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Catalytic asymmetric hydroxylative dearomatization of 2-naphthols: synthesis of lacinilene derivatives†

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An enantioselective hydroxylative dearomatization of 2-naphthols with oxaziridines has been accomplished using a *N,N'*-dioxide–scandium(III) complex catalyst. Various substituted *ortho*-quinols could be obtained in high yields (up to 99%) and enantioselectivities (up to 95 : 5 er). This methodology could be applied in the synthesis of bioactive lacinilenes in a gram-scale reaction. Based on the experimental investigations and previous work, a possible catalytic model was proposed.

Introduction

Substituted *ortho*-quinols are essential structural motifs in a number of natural products and pharmaceuticals.¹ For instance, chiral lacinilene derivatives (Fig. 1), a series of phytoalexins isolated from cotton plants, have been utilized for inhibiting the growth of cotton bacterial pathogens, such as *Xanthomonas campestris* or *malvacearum*.² Studies have showed that the (*S*)-enantiomer of lacinilene C is more active than the (*R*)-enantiomer.^{2c} While these biological activities provide a justification for the development of approaches to the synthesis of enantiomerically enriched lacinilene derivatives, novel catalytic enantioselective methods remain limited.^{2b,d}

Optically active lacinilene derivatives in nature were proposed to be produced enzymically from the oxidation of dihydroxycadalenes, thus it is of practical interest to discover a catalytic asymmetric oxidative dearomatization route to the synthesis of these cadinanes.³ Compared with other successful dearomatization events of phenols or naphthols,^{4,5} controlling the chemo-, regio- and enantioselectivity of the asymmetric

hydroxylative dearomatization is more difficult,⁶ as there might be serious side reactions in the presence of oxidants including overoxidation of alkene functions, competitive *para*-oxidation and homocoupling.^{6a,e} Additionally, the *ortho*-quinol product could undergo an unexpected α -ketol rearrangement, which enhances the difficulty of controlling the reactivity and selectivity.^{6a,7} In this respect, only a few reports related to asymmetric hydroxylative dearomatization of phenols or naphthols have been reported. Asymmetric oxidative dearomatization of phenolate mediated by copper–sparteine–dioxygen complexes followed a [4 + 2] dimerization cascade, giving bicyclo[2.2.2] octenones as the final products.^{6a} Several chiral hypervalent organoiodine compounds were developed for the asymmetric hydroxylative dearomatization of phenols and 1-naphthols.^{6b–e} Taking these examples into account, we want to engage in discovering new enantioselective strategies for the synthesis of *ortho*-quinol moieties with improved efficiency and selectivity. Here, we present an efficient asymmetric hydroxylative dearomatization of 2-naphthols catalyzed by a chiral *N,N'*-dioxide–scandium(III) complex catalyst.⁸ The process could be applied to the synthesis of various 1-hydroxy-1-alkyl-naphthalen-2-one derivatives including lacinilene C methyl ether and lacinilene D, in high to excellent yields and good enantioselectivities under mild reaction conditions (Scheme 1).

Results and discussion

We selected the hydroxylative dearomatization of 1-methylnaphthalen-2-ol **1a** as the model substrate using 3-phenyl-2-tosyl-1,2-oxaziridine **2a** as the oxidant which was proven to be chemoselective as a phase-transfer-catalyst under basic conditions (Table 1).^{7a} Initially, the catalytic asymmetric reaction was performed with 10 mol% of chiral *N,N'*-dioxide **L-PiPr₂-Sc(OTf)₃** complex in DCM at 30 °C, and the desired product **3a** could be obtained dominantly with 80 : 20 er while the α -ketol rearrangement byproduct **4a** was isolated in around one-fourth of

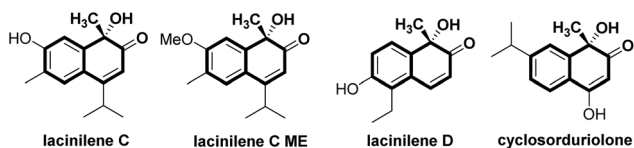


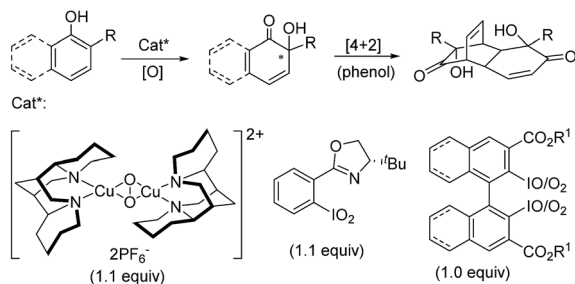
Fig. 1 Representative active lacinilene derivatives bearing *ortho*-quinol structures.

