RSC Advances



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. This Accepted Manuscript will be replaced by the edited, formatted and paginated article as soon as this is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/advances

Journal Name

COMMUNICATION

2'-N-Guanidino,4'-C-Ethylene Bridged Thymidine (GENA-T) modified Oligonucleotide Exhibits Triplex Formation with Excellent Enzymatic Stability

Jharna Barman,^a Deepanjali Gurav,^{b,c} Oommen P. Oommen^b and Oommen P.

Received ooth January 2012, Accepted ooth January 2012

Cite this: DOI: 10.1039/xoxxooooox

DOI: 10.1039/x0xx00000x

www.rsc.org/

Here we present the synthesis and characterization of new 2'-*N*-Guanidino,4'-*C*-Ethylene Bridged Thymidine (GENA-T) modified oligonucleotide possessing North-locked sugar conformation. Incorporation of GENA-T nucleotide though did not change the thermal stability of the oligonucleotides toward the complementary RNA; it has significantly increased the stability of parallel triplex at pH 7. The melting temperature of the triplex was increased by + 9.5 °C as compared to that of the isosequential unmodified sequence. Moreover this modification imparted exceptional nuclease stability to the oligonucleotides for over 33 h. This study clearly demonstrates that GENA-T modified oligonucleotides could improve triplex formation with phenomenal enzymatic stability and could be used for various biomedical applications.

Varghese^b*

Chemically modified nucleotides have generated enormous interest in recent years as they improve the enzymatic stability and efficacy of oligonucleotides, transforming them into drug molecules for various therapeutic applications such as antisense, antigene, siRNA, microRNA, aptamer etc.¹ Of particular interest has been North-type (C3'-endo) conformationally locked nucleotides. namely. Locked/Bridged Nucleic Acid (LNA/BNA). These modified nucleotides have an extra ring fused to the furanose structure at 2'-4' position.^{2,3} These types of 2'-4' fused LNA analogues increase oligonucleotide (ON) rigidity which enhances the binding toward complimentary sequence by either preorganization or improved stacking (entropy or enthalpy).^{4,5} This stabilization can also be attributed to the presence of heteroatom at the 2' position which influences the pseudoaromatic character of the nucleobase, as evident from the change in nucleobase pK_a^{6} LNA analogues have been used for different gene silencing applications such as antisense,⁷ siRNA⁸ and miRNA⁹ therapeutics in vivo. These modifications are also known to favour triplex formation when incorporated into triplex forming oligonucleotides (TFOs) for genetargeted mutagenesis as in antigene approach.¹⁰ DNA TFOs bind purine-rich strand of a target double stranded DNA (dsDNA) either by parallel motif, as in pyrimidine-rich TFOs, or by antiparallel

motif, as in polypurine TFOs. Though this technology is very promising, one challenge that remains elusive is to make these therapeutic ONs more resilient towards enzymatic degradation without altering the backbone structure or compromising their efficacy. Enhancing nuclease resistance of ONs by changing the phosphodiester backbone to phosphorothioates¹¹ or N3'-->P5' phosphoramidates¹² is well established but often at the cost of non-specific binding to several proteins.¹³ We have earlier reported synthesis and properties of 2'-*N*,4'-*C* ethylene bridged thymidine analogue (aza-ENA-T) which showed significant improvement in enzymatic stability.¹⁴ Since the amino group (positive charge) near phosphate offers extra enzymatic stability (as observed in azetidine Vs oxetane comparison),¹⁵ we anticipated that converting amino residue of aza-ENA to guanidino group could impart interesting features to the existing properties of aza-ENA.

