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Mechanical pretreatments of lignocellulosic biomass: towards facile and environmentally sound technologies for biofuels production

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The transformation of lignocellulosic biomass into biofuels represents an interesting and sustainable alternative to fossil fuel for the near future. However, one still faces some major challenges for the technology to be fully realized including feedstock costs, novel pretreatment processes, production, transportation, and environmental impact of the full chain. The development of new technologies focused to increase the efficiency of cellulose conversion to biofuels determines successful implementation. Mechanical fractionation is an essential step in order to increase final carbohydrate output, appropriate particle sizes and densification, enzymatic accessibility, and bioconversion affectivity without the production of toxic side streams. In this review article, we surveyed a substantial amount of previous work in mechanical fractionation or pretreatments of a variety of lignocellulosic biomasses; these include numerous milling schemes and extrusions, and their impacts on the physical and physicochemical properties of the lignocellulosic matrix (crystallinity, surface area, particle size, etc). We have also compared results with other pure chemical and physicochemical pretreatments in order to show the new aspects and advantages/disadvantages of such an approach. Last, but not least, the effect

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of mechanical treatment and physical properties on enzymatic hydrolysis and bioconversion has been discussed, with potentially interesting dry lignocellulosic biorefinery schemes proposed.

1. Introduction

Mechanical size reduction is a crucial step for the transformation of feedstock into energy and polymer biomaterials in the field of bio-based products (bioenergy and biomaterials) from renewable biomass resources.1-4 Size reduction has many advantages: (i) it increases the volume calorific value of biomass and simplifies the densification processes,5 (ii) it simplifies the supply chain of raw materials,6 and their storage conditions, (iii) it increases the total accessible surface area and, thus, improves the bio-accessibility of constituents⁷ and the conversion of saccharides during hydrolysis and (iv) it reduces the



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Rafael Luque has extensively contributed to the areas of biomass and waste valorisation practises to materials, fuels and chemicals as well as nanoscale chemistry over the past 10 years, with >220 research articles, 3 patent applications and 7 edited books as well as numerous contributions to book chapters keynote and plenary and lectures in scientific events worldwide. Among recent

awards, Rafael has received the RSC Environment, Sustainability and Energy Early Career Award (2013), 2013 Distinguished Engineering Fellow from CBME at HKUST in Hong Kong and currently honoured as Chinese Academy of Sciences Visiting Professor at the Changchun Institute of Applied Chemistry. Rafael combines his academic duties with his activities as young entrepreneur after cofounding Green Applied Solutions S.L.U in Cordoba, Spain and more recently involved in Posidonia Oceanica S.L.

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Cellulose-Hemicelluloses 🌠 lignin Plant Structure Lignocellulosio biomass 500-100um 50-30μm <1µm m. cm Mechanical Coarse Intermediate Fine **Ultra-fine** Nano-Grinding Operations Cutting milling micronozations Grinding Grinding

The different mechanical operations for size reduction of constituents related to plant structure

or crushing

mass and heat transfer limitations during the hydrolysis reactions8 and consequently reduces energy inputs.1 We can distinguish different types of size reduction that are generally differentiated, like cutting or crushing (meter to centimeter range in size), coarse milling (cm to mm, cm to 500 µm), intermediate micronization (cm to 100 µm), fine grinding (<100 μm), ultra-fine grinding (<30 μm) and nanogrinding (<1 μm).1,9 However, nanogrinding could only be achieved through wet grinding which is not addressed in this contribution due to the associated energy consumption, particularly to dry biomass after the grinding step, far too substantial to consider it a worthwhile pretreatment step (Fig. 1).

The reduction of raw material size is achieved using a combination of different mechanical stresses such as impact, compression, friction, and shear (Fig. 2) – all may coexist in one commercial equipment. 1,10,11 For example, in a jet mill, the particles are projected against each other in an air stream; major mechanical stresses generated are impact and friction between particles (Fig. 2). Different mill tools are used to fragment and dissociate lignocellulosic biomass: knife mill, hammer mill, pin mill and centrifugal mill, which consist of a rotor driving different tools. The rotor speed is generally adjustable. A sieve or a screen allows control of the particle size of the final product. These mills generate more impact and shear. In ball mills including vibratory

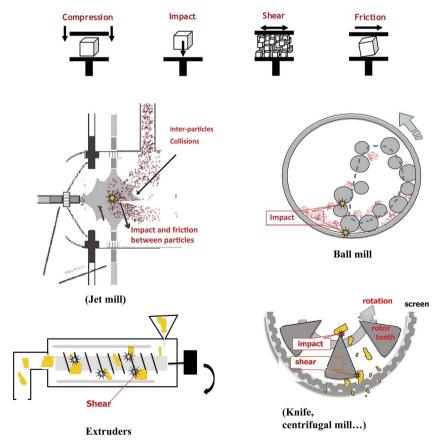


Fig. 2 Schematic representation of some commercial milling equipment with the different mechanical stresses generated.

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ball mill and tumbling ball mills (or planetary ball mills), the raw materials suffer impact and compression stresses when collisions between balls and walls occur. Finally in an extruder, the main mechanical stress is shear occurring between the screw and the walls of the extruder. The choice of equipment depends on many parameters: physical and chemical properties of the biomass, the moisture content, final particle size, the particle size distributions and application targets. Colloid mills and extruders are suitable only for comminuting wet materials with moisture contents over 15-25%, whereas hammer and knife mills are suitable to pretreat dry biomass with moisture contents up to (10-15%). 1,10 Extruders, in comparison with disc and ball mills, have advantages in terms of continuous processing, easy adjustment on-line, and usage in large-scale applications with high throughput. The fluidized bed as superfine grinder has been widely used in various industrial fields for its excellent ability to improve the surface area and enhance the bioavailability of the materials through micronizations, without sacrificing the natural physical-chemical proprieties of the materials.12-14

The energy requirement in relation to final particle size is one of the most important economical parameters in the choice of milling equipment. It mainly depends on (i) machine specifications such as motor speed, (ii) storage capacity of the milling chamber, (iii) material throughput characteristics, (iv) initial biomass structure and physical-chemical proprieties (moisture content, chemical composition, tissue composition, postpretreatment etc.); and (vi) particle sizes. 1,4,6,10,15 However, the equipment could also be selected for steering the reactivity of biomass. As an example, several studies have shown that BM could be described as a mechanical-chemical treatment because the prolonged milling efficiently breaks chemical bonds between lignin and hemicelluloses,16 decrease particle size,17,18 decrease the CrI (from 69.9 for raw wheat straw to 23.7 after a BM step), 19 increase enzymatic hydrolysis16,17 and increase the SSA (from 0.64 for raw wheat straw to 2.3 m² g⁻¹ after a BM step).¹⁹

In this review, we address unique features of extrusion and mechanical size reduction as mechanical pretreatment in lignocellulosic biorefineries. First of all,

bioconversion pathways of lignocellulosic materials and we discuss the effect of mechanical treatment compared to the purely chemical and physicochemical treatments with respect to surface area (in relation to enzymatic accessibility) and CrI. In the second part, we discuss the effect of mechanical treatment on enzymatic hydrolysis and the factors that can influence the performances of enzymatic hydrolysis and bioconversion.

Bioconversion of lignocellulosic biomass: from heterogeneous particles to biofuels

The bioconversion of lignocellulosic biomass has been extensively studied in the past 30 years. In spite of such research endeavors, enzymatic degradation of lignocellulose is still poorly understood because of competing effects including physical properties of the substrate, enzyme synergy and mass transfer. The structural heterogeneity and complexity of cell wall constituents such as crystallinity of cellulose microfibrils, specific surface area of particles and matrix polymers are responsible of the recalcitrance of cellulosic materials (Fig. 3).

Biomass pretreatment is consequently an essential step in order to increase its final carbohydrate output, accessibility, bioavailability and hydrolysis rate (Fig. 3). The objective of pretreatments depends on the process type and biomass structure. For instance, pretreatments aimed to produce biofuels target changes in lignocellulosic matrix properties to make the holocelluloses more accessible to enzymatic attack.20-25

Pretreatment methods can be divided into different categories: mechanical, chemical, physicochemical and biological or various combinations of these. Mechanical pretreatments allow the separation of the main botanical parts of the crop into different fractions (tissues, cell, polymers, etc.), to be used as feedstock for various applications. Such pretreatment greatly reduces biomass particle sizes and possibly affects its molecular structure to facilitate enzymatic accessibility. Palmowski and Muller²⁶ have studied the effect of mechanical operation on

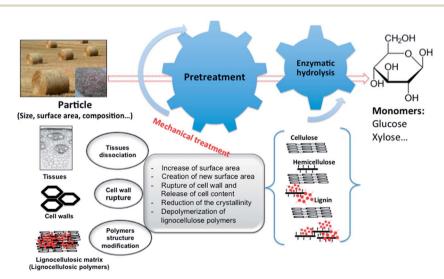


Fig. 3 Different steps of biomass conversion and parameters influencing lignocellulosic particle reactivity.

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different organic samples (apples, rice, sunflower seeds, hay and maple leaves).26 After breakdown of these substrates, a significant particle size reduction was observed which is believed to due to the release of soluble organic compounds in solution (cells destroyed through commination and/or dissolution of organic components through newly generated accessible surfaces).26 The reduction of particle size could enhance the affinity between cellulose polymers and enzymes and thus increase the rate of hydrolysis. The rate can be doubled in a 10 h reaction experiment when the average size of cellulose is reduced from 82 to 38 μm.²⁷ Size reduction also enhances the production of glucose or reducing sugars as illustrated by studies showing a reduction in particle size from 590 to 33 µm resulted in a 55% increase in glucose production after 72 h cellulose hydrolysis.28 It appears that size reduction is an attractive method to increase the yield of hydrolysates from lignocellulosic biomass, especially on the saccharification of plant cell walls by cellulotic enzymes.²⁹ Small particle sizes of untreated cellulosic substrate are more readily hydrolyzed as compared to large ones due to their higher specific surface area which gives the enzymes more surface to interact with the substrate during conversion. These hypotheses were tested using ground corn stover in the size ranges of 425-710 and 53-75 µm.29 The morphology changes in these particles were imaged after treatment with cellulolytic enzymes before and after liquid hot water pretreatment. The smaller corn stover particles of 53-75 µm were 30% more susceptible to hydrolysis as compared to larger 425-710 µm corn stover particles.

Mechanical fractionation is a necessary step in lignocellulosic bioconversion to:

- (i) Decrease particle size and increase total accessible specific surface area (SSA).
 - (ii) Decrease cellulose crystallinity.
- (iii) Increase pore size of particles and the number of contact points for inter-particle bonding in the compaction process.
 - (iv) Dissociation of tissues and rupture of cell wall.

