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Enantioselective synthesis of *iboga* alkaloids and vinblastine *via* rearrangements of quaternary ammoniums[†]

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An efficient and novel strategy for the enantioselective syntheses of various *iboga* alkaloids has been developed. The salient features include a gold-catalyzed oxidation of a terminal alkyne followed by cyclization, a Stevens rearrangement and a tandem sequence that combines the gold-catalyzed oxidation, cyclization and [1,2]-shift. The catharanthine analogs provided by our approach were further converted to the *vinca* alkaloid vinblastine and its analogs, which confirmed the remarkable sensitivity of the cytotoxicity to the C20' substituent of vinblastine.

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Introduction

The total synthesis of complex natural product small molecules invites the examination of various methodologies in a complicated system, which at times reveals current limitations and inspires new advances.¹ Importantly, total synthesis could also provide valuable analogs to explore the structure–activity relationships of targeted chemotypes.² We have been interested in using rearrangement reactions that lead to dramatic changes in molecular skeletons to develop novel and efficient synthetic routes towards various biologically active natural products.³ Herein, we describe a concise and collective synthesis⁴ of *iboga* alkaloids and vinblastine which further substantiates these concepts.

The *iboga* alkaloid family of natural products comprises over 60 members of monoterpene indoles that share a common pentacyclic skeleton of ibogamine (Fig. 1).⁵ Among the various neurological activities of ibogamine (1) and ibogaine (2), the most exciting one is their capability to attenuate the addiction to a number of drugs, although the molecular mechanism of action remains largely elusive.⁶ Ibogaine, as the most abundant alkaloid in the root bark of the shrub *Tabernanthe iboga*, has even been studied in a clinical setting.⁶ Interestingly, both enantiomers of ibogamine are not only active in reducing the

self-administration of cocaine and morphine in rats but are also devoid of tremorigenic activity—a side effect exhibited by ibogaine, which hence deserves further investigation.⁷ Catharanthine (3) and its derivative dihydrocatharanthine (4) have recently been identified as among the most potent TRPM8 antagonists and modulate cold-induced pain signals as well as mammalian thermoregulation.⁸ More importantly, the conversion of catharanthine to the potent anti-cancer drug vinblastine (5) *via* a one-pot procedure has boosted the value of this *iboga* alkaloid, and its derivatives have led to vinblastine analogs revealing insightful structure–activity relationships.⁹

Despite the variety of synthetic approaches towards different *iboga* alkaloids that have been reported, the enantioselective total syntheses remain relatively sparse.^{10–13} Since Trost's group

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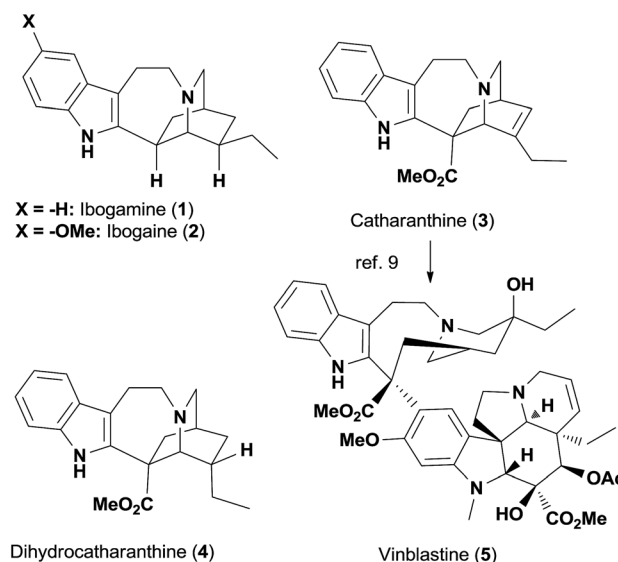


Fig. 1 Representative *iboga* alkaloids and vinblastine.



published the elegant synthesis of enantioenriched ibogamine in 1978,¹¹ the preparation of chiral isoquinuclidine fragments followed by the construction of a C2–C16 bond in the late stages (catharanthine numbering, throughout) has become the focus of asymmetric synthesis studies.¹² The only two exceptions are the efficient syntheses of (–)-ibogamine and (–)-catharanthine by White's group and Oguri's group respectively, both employing the asymmetric Diels–Alder reaction.¹³ An alternative approach to prepare such a privileged skeleton, especially in an enantioselective manner, would be a valuable addition to current synthetic endeavours and more importantly, would enable flexible structural changes of this chemotype.

Results and discussion

While seeking a unified strategy to access *iboga* alkaloids with and without the methoxycarbonyl group at C16, we envisioned two late-stage intermediates **6a** and **6b** (Fig. 2). The C20 carbonyl group of **6** could be a versatile handle for the preparation of bioactive natural products and small-molecule probes. Inspired by the transannular cyclization accomplished by Kutney and co-workers,^{10b} as well as recent advances in the fragmentation of the C16–C21 bond,¹⁴ we decided to explore the [1,2]-Stevens rearrangement of ammonium ylide **7** to construct the C16–C21 bond and give the structurally compact product **6**.¹⁵ Given that zwitterion **7** could be generated from the quaternary ammonium cation **8** upon treatment with base due to the enhanced acidity at C21, **6** would therefore be accessible

from **8** in one step. This key precursor **8** could be prepared by intramolecular alkylation of the tertiary amine **9**, for which we hypothesized that the recently developed gold-catalyzed conversion of terminal alkynes to α -chloromethyl ketones could find application.¹⁶ Thus, the tertiary amine **10** became the precursor for **9**, which could be traced back to the known amide **11** *via* propargylation and reduction.^{17,18}

We commenced our studies with the chiral amide **11a**, which was prepared from tryptamine in 3 steps *via* the organocatalytic Pictet–Spengler reaction reported by Jacobsen's group (Scheme 1).¹⁷ The introduction of the propargyl group was achieved with the protection of the nitrogen atom, and the following deprotection afforded a pair of readily separable diastereomers, where the desired stereoisomer **13a** was isolated as the major product in 52% yield over 3 steps. The subsequent reduction of **13a** by LiAlH₄ produced the tertiary amine **10a** smoothly in 85% yield. We ultimately developed a one-pot procedure that converted **10a** to the quaternary ammonium compound **8a** in good yield (see the ESI† for the determination of the counteranion).

