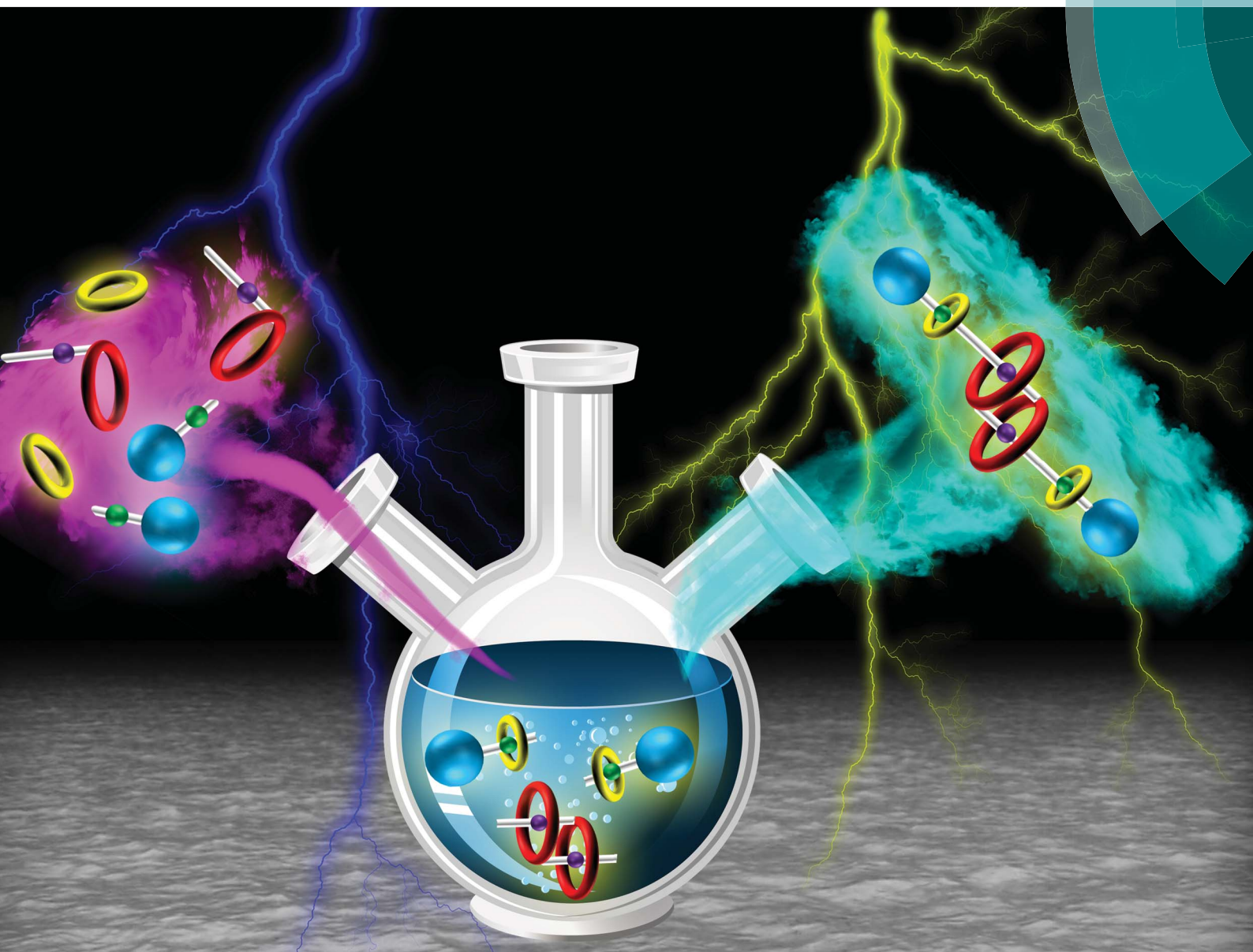


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**EDGE ARTICLE**

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One-pot synthesis of a [c2]daisy-chain-containing hetero[4]rotaxane via a self-sorting strategy