All these parameters improve the digestibility and the conversion of saccharides during hydrolysis. 1,21,23,25,30,31

2.1. Effect of mechanical pretreatment on surface area (SSA) and porosity

The physical properties of the cell wall including its surface specific surface area and porosity play an important role in chemical and biological digestion.³² The study of lignocellulosic porosity and specific surface area is a subject of relevant research since molecular probes have been introduced to advance on such important parameters for the conversion of lignocellulosics to fuels and chemicals. Freeze fracture electron microscopy, solute exclusion, mercury porosimetry, electron microscopy (SEM and TEM), NMR, gas adsorption and related tools have been utilized in this regard.^{32–35} The methods conducted to characterize the porosity and specific surface area (SSA) should be carefully considered.

In general terms, the specific surface area and porous texture of biomass can be measured by the adsorption/desorption of a gas (generally N_2), 32,34,36 and/or intrusion/extrusion of mercury inside the porous texture of the material (mercury porosimetry).

Mercury porosimetry can also determine pore sizes larger than 3 nm by using a new generation of automatic mercury porosimeters based on Washburn theory.37 Unfortunately, both techniques have their own limitations. Textural porosity (soft or hard) is only identified by adsorption/desorption of gas at the temperature of nitrogen liquefaction (77 K). Since the sample is previously degassed, several problems could be encountered during the analysis of biomass using nitrogen physisorption, leading to non-representative results. By comparing the effect of solvent polarity and drying temperature of materials, these factors were found to influence the value of BET (theory Brunauer, Emmett and Teller) measurements.38 Appropriate solvent drying procedures should also be followed to better maintain capillary structure.38-40 The pore volume measured by this technique (adsorption/desorption of gas) is not sufficiently precise for samples containing macropores (pore size greater than 50 nm, corresponding to the relative pressures $P/P_0 > 0.98$, following the Kelvin equation). Lastly, another disadvantage of the nitrogen physisorption relates to the issues to quantify narrow micropores (less than 0.9 nm). In this case, other gases such as argon, CO2 or krypton should be used. Mercury porosimetry is comparatively limited to materials with pore diameters under 3 nm. For samples containing micropores and mesopores, the technique adsorption/desorption of N2 is sufficient to characterize the porosity (SSA and pore size). In the case of materials with macropores and mesopores (greater than 3 nm), the determination of the specific surface area, as well as pore size distribution must be conducted by analyzing the curves of mercury porosimetry. For materials with a hierarchical porosity (micro-, meso- and macropores), the porosity can be determined by combining nitrogen physisorption and mercury porosimetry.41

In general, the purpose of comparing surface areas of materials subjected to different pretreatments can be useful to ascertain whether the pretreatment technology is useful or not. Fig. 4 illustrates the SSA of some treated lignocellulosic substrates, with values within the same range (1-22 m² g⁻¹) among all parameters. Interestingly, milling pretreatment was found to give a larger SSA for both types of lignocellulosic materials (Fig. 4). SSA increased for more than 60% tor wheat straw after BM treatment (4 h) in comparison to untreated and steam exploded samples.36 Piccolo et al.;36 found that a more severe steam explosion (SE) treatment (0.2% w/w H₂SO₄ + SE 210 °C for 10 min) increased SSA from 1.9 to 3.1 m² g⁻¹. On the other hand, Miao et al.;6 investigated the mechanical size reduction of miscanthus and switchgrass using a commercialscale hammer mill. SSA of resulting particles were measured using the geometry and density of particles.6 This method yielded accessible SSA of ca. 20.5 m² g⁻¹, over 5 times increased under hammer milling treatment as compared to a control experiment and steam explosion treatment. Moreover, BET results showed a relatively low SSA for pretreated lignocellulose. As seen in Fig. 4, SSA of treated miscanthus and switchgrass is highly sensitive to particle size since it increases linearly with increasing particle size. Evidences offered by Zhang et al.;42 support this observation; the authors reported a linear correlation of SSA with pan-milling cycles for cellulose powder as a

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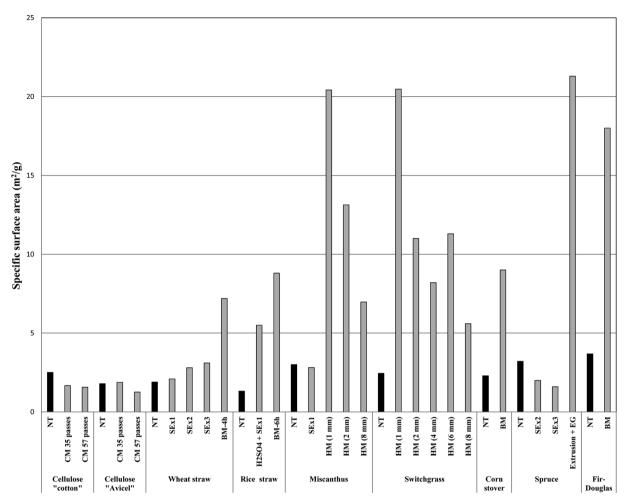


Fig. 4 Effect of mechanical size reduction on specific surface area (SSA) of selected lignocellulosic biomass compared to others pretreatments. NT: not treated; CM: compression milling; BM: ball milling; HM: hammer milling; SEx: steam explosion; EG: ethylene glycol.

consequence of particle size. 42 These findings indicate that the bundle separation and breakage of fibers into small particles result to larger surface areas at the cut edges.42 In a separate study, Miao et al.;6 reported fairly low SSA of switchgrass particles pretreated using a 4 mm milling screen (8.2 m² g⁻¹) as compared to those passed through a 6 mm milling screen (11.3 m^2 g^{-1}). The same phenomenon was observed during compression-milling of cotton and avicell cellulosic materials.43 The compression-milling resulted in a slight drop in SSA compared to the controls of various untreated cellulose samples (Fig. 4). A comparison between compression-milled cellulosic materials showed only small differences in SSA. BET methods provided 1.6 m² g⁻¹ (treated cotton cellulose) as compared to 2.5 $\text{m}^2\text{ g}^{-1}$ for the control, and 1.3 $\text{m}^2\text{ g}^{-1}$ for treated avicell cellulose compared to 1.8 m² g⁻¹ for the control after 57 passes. Particle agglomeration and collapse of capillary structure due to compressive force may be plausible explanations for these phenomena.

BM is considered to be the most effective treatment in increasing SSA. In comparison, the duration of BM treatments of rice straw, corn stover, and fir-Douglas wood significantly increased SSA from 1.3 to 8.8 m² g⁻¹ for rice straw, 2.3 to 9.0 m 2 g $^{-1}$ for corn stover and 3.7 to 18 m 2 g $^{-1}$ for fir-Douglas wood. Extrusion is an alternative method reported to increase the surface area of biomass. Piccolo et al.;36 reported that the fibrillation of spruce wood performed under continuous extrusion at 40 °C and 120 °C in the presence of ethylene glycol significantly increased SSA from 3.2 to 21.3 m² g⁻¹. However, the SSA remained close to 2 m² g⁻¹ when spruce wood was treated under SO₂-steam explosion (2.5% w/w SO₂) at 210 °C for 5 min. The extrusion process largely increases SSA of corn stover biomass for enzymatic adsorption.44 Scanning electron microscopy studies showed that the cellulose network was changed due to the destruction of the lignin sheath. The reported extrusion protocol was a continuous and cost-effective pretreatment method, combining heating with high shear and mixing opening cell walls at the microscopic scale. SSA of pretreated corn stover was significantly improved with respect to control studies, revealing that extrusion could open the cell wall at the microscopic scale (particularly favorable to obtain higher sugar yields). Karunanithy and Muthukumarappan⁴⁵ studied the effect of extrusion parameters on switchgrass bioconversion. The optimum pretreated switchgrass exhibited a 50% higher surface area as compared to those of control

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experiments. These data demonstrate that mechanical treatment or fractionation are most effective methods in increasing specific surface area and improving the efficiency of cellulose digestion. The conflicting results may be due to different methods used to determine accessible surface area and on the type and size of enzymes used.

2.2. Effect of mechanical pretreatment on cellulose crystallinity (CrI)

Intensive pretreatment in polymers is able to cause destruction of macromolecular chains, which leads to the separation of structural constituents. The displacement of structural elements of polymeric chains is firstly accompanied by the distortion of initial chain packing and loss of ordering. It is widely accepted that highly crystalline cellulose is less accessible to cellulase attack as compared to amorphous cellulose. Crystallinity negatively affects the efficiency of enzymatic contact with cellulose. The measurement of crystallinity index (CrI) in lignocellulosics is not a simple task due to the heterogeneous nature of lignocellulosic feedstocks and the contribution of other amorphous components such as lignin and hemicelluloses.

The crystallinity of cellulose in untreated and treated substrates as measured by X-ray diffraction (XRD) is given in

Fig. 5. Data demonstrate that mechanical treatment significantly reduces the crystallinity of lignocellulose biomass, while several chemical and physicochemical pretreatment increase the crystallinity index (CrI) in comparison with untreated substrates. The effect of compression-milling on the crystallinity of cellulose "cotton and avicell" is highly significant. 43,46 After 35 milling cycles, CrI decreased from its original value (84 and 81) to 75 and 17% for cotton and avicell cellulose, respectively. Ouajai and Shanks47 have studied a BM process on cellulose to determine its effect on the crystallinity index. CrI remained almost unchanged during the first 60 minute BM cycles and started to decrease after 100 min. Final cellulose products with a CrI of 36% (330 minute BM treatment) was obtained.47 In a separate study, CrI of poplar wood was found to decreased by 12% after an 8 day BM process. 15 In the initial milling stage (<5 days), CrI decreased rapidly, while it only slightly decreased (or even increased) after 5 days due to agglomeration of fine particles. Chemical and physicochemical pretreatments have comparably reported to provide an increase of CrI for wood cellulose as well as several different biomass feedstocks (Fig. 5). Da Silva et al.;18 compared the effectiveness of BM and WDM in decreasing CrI of sugarcane straw and bagasse. CrI was found to decrease after 30 and 90 min BM of bagasse (from 30% to 20% and 0% respectively)18 which can be

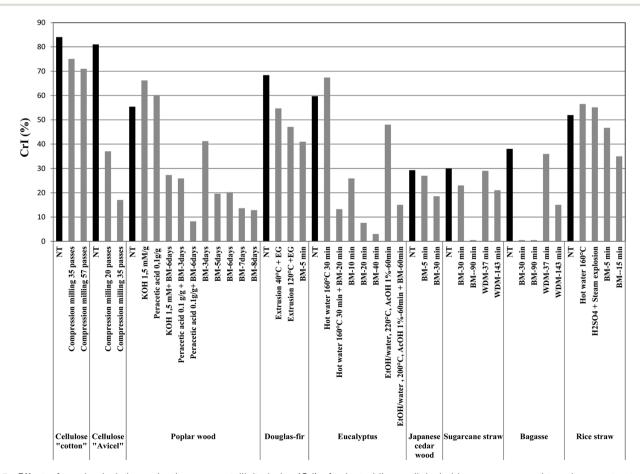


Fig. 5 Effect of mechanical size reduction on crystallinity index (CrI) of selected lignocellulosic biomass compared to others pretreatments (Table 1). NT: not treated; BM: ball milling; WDM: wet disc milling; HM: hammer milling; SEx: steam explosion; EG: ethylene glycol.

