Chemical Science



EDGE ARTICLE

View Article Online



Cite this: Chem. Sci., 2017, 8, 8265

Copper-catalyzed aminoalkynylation of alkenes with hypervalent iodine reagents†

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A copper-catalyzed aminoalkynylation of alkenes is achieved with ethynylbenziodoxolone (EBX) reagents under mild conditions with only 1 mol% copper catalyst. This transformation allows for rapid construction of diverse important azahetereocycles and installation of valuable alkyne groups in one step. The developed method features remarkable substrate scope for both terminal and internal alkenes as well as different alkynyl groups, presenting great potential for broad applications in synthesis, bioconjugation, and molecular imaging.

Received 5th August 2017 Accepted 15th October 2017

DOI: 10.1039/c7sc03420b

rsc.li/chemical-science

Introduction

The carbon-carbon triple bond is one of the most valuable functional groups in organic chemistry. It features versatile reactivity as a synthetic intermediate in organic synthesis and has broad applications as a functional tag in biochemistry and materials sciences (Fig. 1).1 For example, alkynes have been widely used in "alkyne-azide click chemistry" for bioconjugation.2 Recently, alkynes have also been demonstrated as a Raman imaging tag because they show distinct, strong Raman scattering at \sim 2150 cm⁻¹, in a cellular silent region (1800-2800 cm⁻¹) where most endogenous molecules show no Raman scattering.3 Therefore, developing efficient methods for installation of an alkyne group onto organic molecules is important and has attracted intense interests.

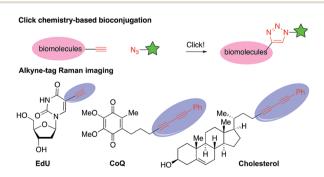
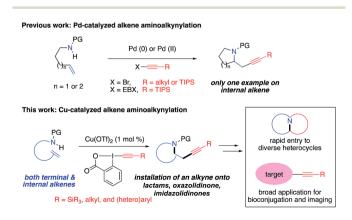


Fig. 1 Representative applications of alkynes in bioconjugation and molecular imaging.

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† Electronic supplementary information (ESI) available: Characterization data and experimental procedures. CCDC 1567005-1567007. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c7sc03420b

Alkene alkynylation represents an attractive and rapid approach to install alkyne groups onto target molecules from readily available alkenes. With azaheterocycles as the most valuable skeletons frequently found in biologically active natural products and pharmaceuticals,4 alkene aminoalkynylation is particularly valued for construction of azaheterocycles and installation of an alkyne group in a single step. Significant advances in this area have been reported by the Waser group (Scheme 1), including Pd-catalyzed alkene aminoalkynylation, as well as oxy- and carbo-alkynylation transformations using ethynylbenziodoxolones (EBX)5 or aliphatic bromoacetylenes.^{6,7} Despite these progress, only one example have been demonstrated on internal alkenes so far, impeding its utility to construct more complex, diverse azaheterocycles. Furthermore, reaction conditions need to be tailored for different alkyne precursors in previous methods, limiting the scope of alkyne groups for broad application in organic synthesis and chemical biology. Motivated by our interest in developing new, efficient methods to access important azaheterocycles8 and inspired by recent success in alkene



Scheme 1 Alkene difunctionalization for installation of an alkyne group onto important azaheterocycles.

functionalization with hypervalent iodine reagents,⁹ we here report our development of a copper-catalyzed selective amino-alkynylation reaction of both terminal and internal alkenes (Scheme 1). This transformation readily occurs with 1 mol% of copper catalyst under mild conditions and enables rapid access to a wide range of alkyne-labelled fused, and bridged azaheterocyles, which are commonly found in pharmaceutically relevant architectures.⁴ The aminoalkynylation method will enable broad applications of using alkynes as a handle for rapid entry to diverse azaheterocycles and as a functional tag in bioconjugation and Raman imaging.