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# One-pot synthesis of a [c2]daisy-chain-containing hetero[4]rotaxane *via* a self-sorting strategy†

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The construction and efficient synthesis of hetero[*n*]rotaxanes with high structural complexity are always attractive challenges. Herein, we demonstrate a facile one-pot preparation of a hetero[4]rotaxane, by employing a self-sorting strategy, which contains an interpenetrated dibenzo-24-crown-8 (DB24C8) based [c2]daisy chain structure and is ended with a benzo-21-crown-7 (B21C7) based rotaxane at each side. The key to the design involved encoding the selective threading using a steric hindrance-related "language", where highly selective self-assemblies occurred in a three-component self-sorting process, which included the threading of a benzylalkylammonium into a B21C7 and interpenetrated dimerized formation of a DB24C8 based [c2]daisy chain simultaneously; the precise pre-assembled system resulted in the efficient synthesis of hetero[4]rotaxane with a high-level of structural complexity under the "CuAAC" reaction.

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## Introduction

In the last twenty years, a variety of mechanically interlocked molecules (MIMs), especially rotaxanes and catenanes, have been delicately designed and constructed.<sup>1</sup> The unique structural features of MIMs are utilized by scientists to birth elegant functional molecules,<sup>2</sup> indicating the great potential of MIMs in functional materials. As a modification platform for functional groups, the structural complexity of MIMs significantly provides infinite possibilities in designing novel functional molecular machines.<sup>3</sup> Hence, the construction and efficient synthesis of MIMs with high structural complexity have been greatly attractive but challenging.<sup>4</sup> Recently, much attention has been focused on the topological,<sup>3a</sup> interpenetrating,<sup>5</sup> knotted<sup>6</sup> structures of MIMs with high structural complexity. In this intriguing family, [c2] daisy chain molecules,<sup>5</sup> a class of rotaxanes constructed from the dimerization of AB-type linear monomers with two self-complementary units, A (host) and B (guest), have shown their talent in mimicking the contraction and extension movements of natural biological machines.<sup>5f</sup> Various elegant [c2] daisy chain molecules have been reported based on different host–guest systems including versatile macrocycles, such as crown ethers,<sup>5b,e,h</sup> cucurbit[*n*]uril,<sup>5i</sup> cyclodextrins,<sup>5a,c,d</sup> pillarenes,<sup>5g,m,p</sup> and cyclophanes.<sup>5o</sup>

A variety of hetero[*n*]rotaxanes, especially those comprising two or more different types of ring moiety, have been designed and constructed,<sup>7</sup> some even in a one-step strategy.<sup>7b,d</sup> This aroused our interest to integrate muscle-like daisy chain structures with hetero[*n*]rotaxanes to breed a hetero[*n*]rotaxane with a novel structural topology. However, it is difficult to synthesize this novel hetero[*n*]rotaxane based on traditional strategies for rotaxane synthesis,<sup>8</sup> which are efficient in the preparation of typical [2]rotaxane but complicated when there is more than one type of ring moiety, due to them being "non-selective". The key challenge is the dramatically increased complexity of the interpenetration process in the formation of a [c2]daisy chain when additional recognition sites and rings are incorporated into the multi-component self-assembling system. Hence, a programming language for highly selective self-assembly is desperately needed to integrate hetero[*n*]rotaxane with the daisy chain structure.

