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Palladium-catalyzed inter- and intramolecular formation of C–O bonds from allenes

Jean Le Bras and Jacques Muzart*

The Pd-catalysed formation of a C-O bond from allenes mainly occurs via the inter- or intramolecular reaction with a hydroxyl group belonging to an alcohol, a phenol or an acid. A carbonyl or a carbonate entity can also be involved. In most cases, the formation of the C-O bond is one step of a domino reaction leading also to a C-C or C-N bond, or to another C-O bond. Thus, a wide range of products, in particular highly functionalized heterocycles, have been synthesised. Catalytic cycles have been proposed but the mechanisms often remain speculative.

1. Introduction

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The synthesis and transformation of 1,2-dienes involve a variety of Q3 procedures.¹ The presence of two contiguous C=C bonds pro-25 vides a unique reactivity, which has been exploited in synthesis. Although most of the studies on the transition metal-catalysed reactions of unsaturated compounds concern reactions of alkenes, alkynes and arenes, those on allenes have nevertheless led to reviews.^{2,3} The purpose of the present review is to highlight 30 the Pd-mediated processes leading to the formation of at least one C-O bond from an inter- or intramolecular reaction with an

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allenyl unit. That is often only one step of a domino reaction. With the assistance of palladium complexes, allenes are indeed capable of undergoing 1,2-addition with both electrophiles and nucleophiles. In most cases, such three-component reactions lead to the attachment of the electrophile to the central carbon of the allene, whereas the nucleophile is bonded to the 1- or 3-carbon. The order of these steps remains usually determinate,⁴ as exem- $\mathbf{Q4}$ plified in Scheme 1 for reactions between buta-2,3-dien-1-ol and aryl, vinyl or allyl halide using either a Pd⁰ or a Pd^{II} catalyst. The attack of the allenyl moiety by the hydroxy group would be an intramolecular Wacker-type reaction requiring the activation of the unsaturated system by a Pd^{II} species, the latter being the PdX_2 catalyst or the RPdX (R = Ar, vinyl, η^{1} - or η^{3} -allyl) obtained by insertion of Pd⁰ into a C-halide bond. After the formation of the C-O bond from 1A or 1B, the resulting intermediate can lead to

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Jean Le Bras

Jean Le Bras obtained his Engineering Diploma from ENSCP-Paris and his MSc degree (DEA) from UPMC. In 1996, he joined the group of Dr Hani Amouri, where he studied iridium mediated functionalization of phenols and obtained his PhD in 1998. He then joined the group of Professor John A. Gladysz in Salt Lake City (USA) and in Erlangen (Germany) and worked on the synthesis of organometallic complexes with polyynediyl

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in Vienne la Ville, a small village in the Argonne area, 200 km east of Paris. He studied chemistry at the Université de Reims Champagne-Ardenne and received his degrees (Doctorat de 3^{ème} cycle in 1972, Doctorat d'Etat in 1976) for his work with J.-P. Pète on photochemical rearrangements of α, β epoxyketones and β -diketones. He spent 15 months as a postdoctoral fellow of the National Science Foundation working with Nobel

Jacques Muzart was born in 1946,

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the formation of the C–C bond with regeneration of the catalyst (paths *a* and *b*). Instead of the Wacker-type reaction, **1B** can evolve *via* the Heck-type addition to afford the allylpalladium complex **1C** or **1D** (path *c*). The formation of the substituted dihydrofuran would then occur from nucleophilic addition to the η^3 -allylpalladium complex (path *d*) or *via* the palladacycle **1E** (path *e*). While **1D** is often proposed as an intermediate, it seems that **1E** is never

suggested for such domino reactions. A side-reaction from **1F** is the protonolysis leading to 2,5-dihydrofuran (path *f*). Given these different possibilities, the mechanism of these Pd-catalysed cyclisation-coupling reactions remains a matter of debate. It has to be however pointed out that the efficient formation, at room temperature, of η^3 -allylpalladium complexes from allenes and alkyl or acvl palladium complexes has been reported.⁵

The oxygen atom of the C–O bond obtained from addition to allenes arises from a hydroxy group (water, alcohol, phenol, acid) or a carbonyl. Moreover, the C–O bond can be formed *via* an inter- or intramolecular reaction. The present review is

40 an inter- or intramolecular reaction. The pres organised in terms of these different angles.

2. Hydrohydroxylation

⁴⁵ Under Pd⁰ catalysis, the reaction of allene with water leads to traces of 3-methyl-2-methylene-3-buten-1-ol.⁶ Inoue and co-workers discovered that this dimerisation–hydroxylation reaction became efficient under carbon dioxide pressure, especially in THF (eqn (1)).⁶ The role of CO₂ has not been determined. An experiment with D₂O has shown that water is the source of the incorporated hydrogen.



3. Intermolecular etherification

3.1. Hydroalkoxylation

In contrast to the above Pd^0 -catalysed water addition–dimerisation of allene, a similar process with methanol occurs with a reasonable yield in the absence of CO_2 (eqn (2)).⁶ According to the authors, the efficiency decrease with the size of the alcohol could be due to steric factors.⁷

R = Me (40%), Et (16%), *n*-Bu (traces)

Subsequently, the Rutjes team reported the Pd^{II}-catalysed reaction of secondary alcohols with methoxyallene under conditions which did not mediate the dimerisation of the substrate (eqn (3)).⁸ Attack of the alcohol onto the complex **2A** formed by the coordination of Pd^{II} with the more electron-rich oxygensubstituted double bond would lead to the vinylpalladium



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complex 2B, the protonolysis of which gives the acetal and 10 regenerates the Pd^{II} catalyst (Scheme 2). The Ruties procedure has been extensively used to obtain acetals from protected propa-1,2-dien-1-ols and a variety of functionalised secondary $(eqn (4), {}^{9}(5)^{10} and (6)^{11})^{12}$ or tertiary $(eqn (6)^{11})$ alcohols.¹³

Some years after the above Rutjes report, Yamamoto et al. 15 used the Pd(PPh₃)₄/PhCO₂H association to catalyse the hydroalkoxylation of arylallenes by primary and secondary alcohols (eqn (7)).¹⁴ This reaction would involve the oxidative addition of



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 R^1 = Me, R^2OH = CH_2 =CHCH₂CH(OH)CO₂Me, reflux, 3-6 h: 98% $R^1 = Bn, R^2OH = CH_2 = CHCH_2CMe(OH)CO_2Me, rt, 16 h: 97\%$ $R^1 = Me$, $R^2OH = Me_3SiC = CCH_2CH(OH)CH_2OBn$, reflux, 4 h: 88%

benzoic acid to Pd⁰ to provide the hydridopalladium species 3A. The subsequent hydropalladation of the allene would afford 20 the η^3 -allylpalladium complex **3B** (Scheme 3), regioselective addition of the alcohol to the latter delivering the allyl ether.



3.2. Hydroaryloxylation

In 1998, Yamaguchi et al. observed weak addition of ArOH to 05 the dimer of undeca-1,2-diene obtained under catalysis with the Pd₂(dba)₃-ArOH system.¹⁵ Much more efficient conditions have then been disclosed by Grigg's team to synthesise 2-(hetero)aryloxymethyl-3-methyl-1,3-butadienes (eqn (8)).¹⁶







Two mechanisms were proposed (Scheme 4, paths *a* and *b*).¹⁶ Path *a* involves the insertion of Pd⁰ into the ArO–H bond and 20 the coordination of allene leading to **4A**; the subsequent hydropalladation gives **4B** which reacts with another molecule of allene to provide **4C**. From **4C**, the product would be obtained either *via* the reductive elimination of Pd⁰, or *via* a η^3 -allylpalladium intermediate and its attack by phenolate 25 occurring externally or by transfer. Path *b* involves the addition 1

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of ArOH to the palladacycle **4D** to afford the Pd^{IV} intermediate **4E**, which also leads to **4C**. Whole etherification of 1,3,5-trihydroxybenzene has also been carried out (eqn (9)).¹⁶With the Rutjes procedure, the hydroaryloxylation of benzyloxyallene effectively happens in a few seconds at room temperature (eqn (10)),^{11,17} while that of methoxyallene requires reflux and results in low yields.¹⁷ The reason for this remarkable difference was unexplained.¹⁷

3.3. Alkoxyarylation

In 1991, Larock et al. disclosed the Pd⁰-catalysed synthesis of the 4-methyleneisochroman derivative shown in eqn (11).¹⁸ The mechanism proposed by them involves the arylation of the central atom of the 1,2-diene followed by regioselective 15 nucleophilic addition of the alcoholate to the more substituted extremity of the n^3 -allylpalladium intermediate (see Scheme 1, path d). (2-Iodo(or bromo)phenyl)methanol has also been used for enantioselective annelations (eqn (12))^{19,20} and for the coupling with heterosubstituted allenes (eqn (13)²¹ and (14)).²²With *o*-iodobenzaldehyde and 2,3-allenols, Ma and co-20 workers reported a domino reaction providing oxa-bridged benzocycloheptanes (eqn (15)).²³ These syntheses would imply the η^3 -allylpalladium intermediate 5A (Scheme 5). The base mediates the formation of the alcoholate to perform the 25 intramolecular addition to the aldehyde, resulting in the



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²⁰ subsequent addition to the η^3 -allylpalladium moiety. This high diastereoselectivity observed from optically active 2,3-allenols (eqn (16)) supports the proposed catalytic cycle.





3.4. Alkoxyvinylation

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Annelation of allenes with vinylic halides bearing an alcohol group led Larock's team to isolate a variety of heterocycles (eqn (17) to (19)) through, probably, η^3 -allylpalladium intermediates, the reaction being enantioselective in using chiral



with PPh₃ (0.05 equiv.), *n*-Bu₄NCI (1 equiv.) and Na₂CO₃ (5 equiv.) $R = n-C_8H_{17}, 72 h: 88\%; R = OMe, 48 h: 95\%$



ligands (eqn (17)).^{19,20,24} With 2-iodo-3-methylbut-2-en-1-ol, the η^1 - or η^3 -allylpalladium intermediate inserts a second allene before closing to a seven-membered heterocycle (eqn (19)).²⁴

OMe

(5 equiv.)

Pd(OAc)₂ (0.05 equiv.) PPh₃ (0.05 equiv.)

n-Bu₄NCI (1 equiv.)

Na₂CO₃ (5 equiv.)

DMF, 40 °C, 20 h



56%

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OMe

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Csl, CsHCO34

Cs₂CO₃

Scheme 7

OH

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0 0

3.5. Aryloxy-arylation

The annelation of undeca-1,2-diene with *m*-iodo-*p*-hydroxyacetophenone has also been documented by Larock's team (eqn (20)).^{18,20} Various examples of a similar domino reaction have then been reported using slightly different experimental conditions (eqn (21),²⁵ (22),²¹ (23)²⁶ and (24) to (26)^{22,27}).









Arl

ArPdl

Pd⁰

Pdl

R = OMe (70%), H (65%), I (59%)

- Recently, Cao and co-workers obtained a 4-amino-2*H*-chromene from the addition of 2-hydroxyphenylboronic acid to an 1-alkoxycarbonyloxy allenamide (eqn (27)).²⁸ The proposed mechanism involves the allenylpalladium intermedi-5 ate **6A**, which undergoes a transmetalation reaction with the boronic acid to afford **6B** (Scheme 6). Reductive elimination of Pd⁰ from the latter provides **6C**. According to the authors, **6C** undergoes spontaneous oxycyclisation leading to the amino-chromene. We however suspect that this intramolecular hydro-10 aryloxylation is mediated by the catalyst, as exemplified with
- the intermolecular reaction depicted in eqn (10).