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attributed to the complete transformation of crystalline cellulose of sugarcane straw to amorphous cellulose, leading to a significant increase in the extent of enzymatic saccharification. However, CrI decreased slightly in sugarcane straw (28 and 21%) upon WDM treatment after 37 and 147 min, respectively. BM shows a much higher efficiency in vitrifying crystalline cellulose. A similar effect has been observed when bagasse underwent BM for 90 min, decreasing CrI from 38 to 0%. Comparatively, BM treatment of rice straw reduces CrI from an initial value of 52% to 12% after 60 min milling.48 The fibrillation of Douglas fir wood performed by continuous extrusion at 40 °C and 120 °C in the presence of additives (ethylene glycol) decreased CrI from 68 to 54 and 47, respectively. CrI also decreased to 41% upon BM treatment (5 min) of Douglas fir wood.49 These results show that a reduction in crystallinity index depends not only on the initial crystallinity, but also on parameters including the supramolecular organizational structure and the degree of polymerization of cellulose as well as the different mechanical pretreatment (ball-, disk, compression-milling, number of passes, time, etc). The increase in crystallinity is generally proportional to the quantity of solubilized substances during the pretreatment process and could be due to the removal of amorphous regions. With proofs and reports evidencing the influence of CrI on enzyme digestibility, future pretreatment methods should pay attention to the reduction of crystallinity in lignocellulosic materials to maximize fractionation and product yields. One way to effectively decrease the crystallinity of lignicellulosic biomass relates to ball-milling pre-processing. In the next section, this review will discuss key findings and observations with further explanation and theories behind these important topics.

Effect of mechanical treatment on the efficiency of enzymatic hydrolysis

The rate and extent of enzymatic hydrolysis of lignocellulosic biomass highly depend on enzyme loadings, time and rate of hydrolysis as well as structural features resulting from pretreatments. The influence on biomass digestibility and structural properties varies with changes in enzyme loading, time of hydrolysis and other structural features. Important parameters affecting the hydrolysis include previously discussed factors such as the crystallinity of the material, the accessible surface area, the particle size and lignin distribution. Table 1 compiles a body of data on monomeric sugars yields after enzymatic hydrolysis or EHGY (enzymatic hydrolysis glucose yields) varying with t structural features of substrate, postpretreatments, mechanical operation and time.

3.1. Milling or mechanical size reduction process

Zhu et al.;3,4 investigated the impact of disk-milling (DM) conditions on the efficiency of enzymatic cellulose saccharification of softwood. EHGY was increased to more than 92% in glucan after 48 h enzymatic hydrolysis when wood chips were pretreated by post-SPORL (Sulfite Pretreatment to Overcome Recalcitrance of Lignocellulose) DM (disk-plate gap of 0.76 mm)

with 2.21% acid charge (Table 1).3,4 Similar effects on EHGY were also achieved using large disk-plate gaps (up to 1.52 mm). The same authors also studied the effect of lignin, acetyl content and biomass CrI on the digestibility of poplar wood with various enzyme loadings using post-chemical DM. Lignin content and biomass crystallinity dominated digestibility, but most importantly lignin content was not relevant for digestibility at low crystallinity. Poplar wood treated with KOH contained 1.5 mmol g⁻¹ of wood and peracetic acid 0.1 g g⁻¹; this resulted in a EHGY of ca. 18% and 6% with a cellulase loading of 2 FPU per g substrate. Cellulose conversion was only improved upon pretreatment with KOH and peracetic acid. Optimum conditions for the enzymatic hydrolysis of cellulose (86%) were achieved using peracetic acid 1 g g^{-1} followed by ball milling for 6 days. Mais et al.; 50 studied enzyme hydrolysis on α-cellulose as a model substrate and SO2-impregnated steam-exploded Douglas-fir wood chips. 50 The softwood-derived substrate was further post-treated with hot water and alkaline hydrogen peroxide to remove over 90% of the original lignin (Table 1). The experiments were evaluated under different reaction conditions, including substrate concentration, enzyme loading, reaction volumes and number of ball beads employed during mechanical milling. The authors demonstrated that the use of BM could improve the overall conversion of α -cellulose by approx. 12%, increasing the total hydrolysis yield from 70 to 82%. It was apparent that the best conditions for the enzymatic hydrolysis of α-cellulose were achieved using a larger number of beads, while the presence of air-liquid interfaces did not seem to affect the rate of saccharification. Similarly, when lignocellulosic substrates were employed, up to 100% hydrolysis was achieved with a minimum enzyme loading (10 filter paper units per g of cellulose) at lower substrate concentrations and with a greater number of reaction beads during milling. These findings support the importance of the simultaneous combination of BM and enzymatic hydrolysis to enhance saccharification rates and/or a reduction in enzyme loading required to attain total hydrolysis of the lignocellulosic matrix.

Inoue et al.;51 examined the enzymatic digestibility of eucalyptus following a combined pretreatment, without using chemicals, comprising ball milling (BM) and hot-compressed water (HCW) treatment. BM treatment simultaneously improved the digestibility of both glucan and xylan, being also effective in reducing enzyme loading as compared to HCW treatment.51 The combination of HCW and BM treatments also reduced the usual BM time (Table 1). Eucalyptus treated with HCW (160 °C, 30 min) followed by BM for 20 minutes had approximately 70% of EHGY with a cellulase loading of 4 FPU per g substrate. This yield was comparable to yields from samples treated with HCW (200 °C, 30 min) or those subjected to BM for 40 min and further hydrolyzed using ten times higher concentration of cellulases (40 FPU per g substrate). HCW treatment was therefore very useful in improving the milling efficiency and its combination with BM treatment can save energy and enzyme loading. Teramoto et. al.;16 developed a sulfuric acid-free ethanol cooking (SFEC) treatment to achieve complete saccharification of cellulosic components in eucalyptus, thereby avoiding the problems associated with the use of

Table 1 Impact of mechanical pretreatment on the chemical composition, physical properties and saccharification of selected biomass feedstocks

	Pretreatments		()	ido	CII		1	DITOVO	
Substrate	1^{st}	2^{nd}	(* nm)	(%)	(%)	(%)	(%)	(%)	Reference
α-Cellulose	Control (25 °C) Ball milling (25 °C) Boll milling (45 °C)			100				38 63	50
Cellulose "cotton"	Dail mining (45 °C) Untreated Compression milling 6 passes Compression milling 35 passes		4 4 4 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	001			84 82 75	78	43
Cellulose "avicel"	Compression milling 37 passes Untreated Compression milling 4 passes Compression milling 30 passes		3 8 8 4 7 8 8 7 8 8 8 8 8 8 8 8 8 8 8 8 8				71 71 37		
Pine chips	Compression mining 33 passes Untreated Hot water, 180 °C 30 min (pH = 5) H_2SO_4 2.2% w/w ((pH = 1.1) 180 °C 30 min)	Disk milling, 2570 rpm Disk milling, 2570 rpm Disk milling, 2570 rpm		42.6 41.1 35.6	22.4 4.9 0.2	27.0 28.3 33.4	à	11.3 33.1 39.6	3 and 4
	b SPORL ((pH = 4.2) 180 $^{\circ}$ C 30 min) b SPORL (fpH = 1.9) 180 $^{\circ}$ C 30 min)	Disk milling, 2570 rpm Disk milling, 2570 rpm		41.2	2.4	22.6		84.1	
Poplar wood	Untreated KOH 1.5 mmol g^{-1}	1 (0	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	44.4	13.9	26.3 24.5	55.4 66.2	18	15
	KOH 1.5 mmol \tilde{g}^{-1} Peracetic acid 0.2 g g^{-1} + KOH 1.5 mmol g^{-1}	Ball milling for 6 days Ball milling for 3 days		49.2 55.8	13.8	24.5 14.8	27.3 30.1	55 72	
	Peracetic acid 0.5 g g ⁻¹ + KOH 1.5 mmol σ^{-1}		$\sim\!\!2$	63.6	16.3	8.9	67.7	09	
	Peracetic acid 0.5 g g ⁻¹ + KOH 1.5 mmol g ⁻¹	Ball milling for 6 days		63.6	16.3	8.9	24.6	82	
	KOH 0.75 mmol g ⁻¹ KOH 0.75 mmol g ⁻¹ Peracetic acid 0.1 g g ⁻¹	Ball milling for 3 days	~ ~ ~ ~	47.5 47.5 47.3	14.8 14.8 14.8	26.0 26.0 23.3	60.0 21.6 60.2	8 44 6	
	Peracetic acid 0.1 g g^{-1} Peracetic acid 0.1 g g^{-1} + KOH 0.07 mmol g^{-1} Peracetic acid 0.1 g g^{-1} Peracetic acid 1.0 g g^{-1}	Ball milling for 3 days Ball milling for 3 days Ball milling for 6 days	~	47.3 46.4 47.3 57.0	14.8 14.8 17.6	23.3 23.1 23.3 6.1	25.9 16.4 8.2 66.1	42 54 65 48	
Douglas-fir wood	Peracetic acid 1.0 g g ⁻¹ Untreated 195 °C, 4.5% SO ₂ -steam exploded, 4.5	Ball milling for 6 days Ball milling (liquid), 45 °C, 7.5% w/v +		57.0 48.9 88.8	17.6 21.3 1.04	6.1 30.0 1.1	17.5	86	50
	45 min	Elloylue; 24 II Ball milling (liquid), 45 °C, 10% w/v + enzyme; 48 h Magnetic stirring 7.5% w/v + enzyme: 24 h		88.8 88.8	1.04	1.1		88 25	
	Untreated Ball milling for 5 min		0.2 0.01-0.1	46.8	28.9	25.3	68.4 41.0	15	49 and 52

