



Cite this: *Nanoscale*, 2023, **15**, 12348

Effects of post-transfer annealing and substrate interactions on the photoluminescence of 2D/3D monolayer WS₂/Ge heterostructures†

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The ultraflat and dangling bond-free features of two-dimensional (2D) transition metal dichalcogenides (TMDs) endow them with great potential to be integrated with arbitrary three-dimensional (3D) substrates, forming mixed-dimensional 2D/3D heterostructures. As examples, 2D/3D heterostructures based on monolayer TMDs (e.g., WS₂) and bulk germanium (Ge) have become emerging candidates for optoelectronic applications, such as ultrasensitive photodetectors that are capable of detecting broadband light from the mid-infrared (IR) to visible range. Currently, the study of WS₂/Ge(100) heterostructures is in its infancy and it remains largely unexplored how sample preparation conditions and different substrates affect their photoluminescence (PL) and other optoelectronic properties. In this report, we investigated the PL quenching effect in monolayer WS₂/Ge heterostructures prepared *via* a wet transfer process, and employed PL spectroscopy and atomic force microscopy (AFM) to demonstrate that post-transfer low-pressure annealing improves the interface quality and homogenizes the PL signal. We further studied and compared the temperature-dependent PL emissions of WS₂/Ge with those of as-grown WS₂ and WS₂/graphene/Ge heterostructures. The results demonstrate that the integration of WS₂ on Ge significantly quenches the PL intensity (from room temperature down to 80 K), and the PL quenching effect becomes even more prominent in WS₂/graphene/Ge heterostructures, which is likely due to synergistic PL quenching effects induced by graphene and Ge. Density functional theory (DFT) and Heyd–Scuseria–Ernzerhof (HSE) hybrid functional calculations show that the interaction of WS₂ and Ge is stronger than in adjacent layers of bulk WS₂, thus changing the electronic band structure and making the direct band gap of monolayer WS₂ less accessible. By understanding the impact of post-transfer annealing and substrate interactions on the optical properties of monolayer TMD/Ge heterostructures, this study contributes to the exploration of the processing–properties relationship and may guide the future design and fabrication of optoelectronic devices based on 2D/3D heterostructures of TMDs/Ge.

Received 1st March 2023,
Accepted 29th June 2023

DOI: 10.1039/d3nr00961k

rsc.li/nanoscale

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†Electronic supplementary information (ESI) available: Additional characterization results of WS₂/Ge by optical imaging, fluorescence imaging, Raman spectroscopy, and AFM. Annealing effects on the PL of 2D WS₂/CVDG heterostructures. Fitting results for the low-temperature PL spectra of as-grown WS₂ and WS₂/Ge. DFT-calculated structure of WS₂/H-Ge(100). See DOI: <https://doi.org/10.1039/d3nr00961k>

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Introduction

Monolayers of two-dimensional (2D) semiconducting transition metal dichalcogenides (TMDs) are attractive candidates for next-generation electronics and optoelectronics, owing to a plethora of impressive physical properties including large on-off ratios, strong and tunable optical absorption/emission, and large excitonic effects.^{1–4} Dangling bond-free TMD surfaces further permit their van der Waals (vdW) stacking with many surfaces or substrates regardless of their dimensionality. This unique feature opens up the capability of effectively engineering properties of 2D TMDs by designing 2D/*n*D (*n* = 0, 1, 2, and 3) heterojunctions with desirable material combinations, leading to emerging functionalities.^{5–7} For example, by either directly synthesizing or mechanically transferring monolayer TMDs on target 3D semiconducting substrates (e.g., silicon (Si), gallium nitride (GaN), germanium (Ge)), 2D/3D semiconducting heterostructures have been prepared, creating a boom in a wide variety of device applications, such as tunnel diodes,^{8,9} tunnel field-effect transistors,¹⁰ photovoltaics,^{11,12} and photodetectors.^{13,14}

Among different types of 2D/3D semiconducting heterostructures, monolayer TMDs/bulk Ge has emerged as an interesting one for optoelectronic applications. Specifically, while monolayer TMDs have demonstrated excellent photoresponsivity, their photoresponse is mainly limited to within the visible range because of their relatively large intrinsic band gap (~1.6–2.0 eV).^{15,16} In comparison, Ge is a commonly used 3D semiconducting substrate with an indirect band gap of ~0.66 eV, which exhibits high mobility and large infrared (IR) absorption, especially suited for IR-related applications.^{17,18} Thus, by constructing 2D/3D vertical heterostructures using TMD/Ge, one can take advantage of the complementary optical properties of TMDs and Ge, so that functional devices such as sensitive IR photodetectors can be fabricated accordingly. For example, photodetectors based on multilayer WS₂/Ge heterojunctions with an ultrathin AlO_x interfacial layer have been constructed, exhibiting high photoresponsivity (634.5 mA W⁻¹), high specific detectivity, ultrafast photoresponse, and ultrabroadband detection ranging from deep ultraviolet to mid-wave IR (200 nm–4.6 μm).¹³ In another report, an ultrasensitive near-IR detector based on few-layer MoTe₂/Ge has also been fabricated by applying an ultrathin GeO₂ interfacial layer to reduce the reverse current and enhance the photoresponsivity.¹⁴

Despite these recent studies demonstrating that TMD/Ge heterostructures are promising candidates for optoelectronics, the investigated TMD materials are usually multilayered, and study on the relationship between preparation methods and optical properties of this 2D/3D system is still in its infancy. For example, it remains unclear how sample preparation conditions and substrate interactions affect the photoluminescence (PL) properties of monolayer TMDs on Ge. In this context, we fabricated 2D/3D WS₂/Ge heterostructures by transferring chemical vapor deposition (CVD)-grown WS₂ monolayers onto the surfaces of Ge(100). Effects of the post-transfer Ar/H₂ annealing process on WS₂/Ge interface morphology and optical properties were studied by atomic force

microscopy (AFM) and PL spectroscopy. Furthermore, we performed a comparative study of temperature-dependent PL responses of as-grown WS₂, WS₂/Ge, and WS₂/graphene/Ge heterostructures. Our results indicate that the WS₂/Ge system exhibits a large PL quenching (~500×) when compared to that of the as-grown WS₂ on SiO₂/Si, and this PL quenching effect becomes even more prominent in the 2D/2D/3D heterostructures of WS₂/graphene/Ge. Density functional theory (DFT) calculations,¹⁹ including vdW corrections (DFT-D2)²⁰ and hybrid functional HSE06²¹ simulations were performed to understand the electronic interaction between WS₂ and Ge. It is found that WS₂ monolayers interact strongly with Ge(100) and this leads to a reduced interlayer separation, though keeping the vdW interaction without forming a chemical bond. Our work helps to establish the structure–processing–property relationship of TMDs/Ge heterostructures by illustrating the roles of post-transfer annealing and sample–substrate interaction, which may provide useful information for the future fabrication of such 2D/3D optoelectronics using the 2D layer transfer technique.