The unique ability of the guanidinium moiety to remain protonated at physiological pH ($pK_a \approx 12.5$) imparts interesting biological function when conjugated to ONs. For example, Sugimoto et al. demonstrated that incorporation of four guanidinium conjugated nucleobases in a 20mer oligonucleotide significantly enhanced cellular uptake.¹⁶ In another study, by Prakash et al., the 2'-amino and also corresponding guanidinium modified ONs have displayed the ability to bind dsDNA to form triplexes through the pyrimidine motif by Hoogsteen hydrogen bonding.¹⁷ Though TFOs are not approved for commercial use as drugs yet, these oligonucleotides have demonstrated great potential for therapeutic and biotechnological applications.¹⁸ Apart from the aforementioned problem of nuclease stability, TFOs face additional problem of intrinsic stability at physiological condition because of ionic repulsion of the three negatively charged DNA strands and the need for N3 protonation of cytosine nucleobase for G•C Hoogsteen Hbond formation.¹⁹ Only purine rich TFOs form stable triplexes at physiological pH.¹⁸ Since the presence of North-sugar puckering is known to favour Hoogsteen H-bonding,²⁰ we envisaged that TFOs modified with GENA could demonstrate good triplex formation at physiological pH along with enhanced enzymatic stability.

Synthesis of GENA derivative was initiated from the previously synthesized aza-ENA intermediate 1.¹⁴ This intermediate was synthesized following a modified strategy of our previously reported protocol (incorporation of cyano intermediate) as described in the

Page 2 of 4

Journal Name

supporting information (SI). Trifluoroacetyl group was deprotected using aqueous CH₃NH₂ in methanol and the product obtained was directly used for the guanylation reaction (Scheme 1). We initially attempted the guanylation reaction using PAC (phenoxyacetyl) protected SMe isothiourea. With this reagent guanylation reaction proceeded smoothly with 76% yield (as a mixture of rotamers). However, the corresponding phosphoramidate reaction was unsuccessful. This is presumably due to highly electrophilic imine bond of guanidine (due to amide protection), which undergo nucleophilic addition reaction with neighbouring trivalent P having non-bonding electrons. We circumvented this problem by using the carbamate protected SMe isothiourea, 2 having N-(2-(cyano)ethoxycarbonyl) or CEOC protecting group as reported by Manoharan et. al.¹⁷ With this reagent the guanylation reaction was sluggish (48% in two steps after 4 days as mixture of rotamers). However, the phosphitylation reaction went smoothly under standard conditions to give the desired phosphoramidite building block with 66% yield. This highlights the significance of carbamate protecting group in phosphoramidite chemistry. Oligonucleotide sequences were constructed by solid phase synthesis using phosphoramidite approach and its mass was confirmed by MALDI TOF spectroscopy (See Table 1 and 2). The CEOC protecting group was removed from ONs using 50% aqueous piperidine to avoid side reactions.²¹



Scheme 1. Synthesis of GENA–T phosphoramidite. Reagents: (i) 40% aqueous CH₃NH₂, CH₃OH, 3 h, rt; (ii) dry CH₂Cl₂, 2, Et₃N, 4 days, rt, (iii) *N*,*N* diisopropylphosphoramido chloridite, DIPEA, dry CH₂Cl₂, 2 h, rt.

Table 1. Melting temperature for the DNA/RNA hybrid duplexes and MALDI-MS analysis $^{\rm a}$

ON	Sequences (3'5')	T _m /° C With RNA	Δ <i>Τ</i> _m (°C)	MALDI-MS Calc/Found [M+H] ⁺
1	3'-d(CTTCTTTTTTACTTC)	44		4448.6/4448.7
2	3'-d(CTTCT <u>T</u> TTTTACTTC)	46.5	+2.5	4489.7/4490.7
3	3'-d(CTTCT <u>T</u> TTTTACTTC	44	0	4533.9/4533.8
4	3'-d(CTTCTTT <mark>T</mark> TTACTTC)	47.5	+3.5	4489.7/4490.7
5	3'-d(CTTCTTT <u>T</u> TTACTTC)	44	0	4533.9/4534.0
6	3'-d(CTTCTTTTT <u>T</u> ACTTC)	48	+4	4489.7/4490.8
7	3'-d(CTTCTTTTT <u>T</u> ACTTC)	45	+1	4533.9/4533.9

^a<u>T</u> = aza-ENA and <u>T</u> = *GENA*. Buffer and condition: Standard deviation in all the melting experiments were below \pm 0.4 °C. Buffer and conditions: 60 mM Tris-HCl pH 7.5, 60 mM KCl, 0.8 mM MgCl₂, and 2 mM DTT; strand concentration = 1μ M.

