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Direct growth of molybdenum disulfide on arbitrary insulating surfaces by chemical vapor deposition[†]

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We report a direct growth approach of large area, uniform and patternable few layer molybdenum disulfide on arbitrary insulating substrates, including polymer and glass. The method can effectively control the number of layers with 100% surface coverage and avoid transferring process.

The two-dimensional transition metal dichalcogenides, e.g., molybdenum disulfide (MoS₂), trigger increasing attentions due to their potential in a wide range of electronic, optical, mechanical applications.¹⁻⁴ The zero-bandgap property makes that graphene is not a real semi-conductor and could not be used directly by the industry. The direct band gap of atomically thin MoS₂ in the visible portion of the electromagnetic spectrum suggests suitable for digital electronics and numerous photonic application.⁵ Recently, various electronic and optoelectronic devices have been fabricated using few-layer MoS₂, such as fieldeffect transistors (FETs),⁶ sensors⁷⁻⁹ and phototransistors.¹⁰ Excellent mechanical flexibility of MoS₂ also makes it an exciting semiconducting material for flexible electronics. The optical properties of MoS₂ can be tuned by stress¹¹ and the ultrathin film FETs on flexible substrates are robust to mechanical bending.¹² Exfoliated MoS₂ and graphene grown by chemical vapor deposition (CVD) was used to build highly flexible and transparent transistors on a polyethylene terephthalate (PET) substrate.¹³ Layer-structured MoS₂ films was stacked and fabricated to Schottky-barrier solar cell though a layer-transfer process on indium tin oxide (ITO) coated glass substrate.¹⁴ MoS₂ offers a wide spectrum response and high photoresponsivity, combined with its optical stability under illumination from ultraviolet (UV) to near infrared (NIR), which can be attractive for touch screen panels for flat panel or flexible display devices.¹⁵ Emerging device applications all require that MoS₂ are supported by an insulating substrate, such as SiO₂, PET, polyimide (PI), ITO or Al₂O₃.

As early as 2005, monolayer MoS₂ was exfoliated using micromechanical cleavage onto insulating substrate.² However,

this method has limitations in controlling the thickness and size. in addition, and it is difficult to realize scaling-up. CVD is the most reprehensive method to bring closer wide spread practical implementation of graphene in term of size, quality and price.¹⁶ Recently, many efforts have been devoted to explore the controllable CVD growth of MoS2 primarily on SiO2/Si substrate using molybdenum compounds or a thin Mo films to react with sulfur powder.^{17–20} The reported methods exist the obstacle that the residual of Mo precursors do not tend to react with sulfur atoms completely, for the reason that transferring process are required to remove the various residuals and contaminated substrates.^{17–20} Very recently. Lee et al. synthesized lavercontrollable and uniform MoS2 films on wafer-scale Si/SiO2 and quartz substrates using highly reactive sulfuric precursor to avoid insufficient sulfurization.²¹ The synthesized MoS₂ films still need to be transferred onto flexible substrates using polymer supports because of high reaction and annealing temperature (750°C and 1000 °C). The transferring process would inevitably introduce impurities, wrinkles and cracks, degrading the performance of the device.²² Wu et al. firstly used MoS₂ powder to directly grow triangular MoS₂ monolayer crystallites on sapphire and silicon oxide substrates at 650°C.²³ However, the discontinuities between the crystals lead to a non-uniform charge carrier mobility. Therefore, it is critical important to control the direct growth of large-area, uniform and patternable MoS₂ on various insulating substrates, especially thermolabile substrates.

Here, we present an effective approach to directly grow largearea, uniform, and patternable MoS_2 on arbitrary insulating substrates. We use silicon with 300-nm oxide layer, quartz, cover glass, and polyimide as representative insulating substrates to demonstrate the advantages with regard to the direct growth of MoS_2 films. High reactive sulfur precursor (H₂S) facilitates sufficient sulfurization and homogeneous MoS_2 films synthesized at low temperature. In addition to the capability to obtain largearea, excellent uniform MoS_2 films with 100% surface coverage on insulating substrates, the simplicity and scalability of our

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method are greatly favorable toward promoting the applications in low-cost, flexible or transparent electrode.

