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Introduction

Osteoporosis is defined as the systemic degradation of bone micro-architecture due to low bone mineral density (BMD) and poor bone quality. As such, this disease is associated with a high risk of fragility fractures.¹⁻³ Indeed, more than 8.9 million cases of osteoporosis-stimulated fractures are reported yearly on a global level. Considering the great influence and serious consequences of osteoporosis, this disease is classified by the World Health Organization (WHO) as a major public health concern.⁴–⁶

To restore proper bone structure and treat osteoporosis, it is important to re-establish the balance between bone resorption and bone formation. Numerous studies have investigated this issue, and many therapeutic agents have been proposed, including antiresorptives such as bisphosphonates, selective estrogen receptor modulator (SERM), and salmon calcitonin.⁷⁻⁹

The parathyroid hormone ($PTH_{1–84}$) secreted by the parathyroid glands has been suggested as an effective agent for the treatment of osteoporosis, as it is implicated in the regulation of

Synthesis and anti-osteoporosis activity of novel Teriparatide glycosylation derivatives†

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Osteoporosis is a metabolic bone disease that is characterized by low bone mass and micro-architectural deterioration of bones. The mechanism underlying this disease implicates an imbalance between bone resorption and bone remodeling. In 2002, the US Food and Drug Administration (FDA) approved Teriparatide for the treatment of osteoporosis, and so far, this compound is the only permitted osteoanabolic. However, as a structurally flexible linear peptide, this drug may be further optimized. In this study, we develop a series of novel N-acetyl glucosamine glycosylation derivatives of Teriparatide and examine their characteristics. Of the analyzed compounds, PTHG-9 exhibits enhanced helicity, greater protease stability, and increased osteoblast differentiation promoting ability compared with the original Teriparatide. Accordingly, PTHG-9 is suggested as a therapeutic candidate for postmenopausal osteoporosis (PMOP) and other related diseases. The successful development of an enhanced osteoporosis drug proves that the method proposed herein can be used to effectively enhance the chemical and biological properties of linear peptides with various biological functions. **PAPER**
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phosphorous and calcium metabolisms, as well as in neuromuscular excitation and blood clotting. In a previous study, it was shown that PTH_{1-84} has osteoanabolic activity, and that daily, subcutaneous, low-dose injections of this hormone could promote bone formation.¹⁰–¹²

Teriparatide (PTH₁₋₃₄) is a human parathyroid hormone Nterminal analogue of $PTH_{1–84}$ that is created using genetic engineering recombination technology. This analogue retains the same properties as the natural hormone, and thus, it is capable of binding to PTH/PTHrP receptors and activating downstream osteoblast signalling, which leads to enhanced bone formation and increased bone density in osteoporosis patients.^{10,11} Today, $PTH₁₋₃₄$ is the only osteoanabolic approved by the U.S. Food and Drug Administration (FDA).¹³⁻¹⁵ However, the osteoanabolic effectiveness of $PTH_{1–34}$ may be further improved, especially considering that linear peptides such as $PTH_{1–34}$ are generally characterized by low oral bioavailability, poor metabolic stability, poor membrane permeability, and rapid clearance.¹⁶ Moreover, linear peptides are usually highly unstructured, resulting in a low intrinsic secondary structure in solution, which constitutes a major problem for target binding.¹⁷

Previously, we had reported that glycosylation can improve the properties and functions of peptides. Specifically, glycosylation increases peptide hydrophilicity and oral bioavailability, forces peptides to adopt certain conformations and/or stabilizes the existing conformations, and enhances peptide resistance to proteolytic cleavage.18,19 The osteoprotegerin (OPG) peptidomimetic and N-acetyl glucosamine glycosylation strategy designed

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by our group leads to more stable secondary structures and stronger chymotrypsin resistance, resulting in greater osteoclast inhibitory activity.²⁰ Based on these results, we believe that glycosylation can also be an effective strategy for improving the chemical and biological properties of Teriparatide. In this study, by replacing all the asparagine and serine residues of Teriparatide one by one with the pre-made O-link and N-link Nacetyl glucosamine amino acid building blocks, we synthesize N-acetyl glucosamine glycosylation derivatives of Teriparatide (Fig. 1B, 4), and we demonstrate that these derivatives exhibit a more stable secondary structure conformation, increased protease resistance, and higher anti-osteoporosis activity, compared with the original compound.

