ChemComm

Accepted Manuscript





This is an *Accepted Manuscript*, which has been through the RSC Publishing peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, which is prior to technical editing, formatting and proof reading. This free service from RSC Publishing allows authors to make their results available to the community, in citable form, before publication of the edited article. This *Accepted Manuscript* will be replaced by the edited and formatted *Advance Article* as soon as this is available.

To cite this manuscript please use its permanent Digital Object Identifier (DOI®), which is identical for all formats of publication.

More information about *Accepted Manuscripts* can be found in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics contained in the manuscript submitted by the author(s) which may alter content, and that the standard **Terms & Conditions** and the **ethical guidelines** that apply to the journal are still applicable. In no event shall the RSC be held responsible for any errors or omissions in these *Accepted Manuscript* manuscripts or any consequences arising from the use of any information contained in them.

RSCPublishing

www.rsc.org/chemcomm Registered Charity Number 207890 Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxx

COMMUNICATION

Dynamic behaviour in giant unilamellar vesicles induced by the uptake of [70]fullerene[†]

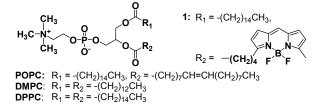
Atsushi Ikeda,* Tomohiro Hida, Tatsuya Iizuka, Manami Tsukamoto, Jun-ichi Kikuchi and Kazuma Yasuhara

s Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX DOI: 10.1039/b000000x

To obtain direct evidence for the fullerene-exchange reactions from the γ -cyclodextrin cavities to the lipid membranes following the addition of a C_{70} + γ -cyclodextrin complex, we

¹⁰ monitored the dynamic behaviours of the giant unilamellar vesicles. A number of the C₇₀ aggregates generated in the lipid membranes moved about vigorously.

- In the last four decades, liposomes have received increasing levels of attention not only for their potential application as drug 15 carriers but also as model cell membrane systems.¹ Fullerenes $(C_x: x = 60 \text{ or } 70)$ -incorporating liposomes $(LMIC_x)^{2,3}$ have been prepared by the fullerene-exchange reaction from the $C_x \bullet \gamma$ cyclodextrin (7-CDx) complex to small unilamellar vesicles (SUVs).⁴⁻¹⁶ LMIC₆₀ prepared by the exchange method does not 20 show any discernible morphological changes in its cryotransmission electron microscopy (TEM) images ([C₆₀]/[dimyristoylphosphatidylcholine (DMPC)] 30 mol%).^{15,16} In contrast, in the cryo-TEM images of LMIC₇₀ $([C_{70}]/[DMPC] = 10 \text{ mol}\%)$, the majority of the liposomes 25 exhibited humps in the lipid bilayer membranes. Furthermore,
- these humps become larger planar structures as the amount of C_{70} present in the system was increased ($[C_{70}]/[DMPC] = 30$ mol%).¹⁶ These self-aggregates of C_{70} grow in a two dimensional manner. Investigations towards developing a deeper
- ³⁰ understanding of the behaviour of C_{70} incorporated in the lipid bilayer membrane are particularly important, not only in terms of the design of effective biomedical materials for applications using C_{70} , but also for identifying any of the potential deleterious effects of C_{70} .¹⁷⁻²⁰ In contrast, giant unilamellar vesicles
- ³⁵ (GUVs)²¹ consisting of phospholipids have been used as models to study the membrane dynamics of living cells.^{22,23} In the present study, we have performed a series of real-time observations of the dynamic behaviour in the fullerene-exchange process using phase contrast and fluorescence microscopies, paying particular to attention to the aggregation of C₂, within the lipid-membrane of
- $_{\rm 40}$ attention to the aggregation of $C_{\rm 70}$ within the lipid-membrane of