Results and discussion

Preparation of monolayer WS₂/Ge heterostructures

Monolayer WS₂ was first synthesized *via* an atmospheric pressure chemical vapor deposition (APCVD) method.²² As schematically shown in Fig. 1a, the APCVD process involves the use of WO₃ and S powders as tungsten and sulfur sources, respectively, and NaBr as a growth promoter (see details in the Methods section).²³ The synthesis was carried out at 825 °C for 10 min, and the heating temperature ramps for the metal precursor and S powder are shown in Fig. 1b. To prepare WS₂-based 2D/3D heterostructures, we utilized a cellulose acetate (CA)-assisted wet-transfer approach to lift-off as-grown WS₂ monolayers and place them onto undoped Ge(100) (Fig. 1c), which had been found to yield improved surface cleanliness and reduced wrinkles/cracks when compared to conventional wet-transfer methods.²⁴ The successful transfer of WS₂ monolayers on Ge(100) was verified by optical microscopy imaging and fluorescence imaging, as shown in Fig. 2a and Fig. S1 in the ESI.† Raman spectroscopy characterization also demonstrated the characteristic vibrational modes from both WS₂ monolayers and Ge (Fig. S2†), further indicating the formation of WS₂/Ge heterostructures. It should be noted that the Raman intensities of characteristic E' and A₁' modes of WS₂ monolayers are significantly lower (~15 times) for WS₂/Ge heterostructures when compared to those of as-grown WS₂ on SiO₂/Si. This may be explained by the optical interference and absorption effects that lead to substrate-dependent Raman enhancement factors, which have already been observed in monolayer TMDs on various substrates.^{25,26}

Effects of post-transfer thermal annealing

Sample preparation conditions can often affect the optical properties of TMD-based heterostructures. Since the preparation



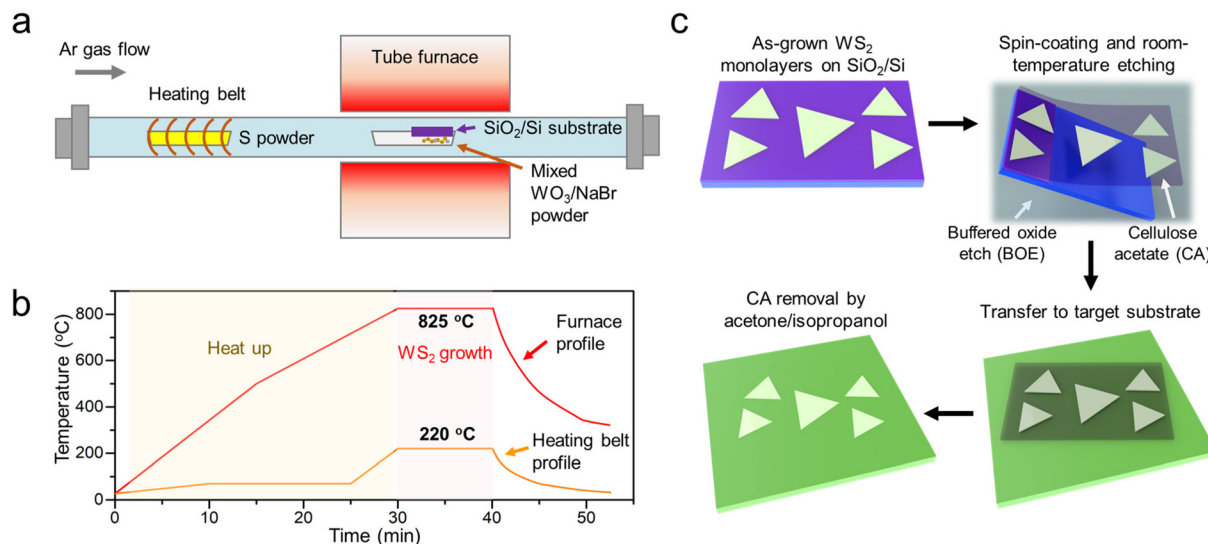


Fig. 1 Synthesis and transfer of WS_2 monolayers for preparing WS_2/Ge heterostructures. (a) Schematic of the APCVD setup for synthesizing WS_2 monolayers on SiO_2/Si . (b) Furnace and heating belt temperature profiles during the synthesis process. (c) Step-by-step schematics of the wet-transfer of WS_2 onto target substrates using a cellulose acetate-assisted approach.

of 2D/3D $\text{WS}_2/\text{Ge}(100)$ heterostructures involves a wet-transfer process, it is inevitable that contaminants and trapped water molecules are introduced at their interface, which may lead to undesired spatial heterogeneity, thus hindering the observation of intrinsic properties of as-prepared heterostructures.^{6,27–29} Thus, despite the fact that CA was used as a cleaner alternative support layer during the transfer

process, a post-transfer treatment is still needed to further improve the interface quality. Several previous studies have shown that post-transfer low-pressure thermal annealing is an effective method for enhancing the interlayer coupling strength in various types of 2D vdW heterostructures, but the effect of thermal annealing has been relatively less investigated in 2D/3D heterostructures.^{27,30–32} In this work, we annealed

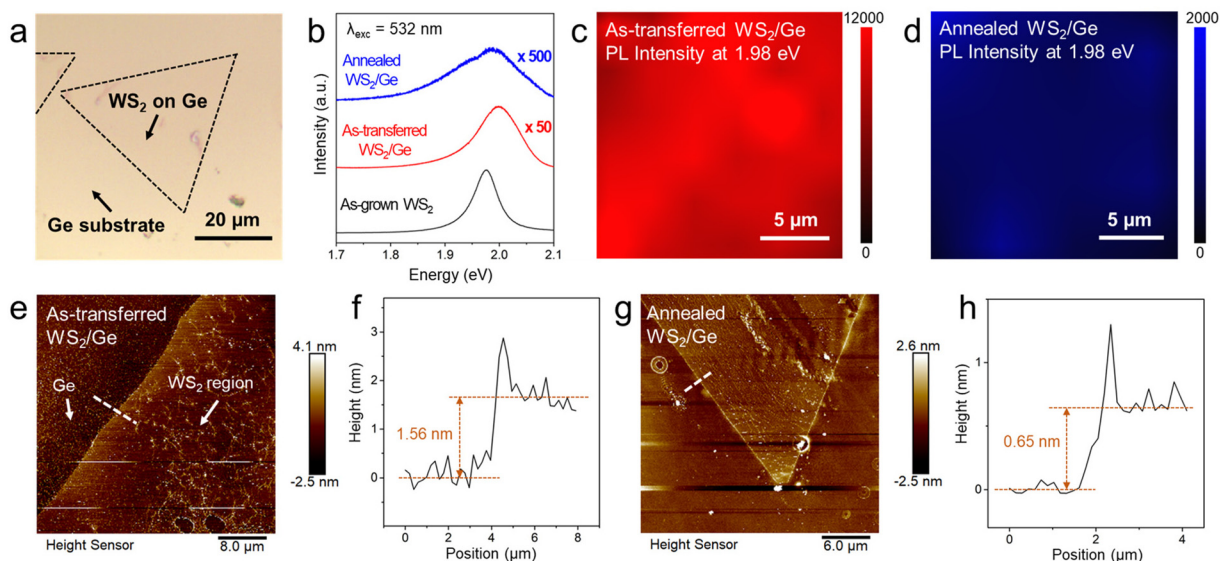


Fig. 2 Effects of post-transfer annealing on the PL and surface morphology of WS_2/Ge heterostructures. (a) Optical image of monolayer WS_2 transferred on Ge. Due to a relatively weak optical contrast, the edges of WS_2 are marked with black dashed lines for clarity. (b) A comparison of PL spectra of as-grown monolayer WS_2 , as-transferred WS_2/Ge , and annealed WS_2/Ge . WS_2/Ge heterostructures display prominent PL intensity quenching compared to as-grown WS_2 . (c) PL intensity mapping of as-transferred WS_2/Ge and (d) annealed WS_2/Ge at 1.98 eV, corresponding to the optical band gap of monolayer WS_2 . (e) AFM topography image of as-transferred WS_2/Ge and (f) the corresponding surface height profile along the white dashed line in (e). (g) AFM topography image of annealed WS_2/Ge and (h) the corresponding surface height profile along the white dashed line in (g).