Results and discussion

Reaction optimization

Our investigation of this alkene aminoalkynylation reaction began with unsaturated amide **1a** (Table 1), a model substrate that was demonstrated to undergo effectively copper-catalyzed aminocyclization in our previous studies. Sc, 9h, 10 We chose ethynylbenziodoxolones (EBX) as the alkynyl precursors, as they have been successfully used in different alkynylation reactions with nucleophiles, 11 C-H bonds, 12 carbon radicals, 13 and olefins. 5 Encouragingly, aminoalkynylation product **3aa** was formed in 69% yield in the presence of Cu(OTf)₂ in CH₃CN at 80 °C (entry 1). Among the various copper salts examined, cationic copper species generally gave higher yields than neutral copper species, with Cu(OTf)₂ most effective (entries 1–

Table 1 Reaction condition optimizations^a

Me NOMe reagent
$$2a-2a''$$
 copper catalyst CH_3CN , temp, 12 h Me NOMe N -OMe N -OM

Entry	Catalyst (mol%)	2	<i>T</i> (°C)	Yield ^b
1	Cu(OTf) ₂ (10)	2a	80	69%
2	$Cu(OAc)_2$ (10)	2a	80	33%
3	Cu(acac) ₂ (10)	2a	80	34%
4	CuCl ₂ (10)	2a	80	26%
5	CuOTf 1/2PhCH ₃ (10)	2a	80	63%
6	$Cu(CH_3CN)_4PF_6$ (10)	2a	80	69%
7	$Cu(CH_3CN)_4BF_4$ (10)	2a	80	67%
8	_	2a	80	15%
9	$Cu(OTf)_2$ (5)	2a	80	73%
10	$Cu(OTf)_{2}(1)$	2a	80	78%
11	$Cu(OTf)_2$ (1)	2a	100	84%
12	$Cu(OTf)_2$ (1)	2a	60	37%
13	Cu(OTf) ₂ (1)	2a'	100	57%
14	$Cu(OTf)_2$ (1)	2a"	100	<5%

 $[^]a$ Reactions conditions: 1 (0.2 mmol, 1 equiv.), 2 (0.24 mmol, 1.2 equiv.), copper catalyst, CH₃CN (1 mL) unless otherwise noted. b Yields were determined by 1 H NMR with CH₂Br₂ as an internal standard.

7). Without a copper catalyst, only trace amounts of **3aa** was observed, indicating the important role of a copper catalyst in this reaction (entry 8). Decreasing the catalyst loading was beneficial for the formation of **3aa** (entries 9 and 10) and elevating the temperature improved the efficiency of reactions (entries 10–12). In comparison to **2a**, other types of alkynylbased hypervalent iodine reagents such as **2a'** and **2a''** resulted in much lower yields (entries 13 and 14). Thus entry 11 was chosen as standard conditions for subsequent studies of this aminoalkynylation reaction.¹⁴

Reaction scope

We examined the generality and efficiency of the aminoalkynylation reaction on different unsaturated amide derivatives (Table 2). Monosubstituted alkenes 1a-f bearing various substitutes on the alkenyl chain all underwent smooth 5-exo cyclization and gave γ -lactam products 3aa-fa. The reactions of substrates bearing substitutes on the backbone were more efficient, likely due to the favorable Thorpe-Ingold effect in the cyclization step. Unsaturated ureas 1g-h and carbamates 1i-j were effective in the aminoalkynylation reactions and provided desired imidazolidinones 3ga-ha and oxazolidinones 3ia-ja. 1,1-Disubstituted alkene 1k led to γ-lactam 3ka bearing a quaternary carbon center. δ-Lactam products 3ma-na were also readily formed, with heteroarenes such as indoles and pyridines well tolerated in the reactions. Note that the diastereoselectivity of this reaction may be influenced by the substitution on the backbone, as observed in the formation of 3ea, 3fa and 3ha. Remarkably, this aminoalkynylation reaction was applicable to internal alkenes, which are known to be challenging in metal-catalyzed alkene difunctionalization due to competing β-H elimination. The reaction of tri-substituted alkene 10 gave aminoalkynylation product 30a bearing a quaternary carbon in 82% yield. The reactions of (E)- and (Z)-1p both led to the formation of product 3pa in the same ratio of 2:1 diasteroselectivity, suggesting that (E)- and (Z)-1p may share the same intermediate leading to the product 3pa. The reactions of cyclic internal alkenes well delivered fused-ring products 3qa and 3ra bridged-ring products 3sa-ua. Overall, this aminoalkynylation reaction proved effective on a broad scope of olefin substrates that encompass diverse substitutions on both alkenes and backbones.

