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#### ARTICLE TYPE

## *In vivo* ratiometric $Zn^{2+}$ imaging in zerbrafish larva using a new visible light excitable fluorescent sensor

Zhipeng Liu,<sup>*a,b,* ‡</sup> Changli Zhang,<sup>*a,c,* ‡</sup> Yuncong Chen,<sup>*a*</sup> Fang Qian,<sup>*a*</sup> Yang Bai,<sup>*a*</sup> Weijiang He,<sup>*a,*\*</sup> and Zijian Guo<sup>*a,*\*</sup>

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A visible light excitable ratiometric  $Zn^{2+}$  sensor was developed by integrating a  $Zn^{2+}$  chelator as ICT donor of fluorophore sulfamoylbenzoxadiazole, which displays the 10  $Zn^{2+}$ -induced hypsochromic emission shift (40 nm) and favors the in vivo ratiometric  $Zn^{2+}$  imaging in zerbrafish larva.

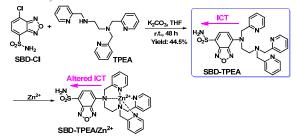
- Labile  $Zn^{2+}$  is attracting much more interests since it is associated with both the physiological progresses such as neurotransmission and gene transcription, <sup>1</sup> and the 15 pathophysiology of certain diseases.<sup>2</sup> Fluorescent  $Zn^{2+}$  imaging with  $Zn^{2+}$  sensors has demonstrated great success in providing temporal-spatial information of  $Zn^{2+}$  homeostasis in live cells.<sup>3</sup> Since the cellular  $Zn^{2+}$  biology is far from the complicated  $Zn^{2+}$ physiology in advanced organisms, the in vivo  $Zn^{2+}$  imaging in
- <sup>20</sup> living animal models is especially demanded. As a valuable vertebrate model of high homology with mammals, zebrafish embryo or larva benefits the studies of development biology, molecular genetics, neuroscience, signal transduction, and pathology from the small size and optically transparency for in <sup>25</sup> vivo imaging.<sup>4</sup> Moreover, the controlled external fertilization
- enables the in vivo imaging during all stages of embryonic development. Therefore, developing  $Zn^{2+}$  sensor especially those of long excitation wavelength to promote the in vivo  $Zn^{2+}$  imaging in zebrafish larvae is one of the most interested areas in
- <sup>30</sup> this field, and several turn-on sensors have been applied for Zn<sup>2+</sup> imaging in live zebrafish larva after our first report.<sup>5</sup> However, this turn-on in vivo Zn<sup>2+</sup> imaging suffers still from the interferences induced by the altered sensor concentration, autofluorescence, bleaching, etc. More accurate Zn<sup>2+</sup> imaging in <sup>35</sup> zebrafish larva is ratiometric Zn<sup>2+</sup> imaging demanding the

ratiometric  $Zn^{2+}$  sensors of visible light/NIR excitability.

In this communication, we report a visible light excitable ratiometric  $Zn^{2+}$  sensor **SBD-TPEA**, which has been utilized for the first in vivo ratiometric  $Zn^{2+}$  imaging in zebrafish larva. This

- <sup>40</sup> sensor was constructed with a mechanism of metal coordination altering ICT (intramolecular charge transfer) effect in fluorophore, which is an effective design rationale for ratiometric metal ion sensors. <sup>6</sup> In this sensor, the  $Zn^{2+}$  chelator, TPEA (*N*,*N*,*N*'tri(pyridin-2-ylmethyl)ethane-1,2-diamine), was incorporated into
- <sup>45</sup> an ICT fluorophore ASBD (4-amine-7sulfamoylbenzo[c][1,2,5]oxadiazole) acting as both the ICT donor group and Zn<sup>2+</sup> ionophore. This sensor was prepared in a moderate yield by reacting SBD-Cl with TPEA via a SN<sub>Ar</sub>

substitution (Scheme 1, please see also Supplementary <sup>50</sup> Information).