Results and discussion

The Teriparatide glycosylation derivatives were synthesized using the key glycoamino acid building blocks, Fmoc-

Scheme 1 Synthetic route of Fmoc-Asn(Ac₃GlcNAc)-OH (4) and Fmoc-Ser(Ac₃GlcNAc)-OH (6). Reagents and conditions: (a) (i) acetyl chloride, r.t., 4 days; (ii) NaN₃, Bu₄NI, DCM/water, r.t., 2 h, 82.4% in 2 steps; (b) (i) H₂, Pd/C, MeOH, r.t., overnight; (ii) Fmoc-Asp-OtBu, HOBt, DIC, DMF, r.t., overnight, 71.2% in 2 steps; (c) DCM/TFA (3 : 1, v/v), r.t., 2 h, 95.3%; (d) (i) 4 Å molecular sieves, $BF_3 \cdot Et_2O$, DCM, 0 °C, 12 h; (ii) Et₃N, Fmoc-Ser-OH, DCM/MeCN (1 : 2, v/v), r.t., 3 days, 32% in 2 steps.

Scheme 2 Synthetic route of PTHG-2. Reagents and conditions: (a) 10% hydrazine hydrate/DMF, r.t., overnight; (b) TFA/EDT/TIPS/water $(95: 2: 2: 1, v/v/v/v)$, r.t., 2 h, 25%; the resin-bound peptides were protected on side chains at asterisk sites. The following protecting groups for amino acid side chains were used: tert-butyl (tBu; for Glu and Ser), 2,2,4,6,7-pentamethyldihydrobenzo-furane-5-sulfonyl (pbf; for Arg), tert-butyloxycarbonyl (tBoc; for Lys and His) and trityl (Trt; for Asn).

Asn(Ac₃GlcNAc)-OH (4)²¹⁻²³ and Fmoc-Ser(Ac₃GlcNAc)-OH (6).²³ To prepare compound 4, N -acetyl- β -p-glucosamine (1) was selected as starting material, and it was reacted with acetyl chloride to obtain the acetyl-protected chloroside intermediate. This intermediate was then reacted with sodium azide, and the two-step reaction was catalyzed by tetrabutylammonium iodide (phase transfer catalyst). The yield of the resulting product (compound 2) was 82.4%. Subsequently, the fully protected glycoamino acid 3 was formed at 71.2% yield by coupling the reduced azide group of compound 2 with Fmoc-Asp-OtBu through HOBt/DIC. Finally, compound 3 was deprotected using a mixture of TFA and DCM, and the resulting building block 4 was obtained in 95.3% yield (Scheme 1A).

Fig. 2 (A) CD spectra of the glycosylation derivatives in 50.0% TFE aqueous solution at 20 \degree C. (B) Proteolytic stability of glycosylation derivatives under a-chymotrypsin treatment. Data points are displayed as the mean value SEM of duplicate independent experiments. The percent residual peptide was monitored via HPLC.

However, building block 6 was generated by direct glycosylation between 1,3,4,6-tetra-O-acetyl-N-acetyl-β-D-glucosamine (5) and Fmoc-Ser-OH, with successive $BF_3 \cdot Et_2O$ and Et_3N treatment (Scheme 1B).

With the key glycoamino acid building blocks in hand, the Teriparatide glycosylation derivatives were synthesized. For example, PTHG-2 was synthesized by incorporating building block 6 into the peptide backbone via standard Fmoc SPPS, using *O*-(7-azabenzotriazol-1-yl)-*N,N,N',N'*-tetramethyluronium hexafluorophosphate (HATU) as a coupling reagent (Scheme 2). The acetyl protection group of the resulting on-resin fully protected peptide 7 was subsequently deprotected using hydrazine hydrate DMF solution, and on-resin intermediate 8 was produced. After acidic cleavage and global deprotection, the crude glycopeptide was obtained (Scheme 2), and then it was purified by preparative high-performance liquid chromatography. The yield of the pure PTHG-2 product was found to be 25.0%.