The extensive optimization of this gold-catalyzed reaction followed by intramolecular alkylation was carried out using racemic **10a** (Table 1 and S1†). The basicity of the tertiary amine **10a** is detrimental to the cationic gold catalysis and needs to be neutralized with the addition of another equivalent of acid.^{16b,19} Using a 10 mol% (Ph₃P)AuNTf₂ catalyst and 2 equiv. of MsOH additive, we examined a variety of oxidants and identified 2-bromopyridine *N*-oxide **14** as the optimal one (Table S1†). The formation of intermediate **15** was supported by LCMS analysis, and the intermediate then underwent facile cyclization upon the treatment of the reaction mixture with a saturated aqueous

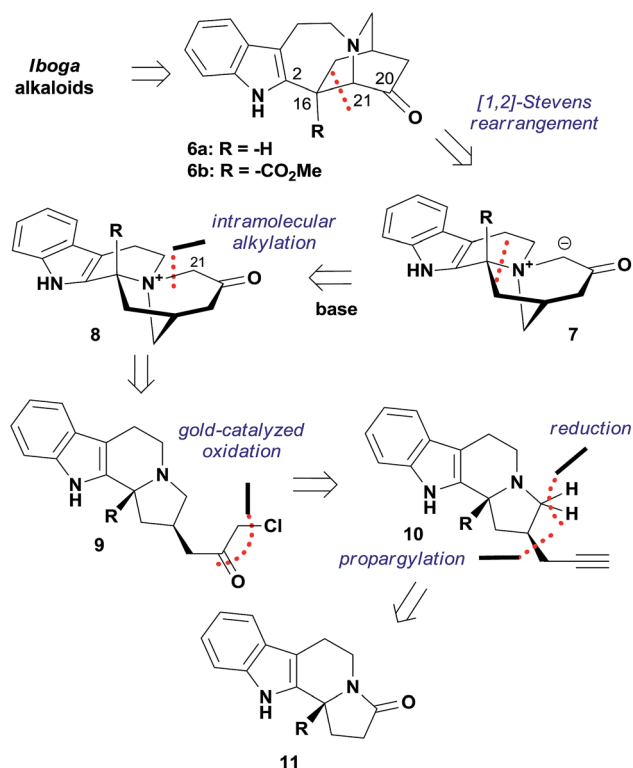
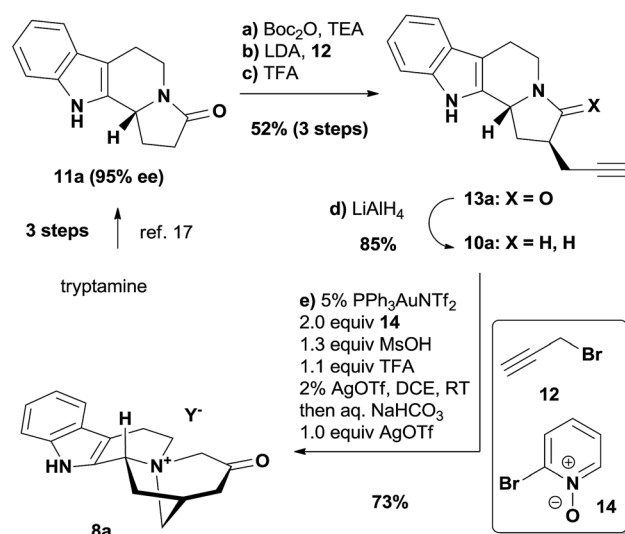
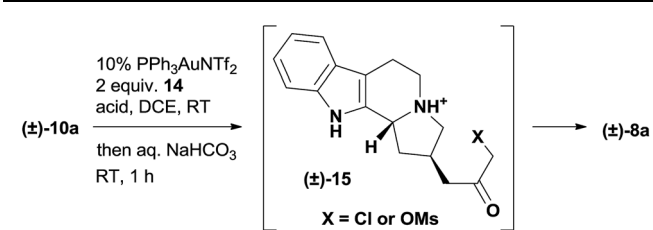


Fig. 2 Retrosynthetic analysis of *iboga* alkaloids based on the [1,2]-Stevens rearrangement reaction.



Scheme 1 Preparation of the quaternary ammonium compound **8a**. Reagents and conditions: (a) Boc₂O (3.0 equiv.), TEA (1.1 equiv.), DMAP (0.2 equiv.), DCM, RT, 14 h, 89%; (b) LDA (1.2 equiv.), propargyl bromide **12** (2.5 equiv.), –78 °C to RT, 2 h, THF; (c) TFA (5.0 equiv.), DCM, RT, 16 h; 58% over two steps; (d) LiAlH₄ (3.0 equiv.), THF, 80 °C, 1 h, 85%; (e) PPh₃AuNTf₂ (5 mol%), **14** (2.0 equiv.), MsOH (1.3 equiv.), TFA (1.1 equiv.), AgOTf (2 mol%), DCE, RT, 6 h; then NaHCO₃ (sat.), AgOTf (1.0 equiv.), RT, 73%.