Next, the scope of alkynylation reagents 2 was examined in the reactions with alkene 1l (Table 2). As seen in the formation of 3lb-lf, both electron-donating and electron-withdrawing groups on the phenyl group were well tolerated. Neither the electronic nor steric properties of the substitutes on the phenyl group significantly influence the reaction efficiency. The thiophene moiety, an electron-rich heterocycles known for direct alkynylation, 12a is compatible in this reaction, affording product 3lg in good yield. Alkynylation reagents bearing alkyl substitutes were also applicable (3lh and 3li), albeit in much lower yields compared to aromatic substitutes. TBS-EBX and TIPS-EBX reagents were also feasible in this transformation (3lj and 3lk). Finally, conjugated dialkyne was readily incorporated into the aminoalkynylation product 3ll. Note that conjugated dialkyne

Table 2 Scope of alkenes and alkynyl groups for the aminoalkynylation reactions a,b

OMe 2a-2l Cu(OTf)₂ (1 mol %) CH₃CN (0.2 M), 100 °C, 12 h 1a-11 Scope of alkenes 3aa 76% 3ca, 76% 3ba. 71% 3da. 35% 3ea, 54%, dr = 2.5:1 3fa. 51%, dr = 1:1 3qa. 75% 3ha, 78%, dr > 20:1 3ia. 42% 3ja, 69% 3ka. 69% 3ma, 46% 3na. 40% -OMe Me 3pa. 71%, dr = 2:19 30a. 82% 3ga. 60%, dr > 20:1 3sa, 65%, dr > 20:1 3ta. 48%, dr = 5:1 3ua. 76%, dr = 2:1 3ra. 92%, dr > 20:1 3pa, 70%, dr = 2:1d

^a Reactions performed with 1 (0.3 mmol, 1 equiv.), 2 (1.2 equiv.), Cu(OTf)₂ (1 mol%) in CH₃CN at 100 °C, unless otherwise noted. ^b Isolation yield. ^c From *trans* alkene. ^d From *cis* alkene. ^e Cu(CH₃CN)₄BF₄ (1 mol%) at 60 °C.

3lg, 73%

3lh, R = t Bu, 48% **3li**, R = n Bu, 37%

groups are known to offer stronger Raman signal compared to other types of alkynes as potential Raman imagine tags. ^{3b,e}

3lb, R = 2-CN, 71% 3lc, R = 3-OMe, 75% 3ld, R = 4-Cl, 73% 3le, R = 4-F, 68% 3lf, R = 4-CF₃, 76%

Mechanism study

Scope of alkynyl groups

A series of control experiments were performed to obtain mechanistic insights on this aminoalkynylation reaction (Scheme 2). In the presence of TEMPO as a radical scavenger, the reaction of model substrates 1a and 2a provided aminoxygenation product 4 in 99% yield (Scheme 2a). The reaction of the trans-D-substituted alkenyl amide D-1a gave a 1:1 mixture of D-substituted diastereomers 5 in 72% yield (Scheme 2b). Both results indicate a radical nature of the intermediate after the intramolecular amino-cupration step, which is consistent with the loss of stereochemistry observed in the formation of 3pa from (E)- and (Z)-**1p** in Table 2. To confirm this speculation, we examined the aminoalkynylation reaction using 1v, an unsaturated amide containing a standard radical clock cyclopropane moiety at the vinyl position (Scheme 2c). The reaction provided ring-opened product 6 in 45% yield with no detection of 3va, confirming the presence of radical alkyl intermediates after the aminocyclization step. Finally, we used 13C labelled Ph-EBX ¹³C-2a to probe the alkynylation step, revealing the exclusive presence of ¹³C atom at its original position of 7 (Scheme 2d).

