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Diastereoselective cyclopropanation of α,β-unsaturated carbonyl compounds with vinyl sulfoxonium ylides†

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Herein, we report a three-component stereoselective cyclopropanation of vinyl sulfoxonium ylides with indane 1,3-dione and aldehydes under mild reaction conditions. In contrast to previous reports, the present work shows that electrophilic addition selectively takes place at the α -position of the vinyl sulfoxonium ylide. The interesting feature of this approach is that the multicomponent reaction selectively proceeds because of the difference in nucleophilic reactivity of vinyl sulfoxonium ylides and indane 1,3 dione with electrophilic partners, such as aldehydes and in situ generated arylidenes. Additionally, density functional theory (DFT) studies were conducted to investigate the difference in the reactivity of these reactants, as well as to unveil the mechanism of this three-component reaction. Furthermore, non-covalent interactions of selectivity-determining transition states explain the origin of the diastereoselectivity of cyclopropanation. **COMMUNICATION**
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Introduction

Vinyl cyclopropanes are important building blocks found in many pharmaceuticals and natural products. Some notable examples of medicinal compounds that include the vinylcyclopropane unit are Danoprevir (used in the treatment of COVID-19) and Simeprevir (used in the treatment of Hepatitis C).¹ Additionally, natural insecticides, such as pyrethrines and chrysanthemic acid esters, also contain this unique structural framework.2 Inherent ring strain with the adjacent olefin makes the vinylcyclopropane a versatile synthetic precursor for various organic transformations, such as ring openings, cycloadditions, and rearrangements.³ Therefore, a significant effort has been made to develop efficient methods for synthesising vinylcyclopropane scaffolds. However, a major challenge in synthesising substituted vinylcyclopropanes is obtaining good diastereoselectivity at both the cyclopropane and the alkene units.4

Our continuing interest in sulfur ylides prompted us to explore the reactivity of vinyl sulfoxonium ylides with electrondeficient alkynes and alkenes.⁵ Vinyl sulfoxonium ylides can undergo electrophilic addition at the α- or γ-carbon due to their dipole structure (Scheme 1b).⁶ Recently, our group developed a method to synthesize 2H-chromenes via catalyst-controlled regioselective $[3 + 3]$ annulation of vinyl sulfoxonium

Scheme 1 (a) Bioactive compounds containing the structural frameworks of vinylcyclopropane, (b) possible electrophilic additions on vinyl sulfoxonium ylide, (c) addition of vinyl sulfoxonium ylides on Michael acceptors, and (d) synthesis of vinylcyclopropanes from vinyl sulfoxonium ylides with Michael acceptors generated in situ.

[†]Electronic supplementary information (ESI) available. CCDC 2349128 and 2349131. For ESI and crystallographic data in CIF or other electronic format see DOI: <https://doi.org/10.1039/d4ob00677a> Department of Chemistry, Indian Institute of Technology Delhi, Hauz Khas, New Delhi, 110016, India. E-mail: tkarmakar@iitd.ac.in, vaitla@chemistry.iitd.ac.in

ylides with quinones. In the absence of a metal catalyst, vinyl sulfoxonium ylide reacts with the quinone at the γ-position of the ylide, followed by cyclization to give 2H-chromene-4-carboxylate.⁷ Similarly, when we used propiolates as electrophiles, the reaction selectively underwent an addition reaction at the γ-position of ylides to give dienyl sulfoxonium ylide (Scheme $1c$). 8

Based on these results, we are interested in developing a multicomponent reaction involving a vinyl sulfoxonium ylide as a nucleophile with electrophiles, such as arylidenes that are generated in situ from aldehydes and compounds containing active methylene groups.⁹ However, the straightforward approach for the generation of vinylcyclopropane scaffolds from vinyl sulfoxonium ylide-mediated multicomponent reaction is challenging because the reaction can produce diverse products due to the presence of nucleophilic centers in both ylide and active methylene compounds (Scheme 2). 10 The possible pathways are: (i) similar to the Corey–Chaykovsky reaction, whereby the addition of the α -carbon of the ylide to the aldehyde is followed by the elimination of DMSO to generate epoxide (path A). (ii) Due to the nucleophilicity at the γ -position, the ylide can attack the aldehyde via vinylogous addition (path B). (iii) Based on the well-known Knoevenagel condensation, active methylene compounds can react with the aldehyde to generate arylidenes (path C). (iv) Similar to our previous approach (Scheme 1c), the γ-attack of ylide at the arylidene gave the γ-substituted ylide (path D) and (v) $α$ -attack of ylide at the arylidene gave the vinylcyclopropane (path E). Interestingly, the addition of these three reactive species in the presence of L-proline (20 mol%) in methanol exclusively produced vinylcyclopropane (Scheme 1d). Previously, vinylcyclopropanes were synthesized using carbenoids derived from diazo compounds¹¹ or Simmons–Smith-like reactions,¹² Michael-initiated ring closure reactions of ylides, 13 and functional group interconversion of appropriately functionalized Organic & Biornolecular Chemistry

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 $cyclopropanes.¹⁴$ However, these methods often have drawbacks, such as poor diastereoselectivity,¹⁵ step economy,¹⁶ as well as the use of expensive metals¹⁷ or toxic reagents.^{13a} Thus, the present mild diastereoselective cyclopropanation prompted us to develop a greener synthetic approach to generate complex cyclopropane scaffolds. In the present report, we have developed multi-component cyclopropanation from vinyl sulfoxonium ylides, aldehydes, and indane 1,3-dione. In addition, we have performed density functional theory (DFT) calculations to investigate the reaction mechanism and selectivity of the cyclopropanation.

We commenced our study by employing vinyl sulfoxonium ylide (1a), benzaldehyde (2a), and 1,3-indandione (3) as model substrates. The optimal conditions to afford substituted cyclopropane $(4a)$ was determined to be the use of 20 mol% of (L) proline as the catalyst and methanol as the solvent at room temperature for 30 min, resulting in an 87% isolated yield with >20 : 1 dr (Table 1, entry 1). Despite the reaction affording excellent yield and diastereoselectivity, we did not observe any enantioselectivity. Next, we investigated the reaction under different reaction conditions. A comparison of the changes introduced in each reaction parameter is summarized in Table 1. Lewis acids, such as $In(OTf)_3$ and $Sc(OTf)_3$, gave poor yields of cyclopropane 4a (entries 2 and 3). The reaction gave a low yield of product in the presence of a base, such as NEt_3 (entry 4). Using a Brønsted acid, such as $HBF₄·OEt₂$, led to the formation of the desired product 4a in 85% yield (entry 5). However, using other Brønsted acids, such as acetic acid, failed to produce the product. Instead, the reaction resulted in the insertion of vinyl sulfoxonium ylides into the O–H bond of the carboxylic acid (entry 6).¹⁸ The reaction afforded a low yield of product 4a when TMSOTf was used as the catalyst. We noticed that protic solvents are critical for this transformation because other commonly used solvents in the laboratory, such as DMSO, acetonitrile, and EtOAc, resulted in a low yield of 4a

Scheme 2 Possible reaction pathways between the vinyl sulfoxonium ylide, aldehyde, and indane 1,3-dione.

Table 1 Catalyst screening and optimization of reaction conditions^a

 a Standard conditions: 1a (0.14 mmol), 2a (0.11 mmol), 3 (0.10 mmol), catalyst (20 mol%), solvent (0.1 M), 30 minutes. $\frac{b}{b}$ The yield was determined by ¹H NMR analysis of crude products using dibromomethane as the internal standard. ^c Determined by crude NMR. ^d Isolated yield of the major diastereomer.

(entries 8–10). Interestingly the reaction failed to afford the product in the absence of the proline catalyst (entry 11).

Following these optimization studies, we explored the scope of this cyclopropanation using a variety of aldehydes. As shown in Scheme 3, an electronically diverse set of aryl aldehydes was tolerated. Aryl aldehydes containing an alkyl group, such as methyl, on the para and meta positions, underwent smooth annulation to afford the desired cyclopropanes 4b–4c in good yield (89–90%) and excellent diastereoselectivity $(20:1)$. Also, the aryl aldehydes containing halogen substituents, such as -Br, -F, -Cl at *ortho*, *meta*, and *para* positions, gave the desired vinylcyclopropane 4d–4h in good to excellent yield (80–86%) with excellent diastereoselectivity (dr $> 20:1$). Electron-rich substituents, such as $-OMe$ and $-OCF₃$, and electron-withdrawing substituents, such as $-NO₂$ and $-CF₃$, at the ortho, meta, and para positions were compatible with this cyclopropanation and afforded corresponding annulated products 4i–4p in 79–93% yields. The relative stereochemistry of the two stereogenic centers of 4l was confirmed by X-ray crystallography (CCDC 2349128†), and the relative configurations of the other cyclopropane products were assigned by analogy. Furthermore, a poly-aromatic aldehyde, 2-naphthaldehyde, was tested in this reaction, which successfully produced 4q in 81% yield without the loss of selectivity. The efficiency of this method was further demonstrated by heteroaryl aldehydes, and the transformation was successful with thiophene-2-carboxaldehyde and furfural to afford the products, 4r (89% yield, 4 : 1 dr) and 4s (87% yield, 9 : 1 dr). The reaction is also successful with aliphatic aldehydes, giving the corresponding cyclopropane 4t in 85% with 5 : 1 dr These observations indicate that the diastereoselectivity is significantly influenced by the aldehyde framework.

Next, we evaluated the scope of substituted vinyl sulfoxonium ylides. Initially, a variation of the aryl ring of the ylide was examined. The reaction demonstrated a high tolerance towards various substituents, such as –Cl, –F, –Me, and –OMe, located at the ortho-, meta-, and para-positions of the phenyl ring and afforded the corresponding cyclopropane products 4u–4aa in excellent yield (83–92%) with excellent diastereoselectivity (dr $> 20:1$). The ylide-containing heteroaromatic ring, such as quinoline, underwent annulations to afford 4ab in 35% yield with excellent diastereoselectivity $(dr > 20:1)$. Subsequently, we assessed the variation in the ester moiety of the ylide. The reaction conditions were suitable for a variety of ester functional groups of ylides, including benzyl (4ac), ethyl trimethyl silyl (4ad), allyl esters (4ae), and cyclohexyl esters (4af). Next, the annulation was carried out in the presence of a ylide containing tertiary butyl ester, resulting in the formation of the annulated product 4ag in 77% yield with excellent dr, which shows that there is no steric influence of the ester group of the ylide on diastereoselectivity. Additionally, the incorporation of natural products, such as L-menthol and nerol, as ester moieties effectively afforded the corresponding cyclopropanes 4ah (85%, dr = 1:1) and 4ai (83%, dr > 20:1). Surprisingly, the reaction failed to afford the vinylcyclopropanes 4aj and 4ak using picolinaldehyde and 1H-pyrazole-4 carbaldehyde. When the reaction was carried out with α,α′-disubstituted vinyl sulfoxonium ylide, the reaction failed to produce the desired vinyl cyclopropane 4al under optimized conditions. **Communication**
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Take and $\left\{\begin{array}{ccc} \text{C}_0\&\text{C}_2\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text{C}_3\&\text$

To expand the scope of the cyclopropanation, we performed the reaction using easily accessible chalcone derivatives, which can provide cyclopropanes with three stereogenic centers (Scheme 4). Surprisingly, the optimized acid-catalyzed conditions for the multicomponent reaction (Table 1) were unable to provide the desired cyclopropanes. However, using cesium carbonate as a base (30 mol%) afforded the desired product 6a in good yield and excellent stereoselectivity.¹⁹ The relative stereochemistry of the three stereogenic centers of 6a was confirmed by X-ray crystallography (CCDC 2349131†). Next, under these base-catalyzed conditions, various substituted chalcones with different electronic properties were investigated, which successfully produced cyclopropanes, 6b–6c, in good yield (76–78%) with excellent selectivity (dr $> 20:1$).