Scheme 1. Synthesis of SBD-TPEA and its  $Zn^{2+}$  complexation.

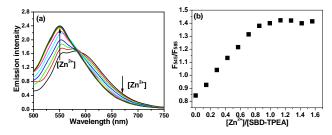
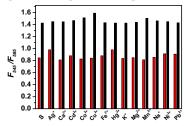


Fig. 1. (a) Emission spectra of 3  $\mu$ M SBD-TPEA ( $\lambda_{ex}$ , 460 nm) in HEPES <sup>55</sup> buffer (50 mM, 0.1 M KNO<sub>3</sub>, pH 7.2, containing 0.15% DMSO) obtained by adding aliquots of Zn<sup>2+</sup> solution (1.2 mM, 1.1  $\mu$ L); (b) the titration profile according to the ratio of emission at 545 nm to that at 585 nm,  $F_{545}/F_{585}$ .

The sensor is able to be dissolved in water with a 60 concentration up to ~ 3.0  $\mu$ M according to a reported determination procedure (Fig. S7).<sup>7</sup> The spectroscopic determination of SBD-TPEA were carried out in HEPES buffer (50 mM HEPES, 100 mM KNO<sub>3</sub>, pH 7.2) containing 0.15 % DMSO. SBD-TPEA exhibits an emission band centered at 585 65 nm, with an excitation maximum at 466 nm. The large Stokes shift (119 nm) is helpful to reduce the excitation interference in imaging. Fluorescence Zn<sup>2+</sup> titration of SBD-TPEA displayed a distinct hypsochromic emission shift from 585 to 545 nm with an isoemission point at 585 nm (Fig. 1a). The ratio of emission at 70 545 nm to that at 585 nm ( $F_{545}/F_{585}$ ) increases linearly from 0.85 to 1.42 with  $[Zn^{2+}]_{total}$  till the  $[Zn^{2+}]_{total}/[SBD-TPEA]$  ratio attains to 1 : 1 (Fig. 1b). Even higher  $[Zn^{2+}]_{total}$  does not lead to any evident change in emission. The excitation and emission maxima for both apo- and Zn<sup>2+</sup>-bound sensors are located in the range of

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visible light, favouring the in vivo imaging in zebrafish larva. Fluorescence pH titration demonstrated that F454/F585 of SBD-TPEA has no pH-dependence in the pH range from 6.5 to 9.0, favoring its ratiometric imaging application in physiological 5 microenvironments (Fig. S8). In addition, the Zn<sup>2+</sup>-induced emission enhancement for SBD-TPEA is limited, suggesting there is no distinct PET (photo-induced electron transfer) effect from Zn<sup>2+</sup> ionophore to the parent fluorophore, ASBD.<sup>6a</sup>



10 Fig. 2. Ratio of emission at 545 to that at 585 nm,  $F_{545}/F_{585}$ , of SBD-TPEA (3 uM) in HEPES buffer (0.15% DMSO, 50 mM HEPES, 100 mM KNO3; pH 7.20) induced by different metal cations. Red bars, ratio for free sensor (S) or in the presence of 1 equiv  $Ag^+$ ,  $Cd^{2+}$ ,  $Co^{2+}$ ,  $Cu^{2+}$ ,  $Fe^{2+}$ ,  $Hg^{2+}$ ,  $Mn^{2+}$ ,  $Ni^{2+}$ ,  $Pb^{2+}$  or 1000 equiv  $Na^+$ ,  $K^+$ ,  $Ca^{2+}$ ,  $Mg^{2+}$ . Black bars, 15 ratio in the presence of  $Zn^{2+}$  (1 equiv), or the indicated metal ions (1 equiv) followed by adding 1 equiv  $Zn^{2+}$ .  $\lambda_{ex}$ , 460 nm.