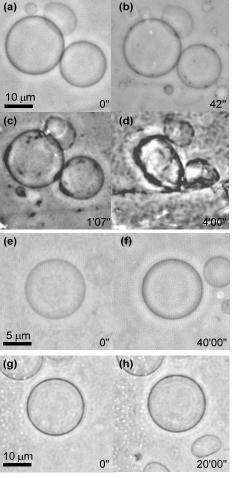


Fig. 1 Time-dependent shape transitions of the GUVs consisting of POPC following the addition of the (a)-(d) C_{70} · γ -CDx complex (0.25 mM) solution, (e) and (f) γ -CDx solution (1 mM) and (g) and (h) C_{60} · γ -CDx complex (0.25 mM) solution at ambient temperature. All of the micrographs are phase contrast images. The time elapsed following the starting injection of the C_{70} · γ -CDx complex solution or γ -CDx solution through the micropipette has been indicated for each image.

50 GUVs.

An aqueous solution of the C_{70} γ -CDx complex (0.25 mM) was added to the GUVs consisting of 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphatidylcholine (POPC) and real-time phase

This journal is $\ensuremath{\mathbb{C}}$ The Royal Society of Chemistry [year] contrast images of the resulting mixture were measured using confocal laser microscopy (Figs. 1a-d and S1 and Movie S1). Following a period of 28 sec, although the shape of the GUVs hardly changed, large aggregates formed in the lipid membrane

- ⁵ with no separation of the individual GUVs (Figs. 1b and S1b).²⁴ The number of aggregates increased over time, and the aggregates themselves were actively integrated through their movements into the lipid membrane (Figs. 1c and S1c and Movie S1). Finally, the integration of the aggregates led to the shrinkage
- ¹⁰ of the GUVs (Figs. 1d and S1e). To confirm that the large aggregates consisted of C_{70} incorporated in the lipid membrane of the GUVs, the changes in the GUVs were investigated via the addition of an aqueous solution of γ -CDx alone (Figs. 1e,f and S2). In this particular case, no aggregation or changes in the
- ¹⁵ shape of the GUV were observed, indicating that the addition of a large excess of γ -CDx alone ([γ -CDx] = 1 mM) could not change the morphology of the lipid membrane (Figs. 1f and S2b). Taken together, these results demonstrated that the large aggregates shown in Fig. 1b and 1c consisted of C₇₀. Furthermore, no
- ²⁰ changes in the shape of the GUVs in the γ -CDx alone case indicated that the shrinkage of the GUVs in the presence of C₇₀ could be attributed not only to the osmotic pressure caused by the large excess of the C₇₀• γ -CDx complex ([total γ -CDx] = 1 mM) but also to the weakening of the vesicles of the GUVs by the ²⁵ integration of the C₇₀ aggregates.
- As shown in Figs. 1g, 1h and S3, the addition of the C_{60} • γ -CDx complex at ambient temperature did not lead to the aggregation or change in shape of the GUVs consisting of POPC.²⁵ Two potential explanations have been provided to explain why no C_{60} -
- $_{30}$ aggregates formed in this experiment, including (i) the C_{60}aggregates in the lipid membrane were too small to be observed by phase contrast microscopy, and (ii) C_{60} was not incorporated into the liposomes. To determine which of these explanations was correct, we collected fluorescence microscopy images using the
- ³³⁵ GUVs consisting of POPC and 2-(4,4-difluoro-5,7-dimethyl-4bora-3a,4a-diaza-*s*-indacene-3-dodecanoyl)-1-hexadecanoyl-*sn*glycero-3-phosphocholine (1) ([1]/[POPC] = 0.25 mol%). When 1 and C₆₀ or C₇₀ coexisted in the liposomes, the C₆₀ or C₇₀ acted as a quencher. The addition of the C₇₀• γ -CDx complex led to the
- ⁴⁰ immediate quenching of the fluorescence (Fig. S4a and S4b), whereas the addition of the C_{60} • γ -CDx complex did not, even after 40 min (Fig. S4c-f). These findings strongly supported the suggestion that C_{60} was not incorporated into the liposomes [i.e., explanation (ii)] and are consistent with the previous result that ⁴⁵ LMIC₆₀ cannot be prepared by the exchange method at ambient

temperature.⁶ We recently reported that the occurrence of a complete C_{70} exchange reaction at temperatures above the phase transition

