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# Amplified inhibition of the electrochemical signal of graphene-thionine nanocomposite by silica nanoprobe for ultrasensitive electrochemical immunoassay

Guosong Lai<sup>a,\*</sup>, Cuiying Yin<sup>a</sup>, Xiangen Tan<sup>a</sup>, Haili Zhang<sup>a</sup>, Aimin Yu<sup>a,b</sup>

<sup>a</sup> Hubei Collaborative Innovation Center for Rare Metal Chemistry, Hubei Key Laboratory of Pollutant Analysis & Reuse Technology, Department of Chemistry, Hubei Normal University, Huangshi 435002, PR China

<sup>b</sup> Faculty of Science, Engineering and Technology, Swinburne University of Technology, Hawthorn VIC 3122, Australia

\* Corresponding author.

Tel.: +86-714-6515602; Tax: +86-714-6573832

E-mail address: laiguosong@hotmail.com (G.S. Lai)

# Abstract:

An ultrasensitive immunoassay method was developed based on the amplified inhibition of the electrochemical signal of graphene-thionine nanocomposite. The graphene-thionine nanocomposite was prepared by one-step reduction of graphene oxide in the thionine solution and used to modify a glassy carbon electrode. The immunosensor was prepared by stepwise assembly of gold nanoparticles (Au NPs) and capture antibody at this nanocomposite modified electrode. The thionine on the immunosensor surface exhibited aood electrochemical signal which was further promoted by the presence of Au NPs. After a sandwich immunoreaction, the current response of the immunosensor decreased due to the formation of dielectric antibody-antigen immunocomplex on its surface. This current decrease could be further amplified by the captured antibody conjugated silica nanosphere with low electric conductivity. Based on this amplified signal inhibition mechanism, a novel detection strategy for the ultrasensitive electrochemical immunoassay was developed. Using human IgG as a model protein, a wide linear range in four orders of magnitude and a low detection limit down to 7 pg/mL were achieved. In addition, the immunosensor has low cost, satisfactory reproducibility and stability, and acceptable reliability, thus providing a promising potential for clinical applications.

# 1. Introduction

Recently, electrochemical immunosensors have shown great success in the accurate determination of protein biomarkers due to their unique advantages including high selectivity, low cost and good portability.<sup>1–3</sup> Typically, various enzyme labels are required to be used in these methods in order to produce corresponding electrochemical signal for the sandwich immunoassays and achieve sensitive analyte measurements. However, the relatively high cost and low storage stability of the enzyme biomolecules may limit their wide applications.<sup>4</sup> Moreover, the electron transfer resistance caused by the dielectric antibody-antigen immunocomplex formed on the immunosensing surface can weaken their electrochemical signal response in some degree, which may further hinder the improvement in analytical performance to achieve lower detection limit and wider linear range.

Except for enzyme, some electroactive indicators such as thionine (TH),<sup>5,6</sup> ferrocene<sup>7,8</sup> and methylene blue<sup>9,10</sup> can be also used for the signal tracing in the electrochemical bioassay field. As the impedance effect of the antibody-antigen immunocomplex can induce obvious signal decrease of these electrochemical indicators modified on the electrode surface, many direct immunoreaction-based label-free electrochemical immunoassay methods have been easily developed in recent years.<sup>5,11–14</sup> Meanwhile, owing to their excellent electron transfer ability, various nanomaterials are commonly used to hybridize with these indicators for promoting their signal responses and achieving higher sensitivity.