Further efforts were made to develop the scope of ylides as substrates by varying their ester moiety. The ylide containing ester functional groups, including cyclohexyl (6d) and L-menthol (6e), gave the desired cyclopropanes.

Notably, the base-catalyzed annulation (Scheme 4) reported here requires a longer reaction time compared to acid-catalyzed cyclopropanation (Scheme 3).

To investigate the mechanism for the multicomponent reaction, we treated the vinyl sulfoxonium ylide 1a with benzaldehyde in the presence of the L-proline catalyst. No conversion of the ylide was observed (Scheme 5a).

The present multicomponent reaction using 5.3 mmol of vinyl sulfoxonium ylide successfully afforded the vinylcyclopropane 4j in an 85% yield (Scheme 5b). Next, we performed

Scheme 3 Scope of cyclopropanation reaction involving ylide, aldehyde, and indane-1,3-dione. Reaction conditions: 1 (0.42 mmol), 2 (0.33 mmol), 3 (0.30 mmol), proline (20 mol%), methanol (0.1 M), 30 minutes at rt. The diastereomeric ratio (>20 : 1 dr, until and unless mentioned) was assigned by ¹H NMR analysis.

gram-scale synthesis using vinyl sulfoxonium ylide 1a with arylidene 7 in the absence of the catalyst (Scheme 5c). The reaction afforded 89% of 4j without loss of selectivity. This observation indicates that the proline catalyst is necessary for the generation of the arylidene intermediate from indane 1,3 dione with aldehyde, but it is not required for $(2 + 1)$ annulation.20 To further explore the utility of the vinyl cyclopropane, we performed selective ketone reduction in cyclopropane 4j using LiAlH₄, which gave the product 8 in 72% yield without cleavage of the strained cyclopropane ring (Scheme 5c).

Next, we focused on the investigation of the regioselective electrophilic addition of vinyl sulfoxonium ylide on arylidenes. In our previous reports, vinyl sulfoxonium ylide underwent Michael addition selectively from the γ-position with benzoquinone and propiolate (as shown in Scheme 1b). In contrast to the previous reports, the present Michael addition selectively undergoes at the α -position of the ylide.

To rationalize this observation, we carried out a noncovalent interaction (NCI) analysis of the transition states using NCIplot software (Fig. 1). Interestingly, we observed that the methyl groups of the DMSO direct the regioselectivity of the ylide for the incoming electrophile. In the case of benzoquinone and propiolate as Michael acceptors, the methyl groups of DMSO exhibit a weak electrostatic attraction with the carbonyl group of electrophiles. Thus, the γ-carbon of the ylide remains in the vicinity of the β-carbon of the electrophile. This interaction favours the Michael addition from the γ -carbon of the ylide. We explored a similar reactivity with the arylidene generated from indane-1,3-dione and aldehyde. In the case of arylidene, the stable transition state possesses predominant C–H– π -interaction²¹ and electrostatic attraction between the two reactants. These interactions bring the β-carbon of electrophile close to the α-carbon of the ylide. Thus, in the present case, Michael addition occurs from the α-carbon of the ylide. In addition, we observed that the tran-

Scheme 4 Scope of cyclopropanation reaction involving ylide and chalcone derivatives. Reaction conditions: 1 (0.4 mmol), 5 (0.3 mmol), and CS_2CO_3 (30 mol%) in MeOH (0.1 M) under argon atmosphere at rt. The diastereomeric ratio $(>20:1$ dr, until and unless mentioned) was assigned by ${}^{1}\mathsf{H}$ NMR analysis.

Scheme 5 Gram-scale synthesis and synthetic utility of cyclopropanation.

sition state corresponding to the γ -addition of ylide to the arylidine was less stable by 4.8 kcal mol⁻¹ than the α-addition transition state (see the ESI for details†).

Fig. 1 Non-covalent interaction analyses of Michael addition transition states.

Then we turned our attention to investigating the selective reaction between nucleophilic reactants, such as sulfoxonium ylide and indane 1,3-dione, with electrophilic partners, such as aldehyde and in situ generated arylidene. In the present one-pot reaction, indane 1,3-dione selectively reacts with aldehyde, whereas in situ generated arylidene undergoes Michael addition with vinyl sulfoxonium ylide.

To rationalize this observation, we analyzed the HOMO– LUMO energy gap between the nucleophiles and electrophiles (Fig. 2). The results revealed that the energy gap between the HOMO of indane-1,3-dione and LUMO of aldehyde is less (1.68 eV) than the HOMO of ylide and LUMO of aldehyde (5.50 eV). This favours the path-C (Scheme 2) for the generation of arylidene intermediate. This arylidene can undergo Michael addition with either vinyl sulfoxonium ylide or indane-1,3 dione. Although the HOMO–LUMO energy gap is small between indane-1,3-dione and arylidene, experimentally, the vinyl sulfoxonium ylide attack generated cyclopropane. This could be due to the complete consumption of indane-1,3 dione for the generation of arylidene or the retro Michael reaction between indane-1,3-dione and arylidene. 22

Further, DFT studies were performed at the M06-2X|def2- TZVPP|SMD (methanol) |M06-2X|def2-SVP|SMD (methanol) level of theory to provide more details on the mechanism of the present multi-component reaction (Scheme 6). Based on experimental observations and the HOMO–LUMO energy gap between nucleophiles and electrophiles (Fig. 2), it appears that

Fig. 2 HOMO–LUMO energy gap of nucleophiles and electrophiles.

proline-catalyzed Knoevenagel condensation initially takes place between aldehydes and indane 1,3-diones. Based on the previous observations, proline may act as a facilitator for condensation under protic solvents, rather than as a reactant to generate iminium ion species.²³ Therefore, in the presence of a proline catalyst, indane-1,3-dione reacts with aldehyde to produce int-1. Further, the counter anion present in int-1 abstracts the acidic proton to generate the enolate ion int-2 via TS-2. Later, dehydration of int-2 generates the arylidene int-3. The reaction energy profile shows that dehydration is the ratelimiting step with a 15.2 kcal mol^{-1} energy barrier. As this is the slowest step in the reaction, indane-1,3-dione will be completely consumed to generate int-1 before generating arylidene, ruling out the possibility of indane-1,3-dione reaction with the arylidene (as previously explained with HOMO–LUMO energy gap in Fig. 2). Arylidene int-3 as Michael acceptor reacts with vinyl sulfoxonium ylide in four possible ways: (i) Si-face of ylide to Re-face of int-3 ($Si-Re$), (ii) Si -face of ylide to Si -face of int-3 $(Si-Si)$, (iii) Re-face of ylide to Re-face of int-3 $(Re-Re)$, (iv) Re -face of ylide to Si-face of int-3 ($Re-Si$). Among these possibilities, interaction (i) and (ii) generate diastereomers, whereas (iii) and (iv) generate enantiomers of (i) and (ii). The transition state corresponding to the $Si-Re$ face attack (TS-4a) is better stabilized than the Si-Si face attack (TS-4b) by 1.9 kcal mol⁻¹. Thus, the int-3 preferably generates the int-4a. Finally, the cyclopropanation of int-4 occurs through the elimination of the DMSO molecule through TS-5a or TS-5b to generate transor cis-cyclopropanes, respectively. Here, the TS-5a is more **Example distribution** $\frac{1}{2}$ Published on 2022. The between the contents of the state o

Scheme 6 DFT computed the energy profile (in kcal mol⁻¹) for the multicomponent reaction between vinyl sulfoxonium ylide, indane 1,3-dione, and aldehyde.

Fig. 3 NCI analyses of stereoselectivity determining transition states, (a) TS-4 and (b) TS-5.

stabilized than TS-5b by 4.1 kcal mol⁻¹. Thus, the reaction proceeds through TS-5a to deliver trans-cyclopropane as an exclusive product.

We performed an NCI analysis to further understand the stability of TS-4a and TS-5a over TS-4b and TS-5b. As highlighted in Fig. 3, TS-4a is stabilized due to the presence of (i) C–H–π-interaction between the methyl group of DMSO and phenyl group of aldehydes, (ii) C–H–π-interaction between the methyl group of DMSO and phenyl group of ylide, and (iii) weak electrostatic attraction between oxygen of indane-1,3 dione with alkyl group of ylide. On the other hand, TS-4b possesses (i) the C–H–π-interaction between the alkyl group of ester and phenyl group of indane-1,3-dione and (ii) the electrostatic attraction between hydrogen of DMSO and oxygen of indane-1,3-dione. Similarly, in the case of TS-5a, (i) an electrostatic interaction develops between the C–H of ylidic carbon and oxygen of indane-1,3-dione during the cyclopropanation step and (ii) a C-H- π -interaction develops between the methyl group of DMSO and phenyl group of aldehydes. On the other hand, TS-5b possesses (i) a weak π - π stacking as a major stabilizing factor and (ii) C–H–π-interaction between the ester group of ylide and phenyl group of indane-1,3-dione. Thus, TS-4a and TS-5a are more stable than TS-4b and TS-5b. **Communication**

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Conclusions

In summary, we have developed stereoselective cyclopropanation using vinyl sulfoxonium ylides with arylidenes under acidic and basic conditions. The present multicomponent reaction selectively proceeds because of the difference in the nucleophilic reactivity of vinyl sulfoxonium ylide and indane 1,3-dione with electrophilic precursors, such as aldehydes and in situ generated arylidenes. This method affords various complex cyclopropane scaffolds in good yields with excellent diastereoselectivity under mild reaction conditions. The origin of excellent regioselectivity (addition at α- vs. γ-position of vinyl sulfoxonium ylide) and diastereoselectivity (cis/trans selectivity of cyclopropanation) was revealed by theoretical studies on non-covalent interactions.

General procedure for the synthesis of spiro[cyclopropane-1,2′ indene]-1′,3′-dione

1,3-Indanedione 3 (0.30 mmol, 1 equiv.) and aldehyde 2 (0.33 mmol, 1.1 equiv.) solutions in methanol (0.1 M) were added to an oven-dried, 25 mL round bottom flask equipped with a magnetic stir bar. A catalytic amount of L -proline (7 mg, 20 mol%) and ylide 1 (0.42 mmol, 1.4 equiv.) was added to the stirring solution. After stirring for 30 minutes at room temperature, spiro[cyclopropane-1,2′-indene]-1′,3′-dione precipitated at the bottom of the round bottom flask. The precipitate was filtered and washed with methanol to yield a chromatographically and spectroscopically pure product.