Grigg and co-workers disclosed a three-component reaction leading to isoxazolidines, which involves the successive addition of an aryl iodide and *N*-(2-hydroxybenzylidene)methanamine oxide to allene, followed by 1,3-dipolar cycloaddition of the nitrone to the C—C bond (eqn (28), Scheme 7).²⁹



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3.6. Other carboalkoxylations

Tanaka *et al.* used the equivalence of 1-bromoallenes to allyl dications to synthesise eight-membered heterocycles from 1-35 bromoallenes bearing a carbon nucleophilic functionality (eqn (29)).³⁰ The proposed mechanism implies the insertion of Pd⁰ into the C-Br bond (Scheme 8). The resulting η^{1} allenylpalladium complex **8A** is in equilibrium with the η^{3} propargylpalladium **8B**, intramolecular nucleophilic addition 40 to the central atom of the latter giving the palladocyclobutene

- 40 to the central atom of the latter giving the palladocyclobutene **8C.** Protonation of **8C** by methanol generates the η^3 allylpalladium intermediate **8D**, attack of the methoxide to the terminal carbon of which delivers the organic compound and regenerates the catalyst.^{30,31}
- 45 Pd⁰-catalysed annelation of the allenylenone of eqn (30) would occur through the formation of the palladacycle **9A** or **9B** (Scheme 9), their protonation by methanol affording the η^3 -





allylpalladium complex **9**C, which, as above, is attacked by the methoxide.³²

3.7. Aminoalkoxylation

The intramolecular Pd⁰-catalysed aminopalladation of 1-bromoallenes 3-substituted with a nucleophilic six- or seven-membered 45 amino-tether can be followed by addition of methoxide (eqn (31) and (32)), *via* a mechanism similar to that of Scheme 8. 30,33,34



R = Me, Pd(PPh₃)₄ (0.2 equiv.), MeONa (1.5 equiv.), MeOH, 50 °C, 3 h: 56% R = Bn, Pd(PPh₃)₄ (0.1 equiv.), NaH (1.5 equiv.), MeOH/THF (1:1), 50 °C, 4 h: 31%

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A different order of the steps would be involved when the nucleophilic amino group is located at the 4-position, the addition of the methylate to the central atom of the allene preceding the cyclisation reaction (eqn (33)).³³







3.8. Dialkoxylation

In contrast to the reaction depicted in eqn (29), the Pd⁰-catalysed reaction of the hydroxy-tethered 1-bromoallene of eqn (34) with methanol led to the dimethoxylation of the allenyl moiety instead of the eight-membered ring, although the experimental conditions were similar.³⁰



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3.9. Alkoxy-alkoxycarbonylation

Alper's team disclosed the effective synthesis of methyl 2methoxymethylacrylate from allene under Pd^{II} -catalysed carbonylation conditions (eqn (35)).³⁵ This methoxy-methoxycarbonylation reaction also occurs from 1,l-disubstituted allenes but with lower yields (eqn (35)). Different, but less efficient, experimental conditions have been reported by Trofimov *et al.* (eqn (36)).³⁶ Various possibilities for the catalytic cycle of the methoxy-methoxycarbonylation can be envisaged from the authors' proposals, all of them involving the methoxycarbonylpalladium chloride **10A** as the key intermediate reacting with the 1,2-diene to afford **10B** or **10C** (Scheme 10).^{36b,37} The reaction of **10B** with methanol would lead to the methoxypalladium complex **10D**, which would suffer the reductive elimination of



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(44)

(45)



Pd⁰ giving the product. From 10C, the reductive elimination of Pd⁰ leading to the allylic chloride 10E would be followed by
reinsertion of Pd⁰ into the C–Cl bond to give 10B. With methoxyallene as the substrate, Trofimov's procedure affords 2-chloro-3,3-dimethoxyprop-1-ene as the main compound (eqn (37)) through a mechanism which would only involve Cu^{II} catalysis.³⁷



central atom of the allenyl unit to give a η^3 -allylpalladium intermediate. Addition of the phenolate to the latter leads to the annelation product.

A similar three-component reaction, but employing carbon monoxide at atmospheric pressure, has been reported by Grigg's team to synthesise 3-methylene-4-chromanone and 3-methylene-1-tosyl-2,3-dihydroquinolin-4-one (eqn (42)).³⁹ Moreover, such domino reactions have also been carried out in the presence of

Pd(PPh3)4 (0.05 equiv.)

CO₂Me

30 3.10. Aryloxy-aroylation

Under carbon monoxide pressure and basic conditions, the Pd⁰-catalysed reaction of *o*-iodophenols with allenes provides 3-methylene-2,3-dihydro- or 3-vinyl-4*H*-1-benzopyran-4-one derivatives in fair to high yields (eqn (38) to (41)).³⁸ As a possible mechanism, Alper and Okura suggested the Pd⁰-catalysed carbonylation of the iodophenol to afford the corresponding aroylpalladium complex, which undergoes addition to the

(3 equiv.)

(1atm)

3 equiv.

CO

(1 atm)

OH (1 atm)

(1 equiv.)

(1.3 equiv.)



60%

 $Ar = Ph (78\%), p-MeOC_6H_4 (83\%)$

Pd(PPh₃)₄ (0.05 equiv.)

K₂CO₃ (2 equiv.) Ag₂O (0.1 equiv.)

DBU (1 equiv.)

Pd(PPh₃)₄ (0.05 equiv.) K₂CO₃ (2 equiv.)

144 h

CO2Me PhMe, 45 °C, 48 h

PhMe, 75 °C.

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nucleophilic amines realising an *in situ* Michael addition (eqn (43)),³⁹ or additives leading to 1,3-dipolar cycloaddition (eqn (44) and (45)) or cyclocondensation products (eqn (46)).⁴⁰



4. Intramolecular etherification

4.1. Hydroalkoxylation

Yamamoto and co-workers synthesised optically active rose oxides *via* the hydroalkoxylation of (3R)- and (3S)-3,7-dimethyl-6,7-octadien-1-ol catalysed by chiral Pd species (eqn (47)).⁴¹



 The Pd⁰-catalysed ring expansion of hydroxy methoxyallenylphthalans depicted in eqn (48) has been reported by Nagao
 and co-workers.⁴² A likely mechanism implies the hydridopalladium intermediate **11A**, which generates the n³-allylpalladium







complex **11B** (Scheme 11). The rearrangement of the latter affords the one-atom ring expanded product and the starting catalyst.⁴²

The Alcaide–Almendros team has observed unexpected hydroalkoxylations when studying the alkoxyarylation of 3-hydroxy-3-(1methyl-1,2-propadienyl)-2-indolinone (eqn (49))⁴³ and 1-(1-methyl-1,2-propadienyl)-2,3-dihydroxypropyl *p*-methoxybenzoate (eqn (50)).^{44,45} We suspect that PhI plays a role in the formation of the six- and five-membered rings. PhI would react with the Pd⁰ catalyst to afford PhPd^{II}I, which promotes the nucleophilic addition of the alcohol in activating the allenyl moiety. This cycloetherification leads to the intermediate corresponding to **1G** (Scheme 1), its subsequent protonolysis liberating the organic compound and PhPd^{II}I. The synthesis of 5,6-dimethyl-2*H*-pyran-4-yl *p*-methoxybenzoate (eqn (50)) arises from the unusual alkoxylation of the central atom of the allene, followed by protonolysis and dehydration (Scheme 12).^{44,45}



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10 4.2. Oxyarylation

4.2.1. From allenols. In 1993, Walkup's team disclosed the synthesis of 2-substituted-5-(1-arylvinyl)-tetrahydrofurans from 6-substituted-1,2-dien-6-ols and aryl halides (eqn (51)).⁴⁶ The authors proposed that the oxypalladation precedes the formation of the C-Ar bond (as in Scheme 1, path *a*), which contrasts with the proposal from Tsuji⁴⁷ and Cazes.^{48,49}



²⁵ The oxypalladation of 3-substituted-1,2-dien-5-ols with phenyl iodide affords 5-benzyl-2,3-dihydrofurans (eqn (52)) in yields depending on the nature of both the base and the solvent.⁵⁰ The cyclisation product was not obtained in the

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$$R^{4}$$
 R^{3} $Pd(PPh_{3})_{4}$ (0.05 equiv.)
 R^{2} PhI $S_{2}CO_{3}$ (4 equiv.)
 $DMF, 70 \ ^{\circ}C$ Ph R^{2} R^{1} R^{1}
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 $R^{1} = H, R^{2} = Me, R^{3} = n-Bu, R^{4} = H: 67\%$
 $R^{1} = R^{2} = H, R^{3} = n-Bu, R^{4} = H: 75\%$
 $R^{1} = R^{2} = H, R^{3} = n-Bu, R^{4} = H: 61\%$
 $R^{1} = R^{2} = Me, R^{3} = n-Bu, R^{4} = H: 61\%$
 $R^{1} = R^{2} = Me, R^{3} = n-Bu, R^{4} = H: 43\%$
 $R^{1} = R^{2} = R^{3} = H, R^{4} = n-Bu: 0\%$
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(52)

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absence of a substituent in C-3 (eqn (52)). The course of these reactions greatly differs from that of the reactions depicted in eqn (51), since the C-Ar bond is formed with the terminal instead of the central carbon of the allene. Ma and Gao proposed the intermediates shown in Scheme 13, the C-O bond formation occurring *via* the 5-*exo-dig* cyclisation of the complex **13A**, and the C-Ar *via* the reductive elimination of Pd⁰ from **13B**.

Recently, Ma and Xie reported the asymmetric couplingoxycyclisation of γ -allenols with aryliodides with ee up to 92% (eqn (53)).⁵¹ The proposed mechanism implies the arylpalladation of the allenyl group, and the nucleophilic attack of the hydroxyl to the resulting chiral η^3 -allylpalladium intermediate.

The unusual regioselectivity of the oxycyclisation–arylation of the above β -hydroxyallenes (eqn (52)) does not seem to arise from the number of carbons between the allene and the hydroxyl groups, because the oxycyclisation–arylation of γ hydroxyallenes (eqn (51)) and α -hydroxyallenes occurs with the same regioselectivity, giving the corresponding (1arylvinyl)oxiranes under similar conditions (eqn (54)).⁵² Instead of DMF, which, according to Ma's reports,^{50,52} would be the optimum organic solvent, Ihara and co-workers used water with a water-soluble ligand and sodium dodecyl sulphate (SDS) (eqn (55)).⁵³

(1-Arylvinyl)oxiranes have also been obtained from α - 25 hydroxyallenes using hypervalent iodonium salts such as diphenyliodonium tetrafluoroborate, instead of aryl iodides (eqn (56)).⁵⁴ With such arylating species, Kang's team has also synthesised 2-(1-arylvinyl)-tetrahydrofuran(pyran)s from 1,2dien-6(or 7)-ols (eqn (57)).⁵⁵ Both catalytic cycles, *i.e.* addition 30 of the hydroxy group to a η^3 -allylpalladium intermediate or to the Pd-activated allenyl moiety, have been considered as plausible by the authors.⁵⁵

The Alcaide–Almendros team observed the formation of both five- and three-membered rings from the reaction of phenyl iodide with 3-hydroxy-3-(1-methyl-1,2-propadienyl)-2indolinone (eqn (49))⁴³ or 3-hydroxy-3-(1-methyl-1,2propadienyl)-2-azetidinone (eqn (58)).⁵⁶ The absence of phenylation of the five-membered ring obtained from the hydroxylindolinone (eqn (49)) is not discussed by the authors. They nevertheless suggested that the cleavage of the β - or γ -lactam



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1 nucleus concomitant with the formation of the oxirane is probably due to the ring strain of the spirocyclic systems.

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$$Pd(PPh_3)_4 (0.05 \text{ equiv.})$$

 R^2
 HO
 $R^1 + R^2 I$
 HO
 $R^2 = Ph (71\%, dr = 96:4), p-MeOC_6H_4 (71\%, dr = 97:3),$
 $E-PhCH=CH (65\%, dr = 92:8)$
 $R^1 = n-6t., R^2 = Ph (78\%, dr = 98:2), p-BrC_6H_4 (46\%, dr = 99:1),$
 $E-1-hexenyl (59\%, dr > 99:1)$

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$$Pd(OAc)_{2} (0.1 \text{ equiv.})$$

$$PhP(C_{6}H_{4}\rho \cdot SO_{3}K)_{2} (0.2 \text{ equiv.})$$

$$SDS (2 \text{ equiv.})$$

$$R^{1} + Arl \xrightarrow{i \cdot Pr_{2}NH (2 \text{ equiv.})}{H_{2}O, 80 \ ^{\circ}C, 2.5 \cdot 96 \ h} \xrightarrow{R^{2}}{R^{2}}$$

$$Ar = Ph \begin{cases} R^{1} = Me, R^{2} = Ph (70\%); R^{1} = R^{2} = C_{3}H_{7} (74\%); \\ R^{1} = R^{2} = C_{5}H_{11} (54\%); R^{1} = R^{2} = Ph (72\%) \end{cases}$$

$$R^{1} \cdot R^{2} = cyclohexyl \begin{cases} Ar = Ph (83\%), p \cdot MeC_{6}H_{4} (33\%), p \cdot AcC_{6}H_{4} (69\%), \\ p \cdot NO_{2}C_{6}H_{4} (86\%), p \cdot MeOC_{6}H_{4} (32\%) \end{cases}$$

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In contrast to the results depicted in eqn (49) and (58) obtained with phenyl iodide, the arylation of these substrates with arylboronic acids proceeds selectively (eqn (59) and (60)).⁵⁷ According to the authors, the Suzuki-type reaction occurs after the heterocyclisation.