The secondary structures of the synthesized glycopeptides were analysed by circular dichroism (CD). As shown in Fig. 2A, the helicity values of the Teriparatide derivatives ranged between 26.1% and 39.2%, compared with 26.1% for the original Teriparatide (PTHG-1). This confirms that N -acetyl glucosamine glycosylation can improve the helicity of peptides. Among the investigated compounds, PTHG-6, PTHG-8, and PTHG-9 showed the highest degrees of helicity (39.2%, 36.7%, and 36.1%, respectively), with values that are 1.50-, 1.40-, and 1.38-fold greater than that of PTHG-1, respectively.

To assess the protease stability of Teriparatide (PTHG-1) and its derivatives, a-chymotrypsin-mediated degradation tests were performed, and the compounds were monitored by HPLC. As shown in Fig. 2B, 50% of PTHG-1 remained after 60 minutes of protease exposure, compared with 25–75% for the synthesized glycopeptides. Among the investigated compounds, only PTHG-5, PTHG-7, and PTHG-9 had half-lives longer than 60 minutes,

Fig. 3 Activity of the glycosylation derivatives at 50 nM concentrations as assessed by alkaline phosphatase (ALP) activity in MC3T3-E1 cells. Data represent the mean \pm SEM from three independent experiments: $*p < 0.01$ as compared with PTHG-1.

and thus, their protease stability is greater than that of the other peptides. These results suggest that the 16- and 30-asparagine residues play extremely significant roles in the process of Teriparatide protease degradation.

The alkaline phosphatase (ALP) activity of Teriparatide (PTHG-1) and its N-acetyl glucosamine glycosylation derivatives (PTHG-2–9) was also analysed. For this purpose, MC3T3-E1 cells were treated with different peptides at a concentration of 50 nM, followed by spectrophotometric and optical density measurements (405 nm). As shown in Fig. 3, the PTHG-1 group presented statistically significant ALP activity, which means that PTHG-1 is capable of promoting osteoblast differentiation, as previously reported in the literature. Compared with PTHG-1, the promoting activities of PTHG-2, PTHG-4, and PTHG-9 were 1.18-, 1.12-, and 1.3-times greater, respectively. However, only the stimulation effect of PTHG-9 was statistically significant. Overall, the results indicate that both protease resistance and conformation stability are important factors contributing to an enhanced osteoblast differentiation ability of Teriparatide and its glycosylation derivatives. This explains the improved activity of PTHG-9 compared with PTHG-1. RSC Advances

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Conclusions

A novel series of Teriparatide glycosylation derivatives have been designed and successfully synthesized, with reasonable yield, via the standard Fmoc SPPS strategy using pre-made glycoamino acid building blocks. In vitro data suggest that the analogue PTHG-9 has improved helicity, greater protease stability, and increased osteoblast differentiation promoting ability compared with the original Teriparatide. Therefore, PTHG-9 has great potential for use in the development of novel therapeutics for postmenopausal osteoporosis. More importantly, the derivatization method proposed herein constitutes a useful strategy for improving the chemical and biological properties of linear peptides with various biological functions. Further optimization of PTHG-9 and more in-depth biological evaluation are ongoing in our laboratory, and the results will be disclosed in due course.

