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Overcoming antibiotic resistance in *Pseudomonas aeruginosa* biofilms using glycopeptide dendrimers†

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Antibiotic resistance in the opportunistic pathogen *Pseudomonas aeruginosa* is partly caused by biofilms forming a physical barrier to antibiotic penetration. Here we focused on modifying tetravalent glycopeptide dendrimer ligands of *P. aeruginosa* lectins LecB or LecA to increase their biofilm inhibition activity. First heteroglycoclusters were investigated displaying one pair each of LecB specific fucosyl groups and LecA specific galactosyl groups and binding simultaneously to both lectins, one of which gave the first fully resolved crystal structure of a peptide dendrimer as LecB complex providing a structural model for dendrimer–lectin interactions (PDB 5D2A). Biofilm inhibition was increased by introducing additional cationic residues in these dendrimers but resulted in bactericidal effects similar to those of non-glycosylated polycationic antimicrobial peptide dendrimers. In a second approach dendrimers displaying four copies of the natural LecB ligand Lewis^a were prepared leading to slightly stronger LecB binding and biofilm inhibition. Finally synergistic application of a LecB specific non-bactericidal antibiofilm dendrimer with the antibiotic tobramycin at sub-inhibitory concentrations of both compounds allowed effective biofilm inhibition and dispersal.

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Introduction

The appearance of widespread antibiotic resistance in pathogenic bacteria is a global public health threat. One approach to overcome antibiotic resistance focuses on preventing the formation of biofilms, by which bacteria escape antibiotics by forming a physical barrier to their penetration. Biofilm inhibitors should not be toxic by themselves to avoid the possibility of an evolutionary pressure towards resistance.¹ One potential application of this approach concerns *Pseudomonas aeruginosa*, a gram negative bacterium causing lethal airway infections in immunocompromised and cystic fibrosis patients by forming antibiotic resistant biofilms.² It has been shown that two *P. aeruginosa* lectins, namely LecA binding specifically to galactosides, and LecB binding specifically to fucosides, are implicated

in biofilm formation by this bacterium,^{3–7} as evidenced by the impaired biofilm formation in deletion mutants^{8,9} and the successful treatment of *P. aeruginosa* infections with lectin-binding saccharide solutions.^{10–12}

Recently we reported glycopeptide dendrimers targeting these lectins and capable of both biofilm inhibition and dispersal of already established biofilms. The first dendrimer **FD2**, a tetravalent fucosylated dendrimer with sequence (fucose- α -CH₂CO-Lys-Pro-Leu)₄(Lys-Phe-Lys-Ile)₂Lys-His-Ile-NH₂ (Lys = branching lysine residue), was initially identified in a glycopeptide dendrimer combinatorial library^{13,14} by screening for binding to a fucose specific plant lectin,^{15–18} and showed tight binding to LecB.^{19–21} Two additional dendrimers were subsequently developed by appending galactosyl groups to the peptide backbone of **FD2**, yielding tight binding LecA ligands **GalAG2** and **GalBG2** (Scheme 1).²² These glycopeptide dendrimers belong to the very few LecA or LecB ligands thus far reported to display *P. aeruginosa* biofilm inhibition properties,²³ among a large number of related multivalent lectin ligands.^{24–30} Structure–activity relationship studies with **FD2** and **GalAG2** showed that multivalency and the nature of the amino acid sequence were critical for biological activity, however in both cases no significant activity increase was achieved beyond the level of a minimal biofilm inhibitory concentration (MBIC) value of 20 μ M, calling for alternative approaches for activity improvement.^{20,21,31–34}

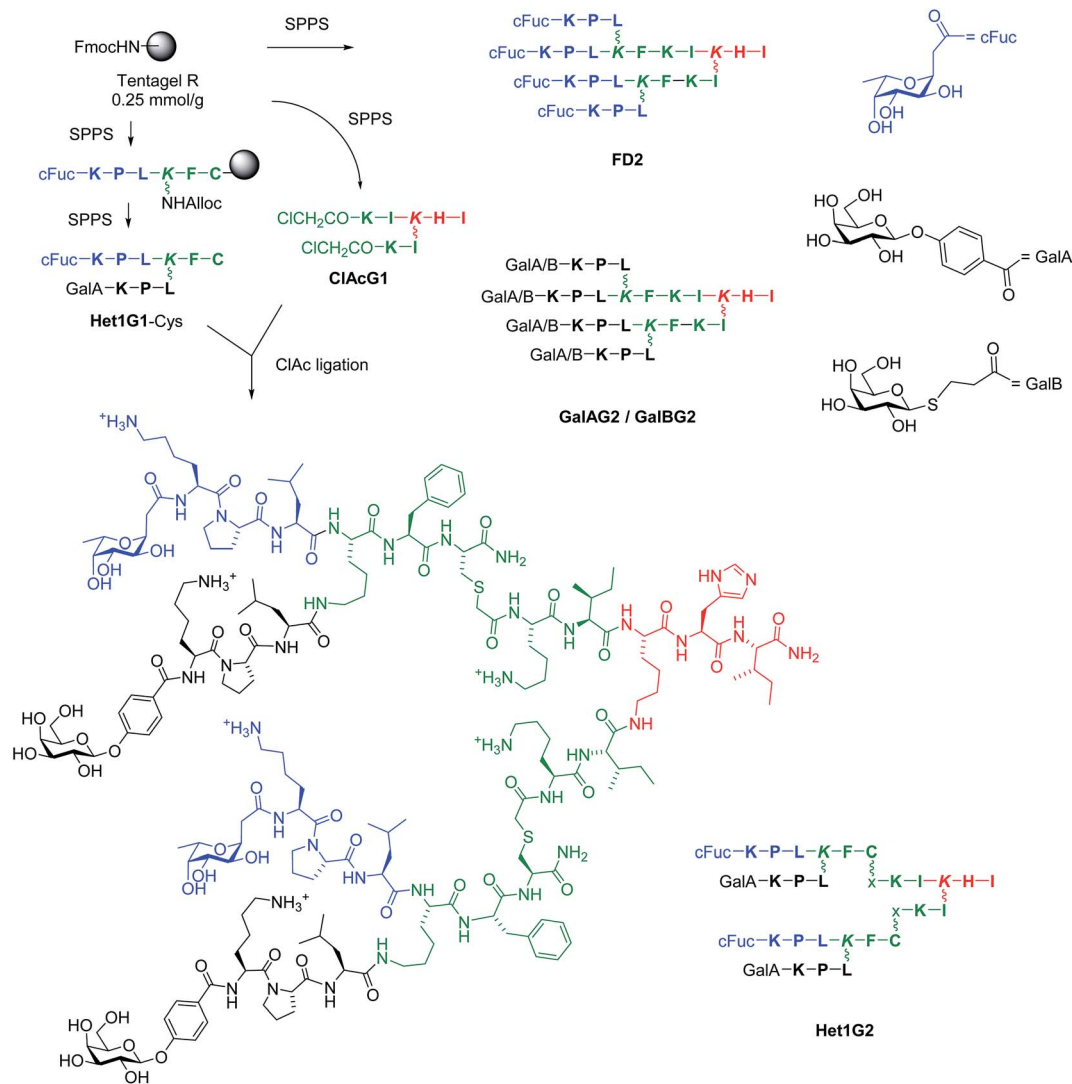
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† Electronic supplementary information (ESI) available: Details of carbohydrate synthesis building blocks and peptide dendrimer synthesis, ITC plots, biofilm inhibition and dispersal assay, MIC determination, cell viability assays and control growth curves, and crystallographic data table for 5D2A. See DOI: 10.1039/c5sc03635f





Scheme 1 Structure and synthesis of glycopeptide dendrimers. Conditions: SPPS: (a) Fmoc deprotection with piperidine/NMP 1 : 4 (v/v), 20 min; alloc deprotection: Pd(PPh₃)₄, PhSiH₃, CH₂Cl₂; amino acid coupling (3 eq. Fmoc-aa-OH, 3 eq. PyBOP, 5 eq. DIEA in NMP), 2–4 hours; carbohydrate coupling: 4 eq. carbohydrate building block, 3 eq. HATU, 5 eq. DIPEA in NMP, overnight; (b) deacetylation: MeOH/25% NH₃/H₂O (8 : 1 : 1, v/v/v); (c) cleavage: TFA/TIS/H₂O (95 : 2.5 : 2.5, v/v/v) or TFA/TIS/H₂O/1,2-ethanedithiol (92.5/2.5/2.5/2.5) for Cys containing peptide; (d) RP-HPLC purification, ClAc ligation: ClAcG1 (1 eq.), Het1G1-Cys (3 eq.), KI (20 eq.) DIPEA (55 eq.) in NMP/H₂O (1 : 1, v/v), RT, overnight under Argon atmosphere. One letter codes for L-amino acids, x = -CH₂-CO-, wobbly bonds mark the side-chain lysine connectivity. The color codes are used to differentiate G0 (red), G1 (green) and G2 (black/blue) branches.

Herein we report the extension of our search towards more potent biofilm inhibitors through further synthetic variations and activity combination approaches. The multivalent chloroacetyl cysteine thioether (ClAc) ligation³⁵ was used as an efficient method to access heteroglycoclusters such as **Het1G2** targeting both LecA and LecB (Scheme 1). One of the heteroglycoclusters provided the first fully resolved crystal structure of a peptide dendrimer as a LecB complex and a structural model for dendrimer–LecB interactions. Heteroglycoclusters incorporating cationic residues showed enhanced biofilm inhibition, however the activity reflected a bactericidal effect similar to that of membrane disrupting polycationic dendrimers.³⁶ In a second approach LecB targeting glycopeptide dendrimers were investigated displaying analogs of the Lewis^a antigen, which is the

probable natural LecB ligand,³⁷ resulting in stronger LecB binding but only a small increase in biofilm inhibition. Finally synergistic application of the LecB specific non-bactericidal antibiofilm dendrimer **FD2** with the antibiotic tobramycin at sub-inhibitory concentrations of both compounds allowed effective biofilm inhibition and dispersal.

Results and discussion

Synthesis and lectin binding of heteroglycoclusters targeting LecA and LecB

Initial biofilm inhibition experiments using the LecB specific dendrimer **FD2** and the LecA specific dendrimer **GalAG2** simultaneously showed simple additive effects between both



dendrimers (see ESI Fig. S109†). Heteroglycocluster analogs of these dendrimers displaying fucosyl and galactosyl groups were prepared to test if a compound targeting both lectins might show enhanced activity compared to the co-application. Due to difficulties in obtaining 8-fold glycosylated G3 peptide dendrimers we focused on the preparation of 4-fold glycosylated G2 peptide dendrimers carrying one pair each of fucosyl and galactosyl groups, which should enable binding to LecB respectively LecA by means of a chelate binding mode on two adjacent carbohydrate binding sites of the same lectin tetramer.^{33,38}

The synthesis of G2 glycoclusters was carried out by ClAc ligation between the bis-chloroacetylated G1 dendrimer **ClAcG1** and bis-glycosylated G1 peptide dendrimers with a cysteine residue at their core (Scheme 1).³⁵ This convergent synthesis was more economical than the previously used direct SPPS of G2 glycopeptide dendrimers, for which the four-fold glycoside coupling consumed large amounts of carbohydrate building block.³⁴ The method was used to prepare analogs of **FD2**, **GalAG2** and **GalBG2**, namely **FD2x**, **GalAG2x** and **GalBG2x**, by

ligation of the G1 intermediates **FD1-Cys**, **GalAG1-Cys** and **GalBG1-Cys** to the bis-chloroacetylated G1 dendrimer **ClAcG1**. Heteroglycoclusters were prepared by the same ligation approach using bis-glycosylated G1 dendrimers bearing a different carbohydrate on each branch, which were obtained using an alloc protected lysine branching residue allowing sequential growth of each G1 branch. All combinations of α -L-fucosylacetic acid (cFuc) with 4-(β -D-galactosyloxy)-benzoic acid (GalA) or 3-(β -D-thiogalactosyl)-propionic acid (GalB) on the main chain tripeptide *versus* side chain tripeptide were realized, yielding intermediates **Het1G1-Cys** to **Het4G1-Cys** and G2 heteroglycoclusters **Het1G2–Het4G2**. The same synthesis was carried out using the tripeptide Lys-Lys-Leu instead of Lys-Pro-Leu in the outer branch of the peptide dendrimer to obtain more polycationic glycoclusters combining lectin binding with a membrane directed antimicrobial effect, yielding G1 intermediates **Het5G1–Het8G1** and the G2 ligation products **Het5G2–Het8G2** (Table 1).

