**Introduction**

The direct generation of chemical fuels from sunlight is a major scientific challenge for the development of a sustainable economy and production of hydrogen through water splitting. In green plants, photosynthesis operates in four steps: (1) light collection, (2) charge separation, (3) charge accumulation and (4) conversion to chemical fuel. Charge accumulation is a key step as most of the redox reactions for fuel generation such as \( \text{H}_2 \) production from water are multi-electronic processes.\(^1\)

While various noble-metal-based complexes (e.g. Pd, Pt...) have been widely employed as reduction catalysts in artificial photosynthetic systems, polyoxometalates have recently emerged in this field owing to their electron reservoir abilities and the activity of their reduced forms in the hydrogen evolution reaction.\(^2\) Polyoxometalates (POMs) form a remarkable class of well-defined molecular nano-scale oxoclusters of the early transition metals with an unmatched diversity of structures and properties.\(^3\) Their implementation in artificial photosynthetic devices requires their association with a visible-range antenna since POMs only absorb in the UV region of the solar spectrum.

Owing to their anionic nature, POMs have mainly been associated with photosensitizer (PS) through electrostatic interactions.\(^6\)\(^\text{–}^9\) By comparison, few covalent organic–inorganic POM-based hybrids have been developed.\(^10\)\(^\text{–}^13\) While the covalent functionalization of POMs is synthetically more demanding, this approach enables fine control between the different subunits of the system, which is required for tuning the kinetics of photoinduced electron transfer between the excited chromophore and the POM.\(^1\)

We previously reported the synthesis and photophysical properties of different POM–PS conjugates.\(^18\)\(^\text{–}^23\) Among them heteroleptic carbocyclometalated iridium(m)–POM dyads offered the most promising photophysical performance.\(^24\) In these compounds, photoinduced charge-separated excited states of various lifetimes (ranging from nanoseconds to hundreds of nanoseconds, the longest values reported for covalently bonded photosensitized POMs) were observed by transient absorption spectroscopy. Furthermore, in the presence of a sacrificial electron donor (triethylamine) and a proton source (acetic acid), the system is capable of photoaccumulating two electrons on the POM and produces hydrogen.\(^25\) However, the POM-Ir(m) system suffered from the presence of a noble metal (Ir) and the low stability of the iridium complex in the presence of strong acid, which is necessary to improve the electron reservoir properties of the POM.

Furthermore, in order to incorporate the POM–PS hybrids into a photocathode, the functionalization of the PS by an appropriate anchoring group (carboxylic acid, phosphonate...) is a required step, which has been very scarcely developed with carbocyclometalated iridium(m) complexes.\(^24\) In this context, bodipy fluorophores have been identified as adequate candidates because of their tunable photophysical properties, high...
chemical stability, easy and versatile chemical modification such as the addition of a grafting function.\(^2\) Here, we present the first examples of Keggin-type POMs that are covalently bound to one or two bodipy units via organic tethers of different lengths (Scheme 1). The series of three hybrids display distinct redox behaviour both on the POM and on the bodipy units. The photophysical properties of these hybrids are reported and discussed on the basis of the required kinetics that should be achieved for their implementation in a photoelectrocatalytic device.