The fully intramolecular oxyarylation shown in eqn (61) has been disclosed by Grigg and co-workers, who proposed a reaction occurring *via* the 6-*exo-dig* carbocyclisation of the ArPdI species onto the allenyl unit, followed by the interception of the resulting η^3 -allylpalladium complex by the alcoholate.⁵⁸

4.2.2. From allenyl alkyl ketones. The selectivity of the phenylation of heptadeca-3,4-dien-2-one highly depends on the reaction conditions. In toluene containing triethylamine, $Pd(PPh_3)_4$ catalysis leads to an almost equimolecular mixture of 2-dodecyl-5-methyl-3-phenylfuran and 2-dodecyl-5-methylfuran



(54)

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(eqn (62)).^{59,60} Co-catalysis with Ag_2CO_3 greatly increased the selectivity towards the cross-coupling adduct (eqn (62)). Thus,

this Pd(PPh₃)₄/Ag₂CO₃ catalytic system has been used for the synthesis of various trisubstituted (hetero)arylfurans (eqn (63)). In contrast, tetrasubstituted arylfurans were better obtained using catalytic amounts of Pd(PPh₃)₄ and *n*-Bu₄NBr in DMA containing K₂CO₃ (eqn (64)).⁶⁰ From control reactions, Ma's team retains two possible catalytic cycles for the synthesis of the cross-coupling compounds (Scheme 14);⁶⁰ both involve the coordination of ArPdX to the allene to generate **14A** followed by







the insertion reaction leading to the η^3 -allypalladium complex 10 14B. Equilibrium of the latter with the palladium dienolate 14C could promote the 5-endo-trig heterocyclisation leading to 14D (path *a*), subsequent β -H elimination liberating the arylfuran. The other possibility is the direct intramolecular nucleophilic attack on 14B by the carbonyl oxygen atom to afford 14E and

 Pd^{0} (path b). Base-mediated deprotonation of 14E would yield 15 the arylfuran. As for the non-arylated product, it appears from a cycloisomerisation reaction. Reports from the teams of Marshall⁶¹ and Hashmi^{62,63} led us to suggest two catalytic pathways diverging from 14F for its formation. This mutual intermediate would arise from the oxygen attack on the η^2 -palladium 2.0 complex 14A. β-H elimination from 14F would generate the Pd^{IV} intermediate 14G (path c), subsequent reductive elimination giving ArPdX and the furan. This compound could also be obtained from 14H, the latter resulting from 14F via a 1,2-H-25 shift (path d).

4.3. Oxyvinylation

4.3.1. From allenols. The intermolecular reaction of vinylic halides with α - and γ -allenyl alcohols has led to the corres-30 ponding epoxides $(eqn (54))^{52}$ and tetrahydrofurans $(eqn (51))^{46}$ substituted by a 1,3-dienic tether, the oxycyclisation-crosscoupling of an optically active *α*-allenyl alcohol occurring without loss of chirality (eqn (65)).⁵²





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The Ohno-Fujii team disclosed the intramolecular alkoxyvinylation of a substrate containing three reactive species, that is the allene, the vinylic bromide and the alcohol (eqn (66)).⁶⁴ The catalytic cycle would entail the addition of the vinylic palladium bromide to the central allene carbon, followed by intramolecular alcoholate attack of the resulting η^3 -allylpalladium complex.64



In the presence of lithium bromide and potassium carbonate, the Pd^{II}-catalysed reaction of acetylene-linked allenyl alcohols can afford bridged tricyclic β-lactams bearing a vinylbromide substituent as shown in eqn (67).⁶⁵ Alcaide and coworkers proposed that this domino reaction implies the addition of bromide to the allenepalladium complex 15A to give the η^3 -allylpalladium complex **15B**, which suffers the intramolecular alkoxy addition leading to 15C (Scheme 15). Insertion of Pd⁰ into the C-Br bond of this five-membered intermediate is followed by a Heck-type addition to the $C \equiv C$ bond giving 15D. According to the authors, the product is obtained from the trapping of 15D by the bromide anion (path *a*).⁶⁵ In fact, **15D** could liberate the isolated compound through the reductive elimination of Pd^{0} (path b). We also suspect that the exo-dig cyclisation could lead to 15E rather than to 15D, its $S_N 2$ by the bromide anion giving the product and Pd⁰ (path *c*).



The vinylation can arise from the reaction with methyl acrylate as reported by the Alcaide-Almendros team in 2005 55 (eqn (68) and (69)).⁵⁷ These syntheses would proceed through

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15 successive Wacker-type and Heck-type reactions and regeneration with Cu(OAc)₂/O₂ of the catalyst required for the heterocyclisation. In fact, such an intramolecular oxypalladation followed by addition to methyl acrylate was reported, as early as 1992 by Gallagher *et al.*, using hepta-5,6-dien-1-ol as the substrate but in a low yield and with a stoichiometric amount of PdCl₂.⁶⁶ Using the same substrate but under catalytic oxidative conditions, Gallagher *et al.* obtained a dimerisation reaction (eqn (70)), which could be considered as the addition of a vinylic system to the oxypalladation intermediate.⁶⁶

$$HO \xrightarrow{PdCl_2 (0.1 equiv.)}_{HO} \xrightarrow{CuCl_2 (3 equiv.)}_{MeOH, 65 °C, 0.5 h} \xrightarrow{O}_{30\%} (70)$$

4.3.2. From allenyl alkyl ketones. The cross-couplingcyclisation reaction of nona-3,4-dien-2-one with methyl (*Z*)-3iodoacrylate (eqn (71)) efficiently arises with the Pd(PPh₃)₄/ Ag₂CO₃ catalytic system already used for the arylationcyclisation of this substrate (eqn (63)).^{59,60} As above (eqn (64)), a modified procedure was favoured when both α - and γ -positions of the allenyl ketone were substituted (eqn (72)).⁶⁰



A vinylic substitution in C-4 of furans can arise from the Pdcatalyzed dimerisation of terminal allenyl ketones disclosed by 55 Hashmi's team. In most cases, these reactions lead to minor amounts of the cycloisomerisation compound (eqn (73)).^{63,67,68}



Nevertheless, the selectivity can greatly depend on the nature of the catalyst, the solvent and the carbonyl substituent as exemplified in eqn (74).⁶⁹ Complete inhibition of the dimerisation



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reaction has been observed from a γ-substituted allenyl ketone⁶²
or aldehyde⁶⁹ (eqn (75)). From the investigation of various plausible reaction pathways, the authors prioritized the catalytic cycle
depicted in Scheme 16.⁶³ The oxypalladation of the allene moiety provides 16A, which could have the structure of either a σ-Pd

30 $\begin{array}{c} R^{2} \\ R^{1} \\ R^{1} \\ \end{array} \xrightarrow{Pd^{||}(cat.)} \\ R^{2} \\ R^{2} \\ C \\ R^{1} \\ R^{2} \\ C \\ R^{1} \\ (75) \\ \end{array}$

 $R^1 = R^2 = Me \text{ or } R^1 = H, R^2 = n-hex$

complex or a carbene palladium complex. This intermediate could lead to the hydridopalladium^{IV} complex 16B via a β-H elimination, or to the 2-alkylfuran via a 1,2-hydrogen-shift. The 2-alkylfuran could also be produced by reductive elimination from 16B. The latter could also carbonylate the terminal C=C bond of another molecule of the substrate to afford another hydridopalladium^{IV} complex (16C), reductive elimination of Pd^{II} giving the dimeric



product. The procedure has been, subsequently, used for the synthesis of macrocycles from diallenyldiketones (eqn (76)).⁷⁰

4.3.3. From monoesters of allenyl phosphonic acids. Ma and co-workers reported the Pd^{II} -catalysed oxycyclisation-Heck reaction of ethyl allenylphosphonic acids with regeneration of the active catalytic species using either benzoquinone or a NaI/ O_2/CaH_2 association (eqn (77)).⁷¹

4.4. Oxyallylation

4.4.1. From allenols. To the best of our knowledge, Ma and Gao were the first to disclose the synthesis of 4-(2'-alkenyl)-2,5dihydrofurans and 3-propenyl-5,6-dihydropyrans from the Pd^{II}catalysed cyclisative coupling reaction of α - and β hydroxyallenes with allylic halides (eqn (78) and (79)).^{72,73} These domino reactions would proceed as underlined in Scheme 1, path *b*, and preserve the chirality of the substrate (eqn (80)).⁷⁴ With allyl bromide, the cyclisation-cross-coupling of 2-methylpenta-3,4-dien-1-ol was however not observed (eqn (78)), whereas that of 1-(propa-1,2-dienyl)cyclohexanol afforded 3-(1-cyclohexylidenemethylvinyl)-1-oxaspiro[4.5]dec-3ene as the major product (eqn (81)), the palladium intermediate **17A** reacting with the substrate to give **17B** (Scheme 17, path *a*)



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(77)

rather than with allyl bromide (Scheme 17, path b).⁷³ Subsequent β -OH elimination leads to the trienic system. According to the 30 authors, this bimolecular cyclisative coupling is caused by the absence of a substituent on the allenyl moiety, hence its high reactivity. The effect of the substitution was also observed with octa-1,2-dien-4-ol, which gave the allylated heterocycle in only 4% yield (eqn (78)). It has been proposed that the substitution of the 1,2-

35 dienyl unit results in steric hindrance favouring the reaction with allyl bromide at the expense of that with another molecule of allenol. This seems to agree with the results depicted in eqn (79): the Heck-type reaction with 3-chlorobut-1-ene occurs with a fair





The Alcaide-Almendros team has generously exploited the above oxycyclisation-Heck addition procedure using, as substrates, lactams substituted by an allenyl alcohol (eqn (83)),⁷⁶ or an allenyl unit and either a hydroxy (eqn (84))⁵⁶ (85)⁷⁷⁻⁷⁹ and $(86)^{43}$) or a hydroxylated tether (eqn $(87)^{44}$).



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 $\frac{Br}{PdCl_2 (0.05 \text{ equiv.})} + (81)$









(eqn (89)).⁴⁵ From the comparison of the reactivity of a γ allenol⁸⁰ and different allendiols,⁴⁵ we observed that the favoured Pd^{II}-catalysed oxycyclisations with one extremity of the allenyl moiety would be 5-*exo* rather than 6-*exo* or 7(or 8)*endo* (eqn (90)), 6-*endo* rather than 4(or 5)-*exo* or 7-*endo* (eqn (91)), and 7-*endo* rather than 5(or 6)-*exo* or 8-*endo* (eqn (88) and (89)). It is however required to point out that this classification is made from substrates without substitution of the terminal carbon and with disubstitution of the other extremity of the allenyl unit.





