Analyst Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/analyst

Interaction of β-cyclodextrin with nile red in a single live CHO cell: An initiative towards developing a prospective strategy for excretion of adsorbed drugs from the cell membrane

Analyst

s Saptarshi Ghosh^a, Shyamtanu Chattoraj^b, and Nitin Chattopadhyay^{a,*}

Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX DOI: 10.1039/b000000x

A successful endeavour has been made to develop a prospective strategy to oust the drug molecules adsorbed on 10 the cell membranes simply by using the non-toxic βcyclodextrin (\beta-CD). For the purpose, fluorescent probes of different geometries and charge characteristics, namely phenosafranin (PSF) and nile red (NR), were exploited for the in-vitro studies using different lipids as model membranes. 15 Considering the success of the in-vitro study, the present work is extended to a live Chinese Hamster Ovary (CHO) cell using the dye NR acting as a model drug system. Steady state and time resolved confocal microscopic studies reveal that with the introduction of β -CD, some of the adsorbed dye 20 molecules are dislodged from the CHO cell membrane to the nanocavity of B-CD resulting in the formation of dve-B-CD inclusion complex. This study promises development of a prospective strategy for removal of the adsorbed drugs from the cell membranes.

In the coming years, prime concern of biophysical and 25 pharmaceutical research is apprehended to be finding avenues to get rid of the drug induced toxicity and the adverse side effects of the excess drugs deposited in our body, especially in the cell membranes. The severe problem can be tackled from two angles. 30 Firstly, to look for proper and effective drug delivery systems to port the drug directly to the targeted zone so that the required dose is reduced, and secondly, to develop a strategy to remove the excess drugs adsorbed on the cell membranes. In the first context, for the last two decades accomplishments have been 35 made in the arena of drug research in developing a wide variety of targeted drug delivery systems (DDS) composed of nanoscale materials like liposomes, micelles, dendrimers, carbon nanotubes etc.¹⁻⁵ The DDS serve as beneficial vehicles for increasing the efficacy of chemotherapeutics through improved 40 pharmacokinetics and biodistribution, reducing the drug toxicity since the drug is only active in the target domain of the body (for example, in cancerous tissues).¹ However, not much work has been done in the second context, i.e., excretion of excess and unused drugs adsorbed in the body system to reduce or eradicate 45 the drug toxicity or side effects. There are, of course, a few longstanding chemical methods for removing the excess drugs from the body system. For example, acidic drugs like barbiturates, salicylates, sulfonamides etc. are known to be excreted by making the urine alkaline by treating with ⁵⁰ bicarbonates and similarly, alkaline drugs like amphetamine, pethidine etc. can be removed from the body by making the urine acidic using ascorbic acid, ammonium chloride etc.⁶ But, to the best of our knowledge, there is no such method for removing the adsorbed drug molecules from the cell membranes. Keeping this

- ⁵⁵ in mind, for the past few years, our group has been interested in developing a strategy to excrete the adsorbed drugs from the cell membrane, rather easily, by using cyclodextrins which are non toxic oligosaccharides and are well known for their ability to form inclusion complexes with variety of probes. Although CDs
 ⁶⁰ have been shown to remove cellular cholesterol by host-guest inclusion complexation mechanism⁷, majority of the literature reports go in favour of the non-destructive character of CDs towards the cell components. On this basis, CDs are often used as DDS for controlled and safe drug delivery.⁸⁻¹⁰ A recent study
 ⁶⁵ from our group has also established CDs to be safe for the serum
- proteins from the structural point of view.¹¹

Since lipids are the principal constituents of the cell membranes, in our early in-vitro works, lipids like dimyristoyl-Lα-phosphatidylglycerol (DMPG), and dimyristoyl-L-a-70 phosphatidylcholine (DMPC) differing in charge characteristics and fluorescent probes of varying geometry and charge characteristics like cationic phenosafranin (PSF) and non-ionic nile red (NR) have been exploited as models of the membranes and drugs respectively.^{12,13} Series of techniques like steady state 75 fluorescence and fluorescence anisotropy, micropolarity determination, fluorescence quenching, dynamic light scattering and time resolved flurometry reveal that addition of β -CD to the probe-lipid complexes leads to the complete or partial removal of the probes from the aforesaid complexes resulting in the ⁸⁰ formation of probe—β-CD inclusion complexes.^{12,13} Looking at the success of removal of the probes (PSF and NR) from the different lipid membranes, we have been inspired to extend the study further to cellular level to find if the strategy is applicable in real biological systems like a live Chinese Hamster Ovary 85 (CHO) cell.

