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Microfluidic synthesis of micrometer-sized photoresponsive actuators based on liquid crystalline elastomers†

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The fabrication of photoresponsive micrometer-sized liquid crystalline elastomer (LCE) particles has been described in this article for the first time. The preparation of the LCE particles using a microfluidic device by a continuous “on the fly” technique allows their fast processing with an irradiation time of less than 2 seconds. In order to accelerate the polymerization and to make the preparation of samples with thicknesses of several hundred microns possible, we modified a NIR-photoinitiation system by adding a bleaching agent. In addition to the photoinitiator, a side-chain liquid crystalline monomer containing an azo-group and a LC-crosslinker were used in the polymerization mixture. For the optimization of the polymerization parameters, LCE films were produced first. These films showed a strong thermoresponsive actuation of more than 40% upon the nematic–isotropic phase transition and changed their length by up to 34% during irradiation with light for just 8 seconds, due to the *trans*–*cis* isomerization of the azo-group. Fabricated particles were investigated with respect to their response to heating, irradiation with light and swelling in a good solvent. A strong photoresponsive actuation of about 24% was found for these particles, which was reversible either by irradiation with red light (8 minutes) or by an isothermal back-isomerization of the azo-group in the dark (17 minutes). The reversibility of the light induced actuation of our particles makes them promising candidates for application as artificial muscles in microrobotics.

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Introduction

Liquid crystalline elastomers (LCEs) attract attention due to their actuation properties, which were already predicted by de Gennes¹ in 1975. They are the subject of many articles and books and are promising candidates for application as artificial muscles in microrobotics.^{2–10} These materials combine the entropy elasticity of an elastomer with the self-organization of a liquid crystal in which the latter acts as an anisotropic solvent for the isotropic polymer chain. In LCEs the mesogens of the liquid crystal either are part of the polymer chain (main-chain LCEs) or they are attached *via* an alkyl spacer (side-chain LCEs) to provide a good interaction. Due to the orientation of the mesogens the polymer chain is deformed into a stretched conformation below the clearing temperature, but when the sample is heated into the isotropic phase the order is lost and the polymer backbone can relax into the random coil conformation. If the liquid crystalline phase is present as a monodomain, the phase transition results in a macroscopic change in the shape of

the whole sample. That is, it expands in a certain direction and contracts in the opposite. With main-chain LCEs, actuations of up to 400% could be obtained¹¹ and thus much more than what is achievable with piezo-actuators. Also compared to high actuating hydrogels, they have the advantage that no mass transfer is necessary. To gain a monodomain within the LCE, an application of an external force is necessary for the alignment of the single domains before crosslinking. This can be for example: stretching of the prepolymerized sample,¹² aligning by magnetic or electric fields^{13,14} and orientation with the aid of a photo-alignment layer.¹⁵ For the orientation of the mesogens in droplets the flow in a high viscous oil in microfluidic devices is the best known way.^{16,17}

So far thermoactuating LCEs, which change their shape (expand and contract) by heating them above the clearing temperature, are investigated mostly. However, a response to a temperature change of the environment is not desirable for devices, because then the whole surrounding needs to be heated up. So actuation induced in a more practical way is desired. Zappe *et al.* solved this problem by incorporating a thin platinum wire into a radial oriented LCE disc.⁹ In this manner they could heat the sample locally with electricity. They used this local stimulus to imitate the movement of a human iris.¹⁰

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Magnetic and electric fields can also be used to trigger LCE actuators, but make the addition of inorganic materials necessary.^{18–21}

Great attention has also been paid to photoresponsive LCEs. Electromagnetic irradiation can here be used in two ways. At first, radiation can be used for local heating caused by light absorption^{22–25} or for a chemical modification of the LC-material by photoisomerization (isothermal shift of the phase transition temperatures due to the creation of less mesogenic compounds). In this respect LCEs containing an azo-benzene group in the mesogenic unit are attractive. In their ground state azobenzenes are present as the *trans*-conformer, but can be switched into the *cis*-isomer by continuous irradiation with UV-light. This *trans*–*cis* isomerization can be reversed either by thermal back-isomerization or by irradiation with light of higher wavelength ($\lambda > 450$ nm).²⁶ While the *trans*-isomer has a rod-like shape and is well compatible with LC-phases, the *cis*-isomer is bent, whereby the liquid crystalline order is reduced. This results in a decrease of the order parameter S and the clearing temperature. Thus, during photoisomerization of an azobenzene containing LCE sample, actuation can be observed (Fig. 1).

