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Sequential bottom-up and top-down processing for the synthesis of transition metal dichalcogenide nanosheets: the case of rhenium disulfide (ReS2)†

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Bottom-up (aerosol-assisted chemical vapor deposition, AACVD) and top-down (liquid phase exfoliation, LPE) processing methodologies are used in tandem to produce colloids of few-layer thick rhenium disulfide (ReS2) in *N*-methyl pyrrolidone. The processing route we use is a potentially robust and scalable pathway to manufacture useful 2D materials.

Rheniite (rhenium disulfide, ReS_2) is a naturally occurring yet extremely rare transition metal dichalcogenide (TMDC).[1] Like other TMDCs it has a layered crystal structure (Fig. 1); with a preference for pinacoidal cleavage. In most common TMDCs such as molybdenum disulfide (MoS_2), the metal spacing is regular and can be simply described by conventional packing. However, the ReS_2 crystal structure is more complex because metal–metal interactions contribute significantly to the interlayer bonding, leading to the formation of a repeating motif of Re-parallelograms of varying sizes within the metal layer.[2]

Very recently, there has been some interest in ReS_2 as a two-dimensional semiconductor. ReS_2 is a direct tunable band gap semiconductor, from 1.47 eV in the bulk to 1.61 eV as a monolayer and shows evidence of birefringent behavior.[3] The band gap energy is predicted to be insensitive to strain, and thus the material may be tolerant to a range of processing methods.[4] It has been proposed that these favourable properties arise from interlayer decoupling caused by Peierls distortion of the 1T structure of ReS_2, which precludes ordered stacking and minimizes wavefunction overlap.[5] Raman spectra of ReS_2 have been shown to be both anisotropic and dependent on the
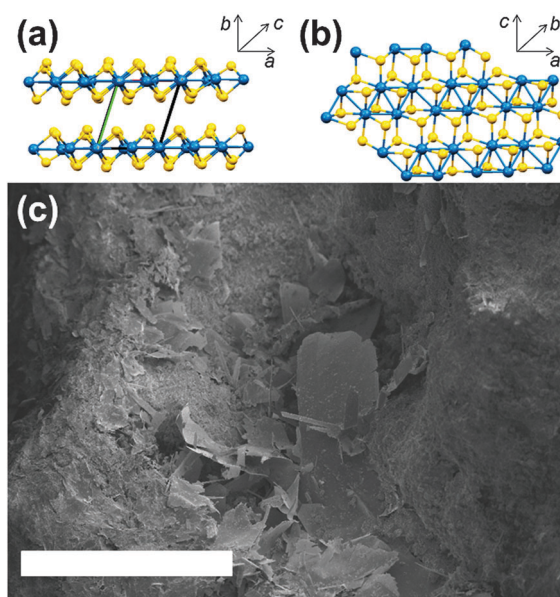

Fig. 1 Structure of rheniite (rhenium disulfide, ReS_2). (a) The crystal structure of ReS_2, showing the triclinic unit cell (a = 6.378 Å, b = 6.417 Å, c = 6.461 Å with α = 91.62°, β = 119.07°, γ = 105.115°). (b) A monolayer of ReS_2 looking along the [010] zone axis. (c) Secondary electron SEM image (6 kV) of a sample of rheniite from Kudriavy volcano type locality showing the sheet-like crystals typical of the layered mineral. Scale bar represents 1 mm.

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number of layers in the material.[6,7] Current routes to ReS_2 monolayers are not ideal as they either rely on chemical vapor deposition processes requiring high vacuum,[8,9] or harsh chemical exfoliation with organolithium species.[10]

Aerosol-assisted chemical vapor deposition (AACVD)[11] from single source precursors is a versatile route to produce pure metal chalcogenide thin films such as tin(II) sulfide (SnS),[12] cadmium sulfide (CdS),[13] pyrite (FeS_2),[14–16] molybdenum disulfide (MoS_2) as well as to produce related (*e.g.* Cr-doped) variants.[17–19] Additionally a range of useful metal oxides[6,7,20–24] and organohalide lead(II) perovskites have been reported.[25–27] AACVD can take place at moderate temperature, at ambient pressure and is inherently

scalable. We have recently reported the production of poly-crystalline thin films of ReS_2 using AACVD from a single source precursor.[28]