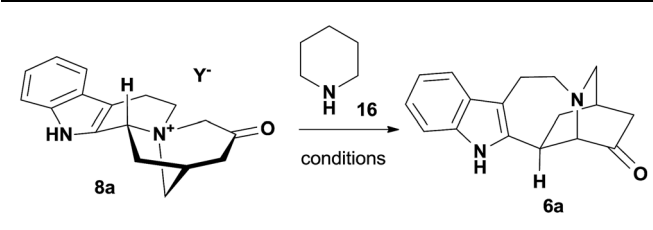
Table 1 Gold-catalyzed synthesis of **8a**: selected optimization^a


Entry	Acid	Time	Yield ^b
1	2.1 equiv. MsOH	24 h	47% ^c
2	1.3 equiv. MsOH, 1.1 equiv. TFA	20 h	63%
3	1.3 equiv. MsOH, 1.1 equiv. TFA	6 h	69% ^d
4	1.3 equiv. MsOH, 1.1 equiv. TFA	8 h	51% ^e
5	1.3 equiv. MsOH, 1.1 equiv. TFA	8 h	74% ^f

^a [**10a**] = 0.1 M (0.12 mmol). ^b Isolated yield after flash chromatography. ^c 39% starting material **10a** was recovered. ^d 200 mg scale reaction, 3% AgOTf was added as an additive. ^e 1 g scale reaction, 5% PPh₃AuNTf₂, 2% AgOTf as an additive. ^f 3 g scale reaction, 5% PPh₃AuNTf₂, 2% AgOTf as an additive, 1 equiv. AgOTf was added with NaHCO₃ (s, aq.) to facilitate the cyclization.

solution of sodium bicarbonate. The end product **8a** was obtained in 47% yield with the recovery of 39% starting material **10a** (Table 1, entry 1). Inspired by the screening of the acid additives to improve the conversion in similar transformations,²⁰ we discovered that complete consumption of **10a** was achieved in 20 h at room temperature with 1.3 equiv. MsOH and 1.1 equiv. TFA as the acid additives to afford **8a** in 63% isolated yield (entry 2). The addition of 3 mol% AgOTf significantly accelerated the reaction,²¹ which gave **8a** in 69% yield on an even larger scale (200 mg scale, entry 3). We intentionally lowered the gold catalyst loading to 5 mol% for the gram scale reaction and found that the first step went to completion in 8 h at room temperature but **8a** was isolated in only 51% yield after workup (entry 4). The conditions for the gram scale reaction were further improved by discovering that the addition of 1 equiv. AgOTf effectively promoted the cyclization step, which eventually afforded **8a** in 74% isolated yield (entry 5).

With abundant **8a** in hand, we proceeded to test the Stevens rearrangement. Initial efforts in base, solvent and temperature screening proved unfruitful, leading to either starting material recovery or decomposition (Table S2†). Inspired by the development of organocatalytic sigmatropic reactions,²² we shifted our focus to exploiting a novel Stevens rearrangement through the intermediacy of an enamine.²³ By examining a variety of amines (Table S3†), we identified that 5 equiv. of piperidine **16** could promote the desired transformation in methanol even if the isolated yield of **6a** was only 11% after heating at 170 °C for 8 h in a sealed tube (Table 2, entry 1).²⁴ We therefore turned to microwave technology and found that it was more effective than conventional thermal conditions (entry 2).²⁵ Through a series of optimization procedures including the amount of **16** (entry 3), solvent (entry 4), concentration and heating sequence (entry 5), the *iboga* alkaloid framework **6a** was eventually obtained from

Table 2 Rearrangement of **8a** to **6a**: selected optimization


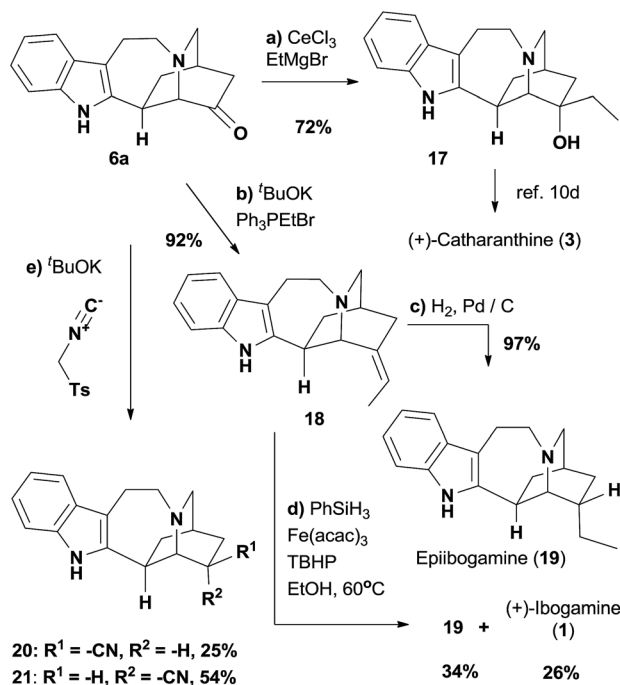
Entry	Conditions ^a	Conversion ^b	Yield ^c
1	5 equiv. 16 , MeOH, 170 °C, 8 h ^d	46%	13%
2	5 equiv. 16 , MeOH, 120 °C (mW), 12 h	N.D.	21%
3	0.4 equiv. 16 , MeOH, 120 °C (mW), 8 h	N.D.	32% ^e
4	0.4 equiv. 16 , HFIP, 150 °C (mW), 3 h	45%	47% ^{e,f}
5	0.4 equiv. 16 , HFIP, 150 °C (mW), 3 h	58%	56% ^{e,f,g}

^a [**8a**] = 0.1 M (0.075 mmol). ^b Conversion was calculated based on the recovery of **8a**. ^c Isolated yield after column chromatography. ^d The reaction was carried out in a sealed tube. ^e The reaction vial was pretreated by *N,O*-bis(trimethylsilyl)acetamide. ^f [**8a**] = 0.5 M (1 mmol). ^g The heating sequence was composed of 12 cycles with each cycle including irradiation at 150 °C for 15 min and at 50 °C for 15 min.