The binding affinities of the different compounds to *P. aeruginosa* lectins LecB and LecA was investigated by isothermal

Table 1 Sequence and synthesis of G2 glycopeptide dendrimers

Compound	Sequence ^a	MS calc./obs. ^b	Yield ^c (mg/%)
FD0	cFuc-KPL	544.34/544.33	19.5/29 ^d
GalAG0	GalA-KPL	638.33/638.2	20.9/55 ^e
GalBG0	GalB-KPL	606.31/606.2	11/30 ^e
FD1	(cFuc-KPL) ₂ KFKI	1586.97/1586.98	123.3/63 ^d
GalAG1	(GalA-KPL) ₂ KFKI	1774.98/1774.8	45.7/43 ^e
GalBG1	(GalB-KPL) ₂ KFKI	1710.93/1710.8	39.5/39 ^e
FD1-Cys	(cFuc-KPL) ₂ KFC	1448.81/1447.80	33.5/18
GalAG1-Cys	(GalA-KPL) ₂ KFC	1636.82/1637.82	79/39
GalBG1-Cys	(GalB-KPL) ₂ KFC	1572.77/1573.78	29.5/15
Het1G1-Cys	cFuc-KPLK(GalA-KPL)FC	1542.82/1543.81	63/33
Het2G1-Cys	GalA-KPLK(cFuc-KPL)FC	1542.82/1543.81	50/26
Het3G1-Cys	cFuc-KPLK(GalB-KPL)FC	1510.79/1510.79	52/28
Het4G1-Cys	GalB-KPLK(cFuc-KPL)FC	1510.79/1510.78	50/26
Het5G1-Cys	cFuc-KKLLK(GalA-KKL)FC	1604.90/1604.90	30/15
Het6G1-Cys	GalA-KKLLK(cFuc-KKL)FC	1604.90/1604.90	32.5/16
Het7G1-Cys	cFuc-KKLLK(GalB-KKL)FC	1572.88/1572.88	66/34
Het8G1-Cys	GalB-KKLLK(cFuc-KKL)FC	1572.88/1572.87	52/26
ClAcG1	(ClAc-KI) ₂ KHI	1030.58/1029.57	89/70
FD2	(cFuc-KPL) ₄ (KFKI) ₂ KHI	3534.16/3534.16	31/8 ^d
GalAG2	(GalA-KPL) ₄ (KFKI) ₂ KHI	3909.17/3911	18.2/7 ^e
GalBG2	(GalB-KPL) ₄ (KFKI) ₂ KHI	3783.07/3783	22.1/10 ^e
FD2x	(cFuc-KPL) ₄ (KFCxKI) ₂ KHI	3854.22/3854.23	7.9/70
GalAG2x	(GalA-KPL) ₄ (KFCxKI) ₂ KHI	4230.24/4231.24	2.9/24
GalBG2x	(GalB-KPL) ₄ (KFCxKI) ₂ KHI	4102.15/4103.15	3.8/32
Het1G2	(cFuc-KPLK(GalA-KPL)FCxKI) ₂ KHI	4042.23/4043.23	10.3/88
Het2G2	(GalA-KPLK(cFuc-KPL)FCxKI) ₂ KHI	4042.23/4042.23	9.3/79
Het3G2	(cFuc-KPLK(GalB-KPL)FCxKI) ₂ KHI	3978.19/3979.18	3.6/31
Het4G2	(GalB-KPLK(cFuc-KPL)FCxKI) ₂ KHI	3978.19/3978.19	7.3/63
Het5G2	(cFuc-KKLLK(GalA-KKL)FCxKI) ₂ KHI	4166.40/4166.40	6.6/54
Het6G2	(GalA-KKLLK(cFuc-KKL)FCxKI) ₂ KHI	4166.40/4166.40	7.1/59
Het7G2	(cFuc-KKLLK(GalB-KKL)FCxKI) ₂ KHI	4102.36/4102.36	11.1/69
Het8G2	(GalB-KKLLK(cFuc-KKL)FCxKI) ₂ KHI	4102.36/4102.36	10.3/57

^a One letter code for amino acids. Branching Lys in italics. GalB is carboxypropyl β -thiogalactoside. GalA is 4-carboxyphenyl β -galactoside. cFuc is α -L-fucoside. x = -CH₂CO-. The peptide C-terminus is carboxamide -CONH₂. See structural formula of **Het1G2** in Scheme 1 for correspondence between line notation and structure. ^b MS calc. for [M + H]⁺/MS obs. determined by ESI⁺. ^c Yields are reported for the RP-HPLC purified product after SPPS or chloroacetyl cysteine (ClAc) ligation. ^d From ref. 15. ^e From ref. 22.



titration calorimetry (ITC). Binding to the fucose specific lectin LecB ranged between $K_D = 430$ nM for the reference carbohydrate methyl- α -L-fucoside³⁹ and $K_D = 66$ nM for the tetravalent dendrimer **FD2** (Table 2). At the level of the monovalent ligands, the *c*-fucosylated tripeptide **FD0** was a poorer ligand than methyl- α -L-fucoside, while the monovalent **Het1G1-Cys** and **Het2G1-Cys** bound even less tightly and showed a too high n value, which might indicate aggregation *via* disulfide bond formation and/or intermolecular β -sheets as observed in the crystal structure (see below), possibly hindering lectin access for part of the glycosyl groups. Binding was not influenced by multivalency, as shown by the relative potency of binding per carbohydrate (r.p./ $n_{\text{Fuc}} < 2$ in all cases), or by additional positive charges in the peptide sequences (**Het7G2**, **Het8G2**). Nevertheless the tetravalent fucosides **FD2** and **FD2x** were the tightest LecB ligands in the series.

The range of binding affinities was much broader in the case of the galactose specific lectin LecA ranging from $K_D = 37$ μ M for the monovalent galactopeptide **GalBG0** down to $K_D = 2.5$ nM for the tetravalent galactoside **GalAG2** (Table 3). Binding was first influenced by the nature of the glycosidic group. All compounds carrying an aromatic aglycone (the reference nitrophenyl- β -galactoside NPG and all GalA containing products) showed a 10–20 fold stronger binding than their GalB analogs containing an aliphatic chain as aglycone, reflecting the previously documented favourable CH- π interaction between the aromatic group and the ϵ -CH of residue His50 on the lectin.³² As for LecB, the monovalent **Het1G1-Cys** and **Het2G1-Cys** bound poorly and gave high n values which might be caused by disulfide bond formation and/or aggregation. Furthermore LecA behaved as a typical lectin by showing enhanced binding to multivalent ligands.⁴⁰ For the aromatic galactosides, the relative potency of binding per galactoside reached r.p./ $n_{\text{Gal}} =$

20 in divalent galactosides **GalAG1**, **Het1G2** and **Het2G2**, and r.p./ $n_{\text{Gal}} = 300$ in tetravalent galactoside **GalAG2** showing the strongest binding. However **GalAG2x** only reached r.p./ $n_{\text{Gal}} = 32$, showing that the ClAc ligated peptide dendrimer backbone was less optimal for LecA binding compared to the original G2 peptide dendrimer. In the weaker binding aliphatic thio-galactosides of the GalB series divalent dendrimers had comparable relative potencies in the range $18 < \text{r.p./}n_{\text{Gal}} < 75$, including the more cationic heteroglycoclusters **Het7G2** and **Het8G2**. The tetravalent dendrimer **GalBG2** reached a relative potency of r.p./ $n_{\text{Gal}} = 230$ corresponding to $K_D = 40$ nM of comparable value to aromatic galactosides.

Overall the ITC data showed that G2 heteroglycoclusters achieved a relatively potent binding of both LecB and LecA at levels comparable to those of G1 glycopeptide dendrimers carrying only one type of carbohydrate. Binding affinities remained 2–10 fold weaker than that of the purely fucose or purely galactose containing dendrimers **FD2**, **FD2x**, **GalAG2**, **GalAG2x** and **GalBG2**, highlighting that the higher binding affinities of these G2 dendrimers required the tetravalent display of four carbohydrates of the same type.

Crystal structure of peptide dendrimer (Het1G1-Cys)₂ as LecB complex

A crystallography screen was undertaken to gain a structural insight into the glycopeptide dendrimer lectin interaction. G2 heteroglycoclusters **Het1G2** and **Het2G2** and the parent G1 dendrimers **Het1G1-Cys** and **Het2G1-Cys** were tested as complexes with either LecB or LecA by direct co-crystallization or by soaking preformed lectin crystals with ligand. A well diffracting crystal was obtained for the **Het2G1-Cys**·LecB complex, which provided a dataset at 2.13 Å resolution. In this structure

Table 2 Thermodynamic parameters and binding data for the interaction of the homo- and heteroglycopeptide dendrimers with LecB^a

Ligand	n_{Fuc}	n^b	ΔH [kcal mol ⁻¹]	$-T\Delta S$ [kcal mol ⁻¹]	ΔG [kcal mol ⁻¹]	K_D [nM]	r.p./ n_{Fuc}^c
FucOCH₃	1	0.77 ± 0.03	-9.87 ± 0.24	1.17 ± 0.24	-8.70 ± 0.02	430 ± 10	1.8
FD0	1	0.80 ± 0.01	-9.93 ± 0.33	0.99 ± 0.05	-8.34 ± 0.02	770 ± 20	1
Het1G1-Cys	1	1.64 ± 0.00	-4.82 ± 0.01	-2.79 ± 0.03	-7.61 ± 0.024	2630 ± 110	0.30
Het2G1-Cys	1	2.14 ± 0.03	-3.53 ± 0.024	-4.11 ± 0.02	-7.63 ± 0.09	2570 ± 400	0.30
FD1	2	0.28 ± 0.01	-23.87 ± 1.21	14.43 ± 1.10	-9.43 ± 0.10	123 ± 2	3.1
FD1-Cys	2	0.65 ± 0.01	-10.71 ± 0.01	1.81 ± 0.02	-8.89 ± 0.05	300 ± 25	1.3
Het1G2	2	0.40 ± 0.05	-15.9 ± 0.05	6.89 ± 0.30	-8.94 ± 0.01	276 ± 1.5	1.4
Het2G2	2	0.66 ± 0.05	-15.2 ± 0.02	6.28 ± 0.03	-8.92 ± 0.01	292 ± 57	1.3
Het3G2	2	0.31 ± 0.01	-25.15 ± 2.75	15.91 ± 2.82	-9.23 ± 0.09	179 ± 27	2.1
Het4G2	2	0.21 ± 0.01	-35.25 ± 0.02	25.79 ± 0.15	-9.43 ± 0.03	121 ± 5	3.2
Het7G2	2	0.30 ± 0.02	-25.1 ± 0.02	16.14 ± 0.24	-8.96 ± 0.02	327 ± 6	1.2
Het8G2	2	0.28 ± 0.02	-24.8 ± 0.06	15.61 ± 0.07	-9.17 ± 0.02	193 ± 4	2.0
FD2	4	0.16 ± 0.01	-40.9 ± 1.25	30.83 ± 1.7	-9.80 ± 0.09	66 ± 1	2.9
FD2x	4	0.59 ± 0.02	-9.72 ± 0.9	0.99 ± 0.9	-9.62 ± 0.02	88 ± 2	2.2

^a Thermodynamic parameters and dissociation constants K_D are reported as an average of two independent runs from ITC in 20 mM Tris, 100 mM NaCl, 100 μ M CaCl₂, pH = 7.5. Isothermal Titration Calorimetry (ITC) data are for LecB binding. Titration concentrations for Ligand/LecB are indicated in brackets: **Me- α -fucose** (0.44 mM/0.0558 mM), **FD0** (0.5 mM/0.044 mM), **Het1G1-Cys** (0.0457 mM/0.5 mM), **Het2G1-Cys** (0.0497 mM/0.95 mM), **FD1** (0.25 mM/0.0496 mM), **FD1-Cys** (0.0497 mM/0.5 mM), **Het1G2** (0.42 mM/0.042 mM), **Het2G2** (0.42 mM/0.042 mM), **Het3G2** (0.15 mM/0.042 mM), **Het4G2** (0.24 mM/0.042 mM), **Het7G2** (0.24 mM/0.042 mM), **Het8G2** (0.24 mM/0.042 mM), **FD2** (0.1 mM/0.028 mM), **FD2x** (0.052 mM/0.5 mM). ^b n defines the stoichiometry of the binding. ^c Relative potency per fucoside r.p./ $n_{\text{Fuc}} = K_D(\text{FD0})/K_D(\text{ligand})/n$, $n_{\text{Fuc}} =$ number of copies of fucosides in the glycopeptide dendrimers. See ESI Fig. S107 for ITC plots.



Table 3 Thermodynamic parameters and binding data for the interaction of the homo- and heteroglycopeptide dendrimers with LecA^a

Ligand	n_{Gal}	n^b	ΔH [kcal mol ⁻¹]	$-T\Delta S$ [kcal mol ⁻¹]	ΔG [kcal mol ⁻¹]	K_D [nM]	r.p./ n_{Gal}^d
NPG ^c	1	0.91 ± 0.03	-10.6 ± 0.45	4.0	6.5	16 000 ± 500	0.2
GalAG0	1	0.65 ± 0.02	-17.8 ± 0.3	10.2	-7.5	3000 ± 50	1
Het1G1-Cys	1	1.17 ± 0.04	-7.83 ± 0.26	0.85	-7.0	7800 ± 1200	0.4
Het2G1-Cys	1	1.37 ± 0.09	-7.36 ± 0.093	0.35	-7.0	7340 ± 300	0.4
GalAG1	2	0.30 ± 0.003	-29 ± 0.5	19.3	-9.7	83 ± 12	18
GalAG1-Cys	2	0.66 ± 0.06	-14.2 ± 0.05	4.9	-9.3	164 ± 0.2	9
Het1G2	2	0.44 ± 0.01	-23.0 ± 0.14	13.3	-9.7	75 ± 17	20
Het2G2	2	0.36 ± 0.03	-25.0 ± 0.15	15.2	-9.7	75 ± 18	20
GalAG2	4	0.14 ± 0.01	-68.6 ± 1.10	57	-11.7	2.5 ± 0.1	300
GalAG2x	4	0.21 ± 0.02	-32.4 ± 0.47	22	-10.4	23 ± 0.3	32
GalBG0	1	0.71 ± 0.01	-14 ± 0.04	7.8	-6.0	37 000 ± 800	1
GalBG1	2	0.37 ± 0.02	-20 ± 0.1	12	-8.2	1060 ± 160	18
GalBG1-Cys	2	0.53 ± 0.06	-13.9 ± 0.82	5.4	-8.5	600 ± 1	31
Het3G2	2	0.25 ± 0.02	-30.3 ± 0.40	21	-9.0	255 ± 14	73
Het4G2	2	0.15 ± 0.00	-42.0 ± 0.21	36	-9.4	120 ± 2	153
Het7G2	2	0.25 ± 0.02	-28.4 ± 0.69	20	-8.8	340 ± 70	55
Het8G2	2	0.25 ± 0.00	-25.5 ± 0.25	16.4	-9.0	250 ± 16	75
GalBG2	4	0.18 ± 0.02	-43 ± 1	33	-10.1	40 ± 1	230