The Spanish team has also used Baylis–Hilman acetates instead of allyl halides to obtain spirocyclic lactams (eqn (92) 55 and (93)).⁸¹

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4.4.2. From allenyl alkyl ketones. Experiments under various conditions led Ma and Li to finalize the selective couplingcyclisation reaction of 1-substituted 1,2-allenyl ketones with allylic bromides (eqn (94)).⁸² The proposed procedure was less selective with a 3-substituted-1,2-dienyl ketone. Using a large excess of allyl bromide in DMF instead of MeCN, the expected product was nevertheless obtained in a fair yield (eqn (95)). These reactions, performed with a Pd^{II} catalyst, could happen as shown in Scheme 18, path *a*. Oxypalladation, mediated by

coordination of PdX₂ to the allene, would afford 18A, elimination of HX followed by a Heck-type reaction giving the product.^{60,82} Having also observed the reaction under Pd₂(dba)₃. CHCl₃ catalysis,⁸³ the authors proposed another plausible catalytic cycle (path *b*), which involves the activation of the allene by the η³-allylpalladium 18B. This would allow the oxypalladation reaction leading, after removal of HBr, to 18C. This complex would undergo a reductive elimination to afford the product.

4.5. Oxydienylation

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The cycloetherification-dienylation of allenols has been documented above with the bimolecular cyclisative coupling of 3-(1cyclohexylidenemethylvinyl)-1-oxaspiro[4.5]dec-3-ene reported

- ¹⁵ by Ma and Gao (eqn (81)). In the presence of 4-bromobuta-1,2-diene, the dienylation mainly occurs with this additive (eqn (96)).⁷³ This reaction occurs following a mechanism similar to the one depicted in Scheme 17, path *a*, the elimination involving the bromide atom instead of the hydroxyl group.
- ²⁰ Ma and co-workers observed that the homodimeric coupling-cyclisation of 2,3-allenols easily occurs, even with a bulky substituent in C-2 (eqn (97)), and without racemisation of optically active substrates when PdCl₂ is associated to NaI as catalyst.⁸⁴ The authors proposed the *in situ* formation of PdI₂ to
- 25 mediate the reaction. The PdCl₂/NaI catalytic system has then been used for the synthesis of 2,5-dihydrofuran-fused bicyclic skeletons from bisallenols, the addition of NaI being however not required when the allenol moieties are connected by a CH₂NTsCH₂ tether (eqn (98)).⁸⁵ Such a reaction also occurs 30 when one hydroxyl group is protected as the acetate
- (eqn (99)).⁸⁵ This Chinese team used PdI_2 with a stoichiometric amount of $BF_3 \cdot Et_2O$ in DMSO for the coupling-cyclisation of two different 2,3-allenols (eqn (100)).⁸⁶



40 $R^1 = H, R^2 - R^3 = (CH_2)_5$: 86%; $R^1 = H, R^2 = R^3 = n$ -Bu: 92% $R^1 = H, R^2 = R^3 = Me$: 66%; $R^1 = n$ -Bu, $R^2 = R^3 = H$: 84% $R^1 = n$ -Bu, $R^2 = Me, R^3 = H$: 83%; $R^1 = Ph, R^2 = n$ -Bu, $R^3 = H$: 75% $R^1 = CO_2Me, R^2 = Et, R^3 = H$: 58%; $R^1 = allyl, R^2 = Me, R^3 = H$: 61%





As for the Alcaide–Almendros team, it obtained various functionalized buta-1,3-dienyl dihydrofurans from the intramolecular oxypalladation of α -allenols followed by coupling with α -allenic esters (eqn (101) and (102)), the efficiency depending on the nature of both the Pd^{II} catalyst and the solvent (eqn (102)).^{87,88}





Use of β , γ -allenols instead of α -allenols provided access to functionalized buta-1,3-dienyl dihydropyrans through a chemoregioselective heterocyclisation-coupling and reaction (eqn (103)).89 This domino process is sensitive to the steric properties of the allendiol, a complex mixture being obtained when the C3 position bears a phenyl substituent (eqn (103)). Surprisingly, a trisubstituted dihydrofuran was isolated from the reaction of 3-methoxy-4-phenylhexa-4,5-diene-1,2-diol with 2-methyl-1-phenylbuta-2,3-dienyl acetate (eqn (104)).⁸⁹ The mechanism proposed by the Alcaide-Almendros team involves the intramolecular oxypalladation of the α -methoxy allenyl ether (Scheme 19). The cross-coupling reaction of the resulting palladium intermediate **19A** with the α -allenic acetate provides 19B. The subsequent deacetoxypalladation⁹⁰ leads to 19C, 30

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(104)

15 hydrolysis of the latter giving the isolated dihydrofuran.



35 4.6. Oxycyclisation-Sonogashira reaction

The palladium intermediate arising from the oxypalladation of α -allenols can participate in a Sonogashira reaction (eqn (105) and (106)).⁵⁷



4.7. Oxy-alkoxycarbonylation

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4.7.1. From a hydroxyallene. In 1987, Walkup and Park disclosed the Pd^{II}-catalysed synthesis of methyl 2-(tetrahydrofuran-2-yl)acrylates from γ -allenols, carbon monoxide and methanol (eqn (107)).^{91–93} The authors considered that the cyclisation precedes the carbonylation reaction. The procedure has been used to prepare methyl 2-(tetrahydropyran-2-yl)acrylates (eqn (108))^{66,94} and bicyclic heterocycles (eqn (109)).⁹⁵









R = Me (60%), CH₂CO*t*-Bu (90%), CH₂COMe (68%), CH₂CH(OH)Me (44%) (110)

4.7.2. From a γ -silyloxy allene. The above protocol led also 55 to methyl 2-(tetrahydrofuran-2-yl)acrylates from γ -silyloxy

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allenes (eqn (110)).^{91,93} Apparently, the experimental condi-1 tions mediate the cleavage of the O-Si bond before the interaction with the allenyl group.93

4.7.3. From a γ -oxoallene. The treatment of 4,5-dienals 5 under the experimental conditions of eqn (110) led to complex mixtures, but addition of propylene oxide and triethyl orthoacetate as acid and water traps, respectively, provided the corresponding methyl 2-(5-methoxytetrahydrofuran-2-yl)acrylates in good crude yields. Unfortunately, these furanosides are unstable under chromatographic conditions, hence the relatively low yields 10

of pure products (eqn (111)).^{96,97} The same procedure is inefficient from β -allenylketones (eqn (111)).⁹⁷



4.8. Oxyaroylation

The cyclisation-arylation of γ -hydroxyallenes has been documented 25 in Section 4.2.1 by Walkup's team. The authors observed the same reaction under a CO atmosphere at 70-80 °C, whereas the selective cyclisation-aroylation occurred at 55-60 °C (eqn (112)).46,98,99



40 4.9. Oxybromination

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In 2000, Bäckvall and co-workers disclosed two Pd^{II}-catalysed Q7 procedures for the cyclisation-bromination of primary γ - and δ -allenic alcohols (eqn (113)).¹⁰⁰ These reactions would imply the addition of bromide to the activated allenyl group to give 45



Procedure A: Pd(OAc)₂ (0.05 equiv.), benzoquinone (2.5 equiv.), AcOH, 40 °C, 8 h 50 $n = 1, R^1 = R^2 = Me (70\%)$

n = 1, R² = H, R¹ = *n*-C₅H₁₁ (75%), *i*-Pr (70%), *n*-BuCHEt (62%) Procedure B: Pd(OAc)₂ (0.1 equiv.), Cu(OAc)₂·H₂O (2.1 equiv.), K₂CO₃ (1.2 equiv.), O2 (1 atm), MeCN, rt, 2.5-4 h

n = 1,
$$R^1 = R^2 = Me(71\%)$$
; n = 1, $R^2 = H$, $R^1 = n-C_5H_{11}(46\%)$
n = 2, $R^2 = H$, $R^1 = n-C_5H_{11}(78\%)$, *i*-Pr(70%)

HO

the corresponding η^3 -(2-bromoallyl)palladium complex. In the presence of benzoquinone, the authors assumed that the quinone acts as a ligand,¹⁰¹ leading to **20A** (Scheme 20). Then, regioselective 5-exo-cycloetherification affords the cyclisation product and the Pd⁰-benzoquinone complex, which reacts with AcOH to regenerate the Pd^{II} catalyst.

The Alcaide-Almendros team has subsequently carried out the synthesis of varied bromoheterocycles from primary, secondary and tertiary allenic alcohols (eqn (114)^{43,57} (115),⁴⁴ $(116)^{44,45}$ and $(117)^{45,80}$) through the regio- and, often, chemioselective 5-, 6- 7- or 8-endo-cycloetherification. The reactions depicted in eqn (116) and (118)^{44,45} show that the regioselectivity can, however, depend on the substitution of the allenic group. According to the authors, this difference could result from "the electron-withdrawing capacities of the phenyl substituent relative to the electron-donating methyl group". The strengthening of the electrophilicity of the benzylic carbon atom by the phenyl substituent would favor the 5-exo cyclization of the primary hydroxy group over the 6-endo cyclization of the secondary hydroxy group.45





Z¹ = Me (55%, dr = 100:0), p-MeOC₆H₄CO (57%, dr = 60:40)

OZ1 (118)

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4.10. Oxyalkoxylation

Bromoallenes, substituted by a heteroatomic or substituted carbon tether terminated by a hydroxyl, have been used in an 25 alcohol under basic conditions and Pd⁰ catalysis to obtain sevenand eight-membered unsaturated rings possessing one or two heteroatoms and an alkoxy or methylalkoxy substituent (eqn (119) to (121)).^{30,33,34} According to Tanaka and co-workers, these reactions would occur via the intramolecular hydroxyl 30 addition to a η^1 -allenylpalladium complex followed by the intermolecular addition of the alcohol to the resulting η^3 -allylpalladium complex. The selectivity however greatly depends on the substitution of the tether as shown with the reaction of 7-bromohepta-5,6-dien-1-ol, which mainly affords the 35 six-membered ring (eqn (122)).³⁰ This tetrahydropyran would be obtained via intermolecular alcoholate addition to the





 η^1 -allenylpalladium complex and subsequent cycloetherification proceeding at the level of the η^3 -allylpalladium complex.³⁰

When the tether possesses two oxygen functionalities susceptible to undergo cycloetherification leading to five- and seven-membered rings, a mixture of these two heterocycles and a bicyclic compound resulting from two intermolecular etherifications has been obtained (eqn (123)).³³ The biscyclisation was the main domino reaction from a bromoallene bearing a tether possessing both hydroxy and benzamide 30 groups as depicted in eqn (124).^{102,103}



The Pd⁰-catalysed reaction of 3,3-dimethylhexa-4,5-dienal in MeOH led to 3-(1-(5-methoxy-3,3-dimethyl-tetrahydrofuran-2yl)vinyl)-5,5-dimethylcyclohex-3-enol as the main product (eqn (125)).¹⁰⁴ According to Tsukamoto *et al.*, the reaction implies the alkenylpalladium complex **21A** formed *via* an anti-Wacker-type addition (Scheme 21). This intermediate would undergo the carbopalladation of the allene to afford the η^3 -allylpalladium species **21B**. Reaction of the latter with MeOH would produce the three isolated products.