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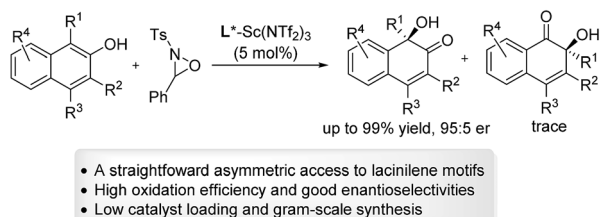
† Electronic supplementary information (ESI) available. CCDC 1536822. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c7sc02809a



Previous work: Asymmetric hydroxylative dearomatization of phenols and 1-naphthols



This work: Asymmetric hydroxylative dearomatization of 2-naphthols



Scheme 1 Catalytic asymmetric hydroxylative dearomatization of phenols and naphthols.

Table 1 Optimization of the reaction conditions^a

1a (±)-2a

metal salt (10 mol%)
L* (10 mol%)
DCM, 30 °C, 3 h

3a 4a

L-PrPr₂: n = 1, R = 2,6-*i*-Pr₂C₆H₃
L-PiPr₂: n = 2, R = 2,6-*i*-Pr₂C₆H₃
L-PiPr₃: n = 2, R = 2,4,6-*i*-Pr₃C₆H₂
L-PiMe₂: n = 2, R = 2,6-Me₂C₆H₃
L-RaPr₂: R = 2,6-*i*-Pr₂C₆H₃

Entry	Metal salt	L*	Yield ^b (%)	Ratio (3a/4a) ^c	er (3a) ^c
1	Sc(OTf) ₃	L-PiPr ₂	96	73 : 27	80 : 20
2	Sc(OTf) ₃	L-PrPr ₂	99	79 : 21	63 : 37
3	Sc(OTf) ₃	L-RaPr ₂	99	75 : 25	53.5 : 46.5
4	Sc(OTf) ₃	L-PiMe ₂	90	>95 : 5	60 : 40
5	Sc(OTf) ₃	L-PiPr ₃	96	89 : 11	73 : 27
6	Sc(NTf ₂) ₃	L-PiPr ₂	99	>95 : 5	92 : 8
7 ^d	Sc(NTf ₂) ₃	L-PiPr ₂	99	>95 : 5	95 : 5
8 ^e	Sc(NTf ₂) ₃	L-PiPr ₂	86	>95 : 5	93.5 : 6.5
9 ^{d,f}	Sc(NTf ₂) ₃	L-PiPr ₂	99	>95 : 5	94.5 : 5.5