^a Thermodynamic parameters and dissociation constants K_D are reported as an average of two independent runs from ITC in 20 mM Tris, 100 mM NaCl, 100 μ M CaCl₂, pH = 7.5. Isothermal Titration Calorimetry (ITC) data are for LecA binding. Titration concentrations for Ligand/LecA are indicated in brackets: **NPG** (3.0 mM/0.3 mM), **GalAG0** (0.5 mM/0.0516 mM), **Het1G1-Cys** (1.2 mM/0.051 mM), **Het2G1-Cys** (0.57 mM/0.05 mM), **GalAG1** (0.25 mM/0.0486 mM), **GalAG1-Cys** (0.255 mM/0.05 mM), **Het1G2** (0.25 mM/0.05 mM), **Het2G2** (0.25 mM/0.05 mM), **GalAG2** (0.03 mM/0.018 mM), **GalAG2x** (0.0909 mM/0.0166 mM), **GalBG0** (1.0 mM/0.091 mM), **GalBG1** (0.25 mM/0.049 mM), **GalBG1-Cys** (0.4 mM/0.0526 mM), **Het3G2** (0.5 mM/0.1 mM), **Het4G2** (0.1 mM/0.05 mM), **Het7G2** (0.175 mM/0.05 mM), **Het8G2** (0.175 mM/0.05 mM), **GalBG2** (0.03 mM/0.08 mM). ^b n defined the stoichiometry of the binding. ^c NPG is 4-nitrophenyl- β -D-galactopyranoside (reference used in the ITC measurements with LecA), (3 mM/0.3 mM, Ligand/LecA). ^d Relative potency per galactoside. r.p./ $n_{\text{Gal}} = K_D(\text{GalAG0 or GalBG0})/K_D(\text{ligand})/n_{\text{Gal}}$. n_{Gal} = number of copies of galactosides in the glycopeptide dendrimer. See ESI Fig. S108 for ITC plots.

each of the four carbohydrate binding sites on each LecB tetramer is occupied by a fucose ligand. Two binding sites are occupied by fucosyl groups attached to a fully visible **Het2G1-Cys** ligand, whereby this ligand is oxidatively dimerized to a disulfide bridged (**Het2G1-Cys**)₂ peptide dendrimer (Fig. 1A, grey stick model and electron density). The other two LecB binding sites are occupied by fucosyl groups where only the terminal fucosylated tripeptide dendrimer arm is visible (Fig. 1A, cyan stick model). This tripeptide adopts a stretched conformation very similar to that of the fully resolved ligand, and distinct from the more folded conformation previously observed in a LecB complex with the fucosylated tripeptide **FD0** (Fig. 1A: red stick model),¹⁹ consistent with attachment to the dendrimer. **Het2G1-Cys** is possibly disordered or not oxidatively dimerized and flexible and therefore not visible at this crystal position. All fucosyl groups adopt the typical position seen in other fucose·LecB complexes, with binding interactions between the C₂(OH), C₃(OH) and C₄(OH) group of fucose and the pair of Ca²⁺ ions and Asp99 on LecB. A tight (2.3 Å) hydrogen bond between the backbone amide of Asn70 and the phenylalanyl-cysteine amide bond in the fully resolved ligand represents the only visible additional protein dendrimer contact.

The electron density of (**Het2G1-Cys**)₂ observed at the fully resolved crystal position is remarkable since it features the first example of a complete crystal structure of a G2 peptide dendrimer. A closer analysis shows that this dendrimer bridges binding sites on two different LecB tetramers. Furthermore, each (**Het2G1-Cys**)₂ interacts with two further symmetrically

positioned dendrimers *via* antiparallel β -sheets involving the C-terminal tetrapeptides Leu-Lys-Phe-Cys (Fig. 1B). The arrangement is repeated around the C₃ symmetry axis such that three dendrimers form a C₃-symmetrical [(**Het2G1-Cys**)₂]₃ supramolecular macrocycle with three fucosyl groups pointing above the ring and three fucosyl groups below the ring and connecting six different LecB tetramers (Fig. 1C-F). The six galactosyl groups are located within the supramolecular macrocycle and are somewhat hidden from solution. Although this supramolecular macrocycle is probably specific of the crystal lattice, one can imagine that similar intermolecular β -sheet interactions might occur in solution and cause steric hindrance around single carbohydrate groups, which could explain the unusually high stoichiometry of binding observed in ITC for some of the dendrimers. The absence of chelate binding mode with any of the LecB tetramers in the present structure might be a general feature of dendrimer LecB interactions, as suggested by the absence of significant multivalency effect with all glycopeptide dendrimers including the tightest binder **FD2**.

Biofilm inhibition by glycodendrimers and antimicrobial compounds

Biofilm inhibition activities were investigated next. In view of a possible direct antimicrobial effect of **Het5G2-Het8G2** incorporating additional cationic residues in their sequences, control non-glycosylated dendrimers were prepared to test the effect of the peptide backbone alone. These were the N-



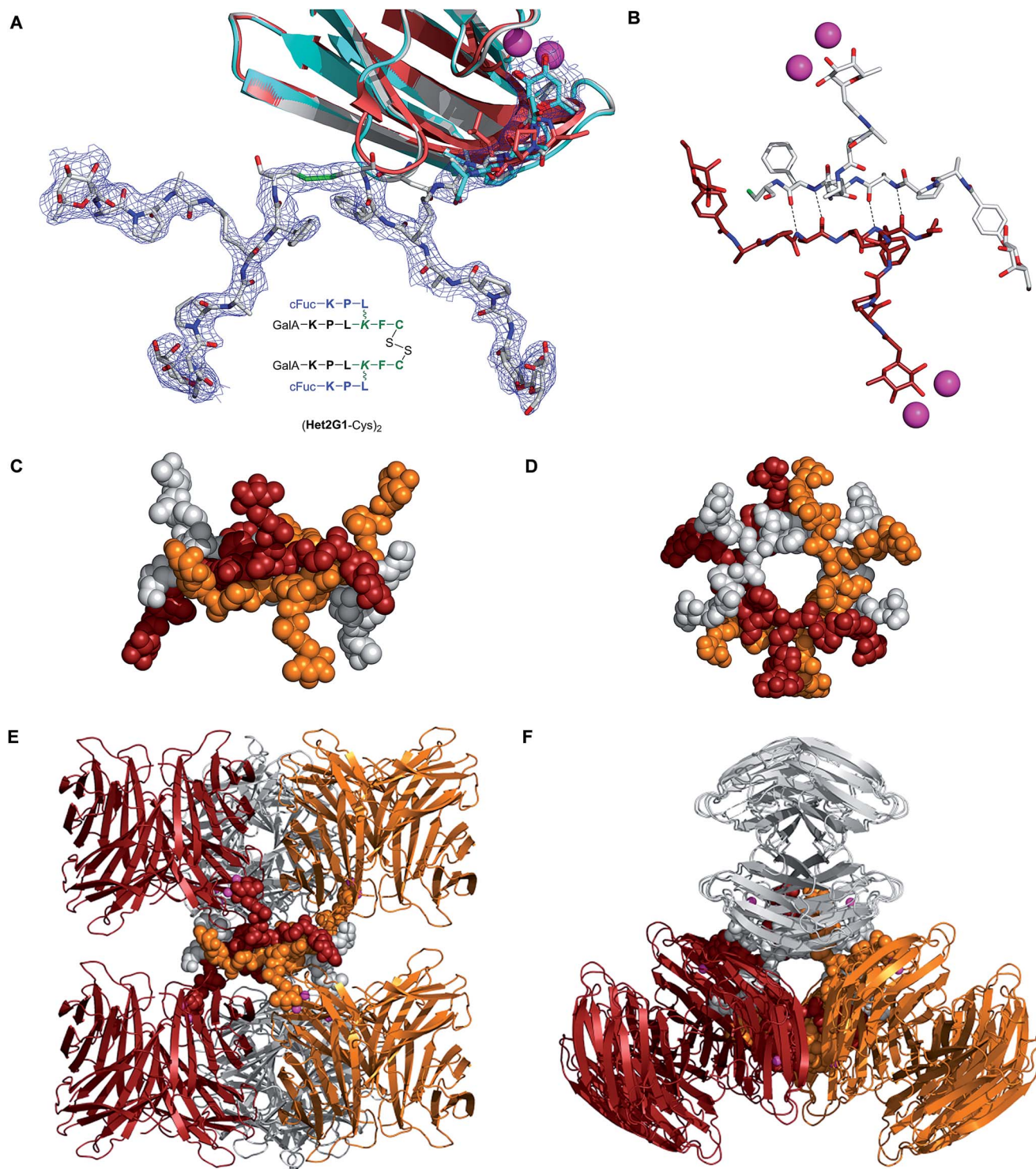


Fig. 1 Crystal structure of dendrimer·LecB complex (PDB 5D2A, see also ESI† for crystal data table). **A**. Overlay view of a single LecB binding site bound with (i) grey model and electron density: fully resolved (Het2G1-Cys)₂ dendrimer; (ii) cyan model: terminal tripeptide at the partially resolved crystal position; (iii) red model: tripeptide FDO, from ref. 19 in each case the peptide backbone is fully visible, but several side chains are disordered and not visible, which is typical for peptides and loop residues in proteins **B**. Detail of the β -sheet interaction connecting two different (Het2G1-Cys)₂ dimers. Dashed lines indicate H-bonded atoms that are 2.7 Å apart. **C–F**. Side and top view of the supramolecular macrocycle [(Het2G1-Cys)₂]₃ binding to six LecB tetramers. Each of the three dendrimers and the two LecB tetramers to which it is bound have the same color: orange, red, or grey.



terminally acetylated peptide dendrimer **AcG2** and its ClAc ligation analog **AcG2x**, as well as the more cationic sequence **AcG2xK** and peptide dendrimer **NG2** with four free N-termini, each carrying four additional positive charges (Table 4). Two recently reported polycationic antimicrobial peptide dendrimers with strong activity against *P. aeruginosa* presumably by a membrane disrupting effect, **G3KL** and **G3LK**,³⁶ were also included in the study, as well as the reference antibiotics polymyxin B and tobramycin (Table 5).

Table 4 Non-glycosylated dendrimers^a

Compound	Sequence	MS calc./obs. ^b	Yield ^c (mg/%)
AcG2	(Ac-KPL) ₄ (KFKI) ₂ KHI	2949.92/2949.92	26/7
AcG1-Cys	(Ac-KPL) ₂ KFC	1156.69/1156.69	38/19
AcG1K-Cys	(Ac-KKL) ₂ KFC	1217.78/1218.78	76/50
AcG2x	(Ac-KPL) ₄ (KFCxKI) ₂ KHI	3269.99/3270.99	4.8/50
AcG2xK	(Ac-KKL) ₄ (KFCxKI) ₂ KHI	3392.43/3393.18	5.9/60
NG2	(KPL) ₄ (KFKI) ₂ KHI	2781.88/2781.88	8/44
G3KL ^d	(KL)(KKL) ₄ (KKL) ₂ KKL	4534.19/4535.10	52.2/8
G3LK ^d	(LK)(KLK) ₄ (KLK) ₂ KLK	4534.19/4534.84	7.8/1

^a One letter code for L-amino acids. *Lys* = branching lysine residue, *x* = -S-CH₂-CO-. Ac = acetyl. C-termini are carboxamides CONH₂. ^b MS calc. for [M + H]⁺/MS obs. determined by ESI⁺. ^c Yields are reported for the RP-HPLC purified products. ^d From ref. 36.

Biofilm inhibition was measured by determining the amount of biofilm formed after 24 h of growth with or without compound at different concentrations. The biofilm was quantified using the WST-8 assay indicating the amount of viable cells as previously described.³¹ To distinguish between biofilm inhibition and a direct bactericidal effect, the number of viable cells in the supernatant was also determined by CFU (colony forming unit) counts and by a growth test in an inoculated full medium (OD₆₀₀ measurement after 7–8 h growth). In a third assay, the ability of the compounds to disperse already established biofilms was tested by preincubating the bacteria under biofilm forming conditions for 24 h, followed by addition of the test compound at a fixed concentration, incubation for an additional 24 h, and quantification of the remaining biofilm as above. Finally, biofilm inhibitory compounds were also tested for their direct bactericidal activity under full medium either by determination of the minimal inhibitory concentration (MIC) by standard serial dilution series, or with a measurement of their effect on bacterial growth in full medium.

