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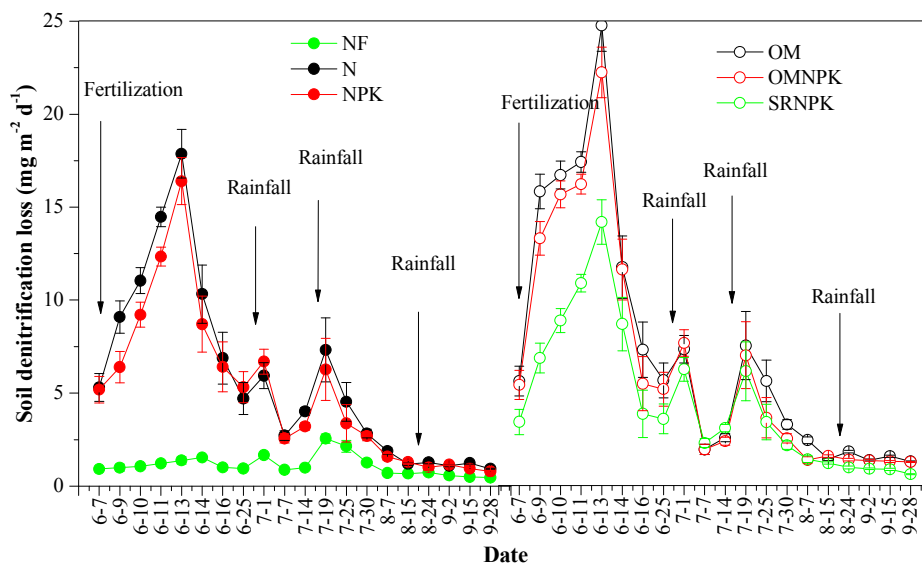
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Crop residue combined with synthetic fertilizer is recommended as an optimal strategy for mitigating N₂O emissions and denitrification-induced N loss.



Environmental impact statement

Agricultural soils have been identified as responsible for approximately 60% of the total anthropogenic N₂O emissions. Nitrogen fertilizer application to agricultural soil will increase N₂O emission. Here we compared N₂O emissions and denitrification from rain-fed cropland under different fertilization regimes involved in single synthetic N fertilizer, organic manure, synthetic N, P, K fertilizer, organic manure with synthetic fertilizer, crop straw residue with synthetic fertilizer. Crop straw residue combined with synthetic fertilizer is recommended as an optimal strategy for mitigating N₂O emissions and denitrification-induced N loss.

The influence of N fertilization regimes on N₂O emissions and denitrification in rain-fed cropland during the rainy season

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Abstract The effects of nitrogen fertilization regimes on N₂O emissions and denitrification rates were evaluated by in situ field incubation experiments with intact soil cores and the acetylene block technique. Intact soil cores were collected from long-term field experiments involving several N fertilization regimes, including single synthetic N fertilizer (N), organic manure (OM), synthetic N, P, K fertilizer (NPK), organic manure with synthetic fertilizer (OMNPK), crop straw residue with synthetic fertilizer (SRNPK) and no nitrogen fertilizer (NF). N₂O was sampled from the head space of the cylinders to determine the daily N₂O emission and denitrification rate. The results showed that the N₂O emissions were greatly influenced by the specific fertilization regime even when the same nitrogen rate was applied. The mean N₂O emissions and denitrification rates from the N, OM, NPK, OMNPK and SRNPK treatments were 2.22, 2.66, 1.94, 2.53, 1.67 and 4.63, 5.96, 4.15, 5.41, 3.65 mg m⁻² d⁻¹, respectively. The application of OM significantly increased the N₂O emission and denitrification compared to the application of NPK because of the high soil organic carbon (SOC) content of OM. However, SRNPK increased the SOC content and decreased the N₂O emissions significantly compared to the OM and OMNPK treatments because the addition of crop straw with a high C/N ratio to soil with a low inorganic N content induced N immobilization. The contents of soil nitrate and ammonium were the main limiting factors for N₂O emissions in a positive regression as follows: $\ln(N_2O) = 2.511 + 1.258 \times \ln([\text{NH}_4^+] + [\text{NO}_3^-])$. Crop straw residue combined with synthetic fertilizer is recommended as an optimal strategy for mitigating N₂O emissions and denitrification-induced N loss in rain-fed cropland.

Keywords N₂O emission; Denitrification; Fertilization regimes; Soil inorganic nitrogen; Mitigation strategy; Rain-fed agriculture.

Introduction

As a potent greenhouse gas, nitrous oxide (N_2O) plays a key role with respect to global warming and climate change, with approximately 60% of global anthropogenic N_2O emissions originating from agricultural soils¹. The magnitude of N_2O emissions from cropland is dependent on environmental factors as well as agricultural practices, such as tillage methods and N fertilizer rates and types^{2, 3}. There are many reports on soil N_2O emission fluxes under different N rates and types^{4, 5, 6, 7}. N_2O is an intermediate gaseous product of denitrification⁸. When different rates and types of N fertilizer are applied, the soil nitrate (NO_3^-) contents are undoubtedly changed. Several studies have demonstrated that the incorporation of organic matter, such as crop residues or livestock manure, could stimulate soil N_2O emissions^{9, 10, 11}. In contrast, previous studies also demonstrated that the incorporation of high C/N ratio organic manure may result in a decrease in the soil N_2O emission during the denitrification process, particularly under conditions of high rainfall or irrigation^{12, 13}. Lopez-Fernandez et al. demonstrated that the application of pig slurry reduced N_2O emissions by 27% compared to the application of urea in a Spanish xerofluent with a sandy loam soil⁵. Yao et al. reported that wheat straw application reduced N_2O emissions from a rice-wheat rotation field¹⁴. However, few reports on the effects of N_2O emissions under different types of N fertilizer were based on the comparison with the same N rates. It remains unclear whether this pattern may also apply to agricultural soils receiving the same amount of N but via different forms of N fertilizer. Meanwhile, the soil organic carbon (SOC) altered by an organic or

inorganic fertilization regime may be another influencing factor on the soil N₂O emission^{15, 16}. The contribution of denitrification to N₂O emissions from different levels of soil inorganic nitrogen and SOC derived from various fertilization regimes has not been well documented.