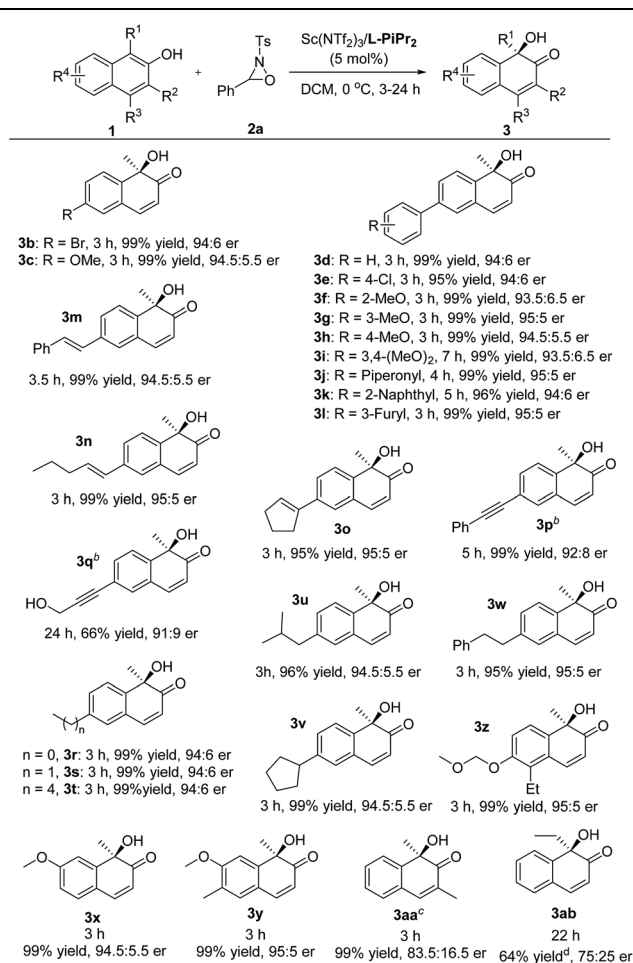
^a Unless otherwise noted, the reactions were performed with L*/Sc(m) (1 : 1, 10 mol%), **1a** (0.10 mmol) and **2a** (2.0 equiv.) in DCM (1.0 mL) under N₂ at 30 °C for 3 h. ^b Isolated yield by silica gel chromatography. ^c Determined by chiral HPLC analysis. ^d 5 mol% catalyst loading at 0 °C. ^e 1 mol% catalyst loading at 0 °C for 4 h. ^f **2a** (1.5 equiv.) was used.

a 96% total yield (Table 1, entry 1). The evaluation of the structure of the *N,N'*-dioxides showed that L-PiPr₂ was the optimal ligand in terms of the enantioselectivity albeit ligand L-PiMe₂ and L-PiPr₃ improved the yield of the desired product **3a** (entries 2–5). Fortunately, changing the counterion of the scandium salt from ⁻OTf to ⁻NTf₂ could suppress the α -ketol rearrangement, delivering the quinol **3a** in a 99% yield with

92 : 8 er (Table 1, entry 6). Further optimization of the reaction conditions, such as decreasing the temperature and the catalyst loading to 5 mol%, resulted in slightly improved enantioselectivity with maintained efficiency (entry 7). Lowering the catalyst loading to 1 mol% or the amount of the oxidant **2a** decreased either the yield or the selectivity a little (entries 8 and 9). We therefore chose the reaction conditions in Table 1, entry 7 for further studies.

We next explored the substrate scope of 2-naphthols (Table 2). The introduction of bromo or methoxy groups at the C6-position of 2-naphthols had no obvious effect on the result. The 6-aryl substituted 2-naphthol derivatives **1d–1l** tethering various electron-donating and electron-withdrawing substituents could undergo the transformations smoothly, providing the products **3d–3l** in 95–99% yield and 93.5 : 6.5–95 : 5 er. It was noteworthy that 6-alkenyl and alkynyl substituted substrates **1m–1q** were compatible with the reaction conditions, and no aminohydroxylation of the unsaturated carbon–carbon bond occurred, giving the hydroxylative dearomatization products **3m–3q** in good to excellent yields and enantioselectivities.⁹

Table 2 Substrate scope for 2-naphthols^a



^a Reaction conditions: the same as entry 7 in Table 1. ^b 10 mol% catalyst loading. ^c L-PiEt₂-Sc(OTf)₃ (1 : 1, 5 mol%). ^d Total yield of **3ab** and **4ab**, **3ab/4ab** = 87 : 13.