Experimental section

General information

All reagents were purchased from Acros, Sigma-Aldrich, Alfa Aesar and Adamas. Amino acids were commercial available from GL Biochem Shanghai Co. Ltd. All solvents used were bought from Sinopharm Chemical Reagent Co. Ltd. Dichloromethane (DCM) and N,N-dimethylformamide (DMF) were distilled over calcium hydride $(CaH₂)$ under argon atmosphere and stored in flask containing 4 Å molecular sieves. All reactions vessels were oven-dried before use. Reactions were monitored by thin-layer chromatography (TLC) and visualized by UV analyzer (254 nm), ninhydrin and/or phosphomolybdic acid. Peptides were analyzed and purified by reverse phase HPLC. A C18 analytic column (Shimadzu Shim-pack VP-ODS, 4.6×250 mm, 5 µm particle size, flow rate 1 mL min⁻¹) was used for analytical RP-HPLC, and a C18 column (Shimadzu

Shim-pack PRC-ODS, 50×250 mm, 15 µm particle size, flow rate 10 mL min^{-1}) was used for preparative RP-HPLC. The solvents systems were buffer A (0.1% TFA in $CH₃CN$) and buffer B (0.1% TFA in water). Data were recorded and analyzed using the software system LC Solution. High resolution mass spectra (HR-MS) were measured on a Waters Xevo G2 QTOF mass spectrometer. ¹H- and ¹³C-NMR spectrum was recorded on a Bruker Avance 300 MHz instrument. Chemical shifts (δ) were reported relative to TMS (0 ppm) for $^1\mathrm{H}\text{-}$ and $^{13}\mathrm{C}\text{-}$ NMR spectra. The coupling constants (J) were displayed in Hertz (Hz) and the splitting patterns were defined as follows: singlet (s); broad singlet (s, br); doublet (d); doublet of doublet (dd); triplet (t); quartet (q); multiplet (m).

(2R,3S,4R,5R,6R)-5-Acetamido-2-(acetoxymethyl)-6-azidotetrahydro-2H-pyran-3,4-diyl diacetate (2)

To N-acetyl- β -D-glucosamine (25.0 g, 113.1 mmol) was added acetyl chloride (50 mL, 707.0 mmol), dropwise over 15 min at 0 °C. The reaction mixture was stirred vigorously at room temperature for 4 day. This mixture was diluted with DCM (100.00 mL) and saturated NaHCO₃ (100 mL) aqueous solution. The organic phase was separated, washed with saturated NaHCO₃ and brine (3 \times 100.00 mL), dried over Na₂SO₄ and concentrated, and used without further purification. A mixture of commercially available NaN_3 (8.82 g, 135.70 mmol) and tetrabutylammonium iodide (16.68 g, 45.23 mmol) in DCM/water $(1:1,$ 200.00 mL) was stirred for 2 hours at room temperature. The organic layer was separated, washed with brine $(3 \times 100.00 \text{ mL})$, dried over $Na₂SO₄$, concentrated and purified by column chromatography (20 : 1–1 : 1, petro ether/EtOAc) to give 2 as a white powder (13.80 g, 82.4%, 2 steps). ¹H-NMR (600 MHz, CDCl₃): δ 5.81 (d, J = 6 Hz, 1H), 5.27 (t, J = 12 Hz, 1H), 5.12 (t, J = 12 Hz, 1H), 4.79 (d, $J = 6$ Hz, 1H), 4.28-4.27 (m, 1H), 4.20 (d, $J = 6$ Hz, 1H), 3.96–3.91 (m, 1H), 3.83–3.81 (m, 1H), 2.12 (s, 3H), 2.05 (d, $J = 6$ Hz, 6H), 2.00 (s, 3H). ¹³C-NMR (600 MHz, CDCl₃): δ 170.48, 170.17, 169.94, 168.77, 87.91, 73.48, 71.65, 67.58, 61.36, 53.67, 22.73, 20.21, 20.12, 20.07. ESI-MS m/z calcd for C₁₄H₂₀N₄O₈ 372.1281; found $[M + H]$ ⁺ 373.29. Paper

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(2R,3S,4R,5R,6R)-6-(3-((((9H-Fluoren-9-yl)methoxy)carbonyl) amino)-4-(tert-butoxy)-4-oxobutanamido)-5-acetamido-2- (acetoxymethyl)tetrahydro-2H-pyran-3,4-diyl diacetate (3)