While the thermal back-isomerization for azobenzenes in isotropic solution usually takes several hours, it could be shown in former studies that this process duration is reduced to several minutes by incorporating them into a liquid crystalline matrix.^{27,28} This faster switching behavior makes azobenzenes even better candidates for use in photoresponsive LCEs and many actuators were prepared already from them. Of course the rate of the switching between *trans*- and *cis*-forms of azobenzenes can also be changed by modifying the nature of the chromophoric system.^{29,30} However, so far, mostly densely crosslinked systems, which work in the bending mode to obtain large macroscopic deformations, were investigated^{31–34} and little attention was paid to the expansion/contraction associated with the phase transition. The group of Ikeda produced a light driven engine for example,³⁵ Fischer *et al.* fabricated biomimetic photoresponsive swimmers³⁶ and the group of Broer synthesized artificial cilia, which bend during UV-irradiation.³⁷

We were interested in the fabrication of photoactuating LCE-micro-particles, which were to be produced in a microfluidic capillary based device, and the investigation of their light-responsive properties. To prepare these particles it became necessary to establish a highly efficient photopolymerization of the azo containing monomer mixture with an optimized NIR-initiator system. With the optimized monomer mixture, particles could then be produced, which showed a thermo-responsive actuation of up to 33% and a fast elongation of 24%

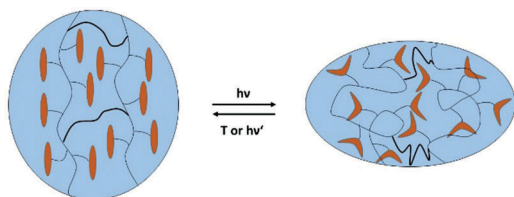


Fig. 1 Schematic drawing of the photoactuation of an azo-LCE.

during irradiation in just a few seconds. Besides a thermal back-deformation we also show that a back-deformation induced by red-light could be achieved. In addition, the swelling behavior of the particles was investigated as a third way of actuation.

Results and discussion

Selection of the initiator system

For the polymerization in a microfluidic device the initiator has to fulfill several criteria. On the one hand a fast curing of the monomer droplets is necessary, so that solid (cross-linked) particles leave the device, while on the other hand the polymerization must not be started prior to droplet formation to prevent clogging of the capillary and tubes. Furthermore, the solid monomer mixture needs to be heated into the liquid crystalline phase before polymerization which prohibits the application of thermally induced initiators. One possible way to meet these requirements is a redox-initiated polymerization, in which both redox components come into contact only during droplet formation.³⁸ A more convenient solution is the use of photoinitiators. Usually UV-photoinitiators are used for this kind of polymerization, but as azobenzenes absorb light strongly in the UV- and blue range of the spectrum, no efficient curing of the mixture is possible. This is, however, necessary, because the particles remain for only a short time in the capillary. In addition, *trans*–*cis* isomerization takes place when azobenzenes are irradiated with UV-light. Thus, the absorption of UV-light might result in a loss of the liquid crystalline phase during the polymerization. To overcome these problems we used 1,3,3,1',3',3'-hexamethyl-11-chloro-10,12-propylene-tricarbocyanine triphenyl-butyl borate (CBC, Fig. 2b) as an initiator which was already described by Keller *et al.* for the synthesis of LCE films made out of the same monomer.^{39,40} Having an absorption maximum of $\lambda = 780$ nm this molecule starts the polymerization by NIR-irradiation. Thus, the wavelength is high enough not to be absorbed by the monomer (Fig. 2a) and no *trans*–*cis* isomerization occurs. As this initiator is already very reactive under dimmed room light, we synthesized it *in situ* (see the Experimental section).

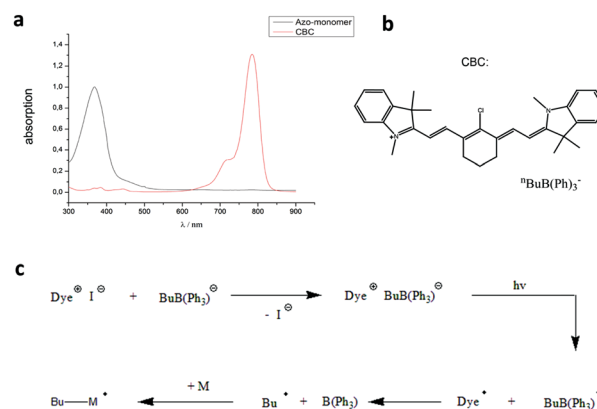


Fig. 2 (a) UV-Vis spectrum of CBC compared to that of the azo-monomer, (b) chemical structure of CBC and (c) assumed mechanism of the initiation step.



Thus, the ion pair of the initiator forms in the reaction mixture and starts the polymerization when the mixture is irradiated with intense NIR/red light. In addition borate V also works as a bleaching agent for the dye so that even thick samples can be cured. The assumed mechanism⁴¹ for the initiation step is shown in Fig. 2c.

Preparation of particles using a microfluidic device

For the fabrication of our photoresponsive LCE particles we used a capillary based microfluidic setup that was already applied for the production of thermoresponsive LCE particles in our group and is shown in Fig. 3a.^{42,43}