3 and 4

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Reference

Ball milling for 20 min MPa Bartision, 25 °C 45-120 pm -415		Tenedulienes			į	(í	
Ball milling for 5 min Ball milling for 5 min Ball milling for 5 min Ball milling for 20 min fibrillated 10 min Ball milling for 20 min fibrillated 10 min Ball milling for 20 min 1 MPa Hot water 140°C, 30 min 1 MPa Bettrusion 120°C 50 rpm + ethylene glycol Extrusion 120°C, 30 min (solid yield 67%) Hot water 140°C, 30 min (solid yield 65%) Hot water 140°C, 50 mi	Substrate	1 st	$2^{ m nd}$	Size (mm) (* nm)	CEL (%)	HC (%)	L1G (%)	Crl (%)	SACH (%)
Ball milling for 5 min fibrillated 10 min abla milling for 20 min fibrillated 10 min abla milling for 20 min 1 MPa Hot water 140°C, 30 min 1 MPa Extrusion, 25°C 45–120 rpm Extrusion 40°C 50 rpm + ethylene glycol Extrusion, 20°C 50 rpm + ethylene glycol Extrusion, 20°C, 30 min (solid yield 65%) Hot water 140°C, 30 min (solid yield 85%) Hot water 140°C, 30 min (solid yield 65%) Ho		Ball milling for 20 min		$\sim\!\!0.01$					
Ball milling for 20 min fibrillated 10 min Ball milling for 20 min 1 MPa Hot water 140 °C, 30 min 1 MPa Hot water 180 °C, 30 min 1 MPa Extrusion, 25 °C 45-120 rpm Ball willing for 10 min Ball milling for 20 min (solid yield 65%) Hot water 140 °C, 30 min (solid yield 65%) Hot water 140 °C, 30 min (solid yield 66%) Hot water 140 °C, 30 min (solid yield 66%) Hot water 140 °C, 30 min (solid yield 66%) Hot water 100 °C, 30 min (solid yield 66%) Hot water 100 °C, 30 min (solid yield 66%) Hot water 200 °C, 30 min (solid yield 66%) Hot water 200 °C, 30 min (solid yield 66%) Hot water 200 °C, 30 min (solid yield 66%) Hot water 200 °C, 30 min (solid yield 66%) Hot water 200 °C, 30 min (solid yield 66%) Hot water 200 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 30 min (solid yield 60%) Hot water 100 °C, 40 min (solid yield 60%) Hot water 100 °C, 50 min (solid yie		Ball milling for 5 min	Fibrillated –2 min	<0.01				49.0	45
Hot water 140°C, 30 min 1 MPa Hot water 140°C, 30 min 1 MPa Hot water 140°C, 30 min 1 MPa Hot water 180°C, 30 min 1 MPa Dutreated Extrusion 40°C 50 rpm + ethylene glycol Extrusion 10°C 50 rpm + DMSO Untreated Ball milling for 10 min Ball milling for 120 min Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Ball milling for 20 min Bactic acid 1 wt% Ball milling for 20 min Bactic aci		Dall milling for 20 min Abrillated 10 min	Fibrillated -10 min	<0.01					27
Hot water 170° C, 30 min 1 MPa Hot water 170° C, 30 min 1 MPa Hot water 180° C, 30 min 1 MPa Extrusion, 25° C 45-120 rpm Extrusion, 120° C 50 rpm + ethylene glycol Extrusion 120° C 50 rpm + ethylene glycol Ball milling for 10 min Ball milling for 10 min Ball milling for 20 min (solid yield 65%) Hot water 140° C, 30 min (solid yield 65%) Hot water 160° C, 30 min (solid yield 65%) Hot water 100° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 200° C, 30 min (solid yield 65%) Hot water 140° C, 30 min (solid yield 65%) Hot water 200° C, 60 min (solid yield 65%) Hot water 200° C, 60 min (solid yield 65%) Hot water 200° C, 60 min AcOH 1% EXCHAWater = 75/25, 220° C, 60 min AcOH 1% Exchange and 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		Hot water 140 °C. 30 min 1 MPa	133 C 101 240 IIIIII, 0.1 IMF	∑ rc	I	22.6	I		7 7
Hot water 180 °C, 30 min 1 MPa Hot water 180 °C, 30 min 1 MPa Hot water 170 °C, 30 min 1 MPa Hot water 180 °C, 30 min 1 MPa Hot water 180 °C, 30 min 1 MPa Extrusion, 25 °C 45-120 rpm Hot water 180 °C, 30 min 1 MPa Ball milling for 20 min Ball milling for 40 min Ball milling for 20 min Ball mil		Hot water 170 °C, 30 min 1 MPa		~ 15	I	15.5	I		ı κ
Hot water 140 °C, 30 min 1 MPa Hot water 170 °C, 30 min 1 MPa Hot water 170 °C, 30 min 1 MPa Hot water 170 °C, 30 min 1 MPa Hot water 180 °C, 30 min 1 MPa Extrusion, 25 °C 45–120 rpm Hot water 40 °C 50 rpm + ethylene glycol Extrusion 40 °C 50 rpm + ethylene glycol Extrusion 40 °C 50 rpm + bMSO Untreated Ball milling for 10 min Ball milling for 120 min Ball milling for 20 min Ball milling, 400 rpm, 60 min Ball milling, 400 rpm, 60 min Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass Untreated Directed Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass		Hot water 180 °C, 30 min 1 MPa		\sim 15	I	14.2	I		4.5
Hot water 170 °C, 30 min 1 MPa Extrusion, 25 °C 45–120 rpm Untreated Extrusion, 25 °C 45–120 rpm Extrusion, 25 °C 45–120 rpm Extrusion, 26 °C 50 rpm + ethylene glycol Extrusion 120 °C 50 rpm		Hot water 140 °C, 30 min 1 MPa	Extrusion, 25 °C 45–120 rpm	~ 15					22.6
Hot water 180 °C, 30 min 1 MPa Extrusion, 25 °C 45-120 rpm Distrested Extrusion 120 °C 50 rpm + ethylene glycol Extrusion 120 °C 50 rpm + bMSO Untreated Ball milling for 10 min Ball milling for 20 min (solid yield 67%) Hot water 140 °C, 30 min (solid yield 65%) Hot water 140 °C, 30 min (solid yield 67%) Hot water 140 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 180 °C, 30 min (solid yield 67%) Hot water 200 °C, 30 min (solid yield 67%) Hot water 200 °C, 30 min (solid yield 67%) Hot water 200 °C, 30 min (solid yield 67%) Ball milling for 20 min (solid yield 67%) Ball milling, 400 rpm, 60 min accet acid 1 wtw. ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwater = 75/25, 200 °C, 60 min AcOH 1% ExoHwate		Hot water 170 °C, 30 min 1 MPa	Extrusion, 25 °C 45–120 rpm	~ 15					21.8
Dutreated Extrusion 40 °C 50 rpm + ethylene glycol Extrusion 40 °C 50 rpm + ethylene glycol Extrusion 40 °C 50 rpm + ethylene glycol Extrusion 40 °C 50 rpm + DMSO Untreated Ball milling for 20 min (solid yield 96%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 500 °C, 30 min (solid yield 85%) Hot water 500 °C, 30 min (solid yield 85%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 100 °C, 60 min (solid yield 85%) Hot water 100 °C, 60 min (solid yield 85%) Ball milling, 400 rpm, 60 min (solid yield 85%) Ball milling, 400 rpm, 60 min (solid yield 85%) Hot compressed water (HCW) 140 °C (solid yield 85%) Ball milling, 1800 rpm, 3 pass HCW 160 °C (solid yield 85%) Ball milling, 1800 rpm, 3 pass HCW 160 °C (solid yield 85%) Ball milling, 1800 rpm, 3 pass		Hot water 180 °C, 30 min 1 MPa	Extrusion, 25 °C 45–120 rpm	~ 15					22.8
Extrusion 40 °C 50 rpm + ethylene glycol Extrusion 100 °C 50 rpm + DMSO Untreated Ball milling for 10 min Ball milling for 10 min Hot water 140 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 85%) Hot water 160 °C, 30 min (solid yield 85%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Ethylwater = 90/10, 200 °C, 60 min Ball milling, 400 rpm, 60 min Ball milling, 400 rpm, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min AcOH 1% Ethylwater = 75/25, 220 °C, 60 min Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass		Untreated		$\sim \!\! 0.2$				68.4	10.8
Extrusion 120 °C 50 rpm + ethylene glycol Extrusion 40 °C 50 rpm + DMSO Untreated Ball milling for 10 min Ball milling for 20 min (solid yield 95%) Hot water 100 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (s		Extrusion 40 °C 50 rpm + ethylene glycol						54.7	62.4
Extrusion 40 °C 50 rpm + DMSO Untreated Ball milling for 10 min Ball milling for 20 min Ball milling for 40 min Ball milling for 20 min Ball milling for 40 min Ball milling for 120 min Hot water 140 °C, 30 min (solid yield 67%) Hot water 140 °C, 30 min (solid yield 67%) Hot water 140 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Ball milling for 20 min (solid yield 67%) Ball milling, 400 rpm, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C, 60 min 400 H 1% EtOH/water = 75/25, 220 °C,		Extrusion 120 °C 50 rpm + ethylene glycol						47.1	38.3
Ball milling for 20 min Ball milling for 120 min Ball milling for 120 min Ball milling for 120 min Hot water 160 °C, 30 min (solid yield 95%) Hot water 140 °C, 30 min (solid yield 95%) Hot water 140 °C, 30 min (solid yield 85%) Hot water 140 °C, 30 min (solid yield 85%) Hot water 160 °C, 30 min (solid yield 85%) Hot water 160 °C, 30 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 85%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) Hot water 200 °C, 60 min (Extrusion 40 °C 50 rpm + DMSO							36.1
Ball milling for 10 min Ball milling for 20 min (solid yield 96%) Hot water 140°C, 30 min (solid yield 85%) Hot water 140°C, 30 min (solid yield 85%) Hot water 160°C, 30 min (solid yield 85%) Hot water 200°C, 30 min (solid yield 85%) Untreated Cutter milling Ball milling, 400 rpm, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200°C, 60 min Ball milling, 400 rpm, 60 min 60 min AcOH 1% EtOH/water = 75/25, 200°C, 60 min AcOH 1% EtOH/water = 75/25, 200°C, 60 min AcOH 1% EtOH/water = 75/25, 200°C, 80 min Ball milling, 1800 rpm, 3 pass HCW 160°C Ball milling, 1800 rpm, 3 pass Untreated Untreated	Eucalyptus	Untreated		2	40.0	10.4	28.8	59.7	0
Ball milling for 20 min Ball milling for 40 min Ball milling for 40 min Ball milling for 40 min Ball milling for 10 min Hot water 140 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 160 °C, 30 min (solid yield 67%) Hot water 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid yield 67%) EtOH/water = 75/25, 200 °C, 60 min (solid		Ball milling for 10 min						25.9	16.0
Ball milling for 40 min Ball milling for 120 min Hot water 140 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 140 °C, 30 min (solid yield 67%) Hot water 140 °C, 30 min (solid yield 67%) Hot water 100 °C, 30 min (solid yield 67%) Hot water 200 °C, 30 min (solid yield 67%) Hot water 200 °C, 30 min (solid yield 67%) Hot water 200 °C, 30 min (solid yield 67%) Untreated EtOH/water = 90/10, 200 °C, 60 min Ball milling, 400 rpm, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min Ball milling, 400 rpm, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min Ball milling, 1800 rpm, 3 pass HCW 160 °C HOt compressed water (HCW) 140 °C Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass Untreated Ball milling, 1800 rpm, 3 pass Untreated		Ball milling for 20 min						9.7	42.8
Ball milling for 120 min Hot water 140 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 67%) Hot water 140 °C, 30 min (solid yield 67%) Hot water 100 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 67%) Untreated Cutter milling, 400 rpm, 60 min acetic acid 1 wt% EtoH/water = 75/25, 200 °C, 60 min Ball milling, 400 rpm, 60 min AcOH 1% EtoH/water = 75/25, 220 °C, 60 min Ball milling, 1800 rpm, 3 pass HCW 160 °C Hot compressed water (HCW) 140 °C Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C		Ball milling for 40 min						3.0	70.4
Hot water 140 °C, 30 min (solid yield 95%) Hot water 160 °C, 30 min (solid yield 85%) Hot water 140 °C, 30 min (solid yield 67%) Hot water 100 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 96%) Hot water 200 °C, 30 min (solid yield 96%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 67%) Untreated Cutter milling for 20 min (solid yield 67%) Untreated Cutter milling, 400 rpm, 60 min acetic acid 1 wt% EtOH/water 50/50, 200 °C, 60 min Ball milling, 400 rpm, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min Ball milling, 1800 rpm, 3 pass HCW 160 °C Hot compressed water (HCW) 140 °C Hot compressed water (HCW) 140 °C Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass Untreated		Ball milling for 120 min						6.4	89.7
Hot water 160 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 67%) Hot water 140 °C, 30 min (solid yield 65%) Hot water 160 °C, 30 min (solid yield 96%) Hot water 200 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 65%) Hot water 200 °C, 30 min (solid yield 65%) Untreated Cutter milling for 20 min Ball milling, 400 rpm, 60 min acetic acid 1 wt% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtoH/water = 75/25, 220 °C, 60 min AcOH 1% EtoH/water = 75/25, 220 °C, 60 min AcOH 1% EtoH/water = 75/25, 220 °C, 60 min AcOH 1% EtoH/water = 75/25, 220 °C, 60 min AcOH 1% EtoH/water = 75/25, 220 °C, 60 min AcOH 1% EtoH/water = 75/25, 220 °C, 60 min AcOH 1% EtoH/water = 75/25, 220 °C, 60 min AcOH 1% EtoH/water = 75/25, 220		Hot water 140 °C, 30 min (solid yield 96%)						62.0	5.9
Hot water 200 °C, 30 min (solid yield 67%) Hot water 140 °C, 30 min (solid yield 66%) Hot water 160 °C, 30 min (solid yield 96%) Hot water 160 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 67%) Untreated Cutter milling Ball milling, 400 rpm, 60 min Ball milling, 1800 rpm, 3 pass HCW 160 °C Hot compressed water (HCW) 140 °C Hot water = 75/25, 220 °C, 60 min Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass Ball milling, 1800 rpm, 3 pass HCW 180 °C		Hot water 160 $^{\circ}$ C, 30 min (solid yield 85%)						67.4	23.3
Hot water 140 °C, 30 min (solid yield 96%) Hot water 160 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 67%) Untreated EtOH/water = 90/10, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% EtoH/water = 75/25, 200 °C, 60 min AcOH 1% E		Hot water 200 $^{\circ}$ C, 30 min (solid yield 67%)			51.9	1.1		74.7	74.7
(solid yield 96%) Hot water 160 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 67%) Untreated EtOH/water = 90/10, 200 °C, 60 min acetic acid 1 wt% EtOH/water 50/50, 200 °C, 60 min 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 2		Hot water 140 °C, 30 min	Ball milling for 20 min					9.1	49.4
Hot water 160 °C, 30 min (solid yield 85%) Hot water 200 °C, 30 min (solid yield 67%) Untreated EtOH/water = 90/10, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acott 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/water = 75/25, 200 °C, 60 min acetic acid 1 wt% EtOH/wate		(solid yield 96%)							
(solid yield 85%) Hot water 200 °C, 30 min (solid yield 67%) Untreated Cutter milling Ball milling, 400 rpm, 60 min EtOH/water = 90/10, 200 °C, 60 min EtOH/water 50/50, 200 °C, 60 min 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min A		Hot water 160 °C, 30 min	Ball milling for 20 min					13.2	63.0
Hot water 200 $^{\circ}$ C, 30 min (solid yield 67%) Untreated Cutter milling Ball milling, 400 rpm, 60 min acetic acid 1 wt% EtOH/water 50/50, 200 $^{\circ}$ C, 60 min 60 min AcOH 1% EtOH/water = 75/25, 200 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH/water = 75/25, 220 $^{\circ}$ C, 60 min AcOH 1% EtOH		(solid yield 85%)							
(Solid yield 67%) Untreated Untreated Cutter milling Ball milling, 400 rpm, 60 min acetic acid 1 wt% EtOHwater 50/50, 200 °C, 60 min AcOH 1% EtOHwater = 75/25, 200 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass Untreated Untreated		Hot water 200 °C, 30 min	Ball milling for 20 min					16.9	49.4
EtOH/water = 90/10, 200 °C, 60 min acetic acid 1 wt% EtOH/water 50/50, 200 °C, 60 min 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min Ball milling, 1800 rpm, 3 pass Ball milling, 1800 rpm, 3 pass Untreated		(Solid yleid 8/%) Untreated		<i>2</i> ∼	42.2	34.1	28.1		
EtOH/water = 90/10, 200 °C, 60 min acetic acid 1 wt% EtOHwater 50/50, 200 °C, EtOHwater 50/50, 200 °C, EtOHwater = 75/25, 200 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% EtOH			Cutter milling Ball milling 400 rnm 60 min	· ~				30.0	8.1
acetic acid 1 wt% EtOH/water 50/50, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% Hot compressed water (HCW) 140 °C HCW 160 °C HCW 160 °C HCW 180 °C Ball milling, 1800 rpm, 3 pass Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass		EtOH/water $= 90/10$, 200 °C, 60 min	Ball milling, 400 rpm, 60 min						82.4
EtOH/water 50/50, 200 °C, Ball milling, 400 rpm, 60 min 60 min AcOH 1% EtOH/water = 75/25, 200 °C, 60 min AcOH 1% EtOH/water = 75/25, 220 °C, 60 min AcOH 1% Hot compressed water (HCW) 140 °C HOT Compressed water (HCW) 140 °C Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass		acetic acid 1 wt%							
60 min AcOH 1% EtOHwater = 75/25, 200 °C, 60 min AcOH 1% EtOHwater = 75/25, 220 °C, 60 min AcOH 1% AcOH 1% Hot compressed water (HCW) 140 °C HOT W 160 °C HCW 160 °C HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass		EtOH/water 50/50, 200 °C,	Ball milling, 400 rpm, 60 min						78.3
AcOH 1% EtOHwater = $75/25$, 220 °C, 60 min AcOH 1% Hot compressed water (HCW) 140 °C Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass Untreated		60 min AcOH 1% EtOH/water = $75/25 200 ^{\circ}$ C. 60 min	Ball milling 400 rnm 60 min	0.05-0.1	68.1	7.	6.3	15.0	100
EtOH/water = 75/25, 220 °C, 60 min AcOH 1% Hot compressed water (HCW) 140 °C Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass Untreated		AcOH 1%	ban minist to thin, or min	1.0	1	2	2		
HCW 160 °C Ball milling, 1800 rpm, 3 pass Ball milling, 1800 rpm, 3 pass Ball milling, 1800 rpm, 3 pass HCW 160 °C Ball milling, 1800 rpm, 5 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass Untreated		EtOH/water = $75/25$, 220 °C, 60 min		$\sim\!\!0.05$				48.0	89
HCW 160 °C Ball milling, 1800 rpm, 3 pass HCW 180 °C Ball milling, 1800 rpm, 3 pass Untreated		HACH 1700 HACH COMPRESSED WATER (HCW) 140 °C	Ball milling, 1800 rpm, 3 pass						∞ ⁶
HCW 180 °C Ball milling, 1800 rpm, 3 pass Untreated		HCW 160 °C	Ball milling, 1800 rpm, 5 pass Ball milling, 1800 rpm, 5 pass						22
Untreated	;	HCW 180 °C	Ball milling, 1800 rpm, 3 pass						42
	Spruce chips	Untreated		8-22	42.8	20.76	28.6		