To a solution of 4 (10.00 g, 26.88 mmol) in MeOH (200.00 mL) was added Pd/C catalyst (1.00 g). Then the mixture was stirred overnight at room temperature under hydrogen. After the Pd/C catalyst was filtered and the MeOH was removed under vacuum, the residue was used without further purification. To a solution of the intermediate in DMF (50.00 mL) was added Fmoc-Asp-OtBu (12.14 g, 29.56 mmol), HOBt (3.99 g, 29.56 mmol) and DCC (3.99 g, 29.56 mmol) mixture of DCM/DMF (4 : 1, 200.00 mL) solution. Then DIC (3.72 g, 29.56 mmol) was added to the solution. The reaction was stirred overnight at room temperature. Then, the reaction was filtrated and the filtrate was washed successively with 1 M HCl (3×100.00 mL), saturated NaHCO₃ $(3 \times 100.00 \text{ mL})$ and brine $(3 \times 100.00 \text{ mL})$. The organic phase was dried over $Na₂SO₄$, filtered, concentrated and purified by column chromatography (100 : 1–10 : 1, DCM/MeOH) to give 3 as a white powder (14.10 g, 71.2%, 2 steps). 1 H-NMR (600 MHz, CDCl₃): δ 7.77 (d, J = 6 Hz, 2H), 7.62 (d, J = 6 Hz, 2H), 7.41 (t, J = 12 Hz, 2H), 7.32 (t, $J = 12$ Hz, 2H), 7.23 (d, $J = 6$ Hz, 1H), 6.18 (d, J $= 6$ Hz, 1H), 5.98 (d, $J = 6$ Hz, 1H), 5.14–5.09 (m, 3H), 4.55 (s, 1H), 4.44–4.43 (m, 1H), 4.33–4.31 (m, 2H), 4.24 (t, $J = 12$ Hz, 1H), 4.16–4.13 (m, 1H), 4.08 (d, $J = 6$ Hz, 1H), 3.78–3.76 (m, 1H); 2.88–2.85 (m, 1H); 2.74–2.71 (m, 1H); 2.54 (s, 2H); 2.09–2.06 (m, 9H); 1.98 (s, 3H); 1.47 (s, 9H). ¹³C-NMR (600 MHz, CDCl₃): d 171.91, 171.47, 170.60, 170.15, 169.45, 168.74, 155.64, 143.42, 143.31, 140.78, 127.20, 126.57, 124.69, 119.47, 81.75, 79.73, 73.07, 72.40, 67.11, 66.66, 61.17, 53.00, 50.51, 46.64, 37.48, 27.40, 22.53, 20.19, 20.07. ESI-MS m/z calcd for $C_{37}H_{45}N_3O_{13}$ 739.2952; found $[M + H]$ ⁺ 740.19.

N_2 -(((9H-Fluoren-9-yl)methoxy)carbonyl)- N_4 -((2R,3R,4R,5S,6R)-3-acetamido-4,5-diacetoxy-6- (acetoxymethyl)tetrahydro-2H-pyran-2-yl)asparagines (4)

3 (10.0 g, 13.53 mmol) was dissolved in TFA/DCM (1 : 1, 50.00 mL) and stirred for 2 hours at room temperature. The reaction mixture was concentrated in vacuo to yield 4 as a white powder (8.77 g, 95.3%) which was used directly in SPPS without further purification. ¹H-NMR (600 MHz, DMSO): δ 8.58 (d, J = 6 Hz, 1H), 7.88 $(d, J = 6$ Hz, 3H), 7.70 $(d, J = 6$ Hz, 2H), 7.50 $(d, J = 6$ Hz, 1H), 7.41 (t, $J = 12$ Hz, 2H), 7.32 (t, $J = 12$ Hz, 2H), 5.17 (t, $J =$ 12 Hz, 1H), 5.10 (t, $J = 12$ Hz, 1H), 4.82 (t, $J = 12$ Hz, 1H), 4.38– 4.37 (m, 1H), 4.28–4.17 (m, 4H), 3.95–3.87 (m, 7H), 2.67–2.64 (m, 1H), 1.99–1.96 (m, 7H), 1.90 (s, 3H), 1.72 (s, 3H). 13C-NMR (600 MHz, DMSO): d 172.91, 170.00, 169.77, 169.46, 169.28, 155.80, 143.76, 140.59, 127.60, 127.05, 125.22, 120.07, 78.05, 73.33, 72.25, 68.36, 65.67, 61.80, 52.09, 49.96, 46.56, 36.82, 22.55, 20.49, 20.36, 20.33. ESI-MS m/z calcd for $C_{33}H_{37}N_3O_{13}$ 683.2326; found $[M + Na]⁺$ 706.54, $[M - H]⁻$ 682.39.