In our device the monomer mixture flow through a thin glass capillary, which was positioned along the main axis of a T-junction, while a high viscous continuous phase (we used a silicon oil with a viscosity of 1000 cSt) was pumped through the perpendicular entrance of the T-junction and flew round the capillary. Due to the high shear rates at the capillary's tip, droplets were formed and flew down the polymerization tube (Fig. 3a). For a good control over the flowrates of continuous and monomer phases (dispersed phase) syringe pumps were used to pump the fluids through the capillaries and tubes. As the monomer, we chose a side-on liquid crystalline azo-monomer with the phase behavior (crystalline 53 °C nematic 84 °C isotropic), which was first synthesized in the group of Keller⁴⁴ and was already applied for the preparation of photoactuating films. Besides the monomer the dispersed phase consisted of 10 mol% of a liquid crystalline crosslinker (Fig. 3d), 1.4 wt% of the initiator dye and 3 wt% of Borate V. The resulting mixture showed a nematic phase with a clearing temperature of 72 °C, but the crystals did not melt before they were heated up to 85 °C. This made a modification of the microfluidic device necessary. The monomer mixture was placed in the tubing from the syringe pump to the T-junction and was pumped through the device by a low viscous silicone oil provided by a syringe. Both the tubing and the T-junction were placed in a water bath which was heated to 90 °C so that the monomer mixture was present as an isotropic liquid. The polymerization tube, after leaving the water bath, was laid on a hotplate that was heated to 65 °C. Thus, the monomer droplets

were cooled into the nematic phase before they were polymerized using a HighPower LED ($\lambda = 623$ nm). In the running set-up the droplets typically flew less than 2 seconds under the light spot.

For photoinitiation with visible light we chose the initiation system presented in Fig. 2b and reported by Keller *et al.* in ref. 40. However the system did – at first – not work effectively under the short time span of the microfluidic experiment. Thus, we increased the amount of Borate V (butyryl choline butyltriphenylborate), which also acts as a clearing agent, by a factor of about 2. With this initiator system the irradiation time of 2 seconds was sufficient to decompose the initiator, as could be seen by the disappearance of the initiator's green color. At the end of the tubing the droplets were collected in a small glass vial and found to be solid monodisperse particles (Fig. 3b). So a rather short time period was sufficient to solidify the LC monomer droplet by VIS-initiated radical polymerization.

Investigation of the best parameters for the system

To optimize thermal and photochemical actuation of the LC-elastomers, independent of the processing parameters in the capillary, we produced films of different thicknesses and crosslinker ratios. For this purpose, a few crystals of the monomer mixture were taken on a glass plate, melted at 100 °C and a second glass plate was put on top, separated from the lower one with a spacer foil of varying sizes. Afterwards, it was cooled to 65 °C into the liquid crystalline phase and irradiated with red light (623 nm). During the cooling and polymerization of the sample a horseshoe magnet was put around the setup in order to achieve a good orientation of the mesogens inside the film. It was shown by the group of Zappe that magnetic fields higher than 100 mT are sufficient for this purpose.¹⁰ For the determination of the actuation we cut little pieces of the LCE film (ca. 500 $\mu\text{m} \times 200 \mu\text{m}$) and observed their change in width during the heating above the clearing temperature or during irradiation for about 10 seconds (continuous wavelength at 70 °C). The actuation was then calculated by dividing the length of the film before the deformation (l_0) by the length afterwards (l_1) (actuation = $100\% \times l_0/l_1$).

Next we characterized the thermal parameters of the prepared liquid crystalline polymer. The ESI† (Fig. S5) contains a DSC

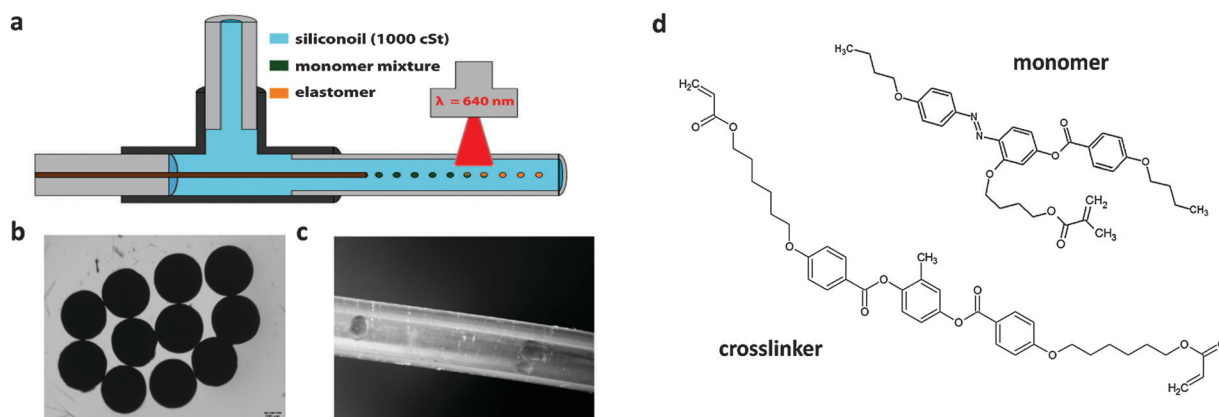


Fig. 3 (a) Schematic drawing of the microfluidic setup, (b) monodisperse LCE particles produced in our device (scale bar: 100 μm), (c) droplet formation at the tip of the capillary in the microfluidic device and (d) chemical structure of the monomer and the crosslinker.