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Table 1 (Contd.)

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	Pretreatments			į	(·	911010	
Substrate	1^{st}	2 nd	Size (mm) (* nm)	(%)	HC (%)	(%)	(%)	SACH- (%)	Reference
	Hot water 100 °C, 10 min, 40% w/v	Disk milling (disk 0.25 mm) 130 $^{\circ}\text{C}, 2.4$ bar	>1.814 0.254- 0.529	46.8 44.3	20.38 20.71	26.3 27.5		12 21	
	Hammer milling atmospheric		<0.127 <0.127 >1.27	29.4	20.54	38.7		w {	
	7% NaOH + 12% urea under freezing condition: $-18^{\circ}\mathrm{C}$ for 24 h (solid yield perc)	Disk milling (disk 0.06 mm) 166 $^{\circ}$ C, 7.4 bar	$^{<0.318}_{0.318^c}$	47.1	15.0	31.4		13 45	
	$9\% \text{ NaSO}_3 + 3.68\% \text{ H}_2\text{SO}_4 \text{ 170 }^{\circ}\text{C for 30}$	Disk milling (disk 0.06 mm) 166 $^{\circ}$ C, 7.4		61.5	69.0	35.4		92	
Japanese	um (sonu yielu o1.0%) Untreated	Dai		43.9	23.9	35.0	29.3		53
cedar wood	Ball milling 30 rps, 30 min Ball milling 30 rps. 5 min			46.9	25.0	34.8	18.5	77	
	Ball milling 15 rps, 5 min			51.1	12.2	33.3	27.0	18	
Sugarcane bagasse	Untreated		₹	44.1	27.7	22.5	31.0		54
	Ball milling 1 h; 25 °C					1	0 0	99	
	Ball milling Z II; Z3 C Ball milling 4 b: 25 °C						0 0	89	
Bagasse	Untreated		<2	38.8	26.0	32.4^{b}	38	22	18
0	Ball milling 30 min; 25 °C; 400 rps						0.9	62	
	Ball milling 90 min; 25 °C; 400 rps						0.0	84	
	Wet disk milling 10 cycle; 37 min; 25 °C; 1800 rps	800 rps 1800 rps					36	32	
Sugarcane straw			<2	33.6	28.9	31.8^{b}	30	24	
)	Ball milling 30 min; 25 °C; 400 rps						23	45	
	Ball milling 90 min; 25 °C; 400 rps						0.0	78	
	1800 r	10					29	56	
	Disk milling 20 cycle 167 min 25 °C 1800 r	rps					21	89	
Bamboo	Untreated		~3~6	45.5	22.8	1		0.0	55
	Ball milling for 5 min		$\sim\!\!0.1$					10.5	
	Steam explosion 243 $^{\circ}\mathrm{C}$ for 5 min; 35 atm		~3-6	45.1	1.4	I		64.3	
	Steam explosion 214 °C for 5 min; 20 atm + 10% NaOH for 60 min		~3–6					9.09	
	Ball milling for 5 min	10% NaOH for 60 min	$\sim\!\!0.1$					63.9	
Corn stover	Hammer milling 30 min		0.053-	39.2	32.7	21.4		24	29
			0.075 0.425	39.0	34.6	19.1		12	
			0.710)	!		 	
	Hammer milling 30 min	Hot water 190 $^{\circ}$ C for 15 min	0.053-					64	
			0.075						
			0.425-					28	
Rice straw	Untreated		0.710 ~2	33.3	15.9	24.3	51.9	23.4	44

Table 1 (Contd.)