The UV-vis Zn<sup>2+</sup> titration of SBD-TPEA demonstrated an absorption shift from 456 (Band A,  $\varepsilon$ , 4.46 × 10<sup>3</sup> M<sup>-1</sup>cm<sup>-1</sup>) to 386 nm (Band B,  $\varepsilon$ ,  $3.29 \times 10^3$  M<sup>-1</sup>cm<sup>-1</sup>, Fig. S9). The linear decrease <sup>20</sup> of Band **A** and increase of Band **B** with  $[Zn^{2+}]_{total}$  can be observed simultaneously till the  $[Zn^{2+}]_{total}/[SBD-TPEA]$  ratio attains to 1 : 1. Even higher zinc concentration does not lead to any further change. The UV-vis titration profiles according to the absorbance of the two bands and the clear isobestic point at 416 nm suggest

- 25 that Zn<sup>2+</sup> addition led to only one reaction to form sole Zn<sup>2+</sup> complex of 1:1 stoichiometry. <sup>1</sup>H NMR titration by Zn<sup>2+</sup> confirmed also the 1:1 Zn<sup>2+</sup> binding stoichiometry of SBD-TPEA (Fig. S10), and all N atoms in TPEA are involved in Zn<sup>2+</sup> coordination directly (Figs. S10-S12, Table S1 and Chart S1). MS
- $_{30}$  determination of **SBD-TPEA**/Zn<sup>2+</sup> complex confirmed again the 1:1 Zn<sup>2+</sup> binding stoichiometry (Fig. S6). The data obtained from the two  $Zn^{2+}$  titrations implied that  $Zn^{2+}$  coordination to TPEA amine N attached to benzoxadiazole decreases the ICT effect of ASBD fluorophore and induces the hypsochromic shift of 35 absorption and emission.

The Zn<sup>2+</sup>-specific ratiometric response of SBD-TPEA was confirmed further by fluorescence titration with biorelated metal cations of interest. As shown in Fig. 2, the presence of Na<sup>+</sup>, K<sup>+</sup>,  $Ca^{2+}$ , and  $Mg^{2+}$ , which are abundant in cells, does not interfere

40 with its ratiometric response to  $Zn^{2+}$ , even though their concentration is 1000 times higher than  $[Zn^{2+}]_{total}$ . In addition, the presence of  $Ag^+$ ,  $Fe^{2+}$ ,  $Hg^{2+}$ ,  $Ni^{2+}$ ,  $Mn^{2+} Co^{2+}$ ,  $Cd^{2+}$ ,  $Cu^{2+}$ , and Pb<sup>2+</sup> (1 equiv) does not interfere with its ratiometric sensing ability to  $Zn^{2+}$ . The  $K_d$  value of  $Zn^{2+}$ /SBD-TPEA complex was

 $_{45}$  estimated to be  $\sim 2.1$  nM via determining the  $Zn^{2+}\text{-induced}$ change of  $F_{545}/F_{585}$  ratio in a series of  $Zn^{2+}$  buffer solutions (Fig. S13). <sup>8</sup> The detection limit ( $3\sigma$ /slope) of this sensor was determined to be 0.5 nM (Fig. S14). All these make SBD-TPEA a suitable candidate of ratiometric imaging agent for intracellular 50 and in vivo Zn<sup>2+</sup>.

The intracellular  $Zn^{2+}$  imaging ability of **SBD-TPEA** was investigated in HepG2 cells using confocal microscope with a