Rain-fed agriculture in southwestern China is an important agricultural practice because of the wet climate and poor irrigation conditions in the hillslope areas. Rain-fed cropland in the Sichuan Basin (the largest agricultural region in southwestern China) supplies 10% of the agricultural products of China⁵, where Regosols (referred to locally as purple soil) are widely distributed over an area of 160,000 km² with dense populations and intensive agriculture during the rainy season¹⁷. Rain-fed agriculture in purple soil is an N-limited system with different practices of N amendments. There are some typical N-fertilization regimes, such as organic manure with pig slurry (OM), which has a long history of use; single synthetic N fertilizer (N), which was used in the 1970s; and synthetic nitrogen, phosphorus and potassium fertilizer (NPK) and OM plus synthetic NPK fertilizer and crop residues returned with synthetic NPK fertilizer, which are currently widely used. Therefore, gathering data on denitrification and its induced N₂O emission in rain-fed cropland during the rainy season is essential for evaluating N₂O pollution at the regional scale. N₂O emissions are 1.72 ± 0.34 kg N ha⁻¹ yr⁻¹ in rain-fed cropland due to the addition of N fertilizer to purple soil², but very few studies have investigated the impacts of fertilization regimes on N₂O emissions and the contribution of denitrification in rain-fed cropland with purple soil. In this study, we conducted in situ soil core

incubation experiments using the acetylene block technique to examine the impacts of N fertilization regimes on N₂O production with the same rate of total N applied to rain-fed cropland with purple soil during the rainy season. We sought to determine a) the effect of the form of N fertilizer, i.e., synthetic fertilizer, manure, manure combined with synthetic fertilizer and crop residue combined with synthetic fertilizer, on N₂O emissions and the key parameters governing N₂O emissions; b) the contribution of denitrification to N₂O emissions; and c) the optimal fertilization regime for the mitigation of N₂O emissions.

Materials and methods

Field site

This study was conducted in Yanting Agro-Ecological Experimental Station of Purple Soil (N31°16', E105°28'), Chinese Academy of Sciences, which is located in the central Sichuan Basin in southwestern China at an altitude of 400-600 m above sea level. The site is characterized by a moderate subtropical monsoon climate with an annual mean temperature of 17.3 °C and a mean precipitation of 826 mm, with rainfall mainly occurring during the summer and rainy season. The croplands are mostly rain-fed because of the topography, with slope gradients of 5-30% and a soil thickness of 30-80 cm, and typically lack an irrigation infrastructure¹⁸.

Long-term fertilization experiments

A long-term fertilization experiment was established in 2003 with 5 fertilization treatments: single synthetic nitrogen fertilizer applied with NH₄HCO₃ (N); mixed synthetic nitrogen, phosphorus and potassium fertilizer (NPK); organic manure with

pig slurry (OM); organic manure with pig slurry and synthetic fertilizer at a ratio of 40%:60% total N applied (OMNPK); and straw residue combined with synthetic fertilizer at a ratio of 15%:85% total N applied (SRNPK). The amount of N that was applied in both synthetic and organic forms was maintained at the same rate of 150 kg N ha⁻¹ during the rainy season when maize (*Zea mays* L.) was planted and 130 kg N ha⁻¹ during the dry season when wheat (*Triticum aestivum* L.) was planted in all of the fertilization treatments. Calcium superphosphate (90 kg P₂O₅ ha⁻¹) and potassium chloride (36 kg K₂O ha⁻¹) were applied to the NPK, OMNPK and SRNPK treatments, and a treatment without fertilizer (NF) was used as the control. Each of the treatments was replicated 3 times with an experimental plot area of 4 m × 8 m, each in a completely randomized design under conventional rain-fed cultivation. The plots were fertilized and sown with the late-maturing maize variety Zhongdan 808 on 7 June 2010. The field observation and incubation experiment spanned an entire rainy season, from 7 June to 28 September 2010. The main soil properties for the topsoil (0-15 cm) of the fertilization experiments in 2010 are listed in Table 1.

Table 1 Physical and chemical properties of the soil under different fertilization regimes in 2010.

Parameter	Fertilization regimes					
	NF	N	NPK	OM	OMNPK	SRNPK
pH	8.35 ^a	8.26 ^b	8.27 ^b	8.28 ^b	8.24 ^b	8.24 ^b
Bulk density (g/cm ³)	1.39 ^a	1.34 ^b	1.30 ^b	1.31 ^b	1.36 ^b	1.26 ^c
TN (g/kg)	0.25 ^a	0.35 ^b	0.56 ^c	0.88 ^d	0.79 ^e	0.73 ^e
SOC (g/kg)	4.24 ^a	4.51 ^b	6.43 ^c	9.89 ^d	8.63 ^e	9.96 ^d
C/N	16.96 ^a	12.89 ^b	11.48 ^c	11.24 ^c	10.92 ^c	13.64 ^d

Different letters in same line represent significant differences ($P < 0.05$) between treatments, according to Duncan test.