All dendrimers except the GalB containing heteroglycoclusters **Het3G2** and **Het4G2** showed substantial biofilm inhibition activity at a level comparable to the reference biofilm inhibitor **FD2** (MBIC = 20 μM). Most of these dendrimers were also able to disperse already established biofilms, the most potent compound still being the parent fucosylated dendrimer **FD2**. Except for **GalAG2x** all dendrimers derived from the parent

Table 5 Inhibition of *Pseudomonas aeruginosa* biofilms

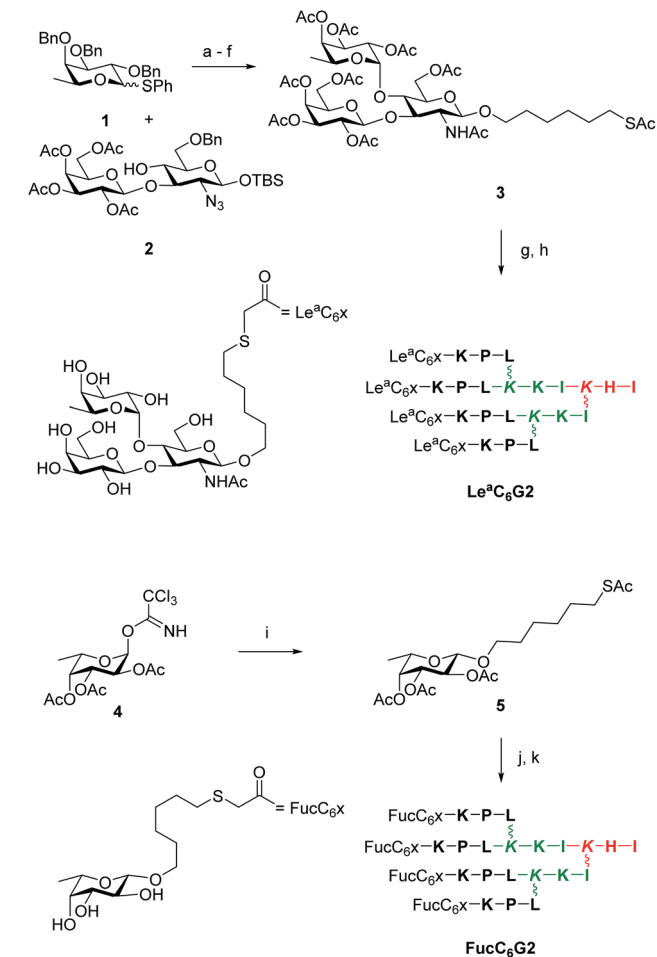
Compound	MBIC ^a [μM]	MBC ^b [μM]	Dispersal ^c	MIC ^d
FD2	20	>30	100% dispersal	n.i. ^e
GalAG2	20	>30	50% dispersal	n.i.
GalBG2	20	nd	60% dispersal	nd
FD2x	45	>45	~35% dispersal	nd
GalAG2x	30	30	100% dispersal	nd
GalBG2x	30	>45	Increase	nd
Het1G2	30	>45	~30% dispersal	nd
Het2G2	30	>45	~35% dispersal	nd
Het3G2	>45	>45	~85% dispersal	nd
Het4G2	>45	>45	No dispersal	nd
(Het2G1-Cys)₂	>45	nd	No dispersal	nd
Het5G2	20	20	100% dispersal	>64 μM
Het6G2	20	20	100% dispersal	>64 μM
Het7G2	13	13	100% dispersal	>64 μM
Het8G2	20	20	100% dispersal	>64 μM
AcG2	30	30	100% dispersal	n.i.
AcG2x	30	30	Increase	nd
AcG2xK	13	13	100% dispersal	>64 μM
NG2	2.5	2.5	100% dispersal @ 22 μM	>64 μM
G3KL	4	6	95% dispersal @ 10 μM	0.44 μM
G3LK	6	9	95% dispersal @ 10 μM	2 μM
Polymyxin B	6	6	~70% dispersal @ 10 μM	2 μg mL ^{-1f}
Tobramycin	0.5	0.5	100% dispersal @ 4 μM	1 μg mL ^{-1g}

^a MBIC is minimal biofilm inhibitory concentration. ^b MBC = minimal bactericidal concentration under biofilm forming conditions, determined by CFU counts in the biofilm supernatant and growth test in full medium. ^c Effect of the compounds at 50 μM or concentration stated on dispersal of already established biofilms. ^d MIC = minimal inhibitory concentration in a standard 2-fold dilution assay for antibacterial activity in full LB medium. n.i. = no inhibition in a growth assay using 50 μM compound, nd = not determined. ^e From ref. 19. ^f From ref. 36. ^g From ref. 41. See also ESI Fig. S110–S114 (biofilm inhibition), S116 (biofilm dispersal), S117–119 (MIC data), Table S1 (cell viability data) and Table S2 (control growth curves).



peptide sequence of **FD2** were not bactericidal at their MBIC concentration, showing that their biofilm inhibition activity was not caused by a simple bactericidal effect. The related disulfide-bridged heterodimer (**Het2G1-Cys**)₂ described above as a LecB complex was not active as a biofilm inhibitor.

Polycationic heteroglycoclusters **Het5G2–Het8G2** also showed good activities, however these compounds were toxic at their biofilm inhibitory concentration as evidenced by a comparable minimal bactericidal concentration (MBC) value determined by CFU counts of the supernatant and growth test. The control, non-glycosylated dendrimers **AcG2**, **AcG2x** and their analogs **AcG2xK** and **NG2** with an additional four positive charges were also bactericidal at their biofilm inhibition concentration. A similar bactericidal antibiofilm effect was detected with the polycationic antimicrobial peptide dendrimers **G3KL** and **G3LK** as well as with antibiotics polymyxin B and tobramycin. All of these bactericidal compounds were furthermore able to disperse already established biofilms, although the effect was not as complete as observed with glycodendrimer **FD2**, and at least 10-fold higher concentrations were necessary compared to their MIC values determined in full medium, where bacteria grow in suspension (planktonic state). The biofilm inhibition study highlighted the remarkable non-toxic biofilm inhibition and dispersal effect of glycopeptide dendrimers such as **FD2** in contrast to the direct bactericidal effect of polycationic dendrimers and antibiotics.



Scheme 2 Synthesis of $\text{Le}^a\text{C}_6\text{G2}$ and $\text{FucC}_6\text{G2}$. Conditions: (a) *N*-Iodosuccinimide, AgOTf, CH_2Cl_2 , 0 °C, 30 min (quant.); (b) H_2 , Pd(OH)₂, THF, AcOH, 17 h, 25 °C; (c) pyridine/ Ac_2O , cat. DMAP/NMM, CH_2Cl_2 , 25 °C, 4 h (55% over 2 steps); (d) TBAF, AcOH, 25 °C, 17 h (60%); (e) Cl_3CCN , DBU, CH_2Cl_2 , 0 °C, 40 min (80%); (f) $\text{AcS}(\text{CH}_2)_6\text{OH}$, TMSOTf, CH_2Cl_2 , 0 °C, 20 min (64%); (g) K_2CO_3 , MeOH, 25 °C, 1 h (75%); (h) **ClAcD2**, KI, DIPEA, $\text{H}_2\text{O}/\text{CH}_3\text{CN}$, 25 °C, 17 h, then prep. HPLC (63%); (i) step f (84%); (j) step g (75%); (k) step h (31%). See ESI† for synthesis details.

Multivalent Lewis^a dendrimers

The above study showed that tetra-fucosylated dendrimer **FD2** remained the most potent non-toxic biofilm inhibitor among the different glycopeptide dendrimers tested. A second variation study was therefore undertaken focusing on variation of the fucosyl group of this dendrimer. Analogs of **FD2** were prepared by the ClAc ligation approach using carbohydrate building blocks appended with a nucleophilic hexylthiol tether. The ligation was performed onto a tetrachloroacetylated dendrimer **ClAcD2** derived from **FD2** in which the phenylalanine residue in the G1 branch had been removed to compensate for the introduction of the alkylthiol tether in terms of hydrophobicity and overall arm length of the dendrimers. This approach did not require excess of non-peptidic reagents and therefore allowed to consider more complex carbohydrates. We focused on preparing a tetravalent peptide dendrimer version of the Lewis^a trisaccharide, which is the tightest monovalent binder known for LecB and its probable natural ligand.³⁷

To obtain the desired Le^a trisaccharide, L-fucosyl donor **1** and the selectively protected disaccharide **2** with a single free hydroxyl group at position C(4) of its glucosamine component were synthesized by published procedures,^{42,43} and coupled by glycosylation to give a protected Lewis^a trisaccharide. Removal

Table 6 Synthesis of **FD2** analogs by ClAc ligation^a

Compound	Sequence	MS calc./obs. ^a	Yield (mg/%)
ClAcD2	(ClCH ₂ CO–KPL) ₄ (KKI) ₂ KHI	2791.64/2791.64	43/17 ^b
FucC₆G2	(Fuc–O–(CH ₂) ₆ –S–CH ₂ CO–KPL) ₄ (KKI) ₂ KHI	3768.27/3768.27	2.5/31 ^c
Le^aC₆G2	(Le ^a –O–(CH ₂) ₆ –S–CH ₂ CO–KPL) ₄ (KKI) ₂ KHI	5228.80/5228.80	8/63 ^c

^a One letter code for amino acids. Branching Lys in italics. The peptide C-terminus is carboxamide CONH₂. ^bMS calc. for [M + H]⁺/MS obs. determined by ESI⁺. Yield is reported for the RP-HPLC purified product after ^bSPPS or ^cClAc ligation.



Table 7 Thermodynamic parameters and binding data for the interaction of FucC6 and Le^a glycopeptide dendrimers with LecB^a

Ligand	n_{Fuc}	n^b	ΔH [kcal mol ⁻¹]	$-T\Delta S$ [kcal mol ⁻¹]	ΔG [kcal mol ⁻¹]	K_D [nM]	r.p./ n_{sugar}
FucOCH ₃	1	0.77 ± 0.03	-9.87 ± 0.24	1.17 ± 0.24	-8.70 ± 0.02	430 ± 10	1
Le ^a	1	1.08 ± 0.01	-8.35 ± 0.02	-0.74 ± 0.002	-9.10 ± 0.02	213 ± 2.0	2
FucC ₆ G2	4	0.22 ± 0.01	-14.7 ± 0.38	5.20 ± 0.28	-9.44 ± 0.09	121 ± 20	0.9 ^c
Le ^a C ₆ G2	4	0.12 ± 0.01	-58.5 ± 1.17	48.5 ± 1.65	-10.1 ± 0.06	39 ± 0.8	2.8 ^c

^a Thermodynamic parameters and dissociation constants K_D are reported as an average of two independent runs from ITC in 20 mM Tris, 100 mM NaCl, 100 μM CaCl₂, pH = 7.5. Isothermal Titration Calorimetry (ITC) data are for LecB binding. Titration concentrations for Ligand/LecB are indicated in brackets: Le^a (0.324 mM/0.0327), FucC₆G2 (0.3 mM/0.08 mM), Le^aC₆G2 (0.07 mM/0.042 mM). ^b n defined the stoichiometry of the binding. ^c Relative potency per fucoside. r.p./ $n_{\text{Fuc}} = K_D(\text{FucOCH}_3)/K_D(\text{ligand})/n_{\text{Fuc}}$. See ESI Fig. S107 for ITC plots.

Table 8 Activity of FucC6 and Le^a glycopeptide dendrimers against *Pseudomonas aeruginosa* biofilms

Ligand	MBIC ^a [μM]	MBC ^b [μM]	Dispersal	MIC ^c
FD2	20	>30	100% dispersal at 50 μM	n.i. ^d
FucC ₆ G2	9	>20	100% dispersal at 30 μM	>64 μM
Le ^a xG2	30	nd	~88% dispersal at 50 μM	nd

^a MBIC is minimal biofilm inhibitory concentration. ^b MBC = minimal bactericidal concentration under biofilm forming conditions, determined by CFU counts in the biofilm supernatant. ^c MIC = minimal inhibitory concentration in a standard 2-fold dilution assay for antibacterial activity in full LB medium. n.i. = no inhibition in a growth assay using 50 μM compound, nd = not determined. ^d From ref. 19. See also ESI Fig. S115/S116 (biofilm data), S120 (MIC data), and Table S1 (cell viability data).

of benzyl ethers and azide reduction by catalytic hydrogenation, acetylation, desilylation of the anomeric hydroxyl group, formation of the trichloroacetimidate and Schmidt glycosylation of 6-hydroxyhexyl-thioacetate then gave the protected thiol tethered Lewis^a trisaccharide 3. Finally, deacetylation and ligation with dendrimer ClAcD2 gave the desired G2 dendrimer Le^aC₆G2. The tethered protected β-fucoside 5 was similarly obtained from intermediate 4.⁴³ Deacetylation and ligation with ClAcD2 as above gave G2 dendrimer FucC₆G2 (Scheme 2, Table 6).

The tetravalent dendrimers Le^aC₆G2 and FucC₆G2 showed the expected binding affinities to LecB as measured by ITC, with simply additive but no multivalency effect on binding compared to their parent monovalent ligand. Nevertheless the slightly tighter LecB binding of Lewis^a compared to simple fucosyl groups and a slightly more than additive effect on LecB binding in the dendrimer resulted in $K_D = 39$ nM for Le^aC₆G2, such that this first multivalent version of the Lewis^a antigen represents one of the tightest LecB ligands reported thus far (Table 7).

Both compounds showed biofilm inhibition and dispersal activities in the same range as the parent FD2, with a slightly stronger biofilm inhibition in the case of FucC₆G2 with MBIC = 9 μM. In both cases the compounds were not bactericidal, confirming the absence of toxicity for this type of glycosylated peptide dendrimer biofilm inhibitors. Considering its relatively straightforward and efficient synthesis, FucC₆G2 stood out as an improved version of biofilm inhibitor FD2 (Table 8).

Synergistic effects between glycopeptide dendrimer biofilm inhibitors and tobramycin

The MBIC value of 9 μM reached with FucC₆G2, although the best value among non-toxic glycopeptide dendrimers, still

represented an activity of only 34 μg mL⁻¹ due to the high molecular weight of the compound (MW = 3.7 kDa), implying that further improvement were necessary to reach a practically useful compound. To search for further activity improvements, possible synergistic effects between our glycopeptide dendrimers and antimicrobial compounds were investigated. Synergistic effects between different antibiotics are well known, and combination therapies with multiple antibiotics are commonly used in the clinic.⁴⁴⁻⁴⁷

Here we focused on synergistic effect with tobramycin, a classical aminoglycoside antibiotic targeting the ribosome, against which *P. aeruginosa* readily acquires resistance. Indeed synergistic effects have been previously reported between tobramycin and iron chelators, which act as *P. aeruginosa* biofilm inhibitors in airway epithelial cystic fibrosis cells.^{48,49} A survey with various glycopeptide dendrimers showed that co-application of FD2 and tobramycin effected a synergistic biofilm inhibition effect, with both compounds acting at sub-inhibitory concentration (Fig. 2A), allowing to use only 5 μM of FD2, which is 4-fold lower than its MBIC value, and 0.1 μM tobramycin, which is 5-fold lower than its MBIC value, thus achieving a remarkable synergistic effect (Table 9). A similar co-application of subinhibitory concentrations of tobramycin with FucC₆G2 also acted as toxic biofilm inhibitors, however without decreasing the MBIC of FucC₆G2 itself (Fig. 2E). In terms of biofilm dispersal application of subinhibitory concentrations of tobramycin also lowered the dispersal concentration of FD2 by 4-fold (Fig. 2B) but not that of FucC₆G2 (Fig. 2F). The effects can be interpreted in terms of triggering a non-toxic dispersal of the biofilm by the glycopeptide dendrimers, allowing facilitated entry of the toxic tobramycin into the cells.