The fluorescent nonionic dye from the phenoxazone family, nile red (NR) (Scheme 1) has been extensively used as a staining agent in biological tissues.¹⁴ The photophysical properties of NR have also been the subject of many studies because of the intense ⁹⁰ solvent-dependence of its fluorescence enabling the probe to be

ARTICLE TYPE

55

56

57

58

59 60 75

used conveniently to study the polarity related aspects in the microheterogeneous environments including micelles, reverse micelles, cyclodextrins, zeolites, etc.¹⁵⁻¹⁹. It has also been exploited to explore the heterogeneity of membranes^{20,21} and to ⁵ study the formation of dendrimer-surfactant supramolecular assemblies.²² The solvent sensitivity of the probe in terms of its fluorometric characteristics provides an easy way to assess the location of it in microheterogeneous media.^{13,15-19}

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57

58

59 60 Single molecule spectroscopy (SMS) provides new avenues to ¹⁰ study various photobiological aspects in a live cell. Phenomena like fluorescence lifetime trajectory during transcription,²³ lipid rafts in live cell,²⁴ diffusion in different regions of cell,²⁵ nanoscale dynamics of membrane lipids in a live cell,²⁶ proteinbinding dynamic²⁷ etc. have been studied through SMS. For our ¹⁵ purpose in the present work, we have exploited steady-state and time-resolved confocal microscopy to study the interaction of βcyclodextrin with nile red in a single live Chinese hamster ovary (CHO) cell. Experimental details including cell preparation, staining procedure and confocal microscopic technique have been ²⁰ described in the Electronic Supplementary Information (ESI).



Scheme 1: Structure of nile red (NR)

Fig. 1(A) shows the confocal FLIM (fluorescence lifetime 25 imaging microscopy) image of a NR stained CHO cell. The different colors in the confocal microscopic image of NR stained CHO cell indicates different lifetimes of NR in the various parts of the cell.²⁸ From the image, different regions of the cell are clearly visible - nucleus, cytoplasm, membrane and a few bright ³⁰ spots which are ascribed to lipid droplets.^{29,30} It is evident that NR is distributed almost uniformly throughout the membrane of the cell suggesting a significant binding of the dye to the constituting lipid of the membrane. This is in tune with our invitro results where NR exhibits a strong binding with both DMPG 35 and DMPC lipids.¹³ Fig. 2(A) depicts the steady state emission spectra of NR in different media like phosphate buffered saline (PBS) solution, 6 mM β-CD in PBS solution, CHO cell membrane and membrane together with β-CD solution. As already stated, the position of the emission band maximum of NR ⁴⁰ is sensitive to the polarity of the medium.¹⁶ In aqueous buffer NR exhibits an emission maximum at around 660 nm.13 The emission maximum of NR in the β-CD medium does not reveal a big shift in the peak position. In the CHO cell membrane, NR shows two emission maxima; one remaining in the same position as that in 45 PBS solution, assigned to the free dye while the other much blue shifted one peaking at ~ 610 nm. This blue-shifted band arises due to the NR bound to the hydrophobic region within the CHO cell membrane. The whole set of data qualitatively agrees our invitro results.13

⁵⁰ The appreciable blue shift in the emission maximum of NR in the cell membrane implies that the dye experiences a much less polar environment within the membrane compared to the aqueous NR, a perceptible change in the relative emission intensities at the ⁵⁵ two band positions is observed (Fig. 2). The 610 nm band is diminished drastically relative to the band peaking at 655 nm, close to the position observed in the β -CD medium.¹³ This drastic modification of the emission maximum in the presence of β -CD is rationalized from the consideration of a competitive binding of ⁶⁰ the probe between the cell membrane and β -CD environment. Thus, in the presence of β -CD, the membrane bound NR molecules are dislodged from the CHO cell to get encapsulated in the nanocavity of β -CD leading to the drastic decrease in the fluorescence intensity of the 610 nm emission band. The small ⁶⁵ blue shift in the emission maximum of the final solution (membrane + β -CD) relative to the position in aqueous buffer (Fig. 2(A)) corroborates the formation of NR— β -CD inclusion complex.¹³

buffer milieu. Upon addition of β-CD on the membrane bound



Fig. 1 Confocal FLIM images of single Chinese Hamster Ovary (CHO) cell stained by NR in absence (A) and in the presence of β -CD (B). The white line shows scale bar. Both the images are taken with same exposure time (at λ_{exc} = 470 nm) and same laser power.



Fig. 2 (A) Normalized emission spectra of NR in buffer (black), β -CD (red), CHO cell membrane (blue) and CHO cell membrane in presence so of 6 mM β -CD (green). (B) Un-normalized emission spectra of NR in CHO cell membrane in the absence (blue) and presence of 6 mM β -CD (green). Both the spectra are taken at same laser power (λ exc = 470 nm).

