega = 0.0025$, and $\tau = 10^3$, for $t_0 = 0$ (top), $t_0 = 500$ (middle), and $t_0 = 1000$ (bottom). All parameter values are given in atomic units.

Upon inclusion of the external driving field, the network of manifolds becomes time-dependent. As shown in Fig. 3, this network exhibits oscillatory motion due to the periodicity in the driving form. The homoclinic and hetereoclinic loops associated with each TS persist, albeit in a time-varying form. This persistence would cease upon inclusion of dissipative forces, such as those experienced by a reactive species in a solvent environment. They do not however require periodic driving as the the arguments for uncovering the manifolds and TS using LDs did not rest on this this assumption. This opens the possibility for addressng aperiodic driving forms, e.g., thermal fluctuations, with the LD approach, and the resulting alterations in the reactive yields in chemical reactions.

At select values of field strength and oscillation frequency ($\varepsilon_0 = 0.03$ a.u. and $\omega = 0.0025$ a.u.), the onset of chaos is observed as characterized by complex stretching and contracting behavior in phase space leading to entanglement of the stable and unstable manifolds. This behavior can be seen in Fig. 4. At these field parameters, a bifurcation threshold has been crossed due to the appearance of additional hyperbolic points connected by stable

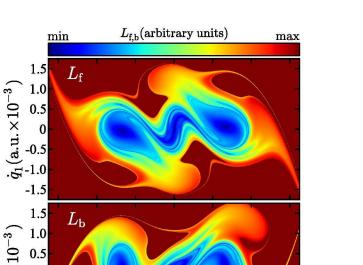


Fig. 5 Contour plots of the forward-time (top) and backward-time (bottom) Lagrangian surfaces $L_{\rm f,b}$ in $q_1 \times \dot{q}_1$ space for Model I with $\mathscr{E}_0 = 0.03$, $\omega = 0.0025$, $\tau = 10^3$, and $t_0 = 0$. All parameter values are given in atomic units.

-1

0

 $q_1(a.u.)$

1

2

3

 $\dot{q}_1({
m a.u.}\,{ imes}10^{-3})$

0

-3

-2

-0.5

-1

and unstable manifolds. While rigorous analysis of this complex structure is beyond the scope of this manuscript, formalisms have been developed to understand the induced phase space partitioning.^{77,78} Using the method of LDs, the complexity of this network can be simplified by constructing both the $L_{\rm f}$ and $L_{\rm b}$ surfaces and subsequent analysis of the dynamics in the respective time direction separately. As shown in Fig. 5, on these unidirectional time surfaces, the corresponding stable and unstable manifolds are visible as distinctive features in the relevant phase space regions. Moreover, the constructed surfaces $L_{\rm f}$ and $L_{\rm b}$ reveal the encapsulated reactive (nonreactive) islands dictating the flow from reactant to product. This method can be applied directly to ketene isomerization with other field parameters, and, most importantly, to other general chemical reactions, to understand how the network of manifolds, and the corresponding reaction dynamics, are affected by an applied field.

Multidimensional reactive systems can also be analyzed using the LD method. In Model II of Sec. 1, the out-of-plane motion of the H atom in oxirene is a a second degree of freedom (q_2) that complements the dynamics on principal reaction coordinate q_1 . The dynamics in these two coordinates are coupled through the potential energy term given by eqn (4). Following the multidimensional study of ketene isomerization rates by Gezelter and Miller in Ref. 72, Ulusoy and Hernandez, $^{62-64}$ and also Mauguière *et al.*, 65 used the GM model to examine the dynamics of ketene in a phase space perspective. In the multidimensional space, with inclusion of field-matter interactions, the onset of chaotic behavior can also be observed. As shown in Fig. 6, phase planes containing trajectories with higher initial en-

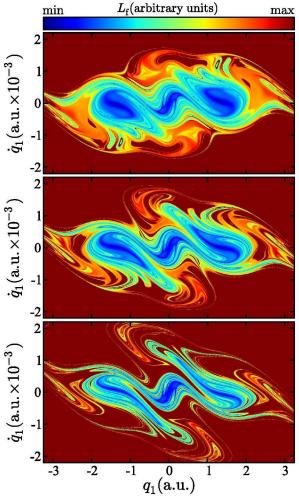


Fig. 6 Contour plots of the Lagrangian surface $L_{\rm f}$ for field-induced ketene isomerization (Model II) in $q_1 \times \dot{q}_1$ space with $\mathscr{E}_0 = 0.03$, $\omega = 0.0025$, $\tau = 2.5 \times 10^3$, and $q_2(t_0) = 0$ for $\dot{q}_2(t_0) = 0$ (top), $\dot{q}_2(t_0) = 0.001$ (middle), and $\dot{q}_2(t_0) = 0.002$ (bottom) with $t_0 = 0$. The measure of $L_{\rm f}$ is taken over q_1 space. All parameter values are given in atomic units.

ergy result in more dissociative-type transformations as the yield of ketene (dark red) increases. This general trend persists with increasing initial velocity in q_1 . The origin of this behavior is that larger values of $\dot{q}_1(t_0)$ increase the initial energy of each trajectory. Thus, the likelihood of a trajectory escaping from the intermediate oxirene and formylmethylene metastable wells is also increased as energy from the velocity space is transferred into the reaction coordinate. Note the the measure of $L_{\rm f}$ is taken over q_1 only. Although not shown, we also observed improved ketene yields with increasing q_2 and \dot{q}_2 . In this case, the energy transfer into q_1 occurs through coupling on the potential energy surface.