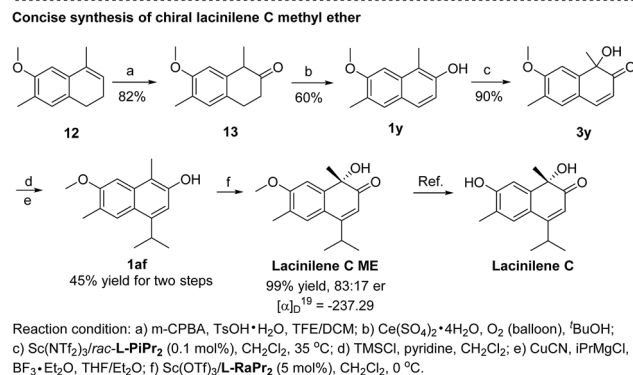
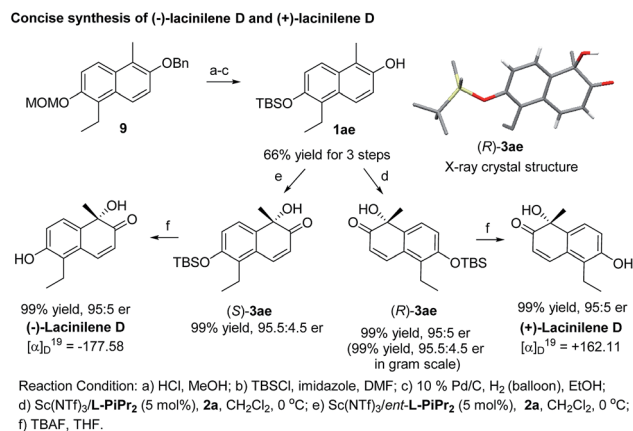
Additionally, 6-alkyl substituted 2-naphthols **3r–3w** bearing methyl, ethyl, and butyl groups were well tolerated, accomplishing the asymmetric hydroxylative reaction with the outcomes of 95–99% yield and 94 : 6–95 : 5 er. The installation of substituents to the 5- and 7-positions did not influence the reaction efficiency (**3x–3z**). The MOM-protected substrate **1z** could deliver the desired product **3z** with good results without any deprotection process occurring under the reaction conditions. However, the increase of steric hindrance at the *ortho*-position of 2-naphthol was harmful as a consequence (**3aa** and **3ab**).

To show the synthetic utility of the current catalyst system, asymmetric synthesis of bioactive lacinilenes was carried out (Scheme 2). Initially, the direct deprotection of the product **3z** under acidic conditions formed the optically active lacinilene D, but an aromatization side product 1-ethyl-5-methylnaphthalene-2,6-diol was obtained.^{24,10} It was anticipated that the TBS protecting group could be easily removed under neutral conditions, which might avoid the occurrence of the aromatization process. As expected, the TBS-substituted 2-naphthol **1ae** could be easily synthesized from **9** in 66% yield after 3 steps, which was further enantioselectively oxidized into the product **3ae** in quantitative yield and 95 : 5 er, even when it was performed at the gram scale. The absolute configuration of **3ae** from **L-PIPr₂**-Sc(NTf₂)₃ complex catalysis was determined to be (*R*) by X-ray crystal diffraction analysis.¹¹ For the benefit of the further differential biological activity study on

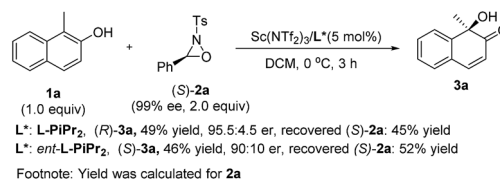
each enantiomer of the chiral lacinilenes,^{2c} (*S*)-lacinilene D was synthesized using an *ent*-**L-PIPr₂**-Sc(NTf₂)₃ complex with a comparable result of 99% yield and 95 : 5 er. Next, the synthesis of optically active lacinilene C methyl ether was explored. The synthetic route began from 1,2-dihydronaphthalene **12**, which could be easily accessed from 2-methoxytoluene through a four-step protocol.^{2d} Subsequent two-step oxidation could afford the 2-naphthol derivative **1y** in 49% yield, which underwent hydroxylative dearomatization catalyzed by 0.1 mol% of the Sc(NTf₂)₃/*rac*-**L-PIPr₂** complex to produce racemic lacinilene **3y** in 90% yield.^{2d} After trimethylsilylation and copper catalyzed 1,4-addition/aromatization, 2-naphthol **1af** could be attained in 45% yield after two steps. By treatment with oxaziridine **2a** in the presence of Sc(OTf)₃/**L-RaPr₂**, chiral lacinilene C methyl ether could be obtained in quantitative yield and 83 : 17 er, which could further transform to lacinilene C according to the literature.^{2b}