In general, two-dimensional (2D) materials such as graphene[29] and 2D TMDCs[30] are produced for demonstrative studies using mechanical (scotch tape) exfoliation. More recently, liquid-phase exfoliation (LPE) has emerged as a more scalable and versatile solution-phase processing methodology to produce industrially

significant quantities of nanomaterials ranging from graphene,[31] TMDCs,[32] phosphorene[33] and 2D tin(II) sulfide.[34] Negatively charged ReS_2 nanosheets have been reported by Chen and co-workers using lithium intercalation-exfoliation.[35] In this paper we present the marriage of a scalable bottom-up process (AACVD) with top-down LPE as a new scalable processing route to produce atomically thin materials, in this case few-layer 2D ReS_2.

ReS_2 thin films were produced by AACVD on glass substrates at 550 °C using the single source precursor $[Re(S_2CC_6H_5)(S_3CC_6H_5)_2]$.[36] Characterization of the films by Raman spectroscopy agreed with the previously reported spectra for ReS_2[5,28] and powder X-ray diffraction (pXRD) measurements showed a preferred orientation in the (010) plane whilst scanning electron microscopy (SEM) revealed the lamellar morphology of the individual crystallites within the thin film. EDX spectrum mapping of the Re L_α (8.6 keV) and S K_α (2.3 keV) lines demonstrated the homogeneous co-localisation of both elements in the thin film. Atomic force microscopy (AFM) at the film-edge was used to measure the thickness of the film as 150 ± 50 nm. The above evidence for ReS_2 film formation is given fully in the ESI.† All of the characterisation data from the thin films were consistent with that observed previously for ReS_2 films produced by AACVD.[28] The thin films were then immersed in N-methyl-2-pyrrolidone (NMP) and ultrasonicated for 36 h, using similar
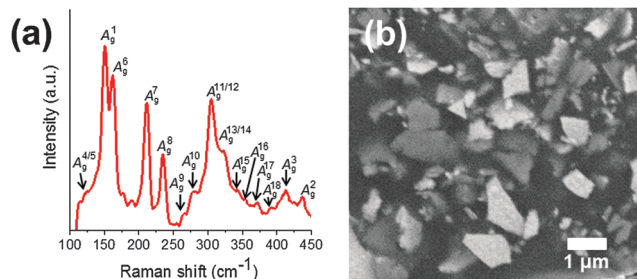