Fig. 1 (B) represents the confocal FLIM image of the single ⁹⁰ CHO cell stained by NR upon addition of β -CD. The reduction in the intensity of the membrane bound dye in the presence of β -CD is also evident from a comparison of the fluorescence spectra of dye in the two environments (Fig. 2(B)). Since in Fig. 1, we have presented the two images of the single CHO cell in the same ⁹⁵ intensity scale the image after addition of β -CD appears relatively pale. The two yellow patches on both sides and away from the cell membrane depict the NR— β -CD inclusion complex. Clustering of β -CD is resulted because of the inhomogeneity of the medium as mentioned in the Experimental Section (ESI). ¹⁰⁰ Thus, the steady state emission study unambiguously divulge that

15

introduction of β -CD leads to removal of the bound probe from the cell membrane resulting in the formation of probe— β -CD complex.

Time resolved measurement serves as a tool to unearth the $_5$ local environment of the dye within microheterogeneous environments. A positive indication from the steady state confocal microscopic study that NR is expelled from the CHO cell membrane upon addition of β -CD prompted us to corroborate it from the fluorescence lifetimes of the probe in the aforesaid ¹⁰ environments. The burst integrated fluorescence lifetime (BIFL) histograms of membrane bound NR in the absence and presence of β -CD are represented in Fig. 3 and the corresponding decay parameters are given in Table 1. In buffer solution lifetime of NR



Fig. 3 Lifetime histogram of NR in CHO cell membrane in the (A) absence and (B) presence of β -CD. Excitation wavelength was 470 nm.

²⁰ *Table 1*. Lifetime values of NR in different conditions in a CHO cell membrane

| | $\tau_1(ns)$ | a 1 | $\tau_2(ns)$ | a ₂ | $\tau_3(ns)$ | a ₃ |
|------------------------------|--------------|------------|--------------|-----------------------|--------------|----------------|
| membrane | 5.23 | 0.92 | 6.72 | 0.08 | | |
| membrane in presence of β-CD | 4.62 | 0.88 | 6.48 | 0.04 | 2.23 | 0.08 |

comes out to be 0.68 ns, in well agreement with the existing ²⁵ literature.^{13,17} In β -CD environment NR shows a bi-exponential fluorescence decay giving two lifetime values, the average being 2.40 ns. In CHO cell membrane also NR gives two lifetime components with values of 5.23 ns and 6.72 ns (Table 1). As is known, multiple lifetimes of the polarity-sensitive probes often ³⁰ originate because of the location of the probe in different polarity

regions within the heterogeneous environments.³¹ A glance on Table 1 reflects that in the presence of added β -CD the values of the two lifetime components of membrane bound NR are reduced marginally. Interestingly, we observe a third lifetime component ³⁵ with a much lower value of 2.23 ns (Fig. 3(B)). This lifetime value qualitatively agrees with the $\tau_{average}$ value of 2.12 (± 0.3) ns observed for NR in β -CD ($\tau_{average} = 2.40$ ns in the present experiment and $\tau_{average} = 1.84$ ns in the previous experiment¹³) and is assigned to the NR encapsulated in the β -CD cavity. The ⁴⁰ observation of this lifetime component is more important than the relative abundance of it since the latter depends on many factors like position of monitoring, relative binding affinities of the probe with the CHO cell membrane and the β -CD environments, absolute concentration of β -CD etc. Of course, one can

⁴⁵ reasonably presume that the extent of removal of the probe from the cell membrane upon addition of CDs should depend on the specific CD chosen for the purpose, since this controls the relative binding affinities of the probe between the two environments. In the present study, we have opted for β-CD over
 ⁵⁰ other CDs because of the suitable matching of the size of the probe with the cavity size of β-CD resulting in the formation of 1:1 inclusion complex.¹⁹

Conclusions

The present study reveals that NR bound to the membrane of a s5 live CHO cell can be expelled by simply using the β cyclodextrin. This is in tune with our previous studies where dyes were removed from DMPG and DMPC lipids. The degree of expulsion can be controlled as well, since it depends on the relative binding affinities of the probe with the cell membrane 60 and the CDs. Non-toxicity of the external agent (cyclodextrins) on one hand and the high water solubility of the CDs on the other promises to make the strategy very effective since the drug molecules extracted from the cell membranes and making the inclusion complexes with the CDs can be excreted from the body 65 easily through the normal water ejection channels. Establishment of the observation to be general for various fluorescent probes / drug molecules and a range of lipids / cell-membranes along with a variation of the extractor (like different CDs) should build a strong foundation for developing a strategy for the possible 70 removal of adsorbed drug molecules from the live cell membranes. However, being the first attempt for the living cell Analyst Accepted Manuscrij