We evaluated the hybridization properties of GENA incorporated oligonucleotide with complementary ssRNA using UV melting analysis. To our surprise, the UV melting temperature with complementary ssRNA showed a moderate decrease of \approx 3 °C compared to that of the isosequential aza-ENA modified oligonucleotide (Table 1). This was however, a marginal increase (\approx +1 °C) with respect to the unmodified sequence. This decrease may be due to the additional steric crowding because of the presence of guanidinium group at the minor groove of DNA-RNA hybrid duplexe. Furthermore, we examined the influence of these Northlocked aza-ENA and GENA on triplex formation with dsDNA, when incorporated into oligo-DNA. The triplex forming property of the modified ONs were determined using 21mer DNA duplex having purine rich triplex binding site (Table 2).

 Table 2. Melting temperature for triplexes and MALDI-MS analysis^a

ON	Sequence (5'3')	<i>T</i> _m /°C	$\Delta T_{\rm m}$	MALDI-MS
		with	(°C)	Calc/Found
		dsDNA		[M+H] ⁺
1	TTTTT ^m CTTT ^m CT ^m CT ^m CT	33		4497.9/4497.8
2	TTTTT ^m CT <mark>T</mark> T ^m CT ^m CT ^m CT	35.5	+2.5	4539.3/4539.4
3	TTTTT ^m CT <u>T</u> T ^m CT ^m CT ^m CT	42.5	+9.5	4581.9/4582 ^b

^aTarget duplex: 5'-d(GCTAAAAAGAAAGAAGAGAGATCG)-3'/-3'd(CGATTTTTCTTTCTTCTCTCAGC)-5'; underlined portion indicates the target site for triplex formation. <u>T</u> = aza-ENA, <u>T</u> = GENA and ^mC = 5methylcytidine; Standard deviation in all the melting experiments were below ± 0.4 °C. Conditions: 7.5 mM Na₂HPO₄ buffer pH 7, containing 140 mM KCl; strand concentration = 1 μ M; scan rate 0.5 °C /min; .^bM+2H.

The UV melting studies with the third strand showed a two-phase dissociation curve with a low triplex melting temperature (33 °C for unmodified TFO) and a high duplex melting temperature (55 °C) which is in agreement with previous reports.^{10b} Interestingly, the GENA containing TFO displayed a triplex thermal stabilization of 42.5 °C at pH 7, which is an increase of + 9.5 °C as compared to the unmodified TFO. The aza-ENA modification on the other hand showed a thermal stabilization of 35.5 °C, which corresponds to an increase of +2.5 °C with respect to that of the unmodified TFO. It has been reported previously that 2'-O-[2-(guanidino)ethyl] modification in isosequential TFO showed a maximum increase of $\Delta T_{\rm m}$ = +4 °C per modification in triplex melting temperature.¹⁷ This clearly suggests that both cationic nature of the guanidinium group near phosphate (forming salt-bridge with neighbouring strand) and the North-conformation of bridged sugar play critical roles in triplex stabilization. Thus high triplex stabilization at neutral pH toward efficient pyrimidine-motif triplex formation was achieved with minimal modification on a TFO (single modification).

We further investigated the enzymatic stability of such modified ON in human serum and with snake venom phosphodiestrase (SVPDE). Human serum mainly comprises of 3'-exonucleases along with some endonucleases and 5'-exonucleases whereas, the SVPDE is a 3'-exonuclease. We anticipated that similar to aza-ENA, GENA could significantly improve the 3'-exonuclease stability of