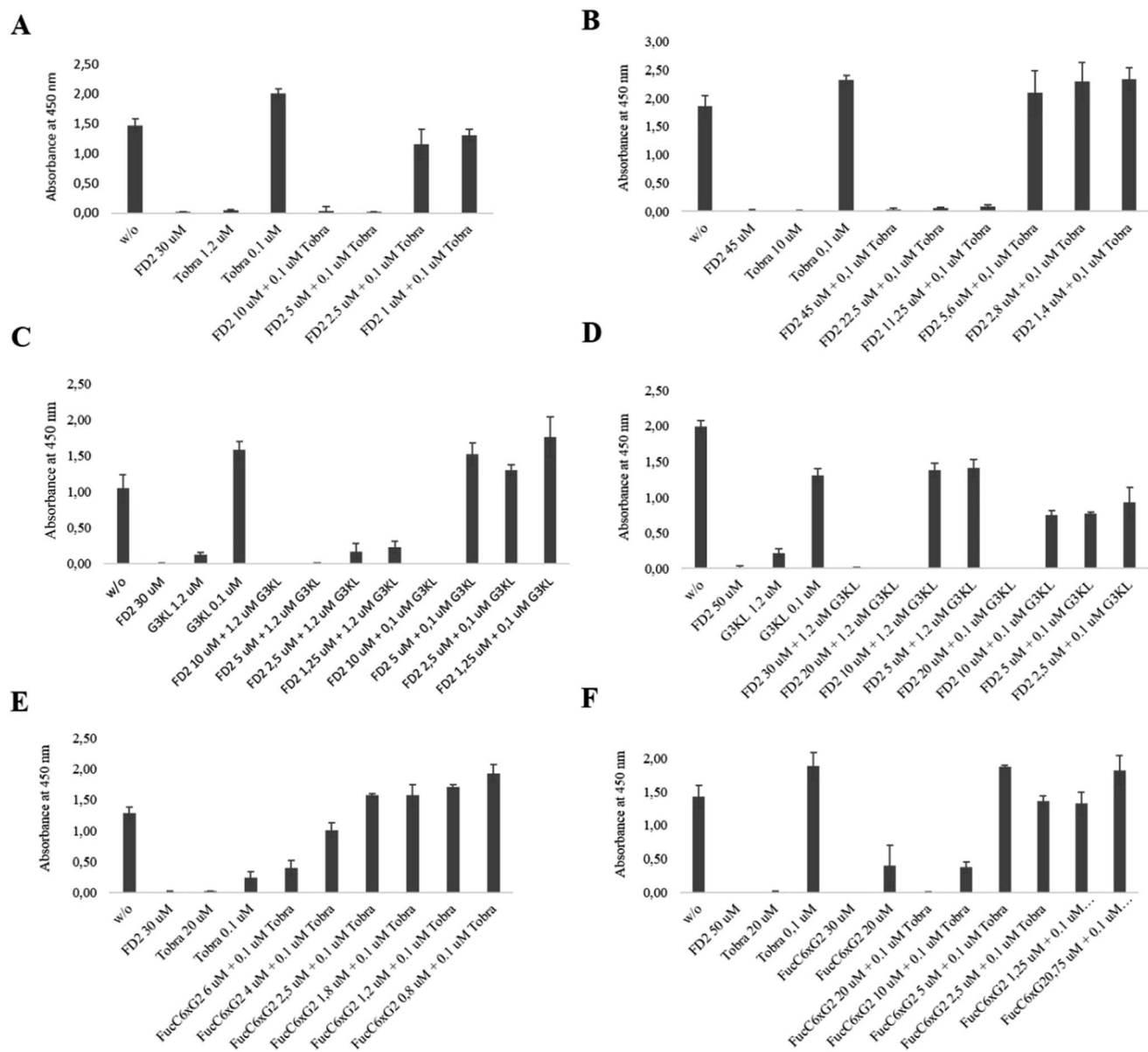


Fig. 2 Inhibition and dispersal of *Pseudomonas aeruginosa* strain PA01 biofilms by synergy. (A) Biofilm inhibition with FD2/Tobramycin. (B) Biofilm dispersal with FD2/Tobramycin. (C) Biofilm inhibition with FD2/G3KL. (D) Biofilm dispersal with FD2/G3KL. (E) Biofilm inhibition with FucC₆G2/Tobramycin. (F) Biofilm dispersal with FucC₆G2/Tobramycin. For inhibition, biofilms were grown on microtiter plates for 24 hours at 37 °C in the presence of the indicated compounds, and viable biofilms were stained with WST-8/PE. For dispersal, biofilms were first grown on microtiter plates for 24 hours at 37 °C in the absence of any compounds, planktonic bacteria were removed, and the biofilms were incubated with compounds for another 24 hours. Viable biofilms were stained with WST-8/PES. See also Table 9.

A similar synergistic effect was obtained by combining FD2 with the antimicrobial peptide dendrimer G3KL lowering of the MBIC value of each compound by 4 to 5-fold. Most strikingly, even a 0.1 μM application of G3KL induced cytotoxic biofilm inhibition at only 10 μM FD2 (Fig. 2C), and dispersal at only 20 μM (Fig. 2D). As for tobramycin, the biofilm disrupting effect of FD2 probably facilitates access of G3KL to the bacteria. Overall these experiments suggest that the non-toxic biofilm inhibitor FD2 might be useful to increase the sensitivity of *P. aeruginosa* to standard antibiotics.

Table 9 Biofilm inhibition and dispersal by co-application of tobramycin or G3KL with FD2

Compound	Antimicrobial compound	MBIC [μM]	Dispersal ^a [μM]
FD2	—	20	50
FD2	Tobramycin 0.1 μM	5	10
FucC ₆ G2	—	9	30
FucC ₆ G2	Tobramycin 0.1 μM	9	20
FD2	G3KL 1.2 μM	5	20
FD2	G3KL 0.1 μM	10	20

^a Concentration necessary to disperse 100% of the biofilm, see also legend of Fig. 2.



Conclusion

In summary an extensive search for improved analogs of our previously reported glycopeptide dendrimer inhibitors of *P. aeruginosa* biofilms **FD2** and **GalAG2** was carried out exploiting the multivalent chloroacetyl cysteine thioether (ClAc) ligation. G2 heteroglycoclusters displaying a pair of fucosyl groups targeting the fucose specific *P. aeruginosa* lectin LecB and a pair of galactosyl groups targeting the galactose specific lectin LecA achieved binding to both lectins at the level of G1 peptide dendrimers. One of the heteroglycoclusters provided the first fully resolved crystal structure of a peptide dendrimer as a LecB complex, revealing a supramolecular arrangement of three dendrimers connected *via* β -sheets and cross-linking six different LecB tetramers without chelate binding, consistent with the absence of multivalency effects by multivalent ligands of this lectin. Heteroglycoclusters **Het1G2–Het4G2** inhibited *P. aeruginosa* biofilms at levels comparable to the parent glyco-dendrimers **FD2** and **GalAG2**. Analogs **Het5G2–Het8G2** with additional positive charges showed stronger biofilm inhibition and dispersal, however the gain in activity reflected a direct bactericidal effect of the dendrimers comparable to other polycationic dendrimers such as the antimicrobial peptide dendrimer **G3KL** and the control, non-glycosylated dendrimer backbone **NG2**. In a second approach G2 glycopeptide dendrimers displaying four copies of the Lewis^a antigen, which is the natural LecB ligand, yielded slightly stronger LecB ligands and biofilm inhibitors. Finally a search for synergistic effects between non-toxic glycopeptide dendrimer biofilm inhibitors and classical antibiotics revealed that biofilm inhibition and dispersal could be achieved by co-application of dendrimer **FD2** and the antibiotic tobramycin at sub-inhibitory concentrations of both compounds. A similar synergistic effect was observed by combining **FD2** with the antimicrobial peptide dendrimer **G3KL**, allowing to inhibit biofilm with both compound significantly below their MBIC value. The synergistic use of non-toxic glycopeptide dendrimer biofilm inhibitors to restore sensitivity to antibiotics might offer new opportunities to address antibiotic resistance in *P. aeruginosa*.

Materials and methods

Synthesis

Amino acids were used as the following derivatives: Fmoc-His(Trt)-OH, Fmoc-Leu-OH, Fmoc-Phe-OH, Fmoc-Ile-OH, Fmoc-Pro-OH, Fmoc-Lys(Boc)-OH, Fmoc-Lys(Fmoc)-OH, Fmoc-Cys(Trt)-OH, Fmoc-Ser(tBu)-OH. Chemicals were used as supplied and solvents were of analytical grade. Analytical RP-UHPLC was performed in Dionex ULTIMATE 3000 RS chromatography system (ULTIMATE 3000 RS photo diode array detector) using a Dionex Acclaim® RSLC 120 C18, 3.0 × 50 mm, particle size 2.2 μ m, 120 Å pore size, flow rate 1.2 mL min⁻¹ column. Eluent A contained water with 0.1% TFA; eluent D contained acetonitrile and water (90 : 10) with 0.1% TFA. Compounds were detected by UV absorption at 214 nm. Preparative RP-HPLC was performed with a Waters PrepLC4000 chromatography system using a Dr Maish GmbH Reprospher

Column (C18-DE, 5 μ m, 100 × 30 mm, pore size 100 Å, flow rate of 40 mL min⁻¹); compounds were detected by UV absorption at 214 nm using a Waters 486 Tunable Absorbance detector or with Waters automatic PrepLC with the four following modules: Waters2489 UV/Vis detector, Waters2545 pump, Waters Fraction Collector III and Waters 2707 Autosampler. For the automatic system, data recording and processing was performed with Waters ChromScope version 1.40 from Waters Corporation. All RP-HPLC were using HPLC-grade acetonitrile and Milli-Q deionized water. The elution solutions were: (A) MilliQ deionized water containing 0.1% TFA; (B) MilliQ deionized water/acetonitrile (50 : 50, v/v); (D) MilliQ deionized water/acetonitrile (10 : 90, v/v) containing 0.1% TFA. MS spectra were provided by the Service of Mass Spectrometry of the Department of Chemistry and Biochemistry, University of Bern. See ESI† for images of analytical HPLC and MS spectra of all compounds.

Solid-phase peptide synthesis

TentaGel S RAM resin (500 mg, loading: 0.22–0.26 mmol g⁻¹) was loaded in a 10 mL polypropylene syringe fitted with a polypropylene frit, a Teflon stopcock, and a stopper. The resin was swollen in DCM (6 mL, 20 min). After removal of the DCM, the Fmoc-protecting group of the resin was removed using a solution of 20% piperidine in NMP. Stirring of the reaction mixture was performed by attaching the closed syringes to a rotating axis. The completion of the reaction and the removal of the Fmoc protecting group were checked using the TNBS test (1 drop of 1% TNBS in DMF and 1 drop 10% DIPEA in DMF). Orange or red beads indicated successful deprotection, colorless beads indicated a complete coupling reaction. In case of proline, the chloranil-test (1 drop of 2% chloranil in DMF and 1 drop 2% acetaldehyde in DMF) for secondary amines was used. In this case, violet or dark green beads indicated successful deprotection as colorless beads indicated complete coupling reaction. Removal of the Fmoc protecting group was performed by using a solution of 20% piperidine in NMP (6 mL, 20 min). After filtration, the resin was washed with NMP (2 × 6 mL), MeOH (2 × 6 mL) and DCM (2 × 6 mL).

Coupling of the Fmoc-protected amino acids was performed using Fmoc-protected amino acids (3 eq.), PyBOP (3 eq.) and DIPEA (6 eq.) in NMP (6 mL). The reaction times for linear peptides were 1 hour, for G1 dendrimers 1.5 hours and for G2 dendrimers 2 hours. The resin was then washed with NMP (2 × 6 mL), MeOH (2 × 6 mL) and DCM (2 × 6 mL). Capping of free amine groups was performed after each amino acid coupling using a solution of acetic anhydride : dichloromethane (1 : 1, v/v).

For allyl/alloc deprotection, the polypropylene syringe was equipped with a septum and dried under high vacuum for one hour. Then it was swollen in dry DCM for 10 min under argon. After removal of the solvent, Pd(PPh₃)₄ (0.15 eq.) and PhSiH₃ (15 eq.) were diluted in 4 mL of dry DCM and added to the resin under argon. The reaction was stirred under argon bubbling for 20 min. The reagents were then removed by filtration and the resin washed with dry DCM. The alloc deprotection procedure was repeated for another 20 min. Finally, the resin was washed



with sodium diethylcarbamate solution (0.02 M in DMF) and then with DCM (8 mL, 2×10 min).

N-Chloroacetylation of the resin was performed with a solution of chloroacetic acid anhydride (10 eq. per free N-terminus) in NMP (5 mL) for 2×10 min. The resin was then washed with NMP (2×6 mL), MeOH (2×6 mL) and DCM (2×6 mL).