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As the latest nanomaterial star, graphene sheet has recently gained increasing interest in a great variety of research fields due to its excellent physical and chemical properties.<sup>15,16</sup> Because graphene is hydrophobic and can not be dispersed in most solvents while hydrophilic graphene oxide (GO) has bad conductivity, chemical reduction of graphene

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oxide along with proper surface functionalization is commonly adopted to produce reduced graphene oxide (rGO) with desired properties and application potentials for the electrochemical biosensing.<sup>17–19</sup> Compared with the covalent and assembly methods, the  $\pi$ -rich surface domains of graphene enable it to be easily conjugated with the electrochemical indicator of TH with a water-soluble planar aromatic structure via the non-covalent  $\pi$ - $\pi$  stacking interaction, and thus obtain a useful electroactive nanocomposite.<sup>20,21</sup> In the previous reports, GO was often reduced first by hydrazine<sup>20</sup> or ascorbic acid<sup>21</sup> and then conjugated with TH. In fact, the good reducing property of thionine indicates that it may be also served as the reducing agent of GO. Herein, we tried the one-step reduction of GO in a thionine solution and achieved the easy preparation of rGO-TH nanocomposite in this work. Based on the electrode modification with the as-prepared rGO-TH and stepwise assembly of gold nanoparticles (Au NPs) and capture antibody on its surface, an immunosensor was successfully constructed for the electrochemical immunoassay (Scheme 1).

In addition, it is worthwhile to note that the conventional label-free immunoassays often have limited sensitivity which makes them difficult to carry out the accurate measurement of low-abundant protein biomarkers in real serum samples. Therefore, the development of some strategies to amplify the electrochemical impendence for promoting the signal decrease induced by the direct immunoreaction should be an effective approach to achieve highly sensitive immunoassay. As a versatile nanomaterial, monodispersible silica nanosphere has shown wide applications in the biomedical and biosensing fields.<sup>22–24</sup> Considering its low semiconducting conductivity in the obstruction of the electron transfer on the electrode surface,<sup>25–27</sup> this work tried the antibody functionalized silica nanoprobe to promote the sensitivity improvement of the rGO-TH nanocomposite based electrochemical immunosensor successfully. After a sandwich immunoreaction, the

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current response of rGO-TH at the immunosensor decreased due to the formation of dielectric antibody-antigen immunocomplex on its surface. This current decrease was further amplified by the captured antibody conjugated silica nanospheres owing to their unique electrochemical impedance effect. Based on this amplified signal inhibition strategy, a novel ultrasensitive electrochemical immunoassay method was thus developed.

Preferred position for Scheme 1

# 2. Experimental section

#### 2.1. Reagent and materials

Human IgG (HIgG), mouse IgG (MIgG), polyclone rabbit anti-human IgG (anti-HIgG) were purchased from Wuhan Boster Biological Technology Ltd. Bovine serum albumin (BSA), human serum albumin (HSA) and (3-aminopropyl)-triethoxysilane were obtained from Sigma-Aldrich Chemical Co. (St. Louis, MO). Glutaraldehyde (25% aqueous solution) was purchased from Alfa Aesar China Ltd. Chloroauric acid (HAuCl<sub>4</sub>·4H<sub>2</sub>O), tetraethoxysilane and TH were obtained from Shanghai Reagent Company (Shanghai, China). The bovine serum sample was obtained from Beijing Solarbio Science & Technology Ltd. Ultrapure water obtained from a Millipore water purification system (Milli-Q) was used in all assays. All other reagents were of analytical grade and used as received.

Phosphate-buffered solution (PBS) of pH 7.0 was prepared by mixing the stock solutions of 50 mM NaH<sub>2</sub>PO<sub>4</sub> and Na<sub>2</sub>HPO<sub>4</sub> and used as working solution. A 50 mM pH 7.0 PBS containing 0.05% (w/v) Tween-20 (PBST) was used as washing buffer, and a 50 mM pH 7.0 PBS containing 3% (w/v) BSA was used as blocking solution.

## 2.2. Apparatus

The electrochemical impedance spectra were recorded at a CHI 660 electrochemical workstation (CH Instruments, USA) in 5.0 mM  $K_3Fe(CN)_6/K_4Fe(CN)_3$  (1:1) containing 0.10 M KCl. All other electrochemical experiments were performed on a CHI 830B electrochemical workstation (Chenhua, China). A conventional three-electrode system containing a platinum wire as auxiliary electrode, a saturated calomel electrode as reference electrode and a modified glassy carbon electrode (GCE, 3 mm diameter) as working electrode was used throughout the electrochemical experiments. The FT-IR and UV-Vis spectra were recorded at a Thermo Scientific Nicolet iD5 spectrometer (USA) and a Hitachi UV-3100 spectrometer (Japan), respectively.