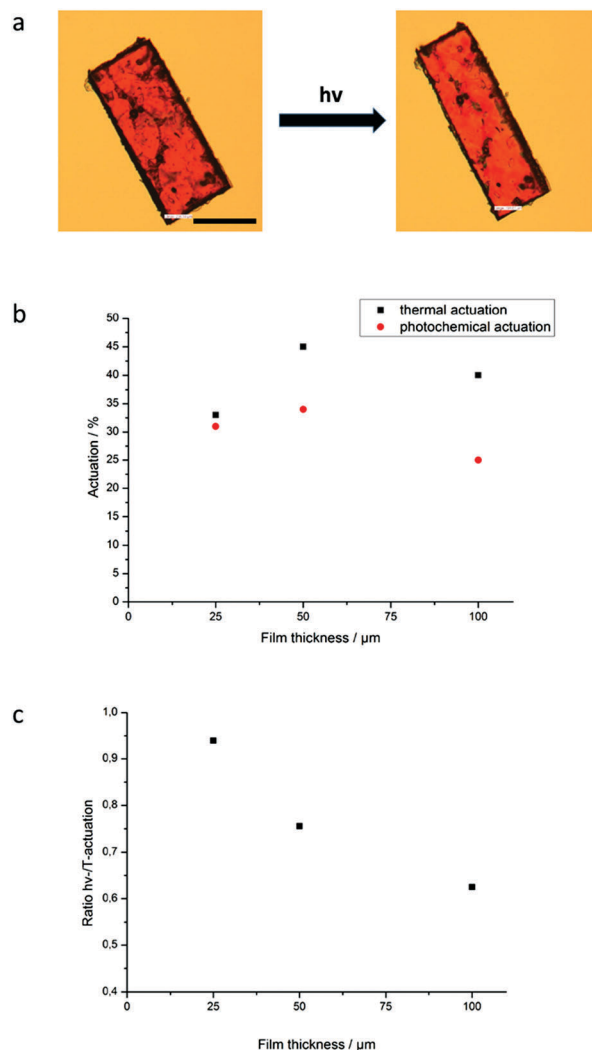


Fig. 4 (a) LCE film before and after irradiation at 70 °C (scale bar: 200 μm), and (b) influence of the film thickness on the actuation and (c) on the ratio between photochemical and thermal actuation (irradiation for the photochemical actuation in (b) and (c) was performed at 70 °C again).

measurement of the polymer prepared. It shows a T_g -value of 37 °C and a clearing temperature (nematic–isotropic) at 90 °C. Thus above 40 °C (or more realistically 50 °C) the crosslinked polymer is in the elastomeric state, and that is where nearly all measurements were done.

One of the actuating films (70 °C) is shown before and after irradiation in Fig. 4a and also a video of this sample is available in the ESI† (S1). For the films, a maximum thermal actuation of 45% and a maximal photo-actuation of 34% by irradiation with a white-light LED (the spectrum is shown in the ESI† (S3)) about 15 °C below the clearing temperature could be achieved. This is, to the best of our knowledge, the strongest response to light of azo-LCEs published in a regular journal till now.

The results for the films of different thicknesses are shown in Fig. 4b and c. For very thin films of just 25 μm thickness thermal actuation is not as good as for the samples with 50 and 100 μm thickness. This is due to a slight rolling of the thin films, which makes actuation less effective. With thermal

actuations around and above 40% the actuation is in the typically range of values for LCEs from side-on fixed mesogens.²

The photo-induced actuation for samples – about 15 °C below the clearing temperature of the all *trans* system – is in a comparable range, but always lower than thermally actuated systems. This is reasonable as *trans* to *cis* isomerization induces in these samples a nearly isotropization of the liquid crystalline phase. Here the achievable actuation is maximal for films of 50 μm thickness and decreasing for thinner and thicker films. For the thin film again slight rolling is observed; for the thicker sample the reduction in actuation is a consequence of the longer distance the irradiation has to take. This reduces the UV-intensity along the path and results in a less strong change in the shape for parts of the sample.

The UV-Vis spectrum of the monomer in solution – before and after irradiation with the white light LED – is shown in Fig. 5. By continuous excitation of the monomer with our lamp the molecules undergo *trans*–*cis* isomerization, which can be seen by the decrease of the $\pi \rightarrow \pi^*$ -band (around 365 nm) of the *trans*-azobenzene. At the same time the $n \rightarrow \pi^*$ band (>450 nm) of the *cis*-conformer increases slightly. Thus, by irradiation with red light the photochemical isomerization of the azo-monomer can be reversed.

Changing the content of the crosslinker in the mixture has a big influence on maximal actuation (Fig. 6). For small amounts of up to 25 mol% the actuation is – with values over 40% – quite stable, but for a high concentration of 50 mol% the deformation of the film is much smaller. With a higher crosslinker concentration the whole film becomes much stiffer and that is probably the reason for a weaker actuation.

For higher crosslinker concentrations also the ratio between the photochemical and thermal actuation decreases strongly. This is logical as the crosslinker does not bear an azo-group and thus cannot contribute to increasing the disorder by bending.