Measurements of the denitrification rate and N₂O emission

Eight intact soil cores (inner diameter 3.2 cm and length 18 cm) from each plot were placed into two different groups of incubation cylinders. All of the incubation cylinders were covered with caps with a small hole in the center of the top cap and sealed with a rubber stopper for gas sampling. Ten percent of the air from one group of cylinders was replaced with C₂H₂ to block the enzymatic reduction of N₂O to N₂ to determine the denitrification rate, while another group of cylinders without C₂H₂ was used to determine the natural N₂O emissions. Before each determination of denitrification rate, the same content of C₂H₂ (purified by bubbling through H₂SO₄ solution and distilled H₂O) had been injected into cylinder and then incubated for 24 h. The cylinders were sealed and buried for in situ incubation in the fertilization plots from which the soil cores were collected daily to measure the denitrification rate and natural N₂O emission. Gas samples of 10 ml were collected from the head space of the cylinders using a syringe for immediate N₂O analysis using a gas chromatograph (HP-5890 Series II, Hewlette Packard, Palo Alto, California, USA) equipped with an ECD. The N₂O emissions were measured daily immediately after fertilization, every other day after a rainfall event, which can trigger N₂O emissions, and once every week during the periods without rainfall. The daily mean emission for each treatment was calculated by averaging the 3 replicates for each measuring day. The total emissions for each treatment were calculated by integrating the daily losses with time.

The denitrification rate or N₂O emission was calculated as follows:

$$F = \frac{\frac{dC_t}{t} \times Mw \times H \times 273.15 \times p \times 1000}{22.41 \times (273.15 + T) \times 1013} \quad 19,$$

where F is the daily denitrification rate ($\text{mg m}^{-2} \text{d}^{-1}$ with C_2H_2) or N_2O emission ($\text{mg m}^{-2} \text{d}^{-1}$ without C_2H_2), $\frac{dCt}{1}$ is the increment of N_2O concentration (ppbv) after 24 h of incubation, M_w (44 g mol^{-1}) is the molar weight of N_2O , H (m) is the height of the cylinder, p (10^5 Pa) is the atmospheric pressure, and T ($^\circ\text{C}$) is the air temperature inside the cylinder.

The cumulative emissions of N_2O were calculated as follows:

$$E = \sum (E_{i+1} + E_i) / 2 \times (t_{i+1} - t_i),$$

where E is the cumulative denitrification or N_2O emissions, i is the i th measurement, and the term $(t_{i+1} - t_i)$ represents the number of days between 2 consecutive measurements.

Measurements of environmental factors

During gas sampling, the soil temperature and volumetric water content at a 5-cm depth were measured, and soil samples from all of the plots were collected for NO_3^- and NH_4^+ analyses. The soil NO_3^- and NH_4^+ were extracted with 2 M KCl and determined using a continuous-flow analyzer (AA3, BRAN+LUEBBE, Germany). The measured water contents were converted into a water-filled pore space (WFPS) using the bulk density given above and a theoretical particle density of 2.65 g cm^{-3} . The data for the daily precipitation, air pressure and temperature were obtained from the meteorological station at Yanting Agro-Ecological Experimental Station, which is 50 m away from the experimental site.

Data analysis and statistical methods

The significance of the differences in the N_2O emission, denitrification rate and soil

properties between the different fertilization treatments was determined using a one-way ANOVA (SPSS Statistics Client 19.0, SPSS, Inc., USA). Pearson's test was used to determine the relationship between the N₂O emission, denitrification rate and soil properties. Graphical outputs were obtained from Origin 8.0 (Origin Lab Ltd., Guangzhou, China).

Results

Climatic and soil conditions

The air temperature was greater than 17.7 °C during the observation period (7 June to 28 September 2010). The mean air temperatures in June, July, August and September were 24.2, 26.8, 27.1 and 24.0 °C (Fig. 1a), respectively. The soil temperature at a 5-cm depth ranged between 19.4 and 33.1 °C, with means of 26.4, 26.4, 25.0, 24.6, 25.2 and 24.6 °C in the NF, N, NPK, OM, OMNPK and SRNPK treatments (Fig. 1b), respectively. During the observed period, there were 55 days of rain and 12 days of heavy rain of more than 10 mm. The amounts of precipitation in June, July, August and September were 56.4 mm, 176.5 mm, 108.2 mm and 89.8 mm, respectively. Continual rain occurred from 23 to 28 June, with a rainfall amount of 11.4 mm; from 13 to 20 July, with 140.9 mm; and from 20 to 25 Aug, with 96.6 mm. The soil water-filled pore spaces (WFPS) ranged from 38.4%-81.6%. In general, the WFPS varied with the rainfall and temperature with over 65% of the WFPS after the three continual rain periods and below 65% for most of the days when the air temperature was greater than 26 °C (Fig. 1c). The SRNPK treatment had the lowest WFPS throughout most of the observation period.