**Results and discussion**

**Synthesis of the different POM–bodipy conjugates**

We previously developed three Keggin-based polyoxometalate hybrid platforms bearing one or two remote iodo aryl functions. These hybrids vary according to the nature of the metal ion (polyoxotungstate vs. polyoxomolybdate) and the nature of the primary functionalization (silyl vs. tin).\(^2\)\(^3\)\(^4\)\(^5\)\(^6\)\(^7\) The organosilyl POM derivative (noted \(K_{Si}^W[I]\)) is more easily reducible than its organotin counterparts due to a lower charge. Furthermore, the reduction processes are more accessible on the polyoxomolybdate organotin derivative (noted \(K_{Sn}^{Mo}[I]\)) compared to acetonitrile, which outlines a very important precipitation. Consequently, we were not able to obtain \(K_{Si}^W[BOD]_2\) with the required purity. The organotin functionalization of POMs leads to the formation of robust hybrids since the lacuna of monovacant POMs is generally well-suited for the incorporation of one Sn\(^{4+}\) atom.\(^3\) We thus investigated the functionalization of the organotin polyoxotungstate and polyoxomolybdate platforms \(K_{Sn}^W[I]\) and \(K_{Sn}^{Mo}[I]\) with \(BOD_2-H\). In both cases stable hybrids of appropriate purity were obtained. The coupling reactions between the POM hybrid platforms bearing the iodoaryl moieties (\(i.e. K_{Si}^W[I]\), \(K_{Sn}^{Mo}[I]\)) were performed following conditions that we previously established. Typically, Sonogashira C–C cross coupling occurs at 70 °C in ca. 30 min under microwave activation, in DMF containing triethylamine (20 equiv.), using [Pd(PPh)\(_3\)Cl] (8–10%) as catalyst source and Cu (8–10%) as co-catalyst. An excess of the alkyne terminated-bodipy is required for the total conversion of the starting POM. For the synthesis of \(K_{Sn}^W[BOD_2]\) and \(K_{Sn}^{Mo}[BOD_2]\) a second loading of \(BOD_2-H\) into the reaction mixture was performed after 15 min, due to the inevitable decomposition of \(BOD_2-H\) under these conditions.

**Electrochemistry**

The redox properties of the POM-bodipy dyads, their related POM hybrids and bodipy references were investigated by cyclic voltammetry and, in cases of irreversible redox processes, by differential pulse voltammetry in deoxygenated dichloromethane (DCM) with tetrabutylammonium hexafluorophosphate (TBAPF\(_6\)) as the supporting electrolyte in a standard three-electrode cell, composed of a glassy carbon working electrode, a platinum counter electrode, and a saturated calomel reference electrode (SCE) (Fig. 1 and Table 1).

First, it is noted that the redox potentials of the starting POM-based platforms are considerably shifted in DCM compared to acetonitrile,\(^1\) which outlines a very important effect of the organic solvent in the reduction processes of the
While the oxidation potential for each are rather similar, reversible monoelectronic oxidation and reduction processes. BOD1-TMS by more than half a volt due to a one-charge difference between both oxoclusters. BOD1-TMS and BOD2-TMS both feature reversible monoelectronic oxidation and reduction processes. While the oxidation potential for each are rather similar, reduction of BOD2-TMS occurs at a significantly more positive potential than reduction of BOD2-TMS (ΔE_red = 300 mV).

Indeed, in BOD2-TMS the phenyl unit is twisted for steric reasons, which decouples it from the π system of the bodipy unit,8 while in BOD1-TMS, the π system of the bodipy unit extends over the alkynyl moieties, favouring its reduction. The redox properties of KSnW[BOD1] are very close to those of the reference compounds KSnW[I] and BOD2-TMS, as is often reported with our POM hybrid platforms.1,12,23,26,30 The shape of the oxidation waves of KSnW[BOD1] is characteristic of adsorption of the oxidized species at the working electrode. The redox properties of KSn- [BOD2] and KSnMo [BOD2] are also similar to those of the reference compounds KSnW[I], KSnMoW[I] and BOD2-TMS. As for BOD2-TMS, the oxidation of KSnW[BOD2] and KSnMoW[BOD2] are irreversible. In the reduction part, KSnW[BOD2] displays two quasi-reversible processes followed by an irreversible process. The first one is attributed of the reduction of the bodipy unit while the two others correspond to monoelectronic reductions of the POM framework. In the reduction part, KSnMo [BOD2] displays two apparent quasi-reversible processes. The first one at −0.57 V vs. SCE is attributed to the reduction of the polyoxomolybdate framework, while the second at −1.04 V vs. SCE is attributed to quasi-simultaneous reduction of the bodipy and the one-electron reduced POM (note that in differential pulse voltammetry experiment these two reduction processes are slightly separated with a maximum at −1.04 V and a shoulder at ca. −1.00 V vs. SCE attributed to the one-electron reduced POM and bodipy reduction respectively, Fig. S14†). The reduction potential comparison of KSnMo [BOD2] versus KSnW[BOD2] thus suggests that the polyoxomolybdate core of KSnMo [BOD2] is more readily reduced than the polyoxotungstate core of KSnW[BOD2] by the BOD2 excited state.