By investigating the parameters of the photoactive LCE films we not only noticed that our system has a stronger photo-responsive actuation than conventional systems, but also has the advantage of showing a fast response to light (full actuation in less than 10 seconds) and moreover is polymerizable in thicker films, a property that is inalienable for the production of particles. In previous works the photopolymerization was

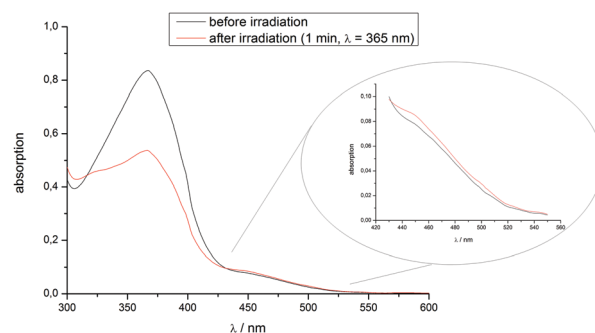


Fig. 5 UV-Vis spectrum of the azo-monomer in solution, before and after irradiation with the white light LED for 1 minute.



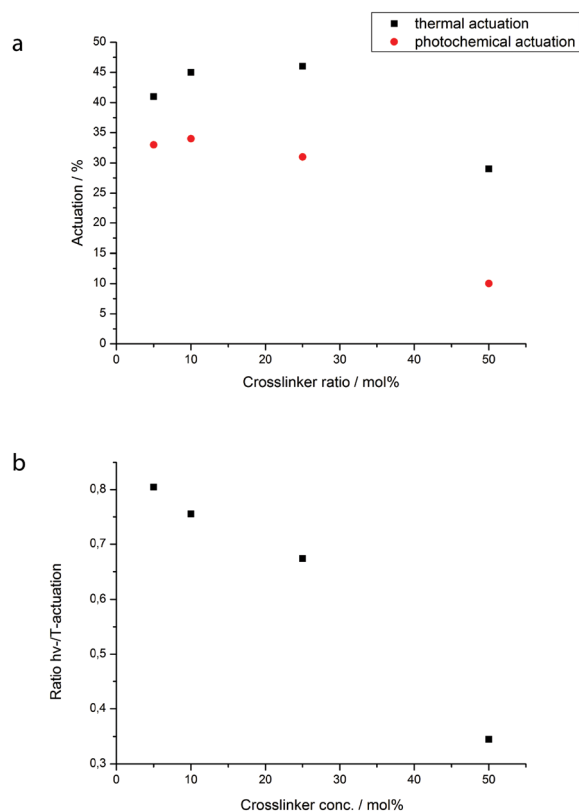


Fig. 6 (a) Influence of the crosslinker concentration on the actuation properties of the film and (b) on the ratio between photochemical and thermal actuation (irradiation for the photochemical actuation in (a) and (b) was performed at 70 °C).

often limited to a sample thickness of a few micrometers, but by adding a higher concentration of the bleaching agent borate V to the initiator mixture this problem could be overcome, as the initiator got decolorized during irradiation.

Another remarkable quality of our films is that they show no bending during irradiation, which is the typical response of densely crosslinked azosystems.^{31,32} Instead they deform uniaxially. This is a big advantage for use in devices since a distortion along just two axes enables the application of distinct higher forces compared to three dimensional bending.

Results of the particle fabrication

Monodisperse LCE particles with diameters of 300–500 μm were produced as described above and then investigated on a precise heating plate under a polarized light microscope. To examine the photoresponsive behavior the particles were heated to a fixed temperature close to the clearing temperature of the all-*trans* sample and then irradiated with a cold-light source (continuous spectrum) to prevent additional warming of the particles. The irradiation was focused by two gooseneck light guides, giving a spot of about 0.5 cm diameter. As a result of their preparation conditions these LCE-particles have a slightly pumpkin like shape and extend along their short axis (see ref. 16). The mesogens in these particles are aligned along a concentric orientation field within the particles with the centers of the concentric rings lying on the axis through the poles.