our WS₂/Ge heterostructures and compared their morphological and optical properties with those of as-transferred systems. The annealing process was carried out under a 150 standard cubic centimeters per minute (sccm) Ar (85%)/H₂ (15%) flow at 250 °C for 10 hours, and the pressure was maintained at 810 mTorr during this process. Similar low-pressure annealing procedures have been reported in the literature, and these do not affect the structural integrity of TMD monolayers.^{33,34}

PL spectra of as-transferred and annealed WS₂/Ge are shown in Fig. 2b and are compared with that of as-grown WS₂ on SiO₂/Si. The PL spectrum of transferred WS₂/Ge displays a blue shift in peak energy, which can be attributed to the relaxation of the tensile strain induced during CVD growth.³⁵ This observation agrees with the blue shift of WS₂ in-plane E' mode in the Raman spectrum of WS₂/Ge (Fig. S2†), which is also a signature of released tensile strain.^{24,35} In addition, it is found that the PL intensity of as-transferred WS₂/Ge was largely quenched (~50 times) when compared to that of as-grown WS₂ monolayers. Through an additional Ar/H₂ annealing process, the PL intensity of WS₂/Ge was further suppressed, which became more than 500 times lower than that of as-grown WS₂ monolayers. We also observed very similar PL intensity quenching phenomena in both CVD-grown MoS₂ transferred to Ge (Fig. S3†) and exfoliated MoS₂ (following a previously reported gold-mediated exfoliation approach³⁶) on Ge (Fig. S4†), noticing a marked effect in CVD-grown materials, and suggesting that the underlying PL quenching mechanisms may be applicable to various TMDs/Ge systems. We believe that the strong PL quenching can be attributed to three independent mechanisms: first, the differences in optical constants (refractive index, reflectivity)^{37,38} and substrate geometries of Ge and SiO₂/Si (*i.e.*, Ge is a single-component bulk substrate, while SiO₂/Si is a double-layer structure consisting of a 300 nm SiO₂ thin film and bulk Si), result in different optical enhancement factors, due to optical interference and absorption effects. Nevertheless, this mechanism is expected to enhance/quench Raman and PL signals by similar orders of magnitude.³⁹ However, this does not match our observations since the Raman signal is much less quenched than the PL (~15 times and ~500 times, respectively), thus suggesting that additional factors are involved in the PL quenching phenomenon of TMDs/Ge. Second, the strong vdW interaction across the interface of TMDs and Ge is responsible for the significant PL quenching, which is anticipated to be more robust in annealed heterostructures when compared to that in as-transferred monolayers. This effect will be discussed in detail below when analyzing the electronic properties across the WS₂/Ge interface *via* DFT calculations. Third, the annealing process leads to the removal of adsorbed H₂O and O₂ molecules from the surface of monolayer WS₂ and/or the WS₂/Ge interface. Because H₂O and O₂ molecules are known as p-type dopants for the n-type as-grown WS₂,^{40,41} their removal may increase the spectral weight of a negative trion (X⁻) when compared to an exciton (X⁰), and further decrease the PL emission intensity in the annealed WS₂/Ge heterostructures.⁴² Apart from the PL quenching, a prominent emergence of a low-energy PL tail

and/or a red shift of PL emission are observed in annealed WS₂/Ge (Fig. 2b) and MoS₂/Ge (Fig. S3d†). This effect is likely due to either a reduction of the electronic band gap of WS₂ induced by the strong vdW interaction with Ge, or the enhancement of the negative trion (X⁻)-to-exciton (X⁰) ratio observed after annealing, as discussed above. However, we were not capable of deconvoluting these two factors at this stage.

To exclude the possibility that the observed PL quenching is due to spatial variation in the optical properties of WS₂/Ge, we performed PL mapping in randomly chosen areas of as-transferred and annealed WS₂/Ge. The PL intensity mappings are shown in Fig. 2c and d, demonstrating relatively uniform PL intensities in a large area, and it can be noticed that annealed WS₂/Ge exhibits uniformly lower PL intensities compared to as-transferred monolayers, thus validating the PL results shown in Fig. 2b. To demonstrate the generality of the low-pressure annealing effect to TMD-based heterostructures, we also fabricated heterostructures of monolayer WS₂ and CVD-grown monolayer graphene (CVDG) by consecutively transferring CVDG and WS₂ onto a clean SiO₂/Si substrate and compared the PL spectra of the as-transferred and annealed heterostructures (see the Methods section for details of graphene synthesis and transfer). Note that monolayer graphene has been known as a substrate that quenches the PL of TMDs, due to the enhanced nonradiative recombination of photo-generated charge carriers in TMDs through charge transfer processes and the dipole-dipole interaction.^{39,43,44} As shown in Fig. S5,† the PL intensities of both transferred WS₂/SiO₂/Si and WS₂/CVDG/SiO₂/Si decreased after low-pressure Ar/H₂ annealing, and the PL intensity uniformity was improved, thus suggesting stronger interfacial interactions that lead to a more efficient substrate-induced charge transfer process after annealing. Future studies are desired to further investigate the tunability of PL emissions as a function of annealing parameters (*e.g.*, annealing temperature, annealing time, Ar and/or H₂ flow rates, and pressure).

In order to verify that the annealing process can improve the WS₂/Ge interface quality and promote physical coupling across the interface, AFM was employed to characterize the surface morphology and interface step height. Fig. 2e depicts the AFM topography image of as-transferred WS₂/Ge, and Fig. 2f shows the surface height profile along the white dashed line in Fig. 2e. Since the surface height profile is taken across the WS₂/Ge interface, the step height (1.56 nm) shown in Fig. 2e represents the separation between the WS₂ layer and the Ge substrate. For annealed WS₂/Ge heterostructures that are similarly analyzed (Fig. 2g and h), it is observed that the WS₂/Ge separation is reduced to 0.65 nm, which matches the expected thickness of TMD monolayers.^{45,46} The reduction of WS₂/Ge separation after the annealing process was also observed in other randomly measured flakes (Fig. S6†). Our comparative AFM studies of as-transferred and annealed WS₂/Ge heterostructures agree well with previous work on transferred 2D heterostructures, where the step height across the interface of as-transferred and annealed MoS₂/WS₂ hetero-



structures was compared, and it was found that the step height was decreased from ~ 1.6 nm to ~ 0.8 nm after annealing.²⁷ Our results reveal an improved layer interaction between WS₂ and Ge that is likely due to the effective removal of interface-trapped molecules and contaminants by low-pressure thermal annealing, and they also imply that the enhanced interfacial interaction is the reason for the strong PL quenching that we observed.