	Pretreatments		()	į	Ç		Ţ	DI TO V O	
Substrate	1^{st}	$2^{ m nd}$	(* nm)	(%)	нС (%)	(%)	(%)	(%)	Reference
	Ball milling 5 min; 1700 rps Ball milling 30 min; 1700 rps Wet disk milling 1 cycle; 3 min; 1800 rps Wet disk milling 5 cycle; 15 min; 1800						41.8 25.2 46.9 46.0	52.2 75.9 46.8 59.4	
	Wet disk milling 10 cycle; 30 min; 1800 rps						48.6	78.5	
	Untreated		0.93	36.5	25.6	12.8		26.3	13 and 14
	Steam explosion 179 °C for 5 min	Coarsely ground	0.59	36.8	23.8	12.5		33.2	
Miscanthus	Steam explosion 179 °C for 5 min Untreated	Jet milling, 4544 rpm, 25 min	0.06	35.8	23.8	12.5		61.4	80
	Milling, 100 rpm	Water, 95 °C	0.22	32.0) 	25.1		15.5)
	ò	NaOH 12% w/w, 4 h, 50 $^{\circ}$ C	0.22	48.9		14.2		38.5	
		NaOH 12% w/w, 4 h, 95 °C	0.22	52.9		9.5		50.7	
		NaOH 12% w/w, 4 h, 70 $^{\circ} C$ (solid yield	1.0	56.5	29.1	10.8		32	
		64%)							
		NaOH 12% w/w, 4 h, 70 $^{\circ}$ C	0.22	54.3	27.7	10.9		38	
		NaOH 12% w/w, 4 h, 70 $^{\circ}$ C (solid yield	0.017	48.8	26.5	14.1		46	
	100000000000000000000000000000000000000	57%)	6		ı	,		,	
	8% W/W NaOH, 25 ⁻ C, 24 h + steam at	Extrusion, 100 °C (solid yield 88%)	20-200	41.4	7.67	16.4		19.6	
	aunospheric pressure for 10 mm Extrusion, 100 °C + NaOH 12% w/w (solid yield 52%)		50-500	59.9	22.1	9.3		34.6	
	Extrusion, 100 °C	NaOH 12% w/w, 4 h, 70 $^{\circ}$ C (solid yield 58%)	50-500	55.6	26.9	8.6		40.2	
Alfalfa	Whole plant			27.4	11.7	4.8		41.3	78
	Hammer milling		0.25-0.5	31.4	15.0	5.6		31.1	
			0.106 - 0.25	23.9	10.5	3.5		45.1	
			0.053-	25.3	5.3	5.2		57.4	
			0-0.053	27.1	7.1	7.3		57.7	
Timothy	Whole plant			28.8	27.2	4.6		39.3	
	Hammer milling		0.25 - 0.5	29.5	37.0	4.9		39.3	
			0.106 - 0.25	29.5	24.4	4.1		38.1	
			0.053-	30.3	19.0	4.0		69.4	
			0-0.053	22.7	13.0	9.9		73.9	

CEL: cellulose; HC: hemicelluloses; LiG: lignin; CrI: crystallinity index; SACH: saccharification." 48 h of enzymatic hydrolysis (% w/w glucose/cellulose in biomass). ^b Lignin and others. ^c Not fractionated.

strong acid catalysts. ¹⁶ Cutter-milled flours were exposed to a mixture of ethanol (EtOH)/water/acetic acid mixture in an autoclave. Enzymatic hydrolysis experiments of the post-chemical size reduction pre-treated samples demonstrated that almost complete conversion (100%) of the cellulosic components to glucose could be achieved under optimum conditions: EtOH 50%, 200 °C for 60 min with 1% AcOH followed by BM for 60 minutes. A large-scale trial revealed that there is little consumption of *in-feed* EtOH during SFEC, with the possibility to recover and reuse the majority of EtOH.

The effect of DM of HCW-treated water insoluble residues from eucalyptus on EHGY was also recently studied.⁵² Glucose production was observed to increase at increasing number of passes, with the highest EHGY (42.8%) obtained for 3-pass DM after HCW treatment at 180 °C. This value can be recalculated to be 101.7% with cellulose content (42.1% of untreated raw material) of the starting material taken as reference, essentially proving all glucan content was digestible.

Studies from our group have been focused in the development of ecofriendly combined treatment using a chemical process and DM as mechanical process.17 An innovative dry NaOH chemo-mechanical pretreatment (TS_{dry}) was conducted at high material concentration (5 kg L⁻¹) with a biomass-liquid ratio of 5:1. The primary objective of this study was to develop a dry chemo-mechanical pretreatment process which minimises waste generation, reducing the environmental impact of the entire process in conjunction with energy savings and increase enzymatic hydrolysis efficiency. NaOH-dilute chemomechanical (TS_{dilute}) pretreatment conducted at low material concentration (0.2 kg L⁻¹) consumed higher amounts of water (5 L water per 1 kg biomass) and energy compared to TS_{dry}. In fact, the lowest energy efficiency obtained was 0.417 kg glucose kW h⁻¹ for TS_{dilute} chemo-mechanical pretreatment as compared to 0.888 kg glucose kW h⁻¹ and 0.197 kg glucose kW h⁻¹ for TS_{dry} and control, respectively.¹⁷

The accessibility of enzymes to exposed cellulose surfaces of Japanese cedar wood tissues pretreated with NaOH, NaOH-Na₂S (kraft pulping), hydrothermolysis, ball-milling, and organosolvolysis was also recently studied.53 Results clarified the linear dependency of the exposure of crystalline and noncrystalline cellulose surfaces for enzymatic saccharification obtained via organosolv and kraft delignification processes. In comparison, BM for 5-30 min, hydrothermolysis for 30-60 min at 180 °C and alkaline (3% NaOH + Na₂S at 180 °C for 30 min) treatments increased the EHGY up to 77%, 15% and 42%, respectively. Buaban et al.;54 in turn studied an integrated process combining mechanical pretreatment by BM, with enzymatic hydrolysis and fermentation of bagasse.54 They reported that BM for 2 h was sufficient to nearly complete cellulose structural transformation into an accessible amorphous form. Pretreated cellulosic residues were hydrolyzed by a crude enzyme preparation method containing cellulase activity combined with complementary β -glucosidase activity. EHGY of 84.0% was obtained after 72 h at 45 $^{\circ}\text{C}$ and pH 5. In another study, da Silva et al.;18 compared the effectiveness of BM and wet disk milling (WDM) on treating sugarcane bagasse and straw.18 EHGY at optimum conditions for BM-treated bagasse and straw

were 84% and 78%, respectively. Maximum yields for bagasse and straw using DM were 49.3% for glucose and 68% for xylose. BM also improved the enzymatic hydrolysis yield by decreasing the crystallinity. The defibrillation effect observed for WDM samples seems to favor enzymatic conversion. Bagasse and straw BM hydrolysates were fermented by Saccharomyces cerevisiae strains. Ethanol yields from total fermentable sugars using a C6-fermenting strain reached 89.8% and 91.8% for bagasse and straw hydrolysates, respectively, and 82% and 78% when using a C₆/C₅ fermenting strain. Yamashita et al.;⁵⁵ studied the post-physicochemical size reduction as a pretreatment method for enzyme saccharification of bamboo biomass. EHGY of 64% was obtained after 48 h of using bamboo steam exploded at 35 atm for 5 min and at 243 °C. In addition, pretreatments using 20 atm and 5 min steam explosion or BM for 5 min followed by a 10% (w/w) sodium hydroxide treatment at 121 °C for 60 min was executed to enhance the digestibility of the holocelluloses component (Table 1). Both pretreatment methods had a large positive effect on the production of sugars subsequent enzymatic hydrolysis. In particular, the combined 20 atm steam explosion and 10 wt% sodium hydroxide treatments obtained the maximum EHGY. Levels obtained were 456 mg g⁻¹ of initial dry sample of glucose and 460 mg g⁻¹ of initial dry sample of reducing sugar. In comparison, the BM process and 10 wt% sodium hydroxide treatment produced 383 and 485 mg g⁻¹ of initial dry sample of glucose and reducing sugar, respectively. This pretreatment method requires severe conditions, namely high pressure and high temperature steam explosion in combination with high concentrations of sodium hydroxide. Apparently, BM pretreatment is a somewhat less effective, but a more environmentally friendly method for the enzyme saccharification of bamboo. Lin et al.;56 studied the chemical BM pre-treatment followed by enzymatic hydrolysis and fermentation for ethanol production from corn stover biomass.56 The obtained results indicated that the yields of glucose and xylose were improved by adding any of the following dilute chemical reagents: H2SO4, HCl, HNO3, CH₃COOH, HCOOH, H₃PO₄, and NaOH, KOH, Ca(OH)₂, NH₃H₂O in the BM pretreatment of corn stover. The optimal enzymatic hydrolysis efficiencies were obtained using BM in an alkali medium, which can be explained by a possible delignification. The said study showed thus that the BM pretreatment is a robust process. Based on microscope image of BM-pretreated corn stover, the particle size of the material was decreased and the fiber structure was more loosely organized. Afterwards, the results indicate that the treatment effect of wet milling is better than that of dry milling. The optimum parameters used for the milling process were ball speed 350 r min⁻¹, solid-liquid ratio of 1:10, raw material particle size with 0.5 mm, and number of balls of 20 (steel ball = 10 mm), and grinding for 30 min. Hideno et al.;48 compared WDM, BM and hot compressed water treatment (HCWT) of rice straw. These authors reported that EHGY by WDM, BM and HCWT were 78.5%, 89.4% and 70.3%, respectively. They also reported that in BM, the EHGY increased with milling time and increased gradually with 10 cycles of DM and leveled off between 10 and 20 cycles.48 It must be emphasized that the EHGY for 10 cycles was 78.5% with an optimal