General procedure for the synthesis of (2-benzoyl-3 phenylcyclopropyl)-3-phenylacrylate

α,β-Unsaturated ketone 5 (0.30 mmol, 1 equiv.) solution in methanol (0.1 M) was added to an oven-dried, Ar-purged 25 mL round bottom flask equipped with a magnetic stir bar under argon atmosphere. A catalytic amount of cesium carbonate (28 mg, 30 mol%) and ylide 1 (0.40 mmol, 1.4 equiv.) were added to the stirring solution. After stirring for 2 hours at room temperature, the reaction mixture was concentrated under reduced pressure. The residue was purified using flash column chromatography (Biotage flash chromatography gradient purification: EtOAc/*n*-hexane = $1:5$) to give the product 6.

Ethyl (E)-3-(1′,3′-dioxo-2-phenyl-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4a). The precipitated product was purified with cold methanol, yield: 113 mg (87%); $dr > 20$: 1; yellow solid.

IR (neat, cm−¹): 3432, 3059, 2982, 1710, 1620, 1495, 1446, 1337, 1287, 1219, 1174, 1095, 1039, 875, 697, 617, 562, 534, 483. ¹H NMR (500 MHz, Chloroform-D) δ 7.86 (s, 2H), 7.77–7.67 (m, 2H), 7.55–7.48 (m, 2H), 7.42–7.31 (m, 7H), 7.31–7.27 (m, 1H), 6.21 (s, 1H), 4.29 (d, $J = 9.6$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.40 (d, $J = 9.6$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.5, 194.2, 164.9, 152.6, 142.0, 141.2, 140.3, 134.3, 134.0, 133.1, 128.7, 128.6, 128.1, 127.9, 127.4, 127.3, 122.3, 121.7, 121.7, 59.7, 50.5, 45.6, 37.9, 13.5. HRMS (ESI) m/z : $[M + Na]$ ⁺ calculated for C28H22O4Na, 445.1410; found 445.1412.

Ethyl (E)-3-(1′,3′-dioxo-2-(p-tolyl)-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4b). The precipitated product was purified with cold methanol, yield: 118 mg (90%); $dr > 20:1$; yellow solid.

IR (neat, cm−¹): 3432, 2925, 2857, 1710, 1620, 1518, 1448, 1339, 1289, 1220, 1174, 1092, 1039, 876, 813, 774, 744, 699, 614, 562, 535, 485. ¹H NMR (500 MHz, Chloroform-D) δ 7.87 $(d, J = 1.9 \text{ Hz}, 2\text{H}), 7.72 \text{ (dd, } J = 6.0, 2.9 \text{ Hz}, 2\text{H}), 7.54 \text{ (t, } J = 3.6 \text{ Hz})$ Hz, 2H), 7.40–7.33 (m, 3H), 7.25–7.19 (m, 3H), 7.12 (s, 1H), 6.22 (s, 1H), 4.31 (d, $J = 9.6$ Hz, 1H), 3.85 (q, $J = 7.1$ Hz, 2H), 3.40 (d, $J = 9.6$ Hz, 1H), 2.35 (s, 3H), 1.00 (t, $J = 7.1$ Hz, 3H).¹³C NMR (101 MHz, Chloroform-D) δ 196.0, 194.7, 165.3, 153.1, 142.4, 141.6, 140.7, 137.5, 134.6, 134.3, 130.4, 129.0, 128.9, 128.9, 128.4, 127.6, 122.6, 122.1, 122.0, 60.0, 51.0, 46.0, 38.2,

Ethyl (E)-3-(1′,3′-dioxo-2-(m-tolyl)-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4c). The precipitated product was purified with cold methanol, yield: 117 mg (89%); $dr > 20$: 1; white solid.

IR (neat, cm−¹): 3431, 2925, 2857, 1711, 1620, 1518, 1448, 1339, 1289, 1220, 1174, 1092, 1039, 876, 813, 774, 744, 700, 614, 562, 535, 485. ¹H NMR (400 MHz, Chloroform-D) δ 7.86 $(s, 2H)$, 7.74–7.67 (m, 2H), 7.53 (s, 2H), 7.36 (dd, $J = 5.1$, 2.0 Hz, 3H), 7.24–7.16 (m, 3H), 7.11 (s, 1H), 6.22 (s, 1H), 4.31 (d, J $= 9.6$ Hz, 1H), 3.85 (q, J = 7.1, 2H), 3.39 (d, J = 9.6 Hz, 1H), 2.34 (s, 3H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.0, 194.7, 165.4, 153.2, 142.5, 141.7, 140.7, 137.8, 134.7, 134.3, 133.4, 129.8, 129.0, 128.7, 128.5, 128.2, 127.7, 126.1, 122.7, 122.1, 122.1, 60.1, 51.0, 46.1, 38.3, 21.4, 13.9. HRMS (ESI) m/z : $[M + Na]^+$ calculated for $C_{29}H_{24}O_4Na$, 459.1567; found 459.1564.

Ethyl (E) -3-(2-(2-bromophenyl)-1',3'-dioxo-1',3'-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4d). The precipitated product was purified with cold methanol, yield: 121 mg (80%); dr > 20 : 1; yellow solid.

IR (neat, cm−¹): 3432, 3060, 2982, 1710, 1622, 1598, 1446, 1407, 1341, 1278, 1220, 1175, 1094, 1039, 880, 767, 736, 698, 622, 567, 496. ¹H NMR (500 MHz, Chloroform-D) δ 7.87 (s, 2H), 7.78–7.69 (m, 2H), 7.54–7.45 (m, 2H), 7.42–7.27 (m, 7H), 6.21 (d, $J = 2.1$ Hz, 1H), 4.24 (d, $J = 9.5$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.33 (d, $J = 9.5$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.6, 194.6, 165.4, 152.6, 142.4, 141.7, 140.6, 135.9, 134.9, 134.6, 132.1, 131.0, 129.8, 129.1, 128.6, 127.7, 127.7, 122.9, 122.4, 122.3, 122.3, 60.2, 50.6, 44.8, 38.2, 14.0. HRMS (ESI) m/z : $[M + Na]$ ⁺ calculated for C28H21O4BrNa, 523.0515; found 523.0515.

Ethyl (E)-3-(2-(3-bromophenyl)-1′,3′-dioxo-1′,3′-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4e). The precipitated product was purified with cold methanol, yield: 126 mg (83%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3432, 3060, 2982, 1710, 1621, 1598, 1446, 1407, 1341, 1278, 1218, 1175, 1094, 1039, 880, 767, 736, 698, 624, 567, 496. ¹H NMR (500 MHz, Chloroform-D) δ 7.90-7.81 (m, 2H), 7.76–7.67 (m, 2H), 7.53–7.47 (m, 2H), 7.41–7.31 (m, 5H), 7.02 (t, $J = 8.5$ Hz, 2H), 6.21 (s, 1H), 4.25 (d, $J = 9.6$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.36 (d, $J = 9.6$ Hz, 1H), 0.99 (t, $J =$ 7.1 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.1, 194.6, 165.4, 152.1, 147.4, 142.3, 141.7, 141.3, 140.5, 135.1, 134.9, 130.0, 129.2, 128.7, 128.6, 127.6, 123.9, 123.5, 122.9, 122.5, 122.4, 60.3, 50.6, 44.2, 38.5, 14.0. **HRMS (ESI)** m/z : $[M + Na]$ ⁺ calculated for $C_{28}H_{21}O_4BrNa$, 523.0515; found 523.0513.

Ethyl (E)-3-(2-(4-chlorophenyl)-1′,3′-dioxo-1′,3′-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4f). The precipitated product was purified with cold methanol, yield: 118 mg (86%); dr > 20 : 1; yellow solid.

IR (neat, cm−¹): 3427, 3060, 2987, 1708, 1605, 1494, 1446, 1400, 1339, 1294, 1214, 1165, 1091, 1036, 1016, 872, 817, 775, 735, 701, 622, 563, 530, 483. ¹H NMR (500 MHz, Chloroform-D) δ 7.87 (s, 2H), 7.78–7.69 (m, 2H), 7.54–7.45 (m, 2H), 7.42–7.27 (m, 7H), 6.21 (d, $J = 2.1$ Hz, 1H), 4.24 (d, $J = 9.5$ Hz, 1H), 3.84 $(q, J = 7.1$ Hz, 2H), 3.33 $(d, J = 9.5$ Hz, 1H), 0.99 $(t, J = 1)$ 7.1 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.7, 194.8, 165.4, 152.7, 142.4, 141.7, 140.7, 134.9, 134.6, 133.7, 132.1, 130.4, 129.1, 128.6, 128.6, 127.6, 122.8, 122.3, 60.2, 50.7, 45.1, 38.4, 14.0. HRMS (ESI) m/z : $[M + Na]$ ⁺ calculated for $C_{28}H_{21}O_4C$ Na, 479.1021; found 479.1018.

Ethyl (E)-3-(2-(4-fluorophenyl)-1′,3′-dioxo-1′,3′-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4g). The precipitated product was purified with cold methanol, yield: 113 mg (85%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3431, 3059, 2981, 2928, 2856, 1710, 1604, 1514, 1448, 1340, 1288, 1223, 1175, 1088, 1038, 878, 828, 738, 699, 612, 537. ¹H NMR (500 MHz, Chloroform-D) δ 7.90–7.81 (m, 2H), 7.76–7.67 (m, 2H), 7.53–7.47 (m, 2H), 7.41–7.31 (m, 5H), 7.02 (t, $J = 8.5$ Hz, 2H), 6.21 (s, 1H), 4.25 (d, $J = 9.6$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.36 (d, J = 9.6 Hz, 1H), 0.99 (t, J = 7.1 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 194.9, 165.4, 161.1, 142.1 $(d, J = 68.2 \text{ Hz})$, 140.7, 134.7 $(d, J = 32.6 \text{ Hz})$, 130.7 $(d, J = 12.6 \text{ Hz})$ 8.1 Hz), 129.1, 128.6, 127.7, 122.8, 122.3, 115.4 (d, J = 21.7 Hz), 60.2, 50.8, 45.2, 38.6, 14.0. ¹⁹F NMR (376 MHz, Chloroform-D) δ -113.85. HRMS (ESI) m/z : [M + Na]⁺ calculated for $C_{28}H_{21}O_4$ FNa, 463.1316; found 463.1313. Organic & Biornolecular Chemistry
 $24.2, 13.9$. HMMS (850) $m\bar{x}$. [M + Na]⁻ calculated for $7.42-7.27$ [m, 7ril, 621 (d, $j = 2.1$ H, 142) (d, $j = 3.4$ Hg, $m\bar{x}$ [M₁, 023 (d, $j = 3.4$ M, 142) (d, $j = 3.4$ M, 142)