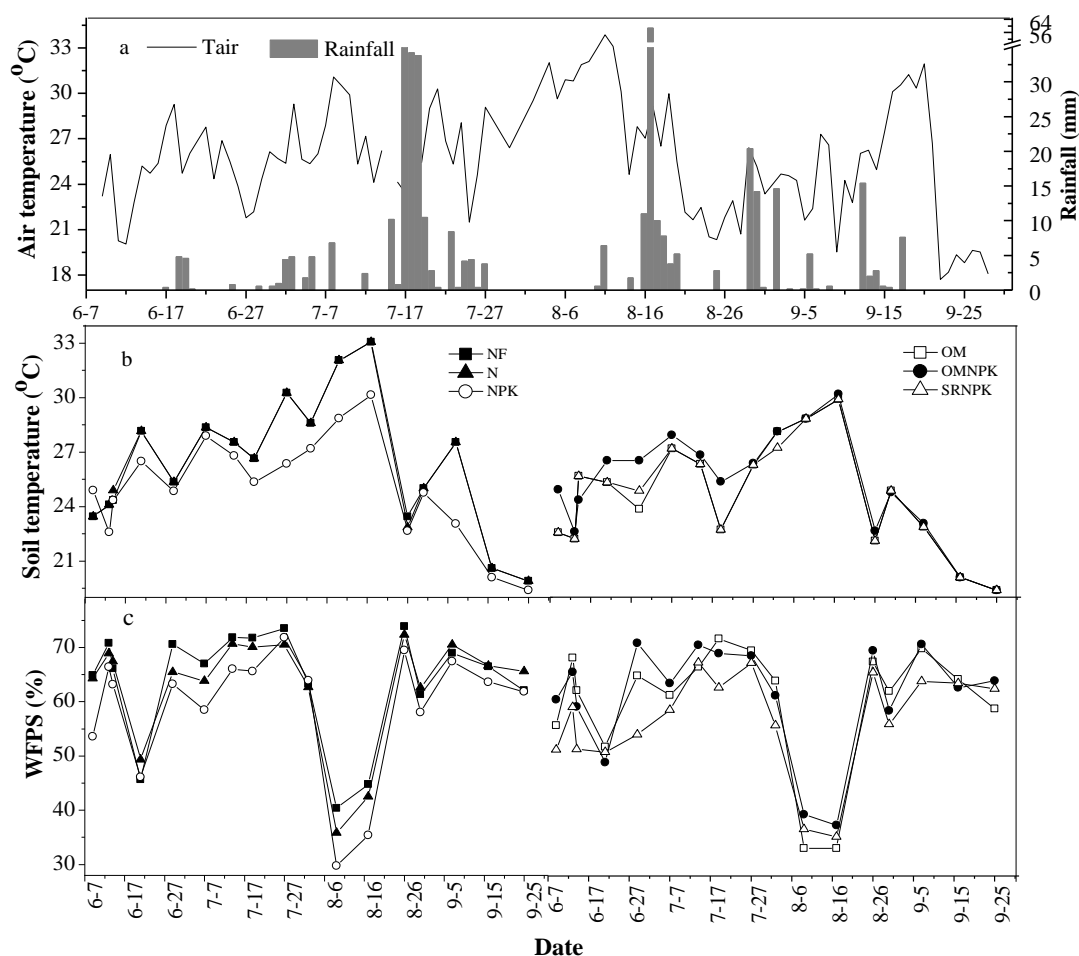


Fig. 1 Daily air temperature ($^{\circ}\text{C}$), rainfall (mm), soil temperature ($^{\circ}\text{C}$) and WFPS during the experimental period. (a) Daily air temperature ($^{\circ}\text{C}$) and rainfall (mm); (b) soil (5 cm depth) temperature ($^{\circ}\text{C}$); and (c) soil moisture content expressed as water-filled pore space (WFPS, %). The standard errors of the soil temperature and moisture are not shown for clarity.

The dynamic soil NH_4^+ and NO_3^- contents during the rainy season are shown in Fig. 2. The inorganic nitrogen content increased immediately after fertilizer application, and the highest soil NH_4^+ was observed 3 days following nitrogen fertilization, with values of up to 1.42-1.80 mg kg^{-1} in the N, OM, NPK and OMNPK treatments, and 7 days after fertilization in the SRNPK treatment, with a lower value of 1.39 mg kg^{-1} (no significant difference). The soil NH_4^+ decreased quickly and

reached the lowest value (approximately 0.5 mg kg^{-1}) at approximately 2 weeks after fertilization (Fig. 2a). The soil NO_3^- peaked ($9.90 - 45.25 \text{ mg kg}^{-1}$, $p < 0.01$, $n = 3$) 7 days after fertilization due to rapid nitrification and continuously decreased until maize harvest, with minimum values lower than 2.0 mg kg^{-1} (Fig. 2b), which can be ascribed to crop absorption, nitrate leaching and denitrification¹⁸. The OM and OMNPK treatments had significantly higher contents of NO_3^- and NH_4^+ than did the SRNPK treatment, especially during the first week following fertilization ($p < 0.01$, $n = 3$), whereas there were no significant differences among the N, NPK and OMNPK treatments (Fig. 2). Both the soil NH_4^+ and NO_3^- contents throughout the entire observation period followed the order $\text{OM} > \text{OMNPK} > \text{N} > \text{NPK} > \text{SRNPK} > \text{NF}$. Thus, organic manure provided more available N, and the inorganic N supplied from crop straw was relatively lower.