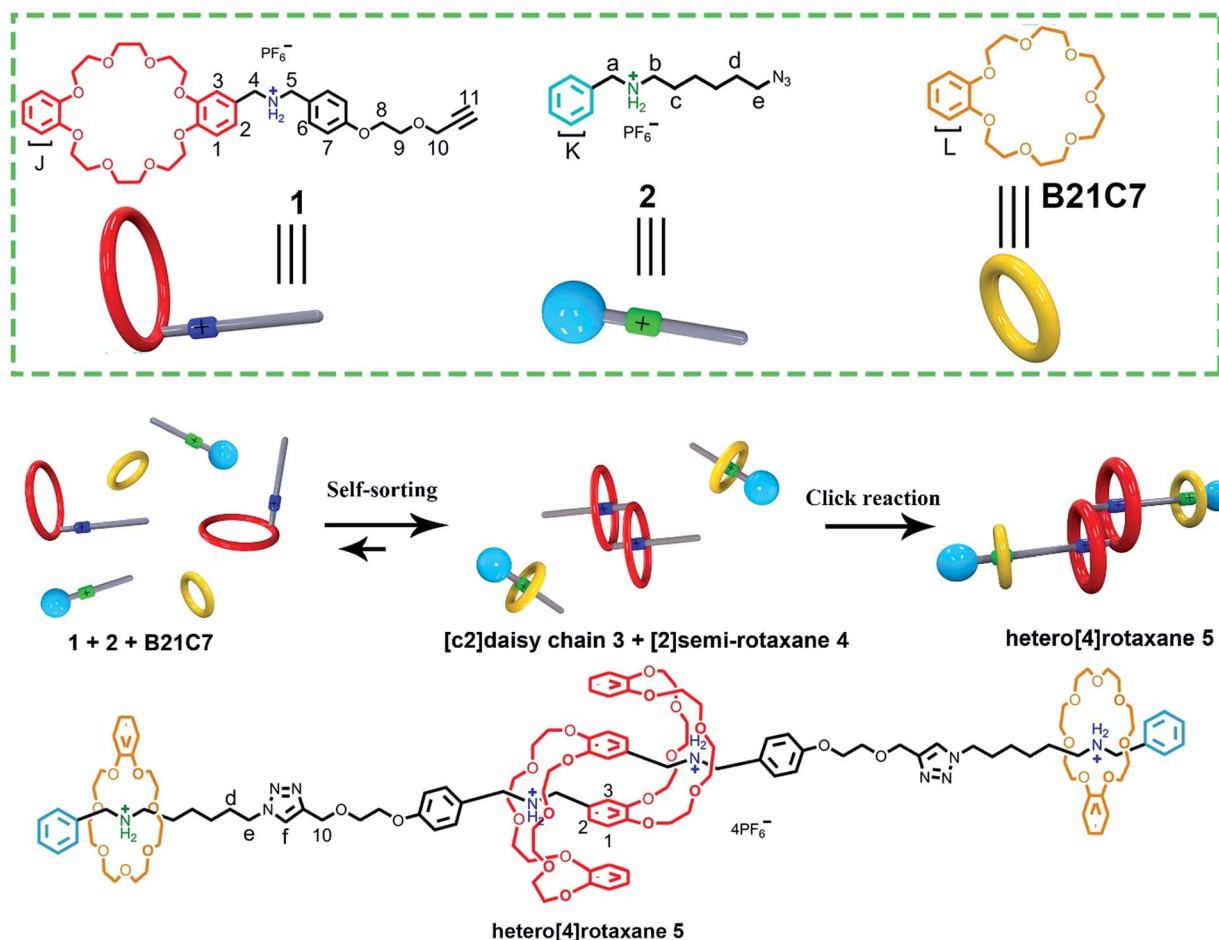
Self-sorting strategies have the unique capability of selective self-assembly in complex supramolecular systems.<sup>9</sup> Schalley and co-workers<sup>10</sup> have developed an efficient integrative self-sorting strategy for selective self-assembly to simultaneously incorporate two kinds of polyether macrocycles into a single axle molecule comprising two kinds of secondary ammoniums which, as a result, form a cascade-stoppered hetero[3]rotaxane. This self-sorting strategy has been proved to be very efficient in constructing many hetero[*n*]rotaxanes with increased structural complexity.<sup>11</sup> However, to the best of our knowledge, this self-sorting strategy has not been proved to be appropriate for a system simultaneously involving both threading and interpenetration, which are known for the formation of pseudorotaxanes and daisy chain structures, respectively.