Ethyl (E) -3-(1',3'-dioxo-2-(2,3,4-trifluorophenyl)-1',3'-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4h). The precipitated product was purified with cold methanol, yield: 115 mg (80%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3436, 3059, 2981, 2928, 2856, 1711, 1604, 1514, 1448, 1340, 1288, 1223, 1175, 1088, 1038, 878, 828, 738, 699, 612, 537. ¹H NMR (500 MHz, Chloroform-D) δ 7.85 (s, 2H), 7.79–7.72 (m, 2H), 7.50 (dd, J = 7.5, 2.0 Hz, 2H), 7.38 (dd, $J = 5.0, 2.5$ Hz, 3H), 7.31 (q, $J = 8.1, 7.3$ Hz, 1H), 7.08-6.99 (m, 1H), 6.21 (s, 1H), 4.12 (d, $J = 9.3$ Hz, 1H), 3.85 (q, $J = 7.1$ Hz, 2H), 3.28 (d, $J = 9.3$ Hz, 1H), 1.00 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.4, 194.8, 165.5, 152.4, 142.0 (d, $J = 31.1$ Hz), 140.7, 134.9 (d, $J = 23.9$ Hz), 129.2, 128.7, 127.7, 123.0, 122.9 (d, $J = 22.6$ Hz), 60.4, 48.9, 37.9, 29.8, 14.1. ¹⁹F NMR (376 MHz, Chloroform-D) δ −133.98 (dt, J = 21.9, 7.7 Hz), -135.88 (dt, $J = 20.8$, 7.4 Hz), -160.07 (td, $J = 20.6$, 7.0 Hz). **HRMS** (ESI) m/z : $[M + Na]$ ⁺ calculated for C₂₈H₁₉O₄F₃Na, 499.1128; found 499.1125.

Ethyl (E) -3- $(1',3'-dioxo-2-(4-(trifluoromethoxy)phenyl)-1',3'-d'$ dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4i). The precipitated product with purified cold methanol, yield: 125 mg (82%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3436, 2927, 1711, 1621, 1447, 1409, 1326, 1291, 1220, 1171, 1123, 1069, 1037, 877, 772, 744, 698, 609, 531. ¹H NMR (500 MHz, Chloroform-D) δ 7.87 (s, 2H), 7.78–7.69 (m, 2H), 7.54–7.45 (m, 2H), 7.42–7.27 (m, 7H), 6.21 $(d, J = 2.1$ Hz, 1H), 4.24 $(d, J = 9.5$ Hz, 1H), 3.84 $(q, J = 7.1$ Hz, 2H), 3.33 (d, $J = 9.5$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (126 MHz, Chloroform-D) δ 195.6, 194.8, 165.5, 152.6, 142.1 (d, $J = 85.0$ Hz), 140.7, 134.9 (d, $J = 39.6$ Hz), 129.5, 129.2, 128.2 (d, $J = 126.5$ Hz), 125.4 (q, $J = 3.7$ Hz), 123.0, 122.5 (d, $J = 9.2$ Hz).60.3, 50.7, 44.9, 38.5, 14.1. ¹⁹F NMR (376 MHz, Chloroform-D) δ –62.42. HRMS (ESI) m/z : [M + Na]⁺ calculated for $C_{29}H_{21}O_5F_3Na$, 529.1233; found 529.1229.

Ethyl (E) -3-(2-(4-methoxyphenyl)-1',3'-dioxo-1',3'-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4j). The precipitated product was purified with cold methanol, yield: 124 mg (91%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3423, 3051, 2981, 2960, 2935, 2838, 1710, 1614, 1519, 1445, 1331, 1274, 1254, 1226, 1177, 1087, 1034, 869, 824, 775, 747, 696, 565, 541, 489. ¹H NMR (500 MHz, Chloroform-D) δ ¹H NMR (500 MHz, Chloroform-D) δ 7.90–7.81 (m, 2H), 7.76–7.68 (m, 2H), 7.54–7.49 (m, 2H), 7.39–7.29 (m, 5H), 6.90–6.84 (m, 2H), 6.21 (d, $J = 2.2$ Hz, 1H), 4.26 (d, $J = 9.6$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.80 (s, 3H), 3.35 (d, $J = 9.6$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 194.8, 165.3, 159.1, 153.1, 142.3, 141.6, 140.7, 134.6, 134.3, 130.1, 128.9, 128.4, 127.6, 125.3, 122.6, 122.0, 122.0, 113.7, 60.0, 55.1, 51.1, 46.0, 38.4, 13.9. HRMS (ESI) m/z : [M + Na]⁺ calculated for C₂₉H₂₄O₅Na, 475.1516; found 475.1517. Communication

Chloroform-19) δ -62.42. H1088 (kS1) $m\bar{x}$ [M + Nel] calculated on 10, 1084, 1283, 309, 1293, 1294, 1295, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294, 1294,

Ethyl (E)-3-(2-(3-methoxyphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4k). The precipitated product was purified with cold methanol, yield: 120 mg (88%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3423, 3051, 2981, 2960, 2935, 2838, 1710, 1614, 1519, 1445, 1331, 1274, 1254, 1226, 1177, 1087, 1034, 869, 824, 775, 747, 696, 565, 541, 489. ¹H NMR (500 MHz, Chloroform-D) δ 7.89–7.81 (m, 2H), 7.74–7.69 (m, 2H), 7.54–7.49 $(m, 2H)$, 7.36 $(dd, J = 5.2, 2.0 Hz, 3H)$, 7.33–7.30 $(m, 2H)$, 6.87 (d, J) $= 8.7$ Hz, 2H), 6.20 (d, $J = 2.2$ Hz, 1H), 4.25 (d, $J = 9.7$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, $2H$), 3.80 (s, $3H$), 3.35 (d, $J = 9.7$ Hz, $1H$), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 194.8, 165.3, 159.1, 153.1, 142.3, 141.6, 140.7, 134.6, 134.3, 130.1, 128.9, 128.4, 127.6, 125.3, 122.6, 122.0, 122.0, 113.7, 60.0, 55.1, 51.1, 46.0, 38.4, 13.9. HRMS (ESI) m/z : $[M + Na]$ ⁺ calculated for $C_{29}H_{24}O_5$ Na, 475.1516; found 475.1516.

Ethyl (E) -3-(2-(3,4-dimethoxyphenyl)-1',3'-dioxo-1',3'-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4l). The precipitated product was purified with cold methanol, yield: 135 mg (93%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3420, 3071, 3009, 2932, 2841, 1710, 1623, 1596, 1523, 1465, 1444, 1334, 1305, 1272, 1238, 1214, 1177, 1091, 1040, 873, 809, 768, 740, 703, 634, 500, 634, 463. ¹H NMR (500 MHz, Chloroform-D) δ 7.87–7.78 (m, 2H), 7.74–7.64 $(m, 2H), 7.52-7.47$ $(m, 2H), 7.33$ $(d, J = 1.5 Hz, 3H), 6.98$ $(d, J = 1.5 Hz)$ 8.3 Hz, 1H), 6.91 (s, 1H), 6.86–6.81 (m, 1H), 6.20 (s, 1H), 4.26 $(d, J = 9.6 \text{ Hz}, 1H), 3.86-3.79 \text{ (m, 8H)}, 3.33 \text{ (d, } J = 9.6 \text{ Hz}, 1H),$ 0.96 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 194.8, 165.3, 153.0, 148.6, 148.5, 142.4, 141.6, 140.6, 134.6, 134.3, 128.9, 128.4, 127.6, 125.8, 122.6, 122.0, 121.9, 121.3, 112.0, 110.6, 60.0, 55.8, 55.7, 51.3, 46.5, 38.5, 13.9. **HRMS** (ESI) m/z : $[M + Na]^+$ calculated for $C_{30}H_{26}O_6Na$, 505.1622; found 505.1624.

Ethyl (E) -3-(2-(2-nitrophenyl)-1',3'-dioxo-1',3'-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4m). The precipitated product was purified with cold methanol, yield: 111 mg (79%); dr > 20 : 1; yellow solid.

IR (neat, cm−¹): 3434, 3061, 2984, 1712, 1601, 1521, 1448, 1404, 1344, 1288, 1220, 1178, 1090, 1037, 920, 878, 852, 770, 735, 699, 631, 562, 533. ¹H NMR (500 MHz, Chloroform-D) δ 8.18 (d, $J = 8.7$ Hz, 2H), 7.86 (s, 2H), 7.76 (dd, $J = 5.9$, 2.6 Hz, 2H), 7.58 (d, $J = 8.5$ Hz, 2H), 7.47 (dd, $J = 6.7$, 3.0 Hz, 2H), 7.38–7.32 (m, 3H), 6.22 (s, 1H), 4.29 (d, $J = 9.5$ Hz, 1H), 3.83 (q, $J = 7.1$ Hz, 2H), 3.41 (d, $J = 9.5$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.1, 194.6, 165.4, 152.1, 147.4, 142.3, 141.7, 141.3, 140.5, 135.1, 134.9, 130.0, 129.2, 128.7, 128.6, 127.6, 123.9, 123.5, 122.9, 122.5, 122.4, 60.3, 50.6, 44.2, 38.5, 14.0. HRMS (ESI) m/z : [M + Na]⁺ calculated for $C_{28}H_{21}O_6NNa$, 490.1261; found 490.1263.

Ethyl (E) -3-(2-(3-nitrophenyl)-1',3'-dioxo-1',3'-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4n). The precipitated product was purified with cold methanol, yield: 114 mg (81%); dr > 20 : 1; yellow solid.

IR (neat, cm−¹): 3431, 3061, 2984, 1711, 1601, 1521, 1448, 1404, 1344, 1288, 1220, 1178, 1090, 1037, 920, 878, 852, 770, 735, 699, 631, 562, 533. ¹H NMR (500 MHz, Chloroform-D) δ 7.90–7.81 (m, 2H), 7.76–7.67 (m, 2H), 7.53–7.47 (m, 2H), 7.41–7.31 (m, 5H), 7.02 (t, $J = 8.5$ Hz, 2H), 6.21 (s, 1H), 4.25 (d, $J = 9.6$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.36 (d, $J = 9.6$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (126 MHz, Chloroform-D) δ 195.2, 194.8, 165.4, 152.2, 148.3, 142.3, 141.8, 140.6, 135.9, 135.2, 135.1, 134.9, 129.3, 129.2, 128.7, 127.6, 124.2, 123.0, 123.0, 122.6, 122.5, 60.3, 50.3, 44.1, 38.5, 14.0. HRMS (ESI) m/ z: $[M + Na]^{+}$ calculated for $C_{28}H_{21}O_6NNa$, 490.1261; found 490.1260.