the ON with single modification near the 3'-end. We therefore performed enzymatic digestion studies with such a sequence and compared with those of the unmodified and aza-ENA-modified ONs. Comparison of enzymatic stability studies with single aza-ENA and GENA modifications incorporated at second position from the 3'-end showed great improvements in stability as compared to the unmodified sequence. Interestingly, the guanidine-residue of GENA seemed to have higher impact on stability of the modified ON as compared to that of the amine residue of aza-ENA (Figure 1). The unmodified deoxy-ON sequence was completely degraded in less than 2 h while the aza-ENA modified sequence was degraded in less than 24 h, when incubated with SVPDE. On the other hand, the GENA modified sequence was still present to some extent even after 33 h. It is noteworthy that both aza-ENA and GENA provided stability to one extra nucleotide towards 3'-end (3'-terminal nucleotide in this case) and prevented degradation of the full-length sequence (Figure 1a). The human serum digestion footprints also indicated similar results as with SVPDE, with the exception that the n-1 band (where 'n' is the full length oligonucleotide) was visible after some time (Figure 1b). The stability of the ONs in human serum could not be monitored after 24 h presumably due to the presence of 5'-phosphodiestease in the serum, which cleaves the 5'-³²P label.²² Degradation of 5'-³²P label is evident from lack of the label in mononucleotide degradation products at higher time points (Figure 1b). Thus our result unequivocally shows that protonated species in the vicinity of 3'-phosphate indeed impart extra stability towards 3'-exonuclease.

a) GENA	Aza-ENA	Native	b) GENA	Aza-ENA	Native
48 33 24 10 4 2 0	48 33 24 10 4 2 0	48 33 24 10 4 2 0	249420	249420	9420
	655		-		1000
111000			Sec. And the		
111000	11284		in the late		_
	11244				
			in the state of the		
1 town	1989				
	4.8.4				
			- **		
-	-	-			-

Figure 1. Autoradiograms of 20% denaturing PAGE, showing the degradation pattern of 5'-³²P-labeled deoxy-ONs. Sequence: 3'-d(C<u>T</u>TCTTTTTACTTC)-5' where <u>T</u> is either, GENA, aza-ENA or T. Incubation was done at 21 °C (a) with SVPDE at 0, 2, 4, 10, 24, 33 and 48h; (b) with human serum at 0, 2, 4, 9, 24h.

In conclusion, we present the synthesis of North-conformationally constrained GENA nucleoside, which upon incorporation into ON showed marginal increase in T_m with complementary RNA ($\Delta T_m \approx$ +1 °C) as compared to that of the unmodified ON. However, such modified ON displayed high affinity for target dsDNA to form triplex with $\Delta T_m = +9.5$ °C and +7 °C, as compared to the unmodified and aza-ENA modified ONs respectively. A single GENA modification near 3'-end of an ON could impart great stability towards 3'-exonucease (SVPDE and human serum). Further

evaluation of such modifications to target genomic DNA without the aid of any intracellular delivery agents is currently underway.

The present work was supported by Swedish Strategic Research grant '*StemTherapy*' and EU Framework Program-7 project '*Biodesign*'.

Notes and references

RSC Advances

^{*a*} Agricultural and Ecological Research Unit, Biological Science Division, Indian Statistical Institute, Kolkata 700108, India

^b Polymer Chemistry Division, Department of Chemistry, The Ångström Laboratory, Uppsala University, Box 538, 751 21 Uppsala, Sweden, Fax: +46 18 471 3477, E-mail: oommen.varghese@kemi.uu.se

^c Department of Chemistry, SavitriBai Phule Pune University, India

†Electronic Supplementary Information (ESI) available: [Experimental details, synthesis and purification of oligos, ³²P labelling experiments, enzymatic digestion studies and uv melting experiments]. See DOI: 10.1039/c000000x/