Temperature-dependent photoluminescence properties

We further compared the PL quenching effect of WS₂/Ge with that of 2D WS₂/CVDG heterostructures, another system that exhibits PL quenching that we have studied, as shown in Fig. S5,† and temperature-dependent PL measurements were performed to investigate the exciton species at reduced temperatures. For this study, all of the WS₂ monolayers were synthesized in the same CVD reaction so that the batch-to-batch variation of sample quality was avoided. It should also be noted that all the measurements of transferred heterostructures here were carried out after low-pressure Ar/H₂ annealing to guarantee enhanced interface quality and uniformity.

Fig. 3a shows comparative PL spectra of as-grown monolayer WS₂ and transferred/annealed WS₂ on various substrates, including WS₂/CVDG, WS₂/Ge, and WS₂/Ge with a graphene interfacial layer (WS₂/CVDG/Ge). When compared to as-grown WS₂ monolayers, transferring monolayer WS₂ to CVDG quenches the PL emission intensity by ~ 120 times. As for WS₂/Ge heterostructures, the PL intensity quenching is even more efficient, where a ~ 700 times reduction is observed when compared to that of as-grown WS₂ monolayers. While the PL spectrum of WS₂/Ge heterostructures contains the contribution of both X⁰ and X⁻ (marked with the dashed lines in Fig. 3a), the

PL of WS₂/CVDG appears to be more red-shifted, which agrees with the observation by Giusca *et al.*, and may indicate a higher density of X⁻ induced by the doping of the underlying graphene layer.⁴³ Furthermore, intrigued by the effective PL quenching effect from both CVDG and Ge, we applied a CVDG interfacial layer between WS₂ and Ge and investigated the PL of resultant WS₂/CVDG/Ge heterostructures. Notably, the PL emission from WS₂/CVDG/Ge is drastically quenched and nearly indistinguishable from the background noise. This indicates a synergistic quenching effect due to the presence of the underlying CVDG layer and the Ge substrate.

Low-temperature PL spectroscopy is an effective method of identifying exciton species and optically probing defect states, because influences of thermal dissociation of bound excitons and the exciton-phonon interaction-induced PL linewidth broadening become less prominent at reduced temperatures, which allows different exciton species to be better resolved.^{47,48} Thus, to gain further insights into the PL properties of WS₂/Ge heterostructures, we compared the temperature-dependent PL spectra of as-grown WS₂, WS₂/Ge, and WS₂/CVDG/Ge in the temperature range of 80–280 K. Fig. 3b and c depict temperature-dependent PL spectra of as-grown WS₂ monolayers and annealed WS₂/Ge heterostructures, respectively. PL spectra of both samples show a gradual blue shift as the temperature decreases (~ 70 meV shift from 280 K to 80 K), agreeing with the temperature-dependent band gap variation.⁴⁹ In addition, by deconvoluting the PL spectra obtained at 80 K (Fig. S7†), it is possible to assign observed exciton species and track their evolution with temperature (black dashed lines in Fig. 3b and c). It is found that X⁰ dominates room-temperature PL spectra of both samples, while at lower temperature, the X⁻ emission starts to emerge. This can be understood by the much smaller trion binding energy (~ 30 – 50 meV) compared to the exciton

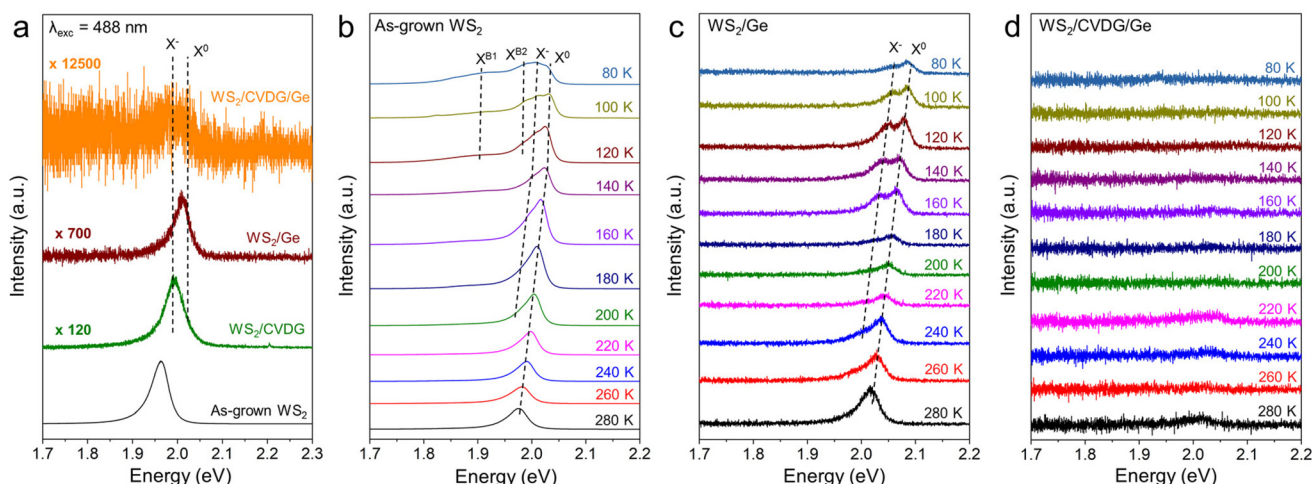


Fig. 3 Comparative PL spectra of as-grown WS₂, and annealed WS₂/Ge and WS₂/CVDG/Ge heterostructures obtained at both room temperature and reduced temperatures, down to 80 K. (a) A comparison of room-temperature PL of as-grown monolayer WS₂, WS₂/CVDG/SiO₂/Si, WS₂/Ge, and WS₂/CVDG/Ge. WS₂/CVDG/Ge heterostructures demonstrate the most prominent PL quenching effect, followed by WS₂/Ge heterostructures. (b–d) Temperature-dependent PL spectra of as-grown WS₂, WS₂/Ge, and WS₂/CVDG/Ge, respectively. The exciton species identified from the PL spectra are marked with black dashed lines in (a–c).



binding energy (a few hundreds of meV), making X^- easier to be thermally dissociated at a higher temperature.⁵⁰ Moreover, multiple defect-bound excitons (X^{B1} and X^{B2}) are present in as-grown WS_2 at low temperatures, which are related to structural defects, such as sulfur vacancy-related defects.⁴⁸ However, it is interesting to note that defect-related emissions are highly suppressed in WS_2/Ge heterostructures. This implies that the defect-induced mid-gap states may be modified or passivated due to the electronic coupling between WS_2 and Ge,⁵¹ which requires follow-up studies for further investigation. Finally, temperature-dependent PL measurement of $WS_2/CVDG/Ge$ was performed and the results are plotted in Fig. 3d. Similar to the room-temperature PL data in Fig. 3a, there is no observable enhancement/recovery of the PL signal at reduced temperatures, and the PL emission remains negligible at all of the measured temperatures, which again indicates the effective PL quenching caused by CVDG and the Ge substrate.