L-Serine,N-[(9H-luoren-9-ylmethoxy)carbonyl]-O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl) (6)

To a solution of $1,3,4,6$ -tetra-O-acetyl-N-acetyl- β -D-glucosamine (5, 1.08 g, 4.74 mmol) and molecular sieve (1.08 g) in DCM (100 mL) was added $BF_3 \cdot Et_2O$ (1 mL, 6.26 mmol) at 0 °C. Then the reaction was stirred at room temperature for 12 h. $Et₃N$ (0.4 mL, 5.4 mmol) was added to the mixture and stirred for 10 min. A mixture of commercially available Fmoc-Ser-OH (1.2 g, 3.67 mmol) in DCM/MeCN $(1:2, 15 \text{ mL})$ was added to the reaction and stirred for 3 days at room temperature. Then the reaction was filtrated and washed with NaHCO₃ (3×100.00 mL) and brine (3 \times 100.00 mL). The organic phase was dried over $Na₂SO₄$, filtered, concentrated and purified by column chromatography $(100:1-10:1, DCM/MeOH)$ to give 6 as a white powder (14.1 g, 32%, 2 step). ¹H-NMR (600 MHz, DMSO): δ 9.32-9.31 (m, 3H), 9.16-9.14 (m, 2H), 8.84 (t, $J = 12$ Hz, 2H), 8.76-8.71 $(m, 3H), 6.51$ $(t, J = 12$ Hz, 1H $), 6.27$ $(t, J = 12$ Hz, 1H $), 6.13$ $(d, J = 12$ 6 Hz, 1H), 5.71–5.61 (m, 6H), 5.44–5.39 (m, 3H), 5.20–5.11 (m, 2H), 3.43 (s, 3H), 3.39 (s, 3H), 3.33 (s, 3H), 3.15 (s, 3H). 13C-NMR (600 MHz, DMSO): d 171.5, 170.03, 169.61, 169.52, 169.23, 155.91, 143.39, 140.69, 127.63, 127.06, 125.24, 125.19, 120.09, 72.49, 70.82, 68.45, 65.83, 61.75, 54.02, 53.02, 46.56, 22.55,

20.45, 20.37, 20.31. ESI-MS m/z calcd for $C_{32}H_{36}N_2O_{13}$ 656.2217; found $[M - H]$ ⁻ 654.24.

General procedures for the Fmoc solid phase peptide synthesis

The amino acid residues were attached to the resin with a single coupling procedure. All peptides were synthesized with a scale of 0.15 mmol.

(a) Standard pre-activation of resin protocol: the resin was swollen in DCM/DMF mixture solvent for 10 minutes.

(b) Standard Fmoc-deprotection protocol: after treatment with 20% piperidine in DMF solution (15 minutes twice) the resin was washed with DMF (\times 5), DCM (\times 5), and DMF (\times 5).