#### 2.3. Preparation of immunosensor

The rGO-TH nanocomposite was first prepared by a one-step reduction method. Briefly, 2.0 mg GO was added to 10 mL of 2.0 mM TH solution and ultrasonicated for 30 min to obtain a homogenous dispersion. Then, this dispersion was heated to 100 °C and reacted overnight under continuously stirring. After thrice centrifugation and washing by water, the resulting rGO-TH nanocomposite was collected and dispersed in water at a concentration of 0.5 mg/mL for further use.

The preparation process of the immunosensor was illustrated in Scheme 1. Before modification, a GCE was polished with alumina slurries of 0.3 and 0.05  $\mu$ m successively followed by rinsing thoroughly with ultrapure water until a mirror-like surface was obtained. After washing by ultrasonication in absolute ethanol and water and then drying at room temperature, 3  $\mu$ L as-prepared rGO-TH nanocomposites were dropped onto the surface of GCE and dried slowly in air. Subsequently, 10  $\mu$ L of 13-nm colloidal Au NPs prepared by the conventional citrate reduction method<sup>4</sup> were cast onto the electrode surface for 8 h assembly. After washing with water to remove the loosely adsorbed Au NPs, 3  $\mu$ L of 0.5 mg/mL anti-HIgG was applied to the GCE surface and incubated in a

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100% moisture saturated environment overnight at 4 °C. The resulting electrode was washed three times with PBST and PBS to remove the loosely adsorbed antibody, and then incubated with the blocking solution for 60 min at room temperature to block possible remaining active sites against nonspecific adsorption. After washing again with PBST and PBS, the resulting immunosensor was finally obtained and stored at 4 °C in a dry environment prior to use.

#### 2.4. Preparation of silica nanoprobe

Firstly, the monodispersible silica nanospheres with an average diameter of about 100 nm were prepared and amino-functionalized followed by the surface-activation with glutaraldehyde according to our previous reports.<sup>24</sup> After centrifugation and repeated washing with PBS, 2.0 mg of the aldehydized silica nanospheres were redispersed in 1.0 mL PBS containing 15  $\mu$ g of anti-HIgG and reacted for 2 h at room temperature by gently mixing. After centrifugation, the obtained bioconjugates were blocked with 3% BSA for 60 min, and then washed thrice with PBS and resuspended in 1.0 mL PBS containing 0.1% BSA as the nanoprobe dispersion.

#### 2.5. Analytical procedure

Based on the sandwich-type immunoassay, the immunosensor was first incubated with a 15- $\mu$ L drop of the HIgG standard solution or serum sample for 50 min at room temperature, followed by washing with PBST and PBS. Then, 15  $\mu$ L of silica nanoprobe dispersion was cast onto the immunosensor surface for another 50 min of incubation. After washing with PBST and PBS again, differential pulse voltammetry (DPV) at a step potential of 4 mV, a pulse amplitude of 50 mV and a pulse period of 0.2 s was performed in pH 7.0 PBS to record the current response for the quantitative analysis.