Mechanisms of photoluminescence quenching in the WS_2/Ge system

DFT calculations were carried out to study the mechanisms of PL quenching in WS_2/Ge from a theoretical perspective. DFT-PBE¹⁹ calculations, including vdW correction DFT-D2,²⁰ were first performed on the bulk WS_2 system to obtain the distance between WS_2 monolayers; it is found that the distance between sulfur atoms in adjacent layers of WS_2 is 3.10 Å (see Fig. 4a). Next, the distance between monolayer WS_2 and the Ge (100) slab was calculated, and the optimized structure of $WS_2/Ge(100)$ is shown in Fig. 4b (side view) and Fig. 4c (top view). It is found that the separation of WS_2 and Ge(100) is 2.50 Å (see Fig. 4b) and, therefore, the vdW interaction in $WS_2/Ge(100)$ is stronger than in bulk WS_2 . It is also well known that the band structure of WS_2 -type TMDs is very sensitive to the vdW interaction. For example, when the layer number is increased from monolayer to multilayer, TMDs become indirect band gap materials, with the direct band gap transition at the Brillouin

zone K -point almost unchanged, but less accessible,^{3,52} which leads to a severely quenched PL in multilayer TMDs when compared to that of monolayers. A similar case with germanene, a 2D form of Ge, also revealed a reduction in the distance between interacting germanene and WS_2 layers.⁵³ Thus, it is possible that the enhanced interfacial interaction in $WS_2/Ge(100)$ can result in our experimentally observed PL quenching phenomenon. To verify this, we investigated the interfacial interaction and electronic behavior in the $WS_2/Ge(100)$ heterostructure. Note that a band gap correction approach is necessary through hybrid functionals as expressed in the HSE06 approach.²¹ First, the band gap was calculated for monolayer WS_2 and bulk Ge, obtaining band gap values of 2.05 eV and 0.68 eV, respectively (Fig. 5a and b). It is worth mentioning that for monolayer WS_2 , the electronic single particle band gap was calculated (no excitonic effects were included). On the other hand, the calculated band gap value for bulk Ge agrees very well with previous experimental and calculated values (0.66 eV).^{54,55} For the $WS_2/Ge(100)$ heterostructure, by calculating the contribution of WS_2 in the density of states (DOS) of $WS_2/Ge(100)$, it is found that the band gap of WS_2 in $WS_2/Ge(100)$ is much reduced to 0.13 eV with multiple additional energy levels exhibited within the WS_2 band gap (~ 2 eV) due to the vdW interaction with the Ge surface, which makes the WS_2 band gap less accessible and is likely responsible for the PL quenching. In addition, the adsorption energy (per unit surface area) was calculated by using the following relationship: $E_{ad} = E_{WS_2} + E_{Ge} - E_{WS_2/Ge}$ where E_{WS_2} , E_{Ge} , and $E_{WS_2/Ge}$ are the total energies for the WS_2 , Ge, and vdW $WS_2/Ge(100)$ system, respectively. The adsorption energy for the $WS_2/Ge(100)$ system was found to be 21.3 meV Å⁻², which in combination with the small distance (2.50 Å) between WS_2 and Ge (100) indicates that PL quenching is expected to occur, similar to the case of WS_2 on sapphire.⁵⁶ The case of hydrogenated Ge (100) [H-Ge(100), using a supercell with the same dimensions as $WS_2/Ge(100)$] was also investigated by adding 32 H atoms at

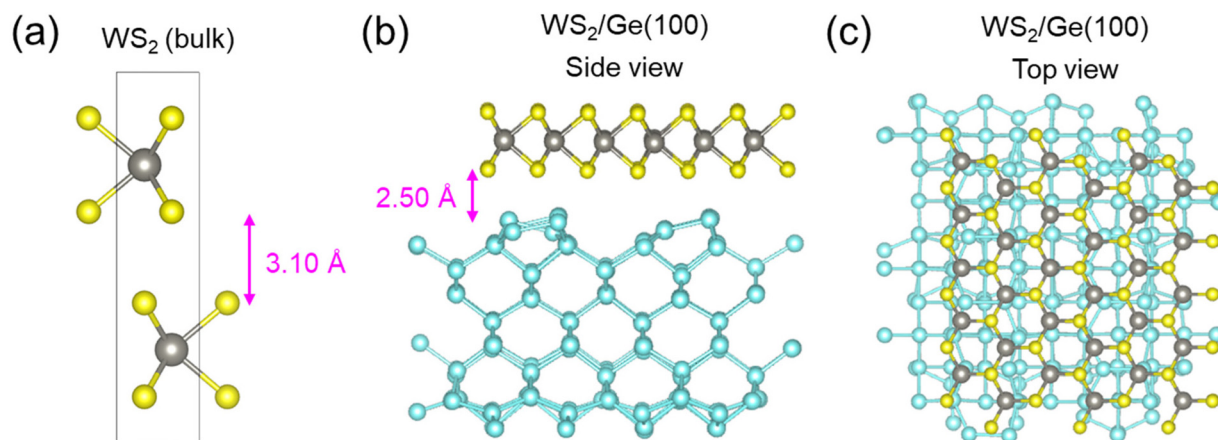


Fig. 4 DFT-calculated structures of bulk pristine WS_2 and monolayer WS_2 on a Ge(100) slab. (a) Structure of bulk WS_2 . The interlayer distance (the distance between sulfur atoms in adjacent layers) is calculated to be 3.10 Å. (b) Side view and (c) top view of monolayer $WS_2/Ge(100)$ heterostructures. The WS_2 -to-Ge distance is calculated to be 2.50 Å.