Electronic absorption and photophysical properties

The electronic absorption spectra for bodipy reference compounds and POM–bodipy hybrids dissolved in DCM are shown in Fig. 2. The absorption spectra of the hybrids are dominated by the bodipy absorption in the visible range, since the POM itself only absorbs in the UV part. The absorption profiles show a typical sharp bodipy-centred S0–S1 absorption with a maximum at 527 nm for BOD1-TMS and KSnW[BOD1]. For

![Figure 1: Cyclic voltammograms of ca. 0.5 mM solutions of reported compounds in DCM containing 0.1 M of TBAPF6.](image)

![Figure 2: Absorbance and normalised emission (inset) of BOD2-TMS (dotted red), BOD2-TMS (dotted blue), KSnW[BOD2] (plain red), KSnMoW[BOD2] (plain blue) and KSnMo[BOD2] (plain purple) in DCM.](image)

<table>
<thead>
<tr>
<th>Compound</th>
<th>[BOD]+/[BOD]−</th>
<th>[BOD]−/[BOD]+</th>
<th>POM/POM + 1 e</th>
<th>POM + 1 e/POM + 2 e</th>
</tr>
</thead>
<tbody>
<tr>
<td>BOD1-TMS</td>
<td>1.00 (80)</td>
<td>−1.35 (80)</td>
<td>−0.49 (70)</td>
<td>−1.00 (70)</td>
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<tr>
<td>KSnW[I]</td>
<td>0.98</td>
<td>−1.34 (80)</td>
<td>−0.49 (80)</td>
<td>−1.00 (70)</td>
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<tr>
<td>KSnW[BOD1]</td>
<td>1.02</td>
<td>−1.05 (80)</td>
<td>−1.10 (80)</td>
<td>−1.58 (90)</td>
</tr>
<tr>
<td>KSnMoW[I]</td>
<td>1.02</td>
<td>−1.00</td>
<td>−1.11</td>
<td>−1.60</td>
</tr>
<tr>
<td>KSnMoW[BOD2]</td>
<td>0.98</td>
<td>−1.00</td>
<td>−0.58 (80)</td>
<td>−1.03</td>
</tr>
</tbody>
</table>

* The standard potentials of irreversible processes were determined by differential pulse voltammetry experiments (Fig. S7–S14).
BOD$_2$-TMS, K$_{Si}^{W}$[BOD$_2$] and K$_{Sn}^{Mo}$[BOD$_2$], the absorption profile is red-shifted due to an increase of the π system resulting from the presence of the alkyne group at the meso-position. Note that the maximum bodipy absorption of K$_{Si}^{W}$[BOD$_2$] and K$_{Sn}^{Mo}$[BOD$_2$] is slightly red-shifted (5 nm) compared to BOD$_2$-TMS due to the participation of the aryl tin unit into the π system of the bodipy unit. A summary of the photophysical data is presented in Table 2.

The fluorescence quantum yield for K$_{Si}^{W}$[BOD$_2$] and K$_{Sn}^{Mo}$[BOD$_2$] decrease by ca. 90% of that of BOD$_2$-TMS and BOD$_2$-TMS, respectively. This quenching of the bodipy fluorescence is consistent with electron transfer to the polyoxometalate. For K$_{Si}^{W}$[BOD$_2$], the fluorescence quantum yield decreases by ca. 60% relative to BOD$_2$-TMS, suggesting charge-transfer (if operating) from bodipy to POM is less efficient (note that the presence of the aryl tin unit that participates into the π system of the bodipy unit slightly modifies its electronic properties and may account for the decrease in quantum yield of K$_{Sn}^{W}$[BOD$_2$]).

To probe the fate of the bodipy excited state in these hybrids, transient optical and infrared absorption spectroscopy was performed on DCM solutions of the hybrids. The results are summarised in Table 3. To extract the intermediate, global analysis of the spectra was performed using the program OPT-MUS, in which the transients at all detection wavelengths are analysed simultaneously with a single set of exponential functions. The transient absorption spectra of K$_{Sn}^{W}$[BOD$_2$] at various time delays after excitation at 540 nm are shown in Fig. 3. At short delay times, a bleach centered at 520 nm forms, consistent with electron transfer to the polyoxometalate. For K$_{Si}^{W}$[BOD$_2$], the first transient species resembles the excited bodipy (spectra for BOD$_2$-TMS and BOD$_2$-TMS are provided in the ESI for comparison, Fig. S16 and 17) and decays with $\tau_1 = 54$ ps, as a second transient species with a sharp absorption at 400 nm grows in over the same timescale and decays with $\tau_2 = 4.8$ ns. The absorption profile of this second transient is consistent with the oxidised bodipy generated by spectrophotometric methods (Fig. S15†). Furthermore, this transient also features a broad absorption starting at 600 nm and extending past 700 nm, which is characteristic of a reduced polyoxometalate, and is hence assigned to the charge-separated state, [BOD$^+$-POM(+1 e$^-$)]$^-$.