- 1 B. Vester and J. Wengel, *Biochemistry*, 2004, 43, 13233–13241.
- 2 C. Zhou and J. Chattopadhyaya, *Chem. Rev.*, 2012, **112**, 3808–3832.
- 3 M. Koizumi, Biol. Pharm. Bull., 2004, 27, 453-456.
- 4 P. M. McTigue, R. J. Peterson and J. D. Kahn, *Biochemistry*, 2004, 43, 5388.
- 5 J. Bhattacharyya, S. Maiti, S. Muhuri, S. Nakano, D. Miyoshi and N. Sugimoto, *Biochemistry*, 2011, 50, 7414–7425.
- 6 S. Chatterjee, W. Pathmasiri, O. Plashkevych, D. Honcharenko, O. P. Varghese, M. Maiti and J. Chattopadhyaya. *Org. Biomol. Chem.*, 2006, 4, 1675–1686
- 7 T. Yamamoto, M. Harada-Shiba, M. Nakatani, S. Wada, H. Yasuhara, K. Narukawa, K. Sasaki, M-A. Shibata, H. Torigoe, T. Yamaoka, T. Imanishi and S. Obika, *Molecular Therapy–Nucleic Acids*, 2012; doi:10.1038/mtna.2012.16
- 8 (a) O. R. Mook, F. Baas, M. B. de Wissel and K. Fluiter, *Mol. Cancer Ther.*, 2007, 6, 833–843. (b) M. A. Campbell and J. Wengel *Chem. Soc. Rev.*, 2011, 40, 5680–5689.
- 9 S. Obad, C. O. dos Santos, A. Petri, M. Heidenblad, O. Broom, C. Ruse, C. Fu, M. Lindow, J. Stenvang, E. M. Straarup, H. F. Hansen, T Koch, D. Pappin, G. J. Hannon and S. Kauppinen, *Nature Genetics*, 2011, **43**, 371–378.
- 10 (a) T. Hojland, S. Kumar, B. R. Babu, T. Umemoto, N. Albæk, P. K. Sharma, P. Nielsen and J. Wengel, *Org. Biomol. Chem.* 2007, 5, 2375. (b) S. M. A. Rahman, S. Seki, S. Obika, S. Haitani, K. Miyashita and T. Imanishi, *Angew. Chem. Int. Ed.*, 2007, 46, 4306 4309.
- 11 B. P. Monia, J. F. Johnston, H. Sasmor and L. L. Cummins J. Biol. Chem., 1996, 271, 14533–14540.
- 12 S. M. Gryaznov Chem. Biodivers., 2010, 7, 477-493.
- 13 P. Hawley, J. S. Nelson, K. L. Fearon, G. Zon and I. Gibson, *Antisense Nucleic Acid Drug Dev.*, 1999, 9, 61–69.
- 14 O. P. Varghese, J. Barman, W. Pathmasiri, O. Plashkevych, D. Honcharenko and J. Chattopadhyaya, J. Am. Chem. Soc., 2006, 128, 15173–15187.
- 15 D. Honcharenko, J. Barman, O. P. Varghese and J. Chattopadhyaya, *Biochemistry*, 2007, 46, 5635–5646.

- 16 T. Ohmichi, M. Kuwahara, N. Sasaki, M. Hasegawa, t. Nishikata, H. Sawai and N. Sugimoto, *Angew. Chem. Int. Ed.*, 2005, 44, 6682 6685.
- 17 T. P. Prakash, A. Pu1schl, E. Lesnik, V. Mohan, V. Tereshko, M. Egli and M. Manoharan, Org. Lett., 2004, 6, 1971–1974.
- (a) M. Duca, P. Vekhoff, K. Oussedik, L. Halby, P. B. Arimondo *Nucleic Acids Res.* 2008, **36**, 5123–5138. (b) J. Y. Chin, F. Reza and P. M. Glazer, *Molecular Therapy*, 2013, **21**, 580–587.
- 19 S. F. Singleton and P. B. Dervan, *Biochemistry*, 1992, **31**, 10995– 11003.
- 20 (a) W. Roberts and D. M. Crothers, *Science*, 1992, 258, 1463. (b) S. M. Gryaznov, D. H. Lloyd, J.-K. Chen, R. G. Schultz, L. A. Dedionisio, L. Ratmeyer and W. D. Wilson, *Proc. Natl. Acad. Sci. USA*, 1995, 92, 5798–5802.
- 21 T. P. Prakash, A. Puschl and M. Manoharan, *Nucleosides, Nucleotides, and Nucleic Acids*, 2007, **26**, 149–159.
- 22 J. Liithje and A. Ogilvie, Clinica Chimica Acta, 1987, 164, 275–284.