The synthesis process of large-area MoS_2 films is illustrated in Fig. 1a–b. Different thickness (typically 1–10 nm) of Mo films deposited on insulating substrates were used to directly grow MoS_2 to demonstrate our strategy in the atmosphere of Ar and H_2S . The photo of direct-growth MoS_2 films on representative insulating substrates with 100% surface coverage is shown in Fig. 1c. The most commonly used methods involve the redox reaction between molybdenum compounds with sulfur powder. The obtained MoS_2 samples are discontinuous flakes (Fig. S1 in the ESI†).



Fig. 1 Schematic illustration of the direct-growth process. Largearea MoS_2 film could be synthesized on arbitrary insulating substrates with 100% surface coverage of MoS_2 .

Raman spectra is used to demonstrate the typical features of MoS₂ directly grown on the insulating surfaces at 900 °C (Fig. 2a). The out-plane vibration of S atoms (A_g^1) at ~404.9 cm⁻¹ and in-plane vibration of Mo and S atoms (E_{2g}^{1}) at ~383.1 cm⁻¹ are exhibited with a frequency ~21.8 cm⁻¹ (grown on SiO₂/Si substrate), reflecting two-layers MoS₂ according to the measurements of exfoliated samples in previous works.²⁴ Our approach to obtain large-area and high-quality films is tolerant to the deposited substrates, which also can be directly grown on transparent insulating substrates, e.g. quartz, as shown in Fig. 2a of the red line. The typical two peaks of E_{2g}^1 and A_g^1 at ~382.2 cm^{-1} and ~404.2 cm^{-1} respectively, with a frequency ~22 cm^{-1} . With the variation of substrates, the quality and thickness of MoS₂ films do not show significant difference. The high crystallinity of the MoS₂ directly grown on SiO₂/Si substrate is explored by high-resolution transmission electron microscopy (HRTEM) (Fig. 2b). The corresponding 2D fast Fourier transformation (FFT) pattern shown in Fig. 2d exhibits only one set of diffraction spots, indicating the single crystalline nature of the film. In Fig. 2e, a false 3D image from a Fourier enhanced TEM micrograph of the marked area in Fig. 2e highlights the perfect atom-scale crystal structure of the MoS₂. The (100) lattice plane, with an interlayer spacing of 0.27 nm is clearly displayed in Fig. 2e, thus MoS_2 in-plane lattice constant a (MoS_2) was determined to be ~ 0.31 nm, which is in agreement with the bulk value of 0.315 nm. The uniformity of MoS₂ films is one of key factors to determine its properties and applications. Fig. 2c displays the typical AFM image of directly grown MoS₂ on 300nm thick SiO_2/Si substrate. No obvious wrinkles and cracks are observed. The corresponding 3D surface image of the direct-growth MoS_2 is shown at Fig. S2a⁺, indicating excellent flatness achieved using our direct growth technique. Based on the above characterization results and analysis, we demonstrate that large-



area and high-quality MoS_2 can be directly grown on insulating surfaces.

Fig. 2 Direct-growth uniform MoS_2 films on 300-nm thick SiO_2/Si , quartz at the growth temperature of 900 °C. (a) Raman spectra of direct-growth MoS_2 on SiO_2/Si and quartz. (b) High-resolution TEM image of two-layer MoS_2 directly grown on SiO_2/Si substrate. (c) AFM profile of uniform MoS_2 films directly grown on SiO_2/Si with 100% full coverage. (d) 2D FFT of the TEM presented in (b). (e) A false 3D image originated from Fourier enhanced TEM micrograph of the marked area in Fig. 2b.