- temperature (T_m) between the gel- and liquid crystal-phases of the SUVs. Interestingly, however, when the temperature was lowered to a value below the T_m , barely any of the C₇₀ was transfer from the γ -CDx cavity to the membranes of the SUVs.^{12,13} To investigate the effect of the T_m of the GUVs on the aggregation process, the C₇₀-exchange reaction was conducted at temperatures
- ss above and below the $T_{\rm m}$ value of using GUVs composed of dipalmitoylphosphatidylcholine (DPPC, $T_{\rm m} = 41^{\circ}$ C) and DMPC ($T_{\rm m} = 23^{\circ}$ C). No shape transitions occurred in the GUV composed of DPPC following the addition of the C_{70} · γ -CDx

complex at ambient temperature ($<T_m$), even after 20 min (Figs. ⁶⁰ 2a, 2b and S5). Fluorescence microscopy of the GUVs consisting of DPPC and **1** ([**1**]/[DPPC] = 0.25 mol%) revealed that the

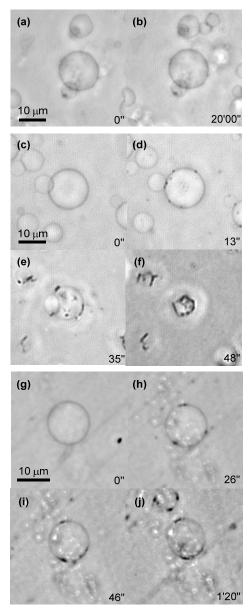


Fig. 2 Time-dependent shape transitions of the GUVs consisting of (a) and (b) DPPC following the addition of the C_{70} */-CDx complex solution at ambient temperature, (c)-(f) DMPC following the addition of the C_{70} */-CDx complex solution at 37°C, and (g)-(j) DPPC following the addition of the C_{70} */-CDx complex solution at 44°C. All of the micrographs are phase-contrast images. The time elapsed following the injection of the C_{70} */-CDx complex solution through the micropipette has been indicated ⁷⁰ for each image.

addition of the C₇₀• γ -CDx complex did not lead to fluorescence quenching (Fig. S6), indicating that the C₇₀ had not been incorporated into the liposomes. To confirm whether the observed differences in behaviour between the POPC and DPPC 75 were caused by the effect of $T_{\rm m}$ or by the structures themselves, DMPC with a structure similar to that of DPPC was also used ($T_{\rm m}$ = 23°C). The GUVs consisting of DMPC at 37°C (> $T_{\rm m}$) behaved in a similar manner to the GUVs consisting of POPC ($T_{\rm m}$ <0°C) at

2 | Journal Name, [year], [vol], 00-00

ambient temperature (Figs. 2e, 2f and S7), suggesting that the different behaviours observed at ambient temperature were related to the $T_{\rm m}$ values of the materials. The conclusion was further supported by the observation that the behaviour in the

- ⁵ GUVs consisting of DPPC at 44°C (> T_m) was similar to that of the GUVs consisting of POPC at ambient temperature (Figs. 2g-j and S8). These results indicated that the C₇₀-exchange reaction only occurred at temperatures greater than the T_m , which was in agreement with our previous results in the SUVs.^{12,13}
- ¹⁰ Attempts to observe changes in the shape of the GUVs following the C_{70} -exchange reaction at higher temperatures failed because the GUVs were not stable, and heating at temperatures in excess of 50°C was required to evaporate the solvent. We therefore investigated the influence of the addition of the C_{70} - γ -
- ¹⁵ CDx complex at 80°C using cryo-TEM images of the SUVs consisting of DMPC. As shown in Fig. 3a, no aggregation or changes in the shape of the SUVs were observed for the exchange reaction when it was conducted at 80°C. Interestingly, however, small aggregates of C_{70} , although unobservable in the cryo-TEM
- ²⁰ images, were formed in the lipid membranes based on the peak broadening of the absorption band (Fig. S9). In a separate experiment, when the solution was heated at 80°C for 2 h following an exchange reaction at 30°C, humps remained on the surfaces of the liposomes (Fig. 3b), indicating that the formation
- $_{25}$ of the large C_{70} -aggregates was an irreversible process and that the large aggregates did not grow in the C_{70} -exchange process at $80^\circ C.$