Ethyl (E) -3-(2-(4-nitrophenyl)-1',3'-dioxo-1',3'-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4o). The precipitated product was purified with cold methanol, yield: 117 mg (83%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3432, 3061, 2984, 1710, 1601, 1521, 1448, 1404, 1344, 1288, 1220, 1178, 1090, 1037, 920, 878, 852, 770, 735, 699, 631, 562, 533. ¹H NMR (500 MHz, Chloroform-D) δ 8.18 (d, $J = 8.7$ Hz, 2H), 7.86 (s, 2H), 7.76 (dd, $J = 5.9$, 2.6 Hz, 2H), 7.58 (d, $J = 8.5$ Hz, 2H), 7.47 (dd, $J = 6.7$, 3.0 Hz, 2H), 7.38–7.32 (m, 3H), 6.22 (s, 1H), 4.29 (d, $J = 9.5$ Hz, 1H), 3.83 (q, $J = 7.1$ Hz, 2H), 3.41 (d, $J = 9.5$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.1, 194.6, 165.4, 152.1, 147.4, 142.3, 141.7, 141.3, 140.5, 135.1, 134.9, 130.0, 129.2, 128.7, 128.6, 127.6, 123.9, 123.5, 122.9, 122.5, 122.4, 60.3, 50.6, 44.2, 38.5, 14.0. **HRMS** (ESI) m/z : $[M + Na]^{+}$ calculated for C₂₈H₂₁O₆NNa, 490.1261; found 490.1253.

Ethyl (E) -3-(1',3'-dioxo-2-(4-(trifluoromethyl)phenyl)-1',3'dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4p). The precipitated product was purified with cold methanol, yield: 124 mg (84%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3435, 2927, 1711, 1621, 1447, 1409, 1326, 1291, 1220, 1171, 1123, 1069, 1037, 877, 772, 744, 698, 609, 531. ¹H NMR (500 MHz, Chloroform-D) δ 7.88 (s, 2H), 7.79–7.70 (m, 2H), 7.59 (s, 2H), 7.54 (s, 4H), 7.38–7.34 (m, 3H), 6.23 (s, 1H), 4.30 (d, $J = 9.5$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.40 (d, $J = 9.5$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.5, 194.7, 165.4, 152.5, 142.0 (d, $J = 67.9$ Hz), 140.6, 134.8 (d, $J = 31.1$ Hz), 129.5, 129.2, 128.1 (d, $J = 100.7$ Hz), 125.3 (q, $J = 3.9$ Hz), 122.9, 122.4 (d, $J = 2.9$ Hz),60.2, 50.6, 44.8, 38.4, 14.0. ¹⁹F NMR (376 MHz, Chloroform-D) δ –62.35. HRMS (ESI) m/z : [M + Na]⁺ calculated for $C_{29}H_{21}O_4F_3Na$, 513.1284; found 513.1290.

Ethyl (E)-3-(2-(naphthalen-2-yl)-1′,3′-dioxo-1′,3′-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4q). The precipitated product was purified with cold methanol, yield: 115 mg (81%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3434, 2926, 1709, 1620, 1446, 1337, 1293, 1271, 1174, 1097, 1028, 867, 777, 738, 698.¹H NMR (500 MHz, Chloroform-D) δ 7.93 (d, J = 7.5 Hz, 1H), 7.88–7.82 (m, 2H), 7.82–7.79 (m, 1H), 7.71 (d, $J = 18.0$ Hz, 3H), 7.62 (d, $J = 7.3$ Hz, 3H), 7.43–7.39 (m, 1H), 7.38–7.28 (m, 4H), 7.18–7.14 (m, 1H), 6.27 (s, 1H), 4.45 (d, $J = 9.5$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.76 (d, $J = 9.5$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.6, 194.2, 165.6, 153.3, 142.2, 140.9, 134.9, 134.4, 133.6, 130.1, 129.1, 129.0, 128.9, 128.7, 127.8, 126.7, 126.6, 125.8, 125.5, 122.9, 122.8, 122.7, 122.3, 60.3, 50.6, 43.5, 38.4, 14.1. **HRMS** (ESI) m/z : [M + Na]⁺ calculated for $C_{32}H_{24}O_4$ Na, 495.1567; found 495.1568.

Ethyl (E) -3-(1',3'-dioxo-2-(thiophen-2-yl)-1',3'-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4r). The precipitated product was purified with cold methanol, yield: 115 mg (89%); dr = 4 : 1; orange solid.

IR (neat, cm−¹): 3432, 3073, 2926, 1709, 1620, 1446, 1337, 1293, 1271, 1174, 1097, 1028, 867, 777, 738, 698. Major product: ¹H NMR (500 MHz, Chloroform-D) δ 7.94–7.90 (m, 1H), 7.86–7.83 (m, 1H), 7.76–7.73 (m, 2H), 7.53 (d, $J = 1.7$ Hz, 2H), 7.40–7.37 (m, 2H), 7.25–7.22 (m, 2H), 7.14 (d, $J = 3.5$ Hz, 1H), 7.01 (dd, $J = 5.2$, 3.5 Hz, 1H), 6.19 (s, 1H), 4.18 (d, $J = 9.3$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.41 (d, $J = 9.3$ Hz, 1H), 0.99 $(t, J = 7.1 \text{ Hz}, 3\text{H})$ ¹³C NMR (101 MHz, Chloroform-D) δ 195.4, 194.0, 165.2, 152.4, 142.3, 141.4, 140.6, 136.7, 134.8, 134.4, 129.0, 128.5, 127.6, 127.4, 127.0, 125.5, 122.8, 122.1, 121.9, 60.1, 50.7, 40.3, 39.8, 13.9. HRMS (ESI) m/z : [M + Na]⁺ calculated for $C_{26}H_{20}O_4$ SNa, 451.0975; found 451.0970.

Ethyl (E) -3- $(2$ -(furan-2-yl)-1',3'-dioxo-1',3'-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4s). The precipitated product was purified with cold methanol, yield: 116 mg $(87%)$; dr = 9 : 1; white solid.

IR (neat, cm−¹): 3436, 2928, 1710, 1621, 1446, 1341, 1293, 1175, 1094, 1037, 879, 775, 739, 698, 598, 525. ¹H NMR (500 MHz, Chloroform-D) δ 7.77 (s, 1H), 7.69 (d, J = 1.6 Hz, 1H), 7.63–7.55 (m, 2H), 7.41–7.37 (m, 2H), 7.26–7.18 (m, 4H), 6.35 (d, $J = 3.3$ Hz, 1H), 6.28–6.21 (m, 1H), 6.05 (d, $J = 2.2$ Hz, 1H), 3.99 (d, $J = 9.3$ Hz, 1H), 3.68 (q, $J = 7.1$ Hz, 2H), 3.08 (d, $J =$ 9.3 Hz, 1H), 0.83 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.4, 194.2, 165.4, 152.2, 148.4, 142.7, 142.4, 141.8, 140.8, 134.9, 134.5, 129.1, 128.6, 127.7, 122.9, 122.3, 122.2, 110.8, 109.4, 60.2, 49.4, 37.6, 37.5, 14.0. HRMS (ESI) m/ z: $[M + Na]^+$ calculated for $C_{26}H_{20}O_5Na$, 435.1203; found 435.1198.

Ethyl (E)-3-(1′,3′-dioxo-2-propyl-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4t). The precipitated product was purified with cold methanol, yield: 99 mg (85%); $dr = 5:1$; white solid.

IR (neat, cm−¹): 3428, 3059, 2961, 2928, 2870, 1709, 1620, 1449, 1368, 1334, 1276, 1209, 1173, 1103, 1038, 878, 769, 746, 699, 612, 562, 531. Major product:¹H NMR (500 MHz, Chloroform-D) δ 7.98 (d, J = 7.4 Hz, 1H), 7.82–7.71 (m, 3H), 7.47–7.42 (m, 2H), 7.37 (dd, $J = 11.6$, 7.2 Hz, 3H), 6.07 (s, 1H), 3.83–3.77 (m, 2H), 3.60 (d, $J = 7.0$ Hz, 1H), 2.16 (dt, $J = 9.1$, 4.5 Hz, 1H), 2.07-1.99 (m, 1H), 1.95 (d, $J = 5.8$ Hz, 1H), 1.25 (s, 2H), 0.96 (t, $J = 7.1$ Hz, 3H), 0.80 (t, $J = 7.4$ Hz, 3H). ¹³C NMR (126 MHz, Chloroform-D) δ 197.3, 197.1, 165.4, 153.8, 142.4, 141.6, 141.4, 134.6, 134.4, 128.8, 128.4, 128.4, 127.8, 126.9, 122.5, 122.1, 122.0, 60.1, 48.4, 43.5, 42.3, 28.6, 22.6, 14.0, 13.7. **HRMS** (ESI) m/z : $[M + Na]^+$ calculated for $C_{25}H_{24}O_4Na$, 411.1567; found 411.1566.

Ethyl (E) -3-(2-(4-methoxyphenyl)-1',3'-dioxo-1',3'-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-(p-tolyl)acrylate (4u). The precipitated product was purified with cold methanol, yield: 125 mg (89%); dr > 20 : 1; yellow solid.

IR (neat, cm−¹): 3429, 2981, 2933, 2838, 1709, 1611, 1515, 1457, 1336, 1290, 1251, 1221, 1171, 1091, 1038, 876, 822, 740, 620, 579, 528, 501. ¹H NMR (400 MHz, Chloroform-D) δ 7.90–7.83 (m, 2H), 7.73–7.67 (m, 2H), 7.43 (d, $J = 8.2$ Hz, 2H), 7.37–7.29 (m, 2H), 7.19–7.13 (m, 2H), 6.88 (d, $J = 8.7$ Hz, 2H), 6.21 (d, $J = 2.1$ Hz, 1H), 4.26 (d, $J = 9.6$, 2.1 Hz, 1H), 3.83 (q, $J =$ 7.1 Hz, 2H), 3.77 (s, 3H), 3.38 (d, $J = 9.6$ Hz, 1H), 2.34 (s, 3H), 0.98 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 194.8, 165.4, 159.1, 153.0, 142.3, 141.6, 139.0, 137.8, 134.6, 134.2, 130.1, 129.1, 127.5, 125.4, 122.5, 122.0, 121.1, 113.6, 59.9, 55.1, 51.1, 46.0, 38.4, 21.2, 13.9. HRMS (ESI) m/z: $[M + Na]$ ⁺ calculated for C₃₀H₂₆O₅Na, 489.1672; found 489.1674. Organic & Biomolecular Chemistry
 $f = 100.7$ Hzi), 123.5 (a, $f = 3.9$ Hz), 122.9, 122.4 (a, $f = 2.9$ Hz (neat, cm⁻¹), 3228, 3293, 2993, 2924, 2379, 1799, 1224

Hz)(a, 266, 448, 384, 146. ¹³ NMC (276 MHz), 1444, 1446,

Ethyl (E)-3-(4-methoxyphenyl)-3-(2-(4-methoxyphenyl)-1′,3′ dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)acrylate (4v). The precipitated product was purified with cold methanol, yield: 131 mg (90%); dr > 20 : 1; yellow solid.