Thinning out the bulk MoS₂ to atomically thin layers can result in the structure of band gap transferring from an indirect to a direct state. The capability of synthesizing uniform MoS₂ with well-controlled layer numbers is especially important for electronics applications. The optical images of few layers directly grown MoS₂ on SiO₂/Si substrates without visible cracks and wrinkles show excellent uniformity and continuity (Fig. S3a-d⁺). Once a molybdenum metal layer with different thickness was deposited on the substrates, a MoS₂ film having a distinct number of layers could be synthesized (Fig. S4[†]). Because the whole substrates were covered by the direct-growth MoS₂ films without obvious cracks, the samples were transferred to SiO₂/Si substrates to determine its thickness. The transfer process for 2D atomic materials is inevitable to introduce wrinkles and cracks, in addition, the residues during the transfer process, e.g. impurities between the films and objective substrates or the remaining PMMA on the surface, which seriously decreases the performance and limits its applications. The transferred MoS₂ on SiO₂/Si substrate show obvious wrinkles and residual impurities compared with those direct-growth samples (Fig. S2b and Fig. S4a, c, e, g^{\dagger}). The step measurement of the MoS₂ film was 1.8 nm in Fig. S4a[†], which matched to the thickness of two layer characterized by Raman. Increasing the thickness of deposited Mo layer lead to the peak positions of E¹_{2g} and A¹_g moved to 382.3 cm⁻¹ and 405.9 cm⁻¹, respectively. And the relative frequency differences increased to 23.6 cm⁻¹, which corresponded to three layered MoS₂ films (red line in Fig. S3e)

with the thickness of 2.2 nm (Fig. S4c). Further increasing the thickness of deposited Mo film, the E¹_{2g} vibration blue shifted 1-2 cm⁻¹, whereas the A_{g}^{1} vibration red shifted ~1 cm⁻¹ (as show in Fig. S3e[†]). The frequency differences increased to $25-26 \text{ cm}^{-1}$, which corresponded to 6 (4.1 nm)-9 layers (6 nm) in Fig. S4e, g[†]. Fig. S4b, d, f, h[†] exhibit the statistical analysis of Raman spectra taken on S3a-d samples dominated by 9, 6, 3, and 2 layers of MoS₂ films, respectively. The relative frequency differences between the modes varied from 22 cm⁻¹ to 26 cm⁻¹, which corresponded to 2 to 9 layered MoS₂ films (as shown in Fig. S4[†]), according to a previous study.²⁴ Raman mapping images of the peak position differences between A_{g}^{1} and E_{2g}^{1} modes over an area of $10 \times 10 \ \mu m$ (from 2 to 9 layers) show nearly uniform color distribution (Fig. S5[†]), revealing the thickness variation is negligible. The high reactivity of H₂S with deposited Mo to form MoS₂ films suggests that it is possible to obtain patterned MoS₂ on diverse surfaces. MoS₂ ribbons arrays were directly grown on SiO₂/Si substrate (as shown in Fig. S6⁺), avoiding transferring process to manufacture devices.

In previous works, MoS₂ films are transferred to transparent insulating substrates including PET, ITO, Al2O3 to build photoelectric devices. Their performance is expected to be improved though optimizing the growth process, for example, develop a direct-growth method to avoiding impurities introduced though transfer process. The absorbance and photoluminescence (PL) of 2, 3, 6, and 9 layered MoS₂ films are directly related to the thickness of MoS₂ grown on quartz, as shown in Fig. 3. The resulting UV-vis spectra exhibit excitonic features of A, B, C peaks. A and B excitonic peaks appear between 600-700 nm, while C excitonic peak appear around 430-450 nm in Fig. 4a. The peak position in each MoS₂ films directly grown on quartzes correspond to the band structures. With the layer thickness increasing, the absorbance intensity of the typical peaks increases. The PL response in the energy range corresponding to a direct transition indicated semiconducting behavior in the 2H-MoS₂ films. In general, the relative intensity of the PL band decreased as the number of MoS₂ layers present increased because the indirect electron transitions dominated the direct transitions.²¹ The transition wavelength of the directly grown samples exhibits red-shifted position and lower intensity having a higher number of layers (Fig. 3b). Obvious absorbance of A and B excitonic peaks still can be observed when the growth



temperature falling into 500–800°C (Fig. S7†), which prove the directly grown samples are tolerant to growth temperature.

Fig. 3 Optical characterization of direct-growth MoS₂ on quartz. (a) UV-vis absorbance and (b) PL spectra of the MoS₂ grown on quartz at 900 °C having different thickness values.