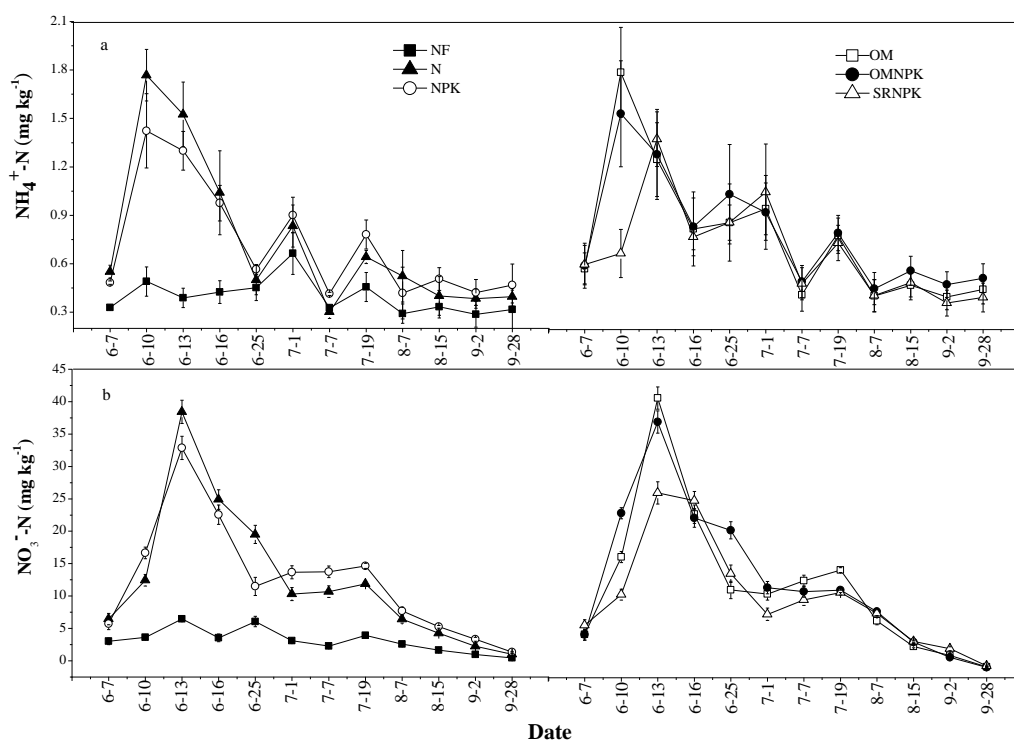


Fig. 2 Dynamics of the soil inorganic nitrogen contents during the rainy season. (a) Soil

ammonium content (mg kg^{-1}); (b) soil nitrate content (mg kg^{-1}). The vertical bars indicate the standard errors (SE) of the measurements from three replicated plots.

There were no significant differences in the pH among the different fertilizer treatments (Table 1). The long-term application of organic manure and crop straw led to significant increases in the SOC contents compared to the synthetic N, P and K fertilization ($p < 0.01$, $n=6$) (Table 1). The soil C/N ratio was greatest in the NF ($p < 0.05$, $n=6$) treatment followed by the SRNPK treatment, whereas the OM and OMNPK treatments had the lowest C/N ratio (Table 1).

Soil N₂O emission

The N₂O emissions under different fertilization regimes are presented in Fig. 3. In general, the fertilizer application greatly stimulated the N₂O emissions during the first week after fertilization. The highest N₂O emission ($8.71\text{-}13.33 \text{ mg m}^{-2} \text{ d}^{-1}$) occurred on the sixth day after fertilization, and the differences in the N₂O emissions among the treatments were significant during the first week after fertilization ($p < 0.01$, $n=3$). The OM and OMNPK treatments showed the highest N₂O emissions, with mean values of 10.58 and $9.35 \text{ mg m}^{-2} \text{ d}^{-1}$, respectively, followed by the N and NPK (mean 6.94 and $4.62 \text{ mg m}^{-2} \text{ d}^{-1}$, respectively) treatments. The SRNPK treatment had the lowest N₂O emission ($3.73 \text{ mg m}^{-2} \text{ d}^{-1}$) of all of the N fertilization treatments, and the N₂O emissions were also extremely low in the NF plots. The N₂O fluxes that were emitted within one week after fertilization comprised 13.8% to 20.7% of the total emissions during the entire maize-growing period (114 days). For the remaining maize season, the response of the N₂O emissions to different fertilization regimes was similar to that

in the first week, but the differences among the treatments diminished (Fig. 3). The average N_2O emissions from the N, OM, NPK, OMNPK, SRNPK and NF treatments during the remaining period of the rainy season were 1.88 ± 0.49 , 2.19 ± 0.54 , 1.71 ± 0.44 , 2.21 ± 0.59 , 1.35 ± 0.34 and $0.53 \pm 0.10 \text{ mg m}^{-2} \text{ d}^{-1}$, respectively. Two N_2O emission peaks were captured due to heavy rainfall events (1 July and 19 July); thereafter, the N_2O emissions decreased continually with a low emission (less than $0.48 \text{ mg m}^{-2} \text{ d}^{-1}$) and lasted until the end of the rainy season.

