Correction: Preparation of Pt–Tl clusters showing new geometries. X-ray, NMR and luminescence studies

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Fig. 5 $^{195}\text{Pt}$ NMR spectra of compounds: 3 + TlPF$_6$ exc. (a) and 1 (b) in CD$_2$Cl$_2$ at variable temperature. Reprinted with permission from Belío et al., Inorg. Chem., 2013, 52, 5627–5629. Copyright 2013 American Chemical Society.

Fig. 6 Comparison between the solid state (above) and solution (below) $^{195}\text{Pt}$ NMR spectra of compounds 1 (blue) and 3 (red). Solid state $^{195}\text{Pt}$ NMR spectra are registered at spinning speeds of 12 kHz. Reprinted with permission from Belío et al., Inorg. Chem., 2013, 52, 5627–5629. Copyright 2013 American Chemical Society.

Fig. S2 Solid state $^{195}\text{Pt}$ NMR spectra of compound 1 registered at spinning speeds of 8 and 12 kHz. Reprinted with permission from Belío et al., Inorg. Chem., 2013, 52, 5627–5629. Copyright 2013 American Chemical Society.

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Table 1 Selected bond lengths (Å) and angles (°) for [Pt(CNC)(tht)]·0.5Me$_2$CO (1·0.5Me$_2$CO). Reprinted with permission from Belío et al., Inorg. Chem., 2013, 52, 5627–5629. Copyright 2013 American Chemical Society.

Table 2 Selected bond lengths (Å) and angles (°) for [{Pt(CNC)(tht)}$_3$Tl]-(PF$_6$)$_3$ (3). Reprinted with permission from Belío et al., Inorg. Chem., 2013, 52, 5627–5629. Copyright 2013 American Chemical Society.

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