4 Reactivity and Final-State Basins

The reactivity and selectivity of ketene isomerization, dynamically modeled through eqn (7a) and (7b), can be examined by mapping initial conditions in phase space to the corresponding final state basin. This procedure involves taking a set of coordinates in the multidimensional phase space as the initial condition for the equations of motion, and following the evolution of tra-

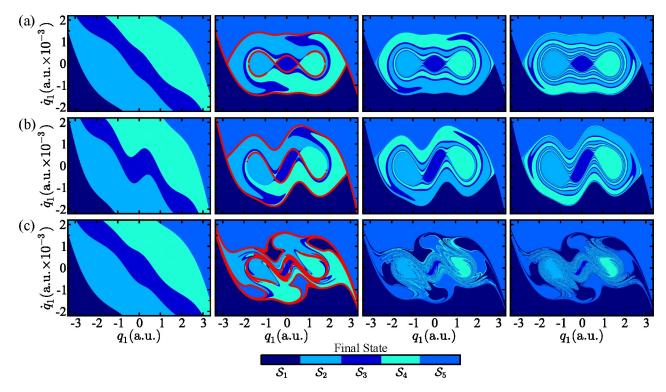


Fig. 7 Final state basins in $q_1 \times \dot{q}_1$ space for Model I and (a) ($\mathscr{E}_0 = 0$), (b) ($\mathscr{E}_0 = 1.0, \omega = 0.0315$), and (c) ($\mathscr{E}_0 = 0.03, \omega = 0.0025$). The initial time is $t_0 = 0$ in all panels. From left to right, the integration times are $\tau \in \{10^3, 10^4, 2.5 \times 10^4, 5.0 \times 10^4\}$. In all panels corresponding to $\tau = 10^4$, pieces of the stable manifold network, as calculated using the method of LDs described in Sec. 2, are shown in red. All parameters values are given in atomic units.

jectories to some final time in order to determine what regions in phase space lead to which products. A large number of trajectories are often necessary in order to adequately sample the basin boundaries as initial conditions are, most commonly, generated using an equally-spaced grid. Creating the mapping from phase space points to final states is analogous to the brute-force computation of basins of attraction in dynamical systems.

For the models of ketene isomerization defined in Sec. 1, we can identify five stable state regions separated by TSs on the GM energy surface. The one-dimensional coordinate q_1 parameterizes the curved reaction coordinate well and can serve as an indicator of the final state as illustrated in Fig. 1. A mapping of each position \mathbf{q}_0 to a given final state S_k can thus be constructed through an indicator function

$$S(\mathbf{q}_{0}(t_{0}), \tau) = \begin{cases} S_{1}, & q_{1}(\tau) < \mathrm{TS}_{1}, \\ S_{2}, & q_{1}(\tau) > \mathrm{TS}_{1} \text{ and } q_{1}(\tau) < \mathrm{TS}_{2}, \\ S_{3}, & q_{1}(\tau) > \mathrm{TS}_{2} \text{ and } q_{1}(\tau) < \mathrm{TS}_{3}, \\ S_{4}, & q_{1}(\tau) > \mathrm{TS}_{3} \text{ and } q_{1}(\tau) < \mathrm{TS}_{4}, \\ S_{5}, & q_{1}(\tau) > \mathrm{TS}_{4}, \end{cases}$$
(12)

according to the location of the final position of the q_1 -mode of the trajectory $\mathbf{q}_0(\tau)$.

For Model I, the state mapping encoded by eqn 12 is shown in Fig. 7 over various parameter values of the driving field and increasing integration times. Only forward-time integration is considered, and thus the reactive regions will be separated by stable manifolds of the TSs. In the nonchaotic regimes of Figs. 7(a) and7(b), the we observe an the regular structure seen earlier for

the manifolds obtained using LDs. In Fig. 7(c), the reactive islands observed through direct calculation of the final state basins are seen to correspond to regions enclosed by the same stable manifolds observed on the forward-time LD surface $L_{\rm f}$ in Fig. 5. Also note that in Fig. 7(c) there is clear evidence of chaotic mixing which was previously posited using information from the LD surface only. As shown in all panels in Fig. 7 corresponding to integration times $\tau = 10^3$ a.u., the stable manifold network, calculated using the method of LDs described in Sec. 2, is in excellent agreement with boundaries that separate regions of different selectivity. This result illustrates that the manifold network dictating reactivity in chemical reactions can be exposed using LDs.

Final state probabilities $P(S_k)$ for several parameter ranges in Model I are shown in Figs. 8(a)-(c). These probabilities are calculated by counting the fraction of trajectories starting in region $\mathscr{R} = [-3.6 \text{ a.u.}, 3.6 \text{ a.u.}] \times [-0.0022 \text{ a.u.}, 0.0022 \text{ a.u.}]$ in phase space that lead to the final states S_k for $k \in \{1, 2, 3, 4, 5\}$ according to the mapping in eqn 12. In Fig. 8(a), the reaction probabilities for the field strength $\mathcal{E}_0 = 1.0$ a.u. are illustrated over varying frequencies. A distinct turnover regime in which oxirene is favored over formylmethylene is observed over the frequencies $\omega \in [1 \text{ a.u.} \times 10^{-5}, 2.5 \text{ a.u.} \times 10^{-5}]$ and also for $\omega \in [7 \text{ a.u.} \times 10^{-5}, 9 \text{ a.u.} \times 10^{-5}]$. Figure 8(b) illustrates reaction probabilities for a reduced field intensity $\mathscr{E}_0 = 0.2$ a.u. over the freqency regime $[1 \text{ a.u.} \times 10^{-3}, 1 \text{ a.u.} \times 10^{-2}]$. For this field strength and frequency $\omega = 0.001$ a.u., almost all trajectories lead to the production of ketene. At an order of magnitude larger frequency $(\omega = 0.01 \text{ a.u.})$, the yields of the autonomous system are ap-