8a in 50–60% isolated yield (over 90% yield based on starting material recovery). Furthermore, when *N*-methylmorpholine was employed in place of piperidine **16** under the optimized reaction conditions, we did not observe the formation of **6a** and the starting material **8a** was recovered in 88% yield, thus suggesting that the formal Stevens rearrangement was not base mediated. To the best of our knowledge, this transformation represents the first example of Stevens rearrangement through secondary amine catalysis.

The stage was set for the late-stage functional group manipulations of **6a** (Scheme 2). First, the addition of an organocerium reagent, prepared from ethylmagnesium bromide, to the ketone afforded **17** as a single diastereomer, effectively completing the formal synthesis of (+)-catharanthine (**3**).^{10d,12c} The Wittig reaction was then employed to convert **6a** to olefin **18**, where the *Z* configuration of the trisubstituted olefin was assigned by a NOESY experiment (see ESI† for details). Hydrogenation of alkene **18** using activated Pd/C as the catalyst afforded epiibogamine **19** in 97% yield.²⁶ The high stereoselectivity could be attributed to the preferential addition of H₂ to the less hindered side of the molecule. Therefore we turned to the radical-based hydrogenation of electron-neutral alkenes initiated by hydrogen atom transfer.²⁷ While manganese and cobalt-based catalyst precursors also produced **19** as the predominant product (Table S4†), we were delighted to find that Fe(acac)₃, the precatalyst reported by Baran and co-workers for reductive alkene coupling,²⁸ afforded separable **19** and (+)-ibogamine **1** in 34% and 26% yields, respectively. The key intermediate **6a** could be expediently decorated to other interesting derivatives with an *iboga* alkaloid skeleton. For instance, the reductive cyanation of ketone **6a** with tosylmethylisocyanide produced a pair of separable diastereomers **20** and **21** in 25% and 54% yields, respectively.²⁹ The structures of racemic **19** and **20** were determined unambiguously by X-ray crystallography,³⁰

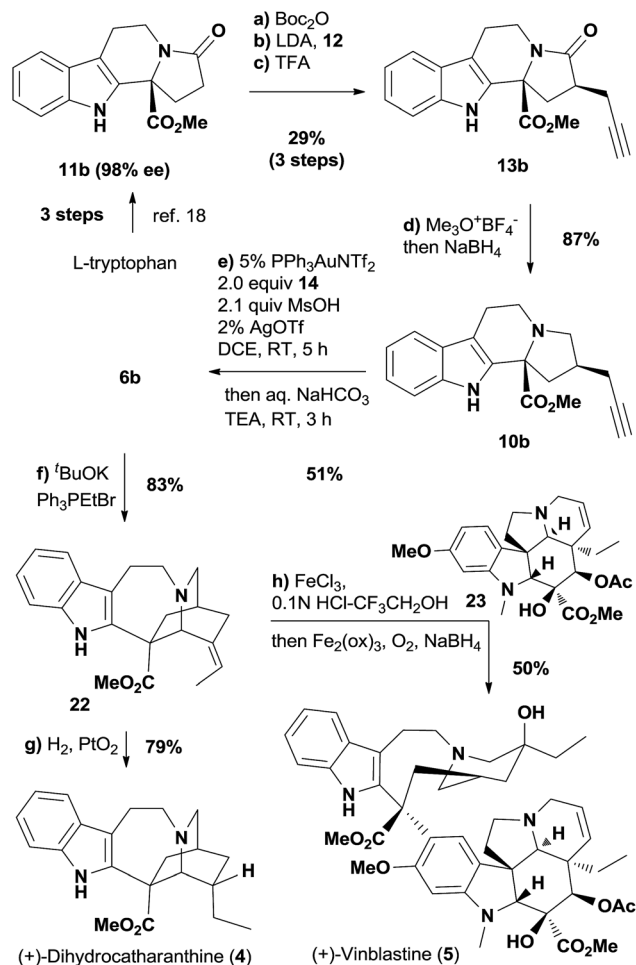




Scheme 2 Formal synthesis of catharanthine (3) and the total synthesis of ibogamine (1). Reagents and conditions: (a) CeCl_3 (2.5 equiv.), EtMgBr (2.0 equiv.), THF, 0.5 h, 72%; (b) $t\text{BuOK}$ (3.0 equiv.), Ph_3PETBr (3.0 equiv.), THF, 2 h, 92%; (c) H_2 , Pd/C (2.0 equiv.), MeOH, 2 h, 97%; (d) PhSiH_3 (2.5 equiv.), $\text{Fe}(\text{acac})_3$ (0.8 equiv.), TBHP (1.5 equiv.), EtOH , 60°C , 6 h; 19, 34%; 1, 26%; (e) $t\text{BuOK}$ (2.5 equiv.), TosMIC (1.3 equiv.), EtOH (1.7 equiv.), DME, 12 h; 20, 25%; 21, 54%.

while the analytical data of 1 and 17 corresponded well with that in the literature.^{12a,c}