dual emission mode (green channel: 510-560 nm; red channel: 580-630 nm) upon excitation at 488 nm, and the ratiometric 55 images were obtained by mediating the green channel image with the related red channel image (Fig. S15). As shown in Fig. 3, the cells stained by SBD-TPEA display the faint green/blue (lower ratio) in cytoplasm in the ratiometric image, indicating the labile  $Zn^{2+}$  level in cytoplasm is low. When exogenous  $Zn^{2+}$  was 60 introduced via incubation the cells with ZnSO<sub>4</sub>/pyrithione solution (5  $\mu$ M, 1:2), the intensive blue was observed inside the cell, indicating the enhanced intracellular Zn<sup>2+</sup> level. According the ratio bar, the red/yellow spots close to the nucleus indicate that the  $Zn^{2+}$  level in these regions is even higher. The followed 65 incubation treatment with the cell membrane permeable Zn2+ chelator, *N*,*N*,*N*',*N*'-tetrakis(2-pyridylmethyl)ethylenediamine (TPEN), displayed very minor green area inside the cells, implying the distinctly suppressed emission ratio and lower Zn2+ level than that in original cells in Fig. 3b. These results also 70 confirmed that the fluorescence ratio enhancement in cells upon ZnSO<sub>4</sub>/pyrithione incubation should be resulted from Zn<sup>2+</sup> binding of SBD-TPEA, and the intracellular Zn<sup>2+</sup> level can be enhanced effectively via Zn2+ incubation. Similar results were obtained also in HeLa cells (Fig. S16), and the higher Zn<sup>2+</sup> level 75 was also observed in the regions close to nucleus when exogenous Zn<sup>2+</sup> was introduced. Co-localization experiments in HeLa cells (Fig. S17) and HepG2 cells via co-staining the cells with SBD-TPEA and Golgi maker BODIPY TR ceramide disclosed that the bright spots of higher Zn<sup>2+</sup> level are Golgi <sup>80</sup> apparatus. In addition, the chelatable  $[Zn^{2+}]$  in Golgi of HepG 2 cells was estimated to be around 0.5 nM (Fig. 3 and Fig. S15). The ratiometric imaging results suggest also that the average chelatable  $Zn^{2+}$  level in HepG2 cells is different from that in HeLa cells. Moreover, the temporal imaging of cells treated by 85 SBD-TPEA displayed no change in cell morphology in 4 h, implying the fine biocompatibility of SBD-TPEA

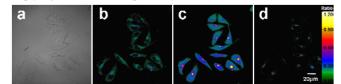


Fig. 3. Confocal fluorescence ratiometric imaging of HepG2 cells stained by SBD-TPEA (10 µM, 20 min) at 25 °C. (a) Bright-field transmission 90 image of the stained cells; (b) ratiometric image of cells in (a); (c) ratiometric image of cells in (b) exposed to ZnSO<sub>4</sub>/pyrithione solution (5 µM, 1:2) for 5 min, followed by staining again with SBD-TPEA solution; (d) ratiometric image of cells in (c) treated by TPEN solution (25  $\mu$ M, 10 min). Ratiometric images were obtained via mediating the fluorescence 95 images collected respectively at green channel (510-560 nm) and red channel (580-630 nm). λex, 488 nm.

Besides the ratiometric Zn<sup>2+</sup> imaging ability in living cells of **SBD-TPEA**, the first ratiometric in vivo Zn<sup>2+</sup> imaging in 3-dayold zebrafish larva was also investigated via staining the larva by 100 SBD-TPEA (50 µM, 1.5 h). As shown in Fig. 4a, the confocal fluorescence images of the larva head exhibit mainly two regions of bright fluorescence, and the overlay of fluorescence and bright-field images discloses the two bright regions are symmetrically located between the two eyes, which was proposed 105 as the neuromasts of the anterior lateral-line system (ALL system) in zebrafish.<sup>9</sup> The ratiometric image obtained via mediating

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fluorescence images obtained respectively from band paths 550-560 and 570-650 nm displays the pale blue regions on the same location, indicating the higher Zn<sup>2+</sup> level in these two bright spots than that in the rest of the head (Fig. 4a-4d). The ratiometric imaging of Zn<sup>2+</sup>-fed zebrafish larva (3-day-old) has also been