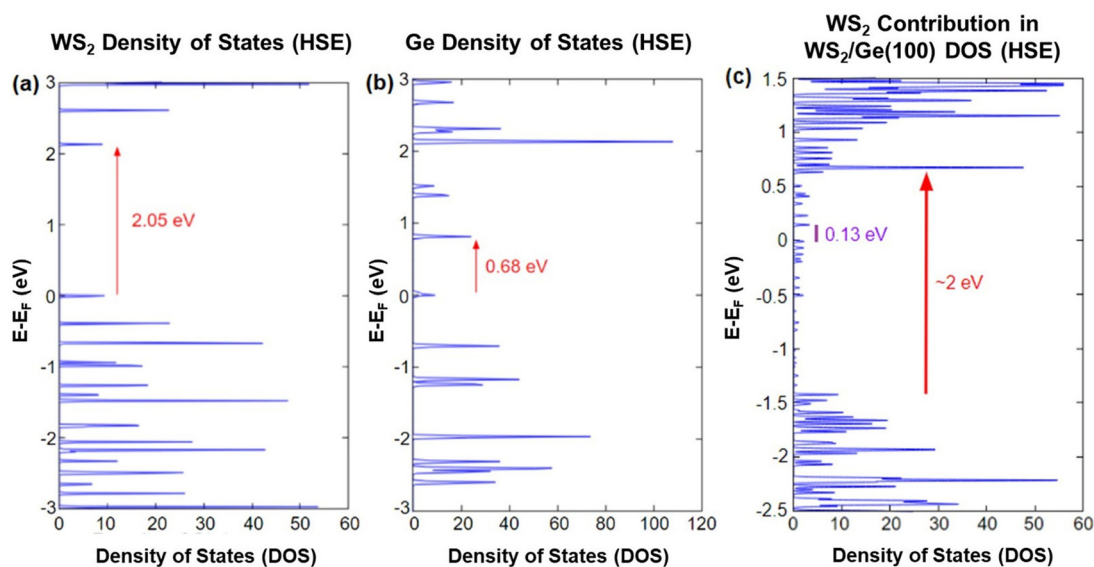


Fig. 5 HSE-calculated DOS and band gaps of monolayer WS_2 , bulk Ge, and $WS_2/Ge(100)$ heterostructures. (a) Calculated DOS and band gap of monolayer WS_2 , indicating a band gap of 2.05 eV. (b) Calculated DOS and band gap of Ge, indicating a band gap of 0.68 eV. (c) Calculated WS_2 contribution in the DOS of $WS_2/Ge(100)$ heterostructures.

the open Ge sites on the top and bottom surfaces of Ge(100). For $WS_2/H-Ge(100)$, the distance between the topmost S atom in WS_2 and the topmost H atom in H-Ge(100) is 2.22 Å (Fig. S8†) and $E_{ad} = 19.4 \text{ meV Å}^{-2}$, which is similar to the previous case with no H-passivation in $WS_2/Ge(100)$. The slight distance reduction in the case of the $WS_2/H-Ge(100)$ system is similar to the effect seen in graphene/hydrogenated hexagonal boron nitride (H-hBN) heterostructures.⁵⁷ Given the strong vdW interaction present in both the $WS_2/Ge(100)$ and $WS_2/H-Ge(100)$ systems, it is likely the PL quenching is caused by strong vdW interactions regardless of whether the Ge(100) substrate is H-passivated or not.

tinguishable from the background noise. DFT calculations were performed to understand the mechanism of PL quenching in $WS_2/Ge(100)$. It is found that the vdW interactions at the WS_2/Ge interface is stronger than in the bilayer or bulk of WS_2 , which leads to large modifications within the band gap of WS_2 . In all, this work demonstrates important roles of post-transfer treatment and substrate vdW interactions in the optical properties of TMDs/Ge heterostructures, paving the way for designing high-performance optoelectronic devices based on 2D/3D TMD/Ge heterostructures by interface engineering.

Conclusions

In this work, we prepared 2D/3D $WS_2/Ge(100)$ heterostructures and investigated in detail the PL properties of WS_2/Ge and how they can be affected by post-transfer low-pressure Ar/ H_2 annealing. We found that WS_2/Ge heterostructures exhibit a significant PL quenching when compared to as-grown WS_2 monolayers, and post-transfer annealing can effectively improve the WS_2/Ge interface quality and enhance the PL signal uniformity. Moreover, the temperature-dependent PL response of WS_2/Ge was further studied and compared with those of as-grown WS_2 and $WS_2/CVDG/Ge$ heterostructures. It is observed that WS_2/Ge heterostructures exhibit consistent PL quenching effects from room temperature down to 80 K, and the defect-related emissions of WS_2 are highly suppressed, possibly due to the modification or passivation of defect-induced mid-gap states. In addition, the PL signals of $WS_2/CVDG/Ge$ heterostructures are highly quenched and hardly dis-

Methods

Synthesis and transfer of monolayer WS_2

For the synthesis of monolayer WS_2 , mixed $WO_3/NaBr$ powders are loaded in a porcelain boat, and a piece of clean SiO_2 (300 nm)/Si substrate is placed on the top of the boat. Another porcelain boat with sulfur powders is placed upstream. During the synthesis, the furnace and S powder are heated up separately to 825 °C and 220 °C, respectively, and held for 10 min. Argon is used as a carrier gas throughout the process. For transferring WS_2 monolayers to fabricate WS_2/Ge heterostructures, the as-grown monolayer WS_2 on SiO_2/Si is first spin-coated with a layer of CA (10 mg mL^{-1} of CA in acetone). Subsequently, CA-coated WS_2 is placed in buffered oxide etch (BOE) solution at room temperature, so that the SiO_2 layer is etched, releasing the CA/ WS_2 film. After etching, the free-floating sample is rinsed in deionized water and transferred to a clean Ge(100) substrate. Finally, the CA film is dissolved in acetone/isopropanol baths, leaving WS_2 transferred on the Ge substrate.



Synthesis and transfer of graphene

To synthesize graphene, a piece of copper foil is first placed in a quartz tube furnace, which is purged with Ar at 0.5 L m^{-1} and a 15% H_2 -Ar mixture at 0.3 L m^{-1} . The furnace is then brought to $1000 \text{ }^\circ\text{C}$ and the Cu is annealed for 30 min. Afterwards, the Ar flow rate is increased to 0.8 L m^{-1} and CH_4 is introduced at 15 sccm for 7 minutes. Then CH_4 is turned off, the Ar flow rate is returned to 0.5 L m^{-1} , and the furnace is allowed to cool naturally to room temperature. The transfer of graphene involves spin-coating graphene on Cu with poly (methyl methacrylate) (PMMA) A6. Afterwards, the Cu is etched away by floating the sample on commercial Cu etchant from Alfa Aesar. Once the Cu has been etched, the supported graphene sample is placed in three separate water baths. After the final water bath, the sample is placed on the desired substrate and allowed to dry on a hotplate at $50 \text{ }^\circ\text{C}$. To remove the PMMA, the sample is placed in acetone for 2 minutes, then dipped in isopropanol for several seconds and blow dried with nitrogen gas.

DFT methods

DFT calculations are performed to understand the experimental observations by using supercells consisting of a layer of WS_2 on a Ge slab. The generalized gradient approximation (GGA) by Perdew, Burke, and Ernzerhof (PBE)¹⁹ is used to obtain the relaxed coordinates. The calculations are performed with a plane wave kinetic energy cutoff of 612 eV with the atomic coordinates relaxed until the forces become smaller than $0.01 \text{ eV } \text{Å}^{-1}$ and a self-consistent field (scf) energy convergence of 10^{-6} eV is achieved. The $\text{WS}_2/\text{Ge}(100)$ system is constructed in an orthorhombic supercell with dimensions of $a = b = 16.3 \text{ Å}$ and $c = 50 \text{ Å}$ with a single layer of WS_2 consisting of 30 W and 60 S atoms on a Ge layer consisting of 144 Ge atoms, and the bulk WS_2 system is constructed as a hexagonal unit cell with dimensions of $a = b = 3.186 \text{ Å}$ and $c = 12.26 \text{ Å}$ consisting of 2 W and 4 S atoms.⁵⁸ The DFT-D2 correction is applied to describe the van der Waals interaction between the WS_2 layer and the Ge slab.²⁰ The density of states (DOS) is calculated to determine the contribution of the WS_2 layer in the $\text{WS}_2/\text{Ge}(100)$ system by applying the HSE06 correction²¹ to the electronic structure obtained using the PBE functional. In order to accurately describe the behavior of $\text{WS}_2/\text{Ge}(100)$ using HSE06, the screening parameter is tuned to $\mu = 0.13$ to best reproduce both the pure WS_2 (2.03 eV)⁵⁹ and Ge (0.66 eV) experimental band gaps.