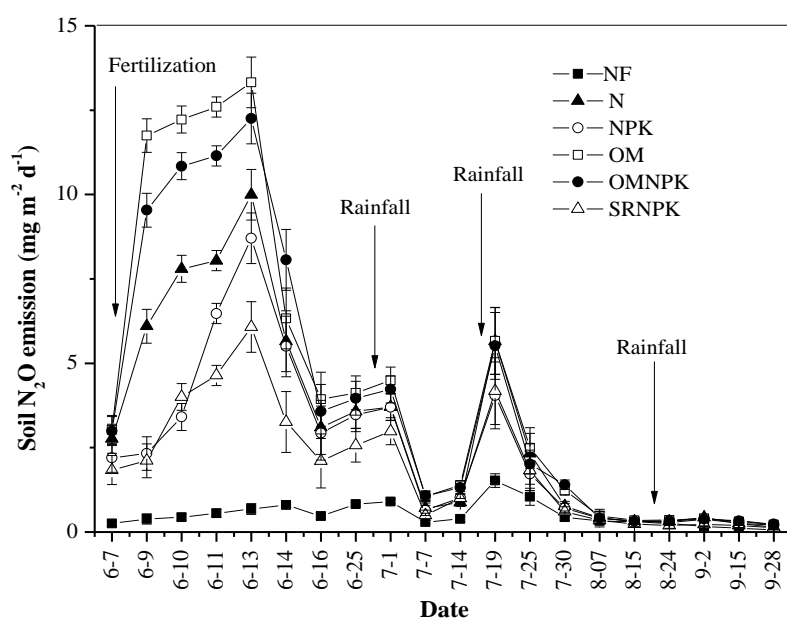
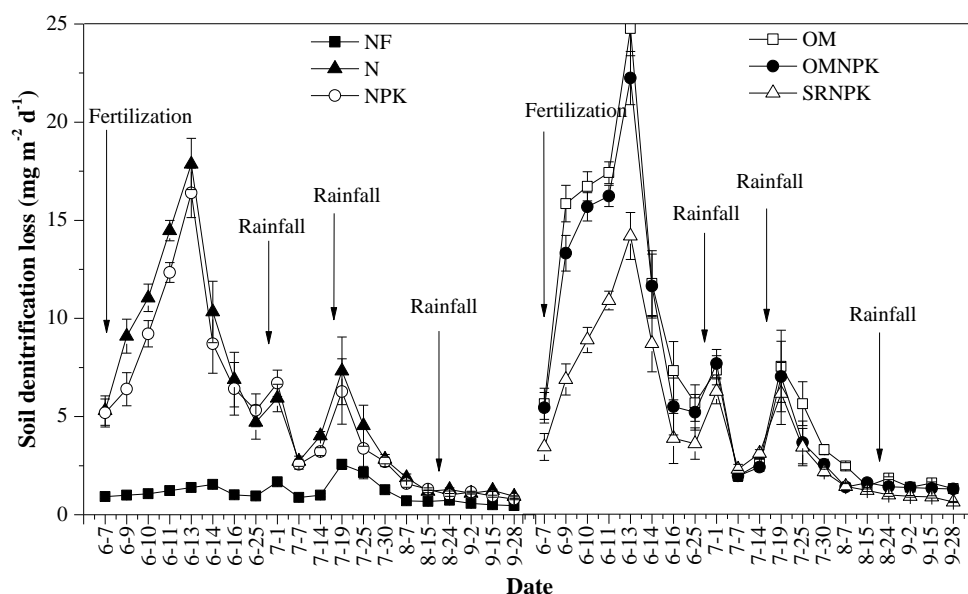


Fig. 3 Seasonal patterns of soil N_2O emissions during the rainy season under different fertilization regimes. The arrows represent the timing of the fertilizer applications and intense rainfall events. The vertical bars indicate the standard errors (SE) of measurements from three replicated plots.

Soil denitrification rate

- 1 The soil denitrification rates during the entire rainy season exhibited similar trends as
- 2 were observed for the soil N_2O emissions but had higher values (Fig. 4). The soil

3 denitrification rates exhibited sharp increases after fertilization or intense rainfall
 4 events in all of the treatments and varied from 0.94 to 17.87 mg m⁻² d⁻¹ in the N plots,
 5 1.28 to 24.77 mg m⁻² d⁻¹ in the OM plots, 0.63 to 14.19 mg m⁻² d⁻¹ in the NPK plots,
 6 1.31 to 22.24 mg m⁻² d⁻¹ in OMNPK plots and 0.79 to 16.40 mg m⁻² d⁻¹ in the SRNPK
 7 plots. The OM treatment showed the highest denitrification rate, with a mean of 5.96
 8 mg m⁻² d⁻¹, followed by the OMNPK (5.41 mg m⁻² d⁻¹), N (4.63 mg m⁻² d⁻¹), NPK
 9 (4.15 mg m⁻² d⁻¹), SRNPK (3.65 mg m⁻² d⁻¹) and NF (0.93 mg m⁻² d⁻¹) treatments. The
 10 differences among the fertilization treatments were significant ($p < 0.05$, $n = 3$).



11
 12 Fig. 4 Seasonal patterns of soil denitrification rates during the rainy season under different
 13 fertilization regimes. The vertical bars indicate the standard errors (SE) of the measurements
 14 from three replicated plots.