Herein, a facile one-pot synthesis of hetero[4]rotaxane **5** is successfully realized *via* a three-component (**1**, **2** and B21C7 in

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**Scheme 1** Schematic representation of the preparation of hetero[4]rotaxane 5 using a three-component modularized self-sorting system of 1, 2 and B21C7. Only one main stereoisomer of hetero[4]rotaxane 5, which is derived from the three possible [c2]daisy chain stereoisomers,<sup>5e</sup> is shown here.

Scheme 1) self-sorting strategy. In principle, these initial components (1, 2 and B21C7 in Scheme 1) could assemble into several self-assembly species, while only two kinds of precursors, [c2]daisy chain 3 and [2]semi-rotaxane 4, were successfully formed through the highly selective self-assembly process as determined by <sup>1</sup>H NMR studies. In this pre-assembly process, both the self-interpenetration of compound 1 to form [c2]daisy chain 3 and the threading of 2 into the cavity of B21C7 to form [2]semi-rotaxane 4 occurred simultaneously. Then the highly symmetrical hetero[4]rotaxane 5 containing [c2]daisy chain could be obtained by the following facile one-pot CuAAC click reaction.<sup>8d</sup> We believe that the facile one-pot synthesis of hetero[4]rotaxane 5 would inspire the successful construction of MIMs with fascinating structures and potential functions.

## Results and discussion

The rational design of block building is a prerequisite to realizing the precise construction of a multi-component system. As shown in Scheme 1, initial substrates include compound 1 (the synthesis information from compounds 7 and 8 can be found in the ESI†), which contains a DB24C8 macrocycle modified with

a secondary dibenzylammonium branched chain; a secondary benzylalkylammonium 2; and B21C7. The key part of the design involves the dibenzylammonium recognition site in compound 1, where the phenyl group encodes the secondary ammonium as a selective site that only can be included by the larger macrocyclic DB24C8, but not by B21C7.<sup>7b,10,12</sup> Therefore, even in one system, B21C7 would not affect the formation of the [c2]daisy chain 3. Meanwhile, this phenyl group also makes the secondary ammonium in compound 1 a preferred recognition site for DB24C8, over that in compound 2.<sup>10</sup> Considering the precise 1 : 1 proportion of DB24C8 and the secondary ammonium in compound 1, compound 2 would not affect the formation of [c2]daisy chain 3 from monomer molecule 1, which has been proved by the <sup>1</sup>H NMR measurements (Fig. S1†). As a result, the self-assembly of the initial molecules was precisely encoded, where the self-sorting process included simultaneous threading and interpenetration, and then [2] semi-rotaxane 4 and [c2]daisy chain 3 were generated, respectively. Finally, in the presence of Cu(MeCN)<sub>4</sub>PF<sub>6</sub>, the subsequent CuAAC click reaction between [c2]daisy chain 3 and [2]semi-rotaxane 4 afforded the formation of hetero[4]rotaxane 5 in a one-pot strategy. In the obtained hetero[4]rotaxane 5, two





**B21C7** rings can be stopped by the outer phenyl groups, while the central [c2]daisy chain structures are cascade-stopped by the **B21C7** rings.