IR (neat, cm−¹): 3429, 2960, 2935, 2838, 1708, 1604, 1514, 1460, 1336, 1292, 1251, 1169, 1091, 1037, 874, 833, 739, 535. ¹H NMR (400 MHz, Chloroform-D) δ 7.89-7.79 (m, 2H), 7.71 $(dd, J = 6.0, 2.8 Hz, 2H$, 7.46 $(d, J = 8.7 Hz, 2H)$, 7.33 $(d, J = 8.6$ Hz, 2H), 6.88 (dd, $J = 8.8$, 2.8 Hz, 4H), 6.16 (s, 1H), 4.22 (d, $J =$ 9.6 Hz, 1H), 3.83-3.74 (m, 8H), 3.38 (d, $J = 9.6$ Hz, 1H), 0.97 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.1, 195.1, 165.6, 160.4, 159.3, 152.7, 142.5, 141.8, 134.7, 134.3, 133.2, 130.2, 129.1, 125.5, 122.7, 122.2, 120.4, 114.0, 113.8, 60.1, 55.4, 55.3, 51.2, 46.2, 38.6, 14.0. HRMS (ESI) m/z: [M + Na]⁺ calculated for C₃₀H₂₆O₆Na, 505.1622; found 505.1621.

Ethyl (E)-3-(3,4-dimethoxyphenyl)-3-(2-(4-methoxyphenyl)- 1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl) acrylate (4w). The precipitated product was purified with cold methanol, yield: 139 mg (92%); dr > 20 : 1; yellow solid.

IR (neat, cm−¹): 3427, 3048, 2943, 2841, 2037, 1706, 1601, 1518, 1463, 1338, 1259, 1224, 1161, 1031, 844, 812, 750, 561, 539. ¹H NMR (500 MHz, Chloroform-D) δ 7.88-7.80 (m, 2H), 7.73–7.66 (m, 2H), 7.11 (d, $I = 2.1$ Hz, 1H), 7.01 (d, $I = 2.2$ Hz, 1H), 6.89–6.81 (m, 3H), 6.20 (d, $J = 2.2$ Hz, 1H), 4.18 (d, $J = 9.7$ Hz, 1H), 3.87 (s, 3H), 3.83–3.79 (m, 2H), 3.78 (s, 3H), 3.71 (s, 3H), 3.36 (d, $J = 9.7$ Hz, 1H), 0.97 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 195.0, 165.4, 159.1, 152.3, 149.8, 148.6, 142.3, 141.6, 134.6, 134.2, 133.3, 130.0, 125.3, 122.5, 122.0, 120.3, 120.1, 113.6, 110.8, 110.6, 59.9, 55.8, 55.6, 55.1, 51.1, 46.3, 38.3, 13.9. HRMS (ESI) m/z : [M + Na]⁺ calculated for $C_{31}H_{28}O_7$ Na, 535.727; found 535.720.

Ethyl (E) -3-(3-methoxyphenyl)-3-(2-(4-methoxyphenyl)-1′,3′dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)acrylate (4x). The precipitated product was purified with cold methanol, yield: 125 mg (86%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3422, 2999, 2967, 2930, 2838, 1709, 1613, 1578, 1515, 1490, 1468, 1338, 1286, 1217, 1182, 1092, 1042, 871, 835, 785, 750, 689, 567, 541, 468. ¹H NMR (500 MHz, Chloroform-D) δ 7.88-7.78 (m, 2H), 7.70 (dd, J = 5.9, 2.6 Hz, 2H), 7.31 (d, $J = 8.5$ Hz, 2H), 7.24 (s, 1H), 7.09 (d, $J = 7.7$ Hz, 1H), 6.85 (d, $J = 8.5$ Hz, 3H), 6.22 (s, 1H), 4.21 (d, $J = 9.7$ Hz, 1H), 3.81 (q, $J = 7.1$ Hz, 2H), 3.77 (s, 3H), 3.70 (s, 3H), 3.35 (d, J $= 9.7$ Hz, 1H), 0.97 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 194.9, 165.4, 159.5, 159.2, 152.8, 142.4, 142.1, 141.7, 134.7, 134.3, 130.1, 129.5, 125.4, 122.6, 122.1, 122.0, 120.0, 115.1, 113.7, 112.7, 60.1, 55.2, 51.1, 46.1, 38.3, 13.9. HRMS (ESI) m/z : $[M + Na]^+$ calculated for $C_{30}H_{26}O_6Na$, 505.1622; found 505.1619. Communication

(101 MHz, Chloroform19) δ 1953, 1953, 1553, 1553, 1532, Mechyl (E)-3-{1,3⁻²diono-2-phenyl-1,3⁻²dibydrospino(yelopno-

143, 143, 143, 141, 163, 143, 143, 143, 143, 153, precipitated pod-old was partic

Ethyl (E) -3-(4-chlorophenyl)-3-(2-(4-methoxyphenyl)-1′,3′dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl) acrylate (4y). The precipitated product was purified with cold methanol, yield: 128 mg (87%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3427, 2969, 2841, 1710, 1614, 1517, 1451, 1331, 1285, 1256, 1174, 1092, 1036, 872, 829, 745, 671, 568, 500. ¹H NMR (500 MHz, Chloroform-D) δ 7.85 (ddd, $J = 8.5$, 6.0, 3.7 Hz, 2H), 7.74-7.68 (m, 2H), 7.44 (d, $J = 8.6$ Hz, 2H), 7.35–7.28 (m, 4H), 6.87 (d, $J = 8.6$ Hz, 2H), 6.18 (s, 1H), 4.24 (d, $J = 9.7$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.78 (s, 3H), 3.31 (d, $J =$ 9.7 Hz, 1H), 0.98 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 194.6, 165.2, 159.3, 151.9, 142.4, 141.6, 139.2, 135.0, 134.7, 134.4, 130.1, 129.0, 128.7, 125.1, 122.7, 122.3, 122.1, 113.8, 60.2, 55.2, 51.2, 45.9, 38.0, 13.9. HRMS **(ESI)** m/z : $[M + Na]^+$ calculated for $C_{29}H_{23}O_5C$ lNa, 509.1126; found 509.1131.

Ethyl (E)-3-(4-fluorophenyl)-3-(2-(4-methoxyphenyl)-1′,3′ dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl) acrylate (4z). The precipitated product was purified with cold methanol, yield: 120 mg (85%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3423, 3066, 2975, 2842, 2051, 1894, 1711, 1607, 1513, 1454, 1329, 1280, 1171, 1092, 1032, 870, 832, 743, 572, 525. ¹H NMR (500 MHz, Chloroform-D) δ 7.88-7.81 (m, 2H), 7.75-7.69 (m, 2H), 7.49 (dd, $J = 8.4$, 5.3 Hz, 2H), 7.31 (d, J $= 8.2$ Hz, 2H), 7.05 (t, $J = 8.4$ Hz, 2H), 6.87 (d, $J = 8.3$ Hz, 2H), 6.16 (s, 1H), 4.23 (d, $J = 9.6$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.79 (s, 3H), 3.31 (d, $J = 9.6$ Hz, 1H), 0.99 (t, $J = 7.1$ Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.0, 194.6, 165.2, 164.3, 161.8, 159.2, 152.0, 141.9 (d, J = 75.9 Hz), 136.8, 136.8, 134.5 $(d, J = 36.2 \text{ Hz})$ 130.1, 129.5 $(d, J = 8.2 \text{ Hz})$, 125.1, 122.4 $(d, J = 1.2 \text{ Hz})$ 58.2 Hz), 121.9, 115.5 (d, J = 21.9 Hz).113.7, 60.1, 55.1, 51.2, 45.9, 38.2, 13.9. ¹⁹F NMR (376 MHz, Chloroform-D) δ -111.94. **HRMS** (ESI) m/z : $[M + Na]^+$ calculated for $C_{29}H_{23}O_5FNa$, 493.1422; found 493.1426.

Methyl (E)-3-(1′,3′-dioxo-2-phenyl-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-(2-fluorophenyl) acrylate (4aa). The precipitated product was purified with cold methanol, yield: 114 mg (83%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3425, 3066, 2975, 2842, 2051, 1894, 1710, 1607, 1513, 1454, 1329, 1280, 1171, 1092, 1032, 870, 832, 743, 572, 525.¹H NMR (400 MHz, Chloroform-D) δ 7.85 (s, 2H), 7.78–7.68 (m, 2H), 7.53–7.45 (m, 2H), 7.36 (td, $J = 5.1$, 3.2 Hz, 5H), 7.02 (t, $J = 8.7$ Hz, 2H), 6.22 (s, 1H), 4.23 (d, $J = 9.5$ Hz, 1H), 3.38 (s, 4H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.8, 194.9, 165.9, 153.2, 142.12 (d, $J = 64.1$ Hz), 140.7, 134.73 (d, $J =$ 28.3 Hz), 130.73 $(d, J = 8.2 \text{ Hz})$, 129.2, 127.7, 122.56 $(d, J = 46.2 \text{ Hz})$ Hz), 121.9, 115.38 (d, $J = 21.4$ Hz), 51.4, 50.8, 45.3, 38.7. NMR (376 MHz, Chloroform-D) δ -113.88. HRMS (ESI) m/z : [M + Na]⁺ calculated for $C_{28}H_{21}O_5$ FNa, 479.1265; found 479.1262.

Methyl (E)-3-(2-(4-methoxyphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-(quinolin-6-yl) acrylate (4ab). The precipitated product was purified with cold methanol, yield: 53 mg (35%); dr > 20 : 1; green solid.

IR (neat, cm−¹): 3427, 3054, 2951, 2840, 1710, 1612, 1516, 1458, 1436, 1291, 1251, 1168, 1121, 1090, 1038, 888, 838, 800, 738, 615, 566, 530, 480. ¹H NMR (400 MHz, Chloroform-D) δ 8.94–8.86 (m, 1H), 8.15–8.05 (m, 2H), 7.98 (s, 1H), 7.86 (dq, $J =$ 4.9, 1.8, 1.4 Hz, 1H), 7.84–7.80 (m, 2H), 7.74–7.68 (m, 2H), 7.41–7.36 (m, 1H), 7.35–7.30 (m, 2H), 6.87 (d, $J = 9.0$ Hz, 2H), 6.33 (s, 1H), 4.31 (d, $J = 9.7$ Hz, 1H), 3.77 (s, 3H), 3.39 (s, 3H), 3.34 (d, $J = 9.7$ Hz, 1H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.0, 194.7, 165.6, 159.3, 152.7, 151.1, 148.2, 142.4, 141.7, 138.9, 136.7, 134.8, 134.5, 130.1, 129.7, 128.8, 128.0, 127.2, 125.1, 122.8, 122.7, 122.2, 121.7, 113.8, 55.2, 51.5, 51.3, 46.1, 38.3. **HRMS (ESI)** m/z : $[M + Na]^+$ calculated for $C_{31}H_{23}O_5NNa$, 512.1468; found 512.1466.

Benzyl (E)-3-(2-(4-methoxyphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4ac). Purified the precipitated product with cold methanol, yield: 135 mg (87%); dr > 20 : 1; yellow solid.