15 Cumulative nitrogen loss via soil N₂O emissions and denitrification

16 The cumulative N₂O emissions during the rainy season varied from 0.56 to 3.01 kg
 17 ha⁻¹ among the fertilization regimes, which accounted for 0.88% to 1.63% of the

18 applied N (Table 2). The accumulated nitrogen loss through denitrification ranged
 19 from 0.78 to 6.74 kg ha⁻¹ in all of the fertilization treatments during the rainy season
 20 (Table 2), which accounted for 2.23%-3.97% of the applied N, respectively. The
 21 denitrification coefficients in the OM-related regimes (OM and OMNPK) were higher
 22 compared to conventional NPK fertilization and crop residue returned with synthetic
 23 fertilization (SRNPK) (Table 2). The seasonal cumulative denitrification-induced
 24 nitrogen losses among the fertilization regimes were significantly different ($p < 0.05$),
 25 with the highest loss in the OM treatment, followed by the OMNPK, N, NPK,
 26 SRNPK and NF treatments. The ratio of cumulative natural N₂O emissions to
 27 denitrification was in the range of 0.45 to 0.48 in the N-fertilization treatments and the
 28 lowest in the OM treatment. However, the ratio was the highest in the NF treatment
 29 with significant differences from those in the N fertilization treatments ($p < 0.01$) and
 30 (Table 2).

31 Table 2 Cumulative N loss through N₂O emissions and denitrification from the rainy season under
 32 different fertilization regimes.

Treatments	Cumulative N ₂ O emission (kg /ha)	Denitrification N loss (kg/ha)	N ₂ O emssion /denitrification	N ₂ O emission in applied N (%)	Denitrification coefficient (%)
NF	0.56 ^a	0.78 ^e	0.72 ^a	-	-
N	2.51 ^{bc}	5.23 ^c	0.48 ^b	1.30	2.97
OM	3.01 ^d	6.74 ^a	0.45 ^{cd}	1.63	3.97
NPK	2.19 ^{ce}	4.69 ^d	0.47 ^{bc}	1.09	2.61
OMNPK	2.86 ^{bd}	6.11 ^b	0.47 ^{bc}	1.53	3.55
SRNPK	1.88 ^e	4.12 ^d	0.46 ^c	0.88	2.23

33 Each value represents the mean of three replications; means followed by different letters are
 34 significantly different among fertilization treatments at $p \leq 0.05$.

35 Discussion

36 In the present study, the N₂O emissions exhibited the same trend as did the
 37 denitrification rates with high ratios of N₂O emissions to denitrification, meanwhile,
 38 the N₂O emission was significantly related to the denitrification rate (Table 3),
 39 indicating that denitrification was a major path of nitrogen loss from the rain-fed
 40 cropland in the Sichuan Basin. The pulse emissions of N₂O that were induced by
 41 heavy rainfall with WFPS greater than 65% were apparently the result of
 42 denitrification. The favorable WFPS levels for denitrification were generally in the
 43 range of 60%-90%²⁰⁻²². Numerous studies have been published to demonstrate the
 44 role of the denitrification process in soil N₂O emission^{6, 15, 22, 23}.

45 Table 3 Correlations between the denitrification rate, N₂O emission and soil physical-chemical
 46 properties.

	SOC (g/kg)	TN (g/kg)	DOC (mg/kg)	C/N	NO ₃ ⁻ (mg/kg)	NH ₄ ⁺ (mg/kg)	N ₂ O flux
Denitrification rate	0.379*	0.266	0.250	0.375*	0.854**	0.805**	0.991**
N ₂ O flux	0.329	0.203	0.226	0.369*	0.862**	0.795**	

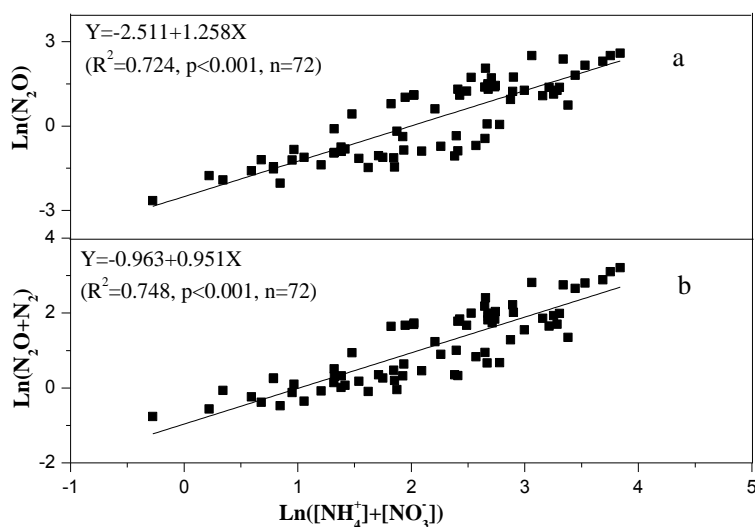
47 Available phosphorus, available potassium and bulk density had no significant correlations with
 48 N₂O emission, which were not shown in the table (n=30)

49 Denitrification is the stepwise reduction of NO₃⁻ to N₂ with an intermediate
 50 gaseous product of N₂O²⁴. The application of nitrogen fertilizer in both organic and
 51 inorganic forms will increase the contents of soil nitrate and ammonium, which are
 52 substrates for denitrification and may increase N₂O emissions. A positive regression
 53 of the natural logarithm of N₂O or denitrification rate was significant compared to the
 54 logarithm of the corresponding soil NH₄⁺ and NO₃⁻ contents (Fig. 5). This result is