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# 3. Results and discussion

#### 3.1. Preparation and characterization of rGO-TH nanocomposite

In this work, the rGO-TH nanocomposite was one-step prepared by using TH as both the reducing and stabilizing agents. Firstly, the as-prepared rGO-TH was characterized by FT-IR spectroscopy (Fig. 1A). Similar to the previous reports,<sup>28,29</sup> GO exhibited a strong C=O stretching vibration at 1735  $\text{cm}^{-1}$ , a strong C=C stretching, skeletal vibrations from unoxidized graphitic domains at 1622 cm<sup>-1</sup>, a broad absorption band at 3000–3500 cm<sup>-1</sup> from O-H stretching vibration as well as weak O-H deformation peak at 1400 cm<sup>-1</sup>, C-OH stretching peak at 1220 cm<sup>-1</sup> and the C-O stretching peak at 1050 cm<sup>-1</sup>. After reduction of GO by TH, the oxygen-containing functionalities of GO such as the C=O and C-O stretching vibration peaks almost disappeared completely while new strong absorption bands at 1600–1302 cm<sup>-1</sup> from the skeletal vibrations of the phenyl ring of TH<sup>30,31</sup> were clearly observed in the IR spectrum of the formed nanocomposite. These results indicate that GO was successfully reduced by TH; meanwhile, TH was also conjugated with graphene through the  $\pi$ - $\pi$  stacking interaction to form an rGO-TH nanocomposite. In addition, from the UV-vis spectrum of rGO-TH shown in Fig. 1B we can also find two obvious absorption peaks of TH at 280 nm and 598 nm.<sup>5</sup> This phenomenon further confirmed the successful preparation of the rGO-TH nanocomposite.

#### Preferred position for Fig. 1

#### 3.2. Preparation of the immunosensor

The as-prepared rGO-TH nanocomposite was then used to modify the electrode for the further assembly of Au NPs and antibody on its surface to form an immunosensor. This step-by-step preparation process was characterized by monitoring the electrochemical behavior of TH modified on the electrode surface. From Fig. 2A we can find that the

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rGO-TH modified electrode showed a pair of well-defined redox peaks at the potentials of -0.080 V and -0.148 V (curve a), which is due to the excellent electrochemical behavior of TH conjugated with rGO. After Au NPs were further assembled on the electrode surface, its peak currents increased obviously (curve b). This phenomenon should be attributed to the excellent electron transfer acceleration from Au NPs introduced onto the electrode surface. The Au NPs assembly also provided an ideal interface for the antibody immobilization through the specific interaction between noble-metal nanoparticles and protein biomolecules.<sup>24</sup> After immobilizing anti-HIgG on the electrode surface and further blocking the possible remaining active sites by BSA, however, the current response of the modified electrode showed an obvious decrease (curve c). This phenomenon should be attributed to the dielectric protein biomolecules attached to the electrode surface, which also indicates the successful preparation of the immunosensor.

#### Preferred position for Fig. 2

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#### 3.3. Electrochemical immunoassay at the immunosensor

As a sensitive electroanalytical method, DPV was used to study the electrochemical response of the immunosensor towards HIgG analyte. As shown in Fig. 2B, the immunosensor showed a sensitive DPV peak at the potential of -0.118 V (curve d). Compared with the current response before immunoreaction, an obvious drop in the peak current of the immunosensor was observed after incubation with 100 ng/mL HIgG (curve e). This is due to the formation of dielectric antibody-antigen immunocomplex on the electrode surface which further hinders the electrochemical signal of the rGO-TH nanocomposite. Further experiment showed that in comparison with this direct antigen-antibody immunoreaction at the immunosensor, when antibody conjugated silica nanoprobe was further used for the sandwich immunoreaction, drastic current decrease occurred at the immunosensor (curve f). This result suggests that the silica nanoprobe

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could greatly increase the electrochemical impendence of the immunosensor resulting in the amplified signal inhibition to the electrochemical indicator of TH.

In addition, electrochemical impedance spectroscopy was also used to characterize the interfacial resistance change during the immunosensor preparation and immunoassay process. As shown in Fig. 3, compared with the bare GCE, we can clearly find the obvious Nyquist diameter decrease after the electrode modification with the rGO-TH nanocomposite and the subsequent assembly of Au NPs on its surface. The results confirmed that both the rGO-TH and Au NPs modified on the electrode surface could greatly promote the electron transfer due to their excellent electrical conductivity and large specific surface area. However, the electrochemical resistance increased drastically after antibody immobilization and BSA blocking to obtain the immunosensor. This phenomenon further demonstrated the immobilization of dielectric protein biomolecules could inhibit the electron transfer on the electrode surface. After incubation of 100 ng/mL HIgG at the immunosensor, obvious resistance increase due to the formation of dielectric antibody-antigen immunocomplex was also observed. Moreover, the Nyquist diameter resistance increased dramatically when the as-prepared silica nanoprobe was further used for the sandwich immunoreaction. These phenomena confirmed that the silica nanoprobe with excellent impedance effect could greatly promote the electrochemical signal inhibition of the immunosensor. Hence, by combination of this amplified signal inhibition strategy and the sensitive electrochemical response of the rGO-TH based immunosensor, a novel ultrasensitive immunoassay method was thus developed.

