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ARTICLE TYPE

Development of a Redox-Free Mitsunobu Reaction Exploiting Phosphine Oxides as Precursors to Dioxyphosphoranes

Xiaoping Tang, Charlotte Chapman, Matthew Whiting and Ross Denton at

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The development of the first redox-free protocol for the Mitsunobu reaction is described. This has been achieved by exploiting triphenylphosphine oxide - the unwanted byproduct in the conventional Mitsunobu reaction - as the 10 precursor to the active P(V) coupling reagent. Multinuclear NMR studies are consistent with hydroxyl activation via an alkoxyphosphonium salt.

Bimolecluar nucleophilic substitution reactions of alcohols are fundamentally important transformations in chemical synthesis. 15 During the last sixty years phosphorus(V) compounds have emerged as widely used activating agents for nucleophilic coupling reactions through a process originally described by Mukaiyama as "redox dehydration". The most familiar of these protocols being the 1967 Mitsunobu reaction² (Scheme 1).

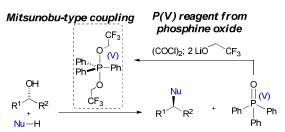
Scheme 1 The 1967 Mitsunobu protocol based on the DEAD/PPh₃ combination

Although this powerful protocol has been widely employed in chemical synthesis it suffers from low atom economy. 25 Furthermore, the diazodicarboxylate oxidants are toxic, high energy compounds and the derived hydrazine and phosphine oxide by-products can complicate product isolation. To this end many creative strategies have been developed to aid purification. For example, tagged³ or polymeric phosphine and 30 diazodicarboxylate reagents have emerged. Notable work in this area includes bipyridyl-tagged phosphine reagents,⁴ fluorous phosphines,⁵ tetraaryl phosphonium salts incorporating phosphines, Rasta Resins, a "waste as catalyst strategy" and "impurity annihilation" methods.9

35 More recently the requirement for both a stoichiometric oxidant and phosphine in the Mitsunobu coupling and other phosphorus-mediated reactions has begun to be addressed. 10 In a pioneering study, Toy described a Mitsunobu system that employed substoichiometric DEAD and an iodine(III)-based 40 terminal oxidant. 11 More recently still Taniguchi has demonstrated an impressive catalytic variant that exploits molecular oxygen as the terminal oxidant. 12 The necessity for an

oxidant in the Mitsunobu procedure is a consequence of the fact that phosphorus(V) reagents have been accessed almost 45 exclusively in an oxidative sense from phosphine precursors e.g. triphenylphopshine and DEAD (Scheme 1). A corollary of this conventional approach is that any phosphorus recycling or indeed catalysis^{13,14} of such reactions must be achieved through the addition of a terminal reductant.15

In contemplating a method to improve the Mitsunobu reaction and eliminate hydrazine by-products we were attracted to a conceptually distinct alternative in which the active phosphorus(V) reagent would be accessed from a phosphorus(V) precursor, 16 specifically triphenylphosphine oxide – the unwanted 55 by-product from the redox-based protocol (Scheme 2). 17



Scheme 2 This work: a redox-free Mitsunobu process which relies upon generation of the active P(V) reagent from triphenylphosphine oxide.

Such a protocol was attractive since it would: (a) eliminate the 60 need for an oxidant; (b) eliminate competing side reactions such as alkylation of the hydrazinedicarboxylate; and (c) result in no net phosphorus waste. To the best of our knowledge this approach has only been investigated to date by Jenkins, whose detailed studies on the Hendrickson reagent (Ph₃POPPh₃•2OTf) 65 indicated that the latter could function in analogy to the Mitsunobu system for esterification of primary alcohols. However, critically, in the case of secondary alcohols the ester products were obtained with retention of configuration¹⁸

In this communication we describe the successful 70 implementation of redox-free Mitsunobu inversion reactions in which the active phosphorus(V) reagent is obtained from triphenylphosphine oxide.

The approach taken (Scheme 2) was contingent upon accessing the required dioxyphosphorane - of which many are ¹⁹/_{20,21,22} from triphenylphosphine oxide. While this was unprecedented at the time of our study we reasoned that the conversion of phosphine oxides chlorophosphonium salts²³,²⁴ would provide a means to access

the required dioxyphosphoranes following addition of the appropriate alkoxides. Gratifying this proved to be the case and several dioxyphosphoranes²⁵ were obtained through this protocol (Scheme 3) including dioxyphosphorane 3 (^{31}P $\delta=-58.2$, $^{2}J_{C-P}$ $_{5}$ J=6.0, 19 F δ =-74.3, 2 J_{F-F}, J=9.0 CDCl₃), which was originally described by Ishikawa and is known to promote a wide range of Mitsunobu-type coupling reactions.²²

Scheme 3 Synthesis of the Ishikawa phosphorane 3 from triphenylphosphine oxide.