55 consistent with previous studies^{15, 25, 26}. However, the cumulative N₂O emissions and
56 denitrification differed greatly among the different fertilization regimes, even when
57 the amount of nitrogen was applied at the same rate. Higher N₂O emissions were
58 observed under long-term amendment with organic manure (Fig. 3, 4, Table 2). In
59 addition to the soil inorganic nitrogen contents, the SOC was found positively
60 correlated with the denitrification rate ($r=0.379$, $p<0.05$, $n=30$, Table 3). Kilian et al.
61 and Chang et al. also observed that repeated applications of organic fertilizers resulted
62 in increased SOC contents and N₂O emissions^{7, 27}. The addition of organic manure
63 increases denitrification because of the additional supply of reactive carbon^{28, 29}. Soil
64 organic carbon may trigger denitrification by enhancing respiration through the
65 consumption of oxygen, which creates anoxic microsites, and by providing energy for
66 denitrifiers²⁹. The synthetic N, P and K fertilizer treatments showed lower cumulative
67 N₂O emissions than the OM and OMNPK treatments due to short supplies of reactive
68 carbon (Table 2, Fig. 2). Interestingly, long-term crop straw residues that were
69 returned to cropland increased the SOC (Table 1) but did not increase the N₂O
70 emissions, as observed in the organic manure treatments. However, this treatment
71 (SRNPK) reduced the N₂O emissions by 60.1%, 52.1% and 13.9% compared to the
72 OM, OMNPK and conventional synthetic NPK fertilization regimes, respectively.
73 This result is consistent with Cai et al., who observed a 70% water- holding capacity
74 (WHC) in a laboratory study with potato soil using the ¹⁵N gas-flux method³⁰. Zou et
75 al. showed that wheat straw residue could slightly reduce the N₂O emissions from rice
76 paddies³¹, whereas Zhao et al. indicated that maize straw residue increased the N₂O

77 emissions from a wheat field³². Generally, straw residues with low C/N ratios may
78 enhance N₂O emissions³³. In contrast, high C/N ratios may reduce N₂O emissions
79 because of temporary N immobilization, which delays the release of N into the soil^{4, 34}.
80 The incorporation of plant residues with C/N ratios greater than 18 may enhance N
81 immobilization³⁵. In this study, the returned crop residues with high C/N ratios (wheat
82 straw has a C/N ratio of 35.7, and maize straw has a C/N ratio of 68.1) most likely
83 stimulated the immobilization of the applied synthetic N fertilizer by soil microbes.
84 This result is further supported by the observed decrease in inorganic N (NH₄⁺ and
85 NO₃⁻) compared to the NPK treatments (Fig. 2), which might decrease the microbial
86 denitrification activity and thus decrease the N₂O emissions³⁶. However, organic
87 manure provided more inorganic nitrogen than did crop straw residues (Fig. 2). The
88 contents of soil nitrate and ammonium were the main limiting factors for the N₂O
89 emissions and denitrification rate in the respective positive regressions as follows: Ln
90 (N₂O) = 2.511 + 1.258 × Ln ([NH₄⁺] + [NO₃⁻]) and Ln (N₂O + N₂) = 0.963 + 0.951 × Ln
91 ([NH₄⁺] + [NO₃⁻]) (Fig. 5). It is noteworthy that soil inorganic nitrogen was the main
92 substrate of denitrification and might be regulated by soil C/N ratios. In soils with
93 high NO₃⁻ contents, the SOC would be the main driver of N₂O production via
94 denitrification, but the NO₃⁻ availability might control the denitrification rates in the
95 low-N soil³⁷. This result may somewhat explain the relatively high N₂O emissions for
96 the OM and OMNPK treatments and the low N₂O emissions for the SRNPK treatment.
97 Thus, the fertilization regime of crop straw residues combined with synthetic N, P and
98 K fertilizers should be recommended for the mitigation of N₂O emissions and for the

99 restoration of the soil SOC levels.



100

101 Fig. 5 Correlation of the soil inorganic nitrogen ($\text{NO}_3^- + \text{NH}_4^+$) contents with the N_2O emissions (a)
 102 and denitrification N loss (b).

103 The body of denitrification rate estimates, especially for terrestrial systems, is
 104 dominated by C_2H_2 -based methods³⁸, and these estimates are reasonably robust,
 105 especially for systems with moderate or high NO_3^- levels³⁸. In our study, fertilizer
 106 provided sufficient NO_3^- for most of the maize growing season. The results are
 107 reasonable when compared with the N_2O emissions as monitored by the static
 108 closed-chamber gas chromatograph method at the same field sites^{1,39}. The acetylene
 109 block technique also could effectively provide information for differences between
 110 the N_2O emission and denitrification rate among the fertilization regimes (Figs. 3, 4).
 111 However, this technique has been criticized due to precluding reliable estimates of
 112 total denitrification losses and has limitation when it is used to identify the N_2O
 113 sources^{38,40,41}, an accurate source of N_2O and the relative importance of denitrification
 114 and nitrification processes can be assessed in the future by isotope enrichment

115 technique.

116 **Conclusions**

117 In this study, different fertilization regimes significantly affected the
118 denitrification rates and N₂O emissions in a rain-fed cropland with purple soil during
119 the rainy season. High N₂O emissions were triggered by N fertilization in the form of
120 organic manure as a result of strong denitrification. The soil C/N ratio that was altered
121 by the fertilization regime appears to regulate denitrification by altering the soil
122 substrate NO₃⁻, NH₄⁺ and SOC contents. Crop straw residue returned with synthetic N,
123 P and K fertilizer is recommended as the best nitrogen management practice for both
124 N₂O mitigation and denitrification-induced N loss compared to the other widely used
125 fertilization regimes of synthetic N, P and K, organic manure and organic manure
126 with synthetic fertilizer at the same amount of applied nitrogen.

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