#### Preferred position for Fig. 3

#### 3.4. Optimization of incubation time

To achieve excellent analytical performance of this method, the effect of incubation time on the DPV response of the immunosensor towards 10 ng/mL HIgG was investigated (Fig.

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4). At room temperature, the DPV current response decreased with increasing incubation time used in the sandwich immunoassay and then tended to a constant value after 50 min, indicating the saturated binding of the sandwich immunoreaction on the electrode surface. Therefore, an incubation time of 50 min was selected for the sandwich immunoassay at room temperature.

# Preferred position for Fig. 4

#### 3.5. Analytical performance

Based on the sandwich immunoassay, the electrochemical responses of the immunosensor towards different concentrations of HIgG were examined. From Fig. 5 we can observe that the DPV responses of the immunosensor decreased with the increasing concentrations of the analyte. The calibration curve showed a good linear relationship between the peak currents and the logarithm values of HIgG concentrations in the range from 0.01 to 100 ng/mL with a correlation coefficient of 0.9984. The detection limit at a signal-to-noise ratio of 3 was estimated to be 7 pg/mL. As shown in Table 1, this method showed excellent analytical performance with wider linear range and lower detection limit not only than many label-free electrochemical immunoassays<sup>32–34</sup> but also than some signal amplification strategy-based electrochemical immunoassays<sup>1,35–37</sup> reported previously, which is very importance for its practical applications.

Preferred position for Fig. 5 and Table 1

#### 3.6. Specificity, reproducibility, stability and reliability

MIgG and HSA were used to investigate the specificity of the immunosensor towards noncognate proteins. As shown in Fig. 6, no obvious current decrease over the blank control was observed when MIgG and HSA were used for the sandwich immunoassay at the immunosensor. However, this immunosensor showed obvious current decrease

towards the target protein of HIgG. These results indicate that the cross-reactivity of the immunosensor towards noncognate proteins was negligible.

#### Preferred position for Fig. 6

In addition, five immunosensors were prepared for the repeated measurements of two different concentrations of HIgG. The coefficients of variation were 3.4% and 4.1% for 0.1 and 10 ng/mL HIgG, respectively. In addition, the immunosensor could retain 92% of the initial response for 10 ng/mL HIgG after a storage period of two weeks in dry air at 4 °C. These results indicate that the immunosensor had satisfactory reproducibility and stability.

#### Preferred position for Table 2

In order to assess the possibility of this method for practical applications, different amounts of HIgG were added into bovine serum for recovery tests. The test results from three repeated experiments were listed in Table 2. The recoveries of the standard addition experiments were between 97% and 109% with relative standard deviation (RSD) lower than 5.8%. These results indicate acceptable reliability of the proposed method for real sample analysis.

# 4. Conclusions

A novel ultrasensitive immunoassay method was developed based on the amplified inhibition of the electrochemical signal of rGO-TH nanocomposite by silica nanoprobe. The one-step reduction of GO in TH solution provided a simple method to prepared the well-dispersed rGO-TH nanocomposite which was successfully used as an ideal material for the electrode modification and an excellent electrochemical indicator for immunoassay. After sandwich immunoreaction, the electrochemical signal of the rGO-TH decreased owing to the impedance effect of the dielectric antibody-antigen immunocomplex formed on the immunosensor surface. This current decrease could be further amplified by the

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captured silica nanoprobes which enabled the development of a novel ultrasensitive immunoassay method. This method showed excellent analytical performance for the protein analyte determination with a wide linear range and low detection limit. In addition, the immunosensor had low cost, good reproducibility and stability, and satisfactory reliability. Thus, this method provides a promising potential for practical applications.