Encouraged by the completion of (+)-ibogamine (1), we moved towards the synthetic study of the *iboga* alkaloids with the methoxycarbonyl group at C16 (Scheme 3). The amide 13b was prepared from the known compound 11b with excellent enantiopurity¹⁸ following the same procedures depicted in Scheme 1, while the undesired diastereomeric amide could also be converted to 13b readily (see ESI† for details). The selective reduction of the amide carbonyl group in 13b subsequently afforded the tertiary amine 10b.³¹ We fortunately isolated a trace amount of the rearranged product 6b after the work-up of the gold-catalyzed oxidation reaction of 10b, indicating that the [1,2]-shift was quite facile in the presence of the C16 methoxycarbonyl group. Therefore, the gold-catalyzed oxidation was followed by the addition of a saturated aqueous solution of sodium bicarbonate and excess triethylamine to promote the cyclization and rearrangement. Gratifyingly, this one-pot procedure afforded ketone 6b in 51% yield from 10b under mild reaction conditions. The Wittig reaction of 6b gave rise to 22—a catharanthine isomer with an exocyclic *versus* endocyclic double bond. Hydrogenation of 22 afforded dihydrocatharanthine (4) in 79% yield. Interestingly, 22 differs from a known compound derived from catharanthine in the olefin geometry.³² Eventually, employing the conditions reported by Boger and coworkers,^{32a} we successfully made vinblastine (5) in 50% yield by coupling 22 with commercially available vindoline 23.



Scheme 3 Syntheses of dihydrocatharanthine (4) and vinblastine (5). Reagents and conditions: (a) Boc_2O (3.0 equiv.), TEA (1.1 equiv.), DMAP (0.2 equiv.), DCM, RT, 14 h, 87%; (b) LDA (1.2 equiv.), 12 (2.5 equiv.), THF, 12 h; (c) TFA (5.0 equiv.), DCM, 16 h, 33% over two steps; (d) trimethyloxonium tetrafluoroborate (2.5 equiv.), 2,6-di-*tert*-butylpyridine (3.5 equiv.), DCM, 12 h; then NaBH_4 (0.5 equiv.), MeOH, 0.5 h, 87%; (e) $\text{PPh}_3\text{AuNTf}_2$ (5 mol%), 14 (2.0 equiv.), MsOH (2.1 equiv.), AgOTf (2 mol%), DCE, RT, 5 h; then NaHCO_3 (sat.), TEA (3.0 equiv.), RT, 3 h, 51%; (f) $t\text{BuOK}$ (3.0 equiv.), Ph_3PETBr (3.0 equiv.), THF, 2 h, 83%; (g) H_2 , PtO_2 (0.3 equiv.), MeOH, 15 h, 79%; (h) vindoline 23 (1.2 equiv.), $\text{HCl}-\text{CF}_3\text{CH}_2\text{OH}$, FeCl_3 (5.0 equiv.), 2 h; $\text{Fe}_2(\text{ox})_3$ (30 equiv.), O_2 ; then NaBH_4 (20 equiv.), 0°C , 0.5 h, 50%.

It is noteworthy that the chiral compound 6b, which was prepared from *L*-tryptophan in 8 steps, would be a valuable synthetic intermediate towards vinblastine analogs. To illustrate this point, compounds 24 and 25, vinblastine analogs differing only in the C20' substituent, were readily prepared by employing the Wittig reaction of 6b followed by biomimetic coupling (Fig. 3). We also prepared fluoroalkene 27 using reagent 26,³³ where the *E* configuration of the olefin was assigned by a NOESY experiment (see ESI† for details). Interestingly, the coupling of 27 with vindoline (23) afforded aldehyde 28 in 68% yield (see Fig. S1† for a proposed mechanism). The cytotoxicities of 24 and 25 were measured in the HCT116 cell line using vinblastine (5) as a positive control. Our data indicated that 24 was over 100-fold less active than vinblastine,



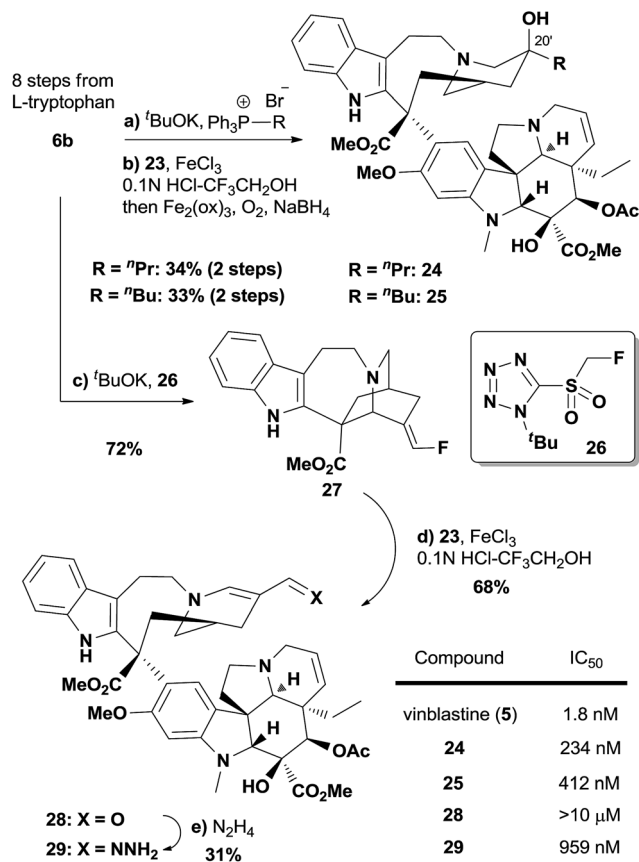


Fig. 3 Synthesis of vinblastine analogs and their cell growth inhibitory activity.

and that **25** was even less active than **24**. Based on a 40-step total synthesis, Fukuyama's group has reported inactive vinblastine analogs with C20' acetylene functionalities that differ significantly in size and shape with the ethyl group of the natural product.³⁴ Herein we showed that even a subtle change—with the C20' alkyl substituent length extended for one (**24**) or two more carbons (**25**)—was enough to dramatically decrease the potency. This could be rationalized by the X-ray crystallographic analysis of the vinblastine–tubulin interactions, in which the C20' ethyl substituent of vinblastine is embedded in a hydrophobic binding site.³⁵ Interestingly, the aldehyde analog of vinblastine, compound **28**, almost lost the ability to inhibit the growth of HCT116 cells. However, compound **29**, obtained by the condensation of **28** with hydrazine, showed decent cytotoxicity (IC₅₀ = 959 nM). This observation implies the necessity of a hydrogen bond donor around the C20' position,³⁶ although further investigation is needed to provide more insight into the hydrazone analog.