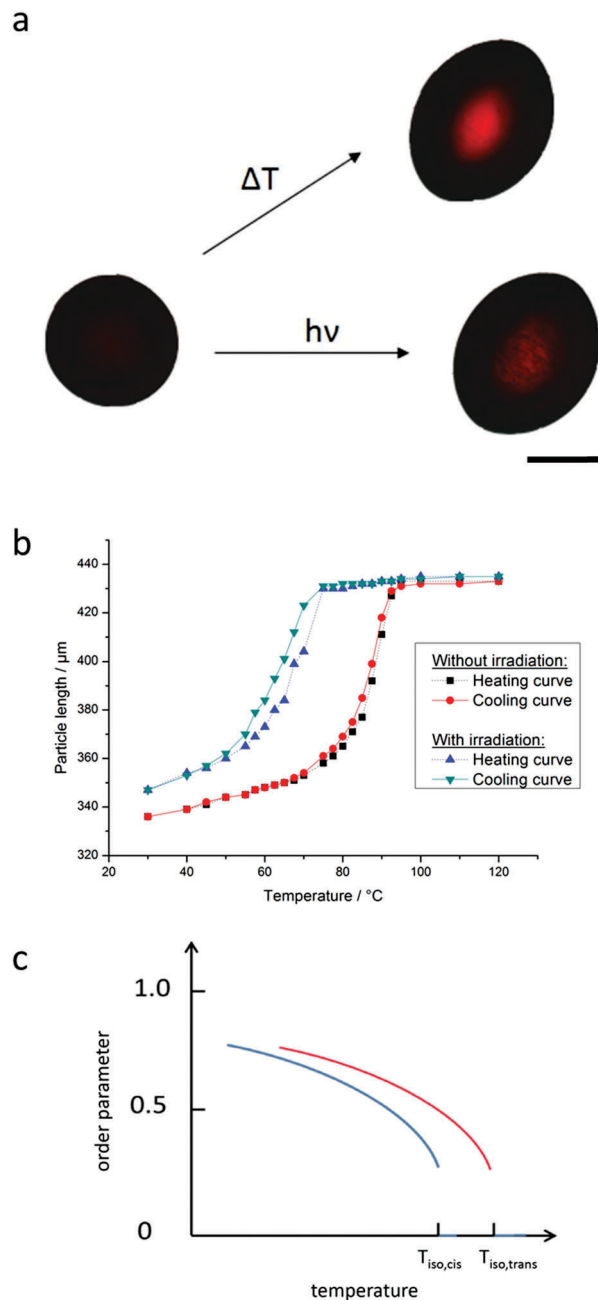


Fig. 7 (a) The two ways of inducing a phase transition and thus a deformation of the particle by either heating above the clearing temperature of the particle or by irradiation at 70 °C with light of a continuous spectrum (scale bar: 200 μm), (b) thermoresponsive behavior of the particle diameter under and without irradiation and (c) schematic drawing of the order parameter's temperature dependence for the LCE with and without irradiation.

Fig. 7a shows the response of one particle to both stimuli. In either case the LCE exhibits a nematic phase at 70 °C before it attains the isotropic phase by the application of the external stimulus. According to the concentric orientation of the mesogens the particle elongates along the axis of its poles during the phase transition and shrinks in the perpendicular direction. The bright spot in the center of the thermally actuated particle indicates a



complete phase transition. Compared to the LCE films the particles showed a smaller change in size, but with a thermal response of up to 33% and a photoresponsive actuation of up to 24% it was still strong. For the latter a video is also available in the ESI† (S2).

The thermoresponsive length change of the particles is shown in Fig. 7b and according to the literature⁴² an s-shaped curve is obtained showing a hysteresis for the heating and cooling curve. Starting from low temperatures the order parameter S of the mesogens in the LCE is expected to get reduced more and more leading to a slight increase of the particle length (see the schematic drawing in Fig. 7c). At the clearing temperature, finally, a drastic loss of orientation occurs (S becomes zero) so that the length of the particles changes strongly by just heating them up a few degrees. Once the LCE is in its isotropic state no further elongation can be observed. By cooling the particle the s-shaped curve is retained and only very slightly shifted to lower temperatures. Furthermore, it can be seen that the hysteresis curve for a particle under irradiation is shifted about 15 °C to lower temperatures. This is a consequence of the lowering of the clearing temperature of the liquid crystalline phase by about 15 °C, induced by the *cis*-isomer created by photoisomerization. The same is valid for the order parameter of the liquid crystalline phase inside the LCE (Fig. 7c). So if the sample in the all-*trans* (dark) state is kept at a temperature not more than 15 °C below the clearing temperature, isotropization and thus a full shape change should be achievable by UV-irradiation (see Fig. 7b).

Also the photoresponsive behavior of the particles was investigated and is shown in Fig. 8. Part a illustrates the photoresponsive actuation as a function of time. By irradiation with the cold-light source at 70 °C the particle elongates linearly before the actuation is completed after about 7 seconds. Then, no further change in the length can be observed for a longer irradiation time. After the light source is shut off the azo-mesogens undergo a thermal back-isomerization to the *trans*-conformer and thus the liquid crystalline phase is formed again which results in a back-deformation of the particle. This return to the original shape takes about 16 minutes at 70 °C as it can be seen in Fig. 7b. As mentioned before, the back-isomerization to the *trans*-form can also be triggered by an excitation of the *cis*-conformer's $n \rightarrow \pi^*$ -band with red light and thus, the necessary time for the back-deformation can be halved to 8 minutes at the same temperature. Here also the cold-light source was used for the irradiation, but with a red light filter (for the transmittance spectrum see S4, ESI†) being held between the lamp and the sample. We also tried to induce back-deformation by the irradiation of the particle with light of a wavelength around the maximum of the $n \rightarrow \pi^*$ -band (450 nm), but surprisingly no back-deformation could be seen at this wavelength. This observation was also made by the group of Keller, who experienced the same result.⁴⁰ We assume that this is due to an equal absorption of the *trans*- and *cis*-conformer at this wavelength. In contrast at wavelengths above 550 nm the absorption of the *cis*-form is indeed weak but probably stronger than the absorption of the *trans*-form. In part (c) of the figure the actuation is shown as a function of temperature during the irradiation. *trans*-*cis* isomerization in photodynamic equilibrium