- carried out by incubating larva with Zn<sup>2+</sup> solution for 1 h (100 µM), and the ratiomertic image demonstrates two bright cyan spots in the same location (Fig. 4e-4h). Moreover, the two cyan regions are larger than the pale blue regions found in the non-
- $10 \text{ Zn}^{2+}$ -fed larva, and the bright evan implies that the neuromast  $Zn^{2+}$  level in  $Zn^{2+}$ -fed larva is higher than that in non- $Zn^{2+}$ -fed larva. The TEPN (50 µM, 0.3 h) treatment of the SBD-TPEA stained 3-day-old zebrafish larva (50 µM, 1.5h) results in almost dim images and the fluorescence in the nuromast is very low. In
- 15 addition, only very minor faint pale blue spots can be found in the corresponding areas in ratiometric image (Fig. 4i-4l). Comparison between the ratiometric images of the normal zebrafish larva and the TPEN treated larva suggests that the two pale blue regions in the normal zebrafish larva should be correlated to the presence of
- 20 higher labile Zn<sup>2+</sup> level. All the altered emission ratio displayed as the variable color in the ratiometric image is really correlated to the variable chelatable Zn<sup>2+</sup> level in live zebrafish larva. The ratiometric imaging results on 5 zebrafish larvae disclosed that the chelatable  $[Zn^{2+}]$  of neuromasts is around 1.3 nM, while the

 $_{25}$  Zn<sup>2+</sup>-incubation made the [Zn<sup>2+</sup>] in the corresponding regions increase to 10.9 nM. As to the TEPN treated larva, the chelatable  $[Zn^{2+}]$  in neuromast can be reduced to 0.1 nM (Fig. S18). All the preliminary imaging data indicate that SBD-TPEA is an effective  $Zn^{2+}$  ratiometric sensor for *in vivo*  $Zn^{2+}$  quantitative imaging.

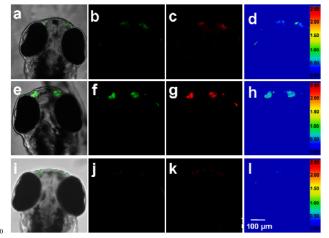


Fig. 4. Confocal fluorescence ratiometric Zn<sup>2+</sup> imaging in the head of three-day-old zebrafish larva at 28.5°C. (a-d) Images of a larva incubated with **SBD-TPEA** (50  $\mu$ M,1.5 h); (e-h) images of a larva fed with Zn<sup>2+</sup> (100  $\mu M,$  1h) solution followed by incubation with SBD-TPEA (50  $\mu M,$ 

 $_{35}$  1.5 h); (i-l) images of a larva incubated with SBD-TPEA (50  $\mu$ M, 1.5 h) followed by 20 min of TPEN incubation (50 µM). (a, e, i) Colocalization of bright-field and fluorescence images for the head (dorsal view); (b, f, j) fluorescence images from band path 500-560 nm; (c, g, k) fluorescence images from the band path 570-650 nm; (d, h, i) ratiometric images

<sup>40</sup> generated from (b, f, j) and (c, g, k).  $\lambda_{ex}$ , 488 nm.

In conclusion, a novel ratiometric Zn<sup>2+</sup> fluorescent sensor SBD-TPEA derived from ICT fluorophore ASBD was developed. This new sensor displays the specific Zn<sup>2+</sup>-induced emission shift from 585 to 545 nm, which provides the sensor

<sup>45</sup> the ratiometric Zn<sup>2+</sup> sensing ability. With the pH-independent sensing behavior in physiological pH range and visible light excitability, SBD-TPEA has been utilized to realize the in vivo ratiometric Zn<sup>2+</sup> imaging in live zebrafish larva for the first time, and the chelatable Zn<sup>2+</sup> level of the nuromasts in 50 the larva head was firstly estimated.

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#### Notes and references

<sup>a</sup> State Key Laboratory of Coordination Chemistry, Coordination Chemistry Institute, School of Chemistry and Chemical Engineering, 210093, Nanjing University, Nanjing Ρ. R China.

60 Email:heweij69@nju.edu.cn; zguo@nju.edu.cn; Fax: +86 25 83314502; Tel: +86 25 83597066.

<sup>b</sup> Department of chemistry, Liaocheng Unviersity, Liaocheng, 252059, P. R. China.