Author contributions

T. Z., A. V. and D. Z. contributed equally to this work. T. Z., A. V., A. L. E., M. R. and M. T. conceived the idea and designed the experiments. T. Z., D. Z., Z. Y. and M. L. synthesized and transferred the WS_2 monolayers. T. G. N. and H. L. synthesized and transferred graphene. T. Z., A. V., D. Z., Z. Y. and K. O. D. A. performed the optical characterization. D. Z. and Y. L. performed the AFM characterization and

analysis. Z. D. W. and H. T. performed the DFT calculations. All authors discussed the results and participated in the preparation of this manuscript.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by Air Force Office of Scientific Research Grant Number FA9550-18-1-0283 and FA9550-21-1-0460. H. T. and Z. D. W. acknowledge the funding support from the USA National Science Foundation (award 2013640) and are grateful to the "Advanced Cyberinfrastructure Coordination Ecosystem: Services & Support" (ACCESS), which is supported by National Science Foundation grant number ACI-1548562 through proposal TG-DMR170008.

References

- 1 D. Lembke, S. Bertolazzi and A. Kis, *Acc. Chem. Res.*, 2015, **48**, 100–110.
- 2 M. Bernardi, M. Palumbo and J. C. Grossman, *Nano Lett.*, 2013, **13**, 3664–3670.
- 3 K. F. Mak, C. Lee, J. Hone, J. Shan and T. F. Heinz, *Phys. Rev. Lett.*, 2010, **105**, 136805.
- 4 D. Y. Qiu, F. H. da Jornada and S. G. Louie, *Phys. Rev. Lett.*, 2013, **111**, 216805.
- 5 D. Jariwala, T. J. Marks and M. C. Hersam, *Nat. Mater.*, 2017, **16**, 170–181.
- 6 M. Hong, P. Yang, X. Zhou, S. Zhao, C. Xie, J. Shi, Z. Zhang, Q. Fu and Y. Zhang, *Adv. Mater. Interfaces*, 2018, **5**, 1800641.
- 7 D. Ruzmetov, K. Zhang, G. Stan, B. Kalanyan, G. R. Bhimanapati, S. M. Eichfeld, R. A. Burke, P. B. Shah, T. P. O'Regan, F. J. Crowne, A. G. Birdwell, J. A. Robinson, A. V. Davydov and T. G. Ivanov, *ACS Nano*, 2016, **10**, 3580–3588.
- 8 S. Krishnamoorthy, E. W. Lee, C. H. Lee, Y. Zhang, W. D. McCulloch, J. M. Johnson, J. Hwang, Y. Wu and S. Rajan, *Appl. Phys. Lett.*, 2016, **109**, 183505.
- 9 S. Aftab, M. F. Khan, K. A. Min, G. Nazir, A. M. Afzal, G. Dastgeer, I. Akhtar, Y. Seo, S. Hong and J. Eom, *Nanotechnology*, 2018, **29**, 045201.
- 10 D. Sarkar, X. Xie, W. Liu, W. Cao, J. Kang, Y. Gong, S. Kraemer, P. M. Ajayan and K. Banerjee, *Nature*, 2015, **526**, 91–95.
- 11 M. L. Tsai, S. H. Su, J. K. Chang, D. S. Tsai, C. H. Chen, C. I. Wu, L. J. Li, L. J. Chen and J. H. He, *ACS Nano*, 2014, **8**, 8317–8322.
- 12 T. Akama, W. Okita, R. Nagai, C. Li, T. Kaneko and T. Kato, *Sci. Rep.*, 2017, **7**, 11967.