While these studies were instructive and guided our choice of phosphorane the stereochemical outcome of intermolecular coupling reactions using 3 remained unknown.²⁶ A further attractive feature of 3 was the presence of the CF₃ groups which 15 allowed us to calculate the molarity of stock solutions of the reagent using ¹⁹F NMR.²⁷

Having developed a new method to access the Ishikawa dioxyphosphorane we next developed a Mitsunobu-type alcohol inversion protocol using reagent 3 and ethyl acetate as solvent 20 (Table 1).

The redox-free coupling reactions took place with inversion of configuration with both non-activated (entries 1 and 2) as well as an activated secondary alcohol (entry 3 and 4). The process depicted in entry 1 was equally efficient on 15 mmol scale (with 25 respect to the alcohol) and, in this case, 86% of triphenylphosphine oxide was recovered for reuse in subsequent coupling reactions. Furthermore, we determined that the stereochemical outcome of the reaction was independent of the order of addition of the alcohol and carboxylic acid.

Additional secondary alcohols underwent esterification in good yield (entries 5 and 6). Acetic acid could also be employed in place of the aromatic carboxylic acids (entry 9); however, the isolated yield in this case was moderate. Finally, menthol underwent esterification with inversion of configuration albeit in 35 low yield (entries 10 and 11).

With this protocol in hand we carried out a multinuclear NMR study in order to gain further insight into the course of the reaction. We were particularly interested in establishing whether or exchange of one of the trifluoroalkoxy ligands at phosphorus 40 would occur in the presence of the alcohol alone to generate a mixed dioxyphosphorane of general Ph₃P(OCH₂CF₃)(OR) as originally proposed by Ishikawa.

The results we obtained (Scheme 4) indicated that this was not the case with (S)-2-octanol. Thus a 0.25 M solution of 3 in ethyl 45 acetate remained unchanged upon addition of one equivalent of the alcohol. However, upon addition of benzoic acid the ³¹P resonance at -59.6 ppm was replaced by a new signal at 53.4 ppm which corresponds to the indicated alkoxyphosphonium salt with a carboxylate counteranion. The structure of this intermediate is 50 corroborated by the presence of a *single* peak in the ¹⁹F NMR, which corresponds to trifluoroethanol and rules out the formation "mixed" dioxyphosphorane of Ph₃P(OCH₂CF₃)(OR). The former species then underwent slow

Arbuzov collapse to the afford ester product with inversion of 55 configuration with concomitant generation of triphenylphosphine oxide.

Table 1 Redox-free Mitsunobu reactions using phosphorane 3.^a

CF ₃ O Ph.,, Ph. → P − Ph 3 1 equiv.							
			Ph Ph	- 	3 1 equiv.	0	
OI <u>=</u>	OH $ \begin{array}{ccccccccccccccccccccccccccccccccccc$		CF ₃			O R ³	
R ¹ 1 equ			EtOAc -78 °C to 70 °C, 18 h		C	₹ ¹ R ²	
							
Entry	Product		Yield e.e.	Entry	Product	Yield %	
1	Me Me	o o o o o o o o o o o o o o o o o o o	73% >99:1 e.r.	7	Me d)g	70%	
2	Me N ₅	\bigcap_{NO_2}	70% >99:1 e.r.	8	Me d) ₉ 6h	72% IO ₂	
3	Me Ph	NO ₂	90% 96:4 er	9	Me Me 6i	50%	
4	Ph 6	Ne O	79%	10	Me O Me	27%	
5	O. N. Boc	o Se	76%	11	6j O Me	`` ^{NO} -30%	
				- /	6k		
6							
6f 72%							

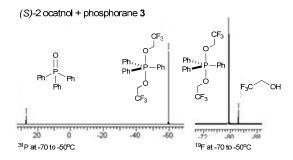
^a For each entry the isolated yield after chromatography is reported.

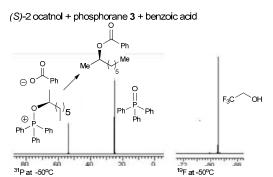
60 Conclusions

In summary we have developed for the first time a redox-free approach to Mitsunobu inversion reactions that relies upon accessing the Ishikawa phosphorane 3 from triphenylphosphine oxide.

This new protocol eliminates the need for diazo-based oxidants and, given that reagent 3 is also known to promote C-N, C-S and other cyclodehydration reactions,²² opens up a wide range of other useful Mitsunobu-type reactions that will not require the addition of an external oxidant. Furthermore, since 70 triphenylphosphine oxide is used to prepare the coupling reagent no net phosphorus waste is generated.

10





Scheme 4 Multinuclear NMR study

Notes and references

- ^a School of Chemistry, University Park, University of Nottingham, 5 Nottingham, United Kingdom. Tel: 441159514194; E-mail: ross.denton@nottingham.ac.uk
- ^b GlaxoSmithKline, Gunnels Wood Road, Stevenage, United Kingdon. † Electronic Supplementary Information (ESI) available: [details of Mitsunobu coupling procedure and full experimental supporting 10 information]. See DOI: 10.1039/b000000x/
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■ Redox Free Mitsunobu Reactions