IR (neat, cm−¹): 3428, 3060, 3033, 2935, 2838, 1709, 1613, 1516, 1452, 1341, 1291, 1251, 1160, 1089, 1036, 877, 826, 739, 699, 621, 564, 539. ¹H NMR (400 MHz, Chloroform-D) δ 7.87 $(s, 1H)$, 7.81 $(s, 1H)$, 7.72 $(s, 2H)$, 7.52 $(d, J = 2.3 \text{ Hz}, 2H)$, 7.38–7.34 (m, 4H), 7.33 (s, 1H), 7.25 (d, $J = 2.3$ Hz, 3H), 7.14 (d, $J = 2.4$ Hz, 2H), 6.89 (d, $J = 8.7$ Hz, 2H), 6.28 (s, 1H), 4.87 (s, 2H), 4.30 (d, $J = 9.6$ Hz, 1H), 3.80 (s, 3H), 3.38 (d, $J = 9.6$ Hz, 1H). 13C NMR (101 MHz, Chloroform-D) δ 196.0, 194.9, 165.1, 159.2, 154.3, 142.4, 141.7, 140.7, 135.9, 134.7, 134.3, 130.2, 129.1, 128.5, 128.4, 128.2, 128.1, 127.7, 125.4, 122.7, 122.2, 121.4, 113.8, 65.8, 55.2, 51.3, 46.1, 38.4. HRMS (ESI) m/z: [M + Na]⁺ calculated for C₃₄H₂₆O₅Na, 537.1672; found 537.1670.

2-(Trimethylsilyl)ethyl (E)-3-(2-(4-methoxyphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4ad). Purified the precipitated product with cold methanol, yield: 139 mg (88%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3431, 2954, 2840, 1711, 1686, 1585, 1589, 1514, 1435, 1347, 1315, 1270, 1250, 1163, 1082, 1034, 862, 837, 775, 737, 698, 564, 532. ¹H NMR (500 MHz, Chloroform-D) δ 7.86 (d, $J = 1.2$ Hz, 2H), 7.77-7.66 (m, 2H), 7.51 (dd, $J = 7.1$, 2.5 Hz, 2H), 7.40-7.28 (m, 5H), 6.87 (d, $J = 8.7$ Hz, 2H), 6.19 (d, $J =$ 2.1 Hz, 1H), 4.26 (d, $J = 9.6$, Hz, 1H), 3.93-3.84 (m, 2H), 3.80 (s, 3H), 3.35 (d, $J = 9.6$ Hz, 1H), 0.75 (ddd, $J = 13.7, 11.1, 5.9$ Hz, 1H), 0.62 (ddd, $J = 13.7, 11.1, 5.9$ Hz, 1H), -0.08 (s, 9H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.8, 194.8, 165.5, 159.1, 152.8, 142.4, 141.7, 140.8, 137.2, 135.0, 134.5, 134.2, 130.1, 128.9, 128.4, 127.6, 125.3, 123.0, 123.0, 122.6, 122.2, 122.0, 114.3, 113.7, 62.2, 55.1, 51.1, 46.0, 38.5, 17.0, −1.6. HRMS **(ESI)** m/z : $[M + Na]$ ⁺ calculated for C₃₂H₃₂O₅SiNa, 547.1911; found 547.1909.

Allyl (E)-3-(2-(4-methoxyphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro [cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4ae). Purified the precipitated product with cold methanol, yield: 118 mg $(84%)$; dr > 20 : 1; white solid.

IR (neat, cm−¹): 3429, 2954, 2840, 1710, 1686, 1585, 1589, 1514, 1435, 1347, 1315, 1270, 1250, 1163, 1082, 1034, 862, 837, 775, 737, 698, 564, 532. ¹H NMR (400 MHz, Chloroform-D) δ 7.91-7.80 (m, 2H), 7.71 (d, $J = 8.5$ Hz, 2H), 7.58-7.50 (m, 2H), 7.34 (dd, $J = 12.0$, 7.9 Hz, 5H), 6.88 (s, 2H), 6.26 (d, $J = 2.1$ Hz, 1H), 5.64 (dd, J = 17.2, 10.4 Hz, 1H), 5.14–4.97 (m, 2H), 4.35–4.26 (m, 3H), 3.78 (d, $J = 0.9$ Hz, 3H), 3.37 (d, $J = 9.6$ Hz, 1H). ¹³C NMR (101 MHz, Chloroform-D) δ 195.9, 194.7, 164.9, 159.2, 153.8, 142.4, 141.6, 140.7, 134.6, 134.3, 131.9, 130.1, 129.0, 128.5, 127.6, 125.3, 122.6, 122.1, 121.5, 118.0, 113.7, 64.7, 55.1, 51.1, 46.0, 38.3. HRMS (ESI) m/z : [M + Na]⁺ calculated for C₃₀H₂₄O₅Na, 487.1516; found 487.1511.

Cyclohexyl (E)-3-(2-(4-methoxyphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4af). The precipitated product was purified with cold methanol, yield: 131 mg (86%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3429, 3058, 2936, 2858, 1708, 1614, 1516, 1450, 1346, 1290, 1251, 1176, 1121, 1090, 1038, 968, 928, 882, 824, 738, 699, 621, 564, 539. ¹H NMR (400 MHz, Chloroform-D) δ 7.89–7.82 (m, 2H), 7.70 (d, $J = 3.2$ Hz, 2H), 7.56–7.51 (m, 2H), 7.37–7.31 (m, 5H), 6.87 (d, $J = 8.7$ Hz, 2H), 6.22 (d, $J = 2.1$ Hz, 1H), 4.45 (dt, $J = 9.1$, 4.9 Hz, 1H), 4.31 (d, $J = 9.7$ Hz, 1H), 3.78 (s, 3H), 3.36 (d, J = 9.7 Hz, 1H), 1.72–1.58 (m, 2H), 1.54–1.48 (m, 1H), 1.42 (s, 1H), 1.30 (s, 1H), 1.27–1.20 (m, 2H), 1.14–1.05 (m, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.0, 194.8, 164.8, 159.1, 152.7, 142.4, 141.6, 140.8, 134.6, 134.1, 130.1, 128.9, 128.4, 127.6, 125.4, 122.7, 122.4, 122.0, 113.7, 72.2, 55.1, 51.1, 46.0, 38.5, 31.4, 31.0, 25.2, 23.7, 23.6. HRMS **(ESI)** m/z : $[M + Na]$ ⁺ calculated for C₃₃H₃₀O₅Na, 529.1985; found 529.1980.

tert-Butyl (E)-3-(2-(4-methoxyphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (4ag). The precipitated product was purified with cold methanol, yield: 112 mg (77%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3430, 3058, 2975, 2930, 1708, 1613, 1517, 1454, 1348, 1251, 1220, 1150, 1090, 1037, 869, 824, 775, 739, 698, 564, 539. ¹H NMR (400 MHz, Chloroform-D) δ 7.84 (d, J = 10.6 Hz, 2H), 7.69 (d, $J = 1.9$ Hz, 2H), 7.52 (s, 2H), 7.34 (d, $J =$ 3.1 Hz, 5H), 6.88 (dd, $J = 8.4$, 1.4 Hz, 2H), 6.16 (s, 1H), 4.29 (d, $J = 9.7$ Hz, 1H), 3.80 (s, 3H), 3.32 (d, $J = 9.7$ Hz, 1H), 1.10 (s, 9H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.3, 195.2, 164.9, 159.2, 151.8, 142.7, 141.7, 140.9, 134.6, 134.1, 130.2, 128.9, 128.6, 127.7, 125.7, 123.5, 122.7, 122.1, 113.8, 80.1, 55.3, 51.0, 45.9, 38.9, 27.7. **HRMS** (ESI) m/z : $[M + Na]$ ⁺ calculated for $C_{31}H_{28}O_5$ Na, 503.1829; found 503.1823.

 $(1R, 2S, 4R)$ -2-Isopropyl-4-methylcyclohexyl (E) -3- $(2-(4-methox$ yphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3 yl)-3-phenylacrylate (4ah). The precipitated product was purified with cold methanol, yield: 144 mg $(85%)$; dr = 1:1; white solid.

IR (neat, cm−¹): 3428, 2952, 2869, 1709, 1621, 1517, 1453, 1350, 1331, 1290, 1251, 1221, 1173, 1089, 1040, 988, 882, 826, 775, 746, 697, 620, 564, 538. ¹H NMR (500 MHz, Chloroform-D) δ 7.82 (s, 2H), 7.77-7.67 (m, 2H), 7.53 (dt, J = 5.3, 2.8 Hz, 2H), 7.39-7.28 (m, 5H), 6.87 (d, $J = 8.2$ Hz, 2H), 6.20 (d, $J = 20.1$ Hz, 1H), 4.49-4.35 (m, 1H), 4.32 (d, $J = 9.6$ Hz, 1H), 3.80 (s, 3H), 3.33 (d, $J = 9.6$ Hz, 1H), 1.83-1.70 (m, 1H), 1.61-1.49 (m, 3H), $1.37-1.12$ (m, 3H), $0.89-0.58$ (m, 10H), 0.23 (d, $J = 6.9$ Hz, 1H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.2, 195.0, 165.1, 165.0, 159.3, 152.9, 141.8, 140.9, 134.7, 134.3, 130.3, 129.1, 128.6, 127.9, 127.7, 125.6, 122.9, 122.3, 122.1, 113.8, 74.1, 55.3, 51.1, 46.7, 46.3, 40.0, 38.7, 34.3, 31.2, 26.4, 23.6, 22.1, 20.7, 16.5. HRMS (ESI) m/z : $[M + Na]^+$ calculated for $C_{37}H_{38}O_5Na$, 585.2611; found 585.2605. **Creamic & Biomolecular Chemistry**
 Published on 12 View Roba Diff, $j = 128.6$, 127.7 , 125.7 , 125.7 , 125.7 , 125.7 , 122.7 , 123.8 , 124.8 , 141.8 , 160.15 , 154.7 , 141.9 , 154.8 (i.i.d.) 141.3 , 14

 (Z) -3,7-Dimethylocta-2,6-dien-1-yl (E) -3-(2-(4-methoxyphenyl)-1′,3′-dioxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3 phenylacrylate (4ai). The precipitated product was purified with cold methanol, yield: 140 mg (83%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3432, 2927, 1709, 1614, 1516, 1447, 1335, 1252, 1222, 1165, 1089, 1037, 965, 879, 827, 749, 699, 618, 563, 539. ¹H NMR (400 MHz, Chloroform-D) δ 7.86 (d, J = 1.8 Hz, 2H), 7.72 (s, 2H), 7.52 (s, 2H), 7.35 (s, 5H), 6.91–6.84 (m, 2H), 6.23 (s, 1H), 5.13–4.99 (m, 2H), 4.35–4.24 (m, 3H), 3.79 (s, 3H), 3.37 (d, $J =$ 9.8 Hz, 1H), 1.97 (s, 3H), 1.66 (d, $J = 4.9$ Hz, 6H), 1.57 (s, 3H), 1.33–1.28 (m, 1H). 13C NMR (101 MHz, Chloroform-D) δ 196.0, 194.9, 165.4, 159.2, 153.3, 142.5, 142.2, 142.1, 141.8, 140.8, 134.6, 134.3, 132.0, 130.2, 129.0, 128.5, 127.7, 125.5, 123.7, 122.7, 122.1, 122.0, 119.1, 113.8, 60.8, 55.2, 51.2, 46.1, 38.4, 32.0, 26.6, 25.7, 23.5, 17.7. HRMS (ESI) m/z : $[M + Na]^+$ calculated for $C_{37}H_{36}O_5Na$, 583.2455; found 583.2459.