- 13 D. Wu, J. Guo, C. Wang, X. Ren, Y. Chen, P. Lin, L. Zeng, Z. Shi, X. J. Li, C. X. Shan and J. Jie, *ACS Nano*, 2021, **15**, 10119–10129.
- 14 W. Chen, R. Liang, S. Zhang, Y. Liu, W. Cheng, C. Sun and J. Xu, *Nano Res.*, 2019, **13**, 127–132.
- 15 H. Wang, C. Zhang, W. Chan, S. Tiwari and F. Rana, *Nat. Commun.*, 2015, **6**, 8831.
- 16 O. Lopez-Sanchez, D. Lembke, M. Kayci, A. Radenovic and A. Kis, *Nat. Nanotechnol.*, 2013, **8**, 497–501.
- 17 W. Hu, H. Cong, W. Huang, Y. Huang, L. Chen, A. Pan and C. Xue, *Light: Sci. Appl.*, 2019, **8**, 106.
- 18 L. Tang, S. E. Kocabas, S. Latif, A. K. Okyay, D.-S. Ly-Gagnon, K. C. Saraswat and D. A. B. Miller, *Nat. Photonics*, 2008, **2**, 226–229.
- 19 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865–3868.
- 20 S. Grimme, J. Antony, S. Ehrlich and H. Krieg, *J. Chem. Phys.*, 2010, **132**, 154104.
- 21 J. Heyd, G. E. Scuseria and M. Ernzerhof, *J. Chem. Phys.*, 2003, **118**, 8207–8215.
- 22 F. Zhang, Y. Lu, D. S. Schulman, T. Zhang, K. Fujisawa, Z. Lin, Y. Lei, A. L. Elias, S. Das, S. B. Sinnott and M. Terrones, *Sci. Adv.*, 2019, **5**, eaav5003.
- 23 N. Higashitarumizu, H. Kawamoto, C. J. Lee, B. H. Lin, F. H. Chu, I. Yonemori, T. Nishimura, K. Wakabayashi, W. H. Chang and K. Nagashio, *Nat. Commun.*, 2020, **11**, 2428.
- 24 T. Zhang, K. Fujisawa, T. Granzier-Nakajima, F. Zhang, Z. Lin, E. Kahn, N. Perea-López, A. L. Elías, Y.-T. Yeh and M. Terrones, *ACS Appl. Nano Mater.*, 2019, **2**, 5320–5328.
- 25 Q. Yue, J. Kang, Z. Z. Shao, X. A. Zhang, S. L. Chang, G. Wang, S. Q. Qin and J. B. Li, *Phys. Lett. A*, 2012, **376**, 1166–1170.
- 26 M. Buscema, G. A. Steele, H. S. J. van der Zant and A. Castellanos-Gomez, *Nano Res.*, 2014, **7**, 561–571.
- 27 S. Tongay, W. Fan, J. Kang, J. Park, U. Koldemir, J. Suh, D. S. Narang, K. Liu, J. Ji, J. Li, R. Sinclair and J. Wu, *Nano Lett.*, 2014, **14**, 3185–3190.
- 28 I. Gasparutti, S. H. Song, M. Neumann, X. Wei, K. Watanabe, T. Taniguchi and Y. H. Lee, *ACS Appl. Mater. Interfaces*, 2020, **12**, 7701–7709.
- 29 M. R. Rosenberger, H. J. Chuang, K. M. McCreary, A. T. Hanbicki, S. V. Sivaram and B. T. Jonker, *ACS Appl. Mater. Interfaces*, 2018, **10**, 10379–10387.
- 30 F. Wang, J. Wang, S. Guo, J. Zhang, Z. Hu and J. Chu, *Sci. Rep.*, 2017, **7**, 44712.
- 31 K. Wang, B. Huang, M. Tian, F. Ceballos, M. W. Lin, M. Mahjouri-Samani, A. Boulesbaa, A. A. Puzos, C. M. Rouleau, M. Yoon, H. Zhao, K. Xiao, G. Duscher and D. B. Geohegan, *ACS Nano*, 2016, **10**, 6612–6622.
- 32 Y. Liu, C. Liu, Z. Ma, G. Zheng, Y. Ma and Z. Sheng, *Appl. Phys. Lett.*, 2020, **117**, 233103.
- 33 H. Fang, C. Battaglia, C. Carraro, S. Nemsak, B. Ozdol, J. S. Kang, H. A. Bechtel, S. B. Desai, F. Kronast, A. A. Unal, G. Conti, C. Conlon, G. K. Palsson, M. C. Martin, A. M. Minor, C. S. Fadley, E. Yablonovitch, R. Maboudian and A. Javey, *Proc. Natl. Acad. Sci. U. S. A.*, 2014, **111**, 6198–6202.
- 34 X. Wang, W. Fan, Z. Fan, W. Dai, K. Zhu, S. Hong, Y. Sun, J. Wu and K. Liu, *Nanoscale*, 2018, **10**, 3540–3546.
- 35 Z. Liu, M. Amani, S. Najmaei, Q. Xu, X. Zou, W. Zhou, T. Yu, C. Qiu, A. G. Birdwell, F. J. Crowne, R. Vajtai, B. I. Yakobson, Z. Xia, M. Dubey, P. M. Ajayan and J. Lou, *Nat. Commun.*, 2014, **5**, 5246.
- 36 F. Liu, W. Wu, Y. Bai, S. H. Chae, Q. Li, J. Wang, J. Hone and X. Y. Zhu, *Science*, 2020, **367**, 903–906.
- 37 I. H. Malitson, *J. Opt. Soc. Am.*, 1965, **55**, 1205–1209.
- 38 D. E. Aspnes and A. A. Studna, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1983, **27**, 985–1009.
- 39 M. Buscema, G. A. Steele, H. S. J. van der Zant and A. Castellanos-Gomez, *Nano Res.*, 2015, **7**, 561–571.
- 40 H. Nan, Z. Wang, W. Wang, Z. Liang, Y. Lu, Q. Chen, D. He, P. Tan, F. Miao, X. Wang, J. Wang and Z. Ni, *ACS Nano*, 2014, **8**, 5738–5745.
- 41 C. Hou, J. Deng, J. Guan, Q. Yang, Z. Yu, Y. Lu, Z. Xu, Z. Yao and J. Zheng, *Phys. Chem. Chem. Phys.*, 2021, **23**, 24579–24588.
- 42 K. F. Mak, K. He, C. Lee, G. H. Lee, J. Hone, T. F. Heinz and J. Shan, *Nat. Mater.*, 2013, **12**, 207–211.
- 43 C. E. Giusca, I. Rungger, V. Panchal, C. Melios, Z. Lin, Y. C. Lin, E. Kahn, A. L. Elias, J. A. Robinson, M. Terrones and O. Kazakova, *ACS Nano*, 2016, **10**, 7840–7846.
- 44 Y. C. Lin, C. Y. Chang, R. K. Ghosh, J. Li, H. Zhu, R. Addou, B. Diaconescu, T. Ohta, X. Peng, N. Lu, M. J. Kim, J. T. Robinson, R. M. Wallace, T. S. Mayer, S. Datta, L. J. Li and J. A. Robinson, *Nano Lett.*, 2014, **14**, 6936–6941.
- 45 C. Lee, H. Yan, L. E. Brus, T. F. Heinz, J. Hone and S. Ryu, *ACS Nano*, 2010, **4**, 2695–2700.
- 46 T. Zhang, K. Fujisawa, F. Zhang, M. Liu, M. C. Lucking, R. N. Gontijo, Y. Lei, H. Liu, K. Crust, T. Granzier-Nakajima, H. Terrones, A. L. Elias and M. Terrones, *ACS Nano*, 2020, **14**, 4326–4335.
- 47 O. A. Ajayi, J. V. Ardelean, G. D. Shepard, J. Wang, A. Antony, T. Taniguchi, K. Watanabe, T. F. Heinz, S. Strauf, X. Y. Zhu and J. C. Hone, *2D Mater.*, 2017, **4**, 031011.
- 48 V. Carozo, Y. Wang, K. Fujisawa, B. R. Carvalho, A. McCreary, S. Feng, Z. Lin, C. Zhou, N. Perea-Lopez, A. L. Elias, B. Kabius, V. H. Crespi and M. Terrones, *Sci. Adv.*, 2017, **3**, e1602813.
- 49 Y. P. Varshni, *Physica*, 1967, **34**, 149–154.
- 50 J. Huang, T. B. Hoang and M. H. Mikkelsen, *Sci. Rep.*, 2016, **6**, 22414.
- 51 K. Zhang, B. M. Bersch, J. Joshi, R. Addou, C. R. Cormier, C. Zhang, K. Xu, N. C. Briggs, K. Wang, S. Subramanian, K. Cho, S. Fullerton-Shirey, R. M. Wallace, P. M. Vora and J. A. Robinson, *Adv. Funct. Mater.*, 2018, **28**, 1706950.
- 52 A. Splendiani, L. Sun, Y. B. Zhang, T. S. Li, J. Kim, C. Y. Chim, G. Galli and F. Wang, *Nano Lett.*, 2010, **10**, 1271–1275.
- 53 S. Sattar, N. Singh and U. Schwingenschlöggl, *Phys. Rev. B*, 2016, **94**, 205415.



- 54 A. M. Al-Sheikh and K. S. Hameed, *Rafidain J. Sci.*, 2011, **22**, 58–68.
- 55 C. A. Klein, *J. Appl. Phys.*, 1968, **39**, 2029–2038.
- 56 H. G. Ji, M. Maruyama, A. S. Aji, S. Okada, K. Matsuda and H. Ago, *Phys. Chem. Chem. Phys.*, 2018, **20**, 29790–29797.
- 57 Z. Liu, R. Z. Wang, L. M. Liu, W. M. Lau and H. Yan, *Phys. Chem. Chem. Phys.*, 2015, **17**, 11692–11699.
- 58 J. Wang and G. P. Srivastava, *Nanomaterials*, 2021, **11**, 534.
- 59 J. Gusakova, X. Wang, L. L. Shiau, A. Krivosheeva, V. Shaposhnikov, V. Borisenko, V. Gusakov and B. K. Tay, *Phys. Status Solidi A*, 2017, **214**, 1700218.