Conclusions

In summary, we have accomplished a unique and general route for the enantioselective synthesis of *iboga* alkaloids by developing a Stevens rearrangement through secondary amine catalysis and an oxidation/cyclization/rearrangement tandem

sequence. The precise mechanism of the rearrangement remains to be investigated to identify whether a radical or an ionic intermediate is involved. Nonetheless, both reactions have the potential to be applied in the synthesis of a myriad of complex alkaloids. This study nicely exemplifies the total synthesis of complex natural products serving as not only a driving force for advancing the synthetic methodology but also as an important source for providing analogs. Furthermore, this practical approach to modify *iboga* alkaloids and vinblastine paves the way for studies into their pronounced pharmacological properties using state-of-the-art chemical biology technologies, which are underway in our group and will be reported in due course.

Acknowledgements

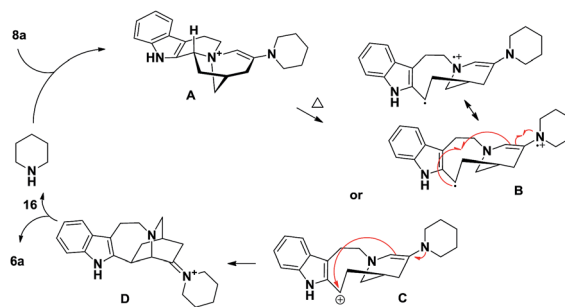
This work was supported by generous start-up funds from the College of Chemistry and Molecular Engineering, Peking University and Peking-Tsinghua Center for Life Sciences, and the National Science Foundation of China (Grant No. 21472003 and 31521004). We thank Dr Nengdong Wang and Prof. Wenxiong Zhang (Peking University) for their help in analyzing the X-ray crystallography data, and Prof. Jian Wang (Tsinghua University) for his help in chiral HPLC analysis.

Notes and references

- R. W. Hoffmann, *Angew. Chem., Int. Ed.*, 2013, **52**, 123–130.
- P. A. Wender and B. L. Miller, *Nature*, 2009, **460**, 197–201.
- (a) W. Ren, Y. Bian, Z. Zhang, H. Shang, P. Zhang, Y. Chen, Z. Yang, T. Luo and Y. Tang, *Angew. Chem., Int. Ed.*, 2012, **51**, 6984–6988; (b) X. Yu, F. Su, C. Liu, H. Yuan, S. Zhao, Z. Zhou, T. Quan and T. Luo, *J. Am. Chem. Soc.*, 2016, **138**, 6261–6270; rearrangement reactions have been used extensively in the organic synthesis of natural products. For some recent examples, see: (c) C. L. Martin, L. E. Overman and J. M. Rohde, *J. Am. Chem. Soc.*, 2010, **132**, 4894–4906; (d) L. Jørgensen, S. J. McKerrall, C. A. Kuttruff, F. Ungeheuer, J. Felding and P. S. Baran, *Science*, 2013, **341**, 878–882; (e) C. W. Plummer, C. S. Wei, C. E. Yozwiak, A. Soheili, S. O. Smithback and J. L. Leighton, *J. Am. Chem. Soc.*, 2014, **136**, 9878–9881; (f) C. P. Ting and T. J. Maimone, *J. Am. Chem. Soc.*, 2015, **137**, 10516–10519.
- S. B. Jones, B. Simmons, A. Mastracchio and D. W. C. MacMillan, *Nature*, 2011, **475**, 183–188.
- K. R. Alper, *Alkaloids: Chem. Biol. Perspect.*, 2001, **56**, 1–38.
- (a) P. Popik, R. T. Layer and P. Skolnick, *Pharmacol. Rev.*, 1995, **47**, 235–253; (b) C. Jenks, *Nat. Prod. Lett.*, 2002, **16**, 71–76.
- S. D. Glick, M. E. Kuehne, J. Raucci, T. E. Wilson, D. Larson, R. W. Keller Jr and J. N. Carlson, *Brain Res.*, 1994, **657**, 14–22.
- Y. Terada, M. Kitajima, F. Taguchi, H. Takayama, S. Horie and T. Watanabe, *J. Nat. Prod.*, 2014, **77**, 1831–1838.
- (a) H. Ishikawa, D. A. Colby and D. L. Boger, *J. Am. Chem. Soc.*, 2008, **130**, 420–421; (b) For a recent review, see: J. E. Sears and D. L. Boger, *Acc. Chem. Res.*, 2015, **48**, 653–662.