(c) Standard coupling of amino acids protocol: after preactivation of 4.0 equivalents of Fmoc-protected amino acid in DMF for 5 minutes using 3.8 equivalents of HCTU and 8.0 equivalents of DIPEA, the solution was added to the resin. After 30 minutes, the resin was washed with DMF $(\times 5)$, DCM $(\times 5)$, and DMF $(x5)$. The coupling reaction was monitored with the ninhydrin test. **PSC Advances**

20.13, 20.37, 20.11. ESI-MS *n*ce calcd for C₁₂11₃₀N₂O₂, 656.2217;

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(d) Standard acetylation protocol: a solution of acetic anhydride and pyridine $(1:1, v/v)$ was added to the resin. After 20 minutes, the resin was washed with DMF $(\times 5)$, DCM $(\times 5)$, and DMF $(\times 5)$.

(e) Standard stapling protocol: after removal of N-terminus Fmoc and acetylation successively, 3.0 equivalents of 1st Grubbs' reagent in dry dichloroethane solution was added to the resin. After overnight reaction, the resin was washed with DMF (\times 5), DCM (\times 5), and DMF (\times 5).

(f) Standard cleavage protocol: the cleavage cocktail (TFA : TIPS : phenol : $H_2O = 88 : 5 : 5 : 2$, $v/v/v/v$) was added to the resin. After stirring for 2 hours, the cleavage cocktail was collected. The solution was bubbled with argon for concentration and the chilled diethyl ether was added to precipitate the crude peptides. The peptide suspensions were centrifuged for 3 minutes at 3000 rpm and then the clear solution was decanted. The step of precipitation, centrifugation and decantation operations was repeated three times. The resulting white residues were dissolved in $CH₃CN/water$, analyzed and purified by RP-HPLC.

PTHG-1. PTHG-1 was obtained as a lyophilized white powder (154 mg, 25% yield according to initial resin load, 99.3% purity). HR-MS m/z calcd for $C_{181}H_{296}N_{57}O_{49}S_2$ 4116.1469; found $[M + 3H]^{3+} = 1373.0696; [M + 4H]^{4+} = 1030.0539;$ $[M + 5H]^{5+} = 824.2456.$

PTHG-2. PTHG-2 was obtained as a lyophilized white powder (161 mg, 25% yield according to initial resin load, 99.2% purity). HR-MS m/z calcd for C₁₈₉H₃₀₉N₅₈O₅₄S₂ 4319.2263; found [M + $3H$]³⁺ = 1440.7517; $[M + 4H]$ ⁴⁺ = 1080.8169; $[M + 5H]$ ⁵⁺ = 864.8535; $[M + 6H]^{6+} = 720.8793$; $[M + 7H]^{7+} = 618.1847$.

PTHG-3. PTHG-3 was obtained as a lyophilized white powder (181 mg, 28% yield according to initial resin load, 99.4% purity). HR-MS m/z calcd for C₁₈₉H₃₀₉N₅₈O₅₄S₂ 4319.2263; found [M + $3H]$ ³⁺ = 1440.7529; $[M + 4H]$ ⁴⁺ = 1080.8223; $[M + 5H]$ ⁵⁺ = 864.8594; $[M + 6H]^{6+} = 721.0512$; $[M + 7H]^{7+} = 618.0464$.

PTHG-4. PTHG-4 was obtained as a lyophilized white powder (168 mg, 26% yield according to initial resin load, 99.1% purity). HR-MS m/z calcd for $C_{189}H_{309}N_{58}O_{54}S_2$ 4319.2263; found [M + $3H$ ³⁺ = 1440.7514; $[M + 4H]$ ⁴⁺ = 1080.8165; $[M + 5H]$ ⁵⁺ = 864.8541; $[M + 6H]^{6+} = 720.8795$.

PTHG-5. PTHG-5 was obtained as a lyophilized white powder (181 mg, 28% yield according to initial resin load, 99.5% purity). HR-MS m/z calcd for C₁₈₉H₃₀₉N₅₈O₅₄S₂ 4319.2263; found [M + $3H]$ ³⁺ = 1440.7524; $[M + 4H]$ ⁴⁺ = 1080.8166; $[M + 5H]$ ⁵⁺ = 864.8556; $[M + 6H]^{6+} = 720.8808$.