Fig. 2 Characterization of ReS_2 nanosheets exfoliated from thin films of ReS_2 grown by AACVD at 550 °C. (a) Raman spectrum of exfoliated nanosheets drop cast from NMP on Si@SiO$_2$ substrates with peaks labelled using the scheme proposed by Pradhan et al.[21] (b) Secondary electron SEM image (6 kV) of ReS_2 nanosheets drop cast from NMP onto Si@SiO$_2$ substrates.
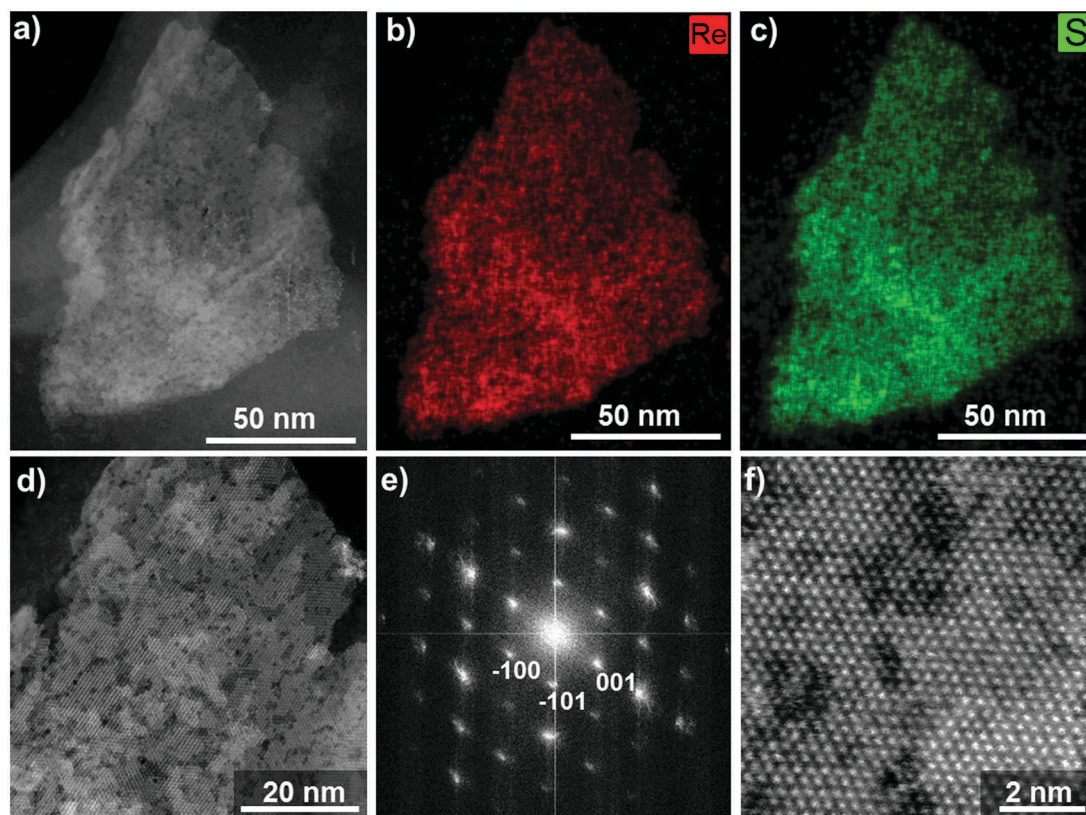

Fig. 3 (a) HAADF STEM image of a representative LPE flake of ReS_2 grown by AACVD at 550 °C. Corresponding EDX elemental maps showing (b) Re L_α (8.6 keV) and (c) S K_α (2.3 keV) X-ray counts demonstrate co-localisation of Re and S, the summed EDX spectrum can be found in the ESI.† (d) Atomic resolution HAADF STEM image of a region of the flake shown in (a). (e) Fourier transform of (d), confirming the expected crystal structure for ReS_2 viewed down the [010] zone axis and demonstrating that the entire nanosheet is a single crystal. (f) Shows a higher magnification image of the same flake so that the atomic structure can be clearly seen.

conditions to that reported previously for liquid exfoliation of other 2D materials.[33,34] The resulting colloidal suspension of nanosheets was drop-casted onto silicon wafers coated with 300 nm of silicon dioxide (Si@SiO₂) for analysis.

Raman spectroscopy of ReS₂ nanosheets on Si@SiO₂ substrates showed signals of a weak intensity associated with ReS₂; optical phonon modes are observed at *ca.* 150, 163, 212, 235 and 305 cm^{-1} (Fig. 2a), as previously reported. A total of 18 phonon modes, which have previously been attributed to vibrations of A_g symmetry,[21] were identified. As all the modes were observed at once, we conclude that a number of randomly oriented flakes were sampled during the measurement, as the Raman response from ReS₂ has been shown to be anisotropic for large mechanically exfoliated flakes.[6,21] There are some subtle differences in the Raman spectrum of the nanosheets compared to that of the bulk thin film: the A_g^1 mode shifts from 148.5 cm^{-1} to 150.4 cm^{-1} whilst the A_g^7 mode shifts from 207.8 to 211.6 cm^{-1} in going from bulk to nanosheet form. These shifts are consistent with those found by Hone and co-workers for micromechanically exfoliated ReS₂ compared with bulk ReS₂.[6] Secondary electron scanning electron microscopy (SEM) images at 6 kV reveal nanosheets of up to 1 μm in length/width (Fig. 2b), though there are many smaller nanosheets found in the sample, typically < 500 nm. The smaller sheet size from LPE sols is typical for TMDCs produced by this method compared to mechanical exfoliation, where sheet size can be in the micron range.[10]