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**Table 1.** Comparison of analytical performance of this method with some otherelectrochemical immunosensors for HIgG measurement.

	Detection stratector	Linear	Detection	Ref.	
Immunosensor	Detection strategy	range	limit		
GCE/rGO/CNTs/anti-HIgG	Enzymatic catalysis by	1–500	0.2	1	
	HRP labels	ng/mL	ng/mL	1	
Gold algotrada/austaina/Au	Dopamine oxidation	0.82.00	0.25		
ND /	current inhibition based	0.82-90	0.25 ng/mI	32	
NPS/anti-migo	label-free immunoassay	ng/mL	ng/mL		
Gold electrode/PTH-	K <sub>3</sub> Fe(CN) <sub>6</sub> signal	10 10000			
methylene blue/Au	inhibition based label-free	10-10000	3 ng/mL	33	
NPs/anti-HIgG	immunoassay	ng/mL			
Gold electrode/protein A	Label-free electrochemical	10-1000	5 ng/mL	34	
	impedance spectroscopy	ng/mL			
	Electrocatalytic reduction	0.05 10			
SPCE/IGO-Au	of oxygen by CNT/Pd	0.05-10	44 pg/mL	35	
NPs/anti-HIgG	NPs labels	ng/mL			
GCE/ferrocene-chitosan-	Engranatio estalucia ha	0.2.500			
ionic liquid/Au	Enzymatic catalysis by	0.2-500	50 pg/mL	36	
NPs/anti-HIgG	rGO-Au NPS-HRP labels	ng/mL			
Gold electrode/	Enzymatic catalysis by	30-1000	25	27	
CNTs-Fe <sub>3</sub> O <sub>4</sub> / anti-HIgG	HRP labels	ng/mL	25 ng/mL	37	
	Amplified signal	0.01 100		T1.:-	
UCE/IGU-IH/AU	inhibition by silica	0.01-100	7 pg/mL	l his work	
NPs/anti-HIgG	nanoprobe	ng/mL			

CNTs: carbon nanotubes; HRP: horseradish peroxidase; PTH: polythionine; SPCE: screen-printed carbon electrode

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Table 2 Recovery	tests of HIgG in bovine	serum sample
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No.	Added (ng/mL)	Found (ng/mL)	RSD (%)	Recovery (%)
1	0.1	0.097	5.8	97
2	1	1.03	5.3	103
3	10	10.9	4.7	109

Figure captions:

**Scheme 1** Schematic representation of the preparation process of the immunosensor and the electrochemical detection strategy based on the sandwich immunoassay.

**Fig. 1** A) FT-IR and UV-vis spectra of GO (a, e), rGO-TH nanocomposite (b, f) and TH (c, d).

**Fig. 2** A) Cyclic voltammograms recorded at the rGO-TH (a), rGO-TH/Au NPs (b) modified GCE and the immunosensor (c); B) DPV responses of the immunosensor before (d) and after immunoreaction with 100 ng/mL HIgG based on the direct (e) and sandwich (f) immunoassay format.

**Fig. 3** Nyquist plots recorded at bare GCE (a), rGO-TH (b) and rGO-TH/Au NPs (c) modified GCE; and immunosensor before (d) and after immunoreaction with 100 ng/mL HIgG based on the direct (e) and sandwich (f) immunoassay format.

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Fig. 4 Effect of incubation time on the DPV response of the immunosensor towards 10 ng/mL HIgG.

**Fig. 5** DPV responses of the immunosensor towards different concentrations of HIgG using the proposed amplified signal inhibition strategy (A), and the calibration curve (B). Curves a–f correspond to HIgG at concentrations from 0.01 ng/mL to 300 ng/mL.

**Fig. 6** DPV responses of the immunosensor towards blank control, 1% HAS, 10 ng/mL MIgG and 10 ng/mL HIgG.











Figure 3



Figure 4







Figure 6



For TOC only:

The amplified inhibition of the electrochemical signal of graphene-thionine nanocomposite by silica nanoprobe enabled a novel ultrasensitive immunoassay method.



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