Carbohydrate coupling was performed by using the protected peracetylated sugar derivative (4 eq. per N-terminus for GalA and GalB, 5 eq. per N-terminus for c-fucoside), HATU (3 eq. per arm) and DIPEA (5 eq. per arm) in NMP (6 mL). The resin was stirred overnight. The resin was washed with NMP (2×6 mL), MeOH (2×6 mL) and DCM (2×6 mL). The deprotection of the carbohydrates was performed by basic methanolysis with a solution of MeOH/25% $\text{NH}_3/\text{H}_2\text{O}$ (8 : 1 : 1, v/v/v). The resin was washed with NMP (2×6 mL), MeOH (2×6 mL) and DCM (2×6 mL) and dried.

Acetylation of the N-termini was performed by using acetic anhydride : dichloromethane (1 : 1, v/v) for 20 min. The resin was washed with NMP (2×6 mL), MeOH (2×6 mL) and DCM (2×6 mL).

TFA cleavage was performed by adding a solution of TFA/TIS/ H_2O (95 : 2.5 : 2.5, v/v/v) to the resin for 2.5–3 hours. Peptide dendrimers containing Cys residues were cleaved with TFA/TIS/ H_2O /1,2-ethanedithiol (92.5 : 2.5 : 2.5 : 2.5, v/v/v/v). The peptide dendrimers were then precipitated with ice-cold TBME and removed from TBME by centrifugation. The crude product was dried at high-vacuum to remove residual TFA and TBME.

For purification, the crude was dissolved in MilliQ deionized water containing 0.1% TFA or in a mixture of MilliQ deionized water and acetonitrile containing 0.1% TFA. The purification was performed by HPLC. The obtained products were analysed by analytical RP-UHPLC, ESI-MS and amino acid analysis.

FD1-Cys. (cFuc-KPL)₂KFC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (33.5 mg, 23 μmol , 18%). Analytical RP-UHPLC: $t_{\text{R}} = 1.398$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{68}\text{H}_{113}\text{N}_{13}\text{O}_{19}\text{S}$ [M + H]⁺: 1448.81, found: 1447.80.

GalAG1-Cys. (GalA-KPL)₂KFC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (79 mg, 48 μmol , 39%). Analytical RP-UHPLC: $t_{\text{R}} = 1.419$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{78}\text{H}_{117}\text{N}_{13}\text{O}_{23}\text{S}$ [M + H]⁺: 1636.82, found: 1637.82 ($z = 1$), 819.41 ($z = 2$).

GalBG1-Cys. (GalB-KPL)₂KFC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (29.5 mg, 18 μmol , 15%). Analytical RP-UHPLC: $t_{\text{R}} = 1.420$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{70}\text{H}_{117}\text{N}_{13}\text{O}_{21}\text{S}_3$ [M + H]⁺: 1572.77, found: 1573.78 ($z = 1$), 786.89 ($z = 2$).

Het1G1-Cys. cFuc-KPLK(GalA-KPL)FC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (63 mg, 41 μmol , 33%). Analytical RP-UHPLC: $t_{\text{R}} = 1.454$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{73}\text{H}_{115}\text{N}_{13}\text{O}_{21}\text{S}$ [M + H]⁺: 1542.82, found: 1543.81 ($z = 1$), 771.91 ($z = 2$).

Het2G1-Cys. GalA-KPLK(cFuc-KPL)FC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (50 mg, 32 μmol , 26%). Analytical RP-UHPLC: $t_{\text{R}} = 1.457$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{73}\text{H}_{115}\text{N}_{13}\text{O}_{21}\text{S}$ [M + H]⁺: 1542.82, found: 1543.81 ($z = 1$), 771.91 ($z = 2$).

Het3G1-Cys. cFuc-KPLK(GalB-KPL)FC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (52 mg, 34 μmol , 28% yield). Analytical RP-UHPLC: $t_{\text{R}} = 1.462$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{69}\text{H}_{115}\text{N}_{13}\text{O}_{20}\text{S}_2$ [M + H]⁺: 1510.79, found: 1510.79 ($z = 1$), 755.90 ($z = 2$).

Het4G1-Cys. GalB-KPLK(cFuc-KPL)FC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (50 mg, 33 μmol , 26%). Analytical RP-UHPLC: $t_{\text{R}} = 1.454$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{69}\text{H}_{115}\text{N}_{13}\text{O}_{20}\text{S}_2$ [M + H]⁺: 1510.79, found: 1510.78 ($z = 1$), 755.90 ($z = 2$).

Het5G1-Cys. cFuc-KKLK(GalA-KKL)FC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (29.4 mg, 18 μmol , 15%). Analytical RP-UHPLC: $t_{\text{R}} = 2.464$ min (A/D 100/0 to 0/100 in 7.5 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{75}\text{H}_{125}\text{N}_{15}\text{O}_{21}\text{S}$ [M + H]⁺: 1604.90, found: 1604.90 ($z = 1$), 802.95 ($z = 2$).

Het6G1-Cys. GalA-KKLK(cFuc-KKL)FC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (32.5 mg, 20 μmol , 16%). Analytical RP-UHPLC: $t_{\text{R}} = 2.483$ min (A/D 100/0 to 0/100 in 7.5 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{75}\text{H}_{125}\text{N}_{15}\text{O}_{21}\text{S}$ [M + H]⁺: 1604.90, found: 1604.90 ($z = 1$), 802.95 ($z = 2$), 535.64 ($z = 3$).

Het7G1-Cys. cFuc-KKLK(GalB-KKL)FC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (66 mg, 42 μmol , 34%). Analytical RP-UHPLC: $t_{\text{R}} = 2.390$ min (A/D 100/0 to 0/100 in 7.5 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{71}\text{H}_{125}\text{N}_{15}\text{O}_{20}\text{S}_2$ [M + H]⁺: 1572.88, found: 1573.88 ($z = 1$), 787.44 ($z = 2$), 525.30 ($z = 3$).

Het8G1-Cys. GalB-KKLK(cFuc-KKL)FC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (52 mg, 33 μmol , 26%). Analytical RP-UHPLC: $t_{\text{R}} = 1.369$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{71}\text{H}_{125}\text{N}_{15}\text{O}_{20}\text{S}_2$ [M + H]⁺: 1572.88, found: 1572.87 ($z = 1$), 766.04 ($z = 2$), 524.90 ($z = 3$).

ClAcG1. (ClAc-KI)₂(KHI) was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (89 mg, 86 μmol , 70%). Analytical RP-UHPLC: $t_{\text{R}} = 1.443$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min^{-1} , $\lambda = 214$ nm). MS (ESI⁺) calc. for $\text{C}_{46}\text{H}_{81}\text{Cl}_2\text{N}_{13}\text{O}_9$ [M + H]⁺: 1030.58, found: 1029.57.

AcG1-Cys. (Ac-KPL)₂KFC was obtained from 500 mg resin (loading 0.25 mmol g^{-1}) as a foamy white solid after preparative RP-HPLC (37.8 mg, 33 μmol , 19%). Analytical RP-UHPLC: $t_{\text{R}} =$



1.497 min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₅₆H₉₃N₁₃O₁₁S [M + H]⁺: 1156.69, found: 1156.69 (z = 1), 578.85 (z = 2).

AcG1K-Cys. (Ac-KKL)₂KFC was obtained from 500 mg resin (loading 0.25 mmol g⁻¹) as a foamy white solid after preparative RP-HPLC (75.9 mg, 62 μmol, 50%). Analytical RP-UHPLC: t_R = 1.359 min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₅₉H₁₀₄N₁₄O₁₁S [M + H]⁺: 1217.78, found: 1218.78 (z = 1), 609.89 (z = 2).

NG2. (KPL)₄(KKI)₂KHI was obtained from 500 mg resin (loading 0.25 mmol g⁻¹) as a foamy white solid after preparative RP-HPLC (8 mg, 2.88 μmol, 44%). Analytical RP-UHPLC: t_R = 1.390 min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. C₁₄₀H₂₄₁N₃₅O₂₃ [M + H]⁺: 2781.88, found: 2781.88 (z = 1), 1391.95 (z = 2), 928.30 (z = 3), 696.48 (z = 4), 557.38 (z = 5).

ClAcD2. (ClAc-KPL)₄(KKI)₂KHI was obtained from 500 mg resin (loading 0.25 mmol g⁻¹) as a foamy white solid after preparative RP-HPLC (42.5 mg, 15 μmol, 17%). Analytical RP-UHPLC: t_R = 1.545 min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. MS (ESI⁺) calc. for C₁₃₀H₂₂₇Cl₄N₃₃O₂₅ [M + H]⁺: 2791.64, found: 2791.64.

Chloroacetyl cysteine (ClAc) ligation

A solution of core dendrimer (sequence containing chloroacetyl endgroups, around 3 mg, 1 eq.) and KI (20 eq.) in MeCN/H₂O (1 : 1, v/v) (300 μL) was prepared in a 2 mL glass vial. The mixture was flushed with Ar during 10 min. In another 2 mL glass vial arm dendrimer (Cys containing sequence, 1.5 eq. per chloroacetyl endgroup in core sequence) was weighed and flushed with Ar during 10 min. The core dendrimer/KI solution was transferred to the glass vial containing the arm dendrimer *via* a gas-tight syringe. DIPEA (55 eq.) was added and the solution stirred overnight at room temperature under argon atmosphere. The reaction was followed by analytical RP-UHPLC. After 16 to 23 hours, the reaction was quenched by the addition of 3.5 mL of eluent A (MilliQ deionized water containing 0.1% TFA). After filtration, the solution was directly purified by preparative RP-HPLC. x denotes the -CH₂-CO- bridge between the Cys side chain of the arm and the N-termini of the dendritic cores. Yields are given for the chloroacetyl cysteine (ClAc) ligation reaction.

FD2x. (cFuc-KPL)₄(KFCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (7.9 mg, 2.05 μmol, 70%). Analytical RP-UHPLC: t_R = 2.062 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₈₂H₃₀₅N₃₉O₄₇S₂ [M + H]⁺: 3854.22, found: 3854.23.

GalAG2x. (GalA-KPL)₄(KFCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (2.9 mg, 0.7 μmol, 24%). Analytical RP-UHPLC: t_R = 2.087 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₂₀₂H₃₁₃N₃₉O₅₅S₂ [M + H]⁺: 4230.24, found: 4231.24.

GalBG2x. (GalB-KPL)₄(KFCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (3.8 mg, 0.9 μmol, 32%). Analytical RP-UHPLC: t_R = 2.052 min (A/D 100/0 to 0/100 in 4.5

min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₈₆H₃₁₃N₃₉O₅₁S₆ [M + H]⁺: 4102.15, found: 4103.15.

Het1G2. (cFuc-KPLK(Gal-KPL)FCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (10.3 mg, 2.55 μmol, 88%). Analytical RP-UHPLC: t_R = 2.065 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₉₂H₃₀₉N₃₉O₅₁S₂ [M + H]⁺: 4042.23, found: 1011.81 (z = 4), 809.65 (z = 5), 674.88 (z = 6).

Het2G2. (GalA-KPLK(cFuc-KPL)FCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (9.3 mg, 2.3 μmol, 79%). Analytical RP-UHPLC: t_R = 2.075 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₉₂H₃₀₉N₃₉O₅₁S₂ [M + H]⁺: 4042.23, found: 4042.23.

Het3G2. (cFuc-KPLK(GalB-KPL)FCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (3.6 mg, 0.9 μmol, 31%). Analytical RP-UHPLC: t_R = 1.442 min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₈₄H₃₀₉N₃₉O₄₉S₄ [M + H]⁺: 3978.19, found: 3978.18.

Het4G2. (GalB-KPLK(cFuc-KPL)FCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (7.3 mg, 1.84 μmol, 63%). Analytical RP-UHPLC: t_R = 1.425 min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₈₄H₃₀₉N₃₉O₄₉S₄ [M + H]⁺: 3978.19, found: 3978.19 (z = 1), 1991.10 (z = 2).

Het5G2. (cFuc-KKLK(GalA-KKL)FCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (6.6 mg, 1.58 μmol, 54%). Analytical RP-UHPLC: t_R = 1.940 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₉₆H₃₂₉N₄₃O₅₁S₂ [M + H]⁺: 4166.40, found: 4166.40 (z = 1), 1390.14 (z = 3), 1042.86 (z = 4), 834.49 (z = 5), 695.57 (z = 6).

Het6G2. (GalA-KKLK(cFuc-KKL)FCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (7.1 mg, 1.7 μmol, 59%). Analytical RP-UHPLC: t_R = 1.938 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₉₆H₃₂₉N₄₃O₅₁S₂ [M + H]⁺: 4166.40, found: 4166.40.

Het7G2. (cFuc-KKLK(GalB-KKL)FCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (8.3 mg, 2.02 μmol, 69%). Analytical RP-UHPLC: t_R = 1.388 min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₈₈H₃₂₉N₄₃O₄₉S₄ [M + H]⁺: 4102.36, found: 4102.36.

Het8G2. (GalB-KKLK(cFuc-KKL)FCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (6.8 mg, 1.66 μmol, 57%). Analytical RP-UHPLC: t_R = 1.912 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₈₈H₃₂₉N₄₃O₄₉S₄ [M + H]⁺: 4102.36, found: 4102.36.

AcG2x. (Ac-KPL)₄(KFCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (4.8 mg, 1.47 μmol, 50%). Analytical RP-UHPLC: t_R = 2.120 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₅₈H₂₆₅N₃₉O₃₁S₂ [M + H]⁺: 3269.99, found: 3270.99.

AcG2xK. (Ac-KKL)₄(KFCxKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (5.9 mg, 1.74 μmol, 60%). Analytical RP-UHPLC: t_R = 1.953 min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, λ = 214 nm). MS (ESI⁺) calc. for C₁₆₄H₂₈₇N₄₁O₃₁S₂ [M + H]⁺: 3392.17, found: 3393.18.