5. Intermolecular acyloxylation

5.1. Hydroacyloxylation

In 1967, Shier reported the Pd-catalysed reaction of allene in AcOH containing AcONa.¹⁰⁵ A small amount of allyl acetate was isolated, the favoured pathway being the formation of dimer derivatives, mainly 3-methyl-2-methylenebut-3-enyl acetate

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25 (eqn (126)). Thirty years later, Yamamoto and Al-Masum disclosed an effective procedure to selectively obtain allyl esters from arylallenes or phenylthioallene and carboxylic acids (eqn (127)).¹⁰⁶ These hydroacyloxylations would proceed by oxidative insertion of Pd⁰ into the RCO₂-H bond to afford the hydridopalladium species 22A (Scheme 22). Addition of 22A to 30 the allene leads to the η^3 -allylpalladium complex 22B, reductive elimination of Pd⁰ from the latter liberating the allylcarboxylate.¹⁰⁶ Such a mechanism agrees with the formation of buta-

1,3-dienylbenzene from 1-phenyl-3-methylallene, the β -H elimination 35

0

being the favoured reaction pathway from the η^3 -allylpalladium 25 intermediate (eqn (128)).¹⁰⁶







R¹ = H, R² = Me: 69%, 63% ee; R¹ = Br, R² = H: 67%, 48% ee

R¹CO₂H

R¹CO₂PdH

22A

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R³



R¹ = (*p*-MeC₆H₄SO₂)NBn, R² = H (63, 1:0), *p*-Me (64, 1:0), *p*-OMe (68, 1:0)

35 5.2. Arylaroyloxylation

In 1984, Larock and co-workers disclosed the formation of isocoumarins from the Pd^{II} -mediated coupling of *o*-thallated



benzoic acids with allenes followed by a basic treatment.¹⁰⁷ As 35 organothallium compounds are toxic, the subsequent synthesis of these heterocycles using allenes, *o*-iodo benzoic acids and Pd^{0} catalysis was of interest. Moreover, optically active compounds were obtained in the presence of chiral ligands (eqn (129)).^{19,20} More recently, Kumara Swamy and co-workers 40 carried out such domino reactions using phenylallenes, allenyl phosphonates, acetates or esters (eqn (130) to (133)).^{22,27,108}

Savic's team has synthesised various arylated allylic acetates from the reaction of both aryl iodides and sodium acetate with buta-2,3-dienyl benzyl ether or *N*-benzyl-*N*-(buta-2,3-dienyl)-4toluenesulfonamide (eqn (134)).^{109,110} This team has also carried out the synthesis of heterocyclic systems using aryl iodides *o*-substituted with an allenic hetero-tether (eqn (135)).¹¹⁰







5.3. Vinylalkenoyloxylation

45 Allenes and (Z)- β -iodo- α , β -unsaturated acids have been used, under Pd⁰ catalysis, to obtain various 5-methylene-5,6dihydropyran-2-one derivatives (eqn (136), (137)²⁴ and (138)^{19,20}). The vinylalkenoyloxylation has also been reported from a three-component reaction (eqn (139)).¹¹⁰

5.4. Carboacyloxylation

Bäckvall and Deng synthesised acyloxylated vinylallenes 40 from allenynes and acetic acid (eqn (140) and (141)).⁷⁵ The coordination of the substrate to Pd(OAc)₂ would lead to the chelate 24A (Scheme 23). A rearrangement of the latter arising through the cleavage of a propargylic C-H bond would afford 24B. Another possibility is the formation of the Pd^{IV} intermediate 24C. Both 24B and 24C could evolve towards the η^3 -allylpalladium complex 24D. Intra- or intermolecular attack of the allyl moiety by AcO⁻ would yield the isolated compound, the regeneration of the catalyst being assumed by benzoquinone and AcOH. 50

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n-Bu

Pd(OAc)₂ (0.05 equiv.) benzoquinone (1.2 equiv.) C

AcOH, 60 °C, 5 h

The carboacyloxylation can also occur in the course of the



6. Intramolecular acyloxylation or alkenoyloxylation

6.1. Hydrolactonisation

Duchêne, Parrain and co-workers obtained pyranones from allenyl stannanes, β -iodo vinylic acids and catalytic amounts of Pd⁰ (eqn (144)).¹¹⁴ In fact, these syntheses involve two reactions in one pot. Firstly, the Stille reaction of the allenyltin with the vinylpalladium complex **26A** affords the allenic acid **26B** and Pd⁰ (Scheme 25). Then, the intramolecular reaction of

dimerisation of allene as depicted in eqn (126)¹⁰⁵ and (142).¹¹¹ The 3,5-dimethylpyranone is obtained from the coupling of two molecules of allene with one molecule of carbon dioxide, while one more molecule of allene is implied in the formation of 3methyl-2-methylenebut-3-enyl methacrylate and 3-methyl-2methylenebut-3-enyl but-2-enoate (eqn (142)). The synthesis of these two esters requires two supplementary hydrogens per molecule through an indeterminate mechanism. Six-

n-Bu

OAc

(141)

52%

membered lactones can be obtained in better yields using
methoxyallene¹¹² or phenylallene (eqn (143)).¹¹³ The Choi/ Sakakura team has recently proposed two likely pathways for formation of 3,5-dibenzylpyranone mediated by the (η³allyl)PdCp/PMe₃ association (Scheme 24).¹¹³ The Pd⁰ complex
would react with either another molecule of allene or
carbon dioxide to afford the five-membered heterocycles 25B or 25C, respectively. The reaction of either allene or carbon dioxide with these species would lead to the seven-membered palladalactone 25D, which would evolve towards the disubsti-

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tuted pyranone.

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0Z

Scheme 26

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the palladium carboxylate **26C** provides the hydrido- $(\eta^3$ -allyl)palladium complex **26D**. Reductive elimination of Pd⁰ from the latter yields the pyranone.

Pd(OAc)₂ (0.05 equiv.)
PPh₃ (0.05 equiv.)
PPh₃ (0.05 equiv.)

$$R^{1} = R^{2} = R^{3} = H: 83\%; R^{1} = Me, R^{2} = R^{3} = H: 85\%$$

R² = H, R¹ = R³ = Me: 85%; R¹ = CH₂OMe, R² = OMe, R³ = H: 82%
(144)

When Ma and co-workers tested the asymmetric lactonisationarylation of 2-methyl-4-phenyl-2,3-butadienoic acid, the expected arylation did not occur: 4-phenyl-2-methyl-2-butenolide was the only product (eqn (145)).¹¹⁵ Given the results below reported in eqn (149),¹¹⁵ we suspect that it is due to the 2-methyl substituent, which would hamper the approach of the allene moiety by the arylating species. Subsequently, Ma's team observed the hydrolactonisation of 2-propyl-4-phenyl-2,3-butadienoic acid under different experimental conditions (eqn (146)).¹¹⁶



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6.2. Lactonisation-arylation

In 1993, Walkup and co-workers disclosed the Pd⁰-catalysed synthesis of 5-(1-arylvinyl)-dihydrofuran-2(3*H*)-ones from 4,5-hexadienoic acid and aryl halides (eqn (147)).⁴⁶ The authors suspected that the cyclisation precedes the arylation. For the lactonisation-arylation of α -allenic acids (eqn (148)), Ma and



Shi reported, in 1998, the increase of the yields by co-catalysis with silver carbonate.^{117,118} They proposed a catalytic cycle involving firstly the arylation when the reaction was performed in the absence of Ag_2CO_3 (as in Scheme 1, paths *c* and *d*), whereas, under Pd/Ag catalysis, they suspected a Ag-mediated cyclisation followed by *trans*-metalation with ArPdX of the resulting 3-silver-2-butenolide intermediate (Scheme 26).



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The use of the Pd/Ag procedure in the presence of chiral ligands led to 3-aryl-2-butenolides with low ee's. This would be due to the Ag-mediated cyclisation (Scheme 26), which would occur without participation of the chiral ligand, hence the enantioselective synthesis of such compounds using a silverfree method with, after some investigation, the chiral ligand (eqn (149))¹¹⁵ already used by Larock (see eqn (129) and (138)). The success of these domino reactions highly depends on the substituent of the allene group. Indeed, a low ee was observed with 4-cyclohexylbuta-2,3-dienoic acid as the substrate (eqn (149)), whereas the arylation of 2-methyl-4-phenyl-2,3-butadienoic acid did not occur (eqn (145)).







Subsequently, Ma and Shi carried out these syntheses using chiral acids (eqn (150)) or their salts (eqn (151) and (152)).¹¹⁹ According to the authors, the dependence of the enantioselectivities on the nature, acid or acid salt, of the substrate is due to a mechanism switch: formation firstly of the Ar–C bond from the acid leading to some loss of the chirality at the level of the η^3 -allylpalladium complex **28A**, and of the O–C bond from the acid salt (Scheme 27).



Bi- and tricyclic lactones have been synthesized using the Walkup⁴⁶-Gallagher¹²⁰ experimental conditions (eqn (153) and (154)).⁹⁵



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The Pd⁰-catalysed lactonisation-arylation has also been carried out using either α -allenic acids and aryl iodides anchored to Merrifield resin to afford polymer-supported butenolides,¹²¹ or a γ (or δ)-allenic acid and diphenyliodonium tetrafluoroborate to provide the corresponding lactone (eqn (155)).⁵⁵

 $Pd(PPh_{3})_{4} (0.05 \text{ equiv.}) \xrightarrow{Ph}_{35}$ $HO + \frac{Ph_{2}IBF_{4}}{(1.2 \text{ equiv.})} \xrightarrow{K_{2}CO_{3} (2.5 \text{ equiv.})}{MeCN, 60 °C, 3 h} \xrightarrow{O}_{n} = 1 (61\%), 2 (65\%) \xrightarrow{O}_{155}$

Cazes and co-workers isolated the substituted γ -lactones shown in eqn (156) from the Pd⁰-catalysed reaction of diethyl 2-(2,5-dimethylhexa-3,4-dien-2-yl)malonate with phenyl iodide under basic conditions.¹²² According to the authors, the anionic η^3 -allylpalladium species **29A** obtained from the addition of the arylating species to the substrate evolves into the sixmembered palladacycle **29B** (Scheme 28). The transformation

of the latter to **29C** will be mediated by the iodide anion, subsequent hydrolysis leading to the substituted lactone.

6.3. Lactonisation-alkenylation

- 50 The Walkup procedure is also effective for the cyclisationvinylation of 4,5-hexadienoic acid (eqn (147)).⁴⁶ The synthesis, under Cazes conditions, of substituted γ -lactones from allenyl malonates led to higher yields when using vinylic bromides (eqn (157)) instead of phenyl iodide (eqn (156)).¹²²
- 55 Ma and Yu obtained butenolides β -substituted with a terminal-unsaturated tether having up to 16 carbons from the

reaction of 2,3-allenoic acids with 1-alkenyl bromides (eqn (158)).¹²³ An experiment using a deuterated homoallylic bromide demonstrated a reaction occurring through successive oxycyclisation, Heck reaction, hydropalladation and debromopalladation (Scheme 29). It is remarkable that the dehydropalladation–hydropalladation sequence can travel through a chain of 14 carbon atoms.

$$Ph \xrightarrow{n-Pr}_{HO} + \underbrace{r}_{(5-10 \text{ equiv.})} \xrightarrow{Pd(OAc)_2 (0.05 \text{ equiv.})}_{DMF, 90-100 \ ^\circ\text{C}, \ 1-21 \ h} \xrightarrow{Ph}_{Ph} \xrightarrow{n-Pr}_{O} \xrightarrow{n}_{O} \xrightarrow{n}$$

Ma and co-workers have also reported the synthesis of butenolides β -substituted with a furanyl or butenolide group. These compounds were obtained from the cyclisation of 2,3-allenoic acids followed by carbopalladation of either a 2,3-allenoic ketone (eqn (159))^{124,125} or another molecule of the substrate (eqn (160)).^{116,126} For the latter reaction, the catalyst was regenerated with alkyl iodide and oxygen, the use of only the alkyl iodide being less efficient (eqn (160)). Nevertheless, most of the alkyl iodide was not consumed in the course of the process (recovery of 94% of (3-iodopropyl)benzene for example). Subsequently, the authors observed that iodo salts or benzoquinone can be effectively used instead of the alkyl iodide.¹²⁶ The above couplings using optically active substrates have been realised with excellent diastereoselectivities (eqn (161)¹²⁵ and (162)¹²⁶).





The Pd^{II}-catalysed reaction of 2,3-allenoic acids with allyl bromide affords the expected β -allenic butenolides (eqn (163)), whereas the use of 1-chloro-2-butene leads to two regioisomers (eqn (164)).¹²⁷ 1

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After analysis of different hypotheses, Ma and Zu concluded that this mixture results from an equilibrium of 1-chloro-3butene with 3-chloro-1-butene under the reaction conditions. Thus, the formation of these β -allenic butenolides involves an oxycyclisation followed by Heck-type reaction and elimination of palladium halide.

$$R^{3} = H, R^{1} = Ph, R^{2} = H (73\%)$$

$$R^{3} = H, R^{2} = Me, R^{3} = H (73\%)$$

$$R^{1} = Ph, R^{2} = Me, R^{3} = Et (83\%)$$
(10)

Subsequently, Ma and Chen carried out the synthesis of β -allenic butenolides using allenyl esters instead of allenyl acids (eqn (165)).¹²⁸ The reaction requires the addition of catalytic amounts of a Lewis acid, in particular FeCl₃, which mediates the oxycyclisation to afford the furanonyl iron species **31A** (Scheme 30). Subsequent *trans*-metalation with 25 PdCl₂ gives **31B**, regenerating Fe^{III} species. Heck reaction of **31B** with the allylic bromide leads to **31C**, the subsequent β -Br elimination delivering the product and regenerating the Pd^{II} catalyst.