- 10 For a recent review, see: (a) G. K. Jana, S. Paul and S. Sinha, *Org. Prep. Proced. Int.*, 2011, **43**, 541–573; (b) J. P. Kutney, R. T. Brown and E. Piers, *J. Am. Chem. Soc.*, 1964, **86**, 2287–2288; (c) G. Büchi, D. L. Coffen, K. Kocsis, P. E. Sonnet and F. E. Ziegler, *J. Am. Chem. Soc.*, 1965, **87**, 2073–2075; (d) G. Büchi, P. Kulsa, K. Ogasawara and R. L. Rosati, *J. Am. Chem. Soc.*, 1970, **92**, 999–1005; (e) J. W. Huffman, G. Shanmugasundaram, R. Sawdaye, P. C. Raveendranath and R. C. Desai, *J. Org. Chem.*, 1985, **50**, 1460–1464; (f) M. E. Kuehne and P. J. Reider, *J. Org. Chem.*, 1985, **50**, 1464–1467; (g) W. Nagata, H. Hirai, T. Okumura and K. Kawata, *J. Am. Chem. Soc.*, 1968, **90**, 1650–1651; (h) S. Hiari, K. Kawata and W. Nagata, *Chem. Commun.*, 1968, 1016–1017; (i) T. Imanishi, N. Yagi, H. Shin and M. Hanaoka, *Tetrahedron Lett.*, 1981, **22**, 4001–4004; (j) T. Imanishi, N. Yagi and M. Hanaoka, *Chem. Pharm. Bull.*, 1985, **33**, 4202–4211; (k) G. R. Krow, D. A. Shaw, B. Lynch, W. Lester, S. W. Szczepanski, K. Raghavachari and A. E. Derome, *J. Org. Chem.*, 1988, **53**, 2258–2262; (l) C. Herdeis and C. Hartke-Karger, *Liebigs Ann. Chem.*, 1991, 99–104; (m) P. Rosenmund, W. H. Haase, J. Bauer and R. Frische, *Chem. Ber.*, 1975, **108**, 1871–1895; (n) S. I. Sallay, *J. Am. Chem. Soc.*, 1967, **89**, 6762–6763; (o) M. Ikezaki, T. Wakamastu and Y. Ban, *J. Chem. Soc. D*, 1969, 88–89; (p) K. J. Henrey, P. A. Grieco Jr and W. J. DuBay, *Tetrahedron Lett.*, 1996, **37**, 8289–8292; (q) W. G. Bornmann and M. E. Kuhene, *J. Org. Chem.*, 1992, **57**, 1752–1760; (r) A. C. Kruegel, S. Rakshit, X. Li and D. Sames, *J. Org. Chem.*, 2015, **80**, 2062–2071.
- 11 B. M. Trost, S. A. Godleski and J. P. Genêt, *J. Am. Chem. Soc.*, 1978, **100**, 3930–3931.
- 12 (a) D. M. Hodgson and J.-M. Galano, *Org. Lett.*, 2005, **7**, 2221–2224; (b) S. Höck and H.-J. Borschberg, *Helv. Chim. Acta*, 2006, **89**, 542–557; (c) L. Moisan, P. Thuéry, M. Nicolas, E. Doris and B. Rousseau, *Angew. Chem., Int. Ed.*, 2006, **45**, 5334–5336; (d) M. Harada, K. N. Asaba, M. Iwai, N. Kogure, M. Kitajima and H. Takayama, *Org. Lett.*, 2012, **14**, 5800–5803; (e) K. Ishihara, H. Yamada and M. Akakura, *Chem. Commun.*, 2014, **50**, 6357–6360; (f) M. Hatano, Y. Goto, A. Izumiseki, M. Akakura and K. Ishihara, *J. Am. Chem. Soc.*, 2015, **137**, 13472–13475.
- 13 (a) J. D. White and Y. Choi, *Org. Lett.*, 2000, **2**, 2373–2376; (b) H. Mizoguchi, H. Oikawa and H. Oguri, *Nat. Chem.*, 2014, **6**, 57–64.
- 14 J. W. Beatty and C. R. J. Stephenson, *J. Am. Chem. Soc.*, 2014, **136**, 10270–10273 and references cited therein.
- 15 (a) T. S. Stevens, E. M. Creighton, A. B. Gordon and M. MacNicol, *J. Chem. Soc.*, 1928, 3193–3197; (b) I. E. Markó in *Comprehensive Organic Synthesis*, ed. B. M. Trost and I. Fleming, Pergamon, Oxford, 1991, vol. 3, pp. 913–974; (c) M. G. Charest, C. D. Lerner, J. D. Brubaker, D. R. Siegel and A. G. Myers, *Science*, 2005, **308**, 395–398; (d) J. A. Vanecko, H. Wan and F. G. West, *Tetrahedron*, 2006, **62**, 1043–1062; (e) P. Tuzina and P. Somfai, *Org. Lett.*, 2009, **11**, 919–921; (f) T. M. Bott, J. A. Vanecko and F. G. West, *J. Org. Chem.*, 2009, **74**, 2832–2836; (g) M. Valpuesta, M. Ariza, A. Díaz and R. Suau, *Eur. J. Org. Chem.*, 2010, 4393–4401; (h) L. Palombi, *Catal. Commun.*, 2011, **12**, 485–488; (i) J. Clayden, M. Donnard, J. Lefranc and D. J. Tetlow, *Chem. Commun.*, 2011, **47**, 4624–4639; (j) S. Harada, M. Kono, T. Nozaki, Y. Menjo, T. Nemoto and Y. Hamada, *J. Org. Chem.*, 2015, **80**, 10317–10333.
- 16 (a) W. He, L. Xie, Y. Xu, J. Xiang and L. Zhang, *Org. Biomol. Chem.*, 2012, **10**, 3168–3171; (b) E. L. Noey, Y. Luo, L. Zhang and K. N. Houk, *J. Am. Chem. Soc.*, 2012, **134**, 1078–1084.
- 17 I. T. Raheem, P. S. Thiara, E. A. Peterson and E. N. Jacobsen, *J. Am. Chem. Soc.*, 2007, **129**, 13404–13405.
- 18 (a) P. Magnus, M. Ladlow, C. S. Kim and P. Boniface, *Heterocycles*, 1989, **28**, 951–956; (b) P. Magnus, J. S. Mendoza, A. Stamford, M. Ladlow and P. Willis, *J. Am. Chem. Soc.*, 1992, **114**, 10232–10245.
- 19 (a) P. de Frémont, N. Marion and S. P. Nolan, *J. Organomet. Chem.*, 2009, **694**, 551–560; (b) R. L. Lalonde, W. E. Brenzovich, D. Benitez, E. Tkatchouk, K. Kelley, W. A. Goddard and F. D. Toste, *Chem. Sci.*, 2010, **1**, 226–233.
- 20 (a) L. Ye, L. Cui, G. Zhang and L. Zhang, *J. Am. Chem. Soc.*, 2010, **132**, 3258–3259; (b) E. P. A. Talbot, M. Richardson, J. M. McKenna and F. D. Toste, *Adv. Synth. Catal.*, 2014, **356**, 687–691.
- 21 J. Han, N. Shimizu, Z. Lu, H. Amii, G. B. Hammond and B. Xu, *Org. Lett.*, 2014, **16**, 3500–3503.
- 22 (a) G. Valero and A. Moyano, in *Comprehensive Enantioselective Organocatalysis: Catalysts, Reactions, and Applications*, ed. P. I. Dalko, Wiley-VCH Verlag GmbH & Co. KGaA, 2013, pp. 1191–1224; (b) A. McNally, B. Evans and M. J. Gaunt, *Angew. Chem., Int. Ed.*, 2006, **45**, 2116–2119.
- 23 (a) H. G. Lindwall and J. S. MacLennan, *J. Am. Chem. Soc.*, 1932, **54**, 4739–4744; (b) S. Mukherjee, J. W. Yang, S. Hoffmann and B. List, *Chem. Rev.*, 2007, **107**, 5471–5569 and references cited therein; (c) D. W. C. MacMillan, *Nature*, 2008, **455**, 304–308 and references cited therein.
- 24 A proposed mechanism for **16**-catalyzed formation of **6a**:



when **8a** and **16** were heated at 80 °C in HFIP, the formation of enamine intermediate **A** was supported by LCMS analysis.

- 25 For a recent review, see: G. B. Dudley, R. Richert and A. E. Stiegman, *Chem. Sci.*, 2015, **6**, 2144–2152.
- 26 The synthesis of racemic epiibogamine has been reported. See: (a) G. Büchi, D. L. Coffen, K. Kocsis, P. E. Sonnet and F. E. Ziegler, *J. Am. Chem. Soc.*, 1966, **88**, 3099–3109; (b) G. K. Jana and S. Sinha, *Tetrahedron Lett.*, 2012, **53**, 1671–1674.



- 27 (a) K. Iwasaki, K. K. Wan, A. Oppedisano, S. W. M. Crossley and R. A. Shenvi, *J. Am. Chem. Soc.*, 2014, **136**, 1300–1303; (b) S. M. King, X. Ma and S. B. Herzon, *J. Am. Chem. Soc.*, 2014, **136**, 6884–6887.
- 28 (a) J. C. Lo, Y. Yabe and P. S. Baran, *J. Am. Chem. Soc.*, 2014, **136**, 1304–1307; (b) J. C. Lo, J. Gui, Y. Yabe, C.-M. Pan and P. S. Baran, *Nature*, 2014, **516**, 343–348.
- 29 O. H. Oldenziel, D. Van Leusen and A. M. Van Leusen, *J. Org. Chem.*, 1977, **42**, 3114–3118.
- 30 CCDC-1433182 (19), CCDC-1433188 (20) and CCDC-1433180 (S4) contain the supplementary crystallographic data for this paper.†
- 31 M. Ito, C. W. Clark, M. Mortimore, J. B. Goh and S. F. Martin, *J. Am. Chem. Soc.*, 2001, **123**, 8003–8010.
- 32 (a) H. Ishikawa, D. A. Colby, S. Seto, P. Va, A. Tam, H. Kakei, T. J. Rayl, I. Hwang and D. L. Boger, *J. Am. Chem. Soc.*, 2009, **131**, 4904–4916; (b) E. Giovanelli, S. Leroux, L. Moisan, H. Carreyre, P. Thuéry, D.-A. Buisson, A. Meddour, J.-M. Coustard, S. Thibaudeau, B. Rousseau, M. Nicolas, P. Hellier and E. Doris, *Org. Lett.*, 2011, **13**, 4116–4119.
- 33 L. Zhu, C. Ni, Y. Zhao and J. Hu, *Tetrahedron*, 2010, **66**, 5089–5100.
- 34 T. Miyazaki, S. Yokoshima, S. Simizu, H. Osada, H. Tokuyama and T. Fukuyama, *Org. Lett.*, 2007, **9**, 4737–4740.
- 35 (a) B. Gigant, C. Wang, R. B. G. Ravelli, F. Roussi, M. O. Steinmetz, P. A. Curmi, A. Sobel and M. Knossow, *Nature*, 2005, **435**, 519–522; (b) Y. Wang, F. W. Benz, Y. Wu, Q. Wang, Y. Chen, X. Chen, H. Li, Y. Zhang, R. Zhang and J. Yang, *Mol. Pharmacol.*, 2016, **89**, 233–242.
- 36 E. K. Leggans, K. K. Duncan, T. J. Barker, K. D. Schleicher and D. L. Boger, *J. Med. Chem.*, 2013, **56**, 628–639.