Chloroacetyl cysteine (ClAc) ligation of Fuc₆ and Le^a

A solution of core dendrimer (sequence containing chloroacetyl endgroups), around 3 mg (1 eq.) and KI (20 eq.) in 130 μL of H₂O/MeCN (1 : 1, v/v) was prepared in a 5 mL pointed round-bottom flask. The solution was degassed under high vacuum for 10 min and pressurized under Ar. In a second 5 mL pointed round-bottom flask, peracetylated 6-mercaptohexyl- α -fucoside or β -mercaptohexyl-Lewis A (10 eq.) and K₂CO₃ (0.3 eq.) were loaded, degassed under high vacuum for 10 min, pressurized under Ar, and dissolved in 200 μL of dry MeOH. The reaction mixture was stirred for 1 hour at RT under an argon atmosphere, which effected complete deacetylation. The previously prepared solution of the core dendrimer and DIPEA (55 eq.) was then added to the reaction mixture which was stirred for 16 to 23 hours at RT under an argon atmosphere. The reaction was followed by RP-UHPLC. After completion the reaction was quenched by adding 10 mL of eluent A (MilliQ deionized water containing 0.1% TFA) and directly purified by RP-HPLC. Yields are given for the chloroacetyl cysteine (ClAc) ligation reaction.

FucC₆G2. (FucC₆-KPL)₄(KKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (2.5 mg, 0.7 μmol, 31%). Analytical RP-UHPLC: $t_R = 1.553$ min (A/D 100/0 to 0/100 in 2.2 min, flow rate 1.2 mL min⁻¹, $\lambda = 214$ nm). MS (ESI⁺) calc. MS (ESI⁺) calc. for C₁₇₈H₃₁₉N₃₃O₄₅S₄ [M + H]⁺: 3768.27, found: 3768.27.

Le^aC₆G2. (LeA-KPL)₄(KKI)₂KHI was obtained as a foamy white solid after preparative RP-HPLC (8 mg, 1.5 μmol, 63%). Analytical RP-UHPLC: $t_R = 2.137$ min (A/D 100/0 to 0/100 in 4.5 min, flow rate 1.2 mL min⁻¹, $\lambda = 214$ nm). MS (ESI⁺) calc. C₂₃₄H₄₁₁N₃₇O₈₅S₄ [M + H]⁺: 5228.80, found: 5228.80. [M + H + TFA]⁺: 5342.80, found: 5342.78. [M + H + 2TFA]⁺: 5456.80, found 5457.78.

Homodimerisation of Het2G1-Cys

G1 dendrimer **Het2G1-Cys** (4.5 mg, 1.0 eq.) was dissolved in 200 μL of degassed mQ-deionized water and flushed for 5 min with argon. Then, Aldrithiol (2,2-dithiodipyridine) (45.5 mM in MeOH, 0.45 eq.) was added. The thiol activation was followed by analytical RP-HPLC. When the reaction was completed, MeOH was evaporated and excess Aldrithiol was removed by extraction with DCM (4 times, 300 μL). The water phase was transferred to a new 5 mL pointed round-bottomed flask and flushed with Ar for 5 min. The pH was then adjusted to 8–9 with (NH₄)HCO₃ buffer solution (400 mM). A second portion of the monomeric dendrimer **Het2G1-Cys** (4.5 mg, 1.0 eq.) was added to the reaction mixture. The reaction mixture was stirred for 18 hours under argon at room temperature and followed by analytical RP-HPLC. After completion of the reaction, the mixture was acidified with 3 mL of water containing 0.1% of TFA. The homodimer (**Het2G1-Cys**)₂ was purified by preparative RP-HPLC.

(Het2G1-Cys)₂. (GalA-KPLK(cFuc-KPL)FC)₂ was obtained as a foamy white solid after preparative RP-HPLC (1.4 mg, 0.45 μmol, 7.1%). Analytical RP-UHPLC: $t_R = 3.009$ min (A/D 100/0 to 0/100 in 7.5 min, flow rate 1.2 mL min⁻¹, $\lambda = 214$ nm). MS (ESI⁺) calc. C₁₄₆H₂₂₈N₂₆O₄₂S₂ [M + H]⁺: 3083.69, found: 3084.60 ($z = 1$), 1028.54 ($z = 3$), 771.66 ($z = 4$).

Isothermal titration calorimetry (ITC)

Recombinant PA-III was purified from *Escherichia coli* BL21(DE3) containing the plasmid pET25pa2l as described previously.³⁰ Lyophilised LecB was dissolved in buffer (20 mM Tris, 100 mM NaCl, 100 μM CaCl₂, pH = 7.5). Protein concentration was checked and adjusted by measurement of absorbance at 280 nm using 1 Abs = 0.576 mg mL⁻¹ (MW = 11 863 g mol⁻¹) using a NanoDrop instrument. Ligands were dissolved using the same buffer. ITC were performed with iTC₂₀₀ calorimeter (MicroCal Inc.). Titration was performed on 0.028–0.50 mM LecB in the 200 μL sample cell using 2 μL injections of ligand every 180 s at 25 °C. The data were fitted with MicroCal Origin 8 software, according to standard procedures using a single-site model. Change in free energy ΔG was calculated from the equation: $\Delta G = \Delta H - T\Delta S$ where T is the absolute temperature, ΔH and ΔS are the change in enthalpy and entropy, respectively. Two independent titrations were performed for each ligand tested. Effective concentration has been determined using α -Me-fucose, a high affinity ligand that gives precise determination of the stoichiometry in active sites. The effective concentration that has been determined for this batch of LecB was of 70% (of mass concentration). The concentrations of LecB were then corrected based on this value.

Recombinant PA-IL was purified from *Escherichia coli* BL21(DE3) containing the plasmid pET25paIL as described previously.³¹ Lyophilised LecA was dissolved in buffer (Tris 20 mM, NaCl 100 mM, CaCl₂ 100 μM, pH = 7.5). Protein concentration was checked and adjusted by measurement of absorbance at 280 nm using 1 Abs = 2.116 mg mL⁻¹ (MW = 12 893 g mol⁻¹) using a NanoDrop instrument. Ligands were dissolved using the same buffer. ITC were performed with iTC₂₀₀ calorimeter (MicroCal Inc.). Titration was performed on 0.0166–0.526 mM LecA in the 200 μL sample cell using 2 μL injections of 200 μM ligand every 120 s at 25 °C. The data were fitted with MicroCal Origin 8 software, according to standard procedures using a single-site model. Change in free energy ΔG was calculated from the equation: $\Delta G = \Delta H - T\Delta S$ where T is the absolute temperature, ΔH and ΔS are the change in enthalpy and entropy, respectively. Two independent titrations were performed for each ligand tested.

Biofilm inhibition and dispersal

A modified version of the method described by Diggle *et al.*⁹ was employed.³¹ 96-well sterile, U-bottomed polystyrene microtiter plates (TPP Switzerland) were prepared by adding 200 μL of sterile deionized water to the peripheral wells to decrease evaporation from test wells. Aliquots of 180 μL of culture medium (0.25% (w/v) nutrient broth no. 2, Oxoid) containing desired concentration of the test compound were added to the internal wells. Compound containing solutions were sterile filtered (pore size 0.22 μm) prior to addition to the wells. Inoculum of *Pseudomonas aeruginosa* strain PAO1 was prepared from 5 mL overnight culture grown in LB broth at 37 °C and 180 rpm shaking. Aliquots of 20 μL of overnight cultures, pre-washed in 0.25% (w/v) nutrient broth and normalised to an OD₆₀₀ of 1, were inoculated into the test wells. Plates were incubated in a humid environment for 24–25 hours at 37 °C



under static conditions. Wells were washed twice with 200 μL sterile deionized water before staining with 200 μL 0.25% (w/v) nutrient broth containing 0.5 mM WST-8 and 20 μM phenazine ethosulfate for 2.5–3 hours at 37 $^{\circ}\text{C}$ under static conditions. Afterwards, the well supernatants were transferred to a polystyrene flat bottomed 96-well plate (TPP Switzerland) and the absorbance was measured at 450 nm with a plate reader (SpectraMax250 from Molecular Devices).

For biofilm dispersal, biofilm was being formed as described above but in the absence of compound for 24 hours. Wells were washed twice with 200 μL sterile deionized water before adding 200 μL 0.25% (w/v) nutrient broth containing the desired concentration of compound. Compound containing solutions were sterile filtered (pore size 0.22 μm) prior to addition to the wells. After another 24 hours of incubation at 37 $^{\circ}\text{C}$ under static conditions, the well supernatants were removed and the wells were washed twice with 200 μL sterile deionized water. The biofilm was stained with 200 μL of 0.25% (w/v) nutrient broth containing 0.5 mM WST-8 and 20 μM phenazine ethosulfate for 2.5–3 hours at 37 $^{\circ}\text{C}$ under static conditions. The resulting absorbance was measured as in the biofilm inhibition experiment.

Broth microdilution method

Glycopeptide dendrimers cytotoxicity was assayed against *Pseudomonas aeruginosa* PAO1. To determine the Minimal Inhibitory Concentration (MIC), broth microdilution method was used. A colony of bacteria from glycerol stock was grown in LB medium overnight at 37 $^{\circ}\text{C}$ and 180 rpm shaking. The compounds were prepared as stock solutions of 64 μM in LB medium, added to the first well of 96-well sterile, F-bottomed polystyrene microtiter plates (TPP, untreated) and diluted serially by 1/2. Compound containing solutions were sterile filtered (pore size 0.22 μm) prior to addition to the wells. The concentration of the bacteria was quantified by measuring absorbance at 600 nm and diluted to an OD_{600} of 0.022 in LB medium. The sample solutions (150 μL) were mixed with 4 μL diluted bacterial suspension with a final inoculation of about 5×10^5 CFU. For each test, two columns of the plate were kept for sterility control (LB medium only), growth control (LB medium with bacterial inoculum, no compound). The positive control, polymyxin B (starting with a concentration of 64 $\mu\text{g mL}^{-1}$) in LB medium with bacterial inoculums, was introduced in the two first lines of the plate. The plates were incubated at 37 $^{\circ}\text{C}$ for ~ 18 hours under static conditions. 15 μL of 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) (1 mg mL^{-1} in sterilized milliQ deionized water) were added to each well and the plates were incubated for 10 minutes at room temperature. The minimal inhibitory concentration (MIC) was defined as the lowest concentration of the glycopeptide dendrimer that inhibits the visible growth of the tested bacteria (yellow) with the unaided eye. For broth microdilution assay, polymyxin B was used as references.¹⁷

Quantification of bacterial cell viability in biofilm inhibition and dispersal assays

96-well sterile, U-bottomed polystyrene microtiter plates (TPP Switzerland) were prepared by adding 200 μL of sterile

deionized water to the peripheral wells to decrease evaporation from test wells. Aliquots of 180 μL of culture medium (0.25% (w/v) nutrient broth no. 2, Oxoid) containing desired concentration of the test compound were added to the internal wells. Compound containing solutions were sterile filtered (pore size 0.22 μm) prior to addition to the wells. Inoculum of *Pseudomonas aeruginosa* strain PAO1 was prepared from 5 mL overnight culture grown in LB broth at 37 $^{\circ}\text{C}$ and 180 rpm shaking. Aliquots of 20 μL of overnight cultures, pre-washed in 0.25% (w/v) nutrient broth and normalised to an OD_{600} of 1, were inoculated into the test wells. Plates were incubated in a humid environment for 24–25 hours at 37 $^{\circ}\text{C}$ under static conditions. The supernatants were used to investigate the bacterial cell viability by two different methods: CFU plating and OD_{600} measurement.

For OD_{600} measurement, the 200 μL of the supernatant were added to 5 mL of LB medium and shaken at 37 $^{\circ}\text{C}$ for 7–8 hours. The OD_{600} was measured with an Ultrospec 10 cell density meter from GE Health Care Life Sciences and compared to the w/o supernatant. For CFU plating, the supernatant was added to the first well of a 96-well sterile, F-bottomed microtiter plate (TPP, untreated) and diluted serially by 1/2 (raw 1 to 6) in 180 μL NaCl 0.9%. 4 μL of raw 6 were plated on LB agar and incubated at 37 $^{\circ}\text{C}$ for 24 hours under static conditions. Wells were washed twice with 200 μL sterile deionized water before staining with 200 μL 0.25% (w/v) nutrient broth containing 0.5 mM WST-8 and 20 μM phenazine ethosulfate for 2.5–3 hours at 37 $^{\circ}\text{C}$ under static conditions. Afterwards, the well supernatants were transferred to a polystyrene flat bottomed 96-well plate (TPP Switzerland) and the absorbance was measured at 450 nm with a plate reader (SpectraMax250 from Molecular Devices).

For biofilm dispersal, biofilm was being formed as described above but in the absence of compound for 24 hours. Wells were washed twice with 200 μL sterile deionized water before adding 200 μL 0.25% (w/v) nutrient broth containing the desired concentration of compound. After another 24 hours of incubation at 37 $^{\circ}\text{C}$ under static conditions, the well supernatants were used to check the bacterial cell viability by CFU plating and OD_{600} measurement. For OD_{600} measurement, the 200 μL of the supernatant were added to 5 mL of LB medium and shaken at 37 $^{\circ}\text{C}$ for 7–8 hours. The OD_{600} was measured and compared to the w/o supernatant. For CFU plating, the supernatant was added to the first well of a 96-well sterile, F-bottomed microtiter plate (TPP, untreated) and diluted serially by 1/2 (raw 1 to 6) in 180 μL NaCl 0.9%. 4 μL of the raw 6 were plated on LB agar and incubated at 37 $^{\circ}\text{C}$ for 24 hours under static conditions. The wells were washed twice with 200 μL sterile deionized water. The biofilm was stained with 200 μL of 0.25% (w/v) nutrient broth containing 0.5 mM WST-8 and 20 μM phenazine ethosulfate for 2.5–3 hours at 37 $^{\circ}\text{C}$ under static conditions. The resulting absorbance was measured as in the biofilm inhibition experiment.