X-ray photoelectron spectroscopy (XPS) is used to further clarify the structure of MoS₂ directly synthesized at different temperature and substrates. The C1s peak is fixed at 284.8 eV. As shown in Fig. 4a, the Mo (IV) show two peaks centered at 229.1~229.6 and 232.3~232.7 eV, attributing to the Mo 3d_{5/2} and Mo $3d_{3/2}$, respectively. And S $2p_{3/2}$ and S $2p_{1/2}$ are centered at 162.3 eV and 163.4 eV, respectively (as shown in Fig. 4b). All the XPS positions of peaks are consistent with the reported values.^{17, 19, 22} The small peak at 235.5 eV indicate the residuals of Mo⁶⁺ grown at 300 °C. The surface oxide is originated from the underlying SiO₂ substrate, and molybdenum oxides are not detected (as shown in Fig. S8[†]). There are no noticeable changes occurring in the peak shape with a decrease in the temperature, in addition, the ratio of S and Mo is close to 2:1, which further suggest the formation of MoS2 is independent of growth temperature in our system.



Fig. 4 The characterizations of XPS to shown the composition of MoS_2 on SiO_2/Si substrate. (a) and (b) show the corresponding the binding energies of Mo and S in the few-layer MoS_2 , which were grown at 300 °C, 500 °C and 900 °C, respectively.

The growth temperature for reported growth methods of MoS₂ varies from 500 °C to 1100 °C, which limits the direct synthesis of MoS₂ films on thermolabile insulating substrates. Those insulating substrates could survive at high temperature, such as SiO₂/Si, quartz and sapphire are more expensive than plastic, and are not suitable for flexible device applications. The optical properties of MoS₂ attached to plastic are tunable though bending the substrates and electronic prosperities can keep stable, which inspires great potential in flexible photoelectric applications. The exfoliated and transferred MoS₂ films are limited by size and inevitable defects. Therefore, it is important to develop a way to directly grow MoS₂ on thermolabile insulating substrates. Our strategy has extremely high tolerance to various kinds substrates including common glass and polymer, such as PI. We succeed to obtain MoS₂ films on cover glass and SiO₂/Si as low as 250 °C (Fig. $S9^{\dagger}$). Fig. $S9b^{\dagger}$ shows the typical peaks of MoS₂ directly grown on cover glasses with difference Mo layer thickness at 300 °C, as shown in the photograph of the inset. The MoS₂ films grown on PI at 250-300 °C have been characterized by XPS (Fig. S10[†]). Obvious signals from Mo(IV) 3d_{5/2} (centered at 229.2 eV) and S 2s peak at 226.7 eV are present. When the temperature is higher than 250 °C, the signal from Mo(VI) disappear gradually.

Conclusions

In summary, we demonstrate the direct growth of large area, uniform and patternable few layer MoS₂ films on arbitrary insulating substrates with 100% surface coverage, and the growth temperature can be as low as 250 °C. The thickness of the synthesized MoS₂ films could be controlled by depositing Mo layers having different thickness values. The AFM characterization exhibit the directly grown MoS₂ films is continuous compared with the transferred samples showing wrinkles, cracks and impurities. The directly synthesized MoS₂ films are also characterized by Raman/PL/UV-vis spectroscopy, XPS, and HRTEM. All analyses reveal almost perfect films uniformity, precisely controlled films thickness values, and high films crystallinity without the presence of an amorphous phase. Our approach is an effective way to directly obtain uniform and layer number controlled MoS₂ films on insulating surfaces especially on low-cost substrates, which would greatly facilitate flexible, stretchable and transparent device applications.

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Notes and references

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1. Q. H. Wang, K. Kalantar-Zadeh, A. Kis, J. N. Coleman, M. S. Strano, *Nat. Nanotechnol.*, 2012, **7**, 699.

2. K. S. Novoselov, D. Jiang, F. Schedin, T. J. Booth, V. V. Khotkevich, S. V. Morozov, A. K. Geim, *P. Natl. Acad. Sci. USA*, 2005, **102**, 10451.