We performed  $^1\text{H}$  NMR experiments to confirm the self-sorting behaviours in this three-component system, as shown in Fig. 1. The  $^1\text{H}$  NMR spectra revealed that **1** existed as a monomer in polar solvents, such as  $[\text{D}_6]\text{DMSO}$  (Fig. 1a), with normal and simple  $^1\text{H}$  NMR signals. While in less polar solvents, such as  $[\text{D}_3]\text{acetonitrile}$  (Fig. 1b), [c2]daisy chain **3** became the predominant species. The  $^1\text{H}$  NMR signals of the  $\text{NH}_2^+$  protons and crown ether moiety shifted and split, becoming much broader and more complicated. As shown in Fig. 1b, two sets of signals for the  $\text{NH}_2^+$  protons were observed at  $\delta = 7.0$ – $7.4$  ppm, attributed to each ammonium facing the two non-symmetrical ends of the axle. According to the previous studies about the analogous [c2]daisy chains,<sup>13,14</sup> there are three dimeric interlocked stereoisomers arising from the unsymmetrical substitution of the **DB24C8** ring, which were also observed in the  $^1\text{H}$  NMR spectrum of [c2]daisy chain **3** (Fig. 1b). As shown in Fig. 1, the aromatic protons of **DB24C8** also become split, especially protons  $\text{H}_1$  and  $\text{H}_3$ .  $\text{H}_1$  splits into two sets of double peaks ( $\text{H}_{1a}$ , 6.4 ppm and  $\text{H}_{1b}$ , 6.9 ppm), and  $\text{H}_3$  splits into two sets of single peaks ( $\text{H}_{3a}$ , 6.8 ppm and  $\text{H}_{3b}$ , 6.2 ppm) with the ratio of 5 : 1, indicating the major supramolecular stereoisomer of the “meso” type, shown in Scheme 1, which is consistent with the previous report.<sup>5e</sup> Meanwhile, the formation of [2]semi-rotaxane **4** from compound **2** and **B21C7** has been detected using the  $^1\text{H}$  NMR spectra shown in Fig. 2. After mixing compound **2** and **B21C7** in a 1 : 1 molar ratio, the signals of the  $\text{NH}_2^+$  protons shifted upfield with a  $\Delta\delta$  of  $-0.97$  ppm, and the signals of  $\text{H}_a$ ,  $\text{H}_b$  and  $\text{H}_c$  shifted downfield with a  $\Delta\delta$  of 0.26, 0.24 and 0.07 ppm, respectively, suggesting the formation of [2]semi-rotaxane **4**.<sup>12</sup> Then, the three compounds **1**, **2** and **B21C7** were mixed in  $[\text{D}_3]\text{acetonitrile}$  in a molar ratio of 1 : 1 : 1.  $^1\text{H}$  NMR spectra (Fig. 3) showed that the signals of  $\text{H}_{1a}$ ,  $\text{H}_{3b}$ ,  $\text{H}_4$  and  $\text{H}_5$  in [c2]daisy chain **3**, as well as those of  $\text{H}_a$ ,  $\text{H}_b$  and  $\text{H}_c$  in [2]semi-rotaxane **4**, remain constant. These observations indicate that the two pseudorotaxanes still exist as the predominant species in this three-component self-sorting system.

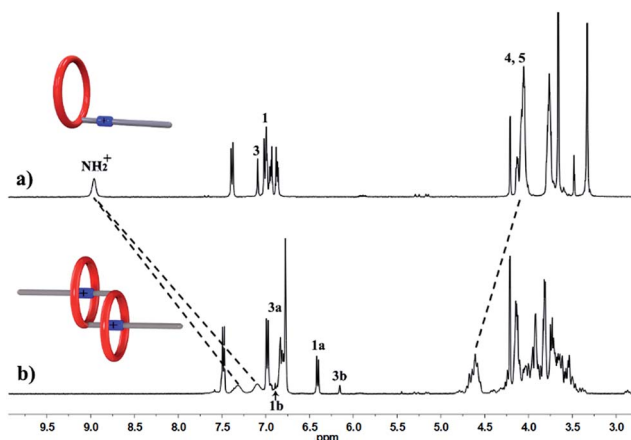


Fig. 1 Partial  $^1\text{H}$  NMR spectra (400 MHz, 298 K) of compound **1** in (a)  $[\text{D}_6]\text{DMSO}$ , and (b)  $[\text{D}_3]\text{acetonitrile}$ .