PdCl₂ (0.05 equiv.) (164)35 35 DMA 50 °C 35 5 h (5 equiv.) 19%, ratio = 17:1 4040 OCO₂Me MeOK NBn 45 45 NBn Θ Pd R¹ 0 MeOPd 33B 33D (a) CO K2CO3 KHCO3 ĸ⊕ (b) 50 MeOPd 50 Θ 0 NBn 0 R 33A NBn **HNBn** NBn MeOK R¹ C 33C 55 55

F









25 6.4. Lactonisation-allenylation

Recently, Ma and co-workers reported the stereoselective synthesis of (*Z*)-5-(1,3,4-alkatrien-2-yl)-4,5-dihydro-2-(3*H*)-furanones from 6-monosubstituted 4,5-allenoic acids and propargylic carbonates (eqn (166)).¹²⁹

Under the experimental conditions they used for the cyclisation-carbonylation of 4,5-dienals (eqn (111)), Walkup and Mosher synthesised the furanone shown in eqn (167) from 4,5-hexadienoic acid.⁹⁶

$$\begin{array}{c} \begin{array}{c} \label{eq:pdCl} \mathsf{PdCl}_2 \left(0.03 \text{ equiv.}\right) \\ \hline \\ \mathsf{CuCl}_2 \left(3 \text{ equiv.}\right), \mathsf{CO} \left(1 \text{ atm}\right) \\ \hline \\ \\ \begin{array}{c} \mathsf{Propylene oxide/MeC(OEt)_{3/}} \\ \mathsf{CH}_2\mathsf{Cl}_2/\mathsf{MeOH} \left(1:1:5:5\right), 25 \ ^\circ\mathsf{C}, 18 \text{ h} \end{array} \right) \end{array} \begin{array}{c} 0 \\ \hline \\ \\ \begin{array}{c} \mathsf{MeO} \\ \\ \\ \\ \begin{array}{c} \mathsf{S5\%} \\ \\ \\ \end{array} \end{array} \begin{array}{c} 2(0) \\ \\ \\ \\ \\ \\ \end{array} \end{array}$$

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6.6. Bromination-lactonisation

In 1998, Bäckvall and Jonasson disclosed the synthesis of the bromolactones shown in eqn (168) from the Pd^{II}-catalysed



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reaction of 4,5- and 5,6-hexadienoic acids with lithium bromide.^{100,130} The mechanism is likely similar to the one proposed for the oxybromination of allenic alcohols (Scheme 20), the bromination preceding the cyclisation and the regeneration of the catalyst being also assumed by benzoquinone or Cu^{II}/O_2 .^{100,130}



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6.7. Lactonisation/Michael-type addition

In 2003, Lu and Liu reported the Pd^{II}-catalysed intramolecular cyclisation of 4,5- and 5,6-hexadienoic acids with conjugate addition to α,β-unsaturated carbonyl compounds (eqn (169)).¹³¹ Although the reactions were carried out in the presence of an excess of lithium bromide, as under the above previous Bäckvall experimental conditions summarised in eqn (168), the formation of bromolactones was not observed.
50 The mechanism suggested for these domino cyclisation-

- conjugate additions involves the 5- or 6-*exo-trig* addition of the carboxylate to the Pd^{II}-coordinated allenyl moiety to afford the vinyl palladium intermediate **32A** (Scheme 31). Addition of the latter to the α , β -unsaturated carbonyl compound provides
- 55 **32B**, which suffers protonolysis. The role of LiBr would be to inhibit the β-H elimination from **32B**;¹³¹ its coordination to Pd

would lead to the absence of a vacant coordination position, which is a prerequisite for the PdH elimination.¹³² In contrast, the absence of a bromide addition to the allenyl moiety remains unexplained.

In contrast to the *exo-trig* cyclisation of 4,5- and 5,6- 25 hexadienoic acids (eqn (169)), the reaction of 5-methylhexa-3,4-dienoic acid under Lu's conditions led to dihydropyranones stemming from a 6-*endo-trig* cyclisation (eqn (170)).¹³¹

Oxazolidinonisation

7.1. Oxazolidinonisation-arylation

Under basic conditions, Ma's team carried out the stereoselective synthesis of the 5-alkenyloxazolidin-2-ones shown in eqn (171) from the Pd-catalysed reaction of 4-monosubstituted 2,3-allenyl amines with carbon dioxide and then phenyl iodide.¹²⁹ Subsequently, these domino reactions have been reported using *N*-benzylbuta-2,3-dien-1-amine and 4,4disubstituted 2,3-allenyl amines (eqn (172)).¹³³



7.2. Oxazolidinonisation-alkenylation

The 5-*exo-trig* cyclisation of *N*-benzylbuta-2,3-dien-1-amine can be also followed by cross-coupling with 1-hexenyl iodide or (2-55 iodovinyl)benzene (eqn (173)).¹³³



= Me(CH₂)₃ (82%), Ph (83%)

(173)

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Carboiminolactonisation or carbolactamisation

The Pd⁰-catalysed reaction of 4,4-disubstituted 2,3-allenamides with aryl iodides affords arylated iminolactones in high yields (eqn (176)).¹³⁴ The domino reaction arises also effectively with vinyl iodides, and, moreover, occurs with preservation of the configuration of the C=C bond (eqn (176)). Instead of such a reaction, 4-monosubstituted 2,3-allenamides provided γ hydroxy-y-lactams (eqn (177)).¹³⁴ These reactions would proceed via carbopalladation of the allenvl unit to give a η^3 allypalladium complex. The intermolecular N- or O-attack of the latter would depend on the steric hindrance at the 4position. The N-attack will be followed by oxidation of the lactam.134

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7.3. Oxazolidinonisation-allenylation

(1 2 equiv

A reaction rather similar to that depicted in eqn (166) has been obtained using 4-monosubstituted 2,3-allenyl amines instead of the 4,5-allenoic acids (eqn (174)).¹²⁹ In adjusting the reaction 25 temperature, an efficient transfer of chirality was observed from enantioenriched 2,3-allenyl amines (eqn (175)). Two plausible reaction pathways have been proposed (Scheme 32).¹²⁹ Path a involves the carbopalladation of the substrate derivative 33A by the allenylpalladium complex 33B obtained from the pro-30 pargylic carbonate. The intramolecular nucleophilic addition

to the resulting η^3 -allylpalladium unit provides the product. Path b occurs via anti-oxypalladation of 33A by 33B to afford

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(178)

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From 4,4-disubstituted 2,3-allenamides, a similar process was obtained using 1,2-allenyl ketones instead of organic iodides (eqn (178)).¹³⁵ The catalytic cycle presented by Ma's 2.0 team (Scheme 33, path a) is oversimplified. We suspect the carbopalladation of the allenvl ketone leading to 34A. The reaction of 34A with the allenamide would give the



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PdCl₂(MeCN)₂ (0.01 equiv.) R benzoquinone (1-1.1 equiv.) AcOH, 25-36 °C, 2-17 h (2 equiv.)

R¹ = Bn, R² = R³ = Me, R⁴ = Me (86%), Ph (56%), Bn (81%), *n*-Bu (76%), *n*-hept (64%) $R^1 = H, R^2 = R^3 = Me, R^4 = Me$ (81%), Ph (55%), *n*-Bu (76%), *n*-hept (59%) $R^1 = H, R^2 - R^3 = (CH_2)_5, R^4 = Me (80\%), n-Bu (70\%)$

35 in Scheme 14, path *a*.

 η^3 -allypalladium complex 34B, which could evolve as proposed tetrafluoroborate.⁵⁴ In these reactions, potassium carbonate 35 has a dual role: base and source of carbon dioxide.

9. Carbocarbonatation

With potassium carbonate instead of cesium carbonate, Kang 40 and co-workers have, in some cases, obtained (1-phenylvinyl)dioxolanones (eqn (179)) rather than (1-arylvinyl)oxiranes (eqn (56)) from α -hydroxyallenes and diphenyliodonium

Pd(OAc)₂ (0.05 equiv PPh3 (0.2 equiv.) Ph₂IBF₄ 40 DMF 60 °C 3 h (1.2 equiv.) (2.5 equiv.) R = n-pent (76%), i-Pr (73%), Cy (68%) (179)

> R⁴ R⁴

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$$R^{5}X = PhI \begin{cases} n = 0 \begin{cases} R^{1} = R^{2} = R^{3} = R^{4} = H; 52\%; R^{1} = R^{2} = R^{3}, R^{4} = Me; 99\%; \\ R^{1} = R^{2} = R^{3} = H, R^{4} = Me; 72\%; R^{1} = R^{2} = R^{3} = H, R^{4} = Me; 99\%; \\ R^{1} = R^{2} = R^{3} = H, R^{4} = Me; 72\%; R^{1} = R^{2} = R^{3} = R^{4} = Me; 99\%; \end{cases}$$
(180)

$$R^{5}X = PhI \begin{cases} n = 0 \begin{cases} R^{1} = R^{2} = R^{3} = R^{4} = H; 52\%; R^{1} = R^{2} = R^{3}, R^{4} = Me; 99\%; \\ R^{1} = R^{2} = R^{3} = H, R^{4} = Me; 82\%; R^{1} = Ph, R^{2} = R^{3} = H, R^{4} = Me; 72\%; \\ R^{1} = R^{2} = Me, R^{3} = H, R^{4} = Me; 72\%; R^{1} = R^{2} = H, R^{3} = R^{4} = Me; 93\% \\ n = 1 \begin{cases} R^{1} = R^{2} = R^{3} = H, R^{4} = Me; 70\% \\ R^{1} = R^{2} = R^{3} = R^{4} = H (at 50 \ ^{\circ}C \text{ for 40 h}); 27\% \\ R^{5}X = PhCH = CHBr; 55\% \end{cases}$$

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- ¹ Cyclic carbonates have also been obtained from α-, β- and γhydroxyallenes by Uemura's team using Pd⁰ catalysis, an aryl or styryl halide and both potassium carbonate and carbon dioxide pressure (eqn (180)).¹³⁶ The η^3 -allypalladium obtained from the
- 5 Pd⁰-mediated addition of the organic halide to the allenyl moiety was presumed as an intermediate.

10. Conclusions

- ¹⁰ Even if this review is limited to the Pd-catalysed formation of C– O bonds from reactions with carbons of allenes, the diversity of the possibilities presented above highlights the interest of these substrates in organic chemistry, especially for the synthesis of heterocycles. While some efficient diastereoselective
- ¹⁵ reactions have been disclosed, only a few examples of enantioselective processes have been reported and, in most cases, better ligand-mediated chirality would be highly desirable. Most of the above methods involve domino reactions, but the order of the successive steps often remains a matter of debate.
- ²⁰ Given the number of related reports in the last few years, we anticipate that further applications and new valuable developments are bound to unfold.