3. H. Li, Jumiati Wu, Z.Y. Yin, H. Zhang, Acc. Chem. Res. 2014, 47, 1067.

4. X. Huang, Z. Y. Zeng, H. Zhang, Chem. Soc. Rev., 2013, 42, 1934.

5. K. F. Mak, C. Lee, J. Hone, J. Shan, T. F. Heinz, *Phys. Rev. Lett.*, 2010, **105**, 136805.

6. B. Radisavljevic, A. Radenovic, J. Brivio V. Giacometti, A. Kis, *Nat. Nanotechnol.*, 2011, **6**, 147.

7. H. Li, Z. Yin, Q. He, H. Li, X. Huang, G. Lu, D. W. H. Fam, A. I. Y. Tok, Q. Zhang, H. Zhang, *Small*, 2012, **8**, 63.

8. C. F. Zhu, Z. Y. Zeng, H. Li, F. Li, C. H. Fan, H. Zhang, J. Am. Chem. Soc., 2013, **135**, 5998.

9. S. X. Wu, Z. Y. Zeng, Qiyuan He, Z. Y. Wang , S. J. Wang, Y. P. Du, Z. Y. Yin, X. P. Sun, W. i Chen, H. Zhang, *Small*, 2012, **8**, 2264.

10. Z. Yin, H. Li, H. Li, L. Jiang, Y. Shi, Y. Sun, G. Lu, Q. Zhang, X. Chen, H. Zhang, *ACS Nano*, 2012, **6**, 74.

11. C. R. Zhu, G. Wang, B. L. Liu, X. Marie, X. F. Qiao, X. Zhang, X. X. Wu, H. Fan, P. H. Tan, T. Amand, B. Urbaszek, *Phys. Rev. B*, 2013, **88**, 121301(R).

12. J. Pu, Y. Yomogida, K. Liu, L. Li, Y. Iwasa, T. Takenobu, *Nano Lett.*, 2012, **12**, 4013.

13. J. Yoon, W. Park, G. Bae, Y. Kim, H. S. Jang, Y. Hyun, S. K. Lim, Y. H. Kahng, W. Hong, B. H. Lee, H. C. Ko, *Small*, 2013, **9**, 3295.

14. M. Shanmugam, C. A. Durcan, B. Yu, Nanoscale, 2012, 4, 7399.

15. W. Choi, M. Y. Cho, A. Konar, J. H. Lee, G. B. Cha, S. C. Hong, S. Kim, J. Kim, D. Jena, J. Joo, S. Kim, *Adv. Mater.*, 2012, **24**, 5832.

16. K. S. Novoselov, V. I. Fal'ko, L. Colombo, P. R. Gellert, M. G. Schwab, K. Kim, *Nature*, 2012, **490**, 192.

17. K. Liu, W. Zhang, Y. Lee, Y. Lin, M. Chang, C. Su, C. Chang, H. Li, Y. Shi, H. Zhang, C. Lai, L. Li, *Nano. Lett.*, 2012, **12**, 1538.

18. Y. Zhan, Z. Liu, S. Najmaei, P. M. Ajayan, J. Lou, Small, 2012, 8, 966.

19. Y. Lin, W. Zhang, J. Huang, K. Liu, Y. Lee, C. Liang, C. Chu, L. Li, *Nanoscale*, 2012, **4**, 6637.

20. Y. Yu, C. Li, Y. Liu, L. Su, Y. Zhang, L. Cao, Sci. Rep. 2013, 3, 1866.

21. Y. Lee, J. Lee, H. Bark, Il. Oh, G. H. Ryu, Z. Lee, H. Kim, J. H. Cho, J. Ahn, C. Lee, *Nanoscale*, 2014, **6**, 2821.

22. I. Song, C. Park, M. Hong, J. Baik, H. Shin, H. C. Choi, *Angew. Chem. Int. Ed.*, 2014, **53**, 1266.

23. S. Wu, C. Huang, G. Aivazian, J. S. Ross, D. H. Cobden, X. Xu, *ACS Nano*, 2013, **7**, 2768.

24. H. Li, Q. Zhang, C. C. R. Yap, B. K. Tay, T. H. T. Edwin, A. Olivier, D. Baillargeat, *Adv. Funct. Mater.*, 2012, **22**, 1385.

TOC:



Direct growth of large area, uniform and patternable few layer molybdenum disulfide is achieved on arbitrary insulating substrates by CVD.