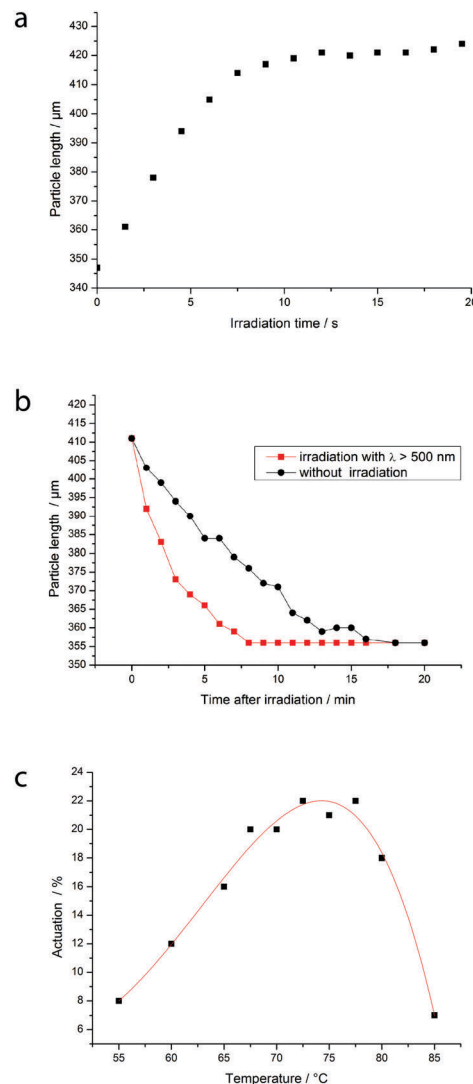


Fig. 8 Photoresponsive behavior of the LCE particles. (a) Actuation as a function of time (at 70 °C), (b) thermal and photochemically induced back-deformation as a function of time (also at 70 °C) and (c) the photoresponsive actuation for different temperatures during irradiation.

as a result of the irradiation leads in the neat nematic phase of our system only to a decrease of the clearing temperature. Thus, at temperatures far below the clearing point just a reduction of the mesogens' order can be achieved which engenders just a small deformation of the LCE. Hence, the actuation increases the closer the temperature is to the clearing temperature during irradiation. Very close to the clearing temperature of the all-*trans* system much of the deformation has already been done in a thermally actuated manner. Thus the contribution of photoactuation gets small again. At intermediate temperatures the effect of photoactuation is maximal. These results match well with the observations of Fig. 7.

In conclusion it could be shown that a fast and remarkable photoactuation can be achieved by irradiation of our LCEs at temperatures close to the clearing point. Drawbacks are the distinctly slower back-deformation and that no actuation at room temperature can be achieved with this system.



Swelling of the particles as a third stimulus

Beside the thermal and photochemical actuation, swelling in a good solvent can also destroy the liquid crystalline phase inside the particle and thus induce anisotropic deformation. As shown in Fig. 9 the crosslinked LCE-particles do not just grow to a multiple of their original size, but also elongate along the same axis – as if they were heated up – when they are swollen in chloroform. Once the solvent around the particle is evaporated the particle shrinks again. However at room temperature they do not relax back to their original shape. This occurs probably because the material of the shell of the particle is too close to the glass transition temperature and hard, when the interior of the particle turns back to the nematic phase due to evaporation of all the solvent. That is the particle as a whole lacks mobility to adopt the original shape.

By heating the particle to 110 °C the mobility to reach the thermodynamically favored conformation is regained, but as the sample is now in the isotropic state it remains elongated. But by cooling the particle slowly down to room temperature, it regains its original shape.

Experimental section

Materials and reagents

Chloroform, dichloromethane and silicone oil (100 cSt and 1000 cSt) were purchased from Sigma-Aldrich. Butyryl choline butyltriphenylborate (Borat V) was purchased from Spectra group limited. The liquid crystalline monomer (4-butoxy-2'-(4-methacryloyloxybutoxy)-4'-(4-butoxybenzoyloxy)azobenzene),⁴⁵ the crosslinker (1,4-bis (4-(6'-acryloxy-hexyloxy)benzoyloxy)-2-toluene)⁴⁶ and the dye for the initiator (1,3,3,1',3',3'-hexamethyl-11-chloro-10,12-propylene-tricarbocyanine iodide)⁴¹ were synthesized according to the literature. Components for the microfluidic device were obtained from WICOM (PTFE tubings) and Postnova Analytics GmbH (fused silica capillary, T-junction, nuts and ferrules).

A red light HighPower LED (623 nm) was purchased from Conrad and the coldlight source KL 1600 (spectrum see S3, ESI†) from Schott.

Preparation of polymer films

The monomer mixture used for the film syntheses was prepared by mixing the monomer, the crosslinker (10 wt%, except for the experiments with varying crosslinker concentrations), 1.4 wt% 1,3,3,1',3',3'-hexamethyl-11-chloro-10,12-propylene-tricarbocyanine iodide and 3 wt% Borate V in dichloromethane. The solvent was removed under vacuum. A few crystals of the monomer mixture were melted at 100 °C on a glass slide which was laid on a precision hot plate (Harry Gerstig GmbH) in the magnetic field of a horse-shoe magnet ($B > 100$ mT). A second glass slide was put on top, separated from the lower one with spacer foil of varying sizes (25 µm, 50 µm, 100 µm and 250 µm). Then the mixture was cooled to 65 °C into the liquid crystalline phase and irradiated with red light (623 nm) for about 5 seconds. After cooling to room temperature the glass slides were separated with a scalpel, with the film just sticking to one of them. For further investigation, we cut little pieces out of the LCE film (ca. 500 µm × 200 µm).