25 Abbreviations

	BINAP	2,2'-Bis(diphenylphosphino)-1,1'-binaphthyl
	cat.	Catalytic
	Ср	η^{5} -C ₅ H ₅
30	Су	Cyclohexyl
	dba	Dibenzylidene acetone
	dppb	1,4-Bis(diphenylphosphino)butane
	dppp	1,3-Bis(diphenylphosphino)propane
	ee	Enantiomeric excess
35	equiv.	Equivalent
00	rt	Room temperature
	TDMPP	Tris(2,6-dimethoxyphenyl)phosphine

References

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- 1 (a) J. March, Advanced Organic Chemistry, Wiley, New York, 4th edn, 1992; (b) R. Zimmer, Synthesis, 1993, 165–178;
 (c) B. A. Trofimov, J. Heterocycl. Chem., 1999, 36, 1469–1490; (d) J. Marshall, Chem. Rev., 2000, 100, 3163–3186; (e) X. Lu, C. Zhang and Z. Xu, Acc. Chem. Res., 2001, 34, 535–544; (f) L. Sydnes, Chem. Rev., 2003, 103, 1133–1150; (g) M. Tius, Acc. Chem. Res., 2003, 36, 284–290;
 (h) L.-L Wei, H. Xiong and R. P. Hsung, Acc. Chem. Res., 2003, 36, 773–782; (i) L. Brandsma and N. A. Nedolya, Synthesis, 2004, 735–745; (j) S. Ma, Chem. Rev., 2005, 105, 2829–2872.
 - 2 (a) R. Zimmer, C. U. Dinesh, E. Nandanan and F. A. Khan, *Chem. Rev.*, 2000, 100, 3067–3125; (b) A. S. K. Hashmi, *Angew. Chem., Int. Ed.*, 2000, 39, 3590–3593; (c) R. W. Bates
- and V. Satcharoen, *Chem. Soc. Rev.*, 2002, **31**, 12–21; (*d*) S. Ma, *Acc. Chem. Res.*, 2003, **36**, 701–712; (*e*) S. Yu

- and S. Ma, *Angew. Chem., Int. Ed.*, 2012, **51**, 3074–3112; (*f*) M. P. Muñoz, *Org. Biomol. Chem.*, 2012, **10**, 3584–3594; (*g*) T. Lu, Z. Lu, Z.-X. Ma, Y. Zhang and R. P. Hsung, *Chem. Rev.*, 2013, **113**, 4862–4904; (*h*) T. Lechel, F. Pfrengle, H.-U. Reissig and R. Zimmer, *ChemCatChem*, 2013, **5**, 2100–2130.
- 3 For reviews not specifically devoted to allenes, see:
 (a) Y. Yamamoto and U. Radhakrishnan, *Chem. Soc. Rev.*,
 1999, 28, 199-207; (b) R. C. Larock, *J. Organomet. Chem.*,
 1999, 576, 111-124; (c) R. C. Larock, *Pure Appl. Chem.*, 1999,
 71, 1435-1442; (d) G. Balme, E. Bossharth and N. Monteiro, *Eur. J. Org. Chem.*, 2003, 4101-4111; (e) S. Ma, *Eur. J. Org. Chem.*, 2004, 1175-1183; (f) I. Nakamura and
 Y. Yamamoto, *Chem. Rev.*, 2004, 104, 2127-2198;
 (g) E. M. Beccalli, G. Broggini, M. Martinelli and
 S. Sottocornola, *Chem. Rev.*, 2007, 107, 5318-5365;
 (h) S. Ma, *Aldrichimica Acta*, 2007, 40, 91-102;
 (i) K. C. Majumdar, B. Chattopadhyay, P. K. Maji,
 S. K. Chattopadhyay and S. Samanta, *Heterocycles*, 2010, 81, 517-584.
- 4 B. Cazes, Personal communication, March, 28, 2013.
- 5 (a) D. Medema and R. van Helden, *Recl. Trav. Chim. Pays-Bas*, 1971, 90, 304–323; (b) R. E. Rülke, D. Kliphuis, C. J. Elsevier, J. Fraanje, K. Goubitz, P. W. N. M. van Leeuwen and K. Vrieze, *J. Chem. Soc., Chem. Commun.*, 1994, 1817–1819.
- 6 Y. Inoue, Y. Ohtsuka and H. Hashimoto, *Bull. Chem. Soc. Jpn.*, 1984, 57, 3345–3346.
- 7 No mechanism has been proposed. A catalytic cycle similar to that with phenol as nucleophilic species could occur, see
 30 Scheme 4.
- 8 F. P. J. T. Rutjes, T. M. Kooistra, H. Hiemstra and H. E. Shoemaker, *Synlett*, 1998, 192–194.
- 9 H. Ovaa, M. A. Leeuwenburgh, H. S. Overkleeft, G. A. van der Marel and J. H. van Boom, *Tetrahedron Lett.*, 1998, 39, 35 3025–3028.
- 10 (*a*) H. Oguri, S. Tanaka, T. Oishi and M. Hirama, *Tetrahedron Lett.*, 2000, 41, 975–978; (*b*) Y. Nagumo, H. Oguri, Y. Shindo, S. Sasaki, T. Oishi, M. Hirama, Y. Tomioka, M. Mizugaki and T. Tsumuraya, *Bioorg. Med. Chem. Lett.*, 2001, 11, 2037–2040; (*c*) H. Oguri, *Bull. Chem. Soc. Jpn.*, 2007, 80, 1870–1883.
- 11 S. S. Kinderman, R. Doodeman, J. W. Van Beijma, J. C. Russcher, K. C. M. F. Tjen, T. M. Kooistra, H. Mohaselzadeh, J. H. Van Maarseveen, H. Hiemstra, H. E. Schoemaker and F. P. J. T. Rutjes, *Adv. Synth. Catal.*, 45 2002, 344, 736–748.
- 12 M. Donnard, T. Tschamber, S. Desrat, K. Hinsinger and J. Eustache, *Tetrahedron Lett.*, 2008, **49**, 1192–1195.
- 13 X. M. Yu, H. Han and B. S. J. Blagg, *J. Org. Chem.*, 2005, **70**, 5599–5605.
- 14 N. T. Patil, N. K. Pahadi and Y. Yamamoto, *Can. J. Chem.*, 2005, **83**, 569–573.
- 15 M. Arisawa, T. Sugihara and M. Yamaguchi, *Chem. Commun.*, 1998, 2615–2616.
- 16 R. Grigg, N. Kongkathip, B. Kongkathip, S. Luangkamin 55 and H. A. Dondas, *Tetrahedron*, 2001, 57, 7965–7978.

5

10

20

30

50

1

5

10

15

- 17 R. Doodeman, F. P. J. T. Rutjes and H. Hiemstra, *Tetrahedron Lett.*, 2000, **41**, 5979–5983.
 - 18 R. C. Larock, N. G. Berrios-Pena and C. A. Fried, J. Org. Chem., 1991, 56, 2615–2617.
- 19 R. C. Larock and J. M. Zenner, *J. Org. Chem.*, 1995, **60**, 482–483.
 - 20 J. M. Zenner and R. C. Larock, J. Org. Chem., 1999, 64, 7312–7322.
 - 21 K. Inamoto, A. Yamamoto, K. Ohsawa, K. Hiroya and T. Sakamoto, *Chem. Pharm. Bull.*, 2005, 53, 1502–1507.
- 22 M. Phani Pavan, M. Chakravarty and K. C. Kumara Swamy, *Eur. J. Org. Chem.*, 2009, 5927–5940.
 - 23 Q. Li, X. Jiang, C. Fu and S. Ma, Org. Lett., 2011, 13, 466-469.
- 15 24 R. C. Larock, Y. He, W. W. Leong, X. Han, M. D. Refvik and J. M. Zenner, *J. Org. Chem.*, 1998, 63, 2154–2160.
 - 25 W. Li and M. Shi, Eur. J. Org. Chem., 2009, 270-274.
 - 26 H.-P. Bi, X.-Y. Liu, F.-R. Gou, L.-N. Guo, X.-H. Duan, X.-Z. Shu and Y.-M. Liang, *Angew. Chem.*, *Int. Ed.*, 2007, 46, 7068–7071.
 - 27 M. Chakravarty and K. C. Kumara Swamy, *J. Org. Chem.*, 2006, **71**, 9128–9138.
 - 28 J. Cao, Y. Kong, Y. Deng, G. Lai, Y. Cui, Z. Hu and G. Wang, *Org. Biomol. Chem.*, 2012, **10**, 9556–9561.
- 25 29 T. Aftab, R. Grigg, M. Ladlow, V. Sridharan and M. Thornton-Pett, *Chem. Commun.*, 2002, 1754–1755.
 - 30 H. Ohno, H. Hamaguchi, M. Ohata, S. Kosaka and T. Tanaka, J. Am. Chem. Soc., 2004, 126, 8744–8754.
 - 31 H. Hamaguchi, S. Kosaka, H. Ohno, N. Fujii and T. Tanaka, *Chem.–Eur. J.*, 2007, **13**, 1692–1708.
 - 32 H. Tsukamoto and Y. Kondo, *Org. Lett.*, 2008, **10**, 2633–2636.
 - 33 H. Ohno, H. Hamaguchi, M. Ohata and T. Tanaka, *Angew. Chem., Int. Ed.*, 2003, **42**, 1749–1753.
- 35 34 H. Ohno, H. Hamaguchi, M. Ohata, S. Kosoka and T. Tanaka, *Heterocycles*, 2003, **61**, 65–68.
 - 35 H. Alper, F. W. Hartstock and B. Despeyroux, J. Chem. Soc., Chem. Commun., 1984, 905–906.
 - 36 (a) B. A. Trofimov, R. N. Kudyakova, A. G. Mal'kina,
- V. V. Nosyreva, N. A. Kalinina and A. I. Albanov, *Doklady Chem.*, 2001, 378, 119–121; translated from *Doklady Akad. Nauk*, 2001, 378, 61–63; (b) A. G. Mal'kina, R. N. Kudyakova, V. V. Nosyreva, A. V. Afonin and B. A. Trofimov, *Russ. J. Org. Chem.*, 2002, 38, 1088–1092; translated
 from *Zhurnal Organicheskoi Khimii*, 2002, 38, 1139–1143.
 - 37 B. A. Trofinov, A. G. Mal'kina, R. N. Kudyakova, O. A. Tarasova, V. V. Nosyrava and A. V. Afonin, *Mendeleev Commun.*, 2004, 61–62.
 - 38 K. Okuro and H. Alper, J. Org. Chem., 1997, 62, 1566-1567.
 - 39 R. Grigg, A. Liu, D. Shaw, S. Suganthan, D. E. Woodall and G. Yoganathan, *Tetrahedron Lett.*, 2000, 41, 7125–7128.
 - 40 R. Grigg, A. Liu, D. Shaw, S. Suganthan, M. L. Washington, D. E. Woodall and G. Yoganathan, *Tetrahedron Lett.*, 2000, 41, 7129–7133.
- 55 41 T. Yamamoto, H. Matsuda, Y. Utsumi, T. Hagiwara and T. Kanisawa, *Tetrahedron Lett.*, 2002, 43, 9077–9080.

- 42 Y. Nagao, A. Ueki, K. Asano, S. Tanaka, S. Sano and M. Shiro, *Org. Lett.*, 2002, **4**, 455–457.
- 43 B. Alcaide, P. Almendros and R. Rodríguez-Acebes, *J. Org. Chem.*, 2006, **71**, 2346–2351.
- 44 B. Alcaide, P. Almendros, R. Carrascosa and T. Martínez del Campo, *Chem.–Eur. J.*, 2009, **15**, 2496–2499.
- 45 B. Alcaide, P. Almendros, R. Carrascosa and T. Martínez del Campo, *Chem.-Eur. J.*, 2010, **16**, 13243–13252.
- 46 R. D. Walkup, L. Guan, M. D. Mosher, S. W. Kim and Y. S. Kim, *Synlett*, 1993, 88–90.
- 47 I. Shimizu and J. Tsuji, Chem. Lett., 1984, 233-236.
- 48 M. Ahmar, B. Cazes and J. Gore, *Tetrahedron Lett.*, 1984, 25, 4505–4508.
- 49 B. Cazes, Pure Appl. Chem., 1990, 62, 1867-1878.
- 50 S. Ma and W. Gao, Synlett, 2002, 65–68.
- 51 X. Xie and S. Ma, Chem. Commun., 2013, 49, 5693-5695.
- 52 S. Ma and S. Zhao, J. Am. Chem. Soc., 1999, 121, 7943-7944.
- 53 M. Yoshida, T. Ishii, T. Gotou and M. Ihara, *Heterocycles*, 2004, **64**, 41–44.
- 54 S.-K. Kang, T. Yamaguchi, S.-J. Pyun, Y.-T. Lee and T. 20 G. Baik, *Tetrahedron Lett.*, 1998, 39, 2127–2130.
- 55 S.-K. Kang, T.-G. Baik and A. N. Kulak, *Synlett*, 1999, 324–326.
- 56 B. Alcaide, P. Almendros, T. Martínez del Campo and
 R. Rodríguez-Acebes, *Tetrahedron Lett.*, 2004, 45, 25
 6429–6431.
- 57 B. Alcaide, P. Almendros and R. Rodríguez-Acebes, *Chem.– Eur. J.*, 2005, **11**, 5708–5712.
- 58 R. Grigg, I. Köppen, M. Rasparini and V. Sridharan, *Chem. Commun.*, 2001, 964–965.
- 59 S. Ma and J. Zhang, Chem. Commun., 2000, 117-118.
- 60 S. Ma, J. Zhang and L. Lu, Chem.-Eur. J., 2003, 9, 2447-2456.
- 61 J. A. Marshall and G. S. Bartley, J. Org. Chem., 1994, 59, 7169-7171.
- 62 A. S. K. Hashmi, Angew. Chem., Int. Ed. Engl., 1995, 34, 1581–1583.
- 63 A. S. K. Hashmi, T. L. Ruppert, T. Knöfel and J. W. Bats, J. Org. Chem., 1997, 62, 7295–7304.
- 64 A. Okano, T. Mizutani, S. Oishi, T. Tanaka, H. Ohno and N. Fujii, *Chem. Commun.*, 2008, 3534–3536.
- 65 B. Alcaide, P. Almendros and C. Aragoncillo, *Chem.–Eur. J.*, 2002, **8**, 1719–1729.
- 66 T. Gallagher, I. W. Davies, S. W. Jones, D. Lathbury, M. F. Mahon, K. C. Molloy, R. W. Shaw and P. Vernon, 45 J. Chem. Soc., Perkin Trans. 1, 1992, 433–440.
- 67 A. S. K. Hashmi, J.-H. Choi and J. W. Bats, *J. Prakt. Chem.*, 1999, **341**, 342–357.
- 68 A. S. K. Hashmi, L. Schwarz and J. W. Bats, *J. Prakt. Chem.*, 2000, **342**, 40–51.
- 69 A. S. K. Hashmi and L. Schwarz, *Chem. Ber./Recl.*, 1997, 130, 1449–1456.
- 70 A. S. K. Hashmi, L. Schwarz and M. Bolte, *Eur. J. Org. Chem.*, 2004, 1923–1935.
- 71 F. Yu, X. Lian and S. Ma, Org. Lett., 2007, 9, 1703-1706.
- 72 S. Ma and W. Gao, Tetrahedron Lett., 2000, 41, 8933-8936.