The transient spectra for K$_{Sn}^{Mo}$[BOD$_2$] in DCM were much simpler (Fig. 5) since electron transfer from the BOD$_2$ to the polyoxotungstate core is thermodynamically not favourable. These transient spectra are initially similar in shape to those of the early transient spectra of K$_{Sn}^{Mo}$[BOD$_2$], in agreement with a bodipy centred excited state and decays with $\tau_1 = 3.4$ ns. A second minor transient signal at time delays exceeding 10 ns, which resembles that of the long-lived species observed with K$_{Sn}^{Mo}$[BOD$_2$], is observed ($\tau_2 = 410$ ns). This suggests a small concentration of the bodipy centered triplet state is formed, possibly caused by a heavy atom effect from the appended POM.

| Table 2 | Photophysical data for bodipy references and POM–bodipy hybrids, recorded in DCM. $E_{0-0}$ was calculated from the intercept between the absorption and emission spectra |
|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|
|                | Absorption $\lambda_{max}$ (nm) | Emission $\lambda_{max}$ (nm) | $E_{0-0}$ (eV) | $\Phi_{FL}$ |
| BOD$_2$-TMS    | 527             | 540             | 2.33           | 0.67           |
| BOD$_2$-TMS    | 574             | 590             | 2.14           | 0.52           |
| K$_{Si}^{W}$[BOD$_2$] | 527             | 541             | 2.33           | 0.07           |
| K$_{Sn}^{W}$[BOD$_2$] | 579             | 600             | 2.10           | 0.22           |
| K$_{Sn}^{Mo}$[BOD$_2$] | 579             | 597             | 2.11           | 0.06           |

Table 3 Driving force for charge-separation ($\Delta G_{CS}$) and recombination ($\Delta G_{rec}$) for POM–bodipy hybrids. $\Delta G_{CS} = E(BOD^+/BOD^*) - E(POM/POM^*)$, $\Delta G_{rec} = E(POM/POM^*) - E(BOD^+/BOD^*)$, $E(D/D^*) - E(D/D^*)$ (Note that the work terms for electrostatic interactions are neglected since they are estimated to be below 0.1 eV).

<table>
<thead>
<tr>
<th></th>
<th>$\Delta G_{CS}$ (eV)</th>
<th>$\Delta G_{rec}$ (eV)</th>
<th>$\tau_{CS}$ (ps)</th>
<th>$\tau_{rec}$ (ns)</th>
</tr>
</thead>
<tbody>
<tr>
<td>K$_{Si}^{W}$[BOD$_2$]</td>
<td>0.86</td>
<td>1.47</td>
<td>54</td>
<td>4.8</td>
</tr>
<tr>
<td>K$_{Sn}^{W}$[BOD$_2$]</td>
<td>-0.03</td>
<td>2.13</td>
<td>—</td>
<td>—</td>
</tr>
<tr>
<td>K$_{Sn}^{Mo}$[BOD$_2$]</td>
<td>0.56</td>
<td>1.55</td>
<td>180</td>
<td>0.52</td>
</tr>
</tbody>
</table>

**Fig. 3** Transient absorption spectra of K$_{Sn}^{W}$[BOD$_2$] in DCM following excitation at 540 nm: (a) evolution associated difference spectra and; (b) decay associated difference spectra derived from global analysis; (c) transient absorption difference spectra at selected time delays after excitation, with single-point decays at 405 nm, 443 nm and 688 nm (inset).
component of the POM–bodipy hybrids and determine the driving forces for charge-separation and recombination, as presented in Table 3.