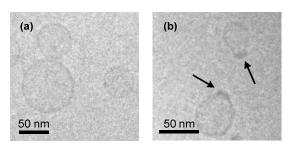


Fig. 3 Cryo-TEM images of SUVs consisting of DMPC following the ³⁰ addition of the C_{70} * γ -CDx complex solution (a) at 80°C, and (b) at 30°C before being heated at 80°C for 2 h ([C_{70}]/[DMPC]= 10 mol%).

In summary, we have directly observed the processes involved in a C₇₀-exchange reaction using GUVs. At temperatures greater than the T_m , the C₇₀-aggregates migrated into the lipid membranes ³⁵ and became progressively larger through coalescence with other similar aggregates. Finally, the resulting huge aggregates led to the shrinkage and, in some cases, occasional bursting of the GUVs. At temperatures below the T_m of DPPC, the GUVs did not change in shape and no C₇₀ aggregates were observed in the

- ⁴⁰ membrane. These results were consistent with our previous findings, where the C_{70} -exchange reaction was found to be dependent on the T_m of the liposomes in the SUVs. No shape transitions or aggregation events were observed in the SUVs when the C_{70} -exchange reaction was carried out at 80°C. Given
- ⁴⁵ that the C₇₀-exchange reaction occurred at this temperature, it has become apparent that large aggregates were not generated at the high temperature. The observation of this behaviour in the artificial lipid membrane could be very important not only in terms of the design of effective biomedical materials for use in ⁵⁰ photosensitisers in photodynamic therapy using C₇₀ but also for

identifying any potential deleterious effects of C₇₀ in human cells. This work was supported by JSPS KAKENHI a Grant-in-Aid

for Scientific Research (B) (Grant No. 25288037), a Grant-in-Aid for Challenging Exploratory Research (Grant Nos. 24655128 and 25550252) and a Creating Aid for Name Scientific (A) (Creat

55 25650053) and a Grant-in-Aid for Young Scientists (A) (Grant No. 24681028).

Notes and references

Graduate School of Materials Science, Nara Institute of Science and Technology, 8916-5 Takayama, Ikoma, Nara 630-0192, Japan.

60 E-mail: aikeda@ms.naist.jp; Fax: +81-743-72-6099; Tel: +81-743-72-6099

† Electronic Supplementary Information (ESI) available: [Experimental procedures, movie of phase contrast microscopy, phase contrast and fluorescence microscopy images and UV-vis absorption spectra]. See 65 DOI: 10.1039/b000000x/