30

35

40

50

5

10

15

35

40

45

50

- 73 S. Ma and W. Gao, J. Org. Chem., 2002, 67, 6104-6112.
 - 74 D. Xu, Z. Lu, Z. Li and S. Ma, *Tetrahedron*, 2004, **60**, 11879–11887.
 - 75 Y. Deng and J.-E. Bäckvall, *Angew. Chem., Int. Ed.*, 2013, **52**, 3217–3221.
 - 76 B. Alcaide, P. Almendros, T. Martínez del Campo, M. C. Redondo and I. Fernández, *Chem.–Eur. J.*, 2011, 17, 15005–15013.
 - 77 B. Alcaide, P. Almendros and T. Martínez del Campo, *Angew. Chem., Int. Ed.*, 2007, **46**, 6684–6687.
 - 78 B. Alcaide, P. Almendros, T. Martínez del Campo,
 E. Soriano and J. L. Marco-Contelles, *Chem.-Eur. J.*, 2009,
 15, 1901–1908.
- 79 B. Alcaide, P. Almendros, T. Martínez del Campo,
 E. Soriano and J. L. Marco-Contelles, *Chem.-Eur. J.*, 2009,
 15, 1909–1928.
 - 80 B. Alcaide, P. Almendros, T. Martínez del Campo,
 E. Soriano and J. L. Marco-Contelles, *Chem.-Eur. J.*, 2009, 15, 9127-9138.
- 81 B. Alcaide, P. Almendros, T. Martínez del Campo and M. T. Quirós, *Chem.-Eur. J.*, 2009, **15**, 3344–3346.
 - 82 S. Ma and L. Li, Org. Lett., 2000, 2, 941-944.
 - 83 Surprisingly, the reaction did not occur under catalysis with $Pd(PPh_3)_4$ or $PdCl_2(PPh_3)_4/4PPh_3$.
- 25 84 Y. Deng, Y. Yu and S. Ma, J. Org. Chem., 2008, 73, 585–589.
 - 85 Y. Deng, Y. Shi and S. Ma, *Org. Lett.*, 2009, **11**, 1205–1208; corrections, 2009, **11**, 2223.
 - 86 Y. Deng, J. Li and S. Ma, *Chem.-Eur. J.*, 2008, 14, 4263-4266.
- 30 87 B. Alcaide, P. Almendros and T. Martínez del Campo, Angew. Chem., Int. Ed., 2006, 45, 4501–4504.
 - 88 B. Alcaide, P. Almendros, T. Martínez del Campo and R. Carrascoa, *Chem.-Asian J.*, 2008, 3, 1140–1145.
 - 89 B. Alcaide, P. Almendros, T. Martínez del Campo, M. T. Quirós, E. Soriano and J. L. Marco-Contelles, *Chem.-Eur. J.*, 2013, **19**, 14233–14244.
 - 90 For the β-elimination competitions, see: J. Le Bras and J. Muzart, *Tetrahedron*, 2012, 68, 10065–10113.
 - 91 R. D. Walkup and G. Park, *Tetrahedron Lett.*, 1987, 28, 1023–1026.
 - 92 R. D. Walkup and G. Park, *Tetrahedron Lett.*, 1988, 29, 5505–5508.
 - 93 R. D. Walkup and G. Park, J. Am. Chem. Soc., 1990, 112, 1597–1603.
 - 94 B. B. Snider and F. He, Tetrahedron Lett., 1997, 38, 5453-5454.
 - 95 C. Shin, Y. Oh, J. H. Cha, A. N. Pae, H. Choo and Y. S. Cho, *Tetrahedron*, 2007, **63**, 2182–2190.
 - 96 R. D. Walkup and M. D. Mosher, *Tetrahedron*, 1993, **49**, 9285–9294.
 - 97 R. D. Walkup and M. D. Mosher, *Tetrahedron Lett.*, 1994, 35, 8545–8548.
 - 98 R. D. Walkup, L. Guan, Y. S. Kim and S. W. Kim, *Tetra*hedron Lett., 1995, 36, 3805–3808.
- 55 99 A similar domino reaction has been obtained using PhCH=CHBr instead of ArX⁹⁸.

- 100 C. Jonasson, A. Horváth and J.-E. Bäckvall, J. Am. Chem. Soc., 2000, 122, 9600–9609. Addition J. Am. Chem. Soc., 2000, 122, 12913–12913.
- 101 The coordination of benzoquinone to Pd^{II} species has been observed by ¹H NMR (J.-E. Bäckvall and A. Gogoll, *Tetrahedron Lett.*, 1988, **29**, 2243–2246.) and ESI-MS spectroscopies. (A. Vasseur, D. Harakat, J. Muzart and J. Le Bras, *J. Org. Chem.*, 2012, **57**, 5151–5158.).
- 102 S. Inuki, Y. Yoshimitsu, S. Oishi, N. Fujii and H. Ohno, *Org. Lett.*, 2009, **11**, 4478–4481.
- 103 S. Inuki, Y. Yoshimitsu, S. Oishi, N. Fujii and H. Ohno, *J. Org. Chem.*, 2010, 75, 3831–3842.
- 104 H. Tsukamoto, T. Matsumoto and Y. Kondo, *J. Am. Chem. Soc.*, 2008, **130**, 388–389.
- 105 (a) G. D. Shier, J. Organomet. Chem., 1967, 10, P15-P17; 15
 (b) G. D. Shier, US Pat., 3567762 A 19710302, 1971.
- 106 M. Al-Masum and Y. Yamamoto, J. Am. Chem. Soc., 1998, 120, 3809–3810.
- 107 R. C. Larock, S. Varaprath, H. H. Lau and C. A. Fellows, J. Am. Chem. Soc., 1984, 106, 5274–5284.
- 108 N. N. Bhuvan Kumar, M. Chakravarty, N. S. Kumar, K. V. Sajna and K. C. Kumara Swamy, *J. Chem. Sci.*, 2009, 121, 23–36.
- 109 S. Husinec, M. Jadranin, R. Markovic, M. Petkovic, V. Savic and N. Todorovic, *Tetrahedron Lett.*, 2010, **51**, 4066–4068.
- 110 S. Husinec, M. Petkovic, V. Savic and M. Simic, *Synthesis*, 2012, 399–408.
- 111 A. Döhring and P. W. Jolly, *Tetrahedron Lett.*, 1980, 21, 3021–3024.
- 112 T. Tsuda, T. Yamamoto and T. Saegusa, *J. Organomet.* 30 *Chem.*, 1992, **429**, C46–C48.
- 113 J.-C. Choi, K. Shiraishi, Y. Takenaka, H. Yasuda and T. Sakakura, *Organometallics*, 2013, **32**, 3411–3414.
- 114 S. Rousset, M. Abarbri, J. Thibonnet, A. Duchêne and J.-L. Parrain, *Chem. Commun.*, 2000, 1987–1988.
- 115 S. Ma, Z. Shi and S. Wu, *Tetrahedron: Asymmetry*, 2001, **12**, 193–195.
- 116 S. Ma and Z. Yu, Org. Lett., 2003, 5, 1507-1510.
- 117 S. Ma and Z. Shi, J. Org. Chem., 1998, 63, 6387-6389.
- 118 Experiments using alkynyl halides instead of aryl halides and carried out with this Pd/Ag procedure did not yield the corresponding butenolides in decent yields: (a) S. Ma, Z. Shi and Z. Yu, *Tetrahedron Lett.*, 1999, 40, 2393–2396; (b) S. Ma, Z. Shi and Z. Yu, *Tetrahedron*, 1999, 55, 12137–12148.
- 119 S. Ma and Z. Shi, Chem. Commun., 2002, 540-541.
- 120 I. W. Davies, D. I. C. Scopes and T. Gallagher, *Synlett*, 1993, 85–87.
- 121 S. Ma, D. Duan and Z. Shi, Org. Lett., 2000, 2, 1419-1422.
- 122 L. Besson, J. Bazin, J. Goré and B. Cazes, *Tetrahedron Lett.*, 50 1994, 35, 2881–2884.
- 123 S. Ma and Z. Yu, Angew. Chem., Int. Ed., 2003, 42, 1955–1957.
- 124 S. Ma and Z. Yu, Angew. Chem., Int. Ed., 2002, 41, 1775–1778.
- 125 S. Ma and Z. Yu, Chem.-Eur. J., 2004, 10, 2078-2087.

10

20

25

35

55

1

126 S. Ma, Z. Yu and Z. Gu, *Chem.-Eur. J.*, 2005, **11**, 2351–2356.
 127 S. Ma and Z. Yu, *J. Org. Chem.*, 2003, **68**, 6149–6152.

- 129 J. Ye, S. Li and S. Ma, Org. Biomol. Chem., 2013, 11, 5370-5373.
- 5 130 C. Jonasson and J.-E. Bäckvall, *Tetrahedron Lett.*, 1998, **39**, 3601–3604.
 - 131 G. Liu and X. Lu, Tetrahedron Lett., 2003, 44, 127-130.
 - 132 (a) P. M. Henry, Acc. Chem. Res., 1973, 6, 16-24;
 (b) Z. Wang, Z. Zhang and X. Lu, Organometallics, 2000,
- **19**, 775–780; (*c*) Q. Zhang, X. Lu and X. Han, *J. Org. Chem.*, 2001, **66**, 7676–7684; (*d*) S. Aït-Mohand, F. Hénin and J. Muzart, *Organometallics*, 2001, **20**, 1683–1686.
- 133 S. Li, J. Ye, W. Yuan and S. Ma, *Tetrahedron*, 2013, **69**, 10450–10456.
- 134 S. Ma and H. Xie, J. Org. Chem., 2002, 67, 6575–6578.
- 135 S. Ma, Z. Gu and Z. Yu, J. Org. Chem., 2005, 70, 6291–6294.
- 136 K. Uemura, D. Shiraishi, M. Noziri and Y. Inoue, *Bull. Chem. Soc. Jpn.*, 1999, 72, 1063–1069.

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25

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35

40

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- 35

40

45

50

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¹²⁸ B. Chen and S. Ma, Chem.-Eur. J., 2011, 17, 754-757.