Growth curve control studies

180 μL of LB medium were added to wells of a 96-well sterile, F-bottomed polystyrene microtiter plate. Compound containing



solutions were sterile filtered (pore size 0.22 μm) prior to addition to the wells. An overnight culture of *Pseudomonas aeruginosa* strain PA01 grown in LB was standardised to an OD_{600} of 0.5 and 20 μL aliquots were inoculated into test wells. The microtiter plate was shaken at 37 $^{\circ}\text{C}$ and 180 rpm. Growth was followed at several time points over a period of 24 hours with a plate reader at 600 nm (SpectraMax250 from Molecular Devices).

Crystallization of Het1G1·LecB

LecB lectin was expressed, cleaned up and dialyzed as previously described.⁵² Co-crystallization of **Het1G1** with LecB lectin was carried out by the sitting drop method. In brief, lyophilized protein was dissolved in water (5 mg mL^{-1}) in the presence of salts (1 mM CaCl_2 and MgCl_2). The compound **Het1G1** was added to the LecB homo-tetramer at a 16 : 1 molar excess. Crystals were obtained within five days after mixing 1.5 μL of LecB ligand-complex with 1.5 μL of reservoir solution at 18 $^{\circ}\text{C}$. Primary crystallization condition was found in Crystal Screens I/II (Hampton Research, Laguna Niguel, CA, USA). Crystals of highest diffraction were obtained from condition 0.1 M sodium acetate trihydrate, 2 M sodium chloride pH 4.6, Crystal Screen II-9. The structures were solved using the XDS,⁵³ CCP4,⁵⁴ the phenix⁵⁵ program suite and the coot⁵⁶ graphical program. Pictures were done with the help of pymol.

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References

- M. N. Hurley, M. Camara and A. R. Smyth, *Eur. Respir. J.*, 2012, **40**, 1014–1023.
- V. E. Wagner and B. H. Iglewski, *Clin. Rev. Allergy Immunol.*, 2008, **35**, 124–134.
- N. Garber, U. Guempel, A. Belz, N. Gilboa-Garber and R. J. Doyle, *Biochim. Biophys. Acta*, 1992, **1116**, 331–333.
- G. Cioci, E. P. Mitchell, C. Gautier, M. Wimmerova, D. Sudakevitz, S. Perez, N. Gilboa-Garber and A. Imberty, *FEBS Lett.*, 2003, **555**, 297–301.
- A. Imberty, M. Wimmerova, E. P. Mitchell and N. Gilboa-Garber, *Microbes Infect.*, 2004, **6**, 221–228.
- E. Mitchell, C. Houles, D. Sudakevitz, M. Wimmerova, C. Gautier, S. Perez, A. M. Wu, N. Gilboa-Garber and A. Imberty, *Nat. Struct. Biol.*, 2002, **9**, 918–921.
- R. Loris, D. Tielker, K. E. Jaeger and L. Wyns, *J. Mol. Biol.*, 2003, **331**, 861–870.
- M. Mewe, D. Tielker, R. Schonberg, M. Schachner, K. E. Jaeger and U. Schumacher, *J. Laryngol. Otol.*, 2005, **119**, 595–599.
- S. P. Diggle, R. E. Stacey, C. Dodd, M. Camara, P. Williams and K. Winzer, *Environ. Microbiol.*, 2006, **8**, 1095–1104.
- H. P. Hauber, M. Schulz, A. Pforte, D. Mack, P. Zabel and U. Schumacher, *Int. J. Med. Sci.*, 2008, **5**, 371–376.
- C. Chemani, A. Imberty, S. de Bentzmann, M. Pierre, M. Wimmerova, B. P. Guery and K. Faure, *Infect. Immun.*, 2009, **77**, 2065–2075.
- A. Bernardi, J. Jimenez-Barbero, A. Casnati, C. de Castro, T. Darbre, F. Fieschi, J. Finne, H. Funken, K. E. Jaeger, M. Lahmann, T. K. Lindhorst, M. Marradi, P. Messner, A. Molinaro, P. V. Murphy, C. Nativi, S. Oscarson, S. Penades, F. Peri, R. J. Pieters, O. Renaudet, J. L. Reymond, B. Richichi, J. Rojo, F. Sansone, C. Schaffer, W. B. Turnbull, T. Velasco-Torrijos, S. Vidal, S. Vincent, T. Wennekes, H. Zuillhof and A. Imberty, *Chem. Soc. Rev.*, 2013, **42**, 4709–4727.
- A. Clouet, T. Darbre and J. L. Reymond, *Angew. Chem., Int. Ed.*, 2004, **43**, 4612–4615.
- E. M. V. Johansson, E. Kolomiets, F. Rosenau, K.-E. Jaeger, T. Darbre and J.-L. Reymond, *New J. Chem.*, 2007, **31**, 1291–1299.
- E. Kolomiets, E. M. Johansson, O. Renaudet, T. Darbre and J. L. Reymond, *Org. Lett.*, 2007, **9**, 1465–1468.
- T. Darbre and J. L. Reymond, *Curr. Top. Med. Chem.*, 2008, **8**, 1286–1293.
- J.-L. Reymond and T. Darbre, *Org. Biomol. Chem.*, 2012, **10**, 1483–1492.
- J. L. Reymond, M. Bergmann and T. Darbre, *Chem. Soc. Rev.*, 2013, **42**, 4814–4822.
- E. M. Johansson, S. A. Cruz, E. Kolomiets, L. Buts, R. U. Kadam, M. Cacciarini, K. M. Bartels, S. P. Diggle, M. Camara, P. Williams, R. Loris, C. Nativi, F. Rosenau, K. E. Jaeger, T. Darbre and J. L. Reymond, *Chem. Biol.*, 2008, **15**, 1249–1257.
- E. Kolomiets, M. A. Swiderska, R. U. Kadam, E. M. Johansson, K. E. Jaeger, T. Darbre and J. L. Reymond, *ChemMedChem*, 2009, **4**, 562–569.
- E. M. V. Johansson, R. U. Kadam, G. Rispoli, S. A. Cruz, K.-M. Bartels, S. P. Diggle, M. Camara, P. Williams, K.-E. Jaeger, T. Darbre and J.-L. Reymond, *MedChemComm*, 2011, **2**, 418–420.
- R. U. Kadam, M. Bergmann, M. Hurley, D. Garg, M. Cacciarini, M. A. Swiderska, C. Nativi, M. Sattler, A. R. Smyth, P. Williams, M. Camara, A. Stocker, T. Darbre and J. L. Reymond, *Angew. Chem., Int. Ed.*, 2011, **50**, 10631–10635.
- G. M. L. Consoli, G. Granata, V. Cafiso, S. Stefani and C. Geraci, *Tetrahedron Lett.*, 2011, **52**, 5831–5834.
- J. J. Lundquist and E. J. Toone, *Chem. Rev.*, 2002, **102**, 555–578.
- C. C. Lee, J. A. MacKay, J. M. J. Frechet and F. C. Szoka, *Nat. Biotechnol.*, 2005, **23**, 1517–1526.
- K. Marotte, C. Preville, C. Sabin, M. Moume-Pymbock, A. Imberty and R. Roy, *Org. Biomol. Chem.*, 2007, **5**, 2953–2961.



- 27 B. Gerland, A. Goudot, G. Pourceau, A. Meyer, S. Vidal, E. Souteyrand, J. J. Vasseur, Y. Chevotot and F. Morvan, *J. Org. Chem.*, 2012, **77**, 7620–7626.
- 28 E. L. Dane, A. E. Ballok, G. A. O'Toole and M. W. Grinstaff, *Chem. Sci.*, 2014, **5**, 551–557.
- 29 S. Zhang, R. O. Moussodia, C. Murzeau, H. J. Sun, M. L. Klein, S. Vertesy, S. Andre, R. Roy, H. J. Gabius and V. Percec, *Angew. Chem., Int. Ed.*, 2015, **54**, 4036–4040.
- 30 S. Cecioni, A. Imberty and S. Vidal, *Chem. Rev.*, 2015, **115**, 525–561.
- 31 R. U. Kadam, M. Bergmann, D. Garg, G. Gabrieli, A. Stocker, T. Darbre and J.-L. Reymond, *Chem.–Eur. J.*, 2013, **18**, 17054–17063.
- 32 R. U. Kadam, D. Garg, J. Schwartz, R. Visini, M. Sattler, A. Stocker, T. Darbre and J. L. Reymond, *ACS Chem. Biol.*, 2013, **8**, 1925–1930.
- 33 R. Visini, X. Jin, M. Bergmann, G. Michaud, F. Pertici, O. Fu, A. Pukin, T. R. Branson, D. M. Thies-Weesie, J. Kemmink, E. Gillon, A. Imberty, A. Stocker, T. Darbre, R. J. Pieters and J. L. Reymond, *ACS Chem. Biol.*, 2015, **10**, 2455–2462.
- 34 M. Bergmann, G. Michaud, R. Visini, X. Jin, E. Gillon, A. Stocker, A. Imberty, T. Darbre and J. L. Reymond, *Org. Biomol. Chem.*, 2015, DOI: 10.1039/c1035ob01682g.
- 35 N. A. Uhlich, T. Darbre and J. L. Reymond, *Org. Biomol. Chem.*, 2011, **9**, 7071–7084.
- 36 M. Stach, T. N. Siriwardena, T. Kohler, C. van Delden, T. Darbre and J. L. Reymond, *Angew. Chem., Int. Ed.*, 2014, **53**, 12827–12831.
- 37 S. Perret, C. Sabin, C. Dumon, M. Pokorna, C. Gautier, O. Galanina, S. Ilia, N. Bovin, M. Nicaise, M. Desmadril, N. Gilboa-Garber, M. Wimmerova, E. P. Mitchell and A. Imberty, *Biochem. J.*, 2005, **389**, 325–332.
- 38 F. Pertici, N. J. de Mol, J. Kemmink and R. J. Pieters, *Chem.–Eur. J.*, 2013, **19**, 16923–16927.
- 39 C. Sabin, E. P. Mitchell, M. Pokorna, C. Gautier, J. P. Utille, M. Wimmerova and A. Imberty, *FEBS Lett.*, 2006, **580**, 982–987.
- 40 Y. C. Lee and R. T. Lee, *Acc. Chem. Res.*, 1995, **28**, 321–327.
- 41 L. R. Hoffman, D. A. D'Argenio, M. J. MacCoss, Z. Zhang, R. A. Jones and S. I. Miller, *Nature*, 2005, **436**, 1171–1175.
- 42 S. Komba, H. Ishida, M. Kiso and A. Hasegawa, *Bioorg. Med. Chem.*, 1996, **4**, 1833–1847.
- 43 A. Toepfer and R. R. Schmidt, *J. Carbohydr. Chem.*, 1993, **12**, 809–822.
- 44 H. Anwar and J. W. Costerton, *Antimicrob. Agents Chemother.*, 1990, **34**, 1666–1671.
- 45 S. D. Aaron, W. Ferris, K. Ramotar, K. Vandemheen, F. Chan and R. Saginur, *J. Clin. Microbiol.*, 2002, **40**, 4172–4179.
- 46 R. Eckert, K. M. Brady, E. P. Greenberg, F. Qi, D. K. Yarbrough, J. He, I. McHardy, M. H. Anderson and W. Shi, *Antimicrob. Agents Chemother.*, 2006, **50**, 3833–3838.
- 47 M. Tre-Hardy, F. Vanderbist, H. Traore and M. J. Devleeschouwer, *Int. J. Antimicrob. Agents*, 2008, **31**, 329–336.
- 48 S. Moreau-Marquis, G. A. O'Toole and B. A. Stanton, *Am. J. Respir. Cell Mol. Biol.*, 2009, **41**, 305–313.
- 49 S. Moreau-Marquis, B. Coutermarsh and B. A. Stanton, *J. Antimicrob. Chemother.*, 2015, **70**, 160–166.
- 50 E. P. Mitchell, C. Sabin, L. Snajdrova, M. Pokorna, S. Perret, C. Gautier, C. Hofr, N. Gilboa-Garber, J. Koca, M. Wimmerova and A. Imberty, *Proteins*, 2005, **58**, 735–746.
- 51 B. Blanchard, A. Nurisso, E. Hollville, C. Tetaud, J. Wiels, M. Pokorna, M. Wimmerova, A. Varrot and A. Imberty, *J. Mol. Biol.*, 2008, **383**, 837–853.
- 52 H. Funken, K. M. Bartels, S. Wilhelm, M. Brocker, M. Bott, M. Bains, R. E. Hancock, F. Rosenau and K. E. Jaeger, *PLoS One*, 2012, **7**, e46857.
- 53 W. Kabsch, *Acta Crystallogr., Sect. D: Biol. Crystallogr.*, 2010, **66**, 125–132.
- 54 M. D. Winn, C. C. Ballard, K. D. Cowtan, E. J. Dodson, P. Emsley, P. R. Evans, R. M. Keegan, E. B. Krissinel, A. G. W. Leslie, A. McCoy, S. J. McNicholas, G. N. Murshudov, N. S. Pannu, E. A. Potterton, H. R. Powell, R. J. Read, A. Vagin and K. S. Wilson, *Acta Crystallogr.*, 2011, **67**, 235–242.
- 55 P. D. Adams, P. V. Afonine, G. Bunkoczi, V. B. Chen, I. W. Davis, N. Echols, J. J. Headd, L.-W. Hung, G. J. Kapral, R. W. Grosse-Kunstleve, A. J. McCoy, N. W. Moriarty, R. Oeffner, R. J. Read, D. C. Richardson, J. S. Richardson, T. C. Terwilliger and P. H. Zwart, *Acta Crystallogr.*, 2010, **66**, 213–221.
- 56 P. Emsley, B. Lohkamp, W. G. Scott and K. Cowtan, *Acta Crystallogr.*, 2010, **66**, 486–501.