High-angle annular dark field scanning transmission electron microscope (HAADF STEM) images of the ReS₂ nanosheets show that they are highly crystalline (Fig. 3). The calculated rhenium positions of octahedral (*T*) ReS₂ as-viewed down the [010] zone axis of can be overlaid on the atomic resolution images and both match perfectly (ESI†). Fast Fourier transforms taken from these atomic resolution HAADF images of individual flakes demonstrate that the sheets are comprised of single crystals of ReS₂. Energy dispersive X-ray (EDX) elemental mapping of the Re L$_\alpha$ (8.6 keV) (Fig. 3b) and S K$_\alpha$ (2.3 keV) (Fig. 3c) X-ray counts show the expected co-localisation of Re and S within the boundary of the flakes.

Atomic force microscopy (AFM) was used to analyse the heights of the ReS₂ flakes produced by our novel method (Fig. 4). It was found that the flakes ranged in height from 1–20 nm. Significant populations of monolayer (*ca.* 1.7 nm), bilayer (3.9 nm) and trilayer (6.0 nm) flakes were observed along with larger flakes with heights up to *ca.* 20 nm, *i.e.* up to 10-layers, though the thinnest flakes do represent the majority of the population analysed under these conditions.

Hence, we conclude that the exfoliation procedure successfully produces ultrathin and crystalline ReS₂ sheets from ReS₂ thin films grown by AACVD. Our ReS₂ nanosheets are identical in phase to those reported previously by Chen and co-workers and have similar thickness,[35] but with the added advantage that there is no lithium associated with them (*i.e.* should be neutral compared to the negatively charged sheets produced by lithium intercalation-exfoliation) and do not require post-exfoliation modification.

In summary, we have coupled the bottom up processing of ReS₂ by AACVD with top-down LPE to produce ultrathin ReS₂
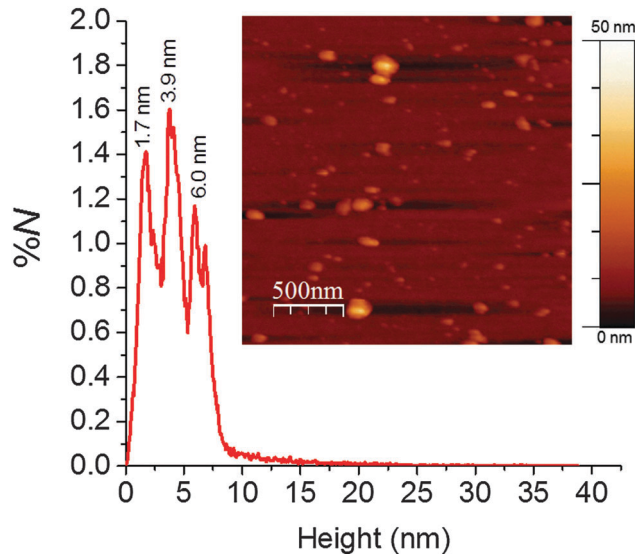

Fig. 4 Particle size distribution of ReS₂ nanosheets exfoliated from thin films of ReS₂ grown by AACVD at 550 °C as-measured by AFM showing significant peaks at 1.7, 3.9 and 6.0 nm. Inset: Representative height profile AFM image of exfoliated ReS₂ nanosheets spin coated from NMP suspension onto a Si@SiO₂ substrate.

nanosheets. In the case of ReS₂, which is a useful 2D semiconductor, this is important as natural sources are scarce and synthetic pathways are required. Additionally our processing methodology is potentially scalable and thus could be a route to commercial exploitation. Finally, the serial processing we report here is potentially applicable to a wide range of layered TMDC materials. It will be useful in the future for the production of a myriad of synthetic TMDC materials with interesting optoelectronic properties which are complementary to graphene. As it has been established that doping of TMDCs is possible by AACVD,[18] there may also be scope to produce doped 2D materials in the future which are interesting for a range of applications.[19]

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