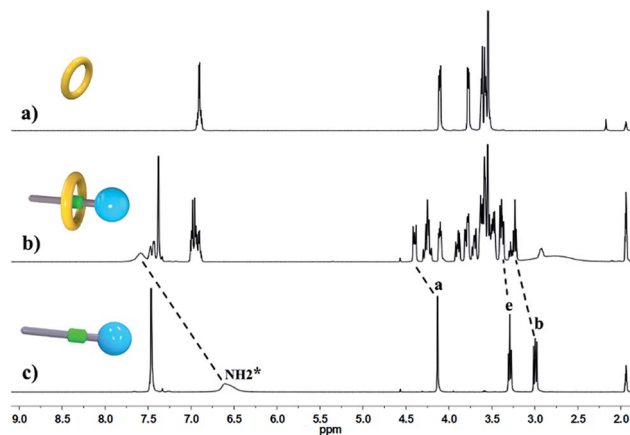


Fig. 2 Partial  $^1\text{H}$  NMR spectra (400 MHz, 298 K,  $[\text{D}_3]\text{acetonitrile}$ ) of (a) **B21C7**, (b) a 1 : 1 mixture of **2** and **B21C7**, and (c) compound **2**.

The synthetic route for hetero[4]rotaxane **5** is outlined in Scheme 1. At room temperature, the CuAAC reaction between [c2]daisy chain **3** and [2]semi-rotaxane **4** successfully afforded the formation of a white solid, hetero[4]rotaxane **5**, with a yield exceeding 50% in the presence of  $\text{Cu}(\text{MeCN})_4\text{PF}_6$ . As shown in Fig. 4a and b, the terminal alkynyl proton  $\text{H}_{11}$  of **3** disappeared after the CuAAC click reaction, indicating the successful formation of the triazole. The resonance for the triazole proton  $\text{H}_f$  appeared at  $\delta = 7.72$  ppm and the adjacent protons  $\text{H}_{10}$ ,  $\text{H}_b$ ,  $\text{H}_d$  and  $\text{H}_e$  shifted downfield with a  $\Delta\delta$  of 0.38, 0.22, 0.12 and 0.89 ppm, respectively. All of these changes demonstrated the importance of the CuAAC click reaction for the formation of hetero[4]rotaxane **5**, as illustrated in Scheme 1.<sup>15</sup> The high resolution mass spectrum (HRMS) of hetero[4]rotaxane **5** showed major signals at  $m/z$  628.0955, 885.7819 and 1401.1543, which correspond to the product after the loss of 4, 3 and  $2\text{PF}_6^-$  ions, *i.e.*  $[\text{M} - 4\text{PF}_6]^{4+}$ ,  $[\text{M} - 3\text{PF}_6]^{3+}$  and  $[\text{M} - 2\text{PF}_6]^{2+}$ , respectively. The molecular ion peaks of  $[\text{M} - 4\text{PF}_6]^{4+}$  and  $[\text{M} - 3\text{PF}_6]^{3+}$ , as well as the results of  $^1\text{H}$  NMR, give enough evidence to show the formation of hetero[4]rotaxane **5**.<sup>15</sup>

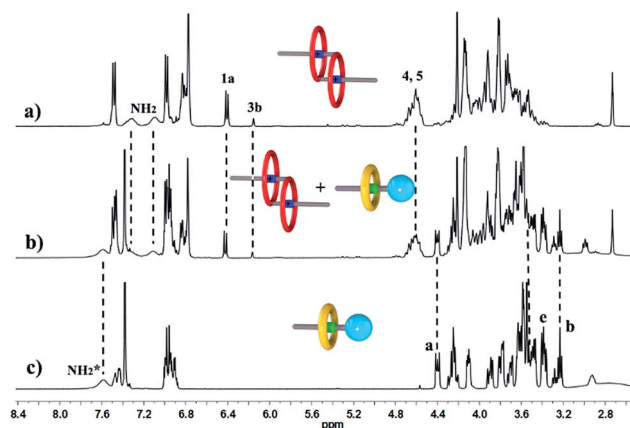


Fig. 3 Partial  $^1\text{H}$  NMR spectra (400 MHz, 298 K,  $[\text{D}_3]\text{acetonitrile}$ ) of (a) compound **1**, (b) a 1 : 1 : 1 mixture of **1**, **2** and **B21C7**, and (c) a 1 : 1 mixture of **2** and **B21C7**.