Methyl (E)-3-(-2-benzoyl-3-phenylcyclopropyl)-3-phenylacrylate (6a). The product was purified using flash column chromatography (Biotage flash chromatography gradient purification: EtOAc/n-hexane = 1:5), yield: 92 mg (79%); $dr > 20:1$; white solid.

IR (neat, cm−¹): 3406, 3061, 3029, 2944, 1711, 1660, 1609, 1492, 1446, 1429, 1374, 1337, 1275, 1226, 1170, 1078, 999, 914, 862, 772, 743, 697, 582, 529, 483. ¹H NMR (500 MHz, Chloroform-D) δ 7.98 (d, J = 7.5 Hz, 2H), 7.44 (dd, J = 20.8, 7.7 Hz, 6H), 7.37-7.31 (m, 5H), 7.23 (d, $J = 6.5$ Hz, 2H), 6.11 (s, 1H), 3.67 (dd, $J = 9.2$, 5.3 Hz, 1H), 3.52 (ddd, $J = 9.4$, 7.7, 1.9 Hz, 1H), 3.46 (s, 3H), 2.82 (dd, $J = 7.8$, 5.3 Hz, 1H). ¹³C NMR (101 MHz, Chloroform-D) δ 196.4, 166.4, 156.3, 141.9, 139.7, 138.3, 132.7, 128.8, 128.8, 128.5, 128.4, 128.3, 127.8, 126.9, 126.4, 120.4, 51.1, 38.9, 35.9, 35.6. HRMS (ESI) m/z : $[M + Na]$ ⁺ calculated for $C_{26}H_{22}O_3$ Na, 405.1461; found 405.1459.

Methyl (E)-3-(-2-benzoyl-3-(4-methoxyphenyl) cyclopropyl)-3 phenylacrylate (6b). The product was purified using flash column chromatography (Biotage flash chromatography gradient purification: EtOAc/n-hexane = $1:5$), yield: 96 mg (76%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3404, 3061, 3029, 2944, 1705, 1660, 1609, 1492, 1446, 1429, 1374, 1337, 1275, 1226, 1170, 1078, 999, 914, 862, 772, 743, 697, 582, 529, 483. ¹H **NMR** (500 MHz, Chloroform-D) δ 8.07–8.02 (m, 2H), 7.52 (s, 3H), 7.45 (s, 2H), 7.38 (s, 3H), 7.21 (s, 2H), 6.94–6.91 (m, 2H), 6.17 (d, $J = 2.1$ Hz, 1H), 3.81 (s, 3H), 3.70-3.65 (m, 1H), 3.54 (t, $J = 8.5$ Hz, 1H), 3.50 (s, 3H), 2.88-2.83 (m, 1H). ¹³C NMR (126 MHz, Chloroform-D) δ 196.2, 166.2, 158.6, 156.2, 141.8, 138.3, 132.5, 131.6, 128.6, 128.3, 128.3, 128.2, 127.7, 127.4, 120.2, 114.1, 55.2, 50.9, 38.8, 35.3, 35.2. HRMS (ESI) m/z : $[M + Na]$ ⁺ calculated for $C_{27}H_{24}O_4$ Na, 435.1567; found 435.1571.

Ethyl (E)-3-(-2-(4-bromobenzoyl)-3-phenylcyclopropyl)-3-phenylacrylate (6c). The product was purified using flash column chromatography (Biotage flash chromatography gradient purification: EtOAc/n-hexane = 1:5), yield: 110 mg (78%); dr > 20 : 1; orange solid.

IR (neat, cm−¹): 3394, 3090, 3059, 2983, 1706, 1662, 1611, 1582, 1490, 1434, 1374, 1271, 1222, 1174, 1070, 1022, 860, 809, 766, 697, 621, 582, 533, 481, 582. ¹H NMR (500 MHz, Chloroform-D) δ 7.86 (d, J = 8.4 Hz, 2H), 7.54 (d, J = 8.4 Hz, 2H), 7.43 (d, J = 1.9 Hz, 2H), 7.38–7.30 (m, 5H), 7.28–7.19 (m, 3H), 6.10 (s, 1H), 3.98 (d, $J = 7.1$ Hz, 1H), 3.91 (d, $J = 7.1$ Hz, 1H), 3.61 (dd, $J = 9.2$, 5.3 Hz, 1H), 3.53 (t, $J = 7.8$ Hz, 1H), 2.85–2.78 (m, 1H), 1.09 (t, $J = 7.2$ Hz, 3H). ¹³C NMR (126 MHz, Chloroform-D) δ 195.2, 165.8, 155.8, 141.7, 139.5, 137.0, 131.6, 130.0, 128.8, 128.7, 128.3, 127.8, 127.7, 127.0, 126.3, 120.9, 59.9, 38.8, 35.9, 35.7, 14.1. HRMS (ESI) m/z : $[M + Na]^+$ calculated for $C_{27}H_{23}O_3BrNa$, 497.0723; found 497.0720. Communication

column chromatography (isotage flash chromatography gradi Eulyl (E)-3(1-4)qlotoxy-2(4-methoxypheryl)-3"soo-1,3"-dily

et perificialities FOA(2026-bene-1=1-15), yield: 501 for disperificialities flash chroma

Cyclohexyl (E)-3-(-2-benzoyl-3-phenylcyclopropyl)-3-phenylacrylate (6d). The product was purified using flash column chromatography (Biotage flash chromatography gradient purification: EtOAc/n-hexane = 1:5), yield: 120 mg (77%); dr > 20 : 1; white solid.

IR (neat, cm−¹): 3441, 3060, 3029, 2933, 2857, 1706, 1668, 1614, 1495, 1449, 1371, 1266, 1219, 1173, 1078, 1018, 872, 750, 697, 534. ¹H NMR (500 MHz, Chloroform-D) δ 7.91 (d, J = 7.2) Hz, 2H), 7.37 (d, $J = 7.1$ Hz, 3H), 7.29 (s, 2H), 7.23 (dd, $J = 14.1$, 7.2 Hz, 5H), 7.14 (s, 3H), 6.00 (s, 1H), 4.45 (s, 1H), 3.60 (ddd, J $= 8.4, 5.3, 2.2$ Hz, 1H), 3.47 (s, 1H), 2.74 (s, 1H), 1.68 (d, $J = 5.7$ Hz, 1H), 1.60–1.52 (m, 1H), 1.45–1.32 (m, 2H), 1.17 (s, 3H), 1.04-0.94 (m, 2H), 0.77 (dt, $J = 15.1$, 7.8 Hz, 1H). ¹³C NMR (126 MHz, Chloroform-D) δ 196.1, 165.3, 155.6, 141.9, 139.8, 138.1, 132.6, 128.7, 128.6, 128.5, 128.5, 128.5, 128.3, 128.3, 128.3, 128.2, 128.2, 127.7, 126.8, 126.3, 121.4, 72.2, 38.8, 35.8, 35.7, 31.7, 31.3, 25.4, 23.8, 23.7. HRMS (ESI) m/z : $[M + Na]^{+}$ calculated for $C_{31}H_{30}O_3$ Na, 473.2087; found 473.2085.

(1R,2S,5R)-2-Isopropyl-5-methylcyclohexyl (E)-3-(-2-benzoyl-3-phenylcyclopropyl)-3-phenylacrylate (6e). The product was purified using flash column chromatography (Biotage flash chromatography gradient purification: EtOAc/n-hexane = $1:5$), yield: 114 mg $(74%)$; dr = 1 : 1; white solid.

HRMS (ESI) m/z : $[M + Na]$ ⁺ calculated for C₃₅H₃₈O₃Na, 529.2713; found 529.2713.

Ethyl (E)-3-(1′-hydroxy-2-(4-methoxyphenyl)-3′-oxo-1′,3′-dihydrospiro[cyclopropane-1,2′-inden]-3-yl)-3-phenylacrylate (8). The product is purified using flash column chromatography (Biotage flash chromatography gradient purification: EtOAc/ n -hexane = 1:4), yield: 100 mg (71%); yellow liquid.

¹H NMR (400 MHz, Chloroform-D) δ 7.72 (d, $J = 0.9$ Hz, 1H), 7.62 (d, $J = 1.7$ Hz, 2H), 7.53–7.49 (m, 2H), 7.38–7.34 (m, 3H), 7.28–7.23 (m, 3H), 6.91 (d, $J = 8.8$ Hz, 2H), 6.18 (d, $J = 2.2$ Hz, 1H), 5.14 $(d, J = 11.7$ Hz, 1H), 4.71 $(d, J = 11.7$ Hz, 1H), 3.81 $(s, 5H)$, 3.70–3.66 (m, 1H), 2.86 (d, $J = 8.5$ Hz, 1H), 0.98 (t, $J =$ 7.1 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-D) δ 199.5, 166.9, 158.9, 156.9, 155.1, 141.5, 136.2, 134.8, 129.1, 129.0, 128.9, 128.6, 128.5, 128.0, 126.3, 122.3, 120.4, 114.3, 69.3, 60.6, 55.4, 53.8, 40.5, 34.6, 13.9. HRMS (ESI) m/z : $[M + Na]$ ⁺ calculated for $C_{29}H_{24}O_4$ Na, 459.1567; found 459.1570.

Methyl (E)-4-(dimethyl(oxo)-l6-sulfaneylidene)-5-oxo-5-phenylpent-2-enoate (1r). H NMR (500 MHz, Chloroform-D) δ 7.64 $(d, J = 15.7 \text{ Hz}, 1\text{H}), 7.57-7.52 \text{ (m, 2H)}, 7.44-7.39 \text{ (m, 2H)}, 5.60$ $(d, J = 15.7 \text{ Hz}, 1\text{H}), 3.69-3.57 \text{ (m, 9H)}.$ ¹³C NMR (101 MHz, Chloroform-D) δ 187.9, 169.1, 139.7, 138.0, 131.0, 128.6, 128.1, 103.2, 88.5, 51.1, 43.5. HRMS (ESI) m/z : $[M + H]^{+}$ calculated for $C_{14}H_{17}O_4S$, 281.0842; found 281.0841.

Author contributions

The manuscript was written with contributions from all authors. All authors have approved the final version of the manuscript. JV designed the project. SD and AP did all the experimental work reported in this manuscript. DKG performed the computation under the supervision of TK.

Data availability

Electronic supplementary information (ESI) comprises experimental procedures, characterisation of new compounds, crystallographic data submitted as CCDC 2349128 (compound 4l) and CCDC 2349131 (compound 6a), and computational procedures.

Conflicts of interest

The authors declare no competing financial interest.

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