<sup>c</sup> Department of chemistry, Nanjing Xiaozhuang College, Nanjing, 65 210017, P. R. China.

Animal Model Research Center, Nanjing University, Nanjing 210061, P. R. China.

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- # Both authors contributed equally to this manuscript.
- [1] (a) J. M. Berg and Y. Shi, Science, 1996, 271, 1081; (b) M. Lu and D. Fu, Science, 2007, 317, 1746.
- (a) C. J. Frederickson, J.-Y. Koh and A. I. Bush, Nat. Rev. Neurosci., [2] 2005, 6, 449; (b) M. Cortesi, R. Chechik, A. Breskin, D. Vartsky, J. Ramon, G. Raviv, A. Volkov and E. Fridman, Phys. Med. Biol., 2009, **54**. 781.
- [3] (a) E. Tomat and S. J. Lippard, Curr. Opin. Chem. Biol., 2010, 14, 225; (b) Z. Xu, J. Yoon and D. R. Spring, Chem. Soc. Rev., 2010, 39, 1996; (c) E. L. Que, D. W. Domaille and C. J. Chang, Chem. Rev., 2008, 108, 1517; (d) P. Jiang and Z. Guo, Coord. Chem. Rev., 2004, 248, 205.
- [4] (a) S. Ellingsen, M. A. Laplante, M. Konig, H. Kikuta, T. Furmanek, E. A. Hoivik and T. S. Becker, Development, 2005, 132, 3799; (b) H. W. Detrich, M. Westerfield and L. I. Zon, The zebrafish: disease models and chemical screens, academic Press, Waltham, 3rd edn, 2011; (c) L. A. Trinh and S. E. Fraser, Dev. Growth Differ., 2013, 55, 434; (d) S. Rinkwitz, P. Mourrain and T. S. Becker, Prog. Neurobiol., 2011, 93, 231; (e) S.-K. Ko, X. Chen, J. Yoon and I. Shin, Chem. Soc. Rev., 2011, 40, 2120.
- [5] (a) F. Qian, C. Zhang, Y. Zhang, W. He, X. Gao, P. Hu and Z. Guo, J. Am. Chem. Soc., 2009, 131, 1460; (b) Z. Xu, K.-H. Baek, H. N. Kim, J. Cui, X. Qian, D. R. Spring, I. Shin and J. Yoon, J. Am. Chem. Soc., 2010, 132, 601; (c) J. E. Kwon, S. Lee, Y. You, K.-H. Baek, K. Ohkubo, J. Cho, S. Fukuzumi, I. Shin, S. Y. Park and W. Nam, Inorg. Chem., 2012, 51, 8760; (d) K. Jobe, C. H. Brennan, M. Motevalli, S. M. Goldup, M. Watkinson, Chem. Commun., 2011, 47, 6036; (e) Y. Xu, Q. Liu, B. Dou, B. Wright, J. Wang and Y. Pang, Adv. Healthcare Mater., 2012, 1, 485.
- (a) Z. Liu, W. He and Z Guo, Chem. Soc. Rev., 2013, 42, 1568; (b) L. [6] Xue, G. Li, D. Zhu, Q. Liu and H. Jiang, Inorg. Chem., 2012, 51, 10842; (c) L. Xue, G. Li, D. Zhu, C. Yu and H. Jiang, Chem. Eur. J., 2012, 18, 1050; (d) Z. Liu, C. Zhang, Y. Chen, W. He and Z. Guo, Chem. Commun., 2012, 48, 8365.
- [7] H. M. Kim, M. S. Seo, M. J. An, J. H. Hong, Y. S. Tian, J. H. Choi, O. Kwon, K. J. Lee and B. R. Cho, Angew. Chem. Int. Ed., 2008, 47, 5167.
- M. Taki, J. L. Wolford and T. V. O'Halloran, J. Am. Chem. Soc., [8] 2004, 126, 712
- [9] K. A. Grant, D. W. Raible and T. Piotrowski, Neuron, 2005, 45, 69.

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