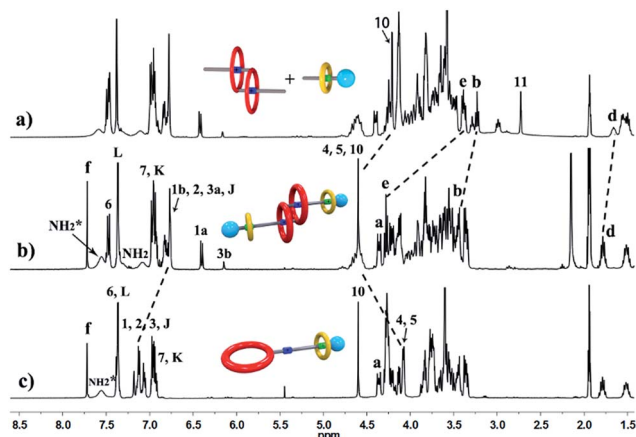
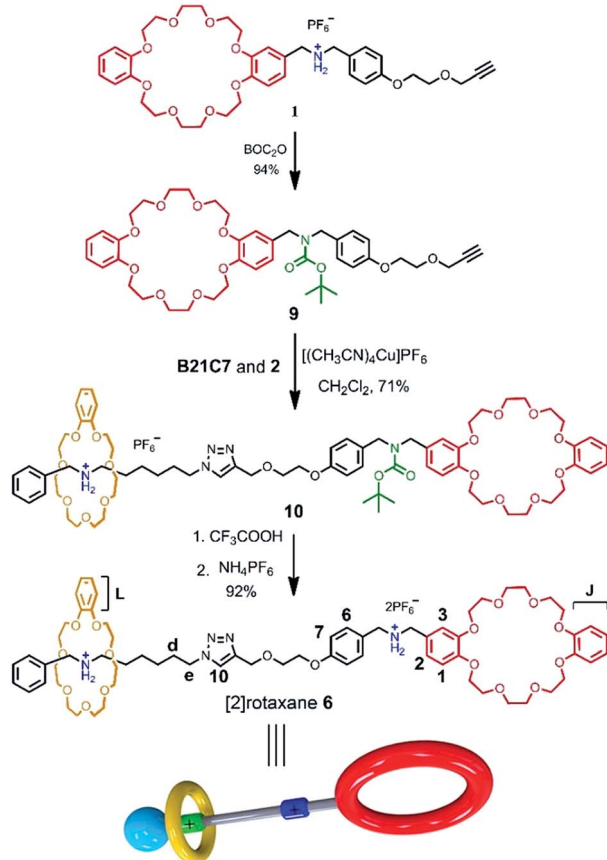


Fig. 4 Partial  $^1\text{H}$  NMR spectra (400 MHz, 298 K,  $[\text{D}_3]\text{acetonitrile}$ ) of (a) a 1 : 1 : 1 mixture of **1**, **2** and **B21C7**, (b) hetero[4]rotaxane **5**, and (c) [2]rotaxane **6**.

For further understanding of the self-sorting process and confirming the chemical structure of hetero[4]rotaxane **5**, the dibenzylammonium site in compound **1** was protected with a BOC group to yield compound **9** (Scheme 2), leading to a recognition failure between macrocyclic **DB24C8** and the DBA site. As a result, there was only one type of secondary



Scheme 2 Synthetic route for [2]rotaxane **6**.