- [‡] This papre is dedicated to the 70th birthday of Professor S. Shinkai.
- Liposomes: A Practical Approach, eds. V. P. Torchilin and W. Weissig, 2nd edn.Oxford University Press, Oxford, 2003.
- H. Hungerbühler, D. M. Guldi and K.-D. Asmus, J. Am. Chem. Soc.,
 1993, 115, 3386–3387.
- 3 R. V. Bensasson, E. Bienvenue, M. Dellinger, S. Leach and P. Seta, J. Phys. Chem., 1994, 98, 3492–3500.
- A. Ikeda, T. Sato, K. Kitamura, K. Nishiguchi, Y. Sasaki, J. Kikuchi, T. Ogawa, K. Yogo and T. Takeya, Org. Biomol. Chem., 2005, 3, 2907-2909.
- 5 A. Ikeda, Y. Doi, K. Nishiguchi, K. Kitamura, M. Hashizume, J. Kikuchi, K. Yogo, T. Ogawa and T. Takeya, *Org. Biomol. Chem.*, 2007, 5, 1158-1160.
- A. Ikeda, Y. Doi, M. Hashizume, J. Kikuchi and T. Konishi, J. Am.
 Chem. Soc., 2007, **129**, 4140-4141.
- 7 M. Akiyama, A. Ikeda, T. Shintani, Y. Doi, J. Kikuchi, T. Ogawa, K. Yogo, T. Takeya and N. Yamamoto, *Org. Biomol. Chem.*, 2008, 6, 1015-1019.
- Y. Doi, A. Ikeda, M. Akiyama, M. Nagano, T. Shigematsu, T. Ogawa,
 T. Takeya and T. Nagasaki, *Chem.-Eur. J.*, 2008, 14, 8892-8897.
- 9 A. Ikeda, T. Sue, M. Akiyama, K. Fujioka, T. Shigematsu, Y. Doi, J. Kikuchi, T. Konishi and R. Nakajima, *Org. Lett.*, 2008, 10, 4077-4080.
- A. Ikeda, M. Nagano, M. Akiyama, M. Matsumoto, S. Ito, M. Mukai,
 M. Hashizume, J. Kikuchi, K. Katagiri, T. Ogawa and T. Takeya, *Chem.-Asian J.*, 2008, 4, 199-205.
- 11 A. Ikeda, M. Akiyama, T. Ogawa and T. Takeya, *ACS Med. Chem. Lett.*, 2010, **1**, 115-119.
- 12 A. Ikeda, Y. Kawai, J. Kikuchi and M. Akiyama, *Chem. Commun.*, 2010, **46**, 2847-2849.
- 13 A. Ikeda, Y. Kawai, J. Kikuchi, M. Akiyama, E. Nakata, Y. Uto and H. Hori, Org. Biomol. Chem., 2011, 9, 2622-2627.
- 15 A. Ikeda, K. Kiguchi, T. Shigematsu, K. Nobusawa, J. Kikuchi and M. Akiyama, *Chem. Commun.*, 2011, 47, 12095–12097.
- 100 16 A. Ikeda, M. Mori, K. Kiguchi, K. Yasuhara, J. Kikuchi, K. Nobusawa, M. Akiyama, M. Hashizume, T. Ogawa and T. Takeya, *Chem.-Asian J.*, 2012, 7, 605–613.
 - 17 G. Jia, H. F. Wang, L. Yan, X. Wang, R. J. Pei, T. Yan, Y. L. Zhao and X. B. Guo, *Environ. Sci. Technol.*, 2005, **39**, 1378–1383
- ¹⁰⁵ 18 C. M. Sayes, A. M. Gobin, K. D. Ausman, J. Mendez, J. L. West and V. L. Colvin, *Biomaterials*, 2005, **26**, 7587–7595.
 - 19 M. Horie, K. Nishio, H. Kato, N. Shinohara, A. Nakamura, K. Fujita, S. Kinugasa, S. Endoh, K. Yamamoto, O. Yamamoto, E. Niki, Y. Yoshida and H. Iwahashi, *J. Biochem.*, 2010, **148**, 289–298.
- 110 20 A. Ikeda, M. Matsumoto, M. Akiyama, J. Kikuchi, T. Ogawa and T. Takeya, *Chem. Commun.*, 2009, 1547–1549.
 - 21 M. I. Angelova, S. Soléau, P. Méléard, J. F. Faucon and P. Bothorel, Prog. Colloid Polym. Sci., 1992, 89, 127–131.
- 22 F. M. Menger and M. I. Angelova, *Acc. Chem. Res.*, 1998, **31**, 789– 797.
 - 23 Giant Vesicles (Perspectives in Supramolecular Chemistry), eds. P. L. Luisi and P. Walde, John Wiley & Sons, Ltd., New York, USA, 2000.

This journal is © The Royal Society of Chemistry [year]

- 24 Delay times until appearances of aggregations on the GUV surfaces were unsettled because each sample had a different distance between the GUV and the tip of micropipette.
- 25 The C₆₀-exchange reaction from a γ -CDx cavity to liposomes hardly

5 occurs at room temperature.

This journal is © The Royal Society of Chemistry [year]