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milling time of 60 min, temperature of 160 $^{\circ}$ C for HCWT and 10 repeated milling operations for WDM. On the basis of the BM treatment of the rice straw, the high glucose yield was achieved in a short operation time compared to eucalyptus.⁵¹ In the said study, the authors suggested that DM is a promising pretreatment for enzymatic hydrolysis. Peng et al.;57 used microcrystalline cellulose as substrate to investigate its potential ability of bioconversion in a novel combined pretreatment of ball milling (BM) and/or microwave irradiation (MWI). To achieve the same or higher glucose yield of BM for 3 h and 6 h, BM for 1 h with MWI for 20 min could save 54.8% and 77.4% energy consumption, respectively.⁵⁷ Moreover, chemicals were not required in this process. It is concluded that the combination of BM and short time MWI is an environment-friendly, economical and effective approach to treat biomass. Lee et al.;52 have developed an energy efficient nanofibrillation method that combines DM and HCW treatment to improve enzymatic accessibility of Eucalyptus wood. In this method, DM fibrillated the residual product of HCW treatment under wet conditions.⁵² The relatively moderate HCW treatment conditions (temperature below 180 °C and reaction time of 30 min) were adopted, and the amount of water used was only five times that of wood. These conditions were sufficient for the partial removal of hemicellulose and lignin from cell walls with supramolecular structures to create nano-spaces between cellulose microfibrils. These morphological characteristics have effectively improved the nanofibrillation by DM. The fibrillated products with a size of less than 20 nm can already be obtained after very short milling time, and this process has significantly improved the enzymatic saccharification yield. The energy consumption is also, notably much lower than that of other mechanical methods for size reduction to give the same monosacchariderecovery yield.

3.2. Twin screw extrusion process

The extrusion of biomass can be a viable pretreatment method due to its ability to simultaneously expose it to a range of disruptive conditions in a continuous flow process. Nevertheless, the extruder screw speed, barrel temperature, and feedstock moisture content are important factors that can influence sugar recovery from biomass.

Lee et al.;49 studied a mechanical micro/nano-fibrillation of Douglas fir by a continuous extrusion process in an attempt to develop a cost-effective pretreatment method for enzymatic saccharification (Table 1). Additives with cellulose affinity (ethylene glycol, glycerol, and dimethyl sulfoxide) were used to effectively fibrillate the wood cell wall up-to submicron- or nano-scale, thus opening up the cell wall structure for improving enzymatic accessibility, and lowering likewise the extrusion torque.49 The fibrillated products were converted into glucose with a high yield by enzymatic saccharification. The maximum cellulose-to-glucose conversion (64%) was achieved when Douglas fir wood was extruded at 40 °C in the presence of ethylene glycol. The EHGY was approximately 6 times higher than that of the untreated raw material. Nonetheless, the fibrillation of wood cell walls into submicron and/or nanoscale

fibers was successfully carried out in the presence of water using a batch-type kneader in combination with the available twinscrew elements. The maximum EHGY was found to be 54.2% in the case when the fibrillated products were kneaded for 20 min after ball milling using a screw combination. The fibrillation also increased the surface area of cellulose and the EHGY was improved by cooking the fibrillated products with water at 135 °C under 0.25 MPa. The authors have reported that only mechanical kneading appears to have some limitations towards exposure of cellulose for complete enzymatic saccharification. de Vrije et al.;58 investigated the biomass conversion pretreatment methods for the production of fermentable substrates from Miscanthus. The obtained results demonstrated an inverse relationship between lignin content and the efficiency of enzymatic hydrolysis of polysaccharides.⁵⁸ The high delignification values were obtained by the combination of extrusion and chemical pretreatment (sodium hydroxide), which is an optimized process, consisted of a one-step extrusion-NaOH pretreatment at moderate temperature (70 °C). A mass balance of this process in combination with enzymatic hydrolysis showed that the pretreatments resulted in 77% delignification, and a cellulose yield of more than 95% and 44% hydrolysis of hemicellulose. After enzymatic hydrolysis, 69% and 38% of the initial cellulose and hemicellulose fractions, respectively, were converted into glucose, xylose and arabinose. Of the initial biomass, 33% was converted into monosaccharides. Normal growth patterns of Thermotogaelfii on hydrolysate were observed and high amounts of hydrogen were produced. Yoo et al.;59 investigated the thermo-mechanical extrusion pretreatment for lignocellulosic biomass using soybean hulls as the substrate. Structural changes in substrate and sugar yields from thermo-mechanical processing were compared with two traditional pretreatment methods that utilized dilute acid (1% sulfuric acid) and alkali (1% sodium hydroxide). Extrusion processing parameters (temperature, moisture, screw speed) and processing aids (starch, ethylene glycol) were studied with respect to reducing sugar and glucose yields. The conditions resulting in the highest EHGY (95%) were screw speed of 350 rpm, maximum temperature 80 °C and moisture content of 40%. Compared to untreated soybean hulls, EHGY of soybean hulls increased by 69.6%, 128.7% and 132.2% when pretreated with dilute acid, alkali and extrusion respectively.⁵⁹ Liu et al.;⁶⁰ investigated the alkaline twin-screw extrusion pretreatment (ATSE) for corn stover. ATSE was conducted with a biomass-liquid ratio of 1/2 (w/w) at a temperature of 99 °C without any addition al heating equipment. The results indicate that ATSE pretreatment is effective in improving the enzymatic digestibility of corn stover. Sodium hydroxide loading is a more influential factor affecting both sugar yield and lignin degradation than heat treatment time. 60 After ATSE pretreatment under the proper conditions (NaOH loading of 0.06 g g⁻¹ biomass during ATSE and 1 hour heat preservation after extrusion), 71% lignin removal was achieved and the conversions of glucan and xylan in the pretreated biomass can reach up to 83% and 89% respectively via subsequent enzymatic hydrolysis (cellulase loading of 20 FPU per g-biomass and substrate consistency of 2%). About 78% of the original

polysaccharides were converted into fermentable sugars. Kang et al.;61 investigated the production of bioethanol from Miscanthus using a specially designed twin-screw extruder with sodium hydroxide. The pretreatment parameters were optimized using a response surface methodology.61 Optimum pretreatment conditions were 95 °C, 0.4 M sodium hydroxide concentration, 80 rpm twin-screw speed, and flow rate of 120 mL min⁻¹. Under these optimum conditions, efficiency of pretreatment through the biomass to ethanol ratio was 66%, as compared to a theoretical maximum of 67%. Duque et al.;62 studied an integrated one-step alkaline extrusion process as pretreatment for sugar production from barley straw (BS) biomass. The influence of extrusion temperature (T) and the ratio NaOH/BS dry matter (w/w) (R) into the extruder on pretreatment effectiveness was investigated in a twin-screw extruder at bench scale. The optimum conditions for a maximum EHGY were determined to be R = 6% and T = 68 °C. At these conditions, glucan yield reached close to 90% of theoretical yields, while xylan conversion was 71% of theoretical values. 62 These values are 5 and 9 times higher than that of the untreated material. Zhang et al.;56,63 investigated the extrusion methodology, as a continuous and cost-effective pretreatment method for corn stover bioconversion. The objective of this study was to examine the effect of extrusion as a pretreatment method and the underlying factors ruling the improvement of sugar yields.56,63 The optimum glucose, xylose, and combined sugar recoveries were 48.79%, 24.98%, and 40.07%, respectively, at 27.5% moisture content and 80 rpm screw speed. These yields were 2.2, 6.6, and 2.6 times higher than those for untreated corn stover. Karunanithy et al.;64 investigated the effects of extrusion parameters on pretreatment of pine wood chips. Pine wood chip at 25, 35, and 45% moisture content were pretreated at various temperatures (100, 140, and 180 °C) and screw speeds (100, 150, and 200 rpm) using a screw with compression ratios of 3:1. The pretreated pine wood chips were subjected to standard enzymatic hydrolysis followed by sugar and byproducts quantification.⁶⁴ Pine wood chips pretreated at a screw speed of 150 rpm and a temperature of 180 °C with a moisture content of 25% resulted in a maximum cellulose, hemicellulose, and total sugar recoveries of 65.8, 65.6, and 66.1%, respectively, which was about 6.7, 7.9, and 6.8 fold higher than the control. Furthermore, potential fermentation inhibitors such as furfural, hydroxyl methyl furfural, and acetic acid were not found in any of the treatment combinations. Karunanithy and Muthukumarappan⁴⁵ used the switchgrass as a substrate to study the effect of moisture content (10, 20, 30, 40, and 50%) and particle size (2, 4, 6, 8, and 10 mm) over a range of temperatures and screw speeds (45-225 °C and 20-200 rpm). Statistical analyses revealed that among the independent variables considered, temperature, screw speed, and moisture content had significant effect on sugar recoveries.45 The optimum pretreatment condition of temperature 176 °C, screw speed 155 rpm, moisture content 20%, and particle size 8 mm resulted in maximum glucose (41.4%), xylose (62.2%), and combined sugar recoveries (47.4%). The optimum pretreated switchgrass samples had 50% higher surface area than that of influence of alkali (NaOH) concentration on *sugar* recovery and optimized the extruder temperature, screw speed, and particle size for maximum sugar recovery. In order to evaluate the sequential effect of alkali soaking and extrusion, prairie cord grass (2–10 mm) was soaked at different alkali concentrations (0.5–2.5%, w/v NaOH) for 30 min at room temperature and then extruded using a lab scale single screw extruder at various temperatures (45–225 °C) and screw speeds (20–200 rpm). All the independent variable had a strong influence on sugar recovery and it was confirmed through statistical analyses. ^{65,66} The optimal pretreatment condition 114 °C, 122 rpm screw speed, 1.70% alkali concentration, and 8 mm particle size resulted in maximum glucose, xylose and combined sugar recoveries of 86.8, 84.5, and 82%, respectively.