For KSn\textsubscript{W}[BOD\textsubscript{1}] and KSn\textsubscript{Mo}[BOD\textsubscript{3}] there is a large driving force for charge-separation (–ΔG\textsubscript{CS}, 0.86 and 0.56 eV respectively), which occurs with a significantly faster rate for KSn\textsubscript{W}[BOD\textsubscript{1}] than KSn\textsubscript{Mo}[BOD\textsubscript{2}]. This is consistent with charge separation in the Marcus normal region (as described for the POM–iridium hybrids).\textsuperscript{23} In KSn\textsubscript{W}[BOD\textsubscript{1}], the organic spacer between the POM and the bodipy introduces an additional phenyl ring compared to KSn\textsubscript{Mo}[BOD\textsubscript{3}] increasing thus the POM–bodipy distance of ca. 4 Å. Furthermore, in KSn\textsubscript{W}[BOD\textsubscript{1}], the organic spacer is electronically decoupled from the π system of the bodipy unit and thus acts as an insulator.

By contrast, in KSn\textsubscript{Mo}[BOD\textsubscript{3}] the π system of the bodipy unit extends over the aryl tin ring \textit{i.e.} at the vicinity of the polynuclear system in the non-adiabatic limit (\textit{i.e.} when its different elements are poorly electronically coupled) can be expressed by the following equation:\textsuperscript{35,36}

\[ k_{\text{ET}} = r \exp \left( \frac{-\Delta G^*}{k_B T} \right) \]

where \( r \) is an electronic factor that is proportional to the overlap between the electronic wavefunctions of the donor and acceptor units. This electronic factor should be considerably more important for KSn\textsubscript{Mo}[BOD\textsubscript{3}] than for KSn\textsubscript{W}[BOD\textsubscript{1}]. The fact that photoinduced charge separation is significantly faster in KSn\textsubscript{W}[BOD\textsubscript{1}] than in KSn\textsubscript{Mo}[BOD\textsubscript{3}], despite a lower electronic factor term, outlines the importance of the driving force (imposed by the redox potentials of the POM and the bodipy) on the photoinduced electron transfer in these molecular systems. In other words, a short organic spacer is not mandatory for efficient charge injection if the energy levels are well positioned. The photophysics (\textit{i.e.} rapid formation and long-lifetime of the charge-separated state) of the photosensitised POM-based hybrids described here, particularly KSn\textsubscript{W}[BOD\textsubscript{1}], are promising for applications in photoelectrochemical solar cells. Dye-
sensitized NiO-based photocathodes are limited by the fast (<ns) rate of charge recombination at the oxide–dye interface.\textsuperscript{17,18} The tune-ability of the electronic properties of the photosensitizer and POM subunits to control the charge-transfer dynamics is, therefore, extremely attractive for implementing photosensitized POM-based hybrids in molecular photocathodes.

Conclusions

New POM–bodipy conjugates were synthesized through post-functionalization of organosilyl and organotin POM derivatives. An advantage of these systems over e.g. dye-sensitized TiO\textsubscript{2}, is that the electronic properties of both the donor and acceptor can be specifically tuned. In these photoactive hybrids the redox properties of the POM, the bodipy and the spacer length were modified in order to evaluate the effect of these parameters on the kinetics of photoinduced electron transfer. The transient absorption spectroscopy unequivocally indicates the occurrence of photoinduced electron transfer from the bodipy to the POM for hybrids displaying the best electron accepting properties, with kinetics up to \(2 \times 10^{-10}\) s\(^{-1}\), constituting the first example of charge-separated state on noble metal-free covalent POM–PS conjugates. While POMs are drawing a growing attention in the field of artificial photosynthesis and molecular electronics, fundamental insights on their kinetics of intramolecular electron injection into the POM unit within POM–PS conjugates are scarce.\textsuperscript{18,23,39,40} For instance the effects of the solvent and their associated counter-ions on their reorganization energy remain unexplored, the only experimental studies being yet limited to outer-sphere chemical and electrochemical reduction of POMs.\textsuperscript{41,42} The present system should allow filling this shortfall owing to their modular design. The long lifetime of the charge separated state is also an exciting prospect for integrating these systems into photocathodes, since electron-hopping between POM units or charge-transfer to a substrate or catalyst would be competitive with recombination.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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Notes and references

\dagger Acronyms used for the POM hybrids: K or D refers to the Keggin or the Dawson-type anion, Mo or W as superscript indicates the metal of the POM backbone, Si, or Sn, as subscript relates to the primary functionalization and the term in brackets corresponds to the remote organic moiety.