Preparation of the particles

Fig. 3a shows a schematic drawing of the microfluidic setup. A glass capillary (inner diameter (ID): 100 µm) was inserted inside a T-junction along its main axis. The capillary's tip, where the droplets are formed, ended in an outlet polytetrafluoroethylene (PTFE) tubing (ID: 750 µm) and the continuous phase (silicone oil, 1000 cSt) was provided by a tubing of the same material (ID: 500 µm) through the perpendicular entrance of the T-junction. For a good control over the flow rate, a syringe pump (Pump 33, Harvard Apparatus) was used to pump the monomer phase at stable flow rates through the setup. The capillary, which provided the monomer mixture, began in another PTFE tubing (ID: 170 µm) on the side of the T-junction opposite to the polymerization tube. Besides the syringe pump, the setup was placed in a water bath that was kept at 90 °C. Between the syringe and the capillary a piece of a PTFE tube (ID: 1/16 inch) filled with the monomer mixture was plugged, so that the monomer mixture was present as a liquid. To pump it through the capillary a low viscous silicone oil (100 cSt) was used *via* the syringe. The polymerization tube was laid on a precision hot plate that was heated to 65 °C. Thus, the monomer droplets were cooled down into the nematic phase before they were polymerized by the light of a HighPower red light LED (623 nm) "on the fly" inside the tube (time under the light spot: ca. 2 seconds). After passing the polymerization tube (length: 30 cm) the crosslinked particles were collected in a small glass vial.

Characterization

The droplet formation at the tip of the capillary was observed using a Zeiss stereomicroscope Stemi 2000-C equipped with an Olympus XM 10 camera. For the investigation of the actuation of films and particles, we used the light microscope Olympus BX51 equipped with a hot-stage (Linkam TMS 94) and the camera Olympus ColorView II. The sizes of the samples were

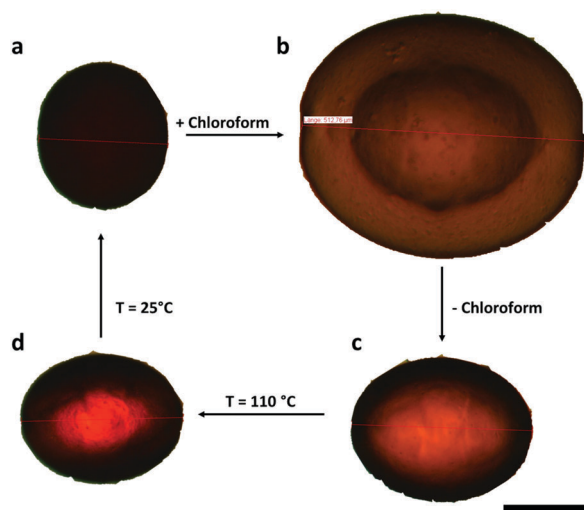


Fig. 9 Swelling behavior of the LCE particles: (a) before any treatment, (b) after swelling in chloroform, (c) after evaporation of surrounding chloroform and (d) after heating of the particle to 110 °C (scale bar: 200 µm).



analysed using the imaging software Cell[^]D. For the irradiation experiments, the sample was heated to the desired temperature and then irradiated with the light of a white-light LED of the coldlight source KL 1600. Light guides were used to focus the irradiation upon the sample giving a light spot with a diameter of about 1 cm. For the actuation of the samples (*trans*–*cis* isomerization) the light source was used without the use of any filters and for the backactuation (*cis*–*trans* isomerization) the same lamp was used, but with a red filter (OG 590 Schott, for the absorption spectrum see S4, ESI[†]) being held in the light beam. UV-Vis spectra were recorded using a Jasco spectrophotometer V-630 in dichloromethane.

Conclusions

For the first time micrometer-sized LCE particles showing a photoresponsive deformation could be synthesized. A microfluidic device could be used to fabricate a high number of these monodisperse actuators in a continuous flow by photopolymerization with red light. For the investigation of the best fabrication parameters LCE films were also produced. These samples showed a thermoresponsive actuation of more than 40% and a fast response of 34% in length change by irradiation with light of continuous wavelength. This is, to the best of our knowledge, the strongest light induced length change (deformation) of azo-LCEs published in a regular journal so far. Even films of several hundred micron thicknesses could be produced by using a bleaching agent in the initiator system. It was remarkable that the films did not show bending but a uniaxial deformation. The then produced particles were examined with respect to their photo- and thermoresponsive behavior. It could be shown that the particles deformed when they were heated above the clearing temperature or when they were irradiated with light of continuous wavelength at temperatures close to the phase transition. Complete light induced deformation of the particles took place in just 8 seconds and could be reversed in the dark or by irradiation with red light. In addition, swelling of the particles in a good solvent was found to be a third stimulus to actuate the particles.

With this work we contribute to an improvement and a better understanding of photoresponsive LCEs. If the drawback of a slow back deformation can be overcome this system will be a good candidate for use as artificial muscles in microrobotics.

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