ammonium recognition site in the mixed systems of compound **9**, compound **2** and **B21C7**, meaning that a non-self-sorting process would occur among these building blocks. Then, a CuAAC click reaction of compound **9** and [2]semi-rotaxane **4** successfully afforded [2]rotaxane **10** in the presence of  $\text{Cu}(\text{MeCN})_4\text{PF}_6$  in a 1 : 1 volume ratio of dichloromethane/acetonitrile at room temperature. After being treated with trifluoroacetic acid, the reference compound [2]rotaxane **6** with an unoccupied **DB24C8** ring was obtained. The HRMS spectrum of [2]rotaxane **6** shows major signals at  $m/z$  1400.6526, corresponding to the product after a loss of the  $\text{PF}_6^-$  ion, confirming the interlocked structure. Furthermore, we compared all of the resonances between hetero[4]rotaxane **5** and [2]rotaxane **6** by analysing their  $^1\text{H}$  NMR spectra (Fig. 4b and c). As shown in the  $^1\text{H}$  NMR spectra of **5** and **6** in  $[\text{D}_3]\text{acetonitrile}$  (Fig. 4b and c), the protons  $\text{H}_1$ ,  $\text{H}_2$ ,  $\text{H}_3$  and  $\text{H}_j$  of [2]rotaxane **6** shifted downfield with a  $\Delta\delta$  of about 0.2 ppm, and the stereoisomer signals of  $\text{H}_{1a}$  and  $\text{H}_{3b}$  disappeared, which was attributed to the fact that the DBA site was not surrounded by the **DB24C8** macrocycle to give a simple chemical shift environment. It is notable that the DBA protons in [2]rotaxane **6** cannot be detected because there are no hydrogen bonding interactions between the crown ether and the DBA hydrogen atoms.<sup>16</sup> Compared with the  $^1\text{H}$  NMR spectrum of hetero[4]rotaxane **5**,  $\text{H}_4$  and  $\text{H}_5$  which are adjacent to the DBA site were observed to shift upfield with a  $\Delta\delta$  of  $-0.53$  ppm in the  $^1\text{H}$  NMR spectrum of [2]rotaxane **6**, which confirmed that the DBA sites were not bound by the host macrocycle **DB24C8**. Therefore, the phenyl-stopped **B21C7** played an important role in inhibiting the dethreading of the daisy chain structure in the middle, this is also called a cascade-stopped strategy.<sup>10,7b</sup> Importantly, this inhibiting behaviour also caused the stable structure of hetero[4]rotaxane **5** instead of a meta-stable pseudorotaxane.

Meanwhile, 2D NOESY spectra of hetero[4]rotaxane **5** and [2]rotaxane **6** validated their chemical structures. The NOESY spectrum of **6** in  $[\text{D}_3]\text{acetonitrile}$  (Fig. S1†) shows two cross-peaks (peak F and G) between the phenyl protons  $\text{H}_L$  and ethylene protons of **B21C7**, thus clearly indicating that **B21C7** is located on the outer ammonium site which is far away from the **DB24C8** ring. In the meantime, no cross-peaks between **DB24C8** and the inner DBA site were found. Nevertheless, in the NOESY spectrum of **5** (Fig. S2†), we found not only cross-peaks (peaks P, Q, R and S) between the phenyl protons  $\text{H}_L$  and ethylene protons of **B21C7**, but also cross-peaks (peak N and O) between the phenyl protons  $\text{H}_7$  and ethylene protons of **DB24C8**, and cross-peaks (peak T and U) between the phenyl protons  $\text{H}_6$  and ethylene protons of **DB24C8**, thus clearly indicating the existence of a [c2]daisy chain structure in **5**.

## Conclusion

In conclusion, a novel hetero[4]rotaxane containing a [c2]daisy chain can be facilely synthesized in one pot by employing a multi-component self-sorting strategy followed by the well-known CuAAC stopping reaction. The highly selective self-assembling process that occurred among the three kinds of building blocks provides the possibility for the formation of



only two kinds of rotaxane precursors, by employing a steric hindrance-related “language”. This work serves as one of few successful examples of the construction of hetero[n]rotaxane with a [c2]daisy chain cored structure directed by a self-sorting strategy. Significantly, the integration of hetero[n]rotaxane and a [c2]daisy chain structure enlarges the family of mechanically interlocked molecules and this synthetic methodology can be utilized in the preparation of mechanically interlocked compounds with increasingly complicated structures and functions.

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