Acknowledgements

The financial supports from the DST and DBT, Govt. of India, are gratefully acknowledged. SG and SC thank UGC and CSIR, Govt. of India, respectively for their fellowships. Thanks are due to Professor K. Bhattacharyya of IACS, Kolkata for allowing us to use his laboratory for the confocal microscopic measurements and for helpful discussion. We thank the reviewers for their constructive suggestions to improve the merit of the work.

system, the present exertion holds a colossal importance from the

perspective of the drug research in the coming future.

110

115

120

125

130

135

Notes and references

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57

58

59 60

- ^aDepartment of Chemistry, Jadavpur University, Kolkata 700 032, India
- ^bDepartment of Physical Chemistry, Indian Association for the
- 5 Cultivation of Science, Jadavpur, Kolkata 700 032, India
- *Corresponding author: Fax: 91-33-2414 6584
- E-mail: <u>nitin.chattopadhyay@yahoo.com</u>
- † Electronic Supplementary Information (ESI) available: Experimental ¹⁰ section. See DOI: 10.1039/b00000x/
- V. V. Ranade, M. A. Hollinger, J. B. Cannon, *Drug delivery systems*, 2nd ed.; CRC, Press: Boca Raton, 2004.
- C. K. Kim, P. Ghosh, C. Pagliuca, Z.-J. Zhu, S. Menichetti, V. M.
 Rotello, Entrapment of hydrophobic drugs in nanoparticle monolayers with efficient release into cancer cells, *J. Am. Chem. Soc.*, 2009, **131**, 1360-1361.
- D. Peer, J. M. Karp, S. Hong, O. C. Farokhzad, R. Margalit, R. Langer, Nanocarriers as an emerging platform for cancer therapy, *Nat. Nanotechnol.*, 2007, 2, 751-760.
- 4 A. Bianco, K. Kostarelos, M. Prato, Applications of carbon nanotubes in drug delivery, *Current opinion in Chemical Biology*, 2005, 9, 674-679.
- R. Mehta, G. Lopez-Berestein, R. Hopfer, K. Mills, R. L. Juliano,
 Liposomal amphotericin B is toxic to fungal cells but not to mammalian cells, *Biochim. Biophys. Acta*, 1984, **770**, 230-234.
 - J. G. Buch, *Quick review of pharmacology*, 3rd Ver, 2010.pp 21.
 R. Zidovetzki, I. Levitan, Use of cyclodextrins to manipulate plasma membrane cholesterol content: evidence, misconceptions and control strategies, *Biochim. Biophys. Acta.*, 2007, **1768(6)**, 1311-1324.
- 8 F. J. Otero-Espinar, J. J. Torres-Labandeira, C. Alvarez-Lorenzo, J. Blanco-Méndez, Cyclodetrins in drug delivery systems, *J. Drug Del. Sci. Tech.*, 2010, 20, 289-301.
- 9 G. Tiwari, R. Tiwari, A. K. Rai, Cyclodetrins in delivery systems: 35 Applications, *J. Pharm. Bioallied. Sci.*, 2010, **2**, 72-79.
- F. Hirayama, K. Uekama, Cyclodextrin-based controlled drug release system, *Advanced Drug Delivery Reviews*, 1999, 36, 125-141.
- S. Ghosh, B. K. Paul, N. Chattopadhyay, Interaction of cyclodextrins with human and bovine serum albumin: A combined spectroscopic and computational investigation, *J. Chem. Sci.*, 2014, **126**, 931-944.
- 12 D. Sarkar, D. Bose, A. Mahata, D. Ghosh, N. Chattopadhyay, Differential interaction of β-cyclodextrin with lipids of varying surface charges: A spectral deciphering using a cationic phenazinium dye, J. Phys. Chem. B, 2010, 114, 2261-2269.
- 45 13 B. Jana, S. Ghosh, N. Chattopadhyay, Competitive binding of nile red between lipids and β-cyclodextrin, J. Photochem. Photobiol. B: Biology, 2013, 126, 1-10.
- P. Greenspan, E. P. Mayer, S. D. Fowler, Nile red: A selective fluorescent stain for intracellular lipid droplets, *J. Cell Biol.*, 1985, 100, 965-973.
- 15 J. L. Meinershagen, T. Bein, Optical sensing in nanopores. encapsulation of the solvatochromic dye nile red in zeolites, *J. Am. Chem. Soc.*, 1999, **121**, 448-449.
- N. Sarkar, K. Das, D. N. Nath, K. Bhattacharyya, Twisted charge
 transfer processes of nile red in homogeneous solutions and in
 faujasite zeolite, *Langmuir*, 1994, **10**, 326-329.
- 17 A. Datta, D. Mandal, S. K. Pal, K. Bhattacharyya, Intramolecular charge transfer processes in confined systems. Nile red in reverse micelles, *J. Phys. Chem. B*, 1997, **101**, 10221-10225.
- 60 18 N. C. Maiti, M. M. G. Krishna, P. J. Britto, N. Periasamy, Fluorescence dynamics of dye probes in micelles, *J. Phys. Chem. B*, 1997, **101**, 11051-11060.
 - P. Hazra, D. Chakrabarty, A. Chakraborty, N. Sarkar, Intramolecular charge transfer and solvation dynamics of nile red in the nanocavity of cyclodextrins, *Chem. Phys. Letts.*, 2004, **388**, 150-157.
- 20 G. Krishnamoorthy, Probing the link between proton transport and water content in lipid membranes, *J. Phys. Chem. B*, 2001, **105**, 1484-1488.
- 21 O. A. Kucherak, S. Oncul, Z. Darwich, D. A. Yushchenko, Y. Arntz,
- ⁷⁰ P. Didier, Y. Mely, A. S. Klymchenko, Switchable nile red based