To elucidate the stereochemical course of the oxidation process, some control experiments were conducted (Scheme 3). The optically pure oxaziridine (*S*)-**2a** reacted with 2-naphthol **1a** in the presence of the Sc(NTf₂)₃/**L-PIPr₂** complex, affording the (*R*)-quinol **3a** in 49% yield and 95.5 : 4.5 er with the recovered oxaziridine (*S*)-**2a** in 45% yield.^{12d} Using *ent*-**L-PIPr₂** as the ligand, (*S*)-quinol **3a** was obtained in 46% yield and 90 : 10 er with the recovered oxaziridine (*S*)-**2a** in 52% yield. This indicates that the chiral matched and mis-matched effect between chiral ligand and chiral oxaziridine was not obvious in this case compared to previous reports,¹² and there might be negligible interaction between the chiral catalyst and oxaziridine.

To probe into the interaction between the catalysts and 2-naphthol, ¹H NMR analysis of the mixture of components was carried out (see ESI† for details). The chemical shift of 1-methyl 2-naphthol **1a** remained nearly unchanged after Sc(NTf₂)₃ was added. There was an obvious high-field shift for most signals of **1a** after mixing with the Sc(NTf₂)₃/**L-PIPr₂** catalyst. This indicates that the chiral catalyst makes the 2-naphthol reactive for hydroxylative reactions. Based on these results and our previous study on the chiral *N,N'*-dioxide–metal complex catalysts,^{8,13} we suggested an enantioselective catalytic model as shown in Fig. 2. The ligand **L-PIPr₂** binds to the scandium(III) center *via* four oxygens to form a polycyclic octahedral metal complex catalyst. The 2-naphthol coordinates to the metal center at one of the vacant sites, with its *Re*-face shielded by one amide unit of the ligand. Therefore, **2a** preferably attacked the α -position of 2-naphthol from the *Si*-face to generate the corresponding *R*-configured product **3ae** and imine byproduct. If a substituent was introduced into the C3 or C4 positions of 2-naphthol, the steric hindrance discrimination between the two sides of the



Scheme 2 Concise synthesis of chiral lacinilene C methyl ether, (–)-lacinilene D and (+)-lacinilene D.



Scheme 3 Control experiments.



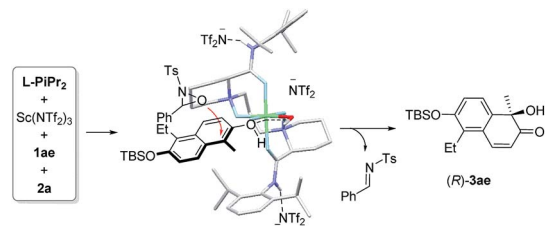


Fig. 2 Proposed enantioselective catalytic model.

hydroxyl group decreases, thus it is difficult to control the face-selection. As a result, the enantioselectivity for the generation of product **3aa** and lacinilene C methyl ether is lower than that for the others.

Conclusions

In summary, we have described a highly chemo- and enantioselective hydroxylative dearomatization of 2-naphthol derivatives with oxaziridine catalyzed by a chiral *N,N'*-dioxide-Sc(NTf₂)₃ complex catalyst. The desired substituted *ortho*-quinols with one quaternary carbon stereogenic center were afforded with high enantioselectivities and reactivity (up to 99% yield and 95 : 5 er). The α -ketol rearrangement byproducts were efficiently suppressed. This new procedure has been successfully applied to the catalytic asymmetric synthesis of the phytoalexins lacinilenes. The application of the *N,N'*-dioxide/metal catalyst system in the synthesis of other bioactive molecules will be explored.

Acknowledgements

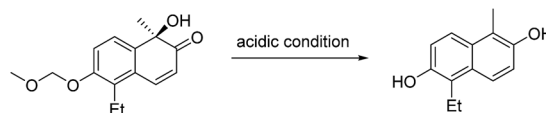
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see ref. 2d for details.

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