PTHG-6. PTHG-6 was obtained as a lyophilized white powder (176 mg, 27% yield according to initial resin load, 99.3% purity). HR-MS m/z calcd for C₁₉₁H₃₁₁N₅₈O₅₅S₂ 4361.2369; found [M + $3H$]³⁺ = 1454.7605; [M + 4H]⁴⁺ = 1091.3238; [M + 5H]⁵⁺ = 873.2618; $[M + 6H]^{6+} = 727.8863$; $[M + 7H]^{7+} = 624.0468$.

PTHG-7. PTHG-7 was obtained as a lyophilized white powder (161 mg, 25% yield according to initial resin load, 99.2% purity). HR-MS m/z calcd for C₁₈₉H₃₀₉N₅₈O₅₄S₂ 4319.2263; found [M + $3H]$ ³⁺ = 1440.7519; $[M + 4H]$ ⁴⁺ = 1080.8170; $[M + 5H]$ ⁵⁺ = 864.8542; $[M + 6H]^{6+} = 720.8797$.

PTHG-8. PTHG-8 was obtained as a lyophilized white powder (183 mg, 28% yield according to initial resin load, 99.6% purity). HR-MS m/z calcd for C₁₉₁H₃₁₁N₅₈O₅₅S₂ 4361.2369; found [M + $3H$ ³⁺ = 1454.7607; $[M + 4H]$ ⁴⁺ = 1091.3232; $[M + 5H]$ ⁵⁺ = 873.2612; $[M + 6H]^{6+} = 727.8855$.

PTHG-9. PTHG-9 was obtained as a lyophilized white powder (176 mg, 26% yield according to initial resin load, 99.1% purity). HR-MS m/z calcd for C₁₉₇H₃₂₂N₅₉O₅₉S₂ 4522.3057; found [M + $3H]^{3+}$ = 1508.1172; $[M + 4H]^{4+}$ = 1131.3393; $[M + 5H]^{5+}$ = 905.2731; $[M + 6H]^{6+} = 754.5621$.

Protease stability experiment

Purified peptides was dissolved in DMSO to a final concentration of 1 mM as solution A. a-chymotrypsin was dissolved in PBS buffer ($pH = 7.4$, containing 2 mM CaCl₂) to a final concentration of 0.5 ng μL^{-1} as solution B. 50 μ L solution A and 1950 μ L solution B were mixed. After $0, 1, 2, 4, 8$ and 12 hours' incubation at 37 \degree C, the percent residual peptide was monitored by HPLC.

CD spectroscopy study

CD measurements were recorded on a JASCO J-820 spectropolarimeter (JASCO Corp., Ltd). Peptides were dissolved in 50.0% TFE aqueous solution with a concentration of 0.1 mg mL^{-1} . UV spectra were recorded at 20 \degree C in a quartz cell of 10.0 mm path length. Percent helicity (α) was calculated by the follow eqn (1).^{24,25}

$$
\alpha = \frac{[\theta]_{222}}{[\theta]_{\text{max}}} \times 100\%
$$
 (1)

 $[\theta]_{222}$ is the molar ellipiticity of 222 nm; $[\theta]_{\text{max}} = (-44000 + 250T)$ $(1 - k/n)$, $k = 4$, *n* is the numbers of amino acids and $T = 20$ °C.

Alkaline phosphatase assay

At 70-80% confluency mice calverial osteoblast were trypsinized and 2×10^3 cells per well were seeded onto 96-well plates for alkaline phosphatase (ALP) measurement. Cells were treated with different concentration of test compound or vehicle for

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48 h in α -MEM supplemented with 10% charcoal treated FCS, 10 mM β-glycerophosphate, 50 μg mL⁻¹ ascorbic acid, and 1% penicillin/streptomycin (osteoblast differentiation medium). At the end of incubation period, total ALP activity was measured using p-nitrophenyl phosphate (PNPP) as substrate and absorbance was read at 405 nm.

Conflicts of interest

There are no conflicts of interest to declare.

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