probe for cholesterol and lipid order at the outer leaflet of biomembranes, J. Am. Chem. Soc., 2010, **132**, 4907-4916.

- 22 D. M. Watkins, Y. Sayed-Sweet, J. W. Klimash, N. J. Turro, D. A. Tomalia, Dendrimers with hydrophobic cores and the formation of supremalagular dandrimer, surfactant assemblies. *Language* 1007
- ⁷⁵ supramolecular dendrimer- surfactant assemblies, *Langmuir*, 1997, 13, 3136-3141.
- 23 M. Sorokina, H.-R. Koh, S. S. Patel, T. Ha, Fluorescence lifetime trajectories of a single fluorophore reveal reaction intermediates during transcription initiation, *J. Am. Chem. Soc.*, 2009, **131**, 9630-9631.
- 24 H. M. Kim, B. H. Jeong, J.-Y.Hyon, M. J. An, M. S. Seo,; J. H. Hong, K. J. Lee, C. H. Kim, T. Joo, S.-C. Hong, B. R. Cho, Two-photon fluorescent turn-on probe for lipid rafts in live cell and tissue, *J. Am. Chem. Soc.*, 2008, **130**, 4246-4247.
- 85 25 S. A. Kim, K. G. Heinze, P. Schwille, Fluorescence correlation spectroscopy in living cells, *Nat. Methods*, 2007, 4, 963-973.
 - 26 C. Eggeling, C. Ringemann, R. Medda, G. Schwarzmann, K. Sandhoff, S. Polyakova, V. N. Belov, B. Hein, C. V. Middendorff, A. Schonle, S. W. Hell, Direct observation of the nanoscale dynamics of membrane lipids in a living cell, *Nature*, 2009, **457**, 1159-1163.
- 27 Y. Philip, V. Kiss, G. Schreiber, Protein-binding dynamics imaged in a living cell, Proc. Natl. Acad. Sci. U.S.A., 2012, 109, 1461-1466.
- 28 S. Chattoraj, S. Saha, S. S. Jana, K. Bhattacharyya, Dynamics of gene silencing in a live cell: Stochastic resonance, *J. Phys. Chem. Lett.*, 2014, 5, 1012-1016.
- 29 S. Ghosh, S. Chattoraj, T. Mondal, K. Bhattacharyya, Dynamics in cytoplasm, nucleus, and lipid droplet of a live CHO cell: Timeresolved confocal microscopy, *Langmuir*, 2013, 29, 7975-7982.
- D. K. Sasmal, S. Ghosh, A. K. Das, K. Bhattacharyya, Solvation
 dynamics of biological water in a single live cell under a confocal microscope, *Langmuir*, 2013, 29, 2289-2298.
 - 31 A. Chakrabarty, A. Mallick, B. Haldar, P. Das, N. Chattopadhyay, Binding interaction of a biological photosensitizer with serum albumins: A biophysical study, *Biomacromolecules*, 2007, 8, 920-927.

Page 5 of 5





¹⁰ Steady state and time-resolved confocal microscopic studies in a single live Chinese Hamster Ovary (CHO) cell suggest that the biological staining dye nile red (NR) adsorbed on the cell membrane can be excreted simply by using β -cyclodextrin.