Based on these results, two possible reaction pathways are proposed for this copper-catalyzed aminoalkynylation reaction

3li Si = TBDMS 46%

311, 55%

Scheme 2 Mechanism investigations

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(Scheme 3).15 In pathway A, the reaction may be initiated by the coordination of copper catalyst with alkene 1a followed by the intramolecular aminocupration.10 The resulting alkyl-Cu intermediate II may undergo a reversible C-Cu(II) homolysis to form radical intermediate III, which would subsequently attack the α -position of the alkynyliodonium salt 2a. Finally, the β-elimination of intermediate IV would lead to product 3 and regeneration of the copper catalyst. In an alternative pathway (B), the reaction may be initiated by the copper-catalyzed activation of alkynyliodonium salt 2a. The aminocyclization would occur via copper intermediates I' and II' to form copper-alkyl complex III', which would undergo similar homolysis to form a radical intermediate IV' and could lead to the formation of product 3. Note that the exclusive presence of ¹³C atom at its original position in the reaction of EBX reagent ¹³C-2a (Scheme 2d) indicated the involvement of α -addition in the alkynylation step in either pathway (A) or (B), rather than the βaddition of the alkyne followed by the 1,2-shift of the Ph group. 11i,l,16

Synthetic applications

We next explored the synthetic utility of alkynyl group for the construction of various bicyclic heterocycles (Scheme 4).¹⁷ First, using 3la, we demonstrate the advantage of the methoxy protecting group of the amide to achieve selective gold-catalyzed oxy- or amino-cyclization onto the alkyne group. For example, upon the removal of methyl group, the resulting hydroxylamide 8 underwent a Au-catalyzed *endo* oxycyclization to form oxazinane 9. On the other hand, the demethoxylation of 3la provided amide 10, which was transformed into dihydropyrole 11 by an

Scheme 3 Proposed reaction pathways.

Scheme 4 Selective construction of diverse fused heterocycles.

Scheme 5 Synthesis of alkyne-labelled derivatives of mesembrane. a **2a** (1.2 equiv.), Cu(OTf)₂ (1 mol%), 100 °C. b **2l** (1.2 equiv.), Cu(CH₃-CN)₄BF₄ (1 mol%), 100 °C. c **2j** (1.2 equiv.), Cu(CH₃CN)₄BF₄ (1 mol%), 60 °C; then TBAF (1.33 equiv.), rt. d (i) SmI₂, THF, rt; (ii) NaH, MeI, THF, rt; (iii) Rh(acac)(cod), PhSiH₃, THF, 50 °C.

analogous Au-catalyzed aminocyclization. Furthermore, we employed the aminoalkynylation product **3aa** to complete the synthesis of a series of important pyrrolizidine derivatives **13–15**. Finally, this aminoalkynylation method was successfully applied for the preparation of alkyne-labelled derivatives of alkaloid mesembrane (Scheme 5). These readily installed alkynyl groups are expected to serve as valuable labelling tools to facilitate future studies regarding understanding its antidepressant potential and mode of action. ¹⁹

Conclusions

In summary, we have developed a new copper-catalyzed alkene aminoalkynylation reaction that is effective on an extensive scope of alkene substrates and EBX reagents. This method enables simultaneous construction of valuable azahetereocyclic skeletons and the installation an alkynyl group, which will greatly advance broad applications of using alkynes as a handle for rapid entry to complex, diverse heterocycles and as a functional tag in bioconjugation and Raman imaging.

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Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We acknowledge financial support provided by Duke University and the National Institute of General Medical Sciences of the NIH (GM118786). Q. W. is a fellow of the Alfred P. Sloan Foundation and a Camille Dreyfus Teacher-Scholar. We thank Dr George Dubay (Duke University) for high-resolution mass spectrometry and thank Dr Roger Sommer (NCSU) for X-ray structural analysis.

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