4. Discussion

Enzymatic hydrolysis of cellulosic biomass depends on many factors such as physical proprieties of the substrate (chemical composition, CrI, degree of polymerization, SSA, lignin content, synergy between enzymes, mass transfer, substrate adsorption, pore diffusion, etc.). 17,30,46,57,67-69 O'Dwyer et al.;70 has proposed an empirical model that describes the roles of wheat straw lignocellulosic properties in enzymatic hydrolysis (digestibility $= 2.04SSA^{0.99}(100 - CrI)LiG^{-0.39}$). Lignin (LiG) content, SSA and CrI have the greatest impact on biomass (for wheat or even for all biomass) digestibility. It is widely accepted that highly crystalline cellulose is less accessible to cellulase attack than amorphous cellulose; therefore, crystallinity negatively affects the efficiency of enzyme contact with cellulose.71-74 A common method of reducing crystallinity is BM (Fig. 5 and 6), which tends to decrease the particle size and increase the SSA simultaneously.75

It can be observed in Fig. 6 that for all the pretreated wood parts, CrI was not directly correlated with the overall saccharification yield. It is hard to deduce whether the increase in enzymatic digestibility was solely due to the reduction in crystallinity or was caused by other physical changes. Some researchers proposed that the effect of reduced crystallinity on the hydrolysis rate might actually be a consequence of increased SSA18,19 or decreased particle size.39 da Silva et al.;18 demonstrated that sugar yield after WDM increased while there was an observed decrease in the CrI value. The authors suggest that the number of operation cycles resulting in reduction of particles size and fiber thickness, was the relevant factor in the improvement of enzymatic digestibility of the pretreated materials. The reduction in fiber thickness, which increases the SSA for the attack of cellulase, may play a role as important as the CrI increase. As a comparison, rice straw that underwent ball-milling presented a CrI of 8% and resulted in glucose yields of 63.6% after digestion, whereas 20-cycles WDM straw presented a CrI of 21%, but resulted in a higher glucose yield of 68%. Chang et al., (1997) have shown that a further reduction of particle size below 40-mesh does not necessarily enhance the hydrolysis rate. Some studies have also shown that accessible SSA is a crucial factor that affects biomass digestibility. 39,76,77 Other studies have reported conflicting results on the effect of

the control. Karunanithy et al.;65,66 investigated also the

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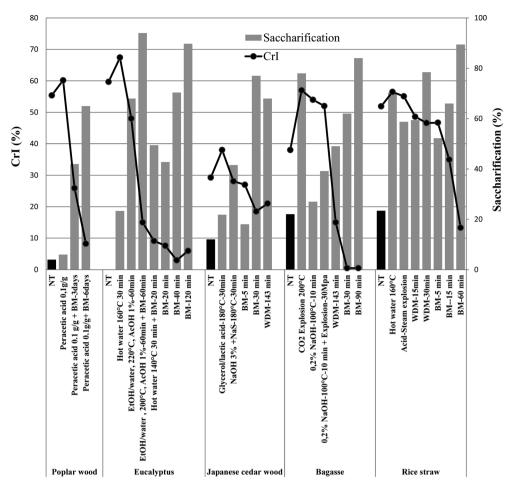


Fig. 6 Relationship between crystallinity index (CrI) and saccharification of selected lignocellulosic biomass (Table 1) NT: not treated; BM: ball milling; HM: hammer milling.

accessible SSA on biomass digestibility. Fan et al.;72 concluded that SSA had no effect on the digestibility of biomass but on limiting enzymatic hydrolysis. 19,76 In contrast, Zhang et al.;63 using extrusion, suggested that X-ray diffraction analysis showed that the CrI was not a good indicator of sugar yield. However, scanning electron microscopy showed that the cellulose network was impacted due to the destruction of the lignin sheath exposing the cellulose even more, making it more susceptible to enzymatic hydrolysis. Peng et al.;57 suggested that the correlation of crystallinity index (CrI), size of crystal, specific surface area (SSA) and degree of polymerization (DP) with the rate of enzymatic hydrolysis is differentiated by an optimized equation that indicates the rate of hydrolysis is much more sensitive as a factor to CrI than to SSA and DP.57 Other studies also suggest that lignin content and biomass crystallinity has dominated digestibility. Lignin removal greatly enhanced the ultimate hydrolysis extent. However, the crystallinity reduction tremendously increased the initial hydrolysis rate and reduced the hydrolysis time or the amount of enzyme required to attain high digestibility. To some extent, the effects of structural features on digestibility were interrelated. At short hydrolysis periods, lignin content was not important for digestibility when the crystallinity was low. Similarly, at long hydrolysis periods,

the crystallinity was not important to digestibility when lignin content was low. These conflicting results may be due to different methods used to determine lignin content, CrI and SSA; for example N₂ adsorption *versus* solute exclusion method, and methods based on geometry of particles and bulk density used to determine SSA. The measurement of accessible SSA and CrI also depends on the size and organized structure of biomass used. Moreover, accessible SSA and CrI are not easily manipulated in pretreatments.

Anyway, to reveal the underlying relationship of structural features and biomass digestibility, it is important to investigate the influence of each one on the rate and extent of hydrolysis. However, the complex biomass structure confounds the understanding of the relative importance of these features, because altering one structural feature often results in substantial changes in others. In addition, previous studies investigating the influence of structural features on digestibility have been restricted to the measurement of either one or two structural features with one cellulase loading. To eliminate the cross effects among structural features, selective pretreatment techniques have been employed to vary one particular structural feature during a pretreatment, while the other two structural features remained unchanged.

5. Conclusion and perspectives for the future: environmental drypretreatment lignocellulosic biorefineries

This contribution has been aimed to demonstrate the versatility of various pretreatment processes for lignocellulosic biomass conversion to biofuels. The complexity of pretreatment steps for an eventual energetic valorization of the biomass was illustrated with a number of examples that provides relevant alternatives to traditional (physic)chemical pretreatments with generally promising (and even improved) results.

Currently, relevant lignocellulosic biomass projects are under development including those of Futurol France, Leuna Germany (http://www.research-in-germany.de/dachportal/en/Research-Areas-A-Z/Plants/Programmes-andInitiatives/Modern-Biorefinery-for-Climate-Protection-and-Resource-

Efficiency.html) and a recently granted COST Action on the conversion of lignocellulosic waste streams to valuable chemicals and fuels (http://www.cost.eu/domains_actions/fps/Actions/FP1306). However, major challenges including feedstock costs, feedstock variability, production, environmental impacts, water recycling, transportation, enzymes costs, preprocessing along with the development of new technologies with increased efficiency of lignocellulosic conversion still need to be resolved before successful implementation of an integrated lignocellulosic valorisation for fuels, materials and chemicals production.

Mechanical pretreatments have been considered to be one of the most expensive processing steps in biorefinery in terms of energy and operating costs. In general, size reduction processes have a high-energy requirement and are often not economically advantageous. The recommended particle size should be less than 3 mm for effective accessibility and hydrolysis of lignocellulosic materials. To counteract these disadvantages, size reduction processes through milling, grinding, and extrusion can be combined with chemical and physicochemical pretreatments to save energy for grinding and enzyme loading.

Chemical or physicochemical pretreatments followed by mechanical processing can be considered as interesting innovative methodologies to reduce energy, particle sizes, increase surface area, decrease cellulose crystallinity and increase biomass digestibility and bioconversion. The benefits in energy savings achieved using this combined physicochemical/ mechanical size reduction pretreatment approach are significant, based on previous studies. Energy consumption for agricultural biomass size reduction was reported to significantly decrease with steam exploded substrates.11 The specific energy required to grind steam exploded oat and wheat straws, using 3.2 and 1.6 mm hammer mill screen sizes, was 52 and 25%, and 90 and 64% lower than untreated straw, respectively. Another study investigated the influence of torrefaction, an oxygen-free mild thermal treatment, on wood grinding energy. 79,80 Results pointed out that the specific energy consumption for grinding was remarkably reduced with increased torrefaction temperatures for pine, spruce and beech chips. Grindability of torrefied pine chips was substantially improved after torrefaction at 275 °C and 300 °C as well as the specific energies required for grinding (24-52 kW h t⁻¹), while the energy consumption to grind non-torrefied spruce and beech chips was as high as 750 kW h $\rm t^{-1}$ and 850 kW h $\rm t^{-1}$, respectively. 79,80 Recently Zhu et al.;81 studied the effects of chemical pretreatments and diskmilling conditions on energy consumption for size-reduction of softwood. They found that combined chemical-sizereduction pretreatments of forest biomass can reduce grinding energy consumption by 20-80%, depending on the utilised pretreatment under conditions corresponding to 20% solids-loading and a disk-plate gap of 0.76 mm under milling. Barakat et al.;17 also reported a reduction in energy consumption (240 kW h t^{-1}) for ball mill-treated wheat straw (NaOH at 25 $^{\circ}$ C for 5 h) as compared to a superior 485 kW h t⁻¹ energy consumed under ball milled wheat straw without chemical treatment.

The coupling of mechanical size reduction with chemical and physicochemical pretreatments strongly suggest that the digestibility is substantially improved (with a concomitant decrease in energy requirements) while preserving

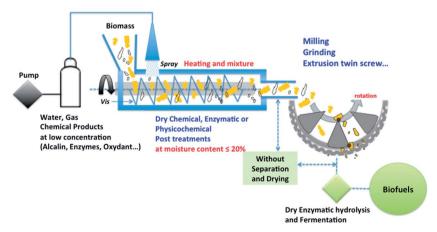


Fig. 7 Future schematic representation of a potentially feasible environmental dry lignocellulosic biorefinery.

lignocellulosic polymers. These would improve the whole process economics, clearly indicating that combined pretreatments are very important for the reduction of energy consumption in *dry* lignocellulosic biorefineries.

The thermal energy consumption for conventional chemical and physicochemical (steam explosion, organosolv, etc) pretreatment is almost linearly proportional to liquid-biomass ratio; thus, reducing this ratio is a prerequisite to improve energy efficiency and limit water utilization and production of (toxic) waste and side streams. In this regard, dry chemical/ physicochemical and biological "solid state" pretreatments combined with dry fractionation steps can be key for advanced pretreatment processes in the future, of direct use in downstream processing in the absence of any solvents/conditioning steps or even separation. Coupling mechanical treatments with dry chemical, physicochemical and/or biological processing (enzymatic hydrolysis and fermentation) in a continuous flow-through process can reduce significantly the energy requirement by 2-5 times, decrease particle size by 2-5 times while increasing lignocellulosic conversion and reduce water utilization by 5-10 times minimizing waste production.

The example illustrated in Fig. 7 showcases the possibilities of the proposed 'dry' biorefinery scheme that may be potentially applicable to the transformation of a wide range of lignocellulosic feedstocks for more efficient and environmentally sound processing.

We believe these technologies can significantly contribute to a more sustainable biomass processing in the future, being part of industrial ventures in our aim to develop multidisciplinary processes equally efficient, cost competitive and with improved environmental footprint to those derived from petrol-based resources that we have relied upon the past 50+ years.

List of abbreviations

BM Ball milling
WDM Wet disc milling
DM Disk milling

HCWT Hot compressed water treatment

SE Steam explosion

EHGY Enzymatic hydrolysis glucose yield

CrI Crystallinity index SSA Specific surface area

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